

***P-V* equation of State, thermal expansion, and *P-T* stability of synthetic zincochromite (ZnCr₂O₄ spinel)**

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ABSTRACT

The elastic properties and thermal behavior of synthetic zincochromite (ZnCr₂O₄) have been studied by combining room-temperature high-pressure (0.0001–21 GPa) synchrotron radiation powder diffraction data with high-temperature (298–1240 K) powder diffraction data. Elastic properties were obtained by fitting two Equations of State (EoS) to the *P-V* data. A third-order Birch-Murnaghan model, which provides results consistent with those from the Vinet EoS, yields: $K_0 = 183.1(\pm 3.5)$ GPa, $K' = 7.9(\pm 0.6)$, $K'' = -0.1278$ GPa⁻¹ (implied value), at $V_0 = 577.8221$ Å³ (fixed). Zincochromite does not exhibit order-disorder reactions at high temperature in the thermal range explored, in agreement with previous studies. The volume thermal expansion was modeled with $\alpha_V = \alpha_0 + \alpha_1 T + \alpha_2/T^2$, where only the first coefficient was found to be significant [$\alpha_0 = 23.0(4) \cdot 10^{-6}$ K⁻¹]. Above 23 GPa diffraction patterns hint at the onset of a phase transition; the high pressure phase is observed at approximately 30 GPa and exhibits orthorhombic symmetry. The elastic and thermal properties of zincochromite were then used to model by thermodynamic calculations the *P-T* stability field of ZnCr₂O₄ with respect to its oxide constituents (Cr₂O₃ and rocksalt-like ZnO). Spinel is expected to decompose into oxides at about 18 GPa and room temperature, in absence of sluggish kinetics.