

Uranium-series Crystal Ages

Kari M. Cooper and Mary R. Reid

Introduction to decay series

Our focus for this session:

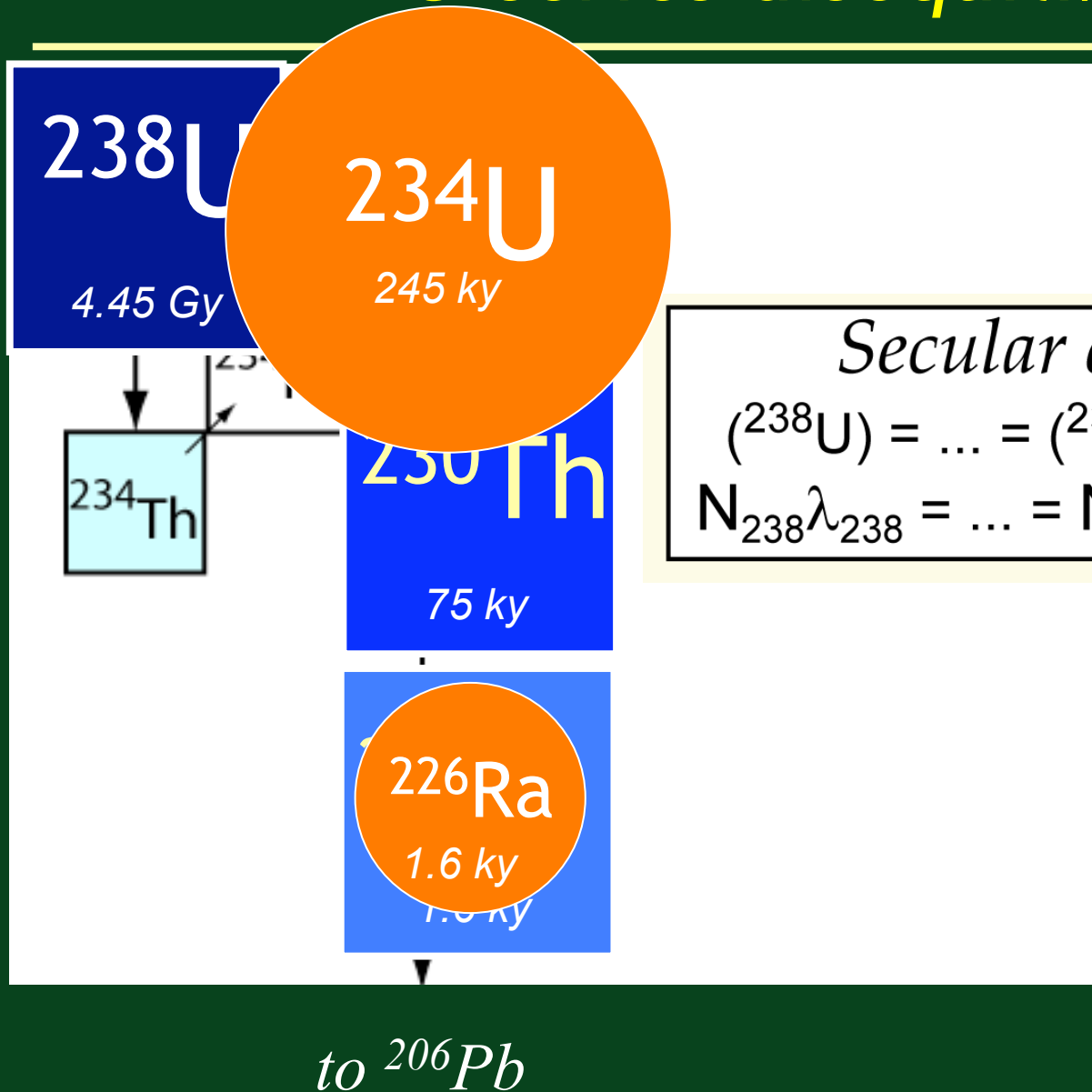
- How to visualize decay series dynamics
- How you'll see these data presented
- How you can obtain these data
- How decay series data can be interpreted

Motivation:

Why would I obtain these ages?

- You want to date an eruption
- You want to learn when in a volcano's history the crystals formed
- You want to determine whether there are multiple crystal populations
- You want to construct time-temperature-composition paths for magma evolution from crystal records

U-series disequilibria



Secular equilibrium:

$$(^{238}\text{U}) = \dots = (^{230}\text{Th}) = (^{226}\text{Ra}) = \dots$$

$$N_{238}\lambda_{238} = \dots = N_{230}\lambda_{230} = N_{226}\lambda_{226} = \dots$$

Decay series equation

Reflects balance between decay and in-growth:

$$N_2 = N_2^0 e^{-\lambda_2 t} + \frac{\lambda_1}{\lambda_2 - \lambda_1} N_1^0 (e^{-\lambda_1 t} - e^{-\lambda_2 t})$$

“Amt daughter present”
decayed”

“Daughter remaining from that present initially”

“Balance between daughter produced and that which has subsequently

Useful approximation:

$$(N_2) = (N_2)^0 e^{-\lambda_2 t} + (N_1)^0 (1 - e^{-\lambda_2 t})$$

At $t \gg t_{1/2}$ of N_2 :

$$(N_2) = (N_1)^0$$

Nuclides:

1: parent

2: daughter

0: initial amount

U-series disequilibria

QuickTime™ and a
decompressor
are needed to see this picture.

^{238}U - ^{230}Th isochron equation

$$\frac{(^{230}\text{Th})}{(^{232}\text{Th})} = \frac{(^{230}\text{Th})_0}{(^{232}\text{Th})} e^{-\lambda_{230}t} + \frac{(^{238}\text{U})_0}{(^{232}\text{Th})} (1 - e^{-\lambda_{230}t})$$

“ Activity of ^{230}Th present ”

“ Activity of ^{230}Th remaining of that present initially ”

“ Balance between ^{230}Th produced and that which has subsequently decayed ”

Age defined by isochron slope (m):

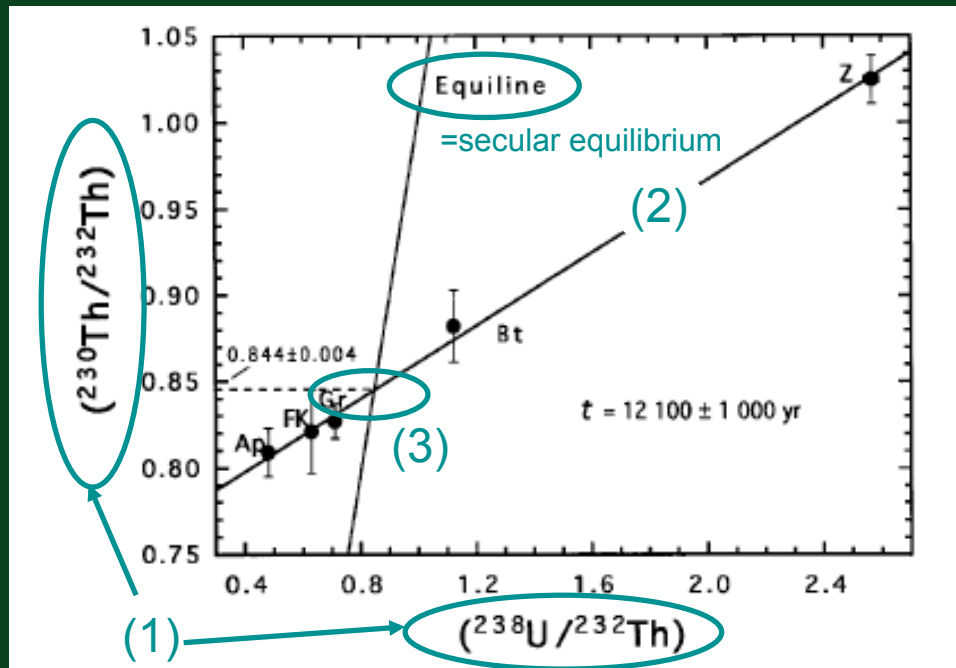
$$t = - \frac{1}{\lambda_{230}} \ln (m - 1)$$

ISOCHRON
EQUATION

At $t \gg t_{1/2}$ of ^{230}Th :

$$\frac{(^{230}\text{Th})}{(^{232}\text{Th})} = \frac{(^{238}\text{U})}{(^{232}\text{Th})}$$

^{238}U - ^{230}Th isochron diagram



Puy du Dome, France

Data for 4 minerals and groundmass:

Ap: apatite

FK: K-fsp

Bt: biotite

Z: zircon

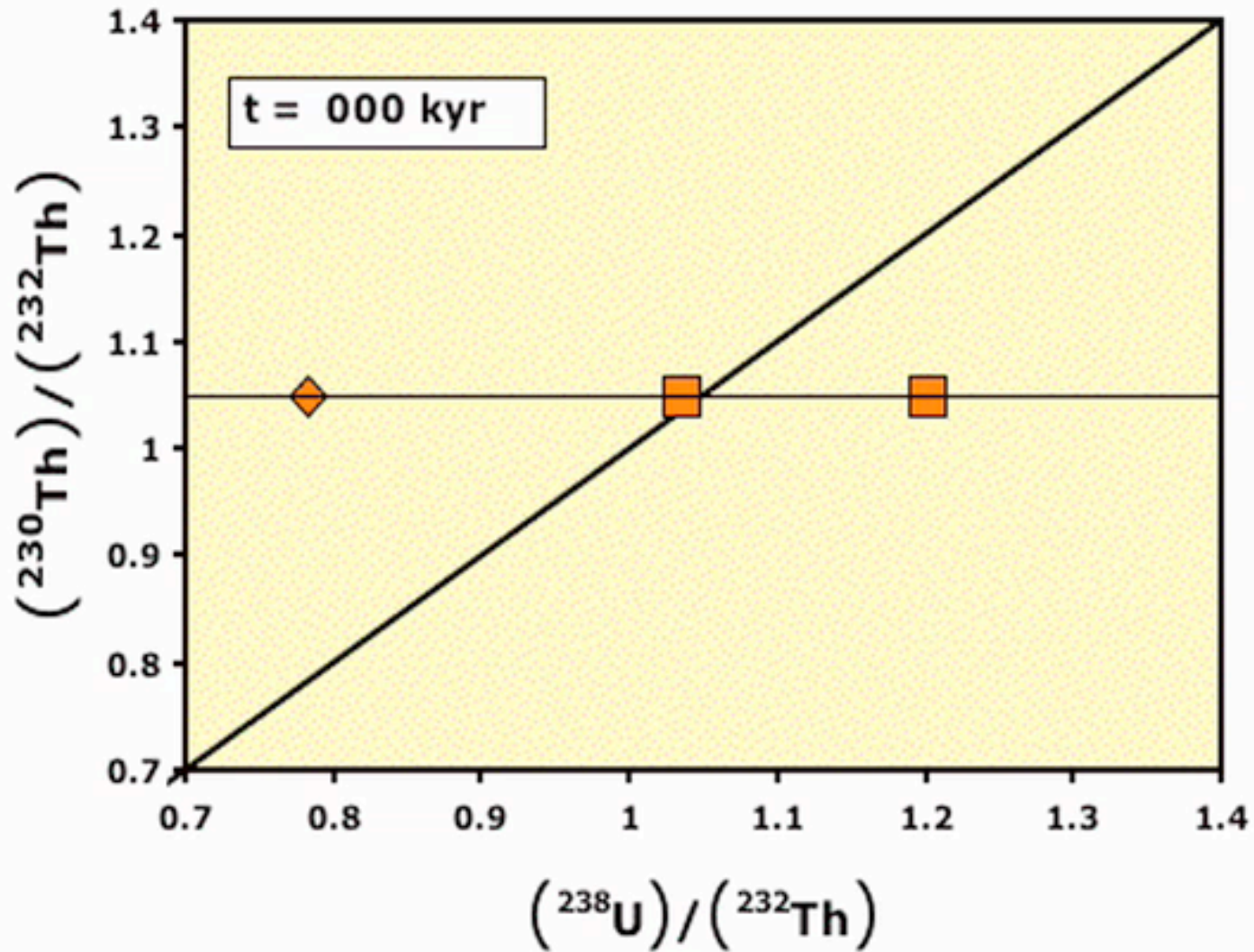
Gr: groundmass

Condomines, 1997

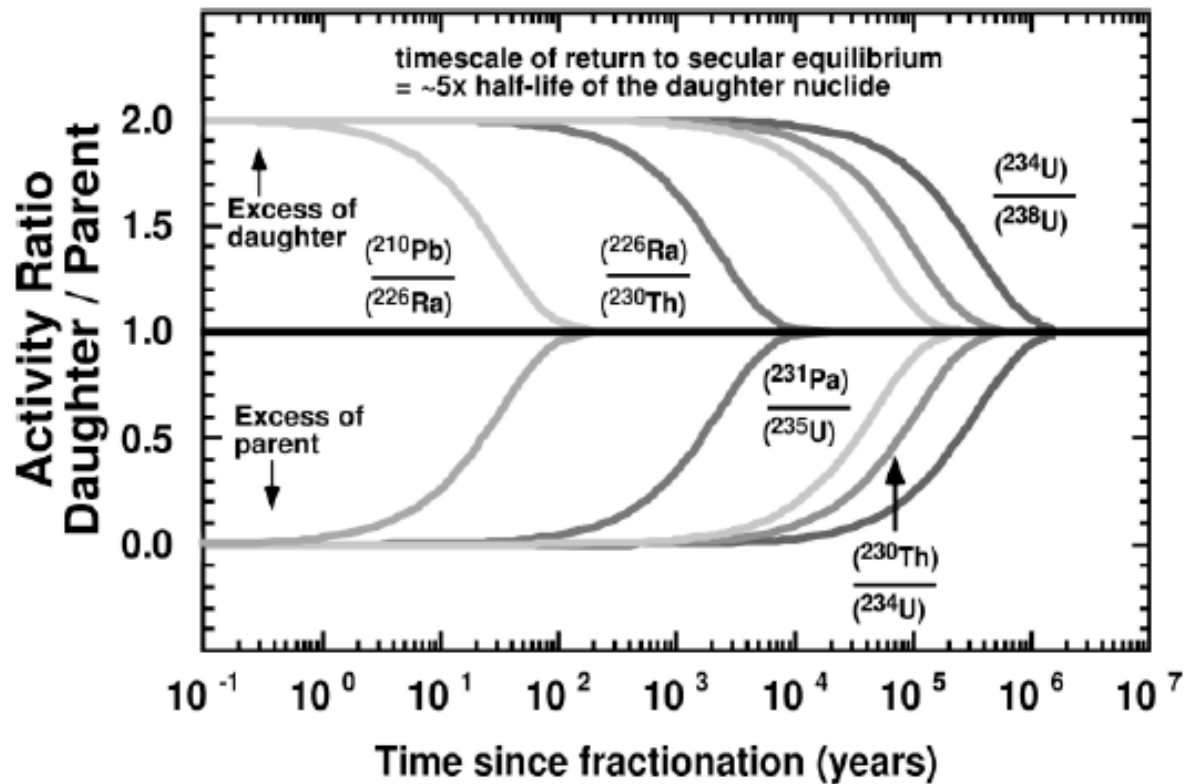
Different from isochron diagram with stable radiogenic daughter:

1. Activities rather than atomic ratios plotted (proportional to amounts)
2. Rather than quasi-linear changes in slope with age, slope increases at an exponentially decrease rate (next illustration)
3. Initial ratio is preserved by intercept with equiline

^{238}U - ^{230}Th isochron



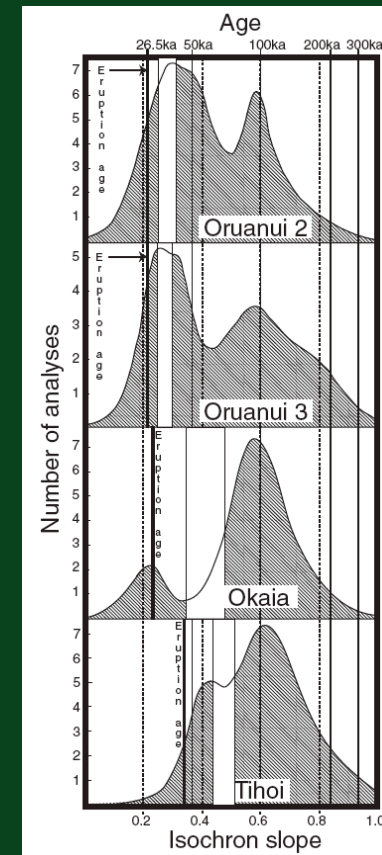
Useful age ranges



from Bourdon et al., 2003

Effects of mixing on U-Th isochrons

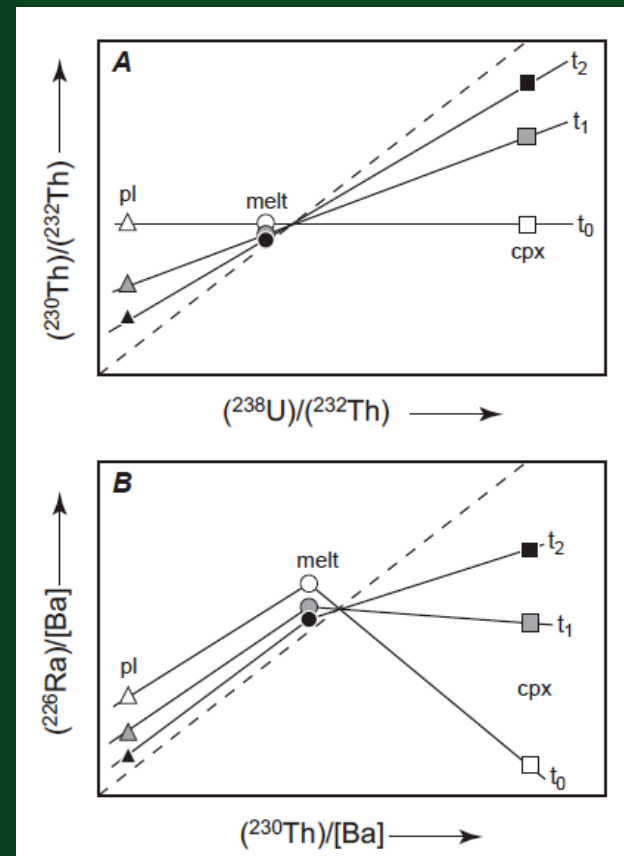
- Mixing of different same-aged phases will not modify the apparent age
- Mixing of different populations of a single phase produces mixed ages that must be interpreted carefully
- For in situ ^{238}U - ^{230}Th analyses (e.g., zircon, allanite, etc.), can calculate model ages (mineral-glass or mineral-whole rock) and identify episodes of crystallization



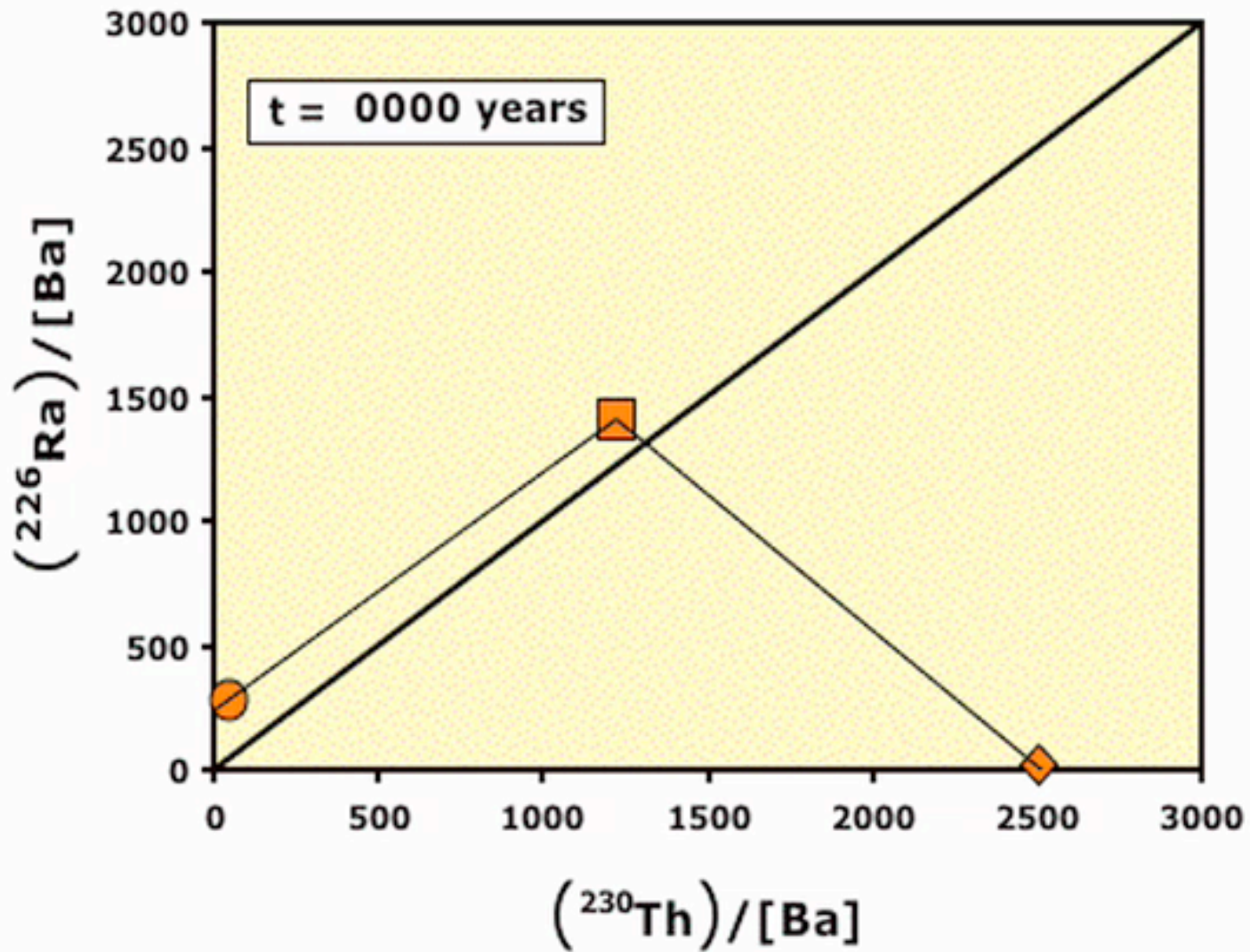
from Charlier et al., 2005

Th-Ra model ages: partitioning matters

- No long-lived reference isotope
- Ra and Ba have different chemical behavior
- Ra-Ba fractionation must be accounted for

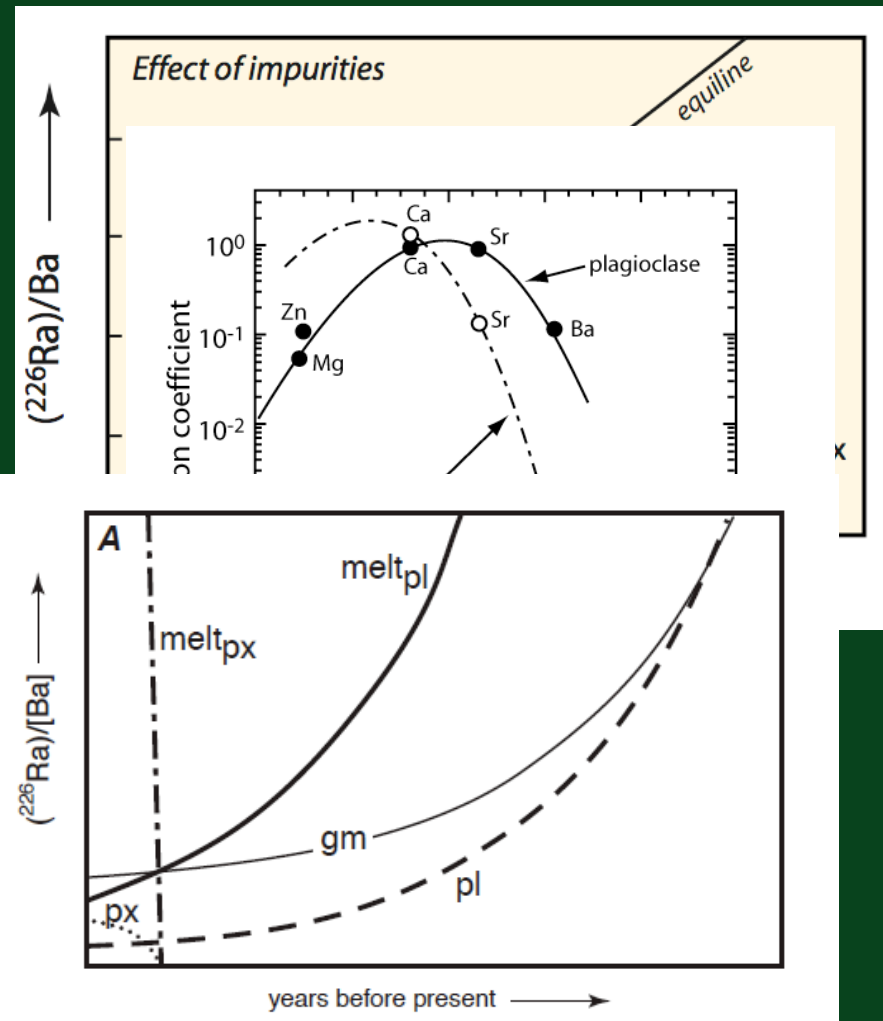


^{230}Th - ^{226}Ra isochron



Calculating ^{230}Th - ^{226}Ra model ages

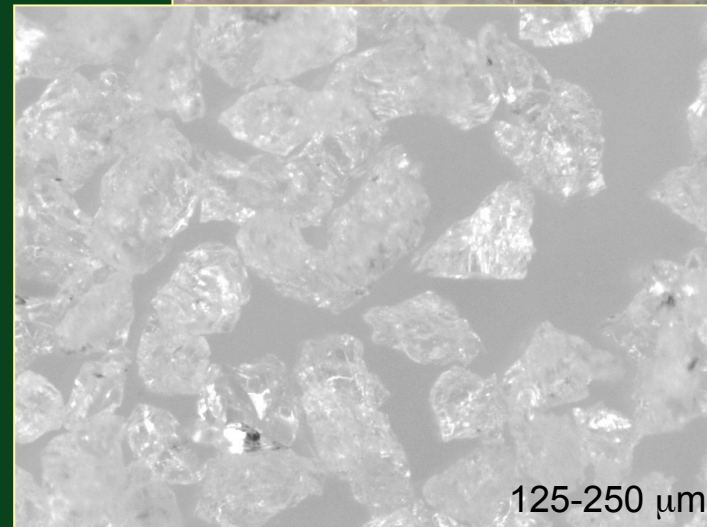
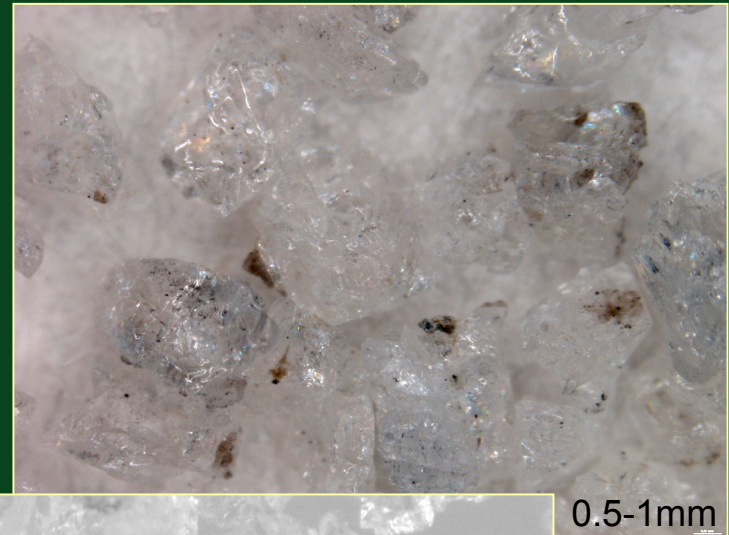
1. Correct for impurities
2. Determine D 's for appropriate conditions
3. Calculate model ages (evolution diagram and/or mathematically)



Calculating ^{230}Th - ^{226}Ra model ages

1. Correct for impurities

- Minimize impurities through careful separation
- Mass balance calculations: solution ICP-MS vs. LA-ICP-MS or SIMS

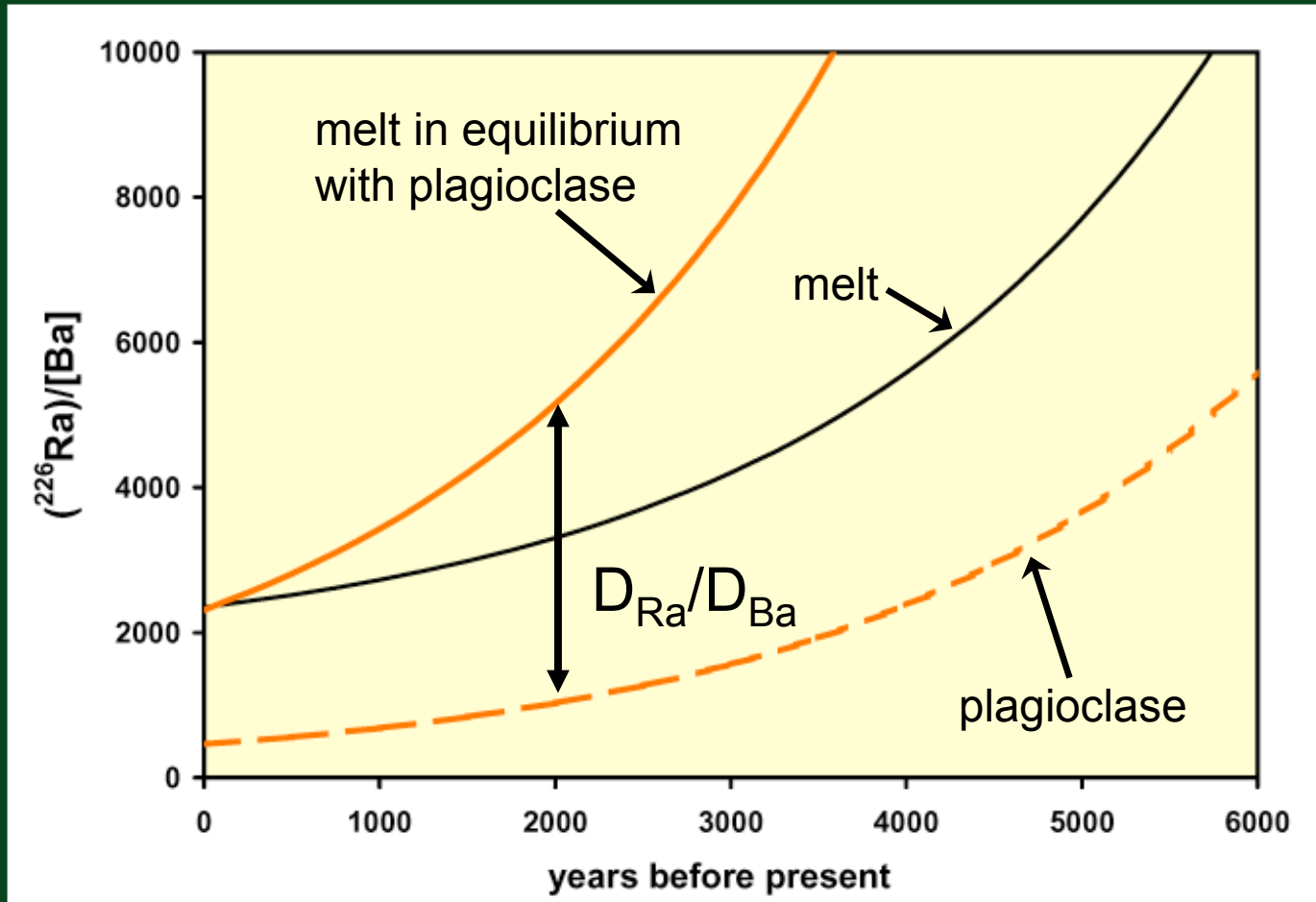


Calculating ^{230}Th - ^{226}Ra model ages

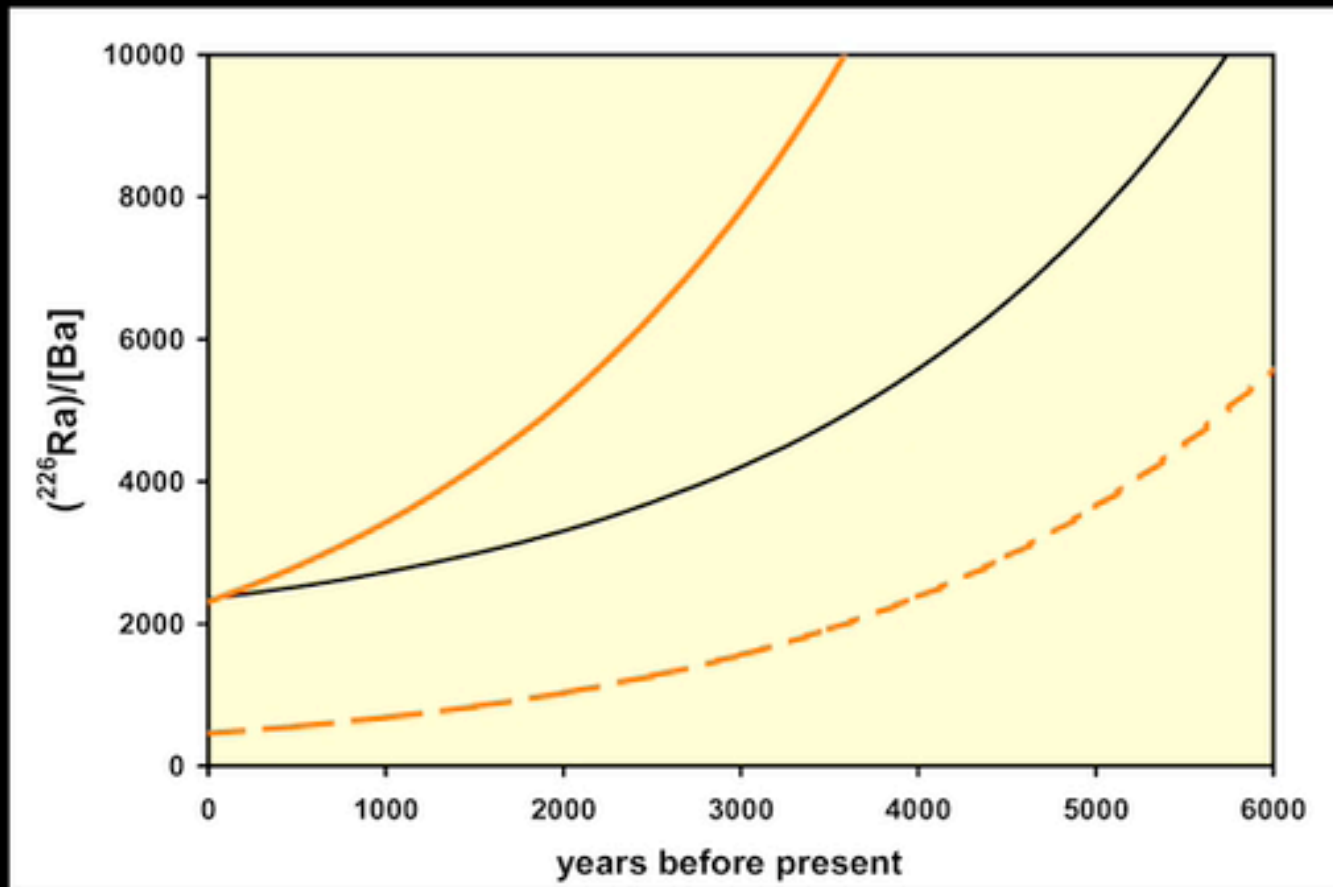
2. Determine D's for appropriate conditions

- Calculate partition coefficients (e.g., Blundy and Wood, 2003)
- Use experimentally-derived partition coefficients (e.g., Miller et al., 2007; Fabbrizio et al., 2008 and in press)
- In almost all mineral phases, $D_{\text{Ra}} \ll 1$ and $D_{\text{Ra}} < D_{\text{Ba}}$
 - Exceptions: Leucite ($D_{\text{Ra}} > 1$, $D_{\text{Ra}} > D_{\text{Ba}}$); K-feldspar ($D_{\text{Ra}} > 1$, $D_{\text{Ra}} < D_{\text{Ba}}$); phlogopite ($D_{\text{Ra}} \sim D_{\text{Ba}}$)
- Most important variables: mineral composition, temperature

Evolution diagrams



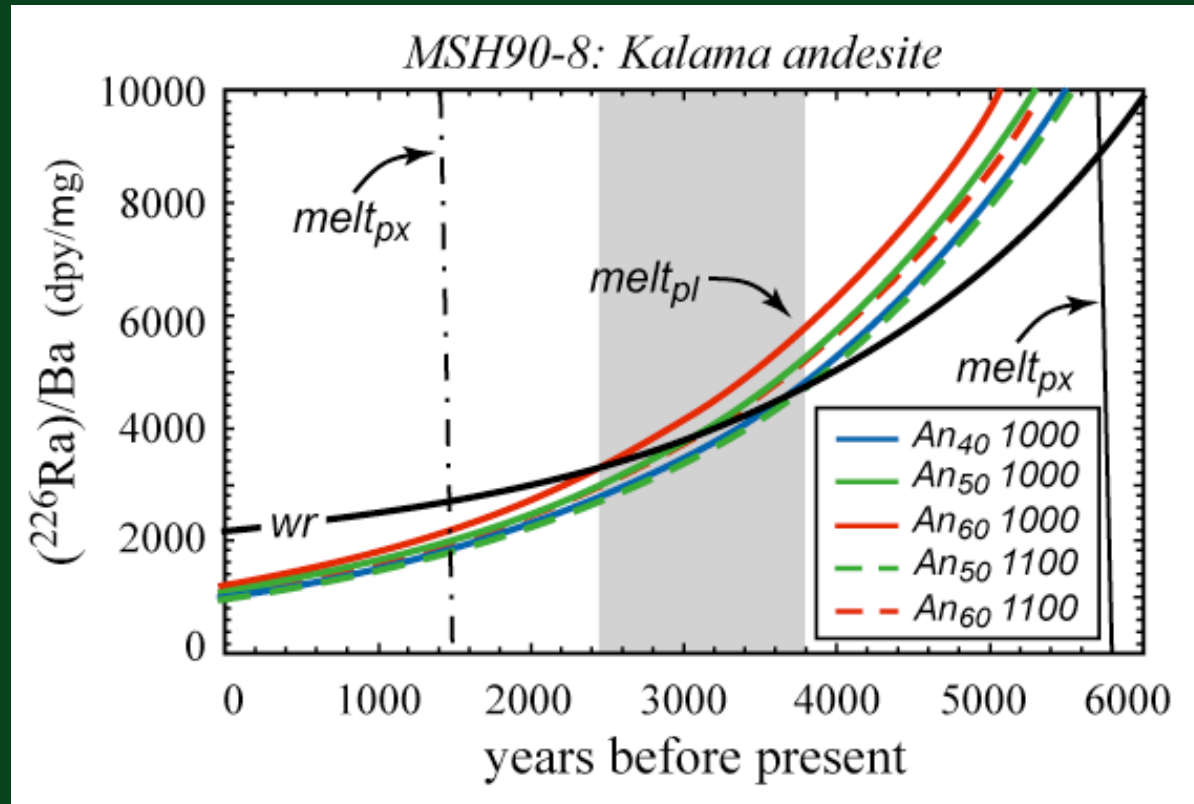
Evolution diagrams



Calculating ^{230}Th - ^{226}Ra model ages

‘Geological uncertainty’

- Crystal composition
- Crystallization T
- Correction for impurities
- Melt-crystal relations



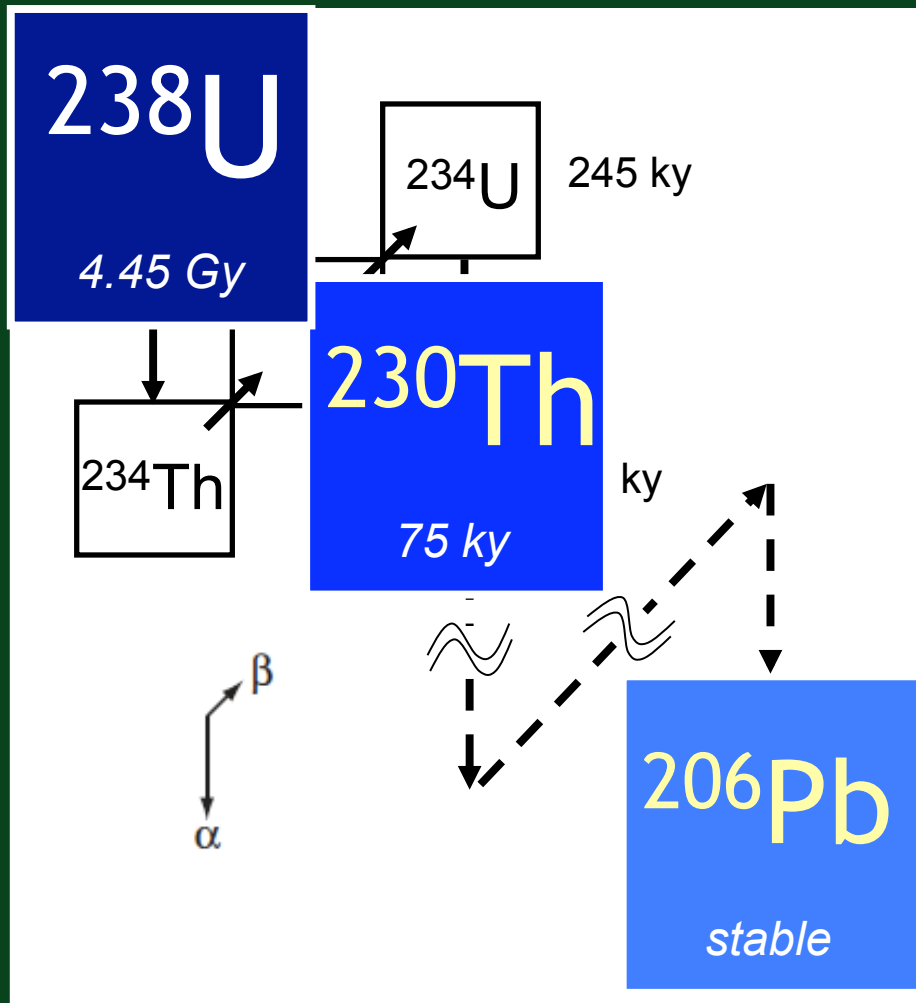
Example spreadsheet available on request

Cooper&Reid, 2003

Effects of mixing on Th-Ra isochrons

- Mixing of different phases will generally produce erroneous ages
- Mixing of different populations of a single phase:
 - Can produce intermediate ages that must be interpreted carefully
 - Can lead to discordant ages between different parent-daughter pairs

The ultimate daughter: ^{206}Pb

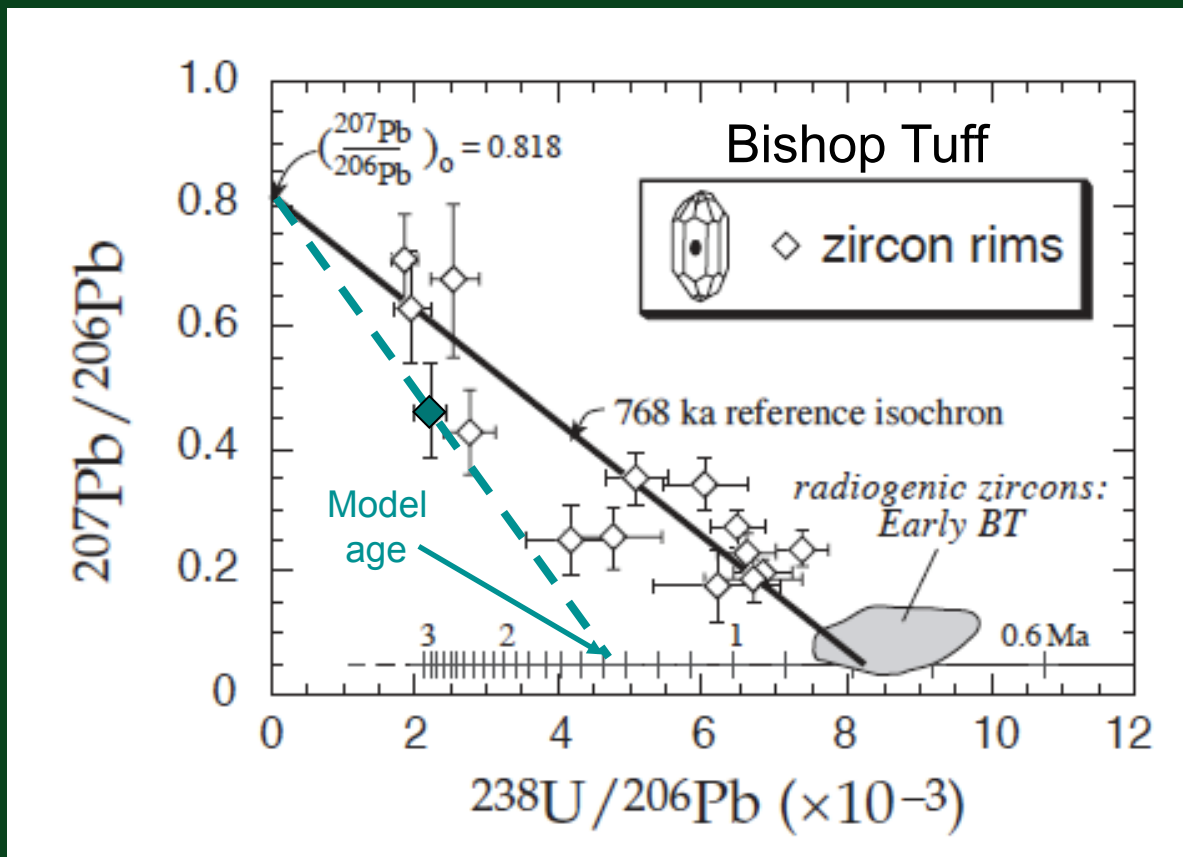


Decay equation:

$$^{206}\text{Pb} = ^{238}\text{U}(e^{\lambda_{238}t} - 1)$$

- Long-standing means of dating igneous rocks via dating of zircon and other accessory phases
- Means of dating domains within individual crystals
- For young rocks, effect of U-series disruptions on ages is significant

Tera-Wasserburg diagram



Reid & Coath, 2000

Corrections for ^{230}Th deficit in zircon

Can essentially define a compound partition coefficient:

$$D(\text{Th}/\text{U})_{\text{crystal/melt}} = \left[\frac{(\text{Th}/\text{U})_{\text{crystal}}}{(\text{Th}/\text{U})_{\text{melt}}^*} \right]_{\text{measured}}$$

$$= f$$

$$^{206}\text{Pb}^*/^{238}\text{U} = (e^{\lambda_{238}t} - 1) + \frac{\lambda_{238}}{\lambda_{232}} (f - 1)$$

“ Measured ratio of parent:daughter “

“ In-growth at secular equilibrium “

“ Reduction due to ^{230}Th deficit “

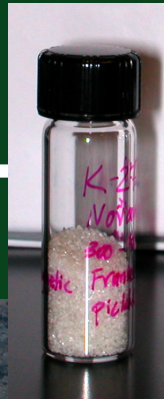
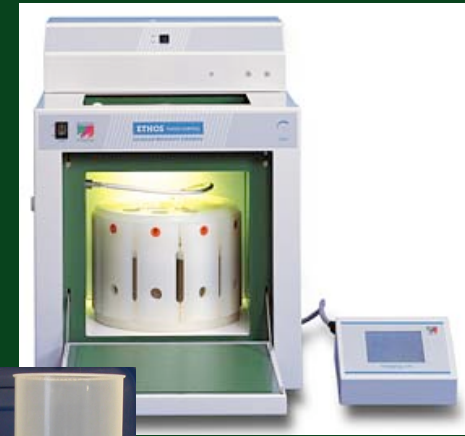
$^*(\text{Th}/\text{U})_{\text{melt}}$ can be estimated from whole rock and/or glass

So - you want to date crystals...

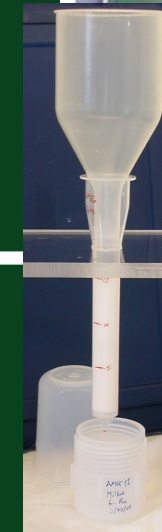
- How do you get from sample to data?
- What do you need to do to prepare samples?
- How much sample will you need?
- How are the samples analyzed?
- How long will this all take?

Collection &
min sep

Analytical
issues - bulk
separates



Dissolution
& chemistry



Th, U, Ra, Ba concentrations
Th, U isotopic compositions

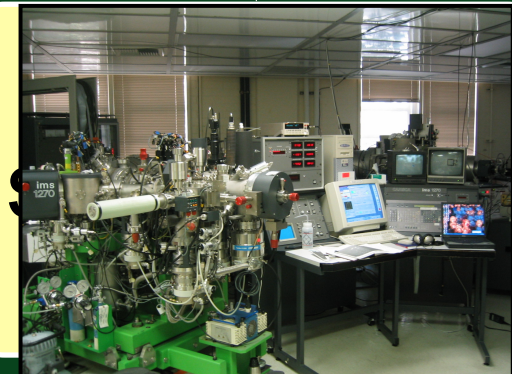
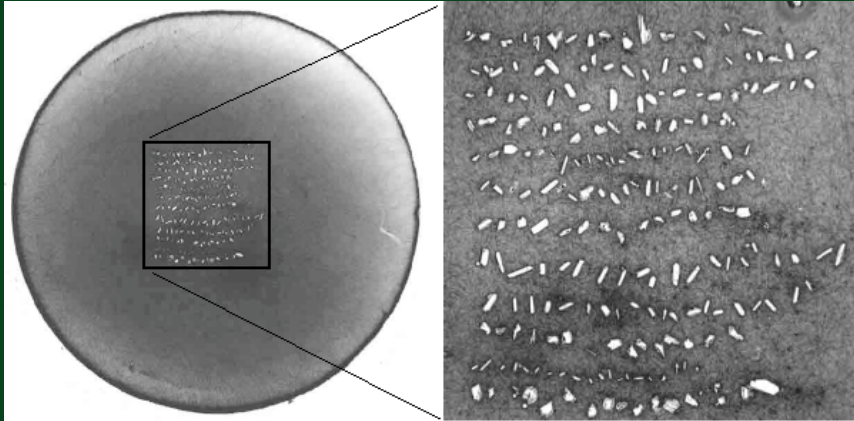


Collection &
min sep

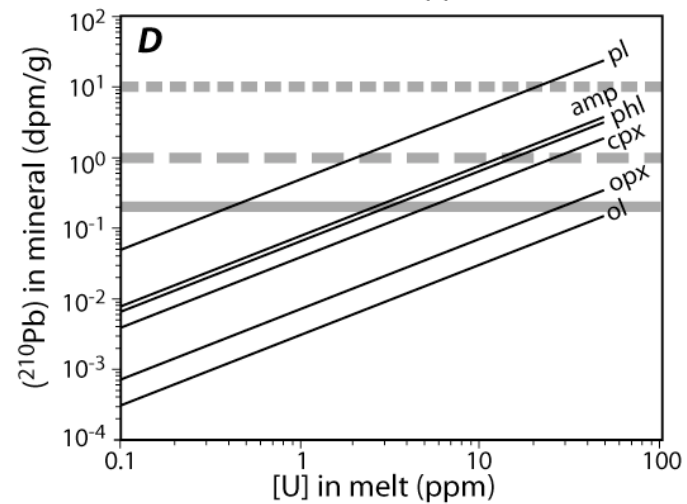
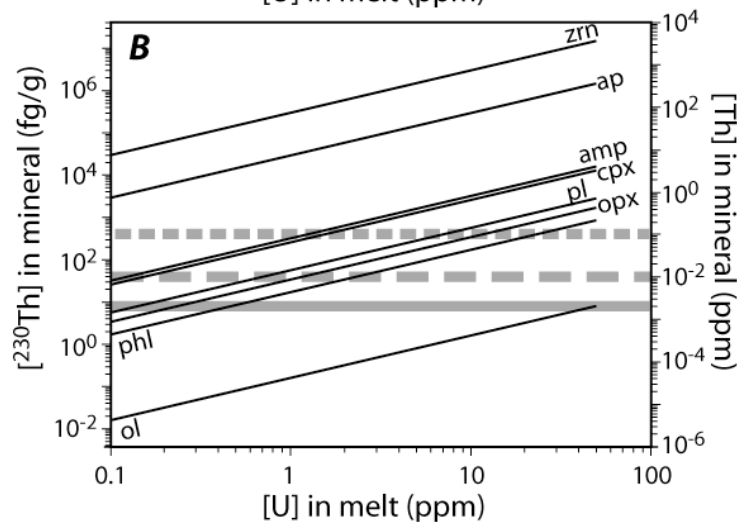
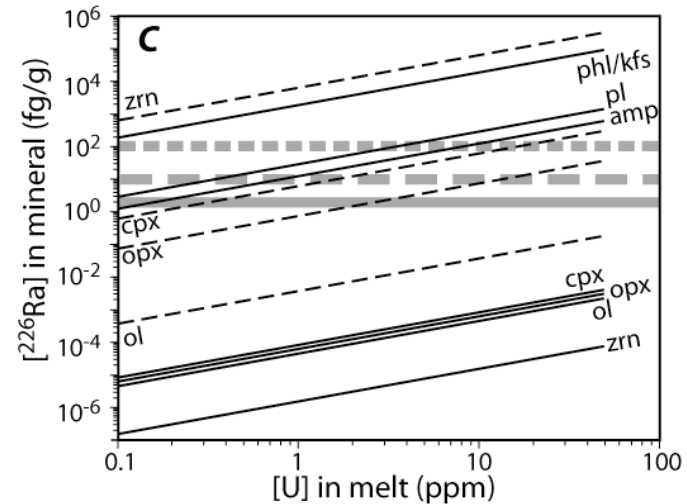
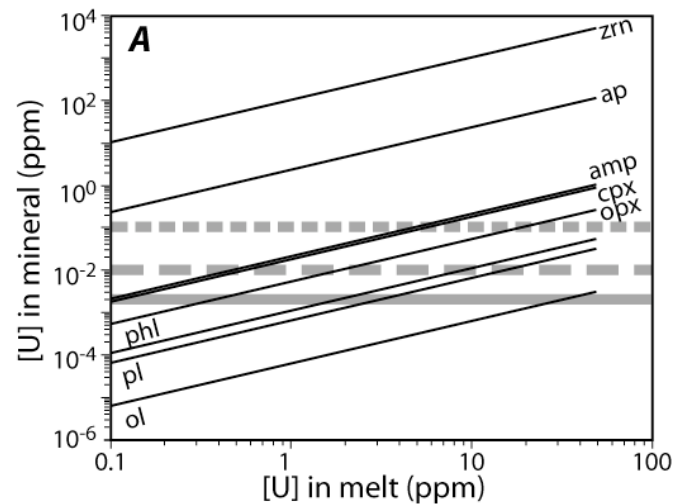
Analytical
issues - in
situ

Grain mount
(no
dissolution)

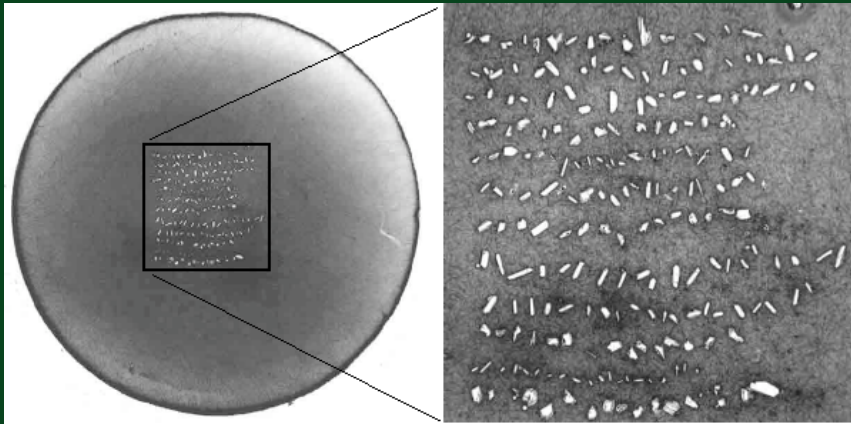
Th, U, Pb isotopic
compositions



Practical matters: how much sample do you need?



Practical matters: how much sample do you need?



10s of grains



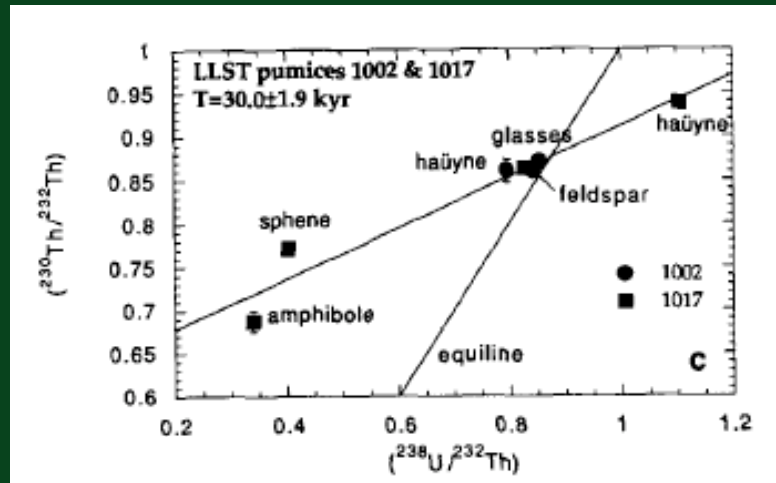
~5 g separate

How can I interpret these ages once I have them?

- Is the age an eruption age?
- What can I say about when in a volcano's history the crystals formed?
- Are there multiple crystal populations?
- How do the ages relate to the time-temperature-composition paths for magma evolution from crystal records

Laacher See Volcano, Germany: Dating Mineral Separates

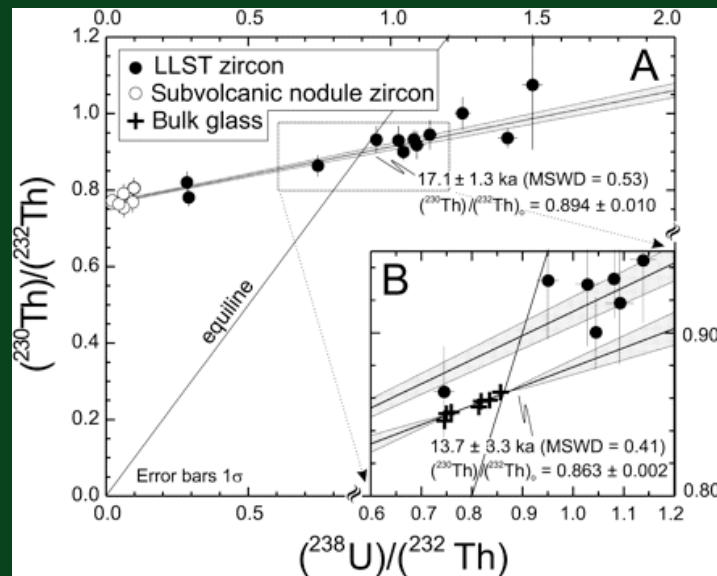
Lower Laacher See tephra



Bourdon et al., 1994

- Eruption from compositionally zoned magma body at 12.9 ka
- ^{238}U - ^{230}Th mineral isochron ages:
 - Similar to eruption age for material erupted later
 - 30 ka for material erupted from the top of magma chamber
- Difference between crystal ages and eruption:
 - Residence of magma for ~17 k.y. before eruption?
 - Entrainment of cumulate material?

Laacher See Volcano, Germany: In Situ Age Dating

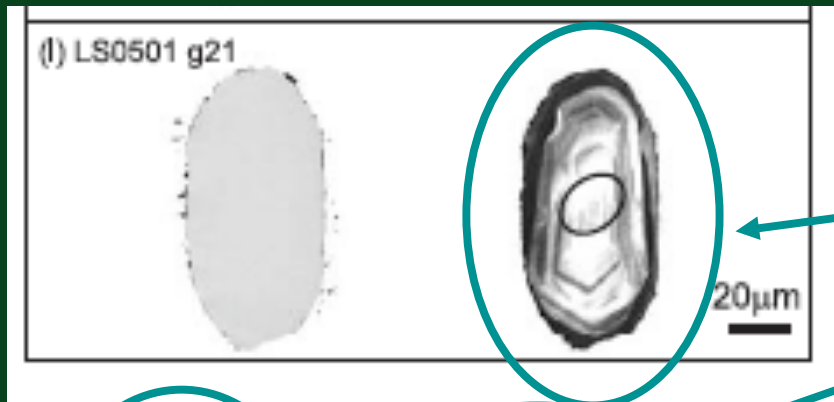


Schmitt, 2006

(xenoliths data not shown)

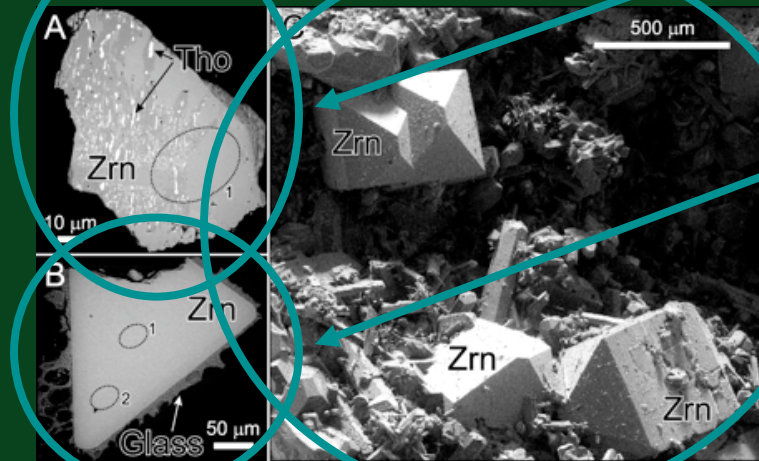
- Isochron age for zircons separated from tephra zircons is 18 ka
- Data for zircons from subvolcanic nodules lie on same isochron and collectively define an age of 17 ka
- Initial ratio higher than host glass (i.e., zircons not in Th isotope equilibrium with host)

Laacher See Volcano, Germany: Zircon-Scale Observations



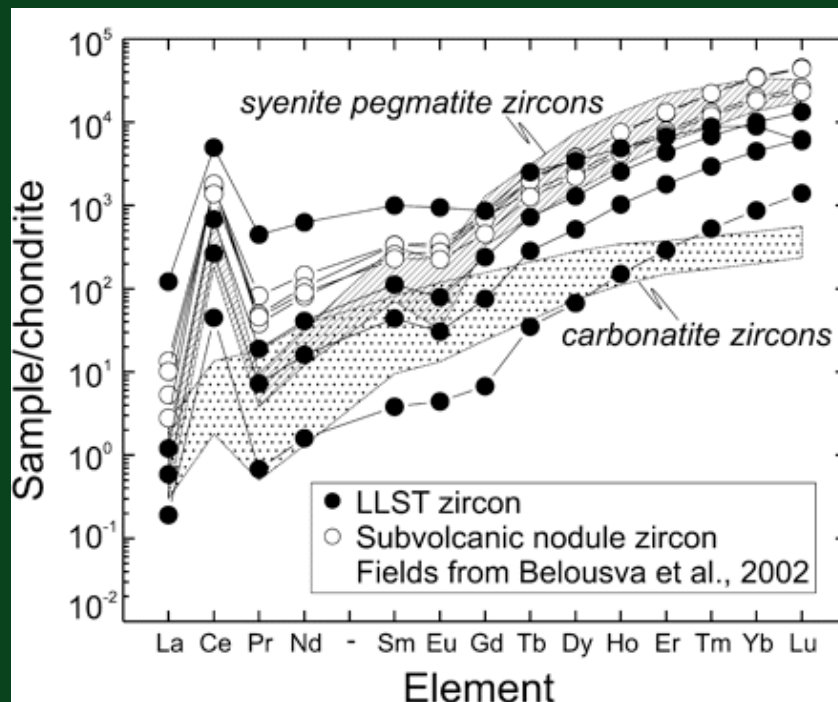
- Petrographic observations:

- Igneous zoning patterns
- Thorite rods in zircon provide evidence of exsolution
- Glass selvages on crystals demonstrate immersion in liquid
- Zircons found in nodule vesicles



Schmitt, 2006

Laacher See Volcano, Germany: In Situ Chemical Observations



Schmitt, 2006

- In situ REE data:
 - similarities between subvolcanic and LLST zircons (HREE enrichment; Ce and Eu anomalies)
 - Patterns more characteristics of syenite than carbonatites
- In situ O isotope data:
 - Overlap between $\delta^{18}\text{O}$ range of nodule zircons (5.0-5.7 ‰) and tephra zircons (5.3-7.1 ‰)
 - Oxygen isotope disequilibrium between zircons and pumice

Laacher See Volcano, Germany: Redux

- Zircons and likely other crystals were scavenged from marginal apophyses of the Laacher See magma chamber
- All units could have differentiated within a few k.y. of eruption

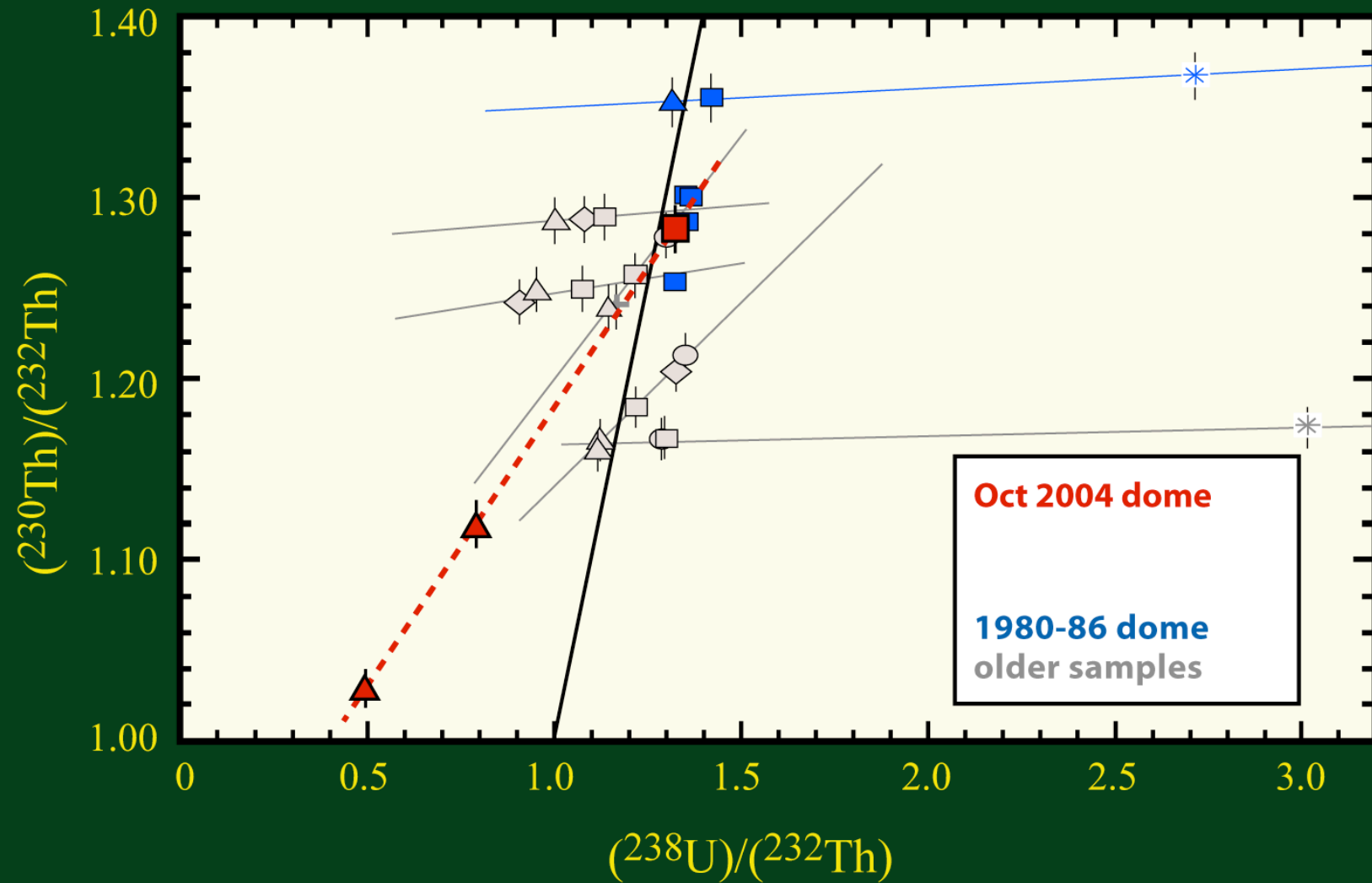
How can I interpret these ages once I have them?

- Is the age an eruption age?
- What can I say about when in a volcano's history the crystals formed?
- Are there multiple crystal populations?
- How do the ages relate to the time-temperature-composition paths for magma evolution from crystal records

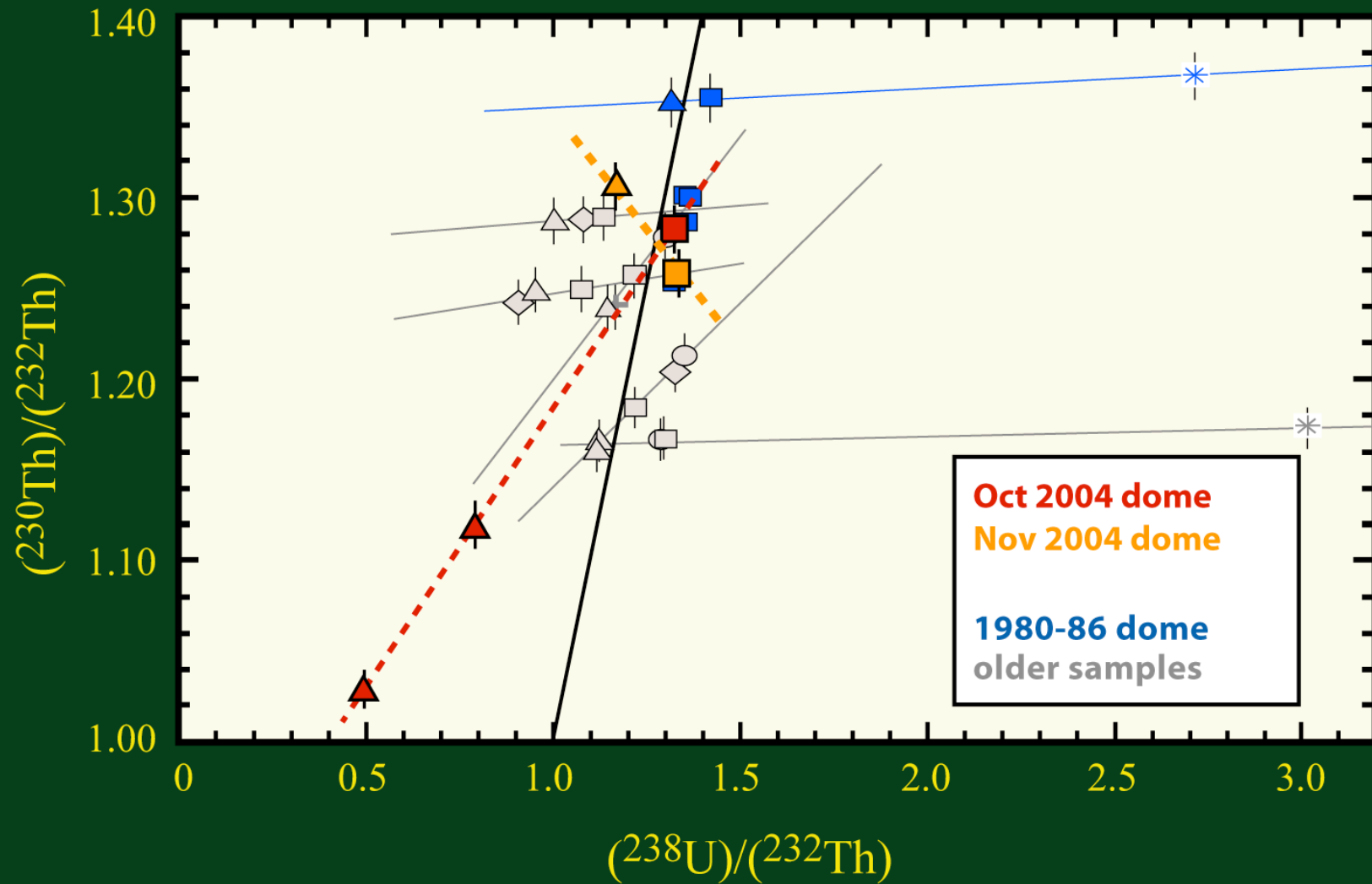
Mount St Helens: 2004-2008 eruption



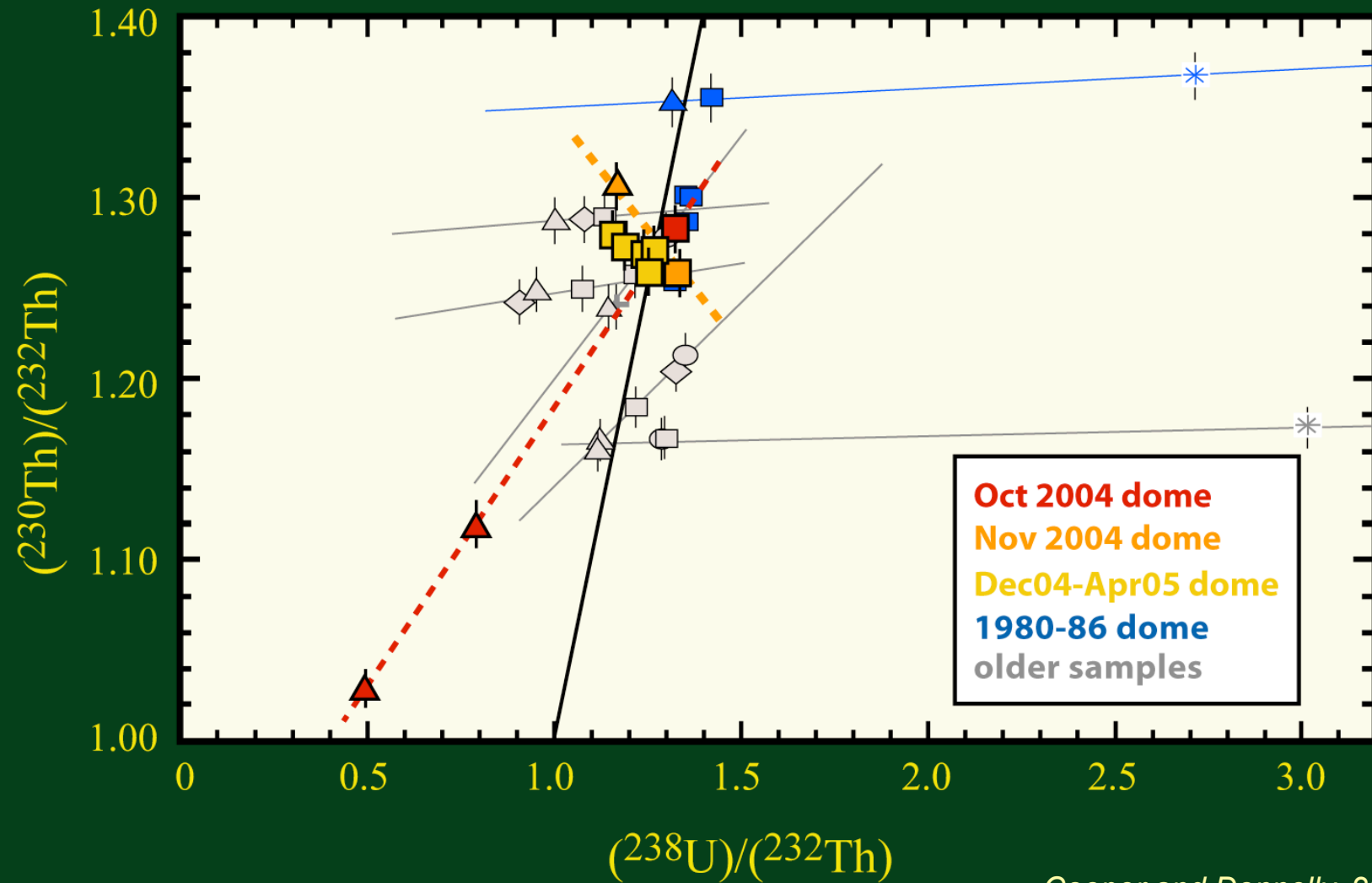
MSH 2004-2005: Th-U



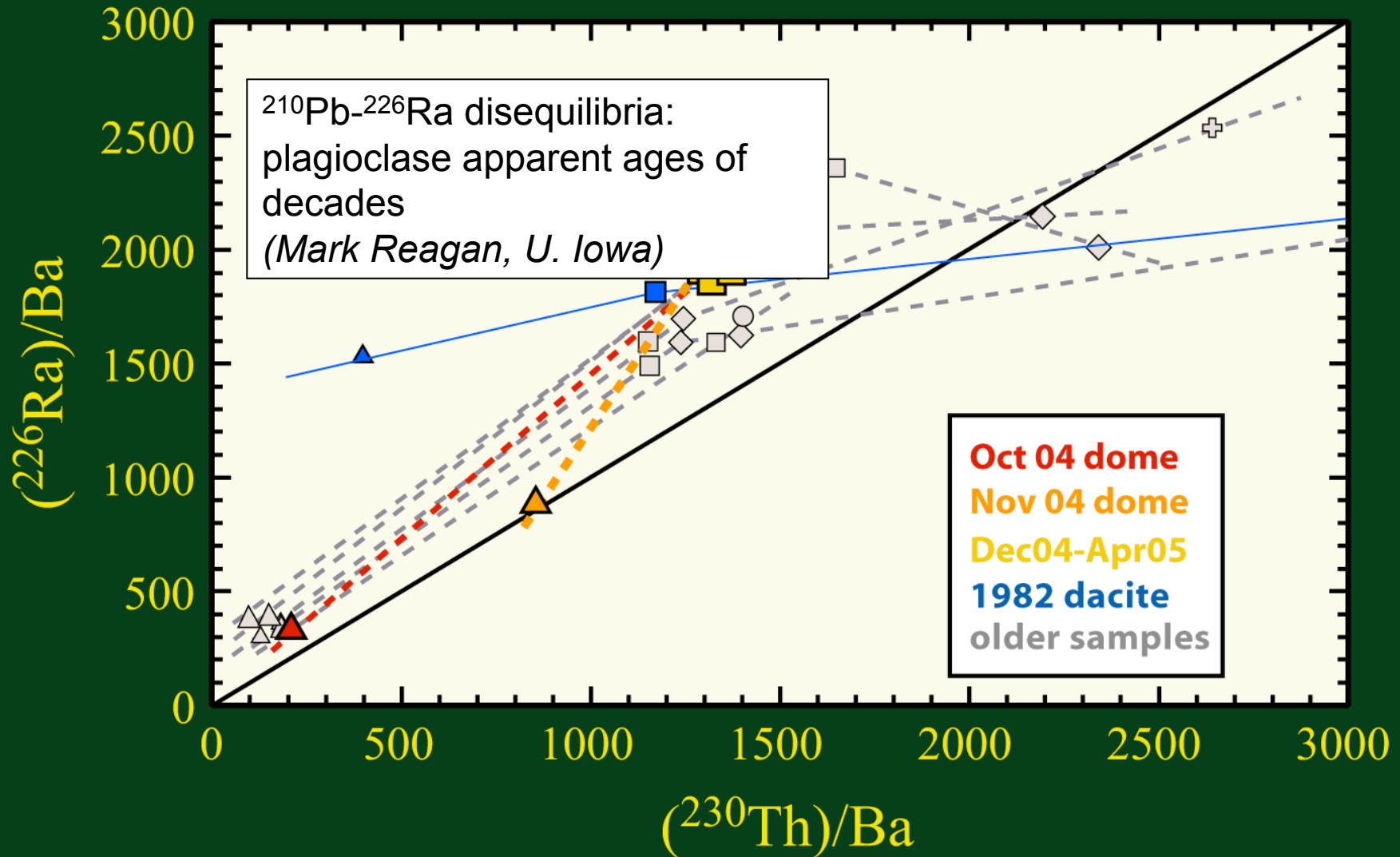
MSH 2004-2005: Th-U



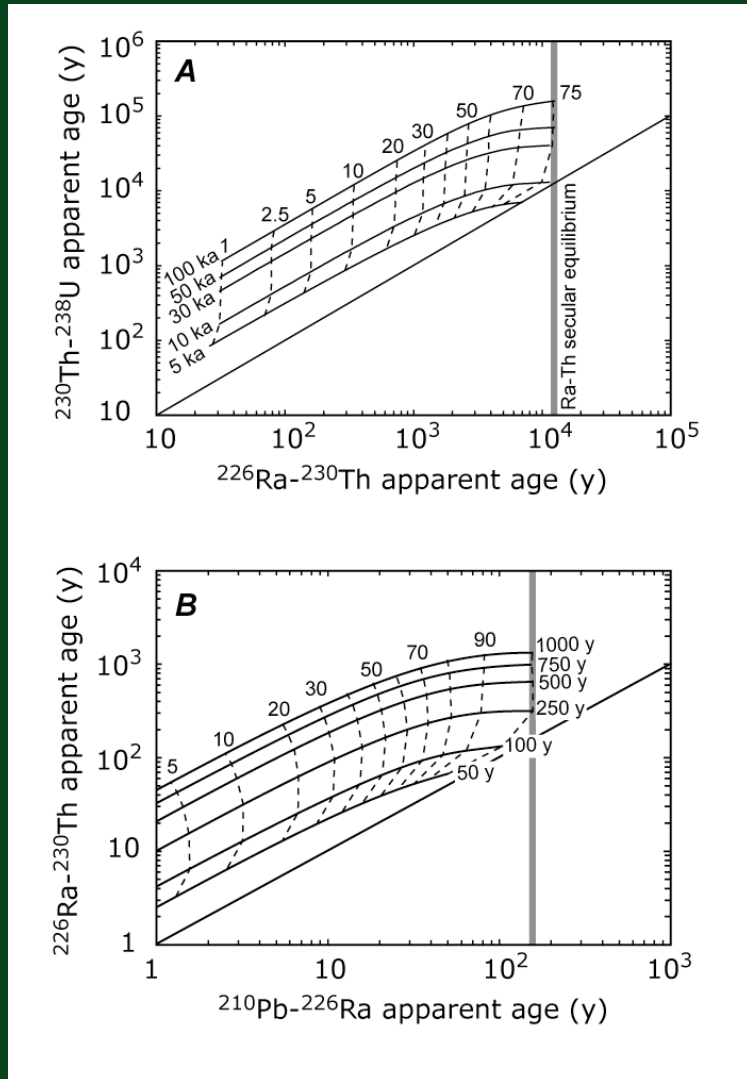
MSH 2004-2005: Th-U



MSH 2004-2005: Ra-Th



Recognizing multiple crystal populations



- Discordant ^{238}U - ^{230}Th - ^{226}Ra - ^{210}Pb ages reflect protracted crystal histories (decades to 10's of ky)
- 2004-05 dome contains new and diverse plagioclase components compared to 1980's

More examples of crystal age results and new directions for research in the volume and AGU sessions!

- You want to date an eruption
- You want to learn when in a volcano's history the crystals formed
- You want to determine whether there are multiple crystal populations
- You want to construct time-temperature-composition paths for magma evolution from crystal records