

1 Highlights and Breakthroughs:

2 Alunite on Mars

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10 **Abstract:** Identification of minerals on the surface of Mars is critical to understanding the geological
11 history of our neighbor planet. In this issue of *American Mineralogist*, Ehlmann et al. report their
12 discovery of alunite ($\text{KAl}_3(\text{SO}_4)_2(\text{OH})_6$) in Cross Crater on Mars. Because terrestrial alunite forms
13 from Al-rich acid sulfate waters, these results strongly suggest the past presence of Al-rich acid saline
14 martian waters. **Keywords:** alunite, Mars, acid, brines, CRISM.

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16 The idea of past water on Mars has intrigued humans for several centuries, since martian canals were
17 first spied through early telescopes. But only in the past few decades has more diagnostic evidence of
18 martian waters been documented (e.g., Carr, 1996). Currently, planetary scientists are gathering
19 sedimentological and mineralogical data to trace types of waters (surface waters versus groundwaters)
20 as well as water composition. Ehlmann et al. (2016), featured in this issue, document their discovery
21 of alunite ($\text{KAl}_3(\text{SO}_4)_2(\text{OH})_6$), contributing significant evidence that Mars once hosted Al-rich acid
22 sulfate waters.

24 A case has been building for past, and perhaps even modern, acid solutions on Mars. Clark (1979,
25 1999) used martian surface conditions, as well as sediment and atmospheric composition, to
26 hypothesize that a sulfuric acid solution would be the most likely stable liquid on the martian surface.
27 Some of the first remotely sensed spectral features from Mars were attributed to acid sulfate minerals
28 jarosite ($\text{KFe}^{3+}_3(\text{OH})_6(\text{SO}_4)_2$) and schwertmannite ($\text{Fe}_8\text{O}_8(\text{OH})_6\text{SO}_4$; Burns, 1987, 1994). Physical
29 sedimentology experiments showed that sulfuric acid solutions make channels and fans
30 morphologically similar to those on Mars (Benison et al., 2008). Abundant recent studies have
31 demonstrated the likelihood of acid saline waters on Mars through: (1) detailed observations of
32 martian surface mineralogy and sedimentology from rovers, landers, and satellites (e.g., Farrand et al.,
33 2009; Sqyures et al., 2004); (2) laboratory experiments involving mineral precipitation, alteration, and
34 dissolution by acid waters (e.g., Bishop and Murad, 2005; Miller et al., 2016); and (3) field and
35 laboratory analyses of potential terrestrial analogs (e.g., Amils et al., 2007; Benison and Bowen, 2006;
36 Bishop and Murad, 2005).

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38 Ehlmann et al. (2016) analyzed images and visible/shortwave infrared spectra of Cross Crater in the
39 Terra Sirenum region of Mars. Spectral data were obtained by CRISM (Compact Reconnaissance
40 Imaging Spectrometer for Mars) on the Mars Reconnaissance Orbiter. Mineralogical results were
41 placed in a topographic and stratigraphic context. Their study discovered a large (10 km x 5 km)
42 sedimentary deposit of alunite, as well as mixed layered alunite and kaolinite ($\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$)
43 deposits (Ehlmann et al., 2016). Alunite-rich and mixed alunite-kaolinite units have been identified
44 over a wide elevation range along the inner crater walls and floor, suggesting that alunite-bearing
45 strata are hundreds of meters thick. The alunite and kaolinite are spatially restricted within Cross

46 Crater; strata outside of the crater contain a different mineral assemblage with no Al-sulfates or Al-
47 phyllosilicates.

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49 Alunite is commonly known as a hydrothermal mineral (e.g., Scott, 1990) or as a weathering product
50 of sulfides (e.g., Amils et al., 2007). However, it also has been documented as a syndepositional
51 mineral precipitated from shallow, Al-rich acid saline groundwaters associated with ephemeral acid
52 saline lakes in Chile, Western Australia, and southeastern Australia (Benison and Gonzalez, 2007;
53 Bowen et al., 2012; Benison et al., 2007; Long et al., 1992). In Chile and Australia, alunite typically
54 forms alongside kaolinite. The spatial characteristics of the alunite documented by Ehlmann et al.
55 (2016), as well as association with kaolinite, strongly suggests that alunite on Mars was formed by
56 long-lived lake waters and/or associated shallow groundwaters.

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58 Alunite is rare on Earth because acid waters rich in Al are needed for its formation. The pH of
59 alunite-producing brines ranges from ~ 2 to ~ 4 (e.g., Benison et al., 2007; Miller et al., 2016).

60 Although acid, sulfurous waters have been documented elsewhere on Mars previously by
61 identification of jarosite (e.g., Klingelhofer et al., 2004), those were Fe-rich acid brines. This study by
62 Ehlmann et al. (2016) is the first to confirm that the past presence of Al-rich, sulfurous acid brines on
63 Mars.

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65 Ehlmann et al.'s discovery of alunite on Mars is a significant contribution to the "follow the water"
66 objective of the planetary geology community. The diverse mineral assemblages in sedimentary strata
67 in different geographic regions on Mars tells us that Mars has had a rich aqueous history that involved

68 surface waters and groundwaters of various water compositions. Mars' hydrologic and mineral
69 history is different from, but as complex as, that of Earth.

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