

## Supplemental Tables

**TABLE OM1.** Complete calculated X-ray powder diffraction data ( $\text{CuK}\alpha$ ) for alumino-oxy-rossmanite.

<i>h</i>	<i>k</i>	<i>l</i>	<i>d<sub>calc.</sub></i>	<i>I<sub>calc.</sub></i>
1	1	0	7.9015	2
<b>1</b>	<b>0</b>	<b>1</b>	<b>6.2938</b>	<b>28</b>
0	2	1	4.9229	23
0	3	0	4.5620	17
<b>2</b>	<b>1</b>	<b>1</b>	<b>4.1783</b>	<b>61</b>
<b>2</b>	<b>2</b>	<b>0</b>	<b>3.9508</b>	<b>85</b>
<b>0</b>	<b>1</b>	<b>2</b>	<b>3.4307</b>	<b>55</b>
1	3	1	3.3461	15
4	0	1	3.0812	4
1	4	0	2.9865	17
<b>1</b>	<b>2</b>	<b>2</b>	<b>2.9236</b>	<b>78</b>
3	2	1	2.8707	8
3	3	0	2.6338	2
3	1	2	2.5904	7
<b>0</b>	<b>5</b>	<b>1</b>	<b>2.5534</b>	<b>100</b>
2	4	1	2.4297	4
0	0	3	2.3626	10
2	3	2	2.3501	13
5	1	1	2.3224	22
0	6	0	2.2810	2
5	0	2	2.1663	9
4	3	1	2.1445	13
0	3	3	2.0979	6
3	0	3	2.0979	5
4	2	2	2.0892	7
2	2	3	2.0277	12
<b>1</b>	<b>5</b>	<b>2</b>	<b>2.0198</b>	<b>39</b>
1	6	1	2.0021	8
4	4	0	1.9754	2
<b>3</b>	<b>4</b>	<b>2</b>	<b>1.8995</b>	<b>30</b>
3	5	1	1.8847	2
1	4	3	1.8529	5
6	2	1	1.8333	8
3	3	3	1.7587	6
1	0	4	1.7573	3
0	2	4	1.7154	2
2	6	2	1.6731	4
0	6	3	1.6410	16

6 0 3	1.6410	3
2 7 1	1.6273	12
5 5 0	1.5803	14
4 0 4	1.5734	3
4 5 2	1.5708	4
8 1 1	1.5624	3
3 2 4	1.5431	2

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*Notes:* Calculated using single-crystal structural data of sample PINK1. Only reflections with

$I_{\text{calc}} > 1$  are listed. The eight strongest reflections are in bold.

**TABLE OM2.** Theoretical compositions of known and potential Al-rich and Li-bearing tourmalines.

	1	2	3	4	5	6	7	8
SiO <sub>2</sub>	31.90	38.65	39.03	38.49	38.40	38.12	37.41	38.12
Al <sub>2</sub> O <sub>3</sub>	54.14	46.45	44.15	40.82	40.73	43.12	47.61	37.73
B <sub>2</sub> O <sub>3</sub>	11.09	11.20	11.30	11.15	11.12	11.04	10.84	11.04
Li <sub>2</sub> O	-	0.80	1.62	2.39	2.39	1.58	-	3.16
Na <sub>2</sub> O	-	-	-	3.31	3.30	3.28	3.22	-
CaO	-	-	-	-	-	-	-	5.93
H <sub>2</sub> O	2.87	2.90	3.90	3.85	2.88	2.86	0.93	2.86
F	-	-	-	-	2.02	-	-	2.01
X site	□	□	□	Na	Na	Na	Na	Ca
Y site	Al <sub>3</sub>	Al <sub>2.5</sub> Li <sub>0.5</sub>	Al <sub>2</sub> Li	Li <sub>1.5</sub> Al <sub>1.5</sub>	Li <sub>1.5</sub> Al <sub>1.5</sub>	Al <sub>2</sub> Li	Al <sub>3</sub>	Li <sub>2</sub> Al
Z site	Al <sub>6</sub>	Al <sub>6</sub>	Al <sub>6</sub>	Al <sub>6</sub>	Al <sub>6</sub>	Al <sub>6</sub>	Al <sub>6</sub>	Al <sub>6</sub>
B site	B <sub>3</sub>	B <sub>3</sub>	B <sub>3</sub>	B <sub>3</sub>	B <sub>3</sub>	B <sub>3</sub>	B <sub>3</sub>	B <sub>3</sub>
T site	Si <sub>5</sub> Al	Si <sub>6</sub>	Si <sub>6</sub>	Si <sub>6</sub>	Si <sub>6</sub>	Si <sub>6</sub>	Si <sub>6</sub>	Si <sub>6</sub>
V site	(OH) <sub>3</sub>	(OH) <sub>3</sub>	(OH) <sub>3</sub>	(OH) <sub>3</sub>	(OH) <sub>3</sub>	(OH) <sub>3</sub>	O <sub>2</sub> (OH)*	(OH) <sub>3</sub>
W site	O	O	OH	OH	F	O	O*	F

Notes: 1: Alumino-oxy-rossmanite, □Al<sub>3</sub>Al<sub>6</sub>(Si<sub>5</sub>AlO<sub>18</sub>)(BO<sub>3</sub>)<sub>3</sub>(OH)<sub>3</sub>O; 2 “oxy-rossmanite”,

□(Li<sub>0.5</sub>Al<sub>2.5</sub>)Al<sub>6</sub>Si<sub>6</sub>O<sub>18</sub>(BO<sub>3</sub>)<sub>3</sub>(OH)<sub>3</sub>O; 3: Rossmanite, □(LiAl<sub>2</sub>)Al<sub>6</sub>Si<sub>6</sub>O<sub>18</sub>(BO<sub>3</sub>)<sub>3</sub>(OH)<sub>3</sub>OH; 4:

Elbaite, Na(Li<sub>1.5</sub>Al<sub>1.5</sub>)Al<sub>6</sub>Si<sub>6</sub>O<sub>18</sub>(BO<sub>3</sub>)<sub>3</sub>(OH)<sub>3</sub>OH; 5: Fluor-elbaite,

Na(Li<sub>1.5</sub>Al<sub>1.5</sub>)Al<sub>6</sub>Si<sub>6</sub>O<sub>18</sub>(BO<sub>3</sub>)<sub>3</sub>(OH)<sub>3</sub>F; 6: Darrellhenryite, Na(LiAl<sub>2</sub>)Al<sub>6</sub>Si<sub>6</sub>O<sub>18</sub>(BO<sub>3</sub>)<sub>3</sub>(OH)<sub>3</sub>O; 7:

Olenite NaAl<sub>3</sub>Al<sub>6</sub>Si<sub>6</sub>O<sub>18</sub>(BO<sub>3</sub>)<sub>3</sub>(O)<sub>3</sub>OH; 8: Fluor-liddicoatite, Ca(Li<sub>2</sub>Al)Al<sub>6</sub>Si<sub>6</sub>O<sub>18</sub>(BO<sub>3</sub>)<sub>3</sub>(OH)<sub>3</sub>F;

\*To produce an ordered end-member formula of olenite, the (OH) group can be assigned to the V site although it occupies the W site.