

1 **REVISION 2**

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3 **Craters of the Moon National Monument Basalts as Unshocked Compositional and**  
4 **Weathering Analogs for Martian Rocks and Meteorites.**

5

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## Keywords

16 Mars, Analog, Craters of the Moon, Basalt, Martian Meteorite, Gusev Crater, Weathering, Age  
17 correlation

18

## Abstract

19

20 The availability of terrestrial sites that are martian analogs allows researchers to  
21 investigate Mars using knowledge gained on Earth. Among the terrestrial analog sites for Mars is  
22 Craters of the Moon National Monument (COTM) in Idaho, USA. Craters of the Moon National  
23 Monument is home to over 60 basalt lava flows, many of which have been dated from 2,050 to  
24 18,340 years before present (y.b.p.). Following previous authors, we examined the chemistry and  
25 petrogenesis of COTM basalts compared to basaltic martian rocks, martian meteorites, and  
26 meteorite clasts, and then examined the results of chemical weathering of the basaltic flows.  
27 Results of our comparative chemical analysis suggest COTM basalts are generally more evolved  
28 than the martian materials, with a few notable exceptions. Several COTM flow basalts, including  
29 rocks of the >18,000 year old Kimama flow, have high FeO, TiO<sub>2</sub>, and P<sub>2</sub>O<sub>5</sub> contents similar to  
30 the Wishstone and Watchtower class rocks analyzed at Gusev Crater, Mars, by the Mars  
31 Exploration Rover *Spirit*. The youngest basalts of COTM, such as those of the Minidoka (3890  
32 y.b.p.) and Blue Dragon (2050 y.b.p.) flows have similarities in SiO<sub>2</sub>, alkali contents, and  
33 mineralogical norms with select clasts in meteorite Northwest Africa – NWA – 7034. These  
34 similarities over a range of flow ages therefore suggest that COTM basalts have the potential to  
35 shed important light on specific igneous processes occurring on Mars.

36 Many of the basaltic rocks measured by rovers on Mars are thought to have experienced  
37 chemical weathering during aqueous interactions; however, few basalt weathering rates exist for

38 terrestrial Mars-relevant field environments to help interpret these processes. COTM, which has  
39 important similarities to some martian rocks discussed above, also represents a basalt flow  
40 chronosequence, and therefore allows for the investigation of basalt weathering as a function of  
41 time. We measured the depth of developed porosity in a suite of basalt flows ranging from 2,050  
42 to 18,340 y.b.p., and compared field weathering relationships at COTM to weathering rinds  
43 developed on the Gusev Crater martian rocks Humphrey, Champagne, Mazatzal, and Wooly  
44 Patch. Our results indicate that depths of incipient weathering in COTM rocks increase with time  
45 at a rate of  $2.32 \times 10^{-2}$  to  $3.04 \times 10^{-2} \mu\text{m y}^{-1}$ , which is comparable to other terrestrial advance  
46 rates. Interestingly, this rate also indicates that chemical weathering strongly outpaces physical  
47 weathering even in this arid to semi arid environment. Weathering primarily of the matrix glass  
48 indicates that glass may be functioning as the profile-controlling mineral, which may have  
49 implications for chemical weathering in glass-rich rocks on Mars. Weathering rates of glass and  
50 other minerals can also help constrain the conditions (pH, temperature) of alteration on Mars. Of  
51 the altered martian rocks we compared to COTM (Humphrey, Champagne, Mazatzal, and Wooly  
52 Patch), altered surfaces of Mazatzal rock at Gusev Crater show the most similarities to weathered  
53 surfaces at COTM. Comparisons of chemical weathering in COTM basalts with altered surfaces  
54 of rocks in Gusev Crater, Mars, indicate Gusev Crater martian rocks have undergone  
55 significantly more aqueous alteration than that experienced by basaltic flows at COTM.

56

57

## Introduction

58

59 Since the 1960's, dozens of robotic missions have been sent to Mars in an effort to better  
60 understand the planet (e.g. Steinbacher et al., 1972; Levinthal et al., 1973; Hess et al., 1976;

61 Snyder and Moroz, 1992; Golombek et al., 1997; Saunders et al., 2004; Zurek and Smrekar,  
62 2007; Brückner et al., 2008; Smith et al., 2008; Soderblom and Bell III, 2008; Grotzinger et al.,  
63 2012). Data from these missions and martian meteorites (e.g. Bogard and Johnson, 1983) have  
64 greatly reshaped our views of the planet's interior and surface processes, potential martian  
65 habitability, and the possibility of life. Robotic missions and meteorite samples, however, have  
66 limitations, and many questions concerning martian interior and surface processes remain  
67 difficult to address. Rovers and landers have only explored a fraction of the martian surface and  
68 by necessity have analytical limitations. Orbiter and flyby missions collect regional and planet-  
69 wide data, but are limited to types of data which can be collected from orbit. Martian meteorites  
70 currently represent our only physical samples of Mars on Earth. However, these meteorites lack  
71 locational context, are not representative of the general martian crust, and have all been shocked  
72 in the processes that made them meteorites in the first place. This shock has altered their  
73 mineralogy and textures in ways that may not be obvious (Stöffler et al., 1986; Gooding, 1992;  
74 Rubin, 1992; Walton and Herd, 2007; McSween et al., 2009; McSween, 2015; Ody et al., 2015;  
75 Adcock et al., 2017). The availability of an unshocked analog to martian meteorites and rocks  
76 could yield numerous insights into both interior and surface processes on Mars.

77 One of the tools employed to advance our understanding of Mars has been the use of  
78 terrestrial martian analog sites. Planetary analogs can potentially fill gaps in our understanding or  
79 can supply reasonable assumptions where direct data are lacking. Martian analogs have allowed  
80 the investigation of Mars-relevant geomorphic, geochemical, petrogenetic, and hydrologic  
81 processes, as well as past or present potential habitability, by studying places on Earth (e.g.  
82 Breed, 1977; Wynn-Williams and Edwards, 2000; Greeley and Fagents, 2001; Arcone et al.,  
83 2002; Greeley et al., 2002; Wierzchos et al., 2006; Amils et al., 2007; Richardson et al., 2012).

84           Among the terrestrial analog sites for Mars is Craters of the Moon National Monument  
85 (COTM) (Peck, 1974; Greeley and King, 1977; Klingelhöfer et al., 2004; Weren et al., 2004;  
86 Brady et al., 2005; Heggy et al., 2006; McHenry, 2008; Usui et al., 2008; Richardson et al.,  
87 2012; Phillips-Lander et al., 2017), a 1600 km<sup>2</sup>, arid to semi-arid basalt lava field located on the  
88 Snake River Plain of Idaho in the continental United States (Figure 1)(Kuntz et al., 1992). The  
89 area is the locale of 60, relatively high-phosphorus (up to 2.6 wt% P<sub>2</sub>O<sub>5</sub>), alkaline a'a' and  
90 pahoehoe basalt lava flows spanning eight eruptive periods and creating a volcanic  
91 chronosequence (Kuntz et al., 1986a; Vaughan, 2008; Vaughan et al., 2011). COTM lava flows  
92 generally consist of a glassy matrix with 0-30% olivine, 70-100% plagioclase, and minor  
93 abundances of pyroxenes (Putirka et al., 2009). A number of mineralogic, geologic, and  
94 landscape features at COTM, such as caves (Peck, 1974; Richardson et al., 2013; Phillips-Lander  
95 et al., 2017), the general basaltic terrain (Greeley and King, 1977; Heggy et al., 2006), igneous  
96 compositional and geomorphic analogs (Weren et al., 2004; Usui et al., 2008), or secondary  
97 minerals similar to some which may be on Mars (Peck, 1974; Klingelhöfer et al., 2004;  
98 Richardson et al., 2012), have made the region a useful martian analog in past studies (Peck,  
99 1974; Greeley and King, 1977; Klingelhöfer et al., 2004; Weren et al., 2004; Brady et al., 2005;  
100 Heggy et al., 2006; McHenry, 2008; Usui et al., 2008; Richardson et al., 2012; Phillips-Lander et  
101 al., 2017).

102           In particular, Usui et al., (2008), suggested that the alkali basalts of COTM may be good  
103 analogs for the high phosphorus Wishstone and Watchtower class rocks at Gusev Crater, Mars.  
104 Usui et al., (2008), however, did not specify which specific COTM lavas might be good analogs.  
105 Many of the alkaline basalt flows at COTM possess elevated concentrations of FeO, TiO<sub>2</sub>, and  
106 P<sub>2</sub>O<sub>5</sub> (Stearns, 1928; Leeman et al., 1976; Kuntz et al., 1992; Stout et al., 1994) similar to Mars,

107 which is generally elevated in FeO (2x higher than Earth) and P<sub>2</sub>O<sub>5</sub> (10x higher than Earth)  
108 (Wanke and Dreibus, 1988; Taylor, 2013). Heterogeneous mantle sources, crustal contamination,  
109 or a combination of both have been proposed as possible explanations for the unusual chemistry  
110 at COTM (Kuntz et al., 1992; Stout et al., 1994; Reid, 1995; Usui et al., 2008), but the origins of  
111 the high-phosphorus basalts at COTM are not conclusively understood (Usui et al., 2008).  
112 Nevertheless, the high content in phosphorus and iron of some of the basalts at COTM suggests  
113 they are good compositional analogs for high-phosphorus rocks at Gusev Crater, Mars.

114 COTM basalts are also of interest to study surface processes such as chemical  
115 weathering. COTM represents an incipiently weathered basalt flow chronosequence with ages  
116 ranging between 2,050 and 18,340 y.b.p. (Vaughan, 2008) in an arid to semi-arid environment  
117 that, while wetter and warmer than present day Mars, may be analogous to a warmer and wetter  
118 martian environment of the distant past. Analysis of weathering processes occurring on these  
119 basalt flows may therefore shed light on the weathering processes that occurred on Mars. The  
120 Mars Exploration Rovers (MER) *Spirit* and *Opportunity* each carried an Alpha Particle X-ray  
121 Spectrometer (APXS) for elemental analyses (Rieder et al., 2003) and a Rock Abrasion Tool  
122 (RAT) for grinding into rocks to reach a "fresh" interior surface for analysis (Gorevan et al.,  
123 2003). APXS elemental analyses show changes in chemistry between the surface and interiors of  
124 rocks, an indication that chemical weathering and mineral dissolution may have occurred,  
125 developing weathering rinds by dissolution of the parent rocks (Klingelhöfer, 2004; Hurowitz et  
126 al., 2006a; Fleischer et al., 2008). The comparison between APXS analyses of outer rock  
127 surfaces and unaltered interiors allows constraints to be placed on chemical weathering and  
128 weathering rinds on the martian rocks. Previous analyses of weathered rocks at Gusev Crater  
129 have indicated dissolution of olivine and phosphate minerals, as well as potentially alteration of

130 glass (Arvidson et al., 2004; Gellert et al., 2004; Hurowitz et al., 2006a; Squyres et al., 2006;  
131 Wang et al., 2006b; Hausrath et al., 2008b; Thomson et al., 2013; Adcock and Hausrath, 2015).  
132 Examination of weathering on the well-defined Craters of the Moon analog basalts may help  
133 shed light on the types of chemical weathering that have occurred on basaltic composition rocks  
134 on Mars.

135 In this study, therefore, we examine the composition of 26 COTM flow basalts and  
136 compare them to a range of martian meteorites, clasts within martian meteorites, and basaltic  
137 rocks analyzed on Mars. For comparisons, we use a sum of squares best-fit method, followed by  
138 elemental and calculated CIPW norm comparisons between rocks with good fits. We then select  
139 a set of COTM basalts spanning the full range of flow ages in the region (2,050 to 18,340 y.b.p)  
140 and investigate the relationship between the extent of aqueous weathering as indicated by  
141 changes in porosity and fracturing with depth over time to determine a field weathering  
142 relationship at COTM. COTM basalts are then compared to weathered rocks at Gusev Crater,  
143 Mars, to investigate aspects of chemical weathering on Mars.

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145

### **Study Area Background**

146

147 The COTM basalts are located in the northern part of the Great Rift volcanic zone and  
148 consist of lava flows, various cinder cones, and eruptive fissures (Kuntz et al., 1982). Russell  
149 (1903) is credited with the first general geologic studies of the COTM region during a regional  
150 geologic survey for the USGS spanning Cinder Butte, Idaho (southeastern Idaho) to Oregon  
151 along the Snake River Plain. Russell detailed much of the basaltic volcanism along the Snake  
152 River Plain, though the main objective of the study appeared to be documenting artesian aquifer,

153 petroleum, and gas reservoirs. The first detailed geologic description specific to COTM was  
154 conducted in 1923 by Stearns (1924; 1928) for the Idaho Bureau of Mines and Geology. Stearns  
155 (1928) described the biota, geology, geomorphology, and reported the first chemical analysis of a  
156 COTM basalt, although it was not spatially located or assigned to a specific lava flow. Since  
157 those initial reports, the chemistry and mineralogy of many COTM flows have been well  
158 documented (Leeman et al., 1976; Kuntz et al., 1985; Kuntz et al., 1992; Stout et al., 1994) and a  
159 number of flows at COTM have been carbon-14 dated with calibrated ages ranging from 2,050 to  
160 >18,000 y.b.p. (Table 1, Supplementary Table S1)(Kuntz et al., 1986b; Kuntz et al., 1992;  
161 Vaughan et al., 2011).

162 Pedogenesis at COTM has also been previously documented, including as part of a Ph.D.  
163 dissertation (Vaughan, 2008; Vaughan and McDaniel, 2009; Vaughan et al., 2011). The  
164 succession of different ages of flows at COTM, and soil associated with them, represents a  
165 chronosequence. Vaughan (2008) observed that volcanic glass plays a significant role at COTM;  
166 flow basalts in the area generally contain 5-40% by volume glass within groundmass, although  
167 the glass content can exceed 65% by volume (Stout et al., 1994). Vaughan (2008) found that the  
168 proportion of weathered basaltic glass within soils associated with flows increases with the flow  
169 age.

170 COTM receives 240 to 380 mm Mean Annual Precipitation (MAP) moving from south to  
171 north (Stearns, 1928; Vaughan, 2008; Kukachka, 2010; Vaughan et al., 2011) making it an arid  
172 to semi-arid environment. The Mean Annual Temperature (MAT) for the region ranges from 6 to  
173 9 °C depending on the specific location. Average monthly high temperatures range from -1.7 to  
174 29 °C and average lows span -12 to 11 °C with sub-freezing lows prevailing much of the year  
175 (Kukachka, 2010).



176 The relatively fresh flows and volcanic structures at COTM led to the naming of the area  
177 and the choice of the region as a lunar surface analog to train astronauts during the Apollo era  
178 (Owen, 2008). Subsequent missions to the Moon and inner planets showed the COTM region to  
179 have features analogous to multiple planetary bodies, especially Mars (Greeley and King, 1977).  
180 The extensive, relatively barren, basalt flows, flow fields, shield volcanoes, and features such as  
181 lava tubes, caves, and channels, present a landscape that has increasingly been useful as an  
182 analog for Mars (Greeley and King, 1977; Weren et al., 2004; Brady et al., 2005; Heggy et al.,  
183 2006; McHenry, 2008; Phillips-Lander et al., 2017). In addition, the discovery of secondary  
184 sulfate minerals in COTM caves, including jarosite, which has also been found on Mars,  
185 suggests COTM caves may also be analogous to past martian environments (Peck, 1974;  
186 Klingelhöfer et al., 2004; McHenry, 2008; Farrand et al., 2009; Richardson et al., 2012;  
187 Cavanagh et al., 2015b; Phillips-Lander et al., 2017).

188

## 189 **Methods**

190

### 191 **COTM and martian material selection and comparison approach**

192

193 Chemical analyses of basalts from 26 different COTM flows spanning all eight eruptive  
194 periods at COTM were sourced from the literature (Supplementary Table S1) (Leeman et al.,  
195 1976; Kuntz et al., 1985; Kuntz et al., 1992; Stout et al., 1994). Selection criteria included basalts  
196 with good locational information, and associated age data. Similar to Kuntz et al. (1982) and to  
197 simplify data handling, we averaged lava composition analyses from the same eruptive ages or  
198 groups (from H – oldest - to A – youngest) for use in comparisons (Table 2). We also included

199 "C-low" and "A-low" groups, which were flows from the A and C eruptive ages with lower MgO  
200 content than other basalts from the same eruptive period.

201 For comparison to the chemical analyses of the COTM basalts described above, 28  
202 representative martian meteorite bulk rock compositions were chosen spanning a range of  
203 martian meteorite types. These all display mafic to ultramafic composition, and include 5  
204 poikilitic shergottites (gabbros and peridotites), 8 basaltic shergottites, 7 olivine-phyric  
205 shergottites (picritic basalts), 7 nakhlites (clinopyroxenites), 1 chassignite (dunite), and the  
206 orthopyroxenite Allan Hills (ALH) 84001 (Supplementary Table S2) (Lodders, 1998; Dreibus et  
207 al., 2000; Rubin et al., 2000; Barrat et al., 2002; Jambon et al., 2002; Shirai and Ebihara, 2004;  
208 Gillet et al., 2005; Day et al., 2006; Ikeda et al., 2006; Anand et al., 2008; Lin et al., 2008;  
209 Treiman and Irving, 2008; Basu Sarbadhikari et al., 2009; McSween et al., 2009). The meteorites  
210 vary in age from 4.0 Ga to 165 Ma and originate from different localities on Mars. All of these  
211 meteorites fall within the basalt envelope in the Total Alkali-Silica (TAS) diagram (Figure 2).  
212 We note that nakhlites, chassignites, and poikilitic shergottites are mafic and ultramafic  
213 cumulates rather than basalts and as such would not normally appear on a TAS diagram.  
214 However, because these meteorites are considered in comparisons within this study, we have  
215 included them in Figure 2, similar to some previous studies (e.g. McSween et al., 2009;  
216 McSween, 2015). We also include 17 clasts from within meteorite Northwest Africa (NWA)  
217 7034 (Santos et al., 2015) (Supplementary Table S3). Unlike the other meteorites, which all  
218 display an evident igneous texture, NWA 7034 is a polymict breccia that consists of basaltic,  
219 trachy-andesitic, Fe-, Ti-, P-rich (FTP) clasts, as well as impact melt clasts (Udry et al., 2014;  
220 Santos et al., 2015). These different clasts likely originate from different sources and were  
221 formed by different igneous and impact processes (Santos et al., 2015). The bulk rock

222 composition of NWA 7034 was therefore not considered in this study, because, as a polymict  
223 breccia, it does not represent a true igneous rock.

224 In addition to martian meteorites, COTM chemical compositions were also compared to  
225 igneous compositions measured by *Spirit* (n = 93) using the APXS instrument at Gusev Crater  
226 (Supplementary Table S4) from sols 14 to 470 of the mission (Brückner et al., 2008). We also  
227 included Bounce Rock major element compositions measured by the *Opportunity* rover in  
228 Meridiani Planum; this surface rock is unique as it is the only rock analyzed on Mars that shows  
229 a shergottitic composition (Zipfel et al., 2011). Although not a meteorite, we included Bounce  
230 Rock in the appended martian meteorite table based on its shergottitic composition and  
231 connection with martian meteorites (Supplementary Table S2) (Zipfel et al., 2011). Dust on the  
232 surface of martian rocks is known to influence APXS analyses (e.g. Arvidson et al., 2006; Berger  
233 et al., 2016) and many of the rocks at Gusev Crater have also undergone chemical weathering  
234 and have altered surfaces as discussed above (Arvidson et al., 2006; Hurowitz et al., 2006a;  
235 Hurowitz et al., 2006b; McSween et al., 2006; Squyres et al., 2006; Hurowitz and McLennan,  
236 2007; Hausrath et al., 2008b; Adcock and Hausrath, 2015). Therefore, the chemical composition  
237 comparison with COTM focuses on the 11 rocks that were RAT-abraded, and thus have  
238 available analyses with the lowest degree of alteration or dust contamination (Supplementary  
239 Table S5) (Squyres et al., 2006; Brückner et al., 2008). These RAT-treated rocks comprised  
240 seven rock classes identified at Gusev Crater as the Adirondack, Clovis, Wishstone, Watchtower,  
241 Backstay, and Independence classes (Squyres et al., 2006; Brückner et al., 2008).

242 Finally, we compared the COTM lavas to the fine-grained alkaline rocks (n=3) measured  
243 by the *Curiosity* rover ChemCam instrument in the early Hesperian-aged Gale Crater (Sautter et  
244 al., 2015) (Supplementary Table S6). These rocks are unique among martian surface rocks as

245 they are both enriched in SiO<sub>2</sub> and alkalis. We chose the fine grained “Group 2” class described  
246 as aphyric effusive volcanic rocks as they are texturally more similar to COTM lavas (Putirka et  
247 al., 2009; Sautter et al., 2015).

248 To quantitatively compare COTM lava compositions to martian igneous compositions  
249 from martian meteorites and measurements from Gusev Crater, Meridiani Planum, and Gale  
250 Crater, we used a sum of squares best-fit method, where differences between the mean COTM  
251 basalt major element compositions ( $i$  in Eq. 1 below, where  $i$  = SiO<sub>2</sub>, TiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, FeO<sub>T</sub>, MnO,  
252 MgO, CaO, Na<sub>2</sub>O, K<sub>2</sub>O, and P<sub>2</sub>O<sub>5</sub>) for a given eruptive period ( $x$ ), and analytical values from  
253 martian igneous rocks ( $x'$ ) are squared and summed (after Wheeler and Cook, 2006):

254

$$255 \quad \text{best fit index} = \sum_i (x_i - x'_i)^2 \quad [\text{Eq. 1}]$$

256

257 We obtained a best-fit index for each composition comparison, where a low index represents  
258 similar compositions and a high index indicates different compositions (Supplementary Tables  
259 S2 to S6). We then calculated and compared the normative mineralogies for martian and COTM  
260 compositions with low indices indicating similar compositions using the CIPW norm  
261 (Supplementary Table S7).

262

### 263 **Analysis of weathering through time of high-phosphorus COTM basalts**

264

265 **Materials and sample preparation.** To investigate weathering over time at COTM, six  
266 of the 23 compiled COTM basalt flows were selected and sampled based on previous chemical  
267 characterizations (Kuntz et al., 1992), location confidence, the presence of associated available

268 carbon-14 ages (Kuntz et al., 1992), elevated P<sub>2</sub>O<sub>5</sub> concentrations (based on the generally higher  
269 phosphorus concentrations measured on Mars), and with an effort to span the entire eruptive age  
270 range at COTM (Table 1, Supplementary Table S1). Exposed surface lava flow samples were  
271 collected in order to maximize the probability of collecting in place samples that had been  
272 exposed to weathering for the entire age of the flow. Locations of the sample sites covered a  
273 geographically wide region of COTM to minimize any local climate effects (Figure 1). Published  
274 carbon-14 age data were calibrated using Calib 7.0.4 with InterCal 13 (Table 1) (Stuiver and  
275 Reimer, 1993).

276 From the six collected samples, representative pieces of basalt were removed and  
277 oriented in epoxy mounts so that subsequent thin sectioning would produce sections, which  
278 included exposed/weathered surfaces as well as a cross-section into the rock perpendicular to the  
279 weathered surface. Thin sections were then made with a 0.3 μm final polish using anhydrous  
280 cutting and polishing methods. Epoxy impregnation was also used in an effort to preserve  
281 weathered material and soluble minerals during sample preparation.

282 **Sample and image analyses.** Scanning Electron Microscopy (SEM) analyses were  
283 performed on prepared thin sections using a JEOL JSM 5600 (EMiL Facility, University of  
284 Nevada Las Vegas) at 20 kV, a 30-40 spot size setting, and a working distance of 20 mm in both  
285 Backscattered Electron (BSE) and Secondary Electron Imaging (SEI) modes. The spot size  
286 setting on the JEOL JSM 5600 is a unitless, relative scale. Analysis of acquired images indicated  
287 spot size remained below 200nm and the estimated theoretical minimum beam diameter was 30-  
288 45nm (based on Goldstein et al., 1992). Analysis focused on characterizing the texture of the  
289 rocks, as well as image collection and measurement of any development of apparent alteration  
290 rinds and weathering-induced porosity or fracturing. Energy Dispersive Spectroscopy (EDS)

291 analyses were carried out using an Oxford Instruments detector and Oxford INCA software with  
292 semi-quantitative capabilities to analyze qualitative mineral major element compositions, when  
293 needed to confirm general mineralogy and phases of interest.

294         Subsequent off-line image analysis was performed using *Adobe Photoshop CS6* software.  
295 Thicknesses of apparent weathering rinds were measured and relative changes in porosity and  
296 fracturing with depth were investigated using backscattered electron (BSE) image mosaics. To  
297 construct the mosaics, overlapping images (typically 7) were collected at 250x magnification  
298 from the weathered surface of the samples into unweathered core material to a depth of at least  
299 1200  $\mu\text{m}$  into the sample (well into interior parent material that showed no surface effects).  
300 Location selection was generally random within representative weathered areas with the  
301 exception that the area was first inspected to ensure the mosaic would not intersect large cracks,  
302 fractures, or vesicles. Mosaics were then manually assembled in *Adobe Photoshop CS6* to avoid  
303 any distortions added by automation software. Plots of porosity with depth were constructed  
304 from the mosaics by first aligning, rotating, and cropping the mosaic images such that 10  $\mu\text{m}$   
305 thick "slices" of equal areas of the sample image parallel to the weathered sample surface could  
306 be consecutively isolated, contrast enhanced, and processed to determine the amount of open  
307 micro-porosity as an area % value (example Supplementary Figure S1). This high contrast  
308 porosity mapping method is similar to methods previously used to measure porosity development  
309 with weathering in basalts (Gordon and Brady, 2002). The resulting values were recorded and  
310 then plotted against depth (Supplementary Figure S2 and Table S8). Macro-pores ( $>100 \mu\text{m}$  on a  
311 side) were excluded to avoid counting large vesicles as developed porosity. The surface porosity  
312 was considered to be the average of the top 100  $\mu\text{m}$  of the sample. The average porosity of the  
313 interior was based on the deepest 100  $\mu\text{m}$  of the sample (1100  $\mu\text{m}$  to 1200  $\mu\text{m}$  interval).

314

315

## Results

316

### 317 **Terrestrial and martian composition comparisons**

318

319 The different COTM rock compositions vary from tephrite to trachyte (Figure 2, Table 2,  
320 Supplementary Table S1) and eruptive group chemistries range between 45.2-61.4 wt% SiO<sub>2</sub>,  
321 0.4-4.5 wt% MgO, 3.3-8.1 wt% CaO, 12.9-14.9 wt% Al<sub>2</sub>O<sub>3</sub>, 16.3-19.2 wt.% FeO, and 0.2-2.6  
322 wt% P<sub>2</sub>O<sub>5</sub>. The different eruptive periods become more evolved with time, corresponding to an  
323 increase in SiO<sub>2</sub>, alkalis (Na<sub>2</sub>O+K<sub>2</sub>O), Al<sub>2</sub>O<sub>3</sub>, and FeO, and a decrease in CaO and P<sub>2</sub>O<sub>5</sub> (Figures  
324 2 and 3). The major element compositions of COTM eruptive groups therefore show alkaline  
325 evolutionary trends (Figure 2). The different lavas contain 50-55% normative plagioclase, 13-  
326 22% normative orthoclase, 0-18% normative hypersthene, and 0-15% normative olivine. The  
327 lava groups C, C-low, and A-low are also quartz-normative. In contrast to COTM rock  
328 compositions, most of the martian igneous compositions are located in the basaltic field in the  
329 TAS diagram (Figure 2). Despite this difference, some compositional comparisons between  
330 rocks of specific COTM eruptive periods and martian igneous materials produce good best-fits  
331 (Table 3).

332 As observed in our best-fit calculations and elemental diagrams, martian meteorites and  
333 COTM compositions differ significantly in major element chemistry (Figures 2 and 3,  
334 Supplementary Table S2). Martian meteorites are generally enriched in MgO and FeO and  
335 depleted in Al<sub>2</sub>O<sub>3</sub> and alkalis compared to COTM rocks (Figures 2 and 3). The basaltic  
336 shergottite compositions yield the closest fits when compared to COTM basalts, mainly with

337 older COTM flows of eruptive periods F and E. However, like the other martian meteorites,  
338 these meteorites also show lower alkali, higher  $\text{FeO}_T$  and MgO contents, and generally different  
339 major element chemistries from the COTM rocks (Figures 2 and 3, Supplementary Table S2).

340 The clasts within NWA 7034 are in general, like most martian materials, alkali poor  
341 compared to COTM rocks (Figure 2). NWA 7034 clasts are also generally MgO rich and  $\text{Al}_2\text{O}_3$   
342 poor compared to COTM materials (Figure 3). However, a number of the NWA 7034 clasts (e.g.  
343 clasts 6, 56, 66, 70 and 74F) show a closer fit to COTM lavas than the other martian meteorites  
344 (Figures 2 and 3; Supplementary Table S3). Normative mineralogies of these clasts are the most  
345 similar to COTM lavas A and B, although they contain higher normative hypersthene  
346 (Supplementary Table S7). Clasts 77, 31, FTP 15, and FTP 64 have somewhat poorer fits than  
347 those discussed above, but do have  $\text{SiO}_2$  and MgO contents comparable to some COTM basalts  
348 (Supplementary Table S3). Clast FTP 15 is also similar in  $\text{Al}_2\text{O}_3$ , FeO, and CaO to COTM  
349 basalts, though very high in  $\text{P}_2\text{O}_5$  relative to COTM rocks (8.65 wt% versus 2.64 wt%).

350 COTM basalts differ significantly in chemical composition when compared to the Gusev  
351 igneous rocks except for the RAT-treated Gusev Crater rocks Champagne and Wishstone (both  
352 Wishstone class rocks) and Watchtower. This result is illustrated in the TAS diagram as well as  
353 other comparison plots where Wishstone class rocks fall within the groupings of COTM basalts  
354 (Figures 2 and 3). The best-fit calculations and normative mineralogies of COTM lavas of  
355 episode H are overall most similar to these Gusev Crater rocks (Supplementary Tables S4, S5  
356 and S7). However, FeO in the Gusev rocks is generally slightly lower (12.2 wt.%, n=3, versus  
357 15.1 wt.%, n=23 in COTM lavas) and  $\text{P}_2\text{O}_5$  is higher in Wishstone class rocks (5.1 wt%, n=3 vs.  
358 2.64 wt.% in COTM lavas) (Figures 3A and 3B, Supplementary Table S5).



359 The alkaline “Group 2” rocks at Gale Crater are generally different from the COTM  
360 flows in major element composition due to their evolved nature and high SiO<sub>2</sub> and Na<sub>2</sub>O content  
361 (Supplementary Table S6). However, despite higher alkali and Al<sub>2</sub>O<sub>3</sub>, in the Group 2 rocks, both  
362 major element compositions and normative mineralogies (especially in Group 2 rock Becraft) are  
363 similar to COTM A-low and C-low lavas (Supplementary Tables S6 and S7).

364

### 365 **Weathering of high-phosphorus COTM basalts**

366

367 Observations of thin sections by SEM BSE imaging revealed textures that are generally  
368 vitrophyric and aphanitic (Figure 4). The grain size and texture of the matrix material varies  
369 between the different basalts, with samples from flows Minidoka and Blue Dragon possessing a  
370 glassy matrix (Figure 4A and B), and samples from Lava Point, Prong Horn, Sunset, and  
371 Kimama exhibiting a lower proportion of glass in the groundmass (Figure 4C through E). Silica-  
372 rich surface coatings, some showing laminations (e.g. Figure 4C and D, Figure 5), were observed  
373 on exposed surfaces of all samples. The coatings were up to 100 μm thick. These features are  
374 similar to coatings previously observed in Antarctica, northern Scandinavia, Svalbard,  
375 Karkevagge, and cooler semi-arid regions of Hawaii (e.g. the Ka'ū desert), and are thought to  
376 result from water-rock interactions in relatively cold and arid or semi-arid environments (Farr  
377 and Adams, 1984; Curtiss et al., 1985; Dixon et al., 2002; Hausrath et al., 2008c; Salvatore et al.,  
378 2013). The coatings were not continuous over significant distances and sharp breaks in the  
379 coatings suggested that in many places they had been removed by spalling (e.g. Figure 5).

380 The Blue Dragon sample, from the youngest of the flows sampled, displayed dark layers  
381 (in BSE imagery) in some places that were distinctly different from the coatings. These layers

382 were ~10  $\mu\text{m}$  thick, always into glassy matrix, and always at or near the exposed surface of the  
383 samples (Figure 6). The chemistry of the dark layers as measured by EDS compared to the glass  
384 did not appear significantly different.

385 Measurements of porosity from exposed surfaces into the interiors of the samples  
386 indicated that increases in fractures and porosity occur to greater depths than the thickness of the  
387 coatings (which were up to 100  $\mu\text{m}$  thick) (Figure 5). Surfaces of older rocks showed generally  
388 greater depths of porosity than younger rock surfaces (Supplementary Figure S1). Average  
389 porosities of exposed surface material ranged from 0.58 - 7.5% with an overall average porosity  
390 of 3.8%. The porosity of interior material varied from 0.08 % to 0.98 % with an average value of  
391 0.6%. While these values may seem low, the technique used here does not measure  
392 macroporosity and the values are within the range of porosities of basalts measured by others  
393 (Freeze and Cherry, 1977; Sato et al., 1997; Rejeki et al., 2005).

394

395

## Discussion

396

### COTM lavas as analogs for martian igneous rocks.

398

399 The relative rarity and shock history of all martian meteorites make the potential  
400 availability of unshocked terrestrial analogs significant. Such analogs to a martian meteorite or  
401 rock could yield numerous insights into both martian interior and surface processes. In addition,  
402 the analogs could shed light on the implications of shock on martian meteorites, or meteorites in  
403 general. There are, however, overall crucial differences between COTM lavas and martian  
404 meteorite compositions, notably the alkali enrichments and lower MgO content of COTM lavas

405 (Figures 2 and 3). COTM lavas also have generally higher  $P_2O_5$  and  $TiO_2$  contents than martian  
406 meteorites. In addition, the mineralogies and textures (mostly plagioclase and olivine  
407 phenocrysts in a microcrystalline to glassy/vitrophyric matrix) observed in the COTM lavas have  
408 not been observed in martian meteorites (McSween, 2015). These differences are almost  
409 certainly due to the evolved nature of the COTM parent magmas. COTM lava compositional  
410 trends observed in Figures 2 and 3 represent fractional crystallization and assimilation of the  
411 crust (intermediate to felsic composition) by the parental magma, which likely originated from  
412 the upper mantle/lower crust (Kuntz et al., 1992; Putirka et al., 2009). However, the martian  
413 crust is mostly basaltic (McSween et al., 2009), and thus, assimilation of the crust on Mars would  
414 not generally result in such evolved lava compositions.

415 One potential exception to the contrasts of martian meteorite materials and COTM basalts  
416 are clasts within NWA 7034. A number of NWA 7034 clast compositions and normative  
417 mineralogies are closer to the COTM lavas than the martian meteorites (Figure 2). Most notably,  
418 clasts 56 and 74F show elevated alkali contents compared to most martian igneous compositions  
419 and are compositionally close to younger lavas at COTM (Supplementary Table S3). Clasts in  
420 NWA 7034 (and paired meteorites) are from a variety of sources and COTM basalts make good  
421 analogs on a case by case basis.

422 The alkaline rocks of Gale Crater show geochemical similarities to the most evolved  
423 COTM lavas. However, the rocks may not be petrogenetically similar. The high- $SiO_2$  and alkali  
424 signatures of Gale Crater rocks are likely due to extensive fractional crystallization for the  
425 martian rocks (Gazel and McSween, 2016) rather than crustal assimilation as is the case for the  
426 COTM rocks (Kuntz et al., 1992).

427 In contrast to the alkaline rocks of Gale Crater, COTM bulk-rock chemistries compared  
428 to RAT-treated APXS rock analyses at Gusev Crater indicate that most of these martian rocks  
429 come from much less evolved parent sources than the COTM rocks. Wishstone class rocks at  
430 Gusev Crater, however, show a good compositional fit with COTM lavas (Figures 2 and 3).  
431 These results are consistent with Usui et al. (2008), who suggested COTM rocks were potential  
432 analogs to Wishstone class rocks. Usui et al. (2008) did not specify a particular flow or eruptive  
433 period as the best analog for Wishstone class rocks, and analyses from multiple eruptive periods  
434 are good matches with the Wishstone rock class (Figures 2 and 3, Supplementary Table S5).  
435 Sum of squares calculations from analytical comparisons in this study indicate rocks of eruptive  
436 period H may be the closest in composition to martian Wishstone class rocks, suggesting  
437 Kimama basalt flow rocks as good chemical analogs (Supplementary Table S5). These lavas are  
438 among the oldest and least evolved at COTM and enriched in FeO and P<sub>2</sub>O<sub>5</sub> compared to the  
439 other eruptive periods. Specific flows from other eruptive periods are also close in chemistry,  
440 including Pronghorn (eruptive period F) and Lava Point basalts (eruptive period E). The high-Mg  
441 group basalts of eruptive period C are additional good matches with Wishstone class rocks. The  
442 similarities between COTM and Wishstone class rocks are due to the elevated FeO, TiO<sub>2</sub>, and  
443 P<sub>2</sub>O<sub>5</sub> contents, above what is typical in terrestrial and martian basalts, respectively. While Mars  
444 rocks in general are elevated in FeO and P<sub>2</sub>O<sub>5</sub> compared to terrestrial basalts (Taylor, 2013),  
445 Wishstone class rocks are even more so, and the enrichments in P<sub>2</sub>O<sub>5</sub> in COTM and Wishstone-  
446 class rocks are both likely due to metasomatization of their mantle source by CO<sub>2</sub>-rich fluids and  
447 the formation of xenocrystic phosphate minerals. This suggests a potential common history  
448 between the rocks (Ming et al., 2006; Usui et al., 2008).

449 Caution, nevertheless, should generally be applied when considering terrestrial analogs  
450 like COTM basalts. Wishstone class rocks, for instance, are often discussed as basaltic igneous  
451 rocks, tephrites, or mafic rocks, which have been metasomatized (McSween et al., 2006; Ming et  
452 al., 2006; e.g. Usui et al., 2008; Adcock and Hausrath, 2015). Consistent with this interpretation,  
453 CIPW norms have been used to investigate Wishstone class rocks by several authors (e.g.  
454 McSween et al., 2006; Ming et al., 2006; Ruff et al., 2006; McSween et al., 2008; Usui et al.,  
455 2008). However, angular textures in RAT abraded surfaces and the lighter color tone of  
456 Wishstone class rocks have also been interpreted as pyroclastic, and thus the rocks may be tuffs  
457 (Arvidson et al., 2006; Ming et al., 2006; Squyres et al., 2006). Further, their occurrence as only  
458 float rocks or as clasts in a geologic sub-units suggests they are impact excavated rocks from a  
459 deeper stratigraphic unit at Gusev Crater (Crumpler et al., 2011). Therefore they are potentially  
460 shock altered or even impact derived (Squyres et al., 2006). Without additional data,  
461 petrogenesis of these martian rocks is difficult to constrain. The fact that both rock types have  
462 likely experienced similar metasomatization, are igneous in origin, and have chemical  
463 similarities, however, suggests that the COTM rocks are useful analogs for Wishstone class  
464 rocks in applications where an exact petrogenetic match is not required.

465

#### 466 **Weathering of COTM basalts and implications for rock weathering at Gusev Crater, Mars**

467

468 In order to examine the effect of weathering with time in the basaltic chronosequence  
469 present at COTM, we compared the weathering present at exposed surfaces of samples of six  
470 COTM flows of different ages. Coatings were observed on the surfaces of each of the samples  
471 (e.g. Figures 4C, 4D, and 5). These silica-rich coatings were up to 100  $\mu\text{m}$  thick and consistent

472 with coatings seen on other terrestrial basalts (Farr and Adams, 1984; Curtiss et al., 1985; Dixon  
473 et al., 2002; Hausrath et al., 2008c; Salvatore et al., 2013). Boundaries between coatings and  
474 underlying surfaces were sharp. The origin of the coatings is thought to be tied to weathering  
475 processes and aqueous interactions, with the most commonly proposed formation mechanisms  
476 involving mineral dissolution and subsequent precipitation of the coatings onto the rock surface  
477 (Farr and Adams, 1984; Curtiss et al., 1985; Dixon et al., 2002; Hausrath et al., 2008c; Salvatore  
478 et al., 2013). This type of mechanism could explain the layered appearance and why the coatings  
479 do not seem influenced by underlying material. The exact source of the chemistry for coatings is  
480 not conclusively known and likely variable (Farr and Adams, 1984; Salvatore et al., 2013). In  
481 some cases the source of the chemistry of the coatings is thought to be the rock itself (e.g. Dixon  
482 et al., 2002), and in other cases the source is proposed to be external, such as from eolian  
483 deposition or solution transport (Farr and Adams, 1984; Curtiss et al., 1985). Such coatings are  
484 susceptible to spalling (Farr and Adams, 1984; Curtiss et al., 1985), and those on COTM samples  
485 were no exception; no correlation between age and thickness could be confirmed likely due to  
486 spalling off of the rock coatings.

487         The Blue Dragon sample, from the youngest of the flows sampled (Blue Dragon) known  
488 for its blue sheen, also possessed rock coatings like the other samples. However, it additionally  
489 displayed dark layers (in BSE imagery) in some places that were different than the coatings  
490 (Figure 6). These layers were only observed in contact with glassy matrix and at or near exposed  
491 surfaces, suggesting they are weathering related. However, they are chemically indistinguishable  
492 from the underlying glass by EDS. This, along with the lower backscatter signal observed in  
493 BSE imagery, may indicate they are of a lower density than the underlying glass material and

494 may be the product of hydration (e.g. Bindeman and Lowenstern, 2016). Blue Dragon is the only  
495 sample where these dark layers were observed.

496 Rock weathering generally results in the loss of the most rapidly dissolving or soluble  
497 minerals in the rock first (Goldich, 1938). Examination of weathered surfaces at COTM indicate  
498 that dissolution of glass resulting in increased porosity has occurred (Figure 5), but little  
499 alteration of crystalline materials is observed, including apatite or olivine. This dissolution of  
500 glass is consistent with observations by Vaughan (2008) who noted mainly glass weathering at  
501 COTM. This is also a similar scenario to previous results in arid environments, including those  
502 of Hausrath et al. (2008b; 2008c) who observed dissolution of glass matrix surrounding  
503 crystalline material (including olivine) within the rocks at Svalbard, Norway. Therefore, we  
504 interpreted enhanced porosity as resulting from glass dissolution due to chemical weathering,  
505 and measured changes in porosity with depth as a function of age of the lava flow, which was  
506 assumed to correlate with duration of exposure to weathering conditions.

507 The depth of weathering-enhanced porosity within samples was interpreted in three ways  
508 (Figure 7). First, BSE image mosaics that had been contrast enhanced to reveal porosity (see  
509 methods) were visually examined, and the apparent depth of enhanced porosity was judged by  
510 the viewer visually (Supplementary Figure S3). Similarly, the apparent depth of enhanced  
511 porosity was estimated visually from plotted profiles of porosity measured as described above in  
512 the methods (Supplementary Figure S1). Finally, the apparent depth of enhanced porosity was  
513 estimated as occurring when three consecutive 10  $\mu\text{m}$  slices gave measured porosities of less  
514 than the average porosity documented from the 700  $\mu\text{m}$  to 1200  $\mu\text{m}$  interval of the BSE mosaic  
515 (i.e. the porosity of the unweathered material) plus the standard deviation of the porosity  
516 measurements. The second of the three consecutive values was then selected as the depth of

517 weathering-induced porosity (Supplementary Table S8). Because the first two techniques are  
518 subject to visual interpretation, they were carried out four times by the same individual  
519 (Supplementary Table S9). We used an average of all four observations, which also allowed us  
520 to calculate a standard deviation for the repeated observations.

521 We then examined the relationship between lava flow age and depth of weathering-  
522 induced porosity in COTM material by plotting porosity depth observations for each thin  
523 section/sample against age and then analyzing by linear regression. Linear regression trends were  
524 forced through zero (i.e. no depth = no duration of weathering). Standard deviations (or other  
525 error) of observations were not given weight in the regression. Calculations were performed in  
526 *OriginPro 2017* software.

527 Linear regression fits of depth of developed porosity versus time in COTM rocks  
528 produced  $R^2$  values indicating a correlation of depth of developed porosity with age (Figure 7).  
529 The two visual approaches of determining the depth of weathering-induced porosity had  
530 exceptionally high  $R^2$  values (0.88 and 0.86) (Figure 7C & D). This could be, in part, an effect of  
531 biasing by the observer since the same person made all of the determinations. However, although  
532 the  $R^2$  value from the interpretation using numerical values for internal porosity is lower (0.59),  
533 it groups well with, and produces a similar slope to, the other interpretations. While the different  
534 techniques used to interpret depth of weathering porosity produced slightly different results  
535 (Figure 7), they all showed the same general correlation between depth of developed porosity  
536 and age (Figure 7A). Regression slopes were all similar and produced advance rates between  
537  $2.32 \times 10^{-2}$  and  $3.04 \times 10^{-2} \mu\text{m y}^{-1}$  (Figure 7 B-D).

538 These findings indicate that, even in an arid environment such as Craters of the Moon,  
539 chemical weathering outpaces physical weathering. Previous examinations of arid environments



540 have indicated the strong importance of physical erosion. Typical erosion rates for basaltic rocks  
541 in arid environments on Earth range from  $1 \times 10^{-1}$  to  $3 \times 10^1 \mu\text{m y}^{-1}$  (Greeley et al., 1984;  
542 Nishiizumi et al., 1986; Nishiizumi et al., 1991; Bierman, 1994; ). However, a broad range of  
543 factors control physical erosion rates, making them highly variable (Sharp, 1964; Sharp, 1980;  
544 Greeley and Iversen, 1987; Millot et al., 2002). Sharp (1964), for instance, observed almost no  
545 physical weathering of crystalline rocks in over a decade of observation at Coachella Valley,  
546 California, one of the most vigorous eolian abrasion environments known (Greeley and Iversen,  
547 1987). That study determined limited abrasive supply in the location to be the cause - a situation  
548 possible at COTM. Thus, the exact physical weathering contribution occurring at COTM is  
549 difficult to constrain.

550 The presence of surface coatings on samples and the correlation between age and depth  
551 of developed porosity suggests erosional weathering rates are low at COTM. Discontinuous  
552 coatings and sharp breaks in coatings suggest spalling off of these coatings. However, in contrast  
553 to Hausrath et al. (2008b), who proposed significant spalling of enhanced porosity due to glass  
554 dissolution in their rocks from Svalbard, we see no evidence of similar effects at COTM.  
555 Spalling at COTM appears limited to thin surface coatings. This may be due to the freeze-thaw  
556 process being important in arctic arid environments such as Svalbard (Yesavage et al., 2015).  
557 Steady state between chemical and physical erosion has often been assumed (Brantley and  
558 White, 2009). Results here showing increasing depth of developed porosity with time, crucially,  
559 indicate that such a steady state has not yet been developed after 18,000 years at COTM.

560 Weathering advance rates determined by linear regression analysis (Figure 7) produced  
561 rates of glass dissolution into the surface of our rocks of between  $2.32 \times 10^{-2}$  and  $3.04 \times 10^{-2} \mu\text{m}$   
562  $\text{y}^{-1}$ . Comparing our COTM weathering advance rates to weathering advance rates of matrix glass

563 dissolution in other arid environments, Hausrath et al., (2008b) estimated glass dissolution  
564 depths in basalts at Svalbard as 250  $\mu\text{m}$ , which, assuming a deglaciation of <80,000 years ago  
565 (Landvik et al., 1998; Yesavage et al., 2015) and noting the repeated spalling, would indicate a  
566 weathering advance rate of  $> 3 \times 10^{-3} \mu\text{m y}^{-1}$ . This is consistent with our measured COTM  
567 weathering advance rate of between  $2.32 \times 10^{-2}$  and  $3.04 \times 10^{-2} \mu\text{m y}^{-1}$ .

568 Comparing our COTM weathering advance rate to basalt chemical weathering advance  
569 rates in general, Navarre-Sitchler and Brantley (2007) compiled overall weathering advance rates  
570 for basalts based on rind thickness from several studies (Porter, 1975; Colman and Pierce, 1981;  
571 Oguchi and Matsukura, 1999; Sak et al., 2004). In that study, basalt weathering advance rates  
572 ranged from  $6.0 \times 10^{-3}$  to  $2.8 \times 10^{-1} \mu\text{m y}^{-1}$ . Thus advance rates determined here are within the  
573 range of previous studies for basalts in general.

574 Another crucial observation is the very small amount of porosity generated by glass  
575 dissolution in COTM rocks, and the fact that no dissolution of other phases was observed  
576 (Supplementary Table S8 and Figure S1). This suggests that the glass in these rocks is  
577 functioning as the profile-controlling mineral – the first mineral to dissolve and thus generate  
578 porosity allowing water transport and therefore dissolution of other minerals (Brantley and  
579 White, 2009). The presence of a profile-controlling mineral has been inferred from deep profiles  
580 (Jin et al., 2010), but is difficult to observe at this resolution except during very incipient  
581 weathering such as that occurring at COTM. This observation that glass is the profile-controlling  
582 mineral in COTM, similar to glass dissolution in Svalbard (Hausrath et al., 2008), also suggests  
583 that glass dissolution may be important in incipient weathering on Mars.

584 Evidence for the possible presence of glass has been found at multiple locations on Mars  
585 (Christensen et al., 2004; McSween et al., 2008; Bish et al., 2013; Cavanagh et al., 2015). XRD

586 refinements of soils at Rocknest, Gale Crater, for instance, suggest an amorphous component of  
587 ~27 to 45 wt. % best fit by basaltic glass and/or allophane (or similar) and potentially minor  
588 amounts of metal-sulfides or sulfates (Bish et al., 2013; Blake et al., 2013; Dehouck et al., 2014).  
589 Similar amounts were found in Confidence Hills, a sedimentary rock outcrop at Gale Crater  
590 (Cavanagh et al., 2015). At Meridiani Planum, the Mini-Thermal Emission Spectrometer  
591 (MiniTES) on *Opportunity* indicated glass in a number of rock outcrops (Christensen et al.,  
592 2004). The MiniTES on *Spirit* at Gusev Crater also indicated potentially significant glass  
593 components in rocks such as Barnhill class (McSween et al., 2008).

594         Dissolution rates of glass versus other rapidly dissolving minerals such as olivine and  
595 phosphate-bearing minerals have been previously used to infer both temperature and pH  
596 (Hausrath et al., 2008a; Hausrath et al., 2008b; Adcock and Hausrath, 2010; Hausrath and  
597 Tschauer, 2013; Yen et al., 2016). In this case, temperatures at COTM are relatively low (MAT  
598 6° - 9° C) and the region experiences sub-freezing lows for a significant part of the year  
599 (Kukachka, 2010). Vaughan (2008) measured soil pH at COTM and determined values of 3.4 -  
600 5.7, which are extremely acidic soil pHs for an arid soil formed on basaltic parent material. At  
601 these pH conditions, olivine dissolution would be faster than dissolution of a basaltic glass.  
602 However, these were soil pH values associated with COTM flows and the study noted vegetation  
603 had a strong control on pH. The COTM samples examined in this study were not buried in soil,  
604 but exposed, and thus not influenced heavily by vegetation. Under these conditions, pH during  
605 aqueous interactions would likely be higher than those measured at COTM. Hausrath (2008c)  
606 measured pH values of 8.5 and 6.75 in unvegetated weathering basaltic areas in Svalbard. At the  
607 low temperatures present at COTM, glass dissolution would be favored over olivine dissolution  
608 at pH values of ~7.5 or higher (Hausrath et al., 2008c; Schieber et al., 2016). At 25°C glass

609 dissolution is favored over apatite at a pH of ~6 or higher and the crossover pH decreases  
610 somewhat with decreasing temperature (Gislason and Oelkers, 2003; Palandri and Kharaka,  
611 2004; Wolff-Boenisch et al., 2004; Bandstra et al., 2008; Hausrath et al., 2008c; Adcock et al.,  
612 2013).

613 Weathered surfaces that may be similar to those observed at COTM have been previously  
614 studied at Gusev Crater using various approaches. The Specific Grind Energy (SGE) of RAT  
615 operations has been used as an indicator of relative weathering between rocks at Gusev Crater  
616 (Arvidson et al., 2004; Myrick et al., 2004; Squyres et al., 2006; Wang et al., 2006a; Herkenhoff  
617 et al., 2008; Thomson et al., 2013). Schroeder et al. (2006) used Mössbauer spectrometry to  
618 study alteration in the surfaces of Gusev Crater rocks based on which Fe phases were present.  
619 Selective depth Mössbauer using two gamma energies has also been applied to examine  
620 thickness of weathering rinds in Gusev rocks (Fleischer et al., 2008). The data that are most  
621 comparable to our observations of COTM, and on which we focus, are changes in chemistry with  
622 depth from the altered surface of the rock to the less altered RAT-abraded surface (Arvidson et  
623 al., 2006).

624 Changes in chemistry with depth in these RAT abraded Gusev Crater rocks suggest the  
625 development of a weathering profile, and have been used to interpret the dissolving minerals,  
626 pH, and duration of martian alteration (Hurowitz et al., 2006a; Ming et al., 2006; Wang et al.,  
627 2006b; Hurowitz and McLennan, 2007; Hausrath et al., 2008b; Ming et al., 2008; Adcock and  
628 Hausrath, 2015). For example, Mg and Fe depletion on the surface of Humphrey (Supplementary  
629 Table S10) has been used as an indicator of olivine dissolution under acidic conditions (Hurowitz  
630 et al., 2006a; Hurowitz and McLennan, 2007; Hausrath et al., 2008b) and used to constrain the  
631 duration of alteration (Hausrath et al., 2008b). Calcium and P depletions from the surface of

632 Watchtower and Wishstone have been inferred to result from either acidic (Hurowitz et al.,  
633 2006b; Ming et al., 2006) or near-neutral dissolution (Adcock and Hausrath, 2015) of a  
634 phosphate mineral or minerals (Supplementary Table S10). Wooly Patch is a moderately altered  
635 igneous outcrop of basaltic chemical composition, although SGE values suggest it is likely a  
636 competent tuff rather than basalt (Wang et al., 2006b; Ming et al., 2008). Glass dissolution may  
637 have occurred in the Wooly Patch outcrop of Gusev Crater, based on Al and Si depletions at the  
638 surface and evidence of phyllosilicates associated with the outcrop (Supplementary Table S10)  
639 (Wang et al., 2006b). This potential glass dissolution or alteration has been inferred to have  
640 occurred in a mildly (pH 4–6) acidic environment (Wang et al., 2006b).

641         Of the RAT treated rocks at Gusev Crater, Mazatzal (Adirondack class) weathering may  
642 be the most similar to weathering occurring at COTM. Mazatzal is thought to contain both glass  
643 and olivine (e.g. Hamilton and Ruff, 2012), and depletions of Ca and Al at the surface of  
644 Mazatzal relative to the interior (Supplementary Table S10) suggest glass may have been  
645 dissolving from this rock, and depletions in Mg and Fe suggest dissolution of olivine (Hurowitz  
646 and McLennan, 2007). Dissolution of both olivine and glass at near-freezing temperatures would  
647 indicate either near-neutral or highly acidic conditions (Hausrath et al., 2008b; Schieber et al.,  
648 2016).

649         Mazatzal also has a complex rock coating and evidence of salts in vugs and veins of the  
650 rock suggesting it was weathered, at least at some point, under brine conditions (Haskin et al.,  
651 2005). Brines have been shown to inhibit the dissolution of a range of minerals to a similar  
652 extent with decreasing activity of water (Pritchett et al., 2012; Dixon et al., 2015; Olsen et al.,  
653 2015; Steiner et al., 2016), and thus increase the duration of time necessary to achieve a given  
654 weathering depth. Therefore the multiple mm scale depth of the apparent weathering in Mazatzal

655 could suggest a history with a significantly long period of aqueous interactions compared to  
656 COTM.

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658

### **Implications**

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660 With the exception of the Moon, the exploration of other bodies in our solar system has  
661 been restricted to remote observations or robotic missions. The only physical samples of these  
662 other bodies (if any) are meteorites, which are both rare and have undergone shock  
663 metamorphism. Terrestrial planetary analogs, like Craters of the Moon National Monument,  
664 offer a way to study chemical, physical, and environmental aspects of other planets, such as  
665 Mars, in the absence of direct human visitation or sampling.

666 Though COTM lavas flows are generally more evolved than martian materials, an  
667 examination of COTM flow chemistries in this study indicates COTM rocks from select eruptive  
668 periods are potentially good compositional analogs based on chemistry and normative  
669 mineralogy of specific rocks on Mars, martian meteorites, or clasts within martian meteorites  
670 (Table 3). Among the best compositional comparisons are Wishstone class martian rocks found  
671 at Gusev Crater with older basalts of COTM, most specifically, rocks of the ~18,000 year old  
672 Kimama lava flow (eruptive period H). Clast 6 within NWA 7034 also shows some  
673 compositional similarities with Kimama lava flow rocks.

674 Weathering in the basaltic flows at COTM is also informative regarding martian  
675 weathering. The predominantly glass dissolution occurring at COTM sheds light on potential  
676 glass dissolution in multiple rocks on Mars, particularly the enhanced glass dissolution at very  
677 low pH and near-neutral pH conditions at Gusev Crater. The observed correlation of depth of

678 incipient weathering with flow age at COTM, even after ~18,000 years, is unexpected and  
679 indicates the dominance of chemical weathering over physical weathering at COTM. It also  
680 suggests relatively long weathering times for altered rocks at Gusev Crater, Mars, furthering our  
681 understanding of the martian environment.

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1142

1143 **FIGURE CAPTIONS**

1144

1145 **Figure 1. Craters of the Moon National Monument sampling locations and associated flow**  
1146 **ages.** Inset shows general location within the state of Idaho (red box). Enhanced USGS LANDSAT  
1147 8 color shortwave-infrared image, bands 7-4-3 displayed as R-G-B. Different dark colors in flow  
1148 fields represent different individual flows. Ages determined by radio-carbon dating, y.b.p. =  
1149 years before present.

1150

1151 **Figure 2. TAS ( $\text{Na}_2\text{O} + \text{K}_2\text{O}$  versus  $\text{SiO}_2$ ) diagram of COTM basalts, martian meteorites, and**  
1152 **martian surface compositions.** Plot shows most COTM rocks materials are generally higher in  
1153  $\text{Na}_2\text{O} + \text{K}_2\text{O}$  content compared to the martian materials. There are a few exceptions including  
1154 some high-phosphorus rocks analyzed at Gusev Crater (Watchtower and Wishstone class),  
1155 some of the clasts of NWA 7034, and alkali rocks analyzed at Gale Crater. Note trend in COTM  
1156 basalts of increasing alkalis with increasing  $\text{SiO}_2$ . The relationship generally follows the age of  
1157 COTM flows with lower  $\text{SiO}_2$  and alkali contents typical of the older flows.

1158 **Figure 3. A) FeO total versus MgO, B)  $\text{P}_2\text{O}_5$  versus MgO, C) CaO versus MgO, and D)  $\text{Al}_2\text{O}_3$**   
1159 **versus MgO (all in wt.%).** COTM rocks tend to be lower in MgO and higher in  $\text{Al}_2\text{O}_3$  than the  
1160 martian materials, although there are several exceptions (e.g. Wishstone class rocks). The linear  
1161 trend seen in COTM rocks in some plots is a result of the evolution of volcanism at COTM over  
1162 time. The higher MgO, CaO, and FeO contents are typical of the older flows at COTM.

1163 **Figure 4. BSE images of textures typical of the six flows sampled at exposed edges. A) Blue**  
1164 **Dragon, B) Minidoka, C) Lava Point, D) Prong Horn, E) Sunset, and F) Kimama.** Note differences

1165 in matrix material with Blue Dragon and Minidoka being mainly glass and the other samples  
1166 being more crystalline and less glassy. Coatings, like those of Lava Point (C) and Prong Horn (D)  
1167 were found on all samples with exposed surfaces, but were not continuous on the surfaces. Plg  
1168 = plagioclase, Ol = olivine, Mt = magnetite/ilmenite, Px = pyroxene, gls = glass, Ap = apatite. All  
1169 scale bars 100  $\mu\text{m}$ .

1170

1171 **Figure 5. BSE image of Kimama basalt in thin section.** Image shows development of porosity  
1172 (A) below discontinuous surface coating (B). The porosity appears to be from the dissolution of  
1173 glass in the groundmass. Phynocrysts show no obvious signs of dissolution. Pl = plagioclase, Olv  
1174 = olivine, Mt = magnetite/ilmenite, Px = pyroxene, gls = glass. Scale bar = 50  $\mu\text{m}$ .

1175 **Figure 6. BSE image of Blue Dragon basalt in thin section.** Note dark layer (indicated with an  
1176 arrow). The dark layer possessed very similar chemistry by EDS to the brighter glass material  
1177 beneath it. This may indicate the dark material is hydrated. Pl = plagioclase, Ol = olivine, gls =  
1178 glass, Ap = apatite. Scale bar is 50  $\mu\text{m}$ .

1179

1180 **Figure 7: Depth of developed porosity ( $\mu\text{m}$ ) versus age plots (years b.p.) for the three**  
1181 **approaches of interpretation. A)** All methods combined. **B)** Quantitative method  
1182 (Supplementary Figure S2). **C)** Judged visually from BSE image profile (Supplementary Figure  
1183 S3). **D)** Judged visually from plotted profile (Supplementary Figure S1). Regression intercepts  
1184 were tied to zero. No weighting given to associated errors for regression analysis. Note: all axes  
1185 are identical.

1186

1187 **TABLES**

1188

1189 **Table 1.** <sup>14</sup>C ages of flows at Craters of the Moon National Monument

Flow	Eruptive Period	Reference	Age (y.b.p.) <sup>a</sup>	Error +/- (years)	Calibrated Age (y.b.p.) <sup>a,b</sup>
Kimama	H	1	15100	160	18340
Lava Creek	G	1	12760	150	15180
Sunset	G	1	12010	150	13870
Carey	G	1	12000	150	13860
Heifer Reservoir	F	1	10670	150	12580
Bottleneck Lake	F	1	11000	100	12890
Pronghorn	F	1	10240	120	11970
Lava Point	F	1	7840	140	8690
Laidlaw Lake	E	1	7470	80	8280
Grassy Cone	E	1	7360	60	8180
Little Park	D	1	6500	60	7410
Carey Kupuka	D	1	6600	60	7500
Silent Cone	D	2	6500	nr	7410
Sentinel	C	2	6000	nr	6840
Fissure Butte	C	2	6000	nr	6840
Sheep Trail	C	2	6000	nr	6840
Sawtooth	C	2	6000	nr	6880
Indian Wells N.	C	1	6020	160	6880
Rangefire	B	1	4510	100	5150
Minidoka	B	1	3590	70	3890
Devils Cauldron	B	1	3660	60	3990
Deadhouse	A	1	4300	60	4880
Highway	A	2	2400	nr	2460
Serrate	A	2	2400	nr	2460
Big Craters (Green Dragon)	A	1	2400	300	2450
Trench Mortar Flat	A	1	2180	70	2190
Blue Dragon	A	1	2076	45	2050

1190 **References:** 1 = (Kuntz et al., 1986b) 2 = (Kuntz et al., 1992). <sup>a</sup>y.b.p. = years before present. <sup>b</sup>Ages calibrated using  
 1191 Calib 7.0.4 with InterCal 13 (Stuiver and Reimer, 1993).

1192 **Table 2.** Average analyses (wt.%) of basalts by eruptive period at Craters of the Moon National Monument

	<b>Eruptive Period</b>									
	<b>H</b>	<b>G</b>	<b>F</b>	<b>E</b>	<b>D</b>	<b>C</b>	<b>C low</b>	<b>B</b>	<b>A</b>	<b>A low</b>
<b>SiO<sub>2</sub></b>	45.24	46.79	46.05	46.45	50.93	46.19	56.29	48.57	50.71	61.40
<b>TiO<sub>2</sub></b>	3.28	3.12	3.27	3.43	2.28	3.25	1.50	2.95	2.66	0.81
<b>Al<sub>2</sub>O<sub>3</sub></b>	14.23	14.05	13.62	12.94	14.17	13.71	14.89	13.84	13.54	14.22
<b>FeO</b>	15.11	13.39	14.84	10.87	12.76	8.63	10.72	14.17	13.35	7.51
<b>MnO</b>	0.28	0.26	0.26	0.24	0.23	0.26	0.21	0.24	0.24	0.18
<b>MgO</b>	4.53	4.09	4.16	4.25	2.64	4.43	1.70	3.69	3.02	0.40
<b>CaO</b>	7.50	7.61	8.13	8.29	6.05	8.10	4.54	7.02	6.84	3.25
<b>Na<sub>2</sub>O</b>	3.50	3.81	3.71	3.31	4.02	3.42	4.05	3.90	3.58	4.10
<b>K<sub>2</sub>O</b>	1.75	1.96	1.88	1.74	2.46	1.77	3.22	1.98	2.24	4.28
<b>P<sub>2</sub>O<sub>5</sub></b>	2.64	2.21	2.32	2.23	1.40	2.38	0.75	2.00	1.56	0.22
<b>Total</b>	98.06	100.19	99.83	99.30	98.63	100.20	99.50	99.88	99.26	98.39
<b>n</b>	1	3	2	2	2	3	2	2	4	2

1193 n= number of flows average is based on. Based on analyses from Kuntz et al., (1992) and Kuntz et al., (1985)

1194

1195 Table 3. Best matches between martian materials and rocks from COTM eruptive periods

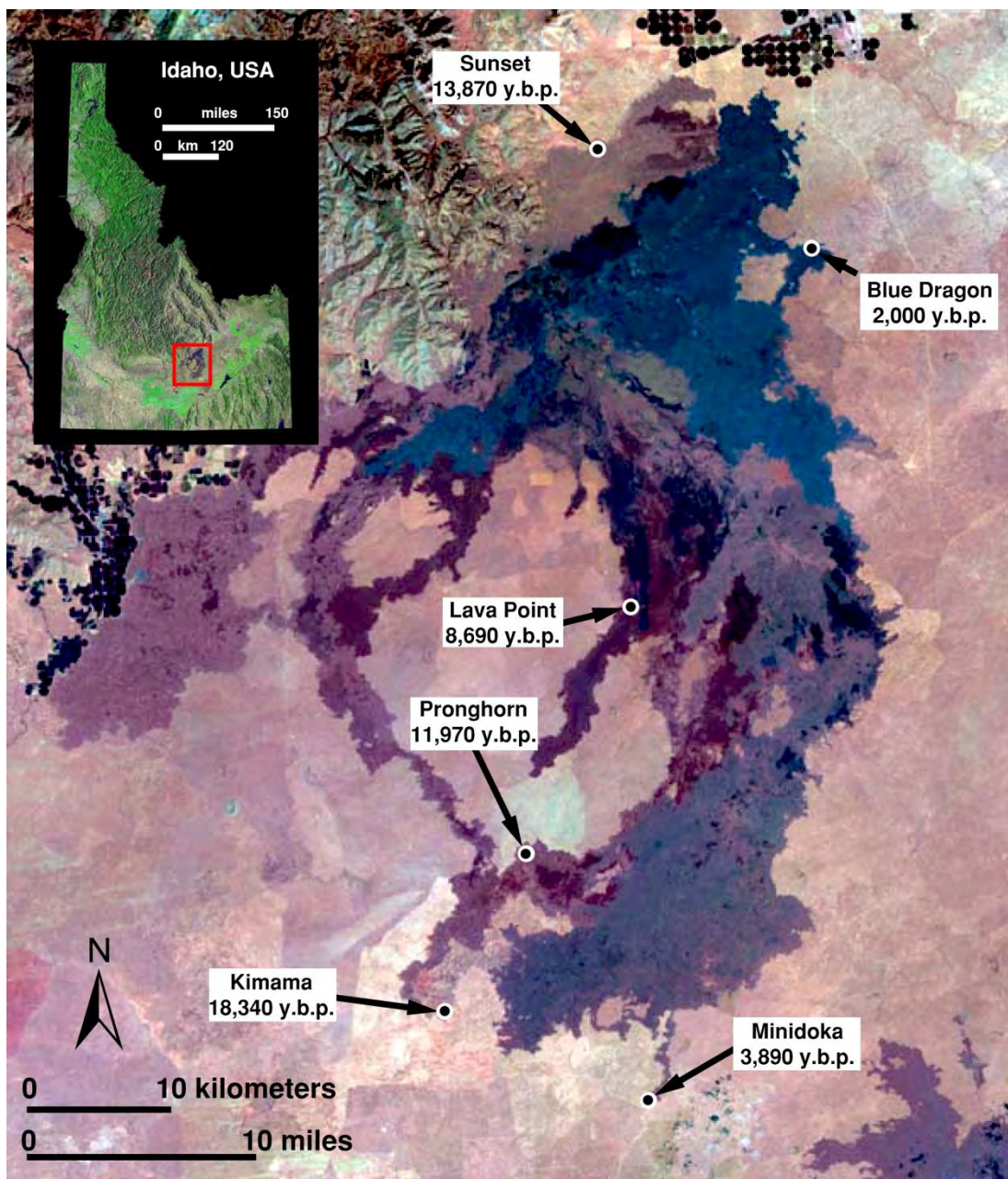
	<b>Martian Materials<sup>1</sup></b>	<b>Most Comparable COTM Eruptive Period<sup>3</sup></b>
<b>NWA 7034 Clasts</b>	6 <sup>2</sup>	Period H
	56	Period C low
	66	Period B
	70	Periods E, F, G, H
	74F	Period A
<b>Gusev Crater</b>	Champagne <sup>2</sup>	Period H
	Wishtone	Period E, G, H
	Watchtower	Period H
<b>Gale Crater</b>	Becraft <sup>2</sup>	Periods A low, C low
	Chakonipau	Period A low
	Sledgers	Period A low

1196 <sup>1</sup> Best matches only. Complete lists of material considered in this study are in Appendix Tables A2 to A7.

1197 <sup>2</sup> Produced best fit for group.

1198 <sup>3</sup> Based on chemical composition and normative mineralogy

1199 **FIGURES**



1200

1201 **Figure 1. Craters of the Moon National Monument sampling locations and associated flow**

1202 **ages.** Inset shows general location within the state of Idaho (red box). Enhanced USGS LANDSAT

1203 8 color shortwave-infrared image, bands 7-4-3 displayed as R-G-B. Different dark colors in flow

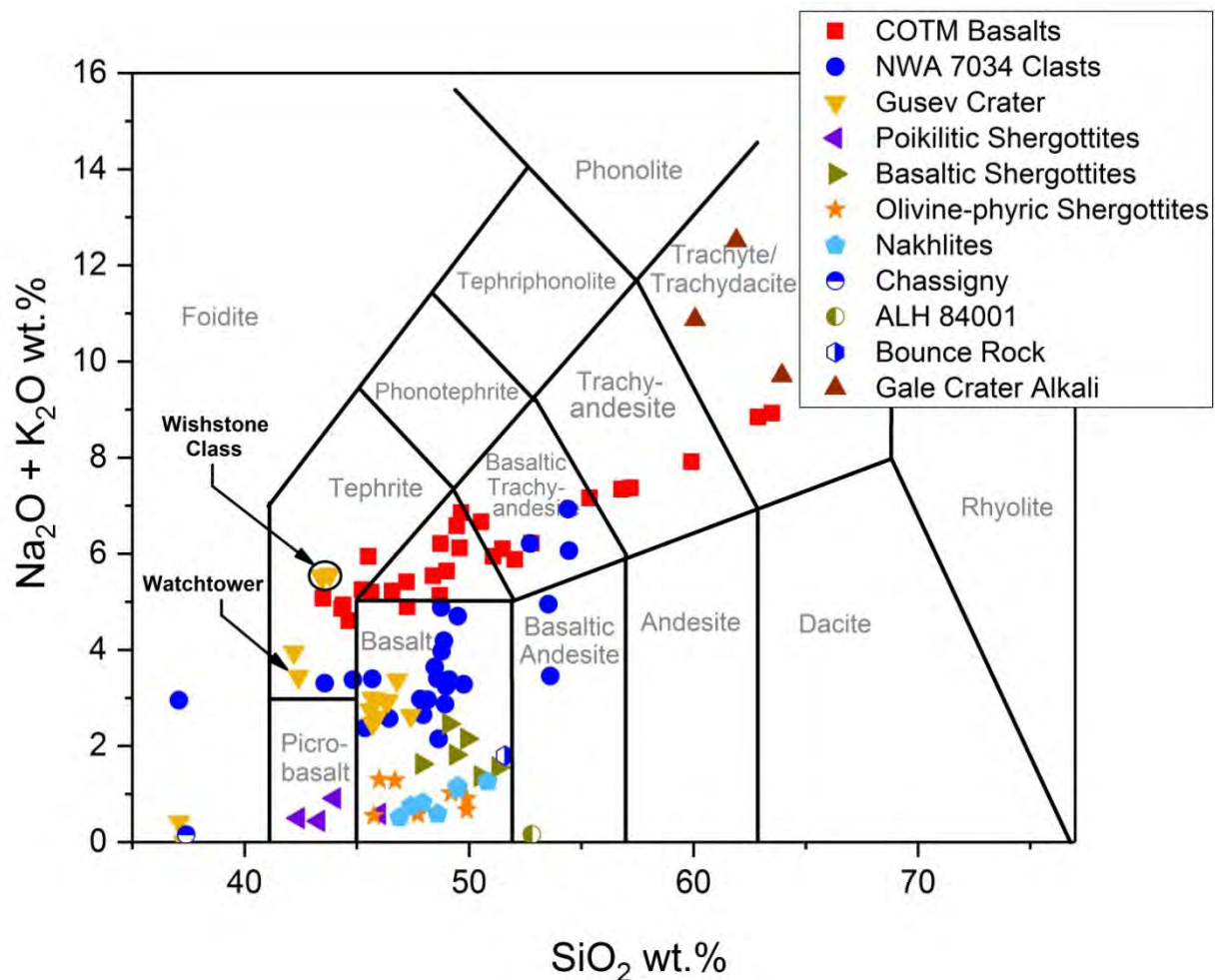
1204 fields represent different individual flows. Ages determined by radio-carbon dating, y.b.p. =  
1205 years before present.

1206

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**Figure 2. TAS ( $\text{Na}_2\text{O} + \text{K}_2\text{O}$  versus  $\text{SiO}_2$ ) diagram of COTM basalts, martian meteorites,**

1211

**and martian surface compositions.** Plot shows most COTM rocks materials are generally higher

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in  $\text{Na}_2\text{O} + \text{K}_2\text{O}$  content compared to the martian materials. There are a few exceptions including

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some high-phosphorus rocks analyzed at Gusev Crater (Watchtower and Wishstone class),

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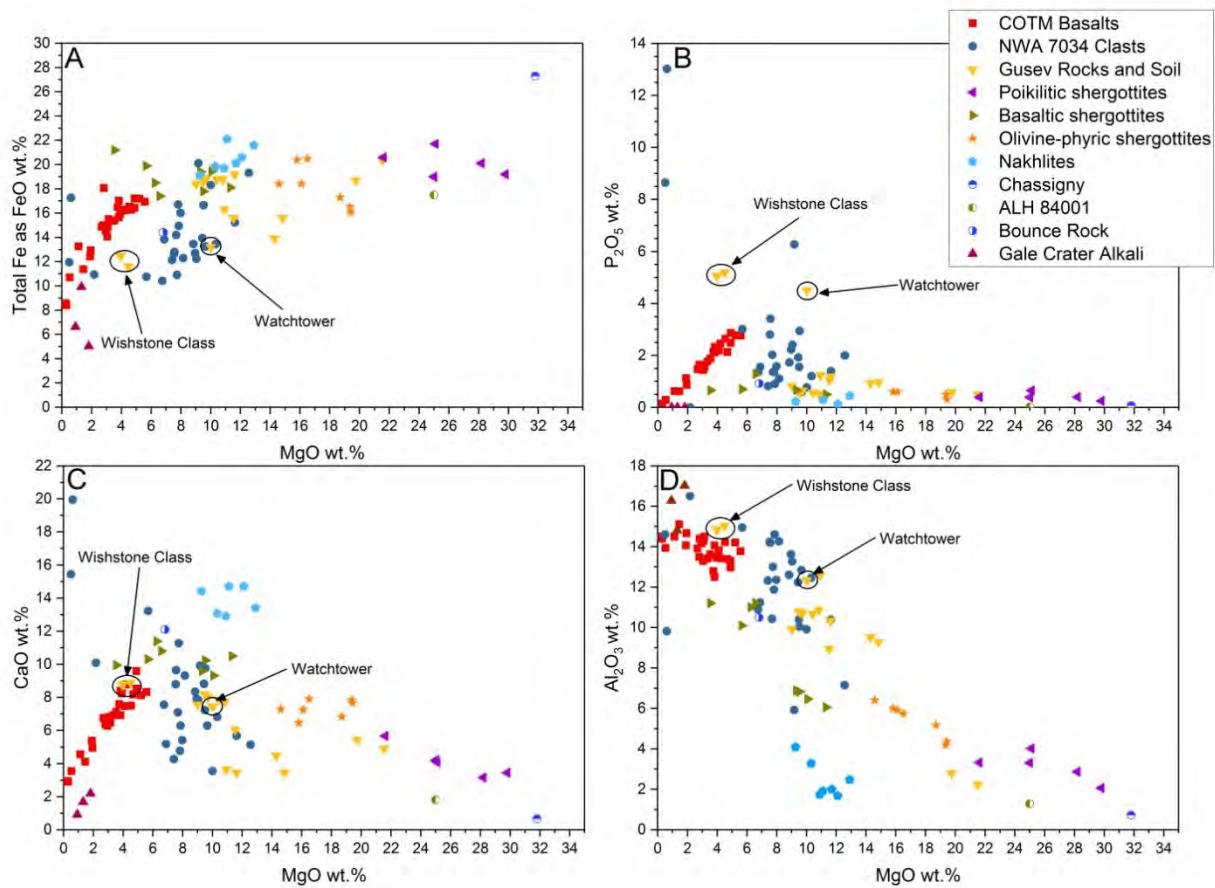
some of the clasts of NWA 7034, and alkali rocks analyzed at Gale Crater. Note trend in COTM

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basalts of increasing alkalis with increasing  $\text{SiO}_2$ . The relationship generally follows the age of

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COTM flows with lower  $\text{SiO}_2$  and alkali contents typical of the older flows.



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1218

**Figure 3. A) FeO total versus MgO, B) P<sub>2</sub>O<sub>5</sub> versus MgO, C) CaO versus MgO, and D) Al<sub>2</sub>O<sub>3</sub>**

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**versus MgO (all in wt.%).** COTM rocks tend to be lower in MgO and higher in Al<sub>2</sub>O<sub>3</sub> than the

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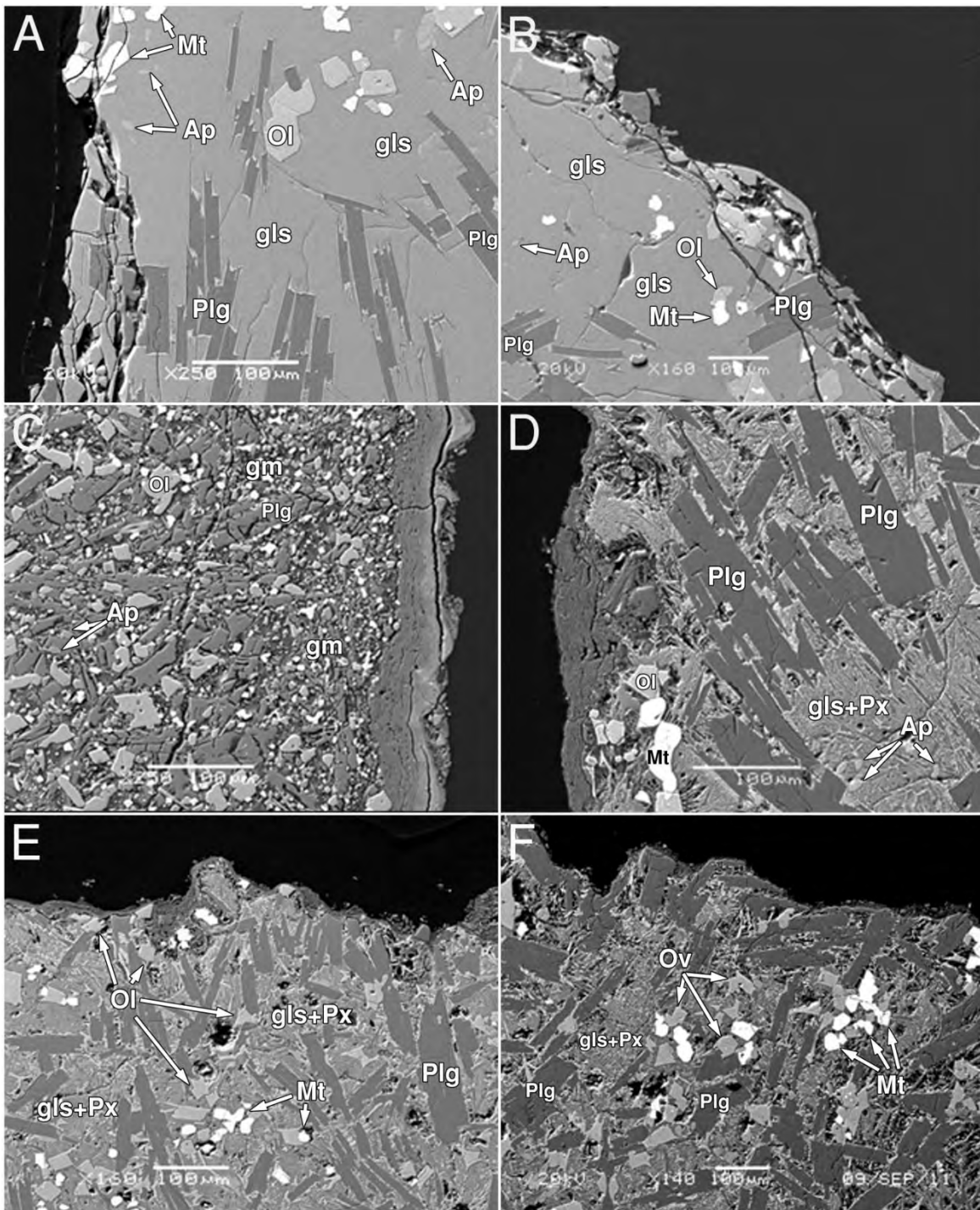
martian materials, although there are several exceptions (e.g. Wishstone class rocks). The linear

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trend seen in COTM rocks in some plots is a result of the evolution of volcanism at COTM over

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time. The higher MgO, CaO, and FeO contents are typical of the older flows at COTM.



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1224

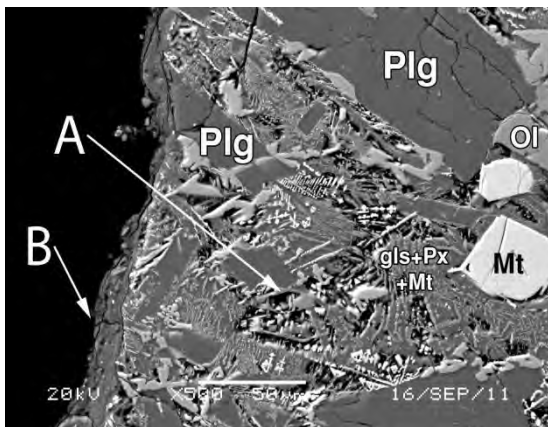
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1226

**Figure 4. BSE images of textures typical of the six flows sampled at exposed edges. A) Blue Dragon, B) Minidoka, C) Lava Point, D) Prong Horn, E) Sunset, and F) Kimama. Note differences in matrix material with Blue Dragon and Minidoka being mainly glass and the other samples**

1227 being more crystalline and less glassy. Coatings, like those of Lava Point (C) and Prong Horn (D)  
1228 were found on all samples with exposed surfaces, but were not continuous on the surfaces. Plg  
1229 = plagioclase, Ol = olivine, Mt = magnetite/ilmenite, Px = pyroxene, gls = glass, Ap = apatite. All  
1230 scale bars 100  $\mu\text{m}$ .  
1231

1232

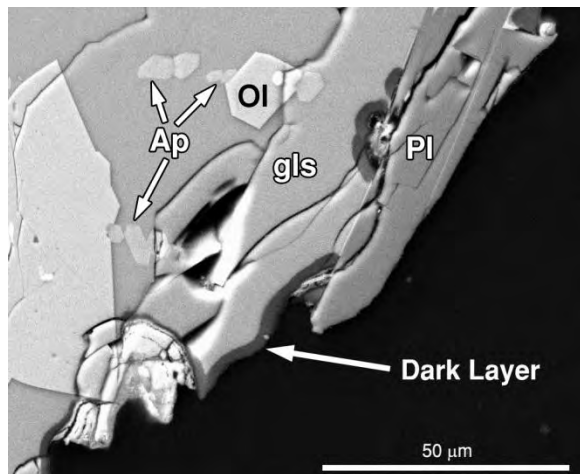


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1234 **Figure 5. BSE image of Kimama basalt in thin section.** Image shows development of porosity  
1235 (A) below discontinuous surface coating (B). The porosity appears to be from the dissolution of  
1236 glass in the groundmass. Phynocrysts show no obvious signs of dissolution. Pl = plagioclase, Olv  
1237 = olivine, Mt = magnetite/ilmenite, Px = pyroxene, gls = glass. Scale bar = 50  $\mu$ m.

1238

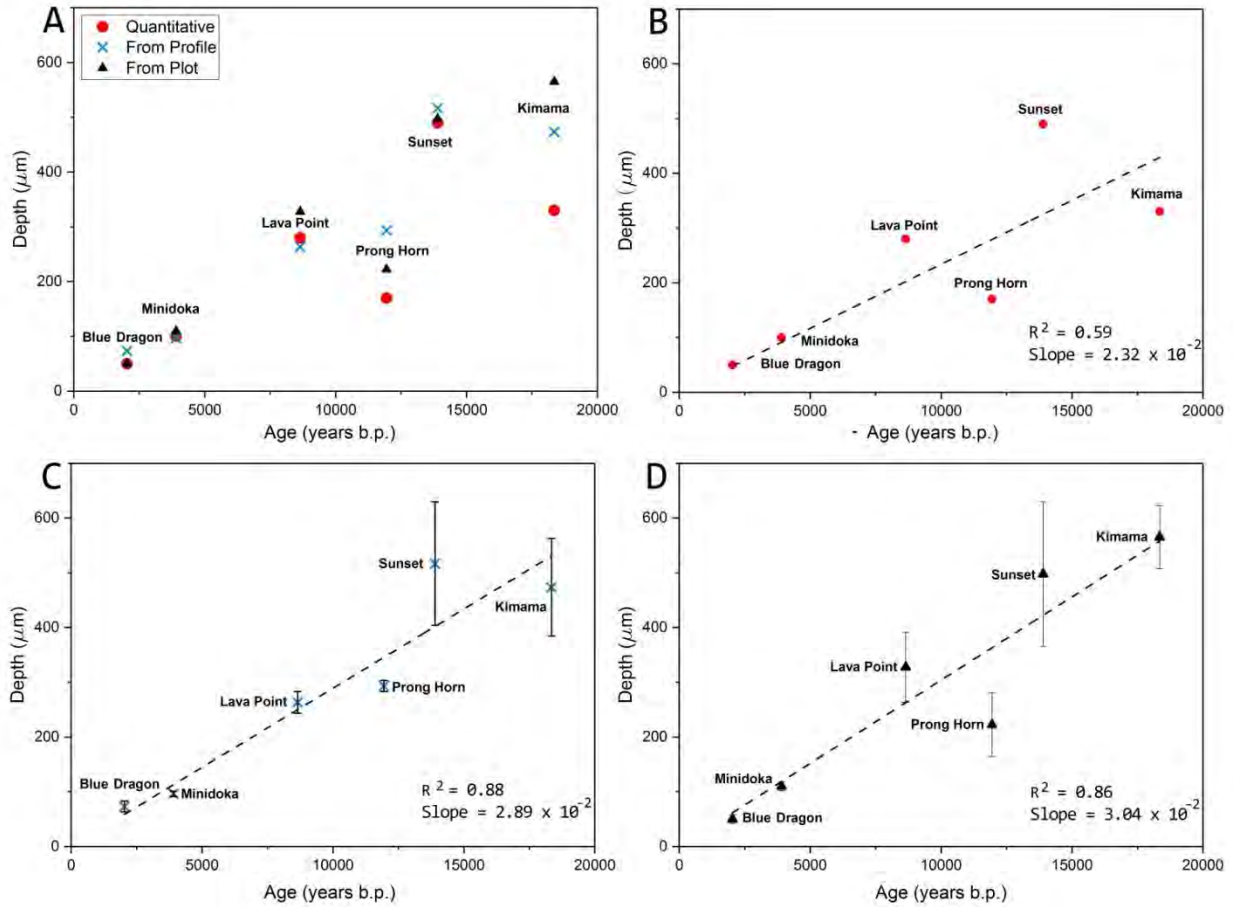
1239



1240

1241 **Figure 6. BSE image of Blue Dragon basalt in thin section.** Note dark layer (indicated with an  
1242 arrow). The dark layer possessed very similar chemistry by EDS to the brighter glass material  
1243 beneath it. This may indicate the dark material is hydrated. Pl = plagioclase, Ol = olivine, gls =  
1244 glass, Ap = apatite. Scale bar is 50  $\mu\text{m}$ .

1245



1246

1247 **Figure 7: Depth of developed porosity ( $\mu\text{m}$ ) versus age plots (years b.p.) for the three**

1248 **approaches of interpretation. A) All methods combined. B) Quantitative method**

1249 (Supplementary Figure S2). **C) Judged visually from BSE image profile (Supplementary Figure**

1250 **S3). D) Judged visually from plotted profile (Supplementary Figure S1). Regression intercepts**

1251 **were tied to zero. No weighting given to associated errors for regression analysis. Note: all axes**

1252 **are identical.**

1253

1254

1255