1	Revision 3 (correction date 08/19/2021)
2	The efficiency of copper extraction from magma bodies:
3	Implications for mineralization potential and fluid-silicate melt
4	partitioning of copper
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ABSTRACT

24 Multiple factors may downgrade the mineralization potential of an intermediate-felsic 25 intrusion, such as the commonly invoked inefficient fluid exsolution and lack of ore-forming 26 species (metals and their ligands) in magmas. However, other factors may affect 27 mineralization potential of a magma body but have poorly understood roles in the formation of magmatic-hydrothermal ore deposits. Here, we present a comparison between two Cu 28 29 mineralizing plutons and a Cu-poor, Fe mineralizing pluton in the Edong district. Efficient 30 fluid exsolution and extraction occurred during the solidification of all three plutons, as 31 evidenced by extensive skarn alteration around them. The results show that the oxidation state of the three plutons is similar (within a range of $\sim \Delta NNO + 0.9$ to $\Delta NNO + 2.5$). A systematic 32 33 comparison of the Cu contents of a certain suite of minerals of the three plutons show that the 34 Cu concentrations of all minerals in the Cu mineralizing plutons are lower than those of the Cu-poor Fe mineralizing pluton. This indicates that the Cu mineralizing plutons underwent 35 36 more efficient copper extraction. Thus, igneous crystals with anomalously low Cu contents may potentially be used as a tool to identify Cu mineralizing magmatic units in a deposit with 37 38 multiphase intrusions. We suggest that the inefficient copper extraction from plutons may be 39 ascribed to the lack of reduced S species during fluid exsolution or different evolution paths 40 of Cu and Cl during magma crystallization.

41 Keywords: geothermobarometry, fluid exsolution, copper deposit, magma, ore-forming
42 potential

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INTRODUCTION

Magmatic-hydrothermal ore deposits, including porphyry (Cooke et al. 2005; Seedorff et 46 47 al. 2005; Sillitoe 2010), skarn (Meinert et al. 2005; Chang et al. 2019) and epithermal type deposits (Hedenquist et al. 1998; Simmons et al. 2005), are the primary source of copper for 48 49 our society (Arndt et al. 2017). There is a consensus that intermediate-felsic intrusions provide most of the copper, chlorine, and water that are necessary to produce hydrothermal 50 51 alteration and ore bodies in porphyry copper systems and associated skarns (Hedenquist and 52 Lowenstern 1994; Candela 1997; Meinert et al. 2003; Richards 2003; Williams-Jones and 53 Heinrich 2005; Sillitoe 2010; Audétat and Simon 2012). The formation of porphyry Cu 54 deposits and associated skarns potentially involves multiple processes (Richards 2013; 55 Wilkinson 2013), such as a contribution from oxidized mantle (e.g., Mungall 2002; Wang et 56 al. 2006), pre-enrichment in the lower crust (e.g., Lee et al. 2012; Chiaradia 2014; Hou et al. 2015; Chiaradia and Caricchi 2017; Zheng et al. 2019), sulfide saturation in shallow magma 57 58 bodies (e.g., Wilkinson 2013), injection of mafic magmas (e.g., Blundy et al. 2015; Yang et al. 2015; Cao et al. 2018), focused fluid flow, and repetitive fluid injections (e.g., Mercer et 59 60 al. 2015; Li et al. 2017). There is growing evidence supporting the view that magma bodies with typical concentrations of metals (e.g., ~50–100 ppm Cu) may sustain the formation of 61 62 economic ore deposits (Cline and Bodnar, 1991; Chelle-Michou et al. 2017; Zhang and Audétat 2017, 2018), because most metals have high fluid/melt partition coefficients, leading 63 64 to significant enrichment of them in exsolved fluids (Zajacz et al., 2008; Audétat 2019). On 65 the basis of this hypothesis, if efficient fluid exsolution occurs, many intermediate-felsic 66 intrusions should have an ability to form economic Cu deposits. Nevertheless, many plutons

showing efficient fluid exsolution also lack Cu mineralization, indicating that other factors 67 may suppress the mineralization potential of intrusions. Whole-rock analyses may not be 68 69 sufficient to provide a better understanding of this question, because some fraction of the 70 metals and volatiles in an intrusion are commonly lost after solidification. Comparison of 71 melt inclusion compositions from barren and mineralized plutons could provide important 72 insights into this question, but available data sets show that there is no obvious difference in 73 metal concentrations between barren and mineralizing melts (e.g., Audétat 2015; Zhang and 74 Audétat 2017, 2018). It is noteworthy that the studied barren plutons contain abundant 75 miarolitic cavities that represent fluid pockets, implying inefficient extraction of the fluids out of the magma (e.g., Audétat and Pettke 2003; Zhang and Audétat 2018). It is thus 76 77 difficult to distinguish whether the lack of mineralization is ascribed to inefficient fluid 78 extraction or other factors.

Here, we present a study of minerals (clinopyroxene, feldspar, amphibole, titanite and 79 80 apatite) from intrusions associated with the Tonglushan Cu-Fe-Au (1.08 Mt Cu, 60 Mt Fe, 70 t Au and 508 t Ag; Li et al., 2014), Tieshan Fe-Cu (160 Mt Fe and 0.67 Mt Cu; Li et al. 81 82 2014), and Jinshandian Fe (200 Mt Fe; Zhu et al. 2015, 2017) skarns in the Edong district (East China), which is one of the most productive Cu-Fe provinces in China. Extensive skarn 83 84 alteration has been found around all three plutons, indicating that efficient fluid exsolution and extraction occurred during their solidification. A comparison of the three plutons 85 provides an opportunity to identify potential factors that downgrade the Cu-mineralizing 86 87 potential of an intermediate-felsic intrusion. Our results suggest that the efficiency of copper 88 extraction from magma bodies by fluids plays a critical role in determining Cu 89 mineralization potential.

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GEOLOGICAL BACKGROUND

92 The three studied skarn deposits are located in the Edong district, which is situated on the 93 northern margin of the Yangtze Craton (Figure 1). Regional tectonic characteristics and 94 magmatism have been described by Zhou et al. (2020a). Magmatic activity in the Edong 95 district began at ~150 Ma and ceased at ~120 Ma when this area was located in an intraplate 96 setting (e.g., Wang et al. 2004; Li et al. 2008; Hou et al. 2013; Pirajno and Zhou 2015; Zhou 97 et al. 2015; Zhou et al. 2020a). Extensive hydrothermal alteration and ore deposit formation accompanied the magmatism (Li et al. 2014; Zhou et al. 2020a). There are more than 70 98 99 economic Cu-Fe-Au-(Mo-W) deposits within the Edong district, which is a very productive 100 Cu-Fe province (Fig. 1; Li et al. 2014). Skarn and porphyry types are the main 101 alteration-mineralization styles. The skarn deposits sampled for this study which include the 102 Tonglushan Cu-Fe-Au, Tieshan Fe-Cu, and Jinshandian Fe deposits (Fig. 2), have different 103 metal endowments.

The Tonglushan Cu-Fe-Au skarn deposit is the largest ore deposit in the Edong district and contains 1.08 Mt Cu (~1.8 wt% Cu), 60 Mt Fe (~41 wt% Fe), 70 t Au (~0.38 g/t Au) and 508 t Ag (Li et al. 2014). The Tonglushan deposit contains twelve ore bodies, which are located at the contact zones between the quartz monzodiorite and carbonate rocks or large carbonate xenoliths within the quartz monzodiorite (Fig. 2C). Most ore bodies have lenticular shapes and are 200–500 m in length and 30–110 m in thickness, with vertical extents of 100–500 m (Yu et al. 1985). In addition, ore-bearing breccia pipes have been found at deep

111 levels (Liu et al. 2005). Skarn alteration developed both within the pluton (i.e., endoskarn) 112 and carbonate rocks (i.e., exoskarn), but ore bodies are mainly distributed in exoskarn zones 113 (Zhao et al. 2012). The minerals formed in the prograde stage are andradite, grossular, 114 diopside, hedenbergite, scapolite, and plagioclase, followed by a retrograde alteration 115 assemblage that includes epidote, actinolite, pargasite, phlogopite, chlorite, fluorite, quartz, 116 serpentine, illite, montmorillonite, kaolinite and dickite (Li et al. 2014; Chen et al. 2019). Ore 117 minerals, including magnetite, hematite, chalcopyrite, bornite, chalcocite, molybdenite, native 118 gold and electrum, are generally associated with retrograde minerals (Li et al. 2014). 40 Ar/ 39 Ar dating of the phlogopite in the skarn gives an age of ~140 Ma (Li et al. 2014). The 119 quartz monzodiorite pluton has a zircon U-Pb age of 142 ± 1 Ma (Li et al. 2014). Several 120 121 albitite dikes cut the pluton and skarns (Fig. 2C), indicating that they were intruded after 122 mineralization.

The Tieshan Fe-Cu skarn deposit contains 160 Mt Fe (~53 wt% Fe) and 0.67 Mt Cu (~0.6 123 wt% Cu) (Li et al. 2014). Six large orebodies have been found along the contact between the 124 quartz diorite and carbonate rocks (Fig. 2A). Most ore bodies are 480-920 m in length and 125 126 10-180 m in thickness, with vertical extents of 200-700 m (Shu et al. 1992). Skarn minerals include garnet, diopside, scapolite, phlogopite, actinolite, albite, plagioclase, epidote, chlorite, 127 tremolite, and pargasite (Li et al. 2014). The phlogopite within the skarns has a 40 Ar/ 39 Ar age 128 of 142 ± 3 Ma (Xie et al. 2011). The ore minerals are dominated by magnetite, pyrite, 129 130 chalcopyrite, pyrrhotite and hematite (Hu et al. 2017). The ore-forming pluton, the quartz 131 diorite, has a zircon U-Pb age of 140.9 ± 1.2 Ma (Xie et al. 2007).

132 The Jinshandian Fe skarn deposit is situated on the southern and western margins of the

133 Jinshandian pluton (Fig. 2B) and contains 200 Mt Fe (~42.3 wt% Fe) (Zhu et al. 2015, 2017), 134 whereas Cu and Au are absent (Shu et al. 1992). More than 130 ore bodies have been found at 135 the contact zones between the Jinshandian pluton and carbonate or clastic rocks (Fig. 2B), and 136 most of them occur as lenses and veins (Zhu et al. 2015). The skarn mineral assemblages 137 include diopside, phlogopite, scapolite, and amphibole, with minor serpentine, garnet, titanite and epidote (Zhu et al. 2015). The phlogopite has a 40 Ar/ 39 Ar age of 127.6 ± 0.9 Ma (Zhu et al. 138 139 2017). The ore mineralogy is dominated by magnetite (Shu et al. 1992), which commonly 140 occurs with diopside and phlogopite (Zhu et al. 2017). In addition, anhydrite and pyrite also 141 exist. Anhydrite is present within country rocks, the contact zones between ore body and 142 country rocks, and retrograde skarns in the form of veins or massive aggregates (Zhu et al. 143 2017). The quartz diorite pluton has zircon U-Pb ages of 127.4 ± 1.2 Ma and 127.6 ± 0.7 Ma 144 (Zhu et al. 2017).

145 The Tieshan plutonic samples, consisting of plagioclase (40-50%), hornblende (10-15%), 146 clinopyroxene (10-15%), K-feldspar (5-10%), quartz (5-10%), and minor magnetite, biotite, and titanite, are adjacent to a endoskarn zone (Fig. 2A). The Tonglushan plutonic samples, 147 148 containing plagioclase (50-60%), K-feldspar (10-20%), quartz (10-20%), hornblende (10-20%) and minor biotite, titanite and magnetite, were collected from an old mining pit, 149 150 which is several hundred meters from the current mining location (Fig. 2C). The Jinshandian 151 plutonic rocks, consisting of plagioclase (30-40%), K-feldspar (20-30%), hornblende 152 (15–20%), clinopyroxene (5–10%), quartz (5–10%), and minor titanite, biotite, and magnetite, 153 are adjacent to the Jinshandian western ore bodies (Fig. 2B). The plutonic samples of the 154 Tonglushan and Tieshan deposits are the same as those reported in Zhou et al. (2020a);

155	previously analyzed samples. All analyses of Jinshandian deposit samples are new. Minerals
156	with homogeneous interiors or regular zoning patterns that indicate magmatic origin were
157	selected for major and trace element analyses. Some secondary titanite crystals were also
158	analyzed for comparison.
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160	ANALYTICAL METHODS
161	Whole-rock compositional analysis
162	Whole-rock major element analyses were performed using X-ray fluorescence
163	spectrometry at Hubei Institute of Geology and Mineral Resource, with analytical errors less
164	than 2 %. Whole-rock trace element analyses were performed using a Perkin-Elmer ELAN
165	6000 inductively coupled plasma source mass spectrometer (ICP-MS) at the State Key
166	Laboratory of Isotope Geochemistry (SKLaBIG), Guangzhou Institute of Geochemistry,
167	Chinese Academy of Sciences (GIG-CAS). The analytical procedures are similar to Li et al.
168	(2002), with an analytical precision better than 2%.
169	
170	Mineral compositional analysis
171	Major element abundances in minerals were analyzed on a JEOL JXA 8100 Superprobe
172	with a 15 kV accelerating voltage, 20 nA beam current, 2 µm probe diameter, at the SKLaBIG
173	GIG-CAS, and a JEOL JXA 8230 electron probe micro-analyzer (EPMA) with a 15 kV
174	accelerating voltage, 20 nA beam current, 1 µm probe diameter, at the Key Laboratory of
175	Mineralogy and Metallogeny of Guangzhou Institute of Geochemistry, GIG-CAS. The EPMA
176	was calibrated using natural and synthetic standards and analytical results were reduced using

177 the ZAF (Z: atomic number; A: absorption; F: fluorescence) correction routines.

178 Mineral trace element analyses were performed using an ELEMENT XR (Thermo Fisher 179 Scientific) inductively coupled plasma sector field mass spectrometry (ICP-SF-MS) coupled 180 with a 193-nm (ArF) Resonetics RESOlution M-50 laser ablation system (LA) in the 181 SKLaBIG, GIG-CAS. All LA-ICP-SF-MS spots was located to overlap a conjugate EMPA spot. The spot size was 33 μ m, at a pulse energy of ~4 J cm⁻² and a laser repetition rate of 5 182 183 Hz. A smoothing device (The Squid, Laurin Technic) was used to smooth the sample signal. 184 For each spot, counting times were 20 s for gas blank collection (laser off) and 30 s for 185 sample signal detection (laser on). The standards, BCR-2G, BHVO-2G and GSD-1G, were 186 analyzed for the construction of the calibration line. Trace element concentrations were 187 normalized to that of SiO₂, as determined by EPMA. The TB-1G (USGS reference glass) was 188 analyzed as an unknown sample and the results are shown in Appendix Table A7. For apatite, NIST SRM 610 and 612 were analyzed for the calibration and unknown, respectively, and 189 190 CaO was used as the internal standard. More detailed experimental procedures and data 191 reduction strategies have been described by Zhang et al. (2019).

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193 High resolution X-ray element mapping

High resolution X-ray element mapping was employed to image the zoning patterns of
titanite, using a Cameca SXFiveFE electron microprobe at the SKLaBIG, GIG-CAS. The
operating conditions were 20 kV accelerating voltage and 80 nA beam current.

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RESULTS

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All of the whole-rock and mineral data used in this study can be found in Appendix Table Al to A6, including data previously published in Zhou et al. (2020a). Representative whole-rock and mineral data are shown in Tables 1–6. The majority of mineral major and trace element analyses both in this study and Zhou et al. (2020a) were performed at the SKLaBIG, GIG-CAS.

204 Whole-rock compositions

205 The representative whole-rock chemical compositions of the Tonglushan, Tieshan and 206 Jinshandian plutonic rocks are in Table 1 and the full list of analyses are provided in Appendix 207 Table A1. The Tonglushan plutonic rocks have $SiO_2 = 62.6-63.7$ wt%, MgO = 0.86-1.44 wt%, $Na_2O = 4.35-4.53$ wt%, $K_2O = 2.71-2.75$ wt%, and Mg# = 26.0-36.9. For trace elements of 208 interest, concentrations or ratios are, respectively: Zr = 131 ppm; Sr = 961-983 ppm; Y =209 17.0–18.3 ppm; $Eu/Eu^* = 0.92-0.94$ ppm; Dy/Dy^* , defined as $Dy/(Dy^*) =$ 210 $Dy_N/(La_N^{(4/13)}*Yb_N^{(9/13)})$ (Davidson et al., 2013), = 0.52-0.54; (La/Yb)_N = 21.1-21.7. The 211 Tieshan intrusion $SiO_2 = 62.4-64.7$ wt%, MgO = 1.42-1.92 wt%, Na₂O = 4.84-5.60 wt%, 212 $K_2O = 2.67-3.30$ wt%, and Mg# = 39.7-47.1. Trace element concentrations or ratios include: 213 Zr = 79-200 ppm; Sr = 1337-2310 ppm; Y = 13.3-16.8 ppm; $Eu/Eu^* = 0.88-1.02$ ppm; 214 $Dy/Dy^* = 0.55-0.64$; (La/Yb)_N = 23.1-33.9. The Jinshandian quartz diorites have SiO₂ = 215 58.1-71.0 wt%, MgO = 0.40-1.71 wt%, Na₂O = 2.53-9.70 wt%, K₂O = 0.18-4.91 wt%, and 216 Mg# = 21.5-59.6. Trace elements concentrations or ratios include: Zr = 98.2-415 ppm; Sr =217 129–392 ppm; Y = 14.0–36.2 ppm; Eu/Eu* = 0.50–1.03 ppm; Dy/Dy* = 0.47–0.60; $(La/Yb)_N$ 218 219 = 8.3 - 28.7.

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221 Mineral major and trace element compositions

222 Clinopyroxene

223 Clinopyroxene is present in the Tieshan and Jinshandian plutons, but no clinopyroxene 224 crystal has been found in the Tonglushan samples. Most of the Tieshan clinopyroxene crystals 225 are small in size (100–500 μ m in length) and most of them are euhedral. Compositionally, 226 they are diopside and in the range of $W_{047-49}En_{36-38}Fs_{14-17}$, with Mg# [Mg/(Mg+Fe_{total})] of 227 70–74 (Zhou et al. 2020a). For certain trace elements of interest, concentrations are: Cu =228 0.02–0.94 ppm; Mo = 0.01–0.74 ppm; Zn = 166–210 ppm; Pb = 0.50–7.26 ppm; Sr = 229 19.7–58.3 ppm. The Jinshandian clinopyroxene crystals are 100–800 µm in length with a euhedral morphology and most are diopside with compositions of Wo₄₉₋₅₃En₃₆₋₄₂Fs₈₋₁₄ and 230 231 Mg# of 69–76. They have Cu = 0.65-0.93 ppm, Zn = 132-179 ppm, Pb = 0.61-1.96 ppm, and 232 Sr = 34.6 - 85.4 ppm.

233

234 Feldspar

Feldspar group minerals are the most common phase in the three plutons. Most 235 236 Tonglushan and Tieshan plagioclase show normal zoning or oscillatory zoning and the cores 237 are in equilibrium with their whole-rock compositions (Zhou et al. 2020a). Setting aside the 238 compositional zoning, the plagioclase An values of the Tonglushan, Tieshan, and Jinshandian plutons are An₂₀₋₅₈, An₁₀₋₄₄, and An₁₄₋₃₀. Concentrations of certain trace elements for the 239 240 Tonglushan, Tieshan, and Jinshandian plagioclase are: Cu = 0.06-1.31, 0.03-1.80 and 241 10.5-12.1 ppm; Zn = 3.16-46.8, 0.56-44.1 and 2.68-5.48 ppm; Pb = 2.90-9.30, 1.04-35.2242 and 2.82–12.8 ppm; Sr = 1088–2752, 1396–4927 and 1605–2962 ppm. A small number of

analyses shows that the Tieshan and Jinshandian K-feldspar crystals have Cu of 0.04–1.80
and 1.81–13.0 ppm, respectively.

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246 Amphibole

247 Amphibole is present in all three plutons. Most amphibole crystals in the Tonglushan pluton are euhedral, with sizes ranging from 0.2 to 1.6 mm in length. Some amphibole 248 249 crystals contain titanite and apatite inclusions. Compositionally, the Tonglushan amphibole 250 crystals have Mg# of 62–72, Cu = 0.08–1.15 ppm, Mo = 0.06–1.14 ppm, Zn = 125–220 ppm, 251 Sr = 9.17-32.9 ppm. Two amphibole populations (euhedral vs. anhedral grains) may be recognized in the Tieshan pluton (Figs. 3E and D). Some amphibole has a close spatial 252 253 relationship with clinopyroxene and they are in contact. Compositions for the euhedral and 254 anhedral amphibole crystals are: Mg# = 62-73 and 60-69; Cu = 0.1-2.06 and 0.03-1.01 ppm; Mo = 0.01-2.07 and 0.01-0.55 ppm; Zn = 65.7-380 and 276-395 ppm; Sr = 26.4-294 and 255 256 18.1-73.6 ppm. Similarly, either euhedral and anhedral amphibole populations may be identified in the Jinshandian pluton. Euhedral and anhedral amphibole crystals occur in 257 258 different hand samples and have Mg# = 57-85 and 57-73, Cu = 3.32-5.28 and 1.39-2.35 ppm, Zn = 152–225 and 260–292 ppm, and Sr = 7.92–51.5 and 15.1–62.4 ppm, respectively. 259

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261 **Titanite**

Titanite is present in the host plutons of all three deposits (Fig. 3). Titanite grains in the Tieshan pluton are interstitial, but the small sizes make microanalysis difficult (Fig. 3C). Analyzed titanite crystals in this study are from the Tonglushan and Jinshandian plutons. On

the basis of a combination of back-scattered electron images and crystal morphologies, 265 266 titanite crystals may be divided into magmatic and secondary crystals. One population of 267 titanite commonly exhibit oscillatory zoning (Figs. 3A and B), sector zoning, or superimposed 268 oscillatory zoning on sector zoning (Fig. 3B), which are interpreted as magmatic textures 269 (Paterson and Stephens 1992). A second population of titanite show no regular zoning patterns 270 and commonly contain Fe-Ti oxide inclusions with irregular shapes (Fig. 3F); this population 271 is interpreted as secondary. Compositionally, the titanite crystals from the Tonglushan pluton 272 have Cu = 1.83–3.61ppm, Mo = 38.9–134 ppm, Zn = 4.68–24.4 ppm, Sr = 2.35–53.6 ppm, Cr 273 = 6.81-19.9 ppm, and Zr = 325-3929 ppm. In the Jinshandian pluton, the magmatic titanite crystals contain Cu = 5.59-6.69 ppm, Zn = 11.0-13.3 ppm, Sr = 44.0-72.0 ppm, Cr = 274 5.55–15.8 ppm, and Zr = 849-7209 ppm, and the secondary titanite crystals have Cu = 275 5.78–6.66 ppm, Zn = 12.0-13.5 ppm, Sr = 71.2-147 ppm, Cr = 87.0-303 ppm, and Zr =276 760–2780 ppm. It is noteworthy that magmatic titanite crystals show sector zoning, because 277 278 this feature exerts important controls on trace element concentrations (Paterson and Stephens 1992). 279

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281 Apatite

Apatite is a minor but ubiquitous phase in the Tonglushan, Tieshan, and Jinshandian plutons. Apatite inclusions commonly occur in other minerals but most grains in the Tonglushan pluton are bigger than those in the other two plutons (Fig. 3A). The larger sizes (up to 200 μ m in length) make microbeam analysis of the Tonglushan apatite crystals feasible. Most Tonglushan apatite crystals are fluorapatite, with F = 2.85–3.69 wt%. They have Cu = 0.03–0.70 ppm, Zn = 0.09–3.19 ppm, Sr = 427–517 ppm, La = 1726–3959 ppm, and Y =
148–403 ppm.

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DISCUSSION

291 Comparison of the magma properties of the three plutons

292 Magma properties play a key role in the formation of magmatic-hydrothermal ore deposits 293 (Hedenquist and Lowenstern 1994; Audétat and Simon 2012). For example, hydrous, oxidized, 294 and S-rich magmas are favorable for producing porphyry copper systems and associated 295 skarns (Sillitoe 2010, 2018). The Tonglushan, Tieshan, and Jinshandian plutons contain 296 abundant amphibole crystals, which record various intensive parameters and have important 297 implications for magmatic processes (e.g., Zhou et al. 2020b). Several thermobarometers were 298 used to estimate the SiO_2 contents of the melt that was in equilibrium with the crystallizing amphibole crystals and the water content, oxygen fugacity, and temperature of the magmas 299 300 (Fig. 4) (Ridolfi et al. 2010; Putirka 2016). The results show that the Tonglushan, Tieshan, and Jinshandian amphibole crystals crystallized from melts with SiO₂ contents of 72.4–75.2, 301 302 71.6-75.8, and 72.1-78.9 wt%, and in temperature ranges of 725-809, 695-801, and 303 720–795 °C, respectively. These relatively low crystallization temperatures and high-silica 304 equilibrium melts, compared to bulk-rock abundances, indicate that the precipitation of amphibole crystals occurred in the late stages during the solidification of the three plutons, 305 306 consistent with the fact that igneous rocks in the Edong district are the products of an initially 307 water-poor intraplate environment (Wang et al. 2004; Zhou et al. 2020a). A decrease in SiO₂ 308 contents in equilibrium melts with decreasing temperature (Fig. 4A) suggests that amphibole

crystallization was accompanied by quartz crystallization (Fig. 3D). High magmatic water 309 310 contents are essential for ore deposit genesis, because fluid saturation and exsolution are key 311 steps in the mineralization processes (Hedenquist and Lowenstern 1994; Candela 1997; 312 Candela and Piccoli 2005; Zajacz et al. 2008; Audétat and Simon 2012; Wang et al. 2014). 313 Application of the amphibole hygrometer (Ridolfi et al. 2010) shows that the Tonglushan and 314 Tieshan amphibole crystallized from melts with H₂O contents of 3.4–4.4 and 3.2–4.6 315 wt%, respectively (Fig. 4B; Zhou et al. 2020a). The anhedral amphibole crystals in the 316 Jinshandian pluton record magmatic H₂O contents of 2.9-3.9 wt%, whereas euhedral 317 amphibole crystals crystallized from the melts with H_2O contents of 0.8–2.0 wt% (Fig. 4B). 318 The H₂O contents in the melts that crystallized the Jinshandian euhedral amphibole crystals 319 increased with decreasing temperature, whereas the H₂O contents of the melts that grew the 320 anhedral amphibole in the Jinshandian pluton, as well as amphibole in the Tonglushan and Tieshan, remained approximately constant with cooling (Fig. 4B). In theory, H_2O is 321 322 incompatible during magma crystallization and will be enriched in residual melts until fluid saturation and exsolution occur. The Jinshandian euhedral amphibole crystals record such a 323 324 crystallization-dominated trend, whereas the other amphibole trends indicate that fluid exsolution accompanied their crystallization, consistent with higher magmatic water contents 325 326 and volatile saturation at higher degrees of pluton crystallinity.

Another critical intensive parameter is magmatic oxygen fugacity, which controls the behavior of sulfur as well as metals during magmatic evolution (Audétat and Simon 2012; Richards 2015). High oxygen fugacities are necessary for generating porphyry copper systems, and reduced ilmenite-series intrusions commonly lack economic Cu mineralization (Ishihara 331 1977, 1981; Sillitoe 2018). The oxidized state of ore-forming magmas is hypothesized to be 332 inherited from subducted oceanic slabs through the transport of slab-derived, oxidized, partial 333 melts or fluids into the mantle wedge, promoting the extraction of metals and sulfur into arc 334 magmas (e.g., Mungall 2002; Evans and Tomkins 2011). In addition, the oxidation state of 335 magmas may also be modified during subsequent magmatic differentiation, leading to either magnetite fractionation-induced oxidation (e.g., Lee et al. 2010) or magnetite fractionation-336 337 and degassing-induced reduction (e.g., Jenner et al. 2010; Kelley and Cottrell 2012). The 338 solubility of copper in silicate melts increases with increasing oxygen fugacity (e.g., Zajacz et 339 al. 2012). More importantly, the solubility of sulfur is strongly controlled by the oxidation state of magmas (Baker and Moretti 2011). The oxidizing S⁶⁺ is much more soluble than the 340 reducing S^{2+} , and there is a dramatic increase of the sulfur solubility in magmas at magmatic 341 oxygen fugacity greater than $\sim \Delta FMQ + 1.0$ (corresponding to $\sim \Delta NNO + 0.4$ at temperatures < 342 343 1000 °C; Jugo et al. 2010). The potential of Cu mineralization is therefore suppressed in magmas with low oxygen fugacities (e.g., Zajacz et al. 2012). Another possible means of 344 lowering ore-forming potential via low oxygen fugacities is the sequestration of metals by 345 346 early sulfide fractionation (e.g., Jenner et al. 2010; Park et al. 2015, 2019; Hao et al. 2017), but this proposal is not widely accepted (e.g., Spooner 1993; Keith et al. 1997; Larocque et al. 347 348 2000; Halter et al. 2002, 2005; Stavast et al. 2006; Nadeau et al. 2010, 2013; Wilkinson 2013; Du and Audétat 2020). Here, we employed the amphibole oxybarometer (Ridolfi et al. 2010) 349 350 to track the oxidation state of the three plutons. As illustrated in Figures 4C and D, the 351 amphibole crystals in the Tonglushan, Tieshan, and Jinshandian plutons record magmatic 352 oxygen fugacities ranging from $\Delta NNO + 1.0$ to $\Delta NNO + 2.5$, $\Delta NNO + 1.0$ to $\Delta NNO + 2.2$

(Zhou et al. 2020a; Duan et al. 2017), and Δ NNO +0.9 to Δ NNO +2.1, respectively. Thus, there is no systematic difference in magmatic oxygen fugacity between Cu and Cu-poor Fe mineralizing magmas. However, it is noteworthy that the oxygen fugacities of euhedral amphibole crystals in the Jinshandian pluton decrease with cooling, whereas those of other amphibole crystals increase with decreasing temperature (Fig. 4D). Combined with the evolving trends of water, we can speculate that magmatic oxygen fugacities will decrease with crystallization but increase with crystallization accompanying fluid exsolution.

360

361 The efficiency of copper extraction

362 The significant enrichment of Cu from magmas with normal concentrations of Cu 363 (~50–100 ppm) (Cline and Bodnar 1991; Richards 2015; Chelle-Michou et al. 2017; Zhang 364 and Audétat 2017) to anomalously Cu-rich fluids (Audétat 2019) may be described by fluid separation due to very high fluid/melt partition coefficients (Zajacz et al. 2008; Audétat 2019). 365 366 Assuming that the common factors such as sulfur contents, host rocks and depths are beneficial for mineralization, fluid exsolution can be observed in many shallow intrusions but 367 368 only a small portion of these produce Cu mineralization, suggesting that there are other factors that downgrade the Cu mineralization potential of a barren intermediate-felsic 369 370 intrusion. The intermediate-felsic intrusions represent residual material after fluid exsolution, 371 and therefore offer clues to the mineralization potential of a given system.

For this purpose, we present a systematic comparison of the compositions of common minerals in the Tonglushan, Tieshan, and Jinshandian plutons, including clinopyroxene, plagioclase, K-feldspar, amphibole, titanite, and apatite. Two important factors should be 375 addressed before such comparisons, (1) exclusion of data that were contaminated by crystal, 376 melt and fluid inclusions; and (2) establishing the influence of disequilibrium zoning patterns 377 (such as sector zoning) on mineral compositions. Small crystals, melt and/or fluid inclusions 378 are easily trapped by large crystals during magmatic crystallization (e.g., Halter et al. 2004) 379 and their presence may affect microanalyses. Examination of the transient signals from analyses is a useful approach for identifying fine-scale inclusions in minerals. All 380 381 microanalytical data for mineral Cu content was checked in transient signals and the data 382 showing the presence of inclusions were excluded. Another problem is the influence of the 383 zoning patterns that were produced by disequilibrium crystal growth. Unlike commonly 384 observed equilibrium zonation in igneous crystals such as normal, reverse and oscillatory 385 zoning patterns, sector zoning is a kinetically induced compositional zoning in which 386 compositionally heterogeneous domains might have crystallized from the same melt at similar conditions (e.g., Zhou et al. 2021). An example of its influence on mineral compositions is 387 388 provided by titanite from the Tonglushan pluton (Fig. 5). Titanite (CaTiSiO₅) is a common Ti-bearing accessory phase in granitoid rocks (e.g., Piccoli et al. 2000) and hydrothermal 389 390 systems (e.g., Chelle-Michou et al. 2015; Song et al. 2019), and has very high contents of 391 certain trace elements (e.g., REE, Y, Zr, Nb, U, Th, etc.). Due to low cation diffusivities, 392 titanite commonly develops sector zoning, even with relatively low crystal growth rates (Paterson and Stephens 1992; Watson and Liang 1995; Kohn 2017). In metaluminous 393 394 granitoids, the crystal habit of titanite is dominated by {111} (Paterson and Stephens 1992). 395 Exactly which faces are cut depends on the angle at which the plane of the thin section 396 intersects the grain, and the resulting zoning patterns are diverse and complex. In general, the

397 dominant {111} sectors have lower grayscale values in back-scattered electron (BSE) images, 398 and lower REE, Zr, U, Pb, Nb and Y contents relative to non-{111} sectors (Paterson and 399 Stephens 1992; Kohn 2017). The results of this study show that non-{111} sectors have 400 higher P, Y, Zr, Nb, Ta, REE, Pb, and U contents than those of {111} sectors by factors of ~2.5, ~ 2 , ~ 3 , ~ 5 , ~ 6 , ~ 2 , ~ 3 , and ~ 5 , respectively. However, there is no pronounced difference in Cu 401 402 contents among different sectors. For example, in the titanite crystal in Figure 5B, the average 403 Cu contents of the {111} and non-{111} sectors are 2.70 and 2.21 ppm, respectively. 404 Although there is no experimentally determined diffusivity of Cu in titanite to date (Kohn 405 2017), high values may be inferred because the low valence of Cu facilitates its diffusion in minerals compared to most other elements (Audétat et al. 2018). The lack of obvious 406 407 difference in Cu content among different sectors in titanite may be ascribed to the low ratios 408 of growth rate to lattice diffusivity (Watson and Liang 1995). Thus, our results indicate that 409 the influence of disequilibrium zoning on titanite Cu contents is limited. Disequilibrium 410 zoning has not been observed in other minerals.

The above assessment demonstrates that it is possible to make meaningful comparisons 411 412 between the mineral compositions of the Tonglushan, Tieshan, and Jinshandian plutons. A systematic comparison of mineral Cu concentrations of the three plutons are illustrated in 413 414 Figure 6 and Table 7. The result shows that almost all minerals in the Jinshandian pluton have higher Cu contents than those in the Tonglushan and Tieshan plutons. The Tonglushan deposit 415 416 contains 1.08 Mt Cu and the Tieshan deposit contains 0.67 Mt Cu, but no economic Cu ore 417 body has been found in the Jinshandian deposit. It is unlikely that the parental magmas of the 418 Jinshandian pluton are more Cu-rich than those of the Tonglushan and Tieshan plutons.

Higher mineral Cu contents may also be attributed to higher partition coefficients of Cu between mineral and melt. However, it is implausible that all minerals in the Jinshandian pluton have higher mineral-melt partition coefficients of Cu at the same time. Thus, systematically lower mineral Cu contents of the Cu- mineralized plutons should be ascribed to Cu being extracted from these plutons by exsolved fluids more efficiently and completely.

424

425 **Possible factors that affect the Cu-mineralizing potential**

426 At the transition from magmatic to hydrothermal processes, a key factor responsible for 427 the lack of economic mineralization in barren plutons is inefficient fluid exsolution and 428 extraction during their solidification (e.g., Zhang and Audétat 2018). On the basis of our 429 comparison of the three skarn associated plutons, that all underwent efficient fluid exsolution 430 and extraction, this study presents evidence that the minerals in Cu mineralizing plutons have distinctly lower Cu concentrations and show more efficient copper extraction relative to 431 432 Cu-poor Fe mineralizing plutons. Our results are consistent with observations in certain other porphyry copper systems. For example, in the Los Bronces-Río Blanco district (central Chile), 433 434 the most productive porphyry Cu province in the world (Sillitoe 2012), plagioclase phenocrysts from fertile porphyries contain significantly lower Cu contents (~0.5 ppm) than 435 436 those from barren intrusions (~6 ppm) (Williamson et al. 2016). Collectively, these results demonstrate that the efficiency of copper extraction from magmas plays a critical role in 437 438 determining Cu mineralization potential (e.g., Cline and Bodnar 1991; Richards 2015; 439 Chelle-Michou et al. 2017; Zhang and Audétat 2017). More significantly, extensive skarn 440 alteration also developed around the Cu-poor Jinshandian Fe mineralized pluton, excluding

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the possibility that the lower efficiency of Cu extraction from the Jinshandian pluton was caused by inefficient fluid exsolution and extraction. It is also unlikely that the initial Cu contents of the Jinshandian pluton are higher than those of the other Cu-mineralized plutons. There must have been other factors that downgraded the Cu-mineralizing potential of the Jinshandian pluton.

446 Here we present two possible explanations for inefficient copper extraction during the 447 solidification of the Jinshandian pluton. Anionic ligands are required to transport Cu from 448 magmas into hydrothermal fluids and chloride complexes are conventionally regarded as the 449 dominant carrier of Cu in aqueous fluids (e.g., Holland 1972; Burnham 1967, 1997; Burnham 450 and Ohmoto 1980; Candela and Holland 1984). Abundant Fe skarn ore bodies around the 451 Jinshandian pluton indicate no shortage of the chloride anion (Cl) in the exsolving fluids 452 because iron is transported as chloride complexes in aqueous fluids (e.g., Simon et al. 2004; Zajacz et al. 2008), as also evidenced by halogen-bearing minerals such as scapolite and 453 454 amphibole in the Jinshandian skarns (Zhu et al. 2015). In addition, many ore bodies in the Jinshandian deposit are distributed within the endoskarns (Zhu et al. 2017) whose protoliths 455 456 are igneous rocks (Meinert et al. 2005), indicating that Cl⁻ was derived from the intrusion rather than from the evaporate-bearing country rocks. Thus, the inefficient transport of Cu 457 458 from magmas into fluids is unlikely a result of a lack of Cl⁻ in parent magmas. Experimental studies suggest that sulfur plays an important role in the transport of Cu from melts into 459 460 alteration-mineralization zones, particularly in exsolving magmatic vapors (e.g., Zajacz et al. 461 2008, 2011; Seo et al. 2009; Zajacz and Halter 2009). It raises a possibility that the lower 462 efficiency of copper transfer from magmas into fluids or vapors may be affected by sulfur.

H₂S may increase Cu partitioning between felsic melts and fluids, however, SO₂ has weak 463 effect on Cu partitioning (Tattitch and Blundy 2017). Thus, the lack of reduced S species 464 465 during fluid exsolution potentially suppresses the extraction of Cu from parent magmas. However, a decrease in fluid-melt Cu partition coefficients caused by the lack of H₂S is still 466 limited (Tattitch and Blundy 2017). Recent studies emphasized that Cu will be extracted 467 468 efficiently by hypersaline liquid at low pressures and high temperatures (Blundy et al. 2021; 469 Tattitch et al. 2021), and Cu mineralization preferentially occurs when both Cu and Cl are 470 enriched in residual magmas concurrently (Tattitch et al. 2021). Thus another possible explanation is that Cu and Cl evolved along different paths during the solidification of the 471 472 Jinshandian pluton and one of them was depleted at the point of fluid saturation.

473

474

IMPLICATIONS

This study presents a comparison of three contrasting types of mineralization associated 475 476 with plutons in the Edong district where two plutons are related to Cu mineralization and the other is a Cu-poor, Fe mineralized pluton. Extensive skarn alteration around the three plutons 477 478 show that efficient fluid exsolution occurred during their solidification. The three plutons had 479 similar oxygen fugacities (within a range of $\sim\Delta NNO + 0.9$ to $\Delta NNO + 2.5$). Almost all 480 minerals in the Cu mineralizing plutons have lower Cu concentrations than those of the Cu-poor Fe mineralizing pluton, indicating the Cu mineralizing plutons underwent more 481 482 efficient copper extraction. Thus, the efficiency of copper extraction from magmas plays a 483 critical role in determining Cu mineralization potential. Our results indicate that a variety of 484 igneous minerals with anomalously low Cu contents could potentially be used as a tool to

485	identify Cu mineralizing magma body in a deposit with multiphase intrusions, nevertheless a
486	suite of igneous mineral compositions from a region should be analyzed for comparison. Our
487	results suggest that the inefficient copper extraction from magma body may be ascribed to the
488	lack of reduced S species during fluid exsolution or different evolution paths of Cu and Cl
489	during magma crystallization.
490	
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- 784

785 Figure captions

- **Fig. 1.** Simplified geological map showing the distribution of Cretaceous intrusions, volcanics and
- 787 magmatic-hydrothermal ore deposits in the Edong District. Modified from Xie et al. (2011) and Li et al.
- 788 (2014). The inset map shows regional-scale characteristics and the location of the Edong District (modified
- 789 from Li et al., 2009).
- 790 **Fig. 2.** (A) Geologic map of the Tieshan Fe-Cu deposit (from Zhu et al., 2019); (B) Geologic map of
- the Jinshandian Fe deposit (modified from Zhu et al., 2017); (C) Geologic map of the Tonglushan Fe
- 792 deposit (from Li et al., 2014).

/93	Fig. 3. Photographs of titanites from the three plutons. (A) Magmatic titanite and apatite grains in the
794	Tonglushan pluton; (B) Magmatic titanite in the Tonglushan pluton; (C) Euhedral amphibole in the Tieshan
795	pluton; (D) Anhedral amphibole in the Tieshan pluton; (E) Magmatic titanite in the Jinshandian pluton; (F)
796	Secondary titanite with Fe-Ti oxide inclusions with irregular shapes in the Jinshandian pluton. Mineral
797	abbreviations: Ttn = titanite, Ap = apatite, Amp = amphibole, Pl = plagioclase, Qz = quartz, Mag =
798	magnetite.
799	Fig. 4. (A) Plot of SiO_2 contents in equilibrium with amphiboles versus temperatures. Equations 5
800	and 10 from Putirka (2016) were employed to estimate temperature and equilibrium melt SiO ₂ contents.; (B)
801	Magmatic water contents versus temperatures. Equation 3 of Ridolfi et al. (2010) was used to calculate
802	water content, and Equation 5 of Putirka (2016) was employed to estimate temperature; (C) and (D)
803	Magmatic oxygen fugacities versus temperatures. Equation 2 of Ridolfi et al. (2010) was used to calculate
804	oxygen fugacity, and Equation 5 of Putirka (2016) was employed to estimate temperature. The calibrations
805	of the nickel-nickel oxide (NNO) buffer is taken from O'Neill & Pownceby (1993).
806	Fig. 5. BSE and X-ray mapping images of two euhedral, magmatic titanite crystals with sector zoning.
807	They are from the Tonglushan pluton. Cu and Ce concentrations are marked in different sectors of crystal

808 (B).

Fig. 6. Box-whisker plots of mineral Cu concentrations for the Tonglushan, Tieshan, and Jinshandian plutons. Number of analyses is marked above each box plot. The top and bottom of the boxes are the first and second quartiles. The black line is the median, and the full circle is the average. The whiskers represent the values within 1.5 times the interquartile range beyond the box edges. The full rhombuses represent outliers.

814

815 Table caption

816	Table 1. R	epresentative	whole-rock	c major and	trace element	compositions of	of the T	onglushan,	Tieshan

- 817 and Jinshandian plutonic rocks
- 818 **Table 2.** Representative clinopyroxene major and trace element compositions
- 819 **Table 3.** Representative feldspar major and trace element compositions
- 820 **Table 4.** Representative amphibole major and trace element compositions
- 821 **Table 5.** Representative titanite major and trace element compositions
- 822 **Table 6.** Representative apatite major and trace element compositions
- 823 **Table 7.** Comparison of the Tonglushan, Tieshan, and Jinshandian plutons

824



Figure 1



Figure 2



Figure 3



Figure 4



Figure 5

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Figure 6

Sample:	HB-008	HB-009	HB001	HB003	HB006	HB007	ts1	ts4
Deposit	Tonglusha	n Cu-Fe-Au deposit		Т	ieshan Fe-Cu s	karn deposit		
Lithology	Ouartz mo	onzodiorite			Ouartz d	iorite		
aior oxides(wt	%)				(
SiO ₂	63.68	62.64	63.77	64.73	64.19	62.84	63.12	62.8
TiO ₂	0.55	0.56	0.54	0.57	0.50	0.56	0.67	0.65
Al_2O_3	15.92	16.17	15.81	15.76	15.98	16.47	16.15	16.1
TFe ₂ O ₃	4.89	4.86	4.30	4.32	4.28	4.18	4.83	4.82
MnO	0.08	0.09	0.05	0.06	0.07	0.05	0.07	0.07
MgO	1.44	0.86	1.62	1.68	1.42	1.88	1.81	1.88
CaO	4.98	5.55	3.99	3.68	4.21	3.94	4.20	4.29
Na ₂ O	4.35	4.53	4.84	5.02	5.57	5.60	5.07	5.07
K ₂ O	2.75	2.71	3.30	2.97	2.67	3.06	2.98	3.17
P_2O_5	0.26	0.26	0.24	0.24	0.23	0.29	0.29	0.29
LOI	1.06	1.73	1.32	0.83	0.73	0.75	0.67	0.59
Total	99.73	99.73	99.53	99.65	99.63	99.43	99.62	99.6
race elements (ppm)							
Cr	6.98	6.65	25.7	25.2	12.9	22.7	19.1	19.2
Ni	5.83	6.52	14.46	15.74	8.01	18.5	14.0	14.9
Со	9.45	9.29	7.93	10.5	8.75	10.5	10.5	10.6
Sc	6.35	5.38	4.98	5.95	5.62	6.29	6.70	6.42
V	70.9	68.6	69.0	63.4	60.1	70.6	69.7	71.0
Pb	8.78	8.53	23.62	11.22	10.75	8.85	9.19	9.62
Rb	93.1	89.3	85.0	72.6	57.3	53.8	95.2	63.9
Cs	1.06	1.11	1.81	1.27	0.56	0.35	0.99	0.57
Ba	734	692	1016	985	1210	2238	1157	124
Sr	961	983	1609	1433	1786	2100	1467	153
Ga	22.3	21.2	24.1	24.5	24.0	23.6	23.9	24.2
Та	1.03	1.08	0.75	0.78	0.80	0.72	0.86	0.82
Nb	18.1	18.0	13.4	14.2	15.0	13.8	14.3	14.0
Hf	4.04	4.19	5.07	5.28	3.56	2.73	5.23	5.1
Zr	131	131	166	168	121	79	187	184
Y	17.0	18.3	13.3	13.8	15.0	14.3	16.3	15.9
Th	13.4	13.1	7.5	8.3	6.4	7.7	10.1	9.9
U	3.24	2.94	2.24	2.35	1.73	1.44	2.24	2.2
La	51.0	51.0	42.6	46.6	53.4	58.7	59.6	61.7
Ce	97.7	96.5	84.8	90.1	105	118	113	116
Pr	10.3	10.0	9.32	9.82	11.4	13.0	13.3	13.0
Nd	37.8	38.2	35.4	36.6	42.2	50.4	50.0	51.3
Sm	6.83	6.94	6.41	6.91	7.76	8.33	7.64	7.6
Eu	1.78	1.78	1.73	1.71	1.91	2.16	1.90	1.92
Gd	4.93	5.03	4.40	4.50	5.08	5.59	4.39	4.3
Tb	0.63	0.62	0.55	0.58	0.66	0.64	0.64	0.6
Dv	3.34	3.43	2.85	2.86	3.12	2.94	3.36	3.3
Но	0.63	0.69	0.49	0.49	0.56	0.54	0.60	0.5
Er	1.88	2.07	1.29	1.46	1.50	1.47	1.61	1.50
Tm	0.25	0.28	0.18	0.18	0.22	0.20	0.22	0.22
Vh	1.71	1.66	1.00	1.20	1.20	1.22	1.40	1 /

Lu	0.24	0.24	0.15	0.16	0.21	0.15	0.23	0.22
Sr/Y	56.4	53.8	121	104	119	147	90.1	96.5
Eu/Eu*	0.94	0.92	0.99	0.94	0.93	0.97	1.00	1.01
Dy/Dy*	0.52	0.54	0.64	0.59	0.61	0.55	0.57	0.56
(La/Yb) _N	21.1	21.7	27.6	27.6	31.6	33.9	30.2	31.1
Data source	Zhou et a	ıl. (2020a)			Zhou et al.	(2020a)		

Notes: LOI is loss on ignition.

•

an plutonic rocks													
JSD-1	JSD-2	JSD-3	JSD-6	JSD-8									
	Jinshandian Fe skarn deposit Quartz diorite												
	Quartz diorite												
(8.24	((57	59.09	(())	(0.(5									
68.24	66.57	58.08	66.38	69.65									
0.65	0.69	0.86	0.85	0.50									
15.11	15.50	14.59	15.//	14.34									
1./1	1.91	9.78	1.89	3.15									
0.02	0.03	0.04	0.03	0.04									
1.11	1.05	1.35	1.10	0.90									
2.43	2.95	3.67	3.29	1.02									
0.80	/.44	4.13	/.01	4.44									
1.64	1.21	2.87	1.41	4.91									
0.15	0.16	0.20	0.23	0.14									
2.00	2.44	4.91	1.34	0.84									
99.92	99.95	100.48	99.90	99.93									
2.80	2.95	1.60	1.89	2.92									
2.34	1.82	2.17	1.52	2.04									
2.49	2.03	5.15	3.43	2.42									
5.38	6.01	6.10	5.17	4.26									
26.8	26.6	43.9	48.2	32.5									
3.51	2.61	2.22	4.18	6.12									
33.4	25.9	75.4	32.4	182									
0.27	0.11	0.93	0.49	0.65									
671	544	815	775	879									
277	180	190	207	266									
19.5	20.0	18.2	20.2	17.8									
2.26	2.10	1.95	2.28	2.16									
29.5	28.4	30.1	34.5	22.3									
8.45	10.67	7.41	9.47	6.34									
305	415	273	367	227									
29.7	30.4	21.4	36.2	20.4									
21.0	19.3	15.2	18.6	23.5									
3.23	3.12	2.39	3.13	5.52									
43.5	49.5	75.4	64.1	35.4									
93.9	103	126	136	69.5									
11.1	11.8	13.3	16.2	7.95									
39.6	41.1	44.9	57.1	28.0									
6.69	6.71	6.59	8.99	4.55									
1.38	1.32	1.20	1.87	0.85									
5.35	5.16	4.21	6.57	3.37									
0.91	0.91	0.74	1.15	0.59									
5.36	5.52	4.07	6.80	3.61									
1.08	1.11	0.77	1.33	0.73									
3.11	3.23	2.06	3.78	2.16									
0.49	0.50	0.29	0.59	0.34									
3.26	3.42	1.86	3.76	2.31									

0.53	0.56	0.30	0.60	0.38								
9.33	5.92	8.88	5.73	13.1								
0.70	0.68	0.69	0.75	0.66								
0.56	0.54	0.53	0.57	0.51								
9.47	10.3	28.7	12.1	10.8								
	This study											

Pluton	1		17	Tieshar	1	1					Jinshandi	an
Sample				TS-1							JSD-2	
Spot	100	101	103	104	107	118	119	38	40	41	45	49
Oxide co	ntents (w	t %)										
SiO ₂	54.00	53.48	53.58	53.76	53.53	53.61	53.99	53.24	52.43	52.60	52.77	52.06
TiO ₂	0.14	0.16	0.06	0.09	0.10	0.14	0.11	0.20	0.11	0.09	0.13	0.39
Al ₂ O ₃	0.63	0.55	0.48	0.47	0.49	0.58	0.45	0.76	0.90	0.61	0.57	1.40
FeO	8.40	8.51	8.68	8.64	8.72	8.81	8.59	8.49	8.25	8.35	8.32	8.80
MnO	0.57	0.54	0.60	0.66	0.55	0.60	0.63	0.29	0.23	0.27	0.29	0.23
MgO	13.07	13.17	12.73	12.73	12.71	12.50	12.93	13.16	13.36	13.11	14.07	12.82
CaO	22.72	22.72	22.55	22.68	22.88	23.16	22.80	24.09	24.17	24.75	23.21	23.82
Na ₂ O	0.43	0.39	0.54	0.60	0.50	0.55	0.54	0.46	0.61	0.52	0.54	0.72
K ₂ O	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	0.01	0.00	0.01	b.d.l.	b.d.l.	0.01	0.01
P_2O_5	0.00	0.00	0.04	0.02	0.06	0.03	0.01	0.01	0.02	0.03	0.03	0.00
Cr ₂ O ₂	0.00	0.01	b.d.l.	0.04	0.00	b.d.l.	0.04	b.d.l.	0.01	0.02	0.05	b.d.l.
F	b.d.l.	b.d.l.	b.d.l.	b.d.l.	h.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.
Cl	0.01	0.01	0.02	b.d.l.	0.00	0.00	b.d.l.	0.01	0.00	b.d.l.	0.01	0.00
Total	99.96	99.53	99.26	99.70	99.54	99.98	100.08	100.71	100.08	100.34	100.00	100.24
Trace ele	ment con	tents (pp	m)									
Cu	0.53	0.07	0.65	0.02	0.10	0.48	0.51	0.77	0.76	0.65	0.74	0.85
Zn	166	171	210	167	172	166	175	145	132	142	172	144
Pb	0.73	0.63	3.41	0.66	0.64	0.60	1.78	0.98	1.46	0.61	1.96	0.62
Р	12.6	6.27	9.24	5.65	18.71	10.9	10.7	11.3	6.56	10.2	8.09	11.4
Li	53.0	55.3	57.7	55.4	62.6	64.5	62.1	57.9	49.1	48.0	65.8	59.2
Sc	101	87.4	90.9	108.0	88.5	92.9	96.6	78.1	72.5	81.4	47.7	87.5
V	95.6	98.7	131	93.2	130	94.6	116	150	152	156	116	183
Cr	12.9	16.1	31.6	26.9	38.8	15.4	24.4	62.9	51.1	68.4	54.6	81.4
Co	38.2	42.8	42.0	44.2	42.3	42.3	40.3	35.7	35.2	35.3	43.3	38.8
N1	23.8	28.4	25.5	30.2	23.5	30.0	21.6	55.0	44.7	46.9	63.4	48.7
Ga	/.38	8.18	12.3	/.12	10.4	/.64	8.85	13.3	11.1	8.79	15.2	15.5
KD S.	54.1	0.03	1.28	0.03	0.00	0.05	0.81	0.00	0.04	0.01 52.0	0.17	0.02
Sr V	34.1 12.1	37.9 7.08	32.7 22.6	23.0 6.74	37.7 15.2	27.2 7.14	41.0	83.4 20.1	/0./	52.9 12.4	54.0 52.2	04.0 26.5
1 7r	22.1	24.0	23.0	21.1	78.0	13.3	14.5	103	51.2	76.6	23.6	20.5
Nh	0.04	0.02	0.20	0.02	0.06	0.01	0.06	0.07	0.05	0.01	0.03	0.10
Cs	0.00	0.00	0.29	0.00	0.00	0.03	0.20	0.01	0.00	0.00	0.02	0.00
Ba	0.05	0.00	1.46	0.00	0.00	0.00	0.63	0.15	0.23	0.22	0.41	0.10
La	14.0	11.8	33.5	9.9	22.3	10.7	16.5	22.9	17.7	10.5	23.0	24.6
Ce	44.2	33.6	94.6	26.1	62.7	27.1	50.1	71.7	57.4	33.0	82.8	80.7
Pr	6.25	4.31	12.7	3.14	8.85	3.39	7.27	10.5	8.39	5.00	13.4	12.3
Nd	29.8	17.9	53.5	15.6	36.4	14.5	32.8	45.7	37.3	21.0	65.9	55.7
Sm	5.11	3.58	10.07	3.08	6.32	2.74	6.20	9.16	7.69	4.68	16.42	11.81
Eu	1.25	0.66	1.94	0.58	1.43	0.54	1.37	2.33	1.77	1.08	2.79	2.49
Gd	3.64	2.63	7.59	1.72	4.86	2.28	4.62	7.03	6.06	3.83	14.08	9.68
Tb	0.48	0.31	0.79	0.20	0.56	0.25	0.58	0.89	0.72	0.51	1.99	1.11
Dy	2.55	1.62	4.57	1.32	2.84	1.41	3.12	4.14	3.41	2.79	10.84	6.02
Ho	0.37	0.27	0.85	0.20	0.57	0.25	0.50	0.64	0.57	0.40	1.87	0.93
Er	1.32	0.75	2.39	0.61	1.25	0.82	1.63	1.72	1.64	1.22	4.94	2.69
Im	0.18	0.12	0.31	0.09	0.18	0.10	0.16	0.24	0.23	0.20	0.64	0.28
Y D	1.30	0.99	2.01	0.67	1.20	0.66	1.51	1.39	1.3/	1.35	4.18	1.80
Lu	0.27	0.17	0.38	0.19	0.30	0.1/	0.24	0.28	0.25	0.23	0.65	0.30
	1.31	1.02	5.48 0.01	1.27	4.81	0.95	5.00	5.70	2.73	5.21 0.01	2.20	5.52 0.01
ra Th	0.00	0.00	0.01	0.00	0.01	0.00	0.00	0.08	0.01	0.01	0.00	0.01
III II	0.04	0.00	0.31	0.12	0.14	0.23	0.07	0.04	0.04	0.04	0.10	0.11
Data	0.01	0.01	0.51	0.05	0.01	0.02	0.07	0.01	0.00	0.01	0.02	0.02
source			Zh	ou et al. (2	2020a)						This stuc	ły

Table 2. Representative clinopyroxene major and trace element compositions

Notes: b.d.l., below detection limit.

54	55
52 91	52 35
0.22	0.27
0.75	0.75
8.04	8.26
0.33	0.31
13.35	13.38
24.02	24.33
0.63	0.55
0.02	0.01
0.06	0.02 h.d.1
0.05 h.d.l	b.d.1.
b.d.l.	b.d.l.
100.34	100.24
0.93	0.72
167	152
10.94	10.79
64.4	50.6
77.8	65.8
203	140
338	113
37.4	35.2 65.8
20.1	12.7
0.00	0.00
69.9	74.6
25.3	23.6
214	108
0.22	0.06
0.00	0.00
37.8	23.8
112.2	75.4
14.7	11.0
58.9	47.0
9.39	9.40 1.87
7.73	7.94
0.89	0.93
4.93	4.79
0.78	0.81
2.28	2.07
2.65	1.70
0.43	0.36
6.54	3.68
0.03	0.00
0.13	0.04
0.01	0.01

Pluton	tepresent	, ,	Fonglusha	n	element		ons	Ties	shan			
Sample			HB008			TS-1	TS-3	HB002	HB004	HB007	HB001	
Spot	24	43	45	46	47	110	206	237	88	163	211	71
Oxide co	ntents (w	t %)	15	10	17	110	200	237	00	105	211	/1
SiO	61.87	53 56	53 56	54 71	55 89	65 44	62.00	63 32	62 67	65 13	62.25	65 11
TiO	b.d.1	b.d.1	b.d.1	b d 1	b.d.1	b.d.1	0.05	b.d.1	0.03	b.d.1	b.d.1	0.03
1102	22.50	20.50	0.0.1.	27.90	0.0.1.	22.56	22.74	22.22	0.05	21.22	22.06	0.05
AI_2O_3	23.59	28.58	27.47	27.80	27.41	22.56	22.74	23.23	23.03	21.23	23.06	21.42
FeO M=O	0.24	0.17	0.10	0.24	0.26	0.13	0.14	0.10	0.16	0.20	0.14	0.20
MaQ	0.01	0.01	0.02	0.03	b.d.1.	0.0.1. b.d.1	0.05 h.d.1	0.0.1. h d 1	0.01	0.04 b.d.1	0.02 h.d.1	0.0.1. h.d.1
MgO CeO	5.82	11 02	0.05	11.28	0.d.1. 10.51	0.d.1. 2.75	0.0.1. 1 27	5.06	4.08	2 22	0.a.i. 4 74	0.d.1. 2 75
Na O	7.03	11.95	5.04	5 47	5 34	5.75 7.42	9.37	5.00 8.40	4.90	9.22	4.74 8.72	9.75
Ka ₂ O	0.52	4.99 0.15	0.19	0.17	0.10	0.26	9.52	0.79	0.77	0.95	0.72	9.55
K ₂ O	0.55	0.15	0.18	0.17	0.19	0.26	0.32	0.30	0.37	0.85	0.34	0.30
P_2O_5	0.01	0.03	0.01	0.01	0.01	0.02	0.02	b.d.l.	0.02	b.d.l.	0.01	b.d.l.
Cr_2O_3	b.d.l.	0.04	0.26	0.35	b.d.l.	b.d.l.	b.d.l.	0.03	0.31	0.00	0.14	b.d.l.
F	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.
	b.d.l.	0.01	b.d.l.	0.00	0.01	0.01	0.00	b.d.l.	0.01	b.d.l.	0.01	b.d.l.
l otal	100.01	99.45	98.24	100.17	99.62	99.59	98.98	100.53	100.05	100.15	99.42	100.43
I race ele			om)	0.68	0.40	1.21	0.65	0.76	0.87	0.41	0.28	11.60
Cu Zn	7.47	0.29	6.08	0.08	0.40	3 30	0.05	0.70	6.40	10.1	0.58	2 72
Dh	0.30	1 80	2 00	5.45	5.53	5.59 8.78	6.94	7.60	0.49	0.06	12 /0	834
P	9.30 44 2	7.09 21.4	191	27.9	21.7	3.5	0.94 24 8	27.9	12.6	34.1	19.49	14 7
Ti	7.26	5 04	0.63	0.91	0.87	2.5 2.77	0.00	12 67	7 92	13.06	25.47	2 49
Sc	2.93	2.14	2.13	2 40	1 71	2.77	2.62	3 20	3 39	2 10	3.07	6.12
V	0.20	3.28	0.00	0.04	0.12	0.00	0.00	0.37	0.18	0.02	2.63	0.02
Ċr	17.5	31.2	7.4	10.6	12.4	18.0	7.8	10.1	8.6	4.5	6.9	13.1
Co	0.30	1.91	0.22	0.51	0.14	0.13	0.00	0.06	0.07	0.08	0.58	0.10
Ni	3.80	0.00	0.00	0.74	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.79
Ga	27.6	22.4	20.4	24.3	21.6	29.9	32.6	25.9	26.1	26.5	26.6	29.2
Rb	9.51	1.29	0.71	0.28	0.83	1.47	3.41	0.00	1.05	4.17	11.43	0.78
Sr	1884	2752	2029	2338	1806	1910	1396	4641	3834	3551	3690	1605
Y	0.09	0.29	0.11	0.06	0.03	0.04	0.04	0.18	0.08	0.05	0.06	0.21
Zr	0.00	0.23	0.00	0.00	0.03	0.10	0.00	0.06	0.00	0.00	0.00	0.01
Nb	0.04	0.03	0.01	0.04	0.02	0.13	0.03	0.08	0.00	0.01	0.00	0.23
Cs	0.17	0.18	0.07	0.00	0.07	0.00	0.00	0.00	0.01	0.19	0.63	0.00
Ba	373	187	186	284	554	308	184	363	907	968	518	250
La	9.59	5.73	6.48	6.50	9.67	6.87	4.93	7.51	6.32	4.87	6.78	8.07
Ce	9.46	7.20	7.32	9.50	10.60	4.79	2.22	9.06	5.99	4.20	7.83	4.89
Pr	0.50	0.70	0.48	0.68	0.58	0.18	0.09	0.61	0.29	0.32	0.40	0.20
Nd	0.68	1.83	0.92	1.44	1.71	0.21	0.08	1.82	0.35	0.49	1.49	0.52
Sm Eu	0.15	0.02	0.00	0.11	0.00	0.00	0.00	0.16	0.00	0.04	0.13	0.00
Eu	0.86	0.78	0.57	0.82	0.94	0.64	0.18	0.91	0.90	0.61	0.95	0.42
Са ТЬ	0.07	0.08	0.00	0.12	0.00	0.03	0.00	0.07	0.00	0.01	0.05	0.04
	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00
Но	0.00	0.00	0.00	0.00	0.05	0.00	0.00	0.00	0.02	0.10	0.05	0.00
Er	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00
Tm	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Yb	0.00	0.11	0.02	0.00	0.00	0.04	0.00	0.02	0.00	0.00	0.00	0.00
Lu	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.01	0.00
Hf	0.02	0.00	0.00	0.03	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00
Та	0.00	0.02	0.02	0.01	0.02	0.19	0.06	0.01	0.00	0.00	0.02	0.34
Th	0.06	0.34	0.00	0.01	0.00	0.00	0.00	0.05	0.00	0.00	0.01	0.00
U	0.04	0.10	0.00	0.00	0.00	0.01	0.00	0.02	0.00	0.01	0.02	0.00
Data		Zho	u et al. (20	020a)				Zhou et a	l. (2020a)			
source		2110							(20200)			

T 11 0 D	•	C 1 1	• •		•.•
Table & Rei	nrocontativo	toldenar	maior and	trace element	compositions
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Notes: b.d.l., below detection limit.

	Jinshandia	n	
	JSD-2		
72	82	83	89
64.92	62.20	63.62	63.90
0.01	b.d.1	b.d.1	0.02
0.01	22.10	22.40	0.02
21.21	23.18	22.49	21.45
0.21	0.16	0.13	0.13
0.01	b.d.l.	b.d.l.	b.d.l.
b.d.l.	b.d.l.	b.d.l.	0.00
3.00	5.17	4.92	4.22
10.07	8.86	9.22	8.18
0.33	0.29	0.22	0.32
0.03	0.00	0.03	b.d.l.
b.d.l.	b.d.l.	b.d.l.	0.20
b.d.l.	b.d.l.	b.d.l.	b.d.l.
b.d.l.	b.d.l.	b.d.l.	b.d.l.
99.78	99.87	100.63	98.41
11.71	11.44	11.69	11.45
3.16	3.57	3.01	3.44
9.40	9.32	8.91	9.78
20.3	19.7	9.9	16.7
4.31	0.00	1.17	0.34
5.24	5.24	5.20	5.01
0.11	0.23	0.00	0.04
9.4	14.7	21.0	18.8
0.06	0.02	0.07	0.01
1.14	0.67	1.21	0.87
27.5	24.8	26.2	25.3
0.71	0.42	1.80	0.70
1622	1866	1839	1865
0.24	0.19	0.07	0.16
0.05	0.00	0.05	0.03
0.18	0.17	0.15	0.15
0.00	0.03	0.04	0.07
440	151	125	267
6.96	7 64	6.02	8 4 9
4 76	4 78	3.22	5.07
0.19	0.19	0.14	0.20
0.65	0.19	0.25	0.20
0.02	0.00	0.00	0.03
0.72	0.00	0.00	0.05
0.12	0.02	0.00	0.12
0.00	0.02	0.00	0.00
0.00	0.00	0.00	0.00
0.00	0.00	0.00	0.00
0.00	0.00	0.00	0.00
0.00	0.00	0.00	0.00
0.00	0.00	0.00	0.00
0.00	0.00	0.00	0.00
0.00	0.00	0.00	0.08
0.01	0.23	0.00	0.31
0.00	0.23	0.01	0.02
0.00	0.07	0.01	0.02
0.00	0.02	0.01	0.00
	This study	y	

Pluton longiusnan liesnan	-	
Sample HB008 TS-1 TS-3 HB002 HB004 HB0	07 HB001	
Spot 9 11 21 22 25 132 200 270 79 129	196	1
Oxide contents (wt %)		
SiO ₂ 50.86 51.13 50.97 50.19 50.61 50.03 50.42 47.92 48.91 48.4	8 48.05	49.39
TiO ₂ 0.77 1.02 0.88 0.94 0.87 0.81 0.70 1.25 1.17 1.20	0.91	0.83
Al ₂ O ₃ 4.73 4.48 4.68 5.02 4.85 5.18 4.29 6.33 5.82 6.00	5.38	4.94
FeO 11.42 11.45 11.38 11.61 11.34 13.12 13.35 13.44 13.23 13.2	6 13.13	13.78
MnO 0.62 0.60 0.54 0.48 0.55 0.38 0.40 0.36 0.47 0.26	0.33	0.18
MgO 15.56 15.61 15.31 14.93 15.27 13.93 14.22 13.68 13.75 13.7	9 13.69	14.32
CaO 11.43 11.70 11.79 12.42 11.67 11.99 11.83 11.56 11.52 11.5	5 12.06	12.13
Na ₂ O 1.23 1.25 1.23 1.23 1.16 1.01 1.22 1.51 1.52 1.60	1.48	1.56
$K_{2}O = 0.49 = 0.47 = 0.49 = 0.53 = 0.50 = 0.58 = 0.51 = 0.75 = 0.68 = 0.60$	0.56	0.70
$R_{20} = 0.47 = 0.47 = 0.47 = 0.55 = 0.50 = 0.56 = 0.51 = 0.75 = 0.08 = 0.05$	0.50	0.70
P_2O_5 b.d.l. 0.02 b.d.l. 0.02 b.d.l. 0.01 0.01 0.01	0.03	0.06
Cr_2O_3 b.d.l. 0.05 0.02 0.02 0.22 0.03 0.04 0.07 0.06 b.d.	. 0.64	0.03
F 0.26 0.11 0.15 0.25 0.14 0.21 0.35 0.37 0.26 0.35	0.50	b.d.l.
Cl 0.10 0.13 0.10 0.11 0.12 0.04 0.08 0.06 0.06 0.1	0.08	0.09
Total 97.33 97.94 97.45 97.59 97.22 97.21 97.25 97.14 97.32 97.2	0 96.59	97.98
Trace element contents (ppm)		
Cu 0.08 0.34 0.18 0.27 0.22 0.53 0.67 0.36 0.32 0.4	5 1.16	1.85
Zn 202 175 164 168 170 332 371 308 354 1	324	276
Pb 0.65 0.38 1.08 1.63 0.25 1.27 0.88 1.16 2.36 4.19	7.50	0.95
P 4.2 11.6 20.6 13.5 12.5 11.8 14.3 25.1 9.6 0.0	31.3	12.9
Li 4.08 4.27 2.29 2.94 3.12 13.2 17.3 11.8 28.3 1.4	86.4	15.0
Sc 67.9 47.3 70.9 75.7 61.7 58.7 49.2 52.9 61.8 2.4	62.0	67.9
V 150 138 139 167 141 191 180 220 164 0.1	216	199
Cr 23.1 19.7 17.6 31.3 15.7 32.5 32.6 167 242 1.74	342	170
Co 52.1 53.8 54.2 52.6 53.4 72.0 74.9 61.9 65.3 0.00	48.5	69.5
Ni 34.5 38.3 36.5 34.7 34.5 72.0 109 75.1 97.4 2.0	96.2	125
Ga 18.7 16.9 17.6 21.5 16.7 23.3 23.1 24.7 19.2 21.4	23.4	22.1
Rb 4.12 3.55 3.43 5.64 3.03 3.69 4.10 1.81 7.35 0.65	21.19	3.87
Sr 10.2 11.4 12.7 16.0 12.8 73.6 56.3 42.8 44.3 1633	.9 93.8	44.2
Y 41.7 21.9 26.2 39.1 31.8 12.7 8.36 13.0 11.0 0.02	17.0	8.43
Zr 24.2 20.2 17.6 29.5 18.6 34.6 31.9 45.3 25.6 0.1	42.6	35.6
Nb 17.6 10.4 10.9 21.2 13.1 18.1 5.99 8.38 5.97 0.00	7.95	7.46
Cs 0.06 0.00 0.07 0.20 0.03 0.01 0.01 0.00 0.34 0.01	3.40	0.02
Ba 3.67 2.26 2.88 6.20 5.00 26.4 11.3 31.8 25.8 340	37.0	12.5
La 35.8 23.5 26.2 46.8 24.8 37.1 20.2 31.0 15.5 5.6	55.9	25.6
Ce 94.2 55.2 66.1 112.6 67.2 90.3 39.5 63.0 34.0 3.9	62.4	51.7
Pr 11.9 6.65 8.22 13.49 8.73 12.39 4.05 6.24 4.10 0.16	5.89	5.25
Nd 53.8 29.5 31.9 52.5 36.9 56.7 16.2 22.4 18.6 0.4	21.9	20.1
Sm 11.8 6.05 7.243 10.6 8.83 9.23 2.96 4.27 3.51 0.00	5.83	3.08
Eu 1.73 1.15 1.32 1.60 1.44 1.61 0.77 1.06 0.73 0.20	1.16	1.10
Gd 9.77 5.19 6.14 8.64 7.23 6.09 2.33 3.77 3.17 0.00	4.85	2.68
Tb 1.36 0.62 0.75 1.13 0.96 0.68 0.24 0.51 0.36 0.00	0.52	0.35
Dy 6.51 3.86 4.49 6.54 6.24 2.59 1.33 2.20 2.19 0.03	3.02	1.60
Ho 1.31 0.68 0.83 1.36 1.05 0.40 0.34 0.36 0.35 0.00	0.56	0.33
Er 4.21 2.54 2.54 3.41 3.07 1.14 0.82 1.30 1.05 0.05	5 1.84	0.79
Tm 0.54 0.25 0.33 0.43 0.41 0.12 0.11 0.11 0.09 0.00	0.32	0.08
Yb 4.17 2.05 2.32 3.75 2.83 0.66 0.69 0.93 1.09 0.00	1.68	0.59
Lu 0.55 0.32 0.36 0.63 0.43 0.11 0.13 0.16 0.13 0.00	0.32	0.14
Hf 2.08 1.36 1.11 2.23 1.41 2.32 1.47 2.65 1.90 0.00	2.27	2.04
Ta 0.18 0.08 0.10 0.20 0.13 0.34 0.04 0.08 0.05 0.00	0.10	0.06
Th 0.17 0.08 0.10 0.21 0.10 0.38 0.17 0.09 0.07 0.00	2.32	0.07
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	2.32	0.03
	2.02	0.00
Source Zhou et al. (2020a) Zhou et al. (2020a)		

Table 4. Representative amphibole major and trace element compositions

Notes: b.d.l., below detection limit.

Jinshandian					
JSD-2	JSD-4				
2	3	133	134		
49.71	51.78	49.60	50.16		
0.88	0.56	1.01	0.85		
4 58	3 56	3 71	3 54		
13 25	12.05	15.80	15.95		
0.19	0.22	0.33	0.34		
14.71	15.66	12.88	12.91		
12.13	12.38	10.93	11.10		
1.49	1.15	2.60	2.50		
0.70	0.39	0.71	0.78		
b.d1	b.d.1	b.d.1	b.41		
0.04	0.05	0.02	0.01		
0.00	0.05	0.02	0.01		
D.d.1.	D.d.1.	0.80	0.87		
0.07	0.05	0.09	0.08		
97.70	97.85	98.11	98.70		
1 92	1 55	3 32	5 28		
279	274	225	208		
2.04	0.66	0.52	2.30		
12.1	10.1	11.6	13.5		
14.9	13.2	12.1	11.6		
66.7	57.7	162.9	104.7		
189	170	44.0	62.1		
141	173	4.88	7.24		
70.0	68.4	41.5	37.1		
126	133	6.26	8.60		
21.3	17.8	32.5	40.2		
3.90	3.29	10.72	15.13		
44.1	21.8	0.46	7.92		
8.38	9.77	360	620		
33.8	41.0	235	165		
7.25	2.76	104	188		
0.03	0.00	0.06	0.01		
11.82	4.89	0.22	2.58		
24.4	16.9	40.3	48.0		
47.6	36.1	165	184		
4.85	3.94	26.81	28.97		
18.4	15.3	128	127		
5.22 1.10	5.00 0.68	49.5	51.4 0.71		
2.06	0.08	0.14 50.5	0.71 65.7		
2.90 0.30	2.31 0.30	11 2	13.0		
1.50	1 40	77 A	96.4		
0.26	0.26	12.7	17.7		
0.66	0.96	33.4	48.3		
0.09	0.13	4.16	6.10		
0.81	0.96	26.1	35.7		
0.13	0.17	3.61	4.19		
1.71	2.51	12.9	15.5		
0.04	0.03	2.81	12.05		
0.12	0.17	0.26	1.11		
0.03	0.03	0.03	0.09		
	This study	7			

1 abic 5. F	tepresen		ne major a	nu trace		ompositio	115		1			
Pluton				Tong	lushan							Jinsha
Sample		HB008 (r	nagmatic)			HB009 (1	nagmatic)			JSD	-4 (magma	ntic)
Spot	18	s11	s12	s13	s34	s35	s38	s39	138	139	143	145
Oxide con	ntents (v	vt %)										
SiO_2	30.21	31.56	31.12	30.17	30.22	30.53	30.75	30.54	30.42	30.41	30.56	30.41
TiO ₂	37.14	37.65	36.76	35.37	34.50	35.84	36.32	36.21	37.09	36.76	36.86	37.14
	0.91	0.98	1 13	1.08	1 31	1 22	1.98	1 23	0.53	0.42	0.67	0.47
F2O3	1.51	1.27	1.15	2.02	2.21	1.22	1.70	1.25	2 12	2.01	1.78	1.82
MnO	0.00	0.17	0.14	2.03	2.21	0.16	0.14	0.16	2.45 h.d.l	2.01 h.d.l	1.70	0.01
MaO	0.09	0.17	0.14	0.24	0.10 h.d.l	0.10	0.14	0.10	0.0.1.	0.02	0.02	0.01
CaO	28 50	0.02	20.25	20.03	20.42	20.10	20.10	20.05	20.15	20.20	20.42	20.66
Va O	20.30 h.d.1	27.62	50.25 h.41	29.62	0.06	29.19	29.10 h.d.1	29.20	29.13	29.20	29.43	29.00
Na ₂ O	D.d.1.	0.01	D.d.1.	0.02	0.06	0.02	0.0.1.	0.05	0.12	0.10	0.11	0.06
K_2O	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	0.01	0.02	b.d.l.	b.d.l.	0.02	b.d.l.
P_2O_5	0.09	0.07	0.05	0.12	0.09	0.07	0.07	0.08	0.08	0.13	0.08	0.11
Cr_2O_3	b.d.l.	0.04	0.04	b.d.l.	0.06	b.d.l.	0.13	0.01	0.01	b.d.l.	0.01	0.01
F	b.d.l.	0.12	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	0.19	0.36	0.17	0.07
Cl	0.01	b.d.l.	b.d.l.	b.d.l.	0.01	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	0.01	0.01
Total	98.46	99.66	100.99	98.88	99.05	99.01	100.31	99.48	100.01	99.27	99.69	99.76
Trace ele	ment co	ntents (pp	m)									
Cu	2.78	3.61	2.83	2.93	2.53	2.39	3.00	2.70	6.18	5.96	6.28	5.93
Zn	7.31	7.86	6.09	13.52	5.70	6.05	6.79	5.09	12.3	12.4	12.3	11.5
Pb	7.48	3.45	3.19	15.57	8.02	5.48	4.82	5.16	4.71	4.97	5.23	4.59
Р	243	163	168	517	312	273	342	268	209	248	254	262
Sc	14.8	12.9	13.0	21.8	24.6	19.2	15.9	18.2	17.3	16.6	25.5	25.4
V	660	591	566	597	767	668	614	629	598	598	550	523
Cr	9.77	13.7	12.6	18.4	17.5	14.5	18.1	18.3	11.9	8.92	5.55	5.58
Со	0.20	0.08	0.07	0.73	0.11	0.12	0.12	0.12	0.12	0.08	0.09	0.04
Rb	1.05	0.43	0.08	3.11	0.86	0.31	0.24	0.31	0.87	0.98	0.39	0.37
Sr	10.3	22.4	21.9	17.5	44.0	42.4	42.8	38.9	47.0	49.7	65.4	67.0
Y	2263	778	797	2720	4352	1898	1342	1355	2070	2248	920	902
Zr	693	343	347	2324	1478	1051	643	924	3068	3603	3929	3884
Nb	1857	560	565	3864	2843	1596	1480	1256	2586	2959	1847	1929
Ba	0.38	0.31	0.04	2.23	0.09	0.09	0.08	0.03	0.00	0.03	0.04	0.02
La	5502	1681	1724	7628	4512	3828	3445	3671	1125	1317	1522	1421
Ce	12816	3852	3879	16239	14768	10883	9540	9780	4384	5071	3029	2899
Pr	1372	436	444	1743	2095	1325	1099	1089	605	687	288	282
Nd	5126	1665	1700	6302	8645	4537	3601	3539	2629	2898	1073	1052
Sm	890	269	276	1001	1405	565	433	413	567	618	201	191
Eu	120	67.0	65.7	147	206	141	111	117	29.2	33.8	25.1	25.7
Gd	643	219	218	786	1039	380	287	274	487	519	184	175
Tb	79.9	26.6	27.1	98.7	142	51.6	36.7	36.4	66.5	72.2	25.1	23.9
Dy	445	136	140	514	782	288	203	199	396	426	156	150
Ho	79.2	28.1	28.3	99.4	156	59.7	41.9	41.2	71.6	78.5	30.9	29.7
Er	226	71.4	73.8	261	425	175	122	125	207	225	93.5	89.3
Tm	30.2	10.5	10.9	39.0	63.7	29.0	20.3	21.2	28.0	31.2	13.6	13.2
Yb	219	66.2	66.7	236	367	185	129	140	186	205	94.3	90.7
Lu	29.1	9.95	10.4	34.2	46.7	26.3	19.4	20.6	23.2	24.7	13.9	12.7
Hf	55.5	24.0	25.5	291	99.8	90.5	39.1	77.6	89.0	89.3	164	180
Та	77.4	14.2	13.7	332	227	106	83.2	69.0	203	211	98.5	107
Th	517	261	265	1040	649	418	343	386	430	468	496	429
U	289	93.7	89.5	382	61.4	44.3	38.5	42.6	26.3	31.7	44.7	42.7
Data				This	study							This s
source				1115	Study							1 1113 3

Table 5. Representative titanite major and trace element compositions

Notes: b.d.l., below detection limit.

ndian							
1.47	JSD-2 (secondary)						
14/	19	20	21				
31.02	30.32	30.56	30.43				
35.97	37.01	36.88	36.05				
0.59	0.78	0.69	1.02				
1.76	1.62	1 42	2 29				
h.d.l.	0.05	0.03	0.07				
0.03	0.03	0.03	0.03				
29.72	29.24	28.99	29.51				
0.07	0.04	b.d.l.	b.d.l.				
b.d.l	0.02	h d l	b d l				
0.07	0.02	0.00	0.04				
0.07	0.10	0.09	0.04				
0.02	0.03	0.01	0.02				
0.22	0.22	0.10	0.35				
b.d.l.	b.d.l.	b.d.l.	b.d.l.				
99.38	99.36	98.74	99.65				
6.69	6.12	6.44	6.03				
13.3	12.3	12.2	13.4				
9.42	6.15	8.74	9.31				
382	413	329	291				
25.4	12.3	10.9	12.5				
698	673	745	670				
6.20	139	188	87				
0.11	0.16	0.09	0.18				
0.51	0.61	0.17	0.18				
72.0	76.5	74.9	71.2				
1219	259	451	535				
2434	962	763	791				
2917	797	1018	1540				
0.73	0.32	0.06	0.31				
2140	2974	3242	4245				
4914	4932	6781	9261				
475	399	672	961				
1663	1141	2475	3441				
284	124	337	458				
41.3	108	96.2	131				
252	88.2	224	295				
33.8	8.77	21.3	26.5				
213	45.5	98.4	119				
40.5	7.7	15.1	19.0				
127	23.0	42.6	49.0				
17.6	3.21	5.24	5.94				
127	25.5	34.9	41.3				
18.5	3.74	4.76	5.71				
98.0	50.2	51.3	60.3				
109	25.9	35.1	44.1				
738	438	622	684				
97.6	106	148	104				
tudy							

Pluton	Tonglushan						
Sample	HB008 HB009						
Spot	s15	s16	s17	s29	s30		
Oxide contents (wt %)							
SiO ₂	0.21	0.05	b.d.l.	b.d.l.	0.24		
TiO ₂	b.d.l.	b.d.l.	0.07	0.03	b.d.l.		
Al_2O_3	b.d.l.	0.00	b.d.l.	0.01	b.d.l.		
FeO	0.09	0.08	0.02	0.06	0.06		
MnO	0.10	0.10	0.06	0.14	0.11		
MgO	0.01	b.d.l.	0.01	b.d.l.	0.00		
CaO	54.72	54.93	55.35	54.87	54.65		
Na ₂ O	0.00	0.02	0.09	0.07	0.05		
K ₂ O	b.d.l.	b.d.l.	b.d.l.	0.01	0.01		
P_2O_5	42.80	42.39	42.46	42.36	41.63		
Cr_2O_3	0.01	b.d.l.	b.d.l.	b.d.l.	b.d.l.		
F	3.20	3.49	3.46	2.85	3.11		
Cl	0.37	0.30	0.23	0.50	0.59		
Total	100.08	99.82	100.24	99.57	99.01		
Trace ele	ement cor	itents (pp	m)				
Cu	0.70	0.59	0.34	0.36	0.50		
Zn	0.51	0.40	0.12	0.54	0.34		
Pb	1.54	1.38	1.18	1.03	2.41		
V	11.3	10.19	9.48	10.8	17.2		
Rb	0.02	0.05	0.02	0.00	0.17		
Sr	504	509	511	432	431		
Y	199	153	148	156	403		
Zr	1.11	0.44	0.45	0.35	2.47		
Ba	0.31	0.26	0.10	0.14	0.25		
La	2635	2125	2037	1726	3959		
Ce	3183	2532	2415	2248	5307		
Pr	254	200	190	193	466		
Nd	754	580	555	602	1469		
Sm	86.9	65.2	62.0	72.2	184		
Eu	17.5	13.3	12.8	10.6	26.3		
Gd	154	121	114	109	265		
Tb	6.63	5.05	4.54	5.34	14.33		
Dy	33.8	24.8	23.8	26.9	69.1		
H0 E	6.51 10-1	4.88	4.61	5.01	13.49		
Er T	18.1	15.4	12.9	14.0	33.5		
1 m	2.39	1.81	1.03	1.01	4.30		
Ү b Т	14.1	11.5	10.5	10.2	27.0		
Lu Th	2.78	2.00	1.94	1./4	4./2		
in T	14.0	03.5	02.2	43.2 14.4	1//		
U Dete	14.0	23.3	∠4.0	14.4	33.2		
Data			This study	T			
Source							

Table 6. Representative apatite major and trace element compositions

Notes: b.d.l., below detection limit.

F	Tonglushan Tieshan Jinshandian						
	1 oligiusliali						
Deposit style	skarn		SK	skarn		skarn	
Metals	Cu-l	Fe-Au	Fe	-Cu	Fe		
Magma properties							
	Rang	Mean	Rang	Mean	Rang	Mean	
Whole-rock SiO ₂	62.6-63.7		62.4–64.7		68.2–69.3		
Whole-rock Sr/Y	54–56		88–169		5-18		
Amp(euh) T (°C)	725-809	760±19	728-801	768±16	743–795	763±21	
Amp(anh) T (°C)			695–778	747±17	720–783	751±16	
Amp(euh) H ₂ O (wt%)	3.4-4.4	3.9±0.2	3.2-4.6	3.9±0.4	1.3-2.0	1.6±0.3	
Amp (anh) H ₂ O (wt%)			3.3-4.4	3.8±0.3	2.9-3.9	3.5±0.2	
Amp(euh) fO_2	1.0-2.5	1.7±0.3	1.0-2.1	1.4±0.3	0.9–2.1	1.3±0.4	
Amp(anh) fO_2			1.1-2.2	1.5±0.3	1.1-2.0	1.6±0.2	
Mineral Cu contents (p	pm)						
	Rang	Mean	Rang	Mean	Rang	Mean	
Срх			0.02-0.94	0.36±0.29	0.65-0.93	0.80 ± 0.08	
Ар	0.03-0.70	0.41 ± 0.21					
Pl	0.06-1.31	0.52 ± 0.34	0.03-1.80	0.64 ± 0.42	1.13-12.09	10.6±2.7	
Kfs			0.04-1.80	0.69 ± 0.77	1.81-13.0	4.40±3.84	
Amp(euh)	0.08-1.15	0.42 ± 0.32	0.10-2.06	0.71±0.52	3.32-5.28	4.19±1.00	
Amp(anh)			0.03-1.01	0.28 ± 0.26	1.85-2.23	1.86±0.25	
Ttn(mag)	1.83-3.61	2.61 ± 0.50			5.59–6.69	6.03±0.30	
Ttn(sec)					5.78-6.66	6.14±0.28	

Table 7. Comparison of the Tonglushan, Tieshan, and Jinshandian plutons

Mineral abbreviations: Amp(euh) = euhedral amphibole, Amp(anh) = anhedral amphibole, Cpx = clinopyroxene, Ap = apatite, Pl = plagioclase, Kfs = K-feldspar, Ttn(mag) = magmatic titanite, Ttn(sec) = secondary titanite.