

1 **Revision 4**

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3 **Majorite–olivine–high-Ca pyroxene assemblage in the shock-melt veins of**
4 **Pervomaisky L6 chondrite**

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13
14 **ABSTRACT**

15 High-pressure minerals - majorite-pyrope garnet and jadeite - were found in the Pervomaisky
16 L6 ordinary chondrite. Majorite-pyrope (79 mol% majorite) was observed within the fine-grained
17 silicate matrix of a shock-melt vein (SMV), coexisting with olivine and high-Ca pyroxene. This is
18 the first report of a garnet-olivine-high-Ca pyroxene assemblage that crystallized from the melt in
19 the SMV matrix of meteorite. *PT*-conditions of the formation of the SMV matrix with olivine
20 fragments are 13.5–15.0 GPa and 1750–2150 °C, the lowest parameters among all known majorite-
21 bearing (H, L)-chondrites. The estimated conditions include the olivine/(olivine + ringwoodite)
22 phase boundary and there is a possibility that observed olivine is the result of
23 wadsleyite/ringwoodite back-transformation during a cooling and decompression stage. In the
24 framework of this hypothesis, we discuss the problem of survival of the high-pressure phases at the
25 post-shock stage in the meteorites and propose two possible *PT*-paths: (1) the high-pressure mineral
26 is transformed to a low-pressure one during adiabatic decompression above the critical temperature
27 of direct transformation; and (2) quenching below the critical temperature of direct transformation

28 within the stability field of the high-pressure phase and further decompression. The aggregates with
29 plagioclase composition ($Ab_{81.1}An_{14.9}Or_{4.1}$) occur in host-rock fragments near (or inside) of the
30 SMV, and have a radial, concentric "spherulite-like" microstructure previously described in the
31 Novosibirsk meteorite, and which is very similar to the texture of tissintite in the Tissint martian
32 meteorite. It is likely that jadeite is related to crystallization of the SMV and could have formed
33 from albitic feldspar (plagioclase) melt at 13.5–15.0 GPa and ~2000 °C.

34

35 Keywords: L6 chondrite Pervomaisky, olivine high-Ca pyroxene majorite-pyroxene assemblage,
36 jadeite, shock-melt vein history.

37

38 INTRODUCTION

39 Heavily shocked chondrites are unusual natural objects, containing a variety of assemblages
40 of high-pressure polymorphs of rock-forming minerals (olivine, pyroxene and feldspar) formed by
41 shock metamorphism during collisions of meteorite parent bodies. The shock events caused the
42 melting of the host-rock, forming shock-melt veins (SMVs). The SMVs are the result of a
43 combination of several processes: the compaction of pore space (e.g. Wünnemann et al. 2008),
44 friction/shear heating (e.g. Langenhorst et al. 2002; Borgert et al. 2003) and localized stress and
45 temperature at the interfaces of mineral grains (e.g. Stöffler et al. 1991). High-pressure and high-
46 temperature phases in the shocked chondrites appear inside or near the SMVs.

47 Majorite is a high-pressure polymorph of pyroxene with garnet structure. Static experiments
48 at high-pressure and high-temperature conditions indicate that the stability field of majorite in the
49 system $MgO-SiO_2 \pm FeO \pm Al_2O_3$ is at about 15–23 GPa at temperatures above 1000 °C (e.g.,
50 Ringwood 1967; Ringwood and Major 1971; Akaogi and Akimoto 1977; Irifune 1987; Ohtani et al.
51 1991; Akaogi et al. 2002; Gasparik 2003). However, the formation of majorite-pyroxene solid

52 solutions related to pyroxene dissolution in a garnet structure begins at ~9 GPa (Ringwood 1967;
53 Akaogi and Akimoto 1977).

54 Majorite-pyrope was found in SMV of L6 chondrite Coorara (Smith and Mason 1970)
55 coexisting with ringwoodite. Majorite has been described in many shocked chondrites along with
56 other high-pressure minerals, such as wadsleyite, ringwoodite, akimotoite, MgSiO₃-perovskite
57 (bridgmanite), magnesiowüstite, lingunite, etc. (Price et al. 1979; Chen et al. 1996; Kimura et al.
58 2000; Xie et al. 2001; Kimura et al. 2003; Tomioka and Kimura 2003; Ohtani et al. 2004; Zhang et
59 al. 2006; Ozawa et al. 2009; Miyahara et al. 2011; Acosta-Maeda et al. 2013; Tschauner et al. 2014).

60 Two types of majorite have been recognized. One forms by melting of the host rock and its
61 crystallization under high-pressure conditions, forming the SMVs with other minerals (e.g., Smith
62 and Mason 1970; Chen et al. 1996; Xie et al. 2001; Ohtani et al. 2004). This type of majorite is
63 enriched in Al, Ca and Na compared with low-Ca pyroxene in the host-rock, and usually forms
64 micron-size idiomorphic grains inside the matrix of the SMVs. Another type of majorite forms by a
65 solid-state transformation of pyroxene in the coarse fragments of the host rock enclosed in the
66 SMVs (Chen et al. 1996; Xie et al. 2001 Ohtani et al. 2004; Zhang et al. 2006; Xie and Sharp 2007).
67 The chemical composition of such majorite is similar to that of pyroxene in the host rock.

68 The meteorite shower of the Pervomaisky L6 ordinary chondrite fell on the territory of
69 Vladimir Region (Russia) on December 26, 1933 near Pervomaisky village. Pervomaisky belongs to
70 the light-black species of chondrites (Britt and Pieters 1994). Semenenko and Golovko (1994)
71 determined compositions of olivine, Ca-rich pyroxene, majorite-pyrope, feldspar, glass, chromite,
72 whitlockite, kamacite and troilite, and they found organic matter (aliphatic hydrocarbons and
73 carbonyl- and N-containing compounds) in the SMVs. Majorite-pyrope in Semenenko and Golovko
74 (1994) was identified as Ca-poor pyroxene. No pressure estimation of majorite-pyrope formation
75 conditions were reported. Later, Collerson et al. (2010) proposed a new majoritic barometer and

76 estimated impact pressure in Pervomaisky as 21.9 GPa on the basis of the chemical composition of
77 majorite-pyrope garnet published by Semenenko and Golovko (1994).

78 In this study, we conducted a detailed mineralogical investigation of the SMVs of the
79 Pervomaisky L6 chondrite by scanning electron microscopy and micro-Raman spectrometry. As a
80 result, we found high-pressure minerals jadeite and majorite-pyrope. The majorite-pyrope coexists
81 with olivine and high-Ca pyroxene in the SMV matrix, which is an unusual assemblage for SMV
82 matrix of chondrites. We discuss the conditions and the sequence of its formation.

83

84 MATERIALS AND METHODS

85 The sample of Pervomaisky was provided from the working collection of G. M. Ivanova
86 (former scientific secretary of the Commission on Cosmic Meteorites and Dust at the Presidium of
87 SB RAS). Sample preparation and investigation were done at the Department of Earth Science,
88 Graduate School of Science, Tohoku University, Sendai, Japan. Petrologic observations were
89 conducted using the polished chip samples under an optical microscope. Fine textural observations
90 were conducted with the Field-Emission Scanning Electron Microscope (FE-SEM), JEOL JSM-
91 7001F. Accelerating voltage and probe current were 15 kV and 1.4 nA, respectively. Chemical
92 compositions of minerals were provided by the Energy-Dispersive X-ray Spectrometry (EDS) using
93 an INCA Energy (Oxford Instruments) microanalysis system attached to the FE-SEM. Counting
94 times for spectra collection were 30–60 seconds. Simple oxides - Al₂O₃ (for Al), MgO (for Mg);
95 silicates - wollastonite, CaSiO₃ (for Si, Ca), albite, NaAlSi₃O₈ (for Na), orthoclase KAlSi₃O₈ (for
96 K); metals – Cr, Ti, Fe, Mn, and GaP (for P), pyrite FeS₂ (for Fe, S) were used as the standards (see
97 Lavrent'ev et al., 2015 for the details of the analytical procedure). Jadeite was additionally analysed

98 by the area scan with shorter collection time (10-20 seconds) to minimize alkali loss during
99 measurements.

100 Phase identification of minerals and glasses was conducted with the laser micro-Raman
101 spectrometer, JASCO NRS-5100 (wavelength 532 nm). The laser power was 6.7 mW, and the size
102 of the laser beam was $\sim 1\text{--}2\ \mu\text{m}$ in diameter. The Raman signal from the sample was collected for
103 60–120 seconds and accumulated twice for each point in the spectral range of $140\text{--}1206\ \text{cm}^{-1}$.
104 Raman shift was calibrated with the peak of silicon standard at $520.5 \pm 0.5\ \text{cm}^{-1}$.

105

106 **RESULTS**

107 The Pervomaisky chondrite belongs to the light-dark type of chondrites (Britt and Pieters,
108 1994). The light part was studied in this work. The host rock has an equilibrated chondrite texture
109 and includes pervasive SMVs with widths of $10\text{--}300\ \mu\text{m}$ (Fig. 1). The major constituent minerals of
110 the host rock are olivine (Fa_{25}), low-Ca pyroxene ($\text{En}_{78}\text{Fs}_{21}\text{Wo}_1$), high-Ca pyroxene ($\text{En}_{47}\text{Fs}_8\text{Wo}_{45}$),
111 maskelynite (former plagioclase) ($\text{Ab}_{83}\text{An}_{10.3}\text{Or}_{6.8}$) (Table 1), Fe-Ni metal, troilite, chromite and rare
112 phosphates. Olivine, pyroxene and chromite are intersected by numerous, irregular cracks. All
113 original plagioclase has been transformed to maskelynite. The SMVs consist of fine-grained silicate
114 matrix with troilite and Fe-Ni metal blobs, and include coarse-grained fragments of the host-rock
115 (Fig. 1b). The chemical compositions of olivine and pyroxene in the fragments are almost identical
116 to those in the host rock (Table 1). The analyses of maskelynite show low totals, which may be
117 related to the deficit of Na_2O due to partial volatile loss under the electron beam. We performed
118 corrections of Na contents in previous experimental work (Shatskiy et al., 2013a,b). A maskelynite
119 analysis with a corrected Na_2O content is shown in Table 1.

120 The Raman spectra of the SMV matrix contain peaks corresponding to majorite-pyropite (at
121 $\sim 928 \text{ cm}^{-1}$), olivine (at ~ 822 and $\sim 855 \text{ cm}^{-1}$) and high-Ca pyroxene (broad peaks at ~ 668 and ~ 1014
122 cm^{-1}) (spectrum R1 in Fig. 2). Olivine and high-Ca pyroxene peaks are present in all spectra
123 containing majorite-pyropite peaks. There are probably micron-sized majorite-pyropite grains
124 submerged in the mixture of submicron olivine and high-Ca pyroxene grains, and it is impossible to
125 get Raman spectra only with garnet peaks due to the larger diameter ($\sim 2 \mu\text{m}$) of the Raman laser
126 beam.

127 Majorite-pyropite occurs as idiomorphic grains with sizes up to $3 \mu\text{m}$ (Fig. 1c). The largest
128 grains occur in the central parts of the SMVs. The average composition of majorite-pyropite is
129 $\text{Na}_{0.07}\text{Ca}_{0.19}\text{Mg}_{2.90}\text{Mn}_{0.03}\text{Fe}_{0.72}\text{Cr}_{0.02}\text{Al}_{0.41}\text{Si}_{3.77}\text{O}_{12}$. It corresponds to about 79 mol% of majorite
130 component in the majorite-pyropite solid solution. Majorite-pyropite is enriched in Al_2O_3 (4.68–5.10
131 wt%), CaO (2.16–2.75 wt%) and Na_2O (0.45–0.59 wt%) relative to low-Ca pyroxene in the host-
132 rock (0.18–0.26 wt% Al_2O_3 ; 0.43–0.94 wt% CaO) (Table 1). In the centers of the SMVs (Figs. 1d
133 and 1e), we found black, rounded, fine grains up to $5 \mu\text{m}$ and light interstitial material between them
134 (Fig. 1e). The FE-SEM-EDS analyses obtained from the center of a zoned grain (Fig. 1e) shows
135 high-Ca pyroxene composition $\text{Na}_{0.05}\text{Ca}_{0.70}\text{Mg}_{1.04}\text{Mn}_{0.01}\text{Fe}_{0.19}\text{Cr}_{0.02}\text{Al}_{0.05}\text{Ti}_{0.01}\text{Si}_{1.95}\text{O}_6$
136 ($\text{En}_{53.8}\text{Fs}_{9.6}\text{Wo}_{36.6}$). Light grains interstitial to the black grains (Fig. 1e) have olivine (Fa_{25-35})
137 composition. Magnesio-wüstite and high-pressure polymorphs of olivine were not found in
138 Pervomaisky.

139 We identified plagioclase-composition aggregates with jadeite (jadeite aggregates) as 5–20
140 μm fragments in the SMVs (Figs. 1b and 1c). Jadeite aggregates are slightly enriched in Fe, Ca and
141 Na and depleted in K compared with maskelynite (Table 1). All jadeite aggregates have radial,
142 concentric, "spherulite-like" microstructures, which are similar to those previously described in

143 jadeite aggregates from the Novosibirsk meteorite (Bazhan et al. 2017). The Raman spectra obtained
144 from the jadeite aggregate (peaks at ~ 1038 , ~ 699 , ~ 579 , ~ 526 and ~ 379 cm^{-1}) also include peaks of
145 majorite-pyrope (at ~ 928 cm^{-1}), olivine (at ~ 823 and ~ 855 cm^{-1}) and a small peak of high-Ca
146 pyroxene (at ~ 665 cm^{-1}) (spectrum R2 in Fig. 2). Raman peaks of majorite-pyrope, olivine and high-
147 Ca pyroxene in jadeite spectrum appears from the surrounding minerals.

148

149 **DISCUSSION**

150 ***PT*-conditions for SMV formation in the Pervomaisky meteorite**

151 The shock event caused melting of the chondrite, forming the SMVs. The growth of
152 majorite-pyrope grains could occur during a cooling stage after crystallization and quenching of
153 olivine–high-Ca pyroxene melt portion. *P-T* conditions of SMV formation in Pervomaisky can be
154 obtained from the phase assemblage of majorite-pyrope + olivine + high-Ca pyroxene in the SMV
155 matrix. The phase diagrams of the CaO–MgO–Al₂O₃–SiO₂ (CMAS) system (Fig.6 in Gasparik
156 1996) show that this assemblage can be stable at 7–16 GPa and 500–2150 °C. The stability field of
157 the present assemblage is limited to 14–15 GPa and 1750–2150 °C due to its content of 79 mol%
158 majorite component in majorite-pyrope garnet (Fig. 6 in Gasparik 1996).

159 The estimated *P-T* conditions are higher than the stability field of olivine (Fa₂₅) in the host-
160 rock fragments in the SMV. Olivine should have transformed to its high-pressure polymorphs
161 (wadsleyite and ringwoodite), according to the relevant phase diagram (Fig. 3; Akaogi et al. 1989).
162 However, these phases are not found in Pervomaisky. To explain this discrepancy we propose two
163 possibilities: (1) the pressure was lower than estimated; and (2) wadsleyite/ringwoodite back-
164 transformed to olivine during a cooling + decompression stage.

165 The upper limit of P - T stability of olivine fragments (Fa_{25} ; Table 1) in the SMV would be
166 ~ 13.5 GPa and ~ 2150 °C, which is at the intersection of olivine/(olivine + ringwoodite) transition
167 line (from Akaogi et al., 1989) and the liquidus curve of olivine in the melting phase diagrams of
168 anhydrous peridotite KLB-1 (Zhang and Herzberg 1994; Fig. 3). The estimated pressure for the
169 SMV matrix assemblage formation (~ 14 – 15 GPa) and the upper pressure limit for olivine survival in
170 the fragments (~ 13.5 GPa) are very close and do not contradict each other. According to the static
171 kinetic experiments of olivine-wadsleyite transformations (e.g. Kubo et al. 2004), micron-size grains
172 of wadsleyite appear after several minutes in experiments at 13.6 – 15.6 GPa and 1000 – 1100 °C. The
173 overpressure range at these conditions was 0.3 – 2.5 GPa (Kubo et al. 2004). Thus, we suggest that
174 the pressure range of the SMV mineral assemblage in Pervomaisky is 13.5 – 15.0 GPa.

175 The majoritic barometer (Collerson et al. 2010) gives pressure estimations of 21 and 22 GPa,
176 using the present and the Semenenko and Golovko (1994) majorite-pyrope compositions (Table 1),
177 respectively. It can be argued that the majoritic barometer from Collerson et al. (2010) does not
178 consider mineral assemblages and may overestimate pressures. For example, in the pressure-
179 composition phase diagram (Gasparik 2003; Fig. 4), it can be shown that the same chemical
180 composition (i.e., 79 mol% of majorite in Pervomaisky) corresponds to five different assemblages at
181 different pressures.

182 We think that our pressure estimations of 13.5 – 15.0 GPa for Pervomaisky are more reliable
183 than the 21 GPa of Collerson et al. (2010), because at such high pressure, wadsleyite or ringwoodite
184 should be present. We plotted pressure estimations versus pyrope contents in majorite-pyrope garnet
185 for different chondrites (Fig. 4). The P - T conditions for shock vein formation in Pervomaisky are the
186 lowest among all known majorite-bearing H- and L- chondrites (Table 2, Fig. 4).

187

188 **The back-transformations of high-pressure minerals to low pressure phases in shocked**
189 **chondrites**

190 We could not avoid the fact that the estimated conditions (13.5–15.0 GPa) include the
191 olivine/(olivine + ringwoodite) phase boundary and there is a possibility that olivine in the
192 fragments is the result of wadsleyite/ringwoodite back-transformation during a cooling-
193 decompression stage. For example, Kimura et al. (2003) reported the back-transformations of the
194 high-pressure minerals in a SMV (ringwoodite, wadsleyite, akimotoite, and majorite) to the low-
195 pressure polymorphs as a result of secondary external heating (up to 1600 °C) of the SMV close to
196 the fusion crust during the atmospheric passage of Yamato 75267 meteorite. The estimated time of
197 the back-transformation in Yamato 75267 was several seconds (Kimura et al. 2003). In addition,
198 Chen et al. (1998) showed that the back-transformation depends on the cooling rate. It was found
199 that no back-transformation of ringwoodite to olivine and majorite to pyroxene in the Sixiangkou
200 meteorite occurred at the cooling rate >10000 °C/s. The partial back-transformation of the same
201 minerals occurred in the Peace River meteorite at the cooling rate of 1000–2000 °C/s and the
202 transformation was mostly completed at cooling rate <500 °C/s in the Mbale meteorite.

203 The finding of the back-transformation features of high-pressure minerals in the meteorites,
204 such as iron zoning in olivine grain as relics of wadsleyite/ringwoodite crystallization from the melt
205 in Mbale (Chen et al. 1998), remnant features of polycrystalline grain boundary nucleation and
206 intracrystalline ringwoodite lamellae formation, or sharp boundaries between areas containing low-
207 pressure and high-pressure minerals (Kimura et al. 2003), are difficult to detect. We did not find
208 clear evidence of the back-transformation of olivine high-pressure polymorphs in Pervomaisky.

209

210 ***PT*-paths for the survival or back-transformation of ringwoodite and wadsleyite**

211 The experimental data of Nishi et al. (2010) show that the time required for the back
212 transformation of majorite to pyroxene is longer than previously reported (Kimura et al. 2003; Chen
213 et al. 1998). According to their data, the growth of 5 μm grains of high-Ca pyroxene during back
214 transformation of majorite at 7 GPa and 1100 $^{\circ}\text{C}$ takes 150 minutes. This is significantly longer
215 relative to the duration of the impact events in different chondrites, which was estimated as 0.04–4
216 seconds (Ohtani et al. 2004; Beck et al. 2005; Xie et al. 2006). The absence of ringwoodite and
217 wadsleyite and survival of majorite-pyroxene in the meteorite are probably related to the different
218 kinetics of the back-transformations of these minerals during the post-shock period of time at rapid
219 decompression and relatively slow cooling of the SMV, i.e. adiabatic decompression stage. In
220 Figure 3, a solid arrow (path 1) schematically shows this scenario.

221 The preservation of metastable phases during pressure and temperature release from initial
222 shock conditions requires quenching below the breakdown temperature before complete pressure
223 release (Sharp et al. 2003, Ohtani et al. 2004). Suzuki et al. (1980) noticed that wadsleyite (β -
224 Mg_2SiO_4) promptly transforms to olivine at 900 $^{\circ}\text{C}$ and ambient pressure. Ming et al. (1991) showed
225 that ringwoodite (Mg_2SiO_4) transforms to olivine in 1.3 h at the same P - T condition. Their studies
226 suggest that a temperature of 900 $^{\circ}\text{C}$ would be a critical threshold for the survival of high-pressure
227 polymorphs of olivine in the chondrites at the cooling-decompression stage, when they passed
228 through the metastable pressure range. In Figure 3, a dashed arrow (path 2) schematically shows this
229 scenario. The transformation of ringwoodite/wadsleyite to olivine occurs quickly during adiabatic
230 decompression above 900 $^{\circ}\text{C}$ (path 1 in Fig.3).

231

232 **Jadeite chemical composition and texture**

233 The chemical composition of jadeite aggregates in Pervomaisky is closer to plagioclase
234 stoichiometry than that of jadeite (Table 1). It is related to silica excess in jadeite compositions and

235 was reported previously for many jadeite-bearing chondrites (e.g. Kimura et al. 2000; Zhang et al.
236 2006; Ozawa et al. 2009, 2014; Miyahara et al. 2013; Bazhan et al. 2017). Silica accompanies
237 jadeite formation by plagioclase (albite) dissociation reaction: albite = jadeite + silica at pressures
238 above 3 GPa and 1000 °C (Birch and LeComte 1960; Bell and Roseboom Jr 1969; Liu 1978;
239 Holland 1980). Kubo et al. (2010) and Miyahara et al. (2013) provided detailed investigations of
240 jadeite formation from albite in experiments and in chondrites, respectively. Kubo et al. (2010)
241 suggested that amorphous plagioclase breaks down into jadeite and SiO₂ at high temperatures and
242 pressures, and nucleation of SiO₂ (stishovite) is significantly delayed compared with jadeite.
243 Miyahara et al. (2013) studied jadeite-bearing chondrites and found that areas containing jadeite
244 consist of two phases: jadeite + residual amorphous material. The average SiO₂ contents in jadeites
245 recovered from the chondrite samples Y-791384 and Y-75100 (Miyahara et al. 2013) and from the
246 synthetic sample s1644 (Kubo et al. 2010) are 61.2 wt%, 61.8 wt%, and 57.4 wt% respectively.
247 Apparently, the excess of SiO₂ in the chemical composition of jadeite in Pervomaisky (65.7 wt%
248 Table 1), is also associated with the presence of SiO₂-rich microinclusions.

249 The textures of "spherulite-like" jadeite aggregates in Pervomaisky (Fig. 1c) and Novosibirsk
250 (H5/6) (Bazhan et al. 2017) chondrites appear to be very similar to each other, with wormy and
251 rimming textural forms of tissintite, - vacancy-rich, high-pressure clinopyroxene (Ca,Na,□)AlSi₂O₆
252 recently reported in the Tissint Martian meteorite (Ma et al. 2015), but are different in chemical
253 compositions. The composition of jadeite in Pervomaisky and Novosibirsk correspond to An₁₅ and
254 An₁₄, respectively, whereas tissintite from Tissint is An₅₈₋₆₉. Jadeite and tissintite aggregates with
255 new textures were found in maskelynite enclosed by, or within ~25 μm of, a shock melt
256 vein/pocket. Ma et al. (2015) proposed two main possibilities of tissintite formation during an
257 impact event via crystallization from a plagioclase melt: (1) as a stable liquidus phase; and (2) as a
258 subliquidus phase due to enhanced nucleation rates. According to Ma et al. (2015), the distance of

259 ~25 μm could be critical to tissintite formation. The grains of maskelynite located more than ~25
260 μm from a melt pocket either did not melt or were not heated sufficiently to allow for tissintite
261 crystallization. The second option is similar to that proposed by Bazhan et al. (2017) for “spherulite-
262 like” jadeite in the Novosibirsk chondrite. They assumed that “spherulite-like” texture of jadeite is
263 the result of nucleation and growth of submicron jadeite crystals from plagioclase melt along the
264 grain boundary with surrounding minerals (olivine and high-Ca pyroxene). “Spherulite-like” jadeite
265 in Pervomaisky and Novosibirsk and wormy and rimming tissintite in Tissint are not the same
266 mineral, since they have different Raman spectra and chemical compositions, Tissintite could be
267 formed only from a Ca-rich plagioclase melt ($> \text{An}_{33}$), and is unlikely to be found in ordinary
268 chondrites (like Pervomaisky) with Na-rich plagioclase (Ma et al. 2015). Despite these differences,
269 they have similar formation mechanisms, P - T conditions and aggregate textures.

270

271 ***PT*-condition of jadeite formation**

272 The minimum pressure of albite dissociation reaction to jadeite + silica is 3 GPa (Birch and
273 LeComte 1960; Bell and Roseboom Jr 1969; Liu 1978; Holland 1980). The silica phase changes
274 from quartz to coesite and stishovite with increasing pressure. At pressures of 19–23 GPa, the
275 jadeite + stishovite assemblage transforms to lingunite ($\text{NaAlSi}_3\text{O}_8$ -hollandite) or to NaAlSiO_4 (with
276 CaFeO_4 -type structure) + stishovite (Tutti 2007). Thus, the pressure range of jadeite stability is 3–19
277 GPa (Fig. 3; Ozawa et al. 2014).

278 Two types of jadeite have appeared in shocked meteorites. The first one has a chemical
279 composition which is nearly identical to albitic feldspar (plagioclase), maskelynite or lingunite in
280 the host-rocks or in the host-rock fragments in SMVs, and forms through a solid-state
281 transformation (sst) of these phases (Ozawa et al., 2009; Miyahara et al., 2013). The SMVs of
282 chondrites containing this type of jadeite (Jd_{sst}) also include majorite-pyropite, wadsleyite,

283 ringwoodite and akimotoite (Table 2). The P - T fields of Jd_{sst} in Figure 3 are $Jd_{sst}+Sti$ and $Jd_{sst}+Coe$,
284 and correspond to the PT -fields of majorite-pyrope, wadsleyite, and ringwoodite. The Jd_{sst} formation
285 does not contradict the survival of the high-pressure minerals by path 2 in Figure 3.

286 The composition of the second type of jadeite is different from that of plagioclase in the
287 host-rocks and forms from albite melt slightly contaminated by surrounding silicates (Jd_{melt} ; Fig.3).
288 Although it is not clearly seen from melt pocket images, jadeite is always accompanied by tiny
289 amorphous SiO_2 -bearing residue and has high- SiO_2 contents (e.g. Miyahara et al., 2013). Jadeite in
290 Pervomaisky belongs to this type and is similar to that in the Chelyabinsk (Ozawa et al. 2014) and
291 Novosibirsk (Bazhan et al. 2017) chondrites. Jadeite in Pervomaisky could have formed at 13.5–
292 15.0 GPa and ~ 2000 °C, as shown in Fig. 3, in the beginning of cooling stage of the paths 1 and 2.

293

294 **IMPLICATIONS**

295 We propose two possible scenarios (PT -paths) for olivine fragments found in majorite-
296 pyrope-bearing SMV silicate matrix using the KLB-1 peridotite melting phase diagram with the
297 boundaries of olivine and albite high-pressure phases (Fig. 3). Path 1 characterizes the history of
298 mineral formation/transformation during cooling and adiabatic decompression stages (Fig. 3). A
299 hypothetical ringwoodite/wadsleyite + majorite-pyrope + high-Ca pyroxene assemblage would form
300 from chondrite melt at 14–15 GPa and 1750–2150 °C at cooling of the SMVs with subsequent
301 jadeite formation in plagioclase-bearing melt pockets. The back-transformation of
302 ringwoodite/wadsleyite to olivine (in the fragments and in the SMV matrix) could occur during an
303 adiabatic decompression stage at the end of path 1 (Fig. 3). The survival of majorite-pyrope can be
304 explained by the different kinetics of the back-transformation of ringwoodite/wadsleyite to olivine
305 and majorite-pyrope to pyroxene. The survival of ringwoodite/wadsleyite could occur along path #2
306 that is an extension of the cooling stage of path #1 below 900 °C and excludes the adiabatic stage

307 (Fig. 3). Whether jadeite formed from the melt or by the solid-state transformation of plagioclase
308 depends on the shock temperature and is consistent with both scenarios (Fig. 3). The SMVs in the
309 Mbale chondrite (Chen et al. 1998) would have evolved along the *PT*-path #1, whereas those of
310 Tenham, Y-75100, Suizhou, Y-75267, Y-791384, Sixiangkou, Y-74445, Taiban and Sahara 98222
311 correspond to the *PT*-path #2.

312 We suggest that our pressure estimation (13.5–15.0 GPa) for majorite-pyrope assemblage
313 formation in the SMV of Pervomaisky is the lowest among known majorite-bearing chondrites. To
314 our knowledge, this is the first report of the coexistence of highly majoritic garnet with olivine and
315 high-Ca pyroxene in the SMV matrix of chondrites.

316

317 **ACKNOWLEDGMENTS**

318 We thank S. Simon and two anonymous reviewers for constructive comments and G.M.
319 Ivanova and N.M. Podgornykh for providing samples of the Pervomaisky meteorite. The authors are
320 also grateful to Y. Ito for preparation and chemical analyses of meteorite samples. The work was
321 conducted under the program of Ministry of Education and Science of Russian Federation (No
322 14.B25.31.0032). KL thanks to support of his research from Russian Science Foundation (project
323 No 15-17-30012).

324

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- 474

475 **Table and Figure captions**

476 **Table 1.** Average chemical compositions of the minerals of the Pervomaisky chondrite.

477 **Table 2.** High-pressure mineral assemblages in different chondrites.

478
479 **Figure 1.** Back-scattered electron images of the Pervomaisky L6 chondrite. **(a)** a shock-melt vein
480 (SMV) that cuts the host rock; **(b)** a SMV with an entrained host rock fragment; **(c)** an aggregate of
481 jadeite in a SMV silicate matrix; R1, R2 and R3 in (c) indicate points where the Raman spectra in
482 Fig. 2 were obtained; **(d)** the matrix of the SMV with majorite-pyroxene – olivine – high-Ca pyroxene
483 assemblage; **(e)** magnified image of the area marked by a rectangle in (d). Cpx, high-Ca pyroxene;
484 En, enstatite; Fa, fayalite; Fs, ferrosilite; Jd, plagioclase composition aggregate with jadeite; Maj-
485 Prp, majorite-pyroxene; Msk, maskelynite; Ol, olivine; Opx, low-Ca pyroxene; Tro, troilite, Wo,
486 wollastonite.

487 **Figure 2.** Representative Raman spectra from different points of fine-grained silicate matrix of the
488 SMV of the Pervomaisky chondrite. Analytical points (R1 and R2) are marked in Figure 1c. R1
489 spectrum from - majorite-pyroxene – olivine – high-Ca pyroxene assemblage. R2 spectrum from
490 jadeite aggregate. Cpx, high-Ca pyroxene; Jd, jadeite; Maj-Pyr, majorite-pyroxene; Ol, olivine.

491 **Figure 3.** *PT*-conditions for majorite-pyroxene + olivine + high-Ca pyroxene assemblage formation
492 (dark grey bar) in the SMV of the Pervomaisky chondrite. Liquidus and solidus of KLB-1 peridotite
493 are from Zhang and Herzberg (1994); Black bold lines represent phase boundaries for olivine (Fa₂₅)
494 (Akaogi et al. 1989) in Pervomaisky. Grey lines represent phase boundaries for albite (after Bell and
495 Roseboom 1969; Akaogi and Navrotsky 1984; Tutti 2007; Ozawa et al. 2014). Light grey area is
496 stability field of jadeite. The solid arrow #1 schematically shows a possible *PT*-path of the back-

497 transformation of ringwoodite and wadsleyite to olivine in Pervomaisky at the post-shock, adiabatic
498 decompression stage. The dashed arrow #2 schematically shows a possible P - T path of the survival
499 of ringwoodite and wadsleyite in chondrite in a post-shock, cooling-decompression stage. Transition
500 temperature from ringwoodite/wadsleyite to olivine at ambient pressure is 900 °C (see text). Ab,
501 albite, Coe, Coesite; Grt, garnet (majorite-pyrope); Jd, jadeite; Jd_{melt}, jadeite formed from the melt;
502 Jd_{sst}, jadeite formed by solid-state transformation (SST) of albite; L_{Ab}, albite melt; Lng, lingunite;
503 Mws, magnesiowüstite; Ol, olivine; Qz, quartz; Rwd, ringwoodite, Sti, stishovite; Wds, wadsleyite.

504 **Figure 4.** A portion of the P-X phase diagram for the enstatite-diopside-pyrope garnet join and for a
505 pyrope content of 0-30 mol% at 1650 °C (Fig. 3.9 in Gasparik 2003). The bars show pyrope/majorite
506 content in majorite-pyrope garnets in the SMVs and pressure estimation of high-pressure mineral
507 assemblage formation in different chondrites (Table 2): (1) Pervomaisky (Present study); (2)
508 Suizhou (Xie et al. 2001); (3) Sixiangkou (Zhang et al. 2006); (4) Yamato 791384 (Ohtani et al.
509 2004); (5) Yamato 75100 (Tomioka et al. 2003); (6) Yamato 75267 (Kimura et al. 2003); (7)
510 Tenham (Langenhorst et al. 1995); (8) Tenham (Tomioka and Kimura 1999).

511

512 **TABLE 1.** Average chemical compositions of the minerals of the Pervomaisky chondrite

oxides wt%	Ol		Cpx		Opx		Melt	Maj-Prp	Opx _{Sem}	Msk	Jd		
	h-rock	fragm	h-rock	fragm	h-rock	fragm	vein	vein	vein	h-rock	fragm		
n	12	7	11	7	24	14	5	3	6	4	4		
SiO ₂	38.3(5)	38.1(5)	54.4(6)	54.5(4)	55.5(6)	55.9(8)	48.9(2.0)	52.9(41)	52.9	65.9(6)	65.7(5)		
TiO ₂	n.d.	n.d.	0.46(6)	0.51(10)	b.d.	b.d.	b.d.	b.d.	0.14	n.d.	-		
Al ₂ O ₃	n.d.	n.d.	0.52(8)	0.55(9)	b.d.	b.d.	3.12(38)	4.91(21)	2.56	20.7(3)	20.3(5)		
Cr ₂ O ₃	n.d.	n.d.	0.83(11)	0.82(11)	n.d.	n.d.	0.31(7)	0.43(13)	0.54	n.d.	-		
FeO	22.7(1)	22.7(6)	5.11(42)	5.43(13)	13.9(5)	14.4(1)	15.2(3)	12.0(1.0)	11.0	0.79(18)	1.56(9)		
NiO	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	b.d.	b.d.	n.d.	n.d.	-		
MnO	0.49(11)	0.45(5)	b.d.	b.d.	0.52(9)	0.54(7)	0.35(13)	0.47(5)	0.33	n.d.	-		
MgO	38.6(6)	38.9(7)	16.7(2)	16.8(3)	28.8(5)	29.0(3)	29.6(2.2)	26.7(72)	29.6	b.d.	0.30(11)		
CaO	n.d.	n.d.	21.9(6)	21.5(3)	0.72(13)	0.65(12)	2.10(57)	2.55(34)	1.88	1.78(8)	2.40(9)		
Na ₂ O	n.d.	n.d.	0.56(9)	0.54(3)	n.d.	n.d.	0.48(28)	0.51(7)	0.29	7.94	9.52(42)		
K ₂ O	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	b.d.	n.d.	n.d.	0.99(5)	0.62(8)		
Total	100.09	100.15	100.48	100.65	99.44	100.49	100.06	100.47	99.24	98.11	100.45		
Si	0.99	0.99	1.98	1.98	1.99	1.98	3.57	3.77	1.90	2.94	2.90		
Ti	-	-	0.01	0.01	-	-	-	-	-	-	-		
Al	-	-	0.02	0.02	-	-	0.27	0.41	0.11	1.09	1.06		
Cr	-	-	0.02	0.02	-	-	0.02	0.02	0.02	-	-		
Fe ²⁺	0.48	0.48	0.16	0.17	0.42	0.43	0.93	0.72	0.33	0.03	0.06		
Ni	-	-	-	-	-	-	-	-	-	-	-		
Mn	0.01	0.01	-	-	0.02	0.01	0.02	0.03	0.01	-	-		
Mg	1.52	1.52	0.91	0.91	1.54	1.54	3.22	2.84	1.59	-	0.02		
Ca	-	-	0.85	0.84	0.03	0.03	0.16	0.19	0.07	0.09	0.11		
Na	-	-	0.04	0.04	-	-	0.07	0.07	0.02	0.69	0.81		
K	-	-	-	-	-	-	-	-	-	0.06	0.04		
Total	3.00	3.00	3.99	3.99	4.00	3.99	8.26	8.05	4.05	4.89	5.00		
Oxygen	4	4	6	6	6	6	12	12	6	8	8		
<i>Fo</i>	75.2	75.3	<i>En</i>	47.3	47.5	77.6	77.2	74.7	75.7	79.6	<i>Ab</i>	83.0	84.6
<i>Fa</i>	24.8	24.7	<i>Fs</i>	8.1	8.6	21.0	21.5	21.5	19.2	16.6	<i>An</i>	10.3	11.8
			<i>Wo</i>	44.6	43.9	1.4	1.2	3.8	5.1	3.8	<i>Or</i>	6.8	3.6
P, GPa									21.3	21.9			

Note: Numbers in parentheses are one standard deviations of the last significant digits. n, number of analyses; h-rock, host-rock; fragm, host-rock fragments in the SMVs; P, GPa - pressure estimation using the majorite barometer of Collerson et.al. (2010); b.d., below detection limit. The detection limit for each element is 0.3 wt.% (Lavrent'ev et al. 2015); n.d., not determined. Abbreviations: Ab, albite; En, enstatite; Fa, fayalite; Fo, forsterite; Fs, ferrosilite; Jd, plagioclase composition aggregate with jadeite; Melt – average chemical composition of interstitials between majorite-pyroxene grains in the SMV matrix; Opx_{Sem} – average chemical composition of low-Ca pyroxene from the SMVs in Pervomayskiy (Semenenko and Golovko, 1994); Maj-Prp, majorite-pyroxene; Msk, maskelynite; Or, orthoclase; Wo, wollastonite.

513 **TABLE 2.** High-pressure mineral assemblages in different chondrites

Meteorite name	Type	High-pressure assemblage	SMV <i>PT</i> conditions		References
			P, GPa	T, °C	
Tenham	L6	Maj, Maj-Prp, Rwd, Akm, Prv, Mws	22–26	2000	Langenhorst et al. 1995; Tomioka and Fujino 1999
Sixiangkou	L6	Maj, Maj-Prp, Rwd, Akm, Mws, Lng, Jd _{sst}	20–24	2000	Chen et al. 1996; Zhang et al. 2006
Y-75100	H6	Maj-Prp + Opx + Ol, Wds, Rwd, Akm, Hol-Jd _{sst}	18–24	1900	Kimura et al. 2000 Tomioka and Kimura 2003; Miyahara et al. 2013
Suizhou	L6	Maj, Maj-Prp, Rwd, Hol, Tuite	18–22	1900	Xie et al. 2001, 2003
Y-75267	H6	Maj, Maj-Prp, Wds, Rwd, Akm, Hol	20	2000	Kimura et al. 2003
Y-791384	L6	Maj-Prp, Rwd, Akm, Hol, Jd _{sst}	18–23	2300	Ohtani et al. 2004; Miyahara et al. 2011, 2013
Y-74445	L6	Maj, Wds, Rwd, Akm, Hol, Jd _{sst}	17–24	2100	Ozawa et al. 2009
Taiban	L6	Maj+Cpx, Wds, Rwd, Lng, Jd _{sst}	17–20	1900	Acosta-Maeda et al. 2013
Pervomaisky	L6	Maj-Prp+Ol+Cpx, Jd _{melt}	13.5–15.0	1750–2150	This work
Sahara 98222	L6	Wds, Jd _{sst}	13–16	1900	Ozawa et al. 2009
Chelyabinsk	LL5	Jd _{melt}	3–12	<1700–2000	Ozawa et al., 2014
Novosibirsk	H5-6	Jd _{melt}	3–15	1400–2150	Bazhan et al., 2017

Note: *PT*-conditions in the SMV were estimated in previous studies. Abbreviations: Akm, akimotoite; Cpx, clinopyroxene; Hol, hollandite; Jd_{melt}, jadeite formed from the melt; Jd_{sst}, jadeite formed at the solid-state transformation of albitic feldspar; Lng, lingunite; Maj, majorite formed at the solid-state transformation of pyroxene; Maj-Prp, majorite-pyrope formed from the melt; Mws, magnesiowüstite; Ol, olivine; Opx, orthopyroxene; Prv, Mg-perovskite (bridgmanite), Rwd, ringwoodite; Wds, wadsleyite.

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