

## Supporting Information

### Amorphous $\text{Mn}_2\text{SiO}_4$ : A potential manganese phase in the stagnant slab

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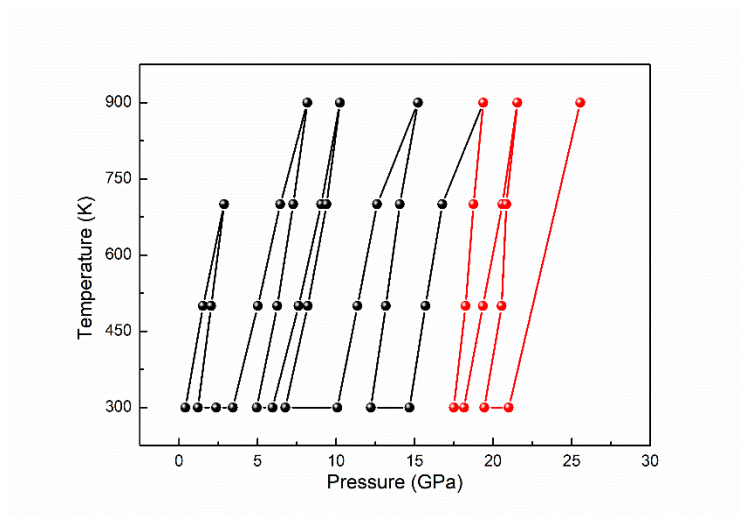
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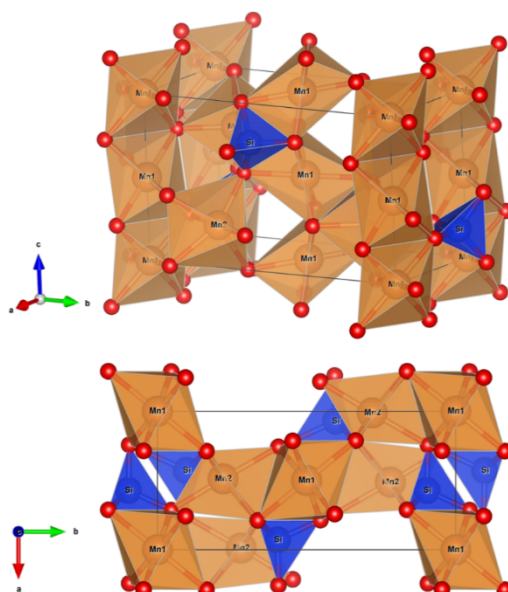
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Figure S1-S6

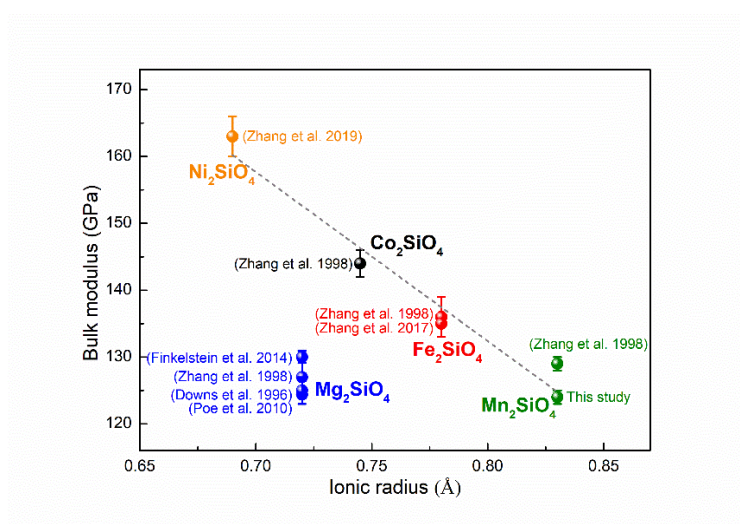
Table S1-S7



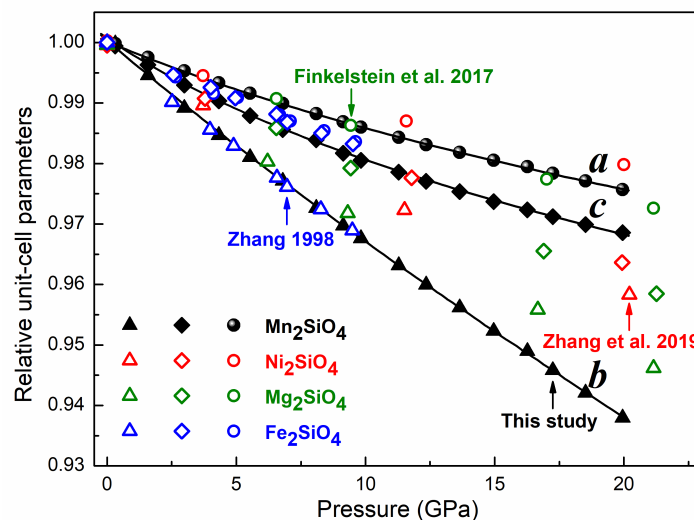
**Figure S1.** High-pressure and high-temperature experimental  $P$ - $T$  path for tephroite. The solid circles represent the collection points of the single-crystal diffraction data. The black and red circles indicate  $\alpha$ -tephroite and amorphous tephroite, respectively.



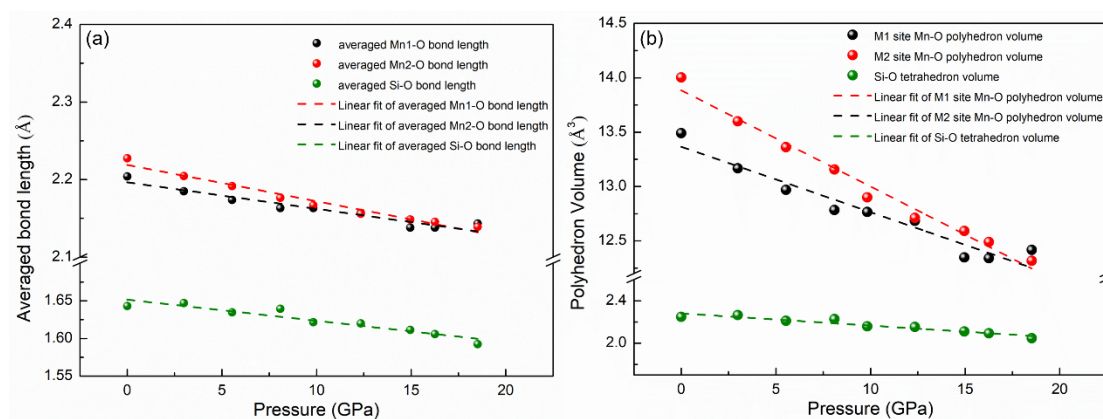
**Figure S2.** Crystal structure of Tephroite at ambient conditions.



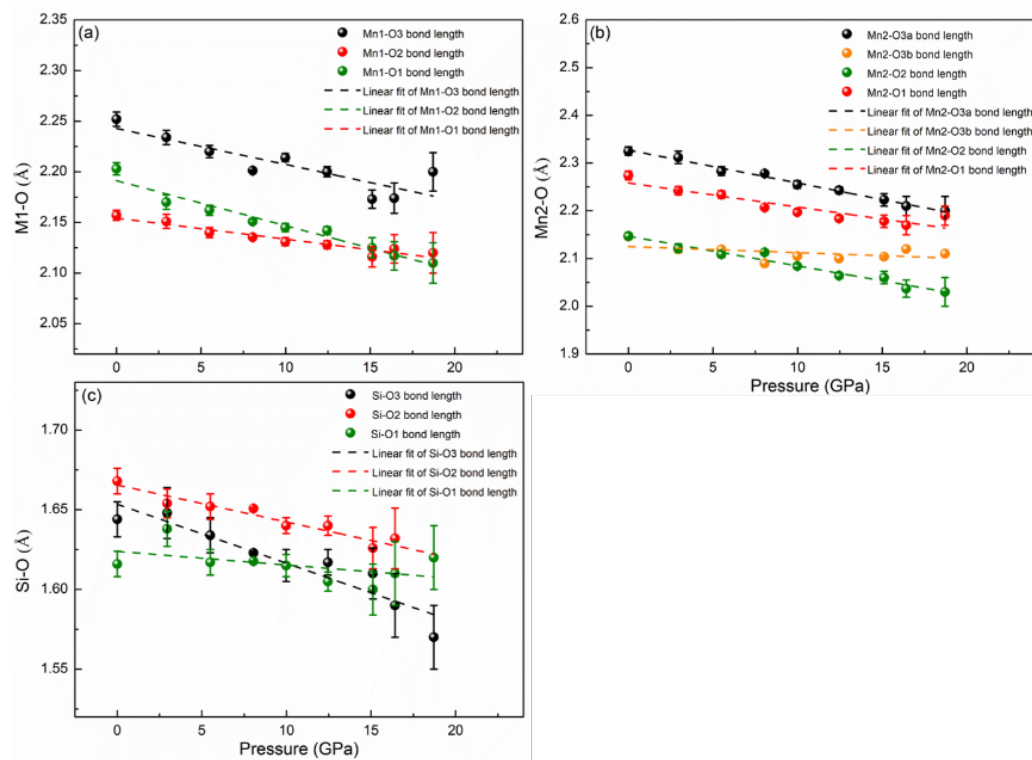
**Figure S3.** Variation in bulk moduli of end-member olivine (Ni, Co, Fe, Mn, Mg)<sub>2</sub>SiO<sub>4</sub> with ionic radius. The bulk moduli of olivine from previous studies are shown for comparison (Downs et al. 1996; Zhang 1998; Poe et al. 2010; Finkelstein et al. 2014; Zhang et al. 2017, 2019). The gray dashed line represents the linear fit.



**Figure S4.** Pressure dependence of the relative unit-cell parameters *a*, *b*, *c* of Tephroite and other end-member olivines at room temperature.



**Figure S5.** Pressure independence of averaged bond length (a) and polyhedron volume (b) in this study. The dashed lines represent the linear fitting.



**Figure S6.** Pressure independence of bond length in this study: (a) Mn1-O, (b) Mn2-O, (c) Si-O. The dashed lines represent the linear fitting.

**Table S1.** Chemical compositions of tephroite based on the average of at least 5-point electron microprobe analyses

Oxide (wt.%)	Tephroite
SiO <sub>2</sub>	29.59(54)
TiO <sub>2</sub>	0.02(1)
Al <sub>2</sub> O <sub>3</sub>	0.02(2)
Cr <sub>2</sub> O <sub>3</sub>	-
FeO	0.51(7)
MnO	72.39(38)
MgO	0.01(0)
CaO	0.01(2)
Na <sub>2</sub> O	0.01(1)
K <sub>2</sub> O	-
Total	102.58 (66)

- Below the detection limit

**Table S2.** Lattice parameters and unit-cell volumes of tephroite at room temperature and high pressure (a) and high temperature and high pressure (b).

<i>P</i> (GPa)	<i>T</i> (K)	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	<i>V</i> (Å <sup>3</sup> )
(a) room temperature and high pressure					
0.0001	300	4.903(1)	10.599(1)	6.258(3)	325.22(13)
0.32(2)	300	4.893(2)	10.597(1)	6.255(1)	324.28(11)
1.58(10)	300	4.880(1)	10.542(1)	6.235(1)	320.77(8)
2.99(17)	300	4.880(1)	10.484(1)	6.214(1)	317.91(6)
4.33(17)	300	4.868(2)	10.437(1)	6.198(1)	314.85(14)
5.54(16)	300	4.857(2)	10.398(2)	6.182(1)	312.22(11)
6.79(16)	300	4.846(1)	10.357(1)	6.168(1)	309.57(6)
8.09(20)	300	4.841(3)	10.309(3)	6.157(1)	307.29(17)
9.13(16)	300	4.831(2)	10.278(2)	6.144(1)	305.06(8)
9.83(2)	300	4.829(1)	10.256(1)	6.136(1)	303.92(6)
11.28(23)	300	4.821(2)	10.208(1)	6.124(1)	301.40(8)
12.35(16)	300	4.818(1)	10.174(1)	6.114(1)	299.71(4)
13.65(14)	300	4.811(1)	10.135(1)	6.104(1)	297.58(4)
14.96(21)	300	4.806(1)	10.094(1)	6.093(1)	295.60(5)
16.26(21)	300	4.795(1)	10.058(1)	6.085(1)	293.48(6)
17.25(21)	300	4.794(1)	10.025(1)	6.078(1)	292.11(6)
18.52(23)	300	4.786(1)	9.985(1)	6.070(1)	290.09(5)
19.95(18)	300	4.778(2)	9.941(1)	6.061(1)	287.88(7)
(b) high temperature and high pressure					
0.41(2)	300	4.895(1)	6.125(2)	10.855(1)	324.07(3)
1.21(1)	300	4.888(1)	6.238(1)	10.546(1)	321.61(4)
2.37(5)	300	4.879(2)	6.221(2)	10.499(1)	318.67(4)
3.44(1)	300	4.867(1)	6.205(1)	10.449(1)	315.59(4)
4.95(1)	300	4.858(2)	6.191(2)	10.401(1)	312.85(5)
5.95(1)	300	4.851(2)	6.175(2)	10.361(1)	310.35(4)
6.78(4)	300	4.846(1)	6.167(1)	10.327(1)	308.62(4)
10.09(2)	300	4.823(1)	6.130(1)	10.218(1)	302.11(4)
12.21(3)	300	4.814(1)	6.109(1)	10.135(1)	298.04(4)

14.68(11)	300	4.799(2)	6.083(2)	10.060(1)	293.66(6)
1.53(4)	500	4.895(2)	6.252(1)	10.575(1)	323.61(5)
2.04(4)	500	4.888(2)	6.243(1)	10.546(1)	321.84(4)
5.02(8)	500	4.866(2)	6.203(2)	10.434(1)	314.9(5)
6.26(8)	500	4.855(2)	6.185(2)	10.376(1)	311.64(4)
7.61(16)	500	4.845(2)	6.166(2)	10.325(1)	308.43(5)
8.20(18)	500	4.843(2)	6.164(2)	10.307(1)	307.69(5)
11.36(9)	500	4.822(2)	6.129(2)	10.198(1)	301.38(4)
13.17(11)	500	4.813(1)	6.110(1)	10.131(1)	297.88(4)
15.70(1)	500	4.798(2)	6.087(2)	10.049(1)	293.50(4)
2.87(7)	700	4.892(3)	6.246(2)	10.551(1)	322.42(8)
6.44(1)	700	4.860(2)	6.198(2)	10.406(1)	313.47(6)
7.28(9)	700	4.854(2)	6.187(2)	10.372(1)	311.44(6)
9.06(10)	700	4.841(2)	6.167(2)	10.309(1)	307.75(5)
9.40(12)	700	4.841(2)	6.160(1)	10.292(1)	306.91(4)
12.61(7)	700	4.821(2)	6.127(2)	10.181(1)	300.72(4)
14.06(18)	700	4.811(2)	6.113(2)	10.126(1)	297.78(5)
16.77(8)	700	4.797(2)	6.088(2)	10.034(1)	293.03(4)
8.17(8)	900	4.858(3)	6.188(3)	10.371(1)	311.77(10)
10.25(5)	900	4.840(3)	6.165(2)	10.288(1)	306.94(10)
15.22(4)	900	4.809(3)	6.112(2)	10.108(1)	297.12(8)
Numbers	in	parenthesis	represent	standard	deviations

**Table S3.** Structure refinement details, fractional coordinates and isotropic displacement parameters for Tephroite at different pressures

<i>P</i> (GPa)	Final R indices [ <i>I</i> > 2σ ( <i>I</i> )]	Number total reflections/ unique reflections/ refined parameters	Atom	<i>x</i>	<i>y</i>	<i>z</i>	U <sub>iso</sub>
0.0001	R1=0.045 wR <sub>2</sub> =0.117	321/82/18 [R(int)=0.0477]	Mn1	0.000000	0.000000	0.000000	0.0074(12)
			Mn2	0.9880(4)	0.28026(15)	0.250000	0.0047(12)
			Si	0.4280(8)	0.0968(3)	0.250000	0.0040(13)
			O1	0.7575(14)	0.0941(8)	0.250000	0.004(2)
			O2	0.2139(15)	0.4538(6)	0.250000	0.004(2)
			O3	0.2878(13)	0.1639(6)	0.040(2)	0.0065(19)
2.99(17 )	R1=0.071 wR <sub>2</sub> =0.191	735/158/18 [R(int)=0.0712]	Mn1	0.000000	0.000000	0.000000	0.0140(6)
			Mn2	0.9860(4)	0.27995(14)	0.250000	0.0112(6)
			Si	0.4269(9)	0.0966(3)	0.250000	0.0128(9)
			O1	0.7625(19)	0.0930(8)	0.250000	0.0115(19)
			O2	0.2126(19)	0.4528(7)	0.250000	0.0113(19)
			O3	0.2865(15)	0.1647(5)	0.038(3)	0.0140(14)
5.54(16 )	R1=0.060 wR <sub>2</sub> =0.161	735/149/18 [R(int)=0.0632]	Mn1	0.000000	0.000000	0.000000	0.0141(5)
			Mn2	0.9850(3)	0.27969(12)	0.250000	0.0112(5)
			Si	0.4273(7)	0.0971(3)	0.250000	0.0122(7)
			O1	0.7597(14)	0.0923(7)	0.250000	0.0114(16)
			O2	0.2132(15)	0.4523(6)	0.250000	0.0120(16)
			O3	0.2844(12)	0.1654(5)	0.040(2)	0.0134(12)
8.09(20 )	R1=0.103 wR <sub>2</sub> =0.257	622/123/18 [R(int)=0.0805]	Mn1	0.000000	0.000000	0.000000	0.0149(13)
			Mn2	0.9846(8)	0.2794(3)	0.250000	0.0097(12)
			Si	0.4278(17)	0.0978(6)	0.250000	0.0133(18)
			O1	0.761(3)	0.0927(13)	0.250000	0.008(3)
			O2	0.211(3)	0.4546(12)	0.250000	0.009(3)
			O3	0.282(3)	0.1663(9)	0.035(5)	0.011(3)
9.83(2)	R1=0.033	724/152/18	Mn1	0.000000	0.000000	0.000000	0.0086(4)

12.35(1 6)	wR <sub>2</sub> =0.095  R1=0.048 wR <sub>2</sub> =0.072	[R(int)=0.0405]  699/126/18 [R(int)=0.0637]	Mn2	0.9830(3)	0.27952(10)	0.250000	0.0048(3)
			Si	0.4264(5)	0.0979(2)	0.250000	0.0051(5)
			O1	0.7603(12)	0.0926(5)	0.250000	0.0045(12)
			O2	0.2121(12)	0.4518(4)	0.250000	0.0025(12)
			O3	0.2848(9)	0.1673(3)	0.0419(19)	0.0064(9)
			Mn1	0.000000	0.000000	0.000000	0.0107(4)
			Mn2	0.9818(3)	0.27945(11)	0.250000	0.0075(4)
			Si	0.4260(5)	0.0978(2)	0.250000	0.0069(6)
			O1	0.7589(10)	0.0925(5)	0.250000	0.0058(14)
			O2	0.2123(12)	0.4505(5)	0.250000	0.0062(14)
			O3	0.2831(9)	0.1678(4)	0.0417(15)	0.0066(11)
			Mn1	0.000000	0.000000	0.000000	0.0163(8)
			Mn2	0.9808(6)	0.2794(2)	0.250000	0.0104(8)
			Si	0.4270(11)	0.0972(5)	0.250000	0.0101(12)
14.96(2 1)	R1=0.083 wR <sub>2</sub> =0.189	695/113/18 [R(int)=0.0868]	O1	0.759(2)	0.0917(11)	0.250000	0.011(3)
			O2	0.213(2)	0.4503(10)	0.250000	0.010(3)
			O3	0.2802(18)	0.1675(7)	0.043(3)	0.0080(19)
			Mn1	0.000000	0.000000	0.000000	0.0142(14)
			Mn2	0.9801(9)	0.2787(4)	0.250000	0.0088(13)
			Si	0.4247(18)	0.0964(8)	0.250000	0.009(2)
			O1	0.760(4)	0.0903(19)	0.250000	0.016(5)
			O2	0.212(4)	0.4481(16)	0.250000	0.011(5)
			O3	0.281(3)	0.1669(13)	0.046(5)	0.008(3)
			Mn1	0.000000	0.000000	0.000000	0.015(2)
			Mn2	0.9784(13)	0.2791(5)	0.250000	0.0084(17)
			Si	0.421(2)	0.0969(10)	0.250000	0.007(3)
			O1	0.758(6)	0.088(2)	0.250000	0.017(7)
			O2	0.211(6)	0.448(2)	0.250000	0.010(6)
			O3	0.285(4)	0.1694(16)	0.049(7)	0.008(5)



**Table S4.** Details of each polyhedron of tephroite under different pressures

<i>P</i> (GPa)	0.0001	2.99(17)	5.54(16)	8.09(20)	9.83(2)	12.35(16)	14.96(21)	16.26(21)	18.52(23)
<b>Mn1</b> Average bond length (Å)	2.2041	2.1849	2.1737	2.1632	2.1633	2.1569	2.138	2.1379	2.1434
Polyhedral volume (Å <sup>3</sup> )	13.491	13.168	12.970	12.784	12.766	12.684	12.348	12.341	12.417
Distortion index	0.01449	0.01495	0.01409	0.01212	0.01554	0.01338	0.01083	0.01211	0.01755
Quadratic elongation	1.0388	1.0373	1.0371	1.037	1.0382	1.0364	1.0367	1.037	1.0382
Bond angle variance	135.0284	130.5926	129.72	129.7995	133.156	127.4935	128.8103	129.7067	132.12
Effective coordination number	5.9304	5.9435	5.9483	5.9627	5.9407	5.956	5.9719	5.9645	5.9239
<b>Mn2</b> Average bond length (Å)	2.2273	2.2044	2.1914	2.1764	2.1669	2.1558	2.1485	2.1453	2.1391
Polyhedral volume (Å <sup>3</sup> )	14.003	13.600	13.362	13.156	12.900	12.712	12.592	12.490	12.318
Distortion index	0.03636	0.03832	0.03452	0.03629	0.03179	0.03146	0.02768	0.02659	0.02767
Quadratic elongation	1.0358	1.0348	1.0344	1.0311	1.0353	1.0347	1.0341	1.0366	1.0403
Bond angle variance	123.4924	119.2335	118.632	107.0648	121.9171	119.8115	118.2012	126.1427	138.4975
Effective coordination number	5.6709	5.6236	5.6985	5.6515	5.7329	5.7244	5.7831	5.7582	5.7369
<b>Si</b> Average bond length (Å)	1.6431	1.6469	1.6345	1.6394	1.6214	1.6199	1.6114	1.6059	1.5925
Polyhedral volume (Å <sup>3</sup> )	2.2484	2.2645	2.2102	2.227	2.1584	2.1524	2.1107	2.0934	2.0463
Distortion index	0.00829	0.00267	0.00547	0.0127	0.0057	0.00618	0.00447	0.01109	0.01636
Quadratic elongation	1.0085	1.0082	1.0094	1.0103	1.009	1.009	1.0116	1.0103	1.0088
Bond angle variance	37.6364	34.8516	41.1174	43.0923	38.7379	39.6335	50.5663	42.2048	33.4747
Effective coordination number	3.9816	3.9984	3.9916	3.9764	3.9942	3.9917	3.9956	3.9815	3.9609

**Table S5.** Equations of state parameters of end-member olivine from single-crystal XRD

Composition	$V$ (Å <sup>3</sup> )	$K_0$ (GPa)	$K'_0$	Medium	$P_{\max}$ (GPa)	Reference
Mn <sub>2</sub> SiO <sub>4</sub>	325.22(13)	124(1)	4.67(29)	Ne	20	This study
	324.3(1)	129(2)	3.0(5)	4:1 meth:eth <sup>a</sup>	10	(Zhang 1998)
	290.14(9)	125(2)	4.0(4)	4:1 meth:eth, Ar, He	17	(Downs et al. 1996)
Mg <sub>2</sub> SiO <sub>4</sub>	289.3(1)	127(4)	4.2(8)	4:1 meth:eth	10	(Zhang 1998)
	291.15(1)	124.4	4.9	4:1 meth:eth	6.7	(Poe et al. 2010)
	290.1(1)	130.0(9)	4.12(7)	He	50	(Finkelstein et al. 2014)
Fe <sub>2</sub> SiO <sub>4</sub>	306.9(1)	136(3)	4.1(7)	4:1 meth:eth	10	(Zhang 1998)
		135 <sup>fixed</sup>	4.0(2)	Ne	30.7	(Zhang et al. 2017)
Co <sub>2</sub> SiO <sub>4</sub>	295.21(7)	144(2)	4.1(5)	4:1 meth:eth	10	(Zhang 1998)
Ni <sub>2</sub> SiO <sub>4</sub>	283.38(7)	163(3)	4.5(3)	He	42.6	(Zhang et al. 2019)

Numbers in parenthesis represent standard deviations.

<sup>a</sup> 4:1 methanol-ethanol mixture

**Table S6.** Best-fit volumetric and linear BM3 parameters of the tephroite

Unit-cell parameter	$V$ (Å <sup>3</sup> )/ $L$ (Å)	$K_0 / M_{l0}$ (GPa)	$K'_0 / M_{l'_0}$
Unit-cell volume $V$	325.22(13)	124(1)	4.67(29)
Lattice parameter $a$	4.90393(3)	553(15)	32(4)
Lattice parameter $b$	10.603(1)	273(4)	5.4(5)
Lattice parameter $c$	6.2626(1)	341(13)	39(4)

Numbers in parenthesis represent standard deviations

**Table S7.** Frequency of Raman modes, corresponding pressure derivatives, and isothermal mode Grüneisen parameters of tephroite

Symmetry	Frequency, cm <sup>-1</sup>	$(\partial \nu_i / \partial P)_T$	$\gamma_{iT}$	Mode assignment
A <sub>g</sub>	934	4.71(6)	0.62(1)	$\nu_3$
B <sub>3g</sub>	891	3.21(2)	0.45(0)	$\nu_3$
A <sub>g</sub>	841	3.15(5)	0.46(1)	$\nu_1 + \nu_3$
A <sub>g</sub>	809	3.15(5)	0.48(1)	$\nu_1 + \nu_3$
A <sub>g</sub>	569	2.73(20)	0.59(1)	$\nu_4$
B <sub>2g</sub>	550	1.66(18)	0.37(1)	$\nu_4$
A <sub>g</sub>	514	1.25(2)	0.3(1)	$\nu_4$
B <sub>1g</sub>	391	3.58(3)	1.13(1)	$\nu_2$
B <sub>1g</sub>	317	2.60(6)	1.02(2)	Mix (SiO <sub>4</sub> rot)
A <sub>g</sub>	302	4.03(7)	1.64(3)	Mix (SiO <sub>4</sub> rot)
A <sub>g</sub>	287	2.98(3)	1.28(2)	M2 translation
B <sub>3g</sub>	276	2.49(8)	1.12(4)	Mix (M2 trans)
A <sub>g</sub>	245	2.25(9)	1.15(5)	M2 translation

## Reference:

- Birch, F. (1947) Finite Elastic Strain of Cubic Crystals. *Physical Review*, 71, 809–824.
- Downs, R.T., Zha, C.-S., Duffy, T.S., and Finger, L.W. (1996) The equation of state of forsterite to 17.2 GPa and effects of pressure media. *American Mineralogist*, 81, 51–55.
- Fei, Y. (1995) Thermal Expansion. In *Mineral Physics & Crystallography: A Handbook of Physical Constants* pp. 29–44.
- Finkelstein, G.J., Dera, P.K., Jahn, S., Oganov, A.R., Holl, C.M., Meng, Y., and Duffy, T.S. (2014) Phase transitions and equation of state of forsterite to 90 GPa from single-crystal X-ray diffraction and molecular modeling. *American Mineralogist*, 99, 35–43.
- Poe, B.T., Romano, C., Nestola, F., and Smyth, J.R. (2010) Electrical conductivity anisotropy of dry and hydrous olivine at 8 GPa. *Physics of the Earth and Planetary Interiors*, 181, 103–111.
- Zhang, D., Hu, Y., Xu, J., Downs, R.T., Hammer, J.E., and Dera, P.K. (2019) High-pressure behavior of liebenbergite: The most incompressible olivine-structured silicate. *American Mineralogist*, 104, 580–587.
- Zhang, J.S., Hu, Y., Shelton, H., Kung, J., and Dera, P. (2017) Single-crystal X-ray diffraction study of Fe<sub>2</sub>SiO<sub>4</sub> fayalite up to 31 GPa. *Physics and Chemistry of Minerals*, 44, 171–179.
- Zhang, L. (1998) Single crystal hydrostatic compression of (Mg,Mn,Fe,Co)<sub>2</sub>SiO<sub>4</sub> olivines. *Physics and Chemistry of Minerals*, 25, 308–312.