

1 **Revision 2**

2 **Metamorphic amphiboles in the Ironwood Iron-Formation, Gogebic Iron Range,**  
3 **Wisconsin: Implications for potential resource development**

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8  
9 **Abstract**

10 The Paleoproterozoic Ironwood Iron-Formation, a Superior-type banded iron formation  
11 located in the western Gogebic Iron Range in Wisconsin, is one of the largest undeveloped iron  
12 ore resources in the United States. Interest in the development of this resource is complicated by  
13 potential environmental and health effects related to the presence of amphibole minerals in the  
14 Ironwood, a consequence of Mesoproterozoic contact metamorphism. The presence of these  
15 amphiboles and their contact metamorphic origin have long been recognized; however, recent  
16 interest in this resource has highlighted the lack of detailed knowledge on their distribution,  
17 mineral chemistry, and morphology. Optical microscopy, X-ray diffraction, scanning electron  
18 microscopy, and electron microprobe analysis were utilized to investigate the origin, distribution,  
19 morphology, and chemistry of amphiboles in the Ironwood.

20 Amphibole is present in the western portion of the study area due to regional-scale  
21 contact metamorphism associated with the intrusion of the 1.1 Ga Mellen Intrusive Complex.

22 Locally amphibole is also present adjacent to diabase and/or gabbro dikes and sills in the lower-  
23 grade Ironwood in the eastern portion of the study area. In both localities, amphiboles in the  
24 Ironwood most commonly developed in massive and prismatic habits, and locally assumed a  
25 fibrous habit. Fibrous amphiboles were recognized locally in the two potential ore zones of the  
26 Ironwood but were not observed in the portion likely to be waste rock. Massive and prismatic  
27 amphiboles show a wide range of Mg# (molar  $\text{Mg}/(\text{Mg}+\text{Fe}^{2+})$ ) values (0.06 to 0.87), whereas  
28 Mg# values of fibrous amphiboles are restricted from 0.14 to 0.35. Factors that influenced the  
29 compositional variability of amphiboles in the Ironwood may have included temperature of  
30 formation, morphology, bulk chemistry of the iron formation, and variations in prograde and  
31 retrograde metamorphism. The presence of amphiboles in the Ironwood is a known issue that  
32 will need to be factored into any future mine plans. This study provides an objective assessment  
33 of the distribution and character of amphiboles in the Ironwood to aid all decision-makers in any  
34 future resource development scenarios.

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## Introduction

37 Metamorphism of banded iron formations (BIFs) – the most important source of iron ore  
38 globally – commonly results in the formation of Fe-rich amphiboles in the ores at appropriate  
39 metamorphic grades. Amphiboles are found in metamorphosed BIFs throughout the world  
40 (Klein, 2005). Various terms have been used to describe the morphology of amphibole particles,  
41 such as massive, equant, prismatic, fibrous, and asbestiform. The definitions for these terms can  
42 be found below in the Methods section. The term “elongate mineral particle” (EMP) has been  
43 used by the regulatory community for a variety of mineral habits including prismatic, fibrous,

44 and asbestiform particles, as well as cleavage fragments (NIOSH, 2011). Because of the links  
45 between *commercial asbestiform* amphibole minerals, mesothelioma, and lung cancer (Gibbs and  
46 Berry, 2008; Berry and Gibbs, 2008), concerns about potential human-health risks due to  
47 exposure non-asbestiform Fe amphiboles during mining and milling of metamorphosed BIF ores  
48 (taconite) have persisted. Interest in potential links between fibrous amphiboles and human  
49 health in the Lake Superior region began in earnest in the early 1970s related to airborne  
50 emissions around a taconite mill on the shore of Lake Superior and the since-banned practice of  
51 disposing of mill tailings at the bottom of the lake, which resulted in amphibole fibers appearing  
52 in nearby municipal drinking water supplies (Berndt and Brice, 2008). This incident has  
53 prompted over four and a half decades of scientific research exploring links between human  
54 health and amphibole particles through a variety of pathways related to the mining and milling of  
55 taconite ores. Allen et al. (2014) found increased risks for mortality from lung cancer,  
56 mesothelioma, and some cardiovascular disease among taconite miners, but could not exclude  
57 non-occupational behaviors, such as smoking, as a contributor. Brunner et al. (2008) found that  
58 most (14 out of 15 with adequate work histories) of the mesothelioma cases found in a case study  
59 of northern Minnesota iron miners had identifiable source exposures to commercial asbestos in  
60 jobs held both inside and outside of the iron mining industry. Other studies investigating the  
61 incidence of lung cancer in Minnesota taconite miners found that miners have shown no  
62 increased risk of mortality due to lung cancer (Berry and Gibbs, 2008; Allen et al., 2015a,b).  
63 Further, a study comparing the incidence of mesothelioma and lung disease in the western  
64 (hematite-rich, non-amphibole-bearing) and eastern (non-asbestiform, amphibole-bearing)  
65 Mesabi Iron Range in northern Minnesota did not find a higher incidence of disease in the  
66 amphibole-bearing portion of the mining district (Mandel and Odo, 2018). Only a few studies

67 have investigated the role of asbestiform and non-asbestiform amphiboles in relation to lung  
68 cancer and mesothelioma. In laboratory studies, Mossman (2008) found that (non-asbestiform)  
69 cleavage fragments are less bioreactive and cytotoxic than asbestiform amphiboles. Gamble and  
70 Gibbs (2008), on the basis of a review of cohort studies, concluded that non-asbestiform  
71 amphiboles do not increase the risk of lung cancer or mesothelioma. In other words, for naturally  
72 occurring amphiboles encountered in taconite mining, a link between non-asbestiform cleavage  
73 fragments and disease has not been established, and a link between naturally occurring  
74 asbestiform amphibole and disease is unclear. Thus, the long history of controversy surrounding  
75 iron mining, amphibole minerals, and human health in northern Minnesota, and the apparent  
76 human-health significance of cleavage fragments of amphibole versus asbestiform amphiboles  
77 highlight the importance of an increased understanding of the nature and distribution of  
78 amphiboles in the nearby Gogebic Iron Range of northern Wisconsin from the perspective of  
79 informed decision-making related to future potential development of these resources.

80         The geological conditions that resulted in the formation of cummingtonite-grunerite in  
81 the Biwabik Iron Formation in Minnesota are analogous to those in the Ironwood Iron-Formation  
82 in Wisconsin. At both localities, the emplacement of a mafic intrusive body truncated the strata  
83 and caused contact metamorphism of the iron formation. Due to the similarity in geological  
84 conditions and the presence of cummingtonite-grunerite in the Ironwood Iron-Formation, many  
85 of the issues faced in Minnesota will be important when considering the possibility of mining in  
86 the Gogebic Iron Range.

87         Past studies have described the geology, petrography, and nature of metamorphism in the  
88 Ironwood (Laybourn, 1979; Cannon et al., 2007). The present study focuses on providing  
89 detailed mineralogical information about the origin, distribution, morphology, and chemistry of

90 amphiboles in the Ironwood. In light of the historical issues associated with taconite mining in  
91 Minnesota and the complex problems posed by amphiboles with respect to human-health risk  
92 evaluation, the presence of amphiboles in the Ironwood warrants detailed characterization. The  
93 purpose of this study is to provide a mineralogical framework for the amphiboles that occur in  
94 the Ironwood to aid regulatory, medical, and mining entities in their evaluation of this potential  
95 resource. This study is based on a systematic sampling of the Ironwood from drill holes covering  
96 35 km of strike length from its contact with the Mellen Intrusive Complex (MIC) in the west  
97 toward lower metamorphic grade in the east. Additionally, surface outcrops were sampled where  
98 available. Optical microscopy, X-ray diffraction, scanning electron microscopy, and electron  
99 microprobe analysis were used to characterize the amphiboles in the Ironwood Iron-Formation.

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101

### **Setting**

102 The Gogebic Iron Range historically has produced significant amounts of “natural”  
103 (supergene) iron ore but has yet to produce any “taconite” (ore that requires concentration of  
104 magnetite prior to smelting, otherwise known as “concentration-grade ore”), which is the focus  
105 of potential future development in the region. From 1877 to 1967, the Gogebic Iron Range  
106 produced approximately 325 million tons of “natural” ores (Cannon et al., 2007). These natural  
107 ores comprise nearly pure iron oxides and hydroxides that formed as secondary concentrations  
108 resulting from the structurally focused flow of deeply circulating, oxygenated groundwater that  
109 both oxidized the original iron minerals and replaced the chert with iron minerals (Cannon et al.,  
110 2007; Irving and Van Hise, 1892). The natural ores produced from the Gogebic are restricted to  
111 areas of very low-grade metamorphism, as even modest metamorphic recrystallization inhibited  
112 their formation (James, 1955; Cannon et al., 2007). Although concentration-grade ore has not

113 been mined from the Gogebic, the western portion of the Ironwood constitutes one of the largest  
114 undeveloped iron resources in the United States. The study area, a 35-km east-west trending  
115 portion of the Ironwood (Figure 1), has been estimated to contain approximately 3.7 billion tons  
116 of taconite (Mardsen, 1978; Cannon et al., 2007).

117         The 1.87 Ga Ironwood Iron-Formation is a classic BIF in the Lake Superior Region that  
118 formed in a shallow sea resulting from extension and subsidence of the Superior craton  
119 associated with the approximately 1.88-1.84 Ga Penokean Orogeny (Cannon et al., 2007; Schulz  
120 and Cannon, 2007). The Ironwood was divided by Hotchkiss (1919) into five members: from the  
121 base upward, these are the Plymouth, Yale, Norrie, Pence, and Anvil Members. The Anvil  
122 Member is only found in the eastern portion of the range, outside of the study area. The  
123 Ironwood conformably overlies the Palms Formation, an argillite unit that transitions upwards to  
124 quartzite, which was deposited in a tidal environment and grades upward over several meters  
125 into the Ironwood, marking the transition from clastic to chemical sedimentation (Ojakangas,  
126 1983; Cannon et al., 2007). The Ironwood is overlain by the Tyler Formation in the study area, a  
127 turbiditic unit primarily composed of black shale and greywacke. The Ironwood was intruded by  
128 diabase and gabbro dikes and sills of Paleoproterozoic and/or Mesoproterozoic age (Cannon et  
129 al., 2007). Several large gabbroic sills are concordant with the Ironwood, whereas numerous  
130 diabase dikes cut the Ironwood at high angles, forming both northeast- and northwest-trending  
131 sets. These mafic intrusions were commonly exposed in mine workings in the formerly active  
132 portions of the central Gogebic Iron Range, where they played an important role in controlling  
133 the location and distribution of natural iron ore bodies (Cannon et al., 2007).

134         In the western portion of the study area, the Ironwood is truncated by Mesoproterozoic  
135 gabbro of the Mellen Intrusive Complex. The Mellen Intrusive Complex was emplaced roughly

136 parallel with the Ironwood and approximately coeval with northward tilting of the region during  
137 events related to the Midcontinent Rift at about 1.1 Ga; these units dip northward at  
138 approximately 70 degrees on average (Cannon et al., 1993; 1996; 2007). The primary effect of  
139 this intrusion was broad contact metamorphism, with metamorphic grade being most intense in  
140 the west, and diminishing eastward as the distance between the Ironwood and the Mellen  
141 Intrusive Complex increases (Cannon et al., 2007). Laybourn (1979) divided the Ironwood into  
142 four metamorphic zones. Zone 1 experienced diagenetic alteration/low-grade metamorphism and  
143 is defined by the presence of the Fe-phyllsilicates, minnesotaite and stilpnomelane, and Fe-  
144 bearing carbonates such as, siderite, dolomite, and ankerite. Zone 2 was characterized by  
145 medium-grade metamorphic conditions that resulted in the disappearance of Fe-phyllsilicates  
146 and Fe-bearing carbonates and the appearance of amphibole. Zone 3 underwent high-grade  
147 metamorphism and is defined by the presence of pyroxenes that developed from amphiboles.  
148 Zone 4 represents the highest grade of metamorphism where the development of fayalite is  
149 observed and occurs within 100 meters of the intrusive contact (Laybourn, 1979). Retrograde  
150 amphibole is also observed in Zones 3 and 4. These zones are similar to the progressive contact  
151 metamorphism documented in the Biwabik Iron-Formation in Minnesota due to the intrusion of  
152 the Duluth Complex (Gundersen and Schwartz, 1962; French, 1968).

153

154

## Methods

155 Sample collection of the Ironwood Iron-Formation encompassed several outcrop  
156 locations and four diamond drill cores ranging from 0 to approximately 5 km from the contact  
157 with the Mellen Intrusive Complex (Figure 1). Each drill hole was collared in the Pence Member

158 and ended in the upper quartzite unit of the Palms Formation, thus providing between 171 and  
159 237 m of continuous core sampling. Drill hole MP-67-1 was directionally drilled at an unknown  
160 azimuth at angles varying from approximately 60° to 80°, drill hole W-156 was drilled at an  
161 azimuth of 180° and an angle of 40°, drill hole 666-1 was drilled at an azimuth of 180° and a 45°  
162 angle, and drill hole 186C was drilled at an azimuth of 156° and an angle of 45°. Two types of  
163 samples were selected from drill cores: a set from each lithologic sub-unit intended to be  
164 representative of the section, and a topical set that could potentially yield insights about the  
165 paragenesis of amphiboles. One hundred fifteen samples were collected and made into polished  
166 thin sections, which were examined using transmitted and reflected light microscopy. Areas of  
167 interest were then examined using a Hitachi SU5000 field emission scanning electron  
168 microscope equipped with an EDAX Octane Plus 30 mm<sup>2</sup> silicon drift detector for energy  
169 dispersive spectroscopy (SEM/EDS). EDS was used to confirm mineralogy. Using SEM  
170 imagery, amphibole morphology was characterized using a standardized grid designed to limit  
171 selective bias (Green, 2017) to provide morphological and geometric data.

172         Amphiboles from 17 representative samples were categorized according to  
173 morphological definitions adapted from Campbell et al. (1977) including: massive, equant,  
174 prismatic, and fibrous. Massive amphibole particles are tightly packed with scarcely  
175 distinguishable grain boundaries and are arranged in a homogenous structure. Equant particles  
176 are those with three approximately equally spaced dimensions, whereas prismatic particles are  
177 those with one elongate dimension and two approximately equal shorter dimensions. Fibrous  
178 particles are those with high aspect ratios, often displaying curvature and occurring in bundles.  
179 The minimum aspect ratio necessary to apply the term fibrous is generally 3:1 according to most  
180 regulatory bodies, such as the International Organization for Standardization and the National



181 Institute for Occupational Safety and Health (International Organization for Standardization,  
182 1995; NIOSH, 2011), but this is often constrained by a maximum particle width of 3  $\mu\text{m}$   
183 (Lowers and Meeker, 2002). A drawback of observing amphibole particles in thin section is the  
184 inability to determine whether particles possess the properties of flexibility or high tensile  
185 strength. Both properties are inherent to the definition of asbestiform minerals, which are a  
186 subset of fibrous minerals that may not otherwise exhibit these features (Campbell et al., 1977).  
187 Having no means of demonstrating the properties of flexibility or high tensile strength, it is  
188 inappropriate to describe the amphiboles identified in this study as asbestiform based on the  
189 information currently available.

190 A total of 437 amphibole particles of prismatic, equant, and fibrous morphologies was  
191 measured using the software program QuartzPCI to determine geometric parameters. Massive  
192 particles were not measured because by definition, their boundaries were indistinct. Fibrous  
193 amphiboles were identified using ISO 10312 criteria for phase-contrast microscopy equivalent  
194 (PCME) fibers (length  $> 5 \mu\text{m}$ , width between 0.25 and 3  $\mu\text{m}$ , and aspect ratio  $> 3:1$ ) which are  
195 the basis for most health studies related to cancers in humans caused by asbestos exposure  
196 (International Organization for Standardization, 1995; U.S. Environmental Protection Agency,  
197 2005).

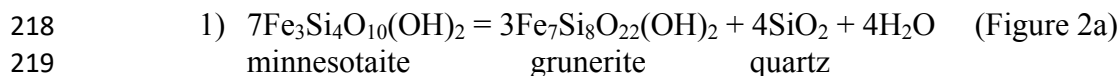
198 A PANalytical X'Pert PRO diffractometer, X-ray diffraction (XRD) was used to  
199 determine the presence of amphiboles and modal mineral abundances, aided by Rietveld  
200 refinement using HighScore Plus v.4.5. A JEOL JXA-8900 electron microprobe analyzer  
201 utilizing wavelength-dispersive spectroscopy (EMPA-WDS) was employed to determine the  
202 chemistry of various minerals. Additional details concerning XRD and EMPA-WDS methods  
203 can be found by consulting Green et al. (2019).

204

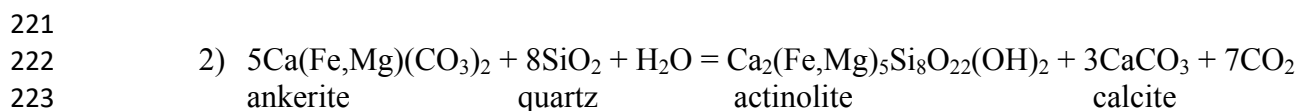
## Results

### 205 **Amphibole formation and distribution**

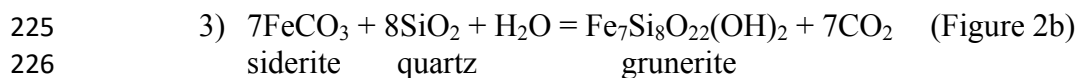
206         The four metamorphic zones of the Ironwood are defined by their predominant mineral  
207 assemblages, which primarily changed with variations in temperature. Magnetite and chert are  
208 ubiquitous throughout the study area, whereas variations in the presence and species of Fe-  
209 silicates indicate changes in metamorphic conditions. Zone 1 is characterized by the presence of  
210 minnesotaite, stilpnomelane, dolomite, ankerite, and siderite. Maximum temperatures in Zone 1  
211 reached approximately 300-340° C during diagenetic alteration/low-grade metamorphism, which  
212 is the upper stability limit of the Fe-phylosilicates and Fe-carbonates characteristic of this zone  
213 (French, 1973; Frost et al., 2007). In Zone 2, which lies approximately 2.5 to 3.5 km from the  
214 Mellen Intrusive Complex, medium-grade metamorphism resulted in the dehydration and  
215 decarbonation of low-temperature minerals and the development of cummingtonite-grunerite and  
216 actinolite-ferro-actinolite by the following reactions (French, 1968; Bonnicksen, 1975; Frost,  
217 1979 (observed in the Mesabi range); Laybourn, 1979 (observed in the Ironwood)):



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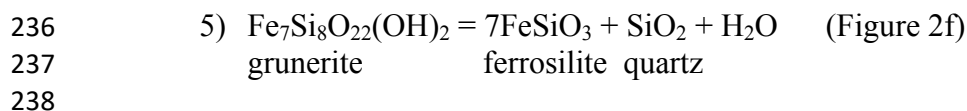
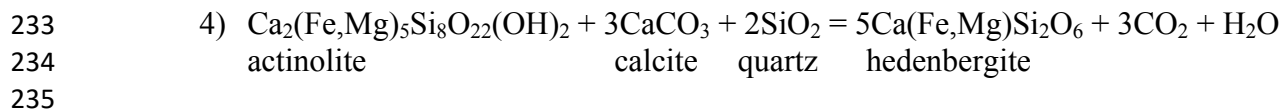


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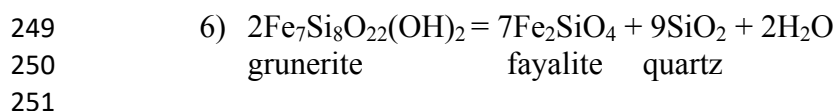
228         Zone 3 occurs in the western portion of the study area where the Ironwood is less than  
229 approximately 2.5 km from the Mellen Complex up to within approximately 100 m from the

230 contact between these units. High-grade metamorphism in this zone resulted in the development  
231 of pyroxenes from amphiboles by the following reactions (Bonnichsen, 1975 (observed in the  
232 Mesabi range); Laybourn, 1979 (observed in the Ironwood)):



239 Retrograde metamorphism in this zone also resulted in the development of fine-grained grunerite  
240 as reaction 5 proceeded in the opposite direction (Bonnichsen, 1969). Textural evidence that this  
241 grunerite is retrograde includes the embayment of orthopyroxene by fine-grained grunerite  
242 (Figure 2d) and the occurrence of fine-grained grunerite aggregates that replace larger prismatic  
243 grunerite crystals (Figure 2e).

244 Zone 4 occurs in the far western portion of the study area within 100 m of the contact  
245 between the Ironwood and the Mellen Complex. Maximum temperatures of approximately 700°  
246 C resulted in the development of fayalite by the following reaction, which is unique to this small,  
247 restricted zone (French, 1968 (observed in the Mesabi Range); Laybourn, 1979 (observed in the  
248 Ironwood)):



252 Retrograde amphibole is observed intergrown with pyroxene and fayalite. Having been  
253 consumed during reaction 6, prograde amphibole is not observed in these samples, although it  
254 has been reported in Zone 4 by Laybourn (1979).

255 The extent and location of each of these zones was determined by the mineralogical  
256 associations observed by optical microscopy and XRD analysis of drill core and outcrop samples

257 throughout the study area. In Zones 2, 3, and 4, amphiboles are present in abundance in each  
258 member of the Ironwood. The prevalence of low-grade mineral assemblages in Zone 1 indicates  
259 that the Ironwood in Zone 1 generally did not reach temperatures high enough to produce  
260 amphibole. Observation of drill core samples in Zone 1 (drill holes 666-1 and 186C), however,  
261 revealed the localized presence of amphiboles among a predominantly low-grade mineral  
262 assemblage comprising Fe-carbonates and Fe-phyllosilicates (Figure 3). Within drill hole 666-1,  
263 the presence of amphibole in minor to trace quantities was observed solely at a depth of  
264 approximately 240-270 m. Within this range, the presence of two diabase dikes are noted in the  
265 core logs. Within drill hole 186C, the presence of amphibole in major quantities was observed  
266 solely at depths of 10 to 40 m; however, no dikes were observed within or near those samples.

267

### 268 **Morphology and geometry**

269 The various habits of amphibole in the Ironwood are commonly intergrown or found in  
270 proximity to one another (Figure 4). In order to provide a representation of the frequency with  
271 which various amphibole habits occur in the Ironwood, amphibole-bearing areas were  
272 characterized using a systematic method of observation. Within the areas that were observed in  
273 thin section from a set of samples in which amphibole is a major constituent, 63% and 25% of  
274 the areas contained massive and prismatic amphiboles, respectively. These are the most common  
275 habits assumed by amphiboles in the Ironwood. Less common forms include equant and fibrous  
276 morphologies, which were each observed in 6% of the areas examined. Fibrous amphiboles,  
277 which have potential human-health and environmental impacts, were identified in each of the  
278 drill holes sampled in this study, but were not observed in outcrop samples. Fibrous amphiboles  
279 were observed in the Pence, Norrie, and Plymouth Members of the Ironwood, which comprise

280 the potential ore zones; whereas, no fibrous amphiboles were observed in the Yale Member,  
281 which is considered to be waste rock due to its high silica and low magnetite content (Figure 3).

282 Cumulative frequency distributions of length, width, and aspect ratios for 437 amphibole  
283 particles are displayed in Figure 5. A difficulty inherent to classifying the morphology and  
284 geometry of amphiboles in thin section is the random orientation of mineral particles, which may  
285 obscure their maximum dimensions. Therefore, measurements of amphibole particles presented  
286 in Figure 5 are potentially minimum values, and represent only prismatic, fibrous, or equant  
287 particles. Massive particles were not measured because, by definition, they are homogenous with  
288 indistinct grain boundaries.

289

### 290 **Amphibole chemistry**

291 The amphiboles that occur in the Ironwood are members of the monoclinic Mg-Fe-Mn-  
292 amphibole group and the Ca-amphibole group (Hawthorne et al., 2012). These amphiboles are  
293 chemically simple and consistent with published values of amphiboles from BIF occurrences  
294 globally (Table 1). The most common species observed include members of the cummingtonite-  
295 grunerite series (Figure 6) and the actinolite-ferroactinolite series; whereas members of the  
296 magnesio-hornblende-ferro-hornblende series are relatively uncommon (Figure 7). Members of  
297 these two groups are frequently intergrown. Mg-Fe-Mn-group amphiboles occur with greater  
298 frequency than do Ca-group amphiboles, the formation of which is dependent upon the presence  
299 of dolomite-ankerite and/or stilpnomelane in the protolith. Among amphiboles of all groups,  
300 variations in their chemistry are observed primarily as changes in the Mg#, ranging from highly  
301 ferrous to highly magnesian. Factors that may influence the Mg# include the presence of  
302 coexisting minerals that can incorporate or buffer Fe and Mg, bulk chemistry of the iron-

303 formation, temperature of amphibole formation, amphibole morphology, and variations in  
304 prograde and retrograde metamorphic reactions.

305         The range of molar  $\text{Mg}/(\text{Mg}+\text{Fe}^{2+})$  values (Mg#) is similar in massive and prismatic  
306 amphiboles and spans from 0.06 to 0.87, representing both the most Fe- and Mg-rich  
307 compositions. In contrast, the Mg# of fibrous amphiboles, regardless of their location, is  
308 restricted to values of 0.14 to 0.35 (Figure 8). Analyses of fibrous grunerite reveal that among  
309 these restricted values, Mg# increases concomitant with true distance (the actual distance  
310 between the units as opposed to the geographic distance observed at the surface) from the Mellen  
311 Intrusive Complex (Figure 9). The relationship between the amphibole Mg# and distance from  
312 the Mellen Intrusive Complex has implications for the peak temperature of metamorphism,  
313 variations in which may result in changes in Mg# values. Because the dips of the Ironwood and  
314 the Mellen Intrusive Complex are roughly parallel, an increase in drill-hole depth results in  
315 increasing distance from the intrusive body accompanied by a decrease in temperature. However,  
316 the magnitude of heat flux within a drill hole as a function of depth is not likely to produce  
317 significant compositional variation in amphiboles. Rather, compositional variation with depth  
318 appears to be a function of bulk composition of the Ironwood. Variations in the range of Mg#  
319 values in amphiboles between various zones may in part be a function of the predominant  
320 mineral assemblage. As amphibole is the primary mineral capable of incorporating Mg in drill  
321 hole W-156 (Zone 2), the wide degree of compositional variation observed in this drill hole may  
322 be explained by the general lack of coexisting minerals capable of accommodating Mg. In  
323 contrast, outcrop samples (Zone 4) and drill hole MP-67-1 (Zone 3) contain abundant pyroxene,  
324 whereas drill holes 666-1, and 186C (Zone 1) contain abundant dolomite. EMPA data indicate

325 that these minerals incorporate significant Mg, which may result in relatively Fe-rich amphiboles  
326 (see Green et al., 2019 for EMPA results).

327 Prograde and retrograde grunerite, distinguished previously on the basis of textural  
328 evidence, can also be identified by chemical composition. Analyses show that prograde  
329 grunerites are relatively enriched in Mn but fall within the range of Mg# values exhibited by  
330 retrograde grunerite (Figure 10).

331

## 332 **Discussion**

### 333 **Amphibole spatial distribution**

334 The development of amphibole in the Ironwood was previously considered to be solely  
335 resulting from pervasive contact metamorphism related to the emplacement of the Mellen  
336 Intrusive Complex. This conclusion was based upon samples from exposed outcrops, which,  
337 while abundant in the western and central portions of the study area, are lacking in the eastern  
338 portion (Laybourn, 1979). Based upon variations in pervasive mineral assemblages, the zone of  
339 amphibole development and the conditions permitting it are theoretically restricted to Zones 2-4,  
340 where temperatures exceeded 300-340° C. Access to drill-core samples in Zone 1, however,  
341 revealed the localized presence of amphibole at depth (Figure 3), in contrast to previous  
342 conclusions about the extent of amphibole development. The mineralogical character of drill  
343 holes 666-1 and 186C is that of a low-grade metamorphic assemblage primarily comprising  
344 dolomite, ankerite, magnetite, chlorite, and chert, indicating that the bulk of this portion of the  
345 Ironwood was not subjected to temperatures in excess of 300-340° C. The presence of grunerite  
346 isolated at 240-270 m depth within drill hole 666-1 is likely dependent upon localized

347 metamorphism of the Ironwood that did not generate a pervasive heat flux. If the heat generated  
348 by the emplacement of the Mellen Intrusive Complex was responsible for grunerite development  
349 at 240-270 m depth in this location, the shallower portions of the drill core, which are closer to  
350 the contact with the Mellen Complex, would have necessarily been subjected to even greater  
351 temperatures, which would have altered the low-grade assemblage and produced pervasive  
352 amphibole-bearing assemblages. The absence of pervasive amphibole development as observed  
353 in Zones 2-4 and the localized development of grunerite at depth in Zone 1 suggest an alternate  
354 mechanism of localized metamorphism, such as the emplacement of mafic dikes and/or sills. In  
355 fact, diabase dikes that cut the Ironwood at nearly right angles to bedding were commonly  
356 identified in underground mine workings in the formerly active part of the central Gogebic Iron  
357 Range. Although seldom seen in natural exposures, the presence of diabase dikes is indicated in  
358 drill core logs from the study area and directly observed in drill core samples (Figure 2c). Based  
359 on this evidence, the presence of amphibole in the Ironwood Iron-Formation cannot be  
360 constrained solely to Zones 2 to 4 but may also result from localized contact metamorphism by  
361 diabase dikes in Zone 1. Previous work indicates that changes in CO<sub>2</sub> content have minimal  
362 effect on expanding the stability field of amphibole; therefore, the development of amphibole in  
363 Zone 1 is likely dependent on localized temperature increases caused by mafic intrusions (Frost,  
364 1979).

365         Whereas the presence of fibrous amphiboles in the Ironwood has not been  
366 unambiguously linked to a particular set of structural features or hydrothermal processes, various  
367 conditions imperative to their formation exist within the study area. High-strain environments  
368 within folds, shear planes, faults, dilation cavities, and at intrusion-host rock boundaries  
369 potentially permit the primary crystallization of fibrous amphiboles from hydrothermal solutions



370 (Ross et al., 2008). Fibrous amphiboles may also form as low-temperature alteration products of  
371 non-fibrous amphiboles (Ross et al., 2008). Thrust faults and folds of various scales that formed  
372 during the Penokean orogeny and Mesoproterozoic northward tilting are widespread throughout  
373 the study area, particularly near Mount Whittlesey where intense deformation resulted in highly  
374 complex structures (Cannon et al., 1993; 2007).

375

376

### **Implications**

377 Previous studies of the Ironwood established the zoned nature of contact metamorphism  
378 which was thought to accurately constrain the extent of amphibole development in the Ironwood.  
379 Results of extensive sampling conducted during this study are in close accord with the original  
380 location of these zones. However, this study presents evidence of amphiboles in the Ironwood  
381 independent of the influence of the contact metamorphic aureole resulting from the emplacement  
382 of the Mellen Intrusive Complex and extends the potential amphibole-bearing areas as far east as  
383 the town of Upson. The identification of localized amphibole development in zones of low-grade  
384 metamorphic conditions suggests localized metamorphism resulting from the intrusion of diabase  
385 dikes approximately normal to bedding in the Ironwood.

386 A salient global issue related to the mining of metamorphosed banded iron formations is  
387 the presence of amphibole minerals and their morphology because of human-health concerns  
388 related to elongate mineral particles. The human-health risks center on particle size and aspect  
389 ratio. The taconite production process generates large volumes of siliceous tailings, which are  
390 typically divided into coarse and fine fractions. The coarse fraction consists of material generally  
391 less than 10 mm in size, and contains less than 10% fine material, which is defined as material

392 that passes through a 200-mesh sieve (approximately 75  $\mu\text{m}$  or less). Processes in which  
393 amphibole particles may be released to the environment include drilling and blasting at the mine  
394 site, loading and hauling to the processing plant, comminution, and tailings disposal. Some  
395 taconite operations allow the coarse and fine fractions to flow together as a slurry to the same  
396 location, rather than separating them (Zanko et al., 2008). Nearly one hundred per cent of  
397 amphibole EMP measured in this study display lengths and widths less than 75  $\mu\text{m}$  (Figure 5a,  
398 5b). This indicates that the comminution (grinding) and separation stages of ore processing  
399 would segregate amphibole EMP in the Ironwood into the fine tailings fraction. Therefore, in  
400 areas where amphibole EMP exist these fine-grained particles that have the potential to be  
401 classified as fibers under regulatory definitions may be produced during taconite processing and  
402 reside in the tailings storage facility.

403         The presence of amphibole within each drill hole examined in this study, and in the  
404 potential ore and waste rock zones of the Ironwood necessitates a comprehensive plan for solid  
405 waste management if a mine is ever put into operation. The comminution process will break  
406 down all types of amphibole particles resulting in the production of fine-grained EMP. However,  
407 it is noteworthy that non-asbestiform amphiboles do not appear to increase risk of lung cancer or  
408 mesothelioma (Gamble and Gibbs, 2008).

409         In the absence of a proposed mine plan, it is only possible to speculate about ore  
410 processing and waste management strategies, both of which will influence potential human-  
411 health risks. The experience in Minnesota with amphibole-bearing taconite tailings will allow for  
412 better informed decisions to be made about the potential future mine development in Wisconsin.  
413 The historical appearance of amphibole particles in drinking water supplies derived from Lake  
414 Superior indicates that disposal of tailings in Lake Superior is untenable (Berndt and Brice,

415 2008). For land-based disposal, erosion of tailings piles and windblown dusts are the main  
416 concerns. Erosion control and dust suppression are common challenges at mines, with which the  
417 mining industry has considerable experience (Mills and Clar, 1976; Evans, 2000; Reed et al.,  
418 2008).

419 As a large undeveloped iron resource capable of solely providing several decades of  
420 domestic supply (U.S. Geological Survey, 2018), the Ironwood will continue to garner interest.  
421 Potential development of this resource must be predicated on a thorough understanding of the  
422 amphiboles in the Ironwood and detailed strategies to mitigate environmental and human-health  
423 risks associated with them. Understanding how various factors contribute to the human-health  
424 risk of exposure to fibrous amphibole particles is a complex challenge that draws upon a variety  
425 of disciplines. The data presented in this study provide a framework upon which further  
426 mineralogical, medical, and biological work can expand to better understand the potential  
427 impacts of developing iron resources affected by metamorphism throughout the world.

428

429

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437

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574

**List of figure captions:**

575 **FIGURE 1.** Geologic map of the study area and sampling locations within the Ironwood Iron-  
576 Formation, after Cannon et al. (1996).

577

578 **FIGURE 2.** Transmitted light photomicrographs in plane-polarized (a, c) and cross-polarized (b,  
579 d, e, f) light of samples from the Ironwood Iron-Formation. **(a)** Mass of grunerite surrounded by  
580 sheaves of minnesotaite (sample 186C-64.0, Zone 1). **(b)** Grunerite developing from siderite  
581 (sample 666-1-813.5, Zone 1). **(c)** Diabase dike (left) cross-cutting relict iron oxide granules in a  
582 quartz matrix at a high angle relative to horizontal bedding (sample 666-1-485.0, Zone 1). **(d)**  
583 Development of retrograde grunerite from orthopyroxene (sample MP-67-1-205.0, Zone 3). **(e)**  
584 Replacement of prograde prismatic grunerite by fine-grained retrograde grunerite (sample MP-  
585 67-1-143.3, Zone 3). **(f)** Ferrosilite and grunerite (sample MP-67-1-532.5, Zone 3).

586

587 **FIGURE 3.** Stratigraphic representation of sampling locations and amphibole occurrences in the  
588 Ironwood Iron-Formation.

589

590 **FIGURE 4.** SEM micrographs showing examples of various amphibole morphologies in the  
591 Ironwood Iron-Formation. **(a)** Massive intergrowth of ferro- and magnesio-hornblende with  
592 magnetite (sample MP-67-1-607.5). **(b)** Interpenetrant prismatic grunerite crystals and an  
593 elongate grunerite crystal (length = 37  $\mu\text{m}$ , width = 1.9  $\mu\text{m}$ , sample MP-67-1-395.5). **(c)** Sprays  
594 of fibrous grunerite crystals (sample W-156-512.5). **(d)** Equant grunerite crystals and an elongate  
595 grunerite crystal (length = 67  $\mu\text{m}$ , width = 4.7  $\mu\text{m}$ , sample MP-67-1-395.5).

596

597 **FIGURE 5.** Cumulative frequency distribution of (a) lengths (b) widths, and (c) aspect ratios for  
598 amphibole particles in the Ironwood Iron-Formation.

599

600 **FIGURE 6.** The lower, iron-rich portion of the Mg-Mn-Fe ternary showing the compositions of  
601 monoclinic Mg-Fe-Mn-group amphiboles from various drill cores and outcrops of the Ironwood  
602 Iron-Formation (after Hawthorne et al., 2012).

603

604 **FIGURE 7.** Classification of the Ca-group amphiboles in the Ironwood Iron-Formation (after  
605 Leake et al., 1997). No Ca-group amphiboles were identified in drill hole 666-1.

606

607 **FIGURE 8.**  $Mg/(Mg+Fe^{2+})$  values of all analyzed amphiboles in the Ironwood Iron-Formation  
608 separated by morphology.

609

610 **FIGURE 9.** Variation of Mg# with true distance from the Mellen Intrusive Complex of fibrous  
611 grunerites from the Ironwood Iron-Formation. Note: True distance accounts for the dip of both  
612 the Ironwood Iron-Formation and Mellen Intrusive Complex.

613

614 **FIGURE 10.** The lower, iron-rich portion of the Mg-Mn-Fe ternary showing the compositional  
615 variation between prograde and retrograde grunerite from drill hole MP-67-1 in the Ironwood  
616 Iron-Formation (after Hawthorne et al., 2012).

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621 **TABLE 1.** Amphibole EMPA data from the Ironwood Iron-Formation compared to other BIF-  
 622 hosted amphiboles. Refer to Green et al., 2019 for individual analyses.

| Analysis<br>(wt %)                  | Ironwood Iron-Formation<br>Wisconsin, USA |                                | Kushaka belt BIF<br>Nigeria    |                         | Penge Mine<br>South Africa     |                          | Bell Lake BIF<br>Slave craton, Canada |       |       |
|-------------------------------------|---|--------------------------------|--------------------------------|-------------------------|--------------------------------|--------------------------|---------------------------------------|-------|-------|
|                                     | Grunerite<br>Avg (n=253)                  | Ferro-actinolite<br>Avg (n=55) | Grunerite<br>Avg (n=2)         | Grunerite<br>Avg (n=10) | Grunerite<br>Avg (n=14)        | Actinolite<br>Avg (n=11) |                                       |       |       |
| SiO <sub>2</sub>                    | 49.37                                     | 50.65                          | SiO <sub>2</sub>               | 49.06                   | SiO <sub>2</sub>               | 50.51                    | SiO <sub>2</sub>                      | 51.70 | 53.22 |
| TiO <sub>2</sub>                    | 0.03                                      | 0.04                           | TiO <sub>2</sub>               | -                       | TiO <sub>2</sub>               | 0.02                     | TiO <sub>2</sub>                      | 0.00  | 0.01  |
| Al <sub>2</sub> O <sub>3</sub>      | 0.73                                      | 0.66                           | Al <sub>2</sub> O <sub>3</sub> | 0.19                    | Al <sub>2</sub> O <sub>3</sub> | 0.20                     | Al <sub>2</sub> O <sub>3</sub>        | 0.23  | 0.54  |
| Cr <sub>2</sub> O <sub>3</sub>      | 0.02                                      | 0.02                           | Cr <sub>2</sub> O <sub>3</sub> | -                       | Cr <sub>2</sub> O <sub>3</sub> | 0.02                     | Cr <sub>2</sub> O <sub>3</sub>        | 0.00  | 0.00  |
| MnO                                 | 1.35                                      | 0.95                           | MnO                            | 2.51                    | MnO                            | 0.63                     | MnO                                   | 0.73  | 0.30  |
| FeO                                 | 36.73                                     | 28.07                          | FeO                            | 40.68                   | FeO                            | 38.78                    | FeO                                   | 35.83 | 23.04 |
| Fe <sub>2</sub> O <sub>3</sub>      | 1.14                                      | 1.35                           | Fe <sub>2</sub> O <sub>3</sub> | 0.46                    | Fe <sub>2</sub> O <sub>3</sub> | -                        | Fe <sub>2</sub> O <sub>3</sub>        | -     | -     |
| MgO                                 | 6.22                                      | 5.48                           | MgO                            | 4.69                    | MgO                            | 6.65                     | MgO                                   | 8.72  | 9.45  |
| CaO                                 | 1.09                                      | 10.79                          | CaO                            | 0.37                    | CaO                            | 0.12                     | CaO                                   | 0.73  | 11.70 |
| Na <sub>2</sub> O                   | 0.12                                      | 0.13                           | Na <sub>2</sub> O              | -                       | Na <sub>2</sub> O              | 0.07                     | Na <sub>2</sub> O                     | 0.03  | 0.07  |
| K <sub>2</sub> O                    | 0.11                                      | 0.06                           | K <sub>2</sub> O               | -                       | K <sub>2</sub> O               | 0.04                     | K <sub>2</sub> O                      | 0.00  | 0.00  |
| F                                   | 0.00                                      | 0.00                           | F                              | -                       | F                              | 0.04                     | F                                     | -     | -     |
| Cl                                  | 0.12                                      | 0.11                           | Cl                             | -                       | Cl                             | 0.02                     | Cl                                    | -     | -     |
| H <sub>2</sub> O                    | 1.88                                      | 1.91                           | H <sub>2</sub> O               | 1.86                    | H <sub>2</sub> O               | 2.91                     | H <sub>2</sub> O                      | -     | -     |
| Total                               | 98.76                                     | 100.10                         | Total                          | 99.80                   | Total                          | 97.09                    | Total                                 | 97.98 | 98.34 |
| <i>Formula assignments</i>          |   |                                |                                |                         |                                |                          |                                       |       |       |
| <i>based on - 24 (OH, F, Cl, O)</i> |   |                                |                                |                         |                                |                          |                                       |       |       |
| Si                                  | 7.90                                      | 7.86                           | Si                             | 7.91                    | Si                             | 8.03                     | Si                                    | 8.05  | 7.95  |
| Al                                  | 0.21                                      | 0.17                           | Al                             | 0.04                    | Fe <sup>2+</sup>               | 5.16                     | Ti                                    | 0.00  | 0.00  |
| Ti                                  | 0.01                                      | 0.01                           | Fe <sup>3+</sup>               | 0.06                    | Mg                             | 1.58                     | Al                                    | 0.04  | 0.10  |
| Fe <sup>3+</sup>                    | 0.30                                      | 0.21                           | Fe <sup>2+</sup>               | 5.51                    | Mn                             | 0.08                     | Cr                                    | 0.00  | 0.00  |
| Cr                                  | 0.00                                      | 0.00                           | Mg                             | 1.10                    | Al                             | 0.04                     | Fe <sup>3+</sup>                      | 0.00  | 0.17  |
| Mn                                  | 0.18                                      | 0.21                           | Mn                             | 0.34                    | Ca                             | 0.02                     | Fe <sup>2+</sup>                      | 4.67  | 2.71  |
| Fe <sup>2+</sup>                    | 4.92                                      | 3.68                           | Ca                             | 0.06                    | Na                             | 0.02                     | Mn                                    | 0.10  | 0.04  |
| Mg                                  | 1.47                                      | 1.26                           | O                              | 22.00                   | Sum T,C,B,A                    | 14.92                    | Ni                                    | 0.00  | 0.00  |
| Ca                                  | 0.19                                      | 1.79                           | OH                             | 2.00                    |                                |                          | Mg                                    | 2.02  | 2.10  |
| Na                                  | 0.04                                      | 0.04                           | Sum T,C,B,A                    | 15.02                   |                                |                          | Ca                                    | 0.12  | 1.87  |
| K                                   | 0.02                                      | 0.01                           |                                |                         |                                |                          | Na                                    | 0.01  | 0.02  |
| O                                   | 22.00                                     | 22.00                          |                                |                         |                                |                          | K                                     | 0.00  | 0.00  |
| OH                                  | 1.98                                      | 1.98                           |                                |                         |                                |                          | Sum T,C,B,A                           | 15.01 | 14.96 |
| F                                   | 0.00                                      | 0.00                           |                                |                         |                                |                          |                                       |       |       |
| Cl                                  | 0.03                                      | 0.03                           |                                |                         |                                |                          |                                       |       |       |
| Sum T,C,B,A                         | 14.99                                     | 15.02                          |                                |                         |                                |                          |                                       |       |       |
|                                     |   |                                | Mücke and Annor, 1993          |                         | Lafuente et al., 2015          |                          | Katsuta et al., 2012                  |       |       |

Figure 1

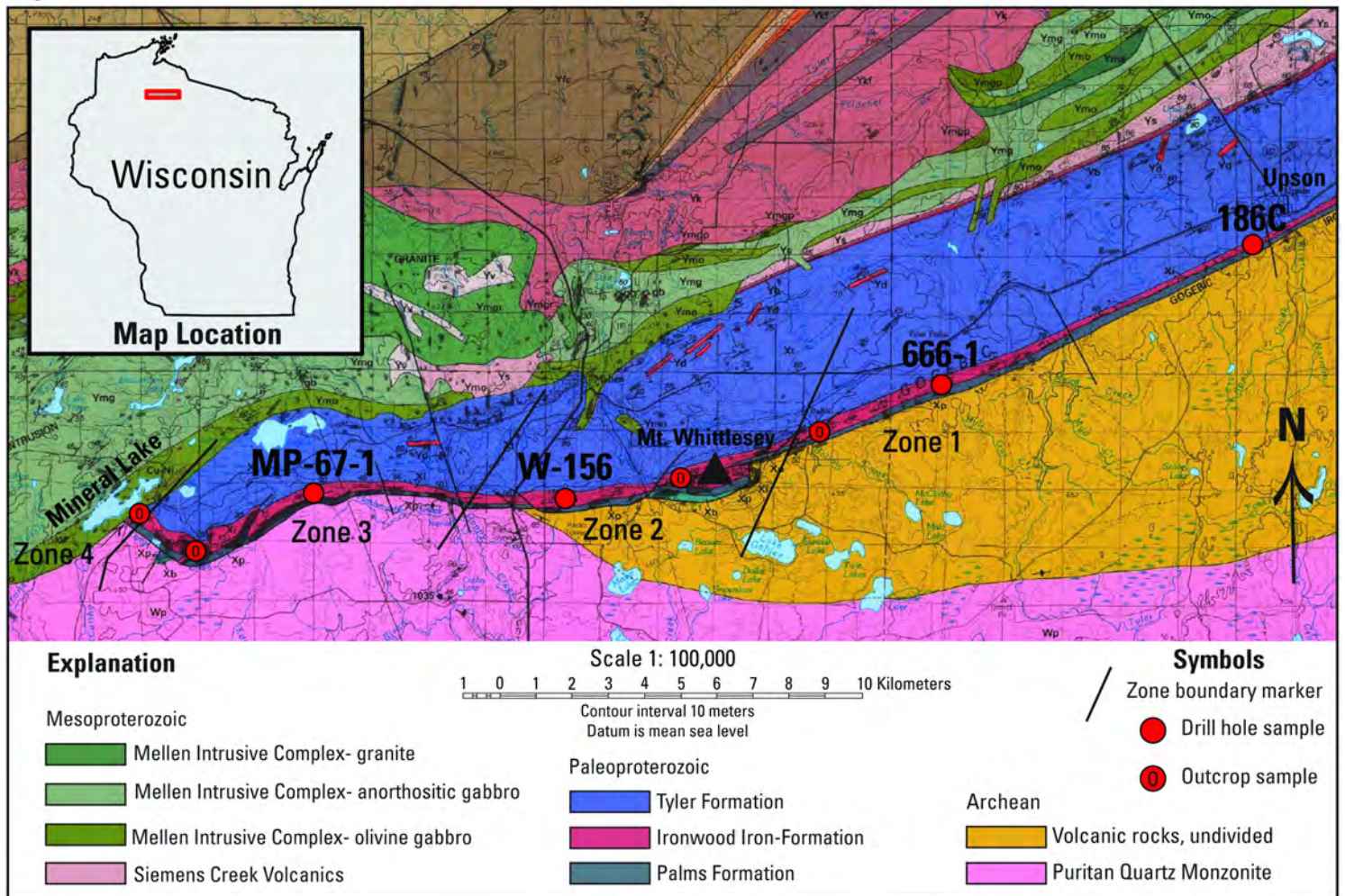


Figure 2

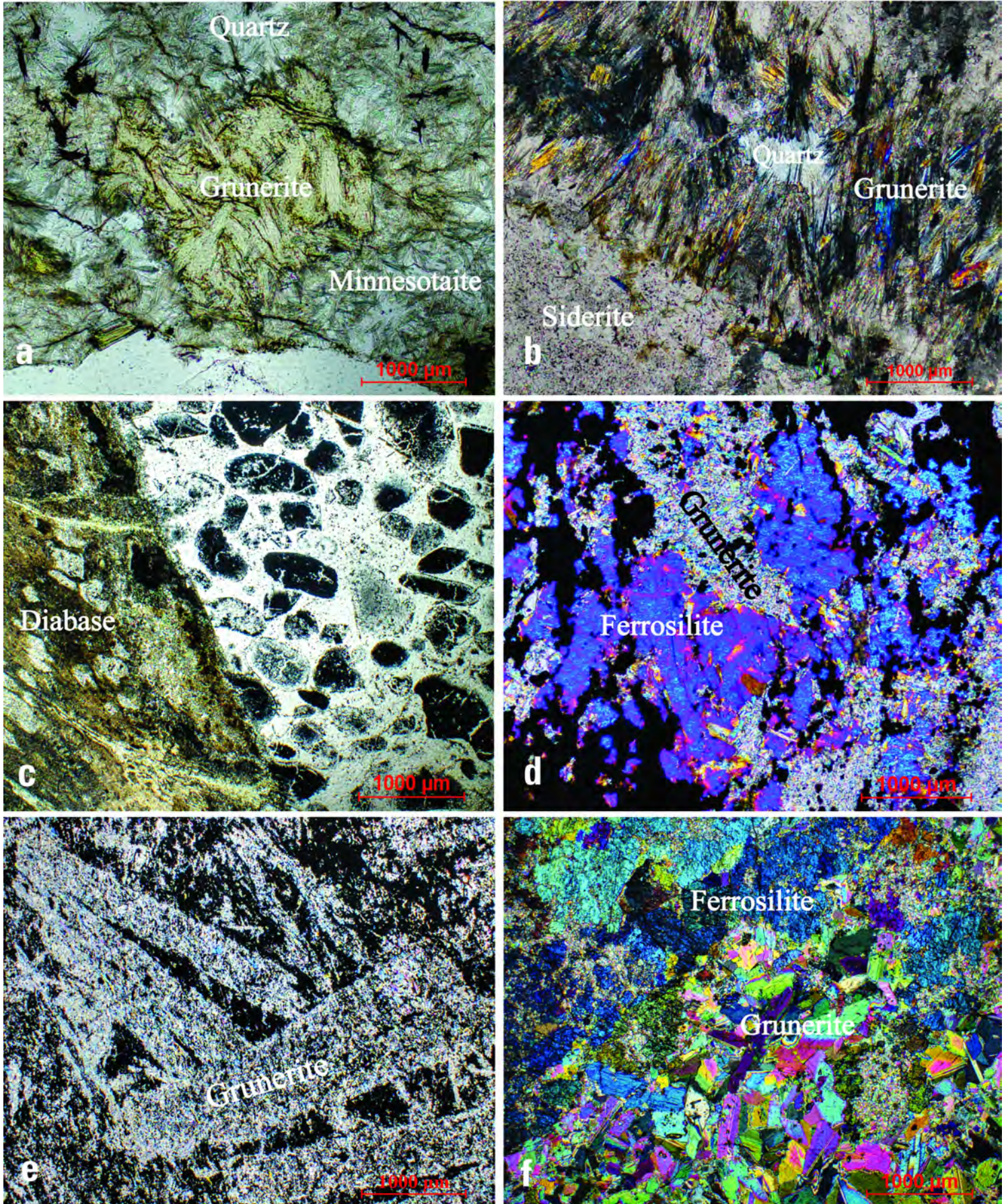


Figure 3

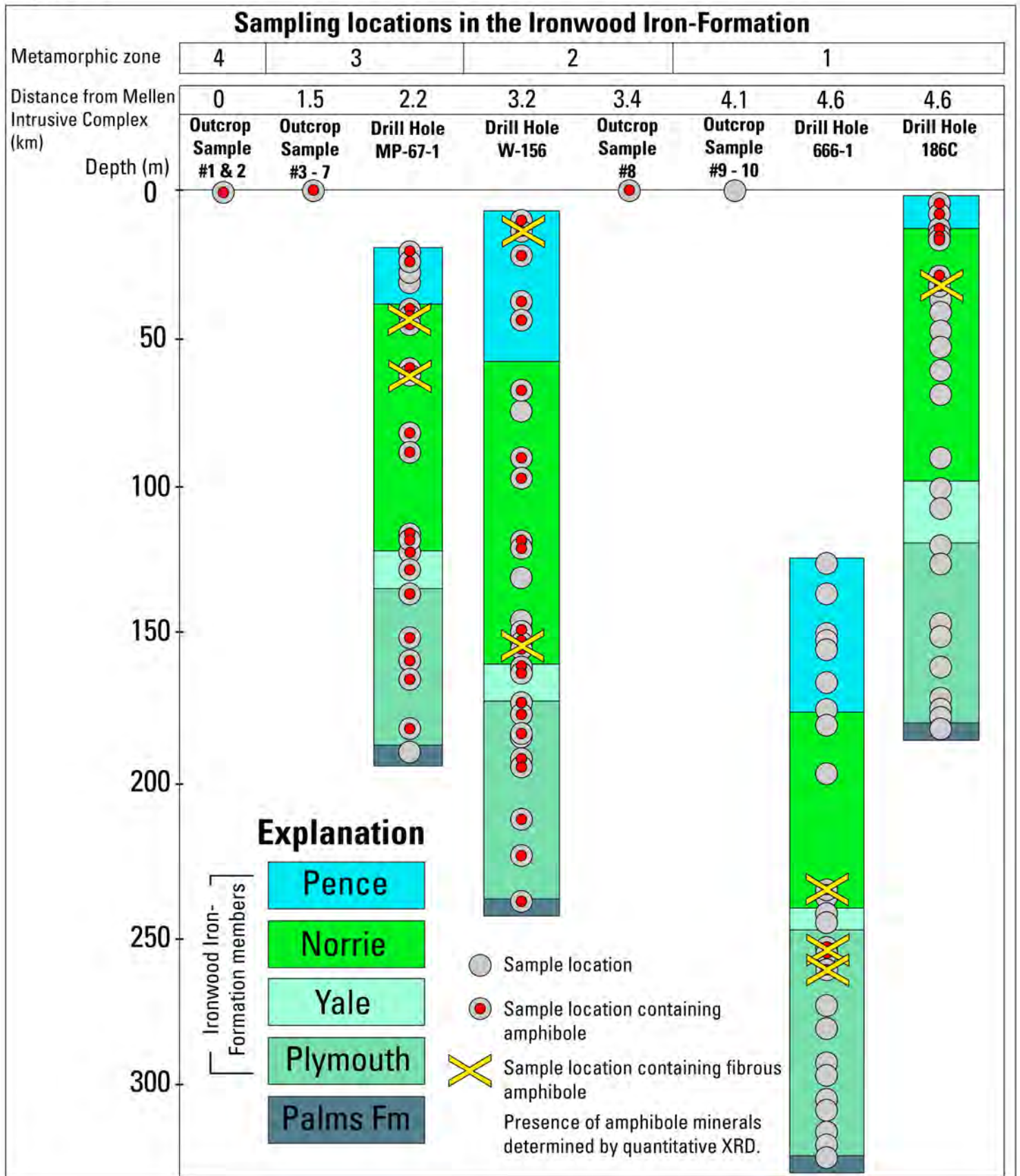


Figure 4

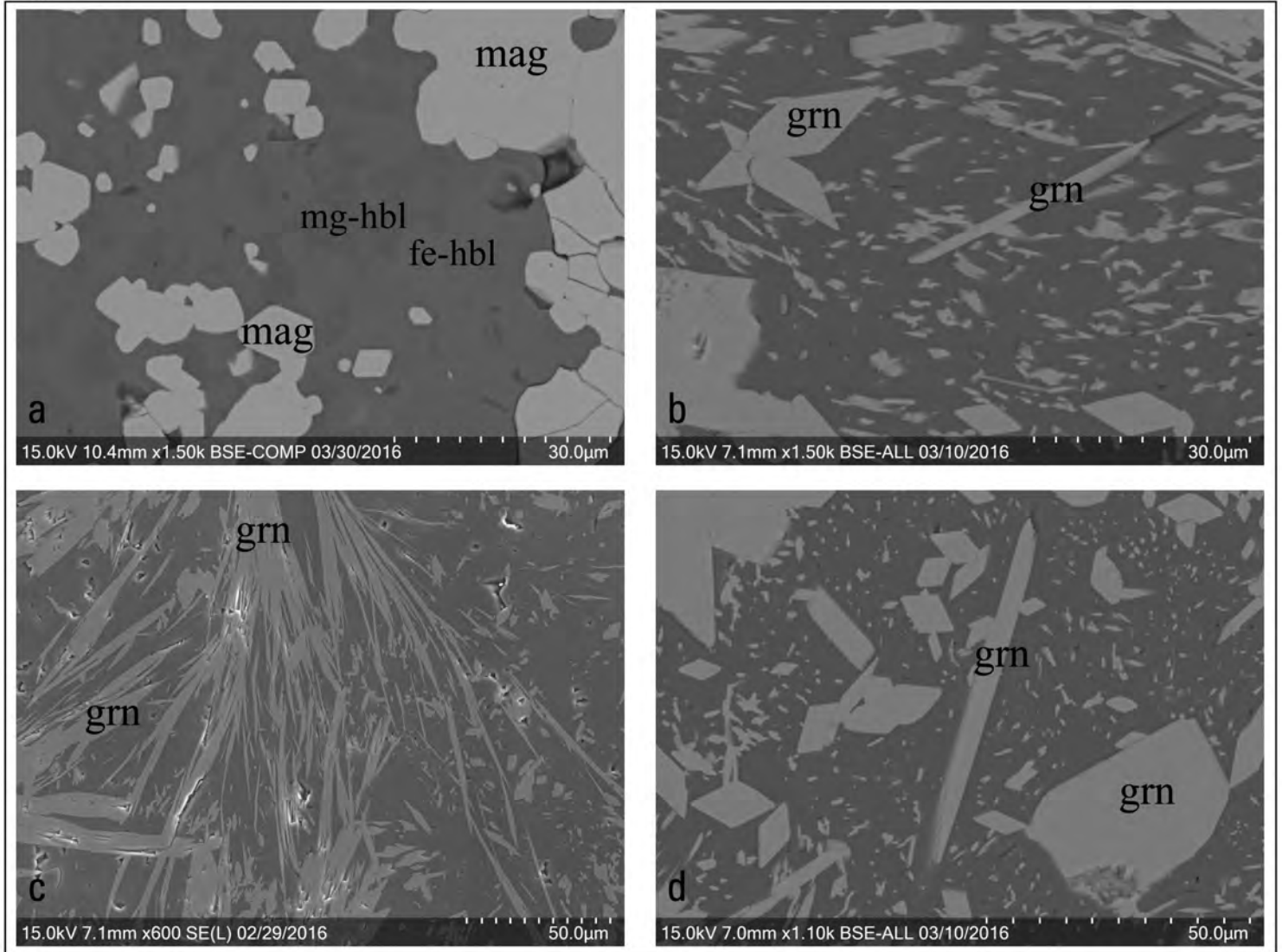


Figure 5

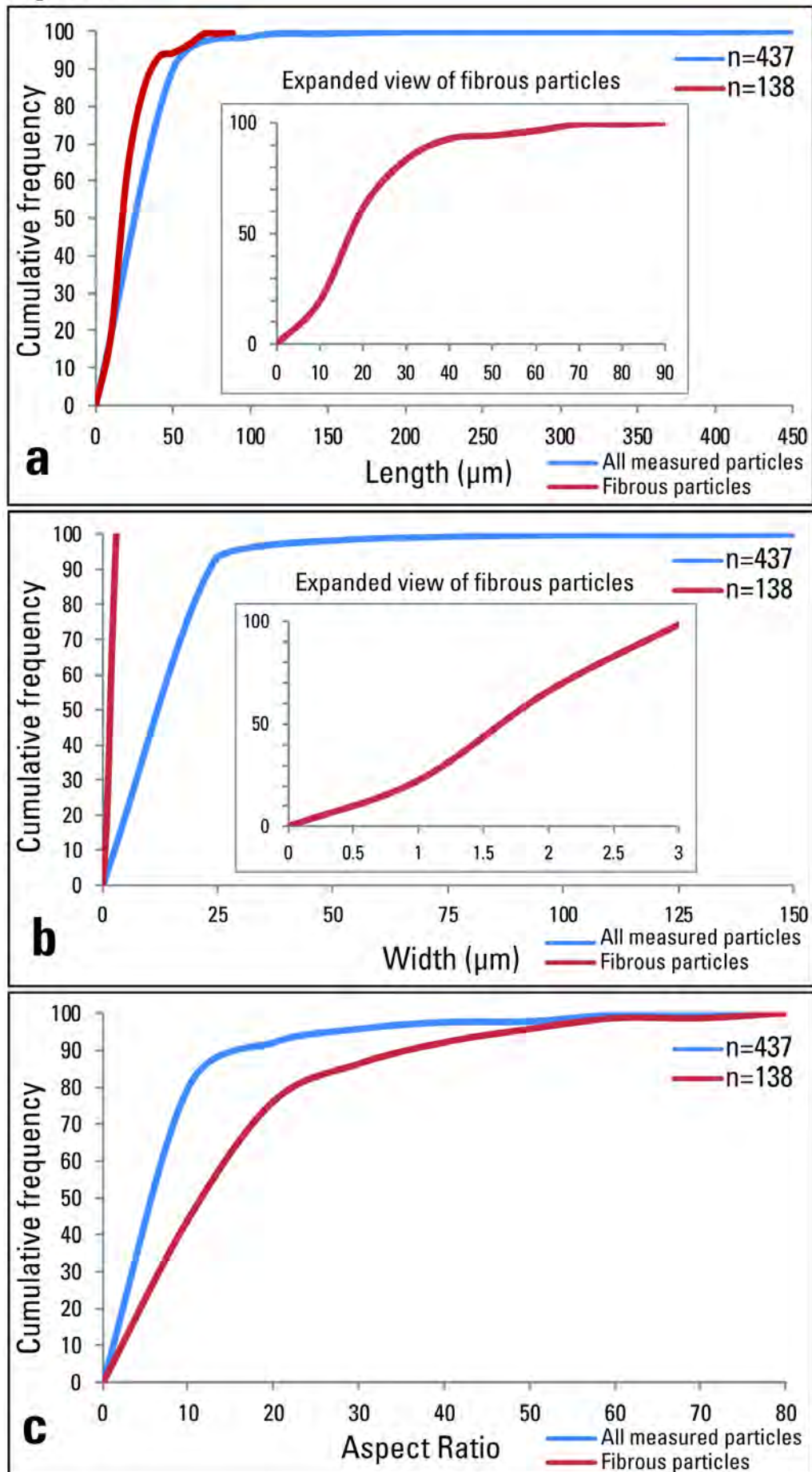


Figure 6

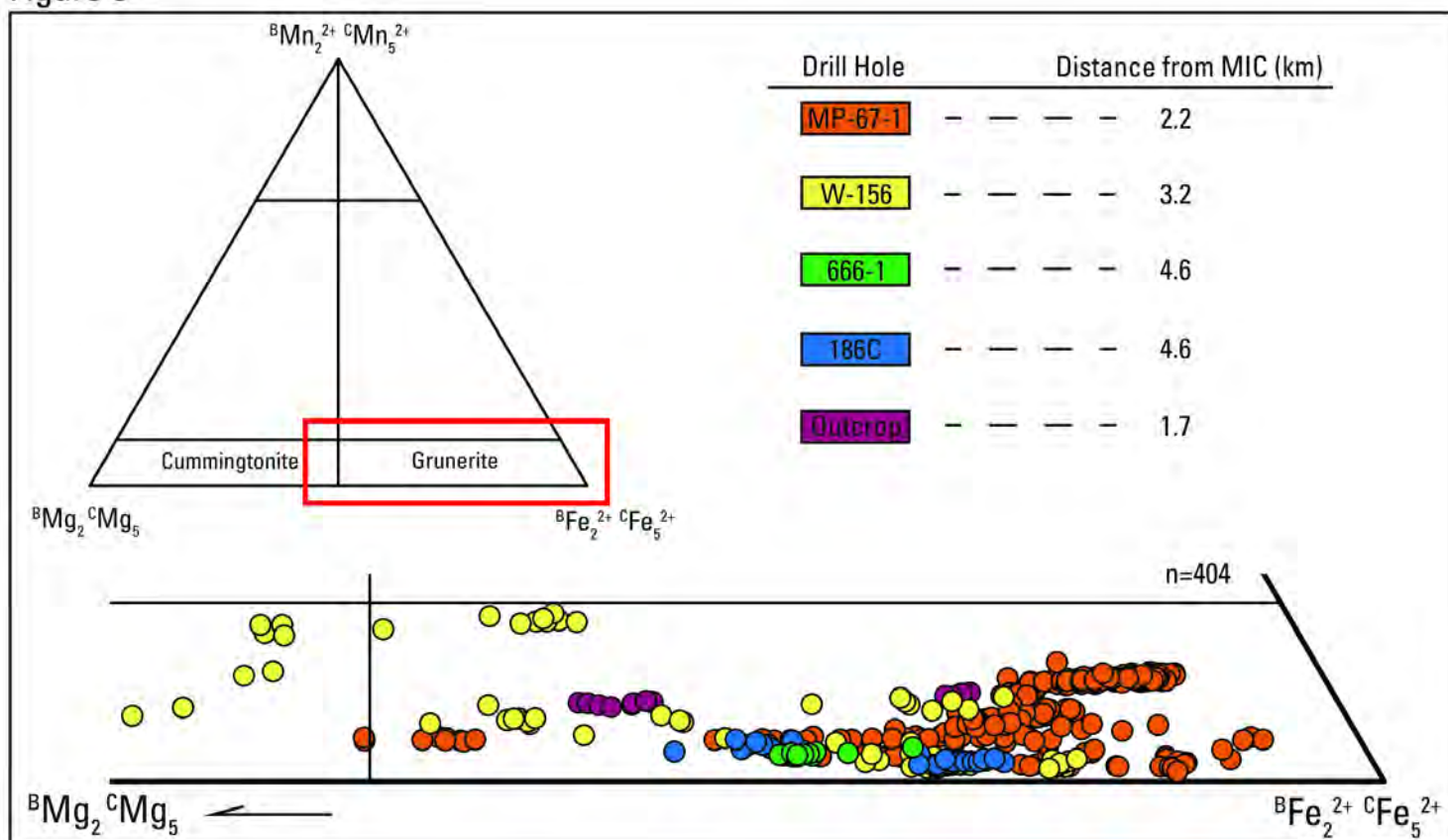


Figure 7

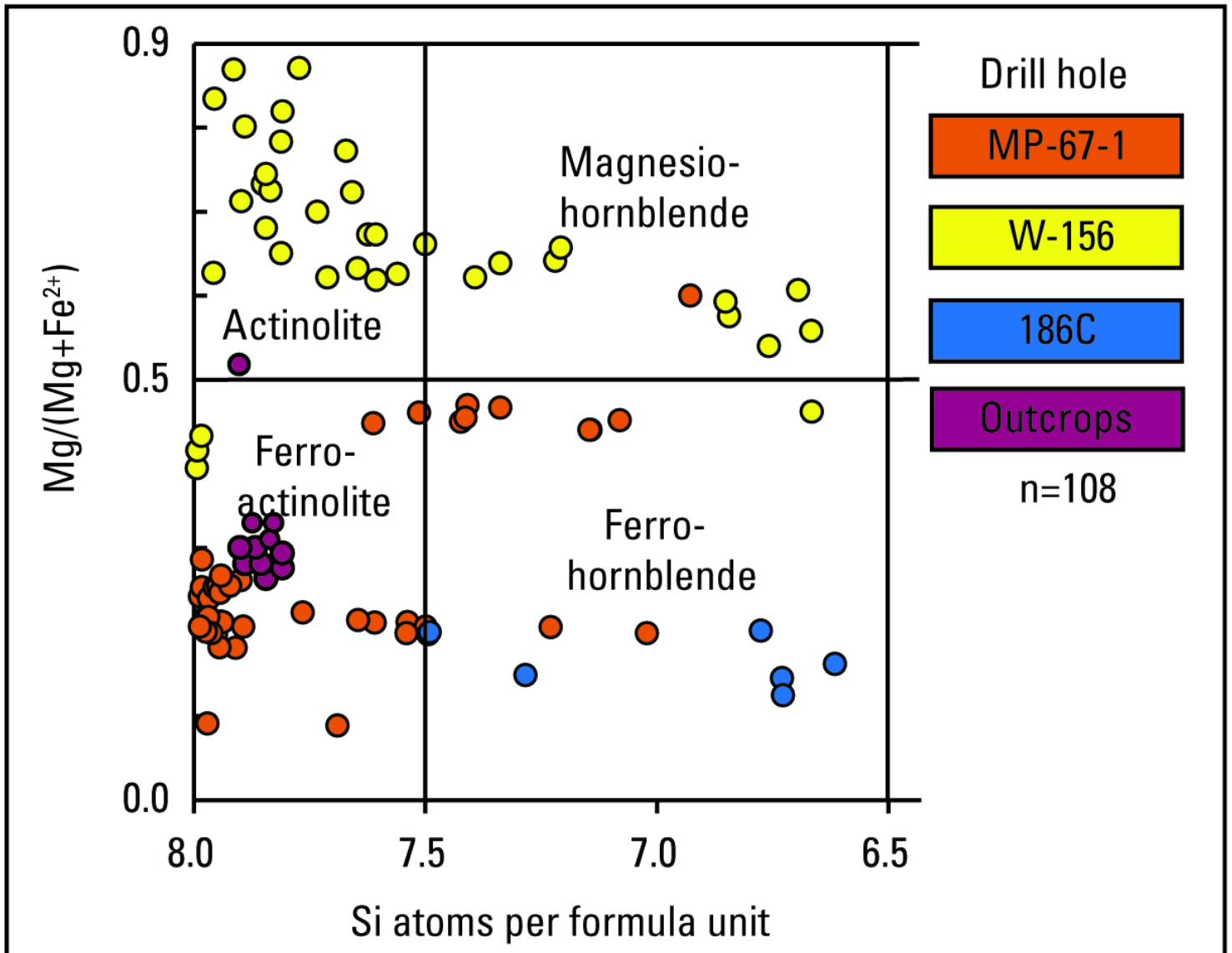




Figure 8

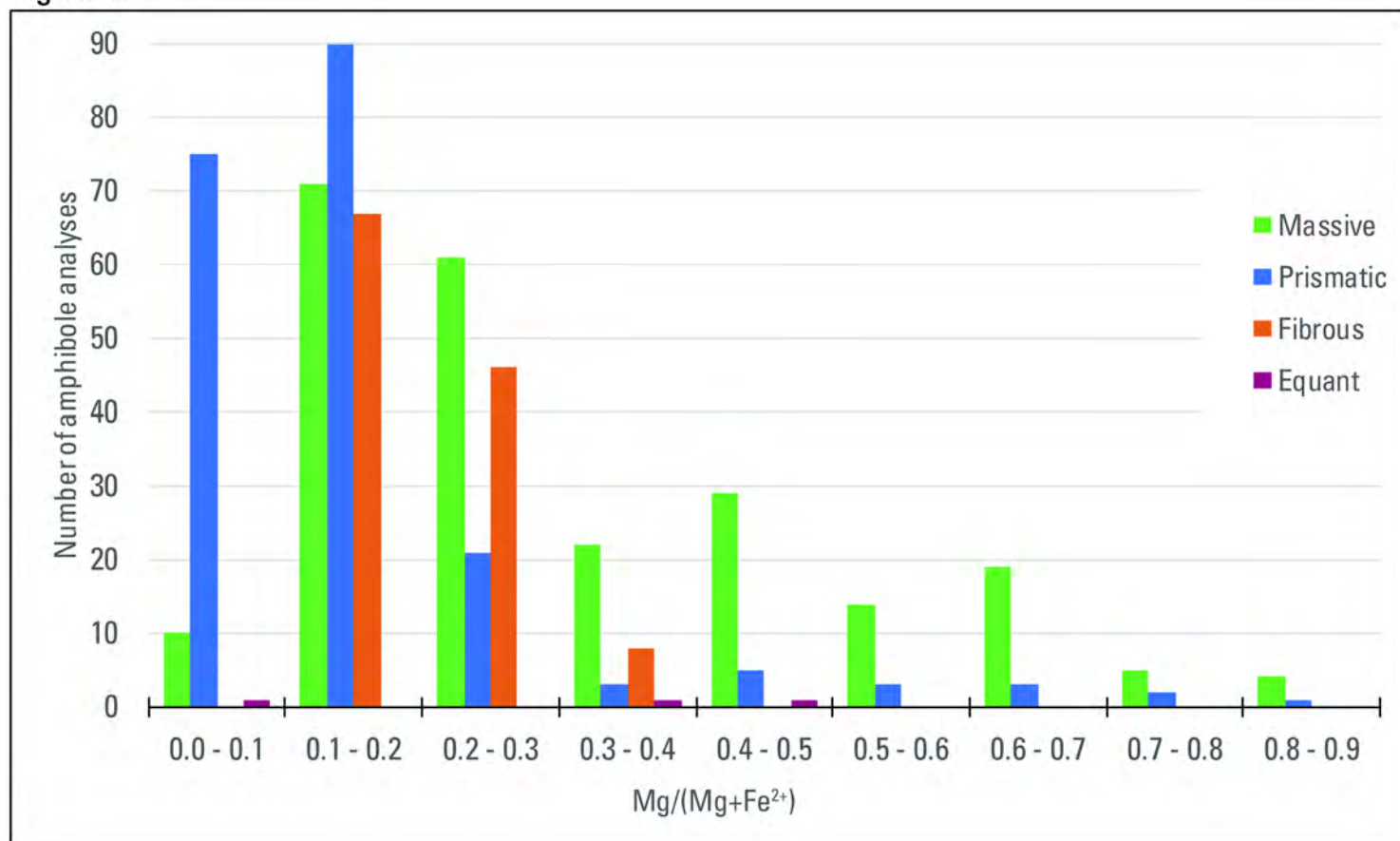


Figure 9

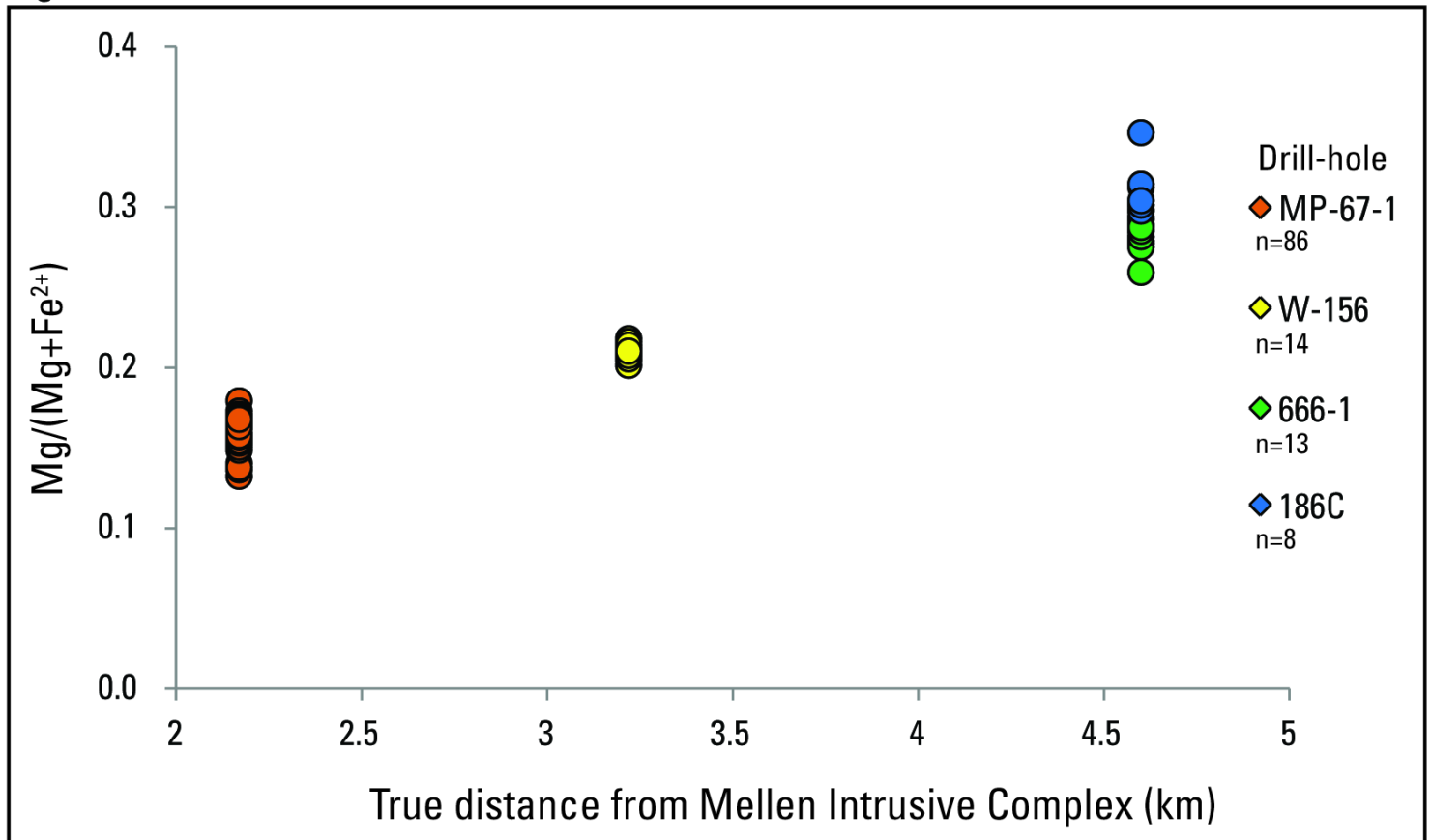


Figure 10

