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Microchemistry and magnesium isotope composition of the Purang

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ophiolitic chromitites (SW Tibet): New genetic inferences

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ABSTRACT

24 New petrographic and microanalytical studies of mineral inclusions in the Purang ophiolitic
25 chromitites (SW Tibet) are used to scrutinize the evolution of the associated Cretaceous
26 sub-oceanic lithospheric mantle section. Silicate inclusions in the chromite grains include
27 composite and single-phase orthopyroxene, clinopyroxene, amphibole, and uvarovite. Most
28 inclusions are sub-rounded or globular, whereas a few inclusions exhibit cubic/octahedral
29 crystal morphologies. The latter are randomly distributed in the large chromite grains, though
30 discrete aggregates are consistently confined to the grain centers. Abundant micron-scale,
31 clinopyroxene inclusions are topotaxially aligned along crystallographic planes. Less
32 abundant sulfide, wüstite, apatite and uvarovite inclusions are observed in some samples.

33 The trace element geochemistry of the Purang chromitite evoke parental MORB- and
34 boninite-like melts, consistent with the supra-subduction zone setting. The $\delta^{26}\text{Mg}$ values of
35 the high-Cr and high-Al chromitites range from -0.25 to -0.29‰ and -0.05 to -0.32‰,
36 respectively. The associated harzburgite has nearly overlapping $\delta^{26}\text{Mg}$ values of -0.13 to
37 -0.37‰, but pyroxenite sills show a rather wider range of the $\delta^{26}\text{Mg}$ values (-0.61 to -0.67‰).
38 The variable Mg isotope signatures, combined with abundant exotic, ultra-high pressure and
39 super reduced (UHP-SuR) mineral inclusions in the chromite grains, suggest that recycling
40 and recrystallization under different mantle conditions played an important role in the genesis
41 and evolution of these rocks. Further, discrete silicate, sulfide, and metal alloy inclusions in
42 the Purang chromitites are comparable to those reported in other Tethyan ophiolites, and
43 collectively suggest a common geodynamic evolution.

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45 **Key words:** Purang ophiolite, SW Tibet, UHP-SuR mineral inclusions, high-Cr and high-Al
46 chromitites, deep mantle recycling, SSZ processes

47

48 INTRODUCTION

49 The occurrence of ultra-high pressure (UHP) minerals, such as diamond, coesite and lamellar
50 exsolutions of clinopyroxene in chromitites from many ophiolitic belts has raised many
51 questions about processes operating in the formation of the oceanic lithosphere (e.g., [Yang et al. 2007, 2014](#);
52 [McGowan et al. 2015](#), [Satsukawa et al. 2015](#), [Su et al. 2015](#), [Gong et al. 2016](#);
53 [Xiong et al. 2018, 2019, 2020a](#)). Micro-diamond occur as inclusions in ophiolitic chromitites,
54 with morphologies and carbon isotopic signatures dissimilar to diamonds in kimberlites and
55 high-grade metamorphic rocks (e.g., [Liou and Tsujimori 2013](#); [Griffin et al. 2016](#)). Unlike
56 diamonds in other geological matrices and environments, the ophiolite-hosted
57 micro-diamonds (and moissanite) are commonly surrounded by amorphous carbon (a hard
58 solid glass, occasionally contains small chromite grains; [Yang et al. 2014](#)).

59 The paragenetic relationships of most ultra-high pressure and super-reduced (UHP-SuR)
60 mineral phases in ophiolites are largely unidentified. Results of numerous experimental
61 investigations indicate that the formation of diamonds and other UHP-SuR minerals in
62 ophiolites occurs at depths between 150 and 300 km ([Yang et al. 2007](#); [Yamamoto et al. 2009](#);
63 [Xiong et al. 2015](#)). Also, peridotites with SuR-UHP mineral inclusions could have been
64 transported to the lithospheric mantle by passive upwelling (forced return flow), likely
65 induced by downward penetration and roll-back of a subducting lithospheric slab (e.g., [Arai](#)

66 [2013](#); [McGowan et al. 2015](#); [Griffin et al. 2016](#)), or related to a super-plume event (e.g.,
67 [Maruyama et al. 2007](#); [Liou and Tsujimori 2013](#); [Xiong et al. 2015](#)).

68 Magnesium isotopes have unique systematics in mantle and crustal materials, offering a
69 new approach for tracing crustal versus mantle source components in the arc magmas. Most
70 significant Mg isotope fractionation is related to low temperature geological processes,
71 whereas little fractionation occurs at high temperatures and pressures ([Teng et al. 2007](#),
72 [2010a,b](#)). Consequently, large deviation from the well-constrained mantle Mg isotope
73 composition in a fresh (unweathered) mafic rock may be indicative of the nature and
74 magnitude of crustal input to the parental magma. Magnesium isotopic values in the arc
75 volcanic rocks ($\delta^{26}\text{Mg} = -0.26 \pm 0.07$) are generally heavier than those of mantle peridotites
76 ($\delta^{26}\text{Mg} = -0.25 \pm 0.07$; [Teng et al. 2010a](#)). Such a deviation in the magnesium isotope
77 signature of these rocks has been attributed to contamination by isotopically heavy Mg in
78 subducting slabs in the mantle wedge (e.g., [Li et al. 2017](#)).

79 Studies of the Mg isotopic composition of mantle rocks can shed light on deep mantle
80 processes and in turn on the accretion history of the Earth (e.g., [Wiechert and Halliday 2007](#)).
81 Theoretical modeling indicates that the Mg isotope fractionation between spinel-group
82 minerals and silicates is dependent on the octahedral ion composition (B_2^{3+}) of the spinel
83 minerals ([Young et al. 2015](#), [Schauble 2011](#)). According to [Schauble \(2011\)](#), inter-mineral
84 Mg isotope fractionations at 727°C (= 1000 K) are 0.6‰ for spinel (MgAl_2O_4), 0.2‰ for
85 magnesiochromite (MgCr_2O_4), and 0.1‰ for magnesioferrite (MgFe_2O_4). The Mg isotope
86 systematics can therefore be used to study the petrogenetic processes operating in the
87 formation of podiform chromitites (e.g., [Dauphas et al. 2010](#); [Teng et al. 2011](#); [Su et al. 2015](#),
88 [2021a](#); [Xiao et al. 2016](#)).

89 In the present study, the microtextural and compositional characteristics of the mineral
90 inclusions in the Purang chromitites are integrated with new magnesium isotope data of these

91 chromitites and the associated peridotites. The data can place constraints on the mantle
92 processes involved in the formation of the studied ophiolite and analogous podiform
93 chromitites in accretionary belts.

94 GEOLOGICAL SETTING

95 The Purang ophiolite is the largest mafic-ultramafic massif (~600 km²) along the western
96 Yarlung Zangbo Suture Zone (YZSZ) in the southern Tibet (Fig. 1). The ultramafic rocks
97 include lherzolite, harzburgite, and subordinate dunite (Su et al. 2019; Xiong et al. 2018, 2019,
98 2020b). Most of the dunite bodies contain irregular masses of deformed chromitites (Fig. 1b).
99 In places, chromitite lenses are 2 to 10 m long and 0.5 to 6 m wide, consistently aligned in a
100 NW-SE direction conformable to the penetrative shear foliation in the host rocks. These
101 lenses are commonly associated with harzburgite, and are surrounded by 5 m thick dunite
102 envelopes (Fig. 2). These chromitite bodies have sharp contacts against dunite, whereas
103 contacts between dunite and harzburgite are obscured or irregular. In the dunite blocks,
104 Cr-spinel grains occur discretely and as disseminations and as inclusions in olivine. The
105 ultramafic section contains rare opicalcite and is cut by dikes and sills of pyroxenite, dolerite,
106 gabbro and basalt (Fig. 1b). The Purang ophiolitic rocks are embedded in a Late
107 Cretaceous mélange of carbonate, shales, and volcanic rocks (Xiong et al. 2013; Liu et al.
108 2011; Fig. 1b).

109 A sequence of basalt, limestone, silty shale and chert overlies the Purang ophiolite rocks
110 (Liu et al. 2015). Fossil Radiolaria in the chert bands indicate an Upper Jurassic-Lower
111 Cretaceous age for these rocks (Huang et al. 2006). In the Purang ophiolite, basalts with
112 mixed enriched-MORB and OIB geochemical affinities were dated at ~137 Ma (Liu et al.
113 2015). Zircons from crosscutting doleritic and gabbroic dikes have U-Pb ages of 144 - 120

114 Ma (Xiong et al. 2020b; Liu et al. 2011), suggesting formation prior to the Early Cretaceous.
115 Peridotites from the Purang ophiolite yielded nearly constant $\epsilon_{\text{Nd}(t)}$ values of +6.1 to +6.2, and
116 $(^{86}\text{Sr}/^{87}\text{Sr})_t$ values of 0.70688–0.70710 (Xiong et al. 2020b).

117 The occurrence of UHP-SuR mineral inclusions in the Purang chromitites was first
118 reported by Yang et al. (2015). In another example of the Tibetan chromitites, Yamamoto et
119 al. (2009) suggested that acicular clinopyroxene inclusions in chromite grains from the
120 Luobusa chromitite were formed under ultra-high pressure (UHP) conditions. Also, the $\delta^{13}\text{C}$
121 and $\delta^{15}\text{N}$ signatures of microdiamonds recovered from Luobusa chromitites and peridotites
122 suggest formation near the mantle transition zone (MTZ) (Xu et al. 2018).

123 ANALYTICAL METHODS

124 For the present study, 68 samples of the Purang chromitites were collected from ten
125 separate ore bodies (Fig. 1b). The mineral chemistry of chromite grains and silicate inclusions
126 were obtained using a JEOL JXA-8100 electron probe micro-analyzer (EPMA) at the Key
127 Laboratory of Deep-Earth Dynamics of Ministry of Natural Resources (Beijing). This EPMA
128 is equipped with three wavelength-dispersive spectrometers and an energy-dispersive Oxford
129 EDX spectrometer. The measuring conditions were 15 kV accelerating voltage, 20 nA beam
130 current, and 2 μm beam diameter. The EPMA data of the studied rocks are given in Tables
131 1-2. Routine quantification to 1% reproducibility was obtained over the many analysis
132 sessions, whereas accuracy remained $\sim 1\text{-}2\%$ for the major elements.

133 The Raman spectra were collected by using a Horiba LabRAM HR Evolution
134 spectrometer at the Institute of Geology, Chinese Academy of Geological Sciences. This

135 spectrometer is equipped with an Olympus BX41 optical microscope. The room temperature
136 of 23 ± 1 °C was maintained during measurements, and a 532 nm air-cooled frequency double
137 Nd: YAG laser was employed under laser power of 80 mW. The scattered light was directed
138 to the spectrograph through a confocal 100 μm pinhole and then was received by a 1800
139 gr/mm diffraction grating. A CCD camera (Syncerity) was consistently cooled to -60 °C
140 while collecting the scattered light. Raman signals were collected for the spectral interval of
141 $100\text{--}600$ cm^{-1} with a spectral resolution of <0.6 cm^{-1} and calibrated with a silicon wafer peak
142 at 520.7 cm^{-1} . The acquisition of the Raman spectra comprised 2 accumulations of 1 s each.
143 Raman shifts were further calibrated with a plane (0001)-parallel section of euhedral
144 pegmatitic α -quartz. The latter showed main Raman peaks at 128, 206, and 464 cm^{-1} (average
145 of 85 standard measurements) during the measuring session.

146 Trace element compositions of chromite grains were obtained by using a laser ablation
147 system with an Agilent 7700x ICP-MS at the State Key Laboratory of Ore Deposit
148 Geochemistry, IGCAS. Laser sampling was performed using a GeoLas Pro 193 nm ArF
149 excimer laser. Argon (900 mL/min) was the make-up gas mixed with the carrier gas, helium
150 (580 mL/min), via a Y-connector before entering the ICP. The spot size was either 60 or 90
151 μm , depending on the size of the analyzed grains.

152 Transmission electron microscopy (TEM) micro-foils ($15 \times 10 \times 0.20$ μm) were prepared
153 using the focused ion beam (FIB) technique (Wirth 2009). A single focused ion beam device
154 (FEI-FIB 200 TEM) at the GFZ Potsdam was used for sample preparation. A Ga-ion beam
155 was applied to protect the surface of the sample and to extract foils (see Wirth 2009). A

156 TECNAI F20 XTWIN model TEM equipped with a Gatan Tridiem™ energy filter, an EDAX
157 Genesis™ X-ray analyzer and a Fishione high angle annular dark field detector (HAADF)
158 were used to study the mineral inclusions. A field emission gun (FEG) was used as the
159 electron source and the applied measuring conditions were 200 kV (Wirth et al. 2009).

160 Fresh representative samples were used for whole-rock Mg isotope analyses. The selected
161 rock samples were crushed in a corundum jaw crusher to 60 mesh, from which ~60 g of each
162 sample was pulverized in an agate ring mill to < 200 mesh. Sample preparation and chemical
163 digestion were carried out in a clean lab environment. Sample powders were weighted and
164 dissolved in a mixture of ultra-pure concentrated acids HF-HNO₃ (~3:1) in Savillex screw-cap
165 beakers. Sealed beakers were placed on a hot plate to 160°C for 24 hrs. The solutions were
166 then evaporated to dryness, dissolved in aqua regia, and heated to 130°C overnight. On the
167 following day, solutions in aqua regia were evaporated to dryness and refluxed with ultrapure
168 concentrated HNO₃ to ensure complete solution. The repeatedly evaporated concentrates
169 yielded a dried residue which was dissolved in 1 N HNO₃ for chromatographic separation.
170 Magnesium isotope compositions of the studied samples were measured by a Neptune
171 MC-ICP-MS at China University of Geosciences (Beijing), following the procedures
172 described in Teng et al. (2010b). The isotopic values were measured by the sample-standard
173 bracketing method in the low-resolution mode. The in-run precision on the ²⁶Mg/²⁴Mg ratio
174 for a single block of 40 measurements was less than ± 0.02‰ (2SD). The USGS standard
175 BHVO-2 gave acceptable δ²⁶Mg values of -0.27 ± 0.04‰ in the measurement session (e.g.,
176 Huang et al. 2015).

177

178

RESULTS

179 **Chromitite bodies and Cr-spinel grains in the Purang peridotites**

180 Irregular chromitite lenses and less common amoeboid chromite patches characterize
181 peridotites of the Purang ophiolite (Xiong et al. 2018). In harzburgite, Cr-spinel grains have
182 variable Cr# and Mg# values of 26.1-44.7 and 54.1-66.6, respectively (Fig. 3). Disseminated
183 Cr-spinel grains in dunite samples have less variable and consistently higher Cr# (~45.9-69.9)
184 and Mg# (49.5-61.8) values (Table 1). In the chromitite masses, Cr-spinel grains have Cr#
185 values of 47.6 - 58.4 and Mg# values of 53.2 - 75.8, classifying them as high-Al chromite
186 (e.g., Melcher et al. 1997; Zhou and Robinson 1997). Also present are Cr-spinel grains with
187 Cr# values of 63.8 to 88.5 and Mg# values of 45.9 - 74.8, typical of high-Cr chromite (Table
188 1).

189

190 **Mineral inclusions in chromite**

191 We identified distinct groups of inclusions in the chromite grains from the Purang chromitite.
192 Discrete globular to anhedral inclusions are composed of pyroxene (Figs. 4, 5, 6).
193 Submicroscopic (~1 μm -thick) acicular silicate inclusions (Fig. 4) seem to be controlled by
194 specific crystallographic planes (e.g., Yamamoto et al. 2009). Although inclusions in the
195 different thin sections are variable in abundance and relative proportions, the two chromite
196 types, high-Cr and high-Al, show no distinct variations in the inclusion types (see
197 Supplementary Material 1).

198 Also observed are euhedral orthopyroxene inclusions (5-20 μm diameter) that exhibit
199 octahedral habits (Figs. 5a - 5c). Other small subhedral to euhedral inclusions (5 - 15 μm)
200 with triangular forms are mainly clinopyroxene (Figs. 5d - 5f). Most of the subhedral
201 inclusions show octahedral morphologies (Fig. 5), and are characterized by ubiquitous
202 interface angles of 54-68° (Figs. 4, 5). Sparse acicular silicate inclusions are arrayed
203 subparallel, though few of them appear to be randomly oriented. The X-ray elemental
204 mapping and EDX of some acicular inclusions show two main peaks, Si and Ca (Figs. 6d),
205 with no apparent Al, Cr and Fe spectra (Figs. 6d; Supplementary Material 2). The abundances
206 of Mg and O are very low (Supplementary Material 2) and some of these acicular inclusions
207 show chemical zoning as indicated by the EDS and WDS results. The Raman spectra
208 collected for these inclusions show scattered peaks superposed on a broad diffuse scattering
209 pattern (Supplementary Material 3). These mixed or composite spectra identify bands with
210 definite M-O stretching at $\sim 322\text{ cm}^{-1}$, Mg-O stretching at $\sim 388\text{ cm}^{-1}$, Si-O-Si bending at ~ 665
211 cm^{-1} , and Si-O stretching at $\sim 1012\text{ cm}^{-1}$ (Supplementary Material 3, Huang et al. 2000).

212 Orientation along specific crystallographic planes and the topotaxial relationship are
213 emphasized by the TEM images of some large chromite grains (Fig. 6). The average
214 composition of many of these inclusions can be formulated as $\text{En}_{52.0}\text{Fs}_{1.5}\text{Wo}_{46.4}$ (Table 2). On
215 the other hand, orthopyroxene inclusions are mainly enstatite with high Mg# values of ~ 96.4
216 to 96.8, and nearly constant Al_2O_3 contents (0.28-0.36 wt.%; Table 2). Moreover, scattered
217 amphibole inclusions exhibit irregular morphologies along or adjacent to micro-fissures in the
218 high-Al chromite grains (Supplementary Material 4). Amphibole inclusions form small

219 subhedral crystals (Figs. 7), commonly associated with phlogopite, apatite, orthopyroxene,
220 and clinopyroxene in composite inclusions.

221 Discrete base metal mineral inclusions, including sulfides and $\text{Fe}\pm\text{Co}\pm\text{Ni}\pm\text{As}$ alloys,
222 were observed in some chromite grains. These inclusions vary significantly in size (from few
223 microns to several hundreds of microns) and exhibit irregular or globular morphologies (Fig.
224 8). Base metal sulfides are commonly associated with metal alloy inclusions. Small, blocky
225 inclusions of pentlandite (Fig. 8f) and a $\sim 5 \mu\text{m}$ -long triangular inclusion of wüstite (FeO)
226 were observed in a single sample (Supplementary Material 5).

227 Dispersed Ni–Cu–Fe (Co) sulfides in the Purang chromitite occur as discrete particles or
228 as inclusions in the chromite grains. Uncommon metal alloys include Ni–Fe, Ni–Cu–Fe, Fe–Co,
229 and Ni–As grains, varying in size from 5 to 200 μm . We observed more abundant Ni–Fe alloy
230 inclusions in the high–Cr chromite than in the high–Al chromite. A single inclusion of an
231 Fe–Co alloy was observed next to a microfracture in a large chromite grain (Fig. 8). Metal
232 alloy inclusions with variable compositions are associated with goethite along microfractures
233 (Fig. 8). In addition, dispersed minute inclusions of Os–Ir–Fe–Ni alloys are intergrown with
234 silicate phases (Supplementary Material 6).

235 The TEM image of an acicular inclusion ($50 \mu\text{m} \times 2 \mu\text{m}$, Fig. 9a) shows bright zones
236 surrounded by a darker domain. Line mapping shows that these bright zones are rich in Si, Fe,
237 O and poor in Al and Mg (Fig. 9b). Another unusual inclusion is the irregular mass of apatite
238 that occupies interstices between chromite grains or perhaps microcracks (Fig. 7b). An
239 irregular grain $\sim 120 \mu\text{m} \times \sim 30 \mu\text{m}$ consists of Ca phosphate and Ti (Fig. 10).

240 An inclusion of uvarovite ($20\ \mu\text{m} \times 100\ \mu\text{m}$) has sharp but irregular contacts with the host
241 chromite grain (Fig. 7c), though its contact appears to be gradational in the TEM image (Fig.
242 10a). The EDX spectrum of this uvarovite inclusion shows prominent peaks for Ca, Si, Al, Cr
243 and O and minor peaks for Fe and Ti. The indexed diffraction pattern (Fast Fourier Transform,
244 FFT) of this inclusion (Fig. 10a) validates its isometric structure.

245 **Minor and trace elements in chromite and Cr-spinel grains**

246 In situ minor and trace element concentration of chromite grains are shown in Table 3. The
247 high-Al chromite grains have Ga contents of 37-61 ppm, 652-2272 ppm Ni, and 186-318 ppm
248 Co, whereas high-Cr chromite has lower Ga (30–41 ppm), Ni (871-1694 ppm) and Co
249 (187-243 ppm) contents. Both the high-Cr and high-Al chromite varieties have similar
250 concentrations of Ti (457-1499 ppm vs 435-1161 ppm), V (777-995 ppm vs 661-1227 ppm),
251 Mn (1043-1348 ppm vs 900-1730 ppm), and Zn (371-571 ppm vs 400-776 ppm) (Table 3).
252 The high-Al and high-Cr chromite varieties show largely overlapping contents of Sc, V, Mn,
253 Zn, and Zr. Within each group, there is a modest positive correlation between Cr# and Sc. The
254 overall data set shows negative correlation between Cr# and both Ga and Ni (Fig. 11). The
255 MORB-normalized trace element patterns of both chromite varieties are similar and show
256 variable depletions in Ga, Ti, and Ni, and relative enrichments in Zn, Co, Mn and V. Most
257 patterns are characterized by depletion in Sc, but some grains are mildly enriched in Sc (Fig.
258 12). Generally, chromite grains from the Purang chromitites show fractionated transition
259 metal compositions, consistent with crystallization from a depleted mantle source. The trace
260 element patterns of the analyzed chromite varieties partially overlap the typical patterns of
261 Cr-spinel from boninites (BON) and high-Cr chromitites of the Ordovician Thetford Mines

262 Ophiolite (RMO; [Pagé and Barnes 2009](#)). The latter is interpreted as a fragment of fore-arc
263 oceanic lithosphere ([Pagé and Barnes 2009](#)).

264 Cr-spinel grains disseminated in the lherzolite and harzburgite members of the Purang
265 ophiolite have similar trace element contents, with variable contents of Ga (16.1-40.2 ppm),
266 Ni (233-1466 ppm), and Co (147-439 ppm) ([Su et al. 2019](#)). The Cr-spinel grains in the dunite
267 samples have similar Cu, Zn, Ca, Rb and Ga contents, but relatively depleted Ti, Co, Nb and
268 Zr.

269 **Magnesium isotopic compositions of the Purang ophiolitic rocks**

270 The Mg isotopic compositions of selected fresh samples from the Purang chromitites and
271 peridotites are given in Table 4. The $\delta^{26}\text{Mg}$ values of five high-Al chromite-dominated
272 chromitites range between -0.32 to -0.05 ‰ ± 0.04 ‰. Three high-Cr chromite-dominated
273 chromitite samples yield overlapping $\delta^{26}\text{Mg}$ values of -0.29 to -0.25 ‰ ± 0.04 ‰. Three
274 harzburgite samples show variable $\delta^{26}\text{Mg}$ values of -0.37 ‰ ± 0.04 ‰ to -0.13 ± 0.02 ‰, and
275 two pyroxenite samples have noticeably light values of -0.67 and -0.61 ‰ ± 0.05 ‰ ([Table 4](#),
276 [Fig. 13](#)). The Mg isotope values of the analyzed samples are distinctively variable for the
277 high-Al chromitite variety compared to the nearly constant values of the high-Cr chromitite
278 and peridotites. The Mg isotopic compositions of the host peridotites are much lighter than
279 those of high-Cr and high-Al chromitites.

280

281

DISCUSSION

282 **Mineralogical and geochemical characteristics**

283 Olivine, orthopyroxene, and clinopyroxene inclusions in ophiolitic chromite grains are
284 commonly interpreted as captive mineral phases (e.g., [Zhou et al. 2014](#)). In the chromite
285 grains from the Purang chromitites, silicate inclusions have Mg# values and Cr₂O₃ and TiO₂
286 contents distinctive from those of olivine and pyroxene in the host peridotites ([Fig. 3](#)). The
287 morphology and distribution of these mineral inclusions clearly suggest an early paragenesis
288 ([Fig. 5](#)). According to [González-Jiménez et al. \(2014\)](#), island arc basaltic magmas can react
289 with other silicate melts or mantle peridotites, leading to an increase in the SiO₂ content to
290 promote chromite grain crystallization. In the mantle peridotite-melt reaction, pyroxene
291 dissolves into the melt and causes variable enrichment in several incompatible elements,
292 including the rare earth elements (REE).

293 The plagioclase, zircon, rutile, amphibole, calcite, apatite and phlogopite inclusions show
294 mostly sharp contact relationships with the host minerals ([Figs. 7, 10](#)). Crystallization of such
295 inclusions may be triggered by late fluid circulation in the subduction zone ([Zaeimnia et al.](#)
296 [2017](#); [Johan et al. 2017](#); [Xiong et al. 2018](#)). The rutile and ilmenite inclusions in the chromite
297 grains can be explained by post-formation high-grade metamorphism of their host rocks. In
298 the subduction zone environment, basaltic melts may alter chromian spinel, and then rutile,
299 ilmenite, and titanite are enclosed during the newly crystallized chromite ([Qiu et al. 2021](#)).

300 The trace compositions of the high-Al and high-Cr chromite varieties from the Purang
301 ophiolite are comparable to the Luobusa chromitite ([Su et al. 2019](#)). The decreasing trends of
302 compatible elements, such as Ni and Ga, in chromite with Cr# suggest that these elements
303 were controlled by both partial melting and melt diffusion ([Fig. 11](#); [Su et al. 2019](#)). The

304 correlation between Zr, V and Sc contents in chromite and Cr# values may suggest diffusion
305 of these elements into melts during melt-rock interaction (Fig. 11). The parental magmas of
306 the Purang dunites and chromitites are suggested to have been intermediate in composition
307 between boninitic and MORB-like melts (Fig. 3). Recycling of both oceanic and continental
308 crustal minerals has been considered as a model of formation of the Luobusa chromitites and
309 peridotites (Zhou et al. 2014; Shi et al. 2007; Yang et al. 2007, 2015; Yamamoto et al. 2009;
310 Wang et al. 2021). Such recycling process is compatible with the high contents of Nb, Zr and
311 Ti in chromitites, as these elements are enriched in the crustal materials. The Sc depletion in
312 the Purang chromitites may be explained by diffusion during melt-rock reaction (Spandler et
313 al. 2007; Pagé and Barnes 2009).

314 Base Metal Sulfide (BMS) and platinum-group element (PGE) mineral inclusions are
315 abundant in chromite grains from the Purang podiform chromitites (Fig. 8). These inclusions
316 are dominated by Ru-rich alloys (Supplementary Material 6; Xiong et al. 2018). According to
317 Prichard et al. (2008), saturation of the PGE is proportional to the extent of reaction between
318 PGE minerals and the melt during chromite formation. This results in PGE minerals
319 (including Ni–Cu–Fe sulfides and PGE alloys) forming micron- to millimeter-scale inclusions
320 in peridotites (e.g., Marchesi et al. 2011; Akmaz et al. 2014). A rise in sulfur fugacity during
321 magma cooling, causes heazlewoodite to transform into millerite and the PGE alloys to
322 dissolve (González-Jiménez et al. 2012), and hence some PGE-rich inclusions could have
323 formed during the recrystallization of chromitites.

324 According to Brenan and Andrews (2001) and Andrews and Brenan (2002), the
325 abundance of Os-bearing laurite indicates high sulfur fugacity values ($\log f_{S_2} = -2$ to -1.3) and
326 temperatures not exceeding 1200°C during chromite crystallization. Microdiamond,
327 moissanite, pseudomorphic octahedral silicates and crustal mineral inclusions (e.g., rutile,

328 monazite, zircon, etc.) have been reported in the Purang lherzolite (Xiong et al. 2019). Also,
329 Yang et al. (2015) described scarce UHP minerals in the Luobusa and Ray-Iz chromitites.

330 The lower-pressure isometric structure chromite phase forms as a consequence of
331 decompression and breakdown of an original high-pressure Ca-ferrite structure chromite
332 phase (CF-type). Griffin et al. (2016) suggested that clinopyroxene and the HP silica
333 polymorph (coesite) develop due to the insoluble nature of Ca and Si in isometric chromite
334 (Figs. 4, 6). The clinopyroxene-chromite association indicates their common paragenesis (Su
335 et al. 2019; Liu et al. 2020). In the Purang lherzolites, Gong et al. (2016) interpreted the
336 clinopyroxene-chromite association to represent symplectites that developed by the
337 breakdown of garnet under the spinel peridotite stability field. Arai (2013) suggested that
338 chromite will revert to the original isometric structure and olivine assemblage during
339 recycling of mantle UHP to low-pressure magmatic conditions. These inclusions may
340 preserve partially obliterated orbicular textures (Fig. 4). The UHP Ca-ferrite type chromite
341 with diamond and other carbon-bearing UHP mineral inclusions can be derived from
342 methane-rich fluid species (Figs. 4 and 6; Yang et al. 2007; Arai 2013). Exsolution lamellae
343 of silica and clinopyroxene may also develop at depths above the mantle transition zone
344 (Yamamoto et al. 2009). The Purang lherzolitite peridotites are interpreted as residues of 10–
345 15% partial melting of a fertile mantle source in a seafloor spreading setting (Xiong et al.
346 2018). Both orthopyroxene and clinopyroxene porphyroclasts in these lherzolites form
347 exsolution lamellae, which can be explained by upwelling of these rocks into shallow mantle
348 depths and associated crystal-plastic deformation (Xiong et al. 2020c). In the Purang
349 chromite, olivine, clinopyroxene and orthopyroxene could have crystallized from anhydrous
350 melts, whereas amphiboles and phlogopite could have formed from water-rich melts (Fig. 7).

351 The conditions estimated for the formation of the Purang lherzolites are 700-1000°C and
352 1.3-2.3 GPa, which correspond to mantle depths of ~100 km (Xiong et al. 2020c). These

353 temperature-pressure estimates are thought to represent the lower limit of conditions under
354 which decomposition of pyroxenes takes place.

355 **Evolution of the Purang ophiolite**

356 Pyroxenes in the Purang lherzolites have compositions typical of pyroxenes in abyssal
357 peridotites (e.g., [Xiong et al. 2013, 2020c](#)). The low Cr# values (~25-30) and TiO₂ contents (≤
358 0.1 wt. %) in disseminated Cr-spinel from the Purang lherzolites suggest 10-15% melting of a
359 N-MORB-type mantle source ([Hellebrand et al. 2001](#)). The clinopyroxene inclusions and
360 blebs in orthopyroxene porphyroclasts in Purang lherzolite suggest direct crystallization from
361 infiltrating/stagnant melts or interaction of metasomatized melts with residual peridotites
362 (e.g., [Suhr et al. 1998; Seyler et al. 2007](#)). For the Purang peridotites, measured ¹⁸⁷Os/¹⁸⁸Os
363 ratios (0.12289 - 0.13194; [Gong et al. 2016](#)) are within the range of those reported for other
364 mantle-hosted podiform chromitites ([Xiong et al. 2018](#)). In addition, the generally low γOs
365 values (< 1) of chromitites indicate that their parental melts were sourced from a
366 heterogeneously depleted mantle region above a downgoing lithospheric slab (e.g., [Xiong et](#)
367 [al. 2018](#)).

368 Water in the podiform chromitites hydrates surrounding silicate minerals in the chromitite
369 and/or penetrates the surrounding wall rocks ([Matveev and Ballhaus 2002; Boudreau et al.](#)
370 [2010; Su et al. 2020](#)). Uvarovite ([Fig. 11](#)) could have crystallized from a metasomatized
371 MORB-like melt that likely infiltrated deformed chromite grains (e.g., [Su et al. 2019](#)).

372 Partial melting occurred at the mid-ocean ridge stage and generated oceanic mafic crust
373 with low δ²⁶Mg values, leaving refractory peridotites with isotopically heavy Mg behind ([Su](#)

374 [et al. 2015, 2021a](#)). The variations of Mg isotopic compositions of the Purang peridotites can
375 be interpreted as a feature of multi-stage melting and metasomatism of the oceanic
376 lithospheric mantle (**Fig. 13: [Su et al. 2015](#)**). Dolerite and gabbro complexes from the Purang
377 ophiolite gave zircon U–Pb ages ranging from 144.2 ± 2.1 Ma to 126.5 ± 0.4 Ma and generally
378 positive zircon $\epsilon_{\text{Hf}(t)}$ values, suggesting a heterogenous juvenile mantle source ([Liu et al. 2011](#);
379 [Xiong et al. 2020b](#)).

380 Oceanic basalts and peridotites commonly show identical $\delta^{26}\text{Mg}$ values since
381 high-temperature mantle processes do not efficiently fractionate Mg isotopes (e.g., [Bourdon et](#)
382 [al. 2010](#); [Pogge von Strandmann et al. 2011](#)). In the Purang ophiolite, the average $\delta^{26}\text{Mg}$ of
383 peridotites (-0.20 ± 0.10 ‰) is slightly higher than both chondrites and oceanic basalts values
384 ([Su et al. 2019](#)). Also, the $\delta^{26}\text{Mg}$ values of the Purang ophiolitic rocks do not correlate with
385 common indices for partial melting and metasomatism (**Supplementary Material 7; [Su et al.](#)**
386 **2019**). In the Luobusa and Zedong harzburgites, the similar $\delta^{26}\text{Mg}$ values (-0.41 to -0.20 ‰)
387 for olivine, orthopyroxene and chromite concentrates have been interpreted to reflect an
388 equilibrium isotope fractionation ([Xiao et al. 2016](#)). In the Purang ophiolite, the high-Cr and
389 high-Al chromitites have Mg isotopic compositions slightly lighter than those of the
390 associated peridotites (**Fig. 13; Supplementary Material 7**). In peridotite xenoliths, olivine
391 with negative Mg–Fe isotopic values has likely experienced Fe–Mg inter-diffusion with
392 metasomatic melts in sub-arc mantle environments ([Huang et al. 2011](#); [Sio et al. 2013](#); [Zhou](#)
393 [et al. 2014](#)).

394

395

IMPLICATIONS

396 The chemical characteristics of the Purang ophiolitic chromitites, including trace element
397 patterns and Mg isotope composition, are comparable to chromitites formed in a fore-arc SSZ
398 setting. Formation of the Purang ophiolites in a SSZ setting is corroborated by the occurrence
399 of basaltic component with mixed enriched-MORB and OIB geochemical affinities.
400 Inclusions of uvarovite, apatite, lamellar clinopyroxene and wüstite in the Purang chromitite
401 may reflect fluid percolation in the lithospheric mantle. Considering the abundant UHP and
402 SuR mineral inclusions in the investigated chromitites, the Purang ophiolite likely
403 experienced a complex petrogenetic history encompassing deep mantle processes. Mantle
404 convection may have transported chromitites and the associated peridotites to shallow mantle
405 levels, leading to partial melting of peridotites, recrystallization of chromitites and
406 encapsulation of crustal and exotic silicates, sulfides, and metal alloys as inclusions.

407 This study reports Mg isotope compositions of Purang ophiolite comparable to other
408 Tethyan ophiolites, implying a common geodynamic evolution. As a part of the Neo-Tethyan
409 ophiolites, the Purang podiform chromitites record superimposed deep mantle and
410 subduction-related processes. Partial hydrous melting of the mantle wedge could have
411 produced island arc tholeiitic and boninitic magmas that reacted with peridotites and endowed
412 parts of this mantle section with chromite ores.

413

414

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665 **Figure captions:**

666 **Figure 1. (a)** Simplified tectonic map of the Tibetan Plateau, with several terranes separated
667 by well-defined sutures. The (Indus-)Yarlung-Zangbo Suture Zone [(I)YZSZ], the
668 southernmost one of these sutures, separates the Lhasa block to the north from the
669 Tibetan-Himalayan orogenic belt to the south. BNS = Bangong-Nujiang Suture; IYZS =
670 Indus-Yarlung-Zangbo Suture. **(b)** Geological map of the Purang ophiolite massif
671 (modified from [Yang et al. 2011](#); [Xiong et al. 2018, 2020b, c](#)). Gab = gabbro, H-dun =
672 Cr₂O₃-rich dunite, L-Dun = Cr₂O₃-poor dunite, Harz = harzburgite, Lhz = lherzolite.

673 **Figure 2.** Field photographs showing the occurrence of chromitite in the Purang massif. **(a)**
674 High-Cr massive chromitite. **(b)** Dunite vein with high-Al massive chromitite laminae.
675 **(c)** Serpentinite with high-Cr massive chromitite. **(d)** High-Al nodular chromitite in a
676 dunitic host rock.

677 **Figure 3. (a)** Compositional variations of chromite grains from the Purang peridotites
678 revealed by the Cr# vs. TiO₂ (wt.%) plot (after [Pearce et al. 2000](#)). **(b)** Cr# vs. Mg#
679 diagram with data for chromite grains from peridotite samples of the Purang ophiolite.
680 Fields for abyssal and forearc peridotites are from [Dick and Bullen \(1984\)](#), [Ishii et al.](#)
681 [\(1992\)](#), [Arai \(1997\)](#), and [Hébert et al. \(2003\)](#). The degree of melt extraction is calculated
682 from the spinel composition using the empirical formula of [Hellebrand et al. \(2001\)](#).
683 Data of spinel grains from the Purang lherzolites are from [Xiong et al. \(2019\)](#).

684 **Figure 4. (a-d)** BSE images of acicular silicate micro-inclusions in chromite grains from the
685 Purang chromitites. **(d)** Plane-polarized light photomicrographs of acicular (Cpx)
686 inclusions in chromite grains from the high-Al chromitites.

687 **Figure 5.** Backscattered (BSE) images of various pyroxene inclusions in chromite grains from
688 the Purang ophiolite. **(a-c)** Octahedral orthopyroxene inclusions. **(d-f)** Subhedral and
689 euhedral clinopyroxene inclusions. **(g-i)** Composite euhedral clinopyroxene and
690 orthopyroxene inclusions in chromite. Cpx = clinopyroxene; Opx = orthopyroxene; Chr
691 = chromite.

692 **Figure 6. (a)** TEM image of a FIB foil showing a single clinopyroxene inclusion in a
693 chromite grain. **(b)** A high-resolution TEM image showing sharp contacts of an acicular
694 inclusion in a high-Al chromite grain. **(c)** EDX spectrum of a Cr-spinel grain. **(d)** EDX
695 spectrum of the lamellar clinopyroxene inclusion. **(e)** EDX spectrum of a high-Cr
696 chromite grain. Note the Ga peaks in the spectrum are due to Ga-ion implantation during
697 the FIB milling and the Cu peaks are due to secondary fluorescence from the copper grid
698 that supports the foil. Cpx = clinopyroxene, Chr = chromite.

699 **Figure 7.** BSE images of mineral inclusions in chromite grains from the Purang chromitites.
700 **(a)** an acicular inclusion about 50 μm long and $<1 \mu\text{m}$, which is interpreted as an
701 Fe-Si-O-bearing fluid inclusion (see Fig. 9). **(b)** Apatite inclusion along cracks between
702 chromite grains. **(c)** A uvarovite inclusion in chromite. **(d)** A composite
703 serpentine-amphibole-phlogopite-millerite inclusion. **(e)** A composite amphibole-apatite
704 inclusion. **(f)** A composite amphibole-apatite-pentlandite inclusion in chromite. **(g)** An

705 orthopyroxene-amphibole composite inclusion. **(h)** A clinopyroxene-amphibole
706 inclusion in the chromitite. **(i)** A uvarovite inclusion in chromite. Cpx = clinopyroxene;
707 Ap = apatite; Amp = amphibole; Pn = pentlandite; Pg = phlogopite; Mil = millerite; Uva
708 = uvarovite; Chr = chromite.

709 **Figure 8.** BSE images of metal alloys and sulfide inclusions in the Purang chromitites. Mil=
710 millerite; Amp = amphibole; Cc = chalcocite; Gn = galena; Chr= chromite.

711 **Figure 9 (a)** FIB foil of the acicular inclusion shown in Fig. 7a with a vertical feature near the
712 center of the foil. **(b)** TEM image of an acicular clinopyroxene inclusion surrounded by
713 open spaces in a chromite grain. **(c)** EDX spectrum of the same inclusion showing peaks
714 of Fe, Si and O as well as Mg, Al and Cr. The black domains may have been filled with
715 fluids, **(d)** & **(e)** X-ray elemental maps along the orange line across the inclusion,
716 showing major Fe, Si and O peaks. The Cr, Al, and Mg are concentrated in the host
717 chrome spinel.

718 **Figure 10. (a)** TEM image of a uvarovite inclusion in the chromite grain shown in Figure
719 10b. **(b)** EDX spectrum of the uvarovite inclusion showing peaks for Si, O, and Ca and
720 minor peaks for Cr and Fe. The indexed diffraction pattern (Fast Fourier Transform,
721 FFT) showing the upper part of the TEM confirming the isometric structure. **(c)** EDX
722 spectrum of the chromite grain with major Al and Cr peaks. The FFT pattern also shows
723 the isometric structure of the same chromite grain.

724 **Figure 11.** Binary plots of Cr# vs trace element contents (ppm) in chromite grains from the
725 Purang high-Cr and high-Al chromitites.

726 **Figure 12.** MORB-normalized trace element patterns of the Purang chromitites compared to
727 patterns of boninite (BON) and Thetford Mines ophiolite (TMO) chromitites. Reference
728 values are from [Barnes et al. \(1985\)](#) and [Pagé et al. \(2008\)](#).

729 **Figure 13.** Histograms of the Mg isotope composition of peridotites from the Purang
730 ophiolites and from other occurrences. **(a)** $\delta^{26}\text{Mg}$ signatures of MORB and OIB lavas.
731 The dashed line represents the average $\delta^{26}\text{Mg}$ values of chondrites (-0.28% , [Teng et al.](#)
732 [2010a](#)). **(b)** $\delta^{26}\text{Mg}$ values of ophiolitic chromitite, mafic and ultramafic rocks and
733 peridotite xenoliths. **(c)** $\delta^{26}\text{Mg}$ compositions of the different lithological units of the
734 Purang ophiolite.

735 Magnesium isotope data sources: MORBs, [Bourdon et al. \(2010\)](#), [Teng et al. \(2010b\)](#);
736 OIBs, [Teng et al. \(2007, 2010a\)](#), [Bourdon et al. \(2010\)](#); peridotite xenoliths, [Yang et al.](#)
737 [\(2009\)](#), [Young et al. \(2009\)](#), [Bourdon et al. \(2010\)](#), [Teng et al. \(2010b\)](#), [Huang et al.](#)
738 [\(2011\)](#), [Pogge von Strandmann et al. \(2011\)](#); mafic and ultramafic ophiolites, [Su et al.](#)
739 [\(2015\)](#).

740

741 **Table captions:**

742 **Table 1.** Representative EPMA data of chromite grains from the Purang chromitites (wt.%).

743 **Table 2.** Representative EPMA data of silicate inclusions in chromite grains from the Purang
744 chromitites (wt.%).

745 **Table 3.** Trace element composition, by LA-ICP-MS, of chromite grains from the Purang
746 chromitites (ppm).

747 **Table 4.** Mg isotopic compositions of selected samples from the Purang chromitites and
748 associated peridotites.

749

750 **Supplementary Material 1.** Lithological subdivision of chromitite-dunite associations in the
751 Purang ophiolite (South Tibet)

752 **Supplementary Material 2.** (a-h) X-ray elemental maps of the acicular inclusions in
753 chromite grains from the Purang chromitites.

754 **Supplementary Material 3.** Raman spectra for an acicular inclusion (same as in Fig. 4), with
755 clinopyroxene and chromite peaks.

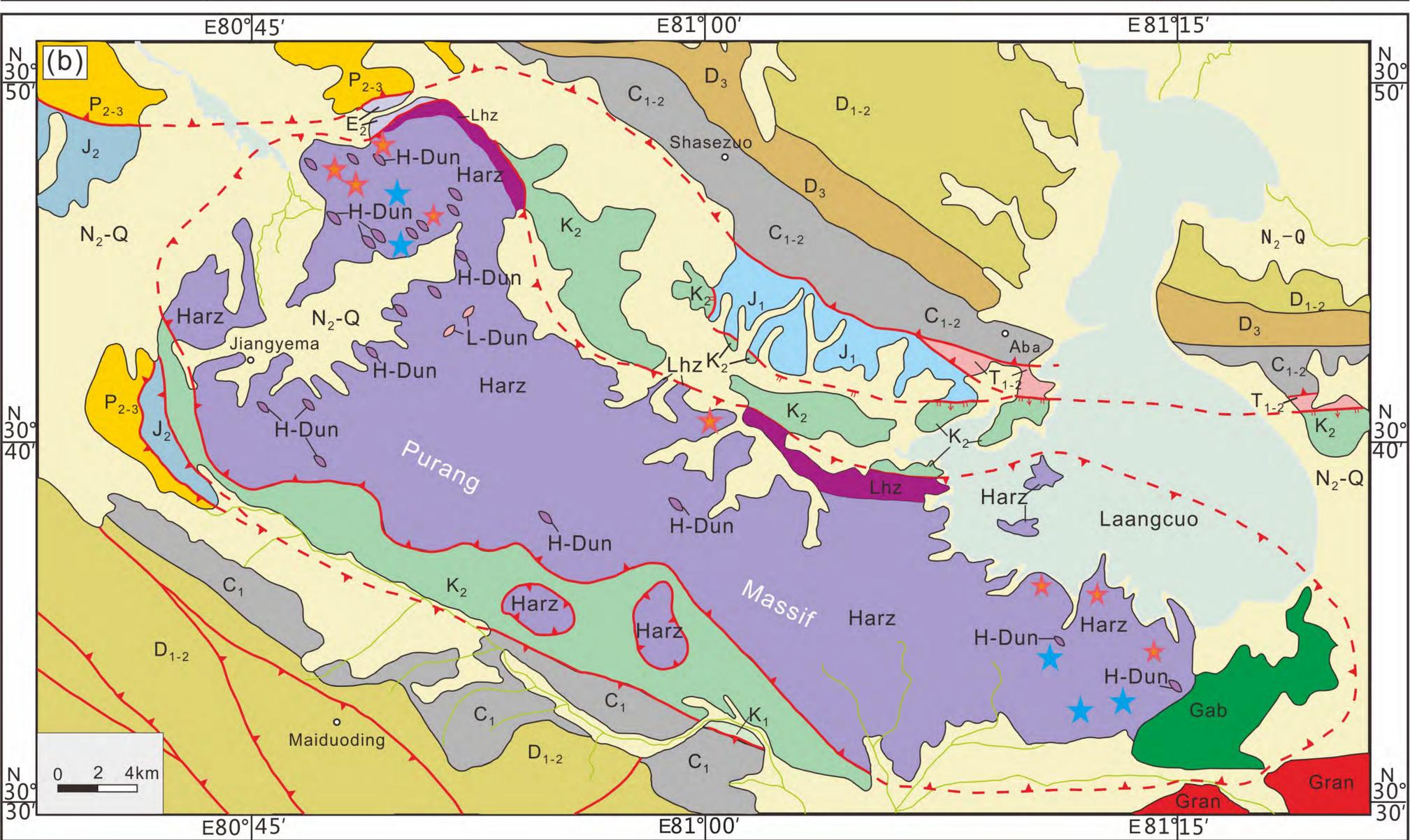
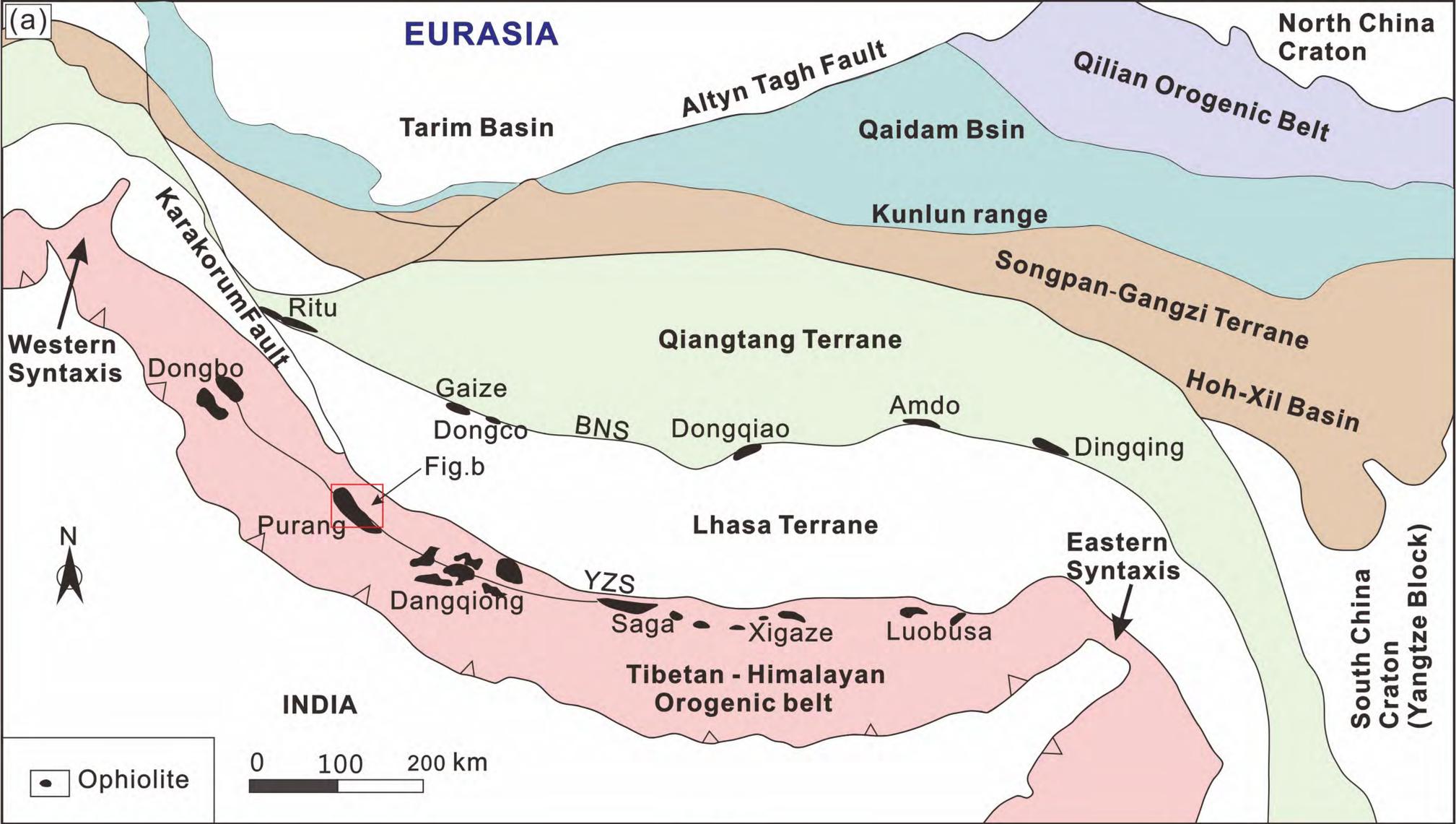
756 **Supplementary Material 4.** Back-scattered-electron (BSE) images of amphibole, phlogopite
757 inclusions from Purang chromitite.

758 **Supplementary Material 5.** X-ray elemental maps of a Ru-Os-Ir-Fe-Ni inclusion in the a
759 chromite grain from the Purang chromitite.

760 **Supplementary Material 6.** BSE images and X-ray elemental maps of a wüstite inclusion in
761 chromite with a major FeO component.

762 **Supplementary Material 7.** Mg isotopic signatures of olivine, orthopyroxene, and chromite
763 concentrates as well as whole-rock data for the associated lithologies ([Xiao et al. 2016](#)
764 and references therein).

765



PURANG MASSIF
(140-120 Ma)

- Gab** Gabbro
- H-Dun** Cr₂O₃-rich Dunite
- L-Dun** Cr₂O₃-poor Dunite
- Harz** Harzburgite
- Lhz** Lherzolite
- K₂** Ophiolitic mélange (Cretaceous)

TETHYAN HIMALAYA OF SOUTHERN TIBET
(Paleozoic-Cretaceous)

- N_{2-Q}** Neogene-Quaternary
- Gran** Leucogranite (28-8 Ma)
- K₁** Cretaceous mélange
- J₂** Upper Jurassic siltstone
- J₁** Early Jurassic mélange
- T₁** Lower Triassic
- P₂₋₃** Upper Permian
- C₁₋₂** Lower Carboniferous
- D₃** Upper Devonian
- D₁₋₂** Lower Devonian

SYMBOLS

- ★** High-Cr chromitite
- ★** High-Al chromitite
- ↔** Thrust fault contact
- ↔** Undifferentiated fault
- River

Figure 1

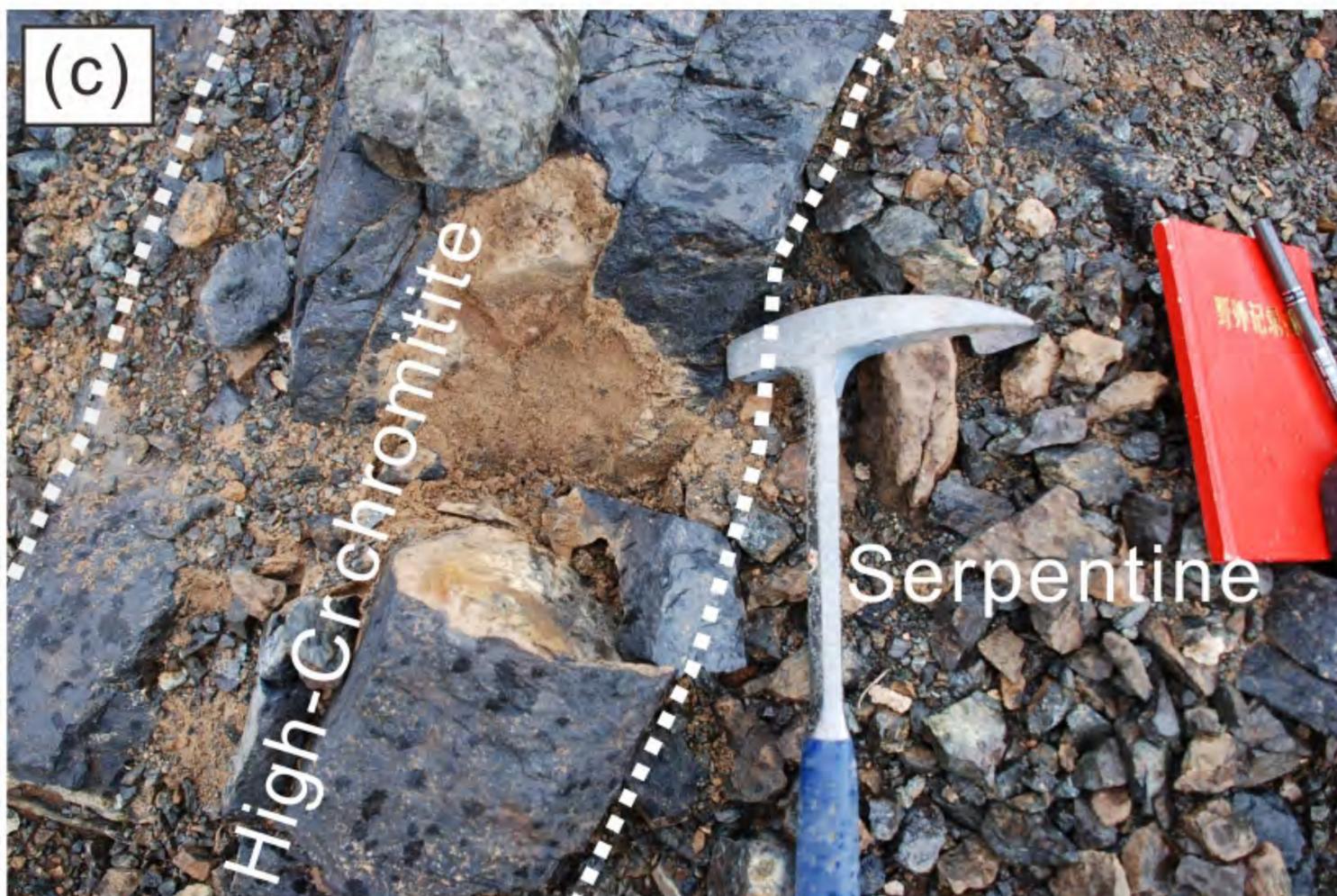


Figure 2

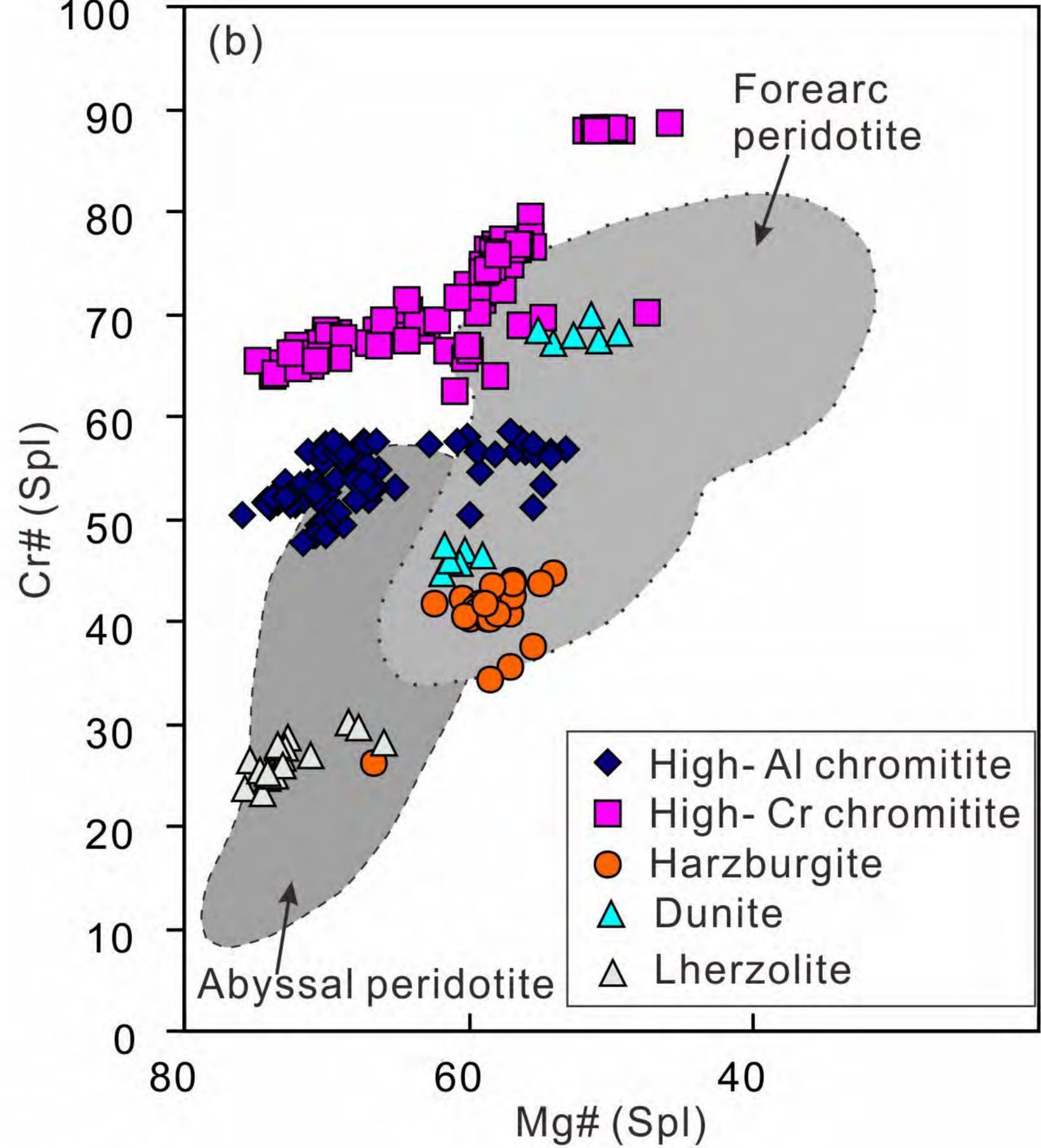
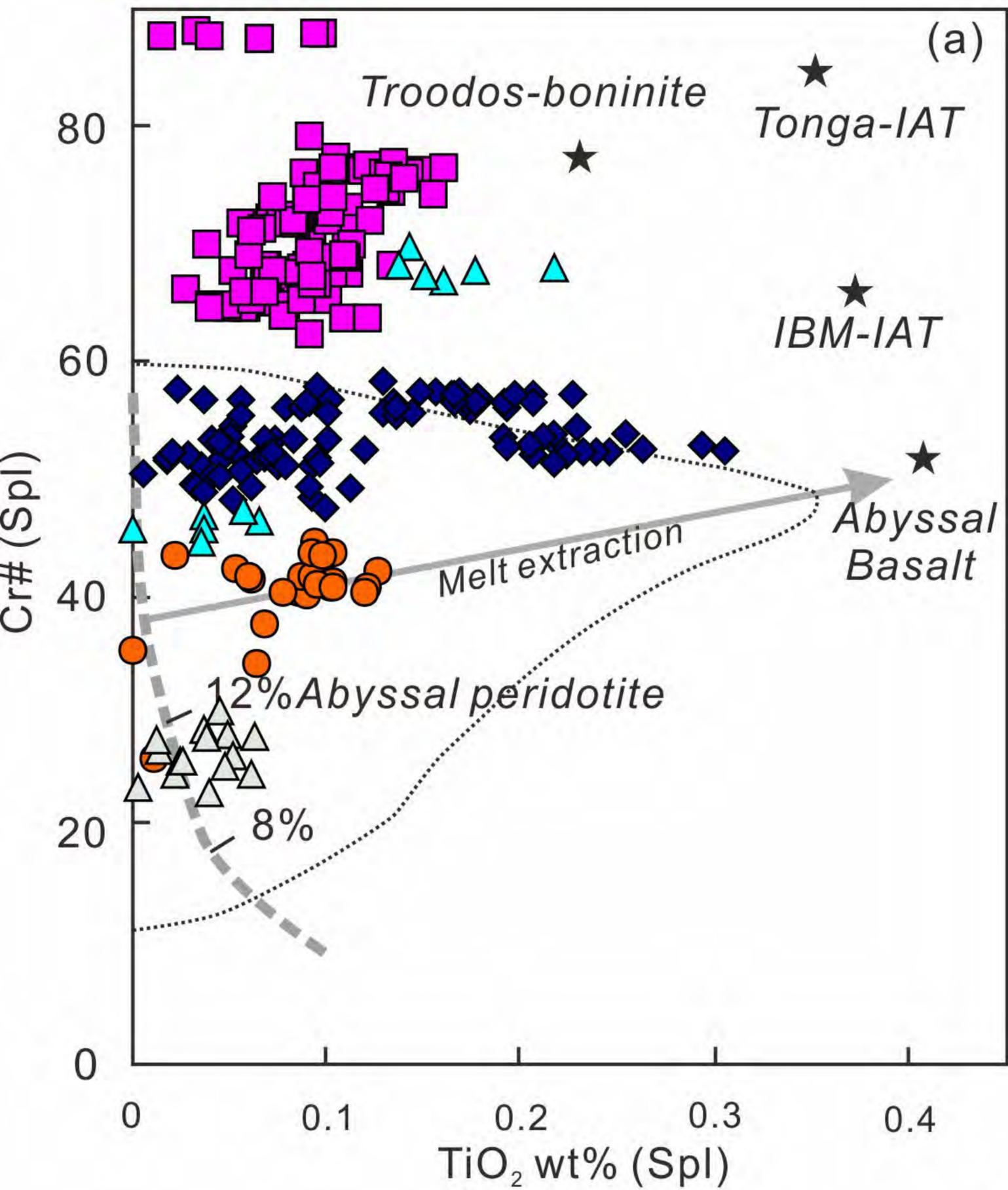


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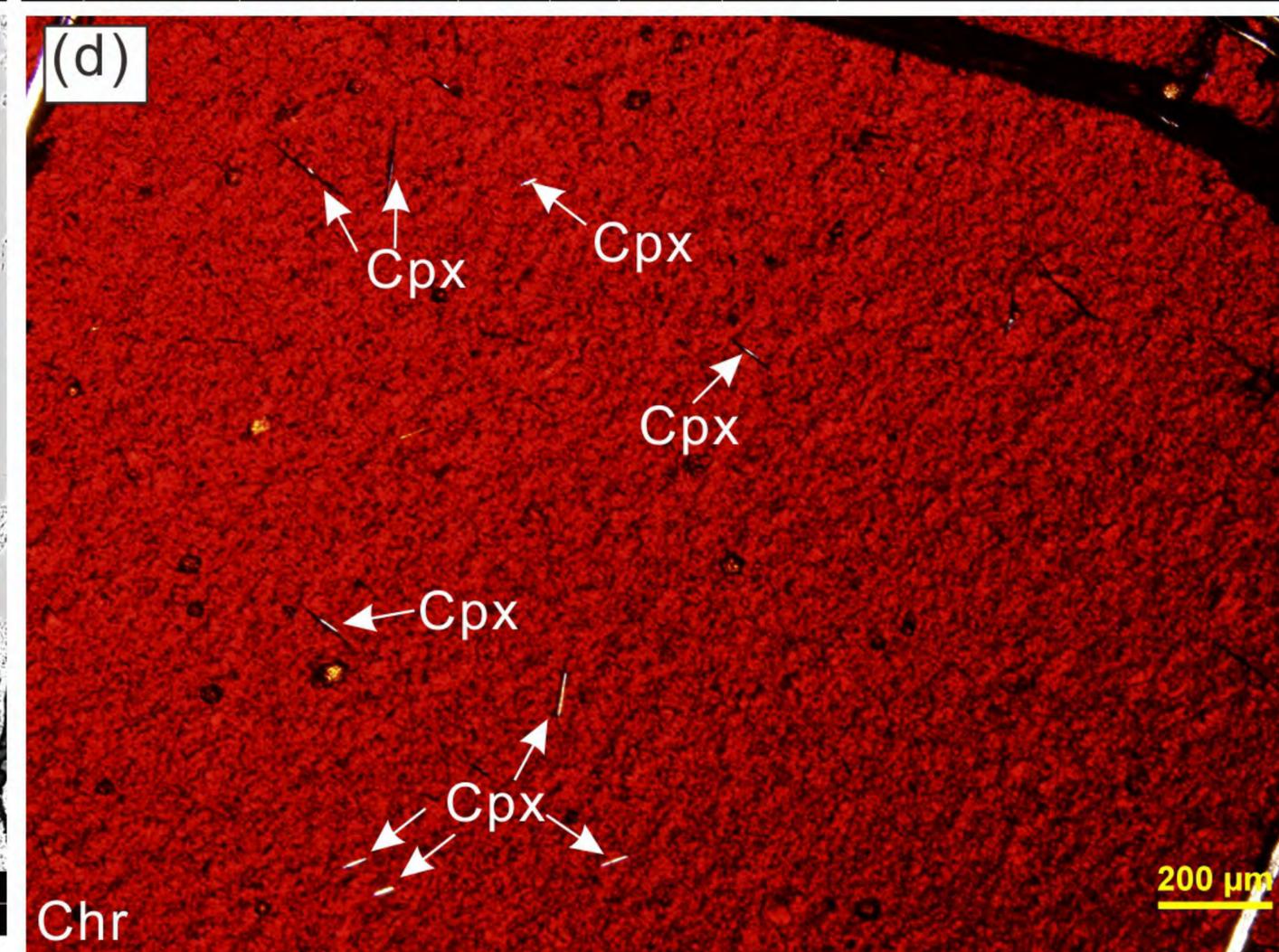
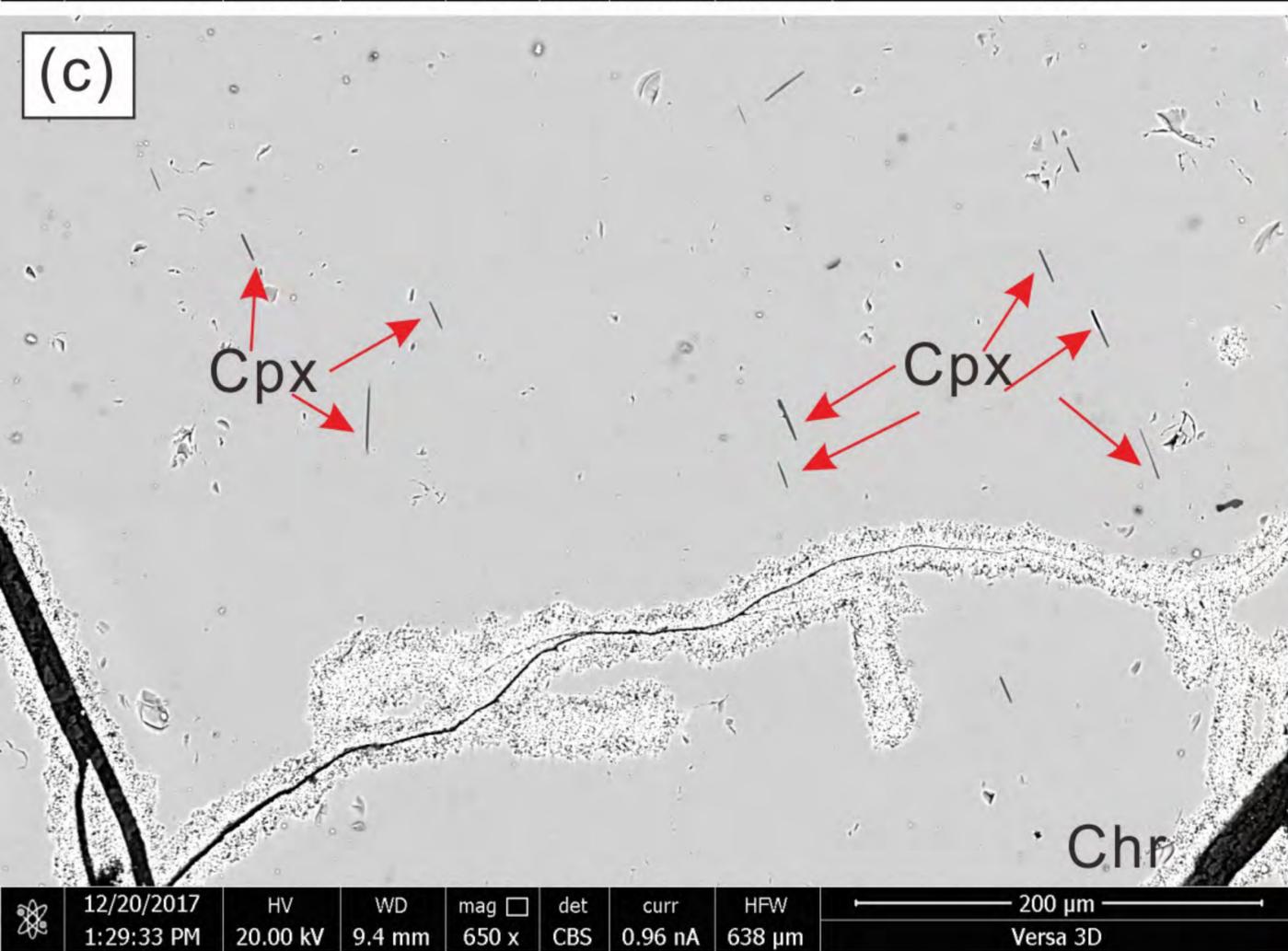
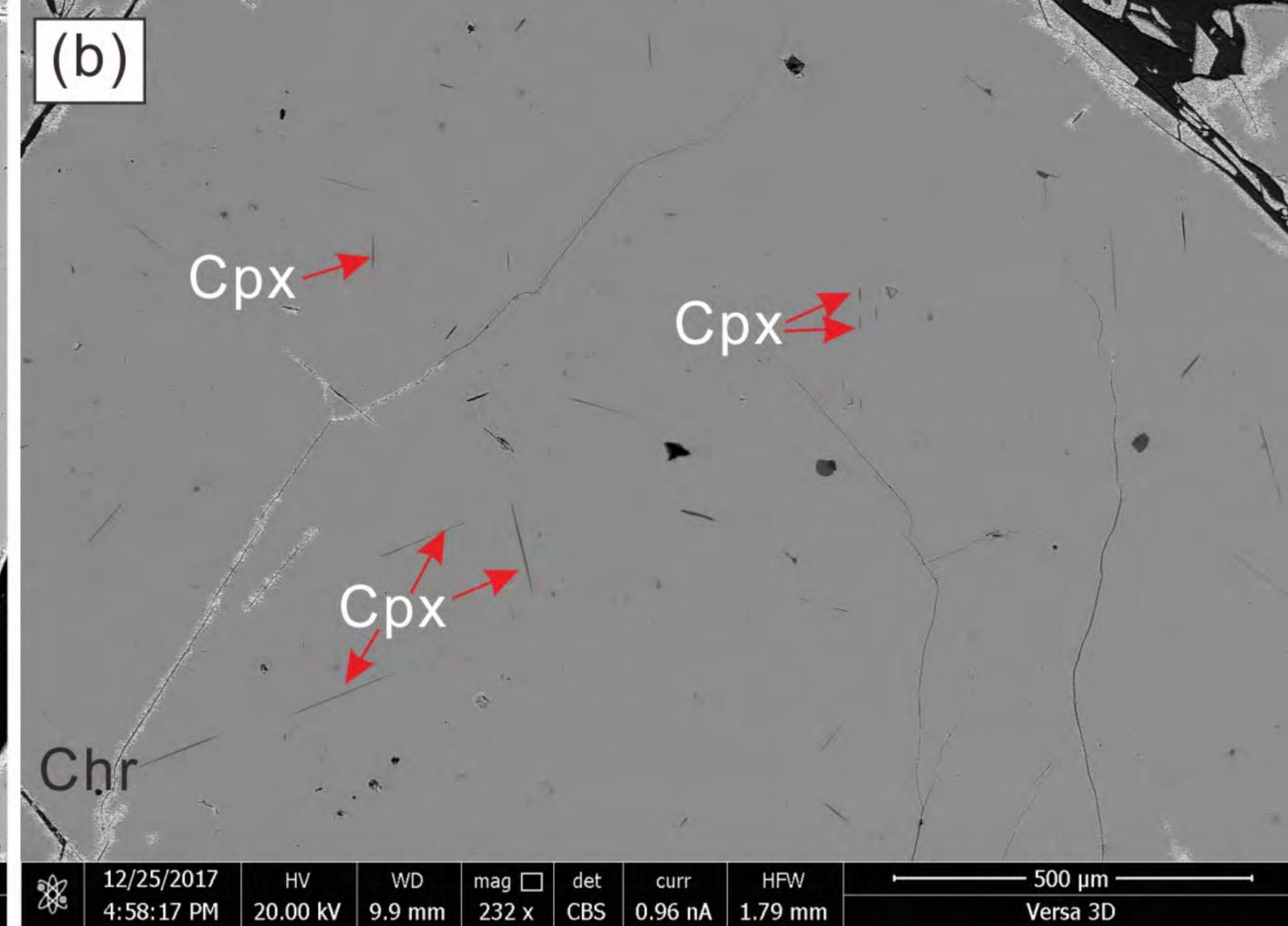
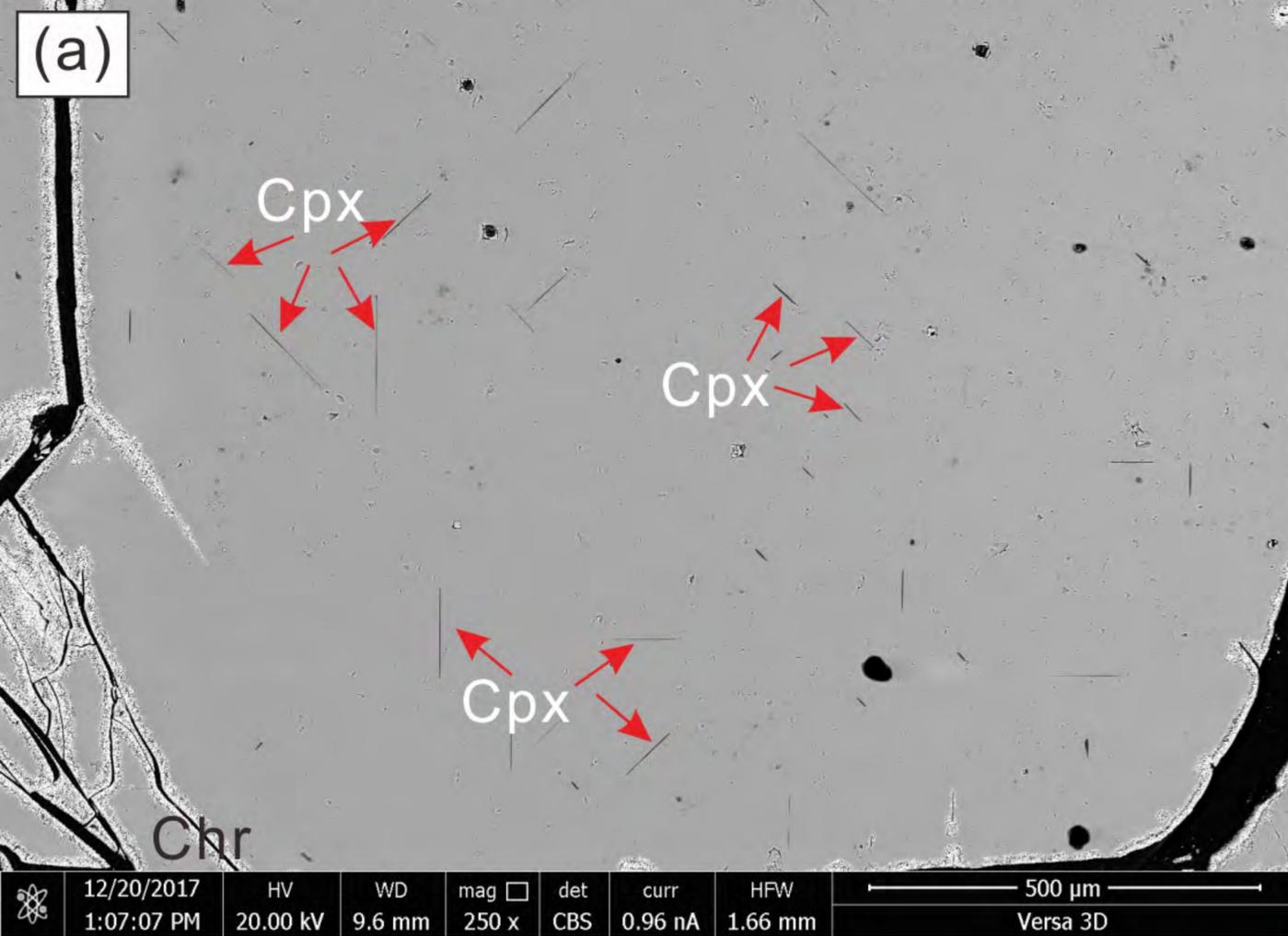


Figure 4

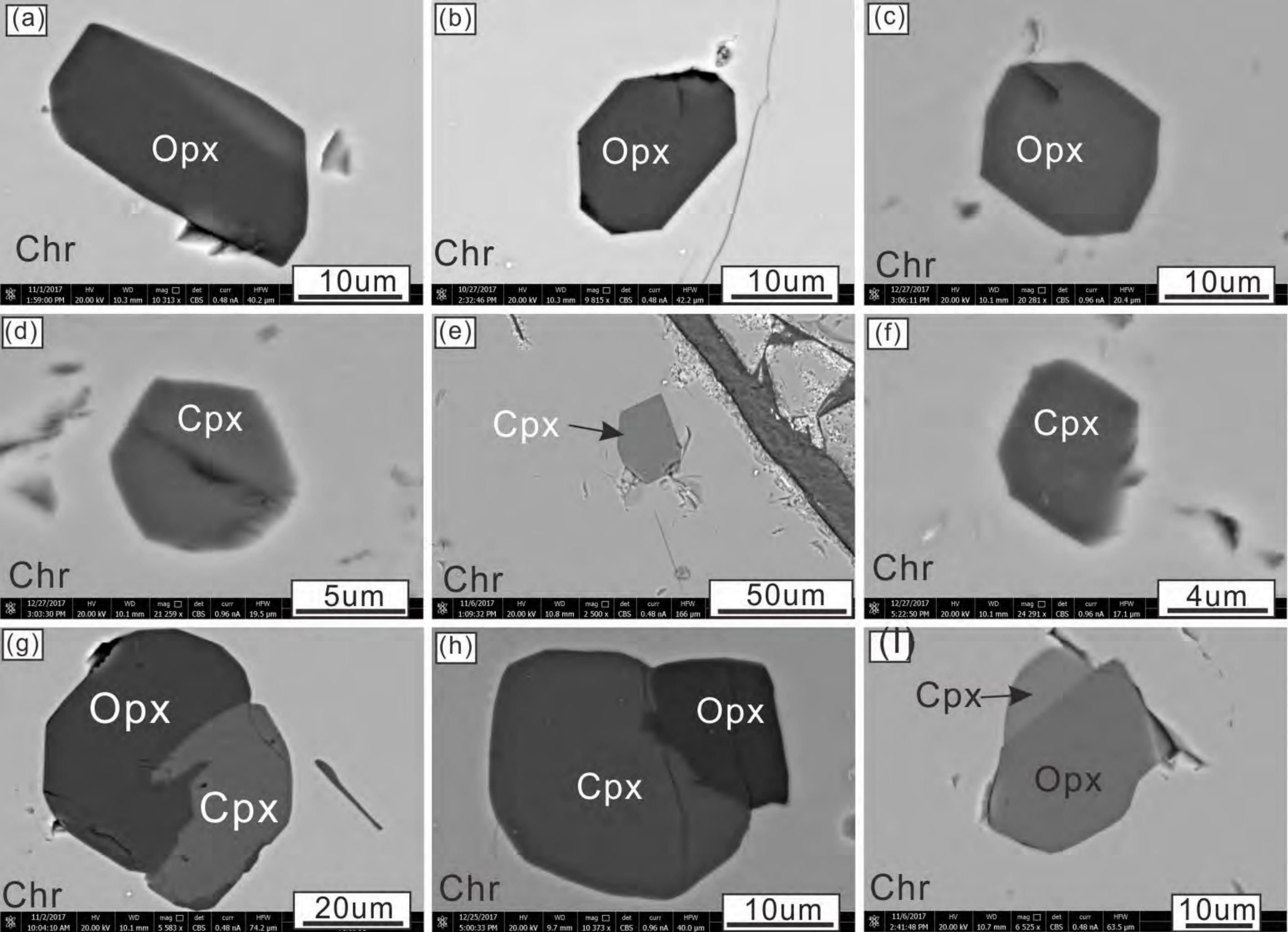


Figure 5

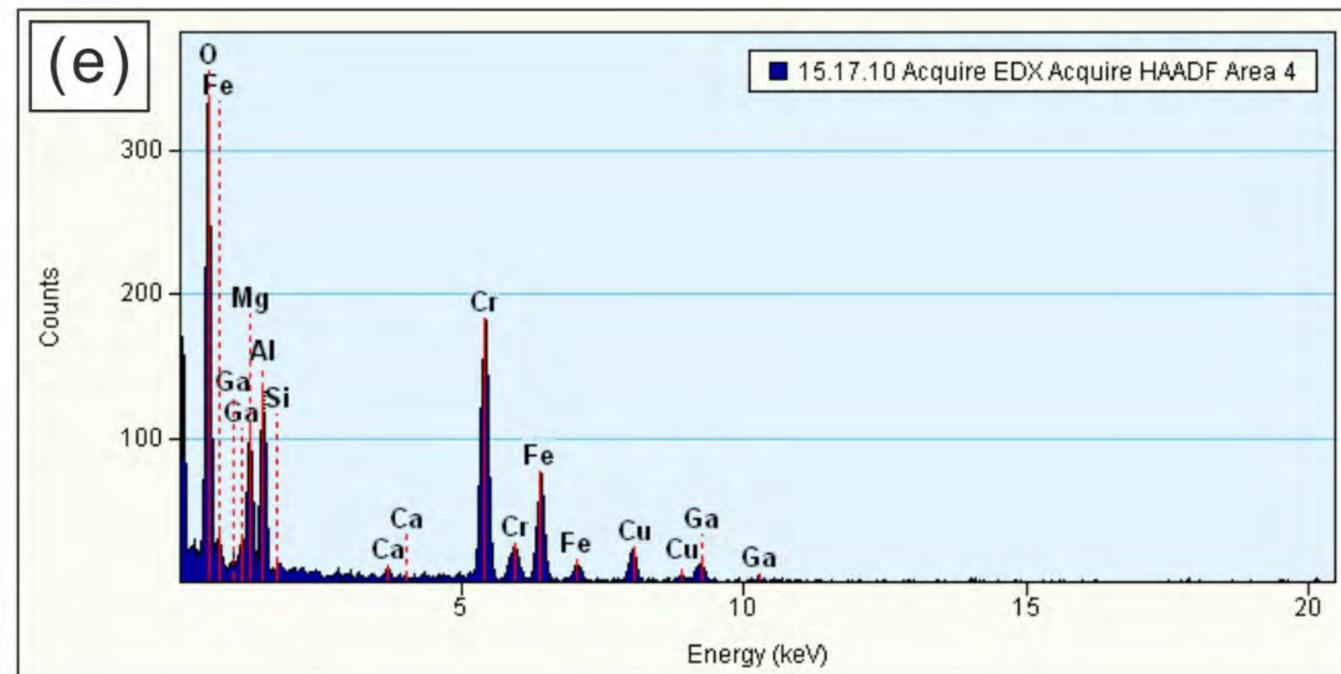
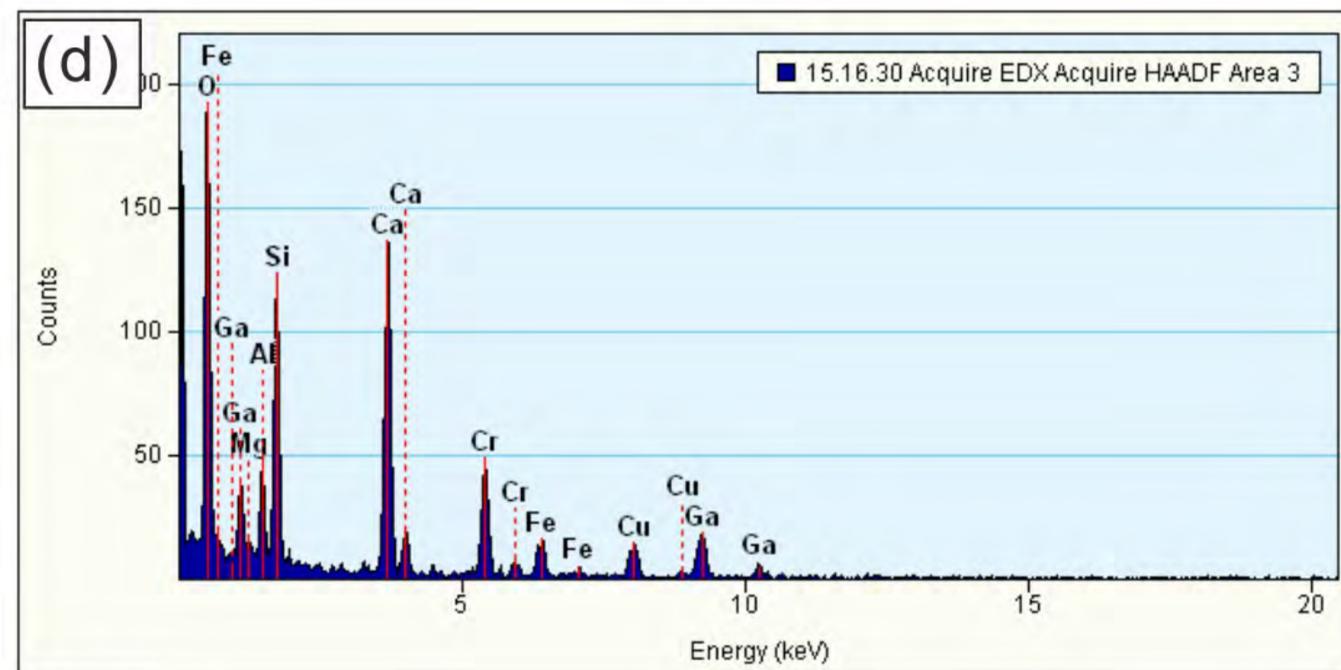
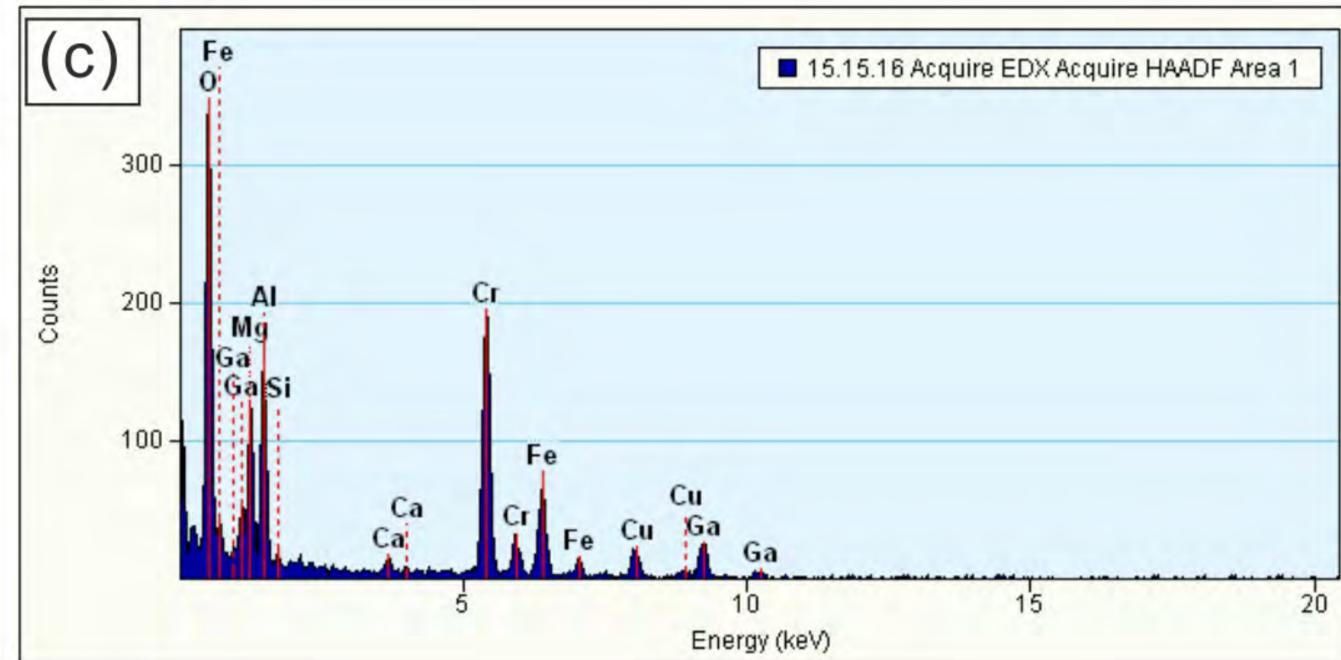
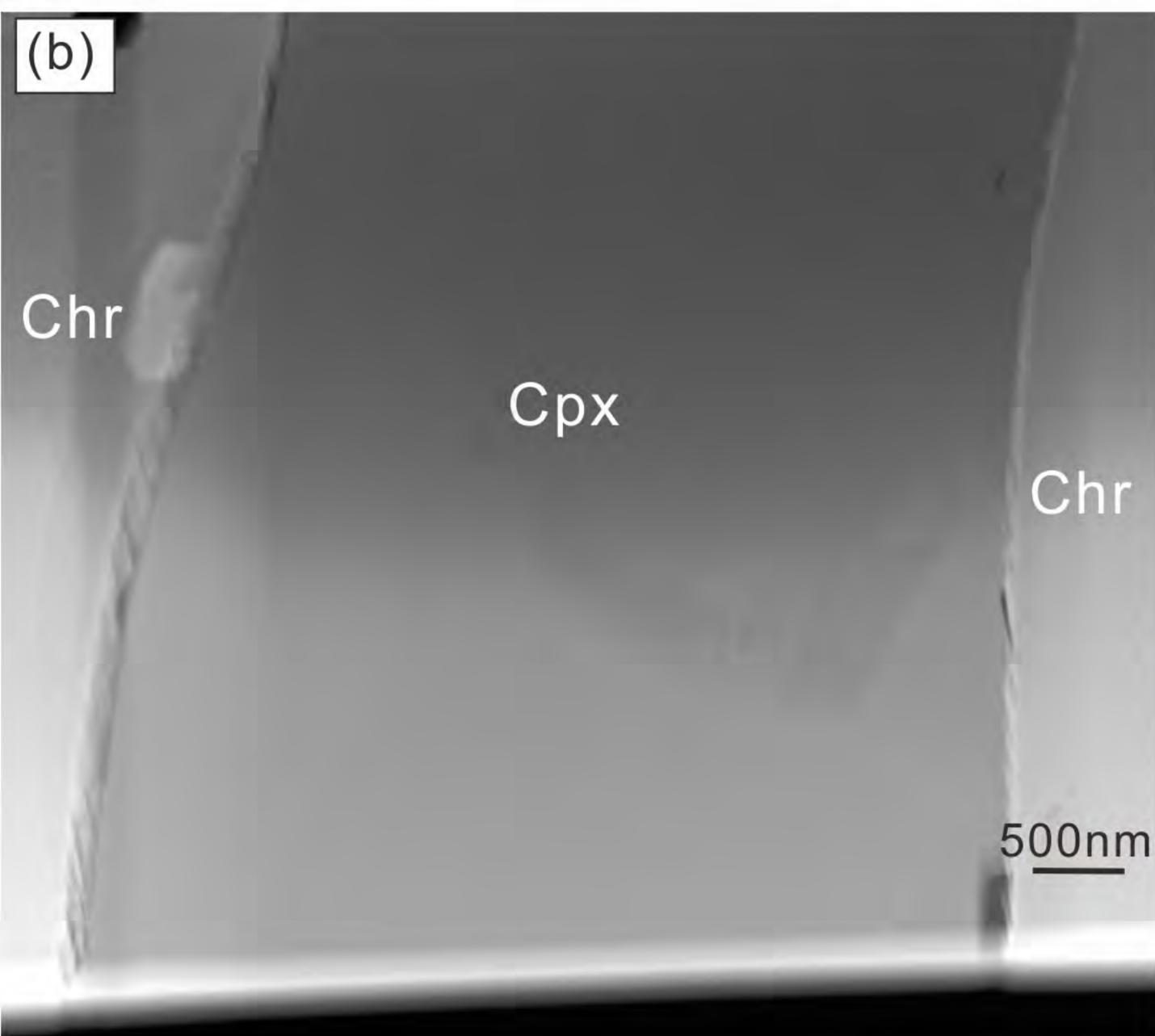
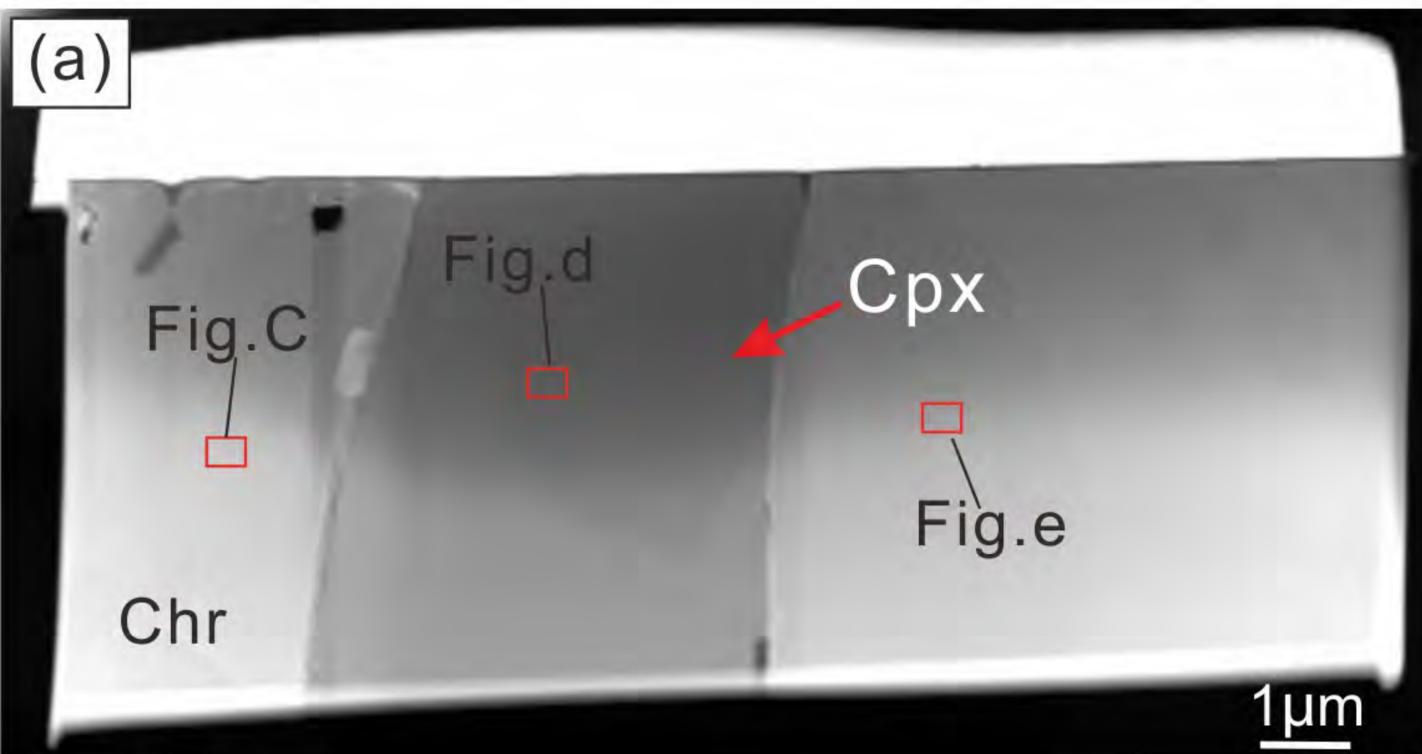


Figure 6

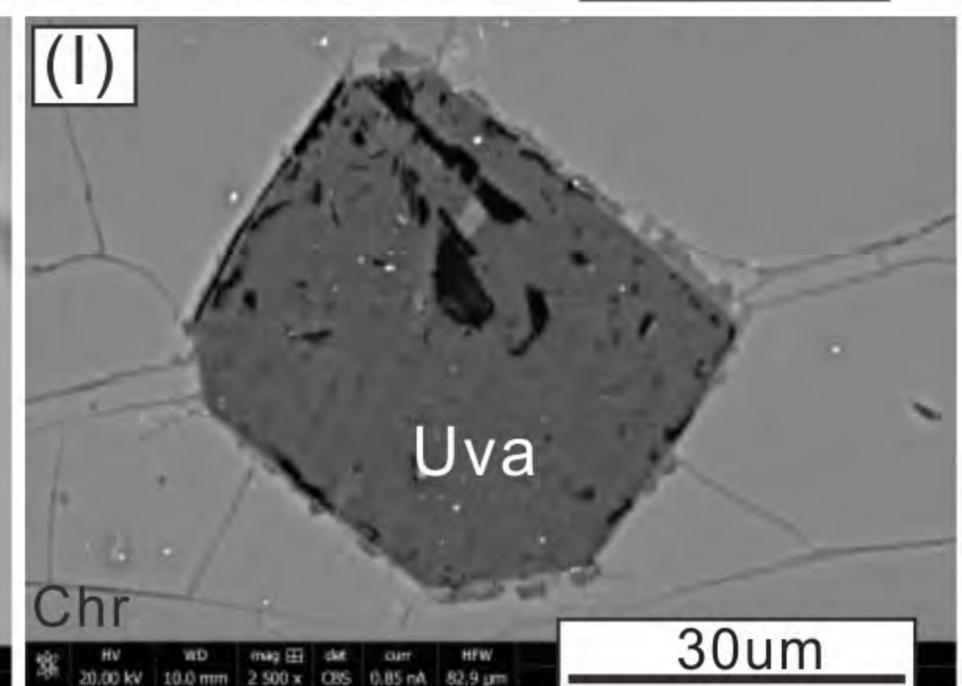
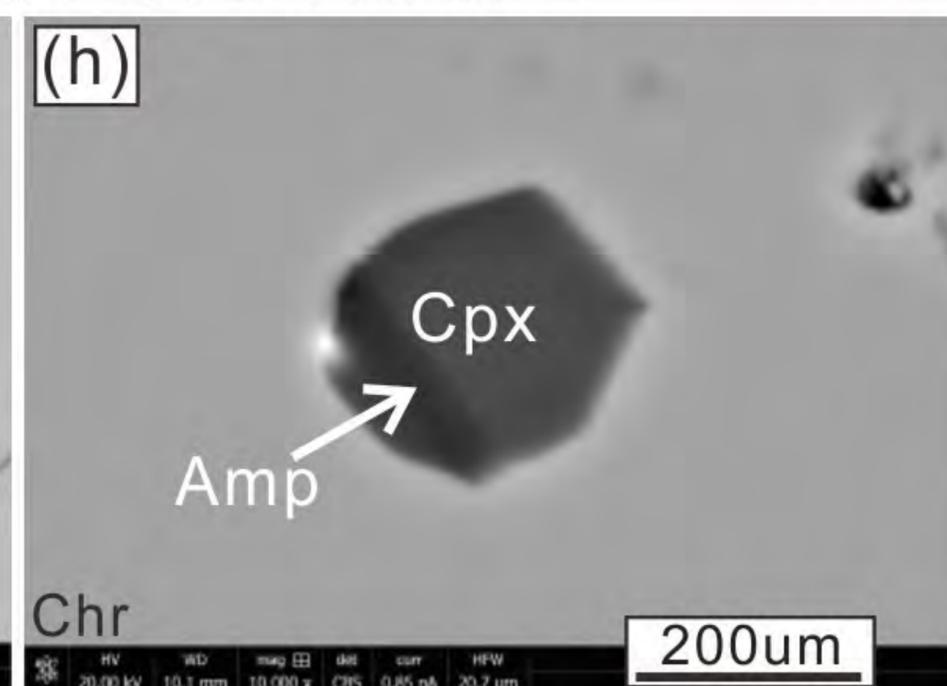
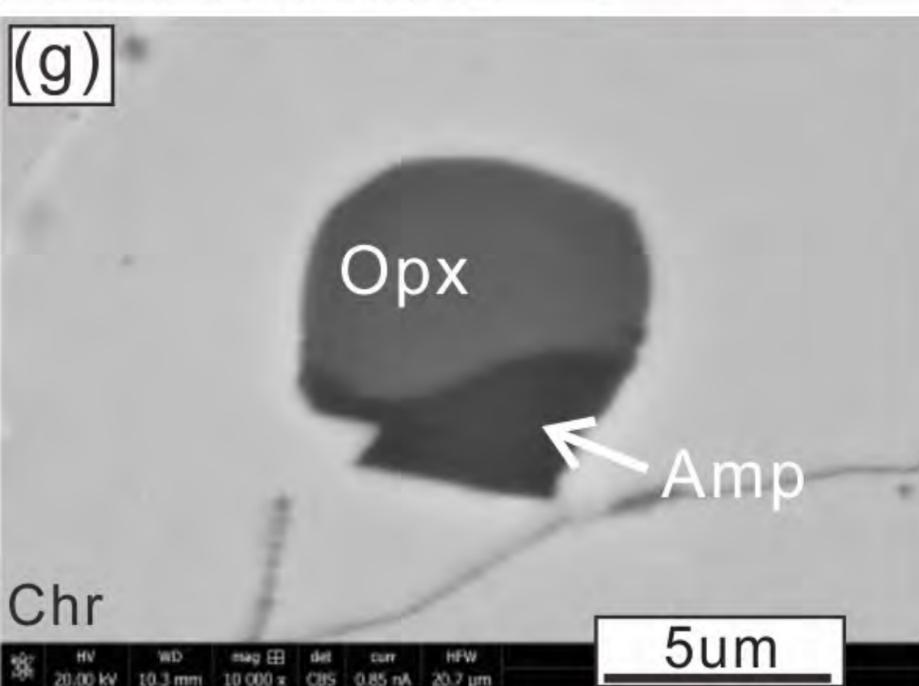
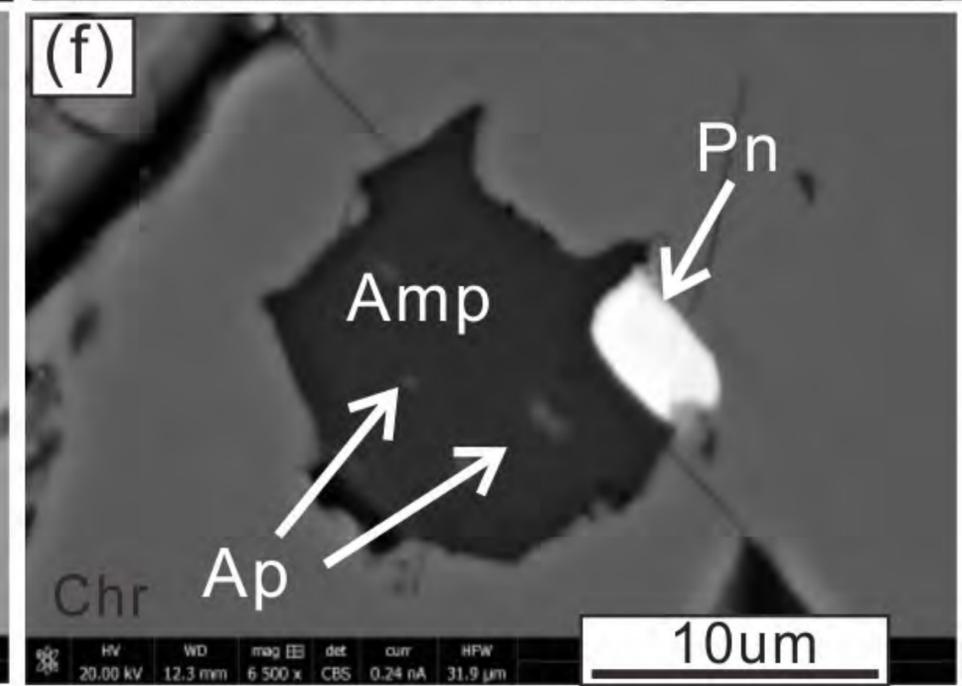
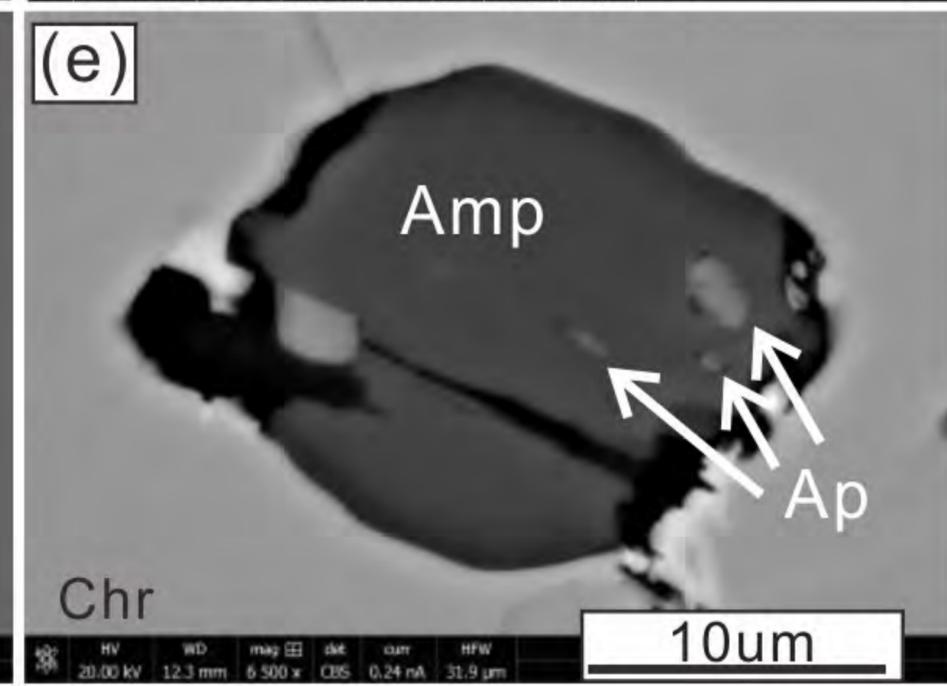
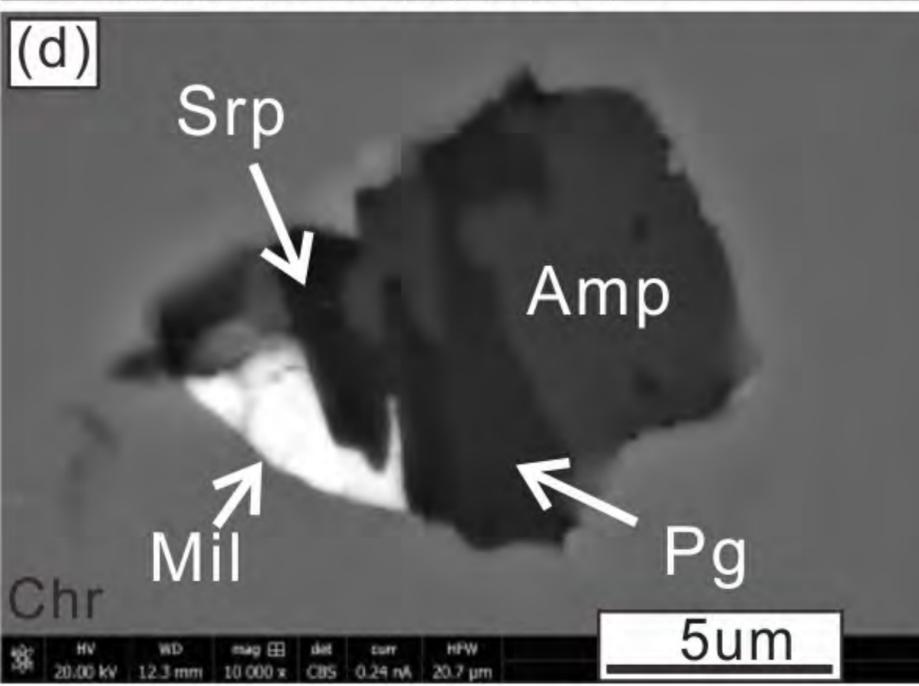
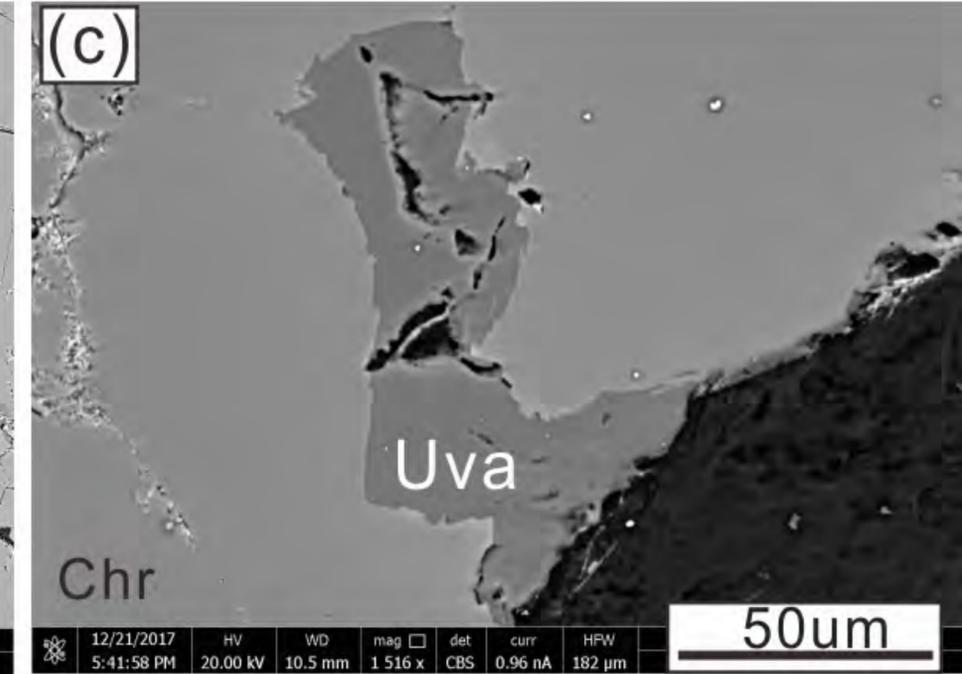
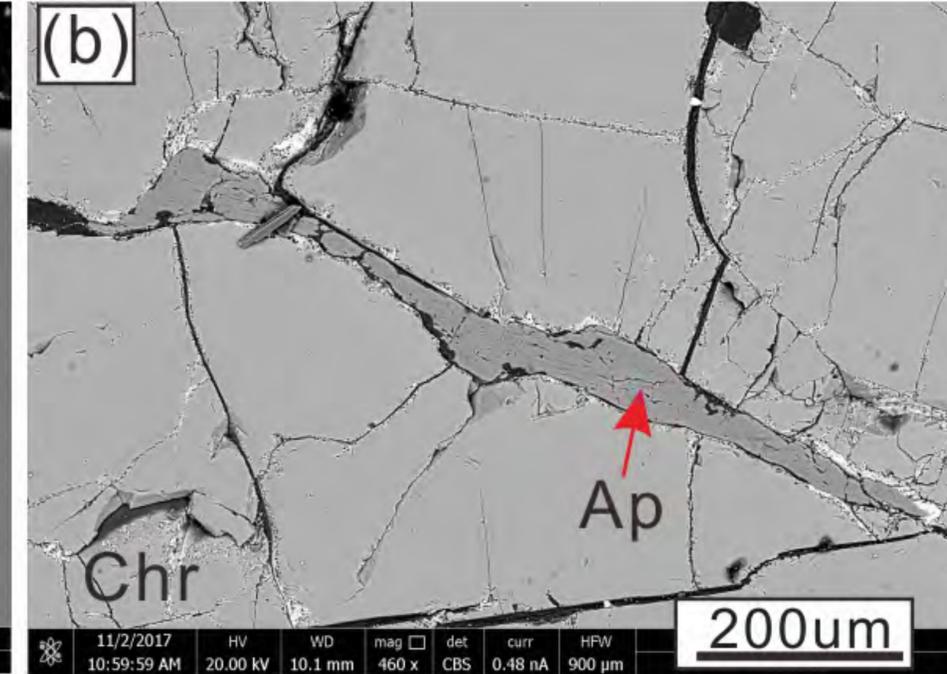
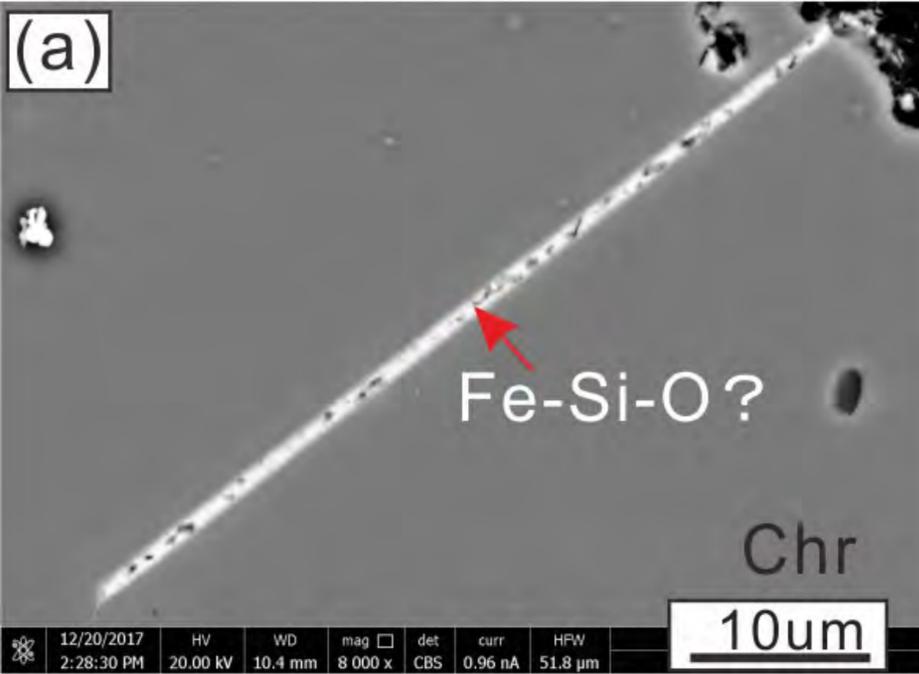


Figure 7

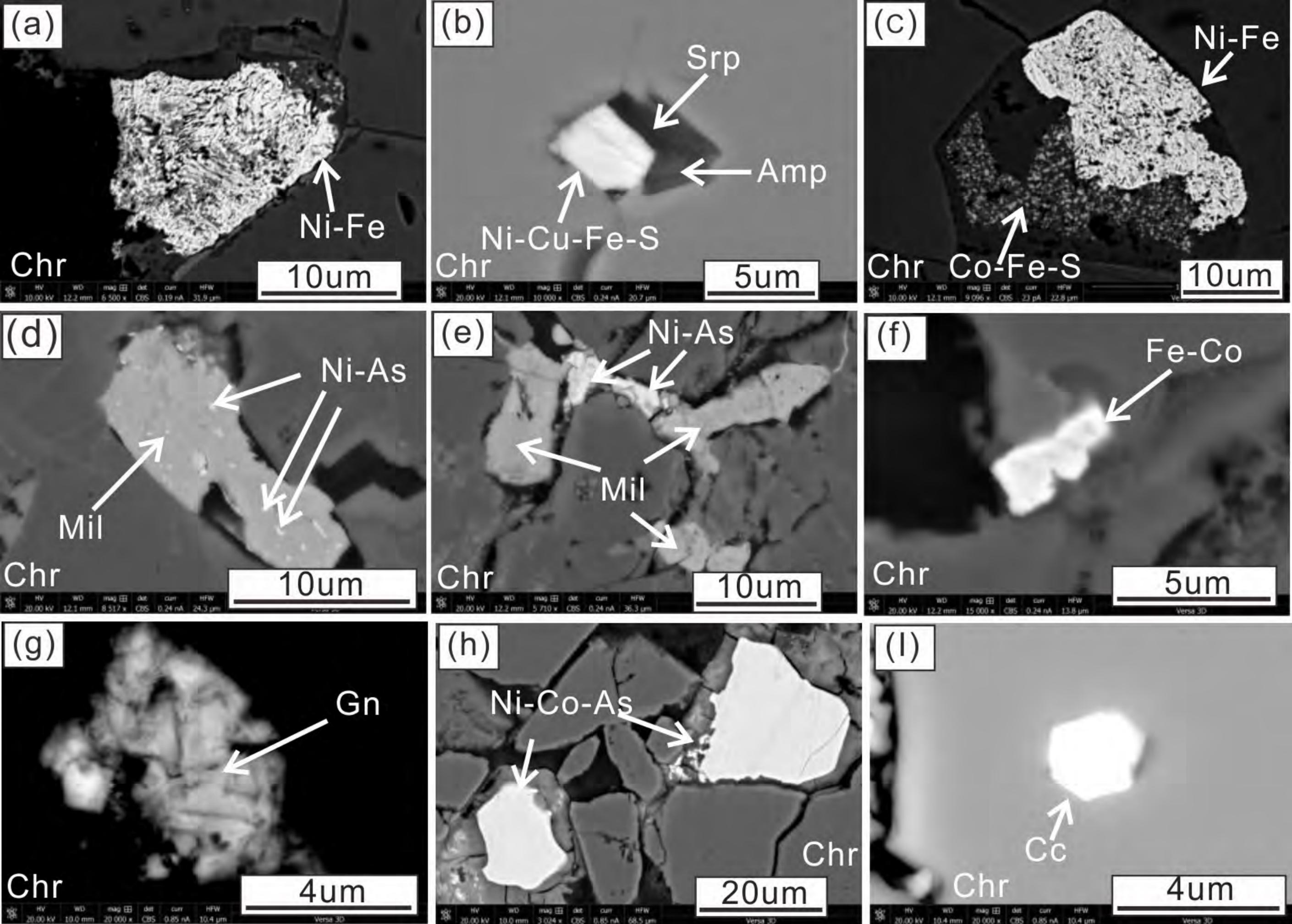


Figure 8

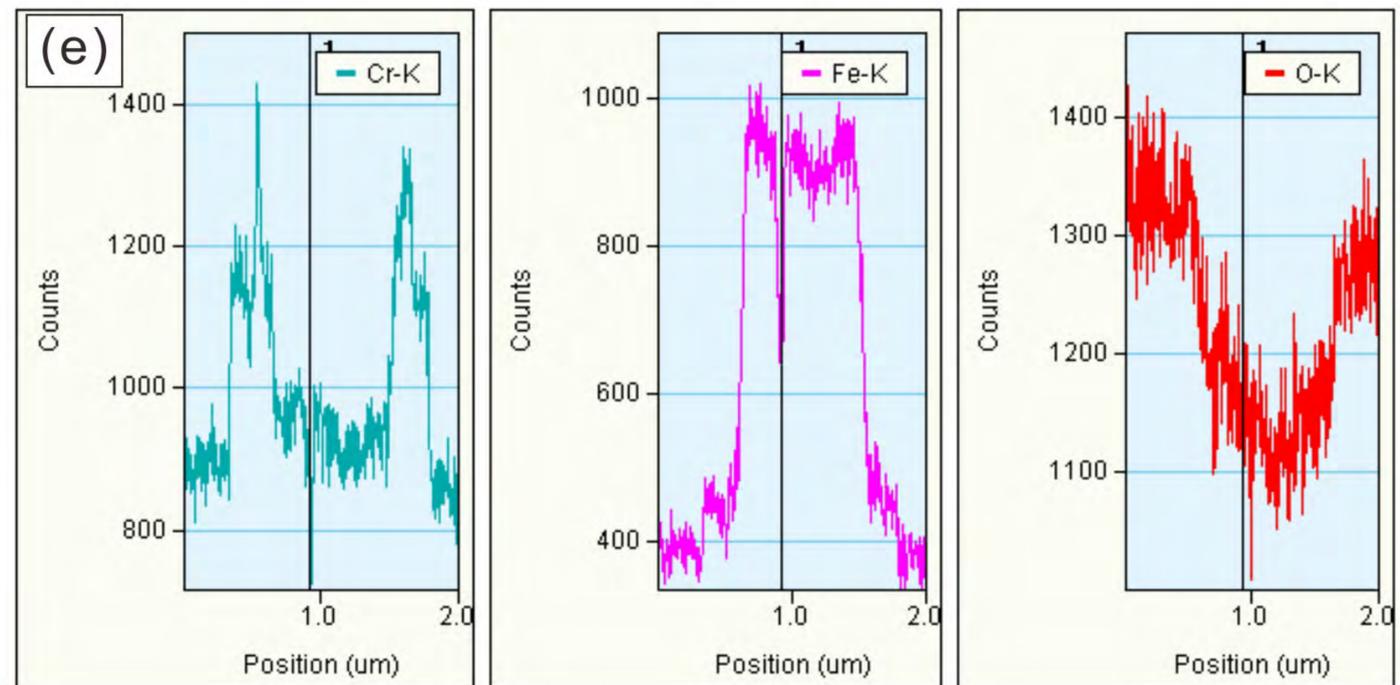
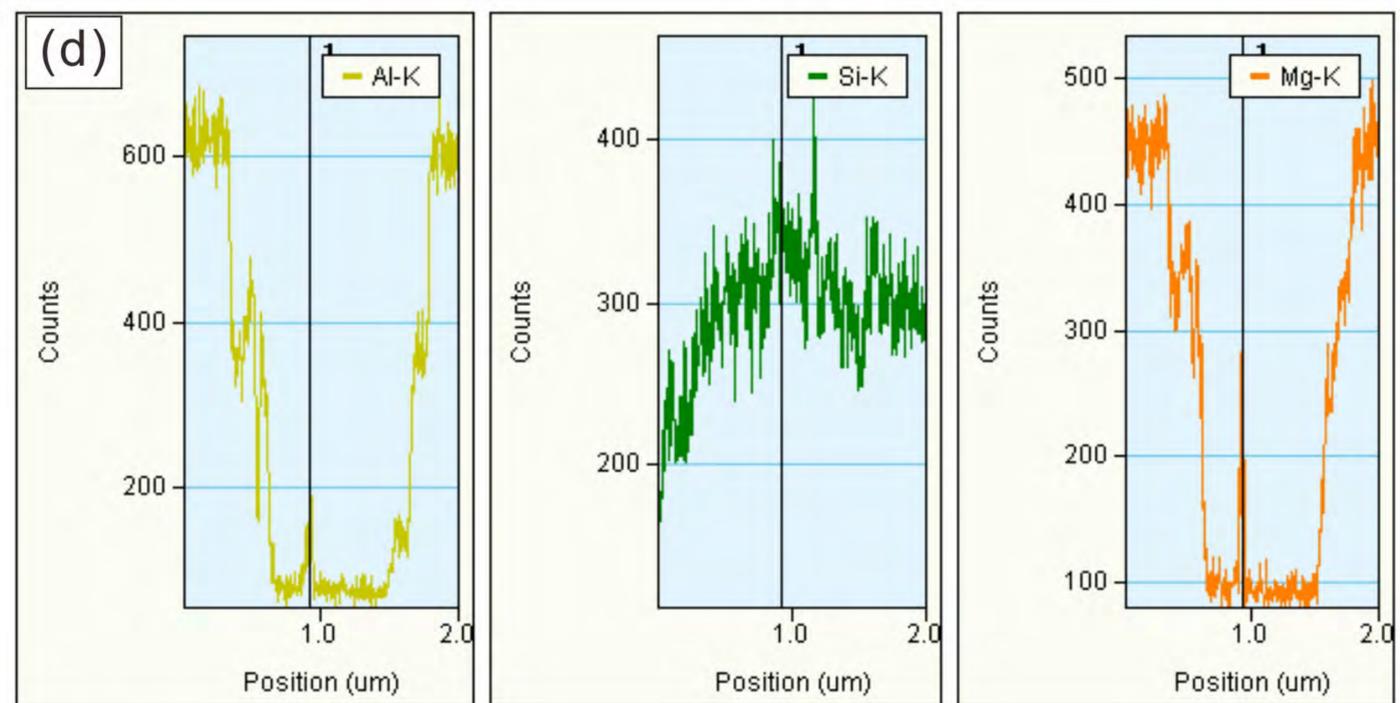
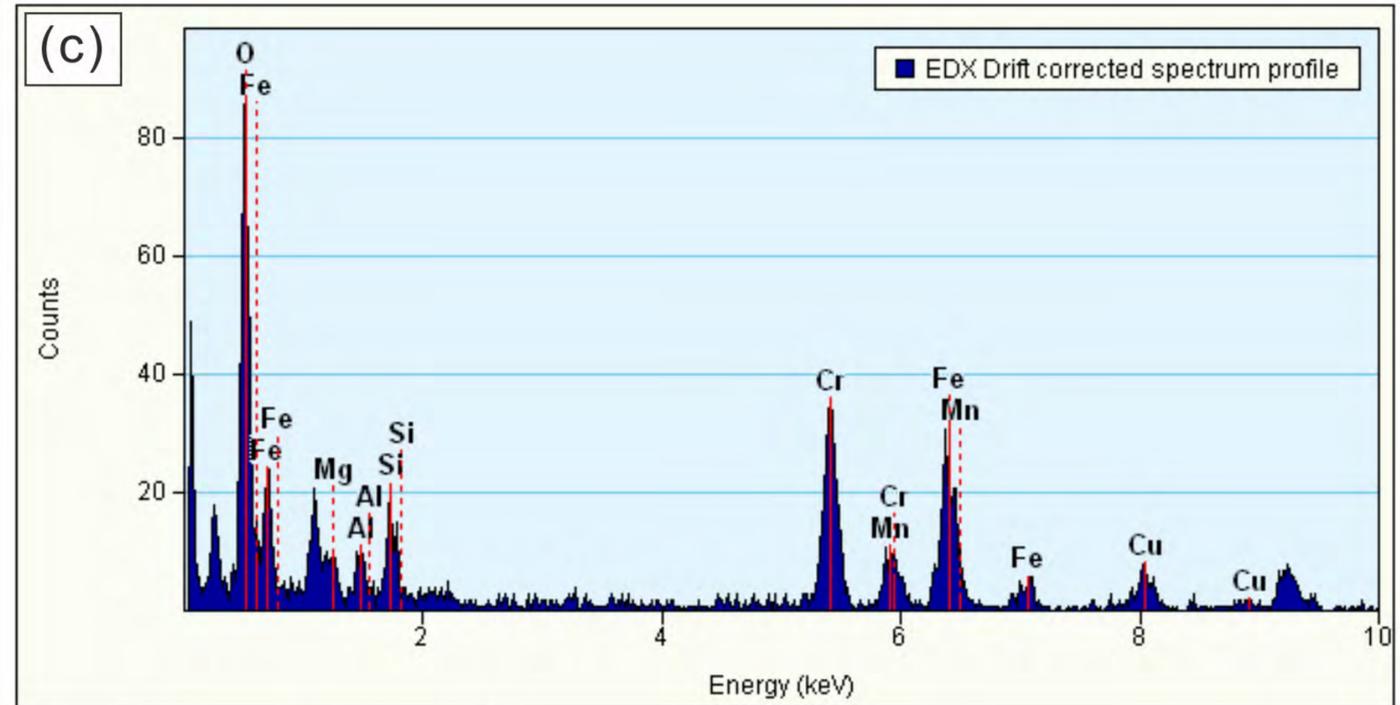
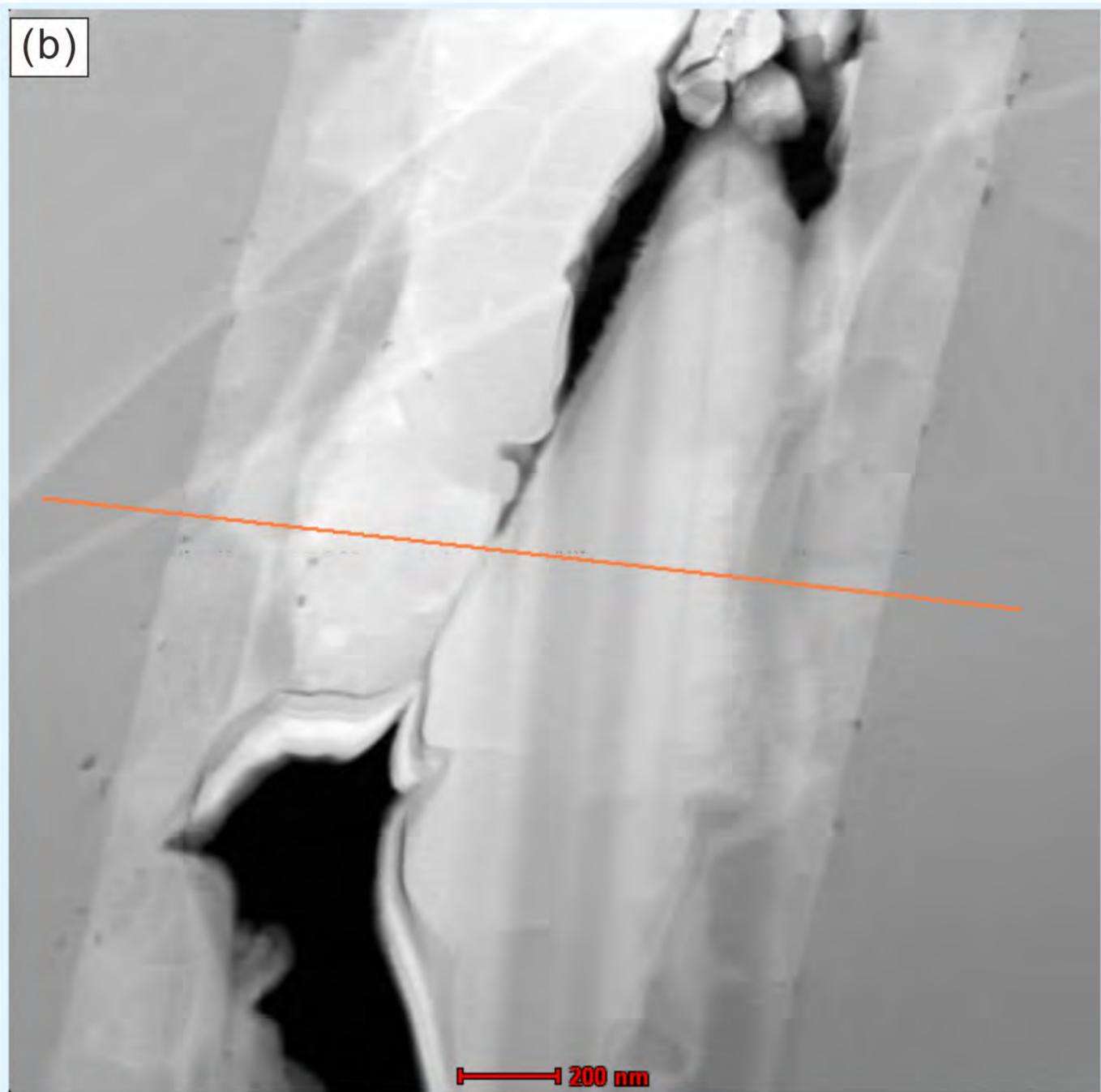
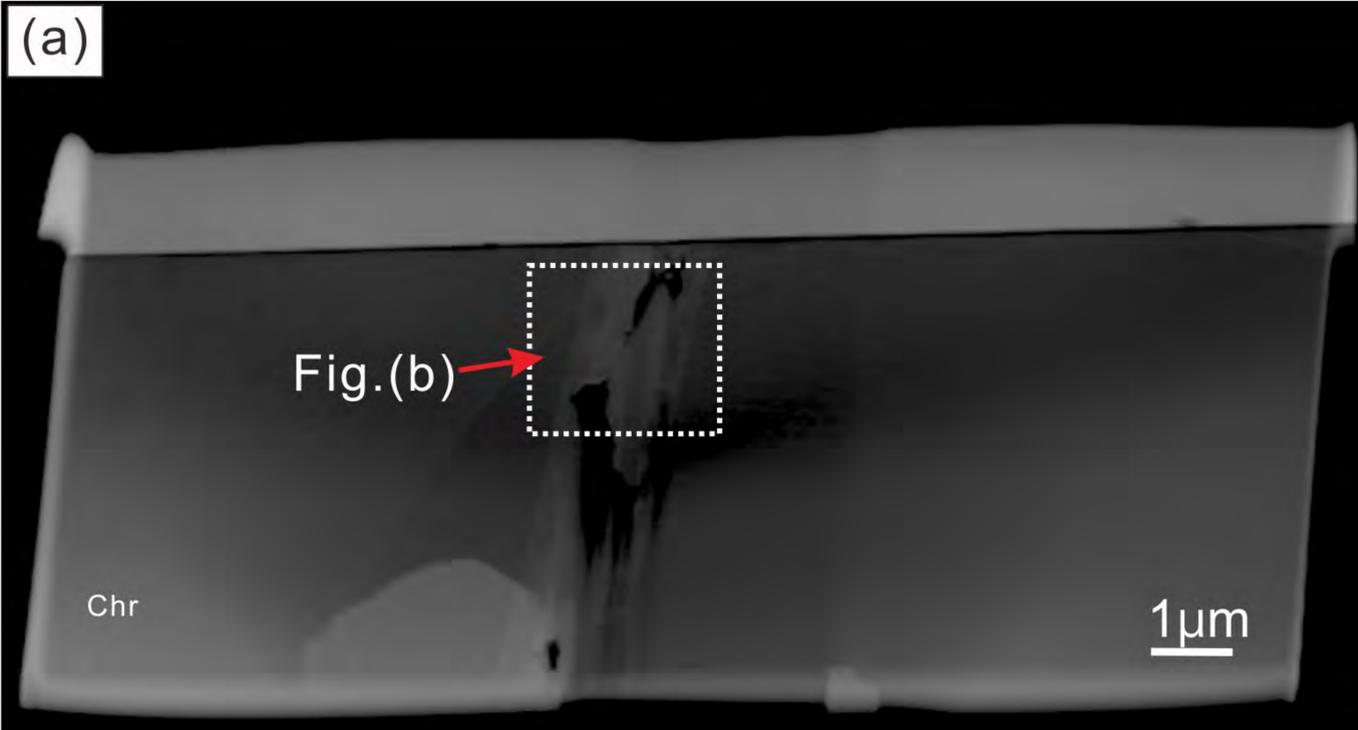


Figure 9

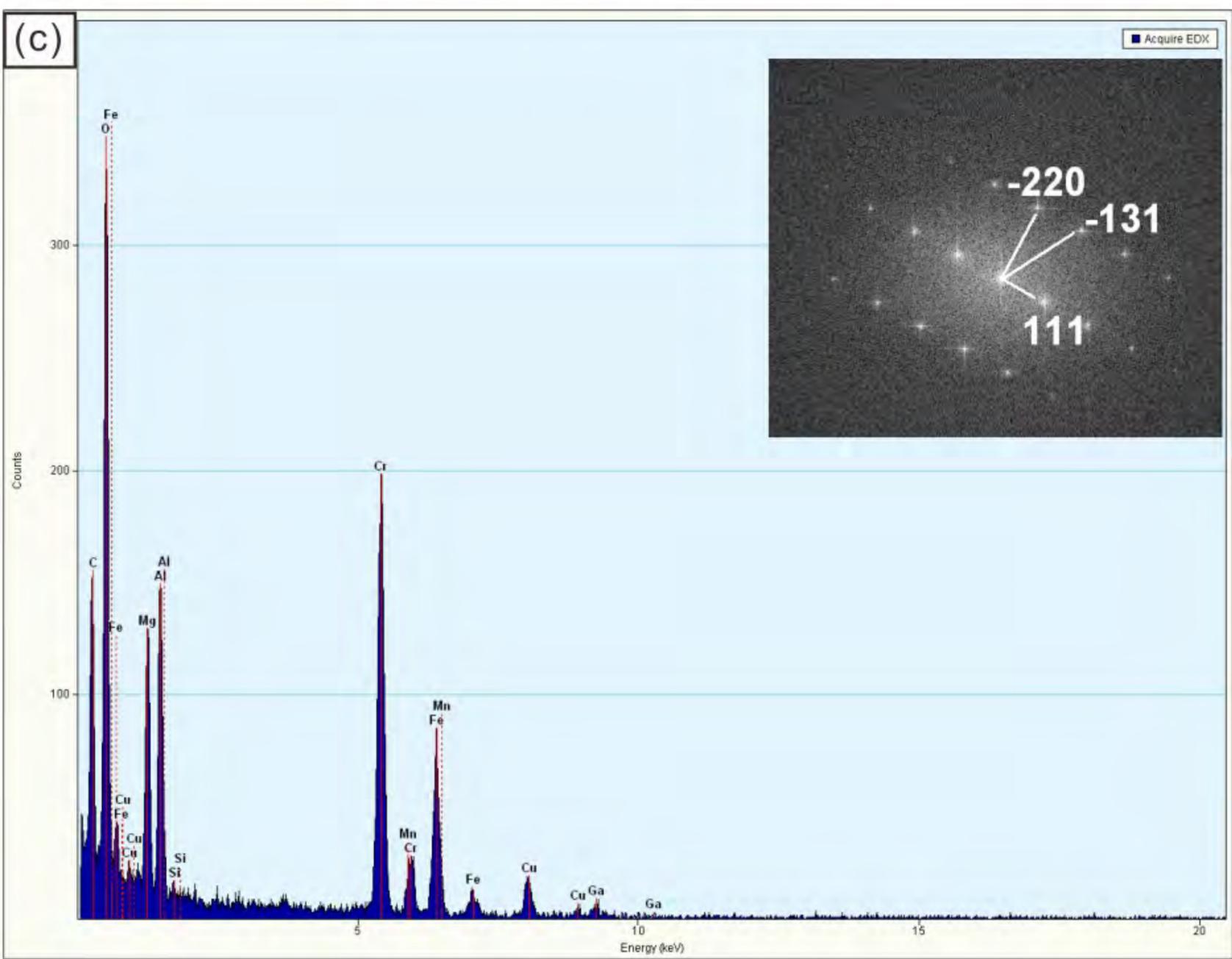
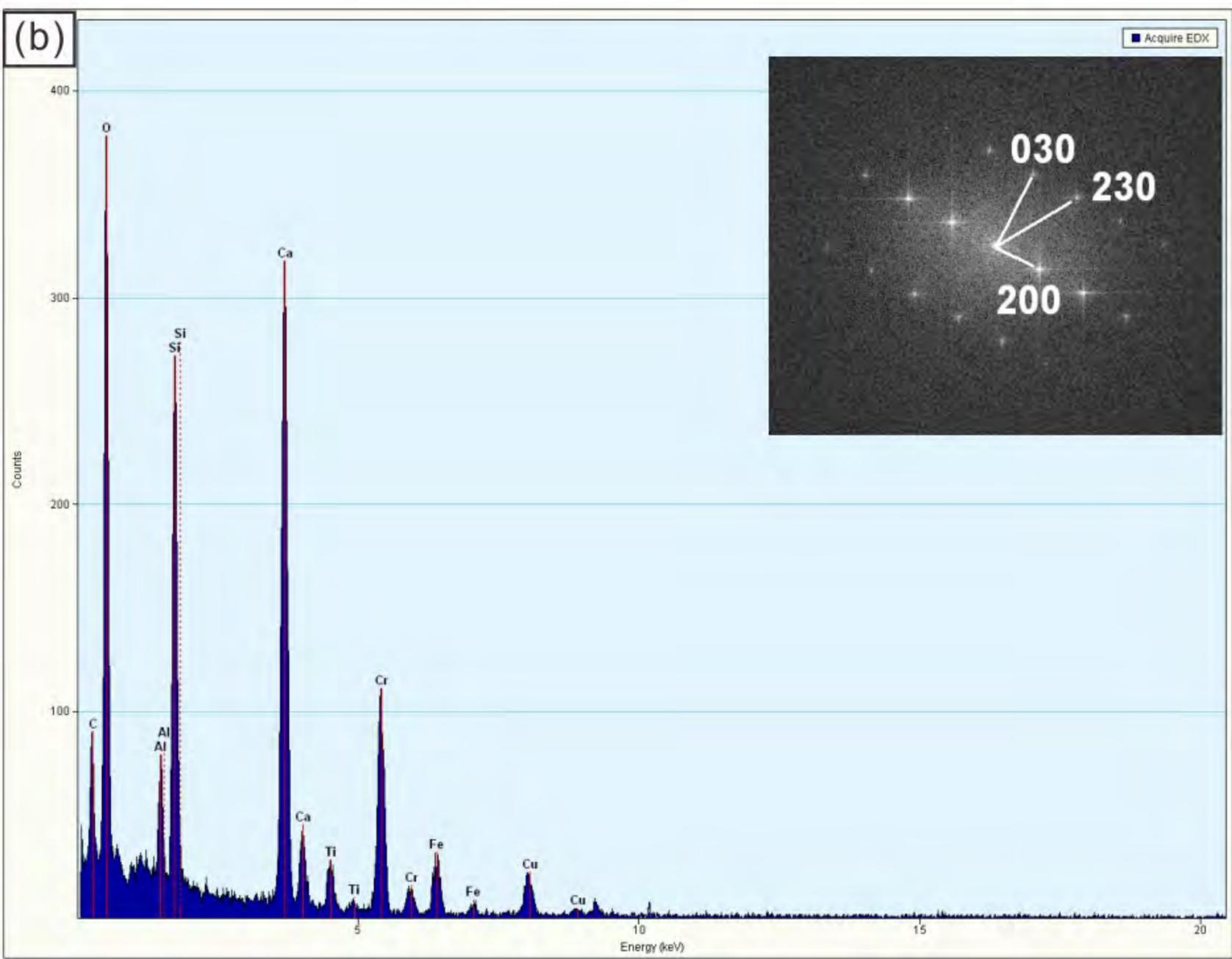


Figure 10

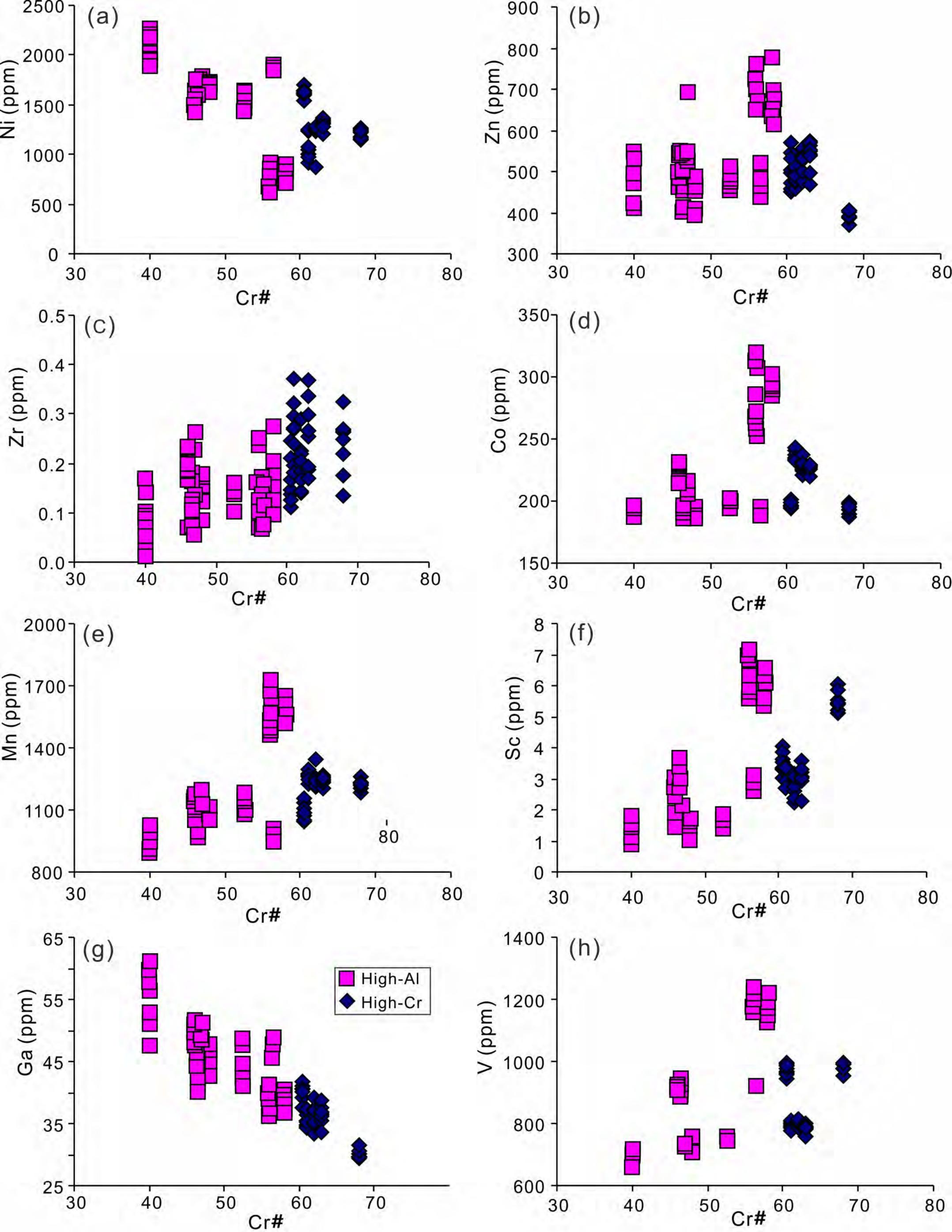


Figure 11

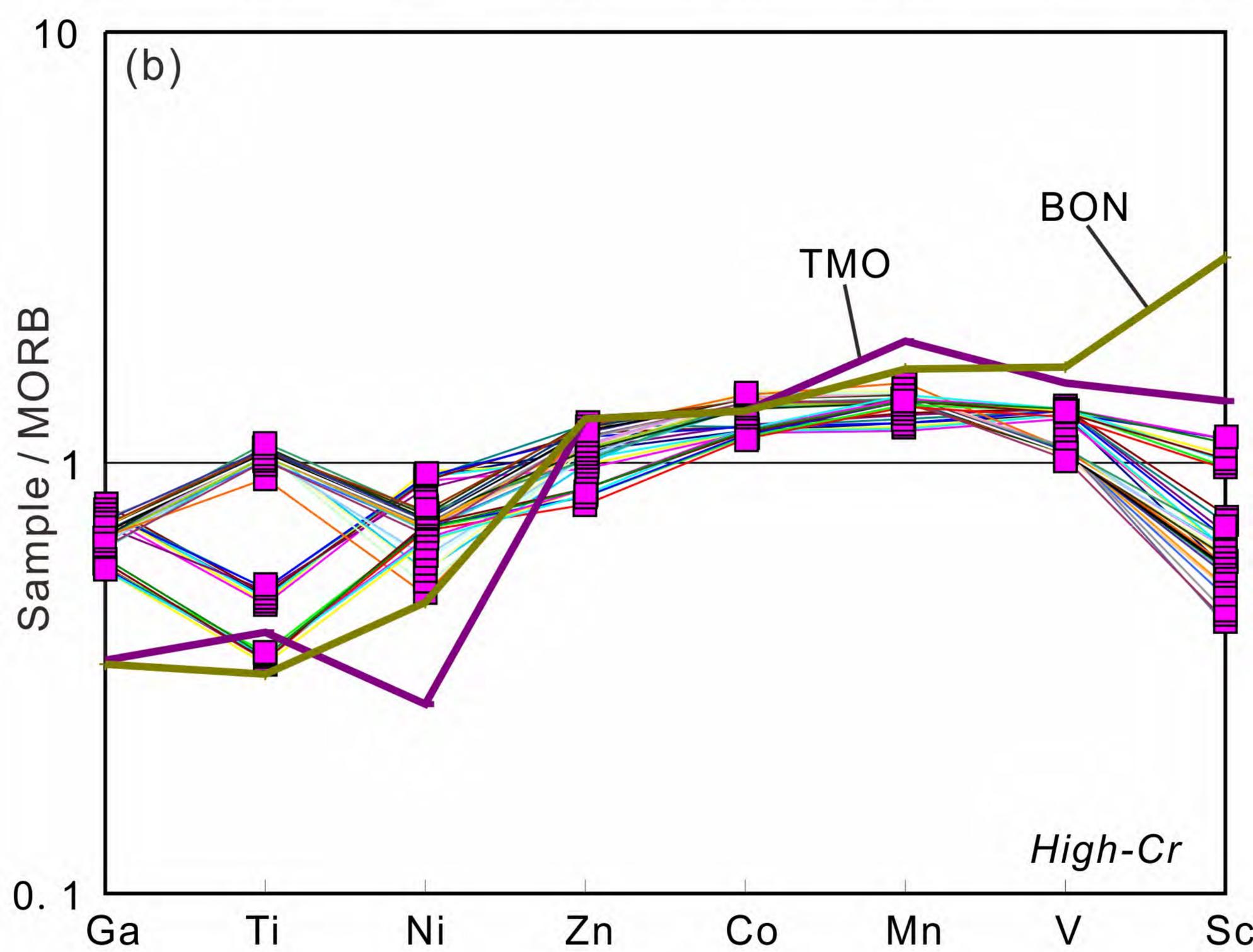
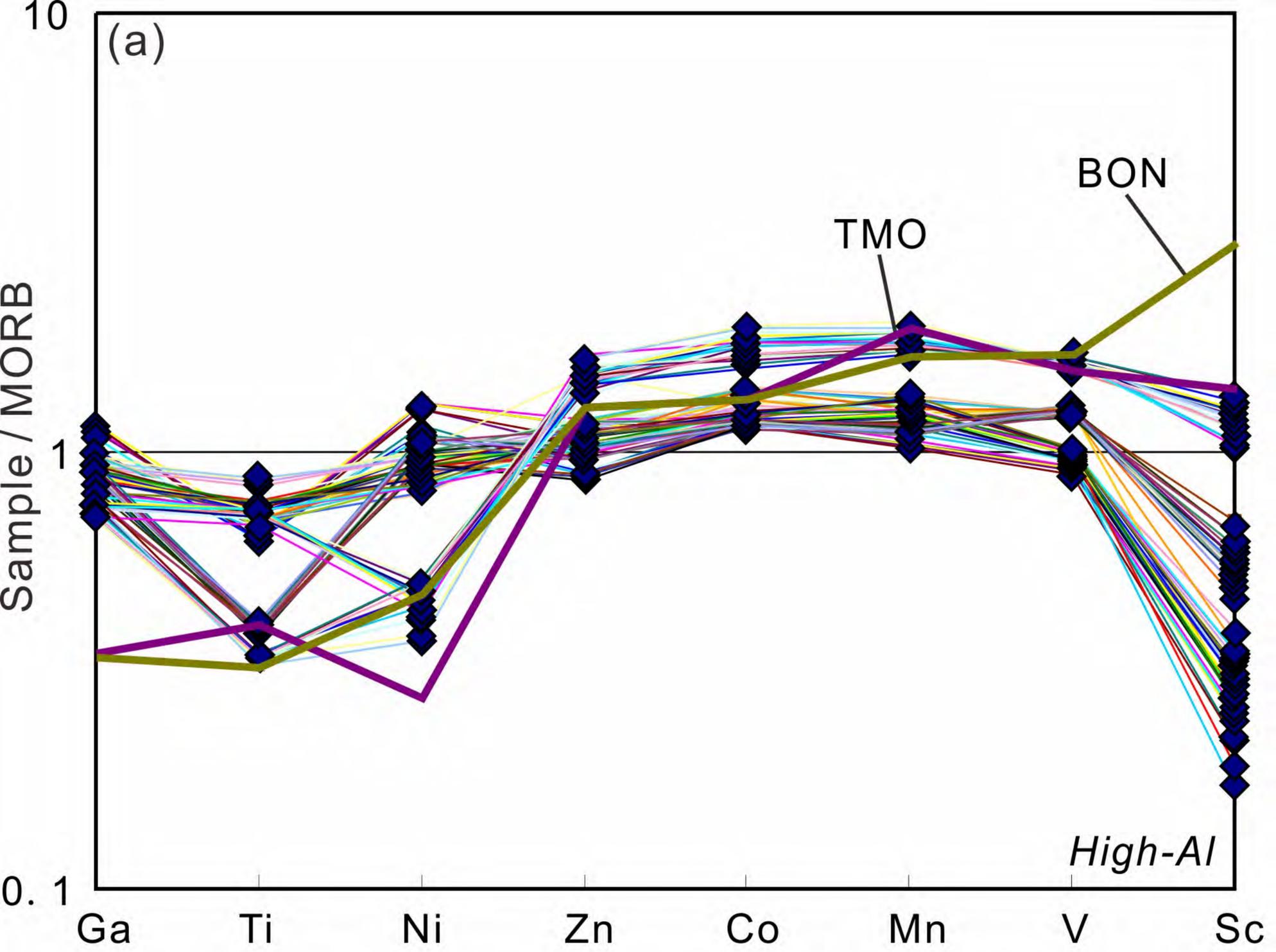


Figure 12

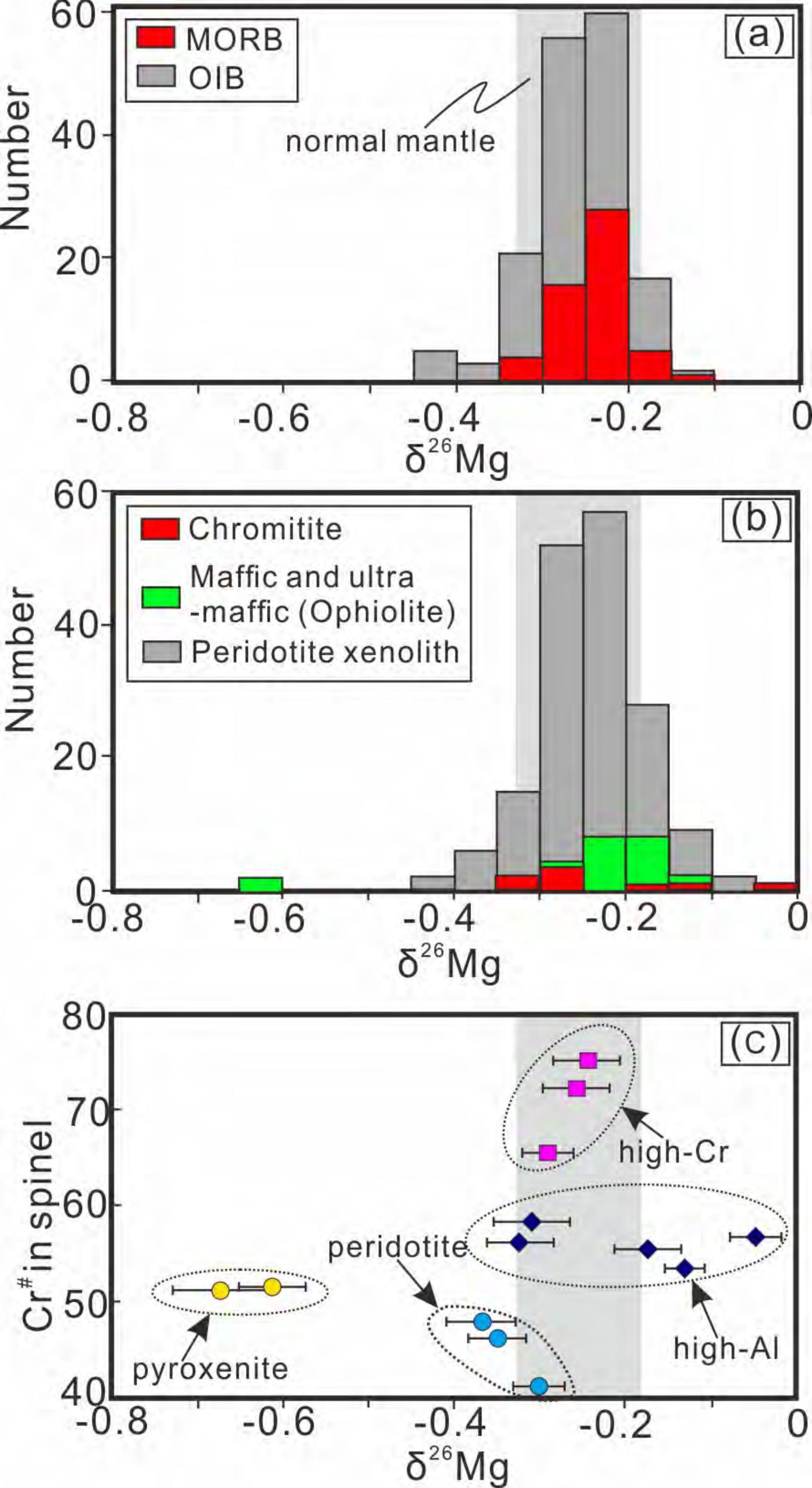


Figure 13

Table 1. Representative EPMA data of chromite grains from the Purang chromitites.

| Lithology | High-Al chromitite | | | | | High-Cr chromitite | | | | | Harzburgite | | | | | Dunite | | | | |
|--------------------------------|--------------------|-------|-------|-------|-------|--------------------|--------------|---------------|---------------|---------------|-------------|-----------|-----------|-----------|-----------|----------|----------|----------|----------|----------|
| | 17x.1 | 17x.2 | 17x.3 | 17x.4 | 17x.5 | 11Y-P214-2.1 | 11Y-P214-2.2 | 11Y-401-10.14 | 11Y-401-10.26 | 11Y-401-10.27 | 09y-636.1 | 09y-636.2 | 09y-636.3 | 09y-636.4 | 09y-636.5 | 1334c2-3 | 1334c2-7 | 1334c2-1 | 1334c2-1 | 1334c2-1 |
| Sample | 17x.1 | 17x.2 | 17x.3 | 17x.4 | 17x.5 | 11Y-P214-2.1 | 11Y-P214-2.2 | 11Y-401-10.14 | 11Y-401-10.26 | 11Y-401-10.27 | 09y-636.1 | 09y-636.2 | 09y-636.3 | 09y-636.4 | 09y-636.5 | 1334c2-3 | 1334c2-7 | 1334c2-1 | 1334c2-1 | 1334c2-1 |
| Cr ₂ O ₃ | 42.24 | 42.59 | 43.29 | 43.04 | 42.92 | 64.17 | 64.46 | 54.50 | 55.14 | 54.86 | 34.268 | 33.67 | 35.005 | 33.698 | 34.941 | 38.889 | 37.681 | 38.76 | 37.255 | 38.228 |
| Na ₂ O | 0.02 | 0.00 | 0.02 | 0.02 | 0.01 | 0.01 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0.03 | 0.01 | 0.00 | 0.00 | 0.03 | |
| K ₂ O | 0.00 | 0.00 | 0.01 | 0.00 | 0.02 | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 | 0 | 0 | 0 | 0.01 | 0.00 | 0.01 | 0.01 | 0.02 | 0.00 | |
| MgO | 16.59 | 15.76 | 16.05 | 16.31 | 16.35 | 10.36 | 10.27 | 12.00 | 12.63 | 12.46 | 13.58 | 13.69 | 13.83 | 13.45 | 14.34 | 13.55 | 13.62 | 13.31 | 14.01 | |
| MnO | 0.22 | 0.22 | 0.25 | 0.28 | 0.21 | 0.39 | 0.38 | 0.37 | 0.33 | 0.33 | 0.30 | 0.26 | 0.22 | 0.24 | 0.25 | 0.22 | 0.21 | 0.21 | 0.28 | |
| CaO | 0.00 | 0.00 | 0.01 | 0.00 | 0.00 | 0 | 0 | 0 | 0 | 0 | 0.00 | 0.00 | 0.00 | 0.04 | 0.00 | 0.00 | 0.00 | 0.04 | | |
| Al ₂ O ₃ | 25.01 | 25.28 | 25.54 | 24.49 | 26.05 | 5.92 | 5.78 | 13.88 | 14.42 | 14.09 | 33.11 | 33.40 | 32.10 | 33.55 | 32.81 | 29.38 | 29.96 | 29.88 | 30.89 | |
| FeO | 14.78 | 15.28 | 14.23 | 14.78 | 14.02 | 19.63 | 19.65 | 18.61 | 17.65 | 18.09 | 19.02 | 18.95 | 19.16 | 19.35 | 17.99 | 17.78 | 18.08 | 18.14 | 17.55 | |
| TiO ₂ | 0.22 | 0.20 | 0.30 | 0.26 | 0.25 | 0.06 | 0.10 | 0.07 | 0.08 | 0.08 | 0.10 | 0.09 | 0.13 | 0.09 | 0.09 | 0.04 | 0.04 | 0.07 | 0.04 | |
| SiO ₂ | 0.04 | 0.04 | 0.02 | 0.02 | 0.04 | 0 | 0 | 0 | 0 | 0 | 0.00 | 0.01 | 0.04 | 0.00 | 0.00 | 0.02 | 0.01 | 0.06 | | |
| NiO | 0.20 | 0.13 | 0.14 | 0.14 | 0.13 | 0.10 | 0.06 | 0.24 | 0.32 | 0.25 | 0.13 | 0.11 | 0.14 | 0.12 | 0.15 | 0.05 | 0.13 | 0.11 | | |
| Total | 99.32 | 99.49 | 99.84 | 99.33 | 99.99 | 100.63 | 100.69 | 99.68 | 100.58 | 100.16 | 100.50 | 100.18 | 100.61 | 100.55 | 100.57 | 99.97 | 99.75 | 100.59 | | |
| Mg# | 67.30 | 65.33 | 67.43 | 66.83 | 68.22 | 51.75 | 51.24 | 57.92 | 60.16 | 59.61 | 59.39 | 59.89 | 60.43 | 58.72 | 62.46 | 60.33 | 60.65 | 59.03 | | |
| Cr# | 53.12 | 53.06 | 53.21 | 54.11 | 52.50 | 87.91 | 88.20 | 72.47 | 71.94 | 72.30 | 40.97 | 40.34 | 42.24 | 40.25 | 41.66 | 47.02 | 45.76 | | | |

[Cr# = $100 \times \text{Cr}^{3+} / (\text{Cr} + \text{Al} + \text{Fe}^{3+})$]; Mg# = $100 \times \text{Mg} / (\text{Mg} + \text{Fe}^{2+})$]

Table 2. Representative EPMA data of silicate inclusions in chromite grains from the Purang chromitites (wt.%).

| Minerals | Cpx | Cpx | Cpx | Cpx | Cpx | Opx | Opx | Opx | Opx | Opx |
|--------------------------------|-------|--------|-------|-------|--------|--------|--------|-------|--------|--------|
| Sample | 706.1 | 706.4 | 706.7 | 706.8 | 706.9 | 706.1 | 706.1 | 706.1 | 706.2 | 706.2 |
| SiO ₂ | 54.36 | 55.21 | 54.22 | 55.34 | 55.24 | 57.54 | 57.75 | 57.58 | 57.29 | 57.59 |
| Na ₂ O | 0.15 | 0.18 | 0.16 | 0.13 | 0.17 | 0.08 | 0.02 | 0.03 | 0.02 | 0.02 |
| Cr ₂ O ₃ | 1.03 | 0.83 | 0.84 | 0.95 | 0.99 | 0.92 | 0.86 | 0.87 | 0.79 | 0.91 |
| K ₂ O | 0.00 | 0.01 | 0.01 | 0.00 | 0.01 | 0.00 | 0.01 | 0.01 | 0.00 | 0.00 |
| MgO | 18.53 | 18.96 | 18.57 | 18.78 | 18.62 | 38.72 | 38.54 | 38.19 | 38.85 | 38.76 |
| Al ₂ O ₃ | 0.66 | 0.65 | 0.69 | 0.72 | 0.82 | 0.29 | 0.32 | 0.35 | 0.28 | 0.36 |
| MnO | 0.04 | 0.06 | 0.04 | 0.05 | 0.02 | 0.06 | 0.07 | 0.05 | 0.06 | 0.07 |
| CaO | 23.53 | 23.72 | 23.32 | 22.36 | 23.14 | 0.35 | 0.29 | 0.27 | 0.33 | 0.37 |
| FeO | 1.02 | 1.05 | 0.86 | 0.98 | 1.01 | 2.59 | 2.45 | 2.27 | 2.52 | 2.39 |
| TiO ₂ | 0.05 | 0.08 | 0.06 | 0.08 | 0.08 | 0.01 | 0.03 | 0.04 | 0.01 | 0.03 |
| NiO | 0.09 | 0.06 | 0.05 | 0.09 | 0.07 | 0.10 | 0.11 | 0.10 | 0.09 | 0.13 |
| Total | 99.46 | 100.81 | 98.82 | 99.48 | 100.17 | 100.66 | 100.46 | 99.76 | 100.24 | 100.63 |
| En | 51.5 | 51.8 | 51.9 | 53.1 | 52.0 | 95.8 | 96.1 | 96.3 | 95.9 | 96.0 |
| Fs | 1.6 | 1.6 | 1.3 | 1.6 | 1.6 | 3.6 | 3.4 | 3.2 | 3.5 | 3.3 |
| Wo | 47.0 | 46.6 | 46.8 | 45.4 | 46.4 | 0.6 | 0.5 | 0.5 | 0.6 | 0.7 |
| Mg# | 97.00 | 96.99 | 97.47 | 97.16 | 97.05 | 96.38 | 96.56 | 96.77 | 96.49 | 96.66 |

Mg# = $[100 \times \text{Mg}/(\text{Mg} + \text{Fe}^{2+})]$

Table 3 Trace element composition, by LA-ICP-MS, of chromite grains from the Puring chromitites (ppm).

| Sample | Lithos | FeO | MgO ₂ | Al ₂ O ₃ | SiO ₂ | TiO ₂ | MnO ₂ | Ga | Ti | Ni | Zn | Co | Mn | V | Sc | Cu | Zr | Nb | Cr | Cr# |
|--------------------|--------------------|-------------|------------------|--------------------------------|------------------|------------------|------------------|----|------|------|-----|-----|------|------|-----|------|------|------|--------|------|
| 662-16-1 | | 11.87175196 | 185517 | 376065 | 4146 | 1407 | 1180 | 60 | 844 | 2260 | 531 | 191 | 914 | 689 | 1.7 | 3.8 | 0.14 | 0.11 | 208112 | 40 |
| 662-16-2 | | 12.06163111 | 184904 | 375313 | 3741 | 1456 | 1203 | 58 | 874 | 2264 | 547 | 197 | 932 | 686 | 1.4 | 2.7 | 0.08 | 0.10 | 207638 | 40 |
| 662-16-3 | | 11.82373195 | 188570 | 377186 | 3747 | 1478 | 1175 | 61 | 887 | 2272 | 531 | 189 | 910 | 681 | 1.3 | 2.6 | 0.07 | 0.13 | 206534 | 40 |
| 662-16-4 | | 11.72831343 | 184180 | 378867 | 4500 | 1395 | 1165 | 56 | 837 | 2159 | 472 | 190 | 903 | 671 | 1.3 | 2.6 | 0.10 | 0.10 | 209376 | 40 |
| 662-16-5 | | 11.64341467 | 187147 | 378294 | 7693 | 1467 | 1164 | 58 | 880 | 2187 | 472 | 189 | 902 | 669 | 1.5 | 2.8 | 0.02 | 0.07 | 204982 | 40 |
| 662-16-6 | | 11.77160677 | 186447 | 379299 | 4224 | 1390 | 1161 | 60 | 834 | 2209 | 494 | 189 | 900 | 661 | 1.2 | 2.5 | 0.09 | 0.12 | 206635 | 40 |
| 692-6-1 | | 12.28386142 | 183765 | 345077 | 1362 | 1403 | 1283 | 53 | 842 | 2020 | 416 | 196 | 994 | 705 | 1.2 | 0.1 | 0.05 | 0.06 | 209674 | 40 |
| 692-6-2 | | 12.297554 | 184009 | 342277 | 2639 | 1404 | 1319 | 53 | 842 | 1957 | 415 | 194 | 1021 | 717 | 1.4 | 1.9 | 0.17 | 0.15 | 223616 | 40 |
| 692-6-3 | | 12.20916705 | 185045 | 346267 | 1187 | 1487 | 1249 | 47 | 892 | 1895 | 412 | 188 | 968 | 709 | 0.9 | 1.7 | 0.09 | 0.16 | 224503 | 40 |
| 692-6-4 | | 12.20725928 | 182415 | 345836 | 2408 | 1486 | 1279 | 51 | 892 | 1948 | 418 | 187 | 991 | 698 | 1.4 | 2.1 | 0.01 | 0.15 | 230407 | 40 |
| 662-27-1 | | 13.14496029 | 173853 | 310537 | 3129 | 1892 | 1458 | 49 | 1135 | 1751 | 549 | 208 | 1129 | 724 | 1.6 | 1.1 | 0.11 | 0.12 | 252128 | 47 |
| 662-27-2 | | 13.2579374 | 171278 | 309558 | 3341 | 1897 | 1552 | 51 | 1138 | 1771 | 694 | 210 | 1203 | 728 | 1.8 | 1.14 | 0.26 | 0.16 | 253023 | 47 |
| 662-27-4 | | 13.2597043 | 172077 | 306745 | 3551 | 1936 | 1394 | 49 | 1161 | 1710 | 521 | 216 | 1080 | 734 | 1.6 | 0.0 | 0.23 | 0.11 | 234853 | 47 |
| 662-27-5 | | 13.1731678 | 172996 | 309402 | 2021 | 1861 | 1384 | 48 | 1116 | 1715 | 486 | 209 | 1072 | 735 | 2.1 | 1.4 | 0.14 | 0.16 | 254928 | 47 |
| 662-27-6 | | 13.17521911 | 173234 | 305668 | 2047 | 1879 | 1379 | 48 | 1135 | 1721 | 532 | 205 | 1068 | 731 | 1.8 | 0.6 | 0.06 | 0.16 | 254513 | 47 |
| 706-8-1 | | 14.18438369 | 162602 | 315281 | 2915 | 1561 | 1520 | 51 | 937 | 1403 | 538 | 231 | 1178 | 909 | 2.9 | 1.2 | 0.07 | 0.16 | 257149 | 46 |
| 706-8-2 | | 14.04933629 | 165080 | 311212 | 2429 | 1542 | 1495 | 49 | 925 | 1416 | 542 | 230 | 1158 | 906 | 2.9 | 1.8 | 0.20 | 0.14 | 251506 | 46 |
| 706-8-3 | | 13.86847647 | 163429 | 315712 | 2533 | 1560 | 1475 | 51 | 936 | 1500 | 548 | 228 | 1142 | 921 | 3.0 | 1.9 | 0.23 | 0.11 | 250917 | 46 |
| 706-8-4 | | 13.62209204 | 161450 | 309602 | 2430 | 1512 | 1434 | 50 | 907 | 1496 | 544 | 227 | 1111 | 905 | 2.8 | 1.8 | 0.23 | 0.14 | 258437 | 46 |
| 706-8-5 | | 13.20825055 | 165671 | 312026 | 1912 | 1568 | 1382 | 47 | 941 | 1624 | 500 | 216 | 1070 | 923 | 1.4 | 1.8 | 0.19 | 0.16 | 257341 | 46 |
| 706-8-6 | | 13.15610487 | 170370 | 317948 | 1885 | 1563 | 1361 | 48 | 938 | 1572 | 462 | 217 | 1054 | 915 | 1.9 | 0.8 | 0.18 | 0.17 | 250175 | 46 |
| 706-8-7 | | 13.29751238 | 164898 | 314522 | 2418 | 1571 | 1419 | 48 | 942 | 1554 | 486 | 229 | 1099 | 925 | 2.4 | 1.5 | 0.17 | 0.15 | 255485 | 46 |
| 599-6-1-1 | | 12.52587291 | 177158 | 291378 | 1746 | 856 | 1276 | 46 | 513 | 1754 | 504 | 196 | 988 | 893 | 3.2 | 2.5 | 0.12 | 0.21 | 268881 | 46.5 |
| 599-6-1-2 | | 12.5112249 | 177041 | 290720 | 2526 | 859 | 1313 | 44 | 516 | 1725 | 480 | 192 | 1017 | 946 | 2.8 | 0.2 | 0.07 | 0.15 | 269298 | 46.5 |
| 599-6-1-3 | | 12.57593438 | 174582 | 291879 | 2810 | 861 | 1256 | 44 | 516 | 1741 | 478 | 194 | 973 | 894 | 2.8 | 0.8 | 0.12 | 0.08 | 268879 | 46.5 |
| 599-6-1-4 | | 12.61613136 | 176414 | 291077 | 3183 | 853 | 1282 | 45 | 512 | 1754 | 544 | 193 | 993 | 899 | 3.2 | 0.1 | 0.10 | 0.12 | 267816 | 46.5 |
| 599-6-1-5 | | 12.62485704 | 174923 | 291642 | 2688 | 846 | 1287 | 46 | 507 | 1748 | 501 | 194 | 997 | 888 | 3.0 | 0.6 | 0.16 | 0.12 | 268915 | 46.5 |
| 599-6-1-6 | | 12.75505568 | 172567 | 287671 | 2448 | 878 | 1324 | 42 | 527 | 1605 | 405 | 189 | 1026 | 931 | 3.4 | 1.5 | 0.18 | 0.14 | 272702 | 46.5 |
| 599-6-1-7 | | 12.68073527 | 173363 | 288006 | 2709 | 880 | 1317 | 40 | 528 | 1640 | 412 | 191 | 1020 | 910 | 3.7 | 0.0 | 0.12 | 0.24 | 271754 | 46.5 |
| 599-6-1-8 | | 12.26943341 | 168950 | 286788 | 2323 | 860 | 1247 | 41 | 516 | 1652 | 450 | 186 | 966 | 894 | 3.4 | 1.9 | 0.11 | 0.16 | 279175 | 46.5 |
| 662-4-1 | | 12.67524986 | 169798 | 304735 | 2844 | 1623 | 1358 | 45 | 974 | 1607 | 409 | 186 | 1052 | 710 | 1.6 | 1.9 | 0.09 | 0.10 | 263166 | 48 |
| 662-4-2 | | 12.72158359 | 173660 | 306595 | 2447 | 1614 | 1366 | 43 | 969 | 1647 | 400 | 189 | 1058 | 708 | 1.4 | 1.6 | 0.14 | 0.13 | 259042 | 48 |
| 662-4-3 | | 13.0321589 | 169767 | 307439 | 4892 | 1750 | 1438 | 48 | 1050 | 1678 | 491 | 195 | 1114 | 752 | 1.7 | 0.9 | 0.14 | 0.13 | 255579 | 48 |
| 662-4-4 | | 12.75604852 | 172100 | 311654 | 2334 | 1714 | 1371 | 45 | 1028 | 1622 | 455 | 194 | 1062 | 738 | 1.0 | 1.7 | 0.18 | 0.13 | 256614 | 48 |
| 662-4-5 | High-Al chromitite | 12.74996509 | 172522 | 311394 | 2687 | 1660 | 1367 | 48 | 996 | 1708 | 472 | 193 | 1059 | 734 | 1.5 | 0.8 | 0.12 | 0.15 | 255950 | 48 |
| 662-4-6 | | 12.52407457 | 172057 | 313387 | 2893 | 1620 | 1385 | 47 | 972 | 1727 | 473 | 194 | 1073 | 742 | 1.6 | 1.1 | 0.15 | 0.18 | 253346 | 48 |
| 662-4-7 | | 13.04575884 | 173075 | 315275 | 2273 | 1652 | 1392 | 46 | 991 | 1715 | 459 | 193 | 1079 | 729 | 1.5 | 1.1 | 0.12 | 0.22 | 254286 | 48 |
| 662-15-1 | | 13.14250613 | 174419 | 289762 | 3189 | 1641 | 1413 | 42 | 984 | 1473 | 457 | 199 | 1094 | 743 | 1.8 | 0.2 | 0.14 | 0.12 | 266576 | 52.5 |
| 662-15-2 | | 13.2672357 | 168471 | 289022 | 3912 | 1626 | 1483 | 41 | 976 | 1476 | 473 | 202 | 1149 | 755 | 1.8 | 2.1 | 0.16 | 0.10 | 268493 | 52.5 |
| 662-15-3 | | 12.9922109 | 174750 | 307966 | 3844 | 1650 | 1473 | 48 | 990 | 1647 | 512 | 202 | 1141 | 752 | 1.4 | 1.9 | 0.16 | 0.14 | 254182 | 52.5 |
| 662-15-4 | | 13.01864086 | 173910 | 313187 | 2070 | 1689 | 1396 | 47 | 1014 | 1633 | 481 | 201 | 1082 | 757 | 1.4 | 0.9 | 0.14 | 0.16 | 252793 | 52.5 |
| 662-15-5 | | 13.27869982 | 171363 | 290352 | 2494 | 1687 | 1493 | 44 | 1012 | 1530 | 506 | 201 | 1157 | 757 | 1.7 | 2.5 | 0.10 | 0.11 | 268248 | 52.5 |
| 662-15-6 | | 13.32887157 | 166656 | 277784 | 3310 | 1719 | 1526 | 42 | 1031 | 1478 | 509 | 195 | 1182 | 754 | 1.7 | 1.8 | 0.14 | 0.18 | 277727 | 52.5 |
| 599-5-1-1 | | 12.49600684 | 178537 | 317624 | 2800 | 886 | 1231 | 48 | 531 | 1904 | 450 | 191 | 954 | 923 | 3.0 | 1.9 | 0.17 | 0.11 | 249871 | 56.5 |
| 599-5-1-2 | | 12.28634768 | 179427 | 319179 | 2514 | 897 | 1227 | 49 | 538 | 1907 | 470 | 192 | 950 | 922 | 2.9 | 0.8 | 0.16 | 0.08 | 249575 | 56.5 |
| 599-5-1-3 | | 12.46620614 | 176911 | 313628 | 2885 | 881 | 1253 | 46 | 529 | 1883 | 483 | 192 | 970 | 921 | 2.9 | 0.7 | 0.07 | 0.12 | 253721 | 56.5 |
| 599-5-1-4 | | 12.4182237 | 184569 | 309754 | 2534 | 1229 | 1229 | 46 | 532 | 1849 | 442 | 190 | 952 | 925 | 2.6 | 1.8 | 0.14 | 0.16 | 251753 | 56.5 |
| 599-5-1-5 | | 12.41031731 | 175938 | 314429 | 2625 | 913 | 1263 | 46 | 548 | 1869 | 466 | 189 | 979 | 928 | 2.6 | 2.8 | 0.11 | 0.11 | 253867 | 56.5 |
| 599-5-1-6 | | 12.63668083 | 178431 | 304129 | 2643 | 871 | 1300 | 48 | 523 | 1853 | 520 | 194 | 1007 | 924 | 3.1 | 2.0 | 0.07 | 0.13 | 257805 | 56.5 |
| 11YP-19-2-b1-Cr-1 | | 17.1025489 | 150856 | 232337 | 2326 | 1615 | 1979 | 39 | 969 | 898 | 620 | 296 | 1533 | 1142 | 6.6 | 5.1 | 0.14 | 0.14 | 293758 | 58 |
| 11YP-19-2-b1-Cr-2 | | 17.55184859 | 150650 | 229441 | 2133 | 1567 | 2038 | 39 | 940 | 884 | 619 | 294 | 1579 | 1163 | 6.5 | 1.5 | 0.25 | 0.14 | 292964 | 58 |
| 11YP-19-2-b1-Cr-3 | | 17.57630239 | 147236 | 229938 | 1717 | 1582 | 2009 | 40 | 949 | 876 | 639 | 286 | 1557 | 1154 | 6.4 | 2.1 | 0.14 | 0.19 | 295242 | 58 |
| 11YP-19-2-b1-Cr-4 | | 17.73446085 | 139984 | 235835 | 1723 | 1609 | 1978 | 39 | 965 | 820 | 681 | 293 | 1532 | 1130 | 6.2 | 1.6 | 0.11 | 0.17 | 293521 | 58 |
| 11YP-19-2-b1-Cr-5 | | 17.8811418 | 139644 | 231960 | 1734 | 1658 | 2123 | 40 | 995 | 823 | 655 | 297 | 1644 | 1161 | 6.6 | 2.0 | 0.15 | 0.16 | 295633 | 58 |
| 11YP-19-2-b1-Cr-6 | | 17.57451136 | 142499 | 235619 | 3433 | 1602 | 2052 | 39 | 961 | 796 | 671 | 297 | 1589 | 1181 | 6.3 | 3.8 | 0.17 | 0.16 | 293403 | 58 |
| 11YP-19-2-b1-Cr-7 | | 17.63907244 | 138853 | 232425 | 1644 | 1577 | 2068 | 40 | 946 | 838 | 696 | 292 | 1602 | 1146 | 6.6 | 6.4 | 0.21 | 0.16 | 298961 | 58 |
| 11YP-19-2-b1-Cr-8 | | 17.16028838 | 140796 | 241529 | 1793 | 1510 | 2020 | 37 | 906 | 754 | 776 | 294 | 1565 | 1209 | 5.4 | 1.5 | 0.12 | 0.18 | 294242 | 58 |
| 11YP-19-2-b1-Cr-9 | | 17.67636666 | 139247 | 239860 | 1541 | 1653 | 2125 | 40 | 992 | 788 | 696 | 303 | 1646 | 1194 | 6.6 | 1.4 | 0.30 | 0.22 | 293252 | 58 |
| 11YP-19-2-b1-Cr-10 | | 17.35752045 | 139423 | 238432 | 2007 | 1659 | 2055 | 39 | 995 | 802 | 652 | 294 | 1592 | 1178 | 5.6 | 1.6 | 0.27 | 0.21 | 296122 | 58 |
| 11YP-19-2-a2-Cr-1 | | 17.01260236 | 144184 | 242280 | 1591 | 171 | 1933 | 39 | 463 | 831 | 725 | 267 | 1497 | 1227 | 7.1 | 1.6 | 0.19 | 0.10 | 292854 | 56 |
| 11YP-19-2-a2-Cr-2 | | 17.00316694 | 145564 | 241771 | 2423 | 755 | 1957 | 41 | 453 | 903 | 689 | 270 | 1516 | 1222 | 7.1 | 1.6 | 0.18 | 0.10 | 291597 | 56 |
| 11YP-19-2-a2-Cr-3 | | 16.97904728 | | | | | | | | | | | | | | | | | | |

Table 4 Mg isotopic compositions of selected samples from the Purang chromitites and associated peridotites.

| Sample Standard | Rock type | $\delta^{25}\text{Mg}$ | 2SD | $\delta^{26}\text{Mg}$ | 2SD | Cr# in spinel |
|-----------------|--------------------|------------------------|------|------------------------|------|---------------|
| GSB | | -1.03 | 0.04 | -2.01 | 0.03 | |
| GSB | | -1.05 | 0.04 | -2.04 | 0.03 | |
| BCR-2 | | -0.12 | 0.04 | -0.25 | 0.03 | |
| BHVO-2 | | -0.13 | 0.05 | -0.27 | 0.04 | |
| 11YP-19-b2 | high-Al chromitite | -0.16 | 0.03 | -0.31 | 0.05 | 58.4 |
| 11YP-19-b1 | high-Al chromitite | -0.15 | 0.02 | -0.32 | 0.04 | 56.2 |
| 10y-22-3 | high-Al chromitite | -0.08 | 0.05 | -0.13 | 0.02 | 53.5 |
| 10y-22-b | high-Al chromitite | -0.09 | 0.03 | -0.17 | 0.04 | 55.5 |
| 11YP-19-1 | high-Al chromitite | -0.02 | 0.02 | -0.05 | 0.03 | 56.9 |
| 11Y-401-1a | high-Cr chromitite | -0.12 | 0.01 | -0.25 | 0.04 | 75.2 |
| 11Y-401-2 | high-Cr chromitite | -0.14 | 0.03 | -0.29 | 0.03 | 65.6 |
| 11Y-401-1b | high-Cr chromitite | -0.12 | 0.04 | -0.26 | 0.04 | 72.3 |
| B3R30P1 | harzburgite | -0.13 | 0.03 | -0.26 | 0.02 | 40.3 |
| B14R665P1c | harzburgite | -0.19 | 0.05 | -0.37 | 0.04 | 48.4 |
| B25R82P3b | harzburgite | -0.17 | 0.05 | -0.35 | 0.03 | 46.8 |
| 11YP-19-a2 | pyroxenite | -0.31 | 0.02 | -0.61 | 0.04 | 51.6 |
| 11YP-19-a1 | pyroxenite | -0.33 | 0.07 | -0.67 | 0.06 | 51.2 |