

FIGURE S1. Representative Raman spectra of materials in the inclusions.

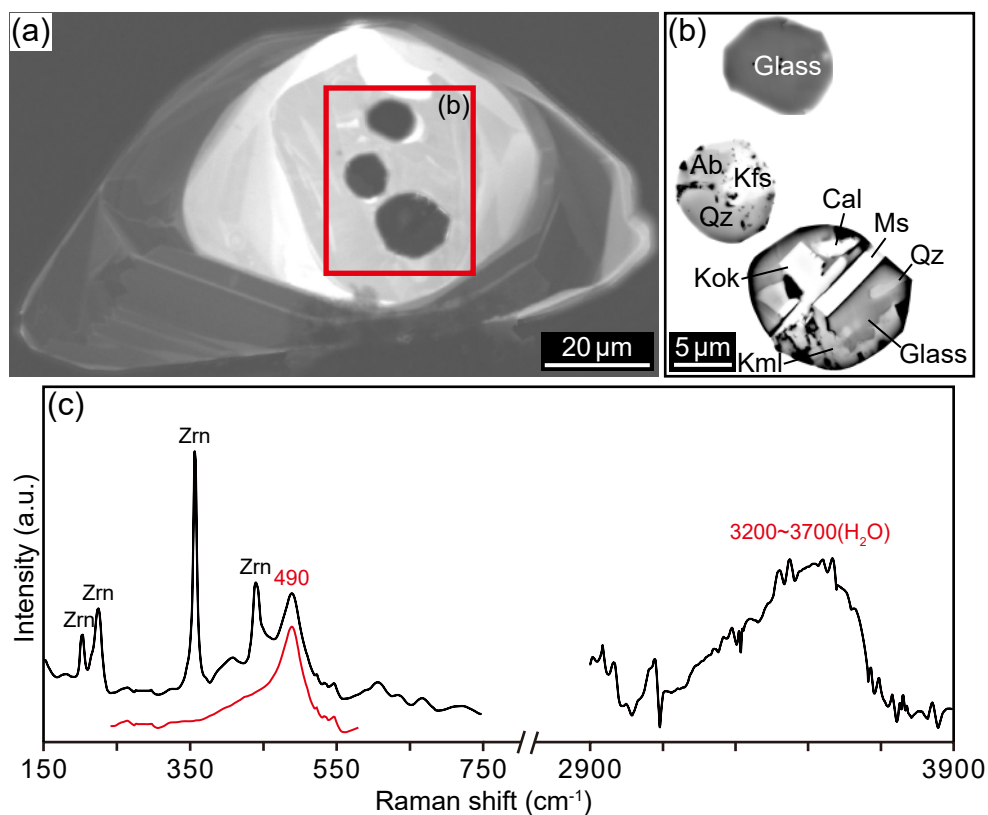


FIGURE S2. (a) CL image showing melt inclusions exposed on the polishing surface (b) BSE image of melt inclusions. (c) Raman spectra of granitic glass in melt inclusion. The glass contains H₂O with a broad band of 3200–3700 cm⁻¹, and has a granitic composition (SiO₂ = 78.23, TiO₂ = 0.07, Al₂O₃ = 12.13, FeO = 0.06, CaO = 0.21, Na₂O = 2.06, K₂O = 1.49, wt.%).

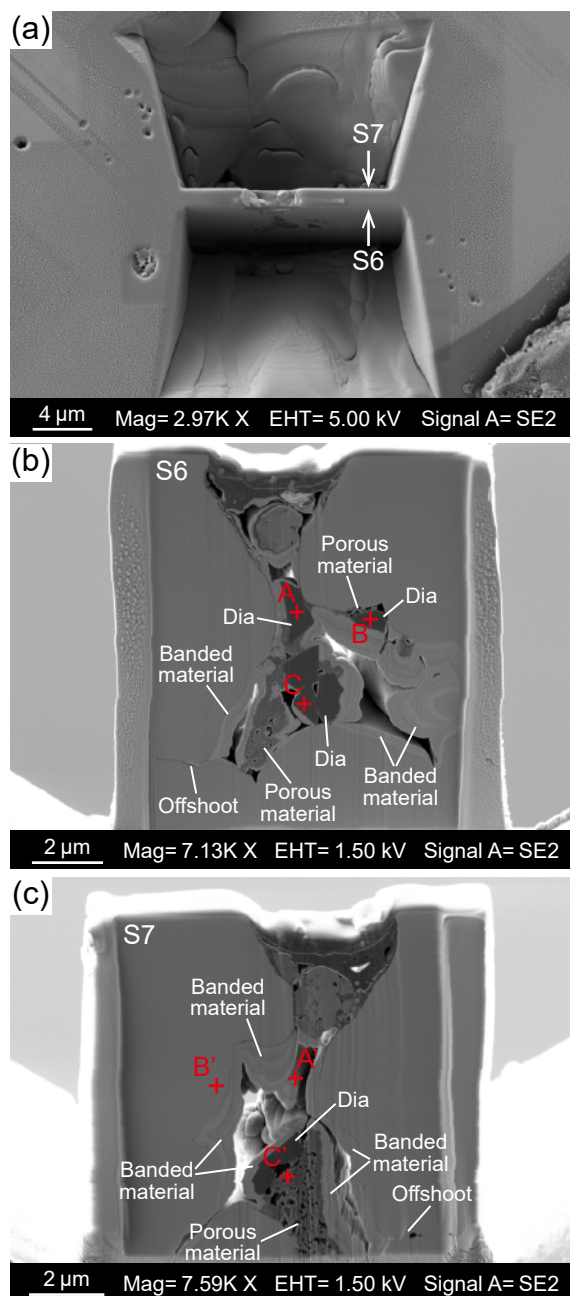


FIGURE S3. SE images obtained from inclusion 5. (a) SE image of air-view showing FIB-milled trench with double-sided vertical cross-sections S6 and S7. (b and c) SE images of vertical cross-sections S6 and S7. A-A', B-B' and C-C' are respectively corresponding positions on S6 and S7 cross-sections. The vertical cross-sections show that the diamonds are enclosed in the banded and porous materials.

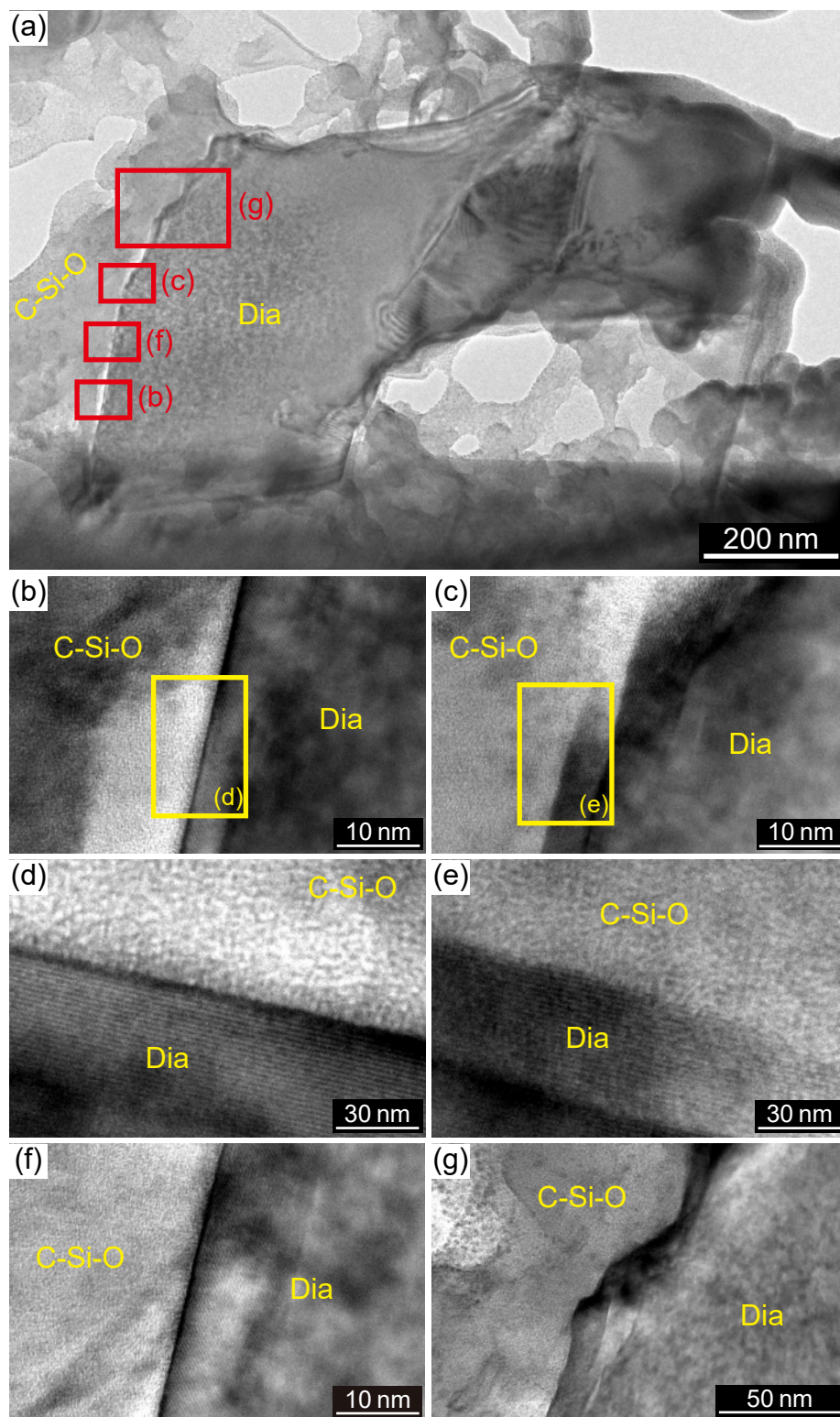


FIGURE S4. BF and HRTEM images showing fully coupled and welded interface between micrometer diamond and amorphous C-Si-O material. **(a)** BF image. **(b-g)** HRTEM images. One set of lattice fringes of diamond are present in images **b-f**. Some discrete black fringes along the interfaces arise from mass-thickness contrast.

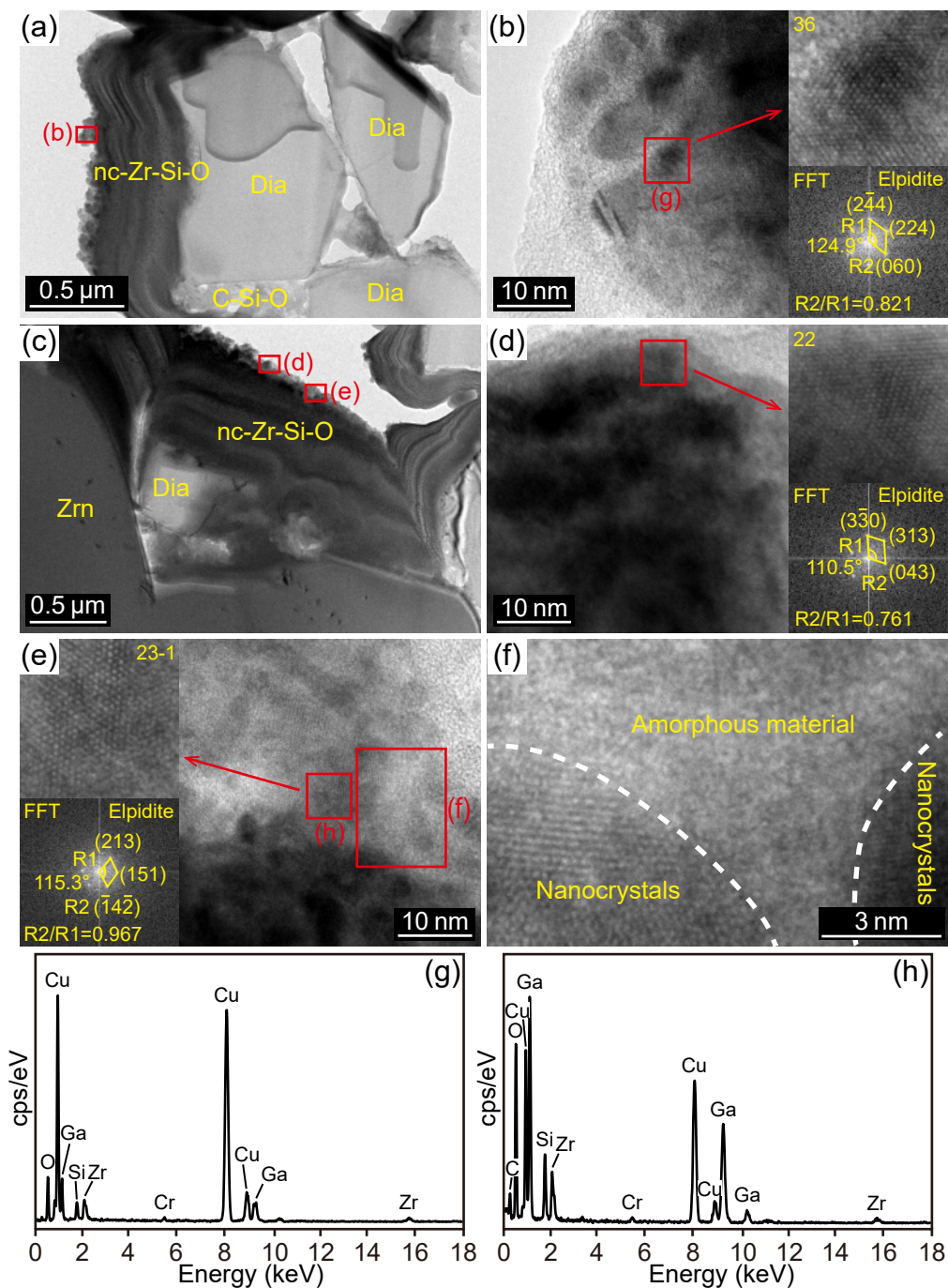


FIGURE S5. TEM observations on the nanocrystals in the banded Zr-Si-O material in inclusion 5. **(a and c)** BF images. **(b, d, e and f)** HRTEM images and FFT diffraction patterns. **(g and h)** X-ray compositional spectra of nanocrystals. The FFT diffraction patterns suggest that the nanocrystals are elpidite.

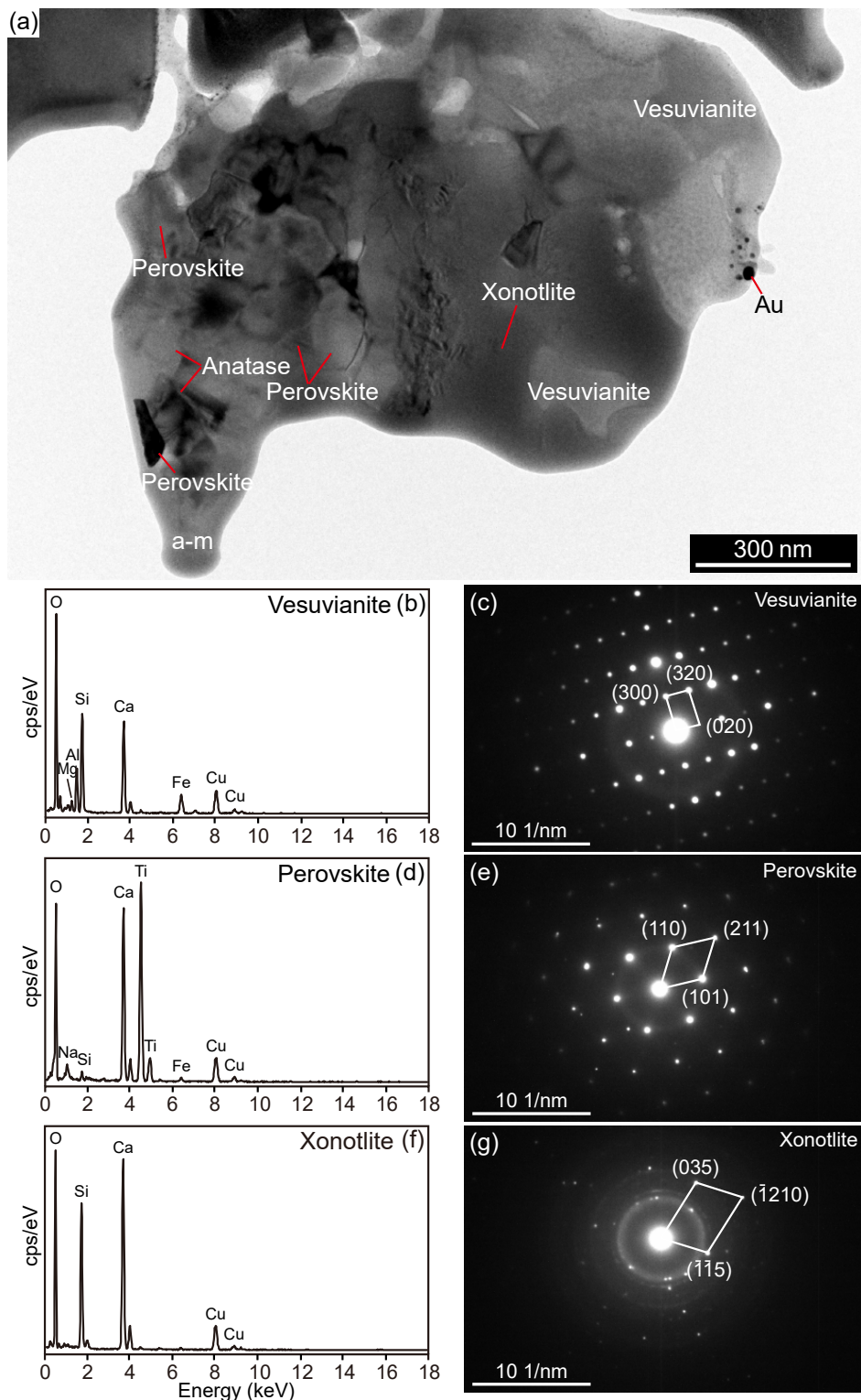


FIGURE S6. BF image, X-ray compositional spectra and SAED patterns showing the spherulite in the upper part of inclusion 5. **(a)** BF image showing nanometer minerals and amorphous material (a-m). **(b and c)** X-ray compositional spectrum and SAED pattern of vesuvianite. **(d and e)** X-ray compositional spectrum and SAED pattern of perovskite. **(f and g)** X-ray compositional spectrum and SAED pattern of xonotlite.

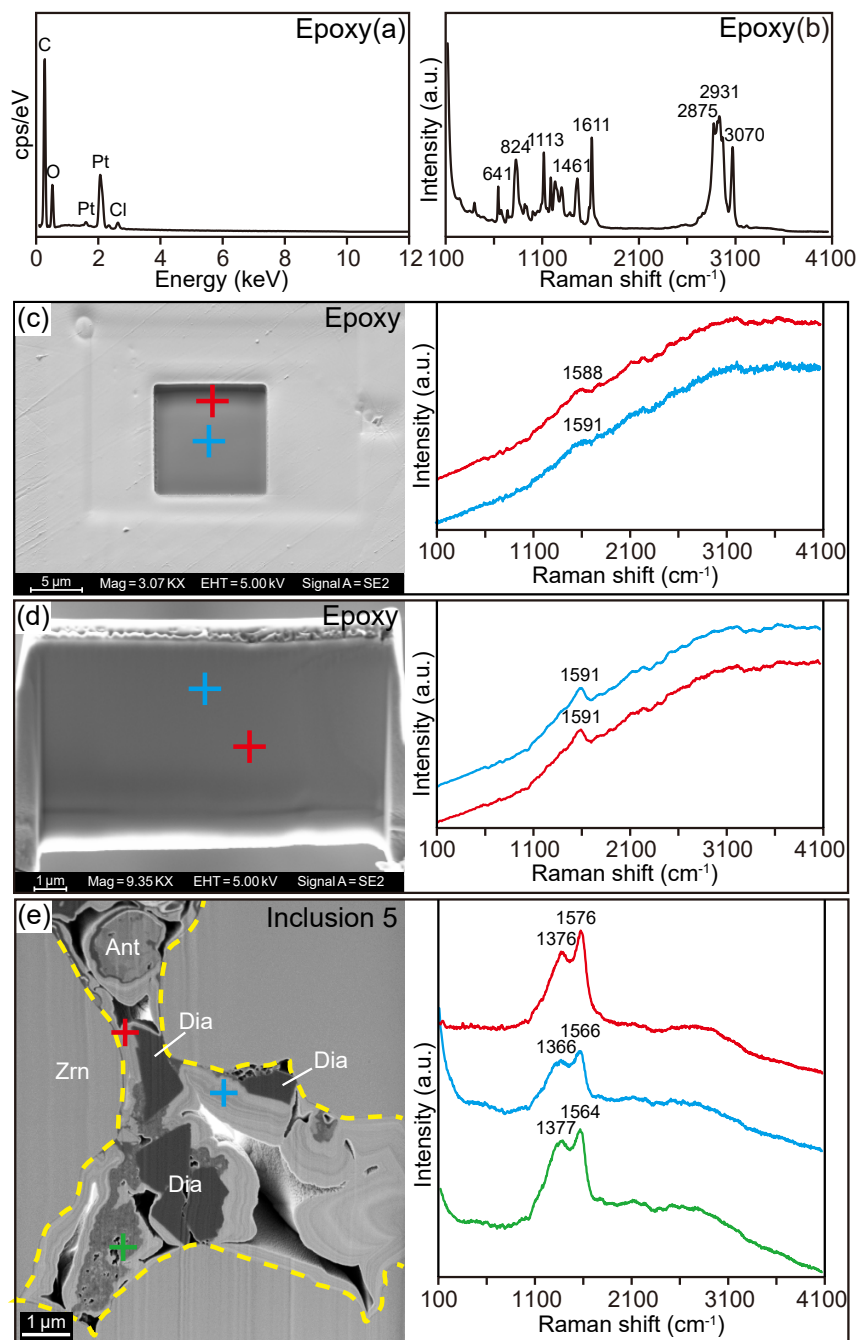


FIGURE S7. (a) X-ray compositional spectrum of epoxy resin, consisting of C, O and Cl. Hydrogen in epoxy resin is indicated from the Raman bands in the region of 2800–3100 cm^{-1} , which are from C–H vibrations (see plate b). Pt is from Pt coating. (b) Raman spectrum of the epoxy resin used in the sample preparation. (c) SE image showing square trench milled by FIB and Raman spectra of FIB-milling epoxy resin in the trench. The Raman signals of FIB-milling epoxy resin almost completely disappear, reflecting that the chemical bonds of epoxy resin were completely destroyed. (d) SE image of FIB-prepared foil of epoxy resin and Raman spectra of the foil. The thickness of the foil is about 1.5 μm . The foil only contains a weak G band, but lacks the D1 band, suggesting that there is not aromatic carbon cluster in FIB-milling epoxy resin. The SE image shows that the vertical section milled by FIB is smooth and flat, indicating that FIB milling cannot induce “evaporation” of epoxy resin. (e) SE image of the foil of inclusion 5 and some representative Raman spectra in the inclusion. The highly disordered CM with characteristic G and D1 bands in the sample show striking difference from FIB-milling epoxy resin.

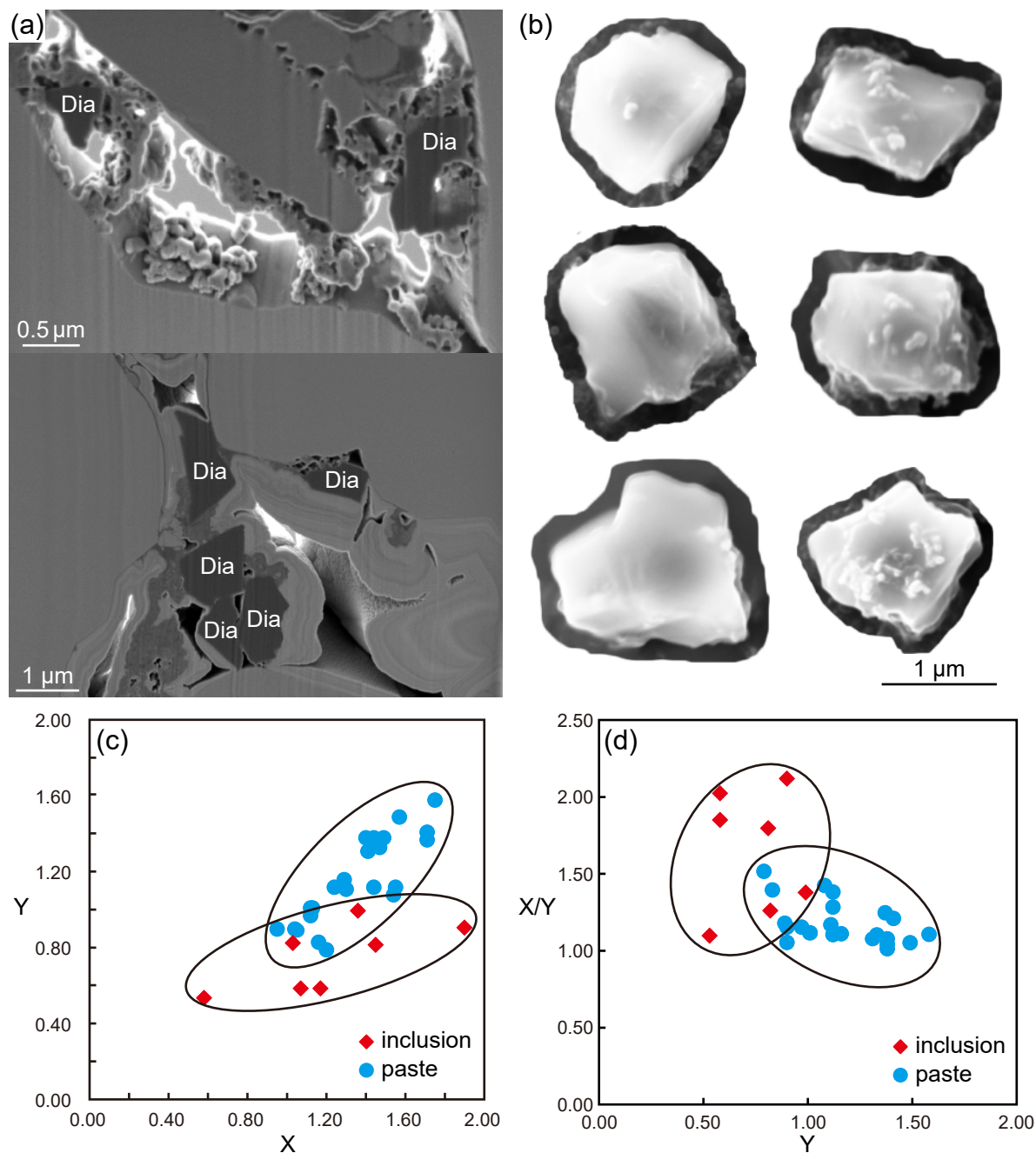


FIGURE S8. (a) SE images of diamonds in the inclusions. (b) SE images of diamonds in the paste. (c) and (d) The relationship between the major axis and the minor axis of diamond, X = the major axis, Y = the minor axis.