1	Revision 1
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3	Genetic implications, composition, and structure of trioctahedral micas in
4	xenoliths related to Plinian eruptions from the Somma-Vesuvius volcano (Italy)
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22 23	RUNNING TITLE: Genetic implications and crystal-chemistry of Vesuvius micas
24	ABSTRACT
25	The present work is part of a systematic mineralogical and petrographic characterization of mica-
26	bearing xenoliths from Somma-Vesuvius volcano (Roman Magmatic Province, southern Italy).

27	Skarns, composite skarns-marbles and cumulates from Pompeii Plinian eruption (AD 79), and
28	skarns and syenite from Avellino eruption (3945 \pm 10 cal yr BP) were investigated to define the
29	crystal chemistry of the Somma-Vesuvius trioctahedral micas and to draw inferences on
30	petrogenetic processes they were subjected. Xenoliths were characterized by means of polarized
31	optical microscopy, scanning electron microscopy (SEM-EDS), X-ray powder diffraction (XRPD)
32	and bulk-rock geochemical analyses. Mica crystals were studied through electron microprobe
33	analysis (EMPA) and single-crystal X-ray diffraction (SCXRD).
34	Micas from skarns are variably associated with Mg+Ca silicates (clinopyroxene, vesuvianite,
35	humite, clinohumite, chondrodite, forsterite, garnet), other sporadic silicates (anorthite, sodalite,
36	titanite, britholite), apatite, calcite, various types of oxides, as well as rare sulfides and halides. In
37	composite skarn-marble rocks, the mineral assemblages show some differences compared to skarns,
38	as lack of clinopyroxene and the presence of dolomite. Cumulate samples consist of mica and
39	clinopyroxene, whereas syenite is mainly composed of mica, K-feldspar, feldspathoids and
40	clinopyroxene. Together with mica, apatite occurs in all the lithotypes.
41	Trace element arrays are scattered for skarn and composite skarn-marble samples. The REE
42	patterns have a general enrichment in light (La, Ce, Pr, Nd) and medium (Sm, Eu, Gd, Tb, Dy) rare
43	earth elements, in some case with slight positive Gd anomaly. Cumulate samples generally have
44	low amounts of Ba, Sr, Zr and Th, while syenite exhibits low concentrations of trace elements,
45	except for Rb, Cs and Tl.
46	Mica crystals occurring in the studied xenoliths are phlogopite with different Al and Mg content at
47	the octahedral site, a negligible tetraferriphlogopite component and variable dehydrogenation
48	degree. All samples belong to the $1M$ polytype ($C2/m$ and $C2$ space group) and have a wide range
49	of unit cell parameters, especially of the <i>c</i> axis $[5.3055(1) \le a \le 5.3218(1) \text{ Å}, 9.1893(1) \le b \le 100000000000000000000000000000000$
50	9.2188(4) Å, 10.1803(2) $\leq c \leq 10.2951(2)$ Å]. The shortest c cell parameter pertains to de-
51	hydrogenated phlogopite from Avellino skarn whereas OH-rich phlogopite from Pompeii composite

53	Overall, it is observed that the crystal chemistry of the micas of the present study extend the known
54	range of the other Vesuvian micas from literature. The Ti-depletion and the wide degree of
55	dehydrogenation of phlogopites from skarns and composite skarns-marbles suggest that the studied
56	samples originated under variable pressure conditions. In addition, the presence of humite in the
57	mineral assemblage seems to indicate the occurrence of devolatilization reactions. The scarce mica
58	occurrence in cumulate and mainly in syenite, instead, may depend on pressure conditions in the
59	magma storage system exceeding the mica stability.
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62	Keywords: phlogopite, crystal chemistry, ejecta, Vesuvius.
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65	INTRODUCTION
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67	Micas are common minerals in magmatic rocks from a wide range of physico-chemical
68	conditions and their importance as petrogenetic indicators is widely accepted (e.g., Lepore et al.,
69	2017, and references therein; Guarino and Brigatti, 2018; Lacalamita et al., 2020). In the peridotite
70	mantle, micas are stabilized at the expense of olivine and pyroxene in presence of hydrous silica-
71	rich metasomatizing agents (e.g., Sekine and Wyllie, 1982) and provide important information on
72	the early fractionation history of ultrapotassic rocks (Fritschle et al., 2013). According to various
73	authors (e.g., Fritschle et al., 2013; Lepore et al., 2017 and references therein), micas of the
74	phlogopite-annite series are particularly useful to define igneous processes leading to the wide
75	compositional spectrum of ultrapotassic rocks. Therein, several crystal chemical studies exist on
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-	micas from different ultrapotassic rocks such as kamafugites (e.g., Lacalamita et al., 2012;
77	micas from different ultrapotassic rocks such as kamafugites (e.g., Lacalamita et al., 2012; Schingaro et al., 2011; Scordari et al., 2012), lamproite-like rocks (e.g., Babushkina et al., 2000;

79 leucitites (e.g., Brigatti et al., 2005; Laurora et al., 2009), Mica-Rich Enclaves (MRE; e.g., Laurora 80 et al., 2007), as well as on micas from alkaline-carbonatite complexes (Guarino and Brigatti, 2018). 81 The role of phlogopite in petrogenetic processes is strictly associated to the multiple concurrent ionic substitutions occurring in the crystal structure: M^{3+,4+}-Tschermak (e.g., Abrecht 82 and Hewitt, 1988; Dymek, 1983; Matarrese et al., 2008; Waters and Charnley, 2002), M^{3+,4+} ↔ 83 vacancy (Waters and Charnley, 2002) and M^{3+,4+}-oxy (e.g., Ventruti et al., 2008 and references 84 85 therein). However, the definition of the substitutions in the crystal structure of phlogopite requires 86 the combination of structural analyses, along with a complete determination of the chemical 87 composition including the analysis of the H₂O content and of the iron speciation, often limiting the 88 application of phlogopite as petrogenetic indicator. In the literature only two complete crystal 89 chemical studies on mica from lamproite rocks (Lepore et al., 2017) and xenolithic ejecta 90 (Balassone et al., 2013) exist. The characterization of phlogopite from three main lamproite rocks -91 Western Alps, Tuscany, and Northern Latium (northern-central Italy) as well as Corsica (France) -92 was undertaken by Lepore et al. (2017), with the aim of shedding light on the possible different 93 origin of micas in relation with the different lamproite-like magmas (e.g., Conticelli et al., 2009). 94 Balassone et al. (2013) proposed the first crystal-chemical and petro-volcanological 95 investigations of trioctahedral micas from the Somma-Vesuvius volcano, one of the most famous 96 worldwide volcanoes located in Italian volcanic belt (at the southern end of the Roman Magmatic 97 Province; Conticelli et al., 2011, and references therein) and associated with the high potassic 98 magmatism developed in the Central Mediterranean area. These authors, by considering different 99 xenolithic ejecta (metamorphic/metasomatic skarns, pyrometamorphic/hydrothermally altered, 100 mafic cumulates) associated to historical lavas of the most recent volcanic activity, AD 1631 -101 1944, found that the variability of the crystal chemical features of micas was consistent with the 102 remarkable variation of their host rocks and magma storage conditions. 103 The present work aims at continuing the previous study on the Somma-Vesuvius micas by 104 focusing on xenoliths from the older and larger eruptions, i.e., the so-called Pompeii and Avellino

105	Plinian eruptions (AD 79 and 3945 ka, respectively). This research wants to improve the knowledge
106	of the crystal chemistry of the Somma-Vesuvius trioctahedral micas and to draw inferences on
107	genetic processes they were subjected to, also offering insights to understanding other similar
108	volcanic systems.
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110	BACKGROUND INFORMATION
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112	Geological and volcanological outline
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114	The Italian Peninsula is unique in the world for the high number, more than fifty, of recent to
115	active volcanoes characterized by extremely different petrochemical features and eruption styles. In
116	this area, the magmatism is alkaline-potassic and developed from Oligocene to present, with the last
117	eruption of leucite-bearing magmas occurring in AD 1944 at Vesuvius (Avanzinelli et al., 2017, and
118	references therein). Vesuvius, or also Somma-Vesuvius (Fig. 1), is a stratovolcano built-up on the
119	Mesozoic carbonate sedimentary sequence of the Campanian Plain (Brocchini et al., 2001). Somma
120	is the old edifice dissected by volcanic collapses consequent the emptying of magma chamber
121	during high-magnitude Plinian eruptions (e.g., Cioni et al., 1999); Vesuvius s.s. is the most recent
122	cone, discontinuously growth inside the caldera until the AD 1944. The volcano produced largely
123	variable magnitude eruptions from high-VEI Plinian and sub-Plinian to low-VEI lava effusion and
124	strombolian to vulcanian events; the last ones characterized the periods of semi-persistent open-
125	conduit - known as "inter-plinian" - volcanism (Santacroce, 1987). The classical and best-studied
126	"inter-plinian" volcanism started after the AD 1631 sub-plinian event and lasted until the most
127	recent eruption in AD 1944 (Arrighi et al., 2001).
128	Based on the rock-geochemistry of the erupted products (Ayuso et al., 1998; Piochi et al.,
129	2006a,b; Di Renzo et al., 2007; Santacroce et al., 2008), different volcanic periods were established:
130	the oldest one (>8-9 ka) produced slightly undersaturated rocks from K-basalt to K-trachyte, the

131 period between 8-9 ka and the AD 79 involved K-phonolite-tephrite to K-phonolite magma 132 compositions, and the period younger than AD 79 characterized for highly undersaturated magmas, 133 from leucititic tephrite to leucititic phonolite. Overall, a wide range of Sr and Nd isotope compositions have been documented, from 0.7066 to 0.7081 and 0.5126 to 0.5124 respectively 134 135 (Ayuso et al., 1998; Piochi et al., 2006a, b; Di Renzo et al., 2007). The compositions mostly reflect 136 evolutionary processes that occurred during upward migration to shallow crustal levels in primary 137 magmas bearing contributions from different mantle reservoirs and slab-derived components. Some 138 authors ascribed the variations of Sr-Nd-Pb-O isotopic compositions in central Italian magmas 139 (including Somma-Vesuvius) to crustal assimilation of carbonates during storage in the upper few 140 kilometres of the crust (e.g., Piochi et al., 2006a; Di Renzo et al., 2007; Iacono-Marziano et al., 141 2008, 2009; Scaillet et al., 2008; Dallai et al., 2011; Pichavant et al., 2014); other researchers favor 142 isotopic variability of associated mantle sources (e.g., Peccerillo, 2005; Moretti et al., 2013). 143 Refilling and mingling processes between different magmas within each volcanic period or eruption 144 event are evident from isotope variations (Cioni et al., 1999; Piochi et al., 2004; Pappalardo et al., 145 2004 and references therein). Recent studies involved the petrological and geochemical evolution of 146 the chemically zoned juvenile samples come from Plinian and sub-Plinian eruptions (Melluso et al., 147 2022). These authors point to the existence of a similar mineralogical paragenesis in the different 148 eruptions, but these minerals have different composition, suggesting the existence of independent, 149 zoned magma batches throughout the activity of the stratovolcano, which possibly started to 150 crystallize at similar depths. 151 However, the processes also involved carbonate contamination and entrapment of crystal 152 mush generated during previous magma storage in the crust (Jolis et al., 2013; Piochi et al., 2006a). 153 The "Avellino" (3945 \pm 10 cal yr BP; Di Vito et al., 2019, and references therein) and 154 "Pompeii" (AD 79) are probably the most famous and widespread plinian eruptions of Somma-155 Vesuvius, with the white and grey pumice deposits associated with sustained pyroclastic column 156 and variable collapsing styles (Gurioli et al., 2005; Sulpizio et al., 2010; Balcone-Boissard et al.,

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157	2012 and references therein; Massaro et al., 2018; Doronzo et al., 2022). White pumices were
158	emitted prior to grey pumices, both phonolitic, but with the white type having a higher
159	differentiation degree (Cioni et al., 2008; Balcone-Boissard et al., 2012). According to the above
160	authors, the volume of fallout tephra is 1.5 km ³ for Avellino and 3.3 km ³ for Pompei. The Avellino
161	products contain sanidine and minor nepheline, clinopyroxene, biotite, amphibole, and oxides;
162	minerals of marialite-meionite ("scapolite") have also found. The Pompeii tephra contains sanidine,
163	pyroxene, phlogopite, K-ferripargasitic amphibole, melanitic garnet and leucite in the groundmass.
164	Similarly to "Pomici di Base" (18.3 ka bp) and "Mercato" (8 ka bp), they determined caldera
165	collapses and dissection of the limestone basement (Cioni et al., 1999) in which their feeding
166	magma chambers were stored and evolved (Cioni et al., 1995; Cioni, 2000; Piochi et al., 2006a, b;
167	Fulignati et al., 2005; Signorelli et al., 1999; Balcone-Boissard et al., 2012; Vona et al., 2020).
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183 entrapment pressure of inclusions of 3.5-5 km and deeper than 7 km (e.g., Marianelli et al., 1999; 184 Cioni, 2000; Fulignati et al., 2005) and Sr-isotope variation through magma storage depths (Piochi 185 et al., 2006a). Based on several authors (e.g., Dallai et al., 2011; Jolis et al., 2013; Piochi et al., 186 2006a), Mesozoic carbonate rocks are the contaminant of magma in the roots of the volcanic 187 structure, mostly at shallow depth. Volcanic rock geochemistry registered carbonate assimilation 188 and is useful tool to study the consequent influence on the magma evolution, its volatile content and 189 degassing modes, the eruption dynamics. However, limestone xenoliths allow detecting the possible 190 physico-chemical changes produced by the magma and carbonate host interaction with implications 191 on rock mechanics, hydrothermal circulation and mineral/metal ore developing. Some authors 192 focused on mineralization and isotope deviance induced by thermo-metamorphism in carbonate 193 xenoliths (e.g., Jolis et al., 2015 and references therein). From petrographic and mineralogical point 194 of view, marble xenoliths mainly consist of calcite with a typical saccharoidal texture. Dolomite is 195 also detected as relic crystals in metadolostone ejecta, characterized by calcite and periclase 196 assemblage; brucite and hydromagnesite commonly occur as secondary retrograde rims around 197 periclase (Gilg et al., 2001). Accessory minerals include diopside, phlogopite, forsterite, 198 chondrodite, humite, clinohumite, norbergite, spinel, graphite, illite, and albite (Barberi and Leoni, 199 1980). 200 Hornfelses, rarely occurring at Somma-Vesuvius, consist of a fine-grained homogeneous 201 granoblastic calc-silicate assemblage of diopside, quartz, feldspars, and wollastonite (Barberi and 202 Leoni, 1980). Marbles and hornfelses are interpreted as thermally metamorphosed carbonate 203 (limestone or dolostone) or marly rocks that have suffered recrystallization and in part 204 devolatilization with modest metasomatic interaction with aqueous fluids or silicate melts (Gilg et 205 al., 2001; Jolis et al., 2015).

Skarns have calc-silicate compositions typically developing in contact-metamorphic aureoles
in response to interaction between limestone or dolostone with a silicate magma. The formation of
these rocks is known to progressively modify the composition of both the magma and the carbonate

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209 wall-rock. Skarn ejecta are common at Vesuvius and display a wide range of mineral compositions 210 and textures also with zoning and metasomatic fronts (also few millimetres in thickness). Typical 211 skarn Ca- and/or Mg-silicate crystal phases are clinopyroxene, phlogopite, wollastonite, olivine, 212 clinohumite, vesuvianite, garnet, gehlenite, and anorthite; other skarn minerals can be calcite, 213 dolomite, meionite, leucite, fluorapatite, titanite, cuspidine, perovskite, baddeleyite, geikelinte, 214 zirconolite, calzirtite, etc. (Gilg et al., 2001; Pascal et al., 2011; Jolis et al., 2015 and references 215 therein). Different varieties of spinels have been also described, i.e., spinel, magnetite, 216 magnesioferrite and qandilite, the last two phases found in metasomatic forsterite-spinel-calcite 217 skarns (Pascal et al., 2011). Many studies characterized pressure and temperature of skarn 218 formation, the fluid nature and the isotope evolution related to the limestone/dolostone-skarn 219 transition, as reported by Jolis et al. (2015 and references therein). Skarn xenoliths in the Vesuvius 220 eruptive deposits can be linked to relatively shallow magma reservoirs ($\leq 4-10$ km; Auger et al., 221 2001; Civetta et al., 2004; Scaillet et al., 2008 and references therein) just developed within the 222 thick limestone and dolostone carbonate sequence which extend from approximately ≥ 2 km to at 223 least 8 km depth under the volcano (Zollo et al., 1996; Bruno et al., 1998; Auger et al., 2001; 224 Brocchini et al., 2001). 225 The igneous ejecta are volcanic to plutonic (or hypabyssal) rocks; the last one consists of 226 pyroxenites and alkali syenites. Pyroxenites are characterized by high proportions of clinopyroxene 227 with minor phlogopite and are likely of cumulate origin. The alkali syenites are fine to medium 228 grained and are dominated by subhedral to anhedral sanidine, garnet, and subordinate amphibole, 229 sodalite group minerals (Balassone et al., 2012, 2016), nepheline, and leucite etc. They have been 230 interpreted as part of a crystallizing magmatic mush. 231 As already stated, the mica-bearing samples considered in this study come from Avellino and 232 Pompeii deposits (Table 1); these eruptions have brought a large amount of ejecta, especially of

- skarn type (Gilg et al., 2001; Jolis et al., 2015), to the surface (Joron et al., 1987). The mica
- varieties commonly observed at Somma-Vesuvius are phlogopite and "biotite" (Russo and Punzo,

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235	2004). As reported by Gilg et al. (2001) for micas from skarn and composite ejecta, Mg-rich and Ti-
236	poor phlogopite, typically occur in zoned skarn ejecta associated to diopsidic/hedenbergitic to (less
237	common) "fassaitic" clinopyroxene, F-bearing vesuvianite, wollastonite, gehlenite, meionite,
238	forsterite, clinohumite, anorthite, and Mg-poor calcite, with many accessory minerals (i.e., apatite,
239	spinel, magnetite, perovskite, baddeleyite, and various undefined REE-, U-, Th-, Zr-, and Ti-rich
240	minerals).
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243	SAMPLES AND EXPERIMENTAL METHODS
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245	The analyzed micas occur in xenoliths collected from outcrops close to Somma Vesuviana,
246	Ercolano and Terzigno towns (Fig. 1), i.e., in skarns, cumulates, syenites, and composite (marble-
247	skarn) lithotypes (see Table 1 and Fig. 2a). After preliminary mesoscopic observations, a
248	combination of polarized optical microscopy, X-ray powder diffraction (XRPD) and scanning
249	electron microscopy equipped with an energy-dispersive spectrometer (SEM-EDS) was used to
250	characterize the mica-bearing xenoliths from the textural, mineralogical, and petrographic point of
251	view.
252	Bulk rock geochemistry was carried out at two laboratories, i.e. Activation Laboratories LTD
253	(Ancaster, Canada) and Bureau Laboratories Ltd. (Vancouver, Canada). At Activation Laboratories
254	(sample set PS40, PS41, PS44, PS50, PS54, PS55, PS57, PS58, PS60, PS80, PS81, PS82, PS84,
255	PSK-R), major elements were analyzed via ICP-OES (inductively coupled plasma-optical emission
256	spectrometry) using a Thermo Jarrell-Ash ENVIRO II ICP. Trace elements were determined by
257	ICP-MS with a Perkin Elmer SCIEX ELAN 6000. Uncertainty is less than 3% for major oxides,
258	less than 15% for Co, Y, Zr, and Tb, and less than 5% for all other trace elements (see
259	www.actlabs.com). At Bureau Laboratories (sample set PS62, PS65, PS68, PS70, PS73, PS77),
260	major elements were analyzed by X-ray fluorescence (XRF) and inductively coupled plasma

261 emission spectrometry (ICP-ES), using LiBO₂/Li₂B₄O₇ fusion. Minor and trace elements were 262 determined by inductively coupled plasma-mass spectrometry (ICP-MS), using a four-acid (HNO₃-263 HClO₄-HF-HCl) digestion. Loss on ignition (LOI) was calculated by weight loss after ignition at 264 1000°C. The uncertainty is less than 3% for major/minor oxides, less than 5-10% for trace elements 265 (see http://acmelab.com). In both cases, loss on ignition (LOI) was calculated by weight loss after 266 ignition at 1000°C. 267 XRPD was conducted on powdered rocks using a Seifert-GE diffractometer ID 3003 268 (DiSTAR, University of Naples, Italy). Intensity profiles were collected in the 20 range of 3-80° 269 using Ni-filtered CuKa radiation at 40 kV and 30 mA, with a step size 0.02°, at a scanning time of 270 10 s/step. The diffraction patterns were processed using the RayfleX software package. A minor set 271 of measurements were performed by using a X'Pert Powder diffractometer by PANalytical (Istituto 272 Nazionale di Geofisica e Vulcanologia, Osservatorio Vesuviano, Napoli, Italy), equipped with a 273 high speed PIXcel detector, Ni-filtered, CuKα radiation (40 kV and 40 mA in the 3-70 °2θ range, 274 0.02° steps at 8 s/step) and a PANalytical B.V. software HIGHScore Plus version 4.9 for the 275 elaboration of the diffraction profiles (Mormone et al., 2014). 276 Electron dispersive spectroscopy (EDS) microanalyses and backscattered electron images 277 (BSE) of the whole mineral assemblages in thin sections were obtained with an INCA X-stream 278 pulse processor and the 4.08 version Inca software (Oxford Instruments detector), interfaced with 279 the JEOL JSM 5310 (DiSTAR, Naples, Italy). Observations were made in backscattered electron 280 mode (BSE), at accelerating voltage 15 kV, 50–100 µA filament current and variable spot size and 281 a working distance of 20 mm. The following reference standards were used: albite (Si, Al, Na), orthoclase (K), wollastonite (Ca), diopside (Mg), almandine (Fe), rutile (Ti), barite (Ba), 282 283 strontianite (Sr), metallic chromium (Cr), rhodonite (Mn), pyrite (S), sphalerite (Zn), galena (Pb), 284 fluorite (F), apatite (P), sylvite (Cl), Smithsonian phosphates (La, Ce, Nd, Sm, Y), gallium arsenide 285 (As), and metallic vanadium (V). Analytical errors are 1% relative for major elements and 3% 286 relative for minor elements.

287	In order to determine the chemical composition by electron microprobe analysis (EPMA) of
288	mica crystals, they were selected under the binocular microscope from the grains of crushed rocks,
289	and then prepared as polished opaque mounts (by means of Bueheler EpoKwich TM epoxy resin, as
290	well as Bueheler polishing cloths and diamond suspensions). Chemical analyses were carried out
291	using a Cameca SX50 WDS instrument (CNR-IGAG, Rome) under the following experimental
292	conditions: 15 nA sample current, 10 kV accelerating voltage, 10 µm beam diameter; no Na loss
293	was detected during the analyses. From each sample, we selected three to five mica grains, and
294	performed five to eight microanalyses, reporting the averages in Table 3. The standards were:
295	wollastonite (Si, Ca), titanite (Ti), corundum (Al), magnetite (Fe), periclase (Mg), jadeite (Na), and
296	orthoclase (K). Data were corrected using the PAP program (Pouchou and Pichoir, 1991). Relative
297	uncertainty on the analytical measurements was 1% for major elements, 4% for minor elements,
298	and 10% for F.
299	Scanning electron microscopy (SEM) examination of morphological features of selected mica
300	crystals were carried out by means of both the JEOL JSM 5310 instrument, and a small subset of
301	samples was also observed bFEI Quanta 200 SEM (IPCB-CNR Pozzuoli, Italy).
302	Single crystal X-ray diffraction (SCXRD) data were collected on mica crystals from selected
303	rocks (PS68, PS77, PS50, PS62, PS73) both from Avellino and Pompeii eruptions by means of a
304	Bruker AXS X8 APEXII automated diffractometer (Dipartimento di Scienze della Terra e
305	Geoambientali, University of Bari, Italy), equipped with a CCD detector and graphite-
306	monochromatized Mo K α radiation ($\lambda = 0.71073$ Å) operating at 50 kV, 30 mA and 40 mm crystal-
307	to-detector distance. The whole Ewald sphere (±h,±k,±l) was recorded up to θ ~45° with a scan
308	width 1.0°/frame and an exposure time 20 s/frame. The APEX program suite allowed to optimize
309	the collection strategy (Bruker, 2010); the SAINT package (Bruker, 2007) was used for the
310	integration of the intensities of reflections and the correction of the Lorentz and polarization effects;
311	the SADABS software (Bruker, 2009) was employed for the empirical absorption correction;

312 XPREP (Sheldrick, 2008) was used for the subsequent analysis of the intensity data and the

313 assignment of the space group.

In the case of PS50 sample, CELL_NOW program (Sheldrick, 2003a) identified two twinned
individuals rotated of 180° around a twin axis parallel to [310]; the TWINABS software provided
for a semi-empirical absorption correction (Sheldrick, 2003b).
Structure refinements were carried out in *C2/m* space group by means of the program

517 Structure remembers were carried out in C2/m space group by means of the program

318 CRYSTALS (Betteridge et al., 2003) considering the reflections with $I > 3\sigma(I)$. Overall scale factor,

atomic positions, cation occupancies and anisotropic atomic displacement parameters were refined,

320 starting from the coordinates of micas belonging to the phlogopite-fluorophlogopite series in

Lacalamita et al. (2020). The occupancy of the tetrahedral site was constrained to 1; the Mg vs. Fe

322 occupancies in the octahedral site were varied with full occupancy constraint for all samples except

for PS50 where only Mg was refined both in M1 and M2; a restraint (1.00 ± 0.08) on the occupancy

324 of the K atoms at the interlayer site was used to allow the occupancy of this site to assume values

higher or lower than 1. The analysis of the difference-Fourier maps for all samples except for PS77

326 evidenced the presence of a residual electron density peak ~ $1 \text{ e}^{-/\text{Å}^3}$ which may be considered as

327 the position of the hydrogen atom, at ~ 0.9 Å from the O4 oxygen. It was included in the refinement

328 using the strategy of refinement described elsewhere (Lacalamita et al., 2020).

For PS77 sample, at the end of the refinement in C2/m space group, an *R*-value of 4% was

obtained; analysis of the difference Fourier synthesis showed two peaks ~ 3.5 and 1.5 e^{-1}/A^3 at 0.57,

331 0, 0.22 and 0, 0.16, 0, respectively. These peaks are hints of $\pm b/3$ shifts of the octahedral sheet,

332 which leads to desymmetrization of the whole structure (Schingaro et al., 2001). Therefore, a

reduction of symmetry from C2/m to C2 was attempted considering extra (disordered) atoms with

respect to the ordered part of the structure. The latter was refined using proper constraints to impose

the C2/m symmetry. The final refinement converged to R 2.9% and no significant residual electron

density was observed. The same disorder affected the PS73 although at minor extent (residual peaks

 $\leq 1.3 \text{ e}^{-1}/\text{Å}^{-3}$). Since the electron density of the disorder atoms was $\leq 0.2 \text{ e}^{-1}$, the disorder modelling

338	was not included in the final refinement of this sample. Crystallographic interchange format (CIF)
339	files are also provided as supplementary materials.
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341	
342	RESULTS AND DISCUSSION
343	
344	The xenoliths
345	
346	Petrographic description
347	The Pompeii xenolithic samples (# 12 samples) are represented by three lithologies, i.e., skarns,
348	composite skarns-marbles and cumulates, whereas the Avellino samples (# 8 samples) mostly
349	consist of skarns and one syenite (Fig. 2b). SEM-BSE micrographs of selected samples in thin
350	sections are shown in Fig. 3.
351	The skarn samples are characterized by a texture variable from granoblastic with equigranular
352	crystals of phlogopite, clinopyroxene and calcite often randomly oriented (decussate texture), to
353	poikiloblastic with small inclusions of various minerals mainly in larger megacrysts (Fig. 3a-h).
354	These samples are mostly composed of Ca±Mg silicates, among which the most common are
355	clinopyroxene, vesuvianite, humite, clinohumite, sodalite, forsteritic olivine, leucite, and garnet,
356	detected in variable amounts (Table 1), locally accompanied by calcite and a wide variety of other
357	μ m-sized mineral phases, difficult to identify with just chemical analyses and optical observations.
358	An association with phlogopite-clinopyroxene±spinel±calcite±apatite - recognized in many skarns
359	of Somma-Vesuvius - was observed both in Pompeii and Avellino samples (Table 1 and Fig. 3a-h).
360	Sodalite and anorthite occur in two Avellino skarns (Table 1 and Fig. 3b). Humite occurs in
361	Avellino (PS55, PS77, PSK-R) and Pompeii (PS70, PS73) skarns, chondrodite in Avellino skarn
362	PSK-R and clinohumite in the Avellino skarn PS41 (i.e., Fig. 3c,f). Titanite was found only in
363	Avellino PS84 skarn, where rare mixed Ca-Ti and Zr-Ti oxides (as perovskite, pyrochlore

14

364 supergroup minerals, and zirconolite) also occur (Fig. 3g). Magnetite is present in Pompeii skarn 365 PS58, PS68, and PS70 (Table 1). Baddeleyite (Fig. 3a,d,h) is quite common and observed in five 366 skarn samples in tiny and disseminated crystals (Avellino: PS41, PS60, PS82, and PSK-R; Pompeii: 367 PS84); thorianite also occurs in Pompeii skarn PS58 (Fig. 3c) in tiny individuals. Carbonates are 368 represented by rare and small calcite crystals, observed in five samples both from Pompeii and 369 Avellino eruptions (Table 1). Tiny pyrite inclusions were observed in Avellino skarns (PS60 and 370 PSK-R) (Fig. 3d). Other trace minerals are cuspidine, as well as likely bismoclite (Table 1). 371 In composite Pompeii skarns-marbles (Table 1), the contacts between the two lithologies can 372 be variously sharp, with the carbonate components relatively more (PS44, PS54, PS62, PS65) or 373 less (PS40, PS50) prevailing over skarn-related assemblages (i.e., sample PS44 in Fig. 2). Calcite 374 displays a granoblastic saccharoidal texture, occasionally very friable, and often with quite fine (< 375 500 µm) crystals (Fig. 3i,l,m); dolomite is recorded in samples PS44, PS50 and PS54 (Fig. 3k-m). 376 A poikiloblastic texture is also observed (partly as in Fig. 31). The silicate minerals include, together 377 with mica, humite, clinohumite, chondrodite, forsteritic olivine, vesuvianite, titanite and meionite 378 (Fig. 3i,j,l,m); Accessory spinel, magnetite, apatite, galena, pyrite, thorianite, baddelevite, and 379 stolzite are locally detected (Table 1 and Fig. 3i). An undetermined Si-Zn-Fe-Pb phase was 380 observed in sample PS44 (Fig. 3j); this could correspond to a mixture of a silicate-rich phases 381 alternating with sulfide-rich phases forming a colloform texture, but no reliable chemical analyses 382 were carried out. 383 Syenite PS57, with holocrystalline texture, is composed of K-feldspar, feldspathoids (leucite, 384 nepheline, and sodalite), phlogopite, clinopyroxene, with small amounts of apatite and britholite 385 (Fig. 3n).

The two cumulate samples (Table 1) are basically formed by clinopyroxene and phlogopite (Fig. 30, p); small inclusions and/or vuggy linings of apatite, sodalite, baddeleyite, bismoclite, and vesuvianite are locally found, together with trace amounts of late fluorite, mimetite, and sylvite (Table 1; Fig. 3,o,p).

390

391 Mineral chemistry

392	The main mineral compositions detected in the investigated mica-bearing lithics are
393	illustrated in Tables S1-S4. Clinopyroxenes, recorded in skarns, syenite and cumulates of both
394	Pompeii and Avellino ejecta (Table S1), have zoned diopside/hedenbergite compositions (i.e., Fig.
395	4b, h), or also subsilicic ferroan aluminian diopside ("fassaite") with significant Ca-Al-Tschermak
396	exchange component (i.e., skarn PS60 and cumulate PS80; table S1). As reported by Gilg et al.
397	(2001), the Vesuvian "fassaitic" skarn pyroxenes generally have higher Ti and lower Mn contents
398	than Al-poor skarn diopsides.
399	Olivine was detected in skarns PS58, PS73, and PSK-R and in composite rock PS62 (Table
400	1); it has a forsteritic composition (Fo ₈₈₋₉₉) comparable to forsterites from metasomatic cumulates
401	and skarns, but Mg-richer than magmatic olivines (Fo ₆₇₋₈₇) (Gilg et al., 2001, and references therein;
402	Fulignati et al., 2005). Among various silicates detected in the analyzed xenoliths, Table S1 also has
403	representative chemical analyses of anorthite, feldspathoids (sodalite, leucite, and nepheline),
404	garnets (grandite), and humite. Anorthite, nepheline, spinel, and garnets have been already reported
405	in skarns from AD 79 (Fulignati et al., 2005). However, apart from sodalite and humite, in our
406	sample set, these silicates mostly occur in the Avellino ejecta (Table 1).
407	F-rich apatite, with low Cl amounts, was furthermore detected in ejecta from both eruptions
408	(samples PS58, PS60, PS80, and PS84), whereas mimetite was found only in sample PS81 as last
409	phase to crystallize (Table S2). In particular, the humite and clinohumite in carbonate-rich rocks is
410	known in the literature (e.g., Rice, 1980; Young and Morrison, 1992), although in our samples not
411	often associated with olivine and pyroxene.
412	Accessory baddeleyite, zirconolite, and pyrochlore supergroup minerals, testifying variable

- 413 leaching/mobilization processes of HFSE elements (such as Ti, Zr, Nb) during metasomatism which
- 414 concentrated these elements in the skarn as oxides (Pascal et al., 2009), are presented in Table S3,

415 together with magnetite, spinel, thorianite, and perovskite; sulfides are also sporadically found and

416 represented by galena and pyrite (Tables 1 and S3).

417 Calcite can be found both in skarns and skarns-marbles (Tables 1 and S4); calcite exhibits the

- following oxide content: 0.09-3.43 wt% of MgO, 0-1.20 wt% FeO, 0-5.58 wt% MnO, and up to
- 419 1.13 wt% SrO and 1.21 wt% BaO. Dolomite, occurring only in three composite samples, has up to
- 420 1.40 wt% FeO, 0.04 wt% MnO and 0.18 wt% BaO.
- 421 The mineralogical assemblage of the Pompeii and Avellino skarn and composite xenoliths
- 422 investigated in this study shows some differences from that reported in Balassone et al. (2013).
- 423 Indeed, the xenoliths investigated here do not contain britholite and glass, instead detected in the
- 424 Ca-Mg-silicate xenoliths from the AD 1631 and AD 1944 eruptions from Balassone et al. (2013);
- 425 also cuspidine was found only in trace amounts in the studied samples.
- 426

427 Xenoliths geochemistry

428 The chemical composition of the studied xenoliths has a wide compositional range in terms of

429 major elements contents, as SiO₂, CaO, and MgO (Fig. 4; Table 2), closely related to their complex

430 mineralogical assemblages. Skarns and skarn-marble xenoliths from the Pompeii eruption are

431 characterized by a broad compositional spectrum, with SiO₂, CaO, and MgO in the ranges 6.4-48.7

432 wt%, 4.2-38.1 wt% and 3.7-30.9 wt%, respectively. The highest CaO concentration is detected in

- 433 rock samples with a composite nature, mainly reflecting the prevailing presence of calcite (and
- 434 minor dolomite) in the marble components. The Avellino skarns have a SiO₂ narrower range of

435 variation (30.1-44.6 wt.%) compared to the Pompeii xenoliths (Fig. 4a), also due to the absence of

- 436 composite rocks in that analysed sample set, and CaO and MgO ranges of 3.7-23.4 wt.% and 8.9-
- 437 31.2, respectively (Fig. 4b). The more frequent occurrence of pyroxene can justify the coupled
- 438 higher SiO₂ and lower CaO content in the Avellino with respect to the Pompeii skarns (Table 1). In
- 439 CaO vs. SiO₂ and MgO vs. CaO diagrams (Fig. 4a, b) Pompeii and Avellino skarns fall
- 440 approximately close to the compositional field of AD 1631 skarns of Balassone et al. (2013, see

441 Fig. 4); some samples also plot in or very close to the AD 79 and AD 472 skarn field of Jolis et al.

442 (2015), while the Pompeii composite ejecta PS44, PS54, and PS62 are close to the marble

443 compositions of these authors.

The Avellino syenite sample (PS57) has the highest SiO₂ values (49.5 wt%) of the analyzed
sample batch, and CaO and MgO values of 6.5 wt% and 4.6 wt%, respectively (Fig. 4a, b); it falls
in the Somma-Vesuvius syenites-foidolites-essexites field from literature (see Fig. 4 in Balassone et
al., 2013, and references therein), as well as in the igneous rocks compositional field of Jolis et al.
(2015).
As regards the two Pompeii cumulate samples (PS80 and PS81), they have SiO₂, CaO, and
MgO in the ranges 41.5-44.6 wt%, 4.1-12.3 wt% and 14.2-16.6 wt%, respectively, which are close

to data from literature (Fig. 4a, b), also considering the igneous rocks compositional field of Jolis etal. (2015).

453 In the main variation diagrams of selected major elements against SiO₂ (Fig. S1 a-f); a semi-

linear trend for K₂O, Na₂O, Al₂O₃, Fe₂O₃, and TiO₂ can be defined for the skarn and skarn-marble

455 samples, with significantly higher values of K₂O for Pompeii sample PS73 and of Fe₂O₃ for

456 Pompeii samples PS58 and PS73. The MgO concentrations in the skarn suite are more scattered, as

457 similarly observed by Jolis et al. (2015). A negative trend can be also observed in Fig. S1 g-j, where

458 K₂O, Na₂O, Al₂O₃, Fe₂O₃ are plotted against CaO concentrations. In general, some samples with

459 composite nature (i.e., PS44, PS62, and PS62) are featured by low amounts in major elements

460 oxides as K₂O, Na₂O, Al₂O₃, Fe₂O₃, and TiO₂, as expected for marble-related rocks. By comparing

the investigated samples and those of Balassone et al. (2013), the skarn samples have similar values

462 (Fig. S1). As already pointed out by Jolis et al. (2015), Somma-Vesuvius skarns typically have a

463 large compositional range in terms of major elements, owing to the occurrence of variable silicates-

- 464 oxides-carbonates assemblages (with different amounts of clinopyroxene, phlogopite, olivine,
- 465 humites, spinel, calcite, etc.). In particular, the investigated Pompeii and Avellino skarn xenoliths
- and Pompeii composite skarn-marble plot between the marble xenoliths and the igneous

467 compositions in Jolis et al. (2015) (Fig. 4), depicting a continuous array; a similar trend can be 468 pointed out in various Harker's diagrams (Fig. S1). The lack of systematic trend for MgO against 469 SiO₂ suggests a combination of limestone and dolostone protoliths (Jolis et al., 2015). 470 Cumulate sample PS80 is quite close to the cumulate samples of Balassone et al. (2013) in 471 terms of K₂O, Na₂O, Al₂O₃, Fe₂O₃ vs. SiO₂ and CaO correlations, whereas sample PS81 plots 472 behaves differently (Fig. S1); it is worth noting that none of the Pompeii cumulates from this study 473 display the high TiO₂ amount of up to 1.9 wt% (Fig. S1 f) in the AD 1944 eruption cumulates 474 recorded by Balassone et al. (2013). 475 Synce 476 wt% K₂O and 4.58 wt% Na₂O), as reported in Fig. S1a,b,c. In any case, cumulate and syenite ejecta 477 display a narrow range of major oxide composition that plots in the intermediate position of the 478 trend depicted by the entire sample set (Fig. 4; Fig. S1). 479 Whole-rock trace element concentrations are reported in Table 2 and a selection of elements 480 is plotted against CaO wt% in Fig. S2. Trace element arrays are scattered for the skarn and 481 composite skarn-marble samples, similarly to literature data. Broadly negative trends can be 482 observed for Ba (with higher values for Pompeii composite sample PS40 and Avellino skarns PS60 483 and PS84), Zr (except to skarns PS55, Avellino, and PS82, Pompeii), Rb, Nb, V, Cs, and Tl when 484 plotted against CaO (Fig. S2 a-h). The higher Sr concentrations are detected in some skarn-marble 485 samples (PS44, PS62, and PS65, Pompeii), usually associated with the occurrence of calcite in the 486 mineral assemblage (Fig. S2b). The Th values are commonly lower than 6 ppm, while it is higher in 487 the skarn samples PS70 and PS73 (Avellino) and PS68 and PS77 (Pompeii) (Fig. S2i), without 488 correlation with Th phases (Table 1). The concentrations of Cr, Ni, and Co are commonly low in 489 the skarn samples (Table 1). Among base metals, Cu is very low, excepting for some Avellino 490 samples (i.e., up to 40 ppm in PS55); Zn concentrations from 110 ppm to 660 ppm in Avellino 491 skarn PS58 (Table 1), and Pb in the range <5-30 ppm up to 240 ppm in Pompeii skarn-marble PS44 492 are measured.

- 493 Cumulate samples PS80 and PS81 have low amounts of Ba, Sr, Zr, and Th, while sample
- 494 PS81 has higher concentrations of Rb, V, Nb, Cs, and Tl compared to both all other xenoliths (Fig.
- 495 S1c, d, f, g, h) and cumulate nodule in Balassone et al. (2013).
- 496 The Pompeii syenite PS57 generally have low concentrations of trace elements, except for
- 497 Rb, Cs, and Tl amounts, which are closer to the cumulate samples.
- 498 The chondrite-normalized REE patterns shown in Fig. 5a display an enrichment in LREE
- 499 (light rare earth elements: La, Ce, Pr, Nd) and MREE (medium rare earth elements: Sm, Eu, Gd,
- 500 Tb, Dy) relative to HREE (heavy rare earth elements: Ho, Er, Tm, Yb, Lu), with a nearly flat
- 501 pattern for HREE. These patterns follow the trend those of recent activity ejecta of Balassone et al.
- 502 (2013, see Fig. 5a), with a wider variation in the LREE values; moreover, the positive Gd anomaly
- 503 found in the cumulate sample of these authors was not detected in the Pompeii cumulates from the
- 504 present study. Moreover, the composite skarn-marbles are more depleted in REE respect to most of
- 505 the samples.
- 506 In the multi-elemental patterns (Fig. 5b), some elements such as Ba and Ta have a deep
- trough for all samples; a similar behavior is observed in most of samples for Th, Sr, P, and Ti, with
- 508 some exceptions represented by skarns and/or composite rocks. A strongly Pb enrichment is
- 509 observed in composite sample PS44 (Pompeii), whereas Pompeii skarn PS82 is characterized by
- 510 higher concentrations of U and REE. Compared to the data of Balassone et al. (2013), a similar
- trace elements trend can be pointed out; higher concentrations of Ba, P, Eu, Tb, Y, Tm, and Yb of
- the AD 1944 cumulate of the cited authors are observed in comparison with the Pompeii cumulatesof this work.
- 514
- 515 Micas

516

517 Mica crystals are commonly light green in color in skarn ejecta and composite rocks, whereas 518 they have variable color from dark green to brown in the cumulates and syenite. Via SEM

20

observation, the micas habit is mainly euhedral, and clusters of hexagonal platy crystals are
common. Micas can be found as laminae reaching up to 1 mm in length in cumulate and syenite
samples, whilst smaller size of tens or few hundreds of µm can be detected in the marbles and
skarns (Fig. 6).

523

524 Chemical composition

525 In the classification diagram *feal* against *mgli* (Fig. 7a), the Pompeii and Avellino micas fall within the compositional field of phlogopite¹. They also define a linear trend between ^[VI]Al-rich and Mg-526 poor "end-member", represented by PS68 (Avellino skarn), and a group of micas characterized by 527 low ^[VI]Al and high Mg contents, mainly belonging to Pompeii composite xenoliths. In the central 528 529 part of this array, PS57 (Avellino syenite) and PS80 and PS81 (Pompeii cumulates) crystals are 530 found. A similar trend was also observed for Vesuvian micas from Balassone et al. (2013), with ^[VI]Al-rich and Mg-poor biotite from cumulate (AD 1944) on one side and micas from skarn (AD 531 532 1631) on the opposite site. The Si vs. Al content in tetrahedral site is nearly close to 4 apfu (atoms per formula unit: Fig. 7b), suggesting negligible $^{[IV]}Fe^{3+}$ occupancy in this site $(^{[IV]}Fe^{3+} = 0.0.07)$ 533 534 apfu); also, the general decrease of the total content of Al with increasing Si (Table 3 and Fig. 7b) is compatible with an Al³⁺-Tschermak substitution. 535

Using the tetrahedral diagram reporting the main octahedral Ti and ^[VI]Al cations compared 536 537 with F and Mg# (Fig. 8a), it is possible to observe that the skarn-related Avellino phlogopites can reach lower Mg#/2 values (up to 0.78, sample PS68) and higher ^[VI]Al (up to 0.25 apfu, sample 538 539 PS55) and Ti (up to 0.15 apfu, sample PS68) values compared to skarn- and composite rocks-540 deriving Pompeii phlogopite (see also Fig. S3a, b). The fluorine amounts vary in the interval 0.09-0.71 apfu for the Avellino phlogopite from skarn and in a slightly higher range (0-0.79 apfu) for 541 542 micas of Pompeii metamorphic-metasomatic samples (Fig. 8a and Fig. S3c). Regarding the Pompeii cumulates, micas PS80 and PS81 have the highest ^[VI]Al (up to 0.27 apfu) and F (up to 0.85 apfu) 543

¹ Possible lithium amounts, not analyzed in this study, may be considered negligible for the classification in the diagram of Fig, 7a.

values, respectively, of the whole analyzed samples (Table 3). Mica from Avellino syenite lies

approximately between the fields of skarn/composite rocks dataset from the two eruptions (Fig. 8a

and Fig. S3). Micas from Avellino skarns and Pompeii cumulates mainly have higher iron contents

547 compared to the other samples (Fig. 8b), as also resulting from bulk-rock geochemistry (Fig. S1).

548 Moreover, most of the investigated micas have a negligible "brittle mica component", due to a Ba

amount lower than 0.005 apfu, except for micas from Avellino skarns PS60, PS84, and PSK-R

which display Ba amounts in the interval 0.02-0.03 apfu (Fig. S4; Table 3).

551 In Fig. 8, the comparison between micas from the present study and those from recent activity

of Vesuvius (Balassone et al., 2013) emphasizes that micas from AD 1631 skarns appear slightly

closer than those from Pompeii eruption (Fig. S3 and S4); micas from

pyrometamorphic/hydrothermally altered ejecta of AD 1872 are chemically rather different (as

exemplified by higher F concentrations), and also mica from cumulate of AD 1944 eruption do not

556 fit with samples from Pompeii cumulates from this study.

557 The substitution of Ba for K is mostly accounted for the coupled substitution $^{[XII]}Ba + ^{[IV]}Al$

558 \leftrightarrow Si + ^[XII]K (Fig. 9), with a linear array and end-members represented by PSK-R (Avellino skarn)

and PS44 (Pompeii composite nodule) micas. A similar trend can be observed for Vesuvius micas

from the most recent volcanic activity (Balassone et al., 2013). It is worth noting that in few

561 Avellino and Pompeii micas small amounts of Sr (0.02-0.03 apfu) were detected (see samples PS41,

562 PS60, PS84, and PSK-R in Table 3).

563 Plotted in the Al₂O₃ vs. TiO₂ and FeO_t (wt%) diagrams (Fig. 10a, b), the chemical feature of

the studied micas from skarns and composite rocks fits most of those from the Vesuvian skarns

565 (mainly Avellino, Pompeii, and AD 472 Pollena eruptions) of Gilg et al. (2001) in terms of Al₂O₃

and TiO₂, whereas have different values compared to micas in various xenoliths (lavas,

567 metamorphic nodules, and composite rocks) from AD 1872 ejecta (Petti et al., 2010). The

568 concentrations of FeO_t (wt%) and to some extent of the Mg# values of micas from skarn of Avellino

and Pompeii (Fig. 10c) eruptions, variously different from the other studied micas (see also Fig.

- 570 8b), likewise differ from the other literature data. The FeO_t content is higher in Avellino than in
- 571 Pompeii micas.
- 572

573 Crystallographic features

- 574 Mica crystals representative of the two clusters in the skarn and composite skarn-marble lithotypes
- 575 identified based on the classification plot in Fig. 7 were selected for a deeper investigation by
- 576 means of SCXRD. Unfortunately, it was not possible to obtain good-quality crystals for SCXRD
- 577 investigations both from the syenite and cumulates.
- 578 Crystals from samples PS68 and PS77 (Avellino skarns) which exhibit slightly low Mg# as
- well as crystal from PS50, PS62, and PS73 (Pompeii composite skarns-marbles and skarns) with
- 580 high Mg# were deeper analyzed. The results of the structure refinements converged to good values
- of the agreement indexes, $R_1 \le 3.29$ % and $wR_2 \le 4.98$ %. All samples are 1*M* polytypes and
- 582 crystallize in the C2/m space group, except for the PS77 which was refined in C2 space group
- 583 (Table 4). The PS50 crystal resulted to be twinned with 0.67(2):0.271(9) refined proportion
- between the two individuals. The samples have a great variability of the lattice constants, especially
- 586 10.2951(2) Å]. The PS68 crystal has the shortest c cell parameters [10.1803(2) Å] which is typical
- 587 of de-hydrogenated phlogopite (Lacalamita et al., 2011; Schingaro et al., 2011; Lacalamita et al.,
- 588 2017). On the contrary PS50 and PS62 crystals from Pompeii have a c cell parameter [10.2932(4)
- and 10.2951(2) Å, respectively], which approximates that of the endmember phlogopite
- 590 (Redhammer and Roth, 2002), in agreement with the trend in Fig. 7. Regarding the bond lengths
- 591 (Table 5), almost all the studied samples are geometrically 'homo-octahedral' (i.e., <M1-O> equals
- the <M2-O> distance within three standard deviations, Weiss et al., 1992). Among the individual
- 593 distances, the PS68 crystal has the M2-O4 bond length remarkably shorter with respect to the other
- two <M2-O3> distances (Table 5) which is one of the clues of the possible occurrence of oxy-type
- substitutions. Consistently, it also has the smallest cell volume (Table 4).

596

597 Crystal-chemistry

598 The structural formulae of the selected mica crystals are listed in Table 6. They were obtained by

- 599 combining the results from the microprobe analysis (Table 3) and structure refinements (Tables 5
- and 7), with estimated H₂O content (Ventruti et al., 2008) following the approach reported

elsewhere (see for instance Schingaro et al., 2012; Lacalamita et al., 2017).

602 The formulae are balanced based on the main substitutions: M^{3+} -Tschermak [$^{VI}M^{2+}$ + $^{IV}Si^{4+}$

$$603 \quad \leftrightarrow^{VI}(Al^{3+}, Fe^{3+}) + {}^{IV}Al^{3+}]; Ti - oxy [{}^{VI}M^{2+} + 2(OH)^{-} \rightarrow {}^{VI}Ti^{4+} + 2(O)^{2-} + H_2\uparrow]; M^{3+} - oxy [{}^{VI}M^{2+} + 2(OH)^{-} \rightarrow {}^{VI}Ti^{4+} + 2(OH)^{-}]; M^{3+} - oxy [{}^{VI}M^{2+} + 2(OH)^{-}]; M$$

604
$$(OH)^{-} \leftrightarrow {}^{VI}M^{3+} + O^{2-} + \frac{1}{2}H_2$$
; $OH^{-} \rightarrow F^{-}$. The tetraferriphlogopite [${}^{IV}Fe^{3+} \leftrightarrow {}^{IV}Al$] substitution also

605 characterizes the PS73 mica.

Fig. 11 exhibits the variation of the *c* cell parameter versus the Mg number $[Mg/(Mg+Fe_{tot})]$. The highest values of the Mg number pertain to the OH-rich phlogopite (PS50 and PS62 Pompeii samples), as well as fluorophlogopite from the AD 1872 eruption of Vesuvius. The latter exhibits *c* shortening typical of dehydrogenation (see below). For the remaining crystals, as well as for other phlogopite and biotite from Vesuvius xenoliths (Balassone et al., 2013), the *c* parameter decreases with the increase of the Fe content, that, alone or with other metals, may be involved in oxy-type substitution in micas.

In the case of the more dehydrated (see section above) PS68 crystal, the entry of Ti and M^{3+} 613 614 cations in the octahedral site by oxy-mechanism involving the deprotonation process at the anionic 615 O4 site, strongly affects the values of geometrical parameters (short c cell parameter, M2-O4 and 616 K-O4 distances; high values of M2 bond lengths distortion, BLD_{M2} ; strong displacement, $shift_{M2}$, of 617 the M2 octahedral cation from the geometric center of the octahedron, etc., see Tables 4 and 5) 618 which are consistent with those found for Ti-oxy phlogopite (Cruciani and Zanazzi, 1994). On the 619 contrary, the very low extent of oxy-mechanisms and $OH^- \rightarrow F^-$ replacement in PS50 and PS62 620 end-member crystals explains the high values of the c cell parameter and the low values of the 621 BLD_{M2} and shif_{M2} parameters (Table 5). For these crystals, the high charge cations are equally

622 distributed between the M1 and M2 octahedral sites as testified by the very close values of the 623 <M1-O> and <M2-O> distances and volumes of the octahedra (Table 5). These crystals also have 624 the highest value (~11° vs 9°, Table 5) of the tetrahedral rotation angle, α , with respect to the other 625 studied crystals. This parameter describes the in-plane rotation of adjacent tetrahedra in opposite 626 directions respect with c^* . The large α parameter of PS50 and PS62 crystals indicate a good 627 congruence between the tetrahedral and octahedral sheet. The PS77 and PS73 samples exhibit the same degree of dehydrogenation ($O^{2-}+F^{-}$ content ~ 628 629 0.8 apfu) resulting in a c cell parameter (~10.23 Å, Table 4) intermediate between those of dehydrogenated (PS68 sample) and OH-rich phlogopite (PS50 and PS62 samples). The high charge 630 cation at the octahedral sites is essentially involved in M³⁺-Tschermak substitutions (Table 6). In 631 632 the case of the PS73 crystal, the best structural formula was obtained by considering the total iron 633 as Fe₂O₃ which was distributed between both the octahedral and tetrahedral sites. 634 In the Shift_{M2} parameter against Ti and F (Fig. S5a) four points fall close to the AD 1631 635 literature data and define a slightly flat pattern, whereas a particularly high Ti and Shift_{M2} values for the mica from Avellino skarn PS68 can be observed. Also in the Shift_{M2} vs. F (apfu) diagram (Fig. 636 637 S5b) sample PS68 plots as an outlier, due to its high value of $Shift_{M2}$, while the other samples 638 behave as in Fig. S5a. Fig. S5c, d shows the in-plane rotation angle, i.e., α parameter (°), as a 639 function of the Ti and F contents; in the α vs. Ti diagram (Fig. S5c) micas from the Pompeii skarn-640 marble samples PS50 and PS62 and from the skarn PS77, as well as the Avellino skarn PS73, are 641 closer to the literature AD 1631 samples, whereas mica from the Avellino skarn PS68 plots very 642 close to micas from the AD 1944. In the α vs. F plot of Fig. S5d a broad negative trend is observed, 643 with the investigated micas close to the AD 1631 and AD 1944 data. 644 645 Crystallization conditions

646

647	Considering the mineralogical assemblages observed in most xenoliths, indicative of crystallization
648	conditions during their formation, and following Guarino and Brigatti (2018, and references
649	therein), we try to estimate the crystallization temperature at which the micas are formed. Using the
650	geothermometer of Luhr et al. (1984) [T (K) = $838/(1.0337 - Ti/Fe^{2+})$] based on the ratio between
651	octahedral Ti and Fe ²⁺ , we have obtained similar ranges of temperatures, i.e., 541 - 583 °C and 545
652	- 596°C for the Pompeii and Avellino xenoliths (Table 3). A higher temperature (824 °C) was
653	obtained for mica in skarn PS68, due to its high Ti/Fe ²⁺ ratio. For the Pompeii composite sample
654	PS62, the Ti/Fe ^{$2+$} value of 0.54 made the above formula inapplicable.
655	The formula proposed by Uchida et al. (2007) for pressure determination cannot be applied because
656	the studied micas have the total aluminum less than 1.5 apfu.
657	We tested other thermobarometers based on different mineral pairs, as apatite-phlogopite
658	(Ludington, 1978; Seifert et al., 2000), garnet-clinopyroxene (Duchêne and Albarède, 1999) and
659	garnet-biotite (Reche and Martinez 1996; Wu, 2015), but unreliable results were obtained because
660	the minerals/rocks are commonly not in equilibrium.
661	
662	
663	IMPLICATIONS
664	
665	The mineralogical features and genetic inferences of the studied mica-bearing ejecta may
666	provide valuable information for reconstructing the environmental conditions at Somma-Vesuvius
667	and in similar geological systems.
668	Firstly, the crystal chemical analysis of micas from different xenoliths related to Avellino and
669	Pompeii Plinian eruptions of Somma-Vesuvius volcano evidenced that they belong to the
670	phlogopite-annite join. They are affected by different cations substitutions and dehydrogenation
671	degrees, that largely extend the known range of the other Vesuvian micas (as in Balassone et al.,
672	2013), providing some genetic indications, that can be of certain utility also for the understanding

the quiescent volcano dynamics. Their host rocks have a wide composition reflecting the abundance
of carbonate or silicate phases (i.e., CaO richer Pompeii skarns-to-marbles or SiO₂ enriched
Avellino ejecta, respectively). The main mineralogical and petrological aspects can be summarized
as follows:

- 678 Micas from skarns and composite skarn-marble rocks. The studied Avellino and Pompeii
- 679 phlogopites from skarns and composite skarn-marbles are generally Ti-depleted, as also found in
- 680 most of phlogopites from Avellino, Pompeii, and Pollena of Gilg et al. (2001), and slightly more Ti-
- depleted than the AD 1631 micas of Balassone et al. (2013). Concurrently, they have variable
- amounts of Al and F, in keeping with the compositional ranges reported by the cited literature.
- 683 However, the micas set of the present work (i.e., both samples of metamorphic/metasomatic and of
- 684 igneous origin) never reach the high F concentrations detected in micas in AD 1872 xenoliths
- 685 (Balassone et al., 2013), which correspond to fluorophlogopite composition and testify for a
- strongly dehydrated condition. Similarly, the studied phlogopites display chemical characteristics
- 687 significantly different from the micas batch related to the AD 1872 nodules of Petti et al. (2010).
- 688 These AD 1872 xenoliths are pyrometamorphic/hydrothermally altered nodules explaining the very
- 689 different nature and crystallization conditions of micas.
- In terms of Mg# and Fe, the micas of this study are more variable compared to the previously
- 691 cited literature data; in particular, three samples from skarn of Avellino eruption, i.e., PS55, PS68,
- and PS77, have higher Fe contents and lower Mg# values. Interestingly, mica PS68 is also distinctly
- highest in Ti concentration of the whole sample set, accompanied by a lower *c* parameter;
- 694 concurrently, the host rock is Fe-rich and have quite high MgO and TiO₂ values, in relation with
- 695 silicate-oxide (mica+clinopyroxene+magnetite+vesuvianite) mineral assemblage. According to
- Laurora et al. (2007), micas with the highest Ti contents can be ideally in equilibrium with a more-
- 697 evolved melt enriched in Fe and Ti.

27

698 The chemical and structural data indicate that micas PS73 (Pompeii) and PS77 (Avellino) 699 samples exhibit the same degree of dehydrogenation, resulting in a c cell parameter intermediate 700 between those of dehydrogenated (PS68 sample; Avellino) and OH-rich phlogopite (P50 and PS62 701 Pompeii samples). It appears that the phlogopite dehydrogenation variability is higher in ejecta 702 from highest VEI Pompeii eruptions, whereas it is more clustered in those from the AD 1631 and 703 1944 eruptive events. Highly dehydrated mica structure coupled with a fluorophlogopite 704 composition found in AD 1872 ejecta (Balassone et al., 2013), characterized by lower P and higher 705 T environment, have never been found in the present study. Following their reasoning, more 706 variable pressure conditions should be consistent with the environment of fluid-carbonate wall rock 707 interaction developed around the Pompeii and Avellino magma chambers. 708 Concerning the temperatures, fluid inclusion studies of Gilg et al. (2001) give a range of 800-709 890 °C for most skarn-derived minerals (vesuvianite, calcite) indicating initial temperatures related 710 to the entrapment of these fluid inclusions during crystal growth, similarly was obtained for mica in 711 sample PS68 (824 °C). Differently the generally low temperatures estimated for micas in the skarns and composite skarn-marbles (541-596 °C) well represent the minimum crystallization temperatures 712 713 related to contact metamorphism. 714 This can be supported by the presence of humite, suggesting that dehydrated micas can occur 715 under high CO₂/H₂O (and F) conditions during their growth by contact metamorphism with with 716 carbonatic limestones. As proposed by Young and Morrison (1992), the humite features of our 717 samples would result by devolatilization reactions driven by addition of aqueous fluid and 718 extraction of heat in endoskarn around plutons at less than 600°C. The presence of humite species 719 in the ejecta paragenesis deserves further studies in order to evaluate stable isotope shift in skarn's 720 calcite correlated with its growth proposed by the Authors. 721 As concerns the Ba content, it is worth noting that micas from Avellino skarn PS60, PS84, 722 and PSK-R have higher Ba amounts compared to the other samples; at the same time, the related 723 host rocks PS60 and PS84 also has the highest Ba concentrations, even though this correlation does

724	not occur for PSK-R sample. Finally, the higher Sr concentrations detected in three of the Pompeii
725	skarn-marble nodules are not reflected in micas composition.
726	Unlike Pompeii, Avellino magmas were usually leucite-free. Leucite in Avellino skarns could
727	suggest that the K enrichment already occurred in the Vesuvius magmatism before the AD 79,
728	either as a primary magma feature or as an effect of carbonate assimilation (Piochi et al., 2006a;
729	Mollo et al., 2010).
730	All the skarns and skarn-marbles contain several F-bearing minerals supporting the role of
731	"fluorine-bearing aqueous-carbonic fluids" in the magma-host rock processes suggested in previous
732	studies (Gilg et al., 2001 and references therein).
733	
734	Micas from cumulates and syenite. Micas related to cumulates (Pompeii eruption) are
735	compositionally different from cumulate sample from the AD 1944 eruption of Balassone et al.
736	(2013); they are distinctly more depleted in Fe, Ti and Ba and enriched in F and Mg, whereas the Al
737	concentration is variable.
738	Mica from syenite (Avellino eruption) appears to record similar low to medium
739	concentrations of selected elements detected in the source rock, as Ti, Ba, Mg, and Fe.
740	Syenite and one cumulate xenolith have the highest content of Cs and Rb, as expected from
741	the major role of the magmatic processes. The low content of crystallizing mica mainly in syenite
742	can determine the observed enrichment of Cs and Rb. This can be associated with pressure
743	conditions exceeding the mica stability, i.e., depth shallower than 4 km (Fabbrizio and Carroll,
744	2008; Scaillet et al., 2001). Similar minimum crystallization temperatures related to contact
745	metamorphism are also obtained for micas in syenite (554 °C) and and other skarn to cumulate
746	xenoliths (558-583 °C).
747	Remarkably, sodalite is common in Avellino and not in Pompeii magmas (Piochi et al., 2006a
748	and references therein). Its occurrence in the Pompeii cumulate should be associated with an
749	Avellino's remnant thrown out by the AD 79 eruption.

750

Finally, this study highlights the importance of micas crystal chemical features in the geneticcontexts.

753 As stated by Lepore et al. (2017), commonly petrologists use only microprobe data and 754 petrographic details to classify mica and the formula is derived by placing the obtained chemical 755 data in worksheets. Often, to get charge neutrality, fanciful cation distributions are hypothezised 756 (e.g., Mg in the tetrahedral site), neglecting crucial information like the role of the light elements (including a possible dehydrogenation) and a good estimation of the Fe^{2+}/Fe^{3+} ratio. Indeed, both 757 the estimation of the Fe^{2+}/Fe^{3+} ratio and the quantification of light elements (H and Li) are also key 758 759 variables to obtain insights on the relationship between the studied phase and the crystallization 760 conditions. 761 In the case of absence of direct estimation of the above variables the crystal chemical details may be of help: the hydrogen content can be estimated from the c parameter and the Fe^{2+}/Fe^{3+} ratio 762 763 the one providing the best agreement between the EPMA-derived and the structure refinement-764 derived mean atomic numbers as well as between calculated and observe bond distances (Ottolini et 765 al., 2012). Structural evidence (i.e., geometrical distortions) of substitution mechanisms can also be 766 employed. This alternative approach is indeed widely recognized and has been pursued in the 767 present work.

768

769

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771

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- 1091
- 1092
- 1093 LIST OF FIGURE CAPTIONS
- 1094
- 1095 Figure 1. Geological sketch map of Somma-Vesuvius (modified after Peccerillo, 2005) with
- samples provenance.

1097

- 1098 Figure 2. (a) Selected xenoliths from Somma-Vesuvius (see Table 1): PS44 (composite skarn-
- 1099 marble) and PS80 (cumulate) are from Pompeii eruption; PS57 (syenite) and PS77 (skarn) are
- 1100 from Avellino eruption (bar scale equal to 5 cm). (b) Distribution of the samples set according

to the lithotype and the occurrence.

1102

1103 Figure 3. Selected SEM micrographs (backscattered electrons, BSE mode) of mica-bearing 1104 xenoliths in thin sections. (a-h) skarn xenoliths; (i-m) skarn-marble xenoliths; (n) svenite 1105 skarns; (o,p) cumulate xenoliths. A thorough description of the mineral associations are given 1106 in the text. Mineral abbreviations (after Warr et al., 2021) are as in Table 1. 1107 1108 Figure 4. Selected major elements variations diagrams for the investigates Somma-Vesuvius 1109 xenoliths compared with literature data, (a) CaO against SiO₂ and (b) MgO against CaO 1110 (wt%). 1111 1112 Figure 5. Geochemical characteristics of the investigated mica-bearing xenoliths. (a) Chondrite-1113 normalized REE and (b) primitive mantle-normalized minor to trace element patterns 1114 (normalization values from McDonough and Sun, 1995, and Lyubetskaya and Korenaga, 1115 2007, respectively). Data from Vesuvius xenoliths (recent activity) of Balassone et al. (2013) 1116 are also reported. 1117 1118 Figure 6. SEM micrographs of micas from Somma-Vesuvius samples. (a) PS54; (b) PS57; (c) and 1119 (d) PS80, (Sdl, sodalite). 1120 Figure 7. (a) Octahedral sheet composition of the studied micas in the *feal-mgli* plot, where *feal* = 1121 ^[IV](Fe_{tot} + Mn + Ti - Al) and $mgli = {}^{[VI]}(Mg + Li)$ of Tischendorf et al. (2007). (b) Plot of 1122 1123 tetrahedral Si vs. Al; the correlation between Si and total Al is reported in the inset. Symbols 1124 used for micas are the same as in Fig. 4. 1125 Figure 8. Discrimination diagrams for the studied micas. Ti-^{VI}Al-F⁻-Mg#/2 (a) and Ba-Fe_{tot}-F⁻-Mg# 1126

|--|

- 1128 2016). Symbols as in Fig. 4.
- 1129 Figure 9. Plot of the ^[XII]Ba + ^[IV]Al \leftrightarrow Si + ^[XII]K substitution mechanism for the studied micas
- (symbols as in Fig. 4).
- 1131 Figure 10. Plot of Al_2O_3 vs. TiO₂ (a) and FeO_t (b), as well as Mg # vs. TiO₂ (c) for micas of the
- 1132 present study, compared to selected Somma-Vesuvius micas composition from literature.
- 1133 Symbols explanation as in Fig. 4; black dotted line field = micas from AD 1872 eruption
- ejecta (Petti et al., 2010); grey solid line field = micas from skarn xenoliths mainly from
- 1135 Avellino, Pompeii and AD 472 Pollena eruptions (Gilg at al., 2001).
- 1136
- Figure 11. Plot of the *c* parameter vs. Mg number [Mg/(Mg+Fe_{tot})] for the investigated trioctahedral
 micas from the Avellino and Pompeii xenoliths (symbols as in Fig. 4). For comparison, values
- of phlogopite, fluorophlogopite and biotite of Vesuvius xenoliths from the AD 1631, AD
- 1140 1872 and AD 1944 eruptions (Balassone et al., 2013) are reported.
- 1141

1142

- 1143 LIST OF TABLE CAPTIONS
- 1144
- 1145 Table 1. Labels, occurrence, host rock and mineral assemblages of the Somma-Vesuvius samples.
- Sample provenances (see Fig. 1) are Somma Vesuviana and Ercolano (Avellino eruption) and
 Terzigno (Pompei eruption).

1148

Table 2. Whole rock composition of the investigated mica-bearing xenoliths (major and minorelements in wt. % and trace elements in ppm).

1151	
1152	Table 3. Chemical composition (EPMA, mean values on multiple point analyses expressed in wt%
1153	of oxides) of the investigated micas. Estimated values of formation temperature for selected
1154	samples are also given (see text for further explanation).
1155	
1156	Table 4. Experimental details and relevant crystallographic data of the investigated micas.
1157	
1158	Table 5. Selected bond distances (Å) and distortional parameters (calculated as reported in
1159	Lacalamita et al., 2011) of the studied mica crystals.
1160	
1161	Table 6. Structural formulas in atoms per formula unit (apfu) of the selected micas for the
1162	crystallographic and crystal-chemical characterization.
1163	
1164	Table 7. Mean atomic numbers (electrons, e ⁻) and average distances (Å) of cation sites derived by
1165	structure refinement (X-ref), and chemical data (EMPA) of the investigated mica crystals.
1166	Average error for mean atomic numbers is $\pm 0.5 \text{ e}^-$.
1167	
1168	
1169	Supplemental Materials
1170	
1171	Supplemental Figure S1. Selected oxides of major and minor elements against SiO ₂ and CaO (wt%)
1172	of Somma-Vesuvius xenoliths. Symbols as in Fig. 4.
1173	
1174	Supplemental Figure S2. Selected trace elements plots against CaO (wt%) of Somma-Vesuvius
1175	xenoliths. Symbols as in Fig. 4.

1176	
1177	Supplemental Figure S3. Plot of octahedral Al (a), Ti (b), and F (c) contents against Mg# for the
1178	studied micas (symbols as in Fig. 4).
1179	
1180	Supplemental Figure S4. Plots of F vs. Ba for the studied micas (symbols as in Fig. 4).
1181	
1182	Supplemental Figure S5. Plots of Shift _{M2} (Å) parameter vs. Ti (a) and F^{-} (b) contents (apfu) and of α
1183	(°) parameter vs. Ti (c) and F (d) contents for the investigated trioctahedral micas from the
1184	Avellino and Pompeii xenoliths (see Tables 3 and 5). For comparison, values of phlogopite,
1185	fluorophlogopite, and biotite of Vesuvius xenoliths from the AD 1631, AD 1872 and AD
1186	1944 eruptions (Balassone et al., 2013) are reported (symbols as in Fig. 4).
1187	
1188	
1189	Supplemental Table S1. Chemical composition (wt% oxide) of miscellaneous silicates in the
1190	studied ejecta (mineral symbols as in Table 1; AV = Avellino eruption, PP = Pompeii
1191	eruption).
1192	
1193	Supplemental Table S2. Chemical composition of phosphates and arsenates (wt% oxide) in the
1194	studied samples (mineral symbols as in Table 1; AV = Avellino eruption, PP = Pompeii
1195	eruption).
1196	
1197	Supplemental Table S3. Chemical composition of oxides (wt% oxide) and sulfides (wt% elements)
1198	detected in the studied samples (mineral symbols as in Table 1; AV = Avellino eruption, PP =
1199	Pompeii eruption).
1200	

- 1201 Supplemental Table S4. Chemical composition of carbonates (wt% oxide) in the studied samples
- 1202 (mineral symbols as in Table 1; AV = Avellino eruption, PP = Pompeii eruption).

1203

1204 **CIF file**: CIF FILES SV PHL.

1205



Figure 1



Figure 2



Fig. 3 (continued)



Fig. 3 (end)



Figure 4



Figure 5



Figure 6



Figure 7



Figure 8



Figure 9



Figure 10



Figure 11

Table 4

Experimental details and relevant crystallographic data of the studied crystals.

Avellino xenoliths Pompeii xenoliths									
	PS68	PS77	PS50	PS62	PS73				
Crystallographic data									
Space group C2/m C2/m									
a (Å)	5.3169(1)	5.3218(1)	5.3072(2)	5.3055(1)	5.3188(1)				
b (Å) 9.2041(2) 9.2188(4) 9.1945(3) 9.1893(1) 9									
<i>c</i> (Å)	10.1803(2)	10.2343(2)	10.2932(4)	10.2951(2)	10.2376(1)				
β (°)	99.983(2)	100.011(2)	99.942(2)	99.902(1)	99.978(1)				
Cell volume (Å ³)	490.65(2)	494.46(2)	494.74(3)	494.45(2)	493.94(1)				
Z	4	4	4	4	4				
Experimental details									
Reflections collected 5033 6066 3816 6048 6117									
Reflections unique 1256 1829 1292 1805 1787									
R _{merging} [R _(int)] (%)	2.79	2.76	2.26	1.79	1.86				
Reflections used ($I>3\sigma(I)$)	953	1333	1124	1585	1432				
No. of refined parameters	56	71	58	56	56				
Goof ^a	1.087	0.983	0.965	0.947	1.045				
R ₁ ^b (%)	2.66	2.85	3.27	1.98	2.63				
wR_2^{c} (%)	3.23	4.23	4.96	2.39	3.50				
$\Delta \rho_{min} / \Delta \rho_{max} (e^{-1} / A^3)$ -0.41/0.71 -0.51/1.01 -0.53/0.96 -0.44/0.49 -0.39/1.38									
a: Goodness-of-fit = $[\Sigma[w(F_o^2 - F_c^2)^2]/(N-p)]^{1/2}$, where N and p are the number of reflections and									
parameters. respectively.									
b: $R_1 = \Sigma[F_0 - F_c]/\Sigma F_0 $.									
c: $wR_2 = \left[\sum \left[w(F_0 - F_0)^2\right]/\sum \left[w(F_0)^2\right]\right]^{1/2}$; w = Chebyshev optimized weights.									

Table 5

Selected bond distances (Å) and distortional parameters (calculated as reported in Lacalamita et al. 2011) of the studied crystals.

Avellino xenoliths Pompeii xenoliths									
	PS68	PS77	PS50	PS62	PS73				
T-O1 T-O1' T-O2 T-O3 <t-o></t-o>	1.656(2) 1.656(2) 1.656(1) 1.670(2) 1.660(4)	1.660(2) 1.661(2) 1.6594(9) 1.661(1) 1.660(3)	1.664(2) 1.666(2) 1.666(1) 1.660(2) 1.664(4)	1.6660(5) 1.6669(5) 1.6668(3) 1.6681(5) 1.6670(9)	1.6572(8) 1.6577(8) 1.6584(5) 1.6504(7) 1.656(1)				
M1-O4(x2) M1-O3(x4) <m1-o></m1-o>	2.031(2) 2.086(1) 2.068(2)	2.048(2) 2.083(1) 2.071(2)	2.043(3) 2.079(2) 2.067(4)	2.0420(6) 2.0737(5) 2.0631(8)	2.0454(8) 2.0853(6) 2.072(1)				
M2-O4(x2) M2-O3(x2) M2-O3'(x2) <m2-o></m2-o>	1.987(2) 2.081(1) 2.112(2) 2.060(3)	2.057(1) 2.071(1) 2.071(1) 2.066(2)	2.057(2) 2.063(2) 2.077(2) 2.065(3)	2.0495(5) 2.0587(5) 2.0721(5) 2.0601(9)	2.0558(6) 2.0741(7) 2.0848(6) 2.072(1)				
<m-o></m-o>	2.063(4)	2.068(3)	2.066(5)	2.061(1)	2.072(1)				
K-O1(x4) K-O2(x2) K-O2(x2) <k-o2'(x2) <k-o>_{inner} <k-o>_{outer} <k-o></k-o></k-o></k-o></k-o2'(x2) 	2.946(2) 3.346(2) 2.948(3) 3.362(3) 2.947 3.351 3.149	2.962(2) 3.368(2) 2.961(2) 3.368(2) 2.962(3) 3.368(3)	2.929(2) 3.428(2) 2.926(3) 3.429(3) 2.928 3.428 3.178	2.9121(5) 3.4566(6) 2.9111(7) 3.4594(9) 2.912 3.458 3.185	2.9834(9) 3.3380(9) 2.981(1) 3.338(1) 2.9826 3.338 3.160				
T _D -O1 T _D -O11 T _D -O2 T _D -O3 <t<sub>D-O></t<sub>		1.65(4) 1.61(4) 1.72(5) 1.66(1) 1.67(7)							
K _D -O1 (x2) K _D -O1'(x2) K _D -O2(x2) K _D -O2'(x2) K _D -O11 (x2) K _D -O11'(x2) <k<sub>D-O11'(x2) <k<sub>D-O>_{inner} <k<sub>D-O>_{outer}</k<sub></k<sub></k<sub>		2.31(1) 2.307(6) 2.32(2) 3.02(5) 2.96(4) 3.36(4)							
$ \begin{array}{l} \alpha \left(^{\circ} \right) \\ \text{Volume}_{\text{T}} \left(\overset{\text{A}^{3}}{\text{A}} \right) \\ \text{BLD}_{\text{T}} \\ \text{Volume}_{M1} \left(\overset{\text{A}}{\text{A}} \right) \\ \text{BLD}_{M(1)} \\ \text{Volume}_{M2} \left(\overset{\text{A}}{\text{A}} \right) \\ \text{BLD}_{M(2)} \\ \text{Shift}_{M2} \left(\overset{\text{A}}{\text{A}} \right) \\ \underline{\Delta}_{K \circ 0} \left(\overset{\text{A}}{\text{A}} \right) \\ \underline{\Delta}_{K \circ 0} \left(\overset{\text{A}}{\text{A}} \right) \\ \end{array} $	8.86 2.344 0.302 11.556 1.172 11.433 2.392 0.062 0.404	8.94 2.347 0.032 11.638 0.733 11.608 0.409 -0.019 0.406	11.06 2.363 0.105 11.581 0.774 11.552 0.343 -0.019 0.500	11.53 2.367 0.190 11.491 0.670 11.446 0.393 -0.017 0.520	8.80 2.340 0.349 11.665 0.857 11.659 0.519 -0.016 0.400				
parameters w volumes and	Errors on distortion parameters, estimated by varying the refined positional parameters within one standard deviation are in the following ranges: < 0.5% for volumes and shifts: 0.1-13% for α and Δ_{KO}								

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Table 6

Structural formulas in atoms per formula unit (apfu) of the selected micas for the crystallographic and crystal-chemical characterization.

	Sample	
Avellino xenoliths	PS68	$(K_{0.89}Na_{0.04})_{\Sigma=0.93}(Mg_{2.03}Fe_{_0.09}^{2+}Fe_{_0.48}^{3+}Mn_{0.01}Al_{0.23}Ti_{0.15})_{\Sigma=2.99}(Si_{2.88}Al_{1.12})_{\Sigma=4.00}O_{10.83}F_{0.25}Cl_{0.01}OH_{0.91})_{\Sigma=0.93}$
	PS77	$(K_{0.94}Na_{0.06})_{\Sigma=1.00}(Mg_{2.32}Fe_{0.37}^{2^{+}}Fe_{0.37}^{3^{+}}Mn_{0.06}AI_{0.16})_{\Sigma=2.98}(Si_{2.80}AI_{1.20})_{\Sigma=4.00}O_{10.03}F_{0.71}OH_{1.26}$
Pompeii xenoliths	PS50	$(K_{0.95}Na_{0.08}Ca_{0.02})_{\Sigma=1.05}(Mg_{2.70}Fe^{3^+}_{0.04}Al_{0.24}Mn_{0.01})_{\Sigma=2.99}(Si_{2.81}Al_{1.19})_{\Sigma=4.00}O_{10.16}OH_{1.84}$
	PS62	$(K_{0.95}Na_{0.02})_{\Sigma=0.97}(Mg_{2.62}Fe_{2.0.09}^{3^{*}}Al_{0.18}Cr_{0.05}Ti_{0.05})_{\Sigma=2.99}(Si_{2.71}Al_{1.29})_{\Sigma=4.00}O_{10.10}F_{0.04}OH_{1.86}$
	PS73	$(K_{0.92}Na_{0.06})_{\Sigma=0.98}(Mg_{2.85}Fe^{3^{*}}_{}0.14}Mn_{0.01})_{\Sigma=3.00}(Si_{2.98}AI_{0.93}Fe^{3^{*}}_{}0.09})_{\Sigma=4.00}O_{10.10}F_{0.69}OH_{1.21}$

Table 7

Mean atomic numbers (electrons, e^{-}) and average distances (Å) of cation sites derived by structure refinement (X-ref), and chemical data (EMPA). Average error for mean atomic numbers is $\pm 0.5 e^{-}$.

	Avellino	xenoliths	Pompeii xenoliths				
	PS68	PS77	PS50	PS62	PS73		
T e ⁻ _{X-ref}	14.00	14.24	14.00	14.00	14.00		
T e [⁻] _{EPM}	13.72	13.70	13.70	13.68	14.04		
e (M1+2M2) X-ref	45.34	41.47	36.94	39.01	38.07		
e (M1+2M2) EMPA	45.72	42.86	36.91	38.42	38.09		
K e⁻ _{X-ref}	17.92	17.61	16.89	18.89	17.23		
K e ⁻ _{EMPA}	17.35	18.52	19.33	18.27	18.14		
<t-o> _{X-ref}</t-o>	1.660	1.660	1.664	1.667	1.656		
<t-o> _{EMPA}</t-o>	1.656	1.659	1.659	1.662	1.655		
<m-o> _{X-ref}</m-o>	2.063	2.068	2.066	2.061	2.072		
<m-o> EMPA</m-o>	2.054	2.075	2.068	2.067	2.079		