

Experimental study of Th-bearing LaPO₄ (780 °C, 200 MPa): Implications for monazite and actinide orthophosphate stability

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ABSTRACT

A complete solid solution has been hydrothermally synthesized between the two end-members LaPO₄ and (Ca_{0.5}Th_{0.5})PO₄ at 780 °C and 200 MPa, indicating that there is no limitation in temperature and pressure conditions corresponding to those of granitic magmas for Th insertion in natural monazites. The composition limits of the (A_{1-2x}³⁺B_x²⁺C_x⁴⁺)PO₄ compounds crystallized in the monazite structure-type are determined by both $r_{\text{average}} = (1 - 2x)^{[9]}r_{\text{A}^{3+}} + x^{[9]}r_{\text{B}^{2+}} + x^{[9]}r_{\text{C}^{4+}}$ and $r_{\text{ratio}} = (1 - x)^{[9]}r_{\text{A}^{3+}} + x^{[9]}r_{\text{B}^{2+}}/(1 - x)^{[9]}r_{\text{A}^{3+}} + x^{[9]}r_{\text{C}^{4+}}$ parameters (where $[9]r_A$ is the ionic radius of the A element in ninefold coordination). The upper and lower values of these parameters are 1.216 Å ≥ $r_{\text{average}} \geq$ 1.107 Å and 1.238 ≥ $r_{\text{ratio}} \geq$ 1. The incorporation of large amounts of trans-uranium elements in the monazite structure is deduced from this model. The limitations and geochronological inferences of this model are discussed.