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### 3 **High-pressure synchrotron single crystal X-ray diffraction study of lillianite.**

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### 13 **Abstract**

14 In the present paper, high pressure data from a synchrotron X-ray diffraction study on a lillianite  
15 ( $\text{Pb}_3\text{Bi}_2\text{S}_6$ ) single crystal up to approximately 21 GPa are presented.

16 A phase transition from lillianite ( $Bbmm$  space group #63, LP lillianite) to the high-pressure form  $\beta$ -  
17  $\text{Pb}_3\text{Bi}_2\text{S}_6$  ( $Pbnm$  space group, HP lillianite) was confirmed and bracketed between 4.90 and 4.92 GPa.

18 The transition is reversible but of first-order with a hysteresis of  $\sim 2.8$  GPa. The phase transition to  $\beta$ -  
19  $\text{Pb}_3\text{Bi}_2\text{S}_6$  showed weak effects of pseudomeroheredral twinning that disappeared upon decompression,  
20 testifying to a full recovery of the single crystal of lillianite. This makes lillianite an interesting shape-  
21 memory material.

22 With a bulk modulus  $K_{4.9} = 78(3)$  GPa and  $K' = 5.1(4)$ ,  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  is markedly less compressible than  
23 lillianite [ $K_0 = 44(2)$  GPa,  $K' = 7(1)$ ]. Compressional anisotropy increased markedly in  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  with  
24 compressibility along the  $b$  axis [ $M_{0b} = 130(6)$  GPa and  $M'_b = 19(3)$  in lillianite,  $M_{4.9b} = 145(4)$  GPa and

25  $M'_b = 16.0(7)$  in  $\beta\text{-Pb}_3\text{Bi}_2\text{S}_6$ ] significantly larger than along the other two axes [ $M_{0a} = 118(5)$  GPa,  $M'_a$   
26  $= 21(3)$ ,  $M_{0c} = 139(12)$  GPa and  $M'_c = 31(10)$  in lillianite,  $M_{4.9a} = 242(12)$  GPa,  $M'_a = 8(1)$ ,  $M_{4.9c} = 242(5)$   
27 GPa and  $M'_c = 29(1)$  in  $\beta\text{-Pb}_3\text{Bi}_2\text{S}_6$ ].

28 The behavior of lillianite at high pressure is an interesting case study in relation to non-quenchable ultra-  
29 high-pressure phases likely occurring in the inner Earth, like post-perovskite  $\text{MgSiO}_3$ , the oxide  
30 homologue  $N = 1$  of the lillianite series. The  $\beta\text{-Pb}_3\text{Bi}_2\text{S}_6$  structure, on the other hand, is the  $N = 3$   
31 homologue of the meneghinite series where also the higher-pressure modification of the post-perovskite  
32 structure belongs (homologue  $N = 1$ ). This makes the two forms of  $\text{Pb}_3\text{Bi}_2\text{S}_6$  potential equivalents of  
33 high- and ultra-high-pressure Mg silicates that could occur both in the deep earth mantle and in other  
34 rocky extrasolar planetary bodies.

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### Keywords

37 Lillianite; high-pressure; synchrotron; single crystal X-ray diffraction; phase transition; shape-memory

38

## Introduction

39

40 During the last decades, the crystal structure and stability of main-group heavy metal chalcogenides have  
41 attracted the interest of the scientific community as likely candidates for renewable energy applications  
42 (Olvera et al. 2015 and references therein). In particular, sulfosalts have been claimed as promising  
43 materials able to both absorb sunlight more efficiently than silicon (Wallace et al. 2017) and to avoid the  
44 toxicity issues found with the lead- and tin-halide perovskite (Kanno et al. 2019) in the thin-film solar  
45 cells technology. Sulfosalts are complex sulfides of As, Sb and Bi with one or more metal cations.  
46 Because of the presence of the  $s^2$  lone electron pair (LEP) of the trivalent metalloids together with the  
47 nature of the metalloids-sulfur bond, sulfosalts have specific semiconducting properties (Olsen et al.  
48 2008).

49 Some of the sulfosalts investigated at high-pressure showed interesting pressure-induced structural phase  
50 transitions, e.g. heyrovskyite  $\text{Pb}_6\text{Bi}_2\text{S}_9$  (Olsen et al. 2011) and chalcostibite  $\text{CuSbS}_2$  (Comodi et al. 2018).  
51 Pressure appears to be an important parameter, which could be used to either tune or significantly change  
52 the transport and structural properties of these thermoelectric materials.

53 The crystal structure of lillianite (Takagi and Takéuchi 1972; Figure 1a) consists of (010) slabs that have  
54 internal topology of a  $\text{NaCl}(\text{PbS})$ -type. The boundary of slabs is cut along the  $\{311\}$  planes of the PbS  
55 archetype. The neighboring slabs are in a twin relation relative to their boundary. The crystal structure is  
56 characterized by two relatively long periods (along  $b$  axis perpendicular to slabs, and along  $a$  axis  
57 corresponding to  $[2-3-3]$  direction of the PbS archetype, respectively) and one short period along  $c$  axis  
58 matching the  $[011]$  direction of the PbS archetype. The symmetry elements of the PbS archetype that  
59 remain parallel in both orientations of slabs are the two-fold axes along the  $[011]_{\text{PbS}}$  and the mirror planes  
60 perpendicular to them, that is, the  $c$  axis and the (001) plane in the lillianite crystal structure, respectively.  
61 The structure contains three distinct metal sites, M1, M2 and M3 as shown in Figure 1a. Based on the

62 bond distances, M3 site is fully occupied by Pb, whereas M1 and M2 sites have a mixed (Pb,Bi)  
63 occupancy with a likely preference for Bi in M1 and for Pb in M2 (Pinto et al. 2006). Due to their similar  
64 and high atomic numbers, it is not possible to refine the amounts of Bi and Pb in the mixed sites from X-  
65 ray diffraction data. Therefore, various authors used a fixed 1:1 occupancy of Bi and Pb on M1 and M2  
66 sites in lillianite crystal structure refinements (Takagi and Takeuchi 1972; Olsen et al. 2008). Lillianite  
67 has an ordered polymorph, the monoclinic  $C2/m$  xilingolite (Berlepsch et al. 2001a). The main difference  
68 is that in a slab M1 sites are occupied predominantly with Pb and the M2 sites with Bi, whereas in a  
69 successive slab the situation is reversed. In other words, all slabs with the same inclination of atomic  
70 planes have Bi predominantly occupying M1, whereas the slabs with the opposite inclination have Bi  
71 dominating in M2.

72 Lillianite is a part of a homologous series where all members share the described topological features but  
73 differ in the thicknesses of slabs. The standard classification of the lillianite homologues (Makovicky  
74 1977) defines the thicknesses of the slabs through the number of the coordination octahedra in a row  
75 running across the slab along the [100] direction of the PbS archetype. In the standard notation that takes  
76 into consideration the existence of homologues with two slabs of different thicknesses, lillianite is the  
77  $4^4L$  member of the series, meaning that it has four octahedra in a row across the slab for both slabs that  
78 are in a twin relation. In the case of homologues with slabs of equal thicknesses, their boundaries are also  
79 mirror planes of the crystal structure, which, combined with the earlier mentioned symmetry elements,  
80 results in an overall orthorhombic symmetry ( $Bbmm$  space group #63).

81 In the pressure range between 3.7 and 4.9 GPa lillianite has been shown to undergo a phase transition  
82 with a volume decrease of approximately 4% (Olsen et al. 2008). The crystal structure of the  $\beta$ - $Pb_3Bi_2S_6$   
83 is shown in Figure 1b. In  $\beta$ - $Pb_3Bi_2S_6$ ,  $n$ -glide planes replace the mirror planes perpendicular to the  $b$ -axis  
84 present in the low-pressure form. The result of the phase transition is a lowering of symmetry from  $Bbmm$   
85 in lillianite ( $LP$  polymorph) to  $Pbnm$  space group in  $\beta$ - $Pb_3Bi_2S_6$  ( $HP$  polymorph), and an increase in the

86 number of independent atomic positions from 10 to 22. The PbS-like slabs in lillianite split in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>  
87 by means of crystallographic shear into double layer rods that extend along the *c* axis. The rods are  
88 rotated around their central 2<sub>1</sub> axes, reducing the inclination angle of their longest diameter to the *b* axis,  
89 which leads to an expansion in the *b*-axis direction and a contraction in the *a*-axis direction. The rotation  
90 is connected to a sliding of the adjacent rods with a relative movement between two parallel rods  
91 corresponding to half a coordination octahedron diameter or the sliding component  $[0 \frac{1}{2} \frac{1}{2}]_{\text{PbS}}$ .  
92 The *LP* and *HP* polymorphs are substantially different in terms of the archetypal structure, which is PbS-  
93 like in lillianite and SnS-like in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>, with an obviously larger stereochemical activity of the LEPs  
94 of M atoms in the latter (Olsen et al. 2008). Accordingly,  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> structurally belongs to the  
95 meneghinite homologous series (Olsen et al. 2008). Due to the lowering of symmetry on phase transition,  
96 some of the symmetry equivalent atomic sites in lillianite split in two symmetry-independent sites in  $\beta$ -  
97 Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>. The here used labeling of atomic sites in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> is different from what Olsen et al. (2008)  
98 used in order to ease the comparison between the *LP* and *HP* forms of Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>. In our notation, all sites  
99 in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> retain the same numeric descriptors as the corresponding sites in lillianite and the split sites  
100 get additionally either A or B descriptors. The sites related to M1 in lillianite get labels M1A and M1B  
101 in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> etc. The reader can consult Table S1 in *Supplementary material* section for the relations  
102 between the present and Olsen et al. (2008) site notations for  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>.  
103 In this work, we extended the investigated pressure range up to ~21 GPa. The new data allow us to  
104 determine the Equation of State of the  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>. Moreover, the crystal structure of the *HP* lillianite  
105 polymorph is revised and accurately described by collecting several high-resolution data both during  
106 compression and decompression. With the present results, the reversibility as well as the shape-memory  
107 character of the phase transition is further clarified.

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109

## Materials and methods

## 110 **Data collection and extraction**

111 The examined sample was a single lillianite crystal ( $31 \times 12 \times 10 \mu\text{m}^3$ ), optically clear at  $10 \mu\text{m}$  thickness,  
112 coming from the same synthesis as the crystals used by Olsen et al. (2008). The crystal quality was  
113 checked by means of in-house Single Crystal X-Ray Diffraction (SC-XRD) analysis at the Department  
114 of Physics and Geology of the University of Perugia (Italy) confirming the sharpness of the diffraction  
115 profiles and the lack of twinning.

116 The HP SC-XRD study was carried out at the ID-15B beamline at ESRF (Grenoble) by means of a  
117 membrane-type Diamond Anvil Cell with an opening angle of 64 degrees, equipped with  $600 \mu\text{m}$  culet  
118 diamonds. Helium was used as a pressure transmitting medium and a Ruby sphere was loaded as a  
119 fluorescent *P* calibrant together with the samples. The pre-indented stainless-steel gasket had  $300 \mu\text{m}$   
120 diameter hole and  $80 \mu\text{m}$  thickness.

121 A fixed monochromatized X-ray beam ( $\lambda = 0.41125 \text{ \AA}$ ) was used, focused down to  $10 \times 10 \mu\text{m}^2$  area.  
122 Data were collected with angular step of  $0.5^\circ$  and counting time of 1s per step and registered by means  
123 of a MAR555 flat panel detector with  $430 \times 350 \text{ mm}^2$  (555mm diagonal) active area. Both the sample-  
124 to-detector distance (279.88 mm) and the wavelength were calibrated using Si standard and Fit2D  
125 software (Hammersley et al. 1996).

126 Pressure was varied in the range from 0 to approximately 21 GPa. Measurements were performed in 10  
127 steps on increasing pressure and 5 steps during decompression to evaluate the reversibly and the likely  
128 occurrence of hysteresis phenomena. Pressure was measured before and after each data collection and  
129 the gasket-relaxation was ensured by waiting approximately 15 minutes after changing the load.

130 The data extraction together with the reflection intensity correction, the merging of reflections and the  
131 refinements of the crystal lattice parameters were done by means of the CrysAlisPro software (Agilent  
132 Technologies UK Ltd, Yarnton, England.). The absorption correction of the extracted data was applied  
133 in two steps: in the first one, by means of the multi-scan method (Blessing 1995) as implemented in

134 WinGX (Farrugia 1999). The correction factors for intensities were optimized without assuming any  
135 physical cause like absorption. In the second step, the ABSORB-GUI software (Angel and Gonzalez-  
136 Platas 2013) was used to correct for DAC absorption and gasket shadowing.

### 137 **Compressibility calculations**

138 The compressibility of the *LP* and the *HP* lillianite phases was calculated from measured unit cell  
139 parameters. The data treatment and Equation of State (EoS) calculations were done using EosFit GUI  
140 (Gonzalez-Platas et al. 2016). The behavior of unit cell parameters with *P* for both lillianite and  $\beta$ -  
141  $\text{Pb}_3\text{Bi}_2\text{S}_6$  is shown in Figure 2 together with data from Olsen et al. (2008). Results of the EoS calculation  
142 are given in Table 1. We calculated the compressibility parameters of both lillianite and  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  using  
143 combined data of our measurements and those published by Olsen et al. (2008) for their crystal *lill2*. Due  
144 to the hysteresis effects and lower quality of refinements, our data obtained during decompression, were  
145 not used for the calculations.

### 146 **Structure refinements and analysis**

147 The structure refinements were performed starting from the crystal structures published in Olsen et al.  
148 (2008) based on  $F^2$  by using ShelXle software (Hübschle et al. 2011) and the scattering factors for neutral  
149 atoms.

150 Given the similar atomic numbers of the heavy Bi and Pb atoms, their mixed occupancies in M1 and M2  
151 in lillianite as well as M1A, M1B, M2A and M2B in  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  were fixed to 0.5 during refinement.  
152 Insights on the likely proportion of Bi in the mixed sites were attempted by applying the model of Brown  
153 and Altermatt (1985) for the relation bond length – bond valence and assuming the following relation:

$$154 \quad xv_{i1} + (1-x)v_{i2} = xv_{c1} + (1-x)v_{c2},$$

155 where  $x$  is the proportion of the first species,  $v_{i1}$  and  $v_{i2}$  are the expected valences of the first and the  
156 second species, and  $v_{c1}$  and  $v_{c2}$  are valences calculated for the site using the bond valence constants for

157 the first and the second species, respectively. Further details on the used procedure are given in the  
158 *Supplementary material* section together with the obtained results.

159 Anisotropic thermal parameters were refined for Pb, Bi and S atomic positions at each pressure, with the  
160 exception of the first data collection after the phase transition (6.3 GPa) as well as data collections of  $\beta$ -  
161  $\text{Pb}_3\text{Bi}_2\text{S}_6$  during decompression, where thermal parameters for sulfur atoms were refined isotropically.

162 The good crystallinity of the lillianite crystal was maintained during the whole experiment. After the  
163 phase transition, the appearance of additional weak reflections was interpreted as twinning induced by  
164 the phase transition. This observation is supported by a successful indexing of the additional reflections  
165 using the unit cell parameters obtained from the main reflections. In Figure S1, in the *Supplementary*  
166 *material* section, the reciprocal lattice images for the crystal at 1.74 GPa (single lillianite crystal), 6.30  
167 GPa (twinned  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$ ) and 2.13 GPa during decompression (the detwinned single lillianite crystal)  
168 are shown. Twinning persisted both during compression and decompression of the HP lillianite phase;  
169 whereas it disappeared after the low-pressure phase recovery. The attempted twin refinements resulted  
170 in inferior R factors compared to single-domain refinements, because the mass proportion of the second  
171 component refined to only 6%. Therefore, we used only the reflections of the main twin component for  
172 the crystal structure analysis.

173 The details of data collection and refinement, together with the crystal structure data, can be found in the  
174 CIF file (deposited). Bond lengths and polyhedral volumes at different pressures, are reported in Table  
175 S2 and S3 in the *Supplementary material* section, for the LP and the HP lillianite, respectively.

176 To calculate the deviations of atomic positions in  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  from those of LP lillianite, we used the  
177 program AMPLIMODES (Orobengoa et al. 2009) from the Bilbao Crystallographic Server  
178 (<https://www.cryst.ehu.es/>). The calculations were made comparing the crystal structure data of lillianite  
179 at 1 MPa and 4.9 GPa with those of  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  at 6.3 GPa. Comparing the data for lillianite for both  
180 room pressure and right before the phase transition to those of  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$ , gives us a possibility also to



181 follow the changes in the structure of lillianite over its stability range by subtracting the two results (see  
182 further explanation below, paragraph “Crystal structure evolution with pressure”). The results are  
183 represented in Table 2.

184 Moreover, the distortion of coordination polyhedra was analyzed with a help of program Ivton (Balić  
185 Žunić and Vicković 1996). Three parameters were taken into account for which a prerequisite is a  
186 calculation of the best-fitting circumscribed sphere to the coordination polyhedron (Balić Žunić and  
187 Makovicky 1996): (i) eccentricity (Balić Žunić and Makovicky 1996) (ii) asphericity (Balić Žunić and  
188 Makovicky 1996; Balić-Žunić 2007) and (iii) volume distortion (Makovicky and Balić-Žunić 1998).  
189 Results are given the *Supplementary material* section (Figure S2).

## 190 **Results and discussion**

### 191 **Compressibility**

192 A discontinuity was observed after 4.9 GPa with a decrease of the  $a$  parameter (Figure 2b) and an increase  
193 of the  $b$  parameter (Figure 2b) of approximately 16% and 14%, respectively, in agreement with data  
194 published by Olsen et al. (2008). The evolution of cell parameters led to a decrease of the unit cell volume  
195 on phase transition from 1057.16(6) to 1006.70(6) Å<sup>3</sup> corresponding to approximately 5%.

196 The well covered  $P$  range both in the lillianite field, thanks to the combination of our compression results  
197 and those of Olsen et al. (2008) obtained from their second crystal labelled *lill2*, and in the  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>  
198 field up to ~21 GPa, allowed us to accurately calculate the EoS for both phases. A comparison of the  
199 compressional behavior of the two polymorphs, together with data from Olsen et al. (2008), is given in  
200 Figure 3 where confidence ellipses are built according to Bass et al. (1981). A negative correlation  
201 between  $K_0$  and  $K'$  is observed for both lillianite and  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> as the marked elongation of the ellipses  
202 with a negative slope. The present data showed that lillianite is more compressible with a higher first  
203 derivative than what suggested by Olsen et al. (2008) (Table 1, Figure 3); whereas, the remarkably

204 smaller bulk compressibility of  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> than that of lillianite at the transition pressure, as proposed  
205 by Olsen et al. (2008), is confirmed (Table 1, Figure 3). Our improved data show an even larger bulk  
206 modulus for the high-pressure polymorph with a smaller first derivative meaning that the compressibility  
207 of the HP lillianite polymorph remains almost the same in the investigated pressure range. The poor  
208 agreement of the presented data with those from Olsen et al. (2008) is likely due to insufficient data  
209 statistics caused by the restricted *P* range that was analyzed by the latter.

210 By the present data the axial moduli were also calculated for both LP and HP polymorphs (Table 1).  
211 Results showed that in lillianite the *a* axis is the most compressible one, whereas in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> the *b*  
212 parameter is markedly more compressible than *a*. The *c* parameter is the least compressible in both  
213 polymorphs with remarkably higher *M'* value expressing a marked stiffening as pressure increases.  
214 Further discussions on the anisotropic compressional behavior of lillianite and in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> are given  
215 below (see paragraph “Anisotropic compression in LP and HP lillianite structures”).

## 216 **Crystal structure evolution with pressure**

217 The indexation of the collected reflections showed that the phase transition from lillianite to  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>  
218 occurs at a pressure higher than 4.9 GPa. Together with the previous results of Olsen et al. (2008), this  
219 helps to bracket the phase transition to a very narrow range between 4.90 and 4.92 GPa. The evolution  
220 of the crystal structure during decompression showed that the  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> completely reverses to the low-  
221 pressure lillianite, with hysteresis since we observed that  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> remained stable during  
222 decompression down to 4.30 GPa and the reverse phase transition occurred between 4.30 and 2.13 GPa  
223 (Figure 2). Therefore, the phase transition is confirmed to be of first order, strongly anisotropic. It is  
224 reversible (Olsen et al. 2008) and involves a hysteresis behavior (present data).

225 **Phase transition of lillianite revisited.** The availability of crystal structure data over a significantly  
226 increased range of pressures, allowed a deep discussion of the lillianite behavior.

227 Displacements of atoms upon the transition are small but indicative, as follows. S2 that lies on the two-  
228 fold axis keeps its position unchanged in accordance with the symmetry constraints ( $2/m$ ). Inside its  
229 coordination, M3 site moves towards the plane defined by the four S4 atoms and away from the S3 atoms.  
230 The latter move in the opposite direction further increasing their distance to M3. S1 atoms (capping atoms  
231 of the trigonal prism) move towards the (010)  $m$  symmetry plane and M3. M2 moves towards the  
232 coordinated S1 on the same  $z$  level thus shortening its longest bond. M1 atoms move in approximately  
233 the same direction as the S4 atoms to which they are bonded. All these changes are a prelude to the phase  
234 transition, during which the displacements of atoms from the positions in the low-pressure phase are  
235 tremendously emphasized, some of them exceeding 1.5 Å (e.g.  $|\Delta|_{M1B} = 1.5304$  Å and  $|\Delta|_{S4B} = 1.5845$  Å,  
236 see Table 2). They are illustrated on Figure 4 and Table 2 where the active modes in both lillianite and  
237  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> are shown to be  $\Delta_x$  and  $\Delta_y$ . The strains accumulated during compression of lillianite are  
238 released upon transition with the spontaneous movement of atomic sites that break the  $Bbmm$  symmetry.  
239 Site M3, which before the phase transition was approaching S4 atoms, now changes the direction of  
240 displacement and moves away from the former mirror plane, becoming a part of a rigid rod. As a  
241 consequence, its coordination changes drastically. The body of its bicapped trigonal prism (prism axis  
242 parallel  $c$ ) is made of two S3 and four S4 atoms, either before and after the transition. The two caps are  
243 the two S1 atoms in lillianite. In  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>, one of them (S1A) is displaced far away from M3 and out  
244 of the coordination, being  $|\Delta|_{S1'} = -0.0285$  and  $|\Delta|_{M3} = 0.0059$  in lillianite,  $|\Delta|_{S1A} = 0.6511$  and  $|\Delta|_{M3} =$   
245  $1.0038$  in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> (Table 2); whereas, M3 moves in the direction of the other (S1B) (Table 2,  $|\Delta|_{S1} =$   
246  $0.0511$  in lillianite and  $|\Delta|_{S1B} = 1.0887$  in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>). An additional S3 atom approaches M3 and now  
247 forms the second cap on the face formed by two S4A and two S4B atoms. This functions as a rotation of  
248 the coordination hole, especially illustrative if we consider the coordination as a square antiprism as  
249 suggested by the volume distortion parameter (Figure S2c). Differently from Olsen et al. (2008) where  
250 the approaching of the additional S3 atom was ignored and the coordination of the M3 in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> was

251 considered as seven-fold, here we consider the coordination as eight-fold, given that it resides in the hole  
252 made by eight S atoms. The movements of M3 sites are in the [100] direction for the sites close to the  
253 same (010) plane and in the opposite direction [-100] in the successive (010) plane (Figure 4). The side  
254 movement from the common (010) plane is successively to the opposite directions for sites close to the  
255 same (010) plane, so that M3 atoms become parts of the rigid rods in the slabs to the left and to the right  
256 of the bordering (010) plane, respectively.

257 The coordination of the M1 and M2 sites changes from octahedral into lying mono-capped trigonal  
258 prismatic (or a split octahedron). The lone electron pairs of M1 and M2 atoms point in the resulting  
259 structure between the rigid rods. This means that some of them moved to the other side of the M atoms  
260 during the phase transition. The coordination number changes for M1 and M2 from 6 in lillianite to 7 in  
261  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> (M1A-M1B and M2A-M2B). Although the structural change guided by the phase transition  
262 produces the same coordination type for atoms M2A, M2B, M1A and M1B – CN7 in the form of a split  
263 octahedron – the movements of atoms that result in this coordination and the changes in bonding  
264 compared to the LP lillianite phase are different for the atoms that now form the pairs of symmetrically  
265 independent sites M2A-M2B and M1A-M1B. M2A and M1A flip their shortest and longest bonds during  
266 the phase transition connected to the formation of rigid rods and decollement of the inter-rod spaces.

267 The asphericity parameter of coordination distortion is lower for M3 in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> than it was in lillianite  
268 structure (Figure S2b), whereas all other coordinations, save M1A, have larger values testifying for larger  
269 distortion of this kind in the high-pressure polymorph, connected with the last distortion parameter, the  
270 volume distortion (Figure S2c). Together with distortions in lillianite and  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>, we indicated on  
271 Figure S2c the values for four ideal coordination types with CN8 (Archimedean and maximum volume  
272 square antiprism) or CN7 (split octahedron and maximum volume split octahedron) (Makovicky and  
273 Balic-Zunic 1998). We can see that M3 coordination fits well with the values for square antiprism both  
274 in lillianite and in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>. For CN7 coordinations, the values fall in the range of ideal split octahedron

275 rather than monocapped trigonal prism (volume distortion = 0.1590), which is a typical description of  
276 CN7 coordinations (e.g. Berlepsch et al. 2001b). The values again correlate with those of asphericity,  
277 M1A having the lowest volume distortion (highest volume efficiency) and M2B the highest distortions  
278 of both types. In lillianite, M1 and M2 had very small volume distortions, as well as asphericities and  
279 approached well ideal octahedron (Figures 3b and 3c). A general conclusion that comes out from the  
280 present data is that most of coordinations increase their distortions on phase transition. On further  
281 increase of pressure, only their eccentricity decreases (Figure S2a), testifying for the main compressional  
282 mechanism in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>. It should be mentioned, however, that M1A coordination stands out as very  
283 regular with even a negligible amount of eccentricity that it keeps throughout the investigated range of  
284 pressures.

285 The volumes of coordination polyhedra of Pb and Bi (Table S2 and S3 in the *Supplementary material*  
286 section) in dependence on pressure are represented in Figure 5. A difference between their evolution with  
287  $P$  is evident: while in M1 (A and B) and M2 (A and B), in the  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> structure, the polyhedral  
288 volumes decrease with a similar trend as observed for their “parent” sites in lillianite, M3 volume  
289 decreases with a clearly slower trend than M3 in lillianite (Figure 5). This is due to its special position in  
290 the structure, on the combination plane of the (010) slabs (Figure 1). The main compression of the  
291 structure occurs in diminishing the spaces between the rigid rods and thus highly affecting the volumes  
292 of all other coordination polyhedra but M3, that shows the lowest variations.

293 **Twining and de-twinning in lillianite under high-pressure.** In the present study, the observed  
294 splitting of reflections of  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> resembles a pseudomerohedral twin (Friese and Grzechnik 2014;  
295 Figure S1b in the *Supplementary material* section). One can observe a rotation of the crystal lattices of  
296 the two twin domains relative to each other by a small angle around the [001]. During decompression, a  
297 de-twinning process occurs together with the restoration of the structure of the original single lillianite  
298 crystal (Figure S2c in *Supplementary material* section). As in phase transition the symmetry of the low-

299 pressure form (*Bbmm*) changes to its sub-group (*Pbnm*) in the high-pressure polymorph, one would  
300 expect a merohedral type of twinning (Friese and Grzechnik, 2014) with a perfect overlapping of the two  
301 components. On the contrary, in the *HP* lillianite they are rotated relatively to each other. As described  
302 earlier, the pressure-induced phase transition in lillianite is characterized by splitting of the (010)  
303 structural slabs into strongly bonded two-atom layers thick rods and intra-rod spaces crossed by weak  
304 bonds. In general, an atomic plane has two arbitrary choices, either making tight bonds or weak bonds  
305 with the neighboring planes, thus defining a rod. Propagated through the crystal structure of the lillianite  
306 during transition, these two choices would make two mirror-related structures of the high-pressure form.  
307 This would produce a merohedral twinning unless an additional effect of the rod rotation were present.  
308 In the contact zone of the two twin domains, a compromise structure can be expected and it can be  
309 assumed that it retains or mimics the features of the low-pressure polymorph. Both polymorphs have  
310 (010) slabs with atomic planes inclined to the boundary (010) planes giving them a repeating fish-bone  
311 pattern when looked along the [001] direction (Figure 1). The inclination of the atomic planes to the  
312 (010) plane changes on phase transition, as mentioned in discussion of compression mechanism. At 4.9  
313 GPa, in the low-pressure phase just before the phase transition, this angle is 26.45°; in the high-pressure  
314 phase at 6.3 GPa, the angle is 31.52°. In  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  at 8.85 GPa, where we could best determine the  
315 orientation matrix for the small twin component, the relative rotation of the two crystal lattices amounted  
316 to 6.5° that is close to the change of inclination angle upon the phase transition. Starting from the contact  
317 zone of the two twin domains, we can see that the only way of saving the structure continuity with each  
318 of the twin domains is a rotation of the *a* and *b* axes around [001]. This is necessary to comply with the  
319 larger inclination angle of atomic planes in their  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  structures relative to the one present in the  
320 lillianite-like transition zone. The rotation of lattices of the two domains relative to their contact zone has  
321 to be in opposite sense (clockwise vs. anti-clockwise) and correspond in magnitude to differences  
322 between the inclination angles in the *LP* and *HP* lillianite polymorphs.

### 323 **Anisotropic compression in *LP* and *HP* lillianite structures**

324 The thorough analysis of the structural characteristics of the  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> here proposed explains the  
325 compressional anisotropy of the two polymorphs. The *c*-axes of the unit cell is directly related to the  
326 archetypal structure in both polymorphs, corresponding to [011]<sub>PbS</sub> period in lillianite and [010]<sub>SnS</sub> period  
327 in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> and the structure is constrained in this direction due to the presence of closely spaced mirror  
328 planes perpendicular to this direction. This makes it the least compressible direction in the crystal  
329 structure. While in the *LP* lillianite we have a relatively homogeneous distribution of strong and weak  
330 bonds inside (010) slabs, in the *HP* lillianite the slabs contain tightly bonded rods separated by spaces  
331 crossed only by weak bonds and containing LEP. The compression along the slab extensions [100] is  
332 larger than their thinning along [010] in lillianite but the situation reverses in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>. This is related  
333 to “harmonica behavior” of the structure: a rotation of the rigid rods around their [001] axes as the main  
334 response to compression. The angle between the extension of rigid rods in the *a/b* plane of the two  
335 adjacent module slabs (see e.g. Figure 1) changes from 63.04° at 6.3 GPa to 62.21° at 20.9 GPa in  
336 accordance with the faster compression along the *b* than along the *a* axis.

### 337 **LEP evolution and the behavior of Pb-Bi sulphosalts at high pressures**

338 This work shows that the stereochemical expression of the LEP of Pb and Bi, both in *LP* and *HP*  
339 polymorphs, evaluated by eccentricity of polyhedral coordinations, decreases with increasing pressure  
340 as observed in other similar studies (Lundegaard et al. 2005; Olsen et al. 2007, 2008, 2011; Comodi et  
341 al. 2019). This applies for both the *LP* and *HP* phases of Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>.

342 In detail, in the *LP* lillianite structure during compression up to the transition point, the LEP diminishes  
343 its expression as the direction of its displacement follows the shortening of the M2 – S1 longest bond  
344 (Figure 1a). In the  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>, the LEP of M2A moves to the opposite side of the nucleus as compared  
345 with the situation in the *LP* structure and increases its displacement from the nucleus; whereas M1A has  
346 the lower eccentricity than its “parent” M1 site in lillianite (Figure S2a). The movements of the

347 coordinating S atoms change the coordination to CN7 split octahedron, but at the same time equalize the  
348 bond lengths so that the stereochemical activity of LEP is suppressed in M1A, whereas it strongly  
349 increased in all others. It is also interesting that this site has the smallest displacement calculated by  
350 AMPLIMODES, below 0.1 Å ( $|\Delta|_{M1A} = 0.0762$ , Table 2), whereas all other atoms have displacements  
351 over 0.5 Å (e.g.  $|\Delta|_{M2A} = 0.6243$ ,  $|\Delta|_{M2B} = 0.9684$  and  $|\Delta|_{M3} = 1.0038$ , Table 2), so in this respect it plays  
352 a role of a “firm point” during the phase transition. The sites M2B and M1B do not flip the LEP during  
353 formation of rigid rods. The changes in coordination in their case produce also the largest distortions of  
354 coordination polyhedra in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> (Figure S2).

355 The increase of the stereochemical activity of LEP during phase transition from the LP to the HP phase  
356 is also observed for heyrovskite (Pb<sub>6</sub>Bi<sub>2</sub>S<sub>9</sub>; Olsen et al. 2011); whereas, galenobismutite (PbBi<sub>2</sub>S<sub>4</sub>)  
357 remains in its ambient pressure structure and does not have phase transitions up to 21 GPa (Comodi et  
358 al. 2019). The eccentricity of its cationic sites continuously decreases in response to compression (Figure  
359 S2a) except M1 (the lowest CN coordination site in galenobismutite) where eccentricity keeps a nearly  
360 constant value as *P* increases. In fact, it becomes the one with the highest eccentricity at high pressure.  
361 It is worth noting that M1 in galenobismutite is characterized as a pure Bi site, whereas the other two  
362 sites were considered mixed Pb/Bi sites (Pinto et al. 2006). Galena (PbS, *Fm-3m* space group) with  
363 octahedral coordination of atoms and a suppressed stereochemical activity of LEP of Pb undergoes phase  
364 transition to an orthorhombic structure at approximately 2.5 GPa with either *Cmcm* (Grzechnik and Friese  
365 2010) or *Pnma* (Bencherif et al. 2011) symmetry and Pb achieves the seven-fold coordination due to  
366 stereochemically active LEP of Pb. On the contrary, bismuthinite (Bi<sub>2</sub>S<sub>3</sub>) has, already at room pressure,  
367 Bi atoms with stereochemically active LEP and seven-fold coordination and does not undergo phase  
368 transition up to approximately 50 GPa (da Silva et al. 2019).

369 We can conclude that the Pb-Bi sulphides with more Pb than Bi atoms in the structure, have a behavior  
370 of LEP dominated by the former. At low pressures, the stereochemical activity is low to suppressed,



371 which obviously is an unstable state at pressures, high enough, where the structures transform to types  
372 that allow a high stereochemical expression of LEP. This does not apply to Bi-dominated compounds  
373 that, already at the lowest pressure, acquire the structures with relatively high expression of LEP. An  
374 interesting conclusion is that, at least considering sulfides of Pb and Bi, the stable structures (up to several  
375 tens of GPa) are those that allow a relatively significant expression of  $s^2$  LEP of cations already at low  
376 pressures, in spite of the general tendency that this expression decreases with pressure, different from  
377 structures with a weak expression, which show a limited pressure stability. This persistence of LEP under  
378 pressure, granted by the structure type, is obviously a property that stabilizes the crystal structure under  
379 high-pressures.

### 380 **Implications**

- 381 1. Our results challenge the existing assumption that the pressure-induced phase transitions in  
382 structures where LEP has a high stereochemical activity are associated with an increase of  
383 coordination number and with the coordination polyhedra becoming more regular (Grzechnik,  
384 2007) meaning that they produce structures with less stereochemical activity of LEP. The here  
385 described cases of Pb-Bi sulphides can be rationalized considering that the coordinations with  
386 higher stereochemical activity of LEP more easily bear the pressure induced strain through  
387 equalization of bond distances and suppression of LEP.
- 388 2. While the pseudomerohedral twinning has been associated to high-pressure experiments (King  
389 and Prewitt 1982; Kudoh et al. 1986; Posse et al. 2011), de-twinning was rarely observed on  
390 decompression (Kaga and Gilman 1969). Recently, the study of de-twinning has become an  
391 attractive topic in the study of shape-memory alloys (e.g. Wang et al. 2018; Rößler et al. 2019),  
392 materials that are capable of bearing large recoverable elastic strain under loading-unloading  
393 (Wang et al. 2018 and references therein). By its characteristics here presented, phase transition  
394 of lillianite qualifies itself as a specifically interesting shape-memory material. Further work

395 devoted to test the likely ability of lillianite to recuperate the original shape not only while  
396 unloading (pseudoelasticity), but also while heating above specific critical temperatures (shape-  
397 memory effect), would open a new scenario where lillianite might have important applications in  
398 several fields such as (i) aerospace, *i.e.* to isolate the micro-vibrations, for low-shock release  
399 devices and self-deployable solar sails (Costanza and Tata 2020), and (ii) photovoltaic panels, as  
400 shape-memory alloy springs that act as actuators by contracting with great strength and speed  
401 when heated and thus changing the position of the solar panel in accordance to the motion of the  
402 sun throughout the day (Gasparetto et al. 2021; Degeratu et al. 2021).

403 3. Sulfosalt structures are interesting due to their similarities with high-pressure silicate phases that  
404 likely occur in the inner Earth, with octahedrally coordinated silicon combined with other cations  
405 with the same or higher coordination number. In particular, post-perovskite  $\text{MgSiO}_3$  (Murakami  
406 et al. 2004, Oganov and Ono 2004) is the oxide homologue  $N=1$  of the lillianite series with  
407 octahedral Si and bicapped trigonal prismatic (or square antiprismatic) Mg polyhedra  
408 (Makovicky and Topa 2014); whereas,  $\text{CaTi}_2\text{O}_4$  (Bertaut and Blum 1956; Rogge et al. 1998;  
409 Yamanaka et al. 2008) and the high-pressure phase  $\text{Fe}_4\text{O}_5$  (Lavina et al. 2011; Guignard and  
410 Chrichton 2014) are the  $N=2$  and  $N=3$  homologues, respectively.

411 While, on the one hand, galenobismutite has the calcium-ferrite structure type, like the aluminous  
412 phases of the inner Earth, on the other hand, we do not know yet a deep-Earth analogue of  $\beta$ -  
413  $\text{Pb}_3\text{Bi}_2\text{S}_6$ . However, recent results of the ultra-high  $P/T$  behavior of  $\text{NaMgF}_3$  (Dutta et al. 2019),  
414 used as a model system for understanding phase transitions in  $\text{ABX}_3$  (e.g.  $\text{MgSiO}_3$ ) systems, show  
415 a  $\text{Sb}_2\text{S}_3$ -type (stibnite-type) post-post-perovskite  $\text{ABX}_3$  phase at 58 GPa, which remains stable to  
416 over 150 GPa and heating to over 2000 K. The  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  structure belongs to the meneghinite  
417 homologous series (Olsen et al. 2008), where also stibnite lies; this increases the interest on the  
418 phase transition of lillianite in relation to the deep-Earth crystallography. Indeed, a similar

419 sequence of transitions of  $\text{NaMgF}_3$  is predicted to occur in  $\text{MgSiO}_3$  system at ultra-high pressures,  
420 implying density changes as well as peculiar anisotropic evolution, and thus likely affecting the  
421 mineralogy and the dynamic in the deep interior of both the earth and the large extrasolar rocky  
422 planets.

423

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424

425

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434

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613

## List of figure captions

614

615

616 **Figure 1** (a) Lillianite crystal structure projected in the  $ab$  plane. Yellow, gray and pink denote S, Pb  
617 and Bi, respectively. The slab thickness is shown by the number of octahedra in the row across a slab  
618 (purple octahedra), defining the homologue order ( $N = 4$ ), with a  $[\text{PbS}_{6+2}]$  bicapped trigonal prism in  
619 gray on the slab boundary. The mirror plane ( $m$ ) perpendicular to the  $b$ -axis is also shown as solid line.  
620 (b)  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  crystal structure projected in the  $ab$  plane. Colors are according to (a) and the  $n$  glide  
621 planes, appearing in the place of  $m$  perpendicular to the  $b$ -axis, are shown as dashed-dotted lines. The  
622 two of the rigid rods in  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  are shaded and, for comparison, a corresponding part in the crystal  
623 structure of lillianite. The crystal structures was were drawn by VESTA software (Momma and Izumi  
624 2011).

625

626 **Figure 2** Evolution of the unit cell volume and unit cell parameters with  $P$ , fitted by the third-order BM  
627 EOSs. (a)  $V$  (black); (b)  $a$  (red),  $b/2$  (blue),  $c*2$  (green). Full circles and triangles refer to lillianite and  $\beta$ -  
628  $\text{Pb}_3\text{Bi}_2\text{S}_6$ , respectively. Data during decompression are shown as stars. Data for *lill2* from Olsen et al.  
629 (2008) are shown as empty symbols.

630

631 **Figure 3** Confidence ellipses at 68.3% confidence level for the  $V$  vs.  $P$  BM3 EoS of lillianite and  $\beta$ -  
632  $\text{Pb}_3\text{Bi}_2\text{S}_6$ . For the  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  the bulk modulus and its first derivative were calculated at 4.9 GPa, in order  
633 to a direct comparison with data by Olsen et al (2008).

634

635 **Figure 4** The displacements of atomic sites of  $\beta$ - $\text{Pb}_3\text{Bi}_2\text{S}_6$  at 6.3 GPa relative to their positions in the  
636 low-pressure lillianite.  $a$  axis vertical,  $b$  horizontal and  $c$  perpendicular to the projection plane. The atoms  
637 at two different  $z$  levels are differently shaded. The displacements are indicated by blue color for cation

638 sites and green for S sites. All displacements are parallel to (001) plane. Yellow shading indicates one of  
639 the rigid rods. A  $2_1$  axis runs through the center of the rod perpendicular to the plane of projection.

640

641 **Figure 5** Volumes of coordination polyhedra at various pressures. Symbols and colors are according to  
642 the legend; the empty symbols are data during decompression.

643

## Deposit Items

644

645

646 **Table S1** Relation of atomic site labeling in this work to Olsen et al. (2008) labeling.

647

648 **Table S2** Bond distances, average bond distances and volumes of coordination polyhedra in lillianite at  
649 various pressures.

650

651 **Table S3** Bond distances, average bond distances and volumes of coordination polyhedra in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>  
652 at various pressures.

653

654 **Figure S1** Collected data extraction and unit cells used for reflection indexing at 1.74 GPa (single  
655 lillianite crystal) (a), 6.30 GPa (twinned  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>) (b) and 2.13 GPa during decompression (the  
656 detwinned single lillianite crystal) (c).

657

658 **Figure S2.** Eccentricities of coordination polyhedra at various pressures. In red, data for coordinations  
659 in galenobismutite (Comodi et al. 2019): squares M1 (CN6), circles M2 (CN7), triangles M3 (CN8) (a);  
660 asphericities of coordination polyhedra at various pressures (b); volume distortion of coordination  
661 polyhedra at various pressures. Squares represent M1 in lillianite (black), M1A (blue) and M1B (green)  
662 in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>; circles M2 in lillianite (black), M2A (blue) and M2B (green) in  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>; triangles M3  
663 in lillianite (black) and  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub> (blue). The empty symbols are data during decompression. The dashed  
664 horizontal lines indicate values for the regular maximum-volume square antiprism (0.0351),  
665 Archimedean square antiprism (0.0535), maximum-volume split octahedron (0.0790) and simple split  
666 octahedron (0.1333) (c).

667

668 **Bond valences.** Details on the used procedure and obtained results of the bond valences calculation.

669

670 The refined crystal structures, the *hkl* Miller indices of the indexed reflections as well as the  $F_o^2$  (squared

671 observed F values) and  $F_c^2$  (squared calculated F values) are deposited in the *cif* file at the publisher.

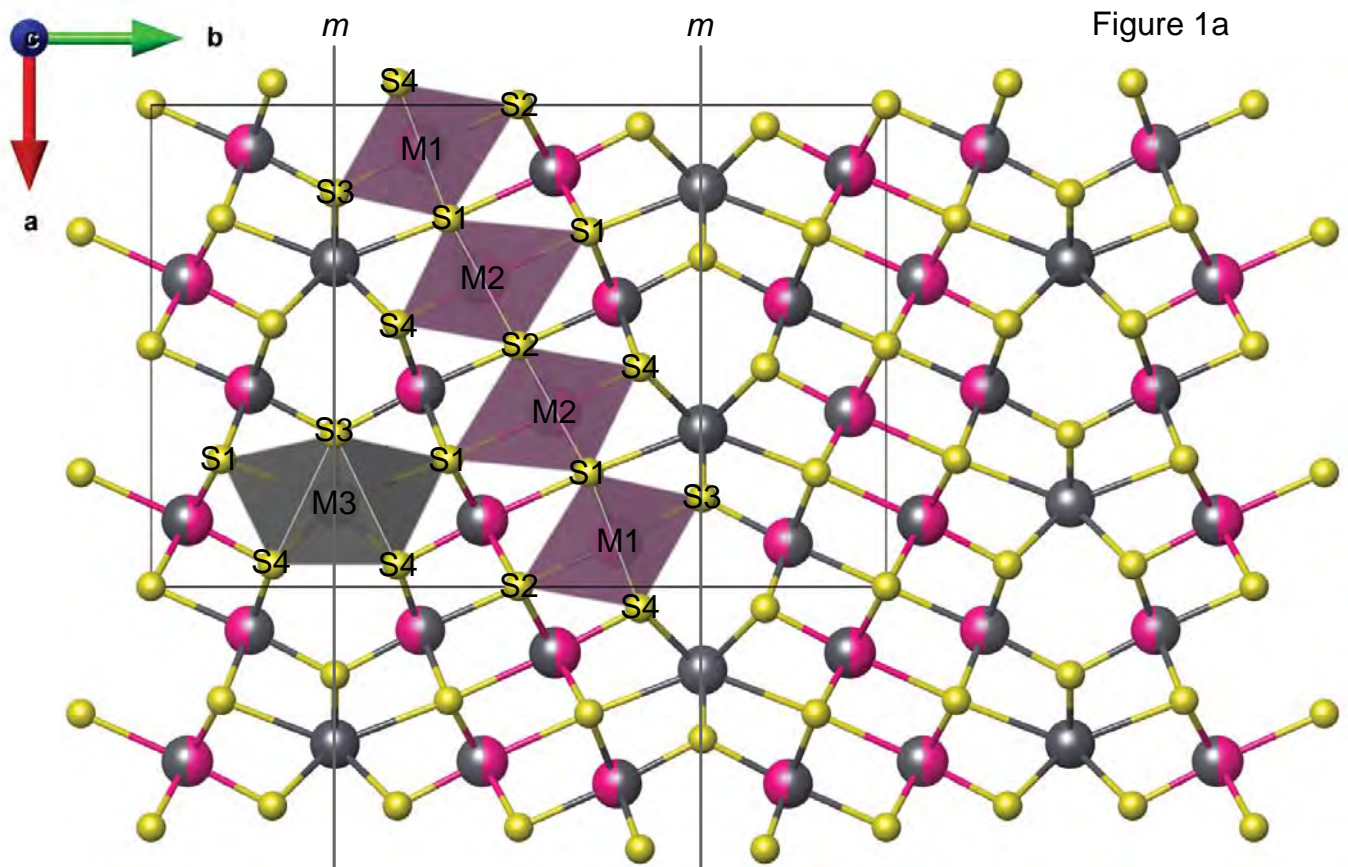


Figure 1b

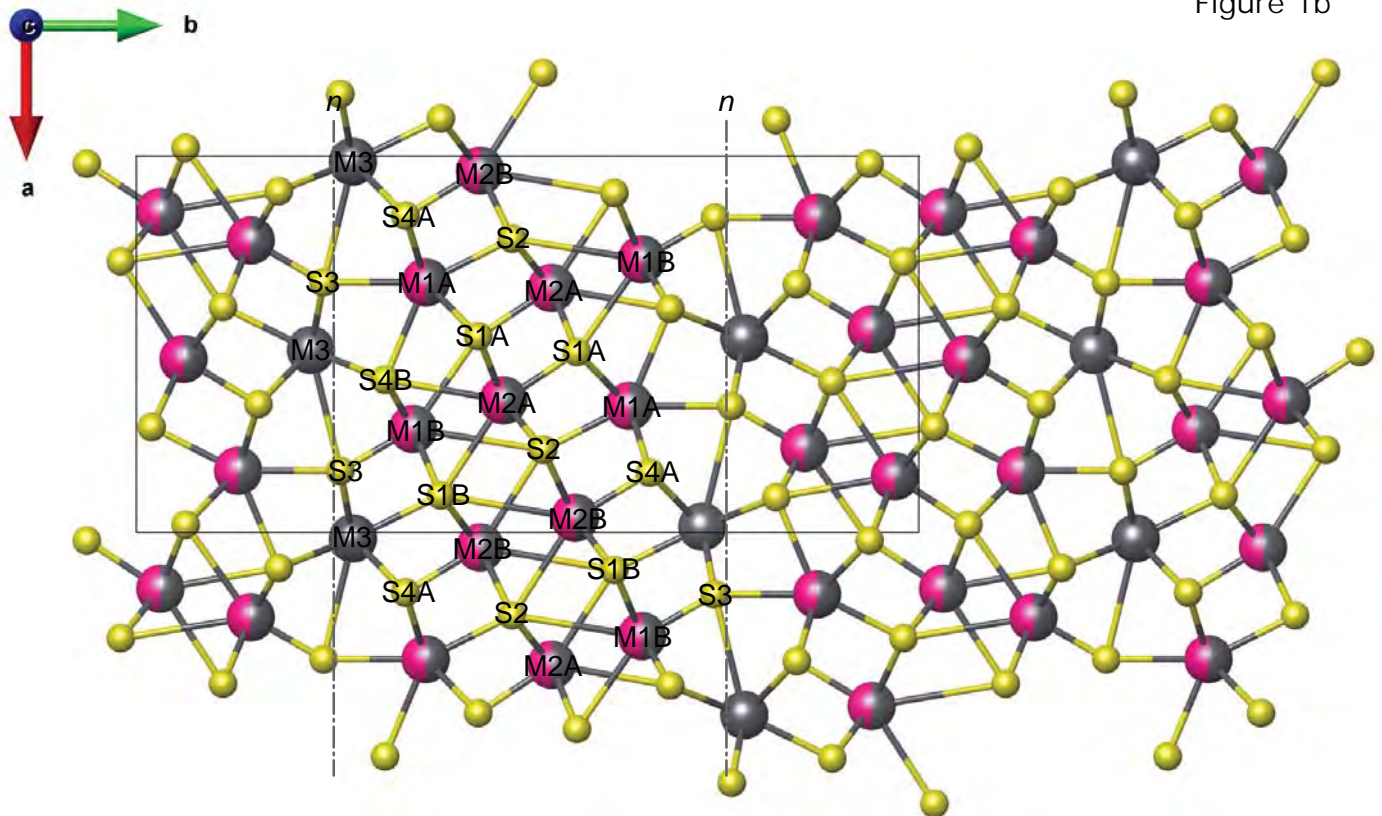




Figure 2a

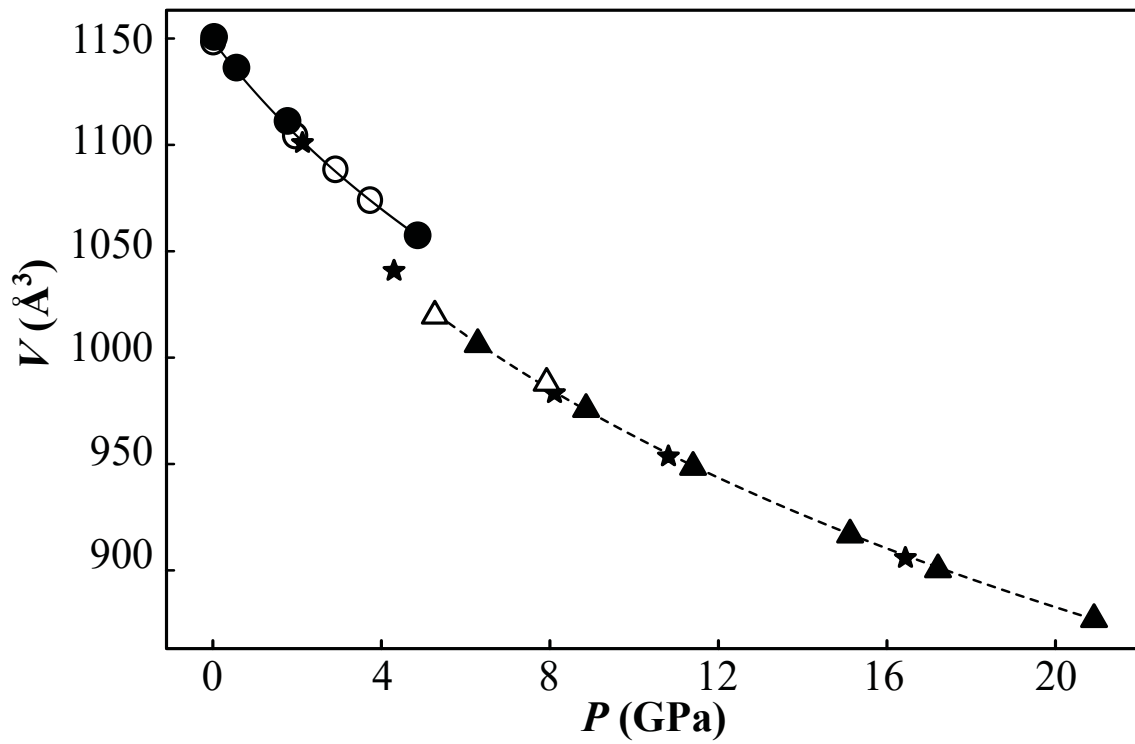


Figure 2b

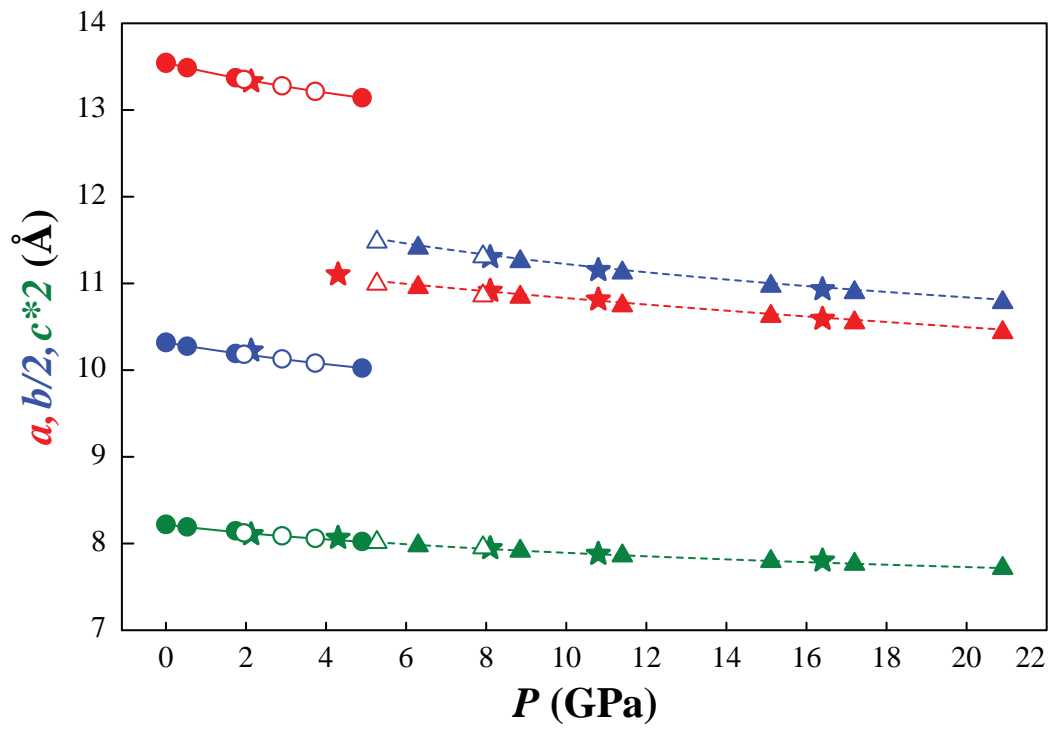


Figure 3

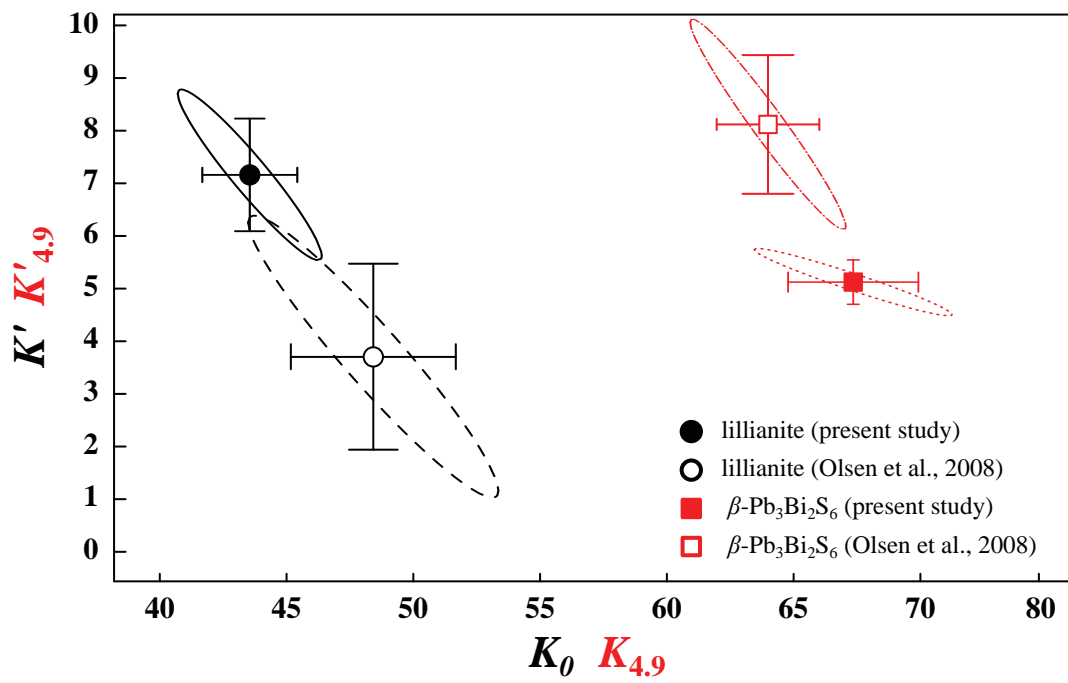


Figure 4

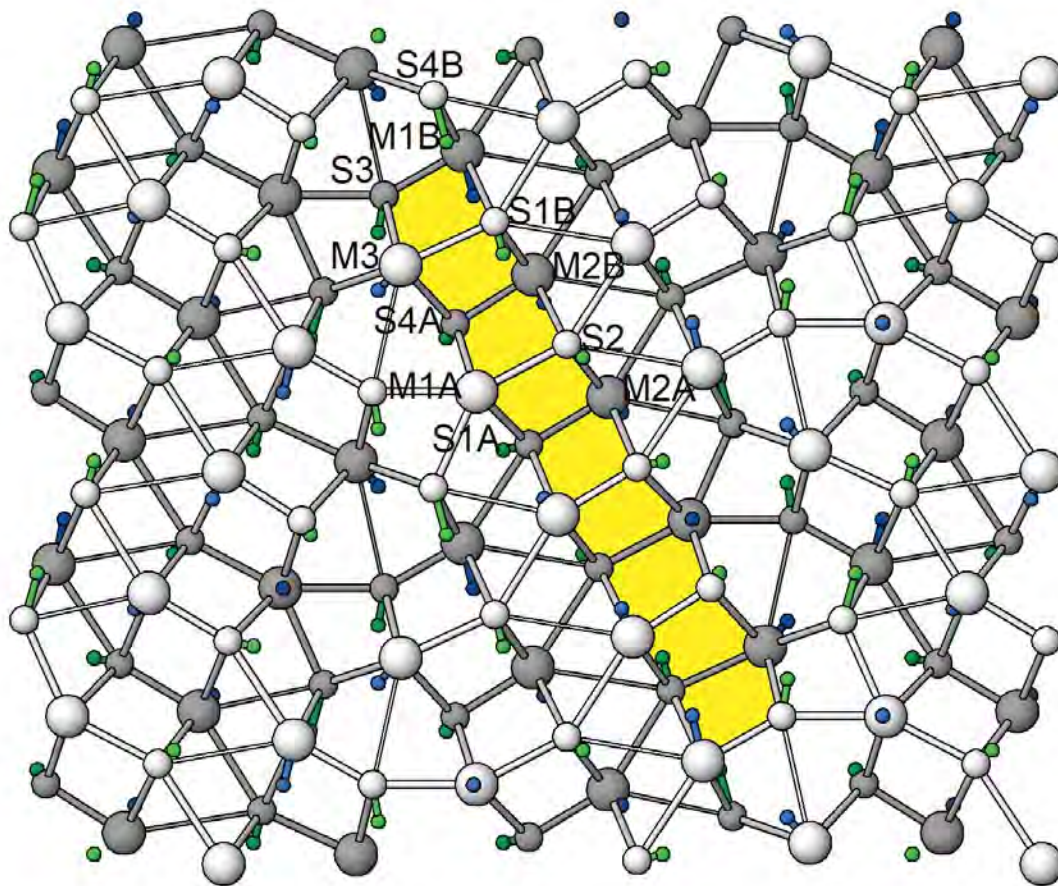
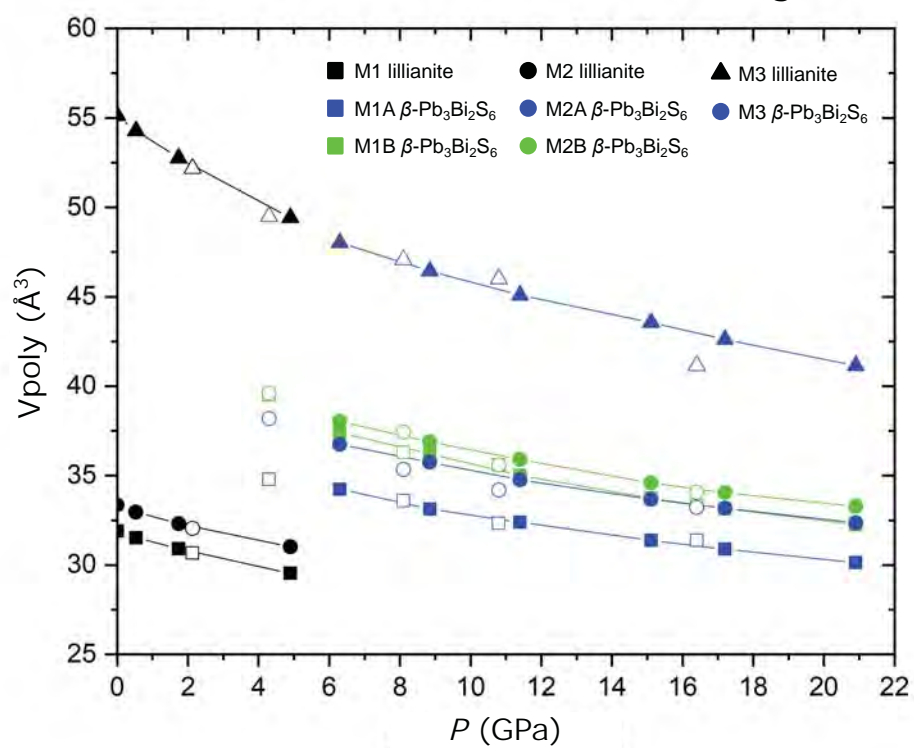


Figure 5



**Table 1** Results of the EoS elaboration for lillianite and  $\beta$ -Pb<sub>3</sub>Bi<sub>2</sub>S<sub>6</sub>. Cell parameters ( $a$ ,  $b$  and  $c$ ) and cell volumes ( $V$ ) are given in Å and Å<sup>3</sup>, respectively; whereas moduli (K and M) are in GPa. First derivatives of bulk and axial moduli are given as  $K'$  and  $M'$ . Results of the EoS calculation from Olsen et al. (2008) are also listed.

	lillianite		$\beta$ -lillianite			
	present work	Olsen et al. (2008)	present work		Olsen et al. (2008)	
$V_0$	1149.3(9)	1148.5(3)	$V_{4,9}$	1027(1)	$V_{4,9}$	1028.8(7)
$K_0$	44(2)	48.4(1.7)	$K_{4,9}$	67(3)	$K_{4,9}$	63.8(3)
$K'$	7(1)	4.0(1.3)	$K'$	5.1(4)	$K'$	8.1(1.0)
$a_0$	13.546(4)		$a_{4,9}$	11.043(7)		
$M_0$	118(5)		$M_{4,9}$	242(12)		
$M'$	21(3)		$M'$	8(1)		
$b_0$	20.638(5)		$b_{4,9}$	23.085(8)		
$M_0$	130(6)		$M_{4,9}$	145(4)		
$M'$	19(3)		$M'$	16.0(7)		
$c_0$	4.1115(3)		$c_{4,9}$	4.028(2)		
$M_0$	139(12)		$M_{4,9}$	242(5)		
$M'$	31(10)		$M'$	29(1)		

**Table 2** The displacements of atomic sites at 4.9 GPa and 6.3 GPa relative to those at 1 MPa, as calculated by AMPLIMODES (Orobengoa et al. 2009).  $\Delta_x$ ,  $\Delta_y$  and  $\Delta_z$  are given in relative units.  $|\Delta|$  is the absolute distance given in Å.

Lillianite at 4.9 GPa					$\beta$ -Pb <sub>3</sub> Bi <sub>2</sub> S <sub>6</sub> at 6.3 GPa				
Site	$\Delta_x$	$\Delta_y$	$\Delta_z$	$ \Delta $	Site	$\Delta_x$	$\Delta_y$	$\Delta_z$	$ \Delta $
M1	-0.0026	0	0.0000	0.0822	M1B	-0.1147	0.0132	0.0000	1.5304
M1'	-0.0026	0	0.0000	0.0175	M1A	-0.0014	0.0037	0.0000	0.0762
M2	-0.0028	-0.0018	0.0000	0.0564	M2B	-0.0707	0.0135	0.0000	0.9684
M2'	-0.0028	0.0018	0.0000	0.0097	M2A	0.0401	0.0167	0.0000	0.6243
M3	0.0022	0	0.0000	0.0059	M3	-0.0640	-0.0273	0.0000	1.0038
S1	-0.0005	0.0025	0.0000	0.0511	S1B	-0.0814	0.0101	0.0000	1.0887
S1'	-0.0005	-0.0025	0.0000	-0.0285	S1A	-0.0141	0.0311	0.0000	0.6511
S2	0	0	0.0000	0.0182	S2	0.0342	-0.0196	0.0000	0.5970
S3	-0.0028	0	0.0000	0.0762	S3	-0.0940	-0.0070	0.0000	1.2433
S4	-0.0048	0.001	0.0000	0.0677	S4A	-0.0356	-0.0106	0.0000	0.5139
S4'	-0.0048	-0.001	0.0000	0.1163	S4B	-0.1181	-0.0159	0.0000	1.5845