# 1 Revision 1

2	High-pressure granulite facies metamorphism (~1.8 GPa) revealed in silica-undersaturated
3	garnet-spinel-corundum gneiss, Central Maine Terrane, Connecticut, U.S.A.
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# 12 ABSTRACT

We quantify the metamorphic pressure-temperature (*P*-*T*) conditions for a newly 13 14 discovered silica-undersaturated high-pressure granulite (HPG) from the Central Maine Terrane 15 (CMT) in northeastern Connecticut, United States. The rocks lie within the Acadian-Neoacadian 16 orogenic belt (Devonian) and form part of the Brimfield Schist. The Brimfield and the adjacent 17 Bigelow Brook Formation contain silica-saturated rocks that have previously been shown to have 18 undergone ~1000 °C metamorphism. Pressure was less well constrained at  $\geq$  ~1 GPa. Silica-19 undersaturated rocks hold underutilized potential for pinpointing peak metamorphic conditions, 20 particularly pressure, because of their resilience to melting and the variety of refractory minerals 21 they contain. The typical silica-undersaturated mineral assemblage is garnet + spinel + corundum 22 + plagioclase + K-feldspar + biotite + ilmenite. Leucosomes are syenites consisting of two 23 feldspars  $\pm$  biotite. Plagioclase is commonly antiperthitic, particularly in feldspathic domains surrounding peritectic garnet; such garnet crystals reach  $\sim 10$  cm in diameter. Alkali feldspars are 24 25 perthitic. The rocks contain remarkable ellipsoidal spinels as much as 5.5 cm long comprising discrete crystallographic domains hosting crystallographically oriented lamellae of a Fe-Ti 26 27 phase, most likely ilmenite. Corundum is usually colorless, but can also be found as sapphire in shades of pink, purple, and blue, particularly in antiperthite-rich domains surrounding large 28 garnets. Some sapphires are concentrically color zoned. We carried out P-T estimation using 29 ternary feldspar reintegration thermometry of metamorphic antiperthites together with 30 31 pseudosection modeling. Samples texturally and chemically record near-eclogite facies equilibration at minimum conditions of  $\sim$ 1040 °C and  $\sim$ 1.8 GPa, establishing the CMT in 32 33 northeastern CT as the first known HPG locality in the United States. These results are consistent with high  $P_2O_5$  levels found in garnet (0.18 wt%), Ti-in-biotite thermometry, regional sillimanite 34

pseudomorphs after kyanite, and preliminary experimental work on melt inclusions in garnet

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36	(Ferrero et al. 2017). The leucosomes provide strong evidence that partial melting of silica-
37	undersaturated rocks at HPG conditions can produce syenitic magmata. Strongly melt-depleted
38	silica-undersaturated rocks may also be protoliths for garnet + spinel + corundum xenoliths
39	reported from kimberlites. The presence of HPG gneisses demonstrates that the large-scale
40	thrusts of the CMT sample the deepest roots of the orogenic belt (60-70 km), and perhaps even
41	deeper subduction zone lithologies as well.
42	Keywords: High-pressure granulite, silica-undersaturated, corundum, spinel, garnet
43	INTRODUCTION
44	High-pressure granulites (HPG) form at or below the base of orogenically thickened crust
45	at pressures ( <i>P</i> ) $\geq$ 1.5 GPa and temperatures ( <i>T</i> ) $\geq$ 850 °C in the stability field of kyanite,
46	commonly at eclogite facies conditions (O'Brien and Rötzler 2003). They provide textural,
47	chemical, and mineralogical records of lower crustal processes and environments, and are
48	particularly valuable tools for evaluating models of how mountain belts form, the depths to
49	which continental material can be underthrust and potentially subducted, and how rocks are
50	exhumed from deep crustal levels. Although HPGs are reported from many orogenic belts
51	worldwide (see O'Brien and Rötzler 2003, Kotková 2007, and references therein) none have
52	been identified in the United States of America. Paleoproterozoic HPGs are present in the
53	Snowbird tectonic zone of Canada (Snoeyenbos et al. 1995; Baldwin et al. 2003), and
54	Neoproterozoic HPGs (and eclogites) are present in the Grenville orogen of eastern Canada
55	(Indares 1993; Indares 1995; Indares 1997; Indares et al. 1998; Indares and Dunning 2001). The
56	lack of high-pressure granulites in the northeastern United States is particularly conspicuous, as
57	granulites are a significant component of the Acadian-Neoacadian orogen in the Central Maine

58 Terrane (Peper and Pease 1975; Fahey and Pease 1977; Chamberlain and England 1985; Schumacher et al. 1989; Robinson et al. 1998). 59 60 Recent experimental evidence suggests, however, that silicic ultrahigh-temperature 61 granulites of the Brimfield Schist and Upper Member of the Bigelow Brook Formation in northeastern Connecticut, part of the Central Maine Terrane (Ague et al. 2013; Axler and Ague 62 63 2015a) underwent early HPG metamorphism (Ferrero et al. 2017). Re-homogenization of quartz-bearing melt inclusions indicates that for  $T 1050 \,^{\circ}\text{C}$  (Axler and Ague 2015a), melt 64 entrapment occurred at  $P \ge 1.7$  GPa (Ferrero et al. 2017). These conditions exceed the canonical 65 pressure limit of UHT metamorphism (Harley 1998; Brown 2006; Kelsey 2008) and easily reach 66 67 the high-pressure granulite facies. This result is preliminary, but nonetheless offers a tantalizing glimpse of higher-pressure metamorphism than has been previously reported in the orogen. 68 69 Independent corroboration of this P estimate is necessary to establish the nature and extent of HPG metamorphism within the orogen. However, the matrix assemblage of the quartz-70 bearing granulites does not provide a robust pressure constraint other than the rutile/ilmenite 71 72 transition, yielding minimum  $P \sim 1$  GPa (Ague et al. 2013). Distinguishing ultrahigh-temperature 73 (UHT) rocks from HPGs represents a particular challenge because unless HPG indicator minerals like kyanite form and are preserved, many intermediate and felsic bulk compositions 74 crystallize quartzofeldspathic garnet-bearing assemblages that are not strongly diagnostic of 75 pressure at  $\sim 1000$  °C and 1–2 GPa. In fact, the type locality granulites from the Granulitgebirge 76 77 of Saxony, Germany were only recently recognized as HPGs, based on petrographic observations coupled with modern geothermobarometric and modeling techniques (e.g. Rötzler 1992; Rötzler 78 and Romer 2001; O'Brien 2006; Müller et al. 2015). Silica-undersaturated rocks, however, have 79

the potential to provide independent *P-T* constraints at UHT and HPG conditions (e.g. Kelsey et
al. 2005; Kelsey 2008; Dorfler et al. 2015; Guevara and Caddick 2016).

82 In particular, silica-undersaturated rocks that contain coexisting garnet, spinel, and 83 corundum hold significant promise for revealing *P*-*T* histories. Such rocks are not commonly 84 described in the literature, but are nonetheless reported from a number of field areas worldwide 85 including in India (Harris 1981; Sengupta et al. 1999; Shimpo et al. 2006), Sri Lanka (Osanai et 86 al. 2006; Dharmapriya et al. 2015), Algeria (Ouzegane et al. 2003), South Africa (Schreyer et al. 87 1984), Antarctica (Asami et al. 1989; Asami et al. 1990), the United States (Cortland complex; 88 e.g. Dorfler et al. 2014, 2015), and Canada (Snoeyenbos et al. 1995). Roof pendants in the 89 Vinalhaven pluton (Maine, U.S.A.) contain garnet, spinel, and corundum, but these minerals do not form a distinct assemblage; garnet is only present in leucosomes with quartz (Porter et al. 90 91 1999). Retrogressed eclogite from the Trans-Hudson orogen contains garnet with spinel  $\pm$ 92 corundum (+ plagioclase) symplectites after kyanite (Weller and St-Onge 2017). Kimberlite pipes have yielded bi- to tri-mineralic garnet  $\pm$  spinel  $\pm$  corundum xenoliths which can contain 93 cm-scale colored corundum and are generally interpreted as mantle cumulates or extremely melt-94 depleted metapelites sourced from slabs (Nixon et al. 1978; Padovani and Tracy 1981; Exley et 95 96 al. 1983; Mazzone and Haggerty 1989).

In this paper, we describe a newly discovered silica-undersaturated lithology from the Central Maine Terrane, Connecticut (CT), which contains porphyroblastic garnet, spinel, and corundum. The rocks are part of the same litho-tectonic rock sequence in which Ague et al. (2013) and Axler and Ague (2015a) determined extreme metamorphic  $T \sim 1000$  °C. We evaluate equilibration *P-T* conditions using ternary feldspar reintegration thermometry and pseudosection modeling in conjunction with textural observations. The results demonstrate that the rocks

103	reached HPG conditions, the first in-situ HPG example we are aware of in the United States. If
104	the silica-saturated rocks studied by Axler and Ague (2015a) and Ferrero et al. (2017) are also
105	HPG, then multiple HPG localities exist in the Central Maine Terrane. We then discuss the
106	tectonic implications of HPG rocks within the Central Maine Terrane and the potential of these
107	rock types to elucidate lower crustal processes.

## 108 GEOLOGIC SETTING

109The Brimfield Schist hosts the rocks of this study and is one of several rock units cut

and/or bounded by west-northwest-dipping thrust faults within the southern Central Maine

111 Terrane (CMT; also known as, for example, the Merrimack Synclinorium, e.g. Rodgers 1981,

and the Central Maine belt, e.g. Robinson et al. 1998). Together with the other rock units in this

area it reached or exceeded granulite facies metamorphic conditions (Tracy et al. 1976; Robinson

114 1978; Rodgers 1981; Ague et al. 2013) (Figure 1). The Brimfield Schist and its internal shear

115 zones contain a wide variety of lithologies including aluminous and quartzofeldspathic gneisses,

116 granitoids, amphibolites, ultramafic blocks, rare metacarbonate rocks, and the silica-

undersaturated rocks described herein (Ague et al. 2013).

Silica-saturated metapelitic gneisses of the Brimfield Schist and the adjacent Bigelow 118 119 Brook Formation record a UHT signature (Ague et al. 2013; Axler and Ague 2015a). These 120 rocks locally preserve a matrix assemblage with spinel and cordierite from this phase of 121 metamorphism, stable at  $\sim 0.6$  GPa. As noted above, however, recent melt-inclusion rehomogenization experiments show that a HPG stage (~1050°C; 1.7–2.0 GPa) likely pre-dated 122 UHT metamorphism in the Bigelow Brook Formation (Ferrero et al. 2017). Sillimanite 123 124 pseudomorphs after kyanite are preserved in metapelitic gneisses of the Brimfield Schist and 125 Bigelow Brook Formation (Peper and Pease 1975; Fahey and Pease 1977; Axler and Ague

126 2015a). This is consistent with overprinting of kyanite-bearing HPG assemblages by later,

127 lower-pressure, UHT metamorphism.

Additional metamorphic overprints have affected the region. A 700–800 °C, 0.5–0.6 GPa granulite facies signature was recognized by Ague et al. (2013) as post-dating the UHT signature of quartz-bearing metapelitic gneisses from the Brimfield Schist. Following this, a kyanite zone overprint is evident from metasomatic kyanite-garnet-quartz-carbonate veins that cut across several lithologies within the Brimfield Schist, together with other regional occurrences of

texturally late kyanite (Schumacher et al. 1989; Ague 1995; Thomson 2001).

Field relationships and geochronology document Acadian (~420–380 Ma) and

135 Neoacadian (~360 Ma) metamorphism in the CMT (e.g. Rodgers 1981; Robinson and Tucker

136 1982; Schumacher et al. 1989; Armstrong et al. 1992; Getty and Gromet 1992; Wintsch et al.

137 1992, 2009; Robinson et al. 1998; Thomson 2001; Massey et al. 2017). Both of these events are

138 recognized in metamorphic zircon from the Brimfield Schist and Bigelow Brook Formation, with

the Neoacadian signal being dominant (Axler and Ague 2015c; Axler, unpublished PhD thesis).

140 Rodgers (1981) suggested that the southern CMT originated in an accretionary prism formed

141 during westward subduction before the Acadian collision, but this interpretation is not

142 universally accepted (Robinson and Tucker 1982).

143Elsewhere in the eastern United States, silica-undersaturated metapelitic or metagranitic

rocks are found only in a handful of localities, including the Ordovician Cortlandt Complex (e.g.

145 Rogers 1911; Barker 1964; Dorfler et al. 2014; Dorfler et al. 2015), the Silurian Cadillac

- 146 Mountain granite (Nichols and Wiebe 1998), and the Siluro-Devonian Calderwood Neck
- 147 Pendant in the Vinalhaven Pluton (Porter et al. 1999).

## 148 ANALYTICAL METHODS

149 Quantitative mineral chemistry and backscattered electron (BSE) images were collected 150 with a JEOL JXA-8530F field emission gun electron probe microanalyzer (EPMA) in the Yale 151 University Department of Geology & Geophysics. Wavelength dispersive spectroscopy (WDS) was used for all mineral chemistry and elemental maps; quantitative analyses employed natural 152 153 and synthetic standards and off-peak background corrections. Operating conditions were 15 kV 154 accelerating voltage and 10.8 mm working distance with a focused beam. Beam current was 10 155 nA for biotite and feldspars, 20 nA for oxides, and 50 nA for garnet. Element mapping used a 156 beam current of 20 nA and a dwell time of 100 ms for corundum, 200 nA and 100 ms for lower-157 resolution garnet maps and the spinel map, and 300 nA and 200 ms for the high-resolution garnet 158 phosphorus map. Spinel host and oxide lamellae were reintegrated using an evenly spaced grid 159 of 24 unfocused (25 µm) beam spots.

160 Ternary feldspar compositions for leucosome and matrix antiperthites were reconstructed 161 following Ague et al. (2013). Compositions of host and lamellae were measured via EPMA, and 162 (BSE) images of each grain were captured. Proportions of host and lamellae were determined by 163 analyzing the BSE images in ImageJ (USA National Institutes of Health, imagej.nih.gov/ij/, 164 accessed Aug 15, 2017). Precursor compositions were then calculated from EPMA data using host and lamellae proportions, assuming a representative cross-section. In the rare case that an 165 166 antiperthite grain rim lacked lamellae, the rim was excluded from the calculation. Perthites were 167 reintegrated using two techniques. The conservative reintegration counted only preserved 168 lamellae in perthite hosts, and used evenly spaced grids of unfocused (25 µm) beam spots (Table 8). Image processing was not used for this group because they have irregularly distributed 169 170 lamellae on several scales, from micrometers to sub-micrometers wide, as well as low BSE

contrast between K-feldspar host and lamellae which complicates image processing. A test case 171 172 for two-feldspar thermometry counted grain boundary plagioclase as well as lamellae for a well-173 preserved example (Fig. 7d, Table 9), and used spot analyses of host and lamellae combined with 174 image analysis. Ternary feldspar temperature estimates were calculated using Theriak/Domino 175 version 3.1 (De Capitani and Petrakakis 2010) with the feldspar activity models of Benisek et al. (2004), and Benisek et al. (2010). Two-feldspar thermometry for antiperthite-176 177 perthite+reintegrated grain boundary plagioclase used the method of Benisek et al. (2010), after 178 Kroll et al. (1993). 179 Bulk rock chemistry (Table 1) was measured via X-ray fluorescence (XRF) of representative samples (150A/268A/269A) by SGS (see Ague 2011 and references therein for 180 analytical details). Measured CaO was reduced assuming all P<sub>2</sub>O<sub>5</sub> is hosted in apatite and LOI 181 (taken to be  $H_2O$ ) was re-calculated from the raw data based on a fixed  $Fe^{3+}/Fe$ (total) molar 182 183 ratio. This ratio was deduced for each sample by comparing measured ilmenite compositions to those predicted by phase equilibrium calculations for a given  $Fe^{3+}/Fe(total)$  molar ratio (see 184 Results section). Assuming ilmenite equilibration between the peak metamorphic conditions 185 estimated herein and later lower granulite facies conditions along the P-T path of the Brimfield 186 Schist (Ague et al. 2013) gives a  $Fe^{3+}/Fe$ (total) molar ratio of 1%. The phase relations are 187 relatively insensitive to the  $Fe^{3+}/Fe^{2+}$ (total) ratio; values up to ~10% have no significant effect 188 on the *P*-*T* estimates. 189

Pseudosection modeling used Theriak/Domino ver. 4.02 (De Capitani and Petrakakis,
2010) and the thermodynamic data file of D.K. Tinkham (version 02,

192 <u>http://dtinkham.net/peq.html#theriak-domino-files-holland-powell-2011-database</u>, August 24,

193 2015) which contains the internally consistent tc-ds62 thermodynamic database of Holland and

194 Powell	(2011; referred to	herein as HP11	) and compatible	activity models.	The following
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- activity models were used: Melt: (White et al. 2014a); Ternary feldspar: (Holland and Powell
- 196 2003); Garnet: (White et al. 2014a,b); White mica: (White et al 2014a); Biotite: (White et al.
- 197 2014a,b); Orthopyroxene: (White et al. 2014a,b); Sapphirine: (Wheller and Powell 2014);
- 198 Cordierite: (White et al. 2014a,b); Spinel: (White et al. 2002); Ilmenite: (White et al. 2014b).
- 199 The clinopyroxene model of Holland and Powell (1996) was used for simplicity and, thus,
- 200 clinopyroxene phase relations should be regarded with caution. This has no impact on our
- 201 conclusions as the *P*-*T* conditions recorded by the rocks are well outside the stability field of
- clinopyroxene. Importantly, HP11 is applicable to HPG conditions (White et al. 2014a).

203 RESULTS

204 Rock Description and Field Relations

205 The dominant mineral assemblage is garnet + spinel + corundum + plagioclase + K-206 feldspar + biotite + ilmenite, with leucosomes of two feldspars  $\pm$  biotite surrounding garnet in some cases. The rock crops out as a single, uneven horizon several meters thick bounded by 207 more siliceous, quartz-bearing gneisses. A fault marks the lower contact with the silicic gneiss. 208 209 The upper contact, with a dark grey gneiss, is obscured by meter-scale leucosomes. Sheared 210 leucosomes document thrust motion with a transport direction toward the modern E-SE (Fig. 2a). The rocks vary from coarsely foliated to massive (Fig. 3a,e). Two textural styles are 211 recognized: Type I is a "salt-and-pepper" variety where alkali feldspar is evenly distributed 212 throughout the matrix, and Type II has uneven alkali feldspar distribution with coarse and 213 214 interconnected leucosomes often defining a foliation (Fig. 3a). Leucosomes in Type I samples

commonly have disaggregated edges. Both types contain the same mineral assemblage, although

216	Type II contains the coarsest garnet, spinel, and corundum. In Type II samples, fine-grained
217	bands several mm thick with numerous idiomorphic garnets and extremely fine-grained biotite
218	suggest localized shearing; in some of these regions garnet porphyroblasts show trails and crusts
219	of idiomorphic garnets in thin section (Fig. 4a). Most Type II samples contain trace pyrrhotite,
220	chalcopyrite, and other sulfides in the matrix and as inclusions in corundum.
221	The chemical characteristics of the rocks likely reflect those of the protolith (e.g. some
222	type of aluminous sediment or volcaniclastic deposit) modified by melt loss leading to silica
223	undersaturation (e.g. Clifford et al. 1975; Lal et al. 1978; Grant 1985a,b) prior to the HPG stage.
224	Protolith identification must await further geochemical study.
225	Mineral Descriptions and Chemistry
226	Garnet
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228 229	corundum with no intervening coronas or reaction zones (Figs. 3d and 4b). Megacrystic (up to $\sim 10$ cm) garnets occur within a sheared portion of the unit (Fig. 2b). In some samples, small
228 229 230	corundum with no intervening coronas or reaction zones (Figs. 3d and 4b). Megacrystic (up to $\sim$ 10 cm) garnets occur within a sheared portion of the unit (Fig. 2b). In some samples, small ( $\sim$ 100–300 µm) subhedral to euhedral garnets are present in addition to larger porphyroblasts
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Representative garnet compositions are given in Table 2. Garnet is almandine rich and
often shows Mn depletion in rims (Table 2) as well as Mg depletion near contacts with biotite
and along infilled cracks (Fig. 6c). Overall, garnets preserve little core-to-rim variation in Fe,
Mg, and Ca. Where zonation of these elements is preserved, it is smooth, almost certainly
reflecting diffusional relaxation (Fig. 6b-d).

242 Notably, some garnets have distinctly elevated phosphorus contents. In particular, the core of garnet in sample 294A-1 contains 0.18 wt% P<sub>2</sub>O<sub>5</sub> (Table 2). Moreover, this garnet 243 244 preserves a detailed textural record of modifications to zonation. The element map reveals that 245 phosphorus is complexly and irregularly zoned in the garnet core, but a more regular core-to-rim 246 zonation pattern is preserved in the bottom-left quadrant (Fig. 6b). The highest-phosphorus area has a sharp and irregular contact with the low-phosphorus area, showing lobate and irregular 247 zoning comparable to that produced by interface-coupled dissolution-reprecipitation (ICDR) in 248 249 garnets from UHP and HP granulites (Ague and Axler 2016). The ICDR process requires the 250 interaction of fluids (defined broadly to also include melts) with minerals (e.g. Putnis and Austrheim 2010; Putnis and John 2010; Harlov et al. 2011). Thus, the isolated zones of 251 phosphorus-rich garnet are likely relics of incomplete reactions with infiltrating fluid. A faint 252 253 core-to-rim zoning of phosphorus is preserved in the bottom-right quadrant of the high-254 phosphorus region (Fig. 6b), suggesting that this is the original growth zoning and the low-255 phosphorus zones have been leached.

The core of the high-phosphorus garnet in sample 294A-1 contains oriented lamellae of rutile, ilmenite, and apatite. The lamellae are restricted to the P-rich portion of the garnet and are absent from immediately adjacent areas with lower phosphorus contents. They are oriented in 4 major directions and several subsidiary directions; the 4 major directions correspond to the

<111> axes of the garnet host (e.g. Griffin et al. 1971; Axler and Ague 2015a). Some lamellae
are composites of two minerals. Pairs of ilmenite with either apatite or rutile are present and
show a range of morphologies (Fig. 4d).

263 Oriented rutile and ilmenite lamellae have been previously documented in garnet cores 264 from the silica-saturated rocks in both the Brimfield Schist and the Bigelow Brook Formation 265 (Ague and Eckert 2012; Ague et al. 2013; Axler and Ague 2015a). Depletion halos of Ti around 266 these lamellae confirm an exsolution (precipitation) origin (Ague and Eckert 2012). For the 267 silica-undersaturated lithology of this study, the restriction of lamellae to portions of garnet 268 apparently unmodified by ICDR, their consistent lamellar habits, and their orientation along garnet zone axes, lead us to the interpretation that the inclusions exsolved from a precursor 269 270 garnet richer in Ti and phosphorus; verification of this must await future crystallographic study.

271 Spinel

272 Spinel is found as black-green ragged, net-like matrix grains ( $\sim$ 500  $\mu$ m – 3 mm) and 273 remarkable elongated, polycrystalline masses that reach nearly 6 cm in length (Fig. 3a,c) with 274 long axes aligned roughly parallel to foliation. The largest grains contain thin ( $\leq 1 \mu m$ ) platelets of Fe-Ti oxide that are oriented in 4 directions and can have broadly trigonal cross-sections. 275 276 Oriented lamellae are largely absent in the rims of spinel masses and were not observed in spinel 277 crystals less than ~1mm wide. Within the large spinel masses, crystallographic domains on the 278 order of several mm are distinguishable in hand sample by their different reflections of light, and 279 in thin section by changes in orientation of oriented oxide platelets (Fig. 5e). Larger spinels typically host inclusion pockets ~50 µm to ~500 µm across containing ternary feldspar with 280 exsolution features, homogeneous feldspar, corundum, and biotite. Isolated feldspar inclusions 281 282 are widespread.

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283	Analyzed spinel is hercynite rich and usually Zn poor ( $\leq 0.5$ wt%; Table 3). Spinel is
284	notably Mg poor relative to many garnet-spinel-corundum-bearing rocks (e.g. Padovani and
285	Tracy 1981; Sengupta et al. 1999; Dharmapriya et al. 2015), especially those containing
286	sapphirine (e.g. Ouzegane et al. 2003). Matrix spinel generally has between 2 and 2.5 wt%
287	MgO; spinels included in garnet have elevated MgO relative to those in the matrix (e.g. sample
288	268A-2; 4.5 wt%), even those in contact with garnet. Matrix spinel from sample 294A-1 has the
289	highest Cr <sub>2</sub> O <sub>3</sub> content (~4.5 wt%), and particularly low MgO content (2.0 wt%).
290	The oriented oxide lamellae are too thin for individual EPMA analysis, but a reintegrated
291	spinel composition was estimated with regularly spaced spot analyses, yielding a bulk ${\rm TiO}_2$
292	content of 0.1 wt% (Table 3). This analysis method may underestimate the proportions of
293	elements preferentially sequestered in the lamellae based on lamellae orientations relative to the
294	sectioned plane, although the beam spot (25 $\mu$ m diameter) was significantly larger than lamella
295	width, and lamellae have a regular distribution within the host. We consider it likely that the
296	lamellae exsolved to reduce spinel Ti content during retrogression.
297	Corundum

298 Corundum is present in two forms: as  $\leq 1$  cm subhedral to euhedral pink, purple, and blue crystals generally in contact with or within leucosomes in Type II samples (Fig. 3b,d), and 299 colorless, jagged matrix grains often rimming spinel in both Type I and Type II samples (Figs. 300 4a, 5c). Both varieties have polysynthetic twinning in thin section and fluoresce red under long-301 wave ultraviolet light, with pink and purple crystals (or zones within crystals) showing stronger 302 303 fluorescence than blue or colorless ones. The largest sapphires have concentric color zoning in hand sample (Fig. 3d). Most corundum crystals are fresh, although some in leucosomes are 304 coated with chlorite. 305

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306	A Cr element map of a representative coarse pink corundum in contact with leucosome
307	surrounding peritectic garnet (in a Type II sample) reveals euhedral concentric zoning suggesting
308	growth in contact with a fluid, most likely silicate melt, which may have supplied the trace
309	elements (Fe, Ti, Cr) responsible for coloration (Figs. 3b and 5d). The presence of large colored
310	corundum crystals, which only occur in Type II samples, appears to be tied to the presence of
311	coarse leucosome networks.
312	Colorless matrix corundum commonly includes biotite and spinel and has few impurities,
313	with negligible SiO <sub>2</sub> , Fe <sub>2</sub> O <sub>3</sub> , TiO <sub>2</sub> , and Cr <sub>2</sub> O <sub>3</sub> contents (Table 4). Impurities are slightly higher
314	in coarse, colored sapphire, but the mole fraction of corundum is >0.99 for all investigated
315	crystals.
316	Biotite
317	Biotite crystals are present in both the leucosomes and the matrix and range from several
318	millimeters across to only $\sim 20 \mu m$ in shear bands in Type II samples. Biotite may include
319	ilmenite and, more rarely, spinel, and is commonly included in garnet cores and rims.
320	Representative biotite analyses are given in Table 5. Matrix biotite Mg# typically ranges from
321	0.35 to 0.43, in keeping with the Fe-rich bulk composition of the rock. Higher values may
322	reflect retrograde Mg-Fe exchange with garnet. Biotite inclusions in the high-phosphorus garnet
323	of sample 294A-1 preserve greatly elevated $TiO_2$ contents exceeding 6.5 wt%. Armoring by
324	garnet may have prevented Ti loss from these biotite crystals such that they retain peak or near-
325	peak Ti contents (see below).

326 Feldspar

327	Plagioclase is found in leucosomes, small leucocratic pockets in the matrix, and as
328	inclusions in garnet, spinel, and biotite. Isolated matrix plagioclase grains are absent.
329	Antiperthite is widespread in leucosomes but homogenous plagioclase is also found, typically
330	adjacent to K-feldspar. The leucosomes surrounding peritectic garnet contain the antiperthites
331	with the coarsest exsolution lamellae (Figs. 3b and 7a). Lamellae of K-feldspar $\sim$ 1 mm wide in
332	the core of the antiperthite with the coarsest lamellae (JAQ278A-1_ap3) are themselves perthitic
333	and contain plagioclase lamellae up to ~2 $\mu m$ across (Fig. 7e,f). Antiperthite host and lamellae
334	compositions of coarse, unaltered grains in leucosomes, primarily surrounding garnet, are given
335	in Tables 6 and 7.

Perthitic and homogeneous K-feldspar are found in leucosomes and as isolated matrix 336 grains. K-feldspar crystals may contain inclusions of spinel and biotite. Some alkali feldspars 337 show evidence of former elevated phosphorus content in the form of oriented apatite needles. 338 339 These appear to be exsolved based on their sharp hexagonal cross-sections, consistent spatial distribution, and alignment along crystallographic planes (Fig. 7b). Alkali feldspar with elevated 340 phosphorous content (up to 2.5 wt%) has been reported from pegmatites and granites (e.g. 341 London et al. 1990; London 1992; Frýda and Breiter 1995; Kontak et al. 1996); usually alkali 342 343 feldspar is significantly enriched in phosphorous compared to coexisting plagioclase feldspar, either as subsolvus separate grains or hypersolvus exsolved lamellae. In Ca-poor felsic rocks, 344 345 alkali feldspars contain most of the bulk-rock phosphorous because apatite is scarce (London 1992). Reintegrated perthite compositions are given in Table 8. 346

Antiperthites can be relict from an igneous precursor (Štípská and Powell 2005). In this study, antiperthites in metamorphic leucosomes (commonly surrounding peritectic garnet and in contact with corundum) were judged to be metamorphic, and therefore preserving solvus

temperatures representative of metamorphic conditions (e.g. Figs. 3b and 7a,e). The consistent

- distribution of antiperthites with coarse exsolution lamellae and little retrogression or resorption
- in leucosomes of different samples also indicates a metamorphic origin.

353 Perthites have several textures that indicate solid-state loss of plagioclase component.

Figure 7c illustrates a typical perthite with plagioclase lamellae in the core of the grain but not in

the rim. This suggests diffusional depletion of the plagioclase component into the matrix.

Plagioclase blebs have a tendency to pool along cracks and grain boundaries (Fig. 7d).

357 Plagioclase crystals adjacent to K-feldspar in leucosomes may represent lost plagioclase

358 component that has been recrystallized and consolidated. These characteristics suggest that the

359 perthites do not retain their peak plagioclase content, and thus do not record peak temperatures

360 for ternary feldspar reintegration thermometry. In contrast, antiperthites normally preserve

361 exsolution lamellae up to their edges (Fig. 7a). Our observation that antiperthites can preserve

362 exsolution lamellae near grain rims while perthites commonly do not has also been reported in

363 previous studies (e.g. Zulauf et al. 2002; Hokada and Suzuki 2006).

## 364 Sillimanite

Sillimanite is found most commonly as inclusions in garnet rims, and occasionally as isolated matrix grains. Where foliation is present, sillimanite is aligned with biotite, the primary matrix mineral defining the fabric, although a later generation of sillimanite cross-cuts the foliation in some samples (Fig. 8). In a few examples, sillimanite inclusions in garnet cores form trails that may record former foliation. These cores may be relict from prograde growth or may have grown during retrogression.

371 Ilmenite

372	Ilmenite is found as discrete, subhedral grains, and as isolated and netlike masses, usually
373	in contact with spinel and corundum. Ilmenite does not deviate far from endmember
374	composition, containing only minor impurities such as Si, Al, Mn, and Mg (Table 4);
375	concentrations could have been higher, however, at peak thermal conditions. Ilmenites have low
376	estimated Fe <sup>3+</sup> contents and lack exsolution textures.
377	Ternary Feldspar Reintegration Thermometry
378	Antiperthites
379	The activity model of Benisek et al. (2010) returns the highest temperatures for
380	reintegrated antiperthites, averaging 1053 °C $\pm$ 38 °C (2 $\sigma$ standard error) (Table 10; Fig. 9). The
381	activity model of Benisek et al. (2004) yields a somewhat lower but still comparable mean of
382	1023 °C $\pm$ 20 °C (Table 10; Fig. 9). Both results use a pressure of 1.5 GPa, although the ternary
383	feldspar solvi are only weakly pressure dependent. Regardless of activity model choice,
384	antiperthites record $T > 1000 \text{ °C}$ (mean ~1040 °C); as these are solvus $T$ estimates, they are
385	minima. These results are in agreement with those obtained for antiperthite and mesoperthite
386	from silicic granulites in the Brimfield Schist and Bigelow Brook Formation (Ague et al. 2013;
387	Axler and Ague 2015a), which gave solvus (minimum) temperatures ranging from ~900 $^{\circ}$ C–
388	1000 °C. Zirconium-in-rutile thermometry likewise gives $T$ in the 1000 °C range (Ague et al.
389	2013), with maximum estimates of nearly 1040 °C (Axler and Ague 2015a).
390	Perthites
391	Perthite T estimates have a significant dependence on activity model, and yield means of

- 392 868 °C  $\pm$  23 °C (Benisek et al. 2010) and 783 °C  $\pm$  44 °C (Benisek et al. 2004) (Table 10; Fig. 9;
- $2\sigma$  standard error). The two activity models differ regarding the behavior of the solvus for K-

394 feldspar-rich ternary compositions (77–85 mol%) near the K-feldspar-albite join. The perthites 395 have textures that clearly indicate loss of plagioclase component (Fig. 7 c,d), leading to T 396 estimates that are erroneously low compared to those obtained from the antiperthites, regardless 397 of activity model. To test this for thin section 278A-1, the perthite and coexisting grain boundary plagioclase in Figure 9d were reintegrated (Tables 9, 10) and the solvus T recomputed, yielding 398 ~915 °C and ~1120 °C using Benisek et al. (2004) and (2010), respectively (Table 10). These 399 estimates are more compatible with those obtained from antiperthites. 400 Furthermore, we tested the assumption that this reintegrated perthite was in equilibrium 401 with reintegrated antiperthites in the same section (278A-1, Table 10) using two-feldspar 402 403 thermometry and the method of Benisek et al. (2010) (after Kroll et al. 1993, which has been 404 applied successfully to HPG rocks such as in O'Brien et al. 1997). This method is powerful in 405 that it corrects for retrograde Na-K resetting and requires convergence of temperature estimates 406 for all three feldspar endmembers. Pairing the reintegrated perthite with each of the three 407 reintegrated antiperthites in 278A-1 (Table 10) gives temperatures of 1059°C, 1057 °C, and a 408 non-converging, but still reasonably constrained, estimate of 1073 °C (Ab and An) to 1081 °C (Or) (Table 10). These estimates agree very well with the antiperthite results discussed above. 409 On this basis, we conclude that antiperthites and perthites were formerly in equilibrium 410 before retrogression. Antiperthites mostly retained their exsolution products and, as a 411 412 consequence, their reintegrated compositions yield reliable HPG solvus temperature estimates. 413 Perthites, on the other hand, lost plagioclase component to grain boundaries and cracks. 414 However, perthite reintegration that accounts for this lost plagioclase yields two-feldspar 415 temperatures that are fully consistent with those from antiperthites alone.

## 416 **Ti-in-biotite thermometry**

417	We employed the Ti-in-biotite thermometer of Wu and Chen (2015) and compared its
418	results to ternary feldspar reintegration thermometry. We used the biotite with the highest
419	measured Ti content, which is an inclusion in the high-phosphorus garnet of sample 294A-1
420	(Table 5). This biotite was likely shielded from retrogression by the surrounding garnet, so as to
421	preserve high Ti concentrations from peak thermal conditions. The thermometer gives a
422	temperature of ~1100 °C for the average pressure of 1.8 GPa obtained from the pseudosections
423	described below. At 1.5 GPa, the result is ~1050 °C. Although the thermometer is only
424	calibrated for T <840 °C, the estimates it returns agree well with the average obtained from
425	ternary feldspar reintegration and two-feldspar thermometry.

## 426 **Pseudosection Modeling**

### 427 General Comments

428 Pressure is estimated based on the phase relations for the solid phase assemblage observed in the rocks: garnet + spinel + corundum + biotite + K-feldspar + ilmenite (the "full 429 430 assemblage"). The phase field, which includes coexisting melt, is relatively narrow and is 431 consistently positioned across bulk composition space, facilitating pressure estimation given 432 constraints on temperature (Fig. 10). The field is also remarkably stable and occupies the same 433 general position in pseudosections constructed for all three representative bulk compositions 434 from Type I and Type II rocks. Despite differences in bulk composition among the samples, 435 including SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, K<sub>2</sub>O, and FeO contents, the stability of this particular assemblage is 436 insensitive to moderate changes in composition.

## 437 *Pressure Estimation*

# The pseudosections for samples 150A, 268A, and 269A predict equilibration at 1.65–2.1 GPa, 1.6–2.1 GPa, and 1.7–2.05 GPa, respectively, for the average reintegrated antiperthite $T \pm$

21

440	$2\sigma$ standard error (Fig. 10). Simultaneous spinel breakdown and corundum growth (Fig. 5c) are
441	predicted only within the full assemblage phase field (and its lower-temperature two-feldspar
442	extension in 150A and 269A). This texture implies that the observed assemblage formed during
443	cooling through the full assemblage field. Ternary plagioclase is absent from this field for each
444	sample, but this is to be expected as it is only present in leucosomes, which were intentionally
445	excluded from bulk analyses to ensure an accurate bulk composition determination for the matrix
446	assemblage (e.g. Guevara and Caddick 2016). For samples 150A and 269A, the down-T, down-F
447	granulite facies two-feldspar extension of the full assemblage field predicts albite-rich
448	plagioclase rather than significantly ternary feldspar and, thus, does not represent the observed
449	mineralogy.

450 Discussion of Pseudosection Results

Charge balance calculations indicate  $Fe^{3+}$  in spinel, ilmenite, and garnet (Tables 2–4). 451 Ferric iron content estimation using ilmenite chemistry may underestimate the  $Fe^{3+}/Fe^{2+}$  ratio of 452 the bulk rock if ilmenites equilibrated in the amphibolite facies. This would not affect results 453 significantly, as pseudosections made assuming higher  $Fe^{3+}$  content up to ~10% yield similar 454 results. The main difference is the expansion of garnet + spinel + corundum + biotite + ilmenite 455 + K-feldspar + melt stability for bulk compositions with more  $Fe^{3+}$ . Likewise, increasing water 456 content does not affect the results. Increasing molar H<sub>2</sub>O by 15% produces nearly identical 457 phase fields. Spinel has low Cr and Zn contents (Table 3). These trace elements can therefore 458 safely be excluded from the modeling system with negligible effect on the stability range of 459 spinel (e.g. Powell and Sandiford 1998; Shulters and Bohlen 1989; Diener and Powell 2010). 460 Garnet and spinel, which can preserve strong chemical growth zonation, show relatively little 461 core-to-rim variation in major or trace elements (Tables 2-3). The impact of prograde zonation 462

in these modally major minerals on the accuracy of the measured effective bulk composition istherefore minimized.

465	In general, preservation of peak garnet compositions at $\sim 1000$ °C is unlikely because
466	intracrystalline diffusion will be rapid (e.g. Chakraborty and Ganguly 1992; Faryad and
467	Chakraborty 2005; Carlson 2006; Chu and Ague 2015). For example, the $2\sqrt{Dt}$ characteristic
468	diffusive length scale for Fe in garnet at 1000 °C is several mm in $10^6$ yr using the diffusion
469	coefficient calibration of Chu and Ague (2015). A further complication is that isopleths of garnet
470	Mg# and grossular content are sub-parallel across much of the modeled phase space for each
471	sample. Moreover, ICDR has clearly affected some garnets (Fig. 6). Nonetheless, sample 150A
472	preserves a garnet core Mg# that intersects the ternary feldspar $T$ estimate within the full
473	assemblage field, and is closely correlated with core grossular content. The latter places
474	minimum garnet core T equilibration at ~975 °C (Fig. 10). The analyzed garnet in sample 150A
475	contains no plagioclase inclusions and the rock overall has much less biotite than samples 268A
476	and 269A; these factors may have restricted retrograde Ca and Fe-Mg exchange.

477 Garnet rim Mg# is consistently lower than core Mg#. Rim Mg# and grossular isopleths tend to cluster in the lower granulite facies, in some cases at lower P than modeled on the 478 pseudosections. Core Mg# and grossular isopleths for sample 268A intersect at  $\sim$ 825 °C and  $\sim$ 0.6 479 GPa, which broadly correlates with the granulite facies overprint identified by Ague et al. (2013; 480 481 700-800 °C, 0.5-0.6 GPa) (Fig. 10). We note in this regard that garnet rims include sillimanite (Fig. 4b) and rare matrix sillimanite cross-cuts foliation. If the rock remained reactive, the 482 pseudosections predict some garnet (and sillimanite) growth during cooling through the granulite 483 facies in the event of even a slight pressure increase following a UHT phase (Ague et al. 2013), 484 which could account for these sillimanite textures. Sillimanite is stable for each bulk composition 485

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below ~800 °C and ~0.8 GPa. Thus, based on general isopleth trends as well as sillimanite
relationships, we conclude that the granulite overprint likely affected the silica-undersaturated
lithology as well.

- For all three analyzed samples, pseudosections predict between 35 and 41 vol% melt at 1040 °C and 1.8 GPa. This would have decreased the strength of the rock substantially and
- 491 helped facilitate the observed widespread ductile shear deformation (Fig. 2a).
- 492 Results obtained using the thermodynamic dataset of Holland and Powell (1998) and
- 493 compatible activity models also yield HPG pressure estimates of 1.55 to 1.7 GPa at 1040 °C. The
- feldspar activity model of Holland and Powell (2003) (which produces results similar to the
- Holland and Powell 1992 model and is standard in the Holland and Powell 2011 dataset) yields
- 496 reintegrated ternary feldspar temperatures >1150 °C. This places pressures as high or even
- 497 higher than our preferred results (Fig. 10).

498 DISCUSSION

499 Spinel Masses

500 The remarkable polycrystalline spinel masses documented herein deserve further 501 comment, especially as they deviate strongly from the expected isometric spinel habit (Fig. 3c). 502 The separation of each mass into crystallographic domains suggests that either a single cubic precursor grain underwent dynamic recrystallization, or that the grains are pseudomorphs or 503 paramorphs after some other phase, perhaps with markedly different symmetry. As no high-504 pressure polymorphs of hercynitic spinel are known, we suggest that strain is responsible for 505 506 recrystallizing formerly cubic spinel into an elongated, non-isometric form. Exsolution of Fe-Ti oxide could have occurred either before or after recrystallization, as each domain retained cubic 507

symmetry (Fig. 5e). To the best of our knowledge, such large, elongated, and polycrystalline
spinel forms are previously undescribed in metamorphic rocks.

510 The oriented Fe-Ti oxide lamellae in spinel are likely ilmenite, because they contain 511 higher Ti than the host spinel, comparable Fe, and negligible Al. Intergrowths of spinel and ilmenite have been reported from UHT granulites (e.g. Sengupta et al. 1999) although they are 512 513 not oriented lamellae of one phase inside of another. Nonetheless, these features have been 514 interpreted as resulting from the expulsion of an ülvospinel ( $Fe_2TiO_4$ ) component from a 515 precursor high-Ti spinel (Sengupta et al. 1999). Given this, and the clear crystallographic shape 516 preferred orientation of the lamellae, we suggest that they formed via exsolution and, thus, the Ti 517 necessary to form the precipitates was soluble in hercynitic spinel at HPG or UHT conditions. This presupposes that the precursor phase was a spinel, but the isometric lattice constraints of the 518 platelets suggest that they precipitated from a host phase with cubic symmetry, so spinel is the 519 520 most likely candidate.

A reintegrated spinel formula shows that the precursor spinel contained 0.2 mol% 521 522 ulvospinel component (277A-3; Table 3; computed following Sack and Ghiorso 1991a,b). An unexsolved spinel from a different sample (294A core; Table 3) has a higher ülvospinel mol% of 523 0.5. Other spinels in this sample show exsolution textures. Pseudosection calculations predict 524 between 2 and 3 mol% ulvospinel component in spinel at peak conditions (1040 °C, 1.8 GPa) 525 and ~2 mol% at UHT granulite facies conditions of 900 °C and 0.9 GPa, so the spinel precursors 526 527 likely equilibrated with decreased Ti content during cooling from granulite facies conditions prior to ilmenite exsolution. The spinel textures deserve further work as they may hold 528 information about pressure and temperature conditions, melt fraction, and strain rate or 529 530 magnitude.

# 531 Trace Elements and ICDR in Garnet

532	Rutile, ilmenite, and apatite lamellae in the core of the phosphorus-rich garnet of sample
533	JAQ294A-1 are interpreted as precipitates from a garnet richer in Ti and phosphorus stable at $T$
534	$\geq$ 1000 °C and $P \geq$ ~1.8 GPa. Similar textures are present in garnet cores from silicic granulites
535	of the Brimfield Schist and Bigelow Brook Formation (Ague and Eckert 2012; Ague et al. 2013;
536	Axler and Ague 2015a). Given that the silica-undersaturated HPG lithology and the silicic
537	gneisses appear to share at least a retrograde history, we expect that the amount of Ti in the
538	precursor garnet would have been stable at HPG or UHT conditions, similar to the conclusion of
539	Ague and Eckert (2012). Solubility of Ti in garnet is thought to be controlled more by
540	temperature than pressure at P $\leq$ 5 GPa, and although UHP majoritic substitutions may change
541	this trend, the pressure dependence of Ti substitution in garnet is not yet clear (e.g. van
542	Roermund et al. 2000; Zhang et al. 2003; Ackerson et al. 2017).
543	Apatite precipitates have been reported from garnets in UHP rocks and kimberlite
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<sup>554</sup> °C, comparable to the highest phosphorus content we have measured (0.18 wt%  $P_2O_5$ ; Table 2). <sup>555</sup> Thus, although UHP conditions may not be required to produce the phosphorus levels we <sup>556</sup> observe, the Konzett (2016) experiments strongly suggest that the *P-T* conditions recorded by the <sup>557</sup> silica-undersaturated rocks are nonetheless unusual for crustal metamorphism.

The striking ICDR record preserved by irregular phosphorus zoning in garnet (Fig. 6a) 558 559 has many parallels to that found in garnet from the HPG Saxony Granulite (Ague and Axler 560 2016). In both samples, phosphorus zonation is highly irregular and shows little diffusional 561 smoothing. In the CT HPG sample, "cut-out" depletion halos around K-feldspar inclusions 562 reveal that the inclusion-host interface may provide an expedited fluid pathway, enabling ICDR 563 reactions to penetrate more efficiently in more poikiloblastic grains. Partial diffusional 564 smoothing of Ca and total smoothing of Mg provide further evidence that P diffuses more slowly 565 than major elements in garnet, making it a valuable recorder of growth zonation and later modifications (e.g. Ague and Axler 2016). Fluid infiltration at HPG, granulite, or amphibolite-566 567 facies conditions could have produced the observed ICDR zonation. Further research is required 568 to understand the extent to which ICDR may be responsible for modifying mineral compositions 569 in high-grade rocks, especially how to distinguish fluid signatures from melt signatures.

570 Petrogenesis

Thermometry and pseudosection modeling place equilibration of the silica-undersaturated HPG lithology in the HPG realm at ~1040 °C and ~1.8 GPa. This pressure requires exhumation from a depth of ~60 km (assuming lithostatic pressure) and thus either burial to the base of orogenically thickened continental crust or thermal equilibration during exhumation from a hot subduction zone. Both of these scenarios have been proposed for high-pressure granulites (O'Brien and Rötzler 2003; O'Brien 2008). Eclogite facies rocks are thought to become high-

577	pressure granulites during exhumation in some cases (e.g. Willner et al. 1997; Yao et al. 2000;
578	Zhao et al. 2001; O'Brien and Rötzler 2003; Kim et al. 2006; Nahodilová et al. 2011). Marschall
579	et al. (2003) suggested that Variscan granulites from the Schwarzwald were heated by basaltic
580	magmas at the base of the orogenic belt and subsequently exhumed by orogenic extension. High
581	levels of mantle heat input have also been invoked to explain high-T Variscan metamorphism
582	(Sorger et al. 2018). Another possible parallel is the Gruf Complex of the Alps, which may have
583	been heated to its UHT conditions by mantle diapirs (e.g. Galli et al. 2011). We conclude that
584	the presence of HPG rocks in the CMT imposes new constraints on metamorphism and
585	tectonism, namely that the thrust sheets of the southern CMT must have sampled the orogenic
586	root and/or deeper settings during the assembly of composite Laurentia.
587	If the rocks formed in a subduction environment, their provenance is a hint that Acadian-
588	Neoacadian terranes may hold similarly deeply subducted rocks to those known from contiguous
589	Caledonian terranes in Greenland and Europe, which contain HPGs as well as the second-largest
590	UHP terrane (by geographic extent) (Gilotti and Elvevold 2002; Gilotti and Krogh Ravna 2002;
591	Hacker et al. 2010; Gilotti 2013; Klonowska et al. 2017). Some Acadian rocks have already
592	been suggested as being indicative of deep subduction (e.g. Peterman et al. 2016).
593	
594	IMPLICATIONS FOR SYENITES AND ALUMINOUS XENOLITHS
595	The partial melting which produced leucosomes has implications for both syenite genesis
596	and the origins of highly aluminous xenoliths.

- 597 Leucosomes in the silica-undersaturated rocks, which were likely generated in situ, are
- 598 effectively syenites consisting of two feldspars and biotite. Notably, syenite bodies with younger

599	crystallization ages than peak metamorphic conditions are reported from several UHT terranes
600	(e.g. Brandt et al. 2003; Karmakar and Schenk 2016) and some UHT localities have syenitic
601	leucosomes (e.g. Hokada and Harley 2004). Moreover, lower-crustal UHT and HPG metapelites
602	have been explicitly considered as syenite producers (e.g. Litvinovsky et al. 2000; Tchameni et
603	al. 2001; Litvinovsky et al. 2002; Hacker et al. 2005; Shaffer et al. 2017). Nonetheless, the
604	prevailing consensus is that some mantle input is responsible for many syenite provinces (e.g.
605	Harris et al. 1983; Bailey 1987; Kramm and Kogarko 1994; Litvinovsky et al. 2002; Markl et al.
606	2010) and some syenite bodies have strong petrological evidence for mantle melt participation
607	(e.g. Lang et al. 1995), such as zoned syenite-pyroxenite dikes. Laboratory results showing that
608	syenites can be produced by HPG silicic rocks offer compelling evidence that high-grade
609	aluminous rocks like the silica-undersaturated rocks of this study may also produce syenites
610	(Litvinovsky et al. 2000).

The density of the CT silica-undersaturated HPGs may provide a key clue to an aspect of syenite genesis and possibly also delamination. Delamination of the lower crust or lithospheric mantle has been the focus of much previous work (e.g. Bird 1978; Bird 1979; Houseman et al. 1981; Rudnick 1995; Zegers and van Keken 2001; Anderson 2005) and delamination has been suggested to follow alkaline igneous melt generation in lower crustal domains (Smithies and Champion 1999). We suggest that the rocks of this study may be a syenite source that did not melt to its full extent and delaminate.

Densities of the samples were calculated using Theriak-Domino (Table 1). At 1040 °C and 1.8 GPa their average density is  $3.0 \text{ g/cm}^3$ , which is roughly as dense as basalt and significantly denser than molten basalt (Ahrens and Johnson 1995). If 50% of the melt calculated to be present at these *P-T* conditions was extracted from the rock, density would

622	increase to 3.0–3.2 g/cm <sup>3</sup> , which is near that of peridotitic or dunitic mantle (Daly et al. 1966;
623	Boyd et al. 1976; Rudnick 1995). If all of the melt was extracted (with no more being generated)
624	densities would reach $3.2 - 3.5$ g/cm <sup>3</sup> , roughly as dense as eclogite, which is negatively buoyant
625	relative to aesthenospheric mantle (Ahrens and Johnson 1995; Rudnick 1995; Zegers and van
626	Keken 2001). If more melting and melt loss occurred beyond what the CT rocks underwent,
627	density would be driven even higher as more feldspar and biotite are lost.

Depending on the mantle rock type directly beneath the base of orogenic crust, and the degree to which that part of the mantle wedge was hydrated, delamination might release the aluminous rock into the mantle (e.g. Hacker et al. 2011). Delamination could be triggered by buoyant upwelling of magma pooling at the base of the crust, or crustal flow in response to orogenic deformation (e.g. Bird 1978; Houseman et al. 1981). If delamination followed melt generation and loss, the syenite source region would founder into the mantle, leaving only the released melt within the crust.

We recognize that there are likely many complex aspects to syenite generation, and that no one model is likely to account for the varied settings in which syenites are found. Further study of syenites in terranes with anatectic UHT/HPG metapelites may well reveal that these magmas can be generated at the base of orogenically thickened crust or in rift zones. Rocks like those of this study may satisfy the requirements of "missing" syenite source regions both chemically and physically.

Melt removal also has implications for the origins of highly aluminous xenoliths in
kimberlites containing various combinations of garnet, spinel, and corundum (e.g. Nixon et al.
1978; Padovani and Tracy 1981; Exley et al. 1983; Mazzone and Haggerty 1989).

644 Pseudosections (Fig. 10) show that large degrees of partial melting and melt removal would

645	leave behind highly aluminous restites enriched in garnet + spinel ± corundum (+ ilmenite)
646	which would bear strong mineralogical similarities to the kimberlite xenoliths. As discussed
647	above, if the restites reside at the base of the crust and are dense enough, they could delaminate
648	and founder into the mantle. They would then be available to become entrained in kimberlites.
649	Alternatively, such residues might also form by extreme melt loss from metasediments in
650	subducted slabs, prior to incorporation in kimberlites (e.g. Mazzone and Haggerty 1989). This
651	melt loss would, in turn, release syenitic melts and associated fluids to arc magma source
652	regions. Either scenario is broadly compatible with current hypotheses for HPG genesis at the
653	base of orogenically-thickened crust or as a consequence of "hot" subduction zone activity (e.g.
654	O'Brien and Rötzler 2003; Marschall et al. 2003; O'Brien 2008). Delamination or subduction
655	would help to explain the relatively uncommon occurrence of these rocks in exhumed orogenic
656	belts.

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1095	<b>FIGURE 1</b> : Geologic map showing (a) Major geologic terranes of New England, U.S.A.
1096	following Axler (unpublished PhD thesis; after Aleinikoff et al (2007) as modified from Hibbard
1097	et al. (2006)). (b) Sample location within thrust slices of the Central Maine Terrane. Formations
1098	colored in red are UHT (Ague et al. 2013; Axler and Ague 2015a; based on geologic maps of
1099	Rodgers 1981,1985). HB=Hartford Basin, RH=Rowe-Hawley, P-N=Putnam-Nashoba.
1100	FIGURE 2: Field exposures of the silica-undersaturated rocks. Grt = garnet. (a) Sheared
1101	leucosome; arrows indicate thrust sense of shear. (b) Garnet megacrysts surrounded by
1102	leucosome.
1103	<b>FIGURE 3</b> : Silica-undersaturated rocks in hand sample. Grt = garnet, spl = spinel, crn =
1104	corundum, $fsp = feldspar$ . (a) Type II sample showing coarse leucosomes and spinel defining
1105	foliation and pink-purple sapphires within leucosomes [296A]. (b) Peritectic garnet including
1106	antiperthite surrounded by antiperthite-bearing leucosome in contact with a pink sapphire; spinel
1107	and corundum are present in the matrix assemblage [295A]. (c) Spinel mass showing irregular
1108	crystallographic domains revealed by differential reflection of light [296A]. (d) Concentrically
1109	zoned sapphire in contact with leucosome and garnet that includes pink corundum [298A]. (e)
1110	Type I sample [150A].
1111	<b>FIGURE 4</b> : Photomicrographs of representative garnet textures in thin section. Grt = garnet, spl =
1112	spinel, $crn = corundum$ , $bt = biotite$ , $rt = rutile$ , $sill = sillimanite$ , $ilm = ilmenite$ , $ap = apatite$ . (a)
1113	Matrix assemblage with typical porphyroblastic garnet and second-generation euhedral garnets,
1114	as well as spinel (dark) rimmed by corundum; plane light [268A-2]. (b) Garnet with sillimanite

inclusions (core) and spinel inclusions (rim) reflecting a granulite facies assemblage; plane light

1116 [296A-3]. (c) Oriented rutile needles in the core of high-phosphorus garnet (Fig. 6; Table 2),

1117 crossed polars [294A-2]. (d) Oriented rutile needles with composite needles of ilmenite and

apatite in the same garnet core; plane light [294A-2].

**FIGURE 5**: Spinel and corundum. Grt = garnet, spl = spinel, crn = corundum, ilm = ilmenite, bt

= biotite. (a) Photomicrograph of ragged spinel and rounded biotite inclusions in a garnet core;

plane light [296A-3]. (b) Photomicrograph of corundum with biotite inclusions cross-cutting and

rimming spinel; plane light [296A-3E2]. (c) Photomicrograph (true color) of oriented Fe-Ti

1123 oxide lamellae in spinel; a domain boundary is marked by the change in lamellae orientation,

1124 condenser lens in, plane light [296A-3]. (d) Element map of Ti in spinel showing orientations of

ilmenite lamellae; brighter colors indicate higher concentration [294B]. (e) Element map of Cr

in a pink corundum in leucosome showing concentric euhedral growth zoning [295A-1].

**FIGURE 6**: Element maps of the high-phosphorus garnet of 294A-1. Ap = apatite, kfs = K-

1128 feldspar. (a) and (b) Phosphorus. (c): Magnesium. (d) Calcium. (e) Backscattered electron

1129 image. Warmer colors indicate higher concentration. Color scale does not correlate across

1130 panels. Black arrow in inset panel indicates the region with measured 0.18 wt%  $P_2O_5$  (Table 2).

**FIGURE 7**: Feldspar textures. Grt = garnet, anti = antiperthite, kfs = K-feldspar, ap = apatite,

1132 plag = plagioclase. (a) Leucosome in contact with peritectic garnet [295A]. (b) Oriented apatite

needles in perthite [294A-5]. (c) Perthite showing characteristic lack of exsolution lamellae in

the rim, crossed polars [294A-1]. (d) Backscattered electron image of perthite (Sample T)

showing plagioclase clustered along cracks and at grain boundaries [278A-1] (e) Composite

backscattered electron image of antiperthites with coarse exsolution lamellae approaching 1 mm

in length [278A-1] (f) Enlarged view of a K-feldspar lamellae in antiperthite of panel e, showing

1138 exsolution lamellae of plagioclase.

**FIGURE 8**: Photomicrograph of texturally late sillimanite in thin section cross-cutting foliation;crossed polars [269A-1].

FIGURE 9: Results of ternary feldspar reintegration thermometry for antiperthites (numbers) and 1141 1142 perthites (letters) plotted on the feldspar ternary diagram. Temperature contours are from the feldspar activity models of Benisek et al. (2010) and Benisek et al. (2004). Sample codes 1-6 1143 1144 correspond to samples 150A-2, 271A-1, 278A-1 (3-5), and 281A-1 respectively. Sample codes 1145 A-F correspond to samples 294A-1 (A-C), 294A-3, and 312A-1(2x) respectively. Points E and F overlap. Points B and C come from different regions of the same grain. Point T is the perthite 1146 pictured in Fig. 7d, reintegrated to include grain boundary plagioclase, for which host and 1147 1148 lamellae compositions are given in Table 9. Temperature estimates are given in Table 10. White 1149 dotted line indicates the tie line for reintegrated antiperthite and perthite including grain 1150 boundary plagioclase from the same sample. FIGURE 10: Pseudosections constructed for samples 150A, 268A, and 269A. Grt = garnet, crn = 1151 corundum, cpx = clinopyroxene, wm = white mica, bi = biotite, ksp = K-feldspar, ilm = ilmenite, 1152 1153 plag = plagioclase, sill = sillimanite, spl = spinel, ky = kyanite, lc = leucite, rt = rutile, gro = leucite, rtgrossular. Yellow band shows the mean  $\pm 2\sigma$  standard error of the antiperthite temperatures of all 1154 analyses (Table 6) using the feldspar activity models of Benisek et al. (2004) and Benisek et al. 1155

1156 (2010).

1150				
1159		Type I	Typ	e II
1160		150A	268A	269A
1161	SiO <sub>2</sub> (wt%)	38.63	40.80	44.10
	TiO <sub>2</sub>	2.38	2.18	1.94
1162	Al <sub>2</sub> O <sub>3</sub>	27.69	24.90	23.00
1163	FeO(tot)	18.74	17.38	16.39
	Fe <sub>2</sub> O <sub>3</sub>	0.21	0.19	0.18
1164	FeO	18.53	17.19	16.21
1165	MgO	2.94	4.01	3.60
1100	MnO	0.19	0.07	0.07
1100	CaO	0.74	0.32	0.88
1167	Na <sub>2</sub> O	0.65	0.66	1.16
1168	K <sub>2</sub> O	6.19	7.97	7.39
1100	P <sub>2</sub> O <sub>5</sub>	0.35	0.14	0.13
1169	LOI	1.79	2.15	2.12
1170	Total	99.94	100.44	100.65
	mol Fe <sup>3+</sup> /Fe(tot)	0.01	0.01	0.01
1171	Density @ 1040 °C, 1.8 GPa (g/cm <sup>3</sup> )	3.09	2.98	2.95
1172	Density (-50% melt)	3.25	3.09	3.05
	Density (-100% melt)	3.52	3.26	3.22
11/3				

**Table 1**. Bulk compositions of samples used for pseudosection modeling

1174 Notes: Compositions were obtained by X-ray fluorescence (XRF). CaO adjusted assuming all 1175  $P_2O_5$  is hosted in apatite. LOI denotes loss on ignition, and is re-calculated from raw values to 1176 account for Fe<sub>2</sub>O<sub>3</sub> created during XRF combustion assuming Fe<sup>3+</sup>/Fe(tot) ratio of 0.01 (see 1177 Analytical Methods). Densities calculated using Theriak-Domino.

1178 <b>Table 2</b> .	Garnet analyses
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1180		150A-3	150A-3	268A-1	268A-2	268A-2	269A-1	269A-1	294A-1	150A-3	268A-1	268A-2	269A-1
1101	Sample	Core	Rim	Gt <sub>u</sub> Core	Core	Rim	Core	Rim	Core	Range	Gt <sub>II</sub> Range	Range	Range
1101		core	Rum		core	Ittill	core	Rum	core	n=13	n=10	<i>n=6</i>	n=11
1182	$SiO_2$ (wt%)	37.86	37.32	37.40	38.01	37.30	37.88	37.10	37.10	1.16	0.26	0.71	1.04
-	TiO <sub>2</sub>	0.06	0.01	0.01	0.01	0.03	b.d.	0.05	0.04	0.06	0.03	0.07	0.05
1183	$P_2O_5$	0.04	0.03	0.05	0.05	0.02	0.03	0.03	0.18	0.02	0.02	0.07	0.04
	$Al_2O_3$	21.35	21.00	21.24	21.47	21.14	21.50	20.97	21.16	0.70	0.10	0.50	0.68
1184	$Cr_2O_3$	0.05	0.01	0.05	0.03	0.04	0.05	0.05	0.09	0.07	0.05	0.05	0.03
1105	FeO(tot)	34.44	36.73	37.80	35.13	37.91	34.59	37.24	36.60	4.01	0.81	3.62	3.53
1105	MgO	4.79	3.36	2.95	4.73	2.64	4.79	2.87	3.90	2.55	0.24	2.09	2.43
1186	MnO	0.93	0.52	0.16	0.46	0.29	0.38	0.41	0.25	0.45	0.06	0.29	0.11
	CaO	1.62	1.29	0.98	1.39	1.48	1.34	1.47	1.16	0.74	0.31	0.48	0.18
1187	Na <sub>2</sub> O	0.03	0.01	0.02	0.01	0.01	0.01	b.d.	0.02	0.02	0.01	0.01	0.01
	Total	101.17	100.28	100.66	101.33	100.86	100.57	100.19	100.50				
1188				Stru	ictural Fo	rmulas (12	<b>2 O</b> )						
1189	Si	2.979	2.994	3.000	2.989	2.991	2.997	2.990	2.962				
1105	Ti	0.004		0.001	0.001	0.002		0.003	0.002				
1190	Р	0.003	0.002	0.003	0.004	0.001	0.002	0.002	0.012				
	Al	1.980	1.985	2.008	1.990	1.998	2.005	1.992	1.991				
1191	Cr	0.003		0.003	0.002	0.003	0.003	0.003	0.006				
1100	Fe <sup>3+</sup>	0.048	0.024		0.021	0.012		0.012	0.042				
1192	Fe <sup>2+</sup>	2.219	2.441	2.553	2.289	2.530	2.295	2.498	2.401				
1193	Mg	0.562	0.402	0.353	0.555	0.316	0.565	0.345	0.464				
1100	Mn	0.062	0.035	0.011	0.031	0.020	0.025	0.028	0.017				
1194	Ca	0.137	0.111	0.084	0.117	0.127	0.113	0.127	0.099				
	Na	0.005	0.001	0.003	0.002	0.002	0.001		0.004				
1195													

1196 Notes: b.d. = below detection. Structural formula  $Fe^{2+}$  and  $Fe^{3+}$  calculated using 8 cations and 12 oxygens.  $Gt_{II}$  is the second

1197 generation noted in the Results section of the text. Range denotes the difference between maximum and minimum values across all

1198 grains.

1200							
1201							
1202	Sample	268A-1 Core	268A-1 Rim	268A-2 Grt incl	269A-1 Core	269A-1 Rim	294A-1 Core
1203	SiO <sub>2</sub> (wt%)	b.d.	0.07	0.02	0.01	b.d.	b.d.
	TiO <sub>2</sub>	0.02	0.04	0.01	0.01	0.02	0.21
1204	$Al_2O_3$	56.30	56.03	57.70	56.16	56.09	55.58
1205	$Cr_2O_3$	0.67	1.18	0.93	0.92	1.29	1.41
1205	FeO(tot)	40.38	39.42	36.28	39.05	39.00	39.60
1206	Fe <sub>2</sub> O <sub>3</sub>	4.02	2.74	2.48	3.26	2.85	2.87
1206	FeO	36.76	36.96	34.05	36.12	36.43	37.01
1207	MgO	2.68	2.27	4.49	2.81	2.62	2.38

b.d.

0.14

0.21

0.06

100.86

1.896

0.015

0.086

0.879

0.114

0.003

0.004

b.d.

0.11

0.42

0.07

99.93

0.002

0.001

1.906

0.027

0.059

0.892

0.098

0.003

0.009

0.01

0.15

0.12

0.06

0.001

1.923

0.021

0.053

0.805

0.189

0.003

0.003

100.02

0.01

0.11

0.36

0.10

99.87

Structural formulas (4 O)

1.905

0.021

0.071

0.870

0.121

0.003

0.008

### Table 3. Spinel analyses 1199

MnO

NiO

ZnO

 $V_2O_3$ 

Total

Si

Ti

Al

Cr

Fe<sup>3+</sup>

 $\mathrm{Fe}^{2+}$ 

Mg

Mn

Ni

Zn

1200

1207

1208

1209

1210

1211

1212

1213

1214

1215	Zn	0.004	0.009	0.003	0.008	0.006	0.004	0.005	0.002
	V	0.001	0.002	0.001	0.002	0.001	0.001	0.001	0.001
1216	Notes: b.d. = below det	tection. Fe	O and Fe	2O3 wt%	and Fe <sup>2</sup>	<sup>+</sup> and Fe <sup>3</sup>	<sup>3+</sup> in strue	ctural for	mulas calculated using 3 cations and 4 oxygens
1217	Range denotes the diffe	erence bet	ween max	ximum ai	nd minin	num valu	les across	s all grain	ns.

b.d.

0.13

0.29

0.06

1.906

0.029

0.062

0.879

0.113

0.003

0.006

99.79

b.d.

0.13

0.17

0.06

99.83

0.005

1.895

0.032

0.063

0.895

0.103

0.003

0.004

277A-3

Reint core

*n*=25

b.d.

0.10

56.59

0.06

39.40

3.91

35.88

3.26

0.01

0.09

0.08

0.04

100.02

0.002

1.909

0.001

0.084

0.859

0.139

0.002

0.002

294A-1

Rim

0.02

0.09

52.62

4.39

2.74

36.76

2.01

0.07

0.04

0.21

0.06

99.01

0.001

0.002

1.830

0.102

0.061

0.907

0.088

0.002

0.001

0.005

39.23

268A-1

Range

n=13

0.06

0.09

1.39

0.76

1.17

0.59

0.04

0.12

0.26

0.03

269A-1

Range

n=7

0.05

0.06

0.96

0.98

1.62

0.39

0.08

0.08

0.16

0.05

294A-1

Range

n=11

0.03

0.21

3.26

4.07

1.84

0.60

0.07

0.15

0.14

0.05

1220							
1221	Sample	296A-3	296A-4A	150A-3	268A-2	269A-1	294A
	Sample	crn	crn	ilm	ilm	ilm	ilm
1222	SiO <sub>2</sub>	b.d.	b.d.	b.d.	0.05	0.05	0.05
1223	TiO <sub>2</sub>	0.01	0.03	51.93	52.08	51.95	50.20
1004	$Al_2O_3$	100.60	99.74	0.04	0.03	0.03	0.05
1224	$Cr_2O_3$	0.03	0.17	0.01	b.d.	b.d.	0.08
1225	$V_2O_3$	b.d.	b.d.	0.37	0.34	0.22	b.d.
1226	FeO(tot)	0.30	0.83	47.64	47.50	47.00	47.66
1220	Fe <sub>2</sub> O <sub>3</sub>	0.33	0.98	1.48	1.07	0.54	4.18
1227	FeO			46.31	46.53	46.52	43.89
1228	MgO	b.d.	b.d.	0.15	0.18	0.06	0.70
1220	MnO	0.01	b.d.	0.07	0.01	0.12	0.03
1229	NiO	b.d.	b.d.	0.02	0.02	b.d.	0.01
1230	ZnO	b.d.	0.01	0.04	b.d.	0.05	b.d.
	CaO	b.d.	0.06	b.d.	0.01	b.d.	0.04
1231	Total	100.98	100.99	100.43	100.30	99.53	99.22
1232			Structural	l formulas	s ( <b>3 O</b> )		
1222	Si				0.001	0.001	0.001
1233	Ti			0.982	0.985	0.991	0.957
1234	Al	1.995	1.984	0.001	0.001	0.001	0.002
1005	Cr		0.002				0.002
1255	V			0.008	0.007	0.005	
1236	Fe <sup>3+</sup>	0.004	0.012	0.028	0.020	0.010	0.080
1237	Fe <sup>2+</sup>			0.973	0.979	0.986	0.931
1237	Mg			0.006	0.007	0.002	0.026
1238	Mn			0.002		0.003	0.001
1239	Ni						
	Zn			0.001		0.001	
1240	Ca		0.001				0.001
							2

1219 **Table 4**. Corundum and ilmenite analyses

1220

1241 Notes: b.d. = below detection. FeO and  $Fe_2O_3$  wt% and structural formula  $Fe^{2+}$  and  $Fe^{3+}$ 1242 calculated using 2 cations and 3 oxygens.

1744											
1277	Sample	150A-3	268A-1	268A-1	269A-1	269A-1	294A-1	294A-1			
1245	$SiO_2$ (wt%)	34.40	34.86	34.84	33.79	34.28	33.85	34.57			
1246	TiO <sub>2</sub>	2.83	4.02	4.49	4.72	4.71	4.84	6.61			
	$Al_2O_3$	20.17	19.75	17.95	18.07	18.22	18.17	16.94			
1247	$Cr_2O_3$	0.11	0.20	0.11	0.16	0.14	0.17	0.22			
1248	FeO	22.45	14.66	20.91	21.53	21.99	23.54	21.00			
1240	MgO	7.53	12.71	8.94	7.81	7.73	6.95	8.09			
1249	MnO	b.d.	0.07	b.d.	0.07	b.d.	0.05	b.d.			
1250	NiO	0.13	0.05	b.d	0.05	0.06	0.06	b.d.			
1251	ZnO	b.d.	b.d.	b.d.	b.d.	b.d.	b.d.	0.02			
1201	BaO	0.11	0.10	0.11	0.10	0.12	0.09	0.11			
1252	Na <sub>2</sub> O	0.07	0.35	0.12	0.13	0.09	0.11	0.18			
1253	K <sub>2</sub> O	9.37	9.65	9.77	9.67	9.84	9.73	9.66			
1254	F	0.10	0.28	0.22	0.19	0.16	0.06	0.22			
1254	Cl	0.02	0.01	0.01	0.03	0.02	0.06	0.01			
1255	Total	97.23	96.58	97.39	96.23	97.29	97.63	97.53			
1756	Structural formulas (11 O)										
1250	Si	2.606	2.570	2.627	2.596	2.606	2.587	2.611			
1257	Ti	0.161	0.223	0.255	0.273	0.270	0.278	0.375			
1258	Al	1.801	1.716	1.595	1.635	1.633	1.636	1.508			
	Cr	0.006	0.012	0.007	0.009	0.008	0.010	0.013			
1259	Fe <sup>2+</sup>	1.422	0.904	1.318	1.383	1.398	1.504	1.326			
1260	Mg	0.850	1.397	1.005	0.895	0.876	0.792	0.910			
1061	Mn		0.004		0.004		0.003				
1201	Ni		0.003		0.003	0.004	0.004				
1262	Zn							0.001			
1263	Ba	0.003	0.003	0.003	0.003	0.004	0.003	0.003			
	Na	0.010	0.049	0.017	0.020	0.014	0.016	0.027			
1264	Κ	0.905	0.908	0.940	0.947	0.955	0.947	0.931			
1265	F	0.024	0.065	0.053	0.046	0.039	0.016	0.053			
1266	Cl	0.002	0.002	0.001	0.004	0.003	0.008	0.002			

**Table 5**. Biotite analyses

1267 Note: b.d. = below detection. The analysis from sample 294A-1 with the highest Ti is an

inclusion in the core of the garnet with 0.18 wt%  $P_2O_5$  (294A-1 Core in Table 2). All Fe as FeO.

1270											
	Samula	150A-2	271A-1	278A-1A	278A-1B	278A-1C	281A-1				
1271	Sample	n=10	n=10	n=5	n=5	<i>n</i> =12	n=5				
4070	SiO <sub>2</sub>	59.48(0.25)	57.24(0.16)	57.73(0.17)	57.93(0.18)	58.22(0.21)	58.28(0.16)				
1272	$Al_2O_3$	26.27(0.15)	27.53(0.19)	27.06(0.14)	26.92(0.05)	27.23(0.14)	27.04(0.07)				
1273	FeO	0.01(0.01)	0.03(0.03)	0.02(0.01)	0.03(0.03)	0.03(0.03)	0.04(0.04)				
1275	MgO	b.d.	0.01(0.01)	b.d.	b.d.	0.01(0.01)	0.02(0.01)				
1274	MnO	0.02(0.03)	0.01(0.02)	0.01(0.02)	0.01(0.03)	0.02(0.03)	0.01(0.02)				
	CaO	7.72(0.08)	9.39(0.10)	8.77(0.11)	8.65(0.12)	8.73(0.09)	8.54(0.10)				
1275	BaO	0.01(0.01)	0.01(0.01)	0.01(0.01)	0.02(0.01)	0.01(0.01)	0.01(0.01)				
1070	Na <sub>2</sub> O	6.99(0.10)	6.11(0.09)	6.41(0.10)	6.42(0.06)	6.48(0.09)	6.56(0.08)				
1276	$K_2O$	0.20(0.01)	0.19(0.02)	0.21(0.01)	0.20(0.03)	0.21(0.02)	0.20(0.02)				
1277	Total	100.7(0.41)	100.51(0.31)	100.21(0.18)	100.18(0.25)	100.92(0.22)	100.70(0.15)				
		Structural formulas (8 O)									
1278	Si	2.635	2.554	2.579	2.588	2.582	2.589				
	Al	1.372	1.448	1.425	1.417	1.423	1.416				
1279	Fe		0.001	0.001	0.001	0.001	0.002				
1200	Mg						0.001				
1280	Mn	0.001	0.001			0.001					
1281	Ca	0.366	0.449	0.420	0.414	0.415	0.407				
-	Ba										
1282	Na	0.600	0.529	0.555	0.556	0.557	0.565				
	K	0.011	0.011	0.012	0.011	0.012	0.011				
1283											

**Table 6**. Host compositions from reintegrated antiperthites

1284 Notes: b.d. = below detection.  $2\sigma$  standard deviations given in parentheses. All Fe as FeO.

1286							
	Samula	150A-2	271A-1	278A-1A	278A-1B	278A-1C	281A-1
1287	Sample	<i>n</i> =12	n=10	<i>n</i> =17	<i>n</i> =12	<i>n</i> =26	n=8
	$SiO_2$	64.40(0.29)	65.37(0.20)	64.90(0.3)	65.32(0.27)	65.59(0.54)	64.88(0.17)
1288	$Al_2O_3$	18.87(0.13)	18.79(0.15)	18.82(0.19)	18.77(0.11)	18.85(0.04)	19.01(0.11)
1280	FeO	0.04(0.03)	0.02(0.02)	0.02(0.03)	0.02(0.02)	0.02(0)	0.05(0.03)
1209	MgO	b.d.	b.d.	b.d.	b.d.	b.d.	A-1C $281A-1$ $=26$ $n=8$ $(0.54)$ $64.88(0.17)$ $(0.04)$ $19.01(0.11)$ $(0)$ $0.05(0.03)$ b.d. $(0)$ $(0,01)$ $0.07(0.01)$ $5(0.02)$ $0.79(0.04)$ $(0.02)$ $0.79(0.04)$ $(0.09)$ $1.17(0.06)$ $3(0.09)$ $14.67(0.13)$ $100.65(0.17)$ $994$ $2.978$ $034$ $1.028$ $001$ $0.002$ $003$ $0.003$ $002$ $0.014$ $087$ $0.104$
1290	MnO	0.01(0.02)	0.02(0.02)	0.01(0.02)	0.02(0.03)	0.01(0)	0.01(0.01)
	CaO	0.07(0.02)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.06(0.11)	0.07(0.01)		
1291	BaO	0.52(0.03)	0.20(0.02)	0.21(0.04)	0.12(0.02)	0.13(0.02)	0.79(0.04)
	Na <sub>2</sub> O	1.10(0.06)	1.01(0.03)	1.01(0.05)	1.01(0.03)	0.99(0.09)	1.17(0.06)
1292	$K_2O$	14.80(0.10)	15.17(0.13)	15.12(0.18)	15.23(0.11)	$278A-1B$ $278A-1C$ $281A-1$ $n=12$ $n=26$ $n=8$ $65.32(0.27)$ $65.59(0.54)$ $64.88(0.17)$ $18.77(0.11)$ $18.85(0.04)$ $19.01(0.11)$ $0.02(0.02)$ $0.02(0)$ $0.05(0.03)$ b.d.b.d.b.d. $0.02(0.03)$ $0.01(0)$ $0.01(0.01)$ $0.07(0.02)$ $0.06(0.11)$ $0.07(0.01)$ $0.12(0.02)$ $0.13(0.02)$ $0.79(0.04)$ $1.01(0.03)$ $0.99(0.09)$ $1.17(0.06)$ $15.23(0.11)$ $15.18(0.09)$ $14.67(0.13)$ $100.56(0.32)$ $100.84(0.18)$ $100.65(0.17)$ $\mathcal{D}$ $2.992$ $2.994$ $2.978$ $1.013$ $1.034$ $1.028$ $0.001$ $0.003$ $0.003$ $0.002$ $0.087$ $0.104$ $0.990$ $0.884$ $0.859$	
1203	Total	99.80(0.43)	100.65(0.35)	100.17(0.42)	100.56(0.32)	100.84(0.18)	100.65(0.17)
1295				Structural formula	as (8 O)		
1294	Si	2.978	2.992	2.986	2.992	2.994	2.978
	Al	1.028	1.013	1.020	1.013	1.034	1.028
1295	Fe	0.001	0.001	0.001	0.001	0.001	0.002
1000	Mg						
1296	Mn	0.001	0.001		0.001		
1297	Ca	0.004	0.003	0.004	0.003	0.003	0.003
1237	Ba	0.009	0.004	0.004	0.002	0.002	0.014
1298	Na	0.098	0.090	0.090	0.090	0.087	0.104
	K	0.873	0.886	0.887	0.890	0.884	0.859
1299							

**Table 7**. Exsolution lamellae compositions from reintegrated antiperthites

1300 Notes: b.d. = below detection.  $2\sigma$  standard deviations given in parentheses. All Fe as FeO.

1302							
1202	Samula	294A-1*	294A-1*	294A-1	294A-3	312A-1	312A-1
1303	Sample	<i>n</i> =25	n=25	<i>n</i> =25	<i>n</i> =25	<i>n</i> =25	<i>n</i> =25
1304	SiO <sub>2</sub>	64.30(0.35)	64.29(0.27)	64.34(0.35)	64.41(0.17)	64.10(0.35)	64.70(0.16)
	$P_2O_5$	0.29(0.03)	0.34(0.04)	0.34(0.22)	0.25(0.03)	0.19(0.02)	0.20(0.03)
1305	$Al_2O_3$	19.48(0.44)	19.39(0.31)	19.30(0.23)	19.19(0.10)	19.35(0.34)	19.61(0.17)
	FeO	0.02(0.02)	0.02(0.03)	0.03(0.03)	0.04(0.03)	0.03(0.04)	0.09(0.03)
1306	MgO	b.d.	b.d.	0.01(0.01)	b.d.	0.01(0.02)	0.02(0.01)
1207	MnO	0.01(0.02)	0.01(0.02)	0.01(0.01)	0.01(0.02)	0.01(0.02)	0.01(0.02)
1307	CaO	0.40(0.43)	0.28(0.30)	0.40(0.30)	0.22(0.04)	0.48(0.25)	0.45(0.12)
1308	BaO	0.14(0.03)	0.15(0.03)	0.14(0.04)	0.19(0.03)	0.40(0.05)	0.38(0.04)
	Na <sub>2</sub> O	2.13(0.48)	1.96(0.32)	2.36(0.35)	1.76(0.06)	3.01(0.60)	3.08(0.35)
1309	$K_2O$	13.43(1.03)	13.76(0.69)	13.17(0.59)	14.01(0.13)	12.02(1.01)	11.91(0.56)
	Total	100.20(0.25)	100.20(0.23)	100.10(0.27)	100.09(0.22)	99.61(0.43)	100.45(0.19)
1310				Structural formulas	s (8 O)		
1311	Si	2.943	2.945	2.946	2.957	2.945	2.944
1311	Р	0.011	0.013	0.013	0.010	0.007	0.008
1312	Al	1.051	1.047	1.041	1.038	1.048	1.052
	Fe	0.001	0.001	0.001	0.002	0.001	0.003
1313	Mg					0.001	0.001
	Mn					0.001	
1314	Ca	0.020	0.014	0.020	0.011	0.024	0.022
1315	Ba	0.003	0.003	0.003	0.003	0.007	0.007
1919	Na	0.189	0.174	0.210	0.157	0.268	0.272
1316	K	0.784	0.804	0.769	0.820	0.705	0.691

1301 **Table 8**. Reintegrated perthite compositions measured by beam grids

1318 Notes: b.d. = below detection.  $2\sigma$  standard deviations given in parentheses. All Fe as FeO. \*Analyses come from two different 1319 portions of the same grain.

-					
1322	Sample	278A-1 host	278A-1 lamellae		
	Sample	n=5	n=5		
1323	SiO <sub>2</sub>	65.32(0.14)	57.60(0.46)		
1324	$Al_2O_3$	18.71(0.15)	26.97(0.10)		
1005	FeO	0.01(0.02)	0.03(0.03)		
1325	MgO	b.d.	b.d.		
1326	MnO	0.01(0.02)	b.d.		
1007	CaO	0.06(0.02)	8.82(0.20)		
1527	BaO	0.20(0.03)	0.01(0.01)		
1328	Na <sub>2</sub> O	1.10(0.06)	6.45(0.09)		
1329	$K_2O$	14.96(0.11)	0.21(0.03)		
1323	Total	100.37(0.27)	100.10(0.36)		
1330	Structural formulas (8 O)				
1331	Si	2.995	2.583		
	Al	1.011	1.418		
1332	Fe	0.001	0.001		
1333	Mg				
1224	Mn				
1334	Ca	0.003	0.421		
1335	Ba	0.004			
1226	Na	0.098	0.558		
1330	Κ	0.875	0.012		
1337					

1320	Table 9. Reintegrated perthite composition with grain boundary plagioclass	e
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1339 Notes: b.d. = below detection.  $2\sigma$  standard deviations given in parentheses. All Fe as FeO.

1341									
1242	Perthite								
1342	Sample	294A-1	294A-1	294A-1	294A-3	312A-1	312A-1	278A-1	
1343	Plotting Code	А	В	С	D	Е	F	Т	
1344	Kfs	79.0	81.0	77.0	83.1	70.2	70.7	77.3	
1245	Ab	19.0	17.6	21.0	15.9	27.6	26.9	16.5	
1345	An	2.0	1.4	2.0	1.1	2.2	2.4	6.2	
1346	T(°C) @ 1.5 GPa [B10]	893	848	885	822	874	886	1118	
1347	T(°C) @ 1.5 GPa [B04]	780	738	792	707	840	843	914	
1017	Antiperthite								
1348	Sample	150A-2	271A-1	278A-1	278A-1	278A-1	281A-1	-	
1349	Plotting Code	1	2	3	4	5	6		
1350	Kfs	16.1	22.8	18.9	19.3	23.2	18.3		
	Ab	52.7	42.7	46.9	47.0	44.9	48.6		
1351	An	31.2	34.5	34.2	33.7	31.9	33.1		
1352	T(°C) @ 1.5 GPa [B10]	979	1114	1055	1054	1084	1034		
1353	T(°C) @ 1.5 GPa [B04] Two-feldspar <i>T</i>	985	1050	1019 1059	1022 1058	1047 1073/1081	1012		

1340 **Table 10**. Reintegrated ternary feldspar compositions plotted in Figure 9

Notes: Abbreviations of feldspar models are B04 = Benisek et al. (2004) and B10 = Benisek et al. (2010). Perthite compositions are
averages of grid analyses (Table 8) except 278A-1 plotting code T, which is based on spot analyses of host and grain boundary

1357 plagioclase (Table 9) reintegrated following the procedure used for antiperthite. K-feldspar (Kfs), albite (Ab), and anorthite (An) are

1358 given in mol%. Two-feldspar *T* estimates all use the same Kfs composition (278A-1 plotting code T). Plotting codes refer to Fig. 9.





Dbl - Littleton Fm.

**DSs - Scotland Schist** 

SOb - Bigelow Brook Fm.

SOh - Hebron Gneiss



**Obr - Brimfield Schist** 



Ota - Tatnic Hill Fm.



Zw - Waterford Group

Intrusive/metaintrusive rocks



Dc - Canterbury Gneiss Other granitic rocks

Quartz diorite, Norite

## Fault

Sample Location





# leucosome


















