1	Isotopic signature of core-derived SiO ₂
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3 4	ms. for resubmission to <i>American Mineralogist</i> : first rev. 7 Feb. 2018; submitted 26 Jan. 2018; original submitted 28 Nov. 2017

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ABSTRACT

9 We apply an experimentally based thermodynamic model of Si+O saturation for the core 10 to determine the saturation level of these elements under the conditions when the core 11 formed. The model limits the bulk Si content of the core to between 0.4 and 3.1 wt% de-12 pending on the pressure, temperature and oxygen content of the metal when it segregated ¹³ from silicate. With knowledge of the core's Si content, the measured ³⁰Si content of the 14 silicate Earth, and the experimentally determined metal-silicate fractionation factor, we 15 can calculate the core's δ^{30} Si, which is between -0.92 to -1.36 ‰. SiO₂ cycled through 16 the core and then released into the mantle might be trapped in inclusions in diamond ¹⁷ formed in the lower mantle. These would be characterized by significantly lighter δ^{30} Si 18 values of -1.12 ± 0.13 (1 σ) ‰, compared to bulk silicate earth values of -0.29 ‰ and a po-19 tentially key indicator of mass transfer from the core to the mantle.

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INTRODUCTION

21 The Earth formed through a relatively rapid process during which approximately chon-22 dritic materials aggregated as a discrete body from the protoplanetary disk and 23 24 *E-mail: george@elsi.jp

25 differentiated into a metallic core and a silicate crust and mantle (Wood et al., 2006). 26 During the melting arising from the accretion process (from impact energy, heat produc-27 tion from short-lived radionuclides, gravitational potential energy release from differenti-28 ation), either a planetary-scale magma ocean or spatially restricted magma lakes arose, 29 imparting a relatively low pressure differentiation signal (compared to the pressures of ei-30 ther the core-mantle boundary (CMB) or the planet's center) on the moderately 31 siderophile elements (Li and Agee, 1996; Wade and Wood, 2005; Siebert et al., 2011; 32 2013; Fischer et al., 2015).

Concurrent with siderophile element partitioning, some major elements also entered the metal destined for the core: Si, O, and Mg are potential candidates (O'Neill et al., Si 1998; O'Rourke and Stevenson, 2016; Badro et al., 2016; Hirose et al., 2017). In particuial ar, some Si is believed to reside in the core because the Mg/Si and Al/Si ratios in the bulk silicate Earth are higher than chondritic values (Palme and O'Neill, 2003). Based on high-pressure experiments at CMB conditions, Hirose et al. (2017) developed a model for SiO₂ saturation in core metal that allows the amount of Si+O potentially ingested by the another of SiO₂ as the core cools. Escape of SiO₂ from the core is virtually certain due to the core imprints core-derived Si with the metal-silicate stable isotope fractionation preta the core imprints core-derived Si with the metal-silicate stable isotope fractionation preta vailing at the conditions of differentiation (Georg et al., 2007; Shahar et al., 2011) rather than bulk silicate Earth values and potentially provides a way to identify SiO₂ previously house of the core. The core's estimated Si content is based on a new, experimentally⁴⁷ based set of constraints not previously exploited, to our knowledge, for making a metal-⁴⁸ silicate isotope balance. We explore core-hosted SiO_2 isotopic signatures here; the physi-⁴⁹ cal mechanism for expelling SiO_2 from the core is described separately (Helffrich et al., ⁵⁰ 2018).

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METHODS

⁵² We use Hirose et al.'s (2017) Si+O solubility model to determine joint Si+O solubility at ⁵³ various pressure (*P*) and temperature (*T*) conditions during the course of core formation ⁵⁴ and cooling. The conditions of core formation are set using a single-stage core formation ⁵⁵ model to approximate the range of pressures and temperatures encountered during accre-⁵⁶ tion. The conditions are set as fractions of the CMB pressure, leading to a *P* range of ⁵⁷ 30-55 GPa (see Rubie et al. (2011) for one reckoning of the range). From *P*, the associ-⁵⁸ ated *T* is obtained from two different equations for the peridotite liquidus and solidus, ⁵⁹ *T*(*P*) (Wade and Wood, 2005; Fiquet et al., 2010), respectively. Figure 1 shows a suite of ⁶⁰ saturation curves at various core formation pressures, calculated by evaluating the joint ⁶¹ Si-O saturation expressions at the *P* and corresponding *T* on a grid and contouring (Hi-⁶² rose et al., 2017).

The method we use to estimate Earth's core's Si content is new, and based on joint 64 Si+O solubilities in core metal, the constitution of the Earth's core, and the high-pressure 65 behavior of the eutectic compositions of Fe-Si and Fe-FeO. In particular, Hirose et al. 66 (2017) note that the Earth's inner core requires that it crystallize essentially pure Fe, 67 which restricts the core liquid composition to the intersection of the SiO₂ saturation con-68 tour and the compositional triangle bounded by SiO₂, the Fe-Si eutectic, and the Fe-FeO



⁷⁰ Figure 1. Saturation of Si and O in metallic iron at a range of pressures corresponding to ⁷¹ core formation. Labels on each saturation line are *P* and *T* of formation; T(P) from Fi-⁷² quet et al. (2010). The Si uptake depends on the O content, and hence yields a range for ⁷³ the mass of Si carried to the core in metal. Quasi-diagonal lines show the SiO₂ loss ⁷⁴ trend; for inner core properties like Earth's, feasible Si+O contents must be more O-rich ⁷⁵ than the Fe-Si eutectic (lower line) and less O-rich than the Fe-FeO eutectic (upper line). ⁷⁶ This limits core Si content to be 0.4-3.1 wt%.

77 eutectic (Fig. 1). The Fe-FeO eutectic moves to higher O content with increasing pres-

78 sure (Komabayashi, 2014; Morard et al., 2017), but the asymptotic behavior of the SiO₂

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⁷⁹ solubility contours with increasing O do not strongly affect the lower bound for Si. In ⁸⁰ contrast, Fe-Si's eutectic moves Fe-ward at high pressure (Fischer et al., 2013; Ozawa et ⁸¹ al., 2016). At the pressure of the Earth's CMB, Fischer et al. (2013) placed the eutectic at ⁸² 5-12 wt% Si, and various estimates of the maximum core Si content by Hin et al. (2014) ⁸³ and Dauphas et al. (2015) placed it at 6-8 and 0-9 wt%, respectively. However, the study ⁸⁴ by Ozawa et al. (2016) of the eutectic dependence on pressure narrowed the bound con-⁸⁵ siderably to ≤1.5 wt%, which we use to define the Fe-Si eutectic in the O-free system. ⁸⁶ Figure 1 depicts the bounds constraining core Si.

A Si isotope balance of the Earth (subscript BE) may be written in terms of its parti-88 tioning between the core's metal (subscript c) and the mantle and lithosphere's silicate 89 (subscript BSE)

90 $\delta^{30}\text{Si}_{BE} = f_c \delta^{30}\text{Si}_c + (1 - f_c)\delta^{30}\text{Si}_{BSE}$. (1) 91 f_c is the mass fraction of Si in the core to the Si in the entire Earth. We use a pyrolitic 92 model for the silicate Earth (McDonough and Sun, 1995) to obtain its Si content (21 93 wt%) and the Si+O solubility model to determine the core Si content. To account for the 94 silicate-metal partitioning during core formation, we use Shahar et al.'s (2011) tempera-95 ture dependent fractionation factor $\Delta^{30}\text{Si}(T) = -7.45(\pm 0.41) \times 10^6/T^2$. Hence,

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$$\delta^{30} \operatorname{Si}_{c} = \Delta^{30} \operatorname{Si}(T) + \delta^{30} \operatorname{Si}_{BSE} \quad .$$
 (2)

97 δ^{30} Si_{BSE} is -0.29 ± 0.02 ‰ (Fitoussi et al., 2009), and the *P*-*T* conditions of core forma-98 tion sets the fractionation factor and f_c from the Si+O saturation model. From (1) and (2) 99 and these values, δ^{30} Si_{BE} may be calculated. We use a single-stage core formation model, but more elaborate methods that track 101 the evolution of the mantle's ³⁰Si content through the accretion process are also possible 102 (Zambardi et al., 2013; Hin et al., 2014). We know experimentally that Si partitioning be-103 tween metal and silicate is not pressure dependent (Fischer et al., 2015; Hirose et al., 104 2017) so the core's Si content is fairly constant during accretion (Tuff et al., 2011), yield-105 ing little difference between multi-stage core formation models and single-stage. Hin et 106 al. (2014) showed that the difference between $\delta^{30}Si_{BE}$ and $\delta^{30}Si_{BSE}$ during their accretion 107 histories is never more than 0.3 ‰. The signal that we predict is 3-4 times larger than 108 this, justifying, post-hoc, the use of the single-stage model for obtaining $\delta^{30}Si_c$.

Xu et al. (2017) investigated Si diffusion in stishovite and provided an expression Xu et al. (2017) investigated Si diffusion in stishovite and provided an expression for the pressure- and temperature-dependent Si diffusion coefficient. Using lower mantle pressure and temperature range of 20 GPa, 2000 K - 135 GPa, 4000K (see, e.g. Helffrich (2017)), leads to Si diffusion times over 1 mm distances of at least 300 Myr to 620 Gyr (due, in part, to the diffusion coefficient's strong pressure dependence). Assuming that (due, in part, to the diffusion coefficient's strong pressure dependence). Assuming that CaCl₂-structure SiO₂, the SiO₂ polymorph stable at higher pressures than stishovite, behaves similarly, Si isotopic disequilibrium may be maintained over the times required for detection in diamond inclusions.

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RESULTS

118 Limits on the uptake of Si by the core may be obtained from Figure 1. The present prop-119 erties of the Earth's core (Hirose et al., 2017), the shape of the saturation contours, and 120 the positions of the Fe-FeO eutectic ($E_{\text{Fe-FeO}}$) and the Fe-Si eutectic ($E_{\text{Fe-Si}}$) control the 121 core's Si content. While the asymptotic nature of the relation at low Si renders the This is a preprint, the final version is subject to change, of the American Mineralogist (MSA) Cite as Authors (Year) Title. American Mineralogist, in press. DOI: https://doi.org/10.2138/am-2018-6482CCBYNCND

25 60 GPa 30 35 40 45 50 55 - 1-4 -1.2 core-derived SiO₂ -1.0 δ³⁰Si, per mil 0 -0 10 0 0 0 0 0 -0.6 enstatite ordinary+carbonaceous 4.0-¥ Ж Ж Ж -0 -0 2507 2755 3294 2633 2873 2985 3093 3196 3053 2658 2789 2921 3184 3316 3447 3579

Figure 2. Bulk Earth (x) and core-hosted SiO₂ (\bullet , \odot)) ³⁰Si fractionation calculated using 124 Si+O saturation at various temperatures corresponding to core formation compared to 125 chondrites. Error bars on each δ^{30} Si_{BE} point correspond to variation due to metal-silicate 126 separation temperature (at pressure given on top scale), O composition of core, and alter-127 native Δ^{30} Si(T) coefficients. Colored bands show reported range of δ^{30} Si of various 128 chondrite classes (Armytage et al., 2011; Fitoussi et al., 2009; Fitoussi and Bourdon, 129 2012). Within the uncertainty of the formation conditions, δ^{30} Si_{BE} is similar to BSE. 130 Core-hosted SiO₂ calculated using different fractionation factors (\bullet - Shahar et al. (2011); 131 \bigcirc - Hin et al. (2014); grey band is $\pm 1\sigma$ of \bullet points) is under most conditions lighter than 132 chondritic meteorites and BSE, making it a useful diagnostic of core-mantle mass trans-133 fer.

Magma ocean T range (K)

134 estimate insensitive to $E_{\text{Fe-FeO}}$, $E_{\text{Fe-Si}}$ and the slope of the SiO₂ loss line define the upper 135 limit of Si saturation. The intersection of the SiO₂ loss line with the saturation contours 136 therefore provides the 3.1 wt% upper bound that we use. These limits set core Si content 137 to be 0.4-3.1 wt%. In turn, the limits place f_c in the range $0.91\% \le f_c \le 6.62\%$. By 138 equation (1), the core mass fractions lead to a δ^{30} Si fractionation range of 139 -0.37 $\le \delta^{30}$ Si_{BE} ≤ -0.30 ‰. Alternative coefficients for Δ^{30} Si(*T*) (Hin et al., 2014; 140 Ziegler et al., 2010) lead to $-0.33 \le \delta^{30}$ Si_{BE} ≤ -0.29 ‰. When compared with the val-141 ues for bulk silicate Earth -0.29 ± 0.02 , we find that the ranges largely overlap. This has 142 implications for the Earth's source materials and formation conditions (Zambardi et al., 143 2013; Hin et al., 2014; Dauphas et al., 2015), but we do not discuss them here.

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DISCUSSION

Figure 2 shows the ³⁰Si fractionation predicted by the Si+O saturation model. The uncertable tainties include the effective pressure of segregation, alternative temperatures of segregatable tainties include the effective pressure of segregation, alternative temperatures of segregatable tainties include the effective pressure of segregation, alternative temperatures of segregatable tainties include the effective pressure of segregation, alternative temperatures of segregatable tainties include the effective pressure of segregation, alternative temperatures of segregatable tainties include the effective pressure of segregation, alternative temperatures of segregatable tainties include the effective pressure of segregation, alternative temperatures of segregatable tainties include the effective pressure of segregation, alternative temperatures of segregatable tainties include the effective pressure of segregation, alternative temperatures associated with core formation.

Depending on the pressure and the solidus temperature model chosen, the Si isotope 151 composition of the core is $-1.12\pm0.13 \% (1\sigma)$, significantly different from the BSE value 152 of $-0.29\pm0.02 \% (2\sigma)$ and all of the chondritic meteorite classes (Figure 2), except at the 153 highest differentiation pressure using the Hin et al. (2014) coefficient. Diamonds are 154 known to trap SiO₂, including those thought to come from the lower mantle (Stachel et 155 al., 2000; Kaminsky, 2012). Some SiO₂ inclusions are likely to be due to deep ¹⁵⁶ subduction of eclogite (Walter et al., 2011) and would have values close to bulk silicate ¹⁵⁷ Earth. The signature of a core source for SiO₂ would be an absence of aluminous phases ¹⁵⁸ and a light δ^{30} Si content of the SiO₂.

The values we report here represent lower bounds on the anticipated δ^{30} Si of coreloo hosted SiO₂ because we are also neglecting any fractionation of ³⁰Si during SiO₂ crystallilization in the core itself, which will shift δ^{30} Si to less negative values. At the end of aclie cretion, the core is likely to be hotter than the mantle and will cool rapidly (Lebrun et al., lie 2013). The SiO₂ crystallization required to run the Earth's dynamo corresponds to a lie cooling rate of 50-100 K/Gyr (Hirose et al., 2017), which is significantly lower than lie ~1000 K/Gyr rates expected in early Earth conditions. Hence the bulk of SiO₂ crystallie lized from the core will have separated under correspondingly higher temperatures than lie factor used, the shift will be within the 1 σ uncertainty depicted in Figure 2 if crystallizalie tion initially occurred at 7000-8000 K. Firmer predictions of uncertainty from this source 170 require more detailed models of early Earth evolution focused on the end-stages of accre-171 tion and evolution past the magma ocean era.

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IMPLICATIONS

173 The Si+O saturation model developed by Hirose et al. (2017) provides a way to estimate 174 the core's bulk Si content, and to estimate the core and bulk earth ³⁰Si fractionation. We 175 find that the core's δ^{30} Si is between -0.92 and -1.36 ‰, depending on the conditions of 176 core formation. The bulk Earth δ^{30} Si is $-0.37 \le \delta^{30}$ Si_{BE} ≤ -0.30 ‰, which overlaps 177 bulk silicate Earth and is marginally heavier than the ordinary and carbonaceous 178 chondrite groups. We also predict that any SiO₂ that originated in the core and was later 179 trapped as an inclusion in diamond should have significantly lighter δ^{30} Si of around -1.12 180 ± 0.13 ‰.

We also described a way to calculate the core's Si content using joint solubility constraints for Si and O in metal that may prove useful to draw up budgets for other stable isotope systems.

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ACKNOWLEDGEMENTS

185 Funding partly provided by MEXT/JSPS KAKENHI Grant numbers 15H05832 and 16H06285. Figures186 and calculations made using R (R Core Team, 2017).

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REFERENCES

188 Armytage, R. M. G., Georg, R. B., Savage, P. S., Williams, H. M. and Halliday, A. N.

- (2011) Silicon isotopes in meteorites and planetary core formation. Geochimica et
- 190 Cosmochimica Acta, 75, 3662–3676.
- 191 Badro, J., Siebert, J. and Nimmo, F. (2016) An early geodynamo driven by exsolution of
- mantle components from Earth's core. Nature, 536, 326–328.
- 193 Dauphas, N., Potraisson, F., Burkhardt, C., Kobayashi, H. and Kurosawa, K. (2015) Plan-
- etary and meteoritic Mg/Si and δ^{30} Si variations inherited from solar nebula chemistry.
- 195 Earth and Planetary Science Letters, 427, 236–248.
- 196 Fiquet, G., Auzende, A. L., Siebert, J., Corgne, A., Bureau, H., Ozawa, H. and Garbarino,
- G. (2010) Melting of peridotite to 140 Gigapascals. Science, 329, 1516–1518.
- 198 Fischer, R. A., Campbell, A. J., Reaman, D. M., Miller, N. A., Heinz, D. L., Dera, P. and
- 199 Prakapenka, V. B. (2013) Phase relations in the Fe-FeSi system at high pressures and

- temperatures. Earth and Planetary Science Letters, 373, 54–64.
- 201 Fischer, R. A., Nakajima, Y., Campbell, A. J., Frost, D. J., Harries, D., Langenhorst, F.,
- ²⁰² Miyajima, N., Pollok, K. and Rubie, D. C. (2015) High pressure metal-silicate parti-
- tioning of Ni, Co, V, Cr, Si, and O. Geochimica et Cosmochimica Acta, 167, 177–194.
- 204 Fitoussi, C., Bourdon, B., Kleine, T., Oberli, F. and Reynolds, B. C. (2009) Si isotope
- 205 systematics of meteorites and terrestrial peridotites: implications for Mg/Si fractiona-
- tion in the solar nebula and for Si in the Earth's core. Earth and Planetary Science Let-
- 207 ters, 287, 77–85.
- ²⁰⁸ Fitoussi, C. and Bourdon, B. (2012) Silicon isotope evidence against an enstatite chon-²⁰⁹ drite Earth. Science, 335, 1477–1480.
- 210 Georg, R. B., Halliday, A. N., Schauble, E. A. and Reynolds, B. C. (2007) Silicon in the
- 211 Earth's core. Nature, 447, 1102–1106.
- 212 Helffrich, G., Ballmer, M. D. and Hirose, K. (2018) Core-exsolved SiO₂ dispersal in the
- Earth's mantle. Journal of Geophysical Research, 122, doi:10.1002/2017JB014865.
- 214 Helffrich, G. (2017) A finite strain approach to thermal expansivity's pressure depen-
- dence. American Mineralogist, 102, 1690–1695.
- 216 Hin, R. C., Fitoussi, C., Schmidt, M. W. and Bourdon, B. (2014) Experimental determina-
- 217 tion of the Si isotope fractionation factor between liquid metal and liquid silicate.
- Earth and Planetary Science Letters, 387, 55–66.
- 219 Hirose, K., Morard, G., Sinmyo, R., Umemoto, K., Hernlund, J., Helffrich, G. and
- Labrosse, S. (2017) Crystallization of silicon dioxide and compositional evolution of
- the Earth's core. Nature, 543, 99–102.

- ²²² Kaminsky, F. (2012) Mineralogy of the lower mantle: A review of 'super-deep' mineral
 ²²³ inclusions in diamond. Earth-Science Rev., 110, 127–147.
- 224 Komabayashi, T. (2014) Thermodynamics of melting relations in the system Fe-FeO at
- high pressure: Implications for oxygen in the Earth's core. Journal of Geophysical Re-
- search, 119, 4164–4177.
- 227 Lebrun, T., Massol, H., Chassfière, E., Davaille, A., Marcq, E., Sarda, P., Leblanc, F. and
- ²²⁸ Brandeis, G. (2013) Thermal evolution of an early magma ocean in interaction with the
- 229 atmosphere. JGR-Planets, 118, 1155–1176.
- Li, J. and Agee, C. B. (1996) Geochemistry of mantle-core differentiation at high pressure. Nature, 381, 686–689.
- 232 McDonough, W. F. and Sun, S.-s. (1995) The composition of the Earth. Chem. Geol.,
 233 120, 223–253.
- 234 Morard, G., Andrault, G., Antonangeli, D., Nakajima, Y., Auzende, A. L., Boulard, E.,
- ²³⁵ Cervera, S., Clark, A., Lord, O. T., Siebert, J., Svitlyk, V., Garbarino, G. and Mezouar,
- ²³⁶ M (2017) Fe-FeO and Fe-Fe₃C melting relations at Earth's core-mantle boundary con-
- ditions: Implications for a volatile-rich or oxygen-rich core. Earth and Planetary Science Letters, 473, 94–103.
- 239 O'Neill, H. St.C., Canil, D. and Rubie, D. C. (1998) Oxide-metal equilibria to 2500°C
- and 25 GPa: Implications for core formation and the light component in the Earth's
- core. Journal of Geophysical Research, 103, 12,239–12,260.
- 242 O'Rourke, J. G. and Stevenson, D. J. (2016) Powering Earth's dynamo with magnesium
- precipitation from the core. Nature, 529, 387–389.

- 244 Ozawa, H., Hirose, K., Yonemitsu, K. and Ohishi, Y. (2016) High-pressure melting exper-
- iments on Fe-Si alloys and implications for silicon as a light element in the core. Earth
 and Planetary Science Letters, 456, 47–54.
- 247 Palme H. and C., O'Neil H. S. (2003) Cosmochemical estimates of mantle composition.
- In H. D. Holland, and K. K. Turekian, Eds., Treatise on Geochmistry, 2.
- 249 R Core Team (2017) R: A language and environment for statistical computing, , R Foun-
- dation for Statistical Computing (https://www.R-project.org/), Vienna, Austria.
- 251 Rubie, D. C., Frost, D. J., Mann, U., Asahara, Y., Nimmo, F., Tsuno, K., Kegler, P.,
- Holzheid, A. and Palme, P. (2011) Heterogeneous accretion, composition and core-
- ²⁵³ mantle differentiation of the Earth. Earth and Planetary Science Letters, 301, 31–42.
- 254 Shahar, A., Hillgren, V. J., Young, E. D., Fei, Y., Macris, C. A. and Deng, L. (2011) High-
- temperature Si isotope fractionation between iron metal and silicate. Geochimica et
- 256 Cosmochimica Acta, 75, 7688–2697.
- 257 Shahar, A., Schauble, E. A., Caracas, R., Gleason, A. E., Reagan, M. M., Xiao, Y., Shu, J.
- and Mao, W. (2016) Pressure-dependent isotopic composition of iron alloys. Science,
 352, 580–582.
- 260 Siebert, J., Badro, J., Antonangeli, D. and Ryerson, F. J. (2013) Terrestrial Accretion Un261 der Oxidizing Conditions. Science, 339, 1194–1197.
- 262 Siebert, J., Corgne, A. and Ryerson, F. J. (2011) Systematics of metal-silicate partitioning
 263 for many siderophile elements applied to Earth's core formation. Geochimica et Cos264 mochimica Acta, 75, 1451–1489.
- 265 Stachel, T., Harris, J. W., Brey, G. P. and Joswig, W. (2000) Kankan diamonds (Guinea)

- II: Lower mantle inclusion parageneses. Contributions to Mineralogy and Petrology,
 140, 16–27.
- 268 Tuff, J., Wood, B. J. and Wade, J. (2011) The effect of Si on metal-silicate partitioning of
- siderophile elements and implications for the conditions of core formation. Geochim-
- ²⁷⁰ ica et Cosmochimica Acta, 75, 673–690.
- 271 Wade, J. and Wood, B. J. (2005) Core formation and the oxidation state of the Earth.
- Earth and Planetary Science Letters, 236, 78–95.
- 273 Walter, M. J., Kohn, S. C., Araujo, D., Bulanova, G. P., Smith, C. B., Gaillou, E., Wang,
- J., Steele, A. and Shirey, S. B. (2011) Deep mantle cycling of oceanic crust: Evidence
- from diamonds and their mineral inclusions. Science, 334, 54–57.
- Wood, B. J., Walter, M. J. and Wade, J. (2006) Accretion and segregation of its core. Nature, 441, 825–833.
- Xu, F., Yamazaki, D., Sakamoto, N., Sun, W., Fei, H. and Yurimoto, H. (2017) Silicon
 and oxygen self-diffusion in stishovite: Implications for stability of SiO₂-rich seismic
 reflectors in the mid-mantle. Earth and Planetary Science Letters, 459, 332–330.
- 281 Zambardi, T., Poitrasson, F., Corgne, A., Méhut, M., Quitté, G. and Anand, M. (2013)
- 282 Silicon isotope variations in the inner solar system: Implications for planetary forma-
- tion, differentiation and composition. Geochimica et Cosmochimica Acta, 121, 67–83.
- 284 Ziegler, K., Young, E. D., Schauble, E. A. and Wasson, J. T. (2011) Metal-silicate silicon
- isotope fractionation in enstatite meteorites and constraints on Earth's core formation.
- Earth and Planetary Science Letters, 295, 487–496.