

1 **Revision 1**

2 **Gamma-enhancement of reflected light images: a rapid, effective tool for**
3 **assessment of compositional heterogeneity in pyrite**

4 QIAOQIAO ZHU^{1,2}, NIGEL J. COOK², GUIQING XIE^{3,*}, CRISTIANA L. CIOBANU⁴, WEI JIAN¹,
5 BENJAMIN P. WADE⁵, AND JING XU^{4,6}

6 ¹ *MNR Key Laboratory of Metallogeny and Mineral Assessment, Institute of Mineral Resources, CAGS,*
7 *Beijing, 100037, China*

8 ² *School of Civil, Environmental and Mining Engineering, The University of Adelaide, Adelaide, SA*
9 *5005, Australia*

10 ³ *Institute of Earth Science, China University of Geosciences, Beijing, 100083, China*

11 ⁴ *School of Chemical Engineering and Advanced Materials, The University of Adelaide, Adelaide, SA*
12 *5005, Australia*

13 ⁵ *Adelaide Microscopy, The University of Adelaide, Adelaide, SA 5005, Australia*

14 ⁶ *Institute of Geology and Geophysics, Chinese Academy of Sciences, Beijing, 100029, China*

15 **ABSTRACT**

16 Trace/minor element variation in pyrite is a feature that has proved invaluable for reconstructing a wide
17 range of geological processes. Routine reflectance observations commonly fail to constrain this
18 variation due to the typically subtle and barely perceptible change in reflectance brought about by
19 deviation from ideal stoichiometry. Such differences may be difficult or impossible to observe in

* Corresponding author. E-mail address: xieguiqing@cags.ac.cn

20 conventional polished sections using standard optical microscopes, at least without oil immersion.
21 Chemical etching and staining, although widely used, are destructive and/or dangerous and the etching
22 process is not completely reproducible. Here we use the gamma correction method to enhance optical
23 digital signal differences obtained in reflected light in order to constrain compositional heterogeneity in
24 pyrite from a representative hydrothermal ore deposit in eastern China. The gamma-enhanced images
25 show significant reflectance variation caused by compositional heterogeneity, confirmed by
26 quantitative electron microprobe analysis and qualitative imaging. Higher reflectance domains in
27 gamma-enhanced images correspond to increases in the effective number of free electrons whereas
28 darker domains are attributed to the decrease of these free electrons by trace/minor element substitution
29 in pyrite (e.g., As). Gamma correction provides a rapid, effective, non-destructive method to constrain
30 compositional heterogeneity of pyrite through enhancement of reflectance variation. Used alone, this
31 method is unable to determine the chemical composition due to simultaneous substitutions, causing
32 disparate increase or decrease of reflectance, in most ore minerals. Nevertheless, gamma correction
33 may be sufficient to predict the substitution of trace/minor elements under the optical microscope prior
34 to scanning electron microscope imaging and quantitative investigation of mineral composition and
35 may help constrain links between textures and compositions of pyrite in evolving ore systems, which
36 could also be applied to other ore minerals with negligible bireflectance.

37 **Keywords:** Compositional homogeneity; reflectance; pyrite; non-destructive analysis; gamma
38 correction.

INTRODUCTION

39

40 Pyrite (FeS_2) is the most common sulfide mineral in the Earth's crust and occurs in various
41 environments ranging from low-temperature sedimentary rocks to high-temperature igneous rocks and
42 high-grade metamorphic terrains (Craig et al. 1998). It commonly incorporates impurities (such as As,
43 Co, Ni, Cu, and Tl) at concentrations ranging from trace levels to a few wt%. Pyrite has a refractory
44 character and can readily retain inherited textural and compositional features during subsequent
45 growth, making it a useful tool for ore petrologists. Applications include tracing changes of pore fluids
46 during progressive diagenesis, tracking physicochemical conditions and enrichment mechanism of
47 hydrothermal mineral systems, assessing trace element abundance in the oceans and atmospheric O_2
48 levels through geological time, as well as providing insights into the geochemical cycle of
49 environmentally hazardous metals (such as As, Cd, Hg, and Tl) during metamorphism and weathering
50 (Fleet et al. 1993; Craig et al. 1998; Chouinard et al. 2005; Deditius et al. 2008, 2016; Cook et al. 2013;
51 Large et al. 2014; Gregory et al. 2015, 2019; George et al. 2018, 2019). Access to a rapid, effective way
52 of estimating trace/minor element variation in pyrite is thus of fundamental importance for subsequent
53 research in the field.

54 Prior to the ready availability of scanning electron microscope or microprobe analysis, methods
55 widely used to constrain trace/minor element variation in pyrite include measurement of reflectance,
56 chemical etching and staining. Reflectance is one of the major diagnostic features of ore minerals in
57 reflected light (Ramdohr 1969; Picot et al. 1982; Criddle and Stanley 1993; Craig and Vaughan 1994).
58 Research has established that reflectance values and reflected color of pyrite may be highly sensitive to
59 variations in its composition even though the relationships between reflectance and impurities remain

60 unclear or inadequately constrained (Saager and Mihalik 1967; Vaughan 1969; Craig and Vaughan
61 1994). For example, the reflectance of pyrite was observed to decrease systematically with increasing
62 of Ni and/or Co concentrations (Vaughan 1969). The variation of reflectance under reflected light is
63 therefore expected to give clues to the substitution of trace/minor elements into pyrite, especially when
64 pyrite concerned display compositional or growth zoning at the grain-scale. However, reflectance
65 variation caused by compositional heterogeneity within pyrite is commonly subtle, and often invisible
66 in conventional polished sections viewed in reflected light, at least without oil immersion (Craig 2001).

67 The application of chemical etching in ore mineralogy can be traced back to ~1885 and is still
68 widely used for the textural characterization of minerals (Ramdohr 1969; Craig and Vaughan 1994;
69 Craig 2001; Criddle 1998; Gregory et al. 2015, 2019). Some internal properties of mineral grains
70 become visible under the optical microscope only through etching (Ramdohr 1969; Craig and Vaughan
71 1994). The different reaction rates between acid chemical etchants (e.g. HNO₃, HCl) and minerals
72 result in selectively removing some domains/phases to accentuate differences between domains/phases
73 that are otherwise near identical in optical properties (Craig and Vaughan 1994). In contrast, chemical
74 staining, refers to the coloration attributed to optical interference by a thin surface layer of elemental
75 sulfur generated during oxidation reactions between chemical etchants (such as H₂SO₄+KMO₄ and
76 NaOCl) and sulfide minerals (Fleet et al. 1993). Although both methods have proven useful to expose
77 compositional zoning, sub-grain boundaries and mineral inclusions in pyrite under reflected light
78 (Saager and Mihalik 1967; Ramdohr 1969; Fleet et al. 1993; Craig 2001; Gregory et al. 2019; Sykora et
79 al. 2019), they are destructive and/or dangerous and the etching process may not be completely
80 reproducible (Craig and Vaughan 1994; Tanner et al. 2016). Furthermore, some experimentations may

81 be required to find the optimum conditions for acid etching and staining since the reaction rate for a
82 certain chemical etchant may differ from mineral to mineral, or between domains within an individual
83 mineral ([Ramdohr 1969](#); [Fleet et al. 1993](#)).

84 Digital image processing technology, based on optical or scanning electron microscope (SEM)
85 images, represents an important advance over traditional techniques in the characterization of ore
86 minerals and underpins the technology used to carry out automatic quantitative mineral analysis and
87 textural discrimination ([Castroviejo et al. 2002](#); [Berrezueta et al. 2016](#); [Bachmann et al. 2017](#)). These
88 techniques, commonly applied to analysis of mineral processing materials, are powerful to analyze
89 large numbers of particles and generate a great variety of information on minerals (species, size, shape,
90 association, etc.), which would be impossible to measure accurately by manual methods
91 ([Pérez-Barnuevo et al. 2012](#); [Berrezueta et al. 2016](#)). Digital image processing also aims to make
92 images easier to interpret and to increase the signal-to-noise ratio for the visualization of specific
93 features with our unaided eyes ([Heilbronner and Barrett 2014](#); [Trauth 2015](#)). Despite this, direct
94 applications of digital image processing technology to constrain the compositional heterogeneity of ore
95 minerals remain scarce. Several authors have attempted to estimate ore grade (Au, In) based on image
96 processing and found that compositional heterogeneity might be a critical reason that led to large
97 discrepancies between estimate data and measured bulk content ([Castroviejo et al. 2002](#); [Bachmann et](#)
98 [al. 2017](#)). This suggests the compositional heterogeneity induced reflectance variation could provide a
99 major challenge for accurate mineral characterization in digital image processing ([López-Benito et al.](#)
100 [2017](#)).

101 Here we use data for pyrite from the Shizilishan Sr-(Pb-Zn) deposit, eastern China, to propose an

102 effective, non-destructive enhancement method to observe the compositional heterogeneity of the
103 pyrite. The methodology is based on gamma correction of the real-time optical digital signal obtained
104 on a petrographic microscope for mineral grains with compositional heterogeneity which can be
105 confirmed by electron probe microanalysis (EPMA). Using the proposed technique, the observer can
106 readily discriminate reflectance variations caused by the presence of trace/minor elements between
107 different domains and obtain high-magnification images highlighting textures comparable with, or
108 better than those obtained using scanning electron microscope (SEM) techniques. Indirectly,
109 application of this new method can also provide indications of future quantitative research avenues on
110 pyrite composition, as well as insights into ore-forming processes interpretable from ore mineral
111 textures, when complemented by additional microbeam techniques.

112 **BACKGROUND ON GAMMA CORRECTION**

113 Gamma correction, also called power-law transformation, is a widely used image enhancement
114 method which improves the interpretability or perception of information in images for human viewers
115 (e.g., [Gonzalez and Woods 2008](#); [Maini and Aggarwal 2010](#); [Heilbronner and Barrett 2014](#)). The
116 general form of the gamma correction is usually defined as:

$$117 \quad s = C \cdot r^\gamma,$$

118 where s and r are the gray values of output and input, respectively; C is a positive weighting constant
119 and used to limit the gray values of s between 0 and 255, and γ is the exponent defining the gamma
120 correction ([Heilbronner and Barrett 2014](#)). Both C and γ are built-in parameters of display equipment
121 or software with certain ranges ([Gonzalez and Woods 2008](#); [Maini and Aggarwal 2010](#)). These
122 power-law transformation functions are shown graphically in [Figure 1](#). It is obvious that the parameter

123 γ controls the contrast adjustment. The condition $0 < \gamma < 1$ produces overall contrast enhancement
124 resulting in a brightened image and may be applied to dark images to produce brightened images. This
125 condition is suitable for underexposed images. If $\gamma > 1$, there will be contrast stretching in such a way
126 that those gray values which are lower than the local mean will be stretched towards black (low gray
127 values) while others are stretched towards white (high gray values) resulting in a balanced image for
128 bright images (Vimal and Thiruvikraman 2012; Heilbronner and Barrett 2014). For example, if the
129 input gray values are chosen as 220 and 240, we expect output values of ~ 67 and ~ 159 , respectively,
130 when using $\gamma = 10$ and $C = 2.51 \times 10^{-22}$ in Figure 1. In this case, the difference of input gray values is
131 stretched from 8.33 % to 57.86 % when using higher values of γ , resulting in significant variation of
132 gray values which is sufficient to distinguish distinct domains that originally had similar gray values.

133

METHODOLOGY

134 Pyrite-bearing samples used in this study were collected from the Shizilishan Sr-(Pb-Zn) deposit,
135 Edong district, eastern China, which occurs at the contact between the ~ 139 Ma Shizilishan quartz
136 diorite porphyry stock and Lower to Middle Triassic limestone and dolomite, as well as interlayered
137 structures within the dolomite (Zhong 1992; Zhu et al. 2017). Lesser Pb-Zn mineralization is hosted by
138 the dolomite, limestone, and Middle Triassic sandstone. Samples containing pyrite are from breccias.
139 These breccias are composed of fragments of dolomite, celestite \pm sphalerite, and sandstone. The pyrite
140 occurs within veins crosscutting the breccia, or as nodules near the breccia margin, with or without an
141 association with calcite (Zhu et al. 2020).

142 Samples were prepared as 1-inch-diameter polished blocks and as polished thin sections. Optical
143 and SEM observation was performed at Adelaide Microscopy (The University of Adelaide). Detailed

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144 textural analysis on sulfides, sulfates and magnetite were performed under reflected light in air using a
145 Nikon Eclipse LV100 POL Petrographic Microscope. Favorable areas were chosen to obtain optical
146 digital images before and after gamma correction. Gamma correlations were refined using the Look-Up
147 Tables (LUTs) function of the Nikon NIS-Elements D software linked to the microscope when minerals
148 were observed at higher magnification (20X, 50X). Since ore minerals commonly have high reflectance
149 values relative to gangue phases and also display colored reflectance (Craig and Vaughan 1994), the
150 condition of $\gamma > 1$ is employed to stretch these high gray values to make different domains more
151 distinguishable in reflected light. A constant γ parameter of 10 was used in this study. The green
152 channel was chosen as the principal color to display the gamma-enhanced images. Detailed operating
153 procedures using Nikon NIS-Elements D and other types of software linked to commonly used optical
154 microscopes are described individually in [Supplementary¹ Appendix A](#). Following gamma
155 enhancement, the samples were coated with carbon and examined using an FEI Quanta 450 SEM
156 equipped with an energy-dispersive X-ray spectrometer to evaluate grain-scale compositional
157 heterogeneity.

158 Quantitative mineral compositional data were obtained using a Cameca SX-Five electron probe
159 microanalyzer. The spatial resolution was $\sim 1 \mu\text{m}$. Count-times were 40 s for unknowns and 8 s
160 background for all elements. X-ray lines and standards for pyrite were: S K α (pyrite), Pb M α (galena),
161 As L α (gallium arsenide), Se L α (Bi₂Se₃), Fe K α (pyrite), Cu K α (chalcopyrite), Zn K α (sphalerite), Co
162 K α (cobalt), Ag L α (silver telluride), Sb L α (stibnite), Ni K α (nickel olivine) and Au L α (gold). Average
163 minimum detection limits (mdl, wt%) were: S (0.02), Pb (0.08), As (0.05), Se (0.05), Fe (0.03), Cu

¹ Deposit item

164 (0.03), Zn (0.04), Co (0.03), Ag (0.05), Sb (0.04), Ni (0.03), Au (0.13). Qualitative data on
165 compositional zoning in pyrite was obtained using a Cameca SX-Five electron probe microanalyzer.
166 X-ray lines for pyrite were: As $L\alpha$, Co $K\alpha$, Ni $K\alpha$, Sb $L\alpha$, and Tl Ma . These elements were collected on
167 separate spectrometers at 20 kV and 300 nA. The spatial resolution was $\sim 0.5 \mu\text{m}$.

168 **RESULTS: PYRITE TEXTURES AND COMPOSITIONAL PATTERNS**

169 Complex intergrown sulfide textures, typical of hydrothermal ores with multi-stage evolution are
170 selected to illustrate how compositionally distinct types of pyrite can be distinguished using the optical
171 microscope. Four sub-types of pyrite (Py1, Py2, Py3, and Py4), and two sub-types of marcasite (Mrc1,
172 Mrc2), are recognized in Shizilishan ores based on characteristic habit, size and mutual relationships as
173 revealed by optical observation when using gamma correction ([Table 1](#)). Comparison of images before
174 and after gamma correction shows that it is far easier to constrain the textural relationships between
175 distinct sub-types of pyrite and marcasite if differences in their reflectance are enhanced. The sub-types
176 of interest here are the two pyrite sub-types (Py1 and Py3) that show contrasting grain-scale textures
177 interpreted as representing different formation stages. An intermediate stage, with marcasite
178 replacement of pyrite is also included for the purpose of comparison. Compositional data obtained from
179 the coarser pyrite generations (Py1 and Py3; [Table 2](#)) shows that Py3 is arsenian, i.e., As content in the
180 range 1.41–4.89 wt% (0.76–2.72 at%), whereas minor Co is measured in both Py1 and Py3. Otherwise,
181 the two pyrite generations are relatively ‘clean’ ([Table 2](#)). A comparison between the two types of
182 pyrite is shown on Fe vs. S and Fe vs. As plots ([Fig. 2](#)).

183 Pyrite 1 is clearly the earliest and has been replaced by fine-grained aggregates of marcasite (Mrc1;
184 [Fig. 3a](#)). Small grains of Py2 occur throughout such aggregates and likely represent relicts of Py1 rather

185 than a separate new growth. Coarser aggregates of marcasite (Mrc2) form along veinlets or larger
186 masses enclosing pre-existing Fe-sulfides. Relationships between these phases are clearly accentuated
187 on the gamma-corrected images (Fig. 3b) and are poorly reproduced on BSE images (Fig. 3c) at
188 comparable magnification. The compositional homogeneity of Py1 is well expressed on all type of
189 images but the presence of relict Py1 within the coarser marcasite is best depicted on the reflected light
190 image (Fig. 3a-c). This draws attention to the fact that presence of the two FeS₂ polymorphs, pyrite and
191 marcasite, one is weakly bireflectant (marcasite) and could hinder identification using
192 gamma-corrected images if the two minerals are intimately intergrown with one another.

193 In contrast, the relationships between pyrite displaying distinct textures represents an ideal case for
194 understanding the effects of gamma correction on reflected light images. Subhedral grains of Py1 occur
195 as preserved cores of larger aggregates of Py3 typified by fine oscillatory rhythms (colloform textures,
196 Fig. 3d-f). The oscillatory rhythms are enhanced on the gamma-corrected images (Fig. 3e) despite their
197 fine-scale, micron-to sub-micron rhythm widths, as resolved by BSE imaging (Fig. 3f). Moreover, the
198 gamma-enhanced image also shows the fibrous, sectorial zoned, growth within the wider, marginal
199 rhythms of the Py3 aggregate (Fig. 3e).

200 Development of more complex textures of Py3, such as coarser aggregates with oscillatory zoning
201 overgrowing or alternating with colloform textures, is barely depicted on the reflected light image but
202 are strongly enhanced on the gamma-corrected image, a feature replicated by chemical patterns on the
203 BSE image (Fig. 4a-c). There is an inverse contrast between As-rich zones on gamma-corrected images
204 and the compositional patterns on the BSE images. However, such contrast could be the cumulative
205 effect of various elements present as traces, as suggested by the distributions of Sb and Tl on EMPA

206 element maps (Fig. 4d-f).

207 Sectorially zoned domains recognized within Py3 also display distinct oscillatory zoning patterns
208 (Fig. 5). In one case, growth develops outside of a colloform core and sectorial zoning is only seen
209 within the coarser outer zones (Fig. 5a-c). The gamma-corrected image reveals all the details in the
210 growth textures and is mimicked by the compositional patterns seen on the corresponding BSE image
211 (Fig. 5b, c). The second case (Fig. 5d-f) illustrates crystal growth with oscillatory and sectorial zoning
212 and also displays a change in morphology from hexagonal to octagonal. The gamma-enhanced image in
213 this case is even clearer than the BSE image (Fig. 5e, f). Such textures, with transition between
214 different types of growth, are important for constraining distinct stages of mineralization against
215 morphological and textural variation of pyrite.

216 DISCUSSION

217 Impurity-induced reflectance variation in ore minerals

218 Optical properties of ore minerals are determined by crystal chemistry, that is, by both the crystal
219 structure (the arrangement of atoms making up the crystal) and by chemical bonding (Vaughan and
220 Wright 1998). Based on molecular orbital/band theory, the energy of visible light is overlapped by that
221 needed by electrons excited from the highest orbitals/valence band to empty orbitals/conduction band
222 of a semiconductor or metal. The reflected light is expected to re-emit when the unstable excited
223 electrons return to the ground state (Vaughan 1978; Craig and Vaughan 1994). Experiments and
224 quantum calculations have suggested that the reflectance of semiconductors or metals will increase
225 with the effective number of free electrons which can be excited (Burns and Vaughan 1970). For

226 example, the incorporation of minor/trace elements, such as Co, Ni, As and Cu, into pyrite has been
227 shown to cause a significant decrease in reflectance (Saager and Mihalik 1967; Brown and Bartholomé
228 1972; Pačevski et al. 2008). This correlation was further constrained by research on the electronic
229 structures of pyrite-type disulfides ($\text{FeS}_2\text{-CoS}_2\text{-NiS}_2\text{-CuS}_2$) and isochemical Pb-chalcogenides
230 (PbS-PbSe-PbTe) (Burns and Vaughan 1970; Vaughan 1978). Bither et al. (1968) noted that the
231 effective number of free electrons per molecule (n_{eff} , a number which reflects the electrons that can be
232 excited by visible light incident on the surface of minerals) of pyrite-type bisulfides decreased
233 systematically from FeS_2 to CuS_2 . This led Burns and Vaughan (1970) to constrain a roughly linear
234 correlation between N_{eff} , the effective number of free electrons per unit volume of pyrite-type bisulfides
235 (100 \AA^3), and their reflectance. The decrease of reflectance through isostructural pyrite-type
236 compounds was attributed to the increase of electrons in their eg^* orbitals, into which the t_{2g} electrons
237 can be excited, from 0 (FeS_2) to 3 (CuS_2), resulting in a decrease in the number of electrons that can be
238 excited to re-emit light when visible light is incident on their surface. The incorporation of impurities
239 into pyrite is therefore expected to change the effective number of free electrons on the mineral surface,
240 causing variation in selective absorption of the incident light, and resulting in an increase or decrease of
241 reflectance compared to the stoichiometric mineral (Craig and Vaughan 1994). It is also expected that
242 reflectance variation will be observed at the grain or sub-grain scale when the impurity elements are not
243 uniformly distributed in pyrite. This is due to a change of physical-chemical conditions at which
244 precipitation takes place, subsequent replacement, and/or superposition of a new generation of
245 hydrothermal mineralization (Vaughan 1969; Craig et al. 1998; Craig 2001; Putnis 2009; Cook et al.
246 2013, 2016; Ciobanu et al. 2019).

247 **Compositional heterogeneity of pyrite from the Shizilishan deposit**

248 The gray values of BSE images, reflecting the mean atomic mass in a given mineral are sensitive to
249 compositional variation (Paterson et al. 1989; Craig et al. 1998; Craig 2001). Therefore, changes in the
250 gray values on BSE images can be used as a reference for the relative enrichment or depletion of
251 trace/minor elements in pyrite. For example, the incorporation of minor As into pyrite increases the
252 mean atomic mass, resulting in increased brightness (higher gray values) for Py3 than Py1 in BSE
253 images of ores from the Shizilishan deposit, as well as As-rich and As-poor domains in Py3 (Figs. 3f,
254 4c, d, and 5c, f). Raman analysis had demonstrated that As in Py1 and Py3 was structurally bound,
255 causing the downshift of Raman band positions (Zhu et al. 2020).

256 According to the molecular orbital model, reflectance is expected to increase since the covalent
257 character of As is stronger than S and the substitution of S by As in pyrite would delocalize electrons,
258 resulting in increasing N_{eff} and reflectance. However, the gamma-enhanced images of pyrite
259 demonstrate that the reflectance variation trend contrasts with the change to gray values on the BSE
260 images (Figs. 3e, f, 4b, c, 5b, c, e, f), suggesting that As-rich domains show lower reflectance than
261 As-poor domains in this study. Incorporation of Co and Ni is expected to decrease the reflectance of
262 pyrite since the CoS_2 and NiS_2 has lower N_{eff} than FeS_2 (Burns and Vaughan 1970; Criddle and Stanley
263 1993). However, it may not be the main reason in our study case as the concentrations of Co (0.04–0.12
264 wt%) and Ni (mdl–0.05 wt%), as well as Sb and Tl are too low to cause such significant reflectance
265 differences (Table 2; Fig. 4d-f). The decrease in reflectance is more likely attributable to the
266 substitution of S by As that not only reduces the S content but also decreases the Fe content (from a
267 mean 33.30 at% in Py1 to a mean 32.73 at% in Py3, Fig. 2), or As substituting for Fe as As^{3+} in Py3

268 (Deditius et al. 2008). Both changes may be expected to reduce the total number of electrons that could
269 be excited to produce reflected light (Burns and Vaughan 1970; Vaughan 1978). This is consistent with
270 previous observations that As, Ni, and Co-bearing pyrite, from the Free State Geduld Mine, South
271 Africa, has a reflectance of 51.1 % at 590 nm in air, whereas relatively clean pyrite has higher
272 reflectance of 53.2 % (Saager and Mihálik 1967). The gamma-enhanced image also demonstrates that
273 some Py₃ grains display a change in growth habit from concentric to sector zoning (Fig. 5b, e),
274 suggesting crystallographic control on trace element incorporation, similar to the concentrically and/or
275 sectorial-zoned auriferous pyrite reported from the high-sulfidation Au-Ag-Cu Pascua deposit,
276 Chile-Argentina (Chouinard et al. 2005). Although reports have shown pyrite can exhibit weak
277 bireflectance, differences in reflectance are <0.2% (Bowie 1962). The variation of reflectance of pyrite
278 caused by impurities would thus dominate over any bireflectance, if present at all.

279 **IMPLICATIONS FOR MINERAL CHARACTERIZATION**

280 Compositional heterogeneity in the pyrite addressed above is expected to be expressed as reflectance
281 variations although most are too weak to be observed under reflected light (Figs. 3d, 4a, 5a, d). Gamma
282 correction of optical digital signals offers a rapid and effective method to reveal subtle compositional
283 heterogeneity (Figs. 2-5). The higher reflectance domains in gamma-enhanced images correspond to an
284 increase of effective number of free electrons whereas darker domains are attributed to relative
285 decrease of these free electrons, compared to stoichiometric pyrite. However, variation in the relative
286 reflectance will differ from one element to another and can be either systematically increased or
287 decreased (as in the case of As, for example), with an increase in mean atomic mass due to minor
288 element incorporation as indicated by BSE images. This method is, if used alone, unable to determine

289 chemical compositions of unknown samples, mainly due to simultaneous substitutions which cause
290 divergent changes in reflectance (Criddle and Stanley 1993; López-Benito et al. 2017). However, a
291 comparison of different reflectance domains under the optical microscope is generally sufficient to
292 predict the compositional heterogeneity or provide indications about trace/minor element substitution
293 that can be followed up by quantitative analysis (Charlat and Lévy 1976). Given that incorporation of
294 trace elements in pyrite is almost certainly crystallographically controlled (Fig. 5b, c, e, f; Chouinard et
295 al. 2005), gamma correction represents an alternative way to observe variations in crystal habit, which
296 are, in turn, believed to reflect different or evolving conditions of formation (Murowchick and Barnes
297 1987; Alonso-Azcárate et al. 2001).

298 The gamma-enhanced images demonstrate clear textural evidence at the grain-scale. These are
299 comparable to, or even clearer than the BSE images, indicating that they can also be used as additional
300 evidence for the variation of texture, which commonly preserved the origin and growth history of pyrite
301 (Craig 2001). Additionally, gamma correction is a non-destructive method, thus giving it clear
302 advantages over chemical etching and staining, which are widely used in texture characterization but
303 are destructive and/or dangerous, and may be difficult to reproduce (e.g., Fleet et al. 1993; Craig and
304 Vaughan 1994; Craig 2001; Gregory et al. 2015, 2019; Sykora et al. 2019). Other ore minerals with
305 negligible bireflectance, including magnetite, wolframite, and tetrahedrite, commonly display
306 compositional heterogeneity at the grain-scale (Charlat and Lévy 1976; Craig and Vaughan 1994;
307 Pačev ki et al. 2007; Ciobanu et al. 2019). Analogous gamma-enhanced imaging might be used to
308 diagnose and interpret such features.

309

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319

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433 **Figure captions**

434 **Fig. 1.** Plot of gamma correction result for various γ using different values of constant C to scale each curve to
435 fit the range shown, where s and r are the gray values of output and input, respectively. Figure modified after
436 [Gonzalez and Woods \(2008\)](#) and [Vimal and Thiruvikraman \(2012\)](#).

437 **Fig. 2.** Binary plots of (a) Fe vs. S, (b) Fe vs. As of EMPA data in Py1 and Py3, expressed in atomic percent
438 (at%).

439 **Fig. 3.** Reflected light (a and d), gamma-enhanced (b and e) and BSE images (c and f) showing textural
440 relationship between generations of pyrite. (a–c) Euhedral Py1 included within fine-grained, spongy
441 aggregates of Py2+Mrc1 which are crosscut by coarser, vein-like Mrc2. In reflected light, Py1 shows a
442 brownish color when surrounded by aggregates of Py2+Mrc1. (d–f) Py3 observed as colloform overgrowths
443 on Py1 which is also crosscut by galena+Py3. Boundaries between sub-types of pyrite and marcasite are
444 marked by white dashed lines in (b) and (e). Abbreviation: Mrc–marcasite; Py–pyrite; Gn–galena.

445 **Fig. 4.** Shift in the growth habit and compositional zonation of Py3. Euhedral grains of Py1 preserved as cores
446 within Py3 featuring fine oscillatory rhythmic zoning which gradually changes to growth zoning combined
447 with sectorial zoning. The shift is clearly demonstrated in the gamma-enhanced (b) and BSE image (c) but is
448 only faintly seen under reflected light image (a). Zoning in Py3 is also revealed by variation in the contents of
449 As (d), Sb (e) and Tl (f). Boundary between Py1 and Py3 marked by white dashed lines in the
450 gamma-enhanced image (b). Abbreviation: Py–pyrite.

451 **Fig. 5.** Growth habit of Py3 changes from an oscillatory-zoned core to a sectorial-zoned rim. The sector zoning
452 displayed in the gamma-enhanced image (b, e) may actually be clearer compared to the BSE image (c, f). a,
453 d–Reflected light image. Abbreviation: Py–pyrite.

Table 1. Summary of features of pyrite and marcasite from the Shizilishan Sr-(Pb-Zn) deposit.

Sub-types	Py1	Mrc1	Py2	Mrc2	Py3	Py4
Reflected color	yellow-white, appears brownish when surrounded by Mrc2 or Mrc1+Py2	greenish-gray to weak brown-white	yellow-white but slightly whiter than Py1	greenish and bluish grays to brown-white	yellow-white to pinkish-grey	yellow-white
Crystal size	10~50 μm	<2 μm	<2 μm	5~35 μm	20~>100 μm	2~5 μm
Growth habit	euhedral, discrete grains, some grains displays concentric, porous core and gradually changes to euhedral surface	aggregates to form spongy texture	aggregates to form spongy texture	anhedral or lath-like crystals	anhedral to euhedral morphology, changes from colloform core to growth zones in rim or from oscillatory-zoned core to sectorial-zoned rim	anhedral aggregates to form veins
Intergrowth relationships	overgrown by later pyrite and marcasite	replacing and overgrowing Py1	likely relicts of Py1 replaced by Mrc1	crosscut or overgrown on Py2+Mrc1 aggregates	overgrown on Py1, Py2+Mrc1 and Mrc2; crosscut Py1 sometimes	crosscut Py3 or cemented broken Py3 crystal

Notes: Py–pyrite; Mrc–marcasite.

Table 2. Electron microprobe data for pyrite (wt%) and minimum detection limits.

Sample	Minimum detection limit	SL2-1 B2-1						SL14-1 C3-1-5									SL14-1 C3-1-7			
		1	2	12	13	14	17	1	2	3	4	5	7	8	8-1	9	6	7	8	
Sub-type		Py1	Py1	Py1	Py1	Py1	Py1	Py1	Py1	Py1	Py1	Py3	Py3	Py3	Py3	Py3	Py3	Py3	Py3	Py3
S	0.02	53.28	53.04	53.01	52.11	53.03	52.83	53.60	53.30	52.46	50.81	50.25	50.48	52.04	51.68	51.13	50.00	51.81	52.09	
As	0.05	<	0.07	<	<	0.06	<	0.12	0.07	0.45	3.90	4.25	4.65	1.77	1.93	3.03	4.89	1.62	1.41	
Se	0.05	<	<	<	<	<	<	<	<	<	<	<	<	<	<	0.05	0.07	<	<	
Fe	0.03	46.32	46.28	46.21	45.86	46.51	46.44	46.13	46.18	45.71	43.94	43.62	43.43	45.37	45.31	44.55	43.38	45.32	45.69	
Cu	0.03	<	<	<	0.04	<	<	<	<	<	<	<	<	<	<	<	<	<	<	
Co	0.03	0.04	0.12	0.06	0.06	0.04	0.10	0.07	0.07	0.06	0.07	0.04	0.07	0.07	0.04	0.08	0.07	0.06	0.08	
Sb	0.04	<	<	<	<	<	<	<	<	<	<	<	<	<	<	<	0.04	<	<	
Ni	0.03	<	<	<	0.05	<	<	<	0.04	<	<	<	<	<	<	<	<	<	<	
Total		99.80	99.61	99.35	98.20	99.68	99.54	99.99	99.71	98.70	98.85	98.25	98.74	99.26	99.05	98.88	98.51	98.91	99.34	
Calculated atomic percent (at%)																				
S		66.64	66.52	66.59	66.34	66.46	66.38	66.82	66.69	66.45	65.33	65.12	65.14	65.97	65.78	65.47	64.86	65.92	65.94	
As		<	0.04	<	<	0.03	<	0.07	0.04	0.25	2.14	2.35	2.57	0.96	1.05	1.66	2.72	0.88	0.76	
Se		<	<	<	<	<	<	<	<	<	<	<	<	<	<	0.03	0.04	<	<	
Fe		33.26	33.32	33.33	33.51	33.47	33.50	33.02	33.17	33.24	32.44	32.46	32.18	33.01	33.11	32.76	32.30	33.11	33.21	
Cu		<	<	<	0.02	<	<	<	<	<	<	<	<	<	<	<	<	<	<	
Co		0.03	0.08	0.04	0.04	0.03	0.07	0.05	0.05	0.04	0.05	0.03	0.05	0.05	0.03	0.05	0.05	0.04	0.06	
Sb		<	<	<	<	<	<	<	<	<	<	<	<	<	<	<	0.01	<	<	
Ni		<	<	<	0.03	<	<	<	0.03	<	<	<	<	<	<	<	<	<	<	

Notes: Py—pyrite; <—below minimum limit of detection; All data for Pb, Zn, Ag, and Au are below minimum detection limit and are thus not show here.

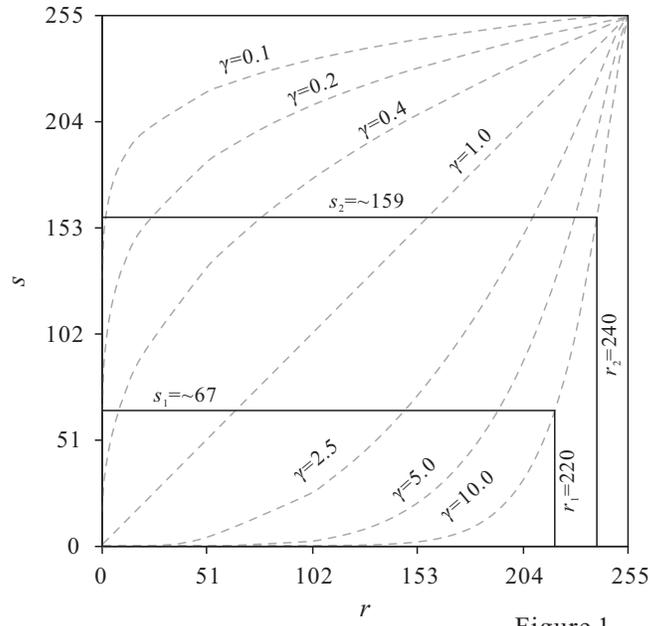


Figure 1

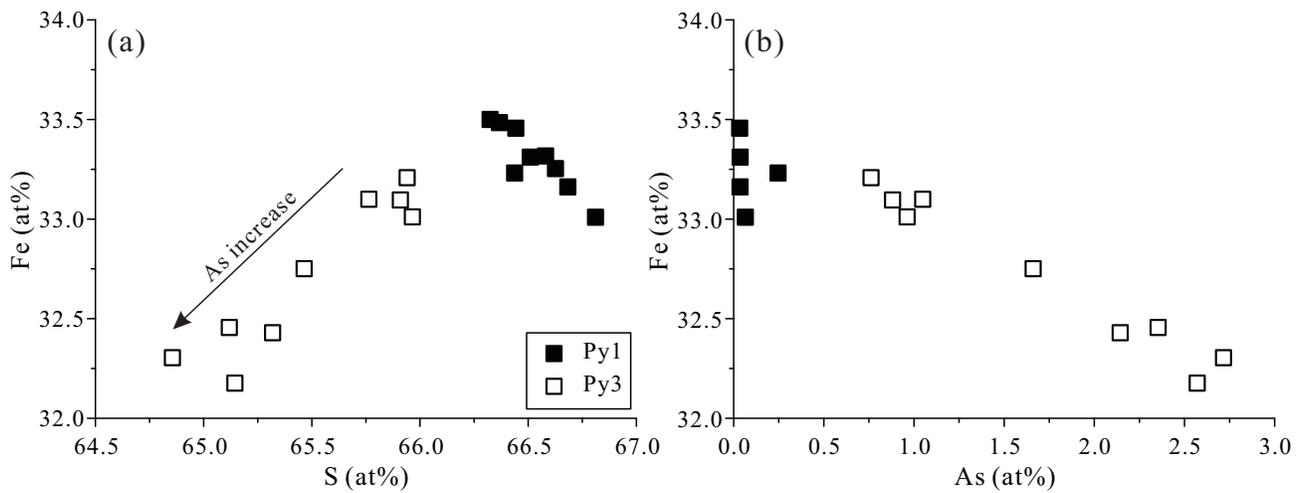


Figure 2

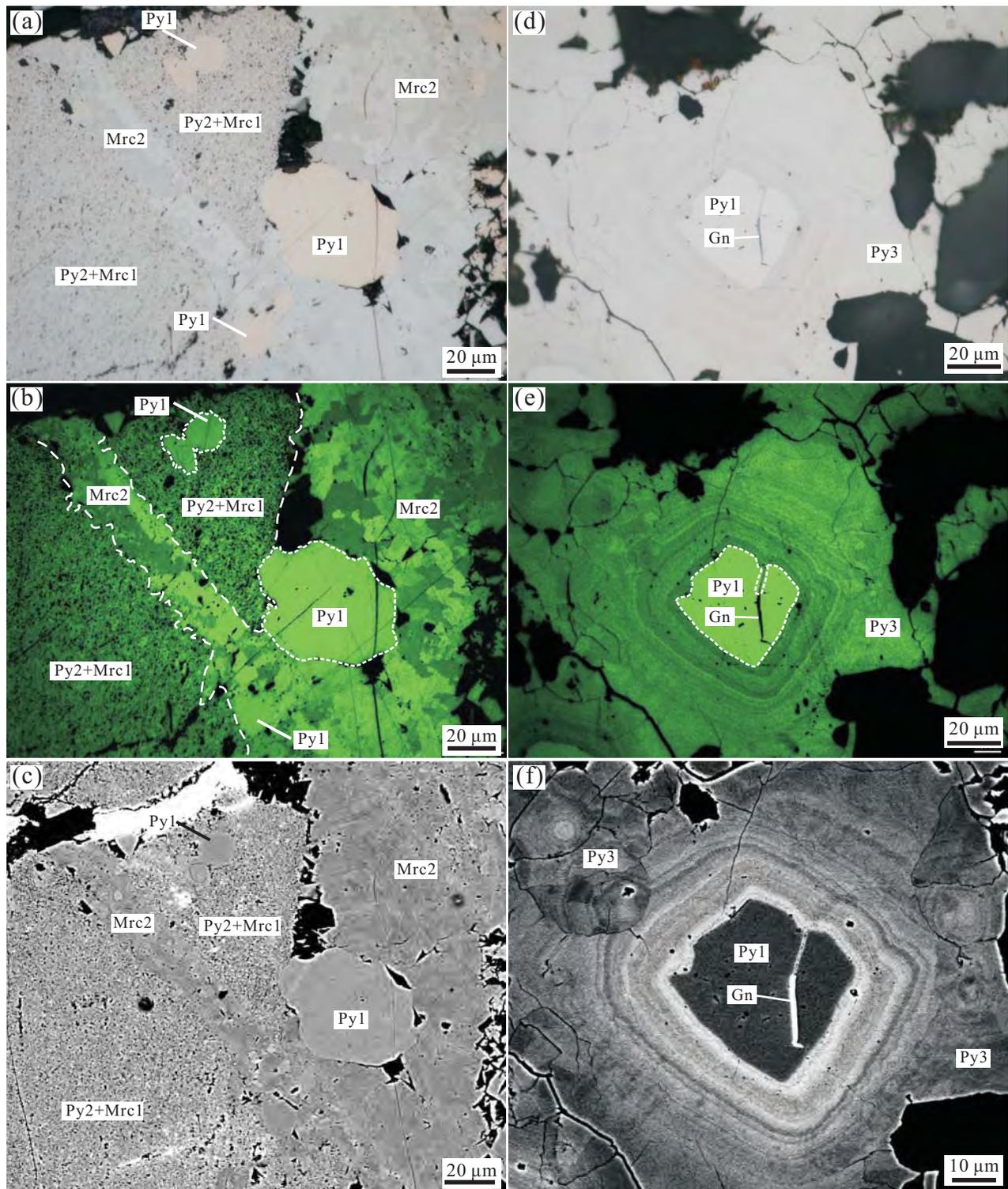


Figure 3

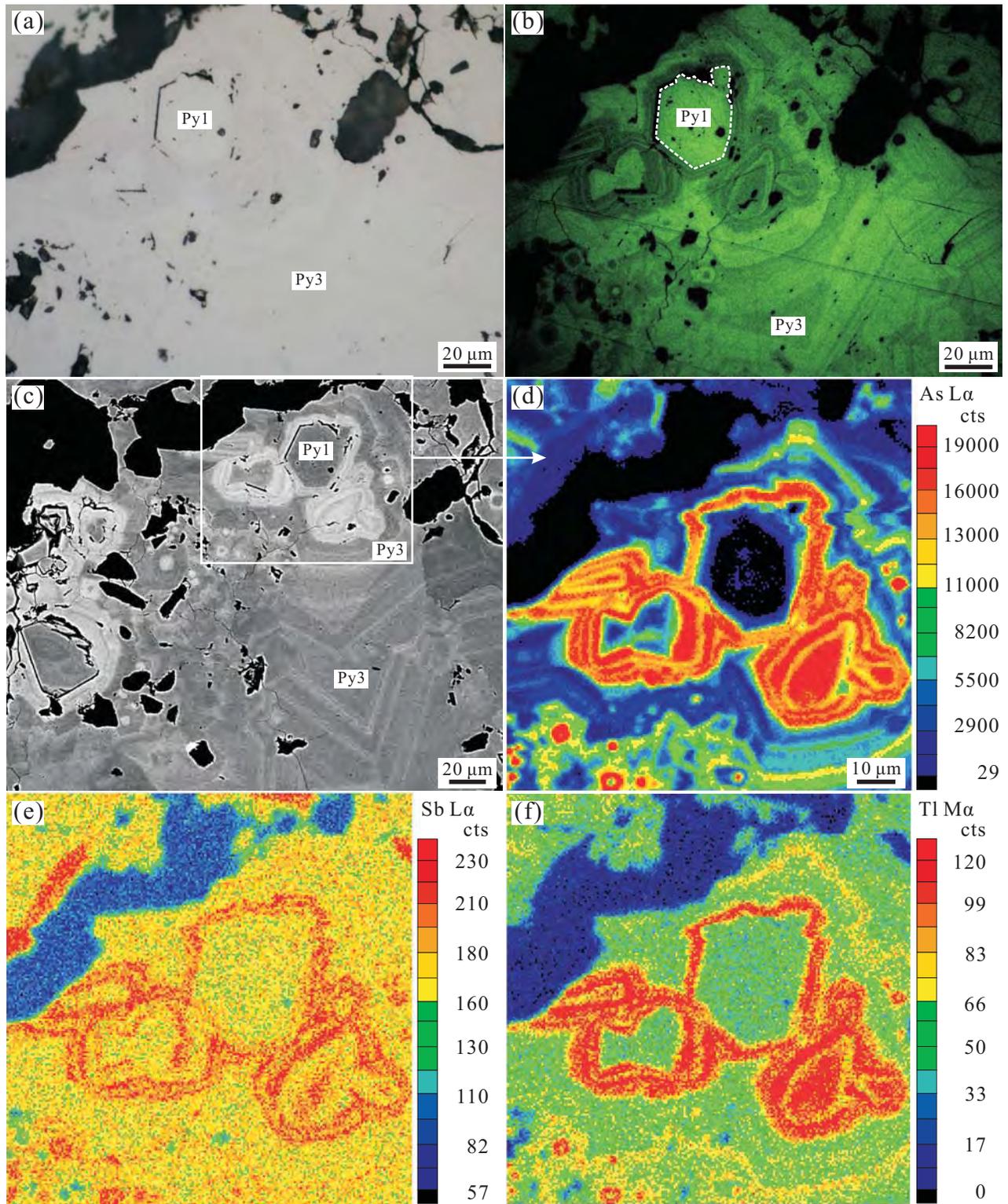


Figure 4

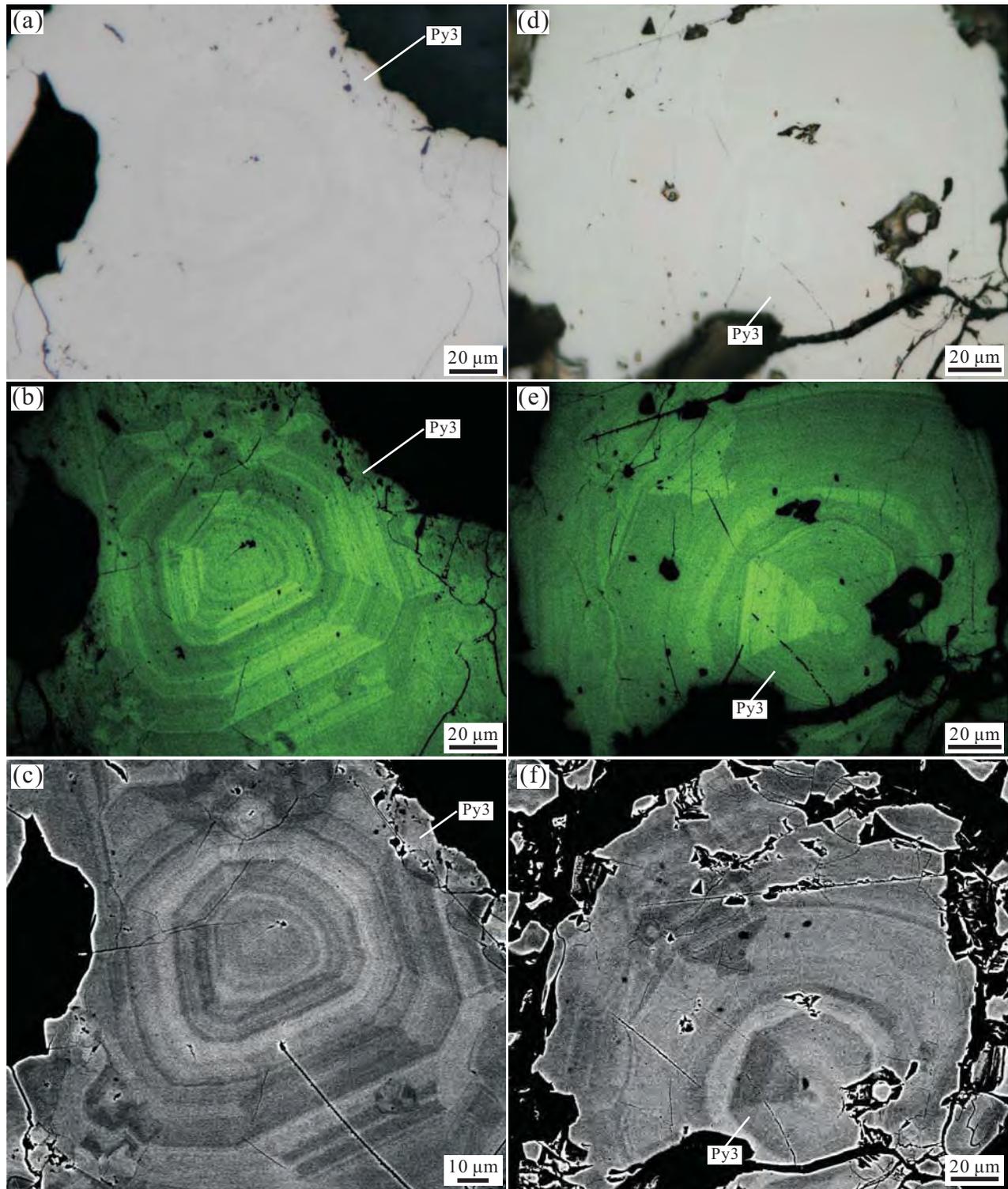


Figure 5