

Thermodynamics of cation ordering in karrooite (MgTi₂O₅)

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ABSTRACT

A thermodynamic model of non-convergent cation-ordering in karrooite (MgTi₂O₅) has been calibrated from the single-crystal X-ray structure refinements of Yang and Hazen (1998) and from estimates of the dependence of the bulk modulus on ordering state (Hazen and Yang 1997). Derived values of the Gibbs free energy, enthalpy, entropy, and heat capacity of disordering of karrooite are reported as a function of temperature at 1 bar and at elevated pressures.