1 Revision 2

2 Elasticity of single-crystal periclase at high pressure and temperature:

3 the effect of iron on the elasticity and seismic parameters of

- 4 ferropericlase in the lower mantle
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16 **ABSTRACT:**

17 In this study, we measured the sound velocities of single-crystal periclase by Brillouin 18 light scattering (BLS) combined with in situ synchrotron X-ray diffraction (XRD) up 19 to ~30 GPa and 900 K in an externally-heated diamond anvil cell (EHDAC). Our 20 experimental results were used to evaluate the combined effects of pressure and 21 temperature on the elastic moduli of single-crystal periclase using third-order Eulerian 22 finite-strain equations. All of the elastic moduli increased with increasing pressure but decreased with increasing temperature, except the off-diagonal modulus C_{12} , which 23 24 remained almost constant up to ~30 GPa and 900 K. The derived aggregate adiabatic 25 bulk and shear moduli (K_{S0} , G_0) at ambient conditions were 162.8(±0.2) GPa and 26 $130.3(\pm 0.2)$ GPa, respectively, consistent with literature results. The pressure

27 derivatives of the bulk $[(\partial K_S / \partial P)_{300K}]$ and shear moduli $[(\partial G / \partial P)_{300K}]$ at ambient 28 conditions were $3.94(\pm 0.05)$ and $2.17(\pm 0.02)$, respectively, whereas the temperature 29 derivatives of these moduli $[(\partial K_{S}/\partial T)_{P}]$ and $(\partial G/\partial T)_{P}$ at ambient conditions were 30 -0.025(±0.001) GPa/K and -0.020(±0.001) GPa/K, respectively. A comparison of our 31 experimental results with the high pressure-temperature (P-T) elastic moduli of 32 ferropericlase (Fp) in the literature showed that all the elastic moduli of Fp were 33 linearly correlated with the FeO content up to approximately 20 mol%. These results 34 allowed us to build a comprehensive thermoelastic model for Fp to evaluate the effect 35 of Fe-Mg substitution on the elasticity and seismic parameters of Fp at the relevant 36 P-T conditions of the lower mantle. Our modeling results showed that both the 37 increase of the Fe content in Fp and the increasing depth could change the 38 compressional wave anisotropy $(AV_{\rm P})$ and shear wave splitting anisotropy $(AV_{\rm S})$ of Fp 39 in the upper parts of the lower mantle. Furthermore, using our modeling results here, 40 we also evaluated the contribution of Fp to seismic lateral heterogeneities of thermal 41 or chemical origin in the lower mantle. Both the thermally-induced and Fe-induced 42 heterogeneities ratios ($R_{S/P}=\partial \ln V_S/\partial \ln V_P$) of Fp from 670 to 1250 km along a representative lower mantle geotherm increased by ~2-5% and ~15%, respectively. 43 44 The thermally-induced $R_{S/P}$ value of Fp20 is ~ 30% higher than Fp10, indicating that 45 the Fe content has a significant effect on the thermally-induced $R_{S/P}$ of Fp. Compared 46 to the seismic observation results ($R_{S/P}=1.7-2.0$) in the upper regions of the lower 47 mantle, the Fe-induced $R_{S/P}$ value of Fp is more compatible than the thermally-induced $R_{S/P}$ value of Fp20 (the expected composition of Fp in the lower 48 49 mantle) within their uncertainties. Thus, we propose that Fe-induced lateral 50 heterogeneities can significantly contribute to the observed seismic lateral heterogeneities in the Earth's lower mantle (670-1250 km). 51

52

53 Keywords: Elasticity, Periclase, Brillouin Light Scattering, Lower Mantle, Diamond
54 Anvil Cell

55 INTRODUCTION

56 Periclase (MgO) is a classic simple oxide that crystallizes in the rock-salt structure 57 with no known experimental phase transition up to 250 GPa at least (e.g., Dorfman et 58 al. 2012; Duffy et al. 1995a; McWilliams et al. 2012). The rock-salt structured MgO 59 also represents one of the most common yet simply structured materials in the deep Earth. The structure can potentially be stable up to approximately 400 GPa based on 60 61 computational studies and laser shock experiments (e.g., Belonoshko et al. 2010; 62 Coppari et al. 2013; Karki et al. 1997; McWilliams et al. 2012; Oganov et al. 2003). 63 MgO also has very high melting temperatures compared to other mantle minerals (e.g., 64 Ito et al. 2004; Kimura et al. 2017; McWilliams et al. 2012; Tateno et al. 2014; Zerr 65 and Boehler 1994). The wide P-T stability of MgO covers relevant high P-T 66 conditions of the deep Earth (Duffy and Ahrens 1995) and possibly terrestrial planets 67 or exoplanets (Bolis et al. 2016; Coppari et al. 2013; Duffy et al. 2015). Therefore, 68 precise knowledge of the elastic properties of MgO under high P-T conditions is 69 crucial for constructing a reliable mineralogical model of the Earth's lower mantle. 70 There have been an increasing number of studies on the elastic behavior of MgO 71 covering a range of high-pressure or temperature conditions using different techniques, 72 including static XRD (e.g., Dewaele et al. 2000; Fei 1999; Jacobsen et al. 2008; 73 Speziale et al. 2001; Utsumi et al. 1998), BLS (e.g., Murakami et al. 2009, 2012; 74 Sinogeikin and Bass 2000; Zha et al. 2000), ultrasonic interferometry (e.g., Chen et al. 75 1998; Li et al. 2006; Reichmann et al. 1998; Yoneda 1990), shockwave compressions 76 (e.g., Coppari et al. 2013; Duffy and Ahrens 1995), as well as theoretical studies (e.g., 77 Dorogokupets 2010; Isaak et al. 1990; Karki et al. 1999; Matsui et al. 2000; Wu et al. 78 2008). To date, however, the elasticity of single-crystal MgO at simultaneous high 79 P-T conditions is limited to 8 GPa and 1600 K (Chen et al. 1998), much lower than 80 the P-T conditions expected in the lower mantle, which extends at depths 81 corresponding to pressures above 23 GPa.

82 MgO forms a solid solution with FeO, and in a MgO-rich composition, it is known 83 as ferropericlase (Fp) (Mg,Fe)O while magnesiowustite is named for a FeO-rich 84 composition. In the Earth's lower mantle, Fp is believed to be the second most 85 abundant mineral, co-existing with the most abundant (Al,Fe)-bearing bridgmanite 86 (Bgm) and Ca-perovskite (e.g., Kung et al. 2002; Lin et al. 2005, 2006; Marquardt et 87 al. 2009a, 2009b; Ringwood 1975). Experiments on iron partitioning between Fp and 88 Bgm (K_D = (Fe/Mg)^{Bgm}/ (Fe/Mg)^{Fp}) at high P-T conditions suggest that the FeO 89 content in Fp lies in the range of 10-20 mol% for potential compositional models of 90 the lower mantle (e.g., Andrault 2001; Irifune et al., 2010; Mao et al. 1997; Wood 91 2000). The influences of Fe-Mg substitution on the physical properties of Fp, 92 especially across its electronic spin transition, have recently attracted significant 93 interest to further our understanding of seismic observations and the geodynamical 94 processes of the lower mantle (e.g., Crowhurst et al. 2008; Fei et al. 2007; Lin et al. 95 2006, 2013; Marquardt et al. 2009; Muir and Brodholt 2015; Wu and Wentzcovitch 96 2014; Yang et al. 2015). The effect of Fe-Mg substitution on the elastic properties of 97 Fp has been investigated for the whole MgO-FeO system (e.g., Jacobsen et al. 2002; 98 Muir and Brodholt 2015). In particular, previous studies on the effect of FeO in Fp 99 showed that the substitution of FeO strongly increased the density and reduced the 100 velocities of Fp (e.g., Badro et al. 2003; Lin et al. 2013; Muir and Brodholt 2015; 101 Tsuchiya et al. 2006; Yang et al. 2015). However, the fundamental question of how 102 FeO substitution affects the elastic moduli of Fp at high P-T conditions still needs to 103 be better addressed experimentally.

Recently, there have been a number of studies attempting to characterize the effects of FeO on the full elasticity of Fp at the high *P* or high *P-T* conditions of the lower mantle (e.g., Jackson et al. 2006; Jacobsen et al. 2002; Marquardt et al. 2009a; Reichmann et al. 2008; Yang et al. 2015, 2016). Single-crystal XRD and ultrasonic interferometry measurements of the full elastic moduli for a wide range of compositions in the MgO-FeO system at ambient conditions have been documented

110 (Jacobsen et al. 2002). In situ ultrasonic interferometry together with synchrotron 111 XRD measurements on polycrystalline MgO have also been conducted up to 23.6 112 GPa and 1650 K, and used to establish an independent absolute pressure scale at high 113 P-T conditions (Kono et al. 2010; Li et al. 2006; Tange et al. 2009). However, 114 ultrasonic technique measurements are limited in pressure (<30 GPa) and are most 115 readily performed on polycrystalline samples (e.g., Kung et al. 2002; Li et al. 2005, 116 2006; Li and Zhang 2005), thus, they only provide limited information on the elastic 117 properties of minerals. Nowadays, BLS coupled with diamond anvil cells (DAC) is 118 the most commonly used technique to measure the single-crystal elasticity of mantle 119 minerals (e.g., Duffy et al. 1995b; Fu et al. 2017; Sang and Bass 2014; Sinogeikin and 120 Bass 2002; Wu et al. 2016; Yang et al. 2014). It has tremendous advantages at 121 extremely high P-T conditions for optically transparent and translucent samples (e.g., 122 Marquardt et al. 2009a; Yang et al. 2015) and to derive the complete elastic moduli of 123 single-crystal materials (Speziale et al. 2014). BLS has been used to study the 124 single-crystal elasticity of MgO at pressures up to 50 GPa at room temperature (Zha et 125 al. 2000) and temperatures up to 2500 K at room pressure (Sinogeikin et al. 2000, 126 2004a). Simultaneous high P-T single-crystal elasticity studies of the candidate 127 minerals (e.g. olivine, pyroxene, garnet, spinel, ringwoodite, Fp) of the Earth's interior 128 were also reported up to 50 GPa and 1300 K using the BLS technique (e.g., Duan et al. 129 2018; Lu et al. 2013; Mao et al. 2012, 2015; Yang et al. 2014, 2016; Zhang and Bass 130 2016). For the single-crystal elasticity of Fp, BLS coupled with an EHDAC was 131 reported up to 50 GPa and 900 K (Yang et al. 2016), while the high-pressure 132 single-crystal elasticity of Fp was reported up to 96 GPa at room temperature 133 (Marquardt et al. 2009a; Yang et al. 2015). However, studying the single-crystal 134 elasticity of Fp with more than 10 mol% FeO (hereafter written as Fp10) is difficult 135 because Fp with a higher amount of FeO is known to be too optically dark to perform 136 the BLS measurement. Using high-pressure inelastic X-ray scattering (IXS) 137 measurements (Figuet et al. 2004) of the acoustic phonon dispersions of Fp to derive

138 the full elastic moduli could be a solution to these problems. However, the reported 139 elastic moduli of Fp17 in high-pressure IXS measurements (Antonangeli et al. 2011) 140 have been inconsistent with other experimental results (impulsive stimulated light 141 scattering, and BLS) (e.g., Crowhurst et al. 2008; Marquardt et al. 2009a, 2009b) and 142 theoretical predictions (e.g., Wentzcovitch et al. 2009; Wu et al. 2009). Thus, the 143 reliability of the IXS experimental results has been questioned because the IXS 144 technique is not a direct method to obtain velocities but uses the acoustic phonon 145 dispersions at very high frequencies to extract the velocities (Figuet et al. 2004), 146 which rises to an inevitable bias in the velocities. Insofar, there is a lack of 147 comprehensive knowledge on the elasticity of MgO and Fp at the relevant high P-T 148 conditions of the lower mantle.

149 In this study, we simultaneously measured the acoustic wave velocities and density 150 of single-crystal MgO at high P-T conditions up to ~30 GPa and 900 K using in situ 151 BLS and XRD in an EHDAC. These results were then used to evaluate the high P-T 152 effects on the sound velocities and elastic moduli of MgO. Using the obtained elastic 153 moduli of MgO at high *P*-*T* conditions as an endmember reference, together with the 154 elastic moduli of Fp from previous BLS studies, we built a comprehensive 155 thermoelastic model of Fp with up to 20 mol% FeO at the high P-T conditions 156 relevant to lower mantle. We also used these results to evaluate the effects of FeO 157 substitution on the elasticity of Fp, including the acoustic wave velocity anisotropies 158 and seismic parameters. Finally, we used our results to interpret the seismic 159 anisotropy and heterogeneities of the Earth's lower mantle.

160 EXPERIMENTAL METHODS

161 A single-crystal MgO platelet of 10 mm \times 10 mm square and 0.5 mm thick with 162 (100) crystallographic orientation was purchased from MTI Corporation, which was 163 grown by an arc melting method and natural cooling. The crystal platelet was >99.95% 164 pure with no visible inclusions or microcracking. The X-ray rocking curve of the 165 platelet showed an orientation of <100>±0.2° on the crystal surface. Analysis of the

166 XRD patterns of the crystal showed a cubic rock-salt structure with a lattice parameter 167 (*a*) of 4.2163(2) Å and a density (ρ) of 3.58(1) g/cm³ at ambient conditions.

168 To prepare our high P-T BLS and XRD experiments in an EHDAC, the platelet 169 was polished down on both sides to approximately 20-30 µm thick in the (100) 170 crystallographic orientation using 3M diamond lapping films with an average 171 grainsize of 0.3 microns. The thin polished platelet was then cleaved into several ~ 70 172 μ m \times 70 μ m square pieces. Round Re disks of 250 μ m thick and 3 mm in diameter 173 were pre-indented to 50-60 um thick using a pair of diamond anvils with 500 um 174 culets. Subsequently, a cylindrical 250 µm diameter hole was drilled in the pre-indented area and used as the sample chamber. An R-type thermocouple was used 175 176 for the high-temperature measurements of the sample chamber in the EHDAC 177 (Kantor et al. 2012; Mao et al. 2015; Sinogeikin et al. 2006; Yang et al. 2014, 2016). 178 The thermocouple was attached to one of the diamond surfaces approximately 500 µm 179 away from its culet and cladded with a ceramic adhesive (Resbond 920), which was 180 thermally conductive and electrically insulating. The EHDAC was also equipped with 181 an alumina ceramic holder that was coiled with two Pt wire strings; each was 200 µm 182 diameter and 48 cm in length (Lu et al. 2013; Mao et al. 2015; Yang et al. 2016). The 183 Pt wire heater had a resistance of approximately 1 Ω at ambient conditions. A 184 single-crystal platelet was then placed into the sample chamber, together with gold 185 (Au) powder as the pressure calibrant for the XRD experiments (Fei et al. 2007) as 186 well as a few ruby spheres as pressure indicators for the neon gas-loading (Mao et al. 187 1986) (Fig. 1a). The neon pressure medium was loaded into the sample chamber using 188 the gas loading system in the Mineral Physics Laboratory of the University of Texas 189 in Austin (Lu et al. 2013; Mao et al. 2015; Yang et al. 2016).

We conducted the high *P-T* BLS combined with XRD measurements at beamline 13-BMD of the GeoSoilEnviroConsortium for the Advanced Radiation Sources (GSECARS) of the Advanced Photon Source (APS), Argonne National Laboratory (ANL). An incident X-ray beam of 0.3344 Å wavelength focused to $7 \times 20 \ \mu m^2$

194 diameter (Sinogeikin et al. 2006) was used to confirm the platelet orientation and to 195 determine the unit cell volume of the crystal in the EHDACs. To avoid oxidation and 196 failure of the diamond anvils and Pt wires at high temperature, the EHDAC was 197 mounted inside a water-cooled holder with 150 µm thick silica glass windows, and Ar 198 with 2% H₂ gas mixture continuously flowed through the EHDAC during heating 199 (Sinogeikin et al. 2006). To minimize pressure instability for each given heating run, 200 the sample chamber was heated to a given temperature and then stabilized for at least 30 201 minutes, and its pressure was continuously monitored using XRD until the pressure 202 was sufficiently stable for the BLS experiments. The pressure was determined from 203 the measured lattice parameters of Au at high P-T using the third-order 204 Birch-Murnaghan equation of state of Au (Fei et al. 2007), and the temperature of the 205 sample chamber was from the thermocouple measurements. BLS measurements were 206 only conducted when no significant change in pressure was confirmed by the analysis 207 of the XRD patterns of the Au calibrant. To ensure the pressure stability of the 208 experiments a series of XRD patterns of the Au pressure calibrant were collected 209 before and after the BLS measurements of pressure determinations at high P-T. The 210 pressure uncertainties were determined using multiple measurements from the 211 pressures before and after the BLS measurements for each *P*-*T* point. Temperatures of 212 the sample chamber were actively stabilized within ± 1 K using the temperature-power 213 feedback program with a remotely controlled Tektronix Keithley DC power supply 214 during the experiments (Sinogeikin et al. 2006). The uncertainties of the unit-cell 215 parameters of MgO were typically in the order of 0.03% (Table 1), indicating that the 216 MgO sample remained sufficiently high quality for the XRD and BLS experiments at 217 high P-T conditions. Analysis of the XRD pattern of the sample was used to determine the density of MgO at each P-T condition before and after the BLS 218 219 measurements.

The single-crystal XRD patterns (Fig. 1b) were also taken to verify the [100] crystallographic direction of the MgO platelet before carrying out the BLS

222 measurements. Then, the motorized DAC rotation stage, which was perpendicular to 223 the focused X-ray beam, was rotated by 45° to get the [110] crystallographic direction 224 of the MgO platelet. Analysis of the XRD spots of the MgO crystal showed a 225 high-quality single crystal nature with a typical FWHM (full width at half maximum) 226 of a diffraction peak at $\pm 0.2^{\circ}$ (Fig. 1b). High *P-T* BLS spectra of the single-crystal 227 MgO along the [100] and [110] crystallographic directions as well as XRD spectra were collected up to ~30 GPa at selected temperatures of 300 K, 500 K, 700 K, and 228 229 900 K (Figs. 2 and 3, and Table 1). The Brillouin system was equipped with a 230 Coherent Verdi V2 solid-state laser with a wavelength of 532 nm, and a laser head 231 power of 0.4 W was used for the BLS measurements. BLS spectra were collected in 232 the symmetric forward scattering geometry with an external scattering angle of 50° 233 using a six-pass Sandercock-type piezoelectrically scanning tandem Fabry-Pérot 234 interferometer (Sandercock 1982). The interferometer was equipped with a Perkin 235 Elmer MP983 photocounting module with a low dark count rate of <2 counts/s at 236 room temperature (Sinogeikin et al. 2006). The laser beam focused on the sample 237 position was approximately 15 µm in diameter. The external scattering angle of the 238 system was calibrated using silicate glass and distilled water standards (Ostwald et al. 239 1977; Polian et al. 2002). The acoustic $V_{\rm P}$ and $V_{\rm S}$ velocities of the sample were 240 derived from the analysis of the measured Brillouin frequency shift as follows:

241
$$V_{P,S} = (\lambda_0 \Delta v_B)/2 \sin(\theta/2)$$
 (1)

where $V_{P,S}$ is the acoustic compressional or shear wave velocity, respectively, λ_0 is the incident laser wavelength of 532 nm, Δv_B is the Brillouin frequency shift, and θ is the external scattering angle of 50°.

245 **RESULTS**

Our collected BLS spectra along the [100] and [110] crystallographic directions showed strong V_P and V_S peaks with high signal-to-noise ratios at high *P*-*T* conditions (Fig. 2). Although two polarized V_S peaks were expected to exist, only one V_S peak

249 was observed in our study. This is a result of the polarization of the $V_{\rm S}$ in the (100) 250 crystallographic orientation of the MgO crystal interacting with the incident laser. 251 Brillouin signals of the neon pressure medium were also observed at pressures below 252 \sim 8 GPa, but they were too weak to be visible in the BLS spectra when the pressure 253 was increased above approximately 8 GPa (Fig. 2). Analysis of the V_P and V_S 254 velocities of MgO as a function of pressure at each given temperature showed that the 255 $V_{\rm P}$ along the [100] and [110] directions as well as $V_{\rm S}$ along the [110] direction 256 (hereafter written as $V_{\rm P}[100]$, $V_{\rm P}[110]$, and $V_{\rm S}[110]$, accordingly) increased with 257 increasing pressure and decreased with increasing temperature. However, Vs along the 258 [100] (hereafter written as $V_{\rm S}$ [100]) remained almost unaffected by increasing P-T up 259 to \sim 30 GPa and 900 K (Fig. 4). These observations indicate strong *P*-*T* effects on the 260 *V*_S anisotropy.

The elastic tensors of cubic MgO can be fully described by three independent elastic moduli, C_{11} , C_{12} , and C_{44} . The use of the (100) platelets in this study permits determination of the elastic moduli (C_{11} , C_{12} , C_{44}) using the velocity determined in the [100] and [110] directions (Fig. 5), respectively, together with the measured density (Table 2). The equations used in the analysis were derived from Christoffel's equation (Grimsditch 2001):

$$267 \quad \rho V^2_{\rm P}[100] = C_{11} \tag{2}$$

$$268 \quad \rho V^2 {}_{\rm S}[100] = C_{44} \tag{3}$$

269
$$\rho V_{P}^{2}[110] = (C_{11} + 2C_{44} + C_{12})/2$$
 (4)

270
$$\rho V^2 {}_{\rm S}[110] = (C_{11} - C_{12})/2$$
 (5)

where ρ is the density of MgO; $V_P[100]$ and $V_S[100]$ are the measured compressionaland shear-wave velocities of MgO along the [100] direction, respectively; $V_P[110]$ and $V_S[110]$ are the compressional- and shear-wave velocities along the [110] direction, respectively; and C_{11} , C_{12} , and C_{44} are the compressional, off-diagonal, and shear moduli of MgO, respectively. We used the least squares regression method

276 utilizing the aforementioned four equations instead of only three equations for the 277 modeling because examination of equations (2-5) showed that the three elastic moduli 278 (C_{11} , C_{12} , and C_{44}) could be determined more accurately. The uncertainties of the 279 elastic moduli (Table 2) were estimated using standard error propagation in these 280 derivations. We should note that these equations were derived from the general 281 Christoffel's equation (Every1980):

$$282 \quad \left|\Gamma_{ij} - \rho V^2 \delta_{ij}\right| = 0 \tag{6}$$

where ρ is the density from XRD measurements, *V* is the measured acoustic velocity, δ_{ij} is the Kronecker delta, and Γ_{ij} is the coefficient in the Christoffel matrix. The values of the Christoffel coefficients (Γ_{ij}) depend on the single-crystal constants (C_{ij}) in the reduced Voigt notation, where the propagation direction of the sound velocity is described by the direction cosines of the acoustic wavevector, n_i .

Using the derived elastic moduli of MgO, we calculated the aggregate adiabatic bulk and shear moduli (K_s and G) (Table 1 and Fig. 6 a, b) using the Voigt-Ruess-Hill averages (Hill 1952):

291
$$K_{\rm S} = C_{11} - 2C/3$$
 (7)

292
$$G = [(C/5 + 3C_{44}/5) + 5C_{44}C/(4C_{44} + 3C)]/2$$
 (8)

293
$$C=C_{11}-C_{12}$$
 (9)

The derived K_{S0} and G_0 at ambient conditions were 162.8(±0.2) GPa and 130.3(±0.2) GPa, respectively. The aggregate velocities of MgO (Fig. 6 c, d) were calculated using the equations:

297
$$V_{\rm P} = \sqrt{(K_{\rm S} + 4G/3)/\rho}$$
 (10)

$$V_{\rm S} = \sqrt{G/\rho} \tag{11}$$

299	The pressure and temperature derivatives of the elastic moduli were obtain	ned by
300	fitting the experimental data using the third-order finite strain equations (Tables	s 3 and
301	4) (Birch 1978; Duffy and Anderson 1989):	
302	$C_{ij}(T) = (1+2f)^{7/2} [C_{ij0}(T) + a_1 f] + a_2 P$	(12)
303	$C_{ij0}(T) = C_{ij0}(300K) + (T-300)(\partial C_{ij}/\partial T)_{\rm P}$	(13)
304	$a_1 = 3K_{T0}(T)[(\partial C_{ij}/\partial P)_T - a_2] - 7C_{ij0}(T)$	(14)

$$305 \quad f=(1/2)[(V_0(T)/V)^{2/3}-1] \tag{15}$$

306
$$V_0(T) = V_0(300K)exp \int_{300}^T \alpha \, dT$$
 (16)

$$307 \qquad (\partial C_{ij}/\partial P)_{T} = (\partial C_{ij}/\partial P)_{300K} \exp \int_{300}^{T} \alpha \, dT \tag{17}$$

308 where P is the measured pressure; $C_{ij0}(300 \text{ K})$ is the elastic modulus at ambient 309 conditions; $C_{ii0}(T)$ is the elastic modulus at temperature T and 1 bar; $C_{ii}(T)$ is elastic 310 modulus at temperature T and high pressures; $(\partial C_{ij}/\partial T)_P$ is the temperature derivative 311 of the elastic modulus at constant P and ambient T; $(\partial C_{ij}/\partial P)_T$ is the pressure 312 derivative of the elastic modulus at constant T and ambient P; $(\partial C_{ij}/\partial P)_{300K}$ is the 313 pressure derivative of the elastic modulus at ambient conditions; a_2 equals 3 for C_{11} 314 and 1 for C_{12} and C_{44} ; and f is the finite strain; V is the volume at high P-T determined 315 by single-crystal XRD; $V_0(T)$ is the volume at 1 bar and high temperature T; $V_0(300K)$ 316 is the volume at ambient conditions; and α is the thermal expansion coefficient at 317 ambient conditions. $K_{T0}(T)$ is the isothermal bulk modulus at temperature T and 1 bar, 318 and can be calculated as:

319
$$K_{\rm T0} = K_{\rm T0} (300 \text{ K}) + (\partial K_{\rm T} / \partial T)_{\rm P} (T-300)$$
 (18)

$$320 \quad K_{\rm T0}(300) = K_{\rm S0}(300 \, \text{K}) / (1 + \alpha \gamma \times 300) \tag{19}$$

$$321 \quad (\partial K_{\rm T}/\partial T)_{\rm P} \cong (\partial K_{\rm S}/\partial T)_{\rm P} / (1 + \alpha \gamma T) - K_{\rm S}({\rm T}) / (1 + \alpha \gamma T)^2 [\alpha \gamma + \gamma T \times (\partial \alpha / \partial T)]$$

$$(20)$$

322 where α is the thermal expansion coefficient at ambient conditions, and γ is the 323 Grüneisen parameter at ambient conditions with a literature value of 1.442 (Tange et

al. 2009). The temperature derivative of $K_{\rm S}$, $(\partial K_{\rm S}/\partial T)_{\rm P}$, and the pressure derivative of $K_{\rm S}$, $(\partial K_{\rm S}/\partial P)_{\rm T}$, were derived by fitting $K_{\rm S}$ at high *P*-*T* using the third-order finite strain equations (Tables 3 and 4):

$$327 K_{S} = K_{S0} (1 + 2f)^{5/2} \{ 1 + [3(\partial K_{S} / \partial P)_{T} - 5]f \} (21)$$

328
$$K_{\rm S0} = K_{\rm S0}(300 \text{K}) + (\partial K_{\rm S}/\partial T)_{\rm P}(T-300)$$
 (22)

329
$$(\partial K_{\rm S}/\partial P)_{\rm T} = (\partial K_{\rm S}/\partial P)_{300} \exp \int_{300}^{T} \alpha \, dT$$
 (23)

where $K_{S0}(300 \text{ K})$ is the adiabatic bulk modulus at ambient conditions, $K_{S0}(T)$ is the adiabatic bulk modulus at temperature *T* and 1 bar, $K_S(T)$ is the adiabatic bulk modulus at temperature *T* and high pressures, and $(\partial K_S / \partial P)_{300K}$ is the pressure derivative of K_S at ambient conditions. We also constrained the pressure and temperature derivatives of the shear modulus *G* of MgO using the third-order finite strain equations:

336
$$G(T) = (1+2f)^{5/2} [G_0(T) + b_1 f]$$
 (24)

337
$$G_0(T) = G_0(300 \text{K}) + (\partial G / \partial T)_P (T-300)$$
 (25)

$$338 b_1 = 3K_{S0}(T)(\partial G/\partial P)_T - 5G_0(T) (26)$$

339
$$(\partial G/\partial P)_{\mathrm{T}} = (\partial G/\partial P)_{300\mathrm{K}} \exp \int_{300}^{T} \alpha \, dT$$
 (27)

where $G_0(300\text{K})$ is the shear modulus at ambient conditions; $G_0(T)$ is the shear modulus at temperature *T* and 1 bar; G(T) is the shear modulus at temperature *T* and high pressures; $(\partial G/\partial P)_T$ and $(\partial G/\partial T)_P$ are the pressure and temperature derivatives of the shear modulus, respectively, and $(\partial G/\partial P)_{300\text{K}}$ is the pressure derivative of the shear modulus at ambient conditions (Tables 3-4).

345 **DISCUSSION**

346 The single-crystal elasticity of MgO at high *P-T*

347 By combining the BLS and single-crystal XRD measurements, we derived the full

348 single-crystal elasticity of MgO at simultaneously high *P*-*T* conditions, including all

349 of the elastic moduli (C_{ii} s, K_s , and G) (Figs. 5 and 6, Tables 1 and 2). All C_{ii} s and K_s 350 exhibit a linear increase with pressure at the given temperatures (Figs. 5 and 6a), 351 whereas G shows a slight pressure-dependent concave curvature (Fig. 6b). Compared 352 with the values from ultrasonic interferometry and BLS experiments, the derived pressure derivatives for the compressional moduli C_{11} , off-diagonal moduli C_{12} , and 353 354 bulk modulus $K_{\rm S}$ at ambient conditions in this study are slightly lower than the 355 corresponding values in the literature (Jackson and Niesler 1982; Kono et al. 2010; Li 356 et al. 2006; Sinogeikin and Bass 2000; Yoneda 1990), except the pressure derivative 357 of C_{12} and K_S from previous BLS studies (Sinogeikin and Bass 2000; Zha et al. 2000), which are consistent with the results in this study. This indicates an enhanced 358 359 temperature effect on these elastic moduli at high pressures. In addition, the pressure 360 derivatives of the shear moduli (C_{44} and G) at ambient conditions are 361 indistinguishable from the values in previous studies within experimental 362 uncertainties (Tables 3 and 4), except a larger pressure derivative of G at ambient conditions was reported by Zha et al. (2000) and a smaller pressure derivative of C_{44} 363 364 was reported by Sinogeikin and Bass (20000). These discrepancies potentially due to 365 the limited experimental pressure range or the possible distortion or deformation of 366 MgO under high-pressure condition. At high temperature, all the elastic moduli 367 follow similar trends with pressures (Figs. 5 and 6): increasing temperature offsets of 368 C_{ijs} to a lower value at a given pressure, with the exception of those for the off-diagonal modulus C_{12} , which remain almost constant up to ~30 GPa and 900 K 369 (Fig. 5). The temperature derivatives of the elastic moduli derived from our 370 371 simultaneously high P-T measurements are consistent with literature experimental and 372 theoretical results within their uncertainties (Tables 3 and 4). In particular, our $(\partial G/\partial T)_{\rm P}$ is -0.020(1) GPa/K, which is in excellent agreement with another result 373 (-0.020(1) GPa/K) from high P-T BLS measurements on polycrystalline MgO at 374 375 T=2700 K and P up to 68.4 GPa (Murakami et al. 2012). However, our $(\partial K_{\rm S}/\partial T)_{\rm P}$ is 376 -0.025(1) (GPa/K), whose absolute value is slightly larger than those of most

377 literature values (Isaak et al. 1989; Karki et al. 1999; Kono et al. 2010; Sinogeikin et 378 al. 2000; Sumino et al. 1983), except the computational value of -0.028 (GPa/K) by 379 Matsui et al. (2000). It is also worth noting that the temperature derivative of our C_{12} 380 is very consistent with an earlier reported value obtained at 1 bar and high 381 temperature (Sinogeikin et al. 2000), but it is substantially lower than previous 382 rectangular parallelepiped resonance (RPR) experimental results (Isaak et al. 1989; 383 Sumino et al. 1983) and a computational result (Karki et al. 1999). These 384 discrepancies may be caused by the potential systematic errors in different 385 experimental or theoretical methods.

386 Compressional wave velocity and shear wave splitting anisotropy of MgO

387 The elastic wave anisotropy of minerals expresses the difference in stiffness of a 388 structure in different crystallographic directions, and shear wave anisotropy can be an 389 indicator of the stability of minerals (Sinogeikin and Bass 2000). Knowledge of the velocity anisotropy of minerals at high P-T conditions may help to explain the seismic 390 391 anisotropy within the Earth's interior (Long and Becker 2010; Mainprice 2015). To 392 understand the evolution of the velocity anisotropy of MgO at high P-T conditions, 393 the $V_{\rm P}$ and $V_{\rm S}$ velocities at different propagation directions and anisotropy 394 distributions were calculated using our derived C_{ii} s and density at each given P-T395 point (Mainprice 1990; Mainprice et al. 2000). The percentage anisotropy for $V_P(AV_P)$ 396 is defined here as:

397
$$AV_{\rm P} = (V_{\rm P,max} - V_{\rm P,min}) / (V_{\rm P,max} + V_{\rm P,min}) \times 200 \%$$
 (28)

398 where $V_{P,max}$ and $V_{P,min}$ represent the maximum and minimum V_P velocities, 399 respectively. The polarization anisotropy factor of V_S (AV_S), also called the shear 400 wave splitting factor, is the anisotropy percentage of the two V_S in a given direction 401 and is defined as:

402
$$AV_{\rm S} = (V_{\rm S1} - V_{\rm S2}) / V_{\rm S} \times 100 \%$$
 (29)

403 where V_{S1} and V_{S2} are the two orthogonally polarized V_S velocities in the given 404 propagation direction and $V_{\rm S}$ represents the aggregate shear wave velocity. A 405 contoured upper hemisphere stereogram of $V_{\rm P}$ and $AV_{\rm S}$ for MgO at two representative 406 *P-T* conditions from our experiments is shown in Figure 7. These results show that, 407 even though MgO is highly anisotropic at ambient conditions, with an AV_P of 11.0% 408 and $AV_{\rm S}$ of 21.5% (Fan et al. 2015; Mainprice 2015), $AV_{\rm P}$ reduces to 7.1% and $AV_{\rm S}$ to 409 14.1% at 7.17 GPa and 500 K in this study (Fig. 7). $AV_{\rm P}$ and $AV_{\rm S}$ decrease to 0.5% and 1.2%, respectively, at the maximum P-T conditions of 29.63 GPa and 900 K in 410 411 our experiments, showing that MgO is almost elastically isotropic at such high P-T 412 conditions (Fig. 7). These results clearly show that high *P*-*T* have a significant impact 413 on the elastic wave velocity anisotropy of MgO.

414 Modeling the elasticity of (Mg,Fe)O ferropericlase at high *P-T* conditions

415 MgO is an endmember of (Mg,Fe)O ferropericlase, the second most abundant 416 mineral in the lower mantle. Since the (Mg,Fe)O solid solution system crystallizes in 417 the rock-salt structure at relevant lower-mantle P-T conditions, it is possible to use our 418 MgO results as an endmember reference for understanding the elasticity of the 419 (Mg,Fe)O system at high P-T conditions. In order to model the elasticity of Fp with 420 different FeO contents at high P-T conditions, we initially compared our MgO results 421 to literature elasticity values of Fp (Fp1, Fp6, Fp10) derived from Brillouin 422 measurements (Marquardt et al. 2009a; Reichmann et al. 2008; Yang et al. 2016) 423 (Figs. 8 and 9). Considering the limited available single-crystal elastic moduli results 424 of Fp from other methods (Antonangeli et al. 2011; Crowhurst et al. 2008; Jacobsen et 425 al. 2002), and avoiding the potential systematic errors in different experimental 426 methods (for example, incompatible results by IXS methods with other available 427 studies; the possible effect on the pressure derivatives of elastic moduli by ultrasonic 428 methods vs Brillouin scattering), we only used literature Brillouin results in our 429 modeling. We also limited the modeling to Fp with up to 20 mol% FeO because of the 430 limited Fp compositions in the literature (Marguardt et al. 2009a; Reichmann et al.

2008; Yang et al. 2016) and the relevant Fp composition of approximately Fp20 in the
Earth's lower mantle (Andrault 2001; Irifune et al. 2010; Mao et al. 1997; Ringwood
1975; Wood 2000).

Our analysis of the single-crystal elastic moduli (C_{ij}) of MgO as a function of the FeO content in mol% at ambient conditions showed that the substitution of FeO in Fp had a positive effect on C_{12} and a negative effect on C_{11} and C_{44} . Within experimental uncertainties, C_{12} linearly increased with FeO content, while C_{11} and C_{44} decreased linearly with increasing FeO content (Fig. 8):

439
$$C_{11} = 296.8(\pm 1.8) - 184.5(\pm 30.4) X_{\rm Fe}$$
 (30)

440
$$C_{12} = 95.6(\pm 0.7) + 71.3(\pm 12.6) X_{\rm Fe}$$
 (31)

441
$$C_{44} = 152.7(\pm 0.2) \cdot 108.4(\pm 2.9) X_{\rm Fe}$$
 (32)

442 where $X_{\text{Fe}} = M_{\text{FeO}} / (M_{\text{FeO}} + M_{\text{MgO}})$, X_{Fe} is the mole fraction, and M_{MgO} and M_{FeO} are the 443 molar content. In addition, Jacobsen et al. (2002) reported the effects of FeO 444 substitution on the C_{ij} values of (Mg,Fe)O using ultrasonic interferometry with 445 high-frequency shear waves at ambient conditions. They fitted their C_{ii} values of the 446 MgO-FeO system using a second-order or a third-order polynomial. A close 447 inspection of the polynomial fits showed that there was a close linear relationship 448 between the C_{ij} and FeO content when the FeO content in Fp is less than 20 mol% (as 449 shown by the dashed lines in Fig. 8). Thus, we used the linear relationship between 450 the C_{ij} and FeO content of Fp at ambient conditions as a fitting parameter in our 451 modeling here.

We also evaluated the high *P*-*T* relationship between the C_{ij} and FeO content by comparing the pressure and temperature derivatives of the single-crystal elastic moduli (C_{ij}) of MgO [$(\partial C_{ij}/\partial P)_{300K}$ and $(\partial C_{ij}/\partial T)_P$] to the literature results of Fp (Fp6, Fp10) (Marquardt et al. 2009a; Yang et al. 2016). Analysis of these results showed that within experimental uncertainties, $(\partial C_{11}/\partial P)_{300K}$ and $(\partial C_{12}/\partial P)_{300K}$ increased

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457 linearly with increasing FeO content, but $(\partial C_{44}/\partial P)_{300K}$ decreased linearly with 458 increasing FeO content (Fig. 9a):

459
$$(\partial C_{11}/\partial P)_{300K} = 8.43(\pm 0.13) + 2.68(\pm 1.89)X_{\text{Fe}}$$
 (33)

460
$$(\partial C_{12}/\partial P)_{300K} = 1.46(\pm 0.09) + 2.00(\pm 1.22)X_{Fe}$$
 (34)

461
$$(\partial C_{44}/\partial P)_{300K} = 1.25(\pm 0.01) - 4.31(\pm 0.09) X_{Fe}$$
 (35)

462 Since the experimental temperature derivatives of $C_{ij} [(\partial C_{ij}/\partial T)_P]$ of Fp were only 463 available for Fp6 (Yang et al. 2016), we calculated the $(\partial C_{ij}/\partial T)_P$ of Fp10 using the 464 C_{iis} at ambient conditions obtained by Marquardt et al. (2009a) and C_{iis} at high 465 temperatures up to 4000 K obtained by Muir and Brodholt (2015). The results above 466 were then used to estimate the effect of FeO on the $(\partial C_{ij}/\partial T)_P$ of Fp. Within 467 uncertainties, these results showed that $(\partial C_{11}/\partial T)_P$ and $(\partial C_{12}/\partial T)_P$ increased with 468 increasing FeO content, while $(\partial C_{44}/\partial T)_{\rm P}$ decreased with increasing FeO content. 469 Therefore, we fitted the data with a linear relationship between the $(\partial C_{ij}/\partial T)_P$ and FeO 470 content in Fp (Fig. 9b):

471
$$(\partial C_{11}/\partial T)_{\rm P} = -0.063(\pm 0.003) + 0.11(\pm 0.03) X_{\rm Fe}$$
 (36)

472
$$(\partial C_{12}/\partial T)_{\rm P} = 0.0010(\pm 0.0006) + 0.08(\pm 0.01)X_{\rm Fe}$$
 (37)

473
$$(\partial C_{44}/\partial T)_{\rm P} = -0.01(\pm 0.001) - 0.10(\pm 0.02) X_{\rm Fe}$$
 (38)

These derived linear relationships for the elastic moduli and their *P*-*T* derivatives were then combined with finite-strain theory equations (equations 7-23) to evaluate the full elasticity of Fp at high *P*-*T* conditions. We should emphasize that these linear equations can only be used within the solid solution range of approximately 0-20 mol% FeO. Evaluation of the relationship between the elastic moduli and a higher FeO content at high *P*-*T* conditions in the (Mg,Fe)O system will require further experimental data in the future.

481 FeO effects on the velocity profiles of Fp in the Earth's lower mantle

482 The Earth's mantle is generally believed to have a pyrolitic composition (Ringwood 483 1975). Based on the pyrolite model, Earth's lower mantle consists of approximately 484 75 vol% Bgm, 20 vol% Fp, and 5 vol% Ca-silicate perovskite (Ca-Pv) (Irifune 1994; 485 Irifune et al. 2010; Ricolleau et al. 2009; Ringwood 1975). Although the exact 486 composition of Earth's lower mantle has yet to be unambiguously determined, the 487 current consensus is that Fp contains approximately 20 mol% FeO, while Bgm 488 contains 10 mol% FeO, together with a few percents of Al₂O₃ in Bgm (e.g., Irifune et 489 al. 2010; Lin et al. 2013; Lin and Tsuchiya 2008; Mattern et al. 2005). The spin 490 transition in both Fp and (Al,Fe)-bearing Bgm can also affect the iron partitioning 491 between these phases (e.g., Irifune et al. 2010; Xu et al. 2017).

492 To better understand the influence of FeO on the velocity behavior of Fp, we 493 calculated its velocity profiles using the Voigt-Reuss-Hill averages based on our high 494 P-T elasticity modeling results for MgO and Fp (Fp10 and Fp20) along an expected mantle geotherm (Katsura et al. 2010) (Fig. 10). These modeled results were then 495 496 compared with the velocity profiles of Bgm10 (with FeO content of 10 mol%) 497 (Chantel et al. 2012; Murakami et al. 2007), Ca-Pv (Kudo et al. 2012), and the 498 Preliminary Reference Earth Model (PREM) (Dziewonski and Anderson 1981) (Fig. 499 10). Briefly, the third-order Eulerian finite-strain equations and the third-order Birch-Murnaghan equation of state (Birch 1978) were used to evaluate the $K_{\rm S}$ and G 500 501 of the relevant minerals by extrapolating the experimentally-derived elastic moduli 502 and their P-T derivatives to relevant P-T conditions. The reference P-T conditions at 503 670 km deep were ~23.8 GPa and ~1980 K, and the modeling depth was limited to 504 1250 km (mid-lower mantle) to avoid the complication of the potential effects of the 505 iron electronic spin transition (e.g., see Lin et al. 2013 for a review).

506 Our modeled velocity profiles showed that substitution of FeO in Fp could 507 significantly reduce both the V_P and V_S velocities at lower mantle conditions; 508 compared to MgO, the V_P and V_S velocities of Fp20 were reduced by ~8% and ~10 %, 509 respectively. That is, $(\partial V_P / \partial X_{Fe})$ and $(\partial V_S / \partial X_{Fe})$ are approximately -0.04 and -0.03

510 $km/s/X_{Fe}$, respectively, in the upper part of the lower mantle. Furthermore, the V_P and 511 $V_{\rm S}$ velocity profiles of Fp20 are ~8% and ~4% lower than the velocity profiles of the 512 PREM, respectively. On the other hand, Bgm10 has the highest $V_{\rm P}$ profile and MgO 513 has the highest V_S profile among all lower-mantle minerals, while Fp20 has the lowest 514 $V_{\rm P}$ profile and Ca-Pv has the lowest $V_{\rm S}$ profile. The depth-dependent velocity profiles 515 of a volume-weighted pyrolitic aggregate for the Earth's lower mantle, which was 516 obtained from the arithmetic mean aggregation of the compositions (75 vol% Bgm10, 517 20 vol% Fp20, and 5 vol% Ca-Pv) (Irifune et al. 2010; Ringwood 1975), were 518 extremely consistent with velocity profiles of the PREM within uncertainties.

519 **IMPLICATIONS**

520 Seismic wave anisotropy of Fp in the Earth's lower mantle

521 Seismic studies of the Earth's mantle have reported seismic anisotropy that could 522 result from the crystallographic lattice preferred orientation (CPO) of the constituent 523 minerals with velocity anisotropy (Blackman and Kendall 2002; Karato et al. 2008; 524 Marquardt and Morales 2012; Marquardt et al. 2009b; Nippress et al. 2004; Niu and 525 Perez 2004; Walpole et al. 2017). The CPO of the constituent minerals could develop 526 due to mantle convection (Boneh and Skemer 2014; Karato 1998; Long and Becker 527 2010; Mainprice 2015; Tackley 2000). Therefore, understanding the seismic 528 anisotropy of the Earth's lower mantle can greatly help us understand the geodynamics of the deep Earth (Long and Becker 2010; Mainprice et al. 2000; 529 530 Romanowicz and Wenk 2017). Specifically, the viscosity of Fp seems to be 531 pressure-dependent (Marquardt and Miyagi 2015), but Fp is still among the weakest 532 rheological phases in the lower mantle and can significantly contribute to the seismic 533 anisotropy of the lower mantle (e.g., Immoor et al. 2018; Marquardt et al. 2009b; 534 Miyagi and Wenk 2016). The elastic anisotropy of Fp combined with its rheological 535 weakness can play a significant role in the interpretation of the observed seismic 536 anisotropy in the lower mantle (e.g., Yamazaki and Karato 2001; Yamazaki et al. 537 2009; Marquardt and Miyagi 2015; Girard et al. 2016).

538 To decipher the potential influence of Fp on the lower-mantle seismic anisotropy, 539 we calculated the $AV_{\rm P}$ and $AV_{\rm S}$ of Fp10 and Fp20 along a representative geotherm 540 using the aforementioned thermoelastic modeling method (Fig. 11). At a 670 km 541 depth, the AV_P and AV_S of Fp20 are 3.7% and 8.86%, respectively, and the difference 542 between the vertically V_{SV} and horizontally V_{SH} polarized shear waves is 0.45 km/s. 543 The $AV_{\rm P}$ and $AV_{\rm S}$ of Fp10 are 4.4% and 9.87%, respectively, and the difference 544 between the orthogonally polarized shear waves $V_{\rm SV}$ and $V_{\rm SH}$ is 0.55 km/s. However, 545 at a depth of ~1250 km, the AV_P is 5.0%, AV_S is 11.42% for Fp20, and the difference 546 between $V_{\rm SH}$ and $V_{\rm SV}$ is 0.66 km/s. The $AV_{\rm P}$ is 3.7% and $AV_{\rm S}$ is 8.09% for Fp10, and 547 the difference between $V_{\rm SH}$ and $V_{\rm SV}$ is 0.51 km/s. These results showed that Fp was clearly V_P and V_S anisotropic in the upper regions of the lower mantle and both the 548 549 increase of Fe content in Fp and the increasing depth from 670 km to 1250 km could 550 change the $V_{\rm P}$ and $V_{\rm S}$ anisotropies of Fp. We should caution that the extrapolated 551 velocity anisotropies were only for the high-spin Fp, and did not take the potential 552 spin transition effects on the elasticity into account. The spin transition of iron has 553 been reported to significantly enhance the elastic shear anisotropy of Fp at high 554 pressure and room temperature (Marquardt et al. 2009b; Yang et al. 2015), making 555 low-spin Fp much more anisotropic than Bgm (the most abundant lower mantle 556 mineral) in the deeper parts of the lower mantle.

557 Origins of seismic lateral heterogeneities at the Earth's lower mantle

558 Seismic lateral heterogeneities have been detected in the lower mantle (e.g., Karato 559 and Karki 2001; Li 2009; Masters et al. 2000; Schumacher et al. 2018; van der Hilst 560 and Kárason 1999; Wu and Wentzcovitch 2017). However, their interpretation is still 561 uncertain. While some studies have suggested that the seismic heterogeneities of the 562 lower mantle was mainly due to thermal effects (e.g., Forte 2000; Forte et al. 1994; 563 Simmons et al. 2010), many other studies have suggested chemically-induced 564 heterogeneities (e.g., Antolik et al. 2003; Garnero and McNamara 2008; Kaneshima 565 and Helffrich 2010; Koelemeijer et al. 2016; Mosca et al. 2012; Resovsky and

566 Trampert 2003). Distinguishing the thermal or chemical origin of the seismic 567 heterogeneities in the lower mantle is crucial because each origin has a drastically 568 different consequence and affects our understanding of the Earth's interior dynamic 569 (Karato and Karki 2001). Experimental elasticity studies of the major minerals in the 570 lower mantle (such as Fp) at relevant high P-T conditions play a significant role in 571 understanding the seismic signatures, and also provide a new constraint on the seismic 572 structures and evolution of the lower mantle (Karato and Karki 2001; Yang et al. 573 2016).

574 The thermal or chemical origins of the seismic lateral heterogeneities can be constrained by comparing the observed ratios of various seismic parameters with 575 576 mineral physics results (e.g., Karato and Karki 2001; Wu and Wentzcovitch 2014; 577 Yang et al. 2016). To investigate the thermal variation of the heterogeneities ratios 578 $(R_{S/P}=\partial \ln V_S/\partial \ln V_P)$ for Fp, we calculated the thermally-induced variations in V_S and 579 $V_{\rm P}$ for MgO, Fp10, and Fp20 (Fig. 12a). In the modeling, the velocities of these 580 phases were first extrapolated along a representative geotherm (Katsura et al. 2010) based on the derived thermoelastic parameters and finite-strain theory. In addition, we 581 582 applied a positive and negative temperature perturbation of 200 K to the velocities to 583 determine the $\partial \ln V_{\rm S}/\partial T$ and $\partial \ln V_{\rm P}/\partial T$ for Fp. The $R_{\rm S/P}$ values of MgO, Fp10, and Fp20 584 show an increase of ~2-5% from 670 km to 1250 km, indicating that high temperature 585 increases the $R_{S/P}$ value of Fp. The $R_{S/P}$ profile of Fp20 is ~ 30% and ~50% higher than Fp10 and MgO, respectively. Therefore, our results showed that the 586 587 thermally-induced $R_{S/P}$ was sensitive to composition and the FeO content had a 588 significant increasing effect on the thermally-induced $R_{S/P}$ of Fp in the upper part of 589 the lower mantle. This also agrees with a previous theoretical prediction, which 590 showed that compositional variations could change the thermally-induced $R_{S/P}$ depth 591 profiles in the lower mantle (Wu and Wentzcovitch 2017).

592 The amount of FeO in the Earth's lower mantle minerals is particularly important 593 because it can affect a wide range of elastic properties, including their densities,

594 sound velocities, bulk and shear moduli (Jacobsen et al. 2002; Yang et al. 2015). 595 Previous studies have also suggested that the variation of FeO content in Fp may be 596 regarded as a potential chemical cause for seismic lateral heterogeneities (van der 597 Hilst and Kárason 1999). Here, we also calculated the Fe-induced variations of $V_{\rm S}$ and 598 V_P for Fp (MgO, Fp10, and Fp20) from 670 km to 1250 km along a representative 599 geotherm (Fig. 12b). Due to change of the Fe content in Fp, the value of $R_{S/P}$ shows a 600 distinctly increasing trend from ~1.92 at 670 km to ~2.21 at 1250 km (Fig. 12b), 601 which is similar with the upward trending thermally-induced variation of Fp (MgO, 602 Fp10, and Fp20) but has significantly steeper slopes. These results indicate that the 603 value of $R_{S/P}$ distinctly increases with increasing Fe content in Fp. Specifically, the 604 value of $R_{S/P}$ due to change of the Fe content (Fe-induced variation) in Fp is ~ 45% and $\sim 20\%$ higher than the thermally-induced variation of MgO and Fp10, 605 606 respectively. The value of $R_{S/P}$ due to change of the Fe content in Fp is ~ 10% lower 607 than the thermally-induced variation of Fp20. Moreover, our value of $R_{S/P}$ due to 608 change of the Fe content in Fp is also $\sim 15\%$ higher than the one from a previous 609 study by Yang et al. (2016) (dot-dashed line in Fig. 12b) within their uncertainties. 610 There are two possible reasons for the slightly different results between this study and 611 Yang et al. (2016). One of them is that we calculated the Fe-induced variations of $V_{\rm S}$ 612 and $V_{\rm P}$ for Fp with up to 20 mol% FeO, whereas Yang et al. (2016) only limited Fp 613 with up to 10 mol% FeO. Another reason is that Yang et al. (2016) calculated the 614 relative variations of velocity with a Fe content using the high pressure and room 615 temperature elasticity data of MgO (Sinogeikin and Bass 2000) as their benchmark 616 data. These factors clearly highlight the importance of our current simultaneously 617 high *P*-*T* elasticity studies of MgO.

In addition, even though some seismic studies showed a decreasing trend of $R_{S/P}$ in the upper part of the lower mantle (e.g., Mosca et al. 2012), most seismic studies demonstrated an increasing trend in the same depth region, where the $R_{S/P}$ was ~ 1.7 at the top of the lower mantle and ~2.0 in the mid-lower mantle (e.g., Robertson and

622 Woodhouse 1996; Romanowicz 2001; Simmons et al. 2010). Compared to the seismic 623 observation results (dashed lines in Fig. 12b), the $R_{S/P}$ value of the thermally-induced 624 variation for Fp10 is consistent with most seismic observations within uncertainty. However, the thermally-induced $R_{S/P}$ value of Fp20, which is the relevant composition 625 of Fp in the lower mantle, is ~50% higher than most seismic observations within 626 627 uncertainty for the upper part of the lower mantle (e.g., Houser et al. 2008; 628 Kowlemeijer et al. 2016; Mosca et al. 2012; Robertson and Woodhouse 1996; 629 Romanowicz 2001; Simmons et al. 2010). On the contrary, the value of $R_{S/P}$ due to 630 change of the Fe content in Fp is fairly consistent with some seismic observations within uncertainty in the upper part of the lower mantle (e.g., Robertson and 631 632 Woodhouse 1996; Romanowicz 2001; Simmons et al. 2010). Thus, based on the 633 results for Fp in this study, we propose that the Fe-induced heterogeneities can offer a 634 significant contribution to the observed seismic lateral heterogeneities in the lower 635 mantle. This also agrees with the results of seismology studies, which showed that the seismic heterogeneities in the lower mantle could not be explained by thermal 636 637 anomalies alone, and suggested the presence of chemical heterogeneities, might be a 638 local variation of Fe content, as an alternative explanation (e.g., Houser et al. 2008; Kellogg et al. 1999; van der Hilst and Kárason 1999; Wysession et al. 1999). We 639 640 should note that the above mentioned Fe-induced heterogeneities in the lower mantle 641 did not take the potential contributions from Bgm into account because Fe 642 preferentially partitions into Fp in the lower mantle (e.g., Auzende et al. 2008; 643 Kaminsky and Lin 2017; Muir and Brodholt 2016; Sakai et al. 2009; Xu et al. 2017) 644 and there are few single-crystal elasticity experimental studies of Bgm at high P-T 645 conditions (e.g., Sinogeikin et al. 2004b). Future single-crystal elasticity studies of Fp 646 and Bgm at the relevant P-T and compositional (the possible Fp and Bgm 647 compositions expected in the lower mantle) conditions are needed to provide a more 648 comprehensive understanding of the seismic anisotropy and lateral heterogeneities of 649 the Earth's lower mantle.

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1056 Figure Captions

Figure 1. Representative sample chamber image and X-ray diffraction image along the principal crystallographic axis [100] of single-crystal MgO platelet at high *P-T* conditions. (a) MgO (100) platelet, together with Au and a ruby sphere calibrant, in Ne medium; (b) Representative X-ray diffraction pattern along the principal crystallographic axis [100] of single-crystal MgO platelet at 11.65 GPa and 700 K. The inset shows the high-quality single crystal nature with a typical FWHM of a diffraction peak of MgO at $\pm 0.2^{\circ}$. (Color online.)

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Figure 2. Representative Brillouin spectra of single-crystal periclase MgO at high pressure and temperature. Open circles: experimental data; solid lines: fitted V_P and V_S peaks, respectively. (Color online.)

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Figure 3. Pressure-volume-temperature relations of single-crystal periclase MgO. The results are derived from analysis of X-ray diffraction measurements in EHDACs. Solid symbols represent experimental data at 300 K, 500 K, 700 K, and 900 K. Solid lines are fits to experimental data using the third-order Birch-Murnaghan equation of state. The inset shows the density of MgO as a function of pressure at the given high temperatures. (Color online.)

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Figure 4. Acoustic velocities of the single-crystal periclase MgO along the [110] and
[100] crystallographic directions as a function of pressure and temperature. Solid
circles: experimental data; solid lines: modeled results using the finite-strain theory.
(Color online.)

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Figure 5. Single-crystal elastic moduli of periclase MgO as a function of pressure and temperature. Solid lines are the fitted results using the third-order finite-strain equation. (Color online.)

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Figure 6. Aggregate elastic moduli and velocities of periclase MgO as a function of
pressure and temperature. Solid lines are fits to the experimental data using the
third-order finite-strain equation. (Color online.)

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Figure 7. Upper hemisphere pole figures of V_P anisotropy and V_S splitting anisotropy of single-crystal periclase MgO at representative pressures and temperatures. Calculations were performed using the petrophysical software UnicefCareware of Mainprice (1990) with the derived single-crystal elastic moduli from this study. (Color online.)

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Figure 8. Single-crystal elastic moduli of ferropericlase (Mg,Fe)O as a function of the FeO content at ambient conditions. Solid symbols are experimental Brillouin light scattering results for MgO (this study), Fp1 (Reichmann et al. 2008), Fp6 (Yang et al. 2016), and Fp10 (Marquardt et al. 2009). Solid lines are linear fits to the Brillouin experimental data. Black dashed lines: ultrasonic results at ambient conditions by Jacobsen et al. (2002) for comparison. (Color online.)

1101

Figure 9. Pressure and temperature derivatives of the single-crystal elastic moduli of
ferropericlase (Mg,Fe)O as a function of the FeO content at high *P-T* conditions.
Solid symbols are derived from experimental Brillouin results for MgO (this study),
Fp6 (Yang et al. 2016), and Fp10 (Marquardt et al. 2009). Solid lines are linear fits to
the Brillouin experimental data. (Color online.)

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Figure 10. Aggregate V_P and V_S of bridgmanite, ferropericlase, and Ca-perovskite in the Earth's lower mantle. These results are compared with the PREM model (Dziewonski and Anderson 1981). For the pyrolite mineralogical model, we used 75 vol% Bgm10 (Chantel et al. 2012; Murakami et al. 2007), 20 vol% Fp20 (This study),

and 5 vol% Ca-Pv (Kudo et al. 2012). Red line: Fp20; magenta lines: Fp10; cyan lines: MgO; blue lines: Bgm10; green lines: Ca-Pv; and crosses: PREM. Error bars represent the propagated uncertainties $(\pm 1\sigma)$. (Color online.)

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Figure 11. Upper hemisphere pole figures of the V_P anisotropy and V_S splitting anisotropy of Fp10 and Fp20 at 670 and 1250 km, respectively. Calculations were performed using the petrophysical software UnicefCareware of Mainprice (1990) with the modeled single-crystal elastic moduli from this study. (Color online.)

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1121 Figure 12. Depth profile of the thermally-induced (a) and Fe-induced (b) 1122 heterogeneities ratios ($R_{\rm S/P}=\partial \ln V_{\rm S}/\partial \ln V_{\rm P}$) for Fp along a representative geotherm 1123 (Katsura et al. 2010). (a) Solid blue line: thermal-induced heterogeneities ratio of 1124 MgO; solid green line: thermal-induced heterogeneities ratio of Fp10; solid red line: 1125 thermal-induced heterogeneities ratio of Fp20; (b) Solid black line: Fe-induced 1126 heterogeneities ratio of Fp; and orange dot-dashed line: Fe-induced heterogeneities 1127 ratio of Fp6 (Yang et al. 2016). Seismic models are also shown as a dashed line. 1128 Magenta dashed line: SP12RTS (Koelemeijer et al. 2016); cyan dashed line: HMSL08 1129 (Houser et al. 2008); wine dashed line: MCDRT12 (Mosca et al. 2012); red dashed 1130 line: GyPSuM (Simmons et al. 2010); blue dashed line: RW96 (Robertson and 1131 Woodhouse 1996); and green dashed line: RB01 (Romanowicz 2001). The 1132 representative error bars (standard deviation, $\pm 1\sigma$) are estimated using the standard 1133 error propagation from the modeled parameters. Vertical lines with arrows represent 1134 the differences in percentage for the $R_{S/P}$ ratios. (Color online.)

<i>T</i> (K)	P(GPa)	$V(Å^3)$	V/V_0	Ks(GPa)	G(GPa)
300	1.34(2)	74.10(4)	0.9918	168(1)	133(3)
	3.01(2)	73.37(2)	0.9821	174(2)	137(3)
	3.72(4)	73.07(5)	0.9780	177(2)	139(2)
	6.71(3)	71.87(2)	0.9620	188(2)	145(2)
	8.68(5)	71.13(5)	0.9521	195(1)	149(3)
	11.97(4)	69.97(3)	0.9366	207(2)	156(2)
	15.56(5)	68.80(3)	0.9209	219(2)	163(2)
	18.36(5)	67.95(4)	0.9095	229(2)	168(2)
	21.18(6)	67.14(3)	0.8987	239(2)	174(3)
500	2.89(2)	73.90(2)	0.9892	169(2)	133(3)
	4.95(2)	73.01(4)	0.9772	176(2)	137(3)
	7.17(2)	72.11(3)	0.9652	184(2)	142(3)
	9.56(3)	71.20(3)	0.9530	193(2)	147(3)
	11.53(2)	70.49(4)	0.9435	201(2)	151(3)
	13.69(4)	69.75(4)	0.9336	209(2)	156(3)
	15.33(3)	69.21(5)	0.9264	214(2)	159(3)
	17.30(5)	68.59(2)	0.9181	221(2)	163(3)
	18.15(4)	68.33(3)	0.9146	224(2)	164(3)
	20.03(4)	67.77(4)	0.9071	231(2)	168(3)
	22.42(5)	67.09(4)	0.8980	239(2)	173(3)
	24.92(6)	66.41(5)	0.8889	248(2)	177(3)
	26.61(4)	65.97(4)	0.8830	254(2)	181(3)
700	6.32(2)	72.95(3)	0.9764	177(1)	136(2)

Table 1 Unit cell volume, adiabatic bulk modulus, and shear modulus of MgO at high *P*-*T*

	9.11(3)	71.83(2)	0.9615	187(2)	142(3)
	11.65(3)	70.88(3)	0.9487	195(2)	147(2)
	14.33(3)	69.94(4)	0.9362	205(2)	153(3)
	16.82(4)	69.12(4)	0.9252	212(2)	157(3)
	17.90(3)	68.78(5)	0.9206	217(2)	159(3)
	19.89(5)	68.17(2)	0.9125	225(2)	164(3)
	22.60(4)	67.38(4)	0.9019	235(2)	169(3)
	24.18(5)	66.94(3)	0.8960	242(2)	173(3)
	26.80(5)	66.29(5)	0.8873	248(2)	177(3)
_	28.14(6)	65.89(4)	0.8819	255(2)	180(3)
900	6.75(2)	73.31(2)	0.9813	172(2)	132(2)
	9.66(3)	72.11(4)	0.9652	182(2)	138(3)
	11.88(3)	71.26(3)	0.9538	190(2)	143(3)
	14.75(4)	70.23(3)	0.9400	200(2)	149(3)
	18.15(5)	69.10(4)	0.9249	212(2)	156(3)
	22.45(5)	67.79(5)	0.9074	227(2)	164(3)
	26.01(4)	66.79(3)	0.8940	241(2)	172(3)
	29.63(5)	65.84(4)	0.8813	254(2)	178(3)

Table 2 Densities, sound velocities of MgO along the [100] and [110] directions, and elastic moduli as a function of pressure at 300 K, 500 K, 700 K, and 900 K

P (MgO)	Density	Vs[100]	$V_{\rm P}[100]$	Vs[110]	$V_{\rm P}[110]$	C_{11}	C_{12}	C_{44}
(GPa)	(g/cm^3)	(km/s)	(km/s)	(km/s)	(km/s)	(GPa)	(GPa)	(GPa)
				300 K				
1.34(2)	3.608(2)	6.559(7)	9.2(1)	5.391(7)	9.96(1)	308(1)	97.8(7)	155(2)
3.01(2)	3.644(3)	6.558(8)	9.4(1)	5.543(7)	10.05(2)	324(1)	99.6(9)	157(1)
3.72(4)	3.659(5)	6.567(6)	9.5(1)	5.585(8)	10.09(2)	329(1)	101(1)	158(1)
6.71(3)	3.720(3)	6.554(7)	9.8(2)	5.812(8)	10.23(2)	355(2)	104(1)	160(2)
8.68(5)	3.759(5)	6.559(9)	9.9(2)	5.931(7)	10.32(2)	371(2)	107(1)	162(2)
11.97(4)	3.821(4)	6.577(9)	10.2(1)	6.124(9)	10.49(2)	399(2)	112(1)	165(2)
15.56(5)	3.886(6)	6.579(8)	10.5(2)	6.324(7)	10.63(2)	426(2)	115(1)	168(1)
18.36(5)	3.935(5)	6.59(1)	10.7(2)	6.464(8)	10.75(2)	448(2)	120(1)	171(1)
21.18(6)	3.982(3)	6.61(1)	10.9(2)	6.612(9)	10.88(2)	471(2)	123(1)	174(1)
				500 K				
2.89(2)	3.618(2)	6.550(8)	9.2(1)	5.383(9)	9.96(2)	308(2)	98.7(8)	155(1)
4.95(2)	3.662(2)	6.554(8)	9.4(1)	5.532(8)	10.07(2)	326(2)	102(1)	157(2)
7.17(2)	3.708(3)	6.544(7)	9.6(2)	5.673(8)	10.16(2)	344(2)	105(1)	159(1)
9.56(3)	3.755(2)	6.542(9)	9.9(2)	5.849(9)	10.28(2)	364(2)	107(1)	161(2)
11.53(2)	3.793(3)	6.551(8)	10.0(1)	5.969(7)	10.38(2)	381(2)	111(1)	163(2)
13.69(4)	3.833(3)	6.57(1)	10.2(2)	6.120(7)	10.50(2)	400 (2)	113(1)	166(2)
15.33(3)	3.863(4)	6.57(1)	10.3(2)	6.186(8)	10.55(2)	411 (2)	116(1)	167(2)
17.30(5)	3.898(3)	6.573(9)	10.5(2)	6.307(9)	10.64(2)	428(2)	118(2)	168(2)
18.15(4)	3.913(5)	6.58(1)	10.5(2)	6.334(7)	10.67(2)	433(2)	119(1)	169(1)
20.03(4)	3.945(6)	6.59(1)	10.7(2)	6.452(9)	10.76(2)	450(2)	121(1)	171(1)

	22.42(5)	3.985(5)	6.60(1)	10.8(1)	6.57(1)	10.86(2)	468(2)	125(1)	173(2)			
	24.92(6)	4.026(4)	6.608(9)	11.0(2)	6.69(1)	10.96(2)	488(2)	128(1)	176(2)			
	26.61(4)	4.053(4)	6.62(1)	11.1(2)	6.77(1)	11.04(2)	502(2)	130(1)	178(2)			
-	700 K											
	6.32(2)	3.665(2)	6.549(9)	9.4(1)	5.466(8)	10.05(2)	323(2)	104(1)	157(1)			
	9.11(3)	3.722(3)	6.534(8)	9.6(2)	5.660(8)	10.18(2)	346(2)	107(1)	159(2)			
	11.65(3)	3.772(3)	6.53(1)	9.8(2)	5.814(9)	10.28(2)	365(2)	110(1)	161(1)			
	14.33(3)	3.823(2)	6.546(9)	10.1(1)	5.990(9)	10.41(2)	388(2)	113(1)	164(2)			
	16.82(4)	3.868(2)	6.549(7)	10.2(2)	6.094(7)	10.49(2)	404(2)	116(1)	166(2)			
	17.90(3)	3.887(3)	6.549(9)	10.3(2)	6.17(1)	10.55(2)	414(2)	118(1)	167(2)			
	19.89(5)	3.922(4)	6.57(1)	10.5(2)	6.33(1)	10.67(2)	435(2)	121(1)	169(1)			
	22.60(4)	3.968(3)	6.57(1)	10.7(2)	6.461(9)	10.78(3)	455(2)	124(1)	171(2)			
	24.18(5)	3.994(3)	6.60(1)	10.9(2)	6.573(8)	10.88(3)	472(2)	127(1)	174(2)			
	26.80(5)	4.033(4)	6.60(1)	11.0(2)	6.64(1)	10.94(2)	485(2)	129(1)	176(2)			
_	28.14(6)	4.058(5)	6.61(1)	11.1(2)	6.74(1)	11.03(2)	501(2)	132(1)	177(2)			
					900 K							
	6.75(2)	3.647(2)	6.532(9)	9.2(1)	5.33(1)	9.96(2)	310(2)	103(1)	156(2)			
	9.66(3)	3.708(3)	6.53(1)	9.5(2)	5.53(1)	10.10(2)	334(2)	107(1)	158(2)			
	11.88(3)	3.752(3)	6.524(9)	9.7(2)	5.667(9)	10.19(3)	350(2)	109(1)	160(2)			
	14.75(4)	3.807(4)	6.525(9)	9.9(1)	5.85(1)	10.32(3)	374(2)	113(1)	162(2)			
	18.15(5)	3.869(5)	6.54(1)	10.2(2)	6.04(1)	10.48(3)	400(2)	118(1)	166(1)			
	22.45(5)	3.944(5)	6.55(1)	10.5(2)	6.29(1)	10.66(3)	435(2)	123(2)	169(2)			
	26.01(4)	4.003(4)	6.58(1)	10.8(2)	6.500(9)	10.84(3)	467(2)	128(1)	173(2)			
	29.63(5)	4.061(3)	6.59(1)	11.0(2)	6.67(1)	10.98(3)	494(2)	133(2)	176(2)			

Table 3 Single-crystal elastic moduli and their pressure and temperature derivatives of MgO at ambient conditions

Study	Methods ^a	C_{11}	C_{12}	C_{44}	$(\partial C_{11}/\partial P)_{300\mathrm{K}}$	$(\partial C_{12}/\partial P)_{300\mathrm{K}}$	$(\partial C_{44}/\partial P)_{300\mathrm{K}}$	$(\partial C_{11}/\partial T)_{\rm P}$	$(\partial C_{12}/\partial T)_{\rm P}$	$(\partial C_{44}/\partial T)_{\rm P}$
		(GPa)	(GPa)	(GPa)				(GPa/K)	(GPa/K)	(GPa/K)
This study	BLS	297.9(6)	95.1(5)	152.8(4)	8.35(5)	1.46(2)	1.26(3)	-0.062(3)	0.001(2)	-0.010(2)
Sinogeikin and Bass (2000)	BLS	298(2)	96(1)	154(2)	9.1(2)	1.3(2)	0.8(2)	b	b	b
Sinogeikin et al. (2000)	BLS	b	b	b	b	b	b	-0.060(2)	0.001(1)	-0.015(1)
Zha et al. (2000)	BLS	297.0(1)	95.2(7)	155.7(5)	b	b	b	b	b	b
Jackson and Niesler (1982)	UI	296.8°	95.3 °	155.8 °	9.17°	1.61 °	1.11 ^c	b	b	b
Yoneda (1990)	UI	297.8 °	95.1 °	155.8(2)	8.76 °	1.81 °	1.31 °	b	b	b
Sumino et al. (1983)	RPR	297(1)	96(1)	156.2(3)	b	b	b	-0.0596 °	0.0068 ^c	-0.0122(2)
Isaak et al. (1989)	RPR	299.0(7)	96.4(6)	157.1(3)	b	b	b	-0.0585 ^c	0.0075 ^c	-0.0126 ^c
Karki et al. (1999)	DFT	300 °	94 °	147 °	b	b	b	-0.0598 °	0.0089 ^c	-0.0088 ^c

^{a)} BLS: Brillouin Light Scattering; UI: Ultrasonic Interferometer; RPR: Rectangular Parallelepiped Resonance; and DFT: Density Functional Theory.

^{b)} The value is not available in the text.

^{c)} The uncertainty is not available in the text.

References	Methods ^a	K_{S0}	$G_{ heta}$	$(\partial K_S / \partial P)_{300\mathrm{K}}$	$(\partial G/\partial P)_{300\mathrm{K}}$	$(\partial K_S / \partial T)_P$	$(\partial G/\partial T)_P$
		(GPa)	(GPa)			(GPa/K)	(GPa/K)
This study	BLS	162.8(2)	130.3(2)	3.94(5)	2.17(2)	-0.025(1)	-0.020(1)
Sinogeikin and Bass (2000)	BLS	163(1)	130(1)	4.0(1)	2.4(1)	b	b
Sinogeikin et al. (2000)	BLS	b	b	b	b	-0.019(2)	-0.024(2)
Zha et al. (2000)	BLS	162.5(7)	130(2)	3.99(3)	2.85(9)	b	b
Murakami et al. (2009)	BLS	b	130(3)	b	1.9(1)	b	b
Murakami et al. (2012)	BLS	b	b	b	b	b	-0.020(1)
Kono et al. (2010)	UI	162.0(4)	128.5(2)	4.27(3)	2.33(1)	-0.0184(4)	-0.0226(2)
Li et al. (2006)	UI	164(1)	129.8(6)	4.2(1)	2.42(6)	b	b
Yoneda (1990)	UI	162.7(2)	131.1 °	4.1(3)	2.41 ^c	b	b
Jackson and Niesler (1982)	UI	162.5(2)	130.9°	4.13(9)	2.53 °	b	b
Sumino et al. (1983)	RPR	163(1)	133.9(2)	b	b	-0.018(2)	-0.021(1)
Isaak et al. (1989)	RPR	163.9(6)	131.8(5)	b	b	-0.0145 °	-0.024 °
Karki et al. (1997)	PP	159.7 °	121.5 °	4.26 °	2.18 °	b	b
Karki et al. (1999)	DFT	162 °	128 °	4.15 °	2.44 ^c	-0.014 ^c	-0.0216 °
Matsui et al. (2000)	MD	161 °	131 °	4.1 °	2.4 °	-0.028 ^c	-0.024 ^c

Table 4 Bulk and shear moduli and their pressure and temperature derivatives of MgO at ambient conditions

^{a)} PP: Pseudopotential; and MD: Molecular Dynamic.

^{b)} The value is not available in the text.

^{c)} The uncertainty is not available in the text.



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