

The nuclear and magnetic structure of “white rust”— $\text{Fe}(\text{OH}_{0.86}\text{D}_{0.14})_2$

**JOHN B. PARISE,^{1,*} WILLIAM G. MARSHALL,² RONALD I. SMITH,² H.D. LUTZ,³
AND HENDRIK MÖLLER³**

¹Center for High Pressure Research and Department of Geosciences and Department of Chemistry, State University of New York,
Stony Brook, New York 11794-3400, U.S.A.

²ISIS Neutron Facility, CLRC Rutherford Appleton Laboratory, Chilton, Didcot, Oxon OX11 0QX, U.K.

³Anorganische Chemie I, Universität Siegen, D-57068 Siegen, Germany

ABSTRACT

The structure of $\text{Fe}(\text{OH}_{0.86}\text{OD}_{0.14})_2$ was refined by applying the Rietveld technique to neutron powder diffraction data collected at 300, 150, and 10 K. The nuclear structure, of the CdI_2 type ($P\bar{3}m1$) and isostructural with Mg, Ca, Mn, Ni, and $\beta\text{-Co}(\text{OH})_2$, is maintained over the range of temperatures studied. At 10 K, the magnetic structure is ordered antiferromagnetically, with the magnetic moments aligned in the basal plane. The refined Fe^{2+} magnetic moment magnitude is $3.50(4) \mu_B$. This magnetic structure (space group $P_{2c}\bar{1}$) is unique amongst those encountered in the transition metal dihydroxides, for which the moment is either directed along the c axis [$\beta\text{-Co}(\text{OH})_2$ and $\text{Ni}(\text{OH})_2$], or at least has a considerable component along this direction [$\text{Mn}(\text{OH})_2$]. The dependence of the strength of potential hydrogen bonds with temperature is discussed.