Neutron powder diffraction studies of phase transitions and kinetics

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In situ high-P/T study essential for phase transitions by neutron scattering

- Neutrons ease in situ study: penetration through complex high-P/T sample environment - maintain fixed/buffered fO₂ add external stress etc.
- Routes to fast neutron scattering studies
- In situ study is often essential if we want to understand real behaviour at deep Earth conditions
- Examples of cation order-disorder *in situ*. Complementarity with X-rays can be exploited, e.g. with different scattering factors
- Future prospects for high-P/T with modulated stress





The need for in situ study....

Eadweard Muybridge, "father of the motion picture", shows that a horse lifts all four hoofs in a gallop (1872):



principles of time-dependent diffraction studies:

•Perturbation (T,P, pH etc.):



•Time-dependent measurement:



Stroboscopic neutron scattering: ic-c p.t. in Rb₂ZnCl₄

Eckold, Hagen & Steigenberger (1997) Physica B 234:151 Steigenberger, Eckold & Hagen (1994) Nucl Inst Meth Phys B 93: 316







Stroboscopic neutron scattering: ic-c p.t. in Rb₂ZnCl₄

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192.7 K

▼ 192.3 K
♦ 192.1 K

0,02

0.00

6 8 10 12 14 16 18 20

t/ms

Stroboscopic scattering enables observations of ferroelectric responses at the phase transition on ms time scales, synching to a pulsed neutron source

strains and tilts in oxide perovskites



strains and tilts in oxide perovskites

Mountstevens, Redfern, Attfield (2005) Phys Rev B 71: 220102 Mountstevens, Attfield, Redfern (2003) J Phys Cond Matt 15: 8315.



High-*T-t* resolution with high flux (D1B)



thermodynamics of I4/mcm-Pm3m transition



HRPD study: structure of lawsonite at low T

(Meyer et al. (2001) Am Min 86: 566)







Neutrons to study order-disorder phase transitions



Neutrons to study cation and anion site occupancies: pyrochlores

Harvey et al. (2006) J Mat Chem 166: 4665 Harvey et al. (2005) J Solid State Chem 178:800

Related to fluorite, with ordered anion vacancies. Potential host for immobilisation of Pu. La₂Zr₂O₇ pyrochlore: neutrons provide essential information on oxygen order/disorder, and to higher scattering vectors





Order-disorder in spinel

Spinel: AB₂O₄

e.g. $MgAl_2O_4$ normal A[4] 2 x B[6] e.g. Mg_2TiO_4 inverse B[4] B[6]A[6]

Neutron scattering • • • • • • lengths Mg Al Mn Fe X-ray scattering • • • • • • •



Order-disorder in spinel: occupancies from neutron powder diffraction





Order-disorder in spinel: occupancies from neutron powder diffraction



free energy difference between fully disordered distribution of metal cations and the equilibrium distribution is:

$$\Delta G = -hQ + \frac{a}{2}(T - T_c)Q^2 + \frac{b}{4}Q^4$$

lower-temperature kinetically-controlled *Q-T-t* pathways are defined by the Ginzburg-Landau equation:

$$\frac{dQ}{dt} = -\frac{\gamma \exp(-\Delta G^*/RT)}{2RT} \frac{\partial G}{\partial Q}$$

Modelling disorder: using spinel





Figure 3. Results of Monte Carlo simulations, compared to results of three neutron scattering experiments and fits of analytic models to experimental data [1]. The order parameter Q is shown as \blacksquare , the energy of ordering as \bigcirc , and the susceptibility $((Q^2) - (Q)^2)/T$ is amplified and shown as \triangle . The experimental values of Q were fitted by Redfern *et al* [1] to a Landau model (.....) and the model of O'Neill and Navrotsky (...) [7]. The experimental data cover both heating and cooling, as indicated by arrows, and are kinetically limited below 1000 K.



 $(Na_{0.065}Ca_{0.205})_2Mg_{5.32}Mn_{1.14}[Si_8O_{22}](OH)_2$

 $P2_1/m$ at room temperature, transforming to C2/m at higher T

Reece et al. (2002) Phys Chem Minerals 29:562



•How does Mn partition over the M sites? What are the equilibrium ordered states in amphiboles at temperatures greater than 500 °C, and can they be applied in geothermometry?

•What is the time-scale of ordering on cooling: does it occur during anneal and quench experiments, and can it be applied as a geospeedometer?

Approach: Structures and direct occupancies from high-T neutron powder diffraction data of Mn-Mg cummingtonite from Talcville.





C2/m transition,

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ii) additional strains +e2, +e3 and -e5 resulting in a volume strain associated with M-site disordering.



Reece et al. (2002) Phys Chem Minerals 29:562

On heating, Mn in M4

decreasing at temperatures

above 800 K (disordering).

All changes are very small,

increases initially (relaxation) before

however.



Temperature (K)

Since $K_D = [M_{123}*(1-M_4)]/[M_4*(1-M_{123})]$ and $\Delta G = RT \ln K_D$ these data give and exchange energy of 21.5±0.7 kJ/mol (cf. A value of 8.95 kJ/mol for Fe-Mg exchange

given by Ghose, 1961)

On cooling, the high-T disorder is frozen in.





Temperature (K)

Neutron diffraction also provides sensitive information on the position of the hydrogen within the structure: it is affected more by the $P2_1/m$ -C2/m transition than by the disordering process. This is a reflection of the rotation-translation coupling at this transition.





Reece et al. (2002) Phys Chem Minerals 29:562

Convergent cation ordering, magnetic transitions and exsolution in ilmenite-hematite





Harrison & Redfern (2001) Phys Chem Minerals 28: 399 Harrison et al. (2000) Am Min 85: 194, 1694.

Cation ordering and exsolution in ilmenite-hematite



Neutron diffraction of convergent ordering in ilmenitehematite

> Ti -3.438 fm Fe +9.45 fm O +5.803 fm

 $Q = \frac{\left(X_{Ti}^B - X_{Ti}^A\right)}{\left(X_{Ti}^B + X_{Ti}^A\right)}$



Fig. 3a, b Neutron diffraction profiles of the $(1\overline{1}2)$ fundamental reflection $(d \approx 3.7 \text{ Å})$ and the $(0\overline{1}1)$ and (003) superlattice reflections $(d \approx 4.2 \text{ Å} \text{ and } d \approx 4.6 \text{ Å}, \text{ respectively})$, as a function of temperature for **a** ilm60 and **b** ilm65

Time of flight data collected on a very simple neutron powder diffractometer (ROTAX)

Harrison & Redfern (2001) Phys Chem Minerals 28: 399 Harrison et al. (2000) Am Min 85: 194, 1694.

Convergent ordering in ilmenite-hematite

Fig. 7a, b Long-range order parameter, Q, as a function of temperature for a ilm60 and b ilm65. Solid lines are a leastsquares fit to the high-temperature data using the modified Bragg-Williams model (Eq. 11). Dashed line is fit by hand to the low-temperature data using the coarsening kinetic model (Eq. 17). After data collection at 500 °C in a, the sample was cooled to room temperature and then reheated. This causes a slight discontinuity in the kinetic relaxation curve. Open symbols show the cooling data





$$Q = \frac{\left(X_{Ti}^B - X_{Ti}^A\right)}{\left(X_{Ti}^B + X_{Ti}^A\right)}$$

Why is ilm70 less ordered at lower T than ilm 80-100? incipient phase separation



R3c

R3c + R3

FM

0.6

0.8

Mole fraction FeTiO₁

0.4

RЗ

1.0

FeTiO₃

Temperature (°C)

500 - CAI

0.0

Fe₃O₅

CAF

0.2



Neutron powder diffraction is very sensitive to Fe-Ti distributions
Magnetic ordering contributions give a measure of the magnetic structure

•Discrepancies can be attributed to allied effects: exsolution

•Data have been employed in macroscopic thermodynamic and microscopic statistical mechanical modelling

Harrison & Redfern (2001) Phys Chem Minerals 28: 399 Harrison et al. (2000) Am Min 85: 194, 1694.

What is the influence of P? **Developing experiments at P & 1**



Anvil Assembly

Gasket Sample



The high-P/T cell at ISIS



The PEARL high pressure facility at ISIS



Paris-Edinburgh cell with internal micro-furnace for use with neutron resonance spectroscopy





An alternative to thermocouples?



- Temperatures measured by neutron resonance spectroscopy
- Pressures determined by compressibility of MgO or NaCl



Can determine sample conditions in dependent of the measured thermocouple emf

Problem:

During experiments the thermocouple failed or withdrew from the sample

Solution:

New design for the sample assembly

Temperature Measurement by NRS

Some difficulties with the use of thermocouples in high P/T experiments

- Significant probability of thermocouple failure
- Compromises the structural integrity of the gasket
- Acts as a heat sink
- Uncertainty of Seebeck coefficients at high P

Solution:

Use the thermally-induced Doppler broadening of neutron absorption resonances



Stone et al. (2005) J Appl Phys 98: 064905, Stone et al. (2005) Nucl Inst Meth A, 547: 601 LeGodec et al (2001) Mineral Mag 65: 737





Temperature measurement by radiography



Transmission spectrum

Effect of temperature

Width of resonance line increases with temperature due to Doppler effect

Width of resonance line can be used to calibrate temperature

Wavelength

Doppler Broadening of Absorption Resonances

University of Cambridge Department of Earth Sciences

Stone et al. (2005) J Appl Phys 98: 064905, Stone et al. (2005) Nucl Inst Meth A, 547: 601 LeGodec et al (2001) Mineral Mag 65: 737

Measurement of P, T, and stucture (including order-disorder)

Harrison, Stone & Redfern (2006) Phys Earth Planet Int 154: 266

Influence of P on high T orderdisorder in ilmenite

- Determine the pressure-dependence of high-T orderdisorder in minerals
- Probe the pressure dependence of the equilibrium high-T order-disorder properties: first neutron measurements of these phenomena at real Earth interior conditions

- Conventional wisdom states that the increase in transition temperature with pressure is a consequence of a negative volume strain
- Assuming a 1% volume strain, the calculated increase in T_c is only 35°C at 0.6 GPa (note: observed strain is <<1%)
 - Increase in T_c must therefore come from increased internal energy of ordering, i.e. increased cation interaction as structure is squeezed
- This implies that the miscibility gap would also increase in temperature with pressure. Estimated increase of up to 150 °C at 1 GPa

