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REVISION 3

Age discordance and mineralogy

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Abstract

Observations of discordant ages, meaning that an age given by one mineral geochronometer is different from the age given by another geochronometer from the same rock, began in the early days of geochronology. In the late 1950s and 1960s, discordant U-Pb zircon ages were unquestioningly attributed to Pb diffusion at high temperature. Later, the mineralogical properties and the petrogenesis of the zircon crystals being dated was recognized as a key factor in obtaining concordant U-Pb ages. Advances in analytical methods allowed the analysis of smaller and smaller zircon multigrain fractions, then the analysis of individual grains, and even pieces of grains, with higher degrees of concordancy. Further advances allowed a higher analytical precision, a clearer perception of accuracy, and a better statistical resolution of age discordance. As for understanding the cause(s) of discordance, belief revision followed the coupling of imaging, cathodoluminescence (CL) and back-scattered electrons (BSE), to in situ dating by secondary ion mass spectrometry (SIMS), or by laser ablation inductively coupled plasma mass spectrometry (LA-ICPMS). Discordant zircon and other accessory minerals (e.g.,

26 monazite, apatite, etc.) often consist of young rims accreted onto/into older cores. Age gradients are
27 sharp, and no Pb diffusion gradients are observed. As U-Pb discordance in crystalline, non-radiation
28 damaged grains is caused by diachronous, heterochemical mineral generations, interpretations of
29 mineral ages, based on the exclusive role of diffusion, are superseded, and closure temperatures of
30 zircon and monazite are irrelevant in geological reality.

31 Other isotopic systems (Rb-Sr, K-Ar) were believed, since the 1960s, to be similarly controlled
32 by the diffusivity of radiogenic daughters. When zircon and monazite discordance were recognized as
33 zone accretion/reaction with sharp boundaries that showed little or no high-temperature diffusive re-
34 equilibration, the other chronometric systems were left behind, and interpretations of mineral ages
35 based on the exclusive role of diffusion survived.

36 The evidence from textural-petrologic imaging (CL, BSE) and element mapping by electron
37 probe microanalyzer (EPMA) or high spatial resolution SIMS or LA-ICPMS provides the decisive
38 constraints. All microcline and mica geochronometers that have been characterized in detail document
39 patchy textures and evidence for mineral replacement reactions. It is important not to confuse causes
40 and effects; heterochemical microstructures are not the *cause* of Ar and Sr loss; rather, they *follow* it.
41 Argon and Sr loss by dissolution of the older mineral generation occurs first, heterochemical textures
42 form later, when the replacive assemblage recrystallizes. Heterochemical mineral generations are
43 identified and dated by their Ca/Cl/K systematics in ^{39}Ar - ^{40}Ar . Replacive reactions adding or removing
44 Cl, such as, e.g., sericite overgrowths on K-feldspar, retrograde muscovite intergrowths with phengite,
45 etc., are detected by Cl/K vs Ar/K isotope correlation diagrams. Calcium-poor reaction products, such
46 as, e.g., young biotite intergrown with older amphibole, adularia replacing microcline, etc., can be
47 easily identified by Ca/K vs Ar/K diagrams supported by EPMA analyses. Mixed mineral generations
48 are observed to be the cause of discordant, staircase-shaped age spectra, while step-heating of crystals
49 with age gradients produces concordant plateaus. Age gradients are therefore unrelated to staircase age
50 spectra.

51 There is a profound analogy between the U-Pb, Rb-Sr, and K-Ar systems. Lead and Ar
52 diffusion rates are both much slower than mineral replacement rates for all $T < 750$ °C. Patchy
53 retrogression textures are always associated with heterochemical signatures (U/Th ratios, REE patterns,
54 Ca/Cl/K ratios). As a rule, single-generation minerals with low amounts of radiation damage give
55 concordant ages, whereas discordance is caused by mixtures of heterochemical, resolvably
56 diachronous, mineral generations in petrologic disequilibrium. This can also include (sub-)grains that
57 have accumulated significant amounts of radiation damage. For accurate geochronology the petrologic
58 characterization with the appropriate technique(s) of the minerals to be dated, and the petrologic
59 context at large, are as essential as the mass spectrometric analyses.

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Introduction

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64 A common observation of the early 1960s was that isotopic ages of different mineral
65 chronometers from the same rock often gave discordant apparent ages, and therefore did not all record
66 the crystallization of the mineral assemblage. Mineral ages of metamorphic rocks were often younger
67 than the expected age of peak metamorphism. The reason for discordance of mineral ages between
68 different dating methods, and within a single dating method, are the focus of the present paper.

69 Age determinations require an accurate solution of one equation, the age equation (Faure 1986),
70 with two unknowns, t and F_0 : $t = (1/\lambda) \times \ln(1 + (F(t) - F_0) / N(t))$, where t is the age, λ is the decay
71 constant, $F(t)$ and $N(t)$ are the present-day concentrations of daughter and parent nuclides, respectively,
72 and F_0 is the concentration of the non-radiogenic daughter isotope initially present in the crystallizing
73 mineral. From a purely mathematical point of view, two samples having the same t and F_0 , such as a
74 pair of cogenetic minerals, are sufficient to uniquely constrain the two unknowns. In order to provide
75 an internal consistency test, it is necessary to rely on overdetermination using four or more minerals

76 from the same rock to calculate a straight line (called an isochron; Faure 1986) that simultaneously
77 solves for t and F_0 . This allows a self-consistency check via statistical criteria (e.g., the chi squared test,
78 or the MSWD parameter (mean square of the weighted deviates); McIntyre et al. 1966), which
79 compares the purely statistical uncertainty with (geological or analytical) systematic artefacts. If
80 $MSWD > 1$, the systematic bias of at least one of the data-points is greater than the random scatter.

81 Some minerals have a very high parent/daughter ratio, dictated by their respective partition
82 coefficients. Methods for dating such minerals were thought to bypass the difficulty to identify, and
83 measure, one or more cogenetic minerals that accurately constrain F_0 , allowing an age calculation from
84 a single mineral. This is done when the importance of the F_0 term in the age equation falls below a pre-
85 set significance level; any systematic age bias due to inaccurate corrections for "initial Pb", "Pb-loss",
86 "initial Sr", and "initial Ar" is therefore neglected.

87 Zircon has an extremely high U/Pb ratio, which makes it well suited for dating without finding
88 a cogenetic mineral. Moreover, U has two long-lived radioactive isotopes. This allows to obtain two U-
89 Pb ages from each measurement: the ^{235}U - ^{207}Pb age, given by $t = (1/\lambda_{235}) \times \ln(1 + ^{207}\text{Pb}^*/^{235}\text{U})$, and the
90 ^{238}U - ^{206}Pb age, given by $t = (1/\lambda_{238}) \times \ln(1 + ^{206}\text{Pb}^*/^{238}\text{U})$. The asterisk denotes the radiogenic Pb, and
91 λ_{238} and λ_{235} are the decay constant for ^{238}U and ^{235}U , respectively.

92 Early geochronological work on zircon showed that in some samples the ^{235}U - ^{207}Pb age and the
93 ^{238}U - ^{206}Pb age agree, in some others they do not. The concordant samples are easily interpreted as ideal
94 chronometers recording a geologically meaningful event (e.g. igneous crystallization), while the
95 discordant samples evidently show deviations from ideal isotopic closure. To display U-Pb data and to
96 explore ways to detect, and explain, discordance, Wetherill (1956) devised the so-called Concordia plot
97 shown here in Fig. 1. The concordia curve is the geometric locus of all points whose two U-Pb ages are
98 concordant.

99 The first working hypotheses that tried to explain the discordant U-Pb ages of zircon samples
100 (Tilton 1960; Wetherill 1963) assumed that age discordance was the effect of one single cause, Pb
101 being lost out of the zircon structure at high temperature by Fickian diffusion after the zircon
102 crystallized. Tilton (1960) calculated the frequency factor expressing Pb diffusion in zircon, $D/a^2 = 25$
103 $\times 10^{-12} \text{ a}^{-1} = 8 \times 10^{-19} \text{ s}^{-1}$. Wetherill (1963) quantified the prediction of a zircon's discordance by a very
104 sophisticated mathematical model as a function of Fickian diffusion of U and Pb in the zircon crystal
105 structure.

106 Dating single minerals is only apparently exempt of the necessity to first and foremost assess
107 petrological equilibrium and cogeneticity. Qualitative imaging by cathodoluminescence (CL) and/or
108 back-scattered electrons (BSE), and X-ray mapping, have increasingly made clear that mineral
109 chronometers are very seldom truly monomineralic. Quantitative proof is provided by compositional
110 characterization at the μm scale. Electron probe microanalysis (EPMA) routinely achieves this
111 resolution.

112 In the past few decades in situ microchemical analyses by laser ablation inductively coupled
113 plasma mass spectrometry (LA-ICPMS) achieved a modest spatial resolution of no better than a few
114 tens of μm . Recent work (e.g., Paul et al. 2012; Rittner and Müller 2012; Ubide et al. 2015; Zhou et al.
115 2017; Petrus et al. 2017; Chew et al. 2017) is pushing the resolution to $<10 \mu\text{m}$, for LA-ICPMS
116 mapping depending on the material analyzed and the elements/isotopes measured. Ion imaging and ion
117 tomography by secondary ion mass spectrometry (SIMS) can be done on terrestrial and other planetary
118 materials to produce ion maps and ion tomography images with a spatial resolution of ca. 2-3 μm (e.g.,
119 Kusiak et al. 2013, Nemchin et al. 2013; Whitehouse et al. 2014).

120 It is only fair to point out that the working hypotheses proposed by Tilton (1960) and Wetherill
121 (1963) were legitimate at the time, given the low-resolution analytical tools and laboratory procedures
122 that were then available. In fact, it is only thanks to improved imaging and in situ dating analytical

123 equipment that the relevance of volume diffusion in mineral geochronometers can be demonstrated on
124 a case-by-case basis. Whether age discordance is due to thermally activated volume diffusion can be
125 verified by a very simple, yet rigorous test. The concentration of the Fick's Law diffusant as a function
126 of distance obeys a sigmoid curve (Crank 1975), called error function or erf(x), whereby the matrix
127 remains inert by definition. Deviations of the diffusant concentration from the sigmoid gradient are
128 mathematical proof that Fickian diffusion was not the only process causing the diffusant's mobility.
129 Such deviations in mineral geochronometers are usually associated with patchy textures revealed by
130 CL and/or BSE imaging. Patchy textures are typical of metasomatic alteration caused by infiltration of
131 hydrothermal fluids leading to recrystallization of the original minerals (Villa and Williams 2013).

132 There are two possible explanations for intra-grain patchy age distributions. In the first
133 interpretation, secondary structures are the *cause* of variable ages; replacive textures form first,
134 followed by diffusion of radiogenic isotopes at a faster rate than in the primary mineral (because the
135 grain size of the reaction products is smaller than that of the unreacted minerals). Thus, even systems
136 recording recrystallization caused by fluids in a fault-plane are considered to preserve information on
137 temperature-dependent diffusion (e.g., Arnaud and Eide 2000). In the second interpretation, the
138 formation of secondary textures requires fluid circulation and coupled dissolution/reprecipitation
139 (Putnis 2002, 2009; Williams et al. 2011; Putnis and Austrheim 2013). In this view, microstructural
140 modifications are not the *cause* of isotope transport but *follow* it. Strictly speaking, both are actually
141 two effects of one cause, namely dissolution/reprecipitation. However, dissolution occurs first, and is
142 simultaneously accompanied by the loss of the radiogenic isotopes previously accumulated (such as Pb
143 and Sr, which are incompatible in the structure of U-rich and of Rb-rich phases, respectively, and Ar,
144 which is incompatible in all mineral structures). The new replacive patches form only later, when
145 dissolution is followed by reprecipitation (Putnis and Austrheim 2013).

146 Dissolution/reprecipitation can manifest itself in several ways, all of which cause the
147 crystallization of a new phase assemblage (Putnis and Austrheim 2013). Dissolution requires a reactive

148 fluid and is often accompanied by open-system exchange with the external environment. At near-
149 surface temperatures, the recrystallization into a different phase assemblage is termed alteration. At
150 higher P-T condition the terms metasomatism and metamorphism are used. On the atomic scale, all
151 three phenomena are equivalent. The perception of metamorphic processes has shifted in the
152 petrological community, leading to the consensus that all paragenetic changes, both prograde and
153 retrograde, require fluids (Putnis 2009; Jamtveit 2010).

154 The quantitative comparison of the dissolution/reprecipitation rate with the diffusion rate of Pb
155 and Ar in two widely used mineral geochronometers, presented in Figure 2, shows that at medium and
156 low grade metamorphic conditions (for $T < 750$ °C, i.e., $1000/T > 0.98$), volume diffusion is the slower
157 process by orders of magnitude. This very large difference in rates explains why intra-grain age
158 variations are mostly observed in close association with patchy retrogression/recrystallization textures
159 in metamorphic minerals, while observations of genuine erf(x)-shaped age gradients, required by
160 diffusion mathematics, are rare exceptions (Villa and Williams 2013; Villa 2016, and references
161 therein). In a grain featuring superposed diffusion and recrystallization (e.g., Labotka et al. 2004, their
162 Fig. 4), the steepest gradient is the upper limit for diffusivity.

163 U-Pb geochronology is discussed in detail in the next section. $^{40}\text{Ar}/^{39}\text{Ar}$ (or, following the
164 general use of indicating the parent-daughter pair in that order, $^{39}\text{Ar}-^{40}\text{Ar}$) geochronology is discussed
165 in detail in the section after that.

166

167 **Age discordance in the U-Pb system**

168

169 The diffusionistic interpretation of discordant U-Pb ages was undermined when *in situ* U-Pb
170 geochronology by SIMS was combined with CL and BSE imaging (Gebauer et al. 1988; Paterson et al.
171 1989, 1992a, 1992b; Vavra 1990; Vavra and Hansen 1991; Hanchar and Miller 1993; Hanchar and
172 Rudnick 1995). Imaging of zircon (and later monazite, apatite, xenotime, allanite, titanite, etc.) grains

173 using these techniques often revealed complex internal structures including growth (sometimes called
174 oscillatory) and sectoral zoning, secondary zoning, inherited magmatic or metamorphic cores that are
175 texturally distinct from subsequent magmatic or metamorphic overgrowths, etc. (e.g., Corfu et al. 2003)
176 Prior to the introduction of those imaging methods, many of the internal features of the zircon were
177 difficult, to impossible, to observe with conventional optical microscopy techniques.

178 Krogh and Davis (1975) improved the detection of the internal zircon structure by mounting
179 grains in epoxy, polishing them to reveal the crystal interiors, etching them using HF acid vapour, and
180 them imaging them using reflected light microscopy (or differential interference contrast microscopy,
181 also known as Nomarski interference contrast or Nomarski microscopy) before analysis. For
182 Proterozoic and Archean zircon, this procedure was quite effective in revealing internal zoning, cores,
183 rims, etc., by preferentially etching the high U regions that had accumulated significant radiation
184 damage. However, in Paleozoic, Mesozoic, and younger zircon, often there is not enough accumulated
185 radiation damage to weaken the zircon structure enough to allow the HF vapour to effectively etch the
186 zircon and therefore rendered this process ineffective in revealing the internal features of the zircon.

187 It is also frequently observed that inherited cores are not in the geometric center of the grain. If
188 a simple sampling strategy was employed, where two analyses were done (one in the presumed core in
189 the center and one in the presumed rim), it was often the case that the core was missed, or the core and
190 rim were both analyzed with some proportion of each other in the analysis, or if the core was in fact
191 close to the rim of the grain, it could appear that the grain had no overgrowth. In fact, some imaged
192 microstructures and angular unconformities (such as zoning in the core truncated by a younger
193 overgrowth) are clear evidence of multiple geologic events (Fig. 3). In the example in Figure 3, from a
194 Cretaceous peraluminous granite (zircon rim region in Fig. 3) with a Proterozoic core (the darker
195 regions in the asymmetrical center of the crystal), even with high resolution BSE imaging it is not
196 obvious what the different regions in the zircon represent until they are dated by an appropriate
197 technique. Analysis A is clearly in the rim and concordant, and analyses B, C, D, and E, are in the core

198 with analysis B the oldest region analyzed (and not obvious that should be the case based on the BSE
199 image revealed internal structures). Analyses F and G could represent mixing of the core and rim ages,
200 or be core ages that experienced Pb loss from the accumulation of radiation damage in the zircon
201 structure. Note also that the core of this zircon grain is not in the geometric center of the crystal, and
202 that there is an outer rim (dark in BSE) that was not analyzed.

203 Figures 4a and b are simplified illustrations of the different trajectories in the Wetherill (1956)
204 Concordia diagram and the Tera-Wasserburg (1972a,b) Concordia diagram. Fig. 4 highlights where the
205 different hypothetical analyses plot and why they plot where they do on each of the diagrams.

206 Within a decade of the first CL observations, the prevailing view regarding zircon
207 geochronology was reformulated by Mezger and Krogstadt (1997); age discordance in zircon is now
208 considered due to discrete events of open-system mass transfer, recorded by chemical heterogeneity
209 (see also Corfu 2013). Microtextures such as growth zoning, fir tree zoning, sectoral zoning,
210 bright/dark patches, etc. (Corfu et al. 2003) underline that corrosion, retrogression, recrystallization,
211 and reprecipitation played an essential role and that the zircon matrix of such samples was neither
212 homogeneous nor inert as required by Fick's Law for volume diffusion to be valid. Each textural zone
213 in the zircon was reproducibly observed to be concordant by in situ dating whenever it had undergone
214 minimal radiation damage and metamictization of the zircon structure from the decay of U and Th, and
215 thereby had remained a closed system. Conclusive proof of the very low intrinsic diffusivity of Pb in
216 zircon was provided by the laboratory measurements of Cherniak and Watson (2000). Möller et al.
217 (2002) indeed observed that in a metamorphic zircon sample from Rogaland (Norway), the relict zones
218 remain concordant despite undergoing UHT conditions ($T > 950$ °C). Nasdala et al. (2010) reported
219 that extremely radiation damaged zircon crystals from the (Grenville age) Saranac Uranium Mine near
220 Bancroft, Ontario, retained their primary U concentrations, with the implication that the U was
221 immobile (but not Pb as many of their SIMS analyses were discordant), as revealed in EPMA X-ray
222 maps (their Fig. 4), even though the crystals were essentially metamict. Deformation, if unaccompanied

223 by recrystallization promoted by an aqueous fluid, can achieve Pb mobility restricted to the scale below
224 1 μm (Piazolo et al. 2012, 2016).

225 One essential characteristic of all laboratory measurements of diffusivity is that they always
226 represent a strict upper limit of the true diffusivity. It is possible (and sometimes likely, as in the case
227 of the mica diffusivity experiments mentioned below) that unanticipated systematic biases affect the
228 laboratory measurements. Such artefacts could pertain to the way that the concentration gradient is
229 measured, or to the way that it is produced. If the calculation of the transport rate is not based on
230 inverting an erf(x) profile (as documented, e.g., by Cherniak and Watson 2003) but on a bulk loss (as in
231 Giletti 1974), there is no immediate proof that transport was due to a single process. If, moreover, the
232 mineral being studied is heated in a sealed capsule containing free water, dissolution occurs, and the
233 measured isotope distribution is controlled by the sum of two processes, diffusion and
234 dissolution/reprecipitation (see, e.g., Villa and Puxeddu 1994; Dohmen et al. 1998, their Figs. 7a-h). In
235 this case the rate of transport is greater than volume diffusion alone. An example of how to deconvolve
236 the original Giletti (1974) data and to separate the dissolution rate in the capsule from true diffusivity
237 was discussed by Villa (2010, p. 5).

238 Observations on the link between petrology and the isotope record have more recently been
239 extended to monazite (e.g., Williams et al. 2007). The general consensus is growing that accurate ages
240 require an accurate interpretation of petrogenesis, which in turn requires imaging by CL and/or BSE,
241 and in the case of monazite BSE images and X-ray mapping using the EPMA, or element/isotope
242 mapping by LA-ICPMS, or ion mapping or ion tomography imaging by SIMS. A given internal feature
243 can be revealed by both CL and BSE imaging methods, whereby CL usually reveals more fine structure
244 (Hanchar and Rudnick 1995; Corfu et al. 2003). Whenever metamorphic minerals are studied in
245 sufficient detail, it becomes evident that few samples show constant (or simple bell-shaped) intra-grain
246 age distributions.

247 In the very first paper introducing the Concordia diagram, Wetherill (1956) did not make an

248 explicit statement on the mechanism causing what he termed (incorrectly) "Pb loss" at high
249 temperature. Radiogenic Pb (hereafter referred to as Pb*) was considered to be continuously or
250 episodically lost from the zircon structure at high temperature during metamorphism or anatexis. In
251 early studies, there was no discussion of the mineralogy of the zircon crystals analyzed. Moreover, in
252 the 1960s there was a paucity of experimental data on volume diffusion in minerals. Therefore, the
253 prevalent mathematical models of the early 1960s were based on the assumption that volume diffusion
254 was more rapid than all other conceivable geologic processes. Phenomena unknown at that time
255 include low-temperature Pb* loss due to metamictization of old and U-rich grains; incorporation of
256 (otherwise incompatible) ^{204}Pb in minute heterochemical inclusions or along cracks or fractures; and,
257 most importantly, complex mineralogical textures due to dissolution/reprecipitation such as
258 metamorphic rims overgrowing inherited cores of a range of different ages. Therefore, such phenomena
259 were not looked for, and therefore not observed.

260 Silver and Deutsch (1963) were the first to systematically consider the context between U-Pb
261 geochronology and the mineralogical properties of zircon. That study followed a different approach
262 than the earlier studies of Weatherill and Tilton, and instead of using elaborate mathematical models to
263 explain the diffusive behavior of Pb*, they tried to empirically understand the properties of zircon.
264 Silver and Deutsch found that the multigrain zircon fractions they studied, separated from a 100 kg
265 granodiorite block, showed heterogeneous U-Pb systematics. They compared the U-Pb data with
266 parameters such as morphology, zoning (in transmitted light and mounted in epoxy and polished to
267 reveal the crystal interiors – not using CL or BSE, as they were unavailable at that time), color,
268 inclusions, size distribution, radioactivity, refractive index, unit cell dimensions, and magnetic
269 properties.

270 Present-day in situ studies of individual grains from samples with similar complex mineralogy
271 document a correlation between petrology and isotope systematics (e.g., Villa and Williams 2013, and
272 references therein). Silver and Deutsch (1963) also proposed that zircon started to crystallize early in

273 the history of the melt and continued to grow over most of the crystallization history. The residual
274 liquid was enriched in incompatible elements, including U and Th, and as a result late crystallizing
275 zircon (i.e., either whole magmatic zircon crystals or magmatic rims on inherited cores) was enriched
276 in those elements and therefore was more affected by the effects of metamictization caused by the
277 radioactive decay of U and Th. They also raised the question of the effects on the U-Pb systematics of
278 metasomatic alteration of zircon by hydrothermal fluids.

279 The next major breakthrough in understanding and minimizing the effects of discordance was
280 achieved a decade later by Krogh, and published in Carnegie Institution of Washington Yearbooks
281 (Krogh 1970, 1971) and journal articles. Krogh (1973) presented a low Pb background dissolution
282 method for the hydrothermal dissolution of zircon. By improving the laboratory contamination in the
283 dissolution procedure, researchers could analyze smaller fractions of zircon (e.g., < 1 mg) which in turn
284 led to improvements that eventually led to the analysis of individual zircon grains and even pieces of
285 zircon (e.g., Steiger et al. 1993) and monazite (e.g., Hawkins and Bowring 1997). The possibility to
286 analyze single zircon grains, and pieces of grains, allows researchers to select well-characterized, high
287 quality, crystals and to contrast the results obtained from them with the less precise and more
288 discordant results obtained from lower quality individual grains from the same rock. It also provides a
289 clear distinction between xenocrysts (inherited crystals formed in a different magma chamber during a
290 previous magmatic episode and passively entrained by the current erupting magma); antecrysts
291 (crystals grown early in the history of the magma, when the magma chamber responsible for the current
292 eruption/intrusion started its maturation); and last-generation autocrysts (crystals still growing at the
293 very end of the current magmatic episode, at the time of eruption/intrusion) (Miller et al. 2007;
294 Schoene et al. 2010).

295 In two subsequent papers, Krogh (1982a, b) presented an improved method for paramagnetic
296 separation and a method for using air to abrade zircon grains. This methodology eliminated 90-100 %
297 of discordance and improved the precision of the analyses to $\pm 1-3$ Ma for zircon grains with an age of

298 2700 Ma. The air abrasion technique preferentially removed the U-rich outer layers of the zircon
299 crystals, because they are weaker due to the higher U contents, which results in accumulation of
300 radiation damage and subsequent metamictization of the zircon crystals. As those regions are also more
301 susceptible to cracking and thus to ingress of ^{204}Pb carried by external fluids, Krogh's method achieved
302 a removal of as much as 80 % of the ^{204}Pb present in the zircon crystals.

303 A further development in the same direction was termed chemical abrasion thermal ionization
304 mass spectrometry (CA-TIMS) by Mattinson (2005). This method endeavors to remove the high U
305 radiation-damaged, and thus more soluble, regions in zircon using high-temperature annealing
306 (typically ca. 900-1000 °C for 48-60 hours followed by leaching in concentrated HF for a few hours at
307 c. 190°C (Mundil et al. 2004). If followed by stepwise leaching, the method allows an easy recognition
308 of Pb loss due to interaction of damaged crystals with low-temperature fluids and often leads to the
309 formation of a geologically meaningful "age plateau" (Mattinson 2011, Huyskens et al. 2016).

310 Depth profiling is an analytical approach sometimes used to study intra-grain age distributions
311 in order to allow diffusion modelling of core-rim age differences, and to analyze thin rims (5-10 μm)
312 on accessory minerals that otherwise cannot be analyzed once the grains are mounted in epoxy and
313 polished to reveal the crystal centers. Due to the generally low number of radiogenic nuclides to be
314 counted, geochronological depth profiling needs larger and deeper sampling volumes than element
315 profiling in geochemical investigations and in diffusion experiments. As pointed out by Villa (2016),
316 sub- μm spatial resolutions are required to positively identify genuine intra-grain diffusion profiles from
317 discrete, diachronous mineral generations. Also, the concept of depth profiling the outside rim of a
318 zircon is in conflict to what Krogh (1982a, b) proposed; that by removing the outer high-U regions in
319 zircon by air abrasion, the resulting zircon showed less dispersion and was more concordant by
320 removing any ^{204}Pb at the surface (in cracks, fractures, etc.), and regions that had experienced Pb loss
321 through the radiation damage produced in the high-U regions. A serious problem with depth profiling is
322 that it cannot be associated with CL/BSE imaging. This forfeits any information on the identity of the

323 phases analyzed and disables constraints on the petrogenetic mechanisms involved in their genesis.
324 Neglecting the petrological characterization of samples led Grove and Harrison (1999) to model rim-
325 core age increases in monazite grains solely as an invertible function of temperature, despite the
326 recognition that monazite grains are predominantly comprised of heterochemical and diachronous
327 patches, whereby zones near the rim can, but need not, be the youngest (DeWolf et al. 1993; Cocherie
328 et al. 1998; Fisher et al. 2017). A useful indicator would be the examination of multi-element profiles
329 (Villa 2016, his Fig. 2b). As the respective diffusivities of ions having different radii and charges are
330 expected to be unrelated to the fluid/mineral partition coefficients, even without CL/BSE imaging it is
331 straightforward to discriminate whether the cause of age discordance is Pb* loss (compatible with
332 smooth multi-element diffusion profiles) or accretion of diachronous and heterochemical mineral
333 generations (identified by patchy, non-monotonic profiles).

334 There appears to be a critical length-scale below which the concept of concordance can no
335 longer be defined in a meaningful way. The combination of U-Pb geochronology with CL/BSE images
336 obtained with (sub-)μm resolution (Kusiak et al. 2013; Valley et al. 2014) reveals that two processes
337 occur at the nm scale: firstly, the recoil of intermediate decay-product nuclides and of associated α
338 particles creates extended defects and at the same time removes the daughter nuclide from the
339 crystallographic location of its parent; and secondly, high-temperature diffusion of Pb* by a few nm
340 (which is irrelevant for the geochronology of grains > 100 μm and therefore usually not considered
341 (Mezger and Krogstadt 1997)) concentrates the interstitial Pb* ions (and Y ions (Valley et al. 2014))
342 into traps corresponding to radiation-created extended defects. Such closed-system redistribution was
343 observed (Kusiak et al. 2013; Valley et al. 2014) to disproportionate the Pb*/U ratio locally by
344 increasing it in the Pb* traps and lowering it in the recoil-depleted source areas. The paradox
345 conclusion is that a single-grain TIMS analysis of zircon that underwent high-temperature
346 metamorphism or anatexis gives a perfectly concordant, undisturbed U-Pb age, but atom-probe
347 analyses of the same grain at a scale < 1 μm can give both positively and negatively discordant "single-

348 atom" ages. It is only the whole-grain average at the 100 μm scale that balances out the nm-scale
349 heterogeneities and results in a concordant age.

350 A similar small-scale disproportionation had been reported by McLaren et al. (1994), albeit
351 with a slightly lower resolution approach (that was appropriate for the time when this study was done)
352 in their investigation of reverse discordance in in situ SIMS U-Pb analyses of zircon. In that study,
353 McLaren et al. found that microstructural features identified in a high U Sri Lankan zircon crystal,
354 using transmission electron microscopy (TEM), could be related to the reversely discordant U-Pb ages.
355 Those authors suggested that in the high U metamict regions, the U is decoupled from the Pb; with the
356 U being partitioned into the zirconia (tetragonal ZrO_2), and the Pb partitioned into the amorphous
357 "silica glass" phase. The zirconia and amorphous silica glass are produced when zircon incongruently
358 breaks down through the metamictization process (e.g., Ellsworth et al. 1994). The glass phase
359 preferentially sputters during the SIMS analysis leading to an apparent excess of radiogenic Pb and
360 producing the observed reverse discordance (McLaren et al. 1994)

361 Another very widely diffused U-Pb and Th-Pb mineral chronometer is monazite, which also is a
362 thermally very retentive mineral chronometer. Monazite is also known to not incorporate significant
363 amounts of ^{204}Pb when it crystallizes (Parrish 1990) making it a suitable target for U-Pb
364 geochronology. It is well known (e.g., Bingen and van Breemen, 1998) that temperatures around 800-
365 850 $^{\circ}\text{C}$ leave the U-Pb age of monazite unaffected. On the other hand, it is also well established that
366 mass transfer during chemical open-system events promotes patchy recrystallization, and
367 corresponding age rejuvenation, at much lower temperatures (Williams et al. 2007, 2011; Laurent et al.
368 2016). The abundant imaging data reported in the literature (including BSE images and EPMA X-ray
369 and maps), demonstrating very widespread intra-grain genetic heterogeneities, have brought about an
370 abandonment of bulk-grain analyses. Analyzing a polygenetic grain, which is very retentive of its
371 radiogenic Pb, is easily predicted to give rise to a discordant data-point.

372 The analytical approach most established in the literature is in situ dating, either by LA-ICPMS

373 (Paquette and Tiepolo 2007; Gregory et al., 2009; Goudie et al. 2014), SIMS (Stern and Bergman,
374 2001; Fisher et al. 2017), or EPMA (Williams et al. 2007). In the case of monazite, however, BSE
375 imaging is often less effective than X-ray maps to reveal the internal structures of monazite crystals
376 (Williams et al. 2011; Fisher et al. 2017). As (multiply) zoned monazite grains are the rule rather than
377 the exception, it would be natural to expect that most studies, even those doing single-grain analyses,
378 would report a great deal of discordant data-points. However, in contrast to zircon, the interpretive
379 paradigm of monazite data evolved away from a description of discordance and towards exploring the
380 tight links between petrology, microchemistry, and in situ geochronology. As the present paper deals
381 with discordance, monazite offers much less pertinent, and less abundant literature than zircon. The
382 analytical protocols just mentioned make discordant data-points very rare, as intra-grain zones, once
383 they are formed, are extremely unlikely to ever lose Pb*. The most important rejuvenation mechanisms
384 for monazite are coupled dissolution and reprecipitation (i.e., recrystallization: Putnis 2009), which
385 once either process occurs, removes an entire (sub-)grain but causes no discordance in any of the
386 surviving (sub-)grains (e.g., Parrish 1990; Hawkins and Bowring 1997; Macfarlane et al. 1997; Harlov
387 et al. 2007; Hetherington and Harlov 2008; Harlov and Hetherington 2010; Harlov et al. 2011;
388 Williams et al. 2011; Goudie et al. 2014; Fisher et al. 2017).

389 In addition to zircon and monazite, several more mineral chronometers are currently used with
390 the U-Pb method, among which are apatite, titanite, and rutile. For all three minerals, there are reliable
391 diffusivity data (Cherniak et al. 1991; Cherniak 1993; 2000; 2010). As in all three cases the diffusion
392 coefficients at temperatures typical of mid-crustal metamorphism are orders of magnitude higher than
393 for zircon, it could be expected that discordance by Pb loss (Tilton, 1960) would be observed very
394 often. Instead, only a small number of papers report discordant apatite, titanite, or rutile (e.g., Romer
395 1996; Corfu and Stone 1998; Bracciali et al. 2013).

396 Discordance of apatite in the samples studied by Romer (1996) occurred below 150 °C and was
397 attributed by him to incomplete hydrothermal recrystallization. Romer (1996) also reported observing

398 two discrete apatite generations; late overgrowths that were separated from the older cores by a
399 corroded grain surface. The apatite and titanite analyses reported by Corfu and Stone (1998) are
400 indistinguishable from concordia in most samples. As titanite occurs in two clearly distinct hetero-
401 chemical generations in the samples studied, Corfu and Stone (1998) proposed local fluid activity as a
402 likely event affecting some, but not all, of their samples and effecting a complete resetting of apatite,
403 titanite, and rutile (as well as causing anomalous Sr isotopic compositions reported in the preceding
404 literature). Such an open-system fluid circulation would also explain why only two apatite samples lie
405 resolvably below concordia whereas at least 25 % of the apatite samples lie reversely discordant above
406 concordia due to U loss and/or Pb addition. Note that, in this (rather unusual) instance, the natural fluid
407 circulation caused the same chemical fractionation in the studied apatite samples as the laboratory
408 artefact attributed to microbeam analyses in Fig. 4a.

409 Discordance of rutile, on the other hand, is attributed by Bracciali et al. (2013) to the
410 incorporation of nonradiogenic Pb (monitored by ^{204}Pb). In the detrital samples that they analyzed, the
411 petrogenetic context is no longer visible and speculations on the mechanisms causing discordance are
412 not possible.

413 The case of rutile resulted in an apparent paradox. Kooijman et al. (2010) reported age gradients
414 in natural rutile crystals obtained by LA-ICPMS, and calculated a higher diffusivity than that measured
415 in the laboratory experiments by Cherniak (2000). As Fick's Law of diffusion is the slowest transport
416 mechanism in most metamorphic environments (Fig. 2), two cases of inequality can occur: (A) the
417 transport rate measured in the laboratory is faster than that estimated basing on non-isotopic constraints
418 in natural samples, and (B) the transport rate in natural samples is faster than in the laboratory.

419 Case (A) is verified in sealed-capsule experiments whose reactants include free water (Villa
420 2010). In these experiments, the transport rate is the sum of a large term, the dissolution rate, and a
421 smaller one, Fick's Law diffusivity. The smallest among all rate estimates is the closest one to the pure
422 diffusivity. Case (B) cannot be due to laboratory artefacts of some kind of "stifled transport", as it is

423 impossible to fix atoms and prevent them from diffusing as T increases. Diffusion occurs in any case in
424 all systems. As in the preceding case, the lowermost transport rate is closest to the true diffusion rate;
425 higher transport rates necessarily derive from a sum of (unaccounted) processes. The spatial resolution
426 of the Kooijman et al. (2010) measurements was limited by the laser pit size, 35 μm diameter by 40 –
427 45 μm depth, i.e., a much longer length scale than the distance of the measurement points in the
428 Cherniak (2000) experiments. This provided Kooijman et al. (2010) a less certain documentation of an
429 erf(x) diffusion profile. As discussed by Villa (2016, his Fig. 2), a diffusive erf(x) profile necessarily
430 affects all atomic species of the mineral, not just the radiogenic isotopes. If at least one major or trace
431 element displays a patchy gradient, or if the element ratio trends antithetically to the prediction of the
432 ionic radius and charge systematics, it means that the interpolation of the observed daughter/parent
433 profile by the erf(x) function is inaccurate. Kooijman et al. (2010) only reported the Pb*/U ratio, and
434 the evidence of partial recrystallization as the cause of the anomalously fast Pb* transport in rutile
435 remains circumstantial.

436

437 **Age discordance in the Rb-Sr and K-Ar systems**

438

439 Other isotopic systems yield discordant age data even more often than the U-Pb chronometers.
440 In a paper long considered a classic, Hart (1964) described massive, ubiquitous rejuvenation of Rb-Sr
441 and K-Ar mineral chronometers of Proterozoic country rocks from the contact aureole associated with
442 the Eocene Eldora (Colorado) pluton. The aureole minerals analyzed gave Rb-Sr and K-Ar ages that
443 varied from sample to sample; no mineral chronometer gave the same age as the others in the same
444 rock. Hart (1964) extrapolated the concept of diffusive Pb loss from zircon to Sr and Ar loss from the
445 aureole minerals, attributing the discordance to thermally driven volume diffusion in an otherwise
446 chemically closed mineral-rock system. We will argue instead that metasomatism, caused by

447 infiltration of hydrothermal fluids in the contact aureole-pluton system, was orders of magnitude more
448 important in causing these age disturbances.

449 Skarns and contact aureoles feature replacive low-*P*, high-*T* mineral assemblages. It is a self-
450 evident consequence of Lavoisier's mass conservation law that secondary assemblages require a
451 chemically open system and fluid-associated mass transfer. As an example, porphyry copper
452 mineralization occurs near the apex of intrusive bodies because the fluids, exsolved from the granite,
453 transported chalcophile elements, not because heat creates Cu. Hart (1964) indeed described retrograde
454 reactions and mass transfer in the Eldora contact aureole, but preferred to attribute the loss of Ar and Sr
455 to pure Fick's Law thermal diffusion in a chemically inert matrix. From the rejuvenation of K-Ar and
456 Rb-Sr ages, he calculated diffusion coefficients for biotite. A re-examination of Hart's (1964) original
457 data allows a reassessment of the validity of his conclusions.

458 Hart's (1964) interpretations, as originally presented, contain a physically incorrect calculation,
459 as his samples had undergone a significant size reduction during contact metamorphism (which in itself
460 is an indication of recrystallization). He was comparing the Ar loss rate, D/a^2 , for large biotite grains
461 away from the contact with that for small grains near the contact. This created an artifact by decreasing
462 the true diffusion coefficient D of the samples furthest from the contact, as one order of magnitude
463 decrease in D/a^2 is accounted for by the factor of 3.1 in their grain size. Thus, the slope of the
464 Arrhenius diagram as used by Hart (1964, his Fig. 8) overestimates the activation energy of the process
465 responsible for Ar loss in biotite. Moreover, Hart was assuming diffusion from a sphere, while
466 subsequent work (Giletti 1974) utilized a more physically realistic cylindrical geometry. From the raw
467 data provided by Hart (1964, Table 2), it is necessary first to transform the calculated D/a^2 into
468 cylindrical D , and then divide D by one single grain radius common to all samples, taken here as $a =$
469 100 μm to make results directly comparable to those by Giletti (1974) discussed below. The D_0/a^2 and
470 E as originally proposed ($5 \times 10^{-7} \text{ s}^{-1}$ and 23 kcal/mol, respectively), together with the correctly
471 recalculated D_0/a^2 and E ($5 \times 10^{-8} \text{ s}^{-1}$ and 19 kcal/mol, respectively), are plotted in Figure 2.

472 Dodson (1973) systematized the diffusionist approach in a mathematical definition of the
473 concept of “closure temperature” T_c for the retention of Ar, Sr, Pb, etc. As an example, the “closure
474 temperature” of a 100 μm radius phlogopite grain cooling at 20 $^\circ\text{C}/\text{Ma}$ can be calculated from E and
475 D_0/a^2 proposed by Hart (1964) as $T_c = 254$ $^\circ\text{C}$. The closure temperature would be a useful concept if
476 the retention of radiogenic isotopes depended solely on T and volume diffusion, as is the case for
477 retention of fission tracks and He in apatite. In this case, it would be possible to constrain cooling
478 histories and, to the extent that the paleo-geothermal gradient is accurately known, provide a proxy for
479 exhumation. However, both the activation energy and the “closure temperature” of phlogopite
480 calculated from Hart's estimates are in gross contrast with the diffusion experiments by Giletti (1974).
481 The question arises why the Eldora “natural laboratory” came to be so discrepant from the controlled
482 experimental laboratory conditions.

483 The qualitative petrological argument that geological processes in contact aureoles are
484 principally controlled by fluids can be quantified by means of the dissolution rates measured by Wood
485 and Walther (1983). Several other studies quantified the dissolution rates of minerals in near-surface
486 conditions and soils. Wood and Walther (1983) determined the silicate dissolution rates over a broad
487 interval at high T , thus estimating an activation energy. It is an essential part of this discussion that the
488 fast dissolution rate proposed by Wood and Walther (1983) still stands as a lower limit, as no
489 subsequent studies (e.g., Gruber et al. 2016) have revised it downwards. Figure 2 quantitatively
490 compares the relative importance of Ar loss by volume diffusion and by dissolution/reprecipitation in
491 the founding example of diffusionism. The rate of Ar loss from biotite from the Eldora aureole
492 (corrected after Hart 1964) is shown in Figure 2 together with the dissolution rate measured by Wood
493 and Walther (1983), the diffusivity of Pb in apatite (Cherniak et al. 1991), and the nominal Ar
494 diffusivity inferred from the measurements by Giletti (1974). Villa (2010) corrected the Giletti (1974)
495 results for systematic deviations from Fick's Law, probably due to partial dissolution caused by the
496 presence of liquid H_2O in the capsule. This correction is validated by its agreement with the later

497 experiment by Giletti and Tullis (1977), which was done in absence of liquid water. The Eldora
498 samples all lie above the nominal Ar diffusivity, defining the same slope as the silicate dissolution line.
499 The Eldora and the phlogopite regression lines intersect at $x = 1.180$ ($T = 574$ °C). At 300 °C ($x =$
500 1.745) the laboratory diffusivity is 6.5 orders of magnitude slower than the Ar loss rate in the field.
501 This rounds off the proof that the mechanism causing the biotite age trend in the Eldora aureole was not
502 heat conduction leading to volume diffusion during the contact metamorphic event, but rather fluid-
503 induced recrystallization, as the latter is a faster process by several orders of magnitude.

504 Muscovite has always been considered the most retentive of the micas (Jäger 1967). Its
505 chronometric behaviour is crucial for understanding K-Ar systematics at large. Field evidence for the
506 very low intrinsic Ar diffusivity in muscovite is provided by the evidence of multiple white mica
507 generations with resolvably different ages in the same sample (Villa et al. 2014) and by the comparison
508 of U-Pb ages of monazite and K-Ar ages of muscovite in a metamorphic terrain infiltrated by aqueous
509 fluids (Tartèse et al. 2011). The latter authors documented a tight parallelism between these two
510 mineral chronometers, which implies that the loss of Pb* and Ar is controlled by the same mechanism.

511 Thus, thermally activated diffusion falls far short of causing discordance not only in the U-Pb
512 system but also in the K-Ar and Rb-Sr systems. In order to understand the true cause of the discordance
513 of K-Ar and Rb-Sr mineral chronometers, it is necessary to scrutinize what these dating methods are
514 based on.

515 Rb-Sr is mainly a “bulk sample” technique, by which the intra-grain isotope distribution is
516 usually not measured. This limits the information on causes and extent of discordance relative to, for
517 example, U-Pb dating of zircon and monazite samples. Even so, it is possible to detect discordance by
518 using overdetermined isochrons on a large number ($n \gg 2$) of minerals from the same rock. The
519 isochron calculation yields the MSWD parameter (McIntyre et al. 1966). If $MSWD \gg 1$, then the
520 sample is unlikely to ever have been in isotopic equilibrium. By linking the petrological information on
521 relict versus neofomed mineral phases to the Rb-Sr systematics, it is possible to unravel the system's

522 discordance (e.g. Glodny et al. 2008). Neoformed phases define a low-MSWD isochron, whereas the
523 petrologic relicts that carry Sr isotopic inheritance deviate from this isochron. A regression of relict
524 phases together with the neoformed phases gives an unacceptably high MSWD, due to a lack of
525 cogeneticity and of initial isotopic equilibrium. In case detailed petrologic groundwork evidences that a
526 given rock contains more than one mineral generation (be it retrogressive or relict phases in addition to
527 the magmatic or metamorphic assemblage), the whole rock data-point should never be used for an
528 internal isochron, as it cannot be in isotopic equilibrium with any of its heterogeneous constituents.

529 Before a detailed discussion of the K-Ar decay system (and its present-day analytical variety by
530 irradiation with fast neutrons, referred to here as ^{39}Ar - ^{40}Ar), it will be beneficial to briefly recall the
531 features that characterized the ^{39}Ar - ^{40}Ar dating technique at the time of its first development. In the
532 1960s, the Berkeley group around physicist John Reynolds developed the I-Xe dating method by
533 stepwise heating of irradiated samples (Jeffery and Reynolds 1961). Reynolds (1963) used three-
534 isotope common-denominator correlation diagrams to identify Xe derived from ^{129}I and ^{244}Pu , together
535 with its carrier phases. A spin-off of I-Xe, a combination of trace-element determinations and K-Ar
536 dating, was proposed by Merrihue (1965). This approach was pursued by Merrihue and Turner (1966),
537 who displayed their data with correlation diagrams. The dating method later came to be called
538 $^{40}\text{Ar}/^{39}\text{Ar}$ by Mitchell (1968), who also devised the age spectrum diagram. This data presentation was
539 perceived as a more immediate type of visualization and became very successful. The shift from
540 isotope correlation diagrams to age spectra puts all emphasis on a single isotope, ^{39}Ar . As graphics
541 limit how much we extract from data, this simplified presentation curtailed Merrihue's (1965)
542 discovery that rare gas isotopes (9 for Xe, 6 for Kr, 5 for Ar) in irradiated samples reveal
543 concentrations and carrier phases of trace elements such as U, I, Ba, Br, Cl, Ca, and K.

544 Turner et al. (1971) endeavored to date a lunar basalt using the ^{39}Ar - ^{40}Ar method. Because of
545 severe sample size limitations, their first analysis was done on an untreated whole-rock chip, which
546 returned a markedly discordant age spectrum. Turner et al. (1971) went on to enrich a plagioclase by

547 handpicking and obtained an age spectrum with a greatly reduced discordance (Fig. 5a). In an
548 interpretive scheme based exclusively on the age spectrum it is difficult to pinpoint the cause of the
549 reduction of the age discordance by handpicking. Information is lost irretrievably if just the two
550 isotopes $^{40}\text{Ar}^*$ and ^{39}Ar are considered. Only the full inventory of isotopes allows the recognition and
551 simultaneous dating of carrier phases. This was done by Turner et al. (1971) by linking the Ca/K
552 signature of the released Ar to plagioclase, groundmass matrix (termed mesostasis by Turner et al.
553 1971), and pyroxene. An alternative presentation of the same data in a common denominator three-
554 isotope correlation diagram is presented here in Fig. 5b.

555 Few studies (e.g., Onstott et al. 1991) extrapolated to terrestrial samples the approach by Turner
556 et al. (1971) that identifies mineral mixtures by heterogeneities in the $^{38}\text{Ar}/^{39}\text{Ar}$ and $^{37}\text{Ar}/^{39}\text{Ar}$ ratios. In
557 most other terrestrial geology studies it was rarely taken into account that carrier phases that have not
558 been completely reequilibrated by diffusion usually yield different mineral ages. The potential of ^{38}Ar
559 and ^{37}Ar to fingerprint the chemically heterogeneous mineral generations is not exploited by age
560 spectra, but only by multi-element-based three-isotope correlation diagrams. In fact, ever since the
561 work by Reynolds (1963) and Turner (1965), it is clear that the halogen-derived ^{128}Xe and ^{38}Ar are
562 essential for the detection of heterochemical phases. Moreover, such phase inventories can nowadays
563 be identified with imaging techniques such as CL, BSE, and X-ray element maps, and quantified by
564 EPMA in situ chemical analyses. Their identification by EPMA in combination with common
565 denominator Ar isotope correlation diagrams, and their association with discordant age spectra, forces
566 the conclusion that discordance in K-Ar, just as in U-Pb and I-Xe, is due to mixed mineral generations
567 (e.g. primary crystallization and later metasomatic alteration zones or regions, produced by the
568 infiltration of hydrothermal fluids).

569 Figure 6 highlights that CL imaging (Figs. 6a, c, e) can offer additional petrological information
570 with respect to optical microscopy (Figs. 6b, d, f). In Figure 6a, the dark regions of the K-feldspar
571 (sanidine) crystals have been partially dissolved and replaced by melt but the crystals have a “fresh”

572 igneous CL emission. In Figure 6c the K-feldspar (microcline) crystal has been partially
573 hydrothermally altered and replaced by low-temperature, non-luminescing adularia (Chafe et al. 2014).
574 In Figure 6e, the green CL plagioclase (albite) shows some alteration and a “core” that is enriched in
575 Ba and Sr (that is not visible in the cross-polarized photograph in Fig. 6f). The blue CL K-feldspar also
576 shows some hydrothermal alteration and partial replacement by adularia.

577 Indeed, monomineralic samples give concordant age spectra even if they feature intracrystalline
578 age gradients (Hodges et al. 1994). It can be noted that the patchy, asymmetric intracrystalline age
579 distributions in the samples documented by Hodges et al. (1994), as was the case with that reported by
580 Phillips and Onstott (1988), are incompatible with true volume diffusion, as the erf(x) function is
581 symmetrical and has its maximum at the center of the grain. Polymineralic samples, formed by mixed
582 mineral generations, are typical in retrogressed rocks, in which replacement reactions resulted in
583 petrologic disequilibrium unaccompanied by diffusive re-equilibration. EPMA X-ray element maps can
584 quantify their characteristic patchy textures. Mineral chronometers (micas, amphiboles, K-feldspar)
585 showing secondary reactions also show Ca/Cl/K signatures that correlate with step ages (e.g., Chafe et
586 al. 2014). Correlation plots linked to EPMA successfully unravel ages of each mineral generation (e.g.,
587 Villa et al. 2000). The recognition that Ca/K and Cl/K ratios can identify the end-members of a phase
588 mixture led to the definition of "isochemical steps" (Müller et al. 2002; Villa et al. 2006) as those steps
589 that have a uniform Ca/Cl/K signature. As it was observed that isochemical steps usually have uniform
590 ages as well, the implication is that isochemical steps monitor the degassing of one isochemical phase
591 characterized by a (nearly) concordant age, the "isochemical age".

592 As a working hypothesis, it can be stated that the coincidence between discordant ^{39}Ar - ^{40}Ar age
593 spectra and the presence of heterochemical, diachronous, mineral phases is not a mere occasional
594 coincidence but a causal necessity. In order to test this hypothesis, three steps are required:

595 (A) Artificially create a mixture of heterochemical, diachronous minerals and examine: (A1) if
596 a discordant age spectrum is produced and (A2) if the age signature of the mixed minerals can be
597 unravelled by the appropriate three-isotope correlation diagrams.

598 (B) Identify natural heterochemical systems, for which sufficiently large grains allow the
599 mapping of intra-grain age gradients and compositional mapping, and compare them to the artificial
600 mixtures.

601 (C) Identify natural monomineralic systems, for which intra-grain K-Ar age gradients both
602 follow the bell-shaped erf(x) profile required by Fick's Law for an inert, unreacted matrix, and are in no
603 way associated with chemical heterogeneities, and compare them with the mixtures described in (A)
604 and (B).

605 An artificial mixture following (A) was first described by Wijbrans and McDougall (1986), who
606 reported that a mixture between a young muscovite and an older phengite, both of which gave
607 comparatively flat spectra, yielded a discordant, hump-shaped age spectrum. As these authors provided
608 an incomplete dataset (no ^{37}Ar or ^{38}Ar were determined), the potential of the experiment was not fully
609 exploited. Subsequent artificial mixtures were reported by Villa et al. (1996) and by Kula et al. (2010).
610 Villa et al. (1996) observed unexpected differences between step-heating and step-leaching age spectra
611 of a natural hornblende separate. By TEM they detected secondary K-feldspar in a cleavage from an
612 amphibole grain. This led to the hypothesis that the distinct differential degassing rate was the signature
613 of a secondary phase. To test this hypothesis, they mixed young K-feldspar with the older amphibole.
614 The age spectrum of the mixture offers no immediately obvious clarification (Fig. 7a). However,
615 unravelling is provided by exploiting all five Ar isotopes. Villa et al. (1996) found that there is an
616 extremely close correspondence between the Cl/Ca/K fingerprint deduced from the common-
617 denominator isotope correlation diagrams and the step ages (Fig. 7b-c). The second instance of an
618 artificial mineral mixture, discussed here, had been incompletely followed through by its proponents,
619 Kula et al. (2010). These authors reported the effect of artificially mixing two biotite samples, PM1 and

620 IV8, in variable mass ratios and concluded that no information could be gained on the mixing end-
621 members, as age spectra were discordant (Fig. 7d). On the contrary, the systematics established in
622 Figures 7b-c can be advantageously applied to the mixture of two heterochemical biotites by a more
623 goal-oriented plot of the original data (Kula et al. 2010, p. 74). While the age spectrum based only on
624 ^{39}Ar is discordant, the correlation between the $^{38}\text{Ar}/^{39}\text{Ar}$ (Cl/K) ratio and the age (Fig. 7e) reveals
625 almost ideal binary mixing (Villa 2001, his Fig. 2) and recovers the age of both end-members if the Ca,
626 Cl, and K concentrations are measured by EPMA. The reason why the age spectrum, based only on
627 ^{39}Ar , is insufficiently informative is that the differential release rates of the two biotite samples are
628 different (Fig. 7f). This distorts the mass balance of the summed Ar deriving in part from PM1 and in a
629 different proportion, variable from step to step, from IV8. The three-isotope correlation diagram in
630 Figure 7e does not suffer from this limitation and simply follows the algebraic rule valid for all binary
631 mixtures (Villa 2001).

632 Natural samples consisting of clear-cut polyminerally mixtures (approach B) display discordant
633 age spectra (Onstott et al. 1990; Villa et al. 2000). The variation range of the age spectrum of the
634 sample studied by Onstott et al. (1990) shows less extreme variations compared to the ^{39}Ar - ^{40}Ar age
635 map obtained by laser ablation. Most importantly, the Cl/Ca/K fingerprint deduced from the common-
636 denominator isotope correlation diagrams follows the same pattern as the artificial mixtures of
637 approach (A). A natural sample that mimics the Wijbrans and McDougall (1986) mixture, when all five
638 Ar isotopes are analyzed, confirms and expands the findings of these authors and provides a
639 quantitative road map to deconvolve the muscovite and phengite end-members responsible for
640 discordant hump-shaped age spectra (Villa et al. 2014, their Fig. 9). A conception frequently heard is
641 that staircase-shaped age spectra reflect Ar loss, whereby it is often assumed (Hart 1964) that Ar loss is
642 always due to thermally activated diffusion. An example of a staircase-shaped age spectrum that
643 illuminates this issue is detrital white mica Rui9978 (Fig. 8a) that was burial-heated to $T < 300$ °C in
644 the sedimentary Houiller Basin in the French Western Alps (Villa et al. 2014, their Fig. 10 a-b). The

645 common denominator three-isotope correlation diagram (Fig. 8b) demonstrates that sample Rui9978
646 consists of two populations. EPMA identifies these populations to be intergrown on the μm scale.
647 These consist of a detrital one, with high Si concentrations, and a metamorphic one, with high Al
648 concentrations. The combination of EPMA analyses with the correlation diagrams establishes that
649 recrystallization in a chemically open system involving mass transfer of cations, such as Al and Si, is
650 the predominant cause of the low-temperature Ar loss in Rui9978.

651 The presence of natural mixtures among micas of different generations in metamorphic rocks
652 opens a side-track on the optimum analytical approach to estimate the age of each generation. The first
653 action required is obtaining microstructural/microchemical imaging and establishing the size,
654 distribution and composition of the heterogeneities. Recent observations (Berger et al. 2017) stress that
655 in sheared rocks the retrogression/recrystallization products have a grain size $< 10 \mu\text{m}$ (Fig. 9a-b),
656 whereby the successive mica generations are tightly intergrown. Perhaps more surprisingly, "static"
657 retrogression of an eclogite in the greenschist-facies also occurs very heterogeneously, on a scale < 10
658 μm (Fig. 9c). Given the impossibility to perform a mineral separation on a sieve fraction of $< 10 \mu\text{m}$,
659 the next action required is choosing an analytical approach that maximizes the accuracy of the obtained
660 information. As already pointed out by Müller et al. (2002, p. 73), ^{39}Ar - ^{40}Ar analyses by laser ablation
661 require a minimum sample size. Two contrasting requirements must be met. The beam size should be
662 the smallest possible, in order to discriminate microstructures within a mineral and the sample size
663 should be as large as possible, so that the counting statistics and the non-zero detection limit for ^{36}Ar
664 have a small relative effect on the overall sum of the propagated uncertainties. The choice by Flude et
665 al. (2014) of ablating areas $\geq (100 \mu\text{m})^2$ of a 470 Ma old K-feldspar is a compromise that gave an
666 acceptable uncertainty of 0.5 %. In order to achieve similar precision on Cenozoic samples, larger
667 beam sizes are necessary (Müller et al. 2002). The best spatial resolution of a laser-ablation ^{39}Ar - ^{40}Ar
668 analysis is not dictated by the laser beam diameter itself, but rather by the minimum sample size
669 requirement, and is at least an order of magnitude larger than the typical size of the microstructures to

670 be dated (Figs. 9b,c). Therefore the volume of mineral ablated in a laser microprobe ^{39}Ar - ^{40}Ar analysis
671 will as a rule give a mixed age, as it contains all of the mineral generations of that sample. A way to
672 disentangle heterochemical mineral generations is by associating stepwise heating of "large" samples
673 (ca. 1-10 mg) to the Ca/Cl/K systematics provided by common-denominator three-isotope correlation
674 diagrams (Fig. 7). Using this technique, Federico et al. (2005) were able to date both mica generations
675 of the sample shown here in Figure 9c.

676 Case (C) has been observed only once in a natural sample, namely the gem-quality K-feldspar
677 from Itrongay, Madagascar (Flude et al. 2014). The mineral is low sanidine (as defined following the
678 structure refinement by Nyfeler et al. 1998), but can also be considered as orthoclase (e.g., Wartho et
679 al. 1999). No stepwise heating spectrum was reported for the crystal rim piece for which the Ar*
680 concentration profile was obtained. Stepheating was performed (Nyfeler et al. 1998; Villa, unpublished
681 data, 1998) on one aliquot of the Itrongay K-feldspar crystal used by Wartho et al. (1999) to quantify
682 the diffusive gradients caused by isothermal laboratory heating. One $(100\ \mu\text{m})^3$ chip was heated in a
683 resistance oven at 900 °C for 15 h. This chip was used by Nyfeler et al. (1998) to compare the
684 respective rates of structural disordering and Ar loss. The chip then was irradiated, and subsequently
685 degassed completely in 13 heating steps between 590 and 1530 °C. The total-gas K-Ar age of 94 Ma
686 corresponds to a bulk Ar loss of 79 %. This bulk Ar loss prompted Nyfeler et al. (1998) to state that Ar
687 loss is slower or equal to the Al,Si disordering rate. Since the experiments by Wartho et al. (1999) and
688 Flude et al. (2014) demonstrate that the Itrongay gem-quality K-feldspar follows Fick's Law, whereby
689 the diffusion length scale coincides with the physical grain size, it follows that in vacuo heating to 900
690 °C of the unirradiated K-feldspar chip that later was irradiated for the present experiment must have
691 resulted in a bell-shaped diffusion profile. The laboratory degassing of Ar, whose spatial distribution
692 follows a bell-shaped concentration gradient, can be modelled under the assumption that the degassing
693 rate follows Fick's Law diffusion. The predicted age spectrum is shown in Figure 10 as a dashed line
694 climbing to ca. 300 Ma.

695 The measured age spectrum, presented here as solid line in Figure 10, shows a subdued
696 staircase. The sub-optimum agreement between the diffusive degassing of the Ar spatial gradient and
697 the shape of the age spectrum is evidence that stepheating does not ensure a satisfactory recovery of the
698 spatial gradients that can only be quantified with laser-ablation ^{39}Ar - ^{40}Ar analysis on a length scale of
699 several mm.

700 The physical mechanism of laboratory degassing is thus directly observed to be different from
701 that causing Ar loss in nature. Most natural samples that underwent rejuvenation are not recording
702 Fickian diffusion anyway (see above), but the question is, do laboratory stepheating experiments? The
703 answer is provided by the context of the Itrongay K-feldspar experiments. The Itrongay pegmatite
704 hosting the K-feldspar did undergo a prolonged medium-temperature history (500-600 °C?), which was
705 so dry that the K-feldspar was not retrogressed to deuterically coarsened perthite (an exceptional
706 instance: Plümper and Putnis 2009), and gave rise to the only bell-shaped ^{40}Ar diffusion profile
707 observed so far (Flude et al. 2014). Degassing by stepheating between 590 °C and 1530 °C did not
708 recover the bell-shaped profile, hence had not followed Fickian diffusion. An explanation originally
709 proposed by Hetherington and Villa (2007) assigns a pivotal role to structural rearrangement of the
710 silica tetrahedra, phonons, and one or more unquenchable phase transitions (Laves 1956). The
711 existence of phonons and phase transitions was unknown to Fick when he formulated his Law
712 quantifying nutrient transport in aqueous solutions in the mid 19th century.

713 The available petrological data show that the discordance of ^{39}Ar - ^{40}Ar age spectra is always
714 dominated by heterochemical phase mixtures, which can only be correctly diagnosed by using isotope
715 correlation plots that take into account both textural images obtained by various techniques (CL, BSE,
716 EPMA X-ray maps) and measuring all five Ar isotopes. In the only instance reported so far of a truly
717 monomineralic sample, in which a bell-shaped age gradient was documented by laser-ablation ^{39}Ar -
718 ^{40}Ar analysis, the stepheating protocol does not yield a staircase-shaped age spectrum reminiscent of
719 the known intra-grain $^{40}\text{Ar}^*$ distribution.

720

721

Implications

722

723 A change of paradigm in geochronology has been slowly emerging in recent years. As more
724 detailed high-resolution petrologic imaging and mapping has shown the complexity of natural
725 materials, early physical models of isotope transport based on simplifying assumptions have been
726 superseded by more articulate views based on the labor-intensive establishment of a context between
727 mineral petrogenesis, microstructural constraints, and intra-grain chemical and isotopic disequilibria. It
728 is still true that the U-Pb, Rb-Sr, and K-Ar isotopic systems share deep-reaching similarities, as already
729 was thought half a century ago by Wetherill (1963) and Hart (1964). However, in complete antithesis to
730 the hypotheses formulated at that time, the similarities do not lie in the fact that in these systems age
731 discordance is caused by Fick's Law diffusion at high temperature, but in the fact that in both U-Pb and
732 K-Ar age discordance is principally due to mixing heterochemical, diachronous phases. Since volume
733 diffusion is orders of magnitude slower than dissolution/reprecipitation at most metamorphic
734 temperatures, it plays a subordinate role whenever circulating fluids are available to promote
735 retrogressive reactions.

736 The mineralogy of the mineral geochronometers reflects first and foremost the petrologic
737 conditions. To the extent that patchy chemical and crystallographic zonations are observed, it is self-
738 evident that diffusive re-equilibration was at most incomplete, if not negligible, whereas fluid-assisted
739 recrystallization was predominant. Therefore, the main control on the isotopic record of mineral
740 geochronometers is the latter, not the former. Microstructural observations associated with chemical
741 microanalyses ubiquitously document that, as a rule, patchy retrogression textures are associated with
742 heterochemical signatures (U/Th ratios, REE patterns, Ca/Cl/K ratios) and thus provide firm evidence
743 of a chemically open system during the reactions that caused petrologic disequilibrium. The geological

744 interpretation of age discordance requires an intensive mineralogical-petrological framework, as
745 discordance is caused by mixtures of heterochemical, resolvably diachronous, mineral generations.

746

747

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748

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1002

1003

Figure Captions

1004

1005 Fig. 1 - Concordia diagram following Wetherill (1956). The solid black curve with the ages (given in
1006 Ma) marked with red dots is referred to as the concordia curve. It is the geometric locus of all points
1007 whose two U-Pb ages are concordant. The green ellipse represents the analysis of such a "concordant"
1008 sample, which crystallized 2000 Ma ago. The orange ellipse, which lies significantly off the concordia,
1009 represents the analysis of a sample whose ^{238}U - ^{206}Pb age is lower than the ^{235}U - ^{207}Pb age; the
1010 chronological information that can be extracted from this sample is not straightforward.

1011

1012 Fig. 2 - Rates of relevant processes affecting the U-Pb and K-Ar systems. Solid lines: Pb diffusivity in
1013 apatite (Cherniak et al. 1991) and corrected Ar diffusivity in phlogopite (Giletti 1974; Villa 2010).
1014 Within stated uncertainty, the slopes of the phlogopite and apatite lines are indistinguishable at the 1
1015 sigma level. Dashed line: dissolution rate of silicates in water (Wood and Walther 1983), which
1016 coincides with the Pb*, Sr, and Ar loss rate of reacting silicates, as incompatible daughter isotopes are
1017 lost upon dissolution. Dash-dotted line: Ar loss rate from the Hart (1964) original data corrected for
1018 variable grain size and cylindrical geometry. Both the *a priori* arguments for a chemically open aureole
1019 mentioned in the text and the parallelism of the Sr and Ar loss rate to the aqueous dissolution rate of the
1020 silicates are conclusive *a posteriori* evidence that Hart (1964) misinterpreted the isotopic evidence for
1021 Fick's Law of diffusion. Argon loss from the Eldora samples was dominated by the aqueous dissolution
1022 quantified by Wood and Walther (1983), as the Ar diffusivity in phlogopite, proposed by Giletti (1974),
1023 is several orders of magnitude slower. Pb diffusivity in apatite is marginally slower.

1024

1025 Fig. 3 - Wetherill (1956) Concordia diagram with back-scattered electron image of a zircon crystal
1026 from a Cretaceous peraluminous granite. Inset: crystal showing locations of SIMS analysis. The

1027 analysis locations were selected intentionally to analyze the rim, core, and to straddle the rim-core
1028 boundary. This crystal illustrates that the core of a zircon crystal is often not in the geometric center of
1029 the grain, that it forms an angular unconformity with its overgrowths, and that the thickness of the rim
1030 varies significantly throughout the crystal.

1031

1032 Fig. 4 - Concordia diagrams showing different ways to plot U-Pb data. (a) Wetherill (1956) Concordia
1033 diagram. The schematic representation of Fig. 1 is augmented by a listing of possible causes of
1034 discordance. Two clusters of concordant data-points are shown, one at 2000 Ma (labelled "igneous or
1035 metamorphic cores") and one at 1000 Ma (labelled "igneous or metamorphic rims"), with reference to
1036 their growth sequence in individual grains. Discordant data-points are color coded with respect to the
1037 geological/mineralogical cause of discordance, as are the trajectories that displace data-points from
1038 concordia. High-temperature diffusive Pb* loss is not represented because it has not been observed in
1039 natural zircon and monazite samples. Instead, the dominant cause of discordance is the physical mixing
1040 of young rims and old cores (e.g., Mezger and Krogstadt 1997), shown by the dashed orange line and
1041 the orange ellipses. The caption "large/small" rims and cores refers to the mass balance of the U in core
1042 and rim, not to their volume: a thin rim with very high U concentration can host most of the grain's U
1043 and will accordingly drag the physical mixture far towards the lower left. The caption "low-T Pb* loss"
1044 points out that old, metamict cores can undergo a loss of Pb* by interacting with aqueous fluids in
1045 near-surface or low-grade conditions. This loss of Pb* can occur at the present day (blue dashed line,
1046 blue ellipses) or can have occurred at some time in the past (orange caption), concurrently with the
1047 formation of metamorphic rims. The grey dashed line represents a mineralogical/geochemical
1048 disturbance, whereby very small inclusions rich in non-radiogenic Pb are included in the U-rich, Pb-
1049 poor chronometer grains. The two-arrowed red line represents an analytical artefact that can be
1050 observed in microbeam analyses (more frequently in laser ablation plasma source mass spectrometry,
1051 sometimes also in secondary ion mass spectrometry): U and Pb are vaporized from the matrix and

1052 ionized at different rates from each other and from those in the reference material used for calibration,
1053 achieving either an artificial U loss+Pb gain, or an artificial Pb loss+U gain. Finally, the two purple
1054 ellipses labelled "ambiguous" are data-points that are analytically indistinguishable from the concordia,
1055 but give a different age than the cluster of green ellipses. Interpreting their age (e.g., in terms of
1056 protracted magmatogenesis) requires establishing a context with non-chronometric indicators, such as
1057 trace element patterns (e.g. Schoene et al. 2010). - (b) Tera-Wasserburg Concordia diagram (Tera and
1058 Wasserburg, 1972a,b). The curve with the ages marked with red dots is also a concordia curve, as it is
1059 the locus of concordant data-points. This alternative diagram is similar in principle to Fig. 4a but is
1060 based on the $^{207}\text{Pb}/^{206}\text{Pb}$ vs. $^{238}\text{U}/^{206}\text{Pb}$ relation. The color-coding of the relevant causes and trajectories
1061 of discordance is the same as in Fig. 4a.

1062

1063 Fig. 5 - (a) Age spectra of two subsamples of lunar rock 14053 (redrawn from the data in Turner et al.
1064 1971). Dashed line, whole rock; solid line, plagioclase. The discordance is strongly reduced when the
1065 phase assemblage is modified, requiring that most of the observed discordance of that sample was due
1066 not to intra-grain age gradients in a homogeneous material but rather to the simultaneous presence of
1067 young matrix (mesostasis) and old plagioclase. (b) Common-denominator three-isotope correlation
1068 diagram (from the data in Turner et al. 1971). Filled diamonds - unpicked whole rock; open triangles -
1069 hand-picked plagioclase. Since the Ca/K signature of pyroxene, plagioclase, and matrix is different, it
1070 is possible to assign a K-Ar age to each of these phases.

1071

1072 Fig. 6 - Real-color cathodoluminescence (CL) and cross-polarized light (XPL) microscopy images of
1073 feldspars in thin sections from different geologic environments. (a), (b) Sanidine crystal (blue CL) from
1074 the Peach Springs Tuff in Arizona, USA. The arrows point to the outside edge of the large previously
1075 euhedral crystal which has experienced partial dissolution but otherwise shows no evidence for
1076 metasomatic alteration by hydrothermal fluids. (c), (d) Microcline crystal from the Chain of Ponds

1077 granite (Chafe et al. 2014) in Maine, USA. The arrows point to regions in the microcline, and at the
1078 grain boundary with plagioclase (green CL), where the microcline has been partly replaced by non-
1079 luminescing adularia. (e), (f) Microcline and plagioclase from the Sweetwater Wash granite, Mojave
1080 Desert, California, USA. The bottom arrows point to regions in the microcline and plagioclase that
1081 have experienced metasomatic alteration by hydrothermal fluids. The upper arrow shows internal
1082 zoning in the plagioclase crystal that is not visible in the XPL image. The CL-revealed zoning is
1083 enriched in Ba and Sr as determined by EPMA (J.M. Hanchar, unpublished results). Mineral
1084 abbreviations: Bt, biotite; Kfs, K-feldspar; Pl, plagioclase; Qz, quartz.

1085
1086 Fig. 7 - Artificial mixtures of heterochemical, diachronous phases. (a) Age spectrum of amphibole–K-
1087 feldspar mixture (redrawn after Villa et al. 1996). Arrows indicate the ages of the pure amphibole and
1088 the pure K-feldspar. The age spectrum alone gives no clear indication on how to deconvolve the K-
1089 feldspar-dominated, artificial mixture. (b) Common-denominator, three-isotope correlation diagram
1090 (Cl/K vs Ca/K) fingerprinting the chemical signatures of the artificial mixture. Filled circles denote the
1091 pure K-feldspar (Cl-free, Ca-poor), the pure amphibole (Ca+Cl-rich), and the bulk sum of all steps
1092 (dominated by the K-feldspar). Open circles are the individual heating steps of the artificial mixture.
1093 The linear correlation (solid line) reflects the artificial binary mixing. (c) Common-denominator three-
1094 isotope correlation diagram (Ar*/K vs Ca/K). Binary mixing is shown by the solid line. By
1095 extrapolating (dashed line) the correlation to the independently determined Ca/K ratio of the pure
1096 amphibole, it is possible to successfully recover its age, even if the individual heating steps (open
1097 circles) never even approach the pure end-member. (d) Age spectra of two biotite samples and two
1098 artificial mixtures in different mass ratios (redrawn after Kula et al. 2010). Red squares - pure biotite
1099 IV8; light blue circles - pure biotite PM1; Mix1 - mixture IV8-PM1 3:1 (open brown squares); and
1100 Mix2 - mixture IV8-PM1 1:3 (open blue-green triangles). The x-axis is the step temperature (following
1101 Fitch et al. 1969), rather than the percentage release of ³⁹Ar, as it more clearly evidences the parallel

1102 differential release patterns of the two pure end-members PM1 and IV8. The age spectra alone,
1103 carrying only incomplete information, allowed Kula et al. (2010) no positive identification of the
1104 mixture's systematic behavior. (e) Common-denominator three-isotope correlation diagram relating the
1105 ages of PM1 and IV8, and the chemical signature of PM1, to those of mixtures Mix1 and Mix2
1106 provided by Kula et al. (2010). The open squares and triangles, representing Mix1 and Mix2, define an
1107 almost ideal binary mixing between PM1 and IV8. As pure biotite, IV8 had been irradiated with Cd
1108 shielding, which destroys all information regarding the Cl/K ratio. Its presumed correct position in the
1109 Cl/K-age common denominator isotope correlation diagram is shown by an open ellipse. (f)
1110 Differential Ar release plot. The release rate k on the ordinate axis is calculated from the percentage
1111 ^{39}Ar released in a given step divided by the temperature increase since the preceding step. It can be
1112 seen that the differential degassing rate of IV8 (red squares) has a different shape from that of PM1
1113 (blue circles). IV8 predominates over PM1 in two temperature intervals, i.e. $T < 700\text{ }^\circ\text{C}$ and $860\text{ }^\circ\text{C} < T$
1114 $< 940\text{ }^\circ\text{C}$, in correspondence with the two age maxima of the curves labelled Mix1 and Mix2 in Figure
1115 7d. PM1 dominates the release for $T > 970\text{ }^\circ\text{C}$, when IV8 is practically exhausted.

1116

1117 Fig. 8 - Natural mixtures of heterochemical, diachronous phases in diagenetically overprinted detrital
1118 white mica Rui9978 (diagrams redrawn after Villa et al. 2014). (a) Staircase-shaped age spectrum. (b)
1119 Common-denominator, three-isotope diagram correlating age with chemical composition. The staircase
1120 shaped age spectrum is caused by the presence of two heterochemical, diachronous phases: a Cl-poor,
1121 *ca.* 300 Ma old detrital muscovite generation, and a much younger Cl-rich metamorphic phengite
1122 over/intergrowing it.

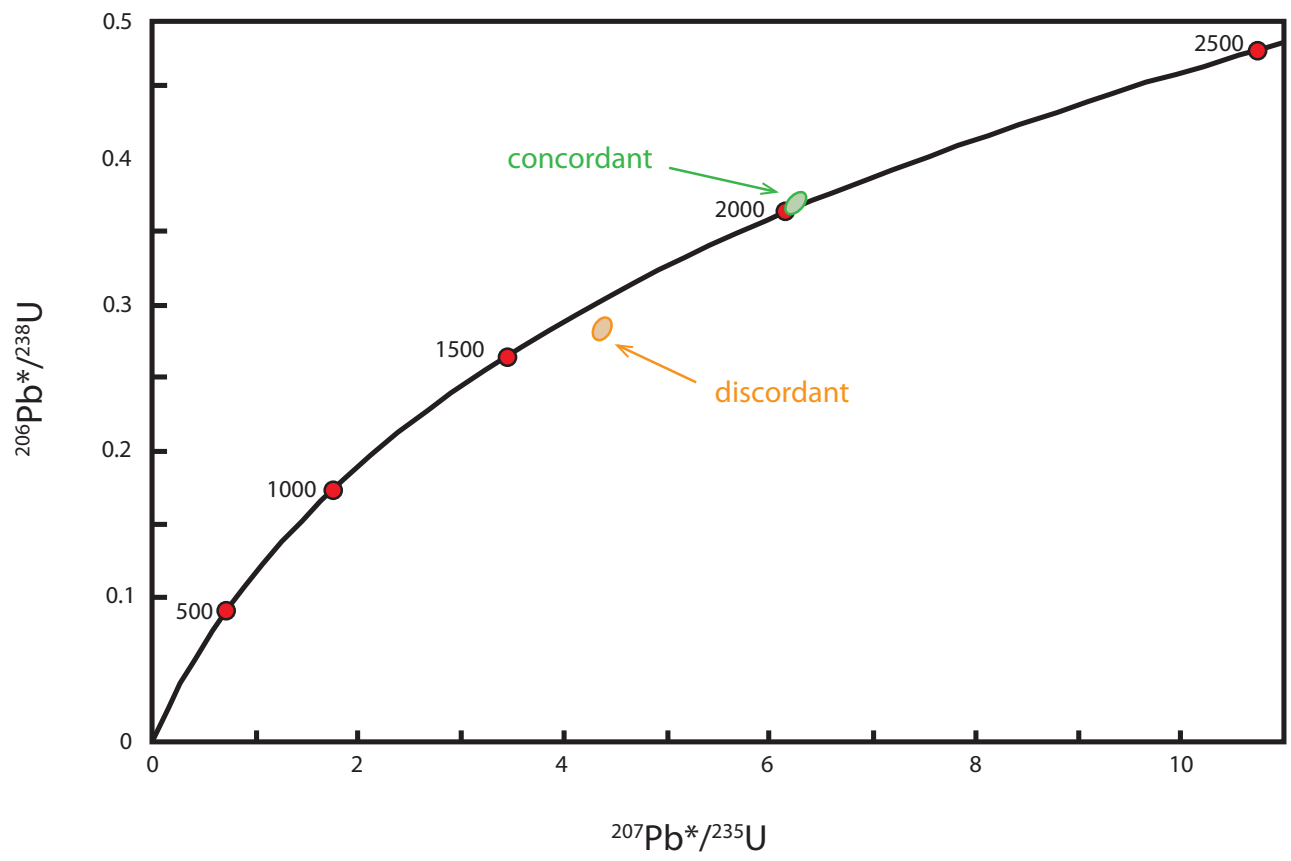
1123

1124 Fig. 9 - Comparison of natural grain sizes of retrogression products of sheared and unsheared rocks. (a)
1125 optical microphotograph of fault rock Ga142, from the Aar shear zone (Berger et al. 2017). (b)
1126 Aluminium element map of Ga142. Magenta and red are high-Al muscovite, green and yellow are

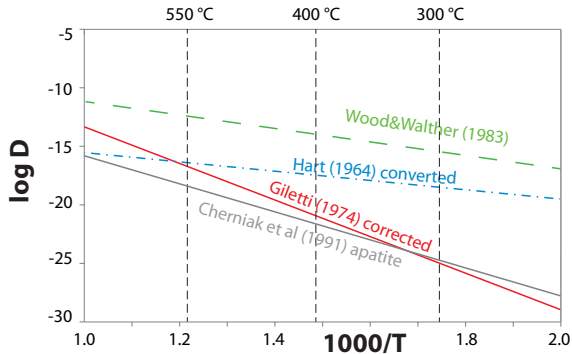
1127 high-Si phengite. The black circle in the middle of the picture has a diameter of 100 μm and represents
1128 an imaginary laser pit (cf. Flude et al 2014). Any 100 μm pit on this sample would include all of the
1129 mica generations at once. (c) BSE image of static retrogression of eclogitic phengite by greenschist-
1130 facies muscovite (modified after Villa 2006). Field of view ca. 40 x 80 μm . Ms3.1 stands for BSE-dark
1131 layers of replacive muscovite, with an average Si concentration of 3.1 atoms per formula unit (apfu)
1132 and an age of ca. 40 Ma. Ph3.6 stands for BSE-bright eclogitic phengite with an average Si
1133 concentration of 3.6 apfu and an age of ca. 49 Ma. Retrogression occurs in an "all-or-nothing" fashion
1134 parallel to the basal cleavage. The red circle has a diameter of 50 μm . The 100 μm pit is much larger
1135 than the entire image. The age of such a hypothetical *in situ* analysis would be 45 Ma, unrelated to both
1136 phengite formation and muscovite retrogression.

1137

1138 Fig. 10 - Age spectrum (solid line) of pre-heated gem-quality K-feldspar from Itrongay, Madagascar.
1139 The diffusivity of the aliquots of the present grain was studied by Wartho et al. (1999) and found to
1140 follow Fick's Law. The predicted age spectrum, corresponding to a bell-shaped core-rim age profile,
1141 and scaled by the volumes of the respective thin spherical shells of the modeled crystal, is shown by the
1142 dashed line. The nearly flat age spectrum fails to correctly extract the age zoning.



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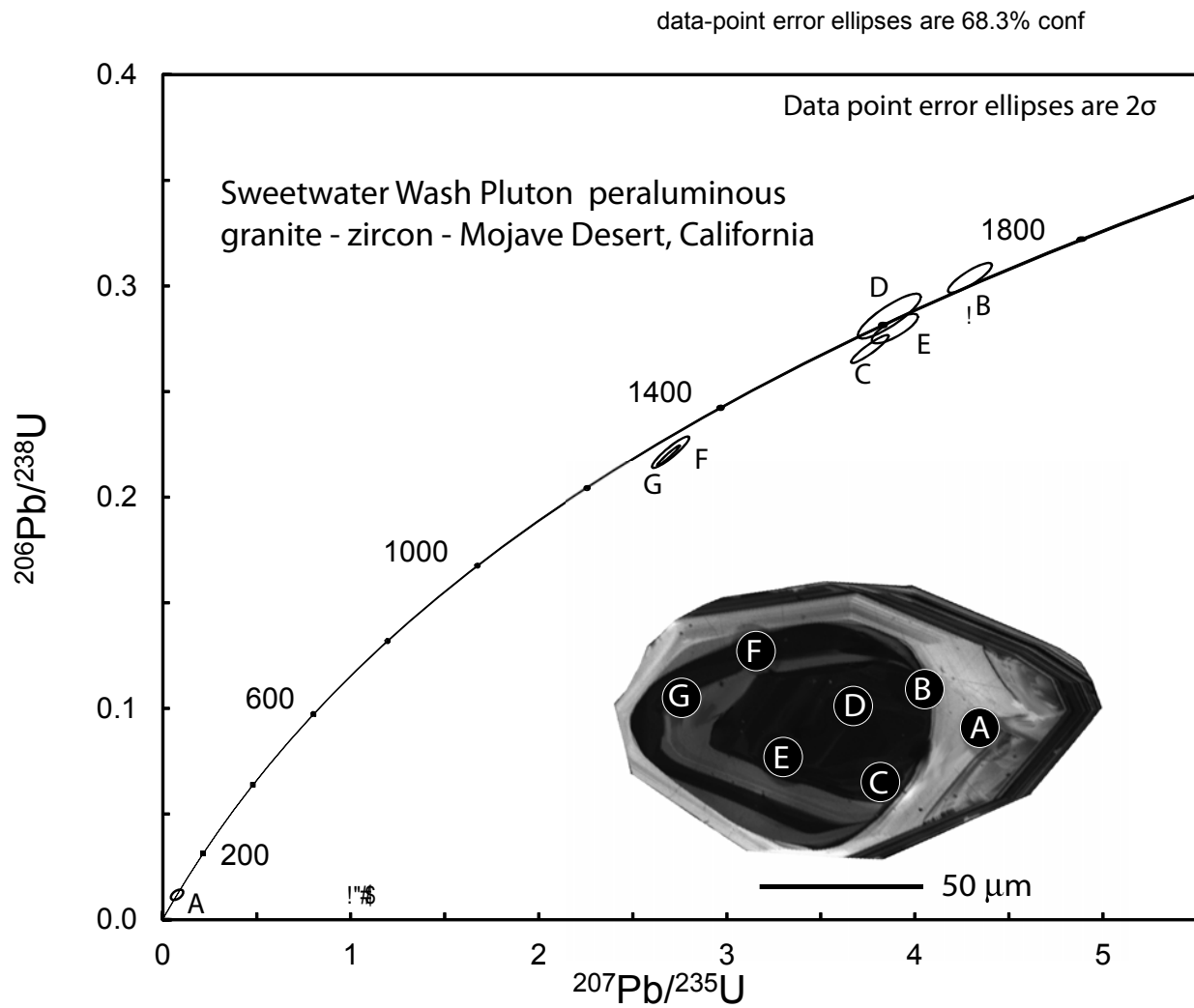
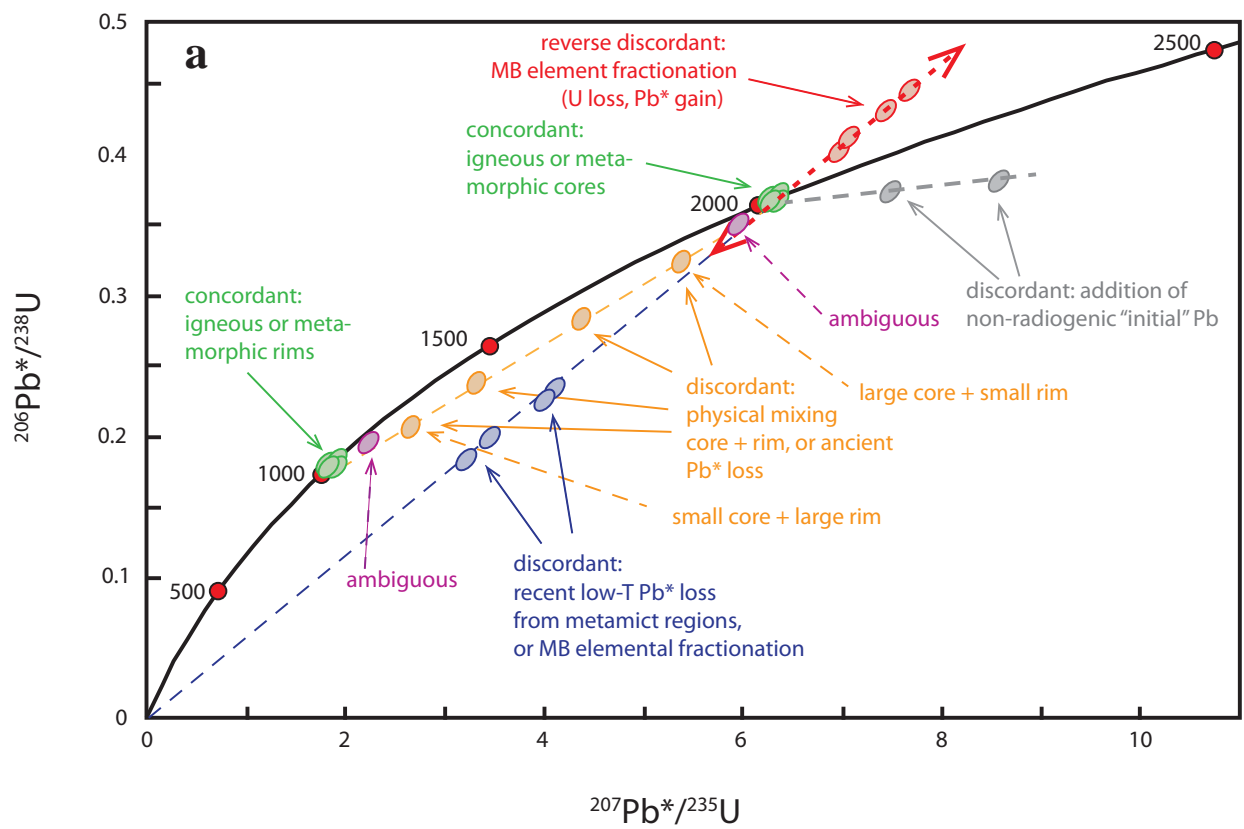
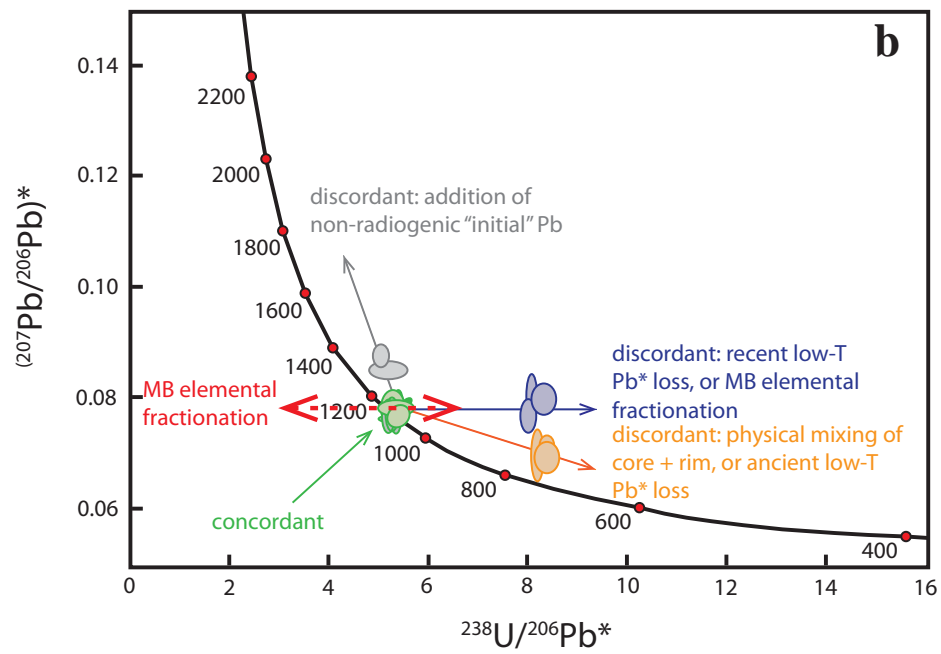


Figure 2





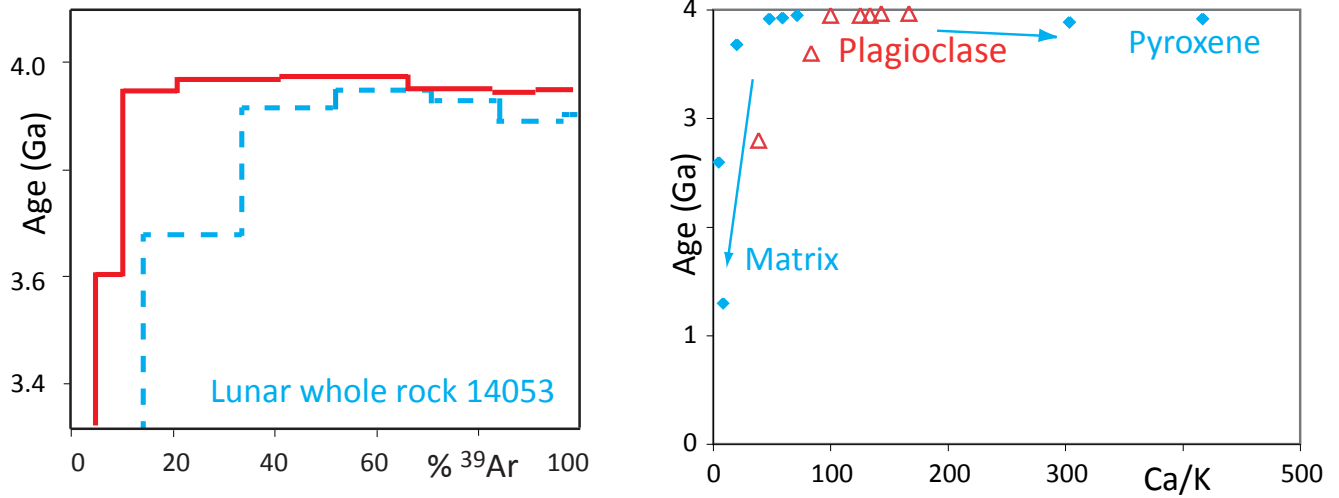


Figure 4

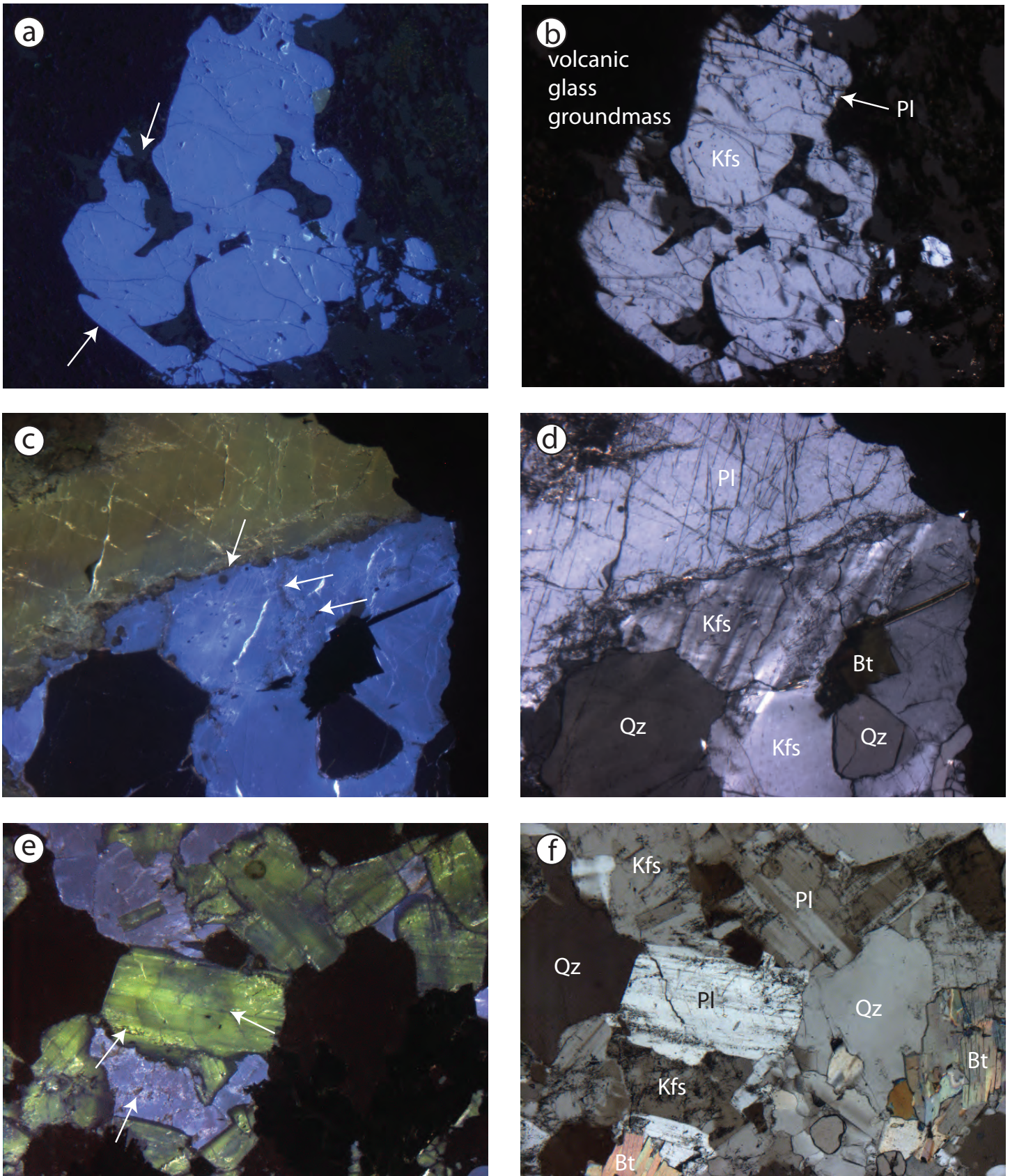


Figure 5

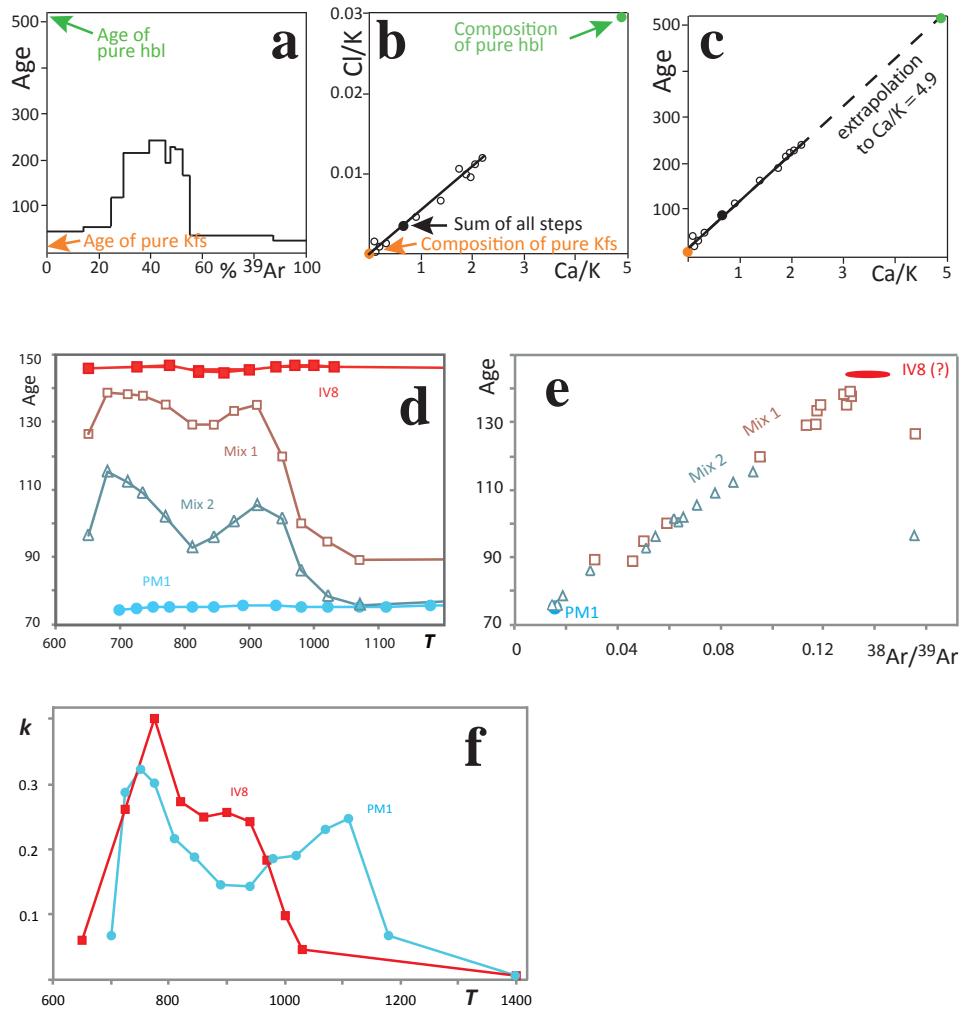


Figure 6

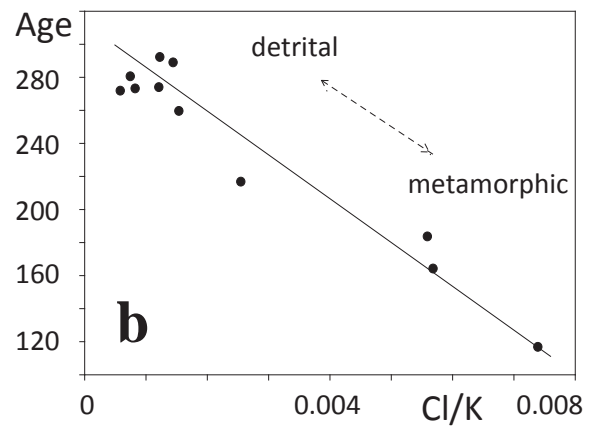
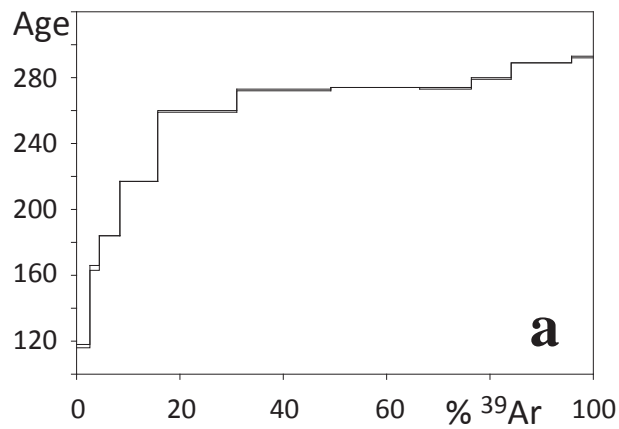


Figure 7

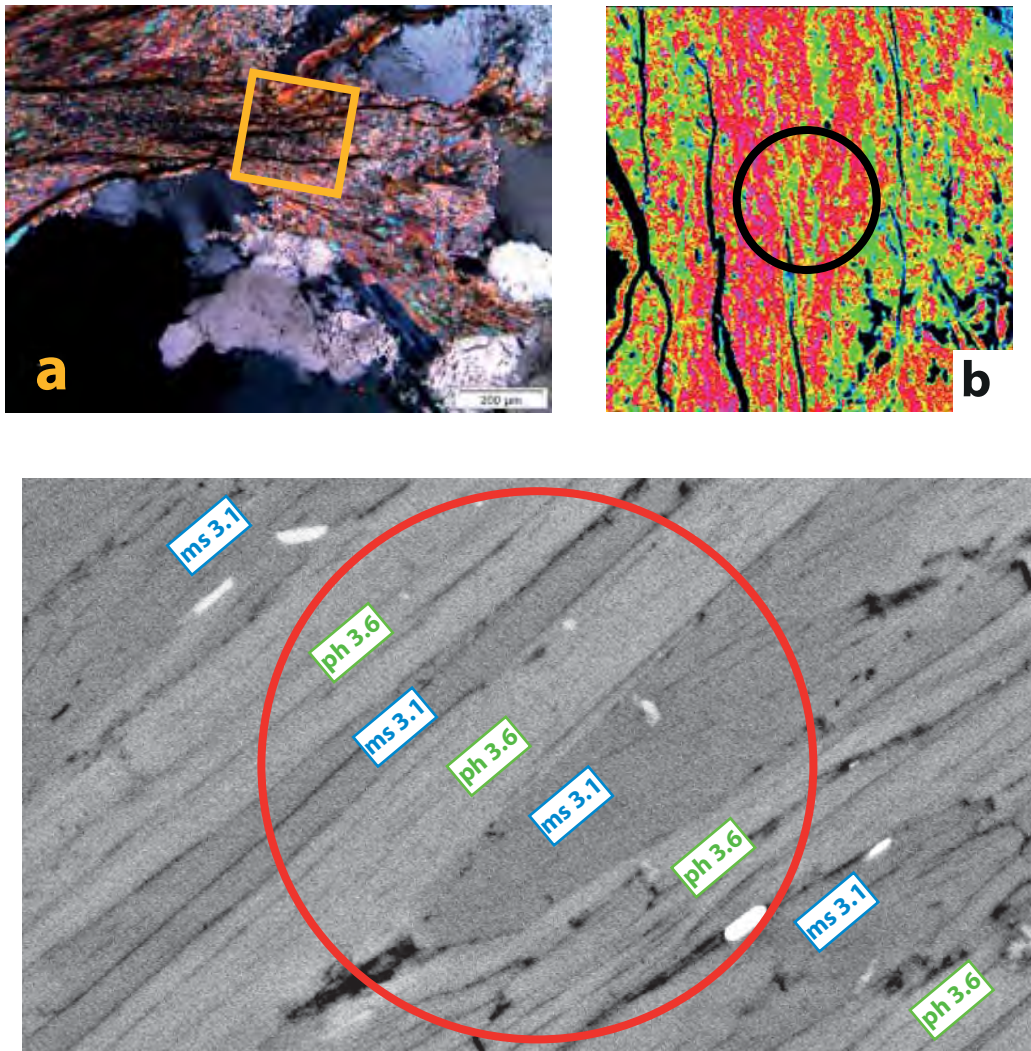


Figure 8

