

CHEMISTRY AND MINERALOGY OF EARTH'S MANTLE

## Recoil-free fractions of iron in aluminous bridgmanite from temperature-dependent Mössbauer spectra†

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### ABSTRACT

Aluminous bridgmanite (Al-Bm) is the dominant phase in the Earth's lower mantle. In this study, the Mössbauer spectra of an Al-Bm sample  $\text{Mg}_{0.868}\text{Fe}_{0.087}\text{Si}_{0.944}\text{Al}_{0.101}\text{O}_{2.994}$  were recorded from 65 to 300 K at 1 bar. The temperature dependence of the center shift was fitted by the Debye model and yielded the Debye temperatures of  $305 \pm 3$  K for  $\text{Fe}^{2+}$  and  $361 \pm 22$  K for  $\text{Fe}^{3+}$ . These values are lower than those of Al-free bridgmanite by 17 and 24%, respectively, indicating that the presence of Fe and Al increases the average Fe-O bond length and weakens the bond strength. At 300 K, the calculated recoil-free fractions of  $\text{Fe}^{2+}$  ( $0.637 \pm 0.006$ ) and  $\text{Fe}^{3+}$  ( $0.72 \pm 0.02$ ) are similar and therefore the molar fractions of  $\text{Fe}^{2+}$  and  $\text{Fe}^{3+}$  are nearly the same as the area fractions of the corresponding Mössbauer doublets. At 900 K, the calculated recoil-free fractions of  $\text{Fe}^{3+}$  is 46% higher than that of  $\text{Fe}^{2+}$ , implying that the molar fraction of  $\text{Fe}^{3+}$  is only 41% for a measured spectral area fraction of 50%, and that the area fractions of iron sites may change with temperature without any changes in the valence state or spin state of iron. We infer that  $\text{Fe}^{3+}$  accounts for  $46 \pm 2\%$  of the iron in the Al-Bm and it enters the A site along with  $\text{Al}^{3+}$  in the B site through the coupled-substitution mechanism. An  $\text{Fe}^{2+}$  component with large quadrupole splitting ( $\sim 4.0$  mm/s) was observed at cryogenic conditions and interpreted as a high-spin distorted iron site.

**Keywords:** Mössbauer spectroscopy, aluminous bridgmanite, ferric iron, recoil-free fraction, Debye temperature, crystallographic site, lower mantle