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2 *Submission to the Special Collection: Mechanisms, Rates, and Timescales of Geochemical*  
3 *Transport in the Crust and Mantle*

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6 **Element mobility during regional metamorphism in crustal and subduction zone**

7 **environments with a focus on the rare earth elements (REE)**

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9 Jay J. Ague<sup>1,2</sup>

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11 <sup>1</sup>Department of Geology and Geophysics

12 Yale University

13 P.O. Box 208109

14 New Haven, CT 06520-8109 USA

15 [jay.ague@yale.edu](mailto:jay.ague@yale.edu)

16

17 <sup>2</sup>Peabody Museum of Natural History

18 Yale University

19 New Haven, CT 06511 USA

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## Abstract

This paper explores bulk-rock geochemical data for a wide array of metamorphosed mafic, quartzofeldspathic, pelitic, and metacarbonate rocks using a quantitative mass balance approach to assess fluid-driven element mobility—particularly of the rare earth elements (REE)—in regional metamorphic and some high pressure subduction zone environments (40 examples; over 240 individual analyses). The results highlight the critical role of mineral reactions in controlling rock compositions and compositional changes. Most examples are from focused fluid flow settings, such as veins and lithologic contacts, where fluxes are large and metasomatic signals are thus strong. A variety of REE behaviors are observed, including little or no REE mobility (roughly 1/3 of the data set); light REE (LREE), mid-REE (MREE), and/or heavy (HREE) mobility; europium “anomalies”; overall REE losses; and local REE redistribution. The REE are typically fractionated by mass transfer, with the exception of several examples that underwent fairly uniform overall losses of REE. The fractionation reflects strong mineralogical controls on REE uptake/loss by rocks by a comparatively small number of phases. Examples include: HREE mass changes associated with garnet, xenotime, and sphene; LREE and MREE changes associated with apatite, monazite, and allanite; and Eu changes associated with plagioclase and lawsonite. As mineralogy is a dominant control, the nature of the metasomatism is not strongly correlated with metamorphic grade, other than obvious mineralogical differences between settings (e.g., plagioclase in Barrovian metamorphism, lawsonite in subduction zones). Extensive mobilization of non-REE major and trace elements can happen without significant open-system transport of the REE. If REE mobility occurs, it is always accompanied by mobilization of other non-REE. When mobile, neighboring REE (e.g., Sm and Nd) typically have strongly correlated mass changes indicating that both were mobilized

44 to about the same degree. Thus, constancy of neighboring REE mass ratios does not mean that  
45 they were immobile. Although individual examples of metasomatism can show correlations  
46 between patterns of mass transfer for the REE and the non-REE, little such correlation is evident  
47 across the entire data set, with the exception of P. Once again, this highlights the importance of  
48 individual minerals in controlling REE systematics. REE mass transfer coupled to gains or losses  
49 of volatiles is also not evident, although more data are clearly needed to document the impact of  
50 F, Cl, CO<sub>2</sub>, sulfur, and other species on REE behavior. Broad correlations of REE and P mobility  
51 suggest REE transport by P complexes, or REE and P transport together by some other  
52 complexing agent. Mass changes for REE and Y are more strongly coupled, reflecting the  
53 geochemical similarity of these elements and perhaps indicating a role for Y complexing as well.

54 The REE are the most mobile of the high-field strength elements (HFSE). On a  
55 percentage mass basis the amount of HFSE mobility decreases roughly in the order:  
56 REE>U>Nb>Ti>Th~Zr. Th mobility is rare but when present is positively correlated with U and  
57 REE mobility. The mobility of the more refractory HFSE is low in aqueous fluids, but is a larger  
58 concern in more extreme environments such as magmatic/magmatic-hydrothermal systems,  
59 charnockite metamorphism, and supercritical fluids in high-pressure/ultrahigh-pressure settings.  
60 The mobility of certain HFSE (e.g., Nb) can be large enough to affect rock plotting positions on  
61 petrotectonic discrimination diagrams.

62 Potassium and related (Rb, Ba, Cs) large-ion lithophile elements (LILE) are typically lost  
63 in association with mica breakdown. As the data set focuses on high fluid flow environments,  
64 LILE changes will likely be smaller for rocks that undergo less fluid-rock interaction. Gains of K  
65 and related LILE are coupled to volatile gain and Na loss, consistent with fluid flow down  
66 temperature gradients, or infiltration from K-rich sources (e.g., pegmatites). In Barrovian

67 settings, two other mass transfer trends are evident in addition to K mass transfer. First, silica  
68 loss is coupled to volatile loss, illustrating the dependence of silica mass transfer on  
69 devolatilization. Most of the silica loss reflects local mass transfer into adjacent vein fluid flow  
70 conduits where additional silica is precipitated; indeed, wallrock inclusions in veins can be highly  
71 silicified. Second, Na and Ca mass gains and losses are well correlated in the Barrovian  
72 examples, reflecting control of plagioclase growth or destruction. Strontium and Pb behaviors  
73 also appear to be largely related to plagioclase behavior (in subduction settings, this role can be  
74 played by phases such as lawsonite or epidote group minerals).

75         Carbon dioxide mass transfer by conventional devolatilization as well as near-  
76 stoichiometric  $\text{CaCO}_3$  dissolution or precipitation are represented in the data set. Determining the  
77 relative roles of these processes as functions of metamorphic grade, intensity of fluid-rock  
78 interaction, and tectonic setting will be an important challenge for future research.

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## Introduction

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Time-integrated fluid fluxes due to crustal devolatilization during regional

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metamorphism of sedimentary rock-dominated sequences can attain  $\sim 10^3 \text{ m}^3 \text{ m}^{-2}$  averaged over

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regional scales (e.g., Wickham and Taylor, 1985; Ague, 1994b, 2014; Ferry, 1994; Skelton et al.,

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1996; Bickle, 1996; Hanson, 1997; Lyubetskaya and Ague, 2009). In other words, a column of

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fluid  $\sim 1 \text{ km}$  long passes through a  $1 \text{ m}^2$  area of rock oriented perpendicular to flow. Fluxes of

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this magnitude will be an unavoidable consequence of prograde fluid generation in such

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sequences. Fluxes upward through subducted oceanic crust are also large and have been

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estimated to be  $\sim 300 \text{ m}^3 \text{ m}^{-2}$  (Zack and John, 2007), but could be even larger depending on the

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hydration state of subducted mantle (e.g., Faccenda et al. 2009) or if focused along the slab-

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mantle wedge interface (e.g., Wilson et al. 2014).

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These fluxes transfer massive quantities of  $\text{H}_2\text{O}$ ,  $\text{CO}_2$ ,  $\text{CH}_4$ , and other volatile species

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from actively metamorphosing regions in the lithosphere to the shallow crust and hydrosphere. In

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so doing, they drive reaction progress (particularly in carbonate-bearing rocks), exert strong

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controls on rock rheology and deformation behavior, determine the onset of partial melting in the

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crust and mantle wedge and, ultimately, influence the long-term global carbon cycle.

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Such fluxes are undeniably large, but whether or not they can regionally transport “non-

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volatile” elements such as the alkalis, alkaline earths, transition metals, and rare earth elements

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(REE) depends strongly on the nature of flow. If the flow is everywhere uniform and pervasive

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at the grain scale, then maximum fluxes in any given part of the rock mass are limited to values

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in the  $10^3 \text{ m}^3 \text{ m}^{-2}$  range. On the other hand, if the flow is focused into some form of conduit, then

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within conduits fluxes can be far larger than regional average values by one or even two orders

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of magnitude (Fig. 1). Of course, this also means that areas outside of a conduit are impoverished

102 in fluid, because fluid is diverted away from them. It is in the more focused flow regimes that  
103 non-volatile element transfer on large scales will occur most readily, at time-integrated fluxes of  
104  $\sim 10^4$  to  $\sim 10^6$  m<sup>3</sup> m<sup>-2</sup>.

105 Focusing generally results from rock permeability variations. High-permeability areas  
106 will have the best potential to evolve into fluid conduits. Examples are many, and include high-  
107 permeability layers, fractures and veins, boudin necks, shear zones, mélange zones, high  
108 permeability regions developed in certain compacting two-phase flow systems (Connolly and  
109 Podladchikov, 2007), fold hinges (e.g., Skelton et al. 1995), and lithologic contacts (for reviews  
110 see, e.g., Oliver, 1996; Ague, 2014). It is straightforward to illustrate how increasing the  
111 permeability and proportions of conduits can drastically alter the hydrologic regime. Consider  
112 unidirectional flow where 10 volume percent of the rock mass is composed of conduits parallel  
113 to the flow direction that are a factor of  $10^2$  more permeable than the surroundings. From  
114 Darcy's law, more than 90% of the total fluid flow is carried by the conduits (assuming uniform  
115 fluid pressure gradient and viscosity across the flow region). Permeability variations of this  
116 magnitude are easily conceivable; for example, for planar cracks, permeability scales as the cube  
117 of the crack aperture (e.g., Norton and Knapp, 1977).

118 The aim of this paper is to examine chemical changes in rocks brought on by focused  
119 fluid flow in the continental crust and in subduction zones (mostly  $< \sim 2.5$  GPa) up to, but not  
120 including, the realm of extensive partial melting. There are many chemical elements to  
121 consider—too many for one paper—so the main theme will be mobility of the rare earth  
122 elements (REE). Other elements such as large-ion lithophile (LILE) and high-field strength  
123 (HFSE) elements will be discussed when appropriate.

124 Historically, REE mobility has commonly been inferred to be relatively limited during  
125 hydrothermal or metamorphic fluid-rock interaction, and this is indeed the case in many  
126 environments (e.g., Taylor and McLennan, 1985; Michard and Albarède, 1986; Banner and  
127 Hanson, 1990; Bau, 1991). Nonetheless, clear examples of transport have been well documented.  
128 REE transport in fluids has broad implications for understanding the nature and timing of a wide  
129 array of lithospheric processes. For example, determining the tectonic setting of magmatism  
130 using trace element discrimination diagrams is predicated on little or no REE mobility  
131 subsequent to igneous crystallization (e.g., Pearce et al. 1984). Dating of garnet growth events  
132 using Sm-Nd or Lu-Hf isotope systematics likewise requires REE immobility in the host rock  
133 (e.g., Baxter and Scherer, 2013). Mobility must also be carefully assessed when interpreting  
134 whole-rock isochrons (e.g., O'Neill et al. 2012). REE mobility during subduction, or lack  
135 thereof, exerts important controls on arc magma chemistry as well as the REE budget of  
136 lithosphere subducted deep into Earth's mantle beyond the arc magma window (e.g., Sorensen  
137 and Grossman, 1989; Spandler et al. 2003; 2004; Zack and John, 2007; Miller et al. 2009;  
138 Bebout, 2014; Bebout and Penniston-Dorland, 2016; Taetz et al. 2016).

139 A wide range of approaches have been applied to assess REE mobility. A common  
140 strategy is to compare the chondrite-normalized REE patterns of a suite of low-grade protoliths  
141 to those of higher-grade rocks. This approach has been used, for example, to deduce little REE  
142 mobility in subduction complex rocks from western Liguria (Tribuzio et al. 1996) and New  
143 Caledonia (Spandler et al. 2003; 2004). Variability of REE patterns can also provide clues to  
144 transport. Zhao et al. (2007) found that light REE (LREE) patterns in UHP eclogites from the  
145 Sulu orogen ranged widely, whereas HREE patterns were more or less constant. They concluded  
146 that the LREE variability resulted from mobilization, but that the HREE were immobile. Similar

147 conclusions regarding larger LREE mobility relative to heavier REE have been reached in a  
148 variety of other studies (e.g., Weaver and Tarney, 1981; Griffin and Brueckner, 1985; Zack and  
149 John, 2007). The presence of isotopic disequilibrium, as evidenced by, for example, negative  
150 Sm-Nd model ages, or extreme ranges of Nd isotopic ratios, are further indicators of REE  
151 transport (e.g., Windrim et al. 1984; Munz et al. 1994).

152 This paper will center on selected studies for which the mass balance assessment of  
153 element transport, including major and trace elements, is feasible. This is related to, but distinct  
154 from, the ever-growing body of work that has illuminated REE partitioning between phases at  
155 the scale of hand samples or thin sections. Such REE redistribution can be the result of  
156 progressive reaction in closed or nearly closed systems, and need not indicate mobility into or  
157 out of a rock mass. Studies of the REE patterns of individual phases are a critical aspect of  
158 understanding the geochemistry of REE, but our focus is mostly on bulk-rock chemistry and the  
159 prospect of larger-scale mobility in focused flow environments including veins, lithologic  
160 contacts, and shear zones. REE mobility in shallow and/or low-temperature settings including  
161 hydrothermal seafloor alteration, burial metamorphism, and contact metamorphic phenomena  
162 such as skarns (e.g., Hellman et al. 1979; Windrim et al. 1984; Bartley, 1986; Gruau et al. 1992;  
163 Marsh et al. 2012; Durand et al, 2015; Floess et al. 2015) is for the most part not considered, the  
164 emphasis being placed on deeper lithospheric processes.

165 Most of the data has been published; some is presented here for the first time (Table 1).  
166 The data set is by no means exhaustive, but it nonetheless attempts to document the more  
167 common sorts of mobility recognized in nature.

## 168 **Some Comments on Determining Chemical Changes in Rocks**



169 Mass balance analysis compares rocks that have been chemically changed by fluid  
170 infiltration to those that have not in order to determine what elements have been gained or lost.  
171 The governing equations describing mass and volume changes have been well studied;  
172 derivations and discussion can be found in, for example, Gresens (1967), Grant (1986), Brimhall  
173 and Dietrich (1987), Brimhall et al. (1988), Ague (1994a), and Baumgartner and Olsen (1995). A  
174 brief summary is given in the appendix. The final mass balance expressions are similar  
175 regardless of treatment, although Brimhall and Dietrich (1987) and Brimhall et al. (1988)  
176 illuminated the relationships between mass change, rock and mineral density, porosity, and  
177 volume change.

## 178 **Closure**

179 Despite the straightforward mathematical nature of the equations, the basic problem of  
180 how to quantify rock mass and volume changes has generated a remarkable amount of confusion  
181 over the years. The source of much of this confusion is the fact that increases or decreases in the  
182 weight percent or ppm of an element need not correlate with actual gains or losses of the mass of  
183 that element. At first, this seems very counterintuitive. If the concentration of, for example, La  
184 increases from 100 to 125 ppm during some fluid-driven mass transfer event, doesn't that have to  
185 mean that the rock gained La?

186 It turns out that there is more than one explanation and La need not have been gained.  
187 Suppose that 20 wt. % silica was removed from the rock due to, for example, local diffusion to  
188 an adjacent quartz vein. Such mass changes are commonly observed (e.g., Ague, 1994b;  
189 Penniston-Dorland and Ferry, 2008). So 100 g of rock now has a mass of 80 g. Let us further  
190 assume that La was immobile during this alteration. Compositions are normalized to 100 wt. %,  
191 so the new La concentration is:  $(100 \text{ ppm}/80) 100 = 125 \text{ ppm}$ . Somewhat surprisingly, then, the

192 La increase from 100 to 125 ppm had nothing to do with La addition in this hypothetical  
193 example. Conversely, if the rock gained mass, the concentration of La would be decreased. The  
194 key point is that the concentrations of immobile elements can change during metasomatism. For  
195 cases where the REE are immobile, REE patterns can be shifted up (rock mass loss) or down  
196 (rock mass gain) (e.g., Masters and Ague, 2005; Wagner et al. 2010; Fig. 2).

197 The complicating factor here is *closure*: the concentrations of all the constituents in the  
198 composition have to add up to 100 wt. % (within analytical error). Thus, due to the 100 %  
199 normalization constraint, changes in one oxide or element will impact one or more of the others  
200 (e.g., Chayes, 1949; 1971). Therefore, compositions are inherently multivariate, and statistical  
201 analysis of compositional data commonly benefits from multivariate statistical techniques (e.g.,  
202 Aitchison, 1986; Woronow and Love, 1990; Ague and van Haren, 1996; Ague, 2011). The core  
203 of this multivariate approach is to use ratios of concentrations, not their absolute values. Because  
204 ratio statistics tend not to be Gaussian and for other technical reasons, it is better to work with  
205 the logarithms of ratios rather than the ratios themselves, but this is easily done. Note that  
206 expressing compositions in moles does not alleviate the closure problem, as the underlying  
207 closure correlations remain.

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### 209 **Relative and absolute chemical changes**

210 Chemical changes can be determined in a relative or absolute sense. Assessing relative  
211 changes only requires consideration of elemental ratios, whereas some type of reference frame is  
212 required to assess absolute changes (see below).

213 Consider the increase in La/Lu along a metapelite layer approaching a fluid conduit in  
214 Figure 3a. There is no ambiguity regarding the precursor rock, so the change in ratio means

215 unequivocally that metasomatic changes have occurred. The percent change in ratio can be  
216 quantified, as can the statistical significance of differences in ratios between groups of samples  
217 (e.g., Woronow and Love, 1990).

218         The effects of metasomatism are undeniable in Figure 3a, but there are multiple (five)  
219 interpretations for what actually happened. (1) La was added and Lu was lost. (2) La was added  
220 and Lu remained constant. (3) La and Lu were both added, but La was added to a greater extent.  
221 (4) La and Lu were both lost, but Lu was lost to a greater extent. (5) La was constant and Lu was  
222 lost. For this example, the mass transfer followed (2), resulting in a nearly order-of-magnitude  
223 increase in Lu/La. Figure 3b depicts a decrease in Eu/Sm toward a fluid conduit in a metarbonate  
224 rock. Here, both Eu and Sm were gained, but Sm was gained to a greater extent.

225         For many purposes, the change in ratio is all that is needed to test a particular geologic  
226 hypothesis. On the other hand, to differentiate between the five possibilities for ratio change,  
227 some type of reference frame is needed to enable quantification of mass changes for each  
228 element individually. Three reference frames are available: (1) zero or known rock mass change;  
229 (2) zero or known rock volume change (e.g. Chowdhury et al. 2013; Centrella et al. 2016); and  
230 (3) zero mass change for an “immobile” element or elements.

231         The geochemical reference frame of option (3) is the most general and permits evaluation  
232 of rock mass, volume, and individual element changes. There have been many discussions of  
233 how best to select the geochemical reference frame to which the reader is referred (e.g., Grant,  
234 1986; Brimhall et al. 1988; Ague, 1994a; 2003; Baumgartner and Olsen, 1995; Ague and van  
235 Haren, 1996; Penniston-Dorland and Ferry, 2008). Typical elements considered in this regard are  
236 the high-field strength elements (HFSE); Zr has received perhaps the most attention. Some  
237 commonly-employed elements, like Al and Ti, have limited mobility in some geologic settings,

238 but can be quite mobile in others. Using the REE as a reference frame without careful evaluation  
239 could be highly problematic as they can undergo significant transport despite being HFSE (see  
240 below). Moreover, in extreme environments, including some ultrahigh-pressure (UHP) rocks and  
241 zones of extensive partial melting, all or nearly all elements can be mobilized, thus limiting the  
242 utility of the geochemical reference frame approach. In what follows, the reference frame used in  
243 the original studies (option 1, 2, or 3) is also used herein unless otherwise noted. Statistical  
244 treatment of data follows Ague (2011).

245 Care must be taken when reading the classic paper by Grant (1986) which describes the  
246 “isocon” method for quantifying metasomatic changes. The method as employed in, e.g., Figure  
247 2 of Grant (1986) is straightforward. However, it is also asserted that for immobile elements, the  
248 change in “geochemical concentration” is equal to zero (p. 1977). Such “geochemical  
249 concentrations” must not be confused with actual concentrations; as shown above, the  
250 concentrations of immobile elements *must* vary due to the closure constraint if there are overall  
251 rock mass changes. It follows that the concentrations of immobile elements will be unchanged  
252 only in situations where there has been no overall bulk rock mass change (e.g., 100 kg of starting  
253 rock is transformed into 100 kg of altered rock). However, this is not the general case in  
254 metasomatism.

255 A further important consideration is that physical mixing of rock types may produce a  
256 very different metasomatic signature than fluid infiltration. This can affect, for example, the  
257 metasomatic rinds on blocks within deforming subduction zone mélange (e.g., Bebout and  
258 Barton, 2002; Penniston-Dorland et al. 2014).

259 Finally, intrinsic variations in pre-alteration rock chemistry due to, for example, original  
260 sedimentary heterogeneities or igneous fractionation, may be large enough to obscure or mask  
261 the geochemical evidence for metasomatism (e.g., Penniston-Dorland and Ferry, 2008).

262

## 263 **REE Behavior During Fluid-Rock Interaction**

### 264 **Rock types and patterns of REE mobility**

265 The data set focuses on representative examples for which quantitative mass balance  
266 analysis is viable. These have direct relationships between precursors and their altered  
267 equivalents together with well-defined reference frames, as well as compositional data for a  
268 reasonably wide range of major, minor, and trace elements. Given these constraints, metasomatic  
269 studies that lack REE analyses, or that have REE and trace element analyses but lack major  
270 elements are, with two exceptions, not included in the data set. New studies are being published  
271 on a regular basis; unfortunately, some are too recent to be incorporated herein but it would be  
272 very instructive to include them in future updates to the data set.

273 To help facilitate discussion, rock types are subdivided somewhat arbitrarily into six  
274 broad categories. (1) Regional metamorphism including mostly Barrovian metamorphism of  
275 pelitic rocks but also mafic lithologies. (2) Regional metamorphism of carbonate rocks. (3)  
276 Subduction zone metamorphism including pelitic and mafic precursor rock types. (4) Subduction  
277 zone metamorphism of carbonate rocks. (5) Greenschist facies shear zones cutting granitic rocks,  
278 Mont Blanc Massif. (6) The final category comprises examples that do not fit neatly into the  
279 other categories: dynamothermal aureole (shear zone), magmatic/pegmatite fluid infiltration, and  
280 charnockite formation. The mass changes for many elements including the REE and other HFSE  
281 can be significant in this category, thus pushing the limits of mass balance-type analysis. The

282 charnockite example is from the classic locality at Kabbaldurga, southern India (Stähle et al.  
283 1987), but is not intended to represent all forms of charnockite-related mass transfer. Rather, it is  
284 included as an illustration of strong HREE metasomatism (see below).

285         Patterns of REE mobility have also been subdivided, as follows. (A) Limited or no REE  
286 mobility. (B) Light REE mobility. (C) Heavy REE mobility. (D) Overall loss of REE. (E)  
287 Europium “anomalies”. The quotation marks are employed to distinguish between the Eu mass  
288 changes due to fluid-rock interaction studied herein and other processes for which the term *Eu*  
289 *anomaly* is typically used such as igneous fractional crystallization. (F) Local redistribution of  
290 REE. Some overlap between categories is unavoidable as samples may be characterized by  
291 multiple patterns of metasomatism.

292         The element with the maximum mass change is referred to as REE<sub>Max</sub> (values are given  
293 in Table 1). The most common REE<sub>Max</sub> elements are La, Yb-Lu, and Eu; they are associated  
294 primarily with LREE, HREE, and Eu “anomaly” mass transfer, respectively, and account for  
295 about half of the total. For the remainder, all REE except Nd and Er are found as REE<sub>Max</sub>. These  
296 relationships should not be taken as representative of all natural systems, as the data set is not a  
297 statistical sampling of every form of REE mass transfer.

298

### 299 **Graphical representation of chemical changes**

300         Two types of plots are used extensively in this paper. The first type is percentage mass  
301 change diagrams for REE and other selected elements (Figs. 4, 6, 8, 10, 12, 14). Some data sets  
302 have small numbers of samples that may not fully represent the chemical variability of  
303 precursors or their altered equivalents, thus limiting the amount of statistical analysis that can be  
304 done. In some cases, only one precursor and one altered analysis are available. When they can be

305 calculated,  $\pm 2\sigma$  standard errors on average mass change values are shown. It is very important to  
306 point out that these are not chondrite-normalized plots, but rather depictions of elemental  
307 percentage mass change (mass gains are positive, losses negative). The second type of plot was  
308 developed herein and is referred to as a “bar code” diagram. These display percentage mass  
309 changes using a color-coding scheme (Figs. 5, 7, 9, 11, 13). Given the statistical limitations  
310 noted above, the mass changes are grouped into broad ranges that attempt to capture the general  
311 patterns of metasomatism while avoiding unwarranted detail that could lead to  
312 overinterpretation.

### 313 **Immobile or limited mobility of REE**

314 While the data set is not a statistically-comprehensive survey of all types of REE  
315 mobility, more examples (about 1/3) fall into this group than any other and include all the rock  
316 type categories noted above except category (6) (Figs. 4, 5). Thorium mobility is very rare in this  
317 group, as it is in most others. On the other hand, there are several examples of significant U  
318 metasomatism in subduction (LM7–LM9) and shear zone settings (LM13). Two examples  
319 underwent large P mass changes (LM2, LM11), but most did not. However, although mean P  
320 mass changes may be indistinguishable from zero statistically,  $2\sigma$  uncertainties can be large  
321 which suggests local scale redistribution of P even though it was not lost or gained overall.  
322 Yttrium was mostly immobile in these examples, although several display small gains or losses.  
323 Importantly, a wide array of other major and trace elements were mobile, illustrating that  
324 significant rock chemical alteration is possible without strongly affecting the REE (Fig. 5). The  
325 patterns of element mobility vary considerably, but most samples lost considerable LILE.

### 326 **Mobility of the LREE**

327 Most examples gained LREE, but losses are also observed (Figs. 6, 7). For samples that  
328 gained LREE, enrichments can be extreme and in some cases exceed +1000 %. Where data are  
329 available, these LREE gains are commonly associated with phosphates, particularly monazite  
330 and/or apatite; allanite can be another important LREE host in this group (Pan and Fleet, 1996).  
331 Gains of P, Y, or both are consistent with LREE concentration in phosphate phases. Most of  
332 these rocks lack garnet or other phases that would concentrate the HREE. For example, LREE2  
333 has only 1-2 volume % garnet. Loss of HREE and Y due to hornblende breakdown to biotite in  
334 the dynamothermal ophiolite aureole example (LREE6) is inferred to have been offset by HREE  
335 gains due to garnet growth (Dostal et al. 1980). In addition, the rock gained substantial Zr, Th,  
336 and U, so growth of metasomatic zircon may also have helped offset HREE losses. Although not  
337 specified in the original paper, the large gains in P strongly suggest that LREE gains were  
338 associated with a phosphate phase or phases. The LREE loss examples are also associated with  
339 phosphates. For example, LREE5 lost P and contains irregular monazite grains interpreted to be  
340 corroded relics. In addition, this example preserves remnants of allanite in kyanite; this could  
341 also have contributed to LREE loss.

342 All examples underwent various forms of major and minor element metasomatism;  
343 clearly, REE transport did not occur without the mobilization of other elements as well (Fig. 7).

344 Calculated mass additions for LREE2 increase smoothly with decreasing atomic number  
345 implying systematic open system behavior. However, it should be noted that due to uncertainties  
346 in the LREE2 geochemical reference frame and intrinsic sample heterogeneity, it is debatable if  
347 the LREE mass additions calculated herein for this example are in fact significant (Penniston-  
348 Dorland and Ferry, 2008). Nonetheless, reference frames using Al, Ti, and two Zr measurements



349 (XRF or ICP-MS) yield a mean mass change for La of  $52.4\% \pm 6.4\%$  ( $2\sigma$  standard error), an  
350 uncertainty roughly the size of the data symbols in Figure 7.

### 351 **Mobility of the HREE**

352 Gain of HREE is the most commonly observed pattern in this group (Figs. 8, 9). Where  
353 data are available, this was due mainly to the growth of garnet or xenotime; hornblende and  
354 sphene may also have been important in some instances (Fig. 8). Gain of HREE is commonly  
355 coupled to gain of Y, which is unsurprising given the importance of phases like garnet or  
356 xenotime in the chemical alteration. Phosphorous behavior is more variable. Example HREE9  
357 that interacted with pegmatite fluids gained Th. Metasomatic apatite in this sample contains Th  
358 (Carson and Ague, 2008), and it was also likely hosted by allanite (e.g., Hermann, 2002).  
359 Otherwise there was little Th mobility in this group. HREE9 also gained U, but the mineral host  
360 is uncertain. The example of incipient charnockite formation (HREE10) has the most striking  
361 REE losses, attributed to zircon, apatite, and monazite destruction by Stähle et al. (1987).

362 Many other major and trace elements were mobile but they do not vary systematically  
363 with HREE mass changes (Fig. 9).

### 364 **Overall loss of REE**

365 Examples which lost all REE are less common than others in the data set (Figs. 10, 11).  
366 They span a diverse range of settings: subduction complex, magmatic-hydrothermal fluid  
367 alteration, charnockite, and Barrovian metapelite. The subduction examples involve high fluid  
368 fluxes and infiltration of fluids with low solute loads capable of dissolving REE and destroying  
369 REE carriers including epidote, titanite, garnet, and/or phosphates (John et al. 2008; van der  
370 Straaten et al. 2008). The magmatic-hydrothermal fluid example is highly altered metapelite  
371 consisting mostly of sillimanite, muscovite, and graphite at the contact with pegmatite and

372 associated quartz veins. REE loss is inferred to be the result of apatite, monazite, and garnet  
373 destruction during alteration. The Barrovian example is a plagioclase-rich alteration selvage  
374 adjacent to a quartz vein. Garnet breakdown led to HREE and Y loss, and there is considerable  
375 evidence for overgrowths on, and dissolution-reprecipitation of, apatite which likely affected the  
376 LREE and mid REE (MREE) (Masters and Ague, 2005; Ague and Baxter, 2007). As noted  
377 above, loss of REE during incipient charnockite formation was attributed to zircon, apatite, and  
378 monazite destruction by Stähle et al. (1987). All examples lost P, Y, and U; extreme fluid-rock  
379 interaction even caused Th loss in two cases. No discernible correlations of REE behavior with  
380 that of other elements are evident (Fig. 11).

### 381 **Fluid-rock interaction and europium “anomalies”**

382         The precipitation or breakdown of plagioclase or lawsonite largely control gains or losses  
383 of Eu with respect to other REE. These Eu mass change “anomalies” are here subdivided into  
384 four categories (Figs. 12, 13). (1) Fluid-rock interactions that destroy plagioclase can deplete Eu,  
385 whereas those that produce plagioclase can enrich it. For example, in Eu1, plagioclase and garnet  
386 grew in response to fluid infiltration. The growth of garnet is inferred to have produced the Y  
387 enrichment. Minor to moderate losses of some REE in this example may have resulted from  
388 phosphate mineral destabilization. In Eu2, plagioclase was destroyed, leading to a negative  
389 “anomaly”. (2) The precipitation of plagioclase from vein-forming metamorphic or pegmatitic  
390 fluids progressively depletes the fluids in Eu relative to neighboring REE. Consequently, these  
391 Eu-depleted fluids can produce metasomatic REE patterns characterized by negative Eu  
392 “anomalies” in the rocks they infiltrate (Eu3, Eu5, Eu6), even if the infiltrated rocks lack  
393 plagioclase (e.g., Eu3; Ague, 2003; Carson and Ague, 2008). Profiles across zones of altered  
394 rock adjacent to conduits for such fluids commonly show progressively deepening Eu

395 “anomalies” as the conduits are approached (Eu3). (3) Redistribution of Eu among feldspars  
396 resulting in little or no overall Eu mass change. For the charnockite example (Eu7), the positive  
397 Eu “anomaly” is very distinct relative to the other REE, most of which were lost due to  
398 destruction of apatite, monazite, and zircon (Stähle et al. 1987). (4) Preferential enrichment of Eu  
399 is also observed in metasomatic lawsonite (Eu6; Vitale Brovarone et al. 2014). Oxygen fugacity  
400 is not quantified for all the examples so its role remains to be established.

401 Most of the examples gained Y, whereas P behavior was more variable. U was either  
402 immobile or was gained; both Th and U were added to two examples at pegmatite contacts (Eu5,  
403 Eu6).

#### 404 **Local redistribution of REE**

405 Local-scale transport of REE can also occur within a rock mass; two examples are shown  
406 in Figure 14. Average mass changes converge to around zero with a large standard deviation  
407 indicative of local redistribution. Fluid flow was highly channelized in both the subduction  
408 example (R1,2) and the granitic shear zone example (R3,4). For the subduction case, HREE loss  
409 due to reaction of pre-existing rutile and sphene was followed by HREE reincorporation into  
410 newly-grown garnet (Beinlich et al. 2010). For the shear zone, LREE losses were associated with  
411 alteration of protolith magmatic minerals whereas gains resulted from precipitation of new  
412 LREE-bearing phases (e.g., monazite, bastnäsite, aeschynite and tombarthite; Rolland et al.  
413 2003).

414

### 415 **Geochemical Relationships Among Elements**

#### 416 **Neighboring REE**

417 Neighboring REE, such as La-Ce, Sm-Nd, and Yb-Lu, are strongly positively correlated  
418 along 1:1 percentage mass change lines (Fig. 15). The LREE have the largest range of mass  
419 changes (Fig. 15a). The HREE have a notably smaller range but nonetheless can undergo  
420 considerable mobility (Fig. 15d). Europium mass change “anomalies” are clearly discernible  
421 when plotted with respect to neighboring Sm (Fig. 15c). Aside from this Eu behavior, a key point  
422 to emphasize is that adjacent REE are in most cases mobilized together. It follows that constancy  
423 of neighboring REE ratios does not mean that the elements were immobile.

#### 424 **Phosphorous and Yttrium**

425 Mass changes for La (a representative LREE) correlate positively with those for P (Fig.  
426 16a). Similar relations hold for REE<sub>Max</sub>-P, although there is somewhat more scatter (Fig. 16b).  
427 On the other hand, there is considerable scatter in HREE (Lu)-P mass changes, and the positive  
428 correlation between them is very weak at best (Fig. 16c). Mass change relations for P-Y are quite  
429 scattered as well, although there is a dispersed positive correlation (Fig. 16d). Positive  
430 correlations between Lu-Y and REE<sub>Max</sub>-Y, however, are much stronger, remarkably so for  
431 REE<sub>Max</sub>-Y (Figs. 16e, f).

432 Five samples fall off the REE<sub>Max</sub>-Y 1:1 correlation line (Fig. 16f). The REE<sub>Max</sub> elements  
433 for the four non-metacarbonate examples are in the LREE (e.g., La); they underwent large P  
434 gains as well. The LREE mass changes are less strongly linked to Y changes than the MREE or  
435 HREE. Two of these five nonetheless also gained significant Y. The metacarbonate sample  
436 (HREE3) has strong gains of HREE, Y, and P.

#### 437 **Thorium, Uranium, and Niobium**

438 Thorium is largely immobile or nearly so for most rocks (Figs. 17a, b). Only five  
439 examples display clear Th mass transfer. (1) A mylonite shear zone in a dynamothermal aureole

440 below an ophiolite remnant that channeled high- $T$  fluids (LREE6; Figs. 6, 7). (2, 3) Two  
441 examples of orthogneiss infiltrated by magmatic-hydrothermal fluids exsolved from pegmatite  
442 dikes (Eu6/HREE9 and Eu7; Figs. 12, 13). (4) Graphitic sillimanite-muscovite rock derived from  
443 metapelite adjacent to or incorporated into a pegmatite-quartz vein complex (LOSS2; Figs. 10,  
444 11). (5) High- $P$  (HP) or possibly ultrahigh- $P$  (UHP) eclogite facies veins derived from blueschist  
445 by prograde devolatilization (LOSS3; Figs. 10, 11). Glaucofane, dolomite, and sphene were  
446 replaced by omphacite, garnet, and rutile;  $\sim 75\%$  of the rock's total carbon was released as  $\text{CO}_2$  to  
447 infiltrating fluids (John et al. 2008). In summary, four of the five examples involve aureoles  
448 and/or magmatic-hydrothermal fluids, and the fifth is from a HP (UHP?) setting in a subduction  
449 zone.

450 A plot of Th-U mass changes shows that U is more mobile than Th such that U mass  
451 transfer can occur without mobilization of Th (Fig. 17a). This is consistent with experimental  
452 evidence showing higher U solubilities relative to Th in aqueous fluids given sufficiently  
453 elevated oxygen fugacity and/or salinity (e.g., Bailey and Ragnarsdottir, 1994; Bali et al. 2011).  
454 In the five examples for which Th was mobile, however, a very clear positive correlation of U-  
455 Th mass changes exists (Fig. 17a). Furthermore, for these rocks, the positive correlation between  
456 Th and  $\text{REE}_{\text{Max}}$  mass changes is also strong (Fig. 17b). Thus, Th mobilization did not occur  
457 without U and REE mobilization in the studied samples.

458 Mass changes for U and  $\text{REE}_{\text{Max}}$  tend to be associated, although there is considerable  
459 scatter (Fig. 17c). U and Nb mass changes appear to be roughly correlated (Fig. 17d); in the  
460 subduction rocks, this may reflect some control of U-Nb geochemical systematics by rutile (e.g.,  
461 Rudnick et al. 2000).

462

## 463 **Alkalis and Alkaline Earths**

464 Relationships among selected alkali and alkaline earth elements are shown in Figures 18–  
465 20. Loss of K together with Na is common (Fig. 18a). Coupled Na gain and K loss, as well as K  
466 gain and Na loss, are also observed. Remarkably, however, there are no examples of both  
467 significant Na and K gain. Overall, there are many more examples of K loss than K gain in the  
468 data set, although it is possible that this reflects some degree of sampling bias.

469 Geochemically, Ca commonly behaves similarly to Na giving rise to sodic-calcic  
470 metasomatism in crustal settings (e.g., Battles and Barton, 1995; de Jong and Williams, 1995;  
471 Ague, 1997). For non-subduction zone rocks, there is a general positive correlation between Na  
472 and Ca mass changes, reflecting the important control of plagioclase as a host for these elements  
473 (Fig. 18c). One low-grade metacarbonate example diverges strongly from the trend as it contains  
474 albite-rich feldspar (HREE3). As expected on geochemical grounds, Ca and Sr mass changes are  
475 positively correlated (Fig. 18e). Thus, although they differ in detail, Ca-K and Sr-Rb systematics  
476 resemble those for Na-K, with few or no examples of coupled Ca-K gain or Sr-Rb gain (Figs.  
477 18b, f).

478 Subduction zone rocks, on the other hand, display no Na-Ca correlations, although the  
479 lack of examples of simultaneous Na-Ca gain is conspicuous (Fig. 18d). In subduction settings,  
480 Na and Ca are not closely coupled in plagioclase, but instead are distributed across multiple  
481 phases (e.g., lawsonite, glaucophane, omphacite, garnet, epidote) facilitating a range of behavior.  
482 The subduction metacarbonate examples are distinct. Their strong Na-gains reflect growth of  
483 sodic phases such as glaucophane, omphacite, and pectolite (LM11, LM12, HREE8). Calcium  
484 loss in two of these examples resulted from stoichiometric carbonate mineral dissolution  
485 occurring simultaneously with silicate precipitation (Ague and Nicolescu, 2014). The dissolved

486 carbonate species are transported away from the site of reaction and may precipitate elsewhere in  
487 the subduction zone (e.g., Piccoli et al. 2016; Scambelluri et al. 2016). The liberated Ca could  
488 also potentially drive calcic metasomatism and the growth of Ca-bearing silicates including  
489 lawsonite (e.g., Alberto Brovarone et al. 2014) or clinopyroxene (e.g., LM10, in which  
490 glaucophane broke down and Ca-bearing sodic clinopyroxene grew).

491       Excellent positive correlations exist between mass changes for K and those for the  
492 geochemically-similar trace elements such as Rb hosted largely by micas and/or K-feldspar (Fig.  
493 19a). Correlations among K, Ba, and Cs are similar (not shown). Pb mass changes do not  
494 correlate well with these elements, but do show a positive relationship with Sr mass changes,  
495 indicating significant coupling of Sr-Pb behaviors (Fig. 19b). For non-subduction zone rocks,  
496 this is probably due to growth or breakdown of plagioclase. The large Pb gain recorded by LM8  
497 that falls off the general trend is inferred to be related to Pb incorporation into abundant sulfide  
498 minerals (mostly pyrite and chalcopyrite). The three examples plotting well above the trend are  
499 characterized by extensive lawsonite (Eu4, Eu5) or epidote metasomatism (LM1). REE<sub>Max</sub> mass  
500 changes (or those of individual REE) do not vary in any clear systematic way with Rb, Sr, K, Ba,  
501 Cs, Ca, or Na mass changes overall across the dataset (not shown).

502

### 503 **Volatiles, REE, and Alkalis**

504       REE<sub>Max</sub> mass changes relative to those of the major volatiles are shown in Figure 20a.  
505 The bulk of the volatile data are actually X-ray fluorescence loss on ignition values, which serve  
506 as proxies mostly for H<sub>2</sub>O and CO<sub>2</sub> contents (but also include organic matter oxidized during  
507 analysis and other complications). Data for actual H<sub>2</sub>O and CO<sub>2</sub> contents, as well as other  
508 volatiles including F, S, and Cl, are available for some examples but are not treated separately as

509 they are not determined in most analyses. The REE<sub>Max</sub>-volatile mass changes range widely and  
510 all fields of the mass change diagram are occupied. The patterns for La, Sm, and Lu differ, as  
511 expected given the differing mass changes for LREE, MREE, and HREE (Fig. 15), but  
512 systematic trends with volatile mass changes are not obvious (not shown).

513 Alkali metal behaviors relative to rock volatile gain or loss are shown in Figure 20b. For  
514 rocks that lost volatiles, only two main types of Na-K behavior are observed: 1) gain of Na and  
515 loss of K or 2) loss of both Na and K. For rocks that gained volatiles, gain of K and loss of Na is  
516 also possible, in addition to the two other types (Fig. 20b). As noted above, the lack of any  
517 examples that gained both K and Na is striking. Ca-K and Sr-Rb systematics are similar to those  
518 of Na-K (not shown). Figure 20b illustrates that K is generally lost when volatiles are lost (also  
519 the case for Ba, Cs, and Rb). Potassium and related elements can also be lost if volatiles are  
520 gained. Notably, however, volatile gains were required to add K to the rocks of the dataset,  
521 coupled to Na loss (Fig. 20b).

522

523

## Discussion

524 The mobilization of REE is always accompanied by the mobilization of an array of other  
525 non-REE major and trace elements. On the other hand, numerous examples exist of extensive  
526 mobilization of non-REE without significant open-system transport of the REE. The patterns of  
527 REE transport show no clear correlations with the general form of major element metasomatism,  
528 such as Si-Al, Mg-Fe, or alkali-alkaline earth metasomatism (Figs. 5, 7, 9, 11, 13). Of course, for  
529 individual examples, there is coupled mass transfer among elements, but with the exception of P  
530 and Y (see below) the overall lack of systematic relationships across the entire data set is  
531 striking.



532           The mobilization of REE is, however, a strong function of the growth or breakdown of  
533 REE-bearing phases. Given the wide range of rock types investigated, the number of key phases  
534 is notably small. Mass transfer involving LREE and MREE in the data set was dominated by  
535 reactions involving apatite, monazite, and allanite/epidote-group minerals; rarer phases including  
536 Ca-LREE-fluorocarbonates can also be important. Metasomatism involving plagioclase or  
537 lawsonite determines Eu mass change “anomalies” relative to neighboring REE. Metamorphic  
538 HREE systematics are controlled largely by garnet and xenotime. Contributions from  
539 hornblende, rutile, sphene, extremely REE-rich phases like tombarthite, and in rare cases even  
540 zircon are documented as well.

541           The REE are locally redistributed among phases, in closed or open systems, and can be  
542 added or lost from rocks during fluid-rock interaction. REE-bearing phases that precipitate take  
543 REE out of fluids, whereas those that breakdown release REE. One possibility is that the host  
544 phase is already present in the rock. For example, garnet that grows during infiltration could pick  
545 up HREE and Y and thus add them to the bulk rock (e.g., HREE1). Garnet breakdown would do  
546 the opposite (e.g., LREE6). Another possibility is that fluid-dominated metasomatism is so  
547 extensive that new REE-bearing phases are produced. Examples include the growth of xenotime  
548 associated with P and Y metasomatism and the consequent enrichment of HREE (HREE3), and  
549 the growth of monazite, bastnäsite, aeschynite and/or tombarthite in example LREE8.

550           REE and P mass changes are broadly correlated (Figs. 16a, b). This reflects the  
551 importance of phosphate minerals—mainly apatite, monazite, and xenotime—to REE  
552 metasomatism in many examples. Furthermore, REE<sub>Max</sub> mass changes for most rocks have an  
553 extremely strong positive correlation with Y mass changes (Fig. 16f). The excellent 1:1  
554 correlation reflects coupling of HREE or MREE mass changes to Y mass changes. Yttrium and

555 the HREE are hosted largely by the same phases (garnet, xenotime), leading to strong Y-HREE  
556 covariance. In addition, the ionic radius of  $Y^{3+}$  is very similar to the MREE (e.g.,  $Gd^{3+}$ ), yielding  
557 Y-MREE correlations. Five samples whose maximum REE gains are in the LREE fall off the 1:1  
558  $REE_{Max}$ -Y correlation line (Fig. 16f). This reflects the strong controls exerted by phases such as  
559 apatite and monazite on LREE budgets. Apatite and monazite can contain considerable Y, so Y  
560 metasomatism can take place concurrently; indeed, three of these five examples gained Y in  
561 addition to LREE.

562 Notably, the uncertainties on mean P mass changes are commonly larger than those for  
563 other elements (Fig. 16). Phosphorous mass changes that are statistically indistinguishable from  
564 zero but with very large error bars are consistent with significant local redistribution that did not  
565 necessarily transport P into or out of the rock. Interface coupled dissolution-reprecipitation (e.g.,  
566 Putnis and Austrheim, 2010; Putnis and John, 2010) is one process by which this could happen  
567 that has implications for the REE systematics of phosphates (e.g., Harlov et al. 2011). For  
568 example, dissolution of original grains would release phosphate-hosted REE to infiltrating fluids  
569 whereas reprecipitation could incorporate a new suite of REE depending on fluid chemistry.

570 The REE are typically fractionated by metasomatism. More or less equal amounts of REE  
571 mass transfer are evident for some REE loss examples, but are otherwise rare. Mobility of just  
572 one or several REE is also uncommon; it is only observed herein for an example of Eu  
573 sequestration in plagioclase (Eu1). The fractionations reflect the preferences that the dominant  
574 REE-host minerals have for particular REE or suites of REE, such as garnet for HREE,  
575 plagioclase for  $Eu^{2+}$ , and so on.

576 REE metasomatism shows no simple correlations with major volatile ( $H_2O$ ,  $CO_2$ ) loss or  
577 gain, but rather a range of behaviors (Fig. 20a). Coupled volatile addition and REE loss suggests

578 leaching by infiltrating fluids. Loss of fluids carrying REE during prograde devolatilization may  
579 have resulted in combined REE and volatile loss. In addition, prograde volatile loss can be  
580 driven by infiltration; this could have either removed REE, or added them depending on fluid  
581 composition and mineralogy. A number of rocks also underwent REE mass transfer with  
582 relatively little overall volatile mass change (Fig. 20a), so in some cases fluids move in and out  
583 of rocks transporting REE without dramatically altering rock volatile contents.

584         The lack of clear correlations between REE mass changes and mass transfer of major  
585 volatiles (H<sub>2</sub>O, CO<sub>2</sub>) or most non-REE elements (except P) suggests that mechanisms of REE  
586 transport differ to some degree from those controlling other mobile constituents. Many REE  
587 carriers have been shown or postulated to be in fluids, including F, Cl, carbonate, hydroxide, and  
588 sulfate species; other variables including oxygen fugacity (for Eu) and pH can also play  
589 important roles (e.g., Michard and Albarède, 1986; Gieré, 1990; Bau, 1991; Gruau et al. 1992;  
590 Gammons et al. 1996; Pan and Fleet, 1996; Bao et al. 2008; Harlov, 2012; Manning and  
591 Aranovich, 2014). For example, LREE gain in example LREE1 was the result of Ca-REE-  
592 fluorocarbonate precipitation, indicating a likely role for F and carbonate anions in REE  
593 transport. Halogen carriers are commonly inferred, but given that Cl complexes can transport  
594 major elements including alkalis and alkaline earths, it is somewhat puzzling that clear,  
595 systematic relationships among these elements and the REE are not evident in the data set. In  
596 addition to the above possibilities, observed correlations between REE and P and/or Y mass  
597 changes could be due to REE transport by P and/or Y complexes, or transport of these elements  
598 together by some other complexing agent (e.g., Gieré, 1990; Lee and Byrne, 1992; Ague, 2003;  
599 Carson and Ague, 2008).

600 It is interesting to note that the strong HREE loss for the charnockite example (HREE10)  
601 differs from the bulk of the data set (Fig. 8). This may reflect fluids with reduced water activity  
602 due to the role of CO<sub>2</sub>, brines, and/or melt in charnockite and granulite genesis (e.g., Newton et  
603 al. 1980; Valley et al. 1990; Harlov, 2012; Rajesh and Santosh, 2012; Manning and Aranovich,  
604 2014), as distinct from predominantly aqueous metasomatic fluids for the other rocks.

605 Major gaps in knowledge remain regarding how dissolved REE are transported in fluids  
606 that need to be addressed by new field, fluid inclusion, experimental, and theoretical studies.

607

## 608 **Implications**

### 609 **Implications for High Field Strength Elements**

610 The data set has clear implications for the mobility of REE and high field strength  
611 elements (HFSE). The conventional wisdom is that the REE are relatively immobile in  
612 metamorphic fluids. This is indeed the case for roughly 1/3 of the data set. As these examples are  
613 all from high-flux environments which mobilized other elements, one would expect little REE  
614 mobility in lower flux, less channelized environments as well. Some high flux environments,  
615 however, are conducive to REE transport, and examples are shown for all settings considered  
616 including Barrovian and subduction zone metamorphism. Moreover, under such circumstances,  
617 the REE are typically the most mobile of the elements traditionally considered to be refractory  
618 including the HFSE (Fig. 21).

619 On a percentage mass change basis, these elements listed in order of decreasing mobility  
620 are: REE>U>Nb>Ti>Th~Zr (Fig. 21). This mobility of REE is broadly consistent with recent  
621 experimental results relevant for subduction (Tsay et al. 2014), but more work is clearly needed  
622 across a range of settings. Uranium mobility is no doubt influenced by redox state, with higher

623 solubilities expected for more oxidizing conditions (e.g., Bailey and Ragnarsdottir, 1994; Bali et  
624 al. 2011). Nb displays less, but still considerable mobile behavior, in accord with experimental  
625 evidence (particularly in halogen-rich fluids; e.g., Tanis et al., 2015). Rutile, a major host for Nb,  
626 has been observed in veins and other metasomatic structures (e.g., Ague, 2003; Gao et al. 2007 ;  
627 Spandler et al. 2011). Care must be exercised, however, to determine if the Ti and Nb were  
628 transported over great distances, or locally from wallrock to vein (e.g., Ague, 2014).

629         At the lowest mobility end of the spectrum, Th and Zr are largely immobile, at least in  
630 the water-dominated fluids studied herein. The lower degrees of mobility of Zr or Th relative to,  
631 e.g., Ti, is well established by experiment (e.g., Hermann and Spandler, 2008; Tsay et al. 2016).  
632 It is true that Zr and/or Th are used as reference frame elements in quite a few of the studies, but  
633 they are not used in all of them and, furthermore, Zr-Th behavior is in most cases strongly  
634 coupled which would not be expected if one or both were mobile. The much larger degree of U  
635 mobility relative to Th can account for at least some of the isotope disequilibrium commonly  
636 observed in arcs, including Tonga-Kermadec, Mariana, and Sunda (e.g., Turner et al. 2001). The  
637 largest degrees of element mobility recognized herein for Th and Zr, as well as other HFSE, are  
638 found in magmatic/magmatic hydrothermal settings as well as the dynamothermal shear zone  
639 aureole, charnockite, and HP (UHP?) rocks (Fig. 17a). The well-known mobility of, for example,  
640 Th and U from charnockite and other granulite facies terranes is consistent with this pattern (e.g.,  
641 Rollinson and Windley, 1980; Hansen and Harlov, 2007; Raith et al. 2016).

642         Notably, one example of HP (UHP?) rock shows considerable HFSE depletion (LOSS3,  
643 Table 1; John et al. 2008). It is tempting to hypothesize that this mobility could have been aided  
644 by solution of refractory phases into supercritical fluids intermediate between aqueous fluids and

645 hydrous melts at extreme pressures (e.g., Manning 2004; Ferrando et al. 2005; Hermann et al.  
646 2013).

#### 647 **Implications for Garnet and the HREE**

648 Garnet has considerable potential to sequester HREE and Y, incorporating these elements  
649 from fluids during growth. Garnet is also nearly ubiquitous in medium and high grade regional  
650 metamorphic settings. These relationships imply that fluids exiting such settings ascending into  
651 shallower parts of the lithosphere will carry with them a geochemical fingerprint depleted in the  
652 HREE and Y. Although speculative, perhaps this helps explain why many REE ores are  
653 noticeably depleted in the HREE (e.g., Long et al. 2010).

#### 654 **Implications for LILE and Alkali Systematics**

655 Coupled gain of K and loss of Na is observed mainly for rocks that gained volatiles,  
656 consistent with retrograde down-temperature (down-*T*) flow which tends to add K and remove  
657 Na (e.g., Brimhall, 1977; Dipple and Ferry, 1992), or infiltration of fluids from K-rich sources  
658 (e.g., pegmatites, micaceous metasediments) (Fig. 20b). Coupled K gains and Na losses are  
659 observed for down-*T* flow in the dynamothermal aureole shear zone (LREE6), flow of  
660 magmatic-hydrothermal fluid from K-rich pegmatite into feldspathic orthogneiss (Eu7),  
661 retrogression of eclogite to blueschist (LOSS4), and replacement of plagioclase by K-feldspar in  
662 charnockite (Eu8/LOSS5/HREE10).

663 Leaching of both K and Na are possible during fluid-rock interaction associated with  
664 either prograde devolatilization, or volatile gains. In either case, large fluid fluxes are likely  
665 required to drive the alkali metasomatism (e.g., Yardley, 1986; Bucholz and Ague, 2010; Ague,  
666 2011).

667           Importantly, if K and related elements are mobilized during prograde devolatilization,  
668 then they are lost, reflecting transport of these elements out of rocks by escaping fluids (Fig. 20b)  
669 (e.g., Melzer and Wunder, 2000), up-*T* fluid flow that also produces Na ± Ca gains (e.g., Dipple  
670 and Ferry, 1992; Ague, 1997), strong alkali leaching (e.g., Yardley, 1986), and/or flow across  
671 lithologic contacts (e.g., Ague, 2003; Breeding et al. 2004; Ague and Nicolescu, 2014; Galvez et  
672 al. 2013) (Fig. 20b). Examples include the destruction of micas and growth of plagioclase (e.g.,  
673 Eu1, HREE3) or, in subduction zones, the destruction of phengite and the growth of glaucophane  
674 and/or omphacite (e.g., LM11, LM12, HREE8).

#### 675 **Comments on LILE and HFSE in Arcs**

676           An important general feature of arc magmas are the elevated concentrations of LILE  
677 relative to HFSE. In particular, the relative “depletion” of Nb relative to K and related LILE has  
678 been extensively studied (e.g., Kelemen et al., 1993). Fluids tend to remove K, Rb, Ba, and Cs  
679 during prograde devolatilization in high flux zones, and thus have the potential to deposit these  
680 elements elsewhere as they migrate along flow paths. As a consequence, fluids ascending from  
681 subducted crust (mafic or metasedimentary) are likely an important source of LILE enrichment  
682 in arcs (e.g., Stolper and Newman, 1994; Noll et al. 1996; Schmidt, 1996; Johnson and Plank,  
683 1999; Becker et al. 2000). Furthermore, these fluids would in general lack HFSE including Nb,  
684 Ti, Th, and Zr, consistent with the observed relative depletions in these elements.

685           At higher temperatures involving partial melting, HFSE including Th are mobilized (e.g.,  
686 Johnson and Plank, 1999). Zircon solubility in silicate melts is considerable, leading to increased  
687 zircon destabilization as temperature increases (e.g., Boehnke et al. 2013). Moreover, Zr  
688 partitions into coexisting rutile with increasing temperature (e.g., Zack et al. 2004; Watson et al.  
689 2006; Tomkins et al. 2007), further destabilizing zircon. Indeed, at ultrahigh-temperature (UHT)

690 conditions in metasedimentary systems, zircon can be completely destroyed (e.g., Kelsey and  
691 Powell, 2011). This is likely the case in more mafic bulk compositions as well, but more work  
692 remains to verify this. Because rutile, a major Nb host, can remain stable during partial melting,  
693 the Nb and Ti anomalies relative to LILE in arcs could be preserved, even though HFSE  
694 associated with zircon (e.g., Zr, Th, Hf) could be strongly mobilized.

### 695 **Implications for Geochronology**

696 The mobility of Nd, Sm, and Lu has implications for geochronology (Hf data are missing  
697 for many examples and are not considered further). Significant mass changes for Sm and/or Nd  
698 are present in about 50% of the data set examples; there are fewer examples of Lu changes but  
699 mobilization is clearly possible (Figs. 15b, d). Garnet-whole rock Sm/Nd or Lu/Hf  
700 geochronology could be affected if, for example, matrix REE values were modified after garnet  
701 growth. Whole-rock isochrons (e.g., O'Neil et al. 2012) could be disrupted by scatter or other  
702 irregularities if some rocks underwent open-system REE transport and others did not. Pervasive  
703 REE metasomatism can take place in large scale, shallow hydrothermal systems (e.g., Hellman et  
704 al. 1979; Windrim et al. 1984; Bartley, 1986; Gruau et al. 1992; Marsh et al. 2012). But at higher  
705 grades of metamorphism at greater depths, mobility appears to be more localized around, for  
706 example, high fluid flux conduits, as discussed herein. Nonetheless, diffusional exchange of Nd  
707 and other REE can take place across lithologic contacts, even in environments that did not  
708 undergo large fluxes (e.g., Watson and Baxter, 2007). Regardless of mechanism, in most cases  
709 Sm and Nd will not be differentially mobilized, such that bulk Sm/Nd ratios remain relatively  
710 constant even though considerable mass transfer of both elements may have occurred (Fig. 15b).

### 711 **Implications for Petrotectonic Geochemical Discrimination Diagrams**



712 Geochemical discrimination diagrams have proven to be invaluable for deducing tectonic  
713 settings of magmatism (e.g., Pearce and Cann, 1973; Pearce et al. 1984). Many of these diagrams  
714 focus on HFSE, due to their general resistance to alteration. As HFSE can be mobile, however, it  
715 is illustrative to see what impact this would have on interpretations of tectonic settings. Consider  
716 two examples: the Nb-Zr-Y diagram for mafic rocks (Meschede, 1986), and the Nb-Y diagram  
717 for granitic rocks (Pearce et al. 1984). The concentrations of Nb and Y were varied within mass  
718 change ranges compatible with Figure 21; Zr was taken to be immobile. Figure 22 shows that  
719 this level of variation can skew considerably the field a given rock will plot in. Plotting positions  
720 can shift over two, three, or more fields. These results illustrate that immobility of HFSE cannot  
721 be taken for granted in petrotectonic studies; this fact becomes increasingly evident as the body  
722 of quantitative mass balance studies grows. Appropriate criteria for selecting the least altered  
723 rocks are essential to implement, as emphasized by, for example, Pearce (1996).

#### 724 **Implications for Major Element Metasomatism During Regional Metamorphism of Pelites**

725 Given the remarkable utility of metapelitic mineral assemblages to elucidate  
726 metamorphic grade (Barrow, 1893; 1912; Thompson, 1957), the chemical evolution of  
727 metapelites during progressive Buchan or Barrovian metamorphism has long been a topic of  
728 interest (e.g., Shaw, 1956). It remains relevant today, as pseudosections are typically constructed  
729 for systems closed to non-volatile major element mass transfer. Nonetheless, focused flow in and  
730 around veins may result in alkali and alkaline earth mass transfer that can stabilize index  
731 minerals such as garnet (Ague, 1997) or staurolite and kyanite (Ague, 1994b, 2011; Bucholz and  
732 Ague, 2010). Based on a statistical examination of over 350 rock chemical analyses from the  
733 literature, Ague (1991) concluded that during progressive metamorphism: (1) in addition to  
734 volatiles, significant silica may also be lost; (2) K is commonly lost; and (3) Na and Ca can be

735 mobile and be either gained or lost. The data set of the current study is useful to test these  
736 findings as it is completely independent of the 1991 data set noted above.

737 Figure 23a confirms the silica loss, and also shows that increasing silica loss is correlated  
738 with increasing volatile loss. Even if REE were immobile, they can still undergo considerable  
739 concentration increases due to silica loss (Fig. 2). Previous studies have demonstrated that much  
740 or all of this silica loss is local, being the result of mass transfer down chemical potential  
741 gradients from altered wallrock selvages to adjacent veins. Vein mass balance, however, clearly  
742 indicates that more silica is in the veins than can be accounted for by local loss, demonstrating a  
743 role for fluid flow in precipitating exogenous silica. Such silica precipitation will in general  
744 result from down-temperature flow (Ferry and Dipple, 1991). The fraction of external silica has  
745 been found to be in the range of ~30 to >90% (Ague, 1994, 2011; Penniston-Dorland and Ferry,  
746 2008; Bucholz and Ague, 2010). Silica can thus be added overall at the outcrop or larger scales  
747 (e.g., Ferry, 1992; Breeding and Ague, 2002). If, however, the amount of veining is small, it  
748 follows that overall silica additions are modest (e.g. Dasgupta et al. 2009). In addition, one  
749 example in the data set gained considerable silica and volatiles (LREE1). It is a heavily altered  
750 and silicified wallrock inclusion directly within a vein; it records large fluxes of fluid capable of  
751 precipitating exogenous silica.

752 The loss of K and the variable loss or gain of Na and Ca are evident in Figures 23b and  
753 23c. To first order, these systematics reflect destruction of micas (Fig. 23b) and Na-Ca mass  
754 transfer largely controlled by plagioclase (Fig. 23c). The example of Ca gain and Na loss  
755 (HREE1) which plots off the general trend in Figure 23c reflects destruction of feldspar together  
756 with uptake of Ca by garnet. The mobility of these elements can, if large enough, help to  
757 determine the stability of aluminous phases including chloritoid, garnet, staurolite, and kyanite

758 relative to micas and feldspars. Behavior can vary widely; in the low-pressure region of the  
759 Barrovian zones, Na-Ca metasomatism, K loss, mica destruction, and feldspar growth was  
760 widespread (e.g., LM1–LM3; LREE1). On the other hand, in some other Barrovian settings (e.g.,  
761 Unst, Shetland Islands, Scotland; Wepawaug Schist, Connecticut) alkali loss and variable Ca  
762 behavior are typically observed (e.g., LREE3, LREE4; HREE1). The reasons for these  
763 differences remain as important research questions; it is interesting to note in this context that the  
764 two latter examples are higher-pressure (~0.8-1.0 GPa) than the first (~0.4-0.5 GPa).

765         The above are broad patterns, but other behavior is of course possible, such as K gain due  
766 to large down-temperature fluid fluxes (e.g., van Haren et al. 1996). Moreover, a wide array of  
767 other elements can be mobilized, including Al, which can be added to vein-selvage systems (e.g.,  
768 Figs 7, 9). Regardless of the elements mobilized, the most pronounced effects are found adjacent  
769 to conduits such as veins. In regions impoverished in fluid flow due to channelization, or in  
770 settings very distal to conduits, much less transfer would be expected. Interestingly, regions of  
771 low fluid flux resulting from channelization will tend to border conduits, leading to the non-  
772 intuitive conclusion that regions of highest and lowest fluxes may be surprisingly closely  
773 associated spatially (Fig. 1). Consequently, inferences about mass transfer will be highly  
774 dependent on where samples are collected in the flow system.

### 775 **Implications for Carbonate Dissolution and Precipitation**

776         Carbon dioxide loss is classically associated with devolatilization reactions such as:  
777 calcite + quartz = wollastonite + CO<sub>2</sub>. More recently, it has also been recognized that  
778 stoichiometric or near stoichiometric carbonate dissolution (e.g., Frezzotti et al. 2011; Ague and  
779 Nicolescu, 2014) or precipitation (e.g., Piccoli et al. 2016; Scambelluri et al. 2016) of carbonate  
780 phases may occur. In metacarbonate rocks in which calcite or aragonite are the dominant

781 carbonate phase, it turns out that stoichiometric carbonate gains or losses will plot near the 1:1  
782 line on a Ca-volatile percentage mass change diagram, facilitating distinction of volatile loss  
783 behavior.

784 Three types of behavior are shown on Figure 23d. A field for devolatilization is clearly  
785 evident, highlighting the important role played by this process. On the other hand, two  
786 subduction zone metacarbonate examples plot near the 1:1 mass change line and underwent  
787 extensive  $\text{CaCO}_3$  dissolution (LM11, HREE8). Finally, a syn-metamorphic carbonate-bearing  
788 vein from the regional greenschist facies records  $\text{CaCO}_3$  precipitation (HREE3). It is tempting to  
789 speculate that carbonate dissolved at deeper levels of the metamorphic system was redeposited at  
790 shallower levels in this vein. Going forward, the answers to these and other questions relating to  
791  $\text{CO}_2$  mass transfer will be critical to resolve in order to obtain a deeper understanding of deep  
792 carbon cycling in the lithosphere.

793

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818

## Appendix

819 The following is a brief summary of the mass balance equations describing the chemical  
820 and volumetric changes of metasomatism. The reader is referred to Gresens (1967), Grant  
821 (1986), Brimhall and Dietrich (1987), Brimhall et al. (1988), Ague (1994a), Baumgartner and  
822 Olsen (1995), Ague and van Haren (1996), Ague (2011), and Durand et al. (2015) for discussion  
823 of various aspects of these equations. The mass change of a mobile constituent can be calculated  
824 using equation (6) for a general system with specified geochemical reference frame, equation  
825 (10) for a constant volume system, or equation (16) for a constant mass system.

### 826 General System

827 The basic mass balance expression relevant for describing chemical changes in rocks is  
828 (e.g., Brimhall et al. 1988; Ague, 1994a):

$$829 \quad V^0 \rho^0 C_i^0 = V' \rho' C_i', \quad (1)$$

830 in which  $V$  and  $\rho$  are rock volume and density respectively (including pore space),  $C_i$  is the  
831 concentration of a reference (immobile) species  $i$  defining the geochemical reference frame  
832 (mass units), and the  $^0$  and  $'$  superscripts refer to the initial and final (altered) states. The volume  
833 strain  $\varepsilon$  is given by:

$$834 \quad \varepsilon = \frac{V' - V^0}{V^0}. \quad (2)$$

835 Substitution of equation (2) into equation (1) and rearranging for volume strain yields:

$$836 \quad \varepsilon_i = \left( \frac{C_i^0}{C_i'} \right) \left( \frac{\rho^0}{\rho'} \right) - 1, \quad (3)$$

837 in which  $\varepsilon_i$  denotes that strain is being computed on the basis of reference element  $i$ . The  
838 fractional change in mass for some mobile constituent  $j$  ( $\tau^j$ ) is given by:

$$839 \quad \tau^j = \frac{V' \rho' C_j' - V^0 \rho^0 C_j^0}{V^0 \rho^0 C_j^0} = \frac{V' \rho' C_j'}{V^0 \rho^0 C_j^0} - 1. \quad (4)$$

840 Noting that  $\varepsilon_i + 1 = V' / V^0$ , substitution yields:

$$841 \quad \tau_i^j = \frac{\rho'}{\rho^0} \frac{C_j'}{C_j^0} (\varepsilon_i + 1) - 1, \quad (5)$$

842 providing a quantitative relationship between changes in rock chemical and physical properties.

843 This expression may be simplified further by substituting explicitly for volume change:

$$844 \quad \tau_i^j = \left( \frac{C_i^0}{C_i'} \right) \left( \frac{C_j'}{C_j^0} \right) - 1. \quad (6)$$

845 The total change in rock mass  $T^m$  is:

846 
$$T^{rm} = \frac{V'\rho' - V^0\rho^0}{V^0\rho^0} = \frac{V'\rho'}{V^0\rho^0} - 1. \quad (7)$$

847 From equation (1), the right side of equation (7) can be recast to give:

848 
$$T_i^{rm} = \left( \frac{C_i^0}{C_i'} \right) - 1. \quad (8)$$

849 Equations (3), (6), and (8) give fractional changes; percentage changes are obtained by  
850 multiplying by 100 (gains are positive, losses negative).

### 851 **Constant Volume or Known Volume Change System**

852 For a system with no volume change, the basic mass balance of equation (1) simplifies to:

853 
$$\rho^0 C_i^0 = \rho' C_i'. \quad (9)$$

854 The mass change for a mobile constituent  $j$  is easily obtained from equation (5) by setting the  
855 volume strain to zero:

856 
$$\tau_i^j = \frac{\rho'}{\rho^0} \frac{C_j'}{C_j^0} - 1. \quad (10)$$

857 As final and initial volumes are equal, the total rock mass change in a constant volume system is  
858 controlled by the density ratio:

859 
$$T^{rm} = \frac{V'\rho' - V^0\rho^0}{V^0\rho^0} = \frac{\rho'}{\rho^0} - 1. \quad (11)$$

860 If the volume change is non-zero, but known, then the change in mass of a mobile  
861 constituent  $j$  can be calculated from equation (5) by substituting in the known value of volume  
862 strain. To calculate the total mass change, we begin by rearranging equation (2):

863 
$$V' = \varepsilon V^0 + V^0. \quad (12)$$

864 Substituting this expression for  $V'$  into equation (7) and rearranging gives the expression for the  
865 total mass change in a system with known, non-zero volume change:

866 
$$T^{rm} = \frac{(\varepsilon + 1)\rho' - \rho^0}{\rho^0}. \quad (13)$$

867 If the volume change is zero, this expression reduces to equation (11).

### 868 **Constant Mass or Known Mass Change System**

869 In systems with no overall rock mass change, the concentrations of a reference species in  
870 the initial and final states are equal. Thus, equation (1) simplifies to:

871 
$$V^0\rho^0 = V'\rho'. \quad (14)$$

872 It follows that the volume strain expression of equation (3) can be written as follows for a  
873 constant mass system:

874 
$$\varepsilon = \frac{\rho^0}{\rho'} - 1. \quad (15)$$

875 Rearranging this expression to give  $\varepsilon + 1$  and then substituting the equivalent density ratio into  
876 equation (5) yields the mass change for mobile constituent  $j$ :

877 
$$\tau^j = \frac{C'_j}{C_j^0} - 1. \quad (16)$$

878 As there is no overall rock mass change,  $T^{rm} = 0$ .

879 In constant mass systems, 1 kg of initial rock has the same mass of 1 kg after alteration.

880 Note, however, that considerable changes in the masses of *individual* elements may have taken  
881 place. The requirement is that the amount of mass lost must equal the amount gained, in order to  
882 keep the overall mass constant.

883 If the total change in rock mass  $T^{rm}$  is non-zero, but known, then equation (8) for the total  
884 rock mass change can be rearranged and substituted into equation (3) for volume strain to yield:

885 
$$\varepsilon = \frac{\rho^0}{\rho'} (T^{rm} + 1) - 1. \quad (17)$$



886 Similarly, substitution of equation (8) into equation (6) yields the following expression for the  
887 mass change of a mobile constituent in a system with known overall mass change  $T^{rm}$ :

$$888 \quad \tau^j = \left( \frac{C_j'}{C_j^0} \right) (T^{rm} + 1) - 1. \quad (18)$$

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1261

## Figure Captions

1262

**Figure 1.** Fluid fluxes calculated from numerical model of flow region cut by two high

1263

permeability conduits. Flow is upward, and conduits are 100 times more permeable than matrix.

1264

Note that focusing of flow into the conduits drastically reduces fluxes in adjacent areas, such that

1265

high and low flux zones are closely associated spatially. Modified from Breeding et al. (2003;

1266

see this paper for calculation details).

1267

**Figure 2.** REE concentration increase in a vein selvage system in which REE were immobile.

1268

Loss of silica from the selvage to an adjacent quartz vein caused residual enrichment of REE

1269

with no REE mass transfer into or out of the system. Data from Penniston-Dorland and Ferry

1270

(2008), site S35-1G (LM4 in Table 1).

1271

**Figure 3.** Geochemical profiles illustrating changes in REE ratios due to REE mobility. **(a)**

1272

Increase in La/Lu in selvage adjacent to a quartz-rich vein cutting metapelitic rock. Increase

1273

reflects strong La gains (site 181 data from Bucholz and Ague, 2010; LREE3 in Table 1). Plus

1274

signs and circles denote least altered and more altered metacarbonate rocks, respectively. **(b)**

1275

Decrease in Eu/Sm ratio in metacarbonate rock toward veined lithologic contact. Both Eu and

1276

Sm were gained, but Sm was gained to a greater extent approaching the veined contact (site 187

1277

data from Ague, 2003; Eu3 in Table 1). Plus signs: least altered metacarbonate rock; light green

1278

circles: diopside-zoisite metacarbonate; dark green circles: hornblende-zoisite metacarbonate;

1279

black square: vein contact zone.

1280

**Figure 4.** Percentage mass change diagrams for examples that underwent little or no REE mass

1281

transfer. Note that these are not chondrite-normalized plots, but instead depict elemental

1282

percentage mass changes for altered rocks relative to their precursors (mass gains are positive,

1283

losses negative). See Table 1 for rock type descriptions.

1284 **Figure 5.** Bar code diagrams illustrating mass changes for examples that underwent little or no  
1285 REE metasomatism. See Table 1 for rock type descriptions.

1286 **Figure 6.** Percentage mass change diagrams for examples that underwent LREE mass transfer.  
1287 Dominant phases controlling REE mass changes are labeled (if known). LREE1 shows mass  
1288 transfer for two silicified metapelite wallrock inclusions in vein. LREE3 samples 181Aii and  
1289 181Aiii are 1.25 cm and 3.75 cm from vein contact, respectively. See Table 1 for rock type  
1290 descriptions.

1291 **Figure 7.** Bar code diagrams illustrating mass changes for examples that underwent LREE  
1292 metasomatism. See Table 1 for rock type descriptions.

1293 **Figure 8.** Percentage mass change diagrams for examples that underwent HREE mass transfer.  
1294 Dominant phases controlling REE mass changes are labeled (if known). HREE3 samples 197Ai,  
1295 197AS, and 197Aii are 0 cm, 1 cm, and 3 cm from vein contact, respectively. See Table 1 for  
1296 rock type descriptions.

1297 **Figure 9.** Bar code diagrams illustrating mass changes for examples that underwent HREE  
1298 metasomatism. See Table 1 for rock type descriptions.

1299 **Figure 10.** Percentage mass change diagrams for examples that underwent overall REE loss.  
1300 Dominant phases controlling REE mass changes are labeled (if known). See Table 1 for rock  
1301 type descriptions.

1302 **Figure 11.** Bar code diagrams illustrating mass changes for examples that underwent overall  
1303 REE loss. See Table 1 for rock type descriptions.

1304 **Figure 12.** Percentage mass change diagrams for examples that underwent Eu mass transfer.  
1305 Dominant phases controlling REE mass changes are labeled (if known). Eu3 samples 187Bii,

1306 187Biii, and 187Mi are 2 cm, 5 cm, and 10 cm from vein contact, respectively. See Table 1 for  
1307 rock type descriptions.

1308 **Figure 13.** Bar code diagrams illustrating mass changes for examples that underwent Eu  
1309 metasomatism. See Table 1 for rock type descriptions.

1310 **Figure 14.** Percentage mass change diagrams for examples characterized by local REE  
1311 redistribution. Dominant phases controlling REE mass changes are labeled (if known). Samples  
1312 JTS-C (R1) and JTS-E (R2) are from blueschist-eclogite transition zone approximately 85 cm  
1313 and 40 cm from vein contact, respectively (see Fig. 2 in Beinlich et al. 2010). For the shear zones  
1314 cutting granitic rocks, MB69 (R3) is in the site of LREE deposition in the shear zone core, and  
1315 MB71 (R4) is in the area of LREE leaching several meters away (see Fig. 6 in Rolland et al.  
1316 2003). R3 and R4 not shown on subsequent diagrams due to lack of full major element data in  
1317 original paper. See Table 1 for rock type descriptions and references.

1318 **Figure 15.** Mass changes for neighboring REE. Note that on this and most subsequent  $x$ - $y$  plots,  
1319 the scaling for negative mass changes (mass losses) has been expanded to make visualization  
1320 easier. This is useful because mass losses are bounded at -100%, whereas mass gains have no set  
1321 upper bound. **(a)** Ce-La. **(b)** Sm-Nd. **(c)** Eu-Sm. Examples that plot off the 1:1 trend have Eu  
1322 “anomalies” relative to neighboring REE. **(d)** Lu-Yb.

1323 **Figure 16.** Mass changes involving P and Y. **(a)** La-P. **(b)** REE<sub>max</sub>-P. REE<sub>Max</sub> denotes the  
1324 maximum mass change for any REE in the rock. **(c)** Lu-P. **(d)** P-Y. **(e)** Lu-Y. **(f)** REE<sub>max</sub>-Y.

1325 **Figure 17.** Mass changes involving Th and U. **(a)** Th-U. **(b)** Th- REE<sub>max</sub>. **(c)** U- REE<sub>max</sub>. **(d)** U-  
1326 Nb.

1327 **Figure 18.** Mass changes involving alkalis and alkaline earths. **(a)** Na-K. Note absence of  
1328 examples that gained both Na and K. **(b)** Ca-K. Note absence of examples that gained both Ca

1329 and K. **(c)** Na-Ca relations for non-subduction zone rocks. **(d)** Na-Ca relations for subduction  
1330 zone rocks. **(e)** Ca-Sr. **(f)** Sr-Rb. Note rarity of coupled Sr and Rb gain, and similarity to parts (a)  
1331 and (b).

1332 **Figure 19.** **(a)** Mass changes for Rb and K. **(b)** Mass changes for Sr and Pb.

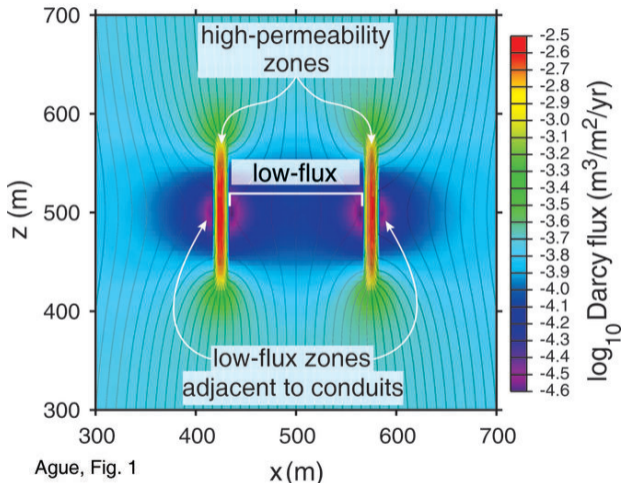
1333 **Figure 20.** Mass changes in involving volatiles, using LOI as a proxy for volatile content. **(a)**  
1334 REE<sub>max</sub>-volatiles. **(b)** Na versus volatile mass changes, contoured for K mass changes. Note: (1)  
1335 examples that gained K are concentrated in the field that lost Na and gained volatiles and (2) the  
1336 general loss of K from all other examples.

1337 **Figure 21.** Comparison of percentage mass changes for HFSE.

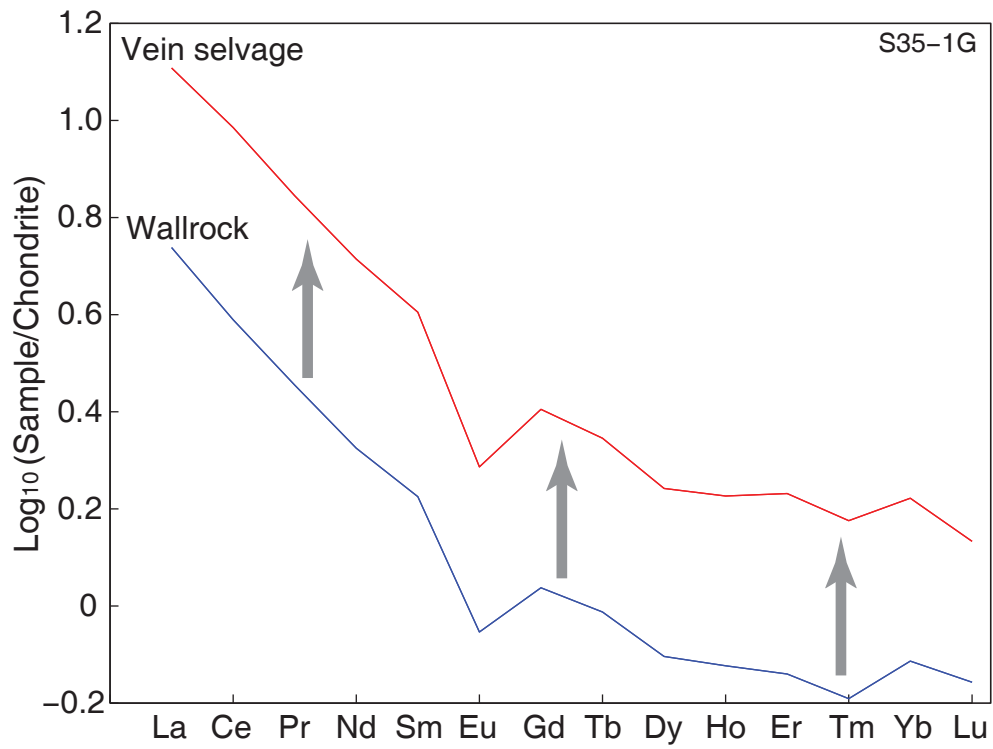
1338 **Figure 22.** Examples of how HFSE mass changes could impact plotting on petrotectonic  
1339 discrimination diagrams. Concentrations of Nb and Y varied to be compatible with Fig. 21. The  
1340 orange ellipses illustrate mass changes of: Y±50% and Nb±25%. The green ellipses illustrate  
1341 somewhat larger mass changes of: Y±75% and Nb ±50%. **(a)** Nb-Zr-Y diagram for mafic rocks  
1342 (Meschede, 1986). **(b)** Nb-Y diagram for granitic rocks (Pearce et al. 1984).

1343 **Figure 23.** Mass changes for Barrovian metapelites and metacarbonate rocks from a range of  
1344 settings. **(a)** Si-volatile mass changes for metapelites. Note coupled loss of silica and volatiles,  
1345 illustrating dependence of silica loss on extent of devolatilization. Example that gained silica and  
1346 volatiles is a silicified wallrock inclusion in a quartz vein (LREE1). **(b)** K-Rb mass changes for  
1347 metapelites. Note preponderance of samples characterized by K-Rb loss. **(c)** Na-Ca mass changes  
1348 for metapelites. General positive correlation due to controls on mass transfer by plagioclase  
1349 feldspar. **(d)** Ca-volatile mass changes for all the metacarbonate rocks of the data set. Examples  
1350 that plot close to the 1:1 line characterized by near stoichiometric dissolution or precipitation of

1351 CaCO<sub>3</sub>. Tan-colored field denotes rocks that lost more CO<sub>2</sub> than Ca, illustrating the role of  
1352 devolatilization.  
1353

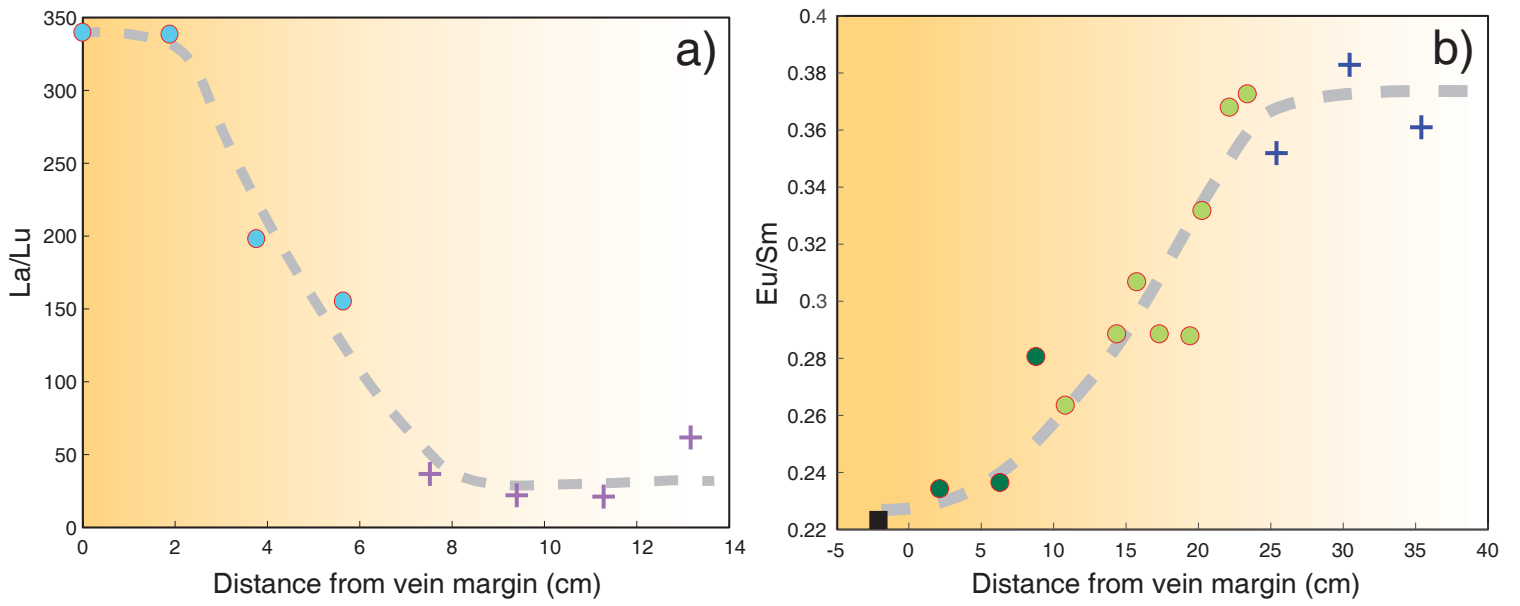


Ague, Fig. 1

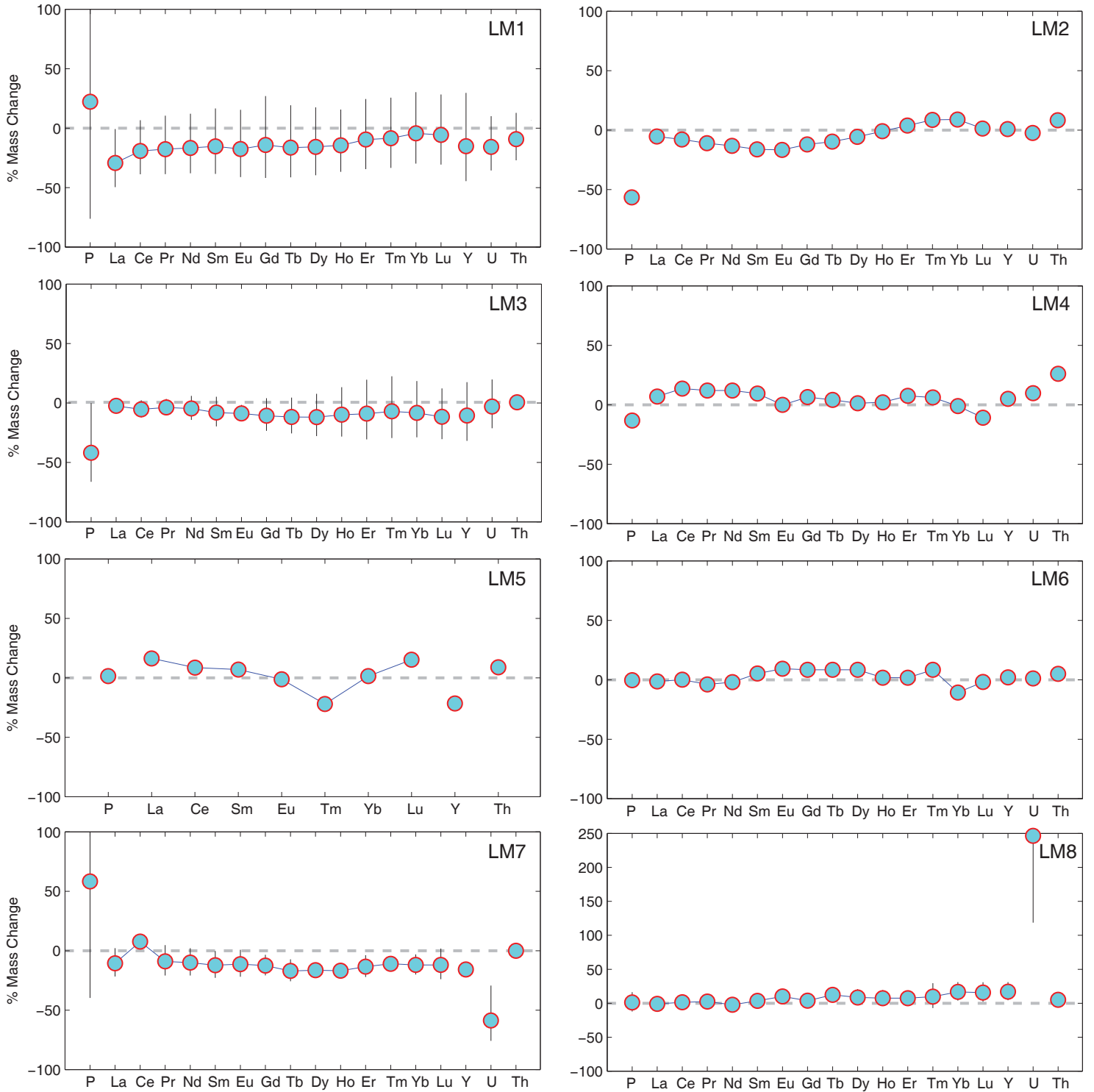


Ague, Fig. 2

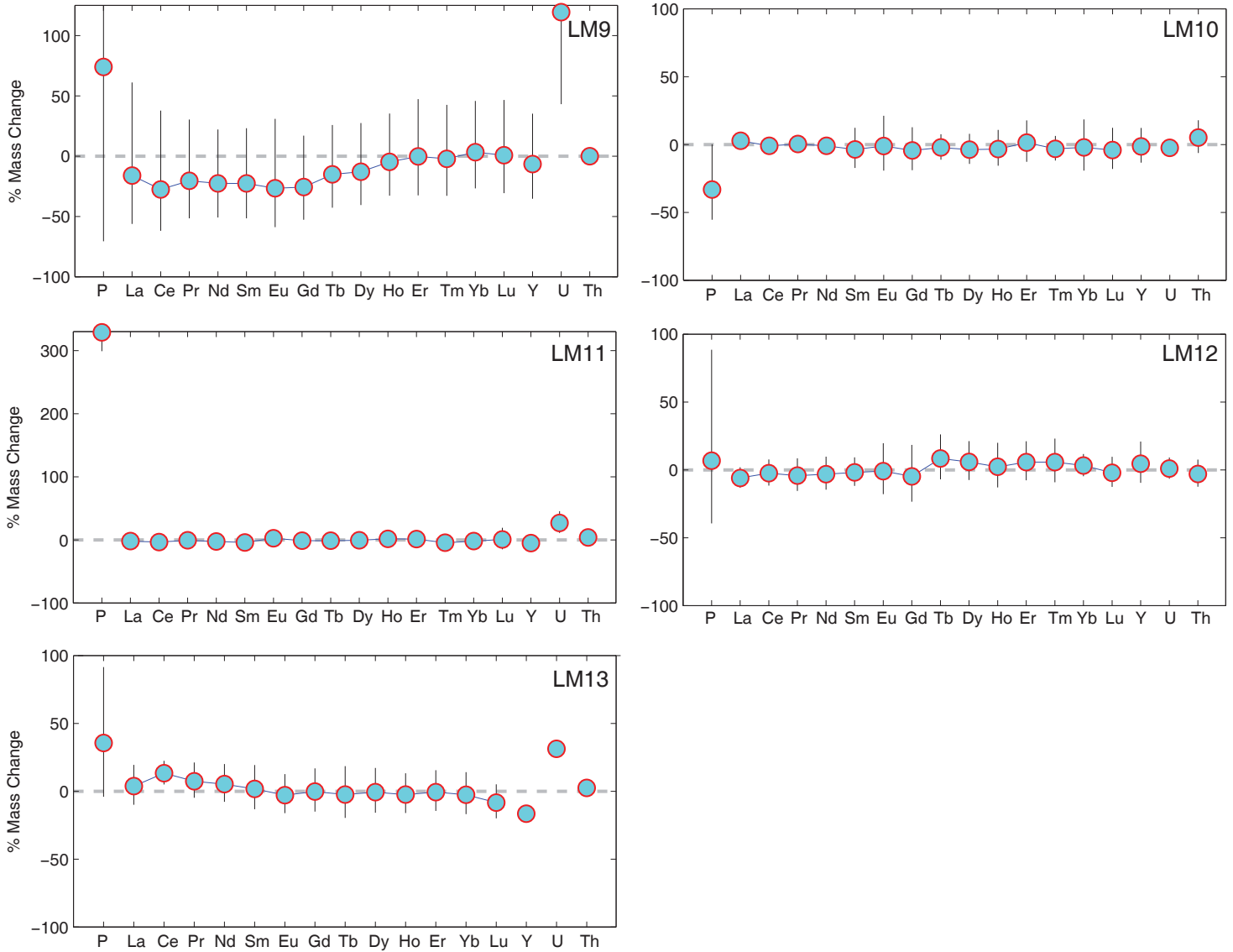




Ague, Fig. 3



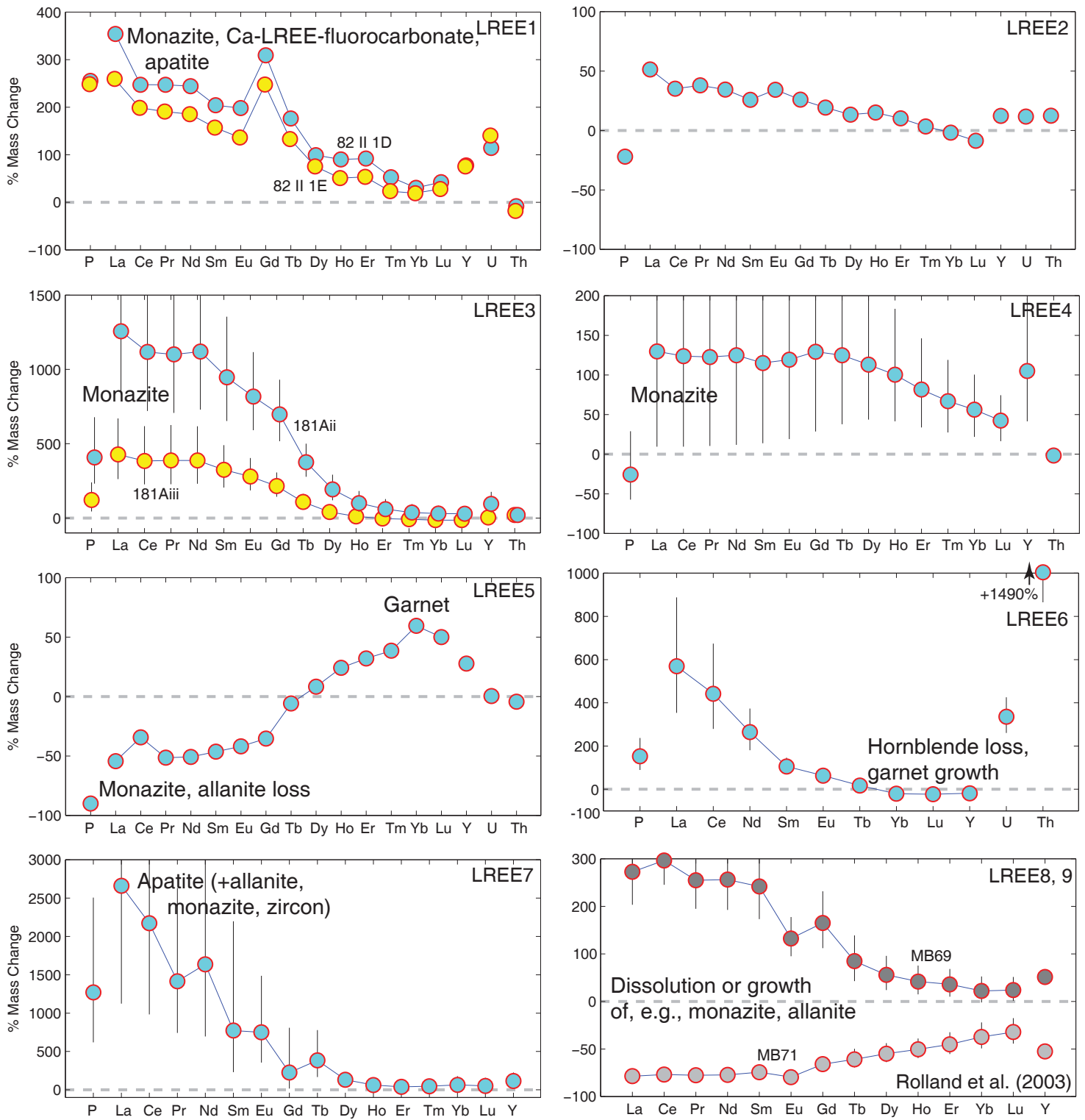
Ague, Fig. 4



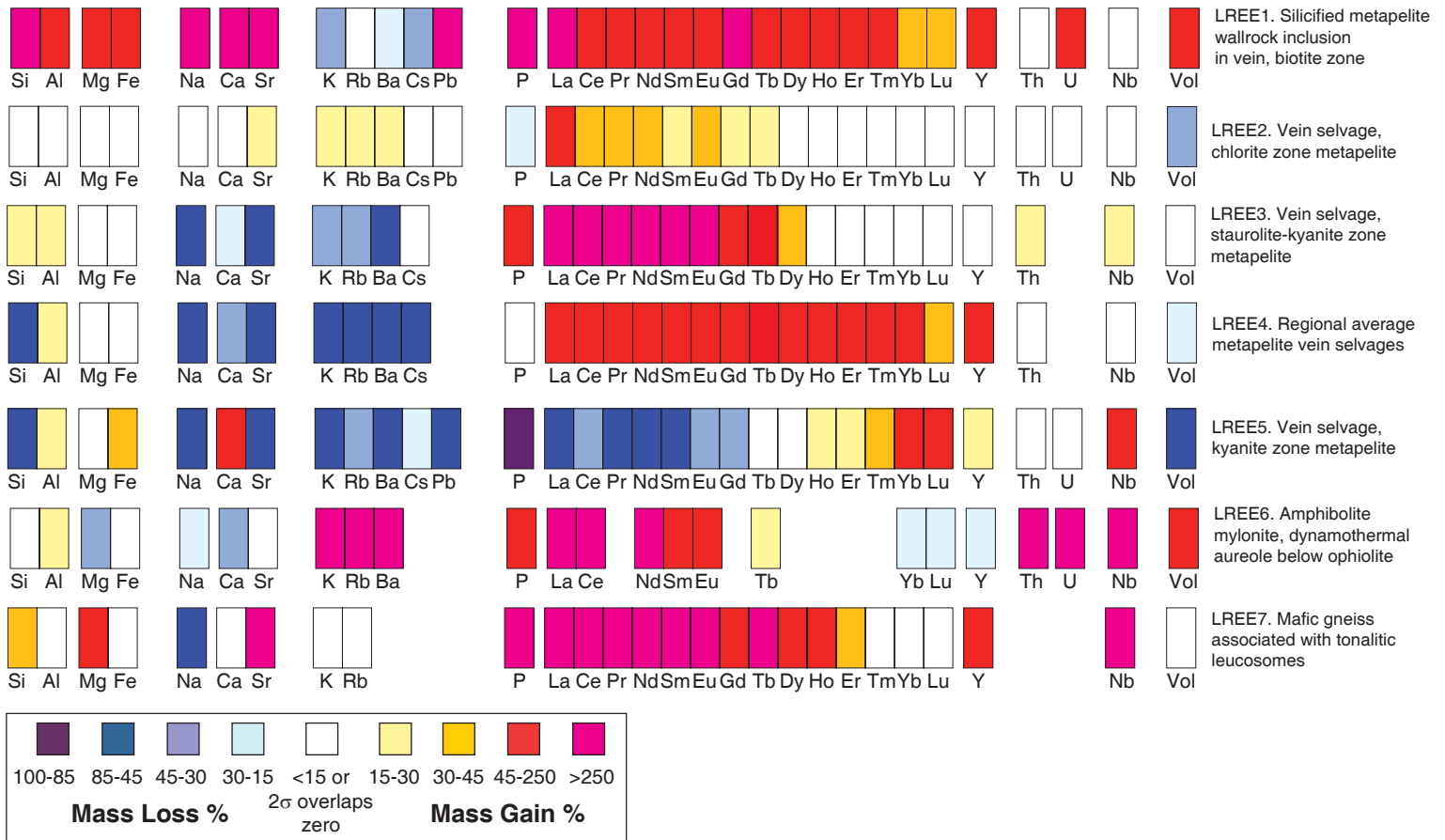
Ague, Fig. 4, continued



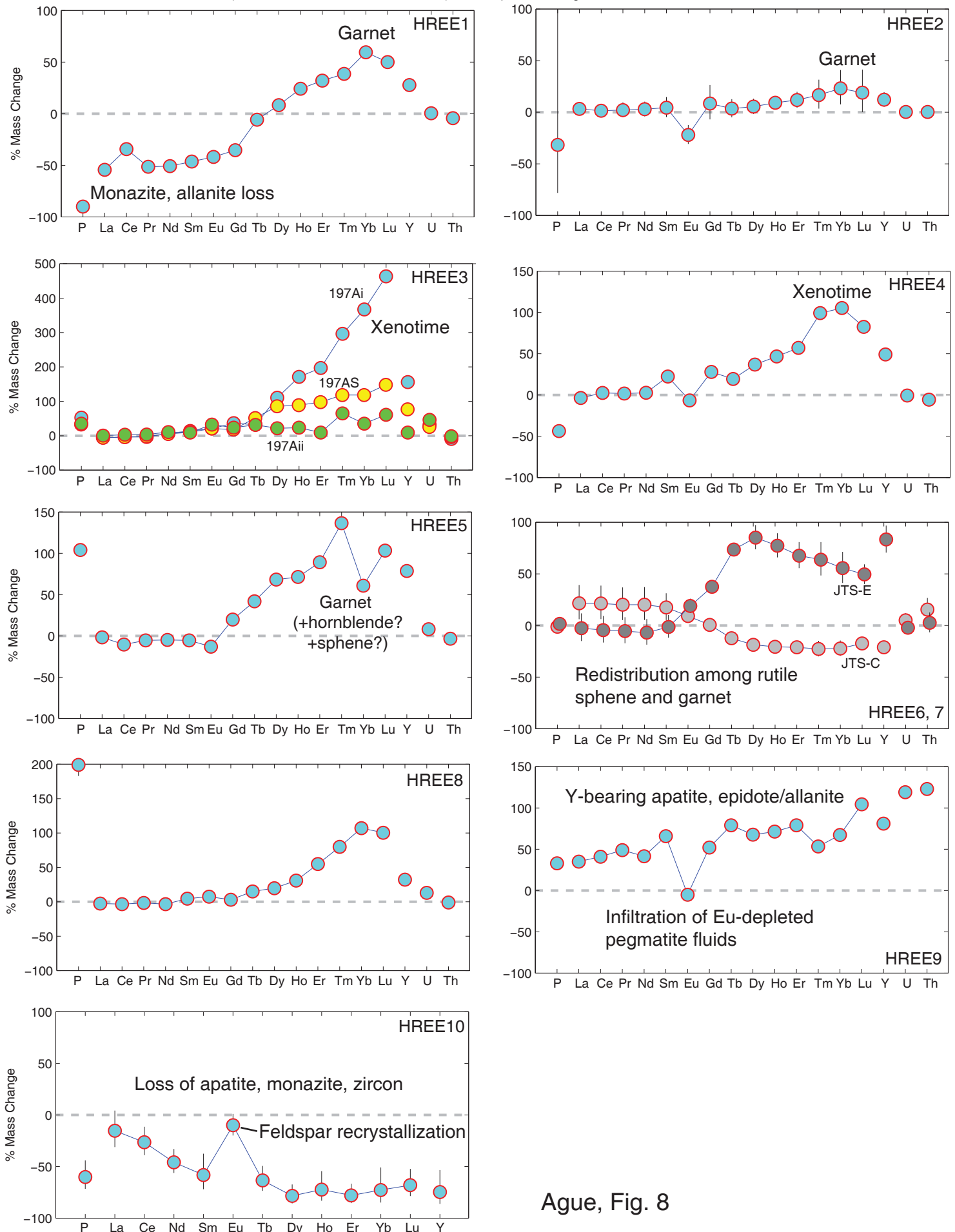
Ague, Fig. 5



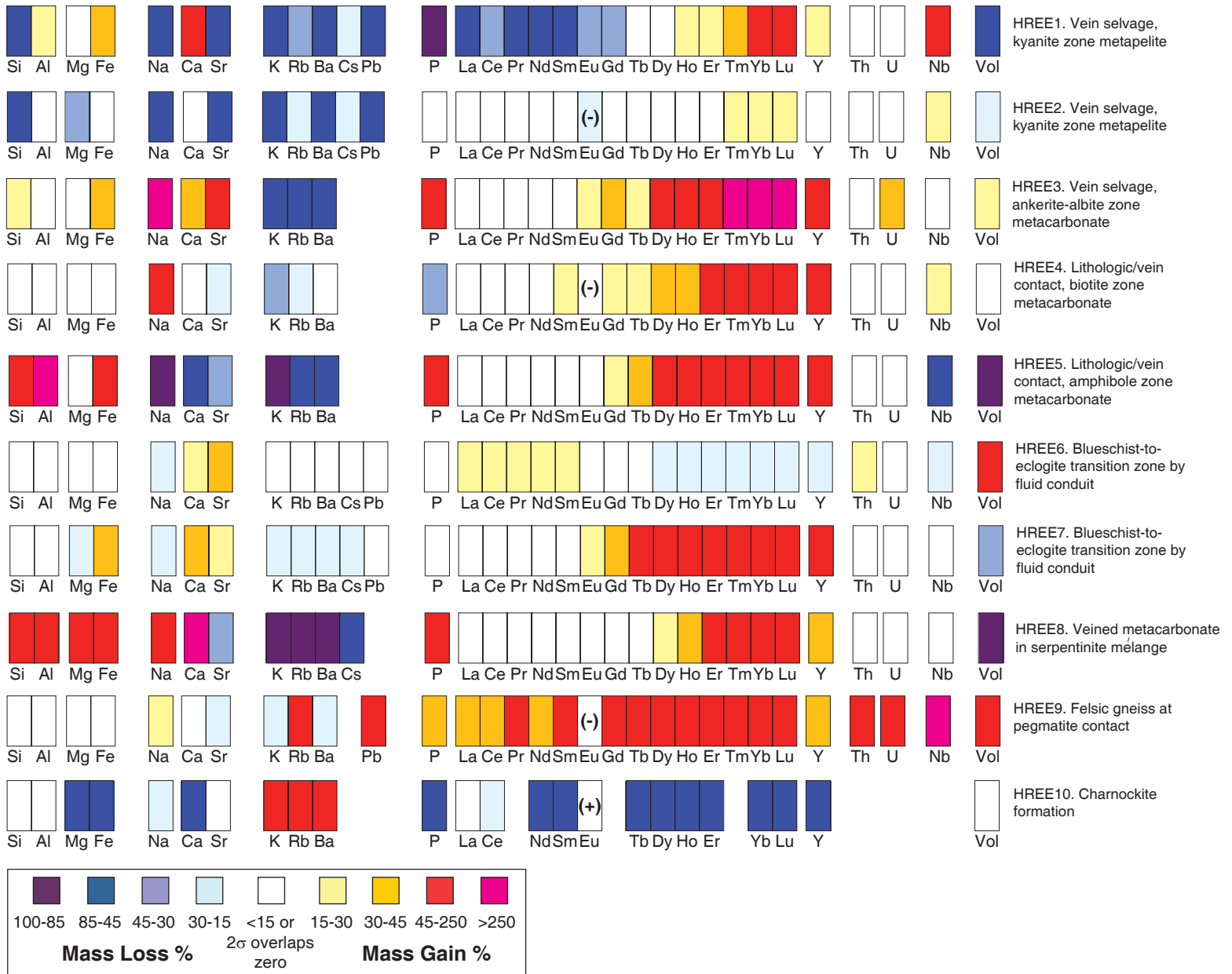
Ague, Fig. 6



Ague, Fig. 7

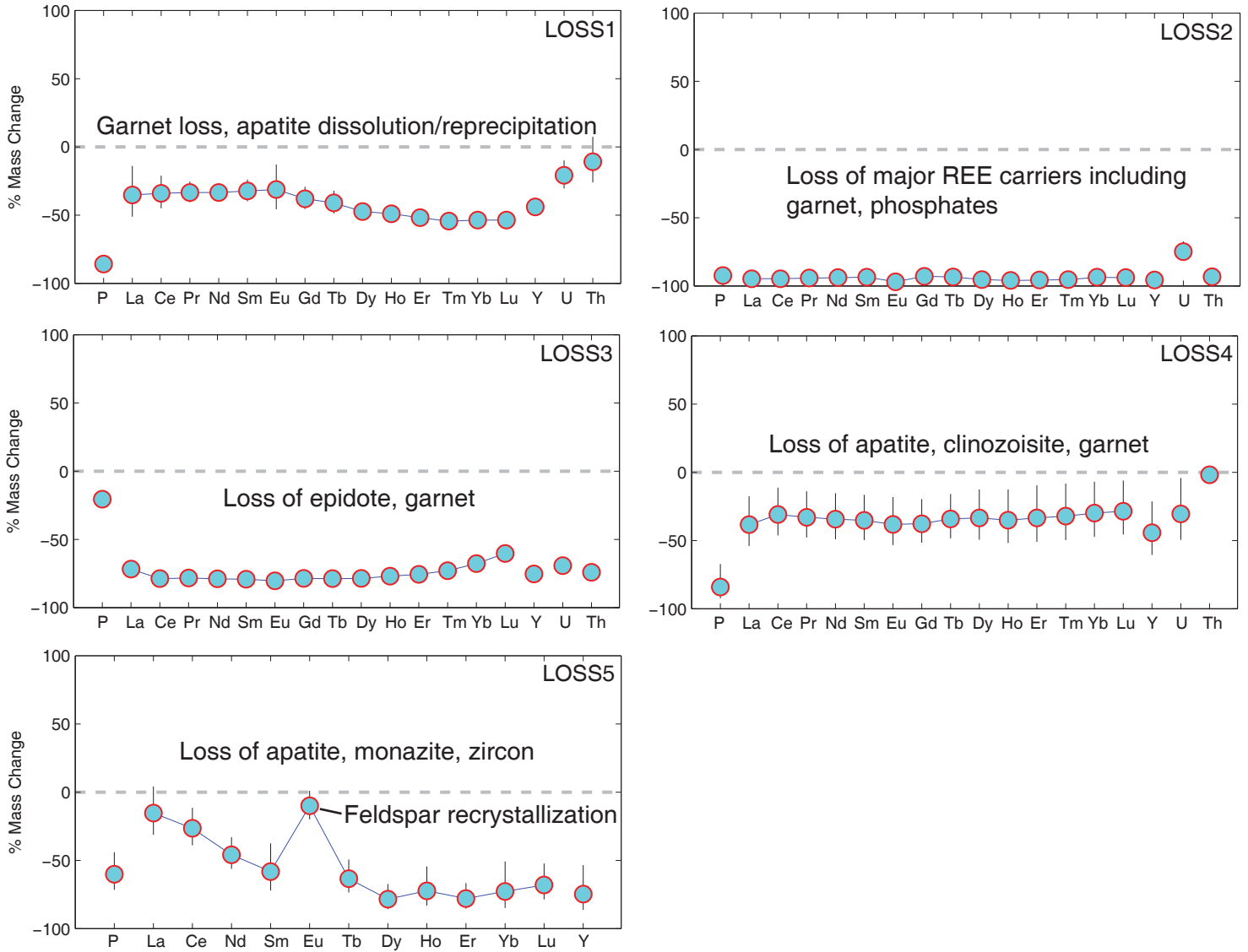


Ague, Fig. 8

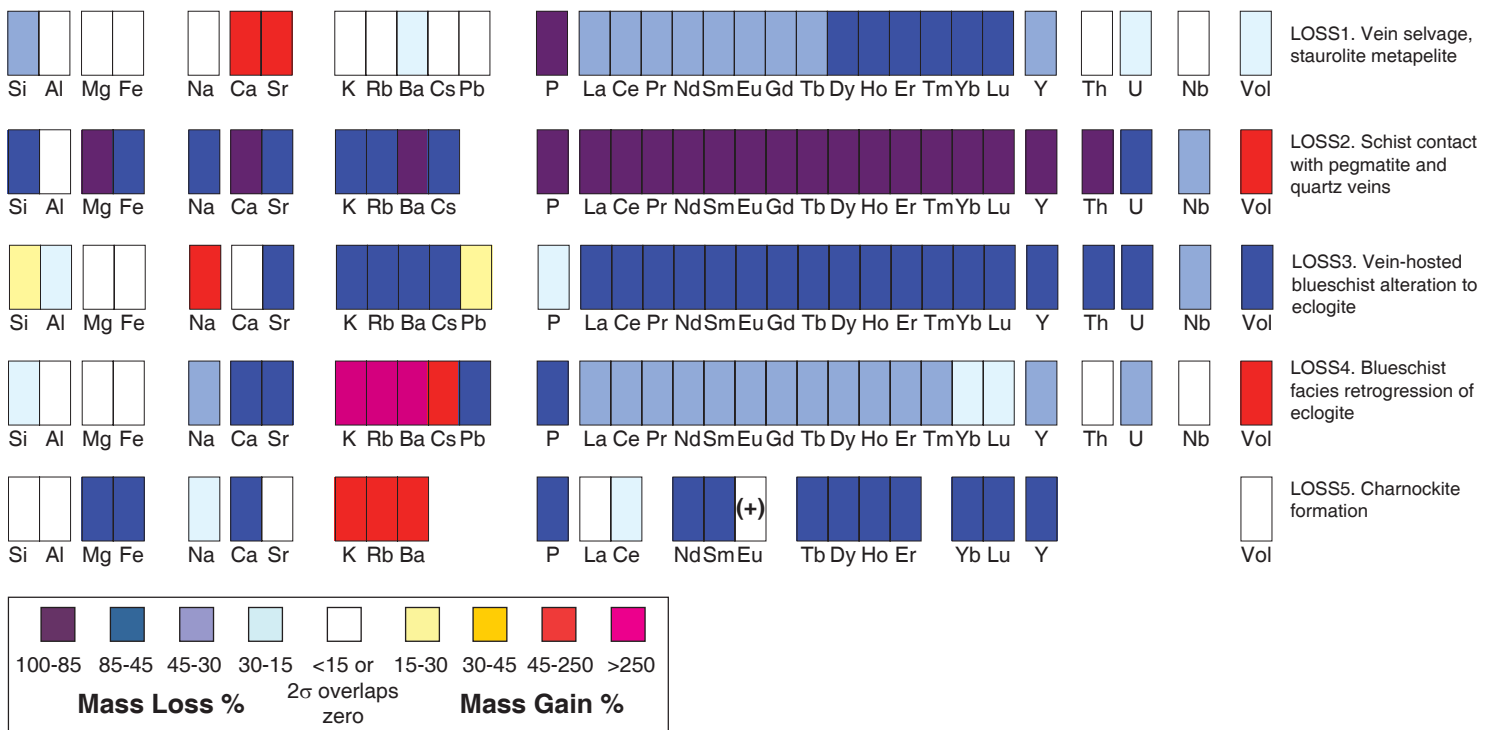


Ague, Fig. 9

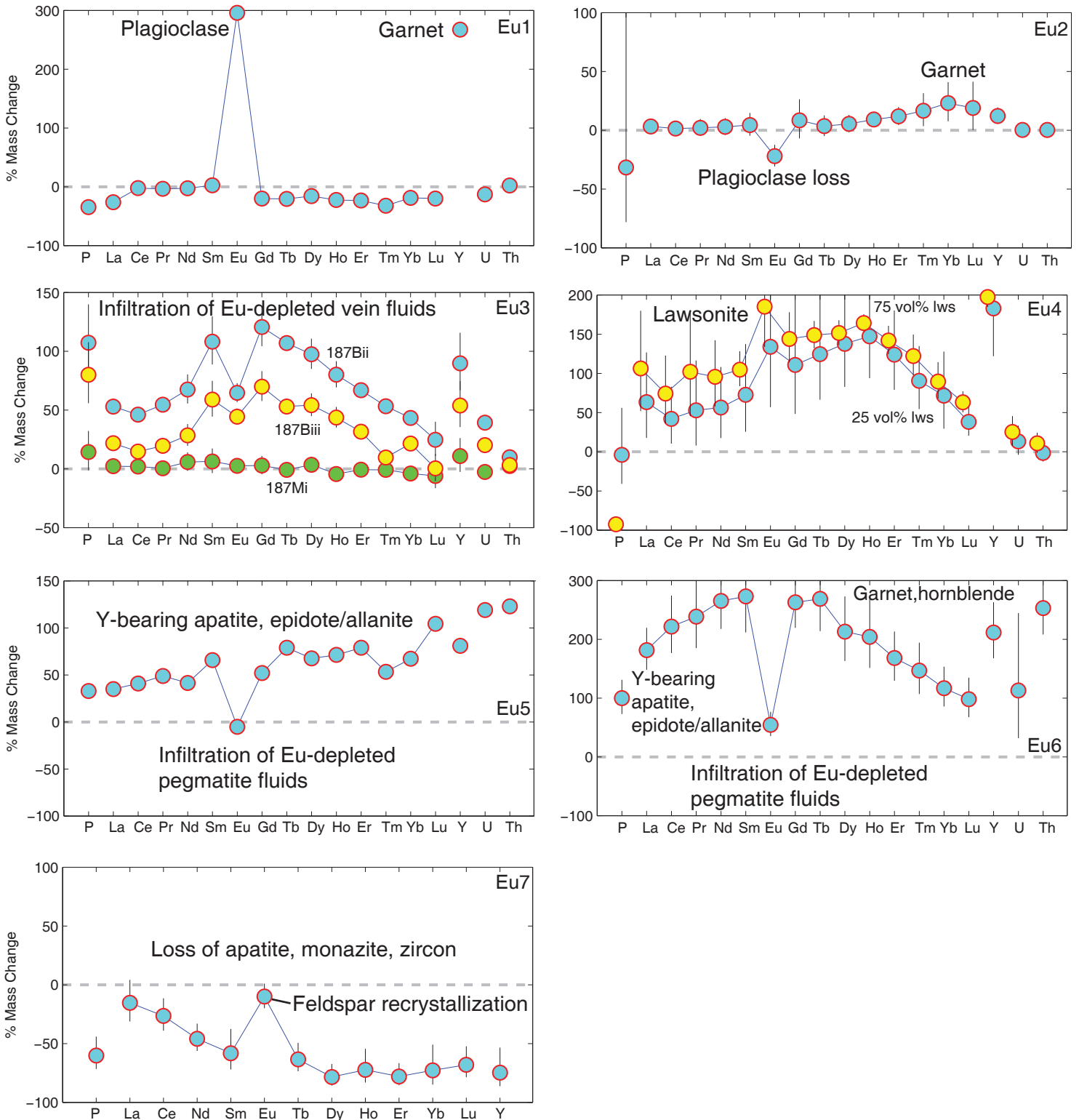




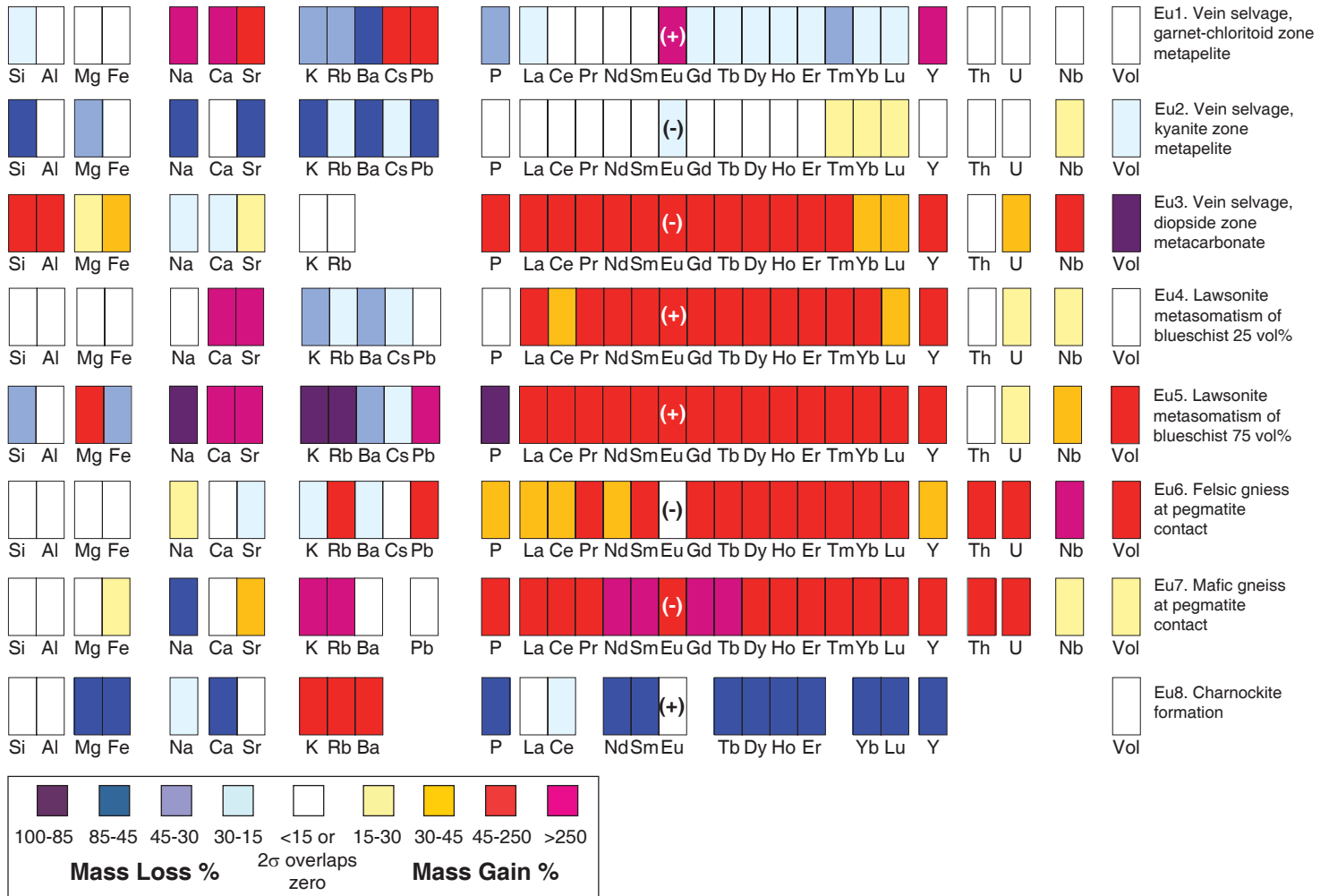
Ague, Fig. 10



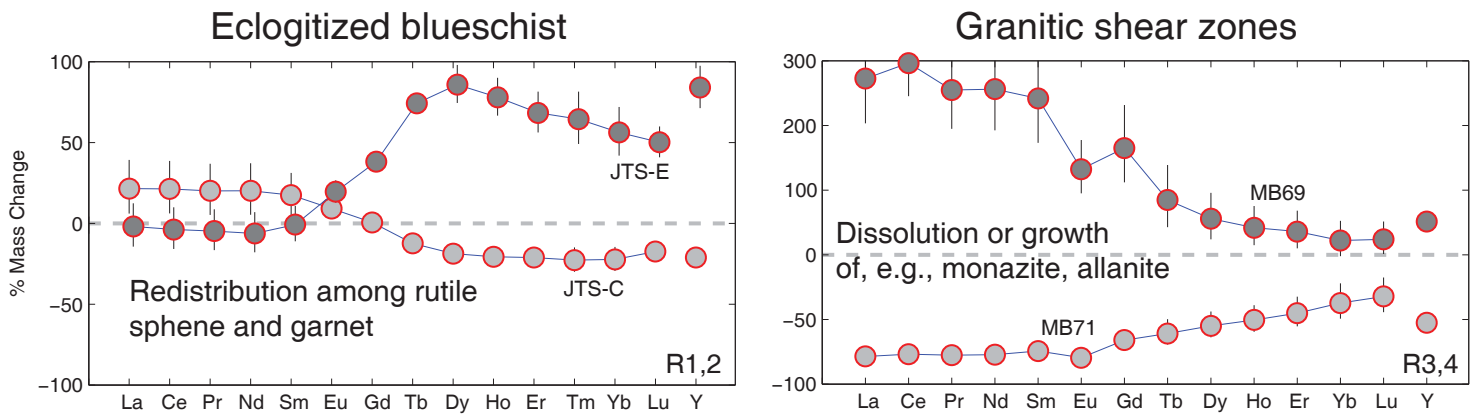
Ague, Fig. 11



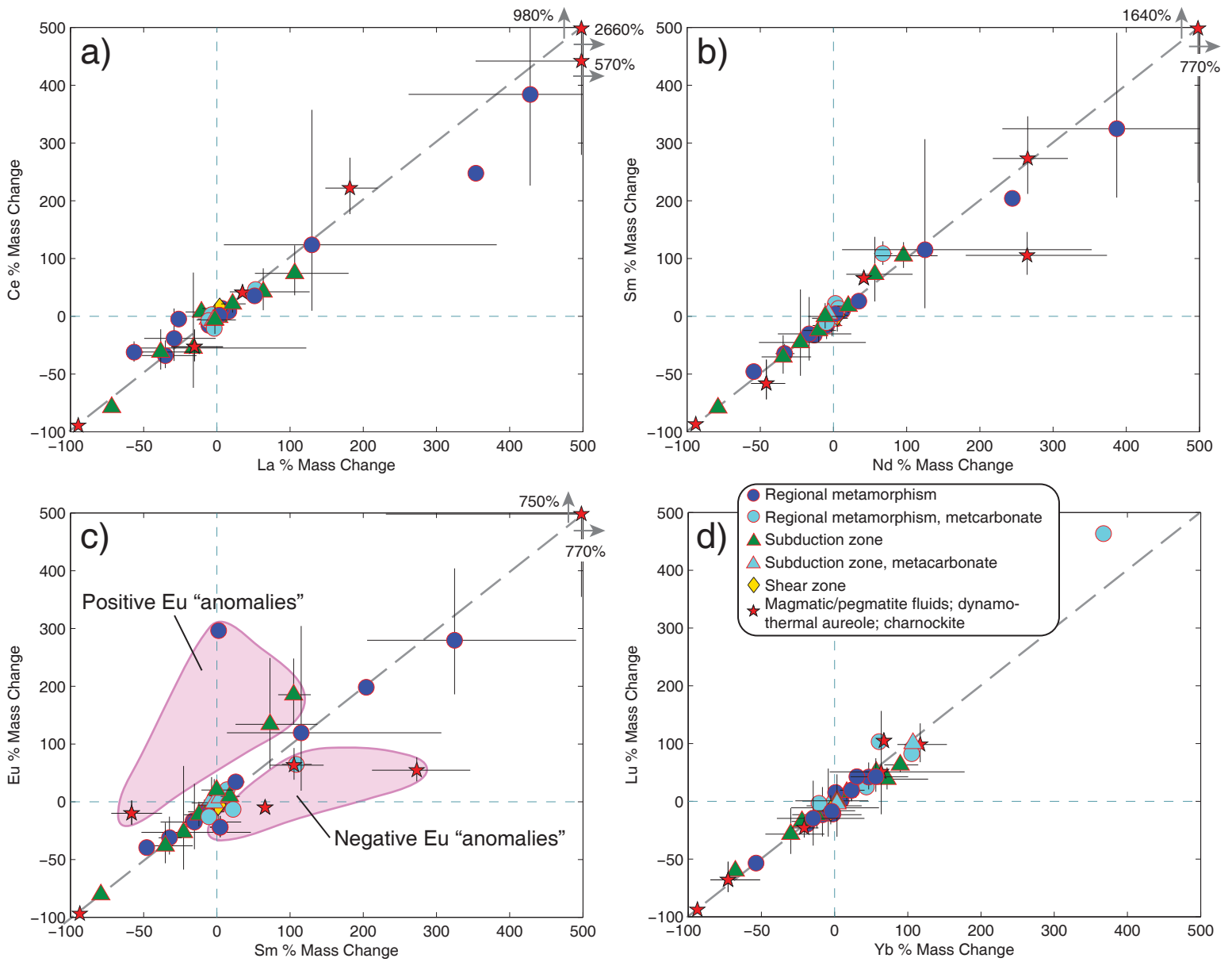
Ague, Fig. 12



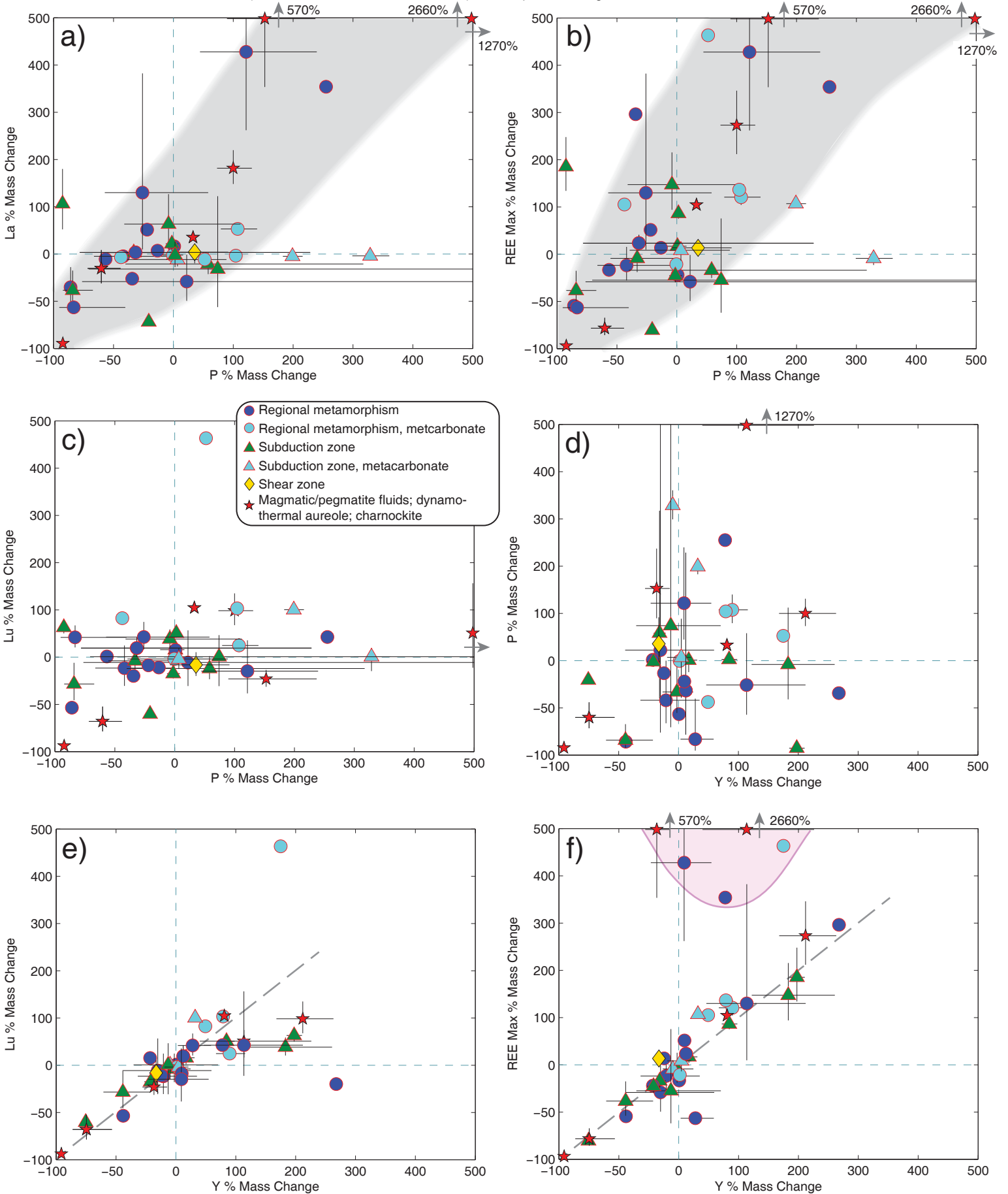
Ague, Fig. 13



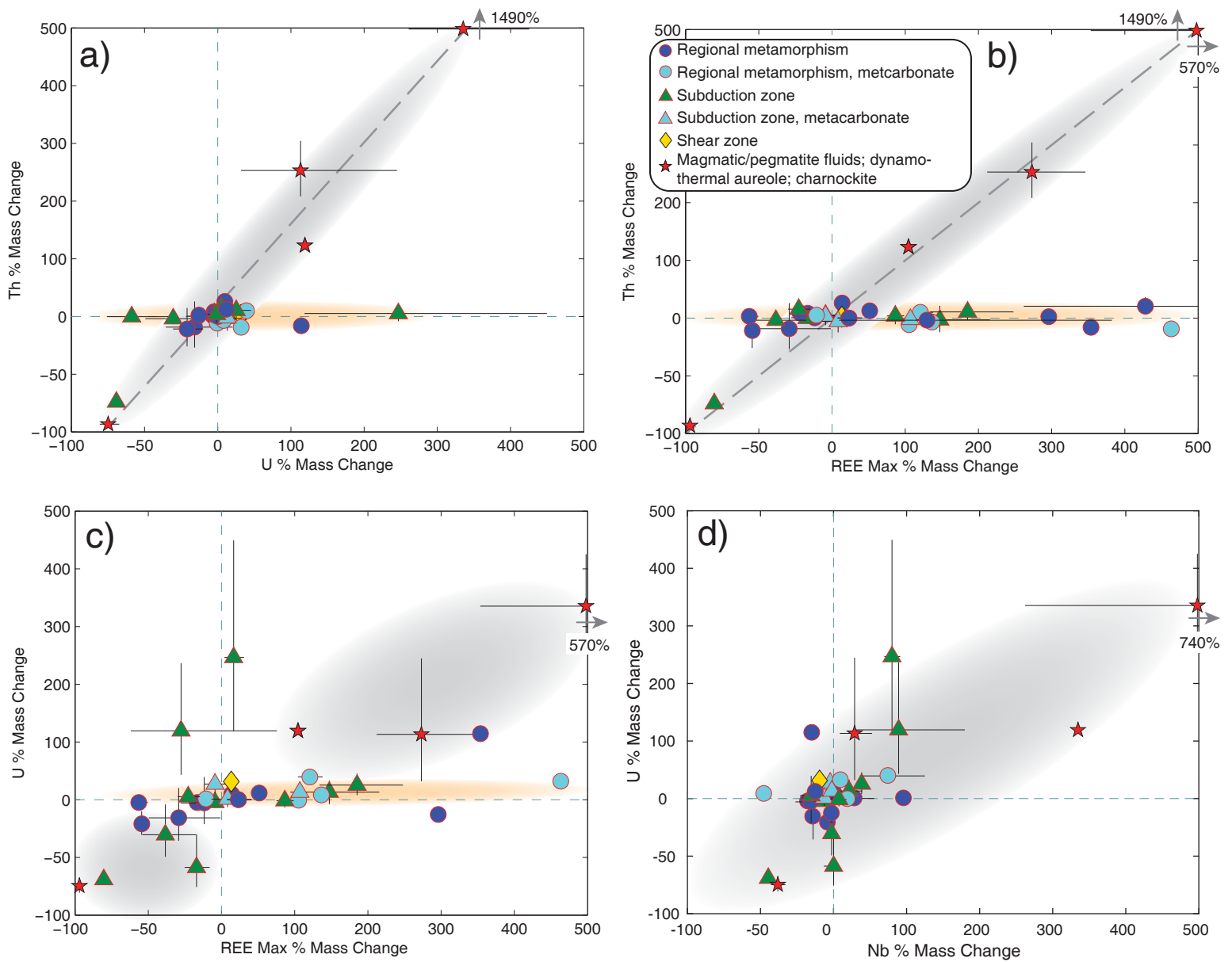
Ague, Fig. 14



Ague, Fig. 15

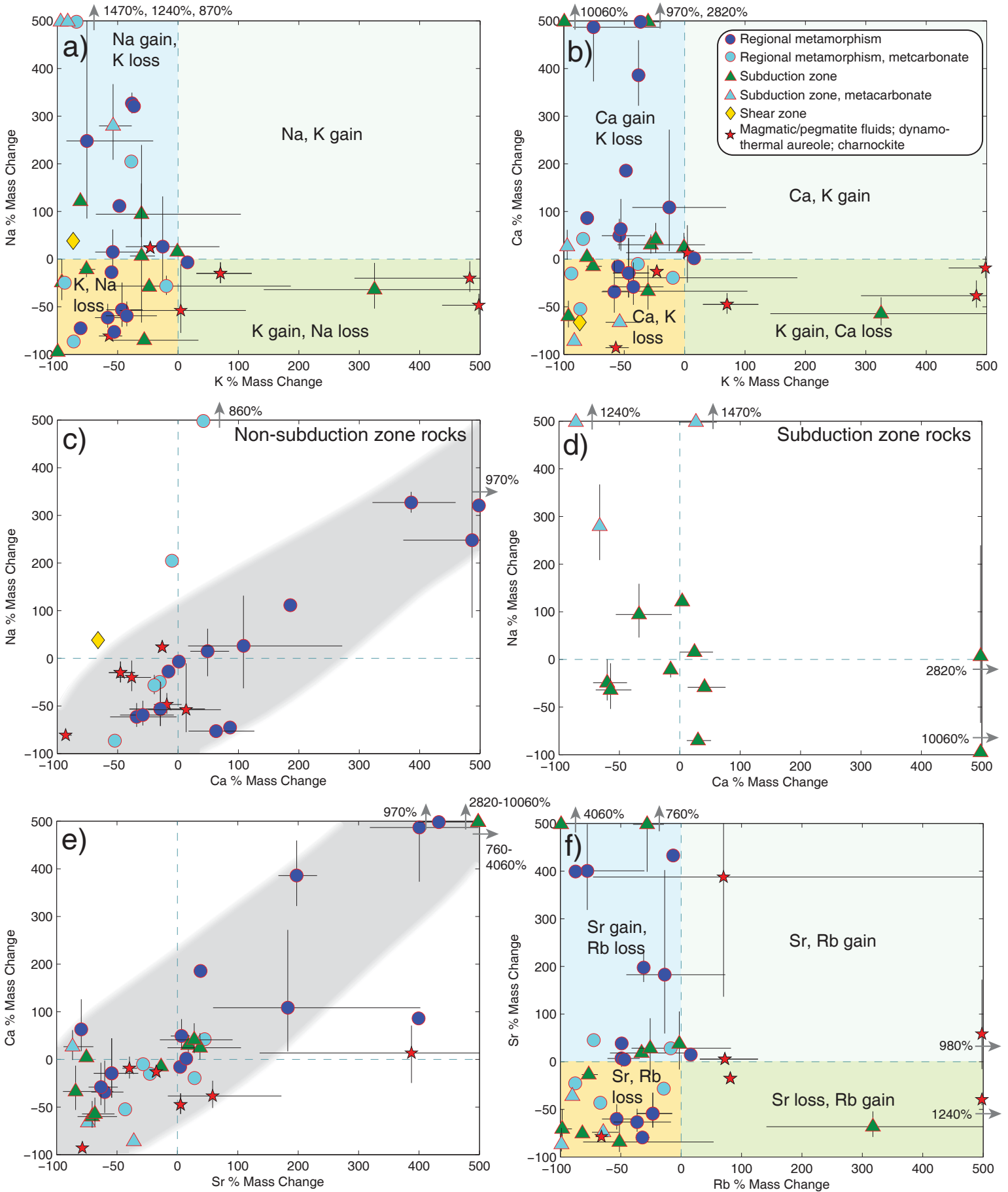


Ague, Fig. 16

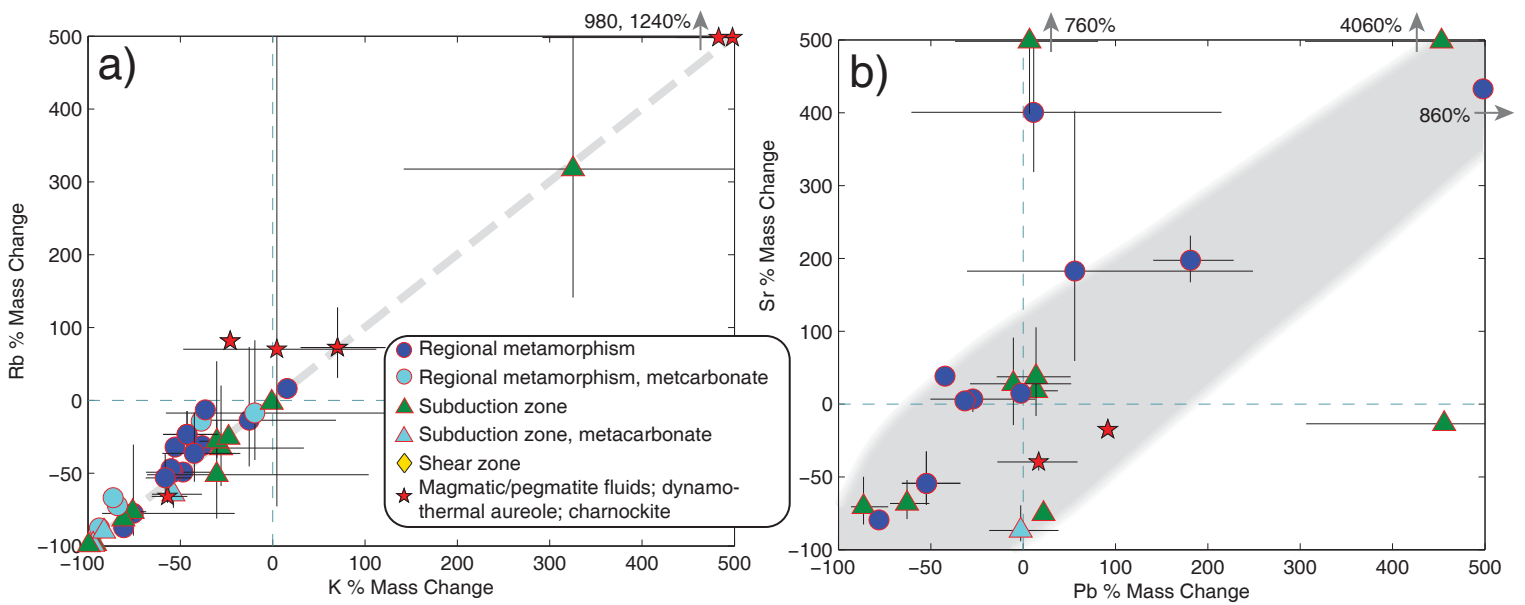


Ague, Fig. 17

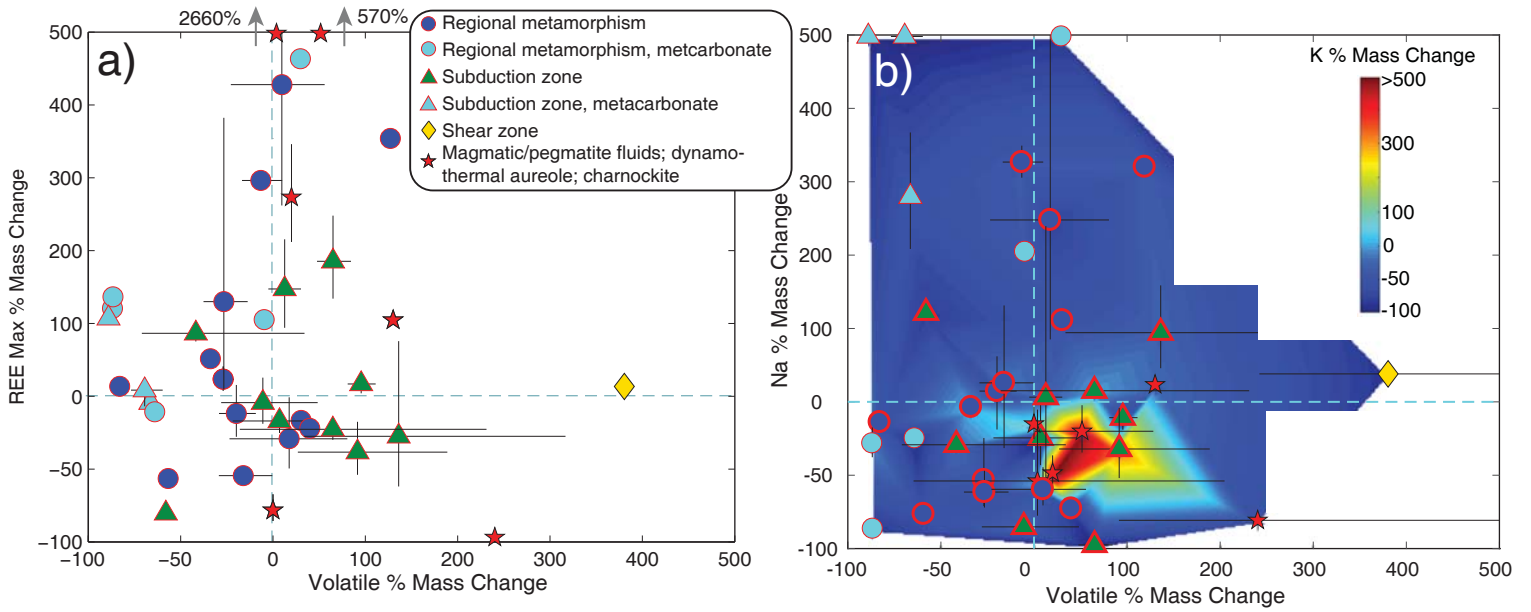




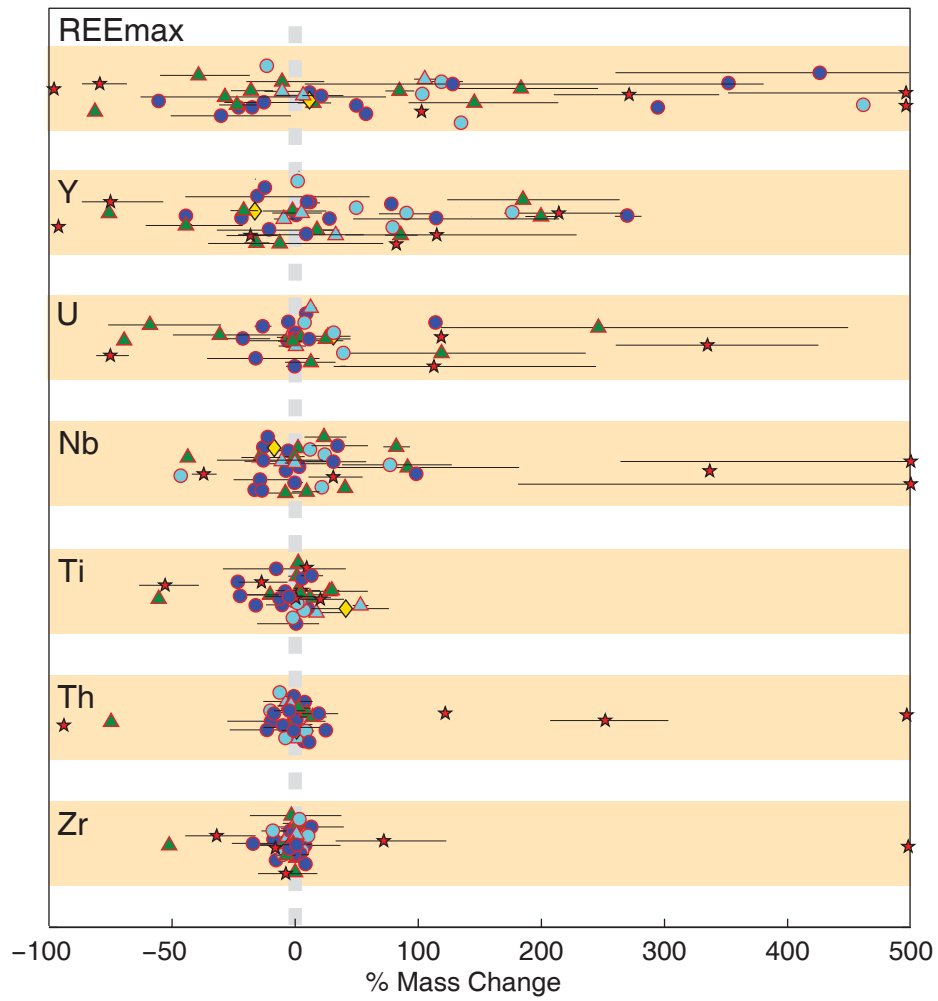
Ague, Fig. 18



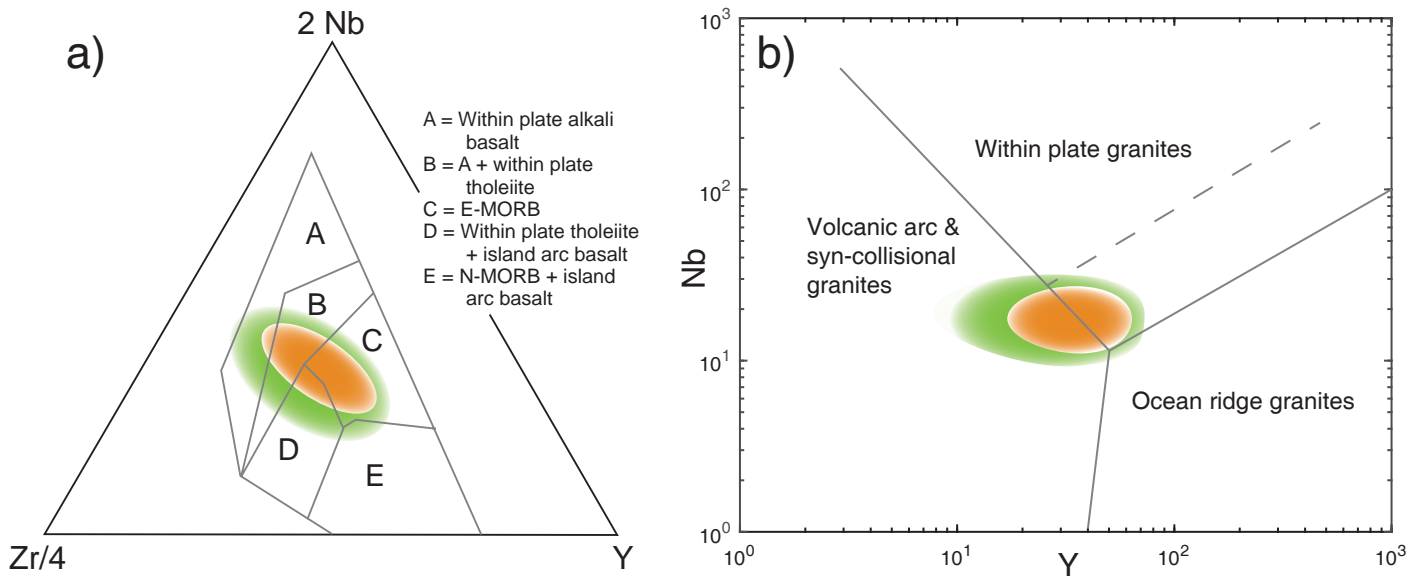
Ague, Fig. 19



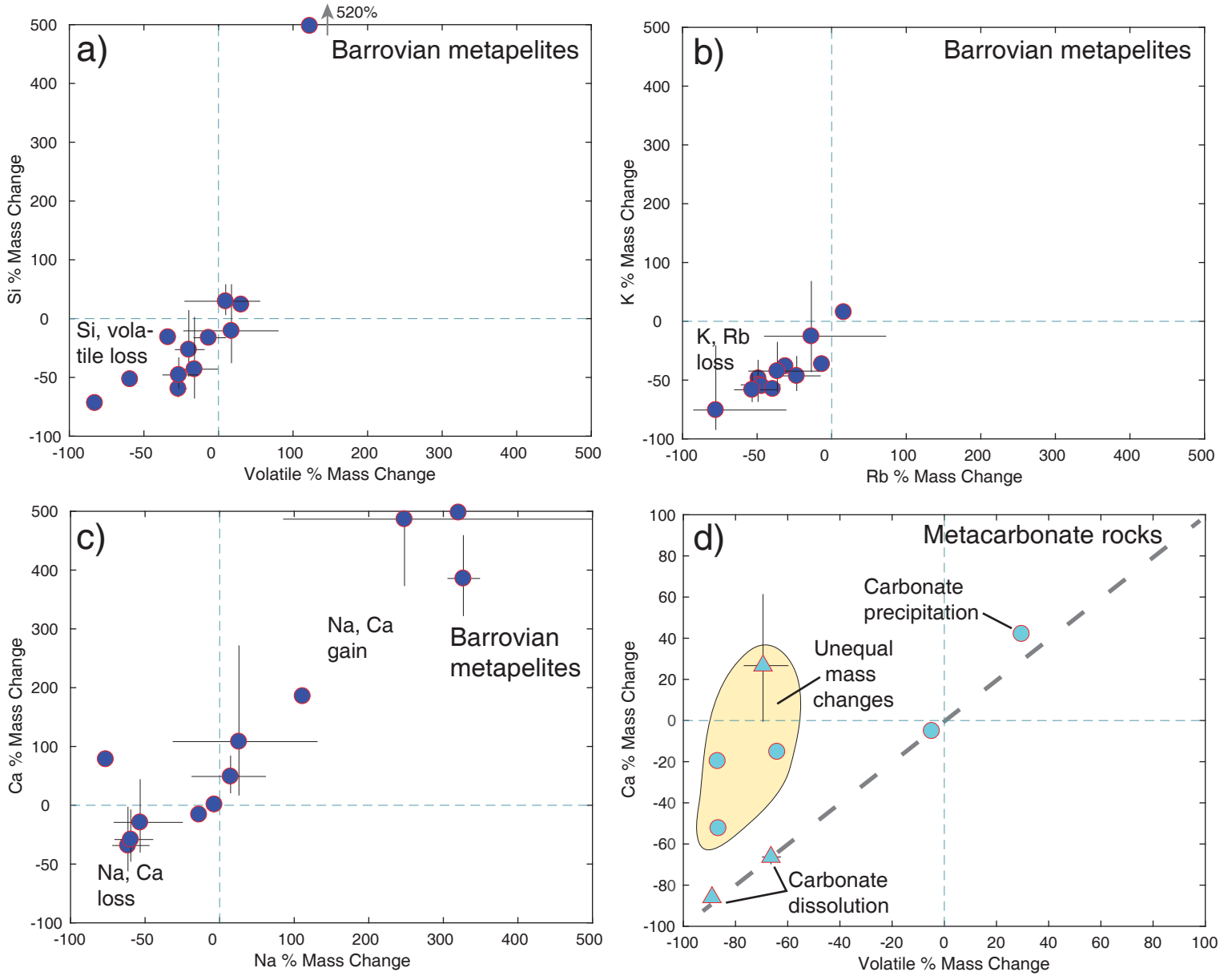
Ague, Fig. 20



Ague, Fig. 21



Ague, Fig. 22



Ague, Fig. 23

**Table 1.** Data sources

Code	Geologic Setting	Rock Category, Notes, and REE <sub>Max</sub>
<b>Limited REE Mobility</b>		
LM1	Vein selvage, Barrovian biotite zone metapelite, Scotland.	(1) Masters and Ague (2005) (site 7) and Supplemental Table 1 <sup>1</sup> . REE <sub>Max</sub> =La (-29%)
LM2	Vein selvage, Barrovian garnet-chloritoid zone metapelite, Scotland.	(1) Masters and Ague (2005) (site 1) and Supplemental Table 2. REE <sub>Max</sub> =Eu (-17%)
LM3	Vein selvages, Barrovian staurolite zone metapelite, Scotland.	(1) Masters and Ague (2005) and (sites 9, 94) and Supplemental Table 3. REE <sub>Max</sub> =Dy (-12%)
LM4	Vein selvage, kyanite zone metapelite, Waits River Formation, Vermont, USA.	(1) Penniston-Dorland and Ferry (2008), S35-1G samples. REE <sub>Max</sub> =Ce (14%)
LM5	Epidosite metasomatism of amphibolite, White Rock Formation, Nova Scotia, Canada.	(1) Muecke et al. (1979), Table 1. Average epidosite lens relative to average metabasite. REE <sub>Max</sub> =Tm (-22%)
LM6	Vein selvage, diopside zone metacarbonate rock, Wepawaug Schist, Connecticut, USA.	(2) Ague (2003), Figure 12F. REE <sub>Max</sub> =Yb (-11%)
LM7	Metapelitic rock at serpentinite mélange contact, Syros Island, Cyclades, Greece.	(3) Breeding et al. (2004). Altered contact samples E1 and E2 relative to distal samples E3–E5. Site JAGSY-51. REE <sub>Max</sub> =Ce (-28%)
LM8	Greenschist retrogression of blueschist, Tinos Island, Cyclades, Greece.	(3) Breeding et al. (2002; 2003) and Supplemental Table 4. REE <sub>Max</sub> =Yb (17%)
LM9	Glaucophanite rind on blueschist-eclogite block in serpentinite mélange, Syros Island, Cyclades, Greece.	(3) Supplemental Table 5. REE <sub>Max</sub> =Tb (-17%)
LM10	Eclogitized blueschist, Tinos Island, Cyclades, Greece.	(3) Supplemental Table 6. REE <sub>Max</sub> =Gd (-4%)
LM11	Metacarbonate rock at veined metasedimentary mélange contact, Tinos Island, Cyclades, Greece.	(4) Ague and Nicolescu (2014), Figure S3b. Site JAGTI-18. REE <sub>Max</sub> =Tm (-4%)
LM12	Metacarbonate rock at serpentinite contact, Malaspina, Corsica, France.	(4) Galvez et al. (2013), Figure 6. REE <sub>Max</sub> =Tb (8%)
LM13	Greenschist facies shear zones cutting granitic rocks, Mont Blanc Massif, Western Alps.	(5) Rolland et al. (2003). Sample C33. REE <sub>Max</sub> =Ce (13%)
<b>LREE Mobility</b>		
LREE1	Silicified metapelite wallrock inclusions in vein, Barrovian biotite zone, Scotland.	(1) Masters and Ague (2005) (site 82) and Supplemental Table 7. 82 II 1D used for main data set herein. Mass transfer for 82 II 1E also shown in Figure 6. REE <sub>Max</sub> =La (350%)
LREE2	Vein selvage, chlorite zone metapelite, Waits	(1) Penniston-Dorland and Ferry (2008). S40-

<sup>1</sup>Deposit item AM-xyz, Supplemental Tables. Deposit items are stored on the MSA web site and available via the American Mineralogist Table of Contents. Find the article in the table of contents at GSW ([ammin.geoscienceworld.org](http://ammin.geoscienceworld.org)) or MSA ([www.minsocam.org](http://www.minsocam.org)), and then click on the deposit link.

	River Formation, Vermont, USA.	1D samples. REE <sub>Max</sub> =La (51%)
LREE3	Vein selvage, staurolite-kyanite zone, Unst, Shetland Islands, Scotland.	(1) Bucholz and Ague (2010) (site 181). 181Aiii selvage relative to less altered samples 181B, C, D, and E used in main data set herein. 181Aii selvage relations also shown in Figure 6. REE <sub>Max</sub> =La (430%)
LREE4	Regional vein selvage average, Unst, Shetland Islands, Scotland.	(1) Bucholz and Ague (2010), Figure 6. REE <sub>Max</sub> =La (130%)
LREE5&HREE1	Vein selvage, kyanite zone metapelite, Wepawaug Schist, Connecticut, USA.	(1) Ague (2011). Sample JAW-114i selvage margin relative to most distal sample 114vi. REE <sub>Max</sub> =Yb (59%); Pr loss has comparable absolute value (-51%)
LREE6	Biotite amphibolite, mylonite shear zone in dynamothermal aureole below an ophiolite remnant.	(6) Dostal et al. (1980). Epidote and quartz amphibolite used as precursors. REE <sub>Max</sub> =La (570%)
LREE7	Clino- and orthopyroxene reaction zones, margins of mafic gneiss lenses/layers associated with tonalitic leucosomes.	(6) Pan and Fleet (1996). REE <sub>Max</sub> =La (2660%)
LREE8 & R3	Shear zones cutting granitic rocks, Mont Blanc Massif, Western Alps.	(5) Rolland et al. (2003). Sample MB69. REE <sub>Max</sub> =Ce (300%)
LREE9 & R4	Shear zones cutting granitic rocks, Mont Blanc Massif, Western Alps.	(5) Rolland et al. (2003). Sample MB71. REE <sub>Max</sub> =Eu (-81%)

#### HREE Mobility

HREE1	Vein selvage, kyanite zone metapelite, Wepawaug Schist, Connecticut, USA.	(1) Ague (2011). Sample JAW-114i selvage margin relative to most distal sample 114vi. REE <sub>Max</sub> =Yb (59%); Pr loss has comparable absolute value (-51%)
HREE2	Average vein selvage, kyanite zone metapelite, Wepawaug Schist, Connecticut, USA.	(1) Ague (2011). Site JAW-165A, south profile. REE <sub>Max</sub> =Yb (23%)
HREE3	Vein selvage, metacarbonate rock, metacarbonate ankerite-albite zone, Wepawaug Schist, Connecticut, USA	(2) Ague (2003), Figure 6A. JAW-197Ai selvage relations used for main data set herein. Selvage mass transfer for samples 197AS and 197Aii also shown in Figure 8. REE <sub>Max</sub> =Lu (460%)
HREE4	Metacarbonate rock at contact with metapelite, metacarbonate biotite zone, Wepawaug Schist, Connecticut, USA.	(2) Ague (2003), Figure 8A. JAW-171A. REE <sub>Max</sub> =Yb (105%)
HREE5	Metacarbonate rock at veined lithologic contact, metacarbonate amphibole zone, Wepawaug Schist, Connecticut, USA.	(2) Ague (2003), Figure 10C. JAW-181Rii. REE <sub>Max</sub> =Tm (140%)
HREE6	Eclogitized blueschist adjacent to fluid conduit, Tian Shan, China.	(3) Beinlich et al. (2010). Blueschist-eclogite transition zone sample C relative to blueschist hosts A and B. REE <sub>Max</sub> =Tm (-23%)
HREE7	Eclogitized blueschist adjacent to fluid	(3) Beinlich et al. (2010). Blueschist-eclogite transition zone sample E relative to blueschist



	conduit, Tian Shan, China.	hosts A and B. REE <sub>Max</sub> =Dy (86%)
HREE8	Veined metacarbonate rock in contact with serpentinite mélange, Syros Island, Cyclades, Greece.	(4) Ague and Nicolescu (2014), Figure S3a. Site JAGSY-37. REE <sub>Max</sub> =Yb (110%)
HREE9	Felsic gneiss at pegmatite contact, Napier Complex, East Antarctica.	(6) Carson and Ague (2008), Figure 6a. Series 28. REE <sub>Max</sub> =Lu (105%)
HREE10, Eu8, & LOSS5	Charnockite formation, incipient granulite facies, Kabbaldurga, India.	(6) Stähle et al. (1987), Table 2. Charnockite and pegmatitic charnockite relative to gneiss. REE <sub>Max</sub> =Dy (-78%)

#### Loss of REE

LOSS1	Vein selvage, Barrovian staurolite zone, Scotland.	(1) Masters and Ague (2005) (site 109) and Supplemental Table 8. REE <sub>Max</sub> =Tm (-54%)
LOSS2	Sillimanite schist in contact with pegmatite and quartz veins.	(6) Supplemental Table 9. REE <sub>Max</sub> =Eu (-97%)
LOSS3	Vein-hosted eclogite facies blueschist alteration zone in blueschist host, Tian Shan, China.	(3) John et al. (2008), Figure 11. REE <sub>Max</sub> =Eu (-80%)
LOSS4	Phengite-ankerite blueschist rehydration of eclogite, Tian Shan, China.	(3) van der Straaten et al. (2008). REE <sub>Max</sub> =La (-38%)
LOSS5, Eu8 & HREE10	Charnockite formation, incipient granulite facies, Kabbaldurga, India.	(6) Stähle et al. (1987), Table 2. Charnockite and pegmatitic charnockite relative to gneiss. REE <sub>Max</sub> =Dy (-78%)

#### Eu “Anomalies”

Eu1	Vein selvage, Barrovian garnet-chloritoid zone metapelite, Scotland.	(1) Ague (1997) (site 101L), Masters and Ague (2005), and Supplemental Table 10. REE <sub>Max</sub> =Eu (300%)
Eu2 & HREE2	Average vein selvage, kyanite zone metapelite, Wepawaug Schist, Connecticut, USA.	(1) Ague (2011). Site JAW-165A, south profile. REE <sub>Max</sub> =Yb (23%)
Eu3	Metacarbonate rock at veined lithologic contact, diopside zone, Wepawaug Schist, Connecticut, USA.	(2) Ague (2003), Figures 17A, B. JAW-187Bii altered zone used in main data set herein. Geochemical relations for selvage sample 187Biii and distal sample 187Mi also shown in Figure 12. REE <sub>Max</sub> =Gd (120%)
Eu4	Lawsonite metasomatism in blueschist facies metapelites at contacts with serpentinite, Schistes Lustrés complex, Corsica, France.	(3) Vitale Brovarone et al. (2013), Table 1; 25 vol% lawsonite. REE <sub>Max</sub> =Ho (150%)
Eu5	Lawsonite metasomatism in metapelites at contacts with serpentinite, Schistes Lustrés complex, Corsica, France.	(3) Vitale Brovarone et al. (2013), Table 1; 75 vol% lawsonite. REE <sub>Max</sub> =Eu (185%)
Eu6 & HREE9	Felsic gneiss at pegmatite contact, Napier Complex, East Antarctica.	(6) Carson and Ague (2008), Figure 6a. Series 28. REE <sub>Max</sub> =Lu (105%)
Eu7	Mafic gneiss at pegmatite contact, Napier	(6) Carson and Ague (2008), Figure 7. Series

	Complex, East Antarctica.	76. Contact sample 76/1B relative to distal samples 76/7B– 76/9B. REE <sub>Max</sub> =Sm (270%)
Eu8, LOSS5,& HREE10	Charnockite formation, incipient granulite facies, Kabbaldurga, India.	(6) Stähle et al. (1987), Table 2. Charnockite and pegmatitic charnockite relative to gneiss. REE <sub>Max</sub> =Dy (-78%)

#### Redistribution of REE

R1 & HREE6	Eclogitized blueschist adjacent to fluid conduit, Tian Shan, China.	(3) Beinlich et al. (2010). Blueschist-eclogite transition zone sample C relative to blueschist hosts A and B. REE <sub>Max</sub> =Tm (-23%)
R2 & HREE7	Eclogitized blueschist adjacent to fluid conduit, Tian Shan, China.	(3) Beinlich et al. (2010). Blueschist-eclogite transition zone sample E relative to blueschist hosts A and B. REE <sub>Max</sub> =Dy (86%)
R3 & LREE8	Greenschist facies shear zones cutting granitic rocks, Mont Blanc Massif, Western Alps.	(5) Rolland et al. (2003). Sample MB69. REE <sub>Max</sub> =Ce (300%)
R4 & LREE9	Greenschist facies shear zones cutting granitic rocks, Mont Blanc Massif, Western Alps.	(5) Rolland et al. (2003). Sample MB71. REE <sub>Max</sub> =Eu (-81%)

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*Notes:* Rock Categories: (1) regional metamorphism (non-subduction); (2) regional metamorphism, metacarbonate rocks (non-subduction) ; (3) subduction zone; (4) subduction zone metacarbonate rocks ; (5) shear zones, granitic rocks; (6) dynamothermal aureole (shear zone), magmatic/pegmatite fluid infiltration, charnockite. Percentage mass change for REE<sub>Max</sub> element given in parentheses.