

ELECTRONIC SUPPORTING INFORMATION FOR:

Ca  $L_{2,3}$ -edge near edge X-ray absorption fine structure of tricalcium aluminate, gypsum and calcium (sulfo)aluminate hydrates

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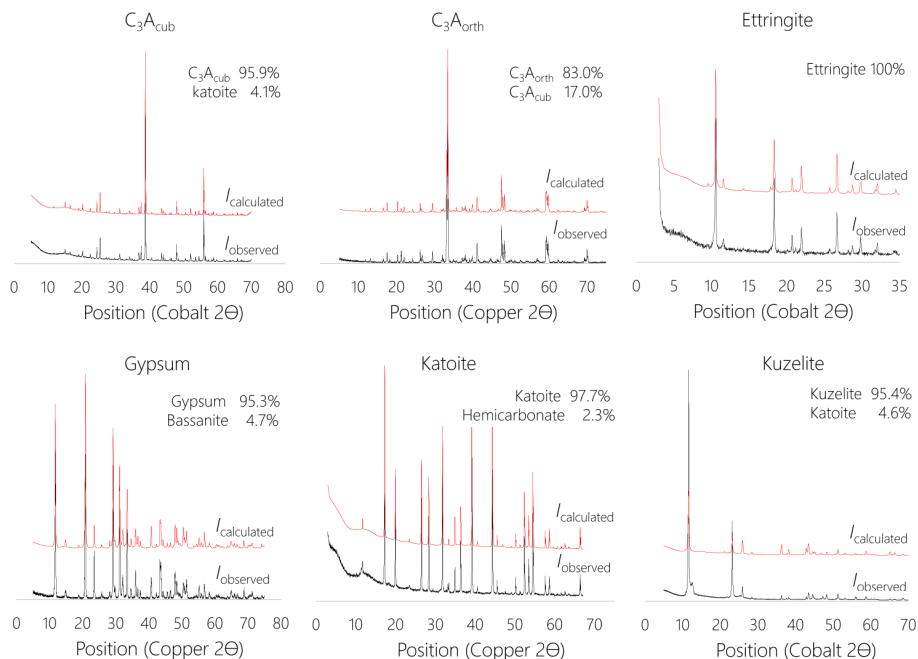
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In the present work,  $C_3A_{cub}$ ,  $C_3A_{orth}$ , ettringite and katoite were purchased from Mineral Research Processing (<http://www.mineralresearchprocessing.fr/>). Gypsum was purchased from Fisher Scientific (#S76764). Kuzelite were synthesized by stoichiometrically mixing  $C_3A_{cub}$ , gypsum and deionized water, and storing in seal plastic bottle for 17 days. The crystal structures and chemical compositions of were verified by powder X-ray diffraction performed with PANalytical X'Pert Pro diffractometer operating at 40 keV and 40 mA with a Cobalt anode for  $C_3A_{cub}$ , ettringite and kuzelite, and a Copper anode for  $C_3A_{orth}$ , gypsum and katoite. Sample purities were further verified by Rietveld refinement with software HighScorePlus and ICSD database (Supplementary Figure 1).

Thermogravimetric analysis (TGA) indicates total weight loss of 2.5% up until 1000 °C. X-ray fluorescence measurement of cubic  $C_3A$  and gypsum were performed with Philips PW2400 Wavelength-Dispersive XRF. The overall cation oxide impurity/CaO weight ratios are 0.02 in  $C_3A_{cub}$  and 0.003 in gypsum.  $C_3A_{orth}$  has atomic ratio of  $Na_2O:CaO:Al_2O_3 = 0.187:2.92:1$ , with 1.07 wt% impurities.

For reading convenience, Supplementary Table 1 lists peak positions of displayed spectra in Figure 3a in the paper. Ca in katoite,  $C_3A_{cub}$  and  $C_3A_{orth}$  are in cubic-like symmetry, while Ca in kuzelite, ettringite, gypsum are in octahedral-like symmetry.

**SUPPLEMENTARY FIGURE 1.**  
Powder XRD and Rietveld refinement of  $C_3A_{cub}$ ,  $C_3A_{orth}$ , ettringite, gypsum, katoite and kuzelite.



**SUPPLEMENTARY TABLE 1.** Peak positions in the Ca L<sub>2,3</sub>-edge NEXAFS spectra shown in Figure 3, eV

Symmetry	Mineral	1	a <sub>1</sub>	2	a <sub>2</sub>	b <sub>10</sub>	b <sub>1</sub>	b <sub>2</sub>	a <sub>2</sub> -a <sub>1</sub>	b <sub>2</sub> -b <sub>1</sub>
Cubic	katoite	346.85	347.43	348.24	349.13	–	351.17	352.38	1.70	1.21
	C3Acub	346.86	347.57	348.31	349.14	–	351.18	352.46	1.57	1.28
	C3Aorth	346.80	347.57	348.14	349.11	–	351.20	352.40	1.54	1.20
Octahedral		1	2	a <sub>1</sub>	a <sub>2</sub>	b <sub>10</sub>	b <sub>1</sub>	b <sub>2</sub>	a <sub>2</sub> -a <sub>1</sub>	b <sub>2</sub> -b <sub>1</sub>
	gypsum	347.00	347.56	348.21	349.06	350.82	351.43	352.33	0.85	0.90
	kuzelite	346.85	347.48	348.08	349.01	–	351.13	352.31	0.93	1.18
	ettringite	346.88	347.54	348.13	349.13	–	351.38	352.45	1.00	1.07

*Note:* Software aXis2000 allows peak position determination with sub-step size accuracy, with uncertainty of ~0.02 eV at the studied energy range.