

Day

Residual lunar sulfide and metal – Revision 2

1 Geochemical constraints on residual metal and sulfide in
2 the sources of lunar mare basalts

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18 **Abstract**

19 Low oxygen fugacity (fO_2) in the lunar interior (one log unit below the iron-wüstite buffer [IW-
20 1]) offers the possibility that stable Fe-metal and sulfide phases exist as restites within lunar
21 mare basalt source regions. Metal and sulfide phases have high metal-melt and sulfide-melt
22 partition coefficients for chalcophile, siderophile (>100), and highly siderophile elements
23 ($\gg 100,000$ - HSE: Os, Ir, Ru, Rh, Pt, Pd, Re, Au). If these phases are residual after mare basalt
24 extraction, they would be expected to retain significant quantities of these elements, likely
25 generating non-chondritic HSE inter-element ratios, including Re/Os in the silicate magma. If
26 such phases were present, the estimated HSE abundances of the bulk silicate Moon (BSM)
27 would be proportionally higher than current estimates ($0.00023 \pm 2 \times$ CI chondrite), and perhaps
28 closer to the bulk silicate Earth (BSE) estimate ($0.009 \pm 2 \times$ CI chondrite). Here I show that
29 relationships between elements of similar incompatibility but with siderophile (W), chalcophile
30 (Cu) and lithophile tendencies (Th, U, Yb) do not deviate from expected trends generated by
31 magmatic differentiation during cooling and crystallization of mare basalts. These results,
32 combined with chondrite-relative HSE abundances and near-chondritic measured $^{187}\text{Os}/^{188}\text{Os}$
33 compositions of primitive high-MgO mare basalts, imply that lunar mantle melts were generated
34 from residual metal- and sulfide-free sources, or experienced complete exhaustion of metal and
35 sulfides during partial melt extraction. Evidence for the loss of moderately volatile elements
36 during lunar formation and early differentiation indicates that the BSM is >4 to 10 times more
37 depleted in S than BSE. Because of an S-depleted BSM, mare basalt melts are unlikely to have
38 reached S saturation, even if sulfide concentration at sulfide saturation (SCSS) was lowered
39 relative to terrestrial values due to low lunar fO_2 . In the absence of residual sulfide or metal,
40 resultant partial melt models indicate that a lunar mantle source with 25 to 75 $\mu\text{g g}^{-1}$ S and high

41 sulfide-melt partition coefficients can account for the chondritic relative abundances of the HSE
42 in mare basalts from a BSM that experienced <0.02% by mass of late accretion.

43

44 Keywords: Moon; siderophile elements; sulfide; metal; saturation; partial melting

45

46 **Introduction**

47 The bulk composition of the Moon places fundamental constraints on its formation (e.g.,
48 [Taylor et al., 2006](#); [Warren & Taylor, 2014](#)). No lunar mantle samples are available for study,
49 making estimates of the bulk silicate Moon (BSM) feasible only by extrapolation of the
50 composition of lunar surface rocks, including ferroan anorthosites and mare basalts. The
51 concentrations of the highly siderophile elements (HSE: Os, Ir, Ru, Rh, Pt, Pd, Re, Au) for the
52 BSM have been estimated by measurements of crustally uncontaminated mare basalts, and
53 regression of these data to an assumed mare basalt source (aka, the lunar mantle) using MgO-
54 regression methods ([Warren et al., 1989](#); [Day et al., 2007](#); [Day & Walker, 2015](#)). Since mare
55 basalts come from partial melting of the lunar interior, and the great majority of the HSE should
56 be in their olivine + pyroxene ± ilmenite bearing sources relative to ultra-low HSE abundance
57 crustal rocks ([Day et al., 2010](#)), then the BSM can be estimated directly to have chondrite-
58 relative HSE abundances $\sim 0.00023 \pm 2 \times$ CI chondrites, with present-day $^{187}\text{Os}/^{188}\text{Os}$ of ~ 0.1293
59 ([Day et al., 2007](#); [2016](#)). The MgO-regression method has also been used to obtain the bulk
60 silicate Earth (BSE) HSE composition from terrestrial lavas (e.g., [Day et al., 2007](#); [2016](#)),
61 resulting in good agreement with the chondrite-relative BSE HSE abundances determined from
62 peridotites ($0.009 \pm 2 \times$ CI chondrite, $^{187}\text{Os}/^{188}\text{Os} = 0.1296 \pm 8$; [Meisel et al., 2001](#); [Becker et al.,](#)
63 [2006](#); [Day et al., 2017](#)) (**Figure 1**).

64

65 The BSM HSE estimate is more than 30 times lower than the BSE HSE estimate, a
66 depletion that has been attributed to a reduced contribution from late accretion materials to the
67 Moon relative to Earth ([Walker et al., 2004](#); [Day et al., 2007](#); [Day & Walker, 2015](#)). In this
68 model, draw-down of the HSE during the formation of the Moon in a giant impact event, and/or
69 during the formation of a small lunar core (e.g., [Weber et al., 2011](#)) led to nearly-quantitative
70 depletion of the HSE in the BSM, followed by limited late accretion of broadly chondritic
71 material to the Moon (~ 0.02 wt.%, or $\sim 1.7 \times 10^{19}$ kg) relative to Earth (~ 0.5 wt.%, or $\sim 2 \times 10^{22}$
72 kg). This observation has been widely used as a constraint for dynamical models of planetary
73 accretion (e.g., [Bottke et al., 2010](#); [Schlichting et al., 2012](#); [Jacobsen et al., 2014](#)) and led to the
74 recent interpretation of W isotopic differences between Earth and the Moon as representing
75 disproportionate late accretion ([Touboul et al., 2015](#); [Kruijer et al., 2015](#)).

76

77 An underlying concern when using mare basalts to determine the BSM HSE composition
78 is that regression methods, similar to those used to determine the BSE HSE composition using
79 terrestrial lavas, may not be effective due to substantially different oxygen (fO_2) and sulfur (fS_2)
80 fugacities for the two planetary bodies. It has been constrained that the Moon's mantle has a low
81 fO_2 , at around one log unit below the iron-wüstite (IW) buffer (e.g., [Sato et al., 1973](#); [Wadhwa,](#)
82 [2008](#)), whereas the Earth's mantle has greater than four orders of magnitude higher oxygen
83 activity, at the fayalite-magnetite-quartz (FMQ) buffer (e.g., [Sato, 1978](#); [Haggerty, 1978](#); [Frost](#)
84 [and McCammon, 2008](#)). It is therefore possible that metal is more stable and residual within
85 lunar mare basalt regions after partial melting than in Earth's mantle. Equally, low fO_2 , might
86 lead to decreased sulfide concentration at sulfide saturation (SCSS) in mare basalts ([Brenan et](#)

87 [al., 2016](#)), and thus favor the survival of residual sulfides during melting. The term SCSS
88 represents the maximum amount of sulfur a silicate melt can dissolve and provides an upper
89 boundary to S concentrations found in melts at reduced fO_2 conditions. Here I examine the
90 possible role of residual metal or sulfide in the source during partial melting of mare basalts,
91 focusing first on constraints placed on these processes by the HSE abundances and $^{187}\text{Os}/^{188}\text{Os}$ of
92 mare basalts.

93

94 **Constraints from the HSE and $^{187}\text{Os}/^{188}\text{Os}$ ratios of mare basalts**

95 Mare basalts are ~3 to 3.9 Ga magmatically-derived rocks that formed by partial-melting
96 of source regions within the Moon with mineralogy dominated by olivine + pyroxene ± ilmenite
97 (e.g., [Snyder et al., 1992](#)). Mare basalt mantle sources are considered to have formed from
98 crystallization of a lunar magma ocean resulting in initial mineralogical stratification followed by
99 later mixing (e.g., [Warren & Taylor, 2014](#)). Two main types of mare basalts have been
100 identified: low Ti variants (<5 wt.% TiO_2) that are considered to make up most of the mare
101 basalts exposed at the lunar surface ([Giguere et al., 2000](#)) and that were primarily sampled at the
102 Apollo 12 and 15 landing sites, and as mare basalt meteorites; and high Ti mare basalts (>6 wt.%
103 TiO_2) that were primarily sampled at the Apollo 11 and 17 landing sites ([Neal & Taylor, 1992](#)).
104 Despite originating from mineralogically variable sources, the most primitive and crustally
105 uncontaminated high-MgO mare basalts from the Apollo 12 (samples 12004, 12040), 15
106 (samples 15016, 15555) and 17 sites (sample 74255) all have chondrite-relative abundances of
107 the HSE and measured $^{187}\text{Os}/^{188}\text{Os}$ within the range of chondrites (0.1265 to 0.1345) ([Day et al.,](#)
108 [2007](#); [Day & Walker, 2015](#)). The $^{187}\text{Os}/^{188}\text{Os}$ ratio of mare basalts is a particularly powerful
109 tracer of composition because the ancient crystallization ages of these rocks (3 to 3.9 Ga) only

110 permit small deviations from a chondritic Re/Os ratio to obtain nearly chondritic $^{187}\text{Os}/^{188}\text{Os}$
111 measured today.

112

113 A histogram plot of measured mare basalt $^{187}\text{Os}/^{188}\text{Os}$ compositions illustrates that many
114 of the samples have $^{187}\text{Os}/^{188}\text{Os}$ ratios that cluster around the range of chondrites, and within the
115 BSE composition, with a tail to more radiogenic ratios (**Figure 2**). The radiogenic $^{187}\text{Os}/^{188}\text{Os}$
116 ratios are found in low-MgO samples and reflect increasing Re/Os during magmatic
117 differentiation (Day & Walker, 2015). The HSE abundances that are in broadly chondritic
118 proportions, and a Re/Os ratio within a few percent of the chondritic ratio are therefore key
119 constraints on any successful partial melting model for generating primitive high-MgO mare
120 basalts. Any fractionation of Re/Os away from the chondritic composition during metal-melt or
121 sulfide-melt partitioning, for example, would not be permitted by the available mare basalt data.

122

123 **Metal in mare basalt source regions**

124 The low $f\text{O}_2$ of the Moon leads to the possibility that residual metal may have been
125 present after partial melting to form mare basalts. Residual metal in the source would lead to
126 near-quantitative retention of the HSE and could account for the low HSE abundance of mare
127 basalts, if the chondritic proportions of the HSE can be reproduced by this process. Evidence for
128 metal formation at low $f\text{O}_2$ during late-stage (>90% crystallization) magmatic differentiation is
129 obtained from mare basalts themselves, which can contain ≤ 0.1 modal% FeNi-metal (e.g., Reid
130 et al., 1970; Day et al., 2006). Formation of metal grains within mare basalts, however, does not
131 equate to the presence of metal in mare basalt mantle sources. Reduction of Fe^{2+} to Fe^0 by
132 eruptive degassing of S or C, or by reactions with Ti^{3+} or Cr^{2+} have all been considered as likely

133 mechanisms for the presence of metal grains in mare basalts (e.g., [Sato et al., 1976](#); [Brett, 1976](#);
134 [Schreiber et al., 1982](#)). Reduction of Fe by H is also possible for lunar mare basalt melts, in the
135 reaction $\text{FeO}^{(+)} + \text{H}_2 \rightarrow \text{Fe}^{(+)} + \text{H}_2\text{O}$.

136

137 Previously, [Day & Walker \(2015\)](#) showed that W, U and Th are not strongly fractionated
138 in Apollo 12 mare basalts, indicating that no residual metal existed in the sources of these mare
139 basalts after their partial melt extraction. The choice of investigating W, U and Th is that these
140 elements behave similarly during partial melting and fractional crystallization, but that W is a
141 moderately siderophile element and U and Th are strongly incompatible lithophile trace
142 elements. Any residual metal in the source after mare basalt partial melt extraction should lead to
143 retention of W, but effective removal of Th and U. On the other hand, a lack of residual metal
144 would lead all three elements to behave as highly incompatible lithophile trace elements.

145

146 To further evaluate this possibility for known mare basalt sources, W, U and Th data are
147 reported for the Apollo 12 mare basalts presented in [Day & Walker \(2015\)](#), and for four mare
148 basalt meteorites (MIL 05035, LAP 02205, NWA 4734 and Dho 287), which, except for NWA
149 4734 and Dho 287, have previously been measured for HSE abundances and Os isotopes ([Day et](#)
150 [al., 2007](#); [Day & Walker, 2015](#)) (for methods and data, see **Table 1**). In a plot of W versus U
151 (**Figure 3**), residual metal in the source after mare basalt melt extraction would be expected to
152 lead to very low W abundances for a given U content. As with the Apollo 12 mare basalts, the
153 mare basalt meteorites fall along a linear correlation for W and U, and all samples have similar
154 U/Th (0.243 ± 0.028). The U/Th ratio constrains the relative proportions of these elements to be
155 approximately the same as those of chondrites (0.283; [Anders & Grevesse, 1989](#)), Earth (0.276;

156 [Taylor & McLennan, 1985](#)), and to the lunar surface (0.269 ± 0.007 ; [Yamashita et al., 2010](#)).
157 These results imply no significant modification of W/U ratios and indicate that metal is not a
158 residual phase in the mantle sources of mare basalts. Published data for W and U contents are
159 also available for the Apollo 11, 12, 15 and 17 mare basalts and, despite considerable scatter
160 relative to the newly presented data, have high W for a given U content implying lack of residual
161 metal during melt extraction.

162

163 Correlations of W versus U or Th, which involves equally incompatible trace elements,
164 argue strongly for either residual metal-free sources, or complete exhaustion of metal during
165 mare basalt partial melting. A collateral consequence of residual metal in the source of mare
166 basalts would also be to strongly fractionate the HSE in partial melts due to the variable but high
167 metal-silicate partition coefficients for the different HSE (e.g., [Mann et al., 2012](#)). Strong
168 fractionations of the HSE are not observed in the primitive high-MgO mare basalts, and HSE
169 patterns of low-MgO basalts can be entirely accounted for by near-surface crystal-liquid
170 fractionation ([Day & Walker, 2015](#)).

171

172 **Sulfur content of mare basalt source regions**

173 In the absence of evidence for residual metal in the source of mare basalts, it is possible
174 that residual sulfide can be retained in source regions during the genesis of mare basalts, due to
175 extremely low fO_2 . For the purposes of understanding mare basalt petrogenesis, there are two key
176 questions that need to be addressed. First, how much S is likely to be in mare basalt sources and,
177 second, how much S can be incorporated into the parental silicate magma without S saturation.
178 Sulfur is a chalcophile and volatile element (50% condensation temperature $[TC_{50}] = 664K$;

179 [Lodders, 2003](#)) and, along with some moderately volatile elements, including Zn and K (TC₅₀ of
180 726 and 1006K, respectively), has been shown to be depleted in the Moon, relative to Earth, due
181 to volatile depletion (e.g., [Paniello et al., 2012](#); [Day & Moynier, 2014](#); [Wing & Farquhar, 2015](#);
182 [Wang & Jacobsen, 2016](#)). Based on current estimates, this depletion is between three
183 ([Bombardieri et al., 2005](#)) and ten times greater in the BSM, relative to the BSE (**Figure 4**). This
184 leads to estimates of S in the BSM, or in the sources of mare basalts of between 75 and 25 $\mu\text{g g}^{-1}$,
185 using a BSE S content of 250 $\mu\text{g g}^{-1}$ ([McDonough & Sun, 1995](#)). These estimates are similar to
186 the inferred heterogeneous abundances of sulfur in the source regions of Apollo 11 (35-120 $\mu\text{g g}^{-1}$
187 ¹), 12 (27-92 $\mu\text{g g}^{-1}$), 15 (10-23 $\mu\text{g g}^{-1}$) and 17 (25-62 $\mu\text{g g}^{-1}$) mare basalts from recent
188 experimental constraints ([Ding et al., 2018](#)).

189

190 Regarding the second question of S-saturation of parental magmas, this is dependent on
191 the major-element composition of the magma, the composition of the immiscible sulfide, and the
192 temperature and pressure (e.g., [O'Neill & Marvogenes, 2002](#); [Holzheid & Grove, 2002](#)).
193 [Mavrogenes & O'Neill \(1999\)](#) showed that sulfur content at sulfide saturation (SCSS) is
194 relatively independent of $f\text{S}_2$ and $f\text{O}_2$, except if the melt composition changes in FeO content.
195 The amount of FeO can vary dramatically in mare basalts as a function of source and parental
196 magma differentiation. SCSS is also dependent on the nature of the immiscible sulfide, which at
197 low $f\text{O}_2$ will have $\text{Fe/S} > 1$, causing SCSS to fall. FeO content is therefore a dominant control on
198 SCSS. An important facet of these observations is that if magmas do not become S saturated
199 during differentiation then primitive parental melts were most likely not S-saturated (i.e., there
200 was no residual sulfide).

201

202 To empirically examine whether sulfide saturation was reached in lunar melts, FeO and S
203 contents of bulk rock mare basalts and of melt inclusions are shown in **Figure 5**, along with S-
204 saturation curves established from several studies. Degassing of S during eruption would drive
205 preserved S contents down in whole-rock mare basalts, but not in melt inclusions. In the bulk
206 rocks, and the majority of melt inclusions with FeO >15 wt.%, none of the low-Ti (Apollo 12,
207 Apollo 15) or high-Ti (Apollo 17) melts would be S-saturated relative to their respective
208 saturation curves, and the likelihood of residual S in the source would be low, consistent with
209 recent experimental evidence for S exhaustion during mare basalt partial melting ([Ding et al.,](#)
210 [2018](#)). It should be noted that melt inclusions with <~15 wt.% FeO have likely experienced
211 significant side-wall crystallization or equilibration with their host crystals, driving their
212 compositions to the left in **Figure 5**. Collectively, these lines of evidence imply that primitive
213 melts parental to mare basalts were likely undersaturated in sulfur.

214

215 Recently, [Brenan et al. \(2016\)](#) have reported - in abstract form - evidence that SCSS falls
216 rapidly as fO_2 approaches metal saturation, resulting in an S-saturated source. So long as metal
217 saturation is not reached, then in this scenario, even the melt inclusions with between 15 to 25
218 wt.% FeO in **Figure 5** would be close to the S-saturation curve, which would likely be below the
219 corrected melt inclusion S-saturation trend of [Bombardieri et al. \(2005\)](#) that considers the effects
220 of degassing of sulfur. Based on their experimental results, [Brenan et al. \(2016\)](#) have suggested
221 that sulfides would remain in the residue during partial melting, retaining the HSE. Following
222 this scenario, they postulate that the BSM HSE abundance may only be ~4 times less than the
223 BSE HSE contents.

224

225 **Melting processes during the formation of mare basalts**

226 To evaluate the role of sulfides during lunar mantle partial melting, melting models are
227 explored to reproduce melting effects on the HSE. Partitioning of the HSE between sulfide melt
228 and silicate melt has been shown to have D values of 10^3 to 10^6 in both experiments and studies
229 of natural samples (e.g., [Mungall & Brenan, 2014](#)). In a recent assessment of sulfide melt-silicate
230 melt partitioning experiments [Mungall & Brenan \(2014\)](#) showed that values of D can be as high
231 as $2\text{-}3 \times 10^6$ for Ir and Pt. Even though these experiments were conducted at ΔFMQ -0.7 to -2.4
232 (i.e., more oxidizing than lunar conditions, at $\sim\Delta\text{IW} +1.6$ to $+3$), they offer the best estimates of
233 sulfide-melt silicate-melt partitioning and the constructed models therefore explore D values
234 from low (10,000) to high (1,000,000; **Figure 6**). Three melting models are investigated: (1)
235 terrestrial conditions with $\sim 250 \mu\text{g g}^{-1}$ S and BSE HSE abundances from [Becker et al. \(2006\)](#) and
236 [Day et al. \(2017\)](#). This model indicates S exhaustion at $\sim 23\%$ partial melting (F); (2) a lunar
237 mantle with BSM HSE concentrations reported in [Day et al. \(2007; 2016\)](#) and source S contents
238 at the high end of the S estimate of the BSM ($75 \mu\text{g g}^{-1}$); (3) a lunar mantle with HSE
239 abundances ~ 4 times lower than in BSE, but with a BSM S content of $75 \mu\text{g g}^{-1}$. The last two
240 models require S exhaustion at $\sim 8\%$ partial melting using estimates of SCSS from [Bombardieri et](#)
241 [al. \(2005\)](#) and 10-14% partial melting for lower SCSS, as illustrated in **Figure 6**.

242

243 The models highlight several important issues for mare basalt petrogenesis. In the first
244 instance, without complete S exhaustion in the source, and with extreme sulfide-melt partition
245 coefficients, the likelihood for HSE inter-element fractionation is significant. Variable sulfide
246 liquid-silicate liquid D values can lead to extreme sulfide-melt partition coefficients, with Pt/Os,
247 Pd/Os and - in particular - Re/Os ratios that can become extremely fractionated (e.g., **Figure 6b**).

248 This effect is apparent from the variable HSE partitioning observed by [Mungall & Brenan \(2014\)](#)
249 for Pt ($D = \sim 4400$ to $1,560,000$), Ir ($D = 48,000$ to $1,900,000$) and Pd ($D = 67,000$ to $536,000$). A
250 firm constraint on viable models for partial melting of mare basalt sources, therefore, is that the
251 melting results in chondrite-relative abundances of the HSE and long-term Re/Os within a few
252 percent of chondrites to satisfy primitive mare basalt compositions. Second, the melting intervals
253 required to explain mare basalt compositions from a source with low SCSS and HSE contents
254 only ~ 4 times lower than in the BSE are highly restricted, with very steep melting curves
255 (**Figure 6**). In contrast, a BSM with low HSE contents (e.g., [Day et al., 2007; 2016](#)) and low
256 sulfide-melt partition coefficients can readily generate the HSE compositions of primitive mare
257 basalts, within the range of acceptable partial melting (5 to 11%). The models are strongly driven
258 by the S contents in the mantle sources of mare basalts and, with high D values, imply generation
259 of mare basalts from source regions that did not retain residual sulfide. This result is consistent
260 with the recent experimental results of [Steenstra et al. \(2018\)](#) indicating that mare basalt and
261 pyroclastic glass bead source regions were not sulfide saturated.

262

263 **Sulfide in mare basalt source regions**

264 As with metal retention in the mantle, a further constraint on any residual sulfide phases
265 in mare basalt source regions can be placed by study of elements with similar incompatibility but
266 differing lithophile or siderophile tendencies. For this purpose, Cu and Yb can be used, as both
267 elements have similar incompatibility during magmatic differentiation. Copper is a chalcophile
268 and moderately volatile element that is estimated to be at lower abundance in the BSM ($\sim 3.3 \mu\text{g}$
269 g^{-1} ; [O'Neill, 1991](#)) relative to BSE ($\sim 27 \mu\text{g} \text{g}^{-1}$; [McDonough & Sun, 1995](#)), whereas Yb is a
270 refractory lithophile element. During magmatic differentiation on the Moon, Yb is incompatible,

271 forming a negative correlation with MgO (**Figure 7a**). The positive correlation between Yb and
272 Cu indicates that Cu variations in mare basalts are a function of crystal-liquid fractionation, with
273 the most primitive Apollo 12 mare basalts having ~6 to 8 $\mu\text{g g}^{-1}$ Cu (**Figure 7b**). The correlation
274 between Cu and Yb argues strongly for undersaturation of S during partial melting, consistent
275 with evidence from Apollo 12 olivine melt inclusions (Bombardieri et al., 2005). Collectively,
276 these results indicate that mare basalt melts were not sulfide saturated.

277

278 **Conclusions and Implications**

279 Current estimates of the siderophile and HSE contents of the lunar mantle assume that the
280 source regions of mare basalts are free of residual metal and sulfide phases. Siderophile and HSE
281 contents would inevitably be underestimated in the lunar mantle if this assumption was invalid.
282 Currently, there is no strong geochemical evidence for residual metal or sulfide saturation after
283 mare basalt partial melt extraction. Low S contents (25-75 $\mu\text{g g}^{-1}$) are likely in the bulk silicate
284 Moon, considering the evidence for loss of moderately volatile elements including Cl, S, K and
285 Zn during formation and differentiation of the Moon, further limiting the likelihood for residual
286 sulfide in mare basalt sources. Potential draw-down of S into a lunar core has not been accounted
287 for but would lead to the inevitable conclusion that the S contents of the BSM are >4 times lower
288 than in the BSE. Melting models for the HSE indicate that high sulfide-melt partition coefficients
289 and an HSE-poor bulk silicate Moon (BSM) can account for the chondrite-relative abundances of
290 the HSE and near-chondritic measured $^{187}\text{Os}/^{188}\text{Os}$ in primitive mare basalts from the Apollo 12,
291 15 and 17 sites. The very low ($0.00023 \pm 2 \times \text{CI chondrite}$) and chondritic relative composition of
292 the BSM HSE remains a robust constraint on the quantity of materials added during late
293 accretion that are present in mare basalt sources. In turn, this suggests that the Moon experienced

294 disproportionately low late accretion compared to the Earth and that S and other volatile
295 elements were not added in significant quantity by late accretion to the Moon.

296

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305

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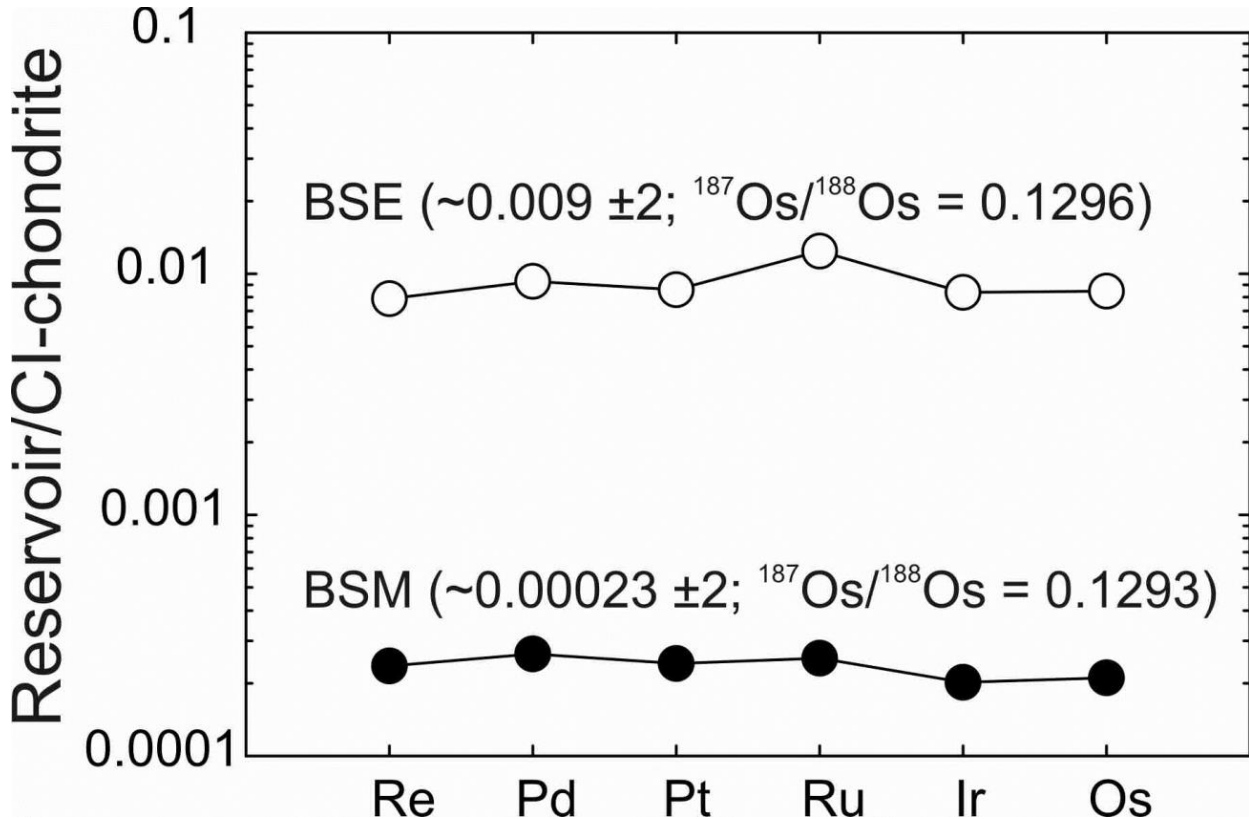
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449 **Figures and figure captions**

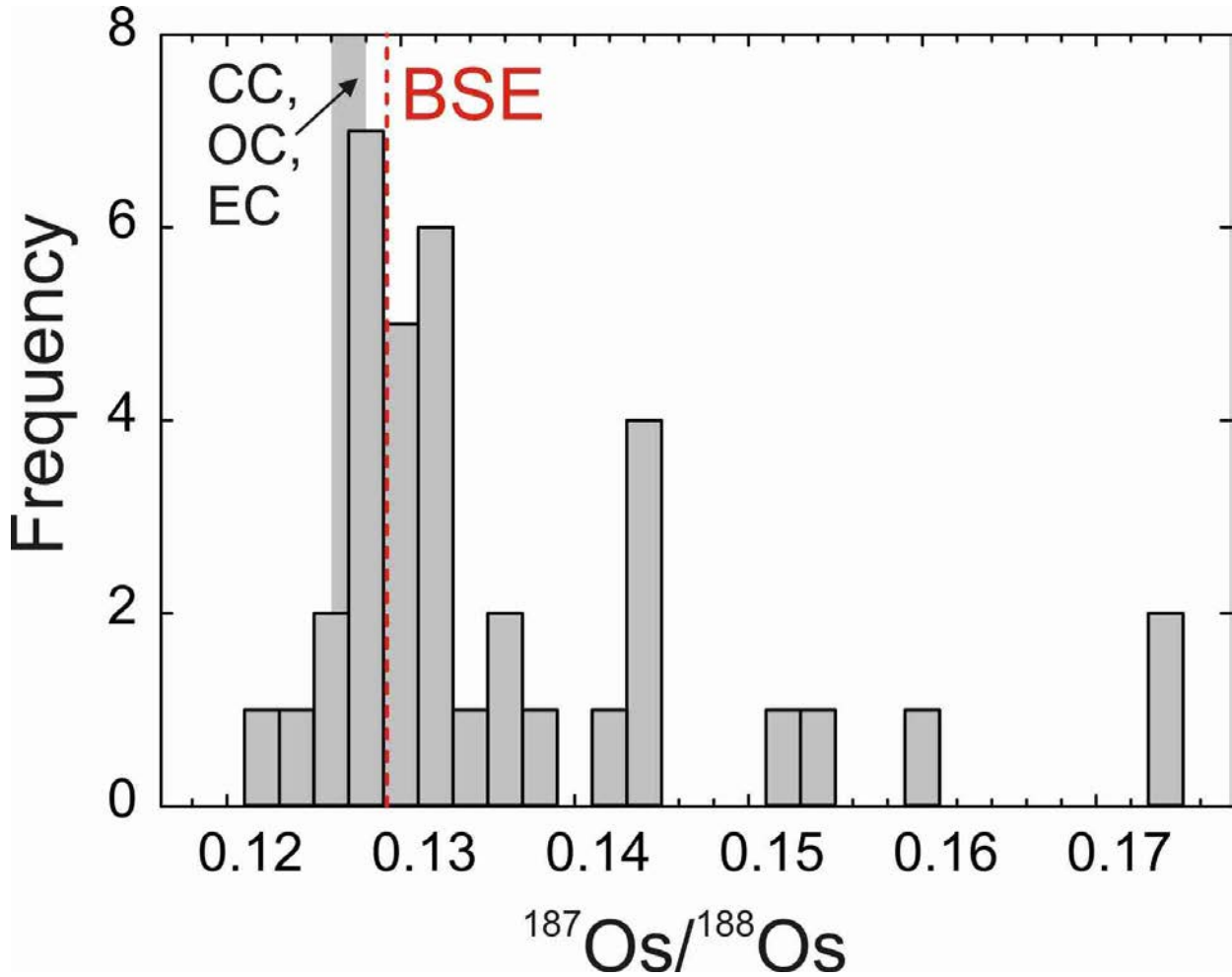
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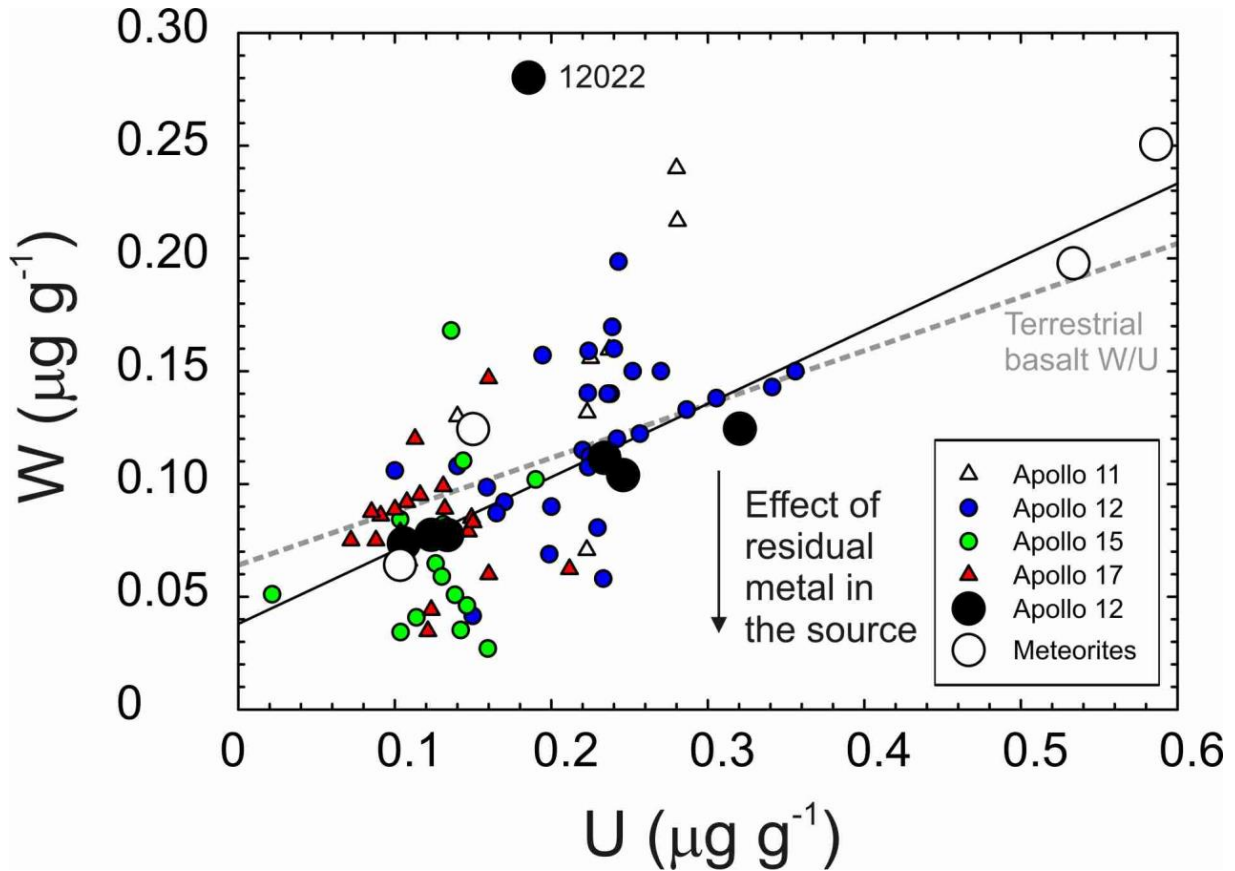
453 **Figure 1:** Estimated highly siderophile elements abundances in the bulk silicate Earth (BSE =
454 primitive mantle) and the bulk silicate Moon (BSM), normalized to Carbonaceous Ivuna (CI)
455 chondrite. BSE and BSM estimates are from Day et al. (2016), and CI-chondrite normalization is
456 from Horan et al. (2003). Here, BSE and BSM are considered to equate to mantle compositions,
457 since crustal reservoirs contain a relatively minor total fraction of the highly siderophile
458 elements.



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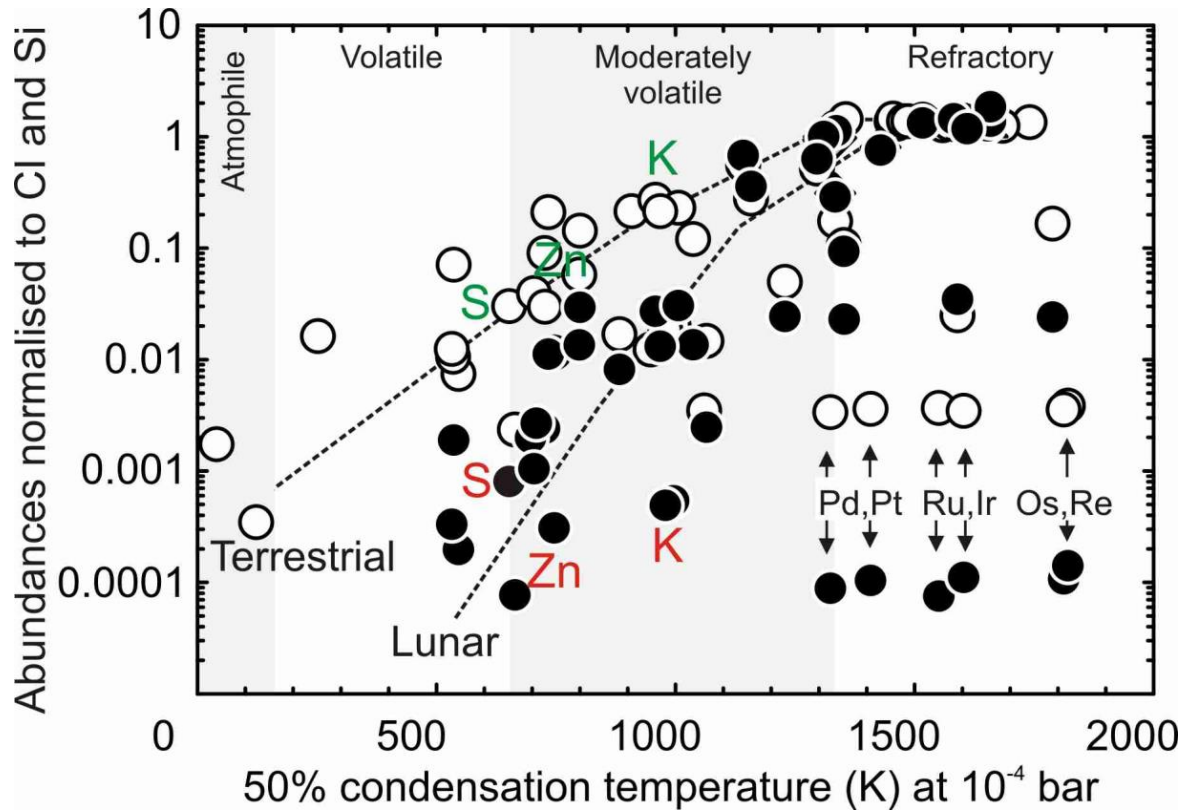
461 **Figure 2:** Histogram of measured $^{187}\text{Os}/^{188}\text{Os}$ ratios for lunar mare basalts from the Apollo 12,
462 15 and 17 sites versus the estimated bulk silicate Earth composition (red dashed line from [Meisel](#)
463 [et al., 2001](#); [Day et al., 2017](#)) and the typical range of carbonaceous, enstatite and ordinary
464 chondrites (CC, EC, OC, shaded region from [Day et al., 2016](#)). Lunar mare basalt data are from
465 [Day et al. \(2007\)](#) and [Day & Walker \(2015\)](#).



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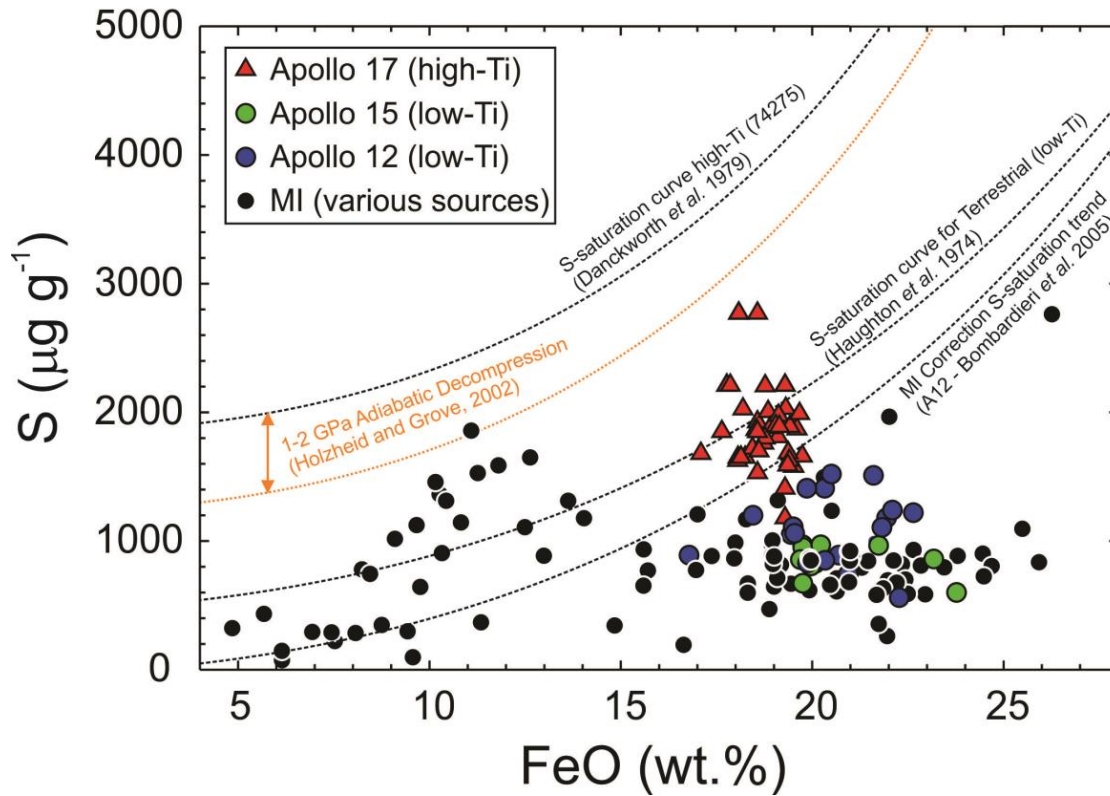
468 **Figure 3:** Uranium versus W concentrations in mare basalts from the Apollo 12 site (Day &
469 Walker, 2015; filled circles), low-Ti mare basalt meteorites (Meteorites; unfilled circles) and
470 published Apollo 11, 12, 15 and 17 mare basalt data (triangles for high-Ti Apollo 11 and 17, and
471 small circles for low-Ti Apollo 12 and 15 samples). Apart from 12022 ($0.28 \mu\text{g g}^{-1}$ W), the new
472 Apollo 12 and low-Ti mare basalt data (large symbols) show linear correlations between W, U
473 and Th and this dataset shows a positive correlation (solid line) with an r^2 of ~ 0.92 , similar to
474 W/U in terrestrial lavas. The high W content of 12022 is inconsistent with metal in the source,
475 which would act to retain W. New data are presented in Table 1, together with a compendium of
476 lunar mare basalt data compiled by C.R. Neal.



477

478

479 **Figure 4:** Estimated abundances of the elements classified according to their 50% condensation
480 temperatures at 10^{-4} bar (from [Lodders, 2003](#)) in the mantles of Earth (unfilled circles) and the
481 Moon (filled circles). Highlighted are the S, Zn and K abundances for Earth (green) and the
482 Moon (red). Data are from [O'Neill \(1991\)](#), [McDonough & Sun \(1995\)](#), with highly siderophile
483 element abundance data for the Moon from [Day et al. \(2007\)](#) and for Earth from [Becker et al.](#)
484 [\(2006\)](#).



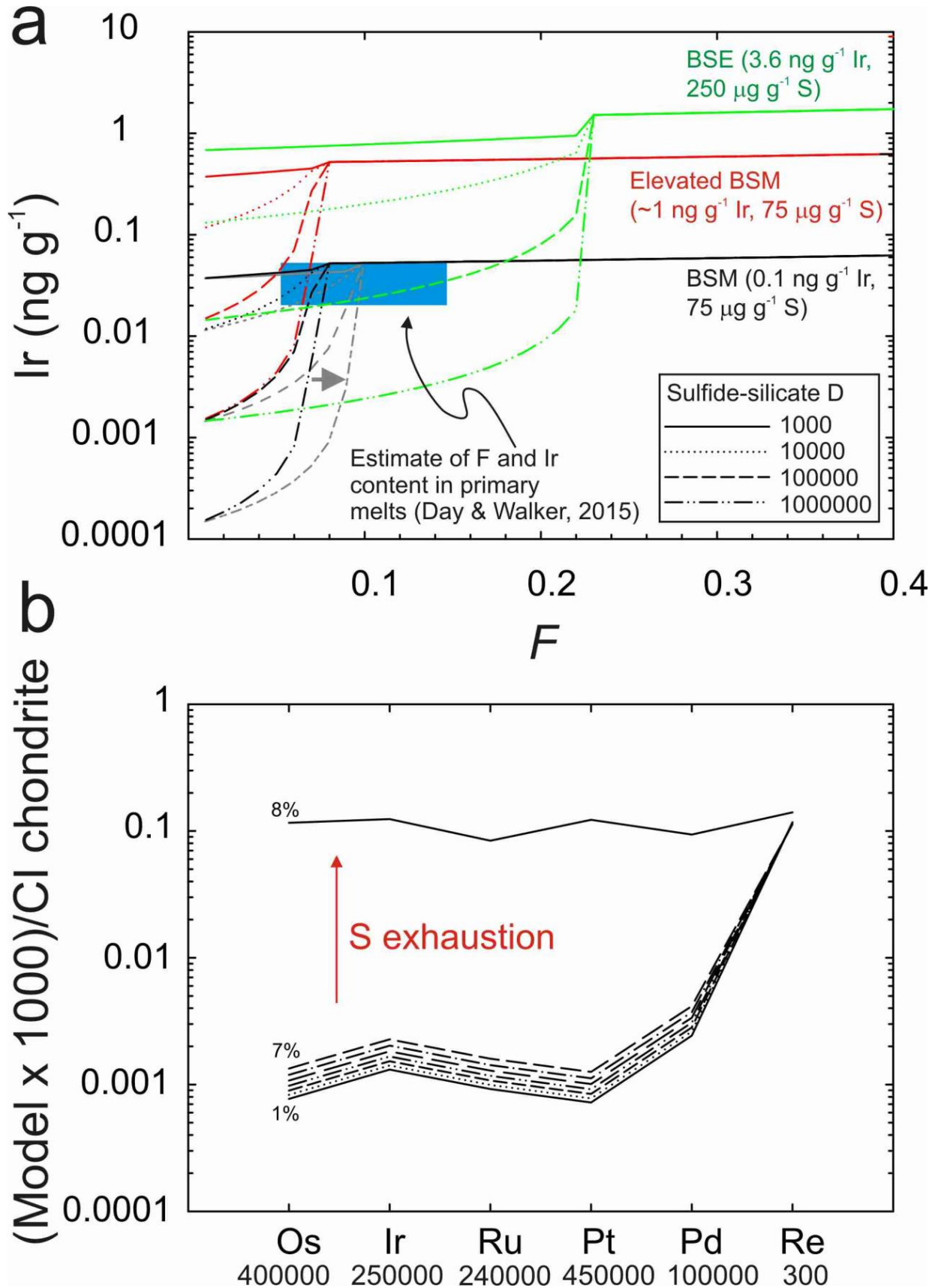
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487 **Figure 5:** Contents of FeO versus S for mare basalt bulk rocks from the Apollo 12, 15 and 17
488 sites and for melt inclusions within olivine grains from Apollo 12 mare basalts and the lunar
489 pyroclastic glass, 74220. Sulfur saturation curves are from [Haughton et al. \(1974\)](#), [Danckworth](#)
490 [et al. \(1979\)](#), [Holzheid & Grove \(2002\)](#) and [Bombardieri et al. \(2005\)](#) with bulk rock data and
491 melt inclusion data from [Rees & Thode, \(1972; 1974\)](#), [Gibson & Moore \(1974\)](#), [Gibson et al.](#)
492 [\(1976; 1977\)](#), [Bombardieri et al. \(2005\)](#), [Hauri et al. \(2011\)](#), [Chen et al. \(2015\)](#), [Ni et al. \(2017\)](#).
493 For reference, the S-saturation curve of high-Ti basalts from [O'Neill & Mavrogenes \(2002\)](#) at
494 higher temperature is 0.4 units above the [Danckwerth et al. \(1979\)](#) curve.

Day

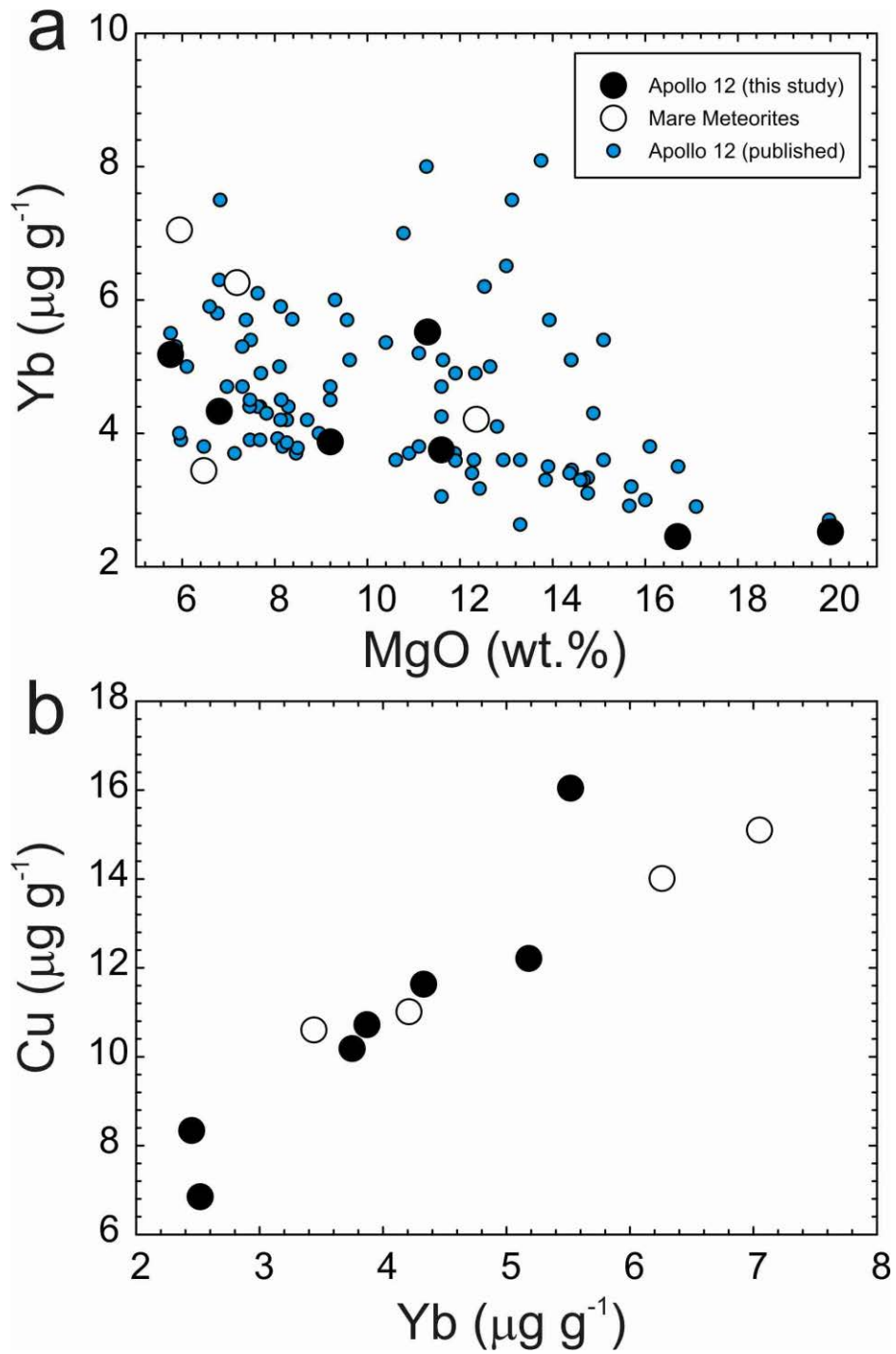
Residual lunar sulfide and metal – Revision 2



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496 **Figure 6:** Melting models for (a) Ir concentrations as a function of partial melting for different
497 sulfide-melt partitioning (1000, 10,000, 100,000, 1,000,000; [Mungall & Brenan, 2014](#)) in a
498 columnar melting regime, with three potential mantle sources, with 1050 $\mu\text{g g}^{-1}$ S in the melt
499 composition ([Bombardeiro et al., 2005](#)) and lower estimates of SCSS for a BSM with low Ir
500 content ($\sim 0.1 \text{ ng g}^{-1}$; grey model curves). The effect of lower SCSS is to drive exhaustion of the
501 HSE during partial melting to the right (arrow). Models are for a terrestrial mantle source with
502 BSE HSE abundances and 250 $\mu\text{g g}^{-1}$ S (BSE Model); a lunar mantle source with BSM HSE
503 abundances (e.g., [Day et al., 2007](#); [Day & Walker, 2015](#)) and 75 $\mu\text{g g}^{-1}$ S in the source (BSM
504 Model); a source with $\sim 25\%$ of the BSE HSE estimate and 75 $\mu\text{g g}^{-1}$ S in the source (Elevated
505 BSM Model). For the models, the behavior of Ir in silicates is assumed to be relatively
506 compatible ($\sim 1-2$; [Day, 2013](#)). The grey shaded region shows the field for the highest HSE
507 abundance Apollo 12 mare basalts and estimated degrees of partial melting to produce them from
508 their source regions ([Day & Walker, 2015](#)). (b) Example of HSE inter-element fractionations that
509 can arise for large and variable sulfide-melt partitioning. Sulfide-melt partitioning values are
510 taken from [Mungall & Brenan, 2014](#), anchoring values (shown beneath respective elements) to
511 Pd sulfide-melt partitioning of 100,000. In this model (a lunar mantle source with BSM HSE
512 abundances and 75 $\mu\text{g g}^{-1}$ S in the source) S exhaustion occurs at $\sim 8\%$ partial melting and HSE
513 patterns are distinctly non-chondritic.



514

515 **Figure 7** - (a) MgO versus Yb and (b) Yb versus Cu for Apollo 12 mare basalts and mare basalt
516 meteorites. Published data are from the compendium of mare basalt data compiled by C.R. Neal.

Table 1: MgO, Yb, Th, U, W and Cu abundances for mare basalts

Sample	Type	MgO (wt.%)	Yb ($\mu\text{g g}^{-1}$)	Th ($\mu\text{g g}^{-1}$)	U ($\mu\text{g g}^{-1}$)	W (ng g^{-1})	Cu ($\mu\text{g g}^{-1}$)	U/Th	W/U
12009, 136	Olivine-norm	11.6	3.75	0.92	0.23	112	10.2	0.25	0.48
12040, 200	Olivine-norm	16.7	2.45	0.47	0.12	77	8.3	0.26	0.63
12039, 38	Pigeonite-norm	5.8	5.18	1.25	0.32	125	12.2	0.26	0.39
12019, 15	Pigeonite-norm	9.2	3.87	0.96	0.25	104	10.7	0.26	0.42
12022, 298	Ilmenite-norm	11.3	5.52	0.72	0.19	280	16.0	0.26	1.51
12005, 63	Ilmenite-norm	20.0	2.52	0.41	0.11	74	6.9	0.26	0.70
12038, 258	Feldspar-norm	6.8	4.33	0.57	0.13	77	11.6	0.24	0.58
MIL 05035	Low-Ti meteorite	6.5	3.44	0.47	0.10	64	10.6	0.22	0.62
LAP 02205	Low-Ti meteorite	5.9	7.05	2.42	0.59	251	15.1	0.24	0.43
NWA 4734	Low-Ti meteorite	7.2	6.26	2.06	0.53	198	14.0	0.26	0.37
Dho 287	Low-Ti meteorite	12.4	4.21	0.89	0.15	124	11.0	0.17	0.83
BHVO-2	(n = 6)	7.2	2.0	1.20	0.40	205	130.8	0.33	0.52
	RSD	3%	1%	3%	3%	9%	3%		

Fifty milligrams of sample powder, remaining after the determination of HSE abundances (Day et al., 2007; Day & Walker, 2015), were digested in HF/HNO₃ in teflon vials, dried down and sequentially taken up in HNO₃ to remove fluorides, and then doped with In and diluted to a factor of 1000 to analyze Yb, Th, U, W and Cu contents, and a factor of 10,000 to analyze MgO, using a Thermo Scientific iCAP Qc inductively coupled plasma mass spectrometer in normal mode at the Scripps Isotope Geochemistry Laboratory. Accuracy and precision were determined using terrestrial rock standards (BHVO-2, BCR-2, BIR-1, AGV-1) that were prepared from powders with samples. Some of the W concentrations were previously reported in Day & Walker (2015).