

1 **Igneous or Metamorphic? Hornblende Phenocrysts as Greenschist Facies**  
2 **Reaction Cells in the Half Dome Granodiorite, California**

3 **Revision 2**

4 **STEPHEN C. CHALLENGER<sup>1</sup> AND ALLEN F. GLAZNER**

5 Department of Geological Sciences, University of North Carolina, Chapel Hill, NC 27599-3315,  
6 U.S.A.

7 **ABSTRACT**

8 The Half Dome Granodiorite, Yosemite National Park, California, is recognized in the  
9 field by euhedral, fresh-looking, black hornblende phenocrysts up to 2 cm in length. This variety  
10 of granodiorite typifies intermediate-age hornblende-phyric units of Cretaceous nested plutonic  
11 suites in the Sierra Nevada batholith. Although only inclusions of feldspar are evident in hand  
12 sample, the phenocrysts are riddled with up to 50% inclusions of every major mineral found in  
13 the host granodiorite plus metamorphic minerals formed during cooling. Amphibole  
14 compositions within single phenocrysts vary from actinolite with less than 1 wt% Al<sub>2</sub>O<sub>3</sub> to  
15 magnesiohornblende with over 8 wt%. Elemental zoning within the amphibole is highly irregular  
16 on the  $\mu\text{m}$  scale, showing patches and polygonal zones with dramatically different compositions  
17 separated by sharp to gradual transitions. The chemical compositions of entire phenocrysts are  
18 equivalent to hornblende plus a small proportion of biotite, suggesting that the non-biotite  
19 inclusions are the result of metamorphism of the phenocrysts. Backscattered electron imaging  
20 shows evidence of brecciation which may have been the result of volume changes as hornblende

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<sup>1</sup> Present address  
Department of Marine, Earth, and Atmospheric Sciences  
NC State University  
Raleigh, North Carolina, USA 27695

21 was converted to actinolite. Pressure calculations using the Al-in-hornblende barometer show  
22 unreasonably wide variations on the  $\mu\text{m}$  scale that cannot have been produced by temperature or  
23 pressure variations during crystallization. These hornblende phenocrysts would thus be  
24 unsuitable for geobarometry, and caution must be used to avoid similarly zoned phenocrysts in  
25 the application of the Al-in-hornblende geobarometer.

## 26 INTRODUCTION

27 What are phenocrysts, and what do they signify? In both plutonic and volcanic rocks,  
28 crystals that are much larger than their surrounding matrix are generally interpreted as having  
29 grown early in a magma's crystallization history (e.g., Harker and Marr 1891; Crosby 1900;  
30 Kelley and Branson 1947; Vernon 1986; Straub and Martin-Del Pozzo 2001). In volcanic rocks  
31 this interpretation is generally unambiguous, because volcanic rocks are cooled rapidly from the  
32 magmatic state and thus freeze in magmatic texture, and experiments can duplicate observed  
33 volcanic phase assemblages and textures (e.g., Lofgren 1980; Johnson and Rutherford 1989).

34 However, textural interpretation is more challenging in plutonic rocks because such rocks  
35 are not quenched but rather cool over periods of up to millions of years (Coleman et al. 2004),  
36 allowing ample time for chemical and textural modification. For example, Davis et al. (2012)  
37 found that some plutons emplaced at shallow levels in the Sierra Nevada of California took up to  
38 10 m.y. to cool below 300 °C. This means that they could have spent millions of years at  
39 temperatures corresponding to the greenschist and amphibolite facies, and that temperature may  
40 have oscillated during cooling owing to emplacement of new increments of magma. In addition,  
41 phase equilibria studies show that some common phenocrystic phases, such as K-feldspar  
42 megacrysts that reach sizes over 10 cm (Gilbert 1906; Bateman 1961; Booth 1968), cannot be

43 early crystallization products (Johnson and Glazner 2010; Glazner and Johnson 2013). Thus,  
44 interpretation of phenocrysts in plutonic rocks can be ambiguous.

45 Hornblende and biotite commonly form large (cm-scale), euhedral phenocrysts in  
46 granodiorites. In contrast to K-feldspar, both experimental studies and the petrography of  
47 volcanic rocks show that they can be near-liquidus phases in intermediate magmas (Piwinski  
48 1968; Naney and Swanson 1980; Devine et al. 1998). For such phenocrysts, unlike K-feldspar  
49 megacrysts, large size and euhedral habit are not surprising, and their textural origin has  
50 remained unquestioned.

51 The Half Dome Granodiorite is the middle unit of the nested Late Cretaceous Tuolumne  
52 Intrusive Suite of Yosemite National Park, California (Fig. 1; Bateman and Chappell 1979;  
53 Bateman, 1992). As with most plutons, this unit was defined and mapped on the basis of its  
54 texture—primarily the presence of large, euhedral hornblende phenocrysts that can reach up to 2  
55 cm in length (Fig. 2; Dodge et al. 1968; Coleman et al. 2005). Hornblende is a particularly  
56 important phase in plutonic rocks because it offers one of the only barometers available for  
57 igneous rocks—the Al-in-hornblende geobarometer (Hammarstrom and Zen 1986; Anderson and  
58 Smith 1995). In a plutonic rock emplaced at a given depth and then cooled, core-to-rim Al  
59 zoning should be a function of slow isobaric cooling, with Al content primarily reflecting the  
60 pressure of emplacement.

61 Hornblende phenocrysts in the Half Dome Granodiorite commonly weather out intact  
62 (Fig. 2), and thus they can be sectioned through their centers in known orientations. In this study  
63 we examined a number of such phenocrysts. Far from displaying a simple story of isobaric  
64 cooling, these phenocrysts contain copious inclusions of other minerals and have extreme,  
65 irregular, fine-scale zoning in Al.

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## METHODS

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Euhedral hornblende phenocrysts ranging from 1 to 2 cm in maximum dimension were collected from grus developed on the Half Dome Granodiorite exposed in glacial pavement west of Tenaya Lake. On glacial pavements throughout this area, weathering releases intact hornblende and biotite phenocrysts (Fig. 2A). Intact phenocrysts also weather out on surfaces that were not glaciated, such as the top of Half Dome.

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Five large, intact, solid, fresh-appearing hornblende phenocrysts were mounted in epoxy, sawn, and polished. Because of their large size, euhedral form and easy removal from matrix, the phenocrysts can be precisely oriented relative to their crystal axes. Four were cut perpendicular to the **c**-axis and one perpendicular to the **a**-axis, all through their centers. The polished phenocrysts were examined in a Tescan VEGA 5136 scanning electron microscope at the University of North Carolina using backscattered electron (BSE) imagery, and X-ray maps were generated using a Sirius Si-drift detector and 4pi Revolution software.

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Element maps were exported as image files and run through a Gaussian low-pass filter with a kernel size of 9x9 to reduce noise. These images were then imported into ENVI GIS software for mapping. Regions of interest were placed on representative mineral inclusions, which were readily identified by their X-ray spectra. A supervised classification was then performed using the minimum distance method to identify and map all mineral inclusions in the images (Fig. 2C; Supplemental Fig. 1).

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Chemical analyses and high-resolution images were collected on a JEOL JXA-8530F field-emission Hyperprobe at Fayetteville State University with an accelerating voltage of 15 kV, calibrated against Astimex standards. Typical beam size was 1  $\mu$ m. Pressures were calculated

88 from these compositions using the Al-in-hornblende geobarometer as implemented by Anderson  
89 et al. (2008).

## 90 **RESULTS**

91 The hornblende phenocrysts host copious inclusions of every other mineral species in the  
92 rock (Figs. 2,3,4). These include abundant biotite largely altered to chlorite, magnetite,  
93 plagioclase, K-feldspar, apatite, and titanite, with minor amounts of quartz, albite, epidote,  
94 clinzoisite, zircon, and muscovite. Inclusions average 45 area % of the five mapped  
95 phenocrysts, with biotite+chlorite comprising 17 area % (Fig. 3).

96 There is little obvious pattern to the distribution of inclusions. Some biotite crystals are  
97 aligned with (001) parallel to crystal faces of the enclosing phenocryst, but most are not. In  
98 phenocryst 3 (Fig. 2), biotite+chlorite crystals seem to form a concentric zone about ½ diameter  
99 out from the center, whereas in phenocryst 5 (Fig. 5) there is a concentric zone of complex  
100 feldspar crystals at about the same distance from center.

101 BSE and microprobe analyses show that the amphibole zones of the phenocrysts are  
102 strongly zoned on a fine scale (Figs. 4,5,6), and that this zoning is not systematic. Spot analyses  
103 range from magnesiohornblende to actinolite (Fig. 6; Table 1). Despite the presence of non-  
104 hornblende amphibole compositions we will continue to refer to the amphiboles collectively as  
105 hornblende. Compositional parameters are tightly correlated above 2 wt% Al<sub>2</sub>O<sub>3</sub>; for example,  
106 MgO shows a strong negative correlation (Fig. 6), and FeO a strong positive correlation, with  
107 Al<sub>2</sub>O<sub>3</sub>. Because FeO and Al<sub>2</sub>O<sub>3</sub> exhibit strong positive correlation, BSE image brightness  
108 correlates directly with Al<sub>2</sub>O<sub>3</sub>, which varies by 5 wt% over a length scale of tens of μm (Fig. 7)  
109 in an irregular, patchy manner. A number of low-MgO, high-FeO points lie off this trend at low

110  $\text{Al}_2\text{O}_3$ , trending toward ferroactinolite (Fig. 6). These latter points are almost exclusively located  
111 immediately adjacent to zones of chloritized biotite.

112 The amphibole areas show a number of small-scale modes of zoning. These vary from  
113 sharply defined polygons to blurred zones with rounded edges to zig-zag patterns. Some blurring  
114 is likely due to slanted interfaces that are sharper than they appear. Some zones are uniform in  
115 composition, whereas others show internal compositional gradients (Fig. 7). In places the zoning  
116 pattern suggests healed fractures in which high-Al materials fill gaps in low-Al actinolite; this is  
117 particularly evident where compositional boundaries are zig-zag (Fig. 7A,B,C).

118 There is no indication of concentric compositional zoning as is commonly seen in  
119 igneous phenocrysts. For example, cm-scale hornblende phenocrysts in Montserrat andesite  
120 display concentric oscillatory zones that likely reflect mixing events (Rutherford and Devine  
121 2003; Humphreys et al. 2009). Such oscillatory zoning is common in hornblende in many  
122 volcanic systems (e.g., Bachmann and Dungan 2002; Sato et al. 2005; Humphreys et al. 2006).  
123 However, the phenocrysts in this study only show patchy, irregular zoning. Although  
124 compositional variability within the hornblende is extreme, there is no apparent pattern of  
125 compositional variation with distance from a phenocryst's center (Fig. 6), even along a smooth  
126 core-to-rim track.

127 Five complete 2-cm-long, euhedral phenocrysts, and three composites of 2-3 euhedral  
128 phenocrysts, were analyzed for major elements by X-ray fluorescence (Table 1). These  
129 phenocrysts were freed of crystals clinging to their faces, although feldspar inclusions  
130 intersecting the crystal surfaces were common. These analyses, which represent the bulk  
131 volumes enclosed by the phenocryst boundaries, lie at the high- $\text{Al}_2\text{O}_3$  end of the spot analyses  
132 for most elements but are significantly higher in  $\text{K}_2\text{O}$  and lower in  $\text{CaO}$  and  $\text{Na}_2\text{O}$ , showing

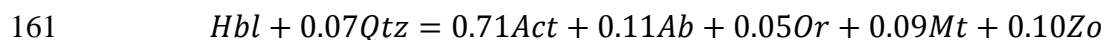
133 values intermediate between high-Al hornblende and biotite analyses from within the  
134 phenocrysts (Fig. 8). P<sub>2</sub>O<sub>5</sub> concentrations averaging 0.6 wt% indicate about 1.5 wt% apatite in  
135 the phenocrysts, assuming that the *P* content of the hornblende is on the order of 500 ppm or less  
136 (Bachmann et al. 2005).

## 137 **DISCUSSION**

138 The maps and imagery show that these hornblende phenocrysts contradict common  
139 assumptions about the nature of phenocrysts. Not only are they heavily included, but many of the  
140 included minerals are late-magmatic or subsolidus products in plutonic systems, such as K-  
141 feldspar. Because K-feldspar crystallizes late in magma of this composition (Johnson and  
142 Glazner 2010) the phenocrysts could not have grown around these inclusions if hornblende  
143 crystallized in a largely liquid magma. The extreme and irregular elemental zoning within the  
144 hornblende also cannot reflect compositional variations occurring during a single crystallization  
145 event. Furthermore, the huge range in amphibole composition within single phenocrysts makes  
146 the Al-in-hornblende geobarometer difficult to apply.

147 Hornblende phenocrysts from the Half Dome Granodiorite do not show the features  
148 predicted for unmodified crystals that grew in a largely liquid magma. Extreme textural  
149 modification is suggested by the alteration of biotite into chlorite, as well as the wide array of  
150 greenschist-grade mineral inclusions, including chlorite, albite and epidote, and the broad range  
151 of amphibole compositions. Slow cooling of incrementally emplaced plutonic rocks may have  
152 left them at pressure (*P*) and temperature (*T*) conditions within the amphibolite and greenschist  
153 facies for extended periods of time (Davis et al. 2012; Glazner and Johnson 2013). The  
154 compositions of whole phenocrysts are also essentially those of the high-Al end of measured  
155 hornblende compositions plus approximately 5-30 wt% biotite (Fig. 8), with the exception of Fe

156 which is slightly higher in the bulk analyses. This suggests suggesting that the included minerals  
157 developed from normal hornblende phenocrysts that contained inclusions of biotite and possibly  
158 magnetite, without any change in bulk composition. Using our measured high-Al hornblende  
159 (*Hbl*) and actinolite (*Act*) compositions with quartz (*Qtz*), end-member albite (*Ab*), K-feldspar  
160 (*Or*), magnetite (*Mt*), and zoisite (*Zo*), one such balanced reaction is



162 This reaction also provides an explanation for why quartz, which is abundant in the host rock, is  
163 rarely found in the inclusion population.

164 Conversion of hornblende to actinolite can induce a decrease in volume, causing  
165 microcracks within the crystal (Okamoto and Toriumi 2005). This may provide an explanation  
166 for the brecciated compositional zones in Figures 4 and 7. In most cases crack filling appears to  
167 consist of higher-Al (and thus higher-T) amphibole than the host material. However, the patterns  
168 are too complicated to reflect a single brecciation event. This suggests that repeated episodes of  
169 brecciation may have been caused by cyclical changes in temperature; cooling would drive the  
170 reaction to the right, causing cracking due to loss in volume, followed by heating that would heal  
171 the cracks with higher-Al amphibole. Such a scenario implies that temperatures oscillated across  
172 the greenschist-amphibolite facies boundary, consistent with the multi-million-year intervals  
173 between hornblende and biotite Ar ages (Davis et al. 2012). Highly angular zones may thus  
174 represent more recent episodes, while rounded or blurred zones show earlier episodes which  
175 have undergone diffusive relaxation.

176 Thus, despite their euhedral shape, the texture and mineralogy of the phenocrysts have  
177 been completely overwritten. Replacement of hornblende by actinolite and patchy zoning  
178 between actinolite and hornblende are typical of amphibole porphyroblasts (Gibbons and Horak



179 1984, Klein 1969, Grapes 1974) rather than phenocrysts, suggesting that the phenocrysts record a  
180 significant metamorphic history. The wide range of large, well-formed secondary minerals  
181 developed from the Half Dome phenocrysts is, however, a unique feature not typical of purely  
182 metamorphic or igneous amphiboles. This may be another anomalous texture caused by  
183 variations in temperature. Rapid crystal coarsening by thermal cycling is a well-documented  
184 phenomenon, and has been suggested as an important coarsening mechanism in petrology  
185 (Johnson and Glazner 2010). It may be that initially small or diffuse reaction products within the  
186 phenocrysts were condensed into larger crystals by this mechanism. It is also possible that the  
187 phenocrysts themselves were coarsened by this mechanism; however, given the common  
188 occurrence of volcanic hornblende phenocrysts it is possible that their initial size was simply the  
189 result of early growth.

190       Whatever their genesis, the strong and aconcentric Al zoning and development of  
191 secondary minerals observed in the phenocrysts indicate that pressure information is masked and  
192 difficult to extract from spot analyses of hornblende phenocrysts in the Half Dome Granodiorite.  
193 The actinolite end of the trend is clearly meaningless, but which Al content records the pressure  
194 is subject to interpretation. Hammarstrom and Zen (1986) discarded crystals which were altered  
195 to actinolite riddled with opaques, and Anderson and Smith (1995) omitted “Al-poor subsolidus  
196 amphiboles.” However, in our samples, where there is a continuum in Al from hornblende down  
197 to clearly metamorphic actinolite, there is no clear break in Al below which analyses can be  
198 confidently discarded (Gray et al. 2008)

199       The best estimate of the predominant hornblende composition lies where the biotite-  
200 addition lines intersect the main amphibole trend (Fig. 8). This intersection occurs at ~7 wt%  
201 Al<sub>2</sub>O<sub>3</sub> for most elements. This value can be used with the Al-in-hornblende barometer (Anderson

202 and Smith 1995) to calculate  $P$ . Figure 9 shows calculated  $P$  as a function of  $T$  for this Al  
203 content. Also shown are (1) hornblende-plagioclase thermobarometry on Half Dome  
204 Granodiorite (Gray et al. 2008) and (2)  $P$ - $T$  estimates from aplite dikes, using the method  
205 outlined in Putnam et al. (2015), which assumes that aplite whole-rock compositions are  
206 equivalent to liquids separated from a quartz-, feldspar-, and zircon-bearing crystalline mush,  
207 and that the magma was water-saturated with a water activity of unity. Aplite analyses from the  
208 Tenaya Lake region are from Glazner et al. (2008). These estimates scatter near the water-  
209 saturated granite solidus at low pressure ( $130 \pm 70$  MPa) using the Zr calibration of Watson and  
210 Harrison (1983); temperatures calculated using the Boehnke et al. (2013) calibration are  
211 systematically  $\sim 60$  °C cooler and well below the solidus. Solution of the hornblende  
212 thermobarometer is  $T$ -dependent, and isobaric cooling should cause a rimward decrease in Al, so  
213 it is difficult to determine an appropriate  $T$ , although it should be higher than those recorded by  
214 the near-solidus aplite dikes. At  $T > 730$  °C (the maximum  $T$  recorded by the aplite dikes),  $P$   
215 estimates for the hornblende are consistent with those of the aplites.

216         Gray et al. (2008) noted that Half Dome hornblende crystals range from  
217 magnesian hornblende to actinolite but did not explain how analyses were screened for  
218 thermobarometry. Presumably they used the higher-Al analyses. Our new data, Gray et al.'s  
219 results, and the aplite thermobarometry indicate cooling at  $P \sim 150$  MPa, and thus a depth of  
220 around  $6 \pm 2$  km assuming a crustal density of  $2700 \text{ kg/m}^3$ . Ague and Brimhall (1988) made a  
221 regional study of pressures calculated using the hornblende barometer. They reported no  
222 analyses of hornblende from the Half Dome Granodiorite, but Tenaya Lake lies in a regional  
223 low- $P$  zone ( $\sim 100$ - $200$  MPa), consistent with our data.

224 We have observed abundant biotite in the cores of hornblende in granodiorites and  
225 tonalites in the Tuolumne Intrusive Suite, and they are not uncommon throughout the Sierra  
226 Nevada (e.g., Frost and Mahood 1987). However, the pervasive occurrence of biotite cores in  
227 hornblende went largely unremarked in previous studies of the Tuolumne Intrusive Suite (e.g.,  
228 Bateman and Chappell 1979; Gray et al. 2008). If this relationship reflects order of  
229 crystallization then it contradicts the common assumption that biotite follows hornblende. For  
230 example, Abbott (1981) noted that “In a series of comagmatic intrusions related by  
231 differentiation, the biotite almost always appears later than the hornblende, in agreement with the  
232 reaction series of Bowen (1928). Certainly the sequence hornblende, biotite + hornblende, biotite  
233 in differentiated granites is sufficiently common that there is little need here for thorough  
234 documentation.” However, in many plutons the textural relationships between biotite and  
235 hornblende are ambiguous (e.g., Speer 1987), and magma mixing can disturb and invert textural  
236 relationships (Frost and Mahood 1987).

237 Smelik et al. (1991) presented transmission electron microscope evidence for fine-scale  
238 exsolution between hornblende and actinolite in a metagabbro. The altered amphibole consists of  
239 lamellae of hornblende and actinolite intergrown on a 5-15 nm scale. The beam size during our  
240 microprobe analyses was 1  $\mu\text{m}$  in diameter (with a somewhat larger activation volume), and thus  
241 would average over 100 or more lamellae of such dimensions. It is likely that the continuous  
242 range in composition from hornblende to actinolite (Fig. 6) reflects varying proportions of  
243 hornblende and actinolite lamellae rather than true compositional variation on the unit cell scale.

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## IMPLICATIONS

245 The Half Dome Granodiorite is considered a classic igneous rock, and yet some of the  
246 minerals in it record extensive recrystallization and reaction at temperatures well below the

247 nominal solidus (~650 °C). The hornblende phenocrysts discussed above, in spite of their large  
248 size, euhedral form, and fresh appearance, have undergone extensive closed-system reaction to a  
249 set of greenschist-facies minerals, leaving behind only remnants of the original hornblende. The  
250 phenocrysts are in effect reaction cells that retained the outward appearance of hornblende in  
251 spite of the complexity within. Other minerals are consistent with this; K-feldspar has  
252 recrystallized down to temperatures on the order of 400 °C or lower (Johnson and Glazner 2010),  
253 magnetite has exsolved essentially all of its ulvöspinel component (Glazner, unpublished data),  
254 and thin films of albite coat most plagioclase-K-feldspar contacts, giving 3-feldspar assemblages  
255 that span the peristerite gap (Glazner and Johnson 2013).

256         The rock is mineralogically in the greenschist facies and yet is undeniably an igneous  
257 rock. We propose that this is a natural consequence of incremental emplacement, because slow  
258 cooling under oscillating temperature conditions promotes exsolution and mineral  
259 reequilibration. Classic plutonic textures, such as that of the Half Dome Granodiorite, reflect the  
260 metamorphic part of their cooling history as well as the supersolidus igneous part.

261

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394

#### FIGURE CAPTIONS

395 **Figure 1.** Geologic map of the Tuolumne Intrusive Suite in Yosemite National Park, California,  
396 adapted from Bateman (1992). Star marks the sampling location at Tenaya Lake where  
397 phenocrysts were collected.

398 **Figure 2.** Half Dome hornblende phenocrysts, despite their euhedral form and unaltered  
399 appearance, show abundant mineral inclusions and extreme compositional zoning. (a) Loose  
400 hornblende phenocrysts are readily collected from the grus. (b) False-color backscatter image of  
401 crystal 3 cut through the center perpendicular to c. (c) Mineral map of a subarea of crystal 3

402 produced by supervised classification of X-ray maps (see Supplementary Figure 1). Inclusions  
403 make up 53 area % of this crystal and include every phase in the granodiorite as well as  
404 greenschist facies metamorphic minerals.

405 **Figure 3.** Proportions of mineral inclusions in five hornblende crystals cut through their centers.  
406 Mineral areas were calculated from supervised classification of X-ray maps (see Supplementary  
407 Figure 1). All but number 6 were cut perpendicular to *c*; 6 was cut perpendicular to *a*. “Other”  
408 includes magnetite, apatite, zircon, epidote, muscovite and quartz.

409 **Figure 4.** A composite BSE image of a traditionally prepared polished section of Half Dome  
410 Granodiorite hosting a prominent hornblende phenocryst. Mineral composition has been  
411 approximated from BSE brightness. Though the hornblende crystal section has not been oriented  
412 relative to its crystal axes, abundant inclusions and prominent aconcentric elemental zoning are  
413 evident. Hbl = amphiboles from hornblende to actinolite, Ab = albite, Kspr = K-feldspar, Mag =  
414 magnetite, Chl = chlorite, Bt = biotite.

415 **Figure 5.** BSE image of crystal #5 showing approximately 36 area % inclusions and locations of  
416 electron microprobe analyses. Mineral labels as in Figure 4. Red and white circles mark the  
417 locations of electron microprobe analyses.

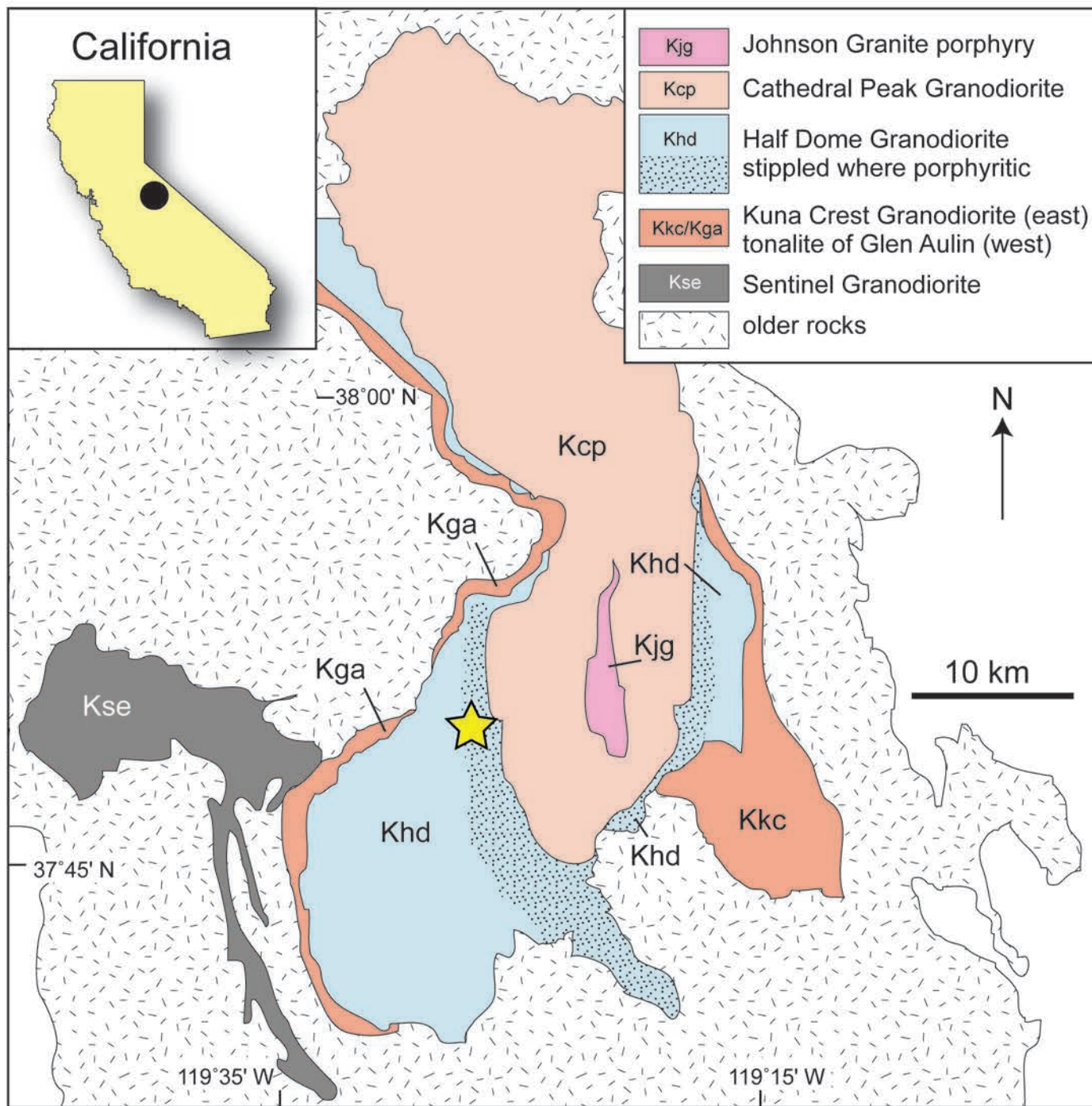
418 **Figure 6.** Large internal compositional variations observed in Half Dome Granodiorite  
419 hornblende phenocrysts. **(left)** Wt% Al plotted against wt% Mg, with approximate mineral  
420 names derived from Leake et al (1997). The phenocrysts include areas of actinolite,  
421 ferroactinolite and ferrohornblende. **(right)** Wt% Al by distance from the center of the crystals.  
422 Different marker types indicate measurements from different crystals. The red-rimmed white  
423 circles indicate measurements from the traverse marked in Figure 5. Variations in Al are large  
424 and show no concentric pattern.

425 **Figure 7.** Elemental variations within Half Dome hornblende phenocrysts show a diverse array  
426 of zoning styles which are difficult to explain within current models of phenocryst formation. **(a)**  
427 Zones have rounded but distinct edges, and an overall pattern suggestive of brecciation. **(b)**  
428 Complex angular zoning combining blurred and sharp-edged zones. **(c)** Sharp-edged angular  
429 zones with lesser internal variations. **(d)** Large angular zones with soft-edged internal zoning.

430 **Figure 8.** Whole crystal compositions plot between the high-Al amphibole (magnesiohornblende  
431 composition) and biotite spot analyses. These results suggest that the phenocrysts were initially  
432 magnesiohornblende with inclusions of biotite and possibly magnetite, and that the other  
433 including minerals formed as hornblende was converted to actinolite under greenschist  
434 conditions without any compositional modification by fluids.

435 **Figure 9.** Thermobarometric estimates for the Half Dome Granodiorite using various methods.  
436 Blue curve is the solution of Anderson and Smith (1995) for hornblende with 7 wt%  $\text{Al}_2\text{O}_3$  (Fig.  
437 8); red circles are from hornblende-plagioclase thermobarometry (Gray et al. 2008); and blue  
438 squares are estimates derived from aplite dikes and zircon saturation using the method of Putnam  
439 et al. (2015). At a given temperature the 7 wt% hornblende curve yields a higher  $P$  than the  
440 hornblende-plagioclase and aplite-zircon estimates, but all indicate crystallization at low  $P$  (<200  
441 MPa; see text for discussion). Water-saturated granite solidus from Holtz and Johannes (1994).

Figure 1



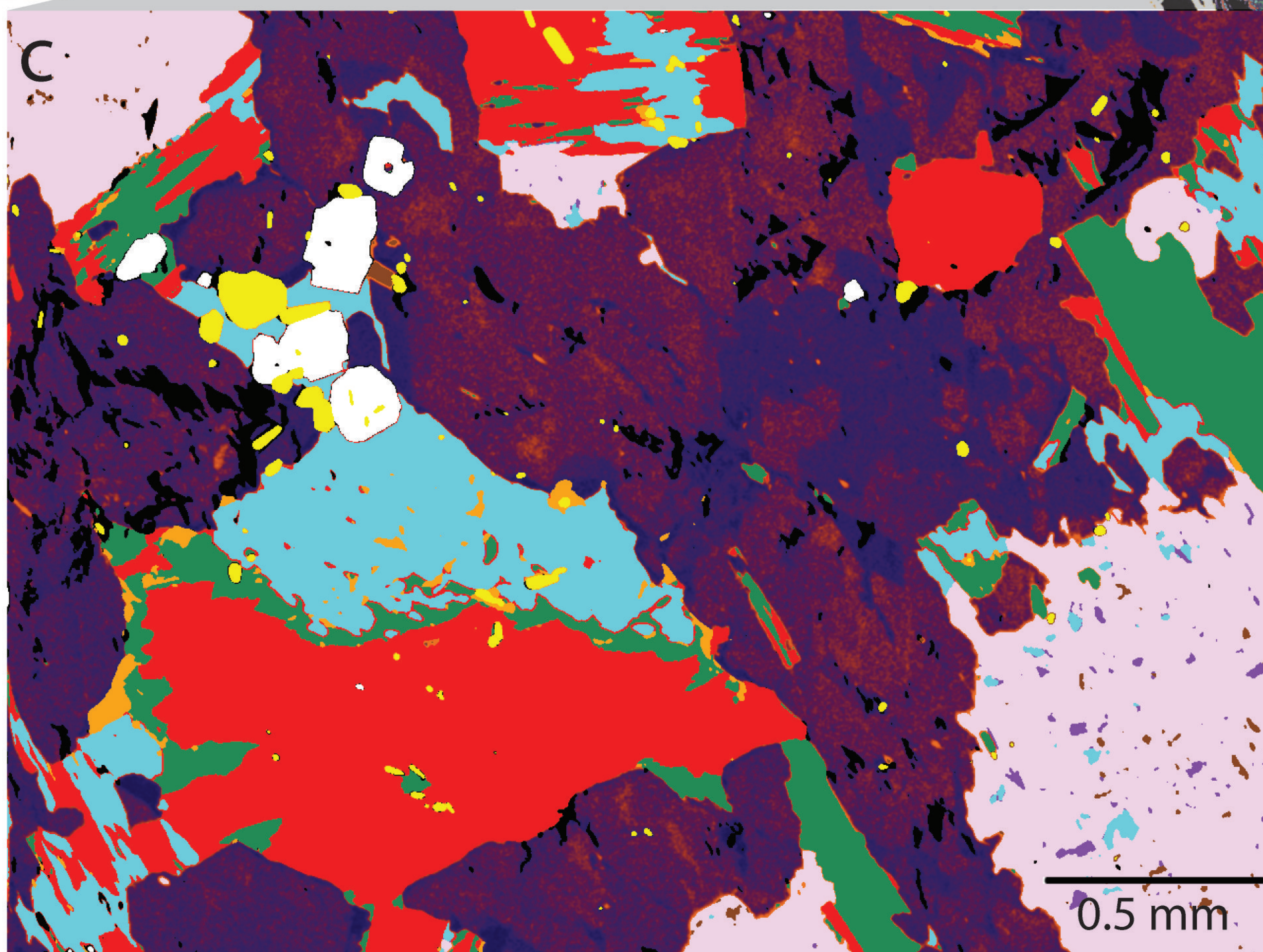
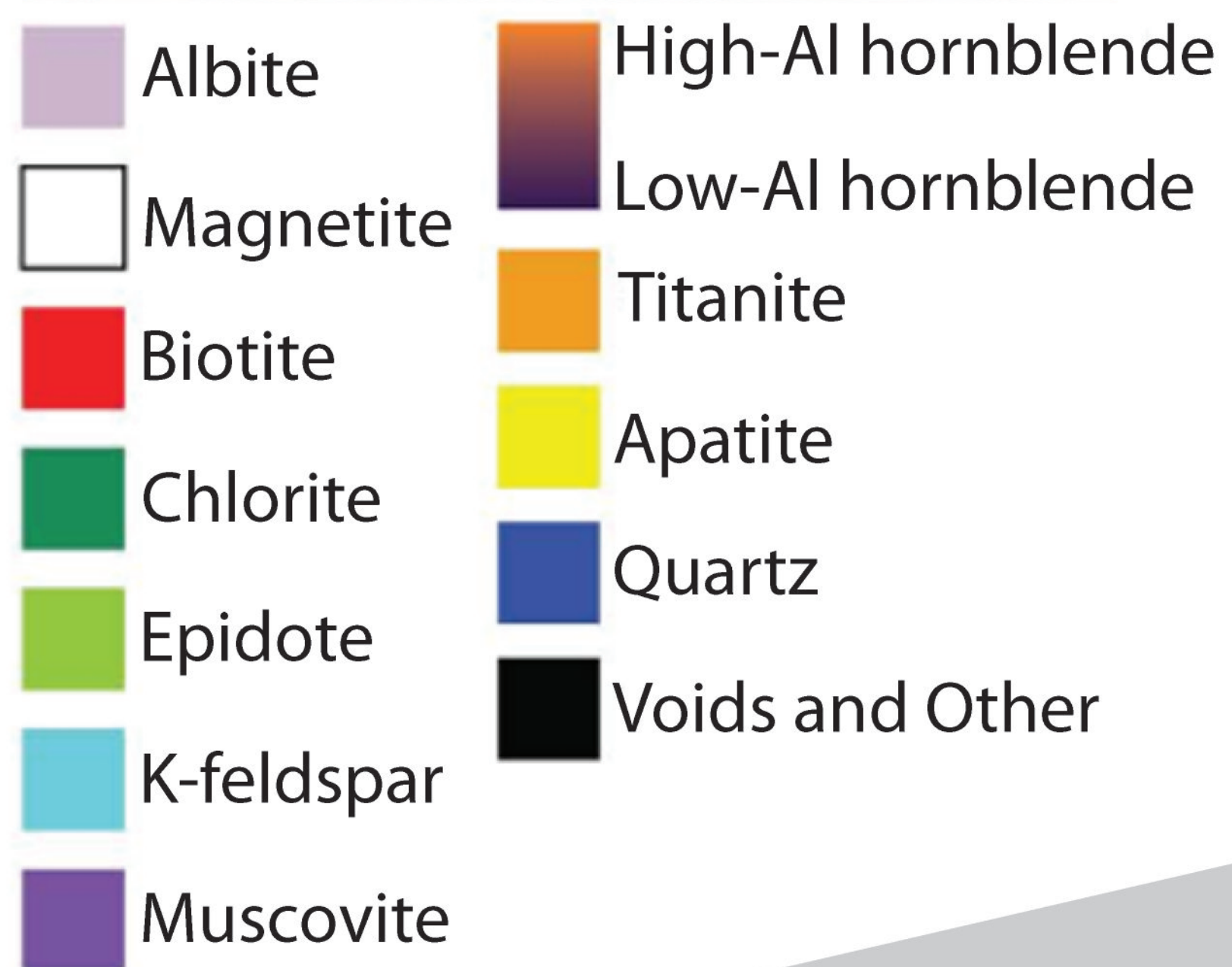
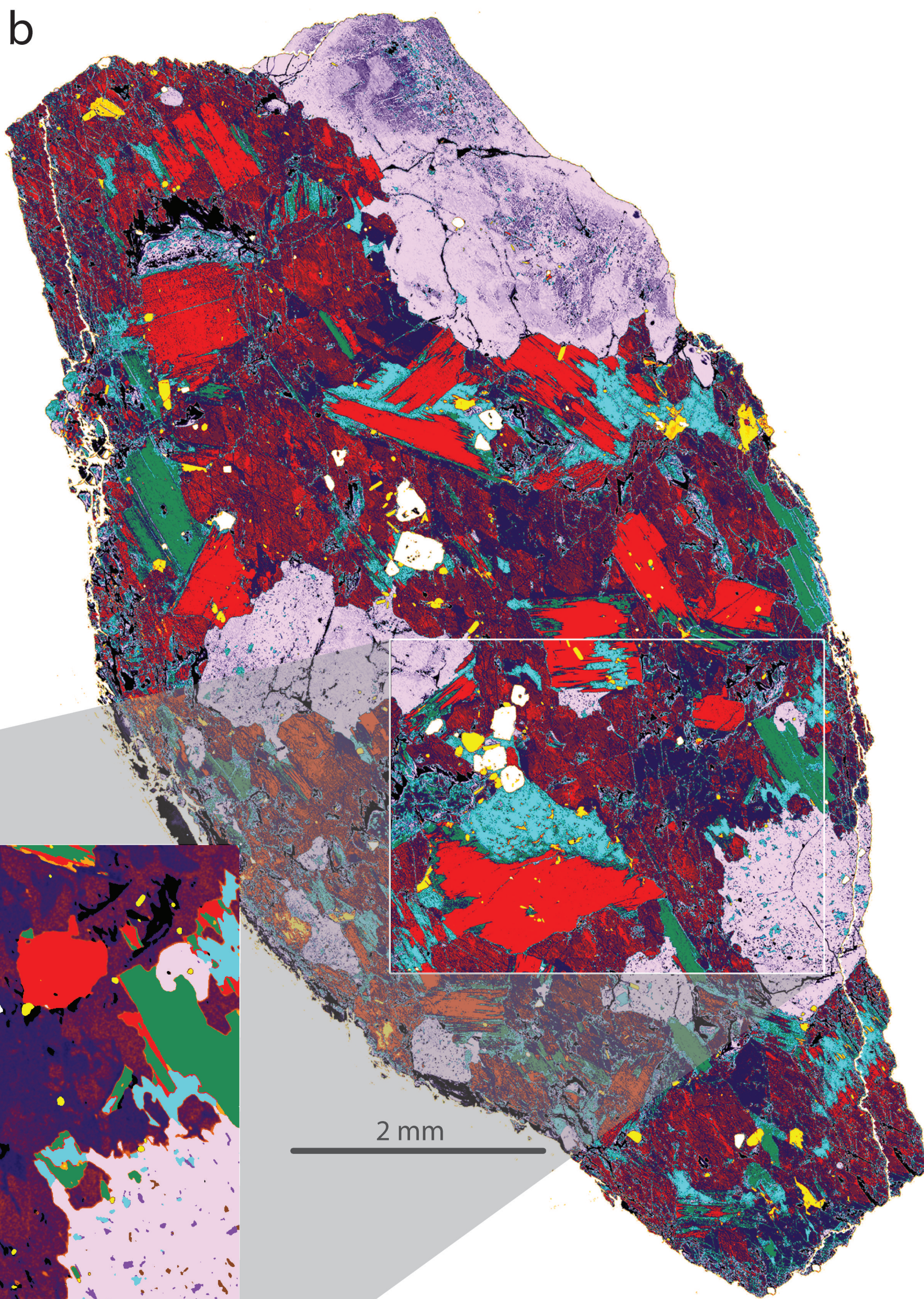


Figure 2

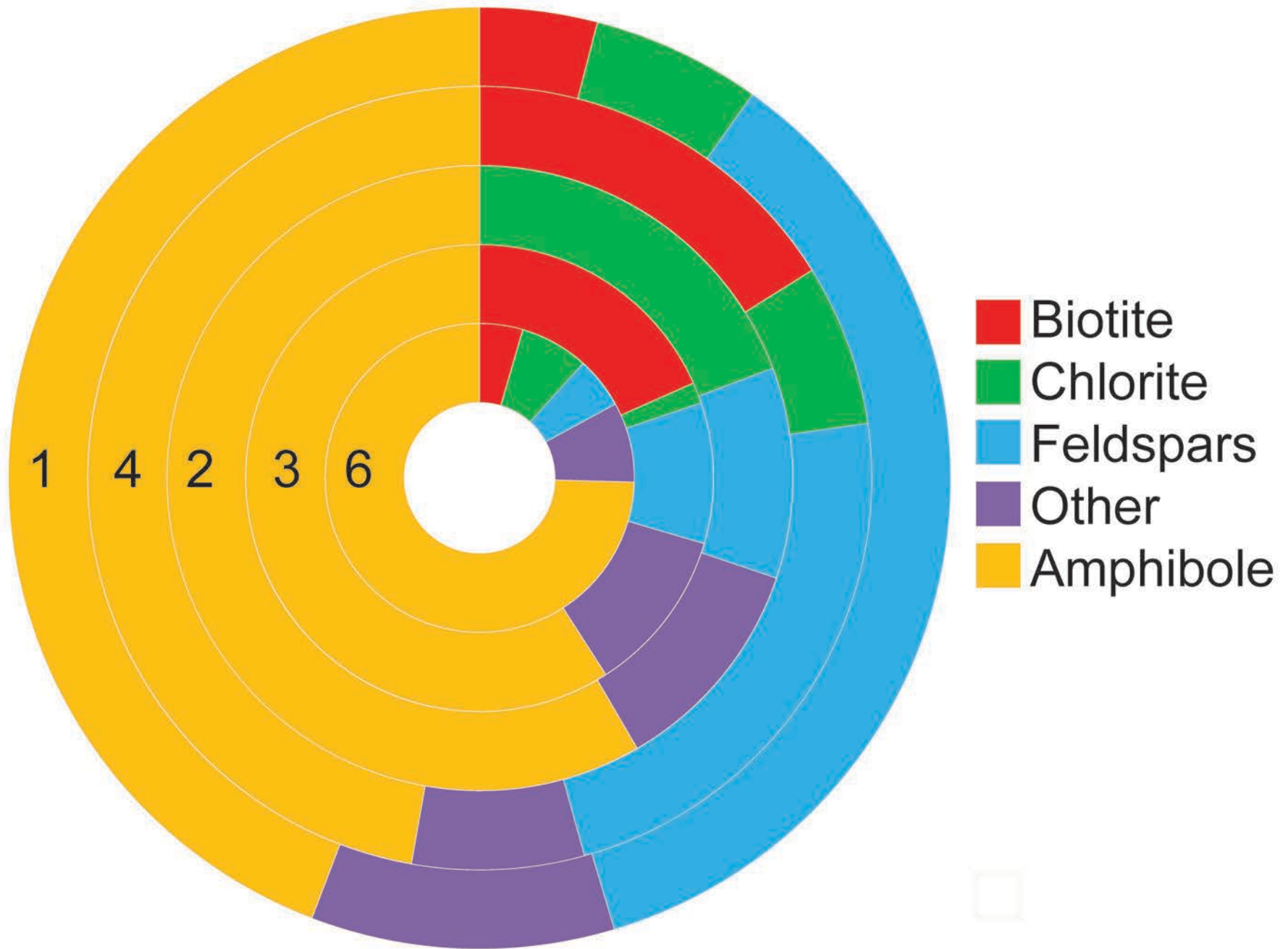


Figure 3

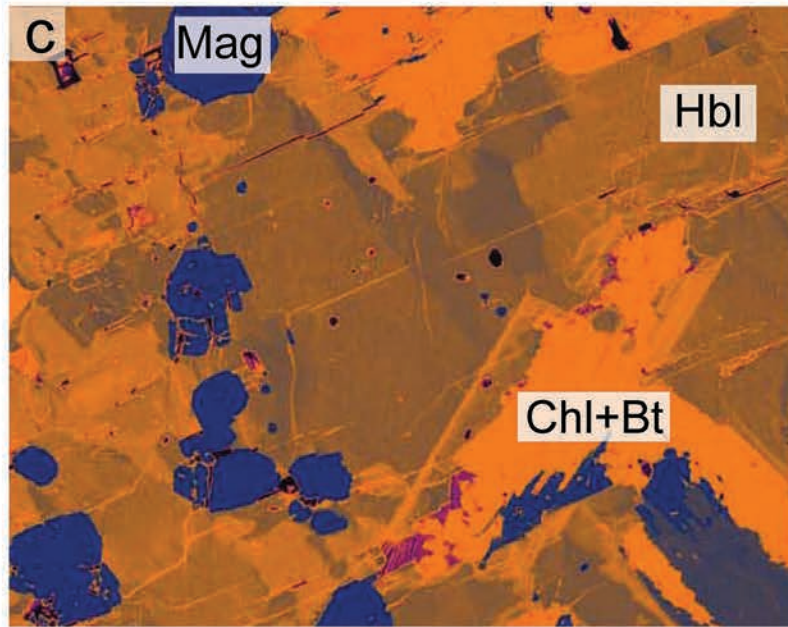
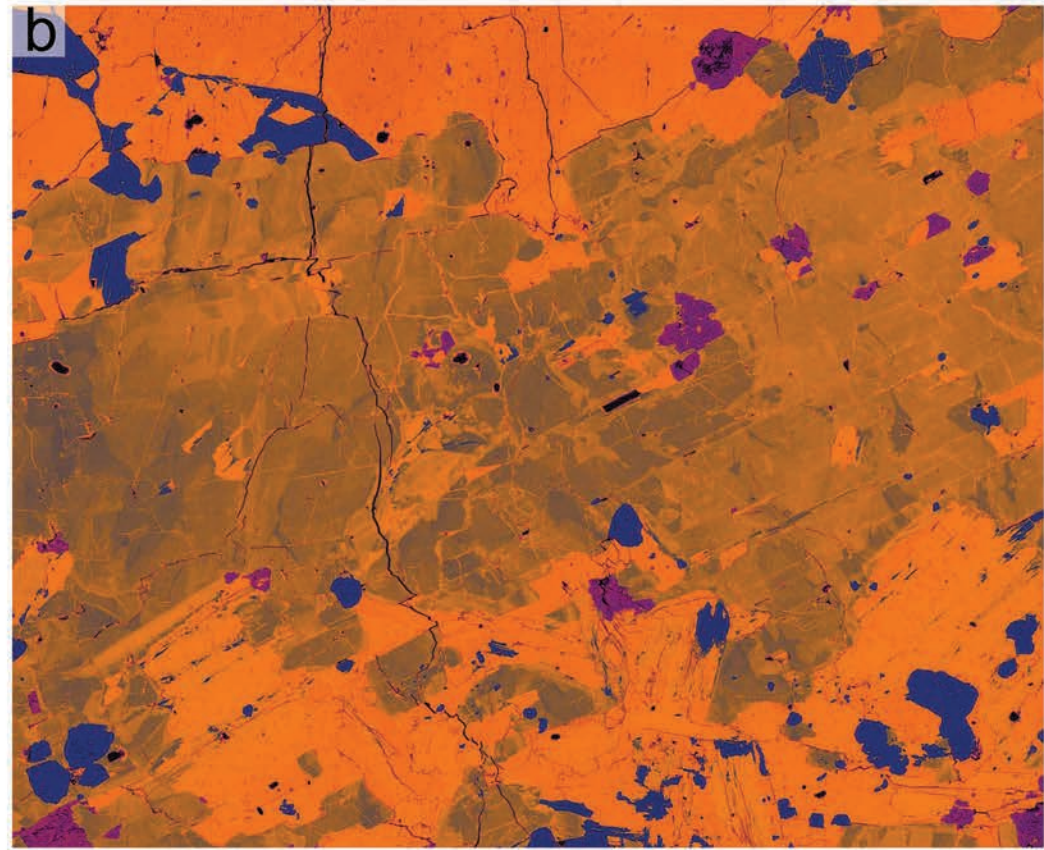
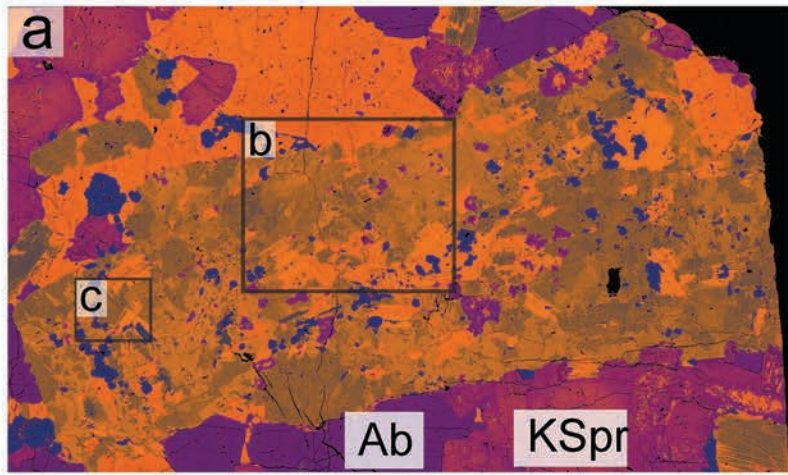


Figure 4



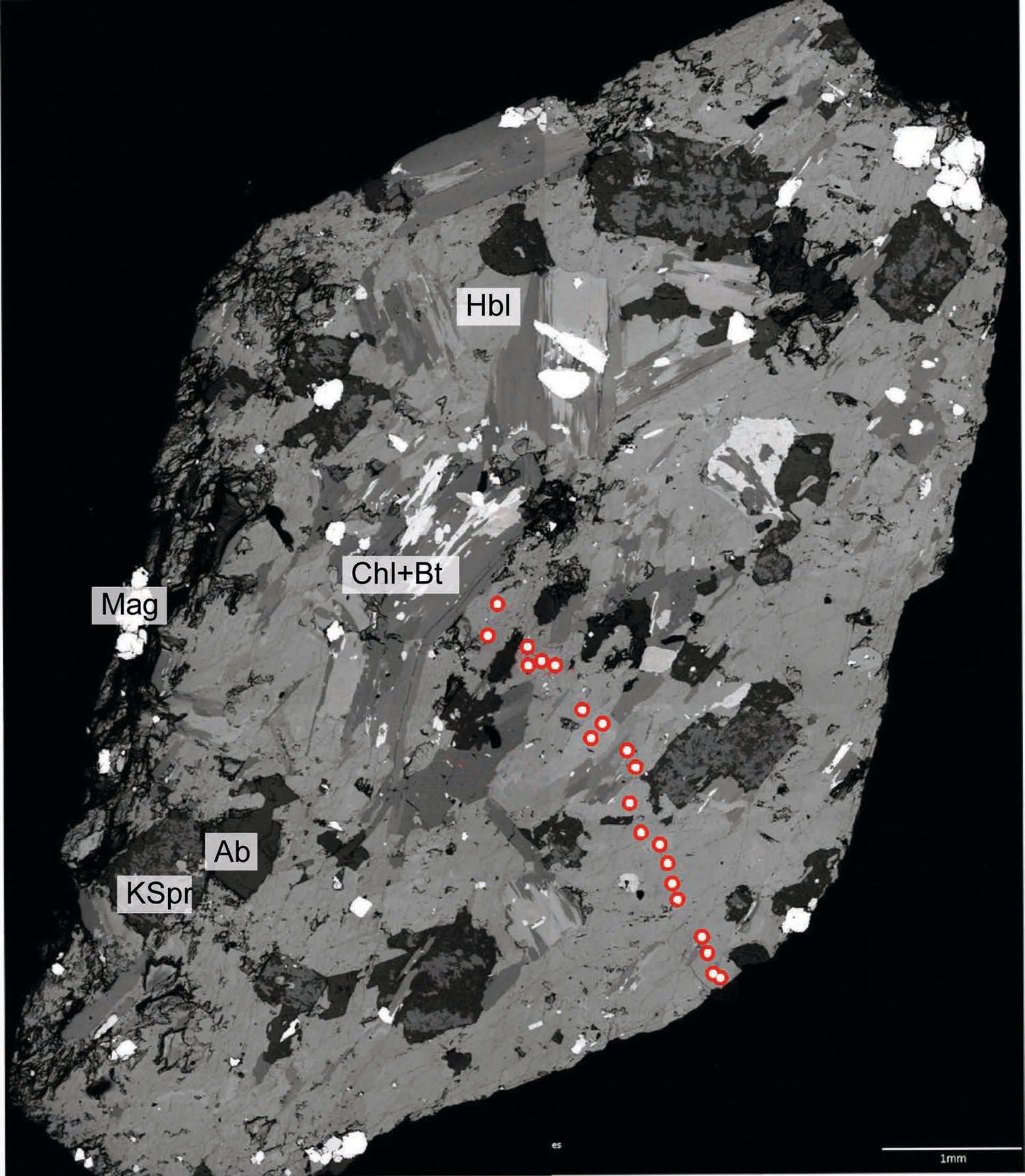


Figure 5

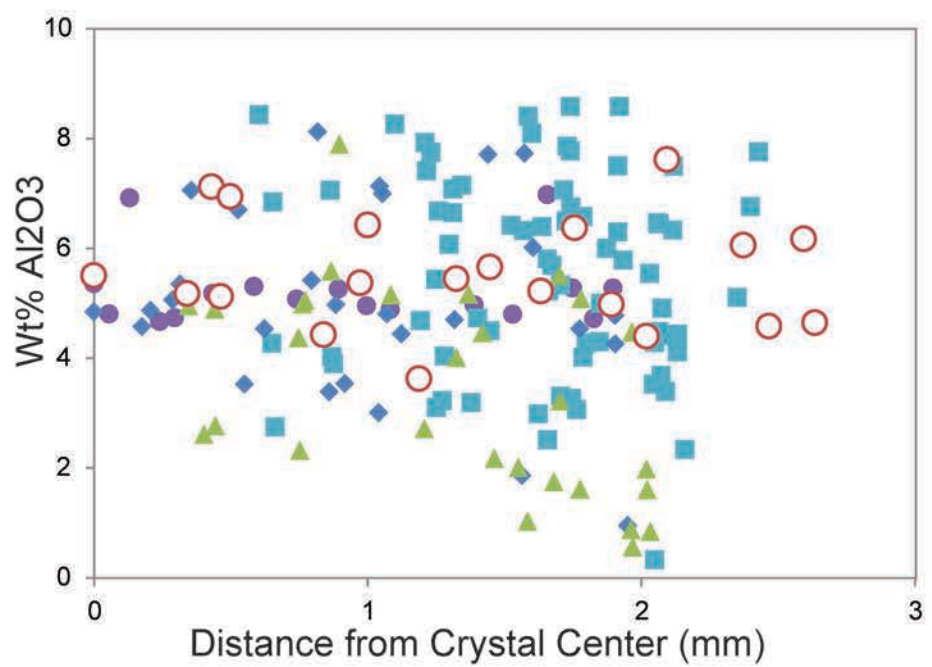
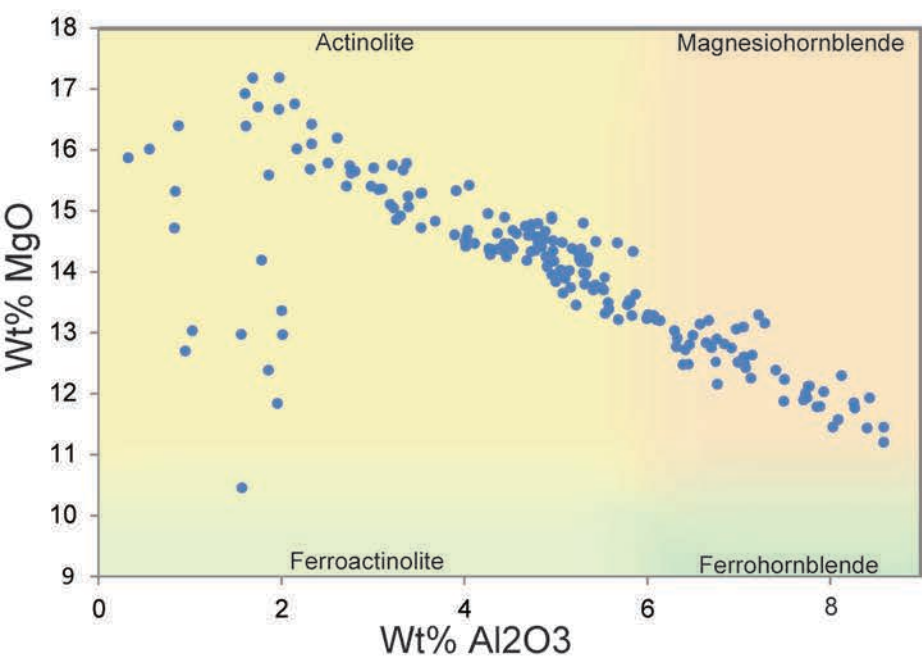


Figure 6

Table 1. Summary of chemical analyses and Structural Formulae

|                                      | Avg Biotite <sup>a</sup> | Avg Whole Xtal | Representative Amphibole Spot Analyses |           |          |          |           |
|--------------------------------------|--------------------------|----------------|--|-----------|----------|----------|-----------|
|                                      |                          |                | Hbl 3 P28                              | Hbl 5 P21 | Hbl 3 P7 | Hbl 2-P6 | Hbl 5 P10 |
| SiO <sub>2</sub> (wt%)               | 37.71                    | 46             | 52.23                                  | 52.07     | 51.33    | 45.88    | 43.84     |
| TiO <sub>2</sub> (wt%)               | 2.69                     | 1.42           | 0.00                                   | 0.13      | 0.33     | 1.01     | 1.19      |
| Al <sub>2</sub> O <sub>3</sub> (wt%) | 14.68                    | 8.21           | 1.79                                   | 2.51      | 4.06     | 7.06     | 8.59      |
| FeO (wt%)                            | 18.41                    | 16.21          | 14.42                                  | 11.26     | 12.11    | 14.83    | 15.31     |
| MgO (wt%)                            | 12.89                    | 12.63          | 14.19                                  | 15.78     | 15.41    | 12.60    | 11.2      |
| MnO(wt%)                             | 0.66                     | 0.65           | 0.56                                   | 0.87      | 0.79     | 0.65     | 0.72      |
| CaO (wt%)                            | 0.01                     | 9.86           | 11.89                                  | 11.90     | 11.80    | 11.63    | 11.24     |
| Na <sub>2</sub> O (wt%)              | 0.07                     | 1.04           | 0.57                                   | 0.32      | 0.69     | 1.06     | 1.23      |
| K <sub>2</sub> O (wt%)               | 8.79                     | 1.7            | 0.10                                   | 0.16      | 0.26     | 0.70     | 0.78      |
| Total Ox.                            | 95.91                    | 97.72          | 95.75                                  | 95.00     | 96.78    | 95.42    | 94.10     |
| Si(T)                                | 2.84                     | *              | 7.73                                   | 7.67      | 7.44     | 6.88     | 6.72      |
| Al(T)                                | 1.30                     | *              | 0.27                                   | 0.33      | 0.56     | 1.12     | 1.28      |
| Σ(T)                                 | 4.14                     | *              | 8.00                                   | 8.00      | 8.00     | 8.00     | 8.00      |
| Al(M1,2,3)                           | 0.72                     | *              | 0.04                                   | 0.10      | 0.14     | 0.13     | 0.27      |
| Ti(M1,2,3)                           | 0.15                     | *              | 0.00                                   | 0.01      | 0.04     | 0.11     | 0.14      |
| Fe <sup>3+</sup> (M1,2,3)            | 0.00                     | *              | 0.23                                   | 0.17      | 0.30     | 0.56     | 0.49      |
| Mg(M1,2,3)                           | 1.45                     | *              | 3.13                                   | 3.46      | 3.33     | 2.82     | 2.56      |
| Mn(M1,2,3)                           | 0.04                     | *              | 0.07                                   | 0.11      | 0.10     | 0.08     | 0.09      |
| Fe <sup>2+</sup> (M1,2,3)            | 1.16                     | *              | 1.53                                   | 1.14      | 1.10     | 1.29     | 1.45      |
| Ca(M1,2,3)                           | 0.00                     | *              | 0.00                                   | 0.00      | 0.00     | 0.00     | 0.00      |
| Σ(M1,2,3)                            | 5.00                     | *              | 5.00                                   | 5.00      | 5.00     | 5.00     | 5.00      |
| Fe(M4)                               | 1.61                     | *              | 0.03                                   | 0.08      | 0.07     | 0.01     | 0.02      |
| Ca(M4)                               | 0.00                     | *              | 1.89                                   | 1.88      | 1.83     | 1.87     | 1.85      |
| Na(M4)                               | 0.39                     | *              | 0.09                                   | 0.05      | 0.10     | 0.12     | 0.14      |
| Σ(M4)                                | 2.00                     | *              | 2.00                                   | 2.00      | 2.00     | 2.00     | 2.00      |
| Ca(A)                                | 0.00                     | *              | 0.00                                   | 0.00      | 0.00     | 0.00     | 0.00      |
| Na(A)                                | 0.01                     | *              | 0.07                                   | 0.05      | 0.10     | 0.19     | 0.23      |
| K(A)                                 | 0.84                     | *              | 0.02                                   | 0.03      | 0.05     | 0.13     | 0.15      |
| Σ(A)                                 | 0.85                     | *              | 0.09                                   | 0.08      | 0.15     | 0.32     | 0.38      |
| OH                                   | 2.00                     | *              | 2.00                                   | 2.00      | 2.00     | 2.00     | 2.00      |
| Σ(Cat.)                              | 12.00                    | *              | 15.09                                  | 15.08     | 15.15    | 15.32    | 15.38     |

<sup>a</sup> A single anomalously phlogopitic spot analysis was excluded from the average biotite calculation; see Supplementary Tables 1 and 2 for all spot analyses.

Average biotite formula calculated assuming 2 OH and all ferrous iron.

Amphibole formulae calculated following the methods of Anderson and Smith (1995).

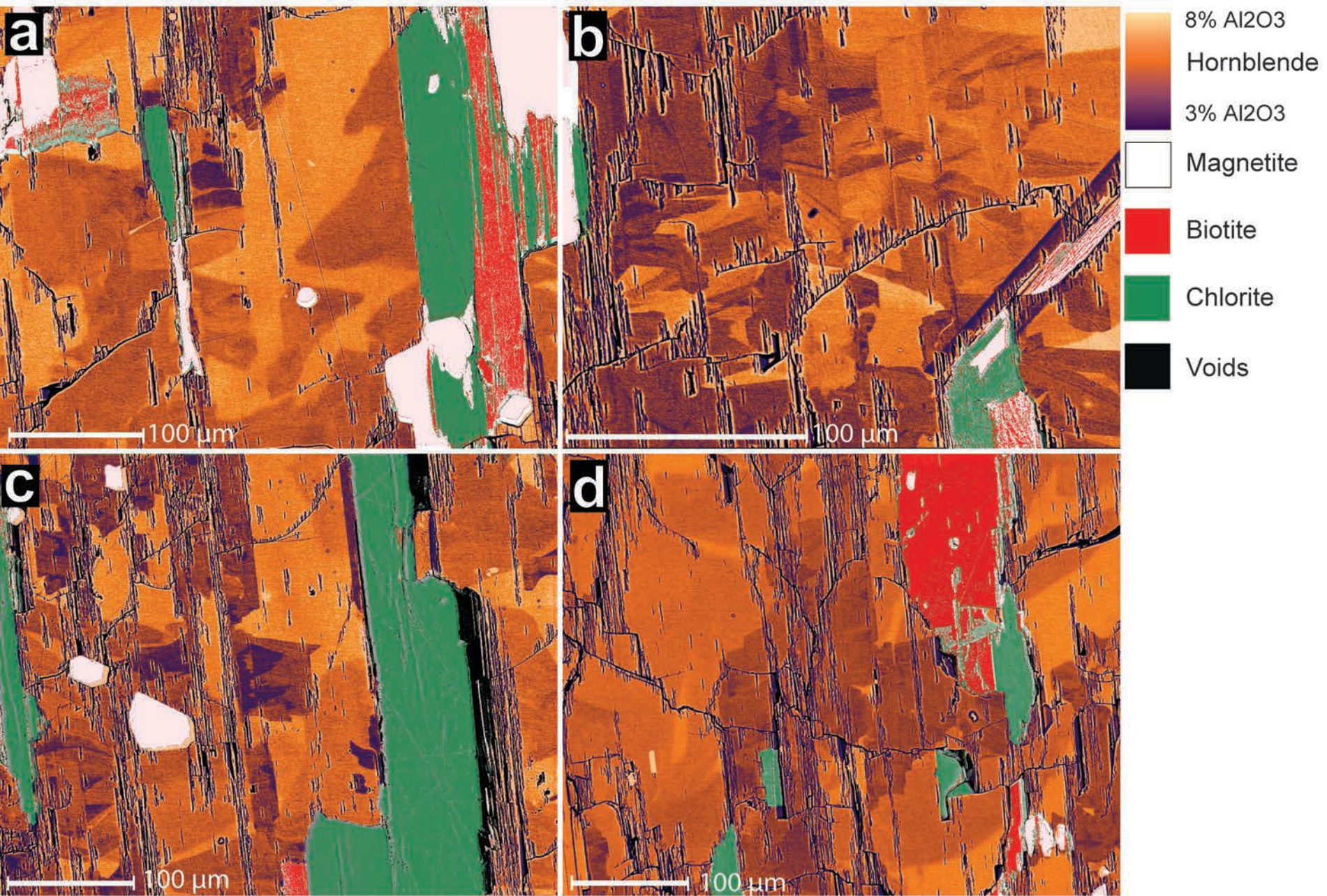


Figure 7

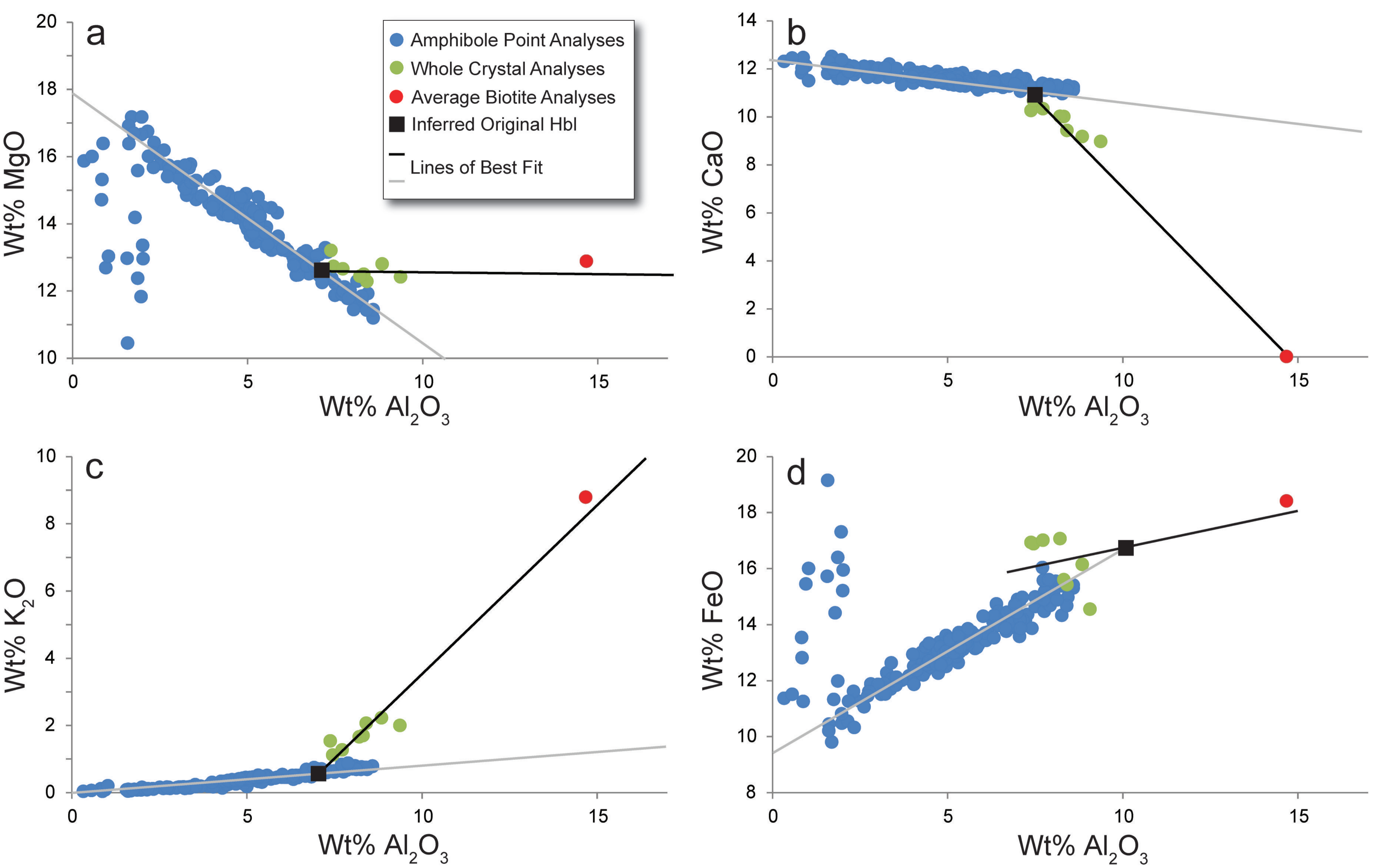


Figure 8

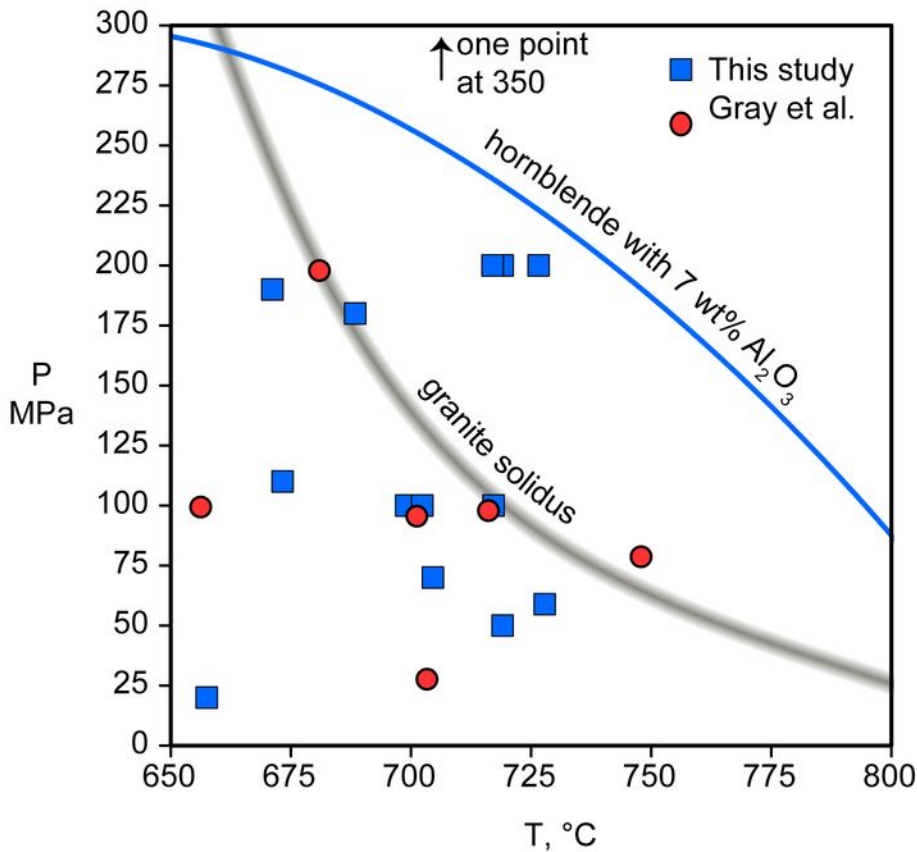


Figure 9