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3	Polyphase solid-inclusions formed by interactions between infiltrating fluids and
4	precursor minerals enclosed in garnet of UHP rocks from the Dabie Shan, China
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ABSTRACT

Three types of polyphase solid-inclusions (PSIs) with distinct mineral assemblages and 16 microstructures were found in garnet of an ultrahigh-pressure (UHP) eclogite-vein system 17 from the Dabie Shan, east-central China. Type-1 PSI contains variable volumes of quartz, 18 K-feldspar, plagioclase \pm other phases, whereas type-2 PSI contains variable volumes of 19 quartz, calcite \pm other phases. Both types display shapes that are compatible with those of 20 euhedral coesite inclusions. Type-3 PSI always contains a rutile core that is surrounded 21 22 by plagioclase \pm quartz and generally displays the morphology of the rutile core. Variable 23 amounts of K-feldspar are embedded within the plagioclase of type-3 PSIs. The three PSI types developed fluid-mediated microstructures which include wedge-like offshoot and 24 25 protrusion textures and inclusion-garnet interfaces controlled by the crystallographic 26 structure of garnet. PSIs in peak minerals of UHP rocks have been previously thought to 27 represent primary supercritical fluid or melt inclusions. Here we propose that the studied 28 PSIs were formed under high-pressure (HP) eclogite-facies conditions during exhumation and represent reaction products between an enclosed mineral, such as coesite and rutile, 29 and external fluids infiltrating the host garnet along fractures that have been healed later 30 31 on. Two immiscible aqueous fluids (i.e., a siliceous and a carbonaceous) were involved in the formation of these PSIs. The siliceous fluid was rich in various large ion lithophile 32 33 elements like Cs, Rb, Ba, K, Pb, Li, and Sr, whereas the carbonaceous fluid was rich in Pb and Sr. The new PSI formation mechanism proposed in this study bears significant 34 implications for tracing fluid evolution and post-entrapment modifications of mineral 35 36 inclusions in HP and UHP metamorphic rocks.

- **Keywords:** Dabie Shan, ultrahigh-pressure, coesite, polyphase solid-inclusion, fluid-rock
- 39 interaction
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INTRODUCTION

Ultrahigh-pressure (UHP) metamorphic rocks represent relicts of deeply-subducted 42 crust and provide a natural laboratory to study the nature and behavior of metamorphic 43 fluids in deep-seated subduction zones. The knowledge of the interaction of these fluids is 44 important for understanding the formation, preservation, rheology and exhumation of 45 UHP rocks and the possible mass transfer in subduction zones (e.g., Prover, 2003; Zhang 46 et al., 2004; Massonne, 2009; Labrousse et al., 2011; Zheng et al., 2011; Hermann et al., 47 2013; Frezzotti and Ferrando, 2015). However, there are considerable difficulties to 48 characterize these fluids at present. One big obstacle is that the metamorphic fluids in 49 deep-seated subduction zones usually escape from their site of interaction and leave little 50 51 traces of their activity in natural rocks. Whereas experimental and theoretical studies are 52 crucial for deciphering the physical and chemical properties of fluids in deep-seated subduction zones (e.g., Zheng et al., 2011; Hermann et al., 2013; Hermann and Rubatto, 53 2014), polyphase solid-inclusions (PSIs) in peak minerals such as garnet have been 54 considered in the past two decades to be a good window for peering into UHP rocks 55 affected by fluid interaction. Such inclusions commonly display an intergrowth of several 56 57 daughter minerals, negative crystal shapes and wedge-like offshoots, and are interpreted to represent precursor supercritical fluid or melt inclusions (e.g., Hwang et al., 2001; 58 Stöckhert et al., 2001, 2009; Ferrando et al., 2005; Korsakov and Hermann, 2006; 59 Malaspina et al., 2006; Frezzotti et al., 2007; Frezzotti and Ferrando, 2015). Similar PSIs 60 (known as nanogranitoids), crystallized from felsic melt, have been also reported in 61 62 anatectic rocks of different crustal levels (e.g., Cesare et al., 2009, 2015; Ferrero et al., 63 2012, 2015; Bartoli et al., 2013). In all the reported cases, PSIs were normally considered 64 primary, i.e. they were captured during growth of the host phases.

Among the reported PSIs in UHP rocks, one group, mainly composed of quartz, 65 K-feldspar and/or plagioclase, has attracted substantial attention. The petrogenesis of this 66 group of PSI has been highly debated in the literature. Such PSIs have been interpreted as 67 the reaction products between coesite inclusions and a K-bearing omphacite host (Yang et 68 al., 1998), although their occurrence in garnet cannot be interpreted in the same way. 69 Another opinion refers to the former presence of composite K-cymrite + coesite 70 inclusions (Massonne, 2001; Massonne and Nasdala, 2003; Song et al., 2003; Zhang et al., 71 72 2009), but it cannot account for the occurrence of larger quantities of plagioclase and/or other minerals within such PSIs. In the light of recent studies, which suggest that these 73 PSIs could represent entrapped melts (e.g., Zeng et al., 2009; Gao et al., 2012, 2013, 2017; 74 Liu et al., 2013: Chen et al., 2014), we present here new perspectives on the formation of 75 quartz-feldspar-bearing PSIs in UHP rocks based on microstructural observations and a 76 77 geochemical study on their counterparts in garnet of an eclogite-vein system from the Dabie UHP terrane, eastern China. Unlike primary PSIs, we demonstrate unequivocally 78 that these PSIs represent reaction products between precursor minerals (coesite and rutile) 79 in the inclusions and external fluids infiltrating the host garnet along fractures that have 80 been healed later on. Instead of melts, two immiscible aqueous fluids (i.e., a siliceous and 81 a carbonaceous) are proposed to have been involved in the formation of the investigated 82 PSIs. 83

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SAMPLE LOCATION AND ANALYTICAL METHODS

The studied sample GJL12-4, collected from the Ganjialing area (N 30°38.666', E

116°19.619') in the Dabie UHP terrane, east-central China, is a decimeter-sized loose 87 boulder of eclogite hosting centimeter-sized quartz-garnet veins (Fig. 1A, B). This 88 boulder coexists with impure marbles in the field but the contact relationship is unclear. 89 Former studies of the impure marbles have led to the discovery of coesite inclusions in 90 dolomite and the derivation of peak P-T conditions of 4.2 ± 0.3 GPa and 745 ± 65 °C (Liu 91 et al., 2015). In addition, intergranular coesite has been found in a metasedimentary rock 92 from the same outcrop (Liu et al., 2017a). Descriptions of the geological background and 93 94 regional geology of this outcrop can be found in Liu et al. (2015, 2017b).

The petrography of sample GJL12-4 was examined on thin sections using an optical
microscope and a FEI Quanta 450 FEG scanning electron microscope (SEM) at the State
Key Laboratory of Geological Processes and Mineral Resources (GPMR), China
University of Geosciences (CUG), Wuhan. The SEM is equipped with an Oxford INCA
X-Max 50 energy dispersive X-ray (EDX) spectrometer, which was used to identify tiny
minerals and to map PSIs. The working conditions were 20 kV accelerating voltage, 5-6
µm spot size, and 12-15 mm working distance.

The major element chemistry of minerals was measured using a JEOL JXA-8100RL 102 electron microprobe (EMP) with four wavelength dispersive X-ray (WDX) spectrometers 103 and an EDX spectrometer at the GPMR. An accelerating voltage of 15 kV and a beam 104 105 current of 20 nA were applied. A beam diameter between 1 and 5 µm was selected 106 depending on the character of the analyzed mineral and its size. The counting times were 10 s for Na, K, Mg, Ca and Al, 20 s for Cr, Ti, Fe and Si, 15 s for Mn, and 30 s for Ba at 107 108 peak and background. The ZAF method was used for data correction. Natural and synthetic silicates and oxides were used as standards. X-ray maps of selected garnet 109

grains containing PSIs were obtained in WDX mode using a CAMECA SX100 EMP at the Institut für Mineralogie und Kristallchemie, Universität Stuttgart. An accelerating voltage of 15 kV, a beam current of 50 nA, a step size of 2-3 μ m, and a dwell time of 100 ms were applied.

The trace element chemistry of garnet and PSIs was analyzed in-situ on thin sections 114 by a laser ablation inductively-coupled mass spectrometer (LA-ICP-MS) at the GPMR. 115 Laser sampling was performed with a GeoLas 2005 system. The laser beam size, 116 117 frequency and energy were set to 44 (for garnet) or 60 µm (for PSIs), 6 Hz and 70 mJ, respectively. An Agilent 7500a ICP-MS was used to determine ion-signal intensities. 118 119 Helium (He) was used as the carrier gas and mixed with Argon (Ar) through a 120 T-connector before entering the ICP. Nitrogen was imported into the central gas flow (Ar + He) of the Ar plasma to improve the detection limits and precision. For each analysis, 121 122 the Agilent Chemstation was used to incorporate a background acquisition of 20-25 s and 123 a subsequent 50 s acquisition for the sample. Off-line selection and integration of background and analytical signals, time-drift correction and quantitative calibration were 124 performed by the software ICPMSDataCal (Liu et al., 2008). Element concentrations 125 126 were calibrated against multiple reference materials BIR-1G, BCR-2G and BHVO-2G 127 without using an internal standard (Liu et al., 2008).

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SAMPLE DESCRIPTIONS

Both the eclogite and the quartz-garnet veins display a fine- to medium-grained texture (Fig. 1C-F). The eclogite partially preserved the peak mineral assemblage of garnet, phengite, coesite, dolomite, rutile, apatite (Fig. 1C, D), omphacite, and magnesite. Garnet

(ca. 40 vol%), occurring as anhedral to subhedral grains (0.1-1.5 mm), contains 133 134 inclusions of coesite (Fig. 1C), omphacite, rutile, apatite, and zircon. Omphacite in the 135 matrix has been totally replaced by amphibole + plagioclase symplectites (Fig. 1C). Phengite is often partially replaced by plagioclase + quartz pools that embed biotite 136 clusters (Fig. 1D). Xenoblasts and coronae (around garnet, phengite, and dolomite) of 137 138 amphibole and epidote are additional retrograde products. Dolomite (ca. 5-10 vol%) 139 constitutes an important rock-forming mineral (Fig. 1C), whereas magnesite only occurs 140 as an accessory phase.

The guartz-garnet veins display a sharp contact to the host eclogite (Fig. 1A, B). At 141 their boundaries, amphibole + plagioclase symplectites in the eclogite are texturally 142 143 preserved, without evidence of interaction with the veins. The veins comprise mainly garnet (ca. 40-45 vol%), quartz (ca. 40-45 vol%), and plagioclase (ca. 5-7 vol%) (Fig. 144 1E). Minor and accessory mineral phases are amphibole, epidote, apatite, phengite, 145 146 K-feldspar, chlorite, calcite, rutile, ilmenite, titanite, sulfide, and zircon. Garnet (0.1-1.8) mm) in the veins displays a similar texture to that in the eclogite. It is partially replaced 147 by amphibole or epidote in places and contains inclusions of quartz, rutile, apatite, and 148 149 zircon. Plagioclase occurs as thin films or wide bands along the grain boundaries of 150 garnet, quartz (Fig. 1F), apatite, and rutile, which grade into irregular pools in places (Fig. 151 1E). Relict coesite and polycrystalline quartz aggregates typically replacing coesite were not found in the vein garnet, but the prismatic and platy morphologies of some quartz 152 inclusions (Fig. 2A, B) still suggest that they inherit their shapes from former coesite. The 153 154 quartz inclusions display specific microstructures like wedge-like offshoot and protrusion textures and inclusion-garnet interfaces controlled by the crystallographic structure of 155

garnet (Fig. 2). In addition, small garnet blebs can be rarely found in quartz inclusions(Fig. 2D).

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PSIS IN GARNET

PSIs are common in the vein garnet. A single garnet grain generally contains one to 160 161 several PSIs that are distributed within this host mineral in a random way. The PSIs can be subdivided into three major types (1, 2, and 3) on the basis of their mineral 162 163 assemblages and microstructures (Table 1). Type-1 PSIs comprise mainly variable 164 volumes of quartz and feldspar (Fig. 3A-K). Amphibole, epidote, titanite, ilmenite, calcite, chlorite, and pyrite also occur within type-1 PSIs, but their occurrence and contents vary 165 166 considerably in these inclusions. Three subtypes were defined for type-1 PSIs (Table 1), 167 which are those containing quartz + K-feldspar \pm other accessory phases (Type-1A: Fig. 3A-D), quartz + plagioclase \pm other accessory phases (Type-1B: Fig. 3E-G) and two 168 169 feldspars \pm other accessory phases (Type-1C: Fig. 3H-K). Type-2 PSIs (Fig. 3L) contain variable volumes of guartz and calcite. In some of them chlorite, epidote, and/or titanite 170 171 form well-defined patches. Type-3 PSIs always contain a rutile core that is surrounded by 172 plagioclase \pm quartz (Fig. 3M-P). Variable amounts of K-feldspar are embedded within 173 the plagioclase of type-3 PSIs. The rutile core is usually partially replaced by ilmenite \pm 174 titanite (Fig. 3M-O). All PSIs contain oval microcavities (Fig. 3C, G, I, L-P) that could represent exposed fluid inclusions. Type-1 and type-2 display regular or irregular shapes 175 and sizes from $5 \times 20 \ \mu m$ to $200 \times 300 \ \mu m$. The regular ones are generally prismatic or 176 177 platy (Fig. 3A, C-E, H, J). Both regular and irregular types developed wedge-like offshoot and protrusion textures (Fig. 3A-D, K, L) with serrated and straight 178

179	garnet-inclusion interfaces (Fig. 3B, C, G, I, L). Type-3 PSIs vary in size with diameters
180	from tens of μm to more than 200 μm . Their general morphology is always similar to that
181	of the rutile core (Fig. 3M-P). Type-3 PSIs developed a protrusion texture as well (Fig.
182	30). Their interfaces with the host garnet are controlled by the crystallographic structure
183	of garnet, as indicated by the angles between some interfaces (Fig. 3M-P). Although this
184	feature is rare, it can be still traced for some type-1 and type-2 PSIs (Fig. 3F, I, L). It
185	should be noted that BSE-brighter vein-like domains occur around the PSIs in garnet (Fig.
186	3E, F, K, L). These domains are cut by extant fractures that are locally sealed by chlorite
187	(Fig. 3Q).
188	Garnet in the eclogite contains rare type-1 and type-2 PSIs similar to those in the vein
189	garnet (Fig. 3Q-S). Type-3 PSIs are common in the eclogite but hardly contain K-feldspar
190	(Fig. 3T). Given that PSIs in the veins are much more abundant and representative, we
191	will focus on the PSIs in the veins hereafter unless specifically stated.
192	EDX mapping were performed on some PSIs in the eclogite-vein system, which reveal
193	more details on the mineral assemblages in the PSIs (Fig. 4). For instance, some mineral
194	phases (e.g., plagioclase and quartz) in single PSIs have a similar color in back scattered
195	electron (BSE) images but they can be easily distinguished by EDX mapping (Fig. 4).
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197	COMPOSITIONS OF GARNET AND PSIS
198	WDX mapping reveals that the rim of garnet in the eclogite-vein system experienced
199	fluid-mediated alteration because, among others, this garnet domain (zone B) displays a
200	completely different composition compared to the unaltered garnet (zone A), where the
201	PSIs are generally included (Fig. 5). Garnet alteration texturally related to the occurrence

202 of the PSIs was not observed (Fig. 5). Regardless of the altered domain, garnet in the eclogite and in the veins is nearly compositionally homogeneous (Fig. 5). This mineral in 203 204 the eclogite contains 44-50 mol% almandine, 32-39 mol% grossular, 14-18 mol% pyrope, and 1-2 mol% spessartine and shows a very similar composition to that in the veins: 205 43-50 mol% almandine, 32-41 mol% grossular, 12-17 mol% pyrope, and 1-2 mol% 206 207 spessartine (Table 2). Also the trace element contents in both garnets are nearly identical. When normalized against chondrite and primitive mantle, the rare-earth element (REE) 208 209 patterns and spider diagrams of garnet are characterized by enrichment of middle and 210 heavy REEs (MREEs and HREEs) and Y and depletion of light REEs (LREEs) and large-ion lithophile elements (LILEs: Cs, Rb, Ba, K, Pb, Li, and Sr; Table 3; Figs. 6A and 211 212 7A).

213 Representative major-element chemistry of minerals constituting the PSIs is listed in Table 2. Both feldspars display a wide compositional range. The composition of 214 215 plagioclase is Ab₇₃₋₁₀₀An₀₋₂₇. The K-feldspar can contain high BaO (0.85-14.5 wt%) but 216 little Na₂O (0.19-1.22 wt%) and CaO (≤ 0.21 wt%) contents. According to the 217 nomenclature of Leake et al. (1997), the amphibole can be classified as pargasite and 218 tschermakite. This mineral contains 2.17-2.64 Al per formula unit (pfu) with Mg/(Mg + Fe^{2+}) ratios between 0.51 and 0.61. Only one spot analysis was obtained for epidote 219 which contains 30 mol% pistacite. Chlorite shows a Mg/(Mg + Fe) ratio of 0.38. Titanite 220 221 hosts 0.08-0.14 Al pfu. Calcite is nearly pure CaCO₃.

Only type-1 and type-2 PSIs were analyzed for their trace element chemistry. The used laser-beam size of 60 μ m could cover the bulk inclusion in most selected cases. For several large PSIs, only a representative portion that contains all the inclusion phases was

225 sampled. As shown later, this will not influence our explanations or change our conclusions. Moreover, variable volumes of garnet ≤ 50 vol% were sampled together 226 227 with the PSIs. In total, thirteen type-1 PSIs (+ garnet) were analyzed (Table 4). Compared with the host garnet, the mixtures commonly exhibit a prominent increase of the contents 228 of LILEs (Fig. 7A). Since these elements are negligible in garnet, they could be assigned 229 230 to the PSIs. The mixtures also exhibit various increase of the contents of LREE and high field-strength elements (HFSEs: Th, U, Nb, Ta, Ti; Table 4; Figs. 6A and 7A), which 231 principally depend on the occurrence of some specific minerals within the PSIs. For 232 233 example, the occurrence of epidote and titanite could cause the increase of the contents of LREE-Th-U and Nb-Ta-Ti, respectively (Table 4; Fig. 7A). In one case, the analyzed 234 235 mixture displayed a remarkable increase of the contents of MREE-HREE-Y (Table 4; Figs. 6A and 7A) because the PSI contained Y-rich epidote proven by EDX analysis (Fig. 236 3C). In another case, high Zr and Hf contents were detected that could be related to 237 238 micro-zircon beneath the thin section surface (Table 4; Fig. 7A).

In order to obtain the bulk compositions of pure type-1 PSIs, chemical contributions of 239 garnet to the mixtures were subtracted based on the mean composition of the unaltered 240 241 garnet domains in the veins and estimated garnet volumes incorporated into single analyses (Tables 4 and 5). Nevertheless, such estimations are semi-quantitative and 242 243 involve large uncertainties. The distribution patterns of the calculated trace element compositions are always similar to those of the mixtures (Figs. 6 and 7). They exhibit a 244 nearly flat MREE-HREE distribution with LREE depletion, LILE enrichment and $U_N >$ 245 246 Th_N (Figs. 6B and 7B). Negative Nb-Ta-Ti and Zr-Hf anomalies were usually observed unless titanite and/or zircon were ablated (Fig. 7B). 247

248	Two large type-2 PSIs were partially analyzed without garnet (Table 4). As calcite and
249	quartz distribute evenly within these two PSIs, the analytical results are supposed to be
250	representative for the bulk inclusions. Compared to type-1 PSIs, these two type-2 PSIs
251	contain much lower contents of trace elements except Pb and Sr (Table 5; Figs. 6B and
252	7B). Their REE patterns are similar to those of type-1 PSIs (Fig. 6B). In the spider
253	diagrams, they display positive Pb and Sr anomalies with the contents of most other
254	LILEs and HFSEs below the primitive mantle values and/or the detection limits (Fig.

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7B).

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257 GENERAL ASPECTS RELATED TO THE FORMATION OF THE

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ECLOGITE-VEIN SYSTEM

Garnet in the eclogite contains coesite inclusions (Fig. 1C), suggesting that the 259 eclogite experienced UHP metamorphism. Applying conventional geothermobarometry, 260 261 Liu et al. (2015) constructed a P-T path for the impure marbles that coexist with the eclogite-vein system. This P-T path is characterized by peak P-T conditions of 4.2 ± 0.3 262 GPa and 745 ± 65 °C that were followed by a nearly isothermal exhumation before late 263 264 amphibolite-facies overprint. Preliminary work on the eclogite P-T evolution points to a similar P-T path but with somewhat lower peak P-T conditions ($P \ge 3.7$ GPa; T around 265 700 °C; Fig. 8). 266

Petrographic features described above suggest that the quartz-garnet veins within the eclogite formed by fluid infiltration during exhumation. The corresponding fluid-rock interactions should have proceeded along fractures that had developed in the eclogite (Fig. 1B). Three different fluid-mediated processes could have been responsible for the 271 formation of the quartz-garnet veins, which include: (1) circulated fluids directly precipitating new minerals in the open fractures; (2) flux melting (i.e. a melting process 272 273 which involves infiltration of external water and other volatile components into a solid rock) of the original eclogite within the fracture zones; and (3) leaching of the original 274 eclogite by passing fluids along the fracture zones. Garnet in the veins contains the same 275 276 mineral (e.g., coesite) inclusions as that in the eclogite (Figs. 1C and 2A, B), and often 277 displays a resorption texture along its rim. In addition, both garnets share virtually the 278 same major and trace element chemistry (Tables 2 and 3; Figs. 6A and 7A). These 279 features indicate that garnet in the veins is an original mineral from the eclogite instead of a newly-precipitated mineral. Thus, process (1) is not supported by evidence. The texture 280 281 of amphibole + plagioclase symplectites in the eclogite is not disturbed by the 282 guartz-garnet veins at the eclogite-vein boundaries indicating that the veins were formed at an eclogite-facies stage before the amphibolite-facies overprint. Extensive occurrence 283 284 of plagioclase in the quartz-garnet veins (Fig. 1E, F) further constrains the formation of the veins at lower pressures than those at the reaction curve of albite = jadeite + quartz. 285 Corresponding to such a late metamorphic stage, the retrograde P-T paths constructed for 286 287 the eclogite and surrounding marble could have overstepped the wet solidus of a common eclogite (Fig. 8). However, a high degree of melting is required to remove a significant 288 289 quantity of the original omphacite (and dolomite) in the fracture zones to produce a garnet-rich vein. Thus, the relatively low temperature around 650 °C leaves a significant 290 291 water influx (i.e., a high water/rock ratio) as the only choice to comprehend the formation 292 of quartz-garnet veins in the eclogite. In such a case, a dissolution-dominated leaching process should have played a dominant role in forming these veins instead of partial 293

294 melting. If the P-T path has never overstepped the wet solidus of a common eclogite, a 295 dissolution-dominated leaching process can be the only candidate as formation 296 mechanism for the veins.

According to the above considerations the quartz-garnet veins formed as follows: (1) fractures were opened in the host eclogite during early exhumation; (2) external aqueous fluids were introduced under eclogite-facies conditions, leading to a local high water/rock ratio in and near the fracture zones; (3) omphacite, dolomite \pm other minerals in these zones were dissolved in the fluids which then precipitated new plagioclase and net quartz; (4) garnet in the fractures zones survived the dissolution process and were enriched in the leached zones to finally form the quartz-garnet veins.

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A NEW MODEL FOR THE FORMATION OF PSIS

PSIs in this study display wedge-like offshoot and protrusion textures and 306 307 inclusion-garnet interfaces controlled by the crystallographic structure of garnet (Fig. 3). These microstructures are similar to those of primary PSIs found in UHP (e.g., Stöckhert 308 et al., 2001, 2009; Ferrando et al., 2005; Korsakov and Hermann, 2006; Frezzotti et al., 309 310 2007; Frezzotti and Ferrando, 2015) and anatectic rocks (e.g., Cesare et al., 2009, 2015; Ferrero et al., 2012, 2015; Bartoli et al., 2013), which could be best explained by 311 312 fluid-mediated shape maturation and decrepitation of fluid-bearing inclusions within garnet (e.g., Roedder, 1984; Perchuk et al., 2005; Ferrero et al., 2012). Nevertheless, the 313 investigated PSIs differ from the previous primary ones in several aspects. Primary PSIs 314 315 are formed by entrapment of a fluid phase in a growing garnet and, thus, usually share the following features: (1) occurrence of inclusions in clusters in a specific domain of garnet; 316

(2) display of a constant polycrystalline mineral assemblage and (3) of a negative crystal 317 shape (an isometric shape with rhombododecahedral geometry). To be compared, PSIs in 318 319 this study are distributed randomly in garnet and a single garnet grain only contains one (Fig. 1F) to several of them. In addition, the kind of the constituting minerals and their 320 proportions vary considerably in our PSIs (Fig. 3). Even though some inclusion-garnet 321 322 interfaces reflect the crystallographic control of the host mineral, PSIs in this study show 323 a typical negative crystal shape rarely (Fig. 3). Therefore, a formation mechanism different to that of the primary PSIs is indicated for the investigated PSIs. 324

325 Type-1 and type-2 PSIs sometimes display a prismatic or a platy shape (Fig. 3A, C-E, H, J, R), which is compatible with that of euhedral coesite inclusions. This leads us to 326 327 believe that external fluids could have infiltrated the host garnet along its fractures and 328 reacted with the precursor coesite inclusions, thereby inheriting their shapes. Such an explanation is validated by the microstructures of type-3 PSIs (Fig. 3M-P, T). The rutile 329 330 $(\pm \text{ ilmenite } \pm \text{ titanite})$ core within these PSIs is too large to be a daughter mineral precipitated from an entrapped fluid. The rutile core is surrounded by feldspars \pm quartz 331 332 and the whole inclusion always exhibits the general morphology of the rutile core. These 333 features suggest that a single rutile grain was once enclosed in garnet and external fluids 334 infiltrated this inclusion along the rutile-garnet interfaces. In response to this infiltration, 335 material was dissolved from the host garnet and/or the rutile core to enlarge the space for accommodating the newly-precipitated feldspars \pm quartz. Whether this process was 336 accompanied by the partial replacement of rutile by ilmenite and titanite must be called in 337 338 question. As outlined above, the fluid infiltration occurred still at conditions of the eclogite-facies. At such conditions, rutile is the stable Ti-mineral, but local equilibria 339

leading to the formation of ilmenite and/or titanite already in the eclogite-facies cannot be 340 ruled out. Concerning PSIs without typical coesite morphologies or rutile cores, the 341 342 former presence of other mineral inclusions (e.g., omphacite and phengite; see Fig. 8 in Gao et al., 2014) is possible. The extant fractures in garnet could have served as the 343 infiltration pathways for the fluids into the former mineral inclusions. If this was the case, 344 345 some of these fractures should have been filled by minerals (such as feldspar) which are also present in the PSIs, not by chlorite which formed during very late alteration (Fig. 346 347 3Q). Therefore, the possibility of fluids infiltrating along the extant fractures can be ruled out. BSE-brighter vein-like domains occur around the PSIs in garnet (Fig. 3E, F, K, L), 348 which are explained to represent early fractures that were healed during fluid-garnet 349 350 interactions. In our opinion these early fractures are to be considered as the pathways for the fluid infiltration. 351

A detailed model on the formation of the studied PSIs is shown in Fig. 8. Coesite, 352 353 rutile, and other metamorphic minerals were captured by growing garnet during prograde metamorphism (before stage A). During early decompression (from stage A to B), early 354 355 fractures were created in garnet because of hydrofracturing and/or differential expansion 356 of mineral inclusions relative to the host garnet (see Whitney, 1996). At stage B when the 357 quartz-garnet veins were formed, aqueous fluids infiltrated the mineral inclusions along 358 the early fractures and interacted with them and the host garnet via a 359 dissolution-reprecipitation mechanism. As microstructural a response, the inclusion-garnet interfaces were adjusted to reduce the surface free energy (Roedder, 360 361 1984), resulting in straight, serrate, and smooth inclusion walls. Later decrepitation could lead to wedge-like offshoot and/or protrusion textures. The external fluid-mineral 362

inclusion-host garnet interaction continued until the amphibolite-facies stage C was 363 reached. The early fractures had been healed and our two representative PSIs were 364 365 formed. It should be noted that some monophase quartz inclusions after coesite display microstructures reflecting a similar fluid-mediated process (Fig. 2). These quartz 366 inclusions and type-1 PSIs are interpreted to represent a continuous product series by 367 368 interaction between the precursor coesite inclusions and the infiltrating fluids. 369 Monophase quartz inclusions were instead formed when no new mineral was precipitated from this interaction. 370

371 Coupled to the variation of constituent minerals and their contents within the PSIs (Fig. 3), the bulk composition of the PSIs varies considerably even if we only consider type-1 372 373 PSIs (Figs. 6 and 7). Such variations can be hardly interpreted as the result of a primary entrapment process but can be explained using our new model: (1) the former mineral 374 (e.g., coesite, omphacite, phengite) inclusions changed from one case to the other; (2) 375 376 infiltrating fluids reacted with the former mineral inclusions and host garnet to different degrees; and (3) the fluid composition evolved with the fluid-rock interaction. As 377 discussed in the subsequent section, more than one type of fluid phase had been involved 378 379 in the PSI formation, which could have also contributed to such variations.

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381 NATURE OF FLUIDS INVOLVED IN THE PSI FORMATION

Recent studies proposed that type-1 PSIs represent silicate melts derived from dehydration melting of phengite and/or paragonite (e.g., Zeng et al., 2009; Gao et al., 2012, 2013, 2017; Liu et al., 2013; Chen et al., 2014; Wang et al., 2016). In addition, calcite in type-2 PSIs was considered to reflect partial melting of subducted carbonates

(Gao et al., 2014). Nevertheless, no good arguments were given in these studies why 386 these PSIs should represent melts instead of aqueous fluids except that a P-T path was 387 388 envisaged to cross the dehydration melting curve of phengite. As seen in Fig. 8, the newly-constructed P-T path is just like earlier ones obtained for different UHP eclogites 389 from the Dabie Shan (e.g., Carswell and Zhang, 1999; Zhang et al., 2003; Rolfo et al., 390 391 2004). These paths do hardly cross the dehydration melting curve of phengite. Instead of flux melting, we propose that a dissolution-dominated leaching process played the key 392 393 role in forming the PSI-bearing veins. This means that aqueous fluids instead of melts 394 should have been formed and dominated the fluid phase in the veins. Compared with silicate melts, aqueous fluids have much lower viscosities (Audétat and Keppler, 2004) 395 396 and could more easily infiltrate the former mineral inclusions along garnet fractures. 397 Silicate melts, if they were really present and have dominated the fluid phase responsible for the formation of PSIs, should have been felsic. Type-3 PSIs should have, thus, 398 399 contained quartz as a common mineral that resulted from the melt crystallization (type-1) and type-2 PSIs cannot give an unambiguous indication in this situation). However, this 400 is not the case (Fig. 3M-P, T). Besides, dark and/or white mica are common 401 402 crystallization products of felsic melt inclusions (e.g., Cesare et al., 2009, 2015; Ferrero et al., 2012, 2015; Bartoli et al., 2013), which are absent in the studied PSIs (Fig. 3; also 403 404 see Zeng et al., 2009; Gao et al., 2012, 2013, 2017; Liu et al., 2013; Chen et al., 2014). K-feldspar within the studied PSIs contains up to 14.5 wt% BaO (Table 2). Such Ba-rich 405 K-feldspar has been also documented for similar PSIs in other studies (e.g., Yang et al., 406 407 1998; Zeng et al., 2009; Gao et al., 2012; Liu et al., 2013; Chen et al., 2014; Wang et al., 2016). It is noteworthy that the occurrence of Ba-rich K-feldspar is often reported to be 408

related to the activities of aqueous fluids (e.g., Harlov et al., 1998; Moro et al., 2001; Shi
et al., 2010; Henry et al., 2015). This relation is compatible with the fact that Ba is highly
compatible in aqueous fluids (e.g., Green and Adam, 2003; Kessel et al., 2005; Tsay et al.,
2017). Thus, Ba-rich K-feldspar within the studied PSIs provides another argument that
fluids involved in the formation of these PSIs were aqueous fluids instead of hydrous
melts.

Related to the fluid nature and the formation of type-3 PSIs, the efficiency of material 415 416 removal from the rutile inclusions and the host garnet (see above) must be considered. At given P-T conditions, the solubility of Ti in aqueous fluids largely depends on the fluid 417 chemistry. For example, high F concentrations can significantly enhance the solubility of 418 419 Ti in aqueous fluids (e.g., Rapp et al., 2010). However, the efficiency of Ti removal from 420 the rutile inclusions by aqueous fluids depends not only on its solubility but also on time. 421 As long as the disequilibrium state was maintained between the infiltrating fluids and the 422 rutile inclusions while the garnet fractures staved open, chemical gradients in the fluids 423 could lead to continuous removal of Ti from the rutile inclusions. This explanation can be 424 applied to the partial removal of garnet as well. Dissolution of rutile and garnet didn't 425 produce any new Ti- or FeMg-bearing mineral in type-3 PSIs probably because chemical 426 gradients in the fluids could not result in saturation of these elements at the dissolving site, 427 i.e. the dissolved elements of Ti-Fe-Mg would diffuse away from the dissolving site.

In the light of our new model described above, recovering the composition of the interacting aqueous fluids through analyzing the bulk composition of the PSIs is nearly impossible unless the bulk distribution coefficient of each element between such fluids and the PSIs is known. Besides, we have to consider the possibility of disequilibrium

between the newly-precipitated minerals and the fluids. Nevertheless, we can still put 432 some compositional constraints on the interacting aqueous fluids. Precipitation of 433 434 feldspars in type-1 and type-3 PSIs indicate that the aqueous fluids were rich in Si-Al-Na-K components. Calcite in type-2 PSIs suggests that a second immiscible 435 aqueous fluid rich in carbon could have been present. Regarding trace elements, Figs. 6 436 437 and 7 indicate that the siliceous aqueous fluids were rich in LILEs (Cs, Rb, Ba, K, Pb, Li, Sr) and that the carbonaceous fluids were rich in Pb and Sr. Such fluids were probably at 438 439 a first stage generated in the subducting oceanic crust at great depths but relatively low temperatures and then infiltrated hotter rocks which were exhumed in a subduction 440 channel (see Massonne, 2012; Liu et al., 2017b) such as the eclogite investigated here. 441

442

443

CONCLUDING REMARKS AND IMPLICATIONS

In this study a new model was proposed for the formation of PSIs, which has 444 445 important implications for tracing fluid evolution and post-entrapment modifications of mineral inclusions (such as coesite) in high-pressure-UHP metamorphic rocks. This new 446 model indicates that PSIs enclosed in peak minerals (particularly those showing fractures 447 448 and/or healed fractures) of such rocks are not always primary as previously thought and that PSIs do not necessarily represent entrapped supercritical fluid or melt inclusions. In 449 450 addition, we conclude that recovering the fluid composition through integrating the bulk composition of this kind of PSIs should be cautious (see Gao et al., 2012, 2013, 2014, 451 452 2017; Liu et al., 2013; Chen et al., 2014) though certain geochemical fingerprints can be 453 qualitatively traced.

455

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649	FIGURE CAPTIONS
650	FIGURE 1. (A) Photograph and (B) scanned image of two sections cut from the
651	eclogite-vein system (sample GJL12-4). The coin in (A) is 2 cm in diameter. (C-E)
652	Photomicrographs and (F) mixed mapping image of eclogite (C, D) and quartz-garnet
653	veins (E, F). Mineral abbreviations in this study: Ab-albite, Alm-almandine,
654	Amp-amphibole, An-anorthite, And-andradite, Ap-apatite, Bt-biotite, Cal-calcite,
655	Chl-chlorite, Cn-celsian, Coe-coesite, Dia-diamond, Dol-dolomite, Ep-epidote,
656	Gra-graphite, Gro-grossular, Grt-garnet, Ilm-ilmenite, Jd-jadeite, Kfs-K-feldspar,
657	Or-orthoclase, Phn-phengite, Pl-plagioclase, Prp-pyrope, Py-pyrite, Q-quartz, Ru-rutile,
658	Sp-spessartine, Ttn-titanite, Zr-zircon.
659	
660	FIGURE 2. BSE images of monophase quartz inclusions within garnet of quartz-garnet
661	veins. Wedge-like offshoot (pink arrows) and protrusion (blue arrows) textures and
662	inclusion-garnet interfaces controlled by the crystallographic structure of garnet (red lines)
663	are indicated. Note that some quartz inclusions display a prismatic (A) or platy (B) shape.
664	
665	FIGURE 3. BSE images of PSIs within garnet of quartz-garnet veins (A-P) and eclogite
666	(Q-T). Three PSI types can be recognized: type-1 (A-K, Q, R), type-2 (L, S), and type-3
667	(M-P, T). See text in details. Healed fractures (black arrows), wedge-like offshoot (pink
668	arrows) and protrusion (light blue arrows) textures, serrate inclusion-garnet interfaces
669	(green arrows), possible crystal faces of garnet (dark blue arrows), and exposed fluid
670	inclusions (red arrows) are indicated.
671	

672	FIGURE 4. Mixed mapping images and single element mapping images of two PSIs
673	(EDX mode). See their BSE images in Fig. 3I and M, respectively.
674	
675	FIGURE 5. Element mapping images of three garnet grains containing PSIs (WDX
676	mode). The garnet rim was altered by fluids with zone A and B representing the original
677	and the altered domains, respectively. Note that PSIs only occur in zone A and did not
678	cause any compositional alteration of garnet. In (D) and (G) are outlined the two garnet
679	grains containing PSIs, where in (A) is only shown part of a single PSI-bearing garnet
680	grain.
681	
682	FIGURE 6. Chondrite-normalized REE patterns of garnet and PSIs: (A) unaltered garnet
683	and mixtures of type-1 PSI + unaltered garnet; (B) calculated type-1 and type-2 PSIs.
684	Chondrite values are from Sun and McDonnough (1989).
685	
686	FIGURE 7. Primitive mantle-normalized spider diagrams of garnet and PSIs: (A)
687	unaltered garnet and mixtures of type-1 PSI + unaltered garnet; (B) calculated type-1 and
688	type-2 PSIs. Primitive mantle values are from McDonnough and Sun (1995).
689	
690	FIGURE 8. Petrogenesis of PSIs in this study. Coesite and rutile were enclosed by
691	growing garnet during prograde metamorphism (before stage A). From stage A to B, early
692	fractures opened in garnet, along which external fluids infiltrated mineral inclusions.
693	Then, interactions between fluid, inclusion, and garnet were initiated and continued until
694	stage C. The early fractures (white lines) had been healed and two representative PSIs

695	were produced. New fractures (black lines) opened later on. C-1 and C-2 are the two PSIs
696	at stage C, which are enlarged to highlight their microstructures. Wedge-like offshoot and
697	protrusion textures and serrate inclusion-garnet interfaces are indicated by pink, light blue
698	and green arrows, respectively. Phase-transition curves of diamond = graphite, coesite =
699	quartz, and albite = jadeite + quartz are after Bundy (1980), Bohlen and Boettcher (1982),
700	and Li et al. (2004), respectively. Curves 1 and 2 are the phengite-dehydration melting
701	curves in metagreywacke (Auzanneau et al., 2006) and eclogite (Liu et al., 2009),
702	respectively. Arrow lines are P-T paths constructed for the Dabie UHP eclogites: pink-this
703	study, green-Carswell and Zhang (1999), grey-Zhang et al. (2003), and yellow-Rolfo et al.
704	(2004). The blue line refers to the wet basalt solidus after Lambert and Wyllie (1972).

The blue solid circle is the second critical point of basalt after Mibe et al. (2011).

PSI types	Mineral assemblages	Notes on mineral assemblages	Key microstructures
Type-1A	Quartz + K-feldspar ± accessory minerals	Accessory amphibole, epidote, titanite, ilmenite, calcite, chlorite,	Type-1 and -2 PSIs have similar microstructures: prismatic, platy,
Type-1B	Quartz + plagioclase ± accessory minerals	and pyrite occur in type-1 PSIs, but their occurrence and contents	and irregular shapes; with wedge-like offshoot and protrusion
Type-1C	Plagioclase + K-feldspar ± accessory minerals	vary considerably.	textures; serrated and straight garnet-inclusion interfaces;
Type-2	Quartz + calcite	Chlorite, epidote, and/or titanite may form well-defined patches in some type-2 PSIs.	surrounded by healed fractures.
Туре-3	Rutile core + surrounding plagioclase ± K-feldspar	Rutile core is usually partially replaced by ilmenite ± titanite. Quartz was only found in one type-3 PSI.	With a general morphology of the rutile core and a protrusion texture; inclusion-garnet interfaces controlled by the crystallographic structure of garnet; surrounded by healed fractures.

TABLE 1. Summary of mineral assemblages and microstructures of PSIs in the eclogite-vein system

TABLE 2. Major element compositions (wt%) of selected minerals in the eclogite-vein system

Rock	Eclogite al Grt		Q-Grt v	Q-Grt vein													
Mineral			Grt Amp				Chl		Ep	Pl		Kfs			Ttn	Cal	
Texture	Unaltered	l	Unaltered		In PSI		In PSI	In PSI		In PSI	In PSI		In PSI			In PSI	
Spot	16-15	16-19	13-6	13-55	13-21	13-99	13-14	13-17	13-19	13-20	13-97	13-16	13-30	13-50	13-15	13-88	13-27
SiO ₂	38.51	38.07	38.85	38.41	44.09	40.85	27.64	27.58	39.00	67.95	60.41	63.00	60.43	51.69	32.46	37.48	-
Al_2O_3	21.63	21.32	21.19	21.14	12.57	14.75	17.68	17.41	22.25	19.83	24.60	18.41	19.58	21.91	6.06	3.47	-
TiO ₂	0.08	0.11	0.14	0.15	0.25	0.88	0.02	0.04	0.06	0.03	0.04	0.11	0.50	0.59	31.70	31.89	-
MgO	4.07	3.59	4.07	4.03	9.81	8.30	11.05	11.11	0.09	0.01	0.09	0.00	0.01	0.03	0.18	0.04	0.02
CaO	12.55	14.50	14.10	14.14	10.50	11.06	0.09	0.22	22.14	0.47	5.56	0.05	0.07	0.08	27.98	26.00	55.51
FeO	22.83	21.01	21.04	21.21	17.79	18.38	31.49	31.94	13.30	0.25	0.48	0.48	0.42	0.62	0.76	0.68	0.58
Cr ₂ O ₃	0.05	0.00	0.00	0.00	0.00	0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	-
MnO	0.46	0.45	0.54	0.50	0.04	0.07	0.18	0.17	0.06	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.05
Na ₂ O	0.06	0.06	0.01	0.02	2.38	2.31	0.13	0.10	0.00	11.31	8.04	0.29	0.29	0.53	0.01	0.00	-
K ₂ O	0.00	0.00	0.00	0.00	0.24	0.95	0.02	0.02	0.01	0.02	0.35	15.19	14.02	9.70	0.07	0.00	-
BaO	-	-	-	-	-	-	-	-	-	-	-	2.15	4.99	14.53	-	-	-
Total	100.25	99.10	99.94	99.60	97.67	97.56	88.31	88.58	96.90	99.86	99.57	99.68	100.31	99.68	99.22	99.58	56.16
0	12.00	12.00	12.00	12.00	23.00	23.00	14.00	14.00	25.00	8.00	8.00	8.00	8.00	8.00	3.00	3.00	1.00
Si	2.98	2.97	3.02	2.98	6.47	6.12	2.98	2.98	6.39	2.98	2.70	2.97	2.88	2.65	0.63	0.71	-
Al	1.97	1.96	1.94	1.94	2.17	2.61	2.25	2.21	4.30	1.02	1.30	1.02	1.10	1.33	0.14	0.08	-
Ti	0.00	0.01	0.01	0.01	0.03	0.10	0.00	0.00	0.01	0.00	0.00	0.00	0.02	0.02	0.46	0.46	-
Mg	0.47	0.42	0.47	0.47	2.15	1.85	1.78	1.79	0.02	0.00	0.01	0.00	0.00	0.00	0.01	0.00	0.00
Ca	1.04	1.21	1.17	1.18	1.65	1.78	0.01	0.03	3.89	0.02	0.27	0.00	0.00	0.00	0.58	0.53	0.99
Fe ³⁺	0.02	0.03	0.05	0.06	0.81	0.54	-	-	1.82	-	-	-	-	-	-	-	-
Fe ²⁺	1.46	1.34	1.32	1.32	1.37	1.77	2.84	2.88	-	0.01	0.02	0.02	0.02	0.03	0.01	0.01	0.01
Cr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	-
Mn	0.03	0.03	0.04	0.03	0.00	0.01	0.02	0.02	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Na	0.01	0.01	0.00	0.00	0.68	0.67	0.03	0.02	0.00	0.96	0.70	0.03	0.03	0.05	0.00	0.00	-
K	0.00	0.00	0.00	0.00	0.04	0.18	0.00	0.00	0.00	0.00	0.02	0.91	0.85	0.64	0.00	0.00	-
Ва	-	-	-	-	-	-	-	-	-	-	-	0.04	0.09	0.28	-	-	-
Total	7.99	7.98	8.02	7.99	15.37	15.63	9.91	9.93	16.45	4.99	5.01	4.99	4.99	5.01	1.84	1.79	1.00
Mg#	0.24	0.24	0.26	0.26	0.61	0.51	0.38	0.38	-	-	-	-	-	-	-	-	-
Alm	0.49	0.45	0.44	0.44	-	-	-	-	Ab	0.98	0.71	0.03	0.03	0.05	-	-	-
Gro	0.34	0.39	0.37	0.36	-	-	-	-	An	0.02	0.27	0.00	0.00	0.00	-	-	-
Prp	0.16	0.14	0.16	0.16	-	-	-	-	Or	0.00	0.02	0.93	0.88	0.65	-	-	-
And	0.01	0.02	0.03	0.03	-	-	-	-	Cn	-	-	0.04	0.09	0.29	-	-	-
Sp	0.01	0.01	0.01	0.01	-	-	-	-	-	-	-	-	-	-	-	-	-

"-" = undetected or uncalculated. Ferric contents are calculated as follows: garnet = 5 six- and eight-fold coordinated cations; Amp = total cations - Ca - Na - K = 13; Ep = total Fe. Mg# = Mg/(Mg+Fe²⁺).

Rock	Spot	Li	Be	В	Р	Κ	Sc	Ti	V	Cr	Со	Ni	Cu	Zn	Ga	Rb	Sr	Y	Zr	Nb	Cs
Eclogite	1	bdl	0.25	0.16	199	2.85	97.2	275	150	31.8	19.2	0.41	0.40	91.5	14.8	bdl	0.058	152	1.16	0.29	bdl
	2	0.21	0.67	bdl	190	bdl	94.2	316	158	59.7	18.7	bdl	bdl	96.5	14.4	0.14	bdl	154	0.56	0.15	bdl
Q-Grt vein	1	0.44	bdl	3.29	94.6	bdl	58.7	583	108	29.6	23.3	1.27	0.23	77.2	14.3	2.67	0.19	121	2.04	bdl	0.080
	2	0.47	0.012	3.17	96.8	bdl	58.8	671	150	66.6	24.6	1.19	0.038	78.9	14.8	0.060	0.11	105	2.35	bdl	bdl
	3	0.068	bdl	3.60	107	0.047	71.9	623	129	43.1	21.9	1.26	0.29	81.1	14.2	bdl	0.14	141	3.17	bdl	bdl
	4	0.19	0.027	3.61	97.9	bdl	65.7	408	126	36.4	27.2	1.27	0.11	85.1	13.3	0.007	0.089	156	1.92	0.003	bdl
	5	0.15	bdl	4.11	105	bdl	59.5	677	146	59.3	23.4	0.93	0.14	78.0	14.3	0.005	0.092	107	2.31	0.007	0.006
	6	bdl	0.057	4.51	75.0	2.54	65.2	199	96.5	60.9	30.4	1.05	0.10	89.4	9.79	bdl	0.041	176	0.89	0.007	0.001
Continued																					
Rock	Spot	Ва	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Но	Er	Tm	Yb	Lu	Hf	Та	Pb	Th	U
Eclogite	1	bdl	0.028	0.004	bdl	0.48	3.78	1.66	13.8	3.56	25.3	5.47	13.7	1.95	17.6	2.29	0.26	bdl	0.060	bdl	0.13
	2	bdl	0.078	bdl	0.11	1.79	5.24	1.75	14.2	3.82	24.6	5.25	15.0	1.97	17.1	2.25	0.34	0.038	0.056	0.12	0.11
Q-Grt vein	1	0.18	0.11	0.27	0.071	0.78	1.66	0.90	6.93	2.09	17.9	4.44	14.3	1.92	13.6	1.93	0.026	bdl	0.034	0.004	0.067
	2	bdl	0.020	0.055	0.052	0.64	1.24	0.72	4.70	1.56	14.4	3.72	12.3	1.89	11.3	1.79	0.041	0.003	0.084	0.009	0.038
	3	bdl	0.002	0.13	0.061	0.86	1.57	1.01	7.61	2.24	20.1	5.13	16.9	2.49	16.3	2.52	0.031	bdl	0.032	bdl	0.075
	4	bdl	0.009	0.14	0.081	1.25	2.87	1.69	10.1	2.94	24.8	5.53	18.5	2.40	16.2	2.32	0.042	bdl	0.054	0.011	0.14
	5	0.090	0.004	0.069	0.029	0.60	1.18	0.72	5.29	1.68	14.3	3.86	12.4	1.80	12.3	1.72	0.026	bdl	0.028	0.018	0.057
	6	bdl	0.006	0.032	0.011	0.69	2.56	2.02	14.0	3.65	28.6	6.42	19.7	2.66	16.9	2.68	bdl	0.003	0.10	0.010	0.035

TABLE 3. Trace element compositions (ppm) of unaltered garnet domains in the eclogite-vein system

"bdl" = below detection limit.

TABLE 4. Trace element compositions (ppm) of polyphase solid inclusion ± garnet in the quartz-garnet veins

PSI types	Ablated minerals	Grt/vol%	Li	Be	В	Р	Κ	Sc	Ti	V	Cr	Co	Ni	Cu	Zn	Ga	Rb	Sr	Y	Zr	Nb	Cs
Type-1A	Kfs+Chl+Q	30	7.27	bdl	3.46	182	16900	41.1	275	49.7	8.12	19.8	0.52	0.91	77.1	9.14	36.6	23.4	189	1.02	bdl	0.27
Type-1A	Q+Kfs+Chl+Y-Ep	25	5.71	21.4	531	299	1112	58.0	1351	100	2.63	18.6	0.87	1.88	75.9	13.1	3.36	2.00	3321	3.96	1.02	0.17
Type-1B	Q+Pl+Ep+Cal+Zr	10	0.99	0.057	40.0	143	699	62.5	1163	159	18.7	3.67	1.12	0.41	13.0	12.8	2.16	81.5	187	72.3	0.30	0.42
Type-1B	Amp+Pl+Ep	50	4.40	0.40	6.04	151	2749	82.2	1020	143	56.5	23.9	2.37	2.25	88.3	13.4	3.72	54.7	105	3.47	0.50	0.23
Type-1C	Pl+Kfs	50	0.99	0.050	3.55	235	113	77.9	729	137	59.1	19.2	1.92	0.75	77.2	14.0	0.42	0.55	113	2.65	0.036	0.018
Type-1C	Pl+Kfs	50	4.29	bdl	22.0	128	21029	45.1	458	37.6	4.28	13.4	0.91	1.27	60.3	10.8	89.7	209	118	1.14	0.26	0.91
Type-1C	Pl+Amp+Kfs	50	4.77	0.15	21.2	148	14503	73.5	840	127	31.7	20.4	1.31	1.18	82.5	10.9	55.0	157	110	1.08	0.15	1.02
Type-1C	Pl+Amp+Kfs	5	5.11	0.42	15.9	120	16423	93.4	1490	179	90.7	21.4	2.68	1.17	103	14.0	64.8	178	115	3.37	2.01	0.70
Type-1C	Kfs+Pl+Cal	50	2.33	0.13	6.87	112	23075	42.7	365	92.3	9.80	16.7	1.25	1.85	59.0	10.4	64.1	21.2	111	1.46	bdl	0.43
Type-1C	Ep+Pl+Kfs+Py	5	0.17	bdl	14.4	98.1	435	59.2	244	89.4	31.0	4.96	0.61	1.61	4.74	11.3	1.13	30.9	117	1.06	0.019	0.25
Type-1C	Pl+Amp+Kfs+Chl	50	11.8	0.20	4.70	175	9401	64.9	730	131	66.8	21.3	0.64	0.14	82.8	13.6	27.5	83.6	104	2.23	0.39	0.10
Type-1C	Pl+Amp+Chl+Kfs	50	1.48	bdl	2.66	225	1168	62.6	552	118	45.3	21.9	1.15	0.66	75.9	13.7	3.70	36.2	116	2.23	0.012	0.35
Type-1C	Kfs+Pl+Chl+Amp+ Ttn	20	22.3	0.35	77.9	133	46167	29.8	18698	110	21.7	14.0	3.97	2.29	80.6	9.62	196	158	52.4	2.42	29.7	1.30
Type-2	Cal+Q	0	0.006	0.077	bdl	41.5	18.2	0.98	6.63	1.20	0.55	0.37	0.82	6.70	7.35	0.20	0.10	65.9	7.48	0.13	0.028	0.055
Type-2	Cal+Q	0	bdl	0.25	0.18	31.0	1.54	0.079	bdl	bdl	bdl	0.070	0.11	0.39	1.35	bdl	0.092	55.0	5.93	bdl	0.003	bdl
Continued																						
PSI types	Ablated PSI minerals	Grt/vol%	Ва	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Но	Er	Tm	Yb	Lu	Hf	Та	Pb	Th	U
Type-1A	Kfs+Chl+Q	30	7554	0.29	0.34	0.075	1.12	2.74	1.57	15.1	3.78	30.5	7.12	22.2	2.88	21.1	3.08	0.007	0.040	0.38	0.017	0.045
Type-1A	Q+Kfs+Chl+Y-Ep	25	264	0.82	2.29	0.41	3.61	13.0	5.07	137	47.7	467	121	386	50.8	292	41.1	0.095	0.053	0.35	1.26	1.12
Type-1B	Q+Pl+Ep+Cal+Zr	10	282	3.58	4.79	0.43	2.38	2.58	1.70	12.7	3.32	28.8	6.61	21.1	2.73	19.5	3.10	1.66	0.021	7.51	0.53	0.56
Type-1B	Amp+Pl+Ep	50	241	0.23	0.57	0.11	1.14	1.40	0.83	7.09	2.08	18.3	3.90	12.0	1.49	9.92	1.48	0.057	0.022	1.31	0.040	0.025
Type-1C	Pl+Kfs	50	23.4	0.036	0.30	0.053	0.48	1.18	0.70	5.20	1.63	15.3	4.02	13.8	2.26	15.2	2.59	0.029	0.004	0.21	0.015	0.026
Type-1C	Pl+Kfs	50	4667	0.21	0.42	0.097	1.01	1.35	0.91	6.62	2.02	16.9	4.80	14.9	1.98	15.1	2.29	0.020	0.007	1.41	0.020	0.021
Type-1C	Pl+Amp+Kfs	50	5074	0.079	0.26	0.055	0.85	2.09	1.21	9.42	2.44	19.2	4.31	12.2	1.68	12.2	1.89	0.078	0.014	1.33	bdl	0.035
Type-1C	Pl+Amp+Kfs	5	5022	0.079	0.19	0.058	0.39	0.61	0.59	5.87	1.80	18.4	4.32	15.1	2.18	12.3	1.77	0.12	0.13	3.21	0.005	0.018
Type-1C	Kfs+Pl+Cal	50	5489	1.95	2.35	0.23	1.04	1.00	0.69	6.02	1.77	15.8	4.30	13.2	2.06	12.5	2.13	bdl	bdl	0.41	0.049	0.085
Type-1C	Ep+Pl+Kfs+Py	5	212	0.075	0.13	0.071	0.52	1.41	1.52	9.28	2.33	18.1	4.07	13.0	1.91	12.1	1.83	0.020	0.002	11.0	0.019	0.062
Type-1C	Pl+Amp+Kfs+Chl	50	6366	0.045	0.18	0.028	0.43	1.04	0.54	5.20	1.42	13.7	3.60	12.8	2.00	13.9	2.18	0.036	0.006	0.37	0.020	bdl
Type-1C	Pl+Amp+Chl+Kfs	50	602	0.017	0.12	0.071	1.02	1.22	0.94	5.56	1.87	17.2	4.06	13.1	1.97	15.0	2.37	0.052	0.006	0.43	0.009	0.083
Type-1C	Kfs+Pl+Chl+Amp+ <i>Ttn</i>	20	22095	0.38	0.97	0.10	0.92	1.28	0.47	6.88	1.18	10.2	1.94	5.11	0.65	4.63	0.62	0.13	5.16	1.03	0.13	0.11
Type-2	Cal+Q	0	5.55	0.038	0.038	0.008	0.049	0.062	bdl	0.24	0.10	0.93	0.21	0.67	0.081	0.48	0.058	0.010	bdl	2.07	bdl	0.011
Type-2	Cal+Q	0	4.40	0.014	0.018	bdl	0.016	0.076	0.011	0.17	0.052	0.52	0.16	0.57	0.079	0.34	0.13	bdl	0.002	0.30	bdl	0.002

Ablated PSI minerals are given in the order of visual modal content. The minerals highlighted in italics are not observed on the thin section surface and their presence is inferred from the analytical results. "bdl" = below

detection limit.

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PSI types	Ablated PSI minerals	Li	Be	В	Р	Κ	Sc	Ti	V	Cr	Со	Ni	Cu	Zn	Ga	Rb	Sr	Y	Zr	Nb	Cs
Type-1A	Kfs+Chl+Q	10.3	bdl	3.35	218	24142	31.6	167	17.2	bdl	17.6	0.25	1.23	75.2	7.31	52.0	33.4	212	0.55	bdl	0.37
Type-1A	Q+Kfs+Chl+Y-Ep	7.53	28.5	706	366	1483	56.2	1626	92.0	bdl	16.5	0.77	2.45	74.0	13.0	4.25	2.64	4383	4.58	1.35	0.22
Type-1B	Q+Pl+Ep+Cal+Zr	1.07	0.059	44.0	149	776	62.4	1233	163	15.2	1.28	1.12	0.44	5.41	12.7	2.32	90.5	193	80.1	0.33	0.46
Type-1B	Amp+Pl+Ep	8.53	0.77	8.36	206	5497	101	1513	160	63.7	22.7	3.58	4.34	95.1	13.4	6.76	109	76.3	4.82	1.00	0.43
Type-1C	Pl+Kfs	1.71	0.068	3.39	373	224	92.4	931	149	68.9	13.3	2.68	1.35	72.7	14.6	0.16	1.00	91.0	3.18	0.067	0.007
Type-1C	Pl+Kfs	8.32	bdl	40.3	161	42056	26.9	390	bdl	bdl	1.56	0.66	2.39	39.0	8.22	179	417	101	0.16	0.51	1.78
Type-1C	Pl+Amp+Kfs	9.27	0.26	38.8	201	29004	83.7	1154	128	14.0	15.7	1.46	2.21	83.3	8.45	109	315	85.1	0.037	0.29	2.01
Type-1C	Pl+Amp+Kfs	5.37	0.44	16.5	121	17288	94.9	1541	182	92.8	21.2	2.76	1.23	104	14.0	68.1	187	114	3.43	2.12	0.74
Type-1C	Kfs+Pl+Cal	4.40	0.23	10.0	128	46149	22.1	204	58.9	bdl	8.34	1.33	3.55	36.3	7.45	127	42.4	87.3	0.81	bdl	0.84
Type-1C	Ep+Pl+Kfs+Py	0.17	bdl	15.0	98.2	458	59.0	229	87.5	30.0	3.89	0.58	1.69	0.69	11.2	1.15	32.5	117	1.00	0.020	0.27
Type-1C	Pl+Amp+Kfs+Chl	23.2	0.37	5.68	253	18800	66.5	934	137	84.3	17.6	0.11	0.13	83.9	13.8	54.2	167	74.1	2.35	0.77	0.17
Type-1C	Pl+Amp+Chl+Kfs	2.70	bdl	1.60	355	2336	61.8	577	110	41.3	18.7	1.13	1.17	70.1	13.9	6.71	72.3	96.6	2.36	0.019	0.67
Type-1C	Kfs+Pl+Chl+Amp+ <i>Ttn</i>	27.8	0.43	96.4	142	57709	21.4	23241	105	14.8	11.2	4.68	2.83	80.4	8.67	245	197	31.9	2.49	37.1	1.62
Type-2	Cal+Q	0.006	0.077	bdl	41.5	18.2	0.98	6.63	1.20	0.55	0.37	0.82	6.70	7.35	0.20	0.10	65.9	7.48	0.13	0.028	0.055
Type-2	Cal+Q	bdl	0.25	0.18	31.0	1.54	0.079	bdl	bdl	bdl	0.070	0.11	0.39	1.35	bdl	0.092	55.0	5.93	bdl	0.003	bdl
Continued																					
PSI types	Ablated PSI minerals	Ba	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Но	Er	Tm	Yb	Lu	Hf	Та	Pb	Th	U
Type-1A	Kfs+Chl+Q	10792	0.40	0.44	0.086	1.26	3.13	1.74	18.1	4.38	35.0	8.09	25.0	3.18	23.9	3.47	bdl	0.056	0.52	0.020	0.035
Type-1A	Q+Kfs+Chl+Y-Ep	353	1.09	3.02	0.53	4.55	16.7	6.37	179	62.8	617	159	509	67.0	384	54.1	0.12	0.070	0.44	1.68	1.47
Type-1B	Q+Pl+Ep+Cal+Zr	313	3.97	5.31	0.47	2.56	2.66	1.76	13.2	3.43	29.8	6.81	21.7	2.79	20.1	3.21	1.84	0.023	8.34	0.58	0.62
Type-1B	Amp+Pl+Ep	481	0.44	1.03	0.17	1.47	0.95	0.49	6.08	1.79	16.5	2.95	8.29	0.78	5.39	0.80	0.080	0.042	2.56	0.071	bdl
Type-1C	Pl+Kfs	46.6	0.047	0.48	0.055	0.17	0.52	0.23	2.30	0.90	10.5	3.18	11.9	2.33	16.0	3.03	0.026	0.005	0.36	0.021	bdl
Type-1C	Pl+Kfs	9333	0.39	0.73	0.14	1.22	0.85	0.64	5.15	1.68	13.7	4.74	14.2	1.77	15.7	2.42	0.006	0.010	2.77	0.031	bdl
Type-1C	Pl+Amp+Kfs	10148	0.13	0.40	0.060	0.90	2.34	1.25	10.7	2.53	18.3	3.77	8.61	1.17	10.0	1.61	0.12	0.026	2.60	bdl	0.002
Type-1C	Pl+Amp+Kfs	5286	0.082	0.19	0.059	0.37	0.54	0.56	5.75	1.78	18.3	4.29	15.0	2.18	12.2	1.75	0.13	0.14	3.37	0.004	0.015
Type-1C	Kfs+Pl+Cal	10978	3.88	4.58	0.41	1.29	0.16	0.20	3.94	1.18	11.7	3.75	10.7	1.92	10.5	2.10	bdl	bdl	0.76	0.089	0.10
Type-1C	Ep+Pl+Kfs+Py	223	0.078	0.13	0.072	0.50	1.39	1.54	9.34	2.33	18.0	4.03	12.8	1.90	12.0	1.81	0.020	0.002	11.5	0.019	0.061
Type-1C	Pl+Amp+Kfs+Chl	12731	0.065	0.24	0.005	0.057	0.24	bdl	2.31	0.48	7.37	2.35	9.99	1.80	13.4	2.20	0.038	0.008	0.68	0.030	bdl
Type-1C	Pl+Amp+Chl+Kfs	1203	0.009	0.13	0.091	1.24	0.59	0.71	3.01	1.39	14.4	3.28	10.5	1.75	15.6	2.57	0.072	0.009	0.81	0.008	0.10
Type-1C	Kfs+Pl+Chl+Amp+ <i>Ttn</i>	27619	0.47	1.18	0.11	0.95	1.14	0.30	6.57	0.88	7.73	1.21	2.46	0.26	2.18	0.24	0.15	6.45	1.28	0.16	0.12
Type-2	Cal+Q	5.55	0.038	0.038	0.008	0.049	0.062	bdl	0.24	0.10	0.93	0.21	0.67	0.081	0.48	0.058	0.010	bdl	2.07	bdl	0.011
Type-2	Cal+Q	4.40	0.014	0.018	bdl	0.016	0.076	0.011	0.17	0.052	0.52	0.16	0.57	0.079	0.34	0.13	bdl	0.002	0.30	bdl	0.002

TABLE 5. Calculated trace element compositions (ppm) of polyphase solid inclusions in the quartz-garnet veins

Ablated PSI minerals are given in the order of visual modal content. The minerals highlighted in italics are not observed on the thin section surface and their presence is inferred from the analytical results. "bdl" = below

detection limit.

Figure 1











Figure 5

