

## **$\text{VIII}(\text{Mg,Fe})_{0.85}\text{VI}(\text{Mg,Fe})_4\text{IV}(\text{Fe,Ge})_3\text{O}_{12}$ : A new tetragonal phase and its comparison with garnet**

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### **ABSTRACT**

Tetragonal  $\text{Mg}_{2.12}\text{Fe}_{3.17}\text{Ge}_{2.56}\text{O}_{12}$  is a new germanate forming at 1 atm pressure in the  $\text{MgO-Fe}_2\text{O}_3\text{-GeO}_2$  system. It is an analogue of the high-pressure silicate mineral TAPP occurring as inclusions in diamonds of lower-mantle origin. Its crystal structure was determined by X-ray diffraction using single-crystals grown from a potassium molybdate flux. It crystallizes in the  $\bar{I}4_2d$  space group with  $a = 6.8153(4)$  Å,  $c = 18.669(2)$  Å,  $Z = 4$ . The refinement of its cation distribution,  $\text{VIII}(\text{Mg}_{0.52}\text{Fe}_{0.33})\text{VI}(\text{Mg}_{1.60}\text{Fe}_{2.40})\text{IV}(\text{Ge}_{2.56}\text{Fe}_{0.44})\text{O}_{12}$ , shows the presence of a partially filled (85%) dodecahedral site characterized by an unusual geometry with two very different bond lengths [2.169 Å ( $\times 4$ ) and 2.609 Å ( $\times 4$ )]. The structure determinations of the  $\text{Ca}^{2+}$ - and  $\text{Y}^{3+}$ -substituted phases show that the larger cations are completely partitioned in the dodecahedral site, increasing its occupancy slightly (up to 92% in the case of  $\text{Ca}^{2+}$ ).

Cubic  $\text{Mg}_{2.35}\text{Y}_{2.00}\text{Fe}_{0.97}\text{Ge}_{2.59}\text{O}_{12}$  is a new garnet phase forming at 1 atm pressure in the  $\text{MgO-Y}_2\text{O}_3\text{-Fe}_2\text{O}_3\text{-GeO}_2$  system. Its crystal structure and cation distribution were also determined by single-crystal X-ray diffraction:  $\text{VIII}(\text{MgY}_2)\text{VI}(\text{Mg}_{1.35}\text{Fe}_{0.56})\text{IV}(\text{Ge}_{2.59}\text{Fe}_{0.41})\text{O}_{12}$ ,  $Ia\bar{3}d$  space group,  $a = 12.232(1)$  Å,  $Z = 8$ . In spite of similarities in their chemical compositions, the tetragonal and garnet phases are structurally distinct, with different ratios of dodecahedral to octahedral sites, viz.,  $\text{VIII}^{\text{A}}\text{VI}^{\text{B}}_4\text{IV}^{\text{T}}_3\text{O}_{12}$  and  $\text{VIII}^{\text{A}}_3\text{VI}^{\text{B}}_2\text{IV}^{\text{T}}_3\text{O}_{12}$  respectively. As a consequence, the tetragonal phase forms in systems containing smaller cations, such as  $\text{Mg}^{2+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Co}^{2+}$ , whereas larger cations, such as  $\text{Y}^{3+}$ , show a strong preference for the garnet phase.