

1 Revision 2

2 Microscopic strain in a grossular-pyrope solution ***anticorrelates*** with excess volume through  
3 local Mg-Ca cation arrangement, more strongly at high Ca/Mg ratio

4

5 WEI DU<sup>1,2</sup>, DAVID WALKER<sup>2</sup>, SIMON MARTIN CLARK<sup>3,4</sup>, XUEFEI LI<sup>5</sup>, AND  
6 BAOSHENG LI<sup>6</sup>

7

8 <sup>1</sup> School of Earth and Space Sciences, Peking University, Beijing100871, China

9 <sup>2</sup> Lamont-Doherty Earth Observatory, Columbia University in the City of New York, Palisades,  
10 New York 10964, USA

11 <sup>3</sup>Department of Earth and Planetary Sciences, Macquarie University, North Ryde, NSW 2109,  
12 Australia

13 <sup>4</sup>The Bragg Institute, Australian Nuclear Science and Technology Organization, Locked Bag  
14 2001, Kirrawee DC, NSW 2232, Australia

15 <sup>5</sup>Key Laboratory of Functional Materials Physics and Chemistry of the Ministry of Education,  
16 Jilin Normal University, Siping 136000, China

17 <sup>6</sup> Mineral Physics Institute, Department of Geosciences, Stony Brook University, Stony  
18 Brook, New York, 11794, USA

19

## 20 **ABSTRACT**

21 Unit cell volume and microstrain of Py<sub>40</sub>Gr<sub>60</sub> garnets vary with synthesis temperature and  
22 annealing time, showing a strong ***negative*** correlation, as is also seen in another garnet solid  
23 solution, Py<sub>20</sub>Gr<sub>80</sub>. This anticorrelation is explained by local Ca-Mg cation arrangement in which  
24 Ca-Ca and Mg-Mg 3<sup>rd</sup>-nearest-neighbor (Same 3NN = S3NN) pairs form at rates other than those  
25 expected from random Ca-Mg distribution. S3NN pairs cause microstrain (Bosenick et al. 2000)  
26 but allow more efficient packing than random Ca-Mg pairings that contribute to excess volume,  
27 hence smaller cell volumes correlate with more microstrain. Both longer annealing time and  
28 higher heating temperature cause more S3NN formation, larger microscopic strain and smaller  
29 unit cell volume. The anticorrelation of microstrain and excess volume is weaker in our previous

30 study of pyrope-rich solutions (i.e.,  $\text{Py}_{80}\text{Gr}_{20}$ , Du et al. 2016) because excess volume varies little  
31 from Ca-Ca S3NN pairings in pyrope-rich solutions, whereas Mg-Mg S3NN pairings in  
32 grossular-rich solutions studied here are effective at reducing excess volume. Heating to 600 °C  
33 under room pressure or cold hydrostatic compression to 10 GPa does not reset microstrain.

34 Margules' formulations for microstrain and volume as a function of Ca/Mg ratio captures  
35 these features, especially the two-peaked distribution of microstrain with composition discovered  
36 by Du et al. (2016). The similar two-peaked distributions of microstrain and excess energies  
37 derived from ab initio calculation with short range ordering of Mg and Ca cations (Vinograd and  
38 Sluiter 2006) indicate that the macroscopic thermodynamic mixing properties of solid solutions  
39 are directly related to arrangement of cations with large size misfit. The observed changes of  
40 microstrain with annealing temperature suggest that mixing properties measured from our  
41 pyrope-grossular garnet solid solutions synthesized at same temperature can serve as better  
42 experimental constrains for computational work.

43 *Key words: pyrope-grossular garnet solid solution, high temperature, FWHM, microstrain, unit*  
44 *cell volume*

## 45 INTRODUCTION

46 The garnet binary pyrope-grossular ( $\text{Mg}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ - $\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ ) provides an iconic solid-  
47 solution system for the study of non-ideal mixing behavior, because of the large size difference  
48 between divalent  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$  (0.89 vs 1.12 Å in 8 coordination) (Shannon 1976). Enthalpy  
49 (Newton et al. 1977), low-temperature heat capacity (Haselton and Westrum 1980; Dachs and  
50 Geiger 2006), and volume (Ganguly et al. 1993; Bosenick and Geiger 1997; Bosenick et al. 2001;  
51 Du et al. 2016) all show positive deviations from idea mixing, whereas both volume derivatives,  
52 thermal expansion and bulk modulus, show negative deviation from ideal mixing (Du et al.

53 2015). Lattice strain arising from local structural heterogeneities in substitutional pyrope-  
54 grossular solid solutions has been thought to produce elastic energies that strongly affect the  
55 macroscopic enthalpy and volume of mixing (e.g., Geiger 2001, 2008; Dapiaggi et al. 2005).  
56 However, a more recent study by Du et al. (2016) shows that microstrain does not correlate well  
57 with either the positive excess mixing volume or enthalpy. Garnet solid solutions with  
58 composition  $\text{Py}_{40}\text{Gr}_{60}$  show the largest positive deviation from ideal mixing volume, but carry  
59 almost no microstrain, unlike the large microstrains of garnets with compositions  $\text{Py}_{80}\text{Gr}_{20}$  and  
60  $\text{Py}_{20}\text{Gr}_{80}$ . In contrast, there appears to be a good anti-correlation between microstrain along  
61 pyrope-grossular garnet and their thermal expansions (e.g., see Figure 7 in Du et al. 2016),  
62 therefore, more work is needed to understand the effect of temperature on the microscopic strain  
63 and the puzzling anticorrelation between microstrain and unit cell volume.

64 The pyrope-grossular garnet structure has the space group  $\text{Ia}\bar{3}\text{d}$ , which contains  $\text{SiO}_4$   
65 tetrahedra and  $\text{AlO}_6$  octahedra that are connected through corners and edges, building a three-  
66 dimensional quasi-framework with the divalent cations ( $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$ ) located in the  
67 dodecahedral cavities. When the divalent cations ( $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$ ) replace one another, the local  
68 structure will be modified and structural heterogeneities may be produced. The structural  
69 distortion around the “foreign” atom gives rise to strain energy. The total elastic strain in the  
70 pyrope-grossular solid solution thus depends on the ratio of Mg and Ca cations and the  
71 distribution of these two different divalent cations. Du et al. (2016) proposed that  $\text{Py}_{40}\text{Gr}_{60}$  garnet,  
72 which has the smallest microstrain and the largest excess volume among the Ca-Mg  
73 compositions studied, has a larger degree of short range ordering of Mg and Ca cations. Ordering  
74 is used in the sense of Bosenick et al. (2000), taken as the avoidance of same cation Mg-Mg or  
75 Ca-Ca pairs as the third nearest neighbors (S3NN) in the dodecahedral sites during substitution

76 (Figure 1).  $\text{Py}_{40}\text{Gr}_{60}$  with Mg:Ca ratio close to 1:1 has less possibility of randomly generated Mg-  
77 Mg or Ca-Ca 3NN pairings, thus shows an intrinsic degree of short-range ordering (e.g.,  
78 Bosenick et al. 1995, 2000, 2001; Dove et al. 2000; Vinograd et al. 2004; Lavrentiev et al. 2006;  
79 Vinograd and Sluiter 2006). This less “strained” structure may have relatively large excess  
80 volume because Ca-Mg pairs pack less efficiently than strain-causing Mg-Mg and Ca-Ca S3NN  
81 pairs. The short range ordering of Mg/Ca in pyrope-grossular garnet was reported to decrease  
82 with increasing synthesis temperature (Bosenick et al. 1999). Thus, garnets synthesized at higher  
83 temperatures might be expected to exhibit larger microstrain as more strain-causing S3NN pairs  
84 are allowed. Therefore, the anticorrelation between microstrain and the short range ordering of  
85 Mg/Ca will be further supported if larger microstrain can be observed in garnets synthesized at  
86 higher temperatures. Thus, temperature would not relieve microstrain but would enhance it as a  
87 consequence of the anti-intuitive sort of disorder proposed and developed by Bosenick et al.  
88 (2000): the allowance of low-volume, strain causing S3NN pairs in proportion to composition  
89 alone.

90 The aim of the present study is to test the effect of temperature and/or annealing time on  
91 microstrain and excess volume, to explore whether there is a connection between microstrain and  
92 the arrangement of cations Ca and Mg. We chose  $\text{Py}_{40}\text{Gr}_{60}$  glass as starting material because  
93 garnet crystals with this composition show the largest deviation from ideal mixing and almost no  
94 microstrain when synthesized at 6 GPa and 1400 °C. We synthesized garnet crystals of  $\text{Py}_{40}\text{Gr}_{60}$   
95 at 6 GPa within the range 1100-1700 °C with different annealing times (0.5 to 48 hours).  
96 Microstrain and unit cell volume of these garnet crystals were calculated from X-ray diffraction  
97 patterns collected with high resolution synchrotron light sources.

98 Different heating temperatures were also applied to another pyrope-poor garnet,  $\text{Py}_{20}\text{Gr}_{80}$ .

99 The result of a reheating experiment on  $\text{Py}_{20}\text{Gr}_{80}$  is compared with a previously studied pyrope-  
100 rich composition  $\text{Py}_{80}\text{Gr}_{20}$  (Du et al. 2016) in order to examine the possible difference in local  
101 environments of Mg and Ca inside garnet structures of variable composition and their effect on  
102 ordering status. We also examine the microstrains observed during thermal expansion  
103 determinations reported by Du et al. (2015) in reheating experiments (25 to 600 °C) at room  
104 pressure on garnet crystals quenched from 1400 °C and 6 GPa. The calculated microstrain value  
105 from these X-ray diffraction data was used to check whether reheating to modestly high  
106 temperature will change the microstrain and ordering status of cations. X-ray profiles at room  
107 temperature, high pressures were also collected to test whether cold elastic compression resets  
108 the microstrain of  $\text{Py}_{40}\text{Gr}_{60}$ .

109  
110  
111  
112

## EXPERIMENTAL METHODS

### Synthesis of garnet crystals

113 Garnet solid solutions with two different compositions ( $\text{Py}_{40}\text{Gr}_{60}$  and  $\text{Py}_{20}\text{Gr}_{80}$ ) were  
114 synthesized from anhydrous glasses of stoichiometric composition. The same garnet glasses were  
115 also used in Du et al. (2015) and Du et al. (2016), which were prepared by melting a finely  
116 ground mixture of  $\text{CaCO}_3$ , MgO,  $\text{Al}_2\text{O}_3$ , and  $\text{SiO}_2$  powders. The compositions of the glasses  
117 agree well with the starting proportion of oxides. More details about the preparation of garnet  
118 glasses can be found in Du et al. (2015).

119 The garnet crystals studied here were synthesized in a multi anvil (MA) device at the  
120 Lamont Doherty Earth Observatory (LDEO). In order to get pure garnet crystals, a Pt capsule  
121 was used to wrap the finely ground garnet glass, and this Pt capsule was then put into an  $\text{Al}_2\text{O}_3$   
122 tube before packing into the cylindrical cavity of a 3 mm inner diameter  $\text{LaCrO}_3$  heater.

123 Crystalline garnets with composition (Py<sub>40</sub>Gr<sub>60</sub>) were synthesized at 6 GPa, but with different  
124 heating time at 1100, 1200, and 1400 °C. The synthesis or annealing conditions are summarized  
125 in Table 1. Garnet Py<sub>40</sub>Gr<sub>60</sub> synthesized at 1400 °C/6 GPa/0.5 hour (experiment TT721, a =  
126 11.722 ± 0.001 Å) was used as a standard to compare with those from various synthesis  
127 conditions. We tried to lengthen the quenching time in one synthesis experiment GG999 in order  
128 to test the assumption that ordering of Ca-Mg and loss of excess volume happened during  
129 quenching. After heating garnet glass (Py<sub>40</sub>Gr<sub>60</sub>) in MA device at 1400 °C for 0.5 hour, we set up  
130 a program to cool it from 1400 to 1100 °C in 15 minutes, much slower than the quenching time  
131 in experiment TT721, which was quenched by switching off powder after heating at 1400 °C for  
132 0.5 hour, taking only a few seconds to drop temperature from 1400 to ~100 °C after  
133 crystallization and initial annealing at 1400 °C. Lower heating temperatures from 1000 to  
134 1200 °C with heating time varied from 2 to 48 hours were applied to Py<sub>40</sub>Gr<sub>60</sub> garnet glass  
135 (GG1001, GG1002, GG1003, GG1004, BB1007, and GG1005) to check the effect of synthesis  
136 temperature on microstrain. At relatively lower temperature (1100 and 1200 °C), longer  
137 annealing time as long as 24 hours is necessary to produce single phase homogeneous garnet.  
138 Garnet synthesized at 1400 °C for 0.5 hours heating was then put back in MA device and  
139 annealed for much longer time (GG1006) to check effect of the annealing time (48 hours) on  
140 microstrain and unit cell volume.

141 Garnet solid solution Py<sub>20</sub>Gr<sub>80</sub> was synthesized (TT889) using the same batch of glass that  
142 was used to synthesize pure garnet at 6 GPa and 1400 °C (Du et al. 2016). Garnet glass was  
143 heated at 1300 °C and 6 GPa for only 5 minutes and then quenched to room temperature and  
144 pressure. The garnet crystal recovered from this short time synthesis experiment was embedded  
145 within KBr powder and put back into the MA device and pressured to 6 GPa and held at 1400 °C

146 for 100 minutes, and then quenched to room temperature (TT895) (Figure 2). The use of KBr  
147 pressure media was intended to limit mechanical damage to the crystal during recompression and  
148 reheating and thereby limit mechanical contributions to any observed microstrains. The KBr  
149 strategy to limit mechanical strain introduction was also used on Py<sub>40</sub>Gr<sub>60</sub> with the GG1005 and  
150 BB1007 pair of experiments. [See caption to Figure 3.]

### 151 **X-ray Diffraction**

152 X-ray diffraction profiles for these Py<sub>40</sub>Gr<sub>60</sub> garnets were collected at the Brookhaven  
153 National Lab (BNL) on beamline X17C in angular dispersive mode using an MAR345® image  
154 plate detector. The diffraction patterns were collected from 5° to 25° in 2θ angle with  $\lambda =$   
155 0.40722 Å (converted from energy ~30 keV) and the distance between sample and detector  
156 (~290 mm) was determined through collection of a standard LaB<sub>6</sub> diffraction pattern. X-ray  
157 diffraction patterns at high pressure up to 10 GPa were also collected for garnet product Py<sub>40</sub>Gr<sub>60</sub>  
158 from experiment GG999. A four-screw symmetric diamond anvil cell was used for these  
159 measurements. More details about the data collection at high pressures can be found in Du et al.  
160 (2017).

161 Unit cell parameter of garnets Py<sub>20</sub>Gr<sub>80</sub> were measured with X-ray diffraction at Advanced  
162 Light Source at Lawrence Berkeley National Lab (ALS) on station 12.2.2 at room temperature  
163 and pressure, in angular dispersive mode with MAR345® image plates as detectors. Garnet  
164 chunks were loaded in the ~120 μm diameter hole of a 60 μm thick steel gasket with the thrust  
165 axis perpendicular to the incident X-ray beam. The steel gasket was flipped over for the second  
166 measurement after the first X-ray reflection pattern was collected. The two measurements show  
167 perfect consistency with each other. The diffraction images were obtained with  $\lambda = 0.48593$  Å  
168 converted from energy ~25 keV, and the distance between sample and the detector was

169 determined for each run to be ~440 nm through collection of a standard LaB<sub>6</sub> diffraction pattern.  
170 The X-ray diffraction patterns were collected from 5° to 30° in 2θ.

171 FIT2D software package was used to integrate the two dimensional diffraction rings into  
172 one-dimensional diffraction patterns (Hammersley et al. 1996). XFIT (Cheary and Coelho 1996)  
173 and REFCEL (Cockcroft and Barnes 1997) were used to perform profile fitting analysis, which  
174 determines the unit cell parameters from least squares analysis of the positions of the X-ray  
175 diffraction peaks.

## 176 DATA ANALYSIS AND RESULTS

### 177 (1) Unit cell volume

178 The unit cell volumes of Py<sub>40</sub>Gr<sub>60</sub> garnets synthesized at different temperatures with  
179 different heating time are summarized in Table 1 and Figure 3a. The refined unit cell parameter  
180 of Py<sub>40</sub>Gr<sub>60</sub> from slow quenching experiment GG999 is  $11.731 \pm 0.001 \text{ \AA}$ , which is larger than  
181 that from rapidly quenched experiment TT721 ( $a = 11.721 \pm 0.001 \text{ \AA}$ ). Garnet crystallized from  
182 the same glass after heating at 1100 °C for 48 hours (GG1004) shows a similarly large unit cell  
183 parameter ( $a = 11.732 \pm 0.001 \text{ \AA}$ ) as GG999, indicating that the relatively lower annealing  
184 temperature (1100 °C) could be the principle reason for the larger unit cell volume. The unit cell  
185 parameter of the garnet crystals measured at room temperature is an indication of the temperature  
186 from which it has quenched suddenly, rather than the cooling rate of the sample to that  
187 quenching temperature. The experiment BB1007 was carried at the same P-T-t condition as  
188 GG1005 (1200 °C for 48 hours), and the unit cell parameter of garnet from BB1007 agrees well  
189 with experiment GG1005, indicating the reproducibility of the current synthesis experiments.  
190 The use of KBr in one experiment to limit mechanical strains, and not the other experiment, yet  
191 achieving the same result, further suggests that mechanical strains are not contributing to the



192 microstrain. Garnet synthesized at 1200 °C shows smaller unit cell parameter ( $a = 11.725 \pm$   
193  $0.001 \text{ \AA}$ ) than those heated at 1100°C ( $a = 11.732 \pm 0.001 \text{ \AA}$ ), indicating that garnets  $\text{Py}_{40}\text{Gr}_{60}$   
194 synthesized at higher temperatures show smaller unit cell volume. XRD measurement result of  
195  $\text{Py}_{40}\text{Gr}_{60}$  garnet from experiment GG1000 gives a unit cell parameter  $a = 11.705 \pm 0.001 \text{ \AA}$ ,  
196 much smaller than the “standard value”  $11.722 \text{ \AA}$  (TT721). The reason for the termination of this  
197 longer time annealing experiment is that temperature control failed and experimental temperature  
198 shot too high during the long-term heating (Table 1). Such loss of temperature control to the high  
199 side was proved by the melting of the Pt capsule used to isolate garnet glass from potential  
200 contamination by  $\text{LaCrO}_3$  heater. Melting temperature of Pt at 6 GPa is about 1950 °C, so we  
201 believe there was a temperature excursion to as high as 1950 °C. The corresponding points for  
202 this experiment in Figure 3 are plotted at 1700 °C with a right-directed horizontal arrow to  
203 indicate that this is probably a minimum. Although we don’t know the exact temperature of  
204 experiment GG1000, it still implies that garnet that was heated at higher temperature has a  
205 smaller unit cell volume.

206       However, experiment GG1006 ( $\text{Py}_{40}\text{Gr}_{60}$ ), which reheated garnet crystals from experiment  
207 TT721 ( $a = 11.722 \pm 0.001 \text{ \AA}$ ) as starting material, at 1400 °C and 6 GPa for much longer time  
208 (48 hours) showed relatively smaller unit cell parameter ( $a = 11.715 \pm 0.001 \text{ \AA}$ ). We did the  
209 same test on garnet  $\text{Py}_{20}\text{Gr}_{80}$ , results of which are presented in Table 2. The longer annealing  
210 experiment TT895 ( $\text{Py}_{20}\text{Gr}_{80}$ ) used product of experiment TT889 (1300 °C for 5 minutes) as  
211 starting material and heated at higher temperature 1400 °C for longer time (100 minutes). And  
212 we observed a slight shrinking of unit cell parameter from  $11.807 \pm 0.001 \text{ \AA}$  to  $11.789 \pm 0.001 \text{ \AA}$   
213 after second round annealing, showing consistent unit cell parameter as another experiment at  
214 1400 °C (TT649). Therefore, longer annealing time at 1400 °C also reduces the unit cell volume

215 slightly. Figure 3a demonstrates that there is a systematic decrease in cell volume with synthesis  
216 temperature of samples that are sufficiently annealed. At 1400 °C, a 0.5 hour anneal is not  
217 sufficient to reach the systematic trend and the descending arrow shows the evolution of cell  
218 parameter with increasing annealing time at 1400 °C to the systematic trend. The experiment  
219 GG1000 listed as a minimum of 1700 °C reached the trend in a few seconds. As expected, the  
220 required annealing time must be strongly temperature sensitive.

221

## 222 (2) Microstrain

223 The method described in Du et al. (2016) was used to calculate microstrain from the XRD  
224 peaking broadening parameter  $\eta$ . The FWHM (full-width at half maximum) of LaB<sub>6</sub>, which  
225 does not change with  $2\theta$  angle up to 30°, was adopted as the instrumental broadening for  
226 synchrotron XRD file,  $B_{\text{inst}} \sim B_{\text{LaB}_6}$ . The XRD peak broadening of garnet sample,  $B_{\text{garnet}}$ , is  
227 then determined by equation (1):

$$228 \quad B_{\text{garnet}} \approx \sqrt{B_{\text{obs}}^2 - B_{\text{inst}}^2} \approx B_{\text{obs}} - B_{\text{LaB}_6} \quad (1)$$

229 where  $B_{\text{obs}}$  is the FWHM of X-ray diffraction peaks fitted with pseudo-Voigt equations  
230 (Langford 1978). The microstrain that contributes to the XRD peak broadening was calculated  
231 through Williamson-Hall plots (Williamson and Hall 1953):

$$232 \quad B_{\text{garnet}} \cos \theta = \frac{K\lambda}{\langle L \rangle \cos \theta} \cos \theta + 4\eta \tan \theta \cos \theta = \frac{K\lambda}{\langle L \rangle} + 4\eta \sin \theta \quad (2)$$

233 where  $\lambda$  is the wavelength of the X-ray, and  $\theta$  is the Bragg angle (in radians),  $K$  is a constant of  
234 value approximately 0.9 (Langford and Wilson 1978),  $L$  is the crystal size (diffraction domain

235 size),  $\frac{K\lambda}{\langle L \rangle \cos\theta}$  is the size effect on XRD peak broadening (Scherrer 1918),  $4\eta \tan\theta$  is the width of  
236 the peak due to microstrain (Stokes and Wilson 1944), and  $\eta$  is the microstrain.

237 In order to minimize any systematic errors caused by different X-ray sources and to use the  
238 same  $\text{LaB}_6$  pattern as instrumental contribution on XRD peak broadening, it is essential that all  
239 the XRD patterns were collected under the same instrumental condition. Therefore, we collected  
240 X-ray diffraction data on the same sample TT721 that was studied in Du et al. (2016) as  
241 reference. The microstrain data of garnets  $\text{Py}_{40}\text{Gr}_{60}$  and  $\text{Py}_{20}\text{Gr}_{80}$  were summarized in Table 1  
242 and Table 2. For  $\text{Py}_{40}\text{Gr}_{60}$  garnets with the same annealing time 48 hours (GG1004, GG1005, and  
243 GG1006), higher temperature 1400 °C caused more microstrain inside garnet structure (Figure 3),  
244 which is consistent with the reheating experiment on  $\text{Py}_{80}\text{Gr}_{20}$  (Du et al. 2016), indicating that  
245 higher temperature annealing causes larger microstrain. Reheating experiments on  $\text{Py}_{40}\text{Gr}_{60}$  show  
246 similar trends: that garnet annealed at higher temperature ( $> 1700$  °C, GG1000) shows larger  
247 microstrain than that annealed at relatively lower temperature (1100 °C, GG999) (Figure 4).  
248 Reheating experiment on  $\text{Py}_{40}\text{Gr}_{60}$  (GG1006), which used garnet crystals from experiment  
249 TT721 as starting material, at 1400 °C and 6 GPa for much longer time (48 hours) showed  
250 relatively larger microstrain 0.048% than that of TT721 (0.038%). Therefore, longer annealing  
251 time at 1400 °C also increases the microstrain slightly. Figure 3b shows a systematic relation  
252 increasing microstrain with temperature for sufficiently annealed samples, analogous to the cell  
253 size relation, but with opposite slope with temperature.

254

### 255 (3) Thermal expansion

256 X-ray diffraction patterns that were collected at room pressure to 600 °C by Du et al. (2015)  
257 from pyrope-grossular solid solutions powders were analyzed here by Williamson-Hall plots to

258 get the microstrain data at high temperatures. The results are present in Figure 5. Upon  
259 increasing temperature to 600 °C, the microstrain value of each garnet solid solution does not  
260 change with temperature. Evidently the microstrain does not reset with modest reheating. The  
261 two-peaked compositional dependence of microstrain along the pyrope-grossular garnet binary  
262 reported by Du et al. (2016) is demonstrated to hold at least up to 600 °C. There remain two  
263 distinct peaks in the microstrains of thermally expanded garnets, one at about Py<sub>80</sub>Gr<sub>20</sub> and the  
264 other near Py<sub>20</sub>Gr<sub>80</sub>.

265

#### 266 (4) Bulk modulus

267 In order to look for any effect of microstrain on compressional elastic properties, Birch-  
268 Murnaghan equation of state analysis was applied to the V-P data obtained for the specimen  
269 from experiment GG999 with the smallest observed microstrain (Table 3 and Figure 6). In the  
270 case of isothermal (room temperature) hydrostatic compression, the pressure can be written as:

$$271 \quad P_V = 3K_0 f(1 + 2f)^{\frac{5}{2}}(1 + \frac{3}{2}(K_0' - 4)f) \quad (3)$$

272 where  $f$  is the Eulerian finite strain (with sign reversed so  $f$  is positive for compression),  $f =$

273  $\frac{1}{2} \left[ \left( \frac{V_P}{V_0} \right)^{-\frac{2}{3}} - 1 \right]$ ,  $V_0$  is the unit cell volume at ambient condition,  $K_0$  is the isothermal bulk

274 modulus at zero pressure, and  $K_0'$  is the first derivative of  $K_0$  versus pressure. As discussed in Du  
275 et al. (2016), there is no constraint on  $K_0'$  of garnet with intermediate composition along pyrope-  
276 grossular garnet binary. Therefore, we tried fixed  $K_0'$  as 4.4 (Zhang et al. 1998) and 5.92 (Pavese  
277 et al. 2001), and calculated  $K_0$  for Py<sub>40</sub>Gr<sub>60</sub> by fitting the measured P-V data (Table 4 and Figure  
278 6) to the Birch-Murnaghan (BM) Equation of State. For  $K_0' = 5.92$ ,  $K_0 = 153.5 \pm 1.0$  GPa;  
279 similarly,  $K_0 = 158.6 \pm 1.1$  GPa for fixed  $K_0' = 4.4$ . Figure 7 shows the “normalized pressure”  $f_E$

280 ( $f_E = P/(3f(2f + 1)^{5/2})$ ) versus the volume Eulerian finite strain  $f$  plot (Angel 2000). The  
281 weighted linear regression through the data points yields the intercept value,  $f_E(0) = 154.7 \pm 1.0$   
282 GPa, which agrees with the result yielded from Birch-Murnaghan Equation of State. Thus, fitting  
283 these data to the third-order Birch-Murnaghan EoS yields a marginally smaller bulk modulus for  
284 GG999 compared with TT721 of  $159.3 \pm 1.0$  GPa (average from Du et al. 2015). This lower  
285 value for GG999 is consistent with the general thought that crystals with larger excess volume  
286 are more compressible, showing a lower value for the bulk modulus.

287 In order to look for any effect of cold elastic compression on microstrain, values for  
288 microstrain at stages in the cold compression of GG999 are recorded in Table 3. There is no clear  
289 trend to increasing or decreasing microstrain with compression. It should be noted that the  
290 uncertainty in the microstrains reported in Table 3 are considerably greater than those reported in  
291 Tables 1 and 2 because of the fewer number of peaks collected at a more restricted range of  $2\theta$   
292 imposed by the diamond anvil pressure cell. There is no resolvable effect of cold compression to  
293 10 GPa on  $\text{Py}_{40}\text{Gr}_{60}$  microstrain.

## 294 DISCUSSION

295 The substitution of Ca or Mg as “foreign” cations in pyrope or grossular garnets produces  
296 local distortions in the host structure because of the large size difference between Mg and Ca  
297 cations (Newton and Wood 1980; Bosenick et al. 2000). The importance of S3NN pairing for  
298 creating strains was demonstrated by Bosenick et al. (2000), but it was not clear from their lattice  
299 energy calculations that a two-peaked distribution of strain with garnet composition could result.  
300 Indeed, Dapiaggi et al. (2005) reported a single, asymmetric peak in strain for pyrope-grossular  
301 garnets synthesized with hydrothermal assistance. By contrast, Du et al. (2016) observed a two-  
302 peaked distribution with composition of microstrain in dry garnets all made at 1400 °C, 6 GPa,

303 in 0.5 hour that could be explained by the local arrangement of Mg/Ca. Our high resolution  
304 synchrotron X-ray diffraction studies on newly synthesized garnets  $\text{Py}_{40}\text{Gr}_{60}$  in this study further  
305 show that both microstrain and unit cell volume change with synthesis and annealing  
306 temperature and time. The relationship between unit cell volume and microstrain is a strong  
307 anticorrelation for grossular-rich garnet solid solution (Figure 8). Higher heating temperature is  
308 observed to produce smaller unit cell volumes. Lower synthesis temperature is consistent with  
309 the relatively smaller microstrain in GG1004, which has a larger unit cell volume. Comparing the  
310 specimens annealed for 48 hours, those annealed at higher temperature show larger microstrain  
311 and relatively smaller unit cell volume (Table 2 and Figure 8). For experiments that use garnet  
312  $\text{Py}_{40}\text{Gr}_{60}$  that was synthesized at 1400 °C and annealed for 0.5 hours as starting material, longer  
313 time annealing at 1400 °C (GG1006) caused larger microstrain. And longer quenching time to  
314 lower temperature (1100 °C) (GG999) partially releases microstrain (Figure 8). If as Du et al.  
315 (2016) proposed, the microstrain data is correlated with the divalent cation (Mg and Ca)  
316 arrangement in the pyrope-grossular garnet solid solutions, the result that we present in this study  
317 implies that the longer annealing time and/or higher temperature processing causes larger  
318 microstrain by forming the less energetic favored Mg-Mg and Ca-Ca pairs in garnet structures  
319 (Bosenick et al. 2000). The smaller unit cell volumes observed for the more strained structures  
320 are a consequence of Mg-Mg and Ca-Ca S3NN pairs packing more efficiently than Ca-Mg 3NN  
321 pairs. Thus the microstrain – volume anticorrelation,

322       However, we observed systematically weaker anticorrelation between microstrain and unit  
323 cell volume as garnet solutions become more pyrope-rich, unlike for grossular-rich garnets just  
324 reported where the anticorrelation is strong. When garnets  $\text{Py}_{80}\text{Gr}_{20}$  (TT890) and  $\text{Py}_{20}\text{Gr}_{80}$   
325 (TT889), that were both synthesized at 1300 °C, were later annealed at 1400 °C (TT895), each

326 gained more microstrain (Table 2 and Figures 8 and 9a). The  $\text{Py}_{20}\text{Gr}_{80}$  (TT889 to TT895a) lost  
327 volume as expected for grossular-rich garnets (Figure 9b). However,  $\text{Py}_{80}\text{Gr}_{20}$ , whose XRD  
328 peaks also became broader after annealing at higher temperature 1400 °C, did so with less  
329 significant changes in unit cell volume (Du et al. 2016). Therefore, the microstrain of S3NN Ca-  
330 Ca pair formation changes the unit cell volume of pyrope-rich garnet ( $\text{Py}_{80}\text{Gr}_{20}$ ) less than Mg-Mg  
331 pairs reduce the unit cells of grossular-rich garnets. This observation could be explained by the  
332 different local environment of Mg and Ca cations between garnets with composition close to the  
333 two different end members pyrope and grossular as proposed by Oberti et al. (2006) and  
334 Quartieri et al. (2008). The local environments around Ca and Mg cations inside pyrope-  
335 grossular garnet are different for different compositions. Formation of Ca-Ca pairs in Mg-rich  
336 environments has less impact on expanding the volume than Mg-Mg pairs' effect on lattice  
337 collapse in Ca-rich environments.

338         This argument is foreshadowed by the 'forbidden region' argument Newton and Wood  
339 (1980) used to explain S-shaped volume curves with negative excess volumes near the small-  
340 end-membered end of a solid solution series and positive excesses near the larger end member.  
341 Large cations in a small cation matrix are inserted with local lattice stretches that do not cause  
342 global lattice expansion until the large cations interact. The local stretch creates an energetically  
343 forbidden environment where another local large cation would be unfavorably sited. The  
344 proposition in Newton and Wood (1980) was that such strains would be encountered between  
345 S1NN pairs. Bosenick et al. (2000) showed that the S3NN pairs were a much more likely pairing  
346 from a lattice energetics perspective to cause local strains. If one shifts perspective from S1NN  
347 to S3NN pairs, the framing of the problem and arguments of Newton and Wood (1980) continue  
348 to make physical sense and to form the basis of an avoidance model. This preferential avoidance

349 (a form of cation ordering) keeps the large cations from interacting globally over the whole  
350 lattice in such a way that lattice expansion is not as much as expected for the proportion of large  
351 cation substituents. Likewise, when small cations enter a large cation matrix, their ability to  
352 cause global lattice shrinkage is diminished, except for the fact that it is more difficult to resist a  
353 collapse against no sustaining outward force than to resist a stretch against strong constraining  
354 forces. Thus the negative deviations from ideality at the small end-member are expected to be  
355 greater than the positive deviations at the large end member of the series. Newton and Wood  
356 (1980) showed a number of mineralogical solutions that give examples of this behavior. The  
357 application to our garnets is that the S3NN pairs behave differently when it is Mg-Mg in a  
358 grossular-rich matrix compared to Ca-Ca in a pyrope-rich matrix. By the Newton and Wood  
359 (1980) arguments we expect volume variations driven by cation avoidances to be more  
360 suppressed at the small-end-member pyrope-rich end of the series.

361 Do microstrain and related excess volume changes cause significant changes in minerals'  
362 physical properties, for example, elastic moduli (Hazen and Navrotsky 1996)? The unit cell  
363 volume of Py<sub>40</sub>Gr<sub>60</sub> garnet GG999 shows compression similar to that from TT721 (Figure 6),  
364 although the former shows relatively smaller microstrain and larger unit cell volume. Fitting  
365 these data to the third-order Birch-Murnaghan EoS yields a marginally smaller bulk modulus  
366 (~154 GPa) for GG999 comparing with TT721 (~158 GPa) (Du et al. 2015). Considering the  
367 fact that the microstrain calculated from FWHM of X-ray pattern does not change with pressure  
368 up to 10 GPa (Du et al. 2016), which means that microstrain does not change during compression,  
369 the smaller bulk modulus of GG999 compared with that of TT721 may be related to the  
370 relatively larger unit cell volume. This suggestion is supported by the observation that garnet  
371 solid solutions with composition closer to the end members, for example Py<sub>20</sub>Gr<sub>80</sub> and Py<sub>80</sub>Gr<sub>20</sub>



372 which show larger microstrain and relatively smaller excess volume, have larger bulk moduli  
373 (Du et al. 2015). Our new data for  $\text{Py}_{40}\text{Gr}_{60}$  is consistent with previous study that pyrope-  
374 grossular garnet solid solution with composition close to 50:50 shows smaller bulk modulus, thus  
375 larger compressibility than the two end members. However, as mentioned above, the parameters  
376  $K_0$  and  $K_0'$  are usually strongly correlated in equation of state fitting (e.g. Nishihara et al. 2003),  
377 and the substitution between Ca and Mg in the dodecahedral X site may affect the  $K_0'$  (Conrad et  
378 al. 1999) but cannot be resolved by the current data.

379 Why does increasing annealing time favor more strain? It is unsettling to have the  
380 evolution with annealing time be to a state of higher microstrain. The usual expectation is that  
381 annealing relieves strains. However, the microstrains observed in garnet are unusual in that they  
382 provide better packing through the formation of S3NN pairs. Thus the volume reduction provides  
383 the incentive at high pressure to increase strain. One might expect that higher temperature should  
384 disorder a system and thus relieve strain. However, increasing the number of S3NN pairs is not  
385 disordering, because it may also increase the departures from random Ca-Mg arrangement. Thus  
386 the clear anticorrelation of  $V_{\text{ex}}$  and microstrain across all the temperatures studied to both higher  
387 and lower values than the disordered glass starting material, suggests that an order/disorder  
388 framework is not a productive perspective here. 'Ordering' increases towards both ends of the T-  
389  $V_{\text{ex}}$ -microstrain spectrum and cannot monotonically drive the evolution of properties we observe.

390 Given that we deduce that neither 'ordering' nor local lattice energetics is a completely  
391 satisfactory explanation for our anticorrelation, or for the two-peaked distribution of microstrain  
392 with composition, we attempt to describe the system as one in which Ca-Mg arrangements are  
393 basically set by bulk composition, with minor strain perturbations that add asymmetry to the  
394 response with composition. The Margules form of mixing properties has as its basis that

395 substitutional elements will interact in proportion to their probability of being paired based on  
396 the composition of the mix. If the placement is random, then the symmetric form of Margules  
397 mixing results. If the pairs form with unbalanced interaction parameters, then the asymmetric  
398 form results. The strains in the garnet solution come from the misfits of Ca into an Mg matrix (or  
399 Mg into a Ca matrix). Ca-Mg pairs intrinsically have excess volume compared to SNN pairs. We  
400 can pass over the issue of whether we do the bookkeeping on S1NN or S3NN pairs as the  
401 population probabilities are the same. The probability of forming Ca-Mg pairs in a mixed garnet  
402 is  $2X_{Py}X_{Gr}$ . The probability is zero at either end member and  $\frac{1}{2}$  at the equimolar intermediate  
403 composition. The remaining half is  $\frac{1}{4}$  each of Mg-Mg and Ca-Ca pairs. The excess volume  
404 strains should then maximize at the Ca-Mg pair maximum in the middle of the series. However,  
405 the XRD microstrains respond to the formation of Ca-Ca strain-producing pairs in an Mg-Mg  
406 matrix (and vice versa). The fraction of these pairs rise as  $X_{Gr}X_{Gr}$  near pyrope (or  $X_{Py}X_{Py}$  near  
407 grossular). This increase in Ca-Ca microstrain is modified away from the end member pyrope by  
408 the erosion of the matrix of Mg-Mg pairs at rate  $X_{Py}X_{Py}$  through the formation of Ca-Mg and Ca-  
409 Ca pairs that do not contribute to the XRD strain contrast between the Ca-Ca pairs and the Mg-  
410 Mg matrix. [Effectively there is a discount rate for the growth of microstrain.] Thus the variation  
411 of microstrain may be expected to go as the difference of the squares of  $X_{Py}$  and  $X_{Gr}$ . This form  
412 of microstrain can lead to a two-peaked maximum in microstrain across the pyrope-grossular  
413 series, unlike the Dapiaggi data or the Bosenick energetics.

414 We propose that the excess volumes with their volume strains form a basis for modeling  
415 the form of microstrains strains that arise through Ca-Mg, Ca-Ca, or Mg-Mg pair misfits. When  
416 asymmetry for unequal Mg-Mg vs. Ca-Ca interactions is allowed, the  $X_{Py}X_{Gr}$  ( $X_{Py}W_{\eta_{Gr}} + X_{Gr}W_{\eta_{Py}}$ )  
417 form gives this basic driver for strain with Margules parameters  $W_{\eta_{Py}}$  and  $W_{\eta_{Gr}}$  reflecting the

418 possibly unequal excess strains associated with forming Ca substitutions into pyrope and Mg  
419 substitutions into grossular. This is, of course, the same form of the Margules volume mixing  
420 equation in which  $V_{ex} = X_{Py}X_{Gr}(X_{Py}W_{V_{Gr}} + X_{Gr}W_{V_{Py}})$ , the  $W_V$  parameters giving the volume  
421 interactions rather than the  $W_\eta$  for strain. The basic Margules form for strain must be discounted  
422 as Ca-Mg random pairs begin to accumulate in the structure and disrupt the distortional and  
423 torsional strains introduced by Ca-Ca and Mg-Mg S3NN pairings. [They do not disrupt the  
424 accumulating volume strain of Ca-Mg misfit.] The strain disruption discount depends on the  
425 difference between Ca-Ca and Mg-Mg pairings, for which we propose  $(|X_{Gr}^2 - X_{Py}^2| - W_0)$ ,  
426 giving:

$$427 \quad \text{Strain} = \eta = X_{Py}X_{Gr}(X_{Py}W_{\eta_{Gr}} + X_{Gr}W_{\eta_{Py}})(|X_{Gr}^2 - X_{Py}^2| - W_0) \quad (4)$$

428  $W_0$  is an adjustable scaling parameter that allows the possibility that strain not go strictly to  
429 zero when the number of Mg-Mg and Ca-Ca pairs are equal. Figure 10 shows the observed  
430 microstrains fit to the Margules-inspired form given above. The Margules form captures the two-  
431 peaked distribution of microstrain with composition in this solution series prepared at 6 GPa,  
432 1400 °C. A small  $W_0 = 0.07 \pm 0.02$  is indicated, although eliminating  $W_0$  degrades the fit only  
433 slightly.  $W_{\eta_{Py}}$  ( $1.44 \pm 0.36$ ) and  $W_{\eta_{Gr}}$  ( $1.19 \pm 0.02$ ) are comparable in magnitude and have a ratio  
434 that is within a few percent of the inverse ratio of the cation radii ( $1.12/0.89$ ). The  $W_\eta$  Margules  
435 parameters for strain have an analogous physical meaning to the  $W_V$  parameters for excess  
436 volume. If the tangent to the excess strain curve at  $X_{Gr} = 1$  is extrapolated to  $X_{Gr} = 0$  (i.e. to  $X_{Py} =$   
437 1), the intercept has the value  $W_{\eta_{Py}}$ . Likewise  $W_{\eta_{Gr}}$  is the intercept at  $X_{Gr} = 1$  of the extrapolated  
438 tangent to the excess strain curve at  $X_{Gr} = 0$ . The one-peaked asymmetric volume strain pattern  
439 seen in Figure 9b drives both volume and local distortional strain by Mg-Ca size misfit. This

440 driver competes with the strain-reducing disruption of a dropping population of Mg-Mg and Ca-  
441 Ca pairs as bulk compositions approach 50:50, resulting in the strains being maximized in  
442 separate domains of composition space near 20% of either end member.

443 This two-peaked strain pattern does not resemble that of Dapiaggi et al. (2005) measured  
444 on garnets prepared differently with hydrothermal assistance. However, the data of Dapiaggi et  
445 al. (2005) can also be fit with this fairly flexible form but would require grossly asymmetric  
446  $W_{\eta_{Py}}$  and  $W_{\eta_{Gr}}$  to accommodate the single peak towards pyrope-rich compositions. Such gross  
447 asymmetry would contradict the prescription of Bosenick et al. (2000) that strain effects should  
448 be symmetric. Symmetric does not preclude two peaks in strain, but this was not a conspicuous  
449 conclusion of the Bosenick et al. (2000) study. Our Margules form can be compatible with the  
450 Bosenick result with large negative  $W_0$ . These differences among theory and two experimental  
451 studies are important to resolve to understand the controls on intracrystalline strains. The  
452 experimental differences may reflect differences in garnet preparation techniques between these  
453 studies, or differences in data reliability. We have reproduced our data on powder samples and  
454 chunks with two synchrotrons and one lab source data collection platforms with acceptable  
455 success, and at least the Margules model provides a rationale for understanding the two-peaked  
456 strain distribution which is only seen by us.

457

458

## IMPLICATION

459 Does the protocol for sample preparation make any difference to the understanding of  
460 solution behavior in mineral solutions? The volume-microstrain relations of dry pyrope-grossular  
461 garnets grown at 6 GPa by Du et al. (2016) are quite different from those reported by Dapiaggi et  
462 al. (2005) grown with hydrothermal assist at lower pressures. Our strains anticorrelate with

463 excess volume and are bimodally peaked in composition. We have been able to reproduce our  
464 bimodal strain-volume-composition relations on different sample sets on two different diffraction  
465 platforms. The strains in hydrothermally assisted garnet solutions are not bimodal, and have a  
466 variation with composition that mimics the excess enthalpy of solution of hydrothermally  
467 assisted garnets of Newton et al. (1977). Our bimodal strains on dry garnets do not correlate with  
468 the excess enthalpy of hydrothermally assisted garnets, but resemble the mixing entropy  
469 calculated for different states of order of the Mg and Ca cations (Vinograd and Sluiter 2006). So  
470 the implied answer to the question is yes.

471 Our new observation of the anticorrelation between microstrain and unit cell volume  
472 indicates that the arrangement of cations with size misfit affects the thermodynamic mixing  
473 properties of a solid solution and the degree of this effect changes with annealing temperature  
474 and hydrothermal assistant. Garnet solid solutions that were synthesized at same pressure and  
475 temperature conditions should serve as better experimental constrains for modern quantitative  
476 computational work. Mixing properties of pyrope-grossular garnet solid solutions that were used  
477 for geothermal barometers need more work to update.

478

479

## ACKNOWLEDGMENTS

480 The authors thank Associate Editor Dr. Haozhe Liu and two anonymous reviewers for their  
481 perceptive reviews. We thank Dr. Bernard J. Wood for his thoughtful review during preparation  
482 of this work. This work was supported by the U.S. National Science Foundation. The Advanced  
483 Light Source is supported by the Director, Office of Science, Office of Basic Energy Sciences, of  
484 the U.S. Department of Energy under Contract No. DE-AC02-05CH11231 at Lawrence Berkeley  
485 National Laboratory. This work was also supported by DOE/NNSA Grant No. DE-NA0001815  
486 and NSF Grant No. EAR1045630 to B. Li. We thank Jean Hanley, Jinyuan Yan, and Zhiqiang  
487 Chen for their technical assistance.

488

489

490

## REFERENCES CITED

- 491 Angel, R.J. (2000) Equations of state. In: Hazen, R.M. and Downs, R.T., Editors, High-pressure  
492 and high-temperature crystal chemistry. Review in Mineralogy and Geochemistry, 41, 35–60.
- 493 Bosenick, A. and Geiger, C.A. (1997) Powder X-ray diffraction study of synthetic pyrope-  
494 grossular garnets between 20 and 295 K. Journal of Geophysical Research, 102, 22649–  
495 22657.
- 496 Bosenick, A., Dove, M.T., and Geiger, C.A. (2000) Simulation studies on the pyrope-grossular  
497 garnet solid solution. Physics and Chemistry of Minerals, 27, 398–418.
- 498 Bosenick, A., Dove, M.T., Heine, V., and Geiger, C.A. (2001) Scaling of thermodynamic mixing  
499 properties in garnet solid solutions. Physics and Chemistry of Minerals, 28, 177–187.
- 500 Bosenick, A., Geiger, C.A., Schaller, T., and Sebald, A. (1995) A  $^{29}\text{Si}$  MAS NMR and IR  
501 spectroscopic investigation of synthetic pyrope-grossular garnet solid solutions. American  
502 Mineralogist, 80, 691–704.
- 503 Cheary, R.W. and Coelho, A.A. (1996) Programs XFIT and FOURYA, deposited in CCP14  
504 powder diffraction library. Engineering and Physical Sciences Research Council, Daresbury  
505 Laboratory, Warrington, England.
- 506 Conrad, P.G., Zha, C.S., Mao, H.K., and Hemley, R.J. (1999) The high-pressure, single-crystal  
507 elasticity of pyrope, grossular, and andradite. American Mineralogist, 84, 374–383.
- 508 Dachs, E. and Geiger, C.A. (2006) Heat capacities and entropies of mixing of pyrope-grossular  
509 ( $\text{Mg}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ - $\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ ) garnet solid solutions: A low-temperature calorimetric and a  
510 thermodynamic investigation. American Mineralogist, 91, 894–906.

- 511 Dapiaggi, M., Geiger, C.A., and Artioli, G. (2005) Microscopic strain in synthetic pyrope-  
512 grossular solid solutions determined by synchrotron X-ray powder diffraction at 5 K: The  
513 relationship to enthalpy of mixing behavior. *American Mineralogist*, 90, 506–509.
- 514 Dove, M.T., Bosenick, A., Myers, E.R., Warren, M.C., and Redfer, S.A.T. (2000) Modeling in  
515 relation to cation ordering. *Phase Transitions*, 71, 205–226.
- 516 Du, W., Clark, S.M., and Walker, D. (2015) Thermo-compression of pyrope-grossular garnet  
517 solid solution: non-linear compositional dependence. *American Mineralogist*, 100, 215–222.
- 518 Du, W., Clark, S.M., and Walker, D. (2016) Excess mixing volume, microstrain, and stability of  
519 pyrope-grossular garnets. *American Mineralogist*, 101, 193–204.
- 520 Du, W., Li, X., and Li, B. (2017) Microstrain in pyrope-grossular garnet solid solution at high  
521 pressure: a case study of  $\text{Py}_{90}\text{Gr}_{10}$  and  $\text{Py}_{10}\text{Gr}_{90}$  up to 15 GPa. *Physics and Chemistry of*  
522 *Minerals*, 44, 377–388.
- 523 Freeman, C.L., Allan, N.L., and van Westrenen, W. (2006) Local cation environments in the  
524 pyrope-grossular  $\text{Mg}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ - $\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$  garnet solid solution. *Physical Review B*, 74,  
525 134203.
- 526 Ganguly, J., Cheng, W., and O'Neill, H.C. (1993) Syntheses, volume, and structural changes of  
527 garnets in the pyrope-grossular join; implications for stability and mixing properties.  
528 *American Mineralogist*, 78, 583–593.
- 529 Geiger, C.A. (2001) Thermodynamic mixing properties of binary oxide and silicate solid  
530 solutions determined by direct measurements: The role of strain. In Geiger, C.A. Ed, *Solid*  
531 *Solutions in Silicate and Oxide Systems*, 3, 71–100. European Notes in Mineralogy, Eötvös  
532 University Press, Budapest.

- 533 Geiger, C.A. (2008) Silicate garnet: A micro to macroscopic (re)view. *American Mineralogist*,  
534 93, 360–372.
- 535 Hammersley A.P., Svensson S.O., Hanfland M., Fitch A.N., and Häusermann D. (1996) Two-  
536 dimensional detector software: from real detector to idealized image or two-theta scan. *High*  
537 *Pressure Research*, 14, 235–248.
- 538 Haselton Jr., H.T. and Westrum Jr., E.F. (1980) Low-temperature heat capacities of synthetic  
539 pyrope, grossular, and pyrope<sub>60</sub>grossular<sub>40</sub>. *Geochimica et Cosmochimica Acta*, 44, 701–709.
- 540 Hazen, R.M. and Navrotsky, A. (1996) Effect of pressure on order-disorder reactions. *American*  
541 *Mineralogist*, 81, 1021–1035.
- 542 Langford, J.I. (1978) A rapid method for analyzing the breadths of diffraction and spectral lines  
543 using the Voigt function. *Journal of Applied Crystallography*, 11, 10–14.
- 544 Langford, J.I. and Wilson, A.J.C. (1978) Scherrer after sixty years: a survey and some new  
545 results in the determination of crystallite size. *Journal of Applied Crystallography*, 11, 102–  
546 113.
- 547 Lavrentiev, M.Y., van Westrenen, W., Allan, N.L., Freeman, C.L., and Purton, J.A. (2006)  
548 Simulation of thermodynamic mixing properties of garnet solid solutions at high  
549 temperatures and pressures. *Chemical Geology*, 225, 336–346.
- 550 Newton, R.C., Charlu, T.V., and Kleppa, O.J. (1977) Thermochemistry of high pressure garnets  
551 and clinopyroxenes in the system CaO-MgO-Al<sub>2</sub>O<sub>3</sub>-SiO<sub>2</sub>. *Geochimica et Cosmochimica Acta*,  
552 41, 369–377.
- 553 Newton, R.C. and Wood, B.J. (1980) Volume behavior of silicate solid solution. *American*  
554 *Mineralogist*, 65, 733–745.



- 555 Nishihara, Y., Takahashi, E., Matsukage, K.N., and Kikegawa, T. (2003) Thermal equation of  
556 state of omphacite. *American Mineralogist*, 88, 80–86.
- 557 Oberti, R., Quartieri, S., Dalconi, M.C., Boscherini, F., Iezzi, G., Boiocchi, M., and Eeckhout,  
558 S.G. (2006) Site preference and local geometry of Sc in garnets: Part I. Multifarious  
559 mechanisms in the pyrope-grossular join. *American Mineralogist*, 91, 1230–1239.
- 560 Pavese, A., Levy, D., and Pischedda, V. (2001) Elastic properties of andradite and grossular, by  
561 synchrotron X-ray diffraction at high pressure conditions. *European Journal of Mineralogy*,  
562 13, 929–937.
- 563 Quartieri, S., Boscherini, F., Dalconi, C., Iezzi, G., Meneghini, C., and Oberti, R. (2008)  
564 Magnesium *K*-edge EXAFS study of bond-length behavior in synthetic pyrope-grossular  
565 garnet solid solutions. *American Mineralogist*, 93, 495–498.
- 566 Scherrer, P. (1918) Bestimmung der Grösse und der inneren Struktur von Kolloidteilchen mittels  
567 Röntgenstrahlen. *Nachrichten der Kgl Gesellschaft der Wissenschaftern zu Göttingen*, 26,  
568 98–100.
- 569 Shannon, R.D. (1976) Revised effective ionic radii and systematic studies of interatomic  
570 distances in halides and chalcogenides. *Acta Crystallographica A*, 32, 751–767.
- 571 Stokes, A.R. and Wilson, A.C. (1944) The diffraction of X rays by distorted crystal aggregates.  
572 *Proceedings of the Physical Society*, 56, 174–181.
- 573 Vinograd, V.L. and Sluiter, M.H.F. (2006) Thermodynamics of mixing in pyrope-grossular,  
574  $\text{Mg}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ - $\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$ , solid solution from lattice dynamics calculations and Monte  
575 Carlo simulations. *American Mineralogist*, 91, 1815–1830.

- 576 Vinograd, V.L., Sluiter, M.H.F., Winkler, B., Putnis, A., Hålenius, J.D.G., and Becker, U. (2004)  
577 Thermodynamics of mixing and ordering in pyrope-grossular solid solution. Mineralogical  
578 Magazine, 68(1), 101–121.
- 579 Williamson, G.K. and Hall, W.H. (1953) X-ray broadening from filled aluminum and wolfram.  
580 Acta Metallurgica, 1, 22–31.
- 581 Zhang, L., Ahsbahs, H., and Kutoglu, A. (1998) Hydrostatic compression and crystal structure of  
582 pyrope to 33GPa. Physics and Chemistry of Minerals, 25, 301–307.
- 583

584 **TABLE 1.** Unit cell parameter and microstrain of garnet solid solution Py<sub>40</sub>Gr<sub>60</sub> synthesized at 6  
 585 GPa and different temperatures with different annealing time, measured with synchrotron X-ray  
 586 diffraction at BNL

Experiment	Temperature (°C)	Heating time (hour)	Unit cell parameter (Å/cell)	Microstrain (%)	Remarks
TT721	1400	0.5	11.722(5)	0.038(10)	Garnet used in previous studies (Du et al. 2015, 2016)
GG999	1400	0.5	11.731(1)	0.022(3)	Heated at 1400 °C for 30 minutes and cooled slowly to 1100 °C
GG1000	*	*	11.705(1)	0.072(5)	Synthesized 1400 °C for 30 minutes and heated at 1200 °C for 1 day, *but ended with blow out; temperature may have been higher than 1700 °C briefly because the Pt capsule melted
GG1001	1200	24	11.725(1)	0.048(9)	Single phase garnet
GG1002	1000	48	n.a	n.a	Garnet with much broader peaks, possible two phases
GG1003	1100	2	n.a	n.a	Pyroxene coexists with garnets and glass
GG1004	1100	48	11.732(1)	0.031(8)	Single phase garnet
BB1007	1200	48	11.720(1)	0.045(3)	Start from big glass chunks in KBr and product is single phase garnet
GG1005	1200	48	11.721(1)	0.048(4)	Start from glass powder and product is single phase garnet
GG1006	1400	48	11.715(1)	0.059(3)	Starting material is product of TT721

587  
 588  
 589  
 590  
 591  
 592  
 593  
 594

595 **TABLE 2.** Unit cell parameter and microstrain of garnets Py<sub>20</sub>Gr<sub>80</sub> and Py<sub>80</sub>Gr<sub>20</sub> measured with  
 596 synchrotron X-ray diffraction at ALS

	Experiment	Starting material	Temperature (°C)	Time (hour)	Unit cell parameter (Å)	Microstrain (%)
Py <sub>20</sub> Gr <sub>80</sub>	TT649	glass	1400	0.5	11.787(1)	0.099(4)
	TT889	glass	1300	0.1	11.807(1)	0.065(2)
	TT895a	Product of TT889	1400	1.6	11.789(1)	0.081(5)
Py <sub>80</sub> Gr <sub>20</sub> (Du et al. 2016)	TT724	glass	1400	0.5	11.542(3)	0.101(3)
	TT890	glass	1300	0.1	11.543(2)	0.036(3)
	TT895b	Product of TT890	1400	1.6	11.540(2)	0.102(3)

597

598

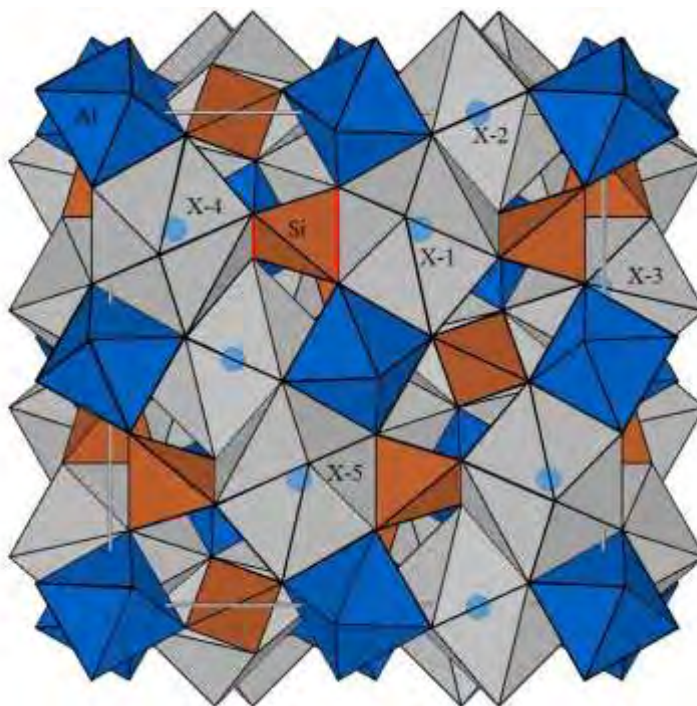
599 **TABLE 3.** Unit cell parameter of garnet Py<sub>40</sub>Gr<sub>60</sub> (GG999 with large excess volume and low  
 600 microstrain) at different pressure measured with synchrotron X-ray diffraction at BNL

P(GPa)*	a(Å)	V(Å <sup>3</sup> )	Microstrain (%)	f <sub>E</sub>	F <sub>E</sub> (GPa)
0	11.730(1)	1614.1(2)	0.032(5)	0.000(1)	0.000
0.9	11.709(1)	1605.4(4)	0.043(26)	0.002(1)	155.5±4.6
1.5	11.694(1)	1599.1(2)	0.012(11)	0.003(1)	156.1±2.5
2.7	11.665(1)	1587.3(5)	0.059(16)	0.006(1)	158.0±2.9
3.6	11.646(1)	1579.4(5)	0.006 (27)	0.007(1)	157.4±2.5
4.0	11.635(1)	1575.2(6)	0.003(21)	0.008(1)	156.0±2.1
4.6	11.624(1)	1570.7(5)	0.006(29)	0.009(1)	159.6±1.7
5.1	11.612(1)	1565.9(5)	0.052(33)	0.010(1)	157.7±1.9
5.5	11.603(1)	1562.2(6)	0.034(17)	0.011(1)	156.1±1.7
6.4	11.584(1)	1554.4(6)	0.100(30)	0.013(1)	156.3±1.9
7.0	11.576(2)	1551.3(7)	0.044(21)	0.013(2)	161.6±1.7
7.4	11.565(1)	1546.9(5)	0.049(21)	0.014(1)	160.6±2.0
8.5	11.544(2)	1538.6(7)	0.002(23)	0.016(2)	160.4±1.2
9.3	11.530(1)	1532.7(5)	0.094(27)	0.018(1)	162.2±1.0
10.3	11.510(1)	1524.9(5)	0.051(42)	0.019(1)	161.4±1.0

601 \*the uncertainties of the pressure values constrained from the pressure measured before  
 602 and after data collection are about 0.1-0.2 GPa

603

604



605

606 **FIGURE 1.** Polyhedral model of the pyrope-grossular garnet structure showing the dodecahedra  
607 (gray) and their linkage to the SiO<sub>4</sub> tetrahedra (orange) within one unit-cell. The nearest  
608 neighbors for two dodecahedra are (X-1) and (X-2). The second nearest neighbors are (X-1) and  
609 (X-3), one dodecahedron is edge shared with SiO<sub>4</sub> tetrahedron and another dodecahedron is  
610 corner shared with the same SiO<sub>4</sub> tetrahedron. The third nearest neighbor (3NN) dodecahedra are  
611 (X-1) and (X-4) that are both edge shared (the red lines) with the same SiO<sub>4</sub> tetrahedron. As  
612 suggested by Bosenick et al. (2000) the third nearest neighbor positions are the least  
613 energetically favored by Mg-Mg or Ca-Ca pairs because the double edge sharing of SiO<sub>4</sub>  
614 tetrahedron between X-1 and X-4 dodecahedra resists axial and torsional distortions imposed by  
615 strain.

616

617

618

619

620

621

622

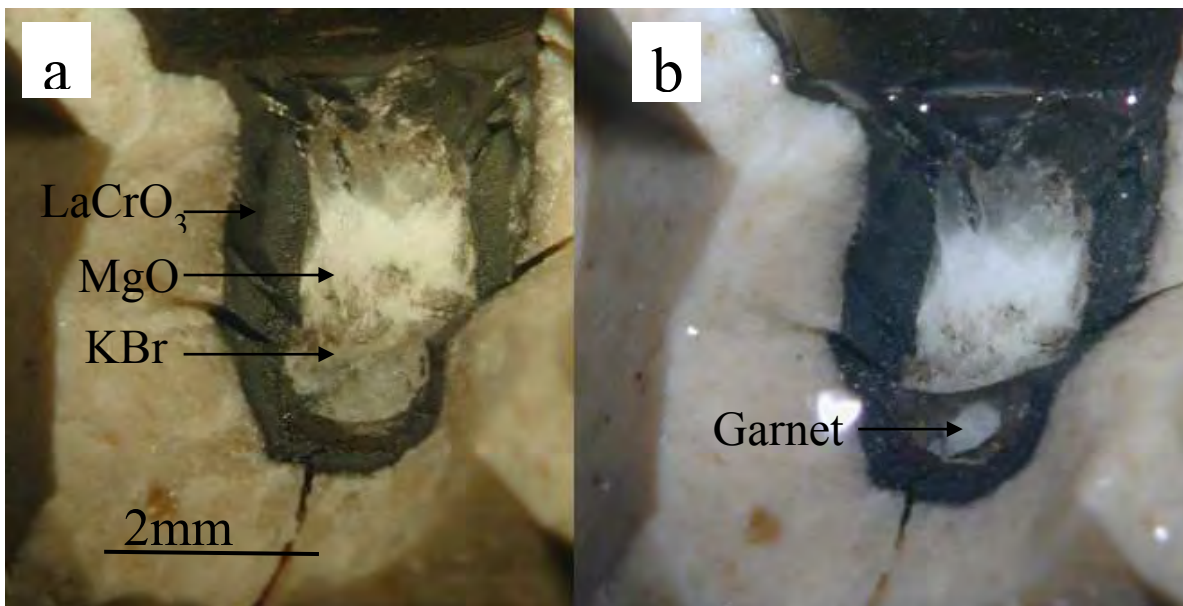
623

624

625

626

627



628

629

630

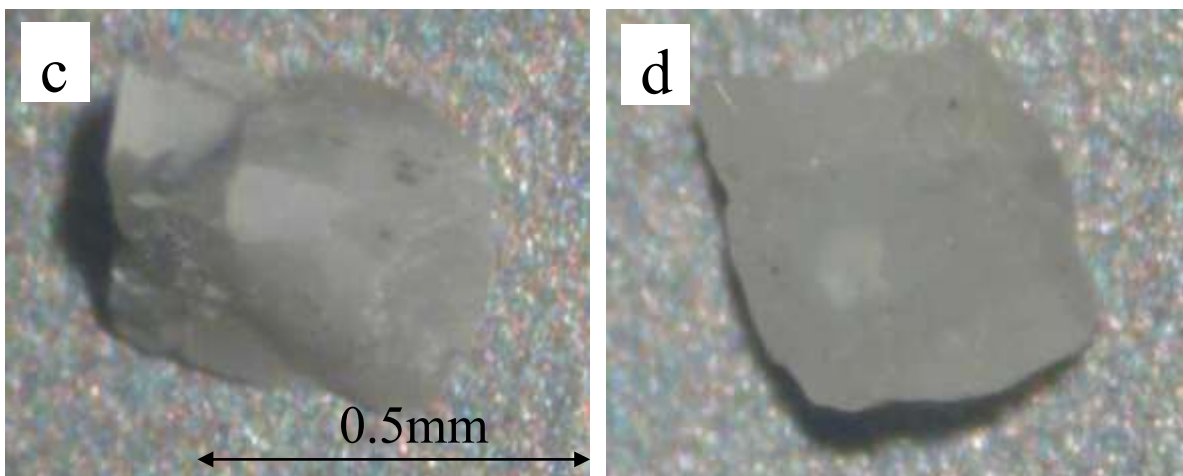
631

632

633

634

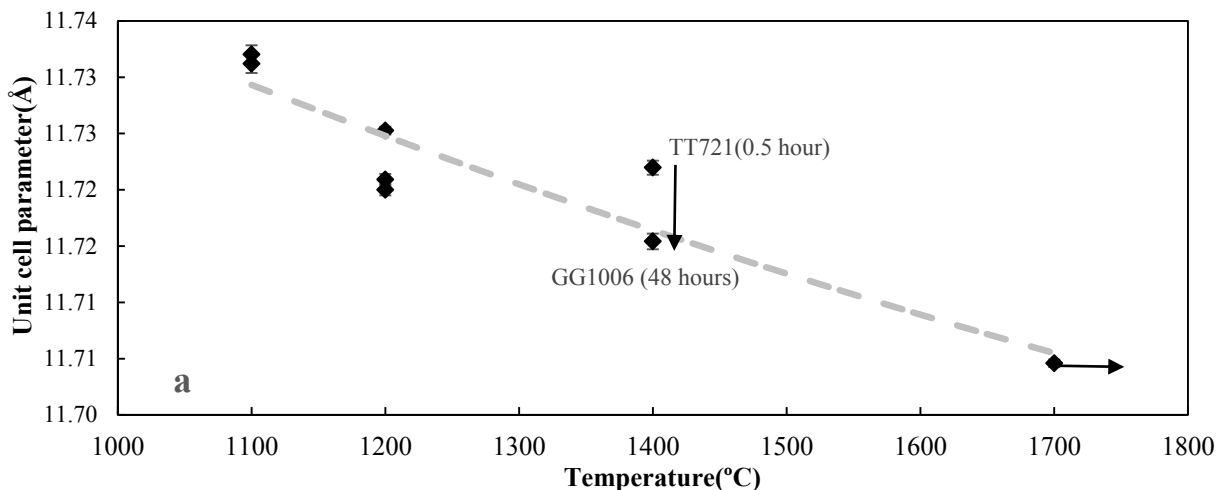
635



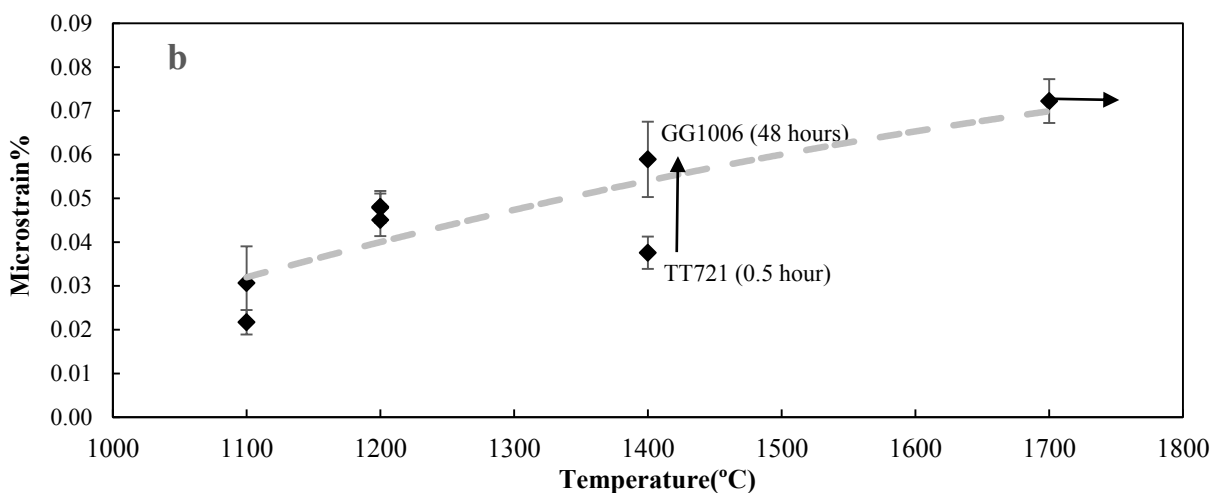
636 **FIGURE 2.** **a** Pictures showing the set-up of reheating experiment (TT895) on Py<sub>20</sub>Gr<sub>80</sub>, KBr  
637 powder was used to protect the garnet crystal from crushing at higher pressure; **b** After washing  
638 away KBr powder by distilled water, garnet crystal was recovered without damage; **c** Garnet  
639 crystal synthesized from glass at 1300 °C and 6 GPa for only 5 minutes heating was used as  
640 starting material for further annealing experiment; **d** recovered garnet sample after annealing at  
641 higher temperature. Note that there is little difference in appearance before and after further  
642 annealing except for smoothing of the sharp edges and subduing of some cracks.

643

644



645



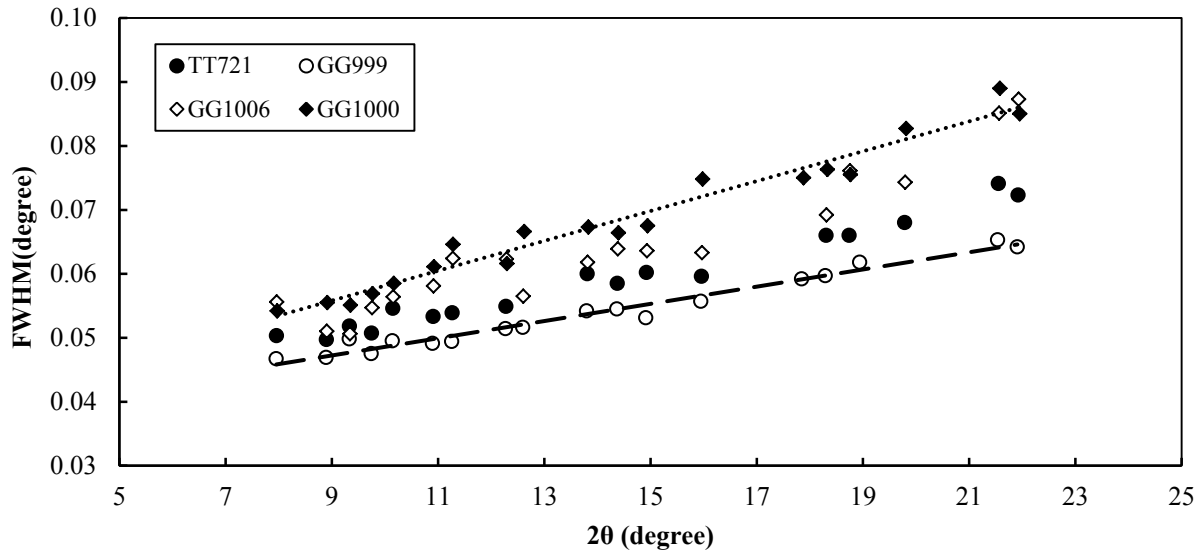
646

647 **FIGURE 3.** Cell parameters (a) and microstrains (b) of garnets (Py<sub>40</sub>Gr<sub>60</sub>) synthesized at  
648 different temperatures and annealed for times up to 48 hours. Higher temperature annealing  
649 causes more XRD peak broadening, and thus larger microstrain. Cell parameters decrease with  
650 both annealing time and higher temperature in a systematic way for sufficient annealing times.  
651 Microstrains grow with both annealing time (vertical arrows at 1400 °C) and higher temperature  
652 (x axis). GG1005 and GG1007, both at 1200 °C, show very good reproducibility. They differ in  
653 that GG1007 used a single chunk of glass starting material surrounded by KBr to prevent  
654 mechanical damage during compression from adding to the microstrains of Ca-Mg  
655 rearrangement. GG1005 was finely ground glass powder without such mechanical isolation from  
656 shear strain during compression. The reproducibility of GG1005 and GG1007 suggest that  
657 mechanical microstrains are not in play.

658

659

660



661

662

663

664

665

666

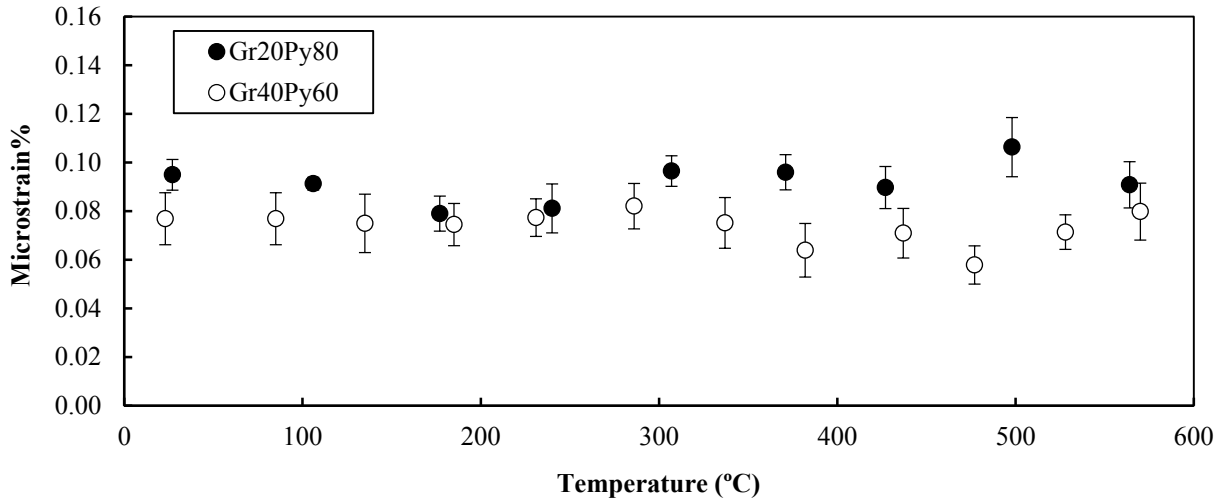
667

668

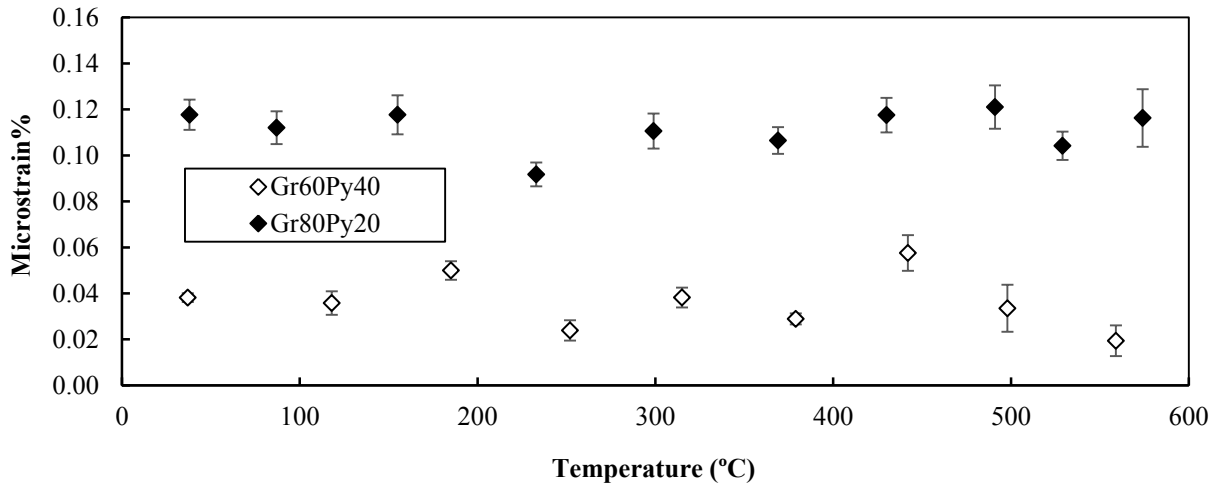
**FIGURE 4.** XRD peak broadening with diffraction angle  $2\theta$  for garnet ( $\text{Py}_{40}\text{Gr}_{60}$ ). Starting materials for experiments GG999, GG1000, and GG1006 are all same as the product of experiment TT721 synthesized at 1400 °C. Different annealing paths change the FWHM of XRD peaks. Higher heating temperature (GG1000) caused more XRD peak broadening [dotted line], and longer quenching time to lower temperature (GG999) releases microstrain [dashed line].



669



670



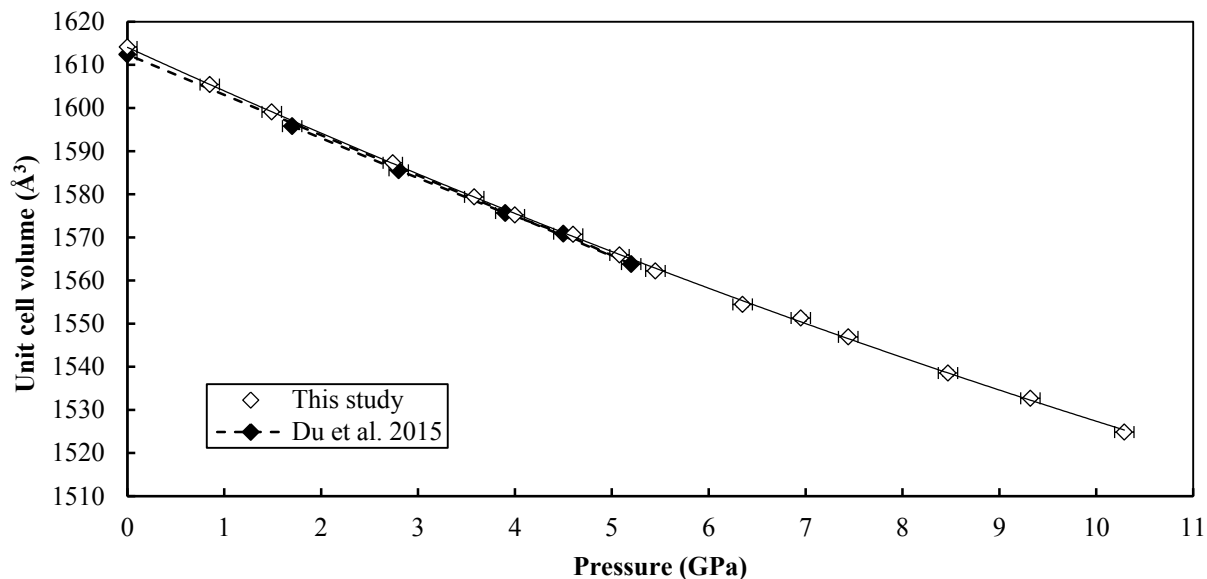
671

672 **FIGURE 5.** Microstrain of garnet solid solutions calculated from Williamson-Hall plot on XRD  
673 peak widening data at different temperatures and room pressure. As temperature increased up to  
674 600 °C, there is no systematic variation in microstrain for each garnet solid solution.

675

676

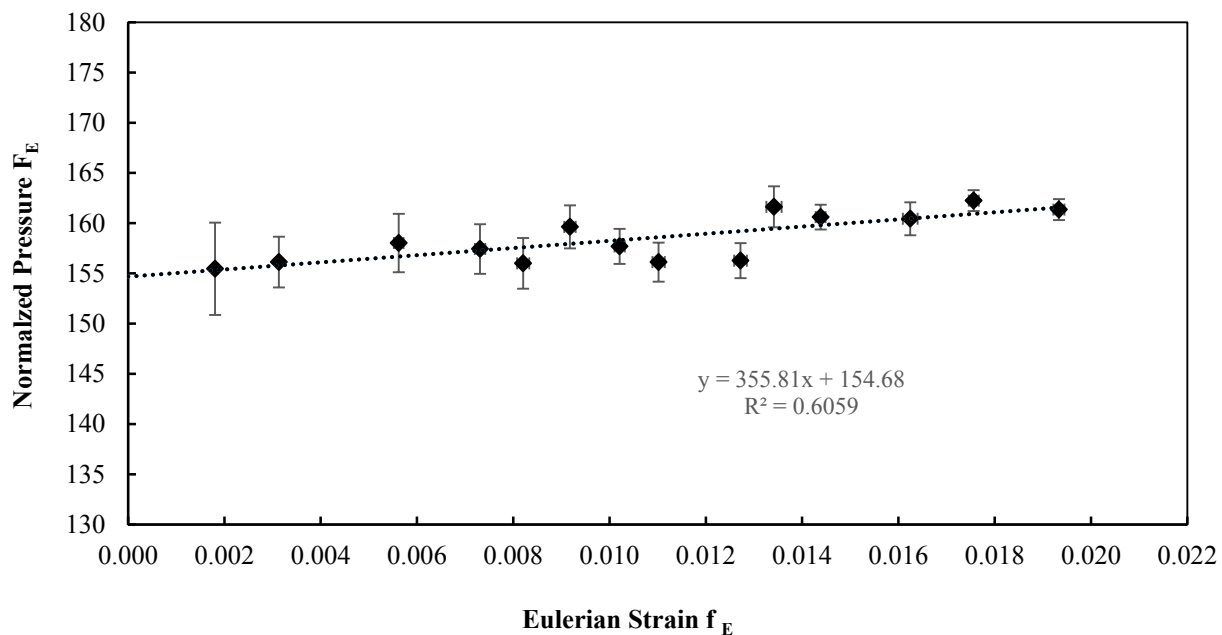
677



678

679 **FIGURE 6.** Volume compression of garnet with composition  $\text{Py}_{40}\text{Gr}_{60}$  from experiment GG999  
680 at high pressure and room temperature. The error bars on the unit cell volume data are smaller  
681 than the symbols. The data from Du et al. (2015) were collected at ALS on garnet  $\text{Py}_{40}\text{Gr}_{60}$   
682 (TT721)

683

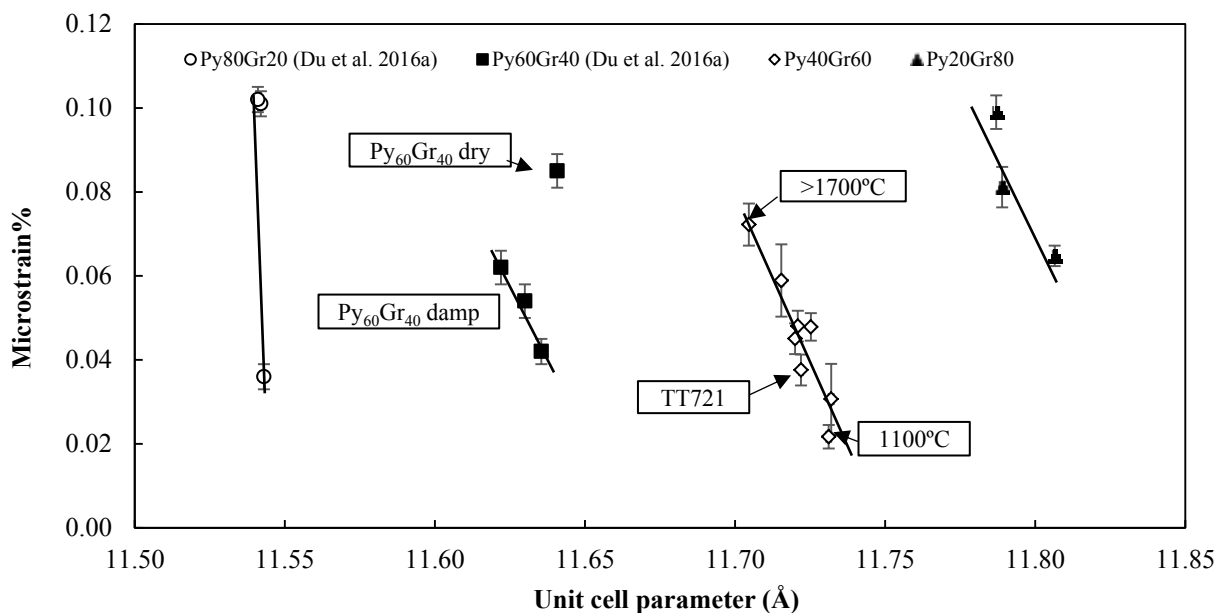


684

685 **FIGURE 7.** Volume Eulerian strain-normalized pressure ( $F_E - f_E$ ) plot of garnet ( $\text{Py}_{40}\text{Gr}_{60}$ ) from  
686 experiment GG999. The dashed line represents the linear fit through the data.

687

688



689

690 **FIGURE 8.** Anticorrelation between microstrain and unit cell parameters for garnets synthesized  
691 from glass at 6 GPa but annealed at different temperatures for different times given in Tables 1  
692 and 2, and from Du et al. (2016). The larger microstrains gained at higher temperature also  
693 correspond to a relatively smaller unit cell parameter. The anticorrelation of cell size and  
694 microstrain systematically becomes more subdued as the garnets become more pyrope-rich. The  
695 offset between the damp and dry  $\text{Py}_{60}\text{Gr}_{40}$  experiments suggests that trace moisture decreases  
696 cell size in addition to the effects of the Mg-Ca ordering anticorrelation that is preserved in the  
697 damp samples.

698

699

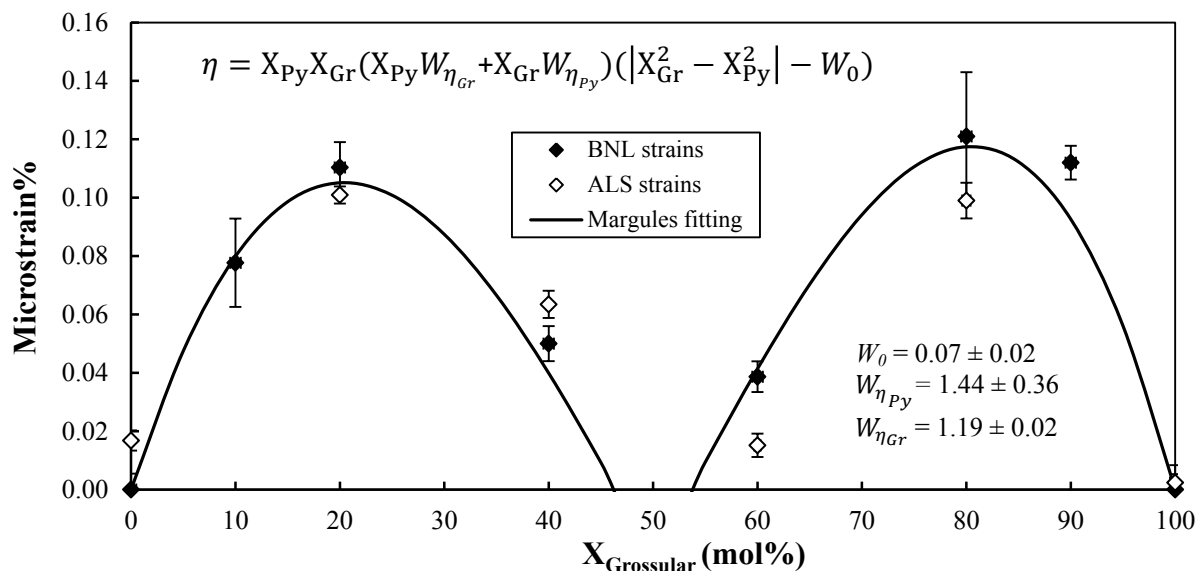
700

701

702



716



717

718 **FIGURE 10.** Microstrain data from ALS and BNL for dry garnets compared to a Margules-  
719 inspired fitting equation. The equation gives a two-peaked distribution of microstrains such as  
720 we observe across two independent data collection exercises at two different synchrotrons (ALS  
721 and BNL) with acceptable reproducibility. Uncertainties of the fitting parameters were difference  
722 calculated from fitting same equation to the two data sets collected at ALS and BNL. The small  
723 positive  $W_0$  pushes the fitting curve below the horizon of physical reality in a narrow region near  
724 equimolar Ca-Mg compositions. This region reflects the disruption in background Mg-Mg or Ca-  
725 Ca lattice structure by the intrusion of too many Ca-Mg pairs to allow a recognizable Mg-Mg or  
726 Ca-Ca lattice against which to introduce microstrains; i.e. the rationale locally fails.  
727