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3	Elastic wave velocity anomalies of anorthite in subducting plate: In situ					
4	experiments					
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6	Kyoko N. Matsukage ^{1,*} , Yu Nishihara ² , Fumiya Noritake ³ , Katsuyuki Kawamura ³ , Noriyoshi					
7	Tsujino ⁴ , Moe Sakurai ⁵ , Yuji Higo ⁶ , Junichi Nakajima ⁷ , Akira Hasegawa ⁷ and Eiichi					
8	Takahashi ⁵					
9						
10	¹ Earth and Planetary Sciences, Kobe University, Rokkoudai, Nada-ku, Kobe 657-8501, Japan					
11	² Geodynamics Research Center, Ehime University, Bunkyocho, Matsuyama, Ehime 790-8577					
12	Japan					
13	³ Environmental Science and Technology, Okayama University, Tsushimanaka, Kita,					
14	Okayama 700-8530, Japan					
15	⁴ Institute for Study of the Earth's Interior, Okayama University, Okayama, Japan					
16	⁵ Earth and Planetary Sciences, Tokyo Institute of Technology, Ookayama, Meguro-ku, Tokyo					
17	152-8551, Japan					
18	⁶ Japan Synchrotron Radiation Research Institute, Hyogo 679-5198, Japan					
19	⁷ Research Center for Prediction of Earthquakes and Volcanic Eruptions, Graduate School of					
20	Science, Tohoku University, Sendai 980-8578, Japan					
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22	*Corresponding author: K.N. Matsukage,					
23	E-mail: kmatsu@people.kobe-u.ac.jp					
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ABSTRACT

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To understand the origin of observed low-velocities in the crustal portion of subducting plates, we performed in situ measurements of elastic wave velocities of anorthite at temperatures up to 1373 K at pressure of ~1 GPa and up to 773 K at 2.0-7.0 GPa. A fine-grained polycrystalline anorthite which was synthesized using a gas pressure apparatus was used for the measurements. The high pressure experiments were performed using the multi-anvil apparatus installed on beam-line BL04B1 at SPring-8. The elastic wave velocity was measured by the ultrasonic pulse method with synchrotron X-ray radiographic imaging and X-ray diffraction techniques. At ~1.0 GPa, elastic wave velocities exhibited a sharp temperature-induced kink at ~500 K. Below 500 K, the elastic wave velocities decrease with increasing temperature. In contrast, above 500 K, the elastic wave velocities show an increasing trend in the range of 500-900 K, and then revert back to a decreasing trend at above 900 K. We also found a pressure-induced velocity anomaly of anorthite. At 300-373 K, $V_{\rm P}$ is constant up to 4 GPa, but decrease above 4 GPa with increasing pressure, while $V_{\rm S}$ decreases monotonously with increasing pressure. These elastic anomalies is considered to be attributable to the tilting behaviour of the corner-sharing TO₄ (T = Al, Si) tetrahedra in three dimensional frameworks of anorthite. Our results suggest the presence of plagioclase feldspar has the potential to causes low velocity anomaly in the subducting oceanic crust when it survives as a metastable phase in the slab at higher pressure and lower temperature conditions.

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Keywords: anorthite, plagioclase, elastic wave velocities, subduction, oceanic crust

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INTRODUCTION

Seismological observations have detected heterogeneous low-velocity structure in the crustal portion of subducting plates (e.g., Kawakatsu and Watada 2007; Rondenay et al. 2010). Plagioclase feldspars are one of the most abundant minerals in the crustal portion. Crusts with MORB-like composition contain large amount of plagioclase, normally more than 50 wt%. Therefore, the knowledge of the physical properties of plagioclase is important in order to resolve the detailed structure of the crust and crustal portions of subducted slabs based on such seismological observations. The purpose of our study is to determine the physical behaviour of anorthite (CaAl₂Si₂O₈, an end member of plagioclase), and to discuss the origin of the low-velocity layer in the crustal portion of subducting plates.

Elastic wave velocity of plagioclase minerals and rock aggregates has been measured using ultrasonic pulse technique (e.g., Ryzhova 1964; Wang et al. 1973; Liebermann and Ringwood 1976; Seront et al. 1993; Kono et al. 2008). Previous studies, however, encounter some experimental difficulties. In this technique, the elastic wave velocities are obtained from travel time and sample length. Thus, accurate measurement of the sample length is crucial for determining accurate elastic wave velocities (Li et al. 1998). In all previous studies for plagioclase, except for one study performed at room pressure (Ryzhova 1964), the sample lengths were not measured directly and were estimated by calculation using equation of state of minerals based on the assumption that only isotropic elastic deformation causes sample strain under high pressure and high temperature. In order to overcome this problem, synchrotron X-ray radiographic imaging technique has been applied in the elastic wave velocity measurements (Kung et al. 2002). When this method is used, direct measurement of the sample length in a high-pressure apparatus can be carried out. Recently, the experimental system for in situ ultrasonic measurement combined with X-ray radiographic imaging

technique has been installed at Spring-8, Japan (Higo et al. 2009). Here, we report the result of elastic wave velocity measurements on anorthite employing the ultrasonic pulse method under high pressure and high temperature with synchrotron X-ray radiographic imaging and X-ray diffraction techniques.

EXPERIMENTAL METHODS

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Synthesis of polycrystalline anorthite

A polycrystalline anorthite was synthesised using the gas pressure apparatus installed at Tokyo Institute of Technology. The starting material was prepared from reagent grade SiO₂, Al₂O₃ and CaCO₃, which were mixed in an appropriate ratio and ground in an agate mortar and pestle with ethanol. The mixed sample was decarbonated in a 1-atm furnace at 1273 K for 24 hours. Then, it was melted at 1873 K and quenched to form glass. The glass sample was ground in ethanol and loaded into a sealed Pt tube container (3.0 mm inner diameter and 0.2 mm thickness). The sample was preheated at 1173 K for 2 hours, and then it was kept at 1373 K for 20 hours at pressure of 0.3 GPa. Figure 1a shows an orientation contrast image of the synthesised polycrystalline anorthite. The anorthite crystals are predominantly anhedral with mosaic texture, and some of crystals exhibit short columnar shape. The average grain size is less than 10 um across. Orientations of grains in the polycrystalline anorthite are estimated to be random based on observation of a thin section under an optical microscope, which showed random extinction of transmitted polarized light. The bulk density of the synthesised anorthite measured by the Archimedes method is 2.72 ± 0.02 g/cm³, which is 98.8% of the X-ray density (= porosity of 1.2%). The sample was formed into cylinders with 2.0 mm in diameter using ultrasonic cutter and ground to ~1.0 mm in length, and both ends of cylinder were polished with 1 µm diamond paste.

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In situ measurements of elastic wave velocities and X-ray diffraction

In situ measurements of the elastic wave velocities (V_P and V_S) of anorthite were performed at pressures between 0.6 and 7.0 GPa at temperatures of up to 1373 K. The experiments were performed using the Kawai-type multi-anvil apparatus (SPEED-1500) on beam-line BL04B1 at the synchrotron facility of Spring-8 (Utsumi et al., 1998). The experimental design for in situ elastic wave velocity measurement using the ultrasonic pulse-echo-overlap method at BL04B1 was presented by Higo et al. (2009). Pressure was generated by eight 26-mm tungsten carbide anvils (Tungaloy F grade) with truncated edge length of 11 mm. A Co-doped semi-sintered MgO octahedron with an 18-mm edge length was used as the pressure medium (Figure 2). A graphite sleeve was used as the heater and was inserted in a ZrO2 thermal insulation sleeve with an MgO window for X-ray path to reduce potential temperature gradients. The temperature gradient of sample was estimated as ~25 ^oC/mm at the maximum (Matsukage and Kawasaki 2014). The sample was enclosed in an hBN sleeve container, and then placed in the central part (hot part) of the furnace. Platinum foils (2.5 µm in thickness) were inserted on both side of the sample for the determination of sample length using the X-ray radiographic imaging technique (Figure 3a). The precision of the sample length measurement is estimated at 0.1-0.3 % of the nominal sample length (Higo et al., 2009). An Al₂O₃ rod (5.3 mm in length and 2.0 mm in diameter) was used as the buffer rod for transmitting ultrasonic waves to the sample. The ultrasonic signals were generated and received by 10°Y-cut LiNbO₃ transducer having thickness of 50 µm and diameter of 3.2 mm. We used ultrasonic waves at frequencies of 30-60 MHz for 3 cycles (Figures 3b and 3c).

The diffracted X-ray from the sample was measured simultaneously with the measurement of elastic wave velocities. A solid-state detector connected to a multi-channel

analyzer combined with incident white X-ray beam was used for data collection. The multichannel analyzer was calibrated with the characteristic X-rays of Cu, Mo, Ag, Ta, Pt, Au and Pb. The X-ray diffractions were collected at a fixed 2θ angle (= 2.961°), which was determined using the unit-cell volume of MgO at room pressure and temperature. The incident X-ray beam was collimated to 0.2 and 0.1 mm in the vertical and horizontal dimensions, respectively. Slits of 2.0 mm in the vertical dimension and 0.1 mm in the horizontal dimension were used between the sample and the detector. MgO was used as a pressure marker, and it was mixed with hexagonal boron nitride (hBN) (MgO:hBN = 1:1 by weight) to prevent grain growth at high temperatures. Exposure times for collecting diffraction patterns of the sample and pressure marker were 300 and 180 s, respectively. Temperature was measured by a W₉₇Re₃-W₇₅Re₂₅ thermocouple. No pressure correction was made on the emf of the thermocouple. The pressure was calculated from the two diffraction peaks of MgO (111 and 200) using the equation of state of MgO presented by Jamieson et al. (1982) in program "XRayAna" available at the beam-line BL04B1.

RESULTS

The elastic wave velocities and X-ray diffraction measurements were carried out under pressures up to 7 GPa in a series of heating and cooling cycles (Table 1, Figure 4). First, we compressed the sample to \sim 2.0 GPa (load = 80 ton), then heated it to 1373 K where it was kept for 60 minutes at 1373 K under constant oil pressure (cycle (1)). The purpose of the first heating and cooling cycle is to relax the stress on the sample and its surrounding materials, and to obtain better mechanical contact between the buffer rod and sample. This compression and preheating process is important for obtaining reliable signals of elastic waves (e.g., Li et al. 2004). After preheating, V_P , V_S and X-ray diffraction measurements were performed in the

145 cooling stage of seven heating and cooling cycles (2)-(8). After each cooling stage, we 146 increased the oil pressure, and then started the next heating and cooling cycle. The data of 147 cycle (2) was collected up to 1373 K at P-T conditions where anorthite is thermodynamically 148 stable (Hays 1966). The data of cycles (3)-(8) were collected up to 773 K at metastable 149 conditions. 150 In this study, we found a kink in elastic wave velocities with discontinuous change in 151 temperature dependence at a temperature of about 500 K in cycle (2) (Figure 5). Below 500 K, 152 the elastic wave velocities decrease with increasing temperature. In contrast, above 500 K, the 153 elastic wave velocities show an increasing trend in the range of 500-900 K, and then revert 154 back to a decreasing trend at above 900 K. Kinks similar to that of cycle (2) is observed for 155 the cycles (3)-(5). At these cycles, temperatures at the kinks increase with increasing pressure, 156 from ~500 K to ~720 K. In the cycles (6)-(8), kinks are not apparent within the studied 157 temperature ranges, and only a monotonously decreasing trend is observed with increasing 158 temperature. Elastic wave velocity measurement of anorthite was performed twice, and the 159 temperature-induced elastic anomalies which are described here were observed in both 160 experiments (see appendix). We also found a pressure-induced velocity anomaly of anorthite 161 (Figure 6). At temperature of 300-373 K, V_P decreases with increasing pressure at above 4 GPa, although it remains constant below 4 GPa. The V_S shows a decrease with increasing 162 163 pressure. At 473 K, both V_P and V_S exhibit increasing trends below 4 GPa while showing a 164 decreasing trend with increasing pressure above 4 GPa. At above 573 K, V_P and V_S decrease 165 monotonously with increasing pressure up to 7 GPa. 166 All X-ray diffraction peaks correspond to anorthite with triclinic structure (Figure 7)

(e.g., Foit and Peacor, 1973). For anorthite, the P(-1) structure is stable at ambient condition,

and it changes to I(-1) structures with increasing temperature (e.g., Redfern and Salje 1987;

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Redfern et al. 1988) and with increasing pressure (e.g., Angel 1988, 1992; Hackwell and Angel 1995) (Figure 4). Because there is no significant difference in the position of atoms between these structures, X-ray diffraction patterns resemble one another. Previous studies (e.g., Hackwell and Angel 1995; Tribaudino and Angel 2012) distinguished I(-1) phase from P(-1) by using the disappearance of 11(-1) reflection (c-type reflection: h + k = even, l = odd) that is absent in I(-1) phase due to the symmetry of body centered structure. In our experiments, 11(-1) reflection was observed at 300-473 K in cycle (2) and at ambient condition before and after the experiments. The X-ray diffraction patterns of sample exhibit no additional peaks at any of the pressure and temperature conditions, although a small amount of reaction products (< 1 %) were observed at the grain boundaries of anorthite crystals from the recovered sample (Figure 1b). Unit cell parameters and volume of the sample are determined by Le Bail method in program GSAS (Von Dreele and Larson 2001) using all diffraction lines shown in Figure 7. The unit cell parameters a, b, c, α and β decrease with increasing pressure and γ increase with increasing pressure (Figure 8). The change in temperature dependence of unit cell angles (β and γ) at about 500-550 K is observed in cycles (2) and (3). The unit cell volume at ambient pressure ($V_0 = 1341 \text{ Å}^3$) agrees well with that of Tribaudino and Angel (2012) ($V_0 = 1340.1 \text{ Å}^3$). The unit cell volume gradually decreases with increasing pressure, though a weak kink is observed at about 4 GPa (Figure 9). The adiabatic bulk modulus (K_S) and shear modulus (G) are represented by the equations: $K_S = \rho(V_P^2 - V_S^2 \times 4/3)$ and $G = \rho V_S^2$, where ρ is density. To calculate the K_S and Gfrom the velocities and density data, we obtained density from the equation, $\rho = (m Z/N_A) \times$ (1/V), where m, Z, N_A and V are formula weight, formula number (Z = 8) for anorthite, the Avogadro constant and unit cell volume, respectively. The calculated ρ , K_S and G are listed in Table 1. In cycles (2), (3) and (4), kinks of K_S and G are observed at ~500-750 K. Adiabatic

bulk modulus (= K_{S0}) and shear modulus (= G_0) at room temperature and room pressure were estimated by polynomial fitting using data at 300 K. We obtained K_{S0} = 83 GPa and G_0 = 38.6 GPa. Based on Angel's (2004) data on plagioclase feldspars, isothermal bulk modulus (K_{T0}) at ambient condition was estimated by Tribaudino et al. (2011) to be K_{T0} = 82.5 GPa. K_{S0} is calculated as 82.7 GPa from the formula K_{S0} = K_{T0} (1 + $\alpha\gamma$ T) where α and γ are thermal expansion coefficient and Grüneisen parameter, and are 1.379× 10⁻⁵ K⁻¹ and 0.46 (Tribaudino et al. 2011), respectively. The K_{S0} determined by our experiments is sufficiently consistent with those of previous study.

DISCUSSION AND CONCLUSIONS

Relationship with the phase transitions and the crystal structure

Anorthite is known for undergoing a reversible phase transition from P(-1) to high-temperature I(-1) at temperature of 510 K at pressures of less than ~2 GPa (e.g., Foit and Peacor 1973; Redfern et al. 1988). Temperature and pressure conditions of the discontinuous elasticity change at 500 K of cycle (2) agree well with the phase boundary of P(-1) and high-temperature I(-1) (Figure 4). Based on this observation, it is thought that anomalous behaviour at > 500 K is caused by successive structural changes in the high-temperature I(-1) phase, followed by the phase transition. In our experiments, the temperature-induced anomaly is observed at higher pressure, at the P-T conditions near the extrapolated phase boundary by Hackwell and Angel (1995) (Figure 4). The structure of anorthite comprises a flexible three dimensional framework of rigid, corner sharing TO_4 tetrahedra (T = Si and Si Al) with Si cocupying interstitial sites of the framework (e.g., Angel et al., 2013). Molecular dynamics (MD) simulation (Noritake et al. 2013) suggests that the temperature-induced anomalies of the Si and Si could be caused by a combination of two different thermal expansion

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mechanisms due to atomic motions and displacements: (1) expansion of inter-atomic distances of near neighbor atoms and (2) increase in T-O-T angles of the framework. Below the phase transition temperature, thermal expansion is mainly caused by the expansion of the inter-atomic distances. In high-temperature I(-1) stability field, two mechanisms contribute to the thermal expansion. The T-O-T angles increase with increasing temperature, which subsequently causes shortening of bridging bond length of T-O (e.g., Newton and Gibbs 1980) that contributes to increase in elastic moduli (K_S and G). This structural change in the I(-1) stability field provides a reasonable explanation why the elastic wave velocities of anorthite increase with increasing temperature in the range of 500-900 K. Above 900 K, thermal expansion is dominated by the expansion of inter-atomic distance and results in the reduction of elasticity. Angel (1992) performed single crystal X-ray diffraction experiments on five samples with various degrees of Al and Si cation disordering, and found that the pressure of phase transition from P(-1) to high-pressure I(-1) was dependent on the state of Si and Al order. When $Q_{\rm OD}$ (where $Q_{\rm OD}$ quantifies the state of order, varying from unity for complete Al and Si order to zero for complete disorder) decreases from 0.92 to 0.78, the pressure of the phase transition increases from 2.5 GPa to 4.8 GPa. In our sample, existence of the 11(-1) reflection seems to have suggested a possibility that the sample was ordered because the 11(-1) reflection should become less intense when the $Q_{\rm OD}$ decreases (Carpenter, 1991). However, for the synthesis method and condition of the anorthite in our study, it can be estimated that the state of Al and Si order should be partly disordered (Carpenter, 1991). The pressure condition of weak kink in the velocities and compression curves at ~4.0 GPa (Figures 6 and 9) is consistent with the phase boundary between P(-1) and high-pressure I(-1) for partly

disordered anorthite for $Q_{\text{OD}} \cong 0.8$. The unit cell angles of α and γ are known to be sensitive to

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the state of order, and the partly disordered anorthite ($Q_{OD} \cong 0.7\text{-}0.8$) has unit cell angles $\alpha = 93.2^{\circ}$ and $\gamma = 91.2^{\circ}$ (Carpenter 1991; Angel 1992). At ambient condition, the sample used in this study has $\alpha = 93.1^{\circ}$ and $\gamma = 91.2^{\circ}$ which are quite consistent with that of previous studies for the partly disordered anorthite. Consequently, we conclude that the anorthite sample in the ultrasonic measurements is possible to be partly disordered and the kink of slope at ~4 GPa is caused by the phase transition.

The velocities decrease accompanying pressure increase was observed in wide pressure and temperature conditions of the experiments at above 4 GPa where I(-1) structures are stable (Figure 6). This pressure-induced velocities anomaly (with the negative pressure dependence) is in contrast to the positive effect for typical minerals such as olivine (e.g., Birch 1961, 1963). First, it is considered a possibility that the anomaly is an experimental artefact originated by an influence of differential stress. The pressure increment induces a differential stress in the sample, and the stress may influence to the elasticity. In this study, a soft material (hBN) was used as the sample container to minimize the differential stress at the compression, and the measurements were performed after heating to reduce the stress. Based on the in situ X-ray diffraction measurements, the reduction of the stress was observed in the sample and pressure maker at the heating. Consequently the influence of the differential stress to the elasticity was not large in our experiments, and we considered that the pressure induced anomaly is not an experimental artefact. Here, we suggest that the pressure induced anomaly with the negative pressure dependence is caused by the tilting behavior of T-O-T angles in the network of TO₄ tetrahedra. The MD simulation indicated that the average T-O-T angle of anorthite decreases with increasing pressure, and this structural behaviour happens especially at the high-temperature phase and the high-pressure phase with I(-1) symmetry (Noritake et al. 2013, 2014). When the T-O-T angle decrease and hence bridging bond length of T-O increase

with increasing pressure, elastic wave velocities are likely to decrease. In our experiments, the anomaly mainly occurs in the I(-1) stability fields irrespective of the existence of P(-1) structure.

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Comparison with previous plagioclase studies

Previous determinations of elastic wave velocities of plagioclase at room temperature in the thermodynamically stability field are compared with our data (Figure 10). Ryzhova (1964) was the first to measure elastic wave velocities of plagioclase single crystals with various chemical compositions at room temperature and room pressure. These data suggest that velocities of plagioclase have a positive linear correlation with anorthite content. After that, Liebermann and Ringwood (1976) and Kono et al. (2008) measured the elastic wave velocities of polycrystalline plagioclase with An100 at 0.75 GPa and with An51 at 1 GPa, respectively, employing an ultrasonic pulse transmission technique using a piston cylinder apparatus without X-ray radiographic imaging. The data of Liebermann and Ringwood (1976) and Kono et al. (2008) do not agree with Ryzhova's data, and were faster. Liebermann and Ringwood (1979) speculated that Ryzhove had underestimated the data, possibly due to the presence of cracks, pores or inclusions in the single crystal measured at atmospheric pressure. Here, we point out another possibility that the velocities measured using piston cylinder apparatus had been the ones overestimated resulting, from overestimation of sample length because the sample lengths could not be measured directly in their studies. Higo et al. (2009) observed that sample length measured by X-ray radiographic imaging is shorter than that estimated from unit-cell volumes in high pressure for their cell assembly. Their observation indicates that the sample deforms not only through isotropic elastic deformation but also through plastic deformation owing to uniaxial compression of the sample by a hard buffer rod.

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This implies that plagioclase sample lengths estimated in previous studies of plagioclase possible to be overestimated if the samples were uniaxially compressed using a piston cylinder apparatus. In contrast, sample lengths measured in our study are considered to be accurate because they were measured directly using synchrotron X-ray radiography.

Our anorthite velocity data are compatible with the trends reported by Ryzhova (1964), and are slower than that of Liebermann and Ringwood (1979). There are two factors which may explain this inconsistency. One is the preferred orientation of the anorthite crystals, which may have developed during the experiment. Anorthite has strong elastic anisotropy, thus the ultrasonic measurements on a highly textured sample would not yield correct aggregate velocities. In this study, the sample experienced not only isotropic elastic deformation but also weak plastic deformation owing to uniaxial compression by a hard Al₂O₃ buffer rod. In order to eliminate this possibility, crystallographic orientation of the recovered sample was measured by electron backscatter diffraction method. As a result, the preferred orientation was not observed (Figure 11). Therefore, it is thought that lowering of velocity accompanying elastic deformation had not occurred during the high pressure experiment. The second is the presence of pores in the sintered polycrystalline anorthite during the measurement of elastic wave velocities. Our sintered polycrystalline anorthite contained ~1.2 % of pore before the experiments. After the experiments (Figure 1b), the pores were not completely closed although they were considerably lessened ($< \sim 0.5$ %). To evaluate the influence of pore on velocity lowering, we estimated elastic wave velocities of pore-free polycrystalline sample using the theory of Takei (2002). Since the shape of pores was fairly globular, calculation was performed assuming a dihedral angle of larger than 70°. The calculated elastic wave velocities of a pore-free polycrystalline anorthite are very close to the original data and are only higher than those of the pore-bearing sample by 0.4 % and 0.8 %

for V_S and V_P , respectively (Figure 10). It is concluded that the elastic wave velocities measured in our experiment is considered to be appropriate as aggregate velocities although it is slightly slower than the value of a sample without pores.

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Implications for the crustal rocks in the subducting slab

Thin low-velocity layer having thickness of about several km has been observed in the upper portion of the subducting plate in the Japan arcs (e.g., Fukao et al. 1983; Hasegawa et al. 1994; Nakajima et al. 2009). It reaches depths of \sim 150 km ($P = \sim$ 5 GPa) in central Japan and ~ 100 km ($P = \sim 3$ GPa) in northeast Japan, and disappears with increasing depths. Nakajima and his co-workers interpreted the low-velocity layer as a metamorphosed crust coexisting with interstitial aqueous fluid, because the observed V_P at depths of < 100 km in this layer is ~10% lower than that of fully-hydrous metamorphic MORB in blueshist facies (Shiina et al. 2013). This model is the basis for the hypothesis that subducting mafic rocks suffer complete metamorphic hydration before subduction. Plagioclase feldspars are one of most abundant minerals in gabbroic rocks that are subducted into the Earth's interior. Therefore the elastic anomaly in plagioclase has the potential to cause the low velocity anomaly in the slab in case of the cold slab in which the hydration reaction from gabbro to hydrous metamorphic rock is slow (Hasegawa et al. 1994). Our experimental results suggest that anorthite is metastable to at least less than 800 K and 7.0 GPa within the timescale of the experiments, although plagioclase will transform to a garnet-bearing assemblage at a depth of ~35 km when thermodynamic equilibrium is achieved at high temperature (>1300 K) (Green and Ringwood, 1967). Consequently, plagioclase may potentially survive in the slab to higher pressures if the slab is sufficiently cold.

Here, seismic velocities of gabbro and eclogite with NMORB composition were

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337 calculated for the range of pressure and temperature assumed to be present in the subducting 338 slab. The one-dimensional geotherms along the slab-wedge interface for the central and 339 northeast Japan arcs were used as reference models for temperature (Iwamori, 2007). The modal and chemical compositions of minerals were estimated using software for thermodynamic calculation "PerpleX667" (Connolly 2012). The value of thermoelastic parameters for coesite, kyanite, olivine, pyroxenes and garnet are after Hacker et al. (2003) and Matsukage et al. (2005). The gabbro with MORB composition contains Na-bearing plagioclase with An55-80. However the pressure dependence of the elastic wave velocity of Na-bearing plagioclase has not been investigated. Therefore, the data of anorthite measured in our experiment was used. Because the slab temperature is consistent with that of high-temperature I(-1) stability field, we modelled temperature and pressure dependent elastic wave velocities of anorthite with high-temperature I(-1) structure using a quadratic expression with least squares fit: $V_X = a_0 + a_1T + a_2T^2 + bP$, where V_X (X = P, S), T and P is the wave velocity, temperature and pressure. The parameters were determined to be $a_0 = 6.407$ km/s, a_1 = 1.085×10^{-3} km/s/K, $a_2 = -5.630 \times 10^{-7}$ km/s/K², b = -0.0280 km/s/GPa at V_P , and to be $a_0 =$ $3.247 \text{ km/s}, a_1 = 7.198 \times 10^{-4} \text{ km/s/K}, a_2 = -3.601 \times 10^{-7} \text{ km/s/K}^2 \text{ and } b = -0.0400 \text{ km/s/GPa at } V_S.$ Elastic wave velocities of An55 are estimated to be slower by ~5 % compared with that of An100 (Figure 10). Since ~60 % of plagioclase is contained in the gabbroic rock, the maximum bulk velocity lowering due to the compositional effect in plagioclase is estimated to be \sim 3%. As shown in Figure 12, the observed $V_{\rm S}$ structures in the Japan arcs (Nakajima et al. 2009) are consistent with those of gabbro but lower than those of eclogite at a depth of $< \sim 100$ km. For central Japan, the V_P structure reported by Nakajima et al. (2009) is consistent with the gabbro data. However, for northeast Japan, the eclogite velocities are more consistent with the V_P by Nakajima et al. (2009). Nakajima et al. (2009) estimated the V_P structure by direct P

wave. Recently, the V_P structure was estimated by using P-to-S-converted wave, which is considered to be useful for the estimation of V_P structure at the upper portion of the subducting plate (Shiina et al. 2013). The data by P-to-S-converted wave is slower than the data by direct P wave. The V_P reported by Shiina et al. (2013) slightly lower than that of gabbro (Figure 12). The above comparison of the mineralogical model with existing seismic observations shows that the presence of metastable gabbro may reasonably explain observed low velocity in the slab without necessitating the introduction of hydration processes or a fluid phase. At present, it is difficult to determine whether the subducting crust is comprised of gabbroic or metamorphic mineral assemblies with aqueous fluid. Further studies are needed on elastic wave velocity measurements for Na-bearing plagioclase and on the reaction kinetics of metamorphic hydration of gabbro and gabbro-eclogite transition to address this issue.

Acknowledgments

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505	
506	Figure captions
507	Figure 1. Orientation contrast images of the synthesized polycrystalline anorthite taken using
508	forescatter detectors in scanning electron microprobe equipped with field emission gun
509	(FE-SEM) with the polished sample inclined at 70° (Prior et al. 1999). a) Before experiment,
510	b) after experiment. Bright particles in (b) are reaction products at high pressure.
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512	Figure 2. A schematic illustration of cell assembly for elastic wave velocity measurements.
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514	Figure 3. A typical example of the X-ray radiographic image (a) and the ultrasonic signals (b
515	and c) at 1373 K and 1.2 GPa. R1, R2 and R3 are shown in Figure 2.
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518	on anorthite. Open diamonds denote the preheating stage. Phase relations of anorthite are after
519	Hackwell and Angel (1995) and Koziol and Newton (1988). The pressure of the phase
520	transition from P(-1) to high-pressure I(-1) was dependent on the state of Si and Al order, and
521	when $Q_{\rm OD}$ (see text) decreases from 0.92 to 0.78, the pressure of the phase transition increases
522	from 2.5 GPa to 4.8 GPa (Angel, 1992). Open circles indicate the pressure and temperature
523	conditions corresponding to the temperature-induced kink in velocities.
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525	Figure 5. Variations in V_P (a-c), V_S (d-f) of cycles (2)-(8) as a function of temperature. Error of
526	velocities is listed in Table 1.
527	
28	Figure 6. Variations in a) V_P , and b) V_S as a function of pressure at temperatures of 300-773 K.

529 Error of velocities is listed in Table 1. 530 Figure 7. A typical diffraction pattern for anorthite at room temperature and room pressure. 531 532 533 Figure 8. Unit cell parameters of anorthite as a function of temperature. Error of the unit cell 534 parameters (one standard deviation) is smaller than symbols. 535 Figure 9. Unit cell volume of anorthite as a function of pressure. Error of the unit cell volume 536 537 (one standard deviation) is smaller than symbols. 538 Figure 10. Elastic wave velocities of plagioclase with various chemical compositions at room 539 temperature and at pressures of ≤1 GPa. The Ryzhova's data are Voigt-Reuss-Hill averages of 540 541 single crystals. Gray circles show the velocities after correction for pore effect. 542 Figure 11. Pole figures showing crystallographic orientations of anorthite grains in recovered 543 sample from \$2708 (equal area lower hemisphere projections of [100] and [(-1)00] directions, 544 (010) and (001) planes). The orientations of grains were measured by electron backscatter 545 546 diffraction method using FE-SEM (Ohuchi et al. 2011). 404 measurements are used. The 547 cylinder axis is vertical. 548 549 Figure 12. Comparison of V_P and V_S in the crustal portion of the subducting slab in the Japan 550 arcs with those of gabbro and eclogite. Solid circles show V_P and V_S values along A-A' and B-B' sections of northeastern and central Japan of Nakajima et al. (2009). Open diamonds 551 denote averaged VP in northeastern Japan estimated using P-to-S-converted waves (Shiina et 552

553 al. 2013)

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Table 1. Experimental conditions, density, elastic wave velocities and elastic parameters for anorthite

for anorthite							
Temperature	Load	Pressure	$V_{\mathtt{P}}$	$V_{ m S}$	ρ	K_{S}	G
(K)	(ton)	(GPa)	(km/s)	(km/s)	(g/cm ³)	(GPa)	(GPa)
Run No. S270	8						
before experin	nent						
300	0	0.0001			2.754		
heating (2)							
1373	150	1.21(10)	6.81(1)	3.51(1)	2.765	82.6(4)	34.1(2)
1273	150	1.09(7)	6.85(1)	3.54(<1)	2.765	83.6(3)	34.6(1)
1173	150	1.07(8)	6.87(1)	3.55(1)	2.767	84.1(7)	34.8(1)
1073	150	0.99(13)	6.89(1)	3.56(1)	2.768	84.4(4)	35.2(2)
973	150	0.95(11)	6.90(2)	3.57(1)	2.769	85.1(7)	35.2(1)
873	150	0.90(9)	6.90(<1)	3.57(1)	2.770	84.9(2)	35.3(1)
773	150	0.82(8)	6.89(1)	3.56(1)	2.771	84.8(4)	35.1(2)
673	150	0.75(9)	6.87(<1)	3.55(1)	2.772	84.2(2)	34.8(2)
573	150	0.66(10)	6.82(1)	3.51(1)	2.773	83.5(3)	34.2(2)
473	150	0.67(7)	6.81(1)	3.48(<1)	2.775	84.0(4)	33.7(1)
373	150	0.62(7)	6.95(<1)	3.66(1)	2.777	84.5(2)	37.3(1)
300	150	0.63(4)	7.01(1)	3.73(<1)	2.778	84.9(4)	38.6(1)
heating (3)							
773	220	2.83(6)	6.84(3)	3.47(1)	2.847	87.7(11)	34.3(2)
673	220	2.82(9)	6.79(3)	3.45(1)	2.850	86.3(10)	33.8(1)
573	220	2.67(10)	6.83(3)	3.46(<1)	2.853	87.6(12)	34.1(1)
473	220	2.62(4)	6.88(3)	3.51(1)	2.855	88.4(12)	35.2(1)
373	220	2.56(6)	6.95(2)	3.60(1)	2.857	88.6(9)	37.1(3)
300	220	2.53(2)	7.00(1)	3.66(1)	2.858	88.8(6)	38.4(2)
heating (4)							
773	250	3.79(18)	6.81(1)	3.44(1)	2.879	88.0(4)	34.1(2)
673	250	3.71(7)	6.80(1)	3.44(<1)	2.880	87.8(4)	34.1(<1)
573	250	3.63(8)	6.83(1)	3.46(<1)	2.882	88.5(5)	34.5(1)
473	250	3.60(4)	6.90(3)	3.53(<1)	2.884	89.2(10)	36.0(<1)
373	250	3.54(3)	6.95(2)	3.61(<1)	2.885	89.5(9)	37.6(<1)

	300	250	3.53(4)	7.00(1)	3.65(1)	2.885	90.1(5)	38.4(2)
1	heating (5)							
	773	280	4.55(14)	6.78(1)	3.41(1)	2.898	88.2(5)	33.7(2)
	673	280	4.42(8)	6.78(3)	3.42(1)	2.898	88.0(13)	33.9(2)
	573	280	4.35(7)			2.900		
	473	280	4.26(3)	6.89(2)	3.53(1)	2.902	89.5(9)	36.1(1)
	373	280	4.28(1)	6.95(7)	3.61(2)	2.902	89.8(28)	37.7(4)
	302	280	4.24(4)	6.97(1)	3.64(1)	2.903	89.7(6)	38.4(2)
	573	280	4.45(6)	6.82(2)	3.46(1)	2.902	88.9(7)	34.7(1)
	300	280	4.37(4)	6.96(2)	3.64(1)	2.907	89.7(8)	38.4(2)
ŀ	neating (6)							
	773	320	5.05(4)	6.74(2)	3.39(1)	2.922	88.1(7)	33.6(1)
	673	320	4.84(7)	6.77(2)	3.41(1)	2.921	88.6(7)	34.1(1)
	573	320	4.76(9)	6.82(2)	3.46(1)	2.923	89.2(8)	35.0(1)
	473	320	4.71(5)	6.87(2)	3.53(1)	2.923	89.5(9)	36.4(2)
	373	320	4.68(1)	6.92(2)	3.59(1)	2.923	89.7(10)	37.7(2)
	300	320	4.64(9)	6.94(3)	3.63(1)	2.926	89.7(12)	38.5(2)
h	eating (7)							
	773	390	5.66(14)	6.72(1)	3.38(1)	2.944	88.2(4)	33.6(1)
	673	390	5.60(3)	6.75(2)	3.41(<1)	2.943	88.6(6)	34.2(1)
	573	390	5.52(5)	6.81(2)	3.47(1)	2.944	89.5(7)	35.4(2)
	473	390	5.33(2)	6.86(3)	3.53(1)	2.945	89.9(12)	36.6(3)
	373	390	5.15(7)	6.89(2)	3.58(1)	2.944	89.7(9)	37.6(2)
	300	390	5.00(35)	6.92(1)	3.61(1)	2.948	89.9(6)	38.4(2)
h	eating (8)							
	773	480	6.99(1)	6.67(1)	3.36(<1)			
	673	480	6.71(12)	6.70(1)	3.40(1)	2.988	88.4(5)	34.5(1)
	573	480	6.59(13)	6.77(2)	3.46(1)	2.988	89.4(7)	35.7(2)
	473	480	6.60(3)	6.81(2)	3.51(1)	2.989	89.5(10)	36.8(2)
	373	480	6.51(8)	6.84(2)	3.55(1)	2.989	89.6(10)	37.7(1)
	300	480	6.52(3)	6.86(2)	3.57(1)	2.991	89.7(9)	38.2(1)

Numbers in parentheses are the one standard deviation (1 σ) in the last digit. 1 σ of density is less than 0.001 (g/cm³)

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APPENDIX

The in situ elastic wave velocity measurements of anorthite were performed twice. In first experiments (S2646), the measurements were carried out under load of 150 ton and 250 ton after preheating process. We, however, could not obtain reliable temperature and pressure data at except for room temperature because thermocouple was broken during preheating process. In second experiment (S2708), we succeeded to measure the temperature directly by thermocouple, and the reliable data in wide pressure and temperature range were corrected (see text). Therefore, only the second experiment was used for the discussion. Here, in order to prove reproducibility of the elastic wave velocity anomalies observed by \$2708, the results of S2646 are compared. Temperature in S2646 was estimated using relation between temperature and electric power at the same loads in S2708. In S2646, pressures at room temperature after heating was calculated from the EOS of MgO as 1.5 GPa for 150 ton and 3.1 GPa for 250 ton, respectively. The pressure at 150 tons was higher than that of 150 tons of S2708, and, in the case of 250 tons, it becomes lower. Therefore there may be larger uncertainty in this temperature estimation. As shown in Figure A1, the elastic wave velocity anomalies were evident in the experiment of S2646. The variation of V_P of S2646 agrees well with that of S2706. In the experiments at 150 ton, the $V_{\rm S}$ of S2646 becomes slower than that of S2708. The lower V_S in S2646 are considered to originate from the difference in pressure because the dV_S/dP shows a negative though dV_P/dP is constant at below ~4 GPa. In conclusion, these comparisons demonstrate the reproducibility of our experiments.

Figure A1. Elastic wave velocities as a function of temperature in S2646 and S2708.
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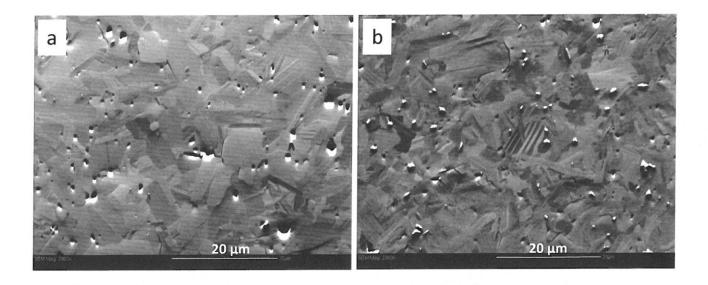


Figure 1. Orientation contrast images of the synthesized polycrystalline anorthite taken using forescatter detectors in scanning electron microprobe equipped with field emission gun (FE-SEM) with the polished sample inclined at 70 ° (Prior et al., 1999). a) Before experiment, b) after experiment. Bright particles in (b) are reaction products at high pressure.

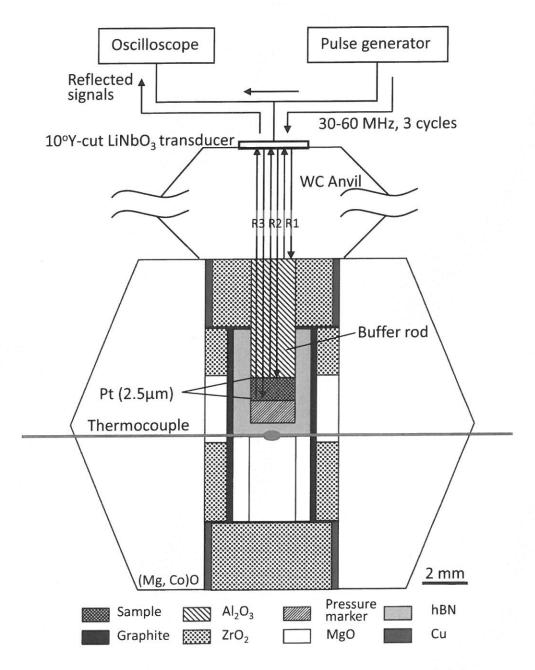


Figure 2. A schematic illustration of cell assembly for elastic wave velocity measurements.

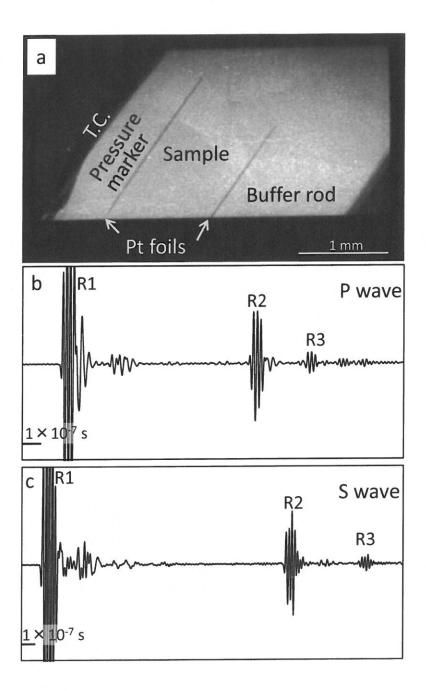


Figure 3. A typical example of the X-ray radiographic image (a) and the ultrasonic signals (b and c) at 1373 K and 1.2 GPa. R1, R2 and R3 are shown in Figure 2.

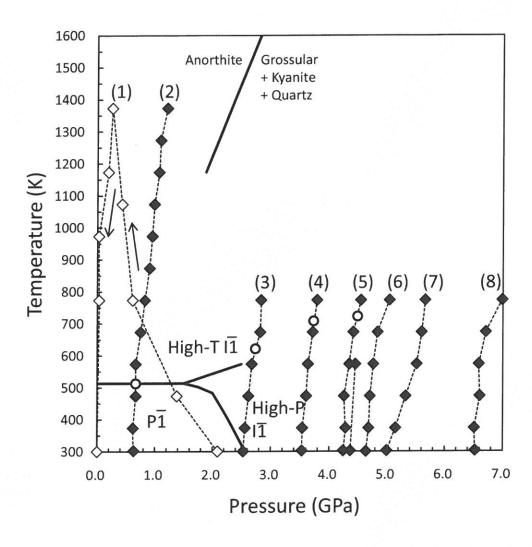


Figure 4. Experimental pressure and temperature paths in elastic wave velocity measurements on anorthite. Open diamonds denote the preheating stage. Phase relations of anorthite are after Hackwell and Angel (1995) and Koziol and Newton (1988). The pressure of the phase transition from P(-1) to high-pressure I(-1) was dependent on the state of Si and Al order, and when $Q_{\rm OD}$ (see text) decreases from 0.92 to 0.78, the pressure of the phase transition increases from 2.5 GPa to 4.8 GPa (Angel, 1992). Open circles indicate the pressure and temperature conditions corresponding to the temperature-induced kink in velocities.

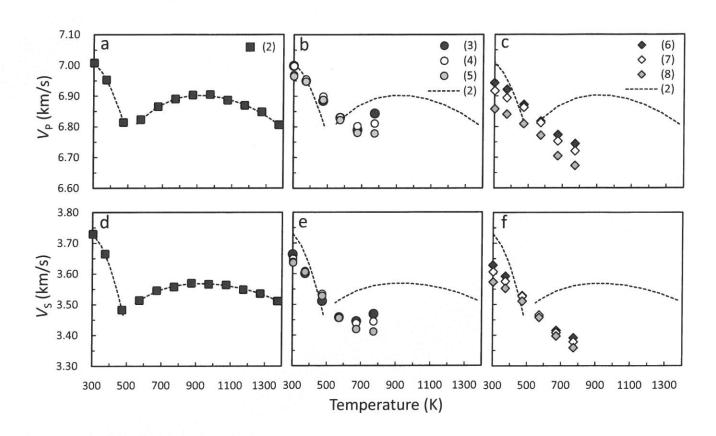


Figure 5. Variations in V_P (a-c), V_S (d-f) of cycles (2)-(8) as a function of temperature. Error of velocities is listed in Table 1.

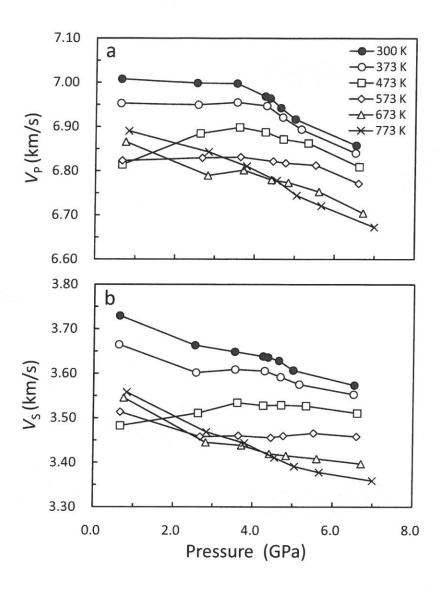


Figure 6. Variations in a) $V_{\rm P}$, and b) $V_{\rm S}$ as a function of pressure at temperatures of 300-773 K. Error of velocities is listed in Table 1.

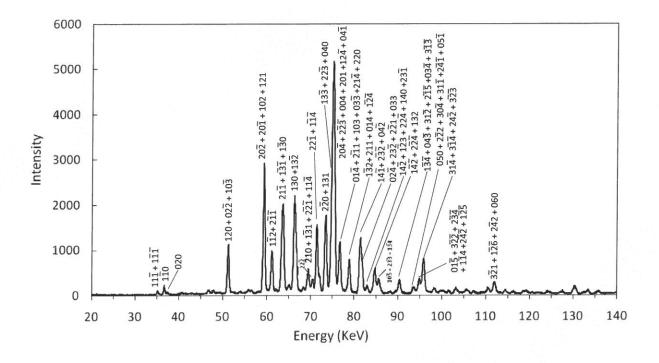


Figure 7. A typical diffraction pattern for anorthite at room temperature and room pressure.

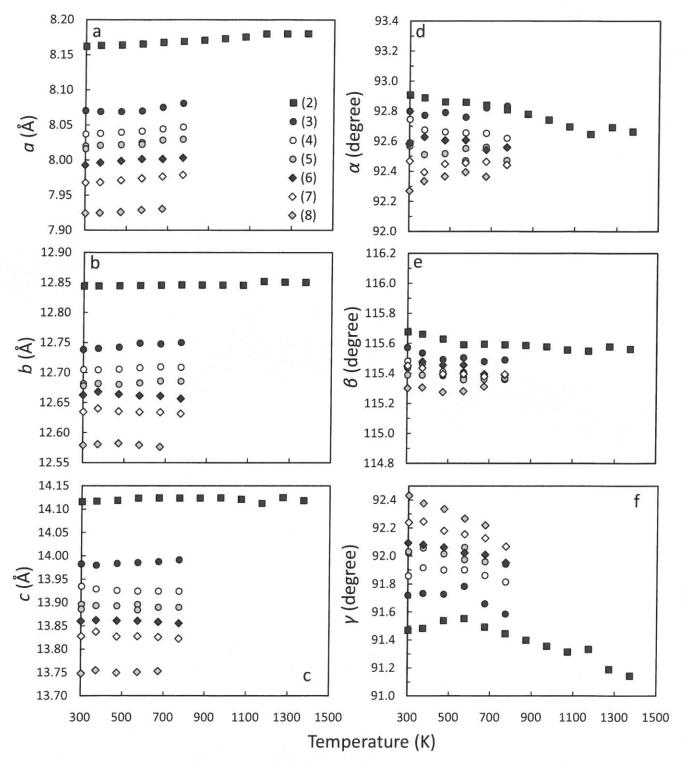


Figure 8. Unit cell parameters of anorthite as a function of temperature. Error of the unit cell parameters (one standard deviation) is smaller than symbols.

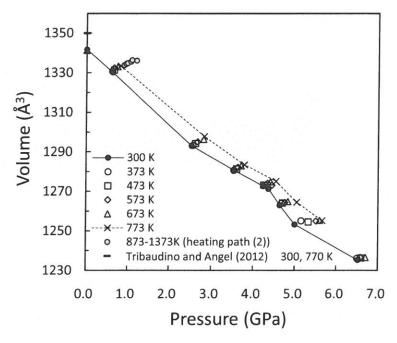


Figure 9. Unit cell volume of anorthite as a function of pressure. Error of the unit cell volume (one standard deviation) is smaller than symbols.

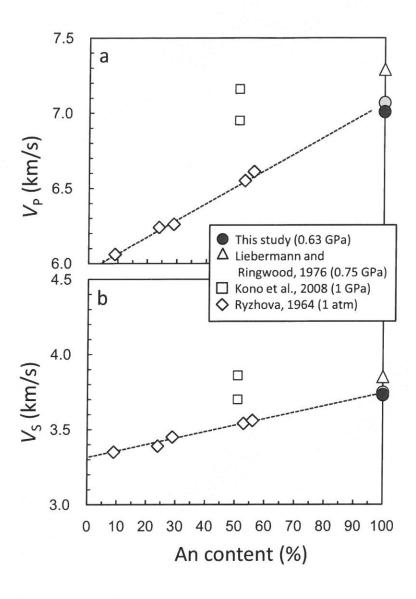


Figure 10. Elastic wave velocities of plagioclase with various chemical compositions at room temperature and at pressures of ≤1 GPa. The Ryzhova's data are Voigt-Reuss-Hill averages of single crystals. Gray circles show the velocities after correction for pore effect.

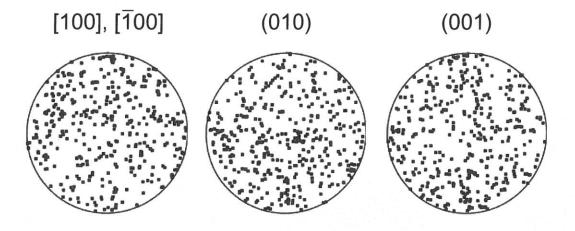


Figure 11. Pole figures showing crystallographic orientations of anorthite grains in recovered sample from S2708 (equal area lower hemisphere projections of [100] and [(-1)00] directions, (010) and (001) planes). The orientations of grains were measured by electron backscatter diffraction method using FE-SEM (Ohuchi et al. 2011). 404 measurements are used. The cylinder axis is vertical.

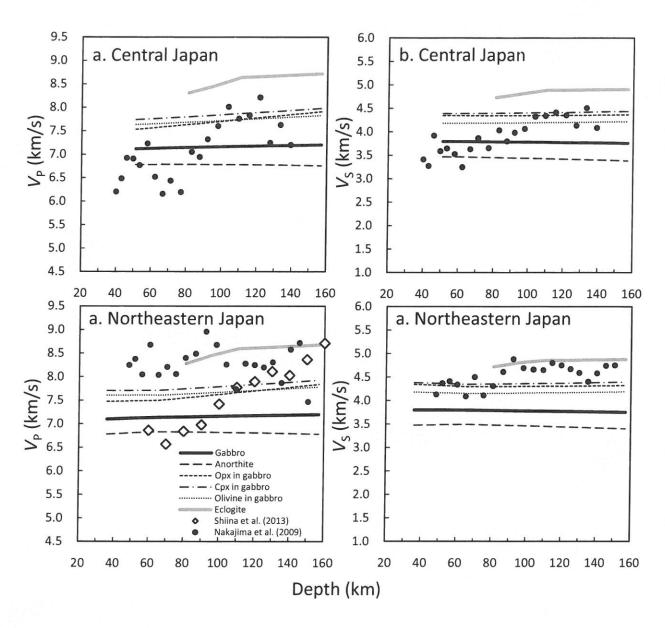
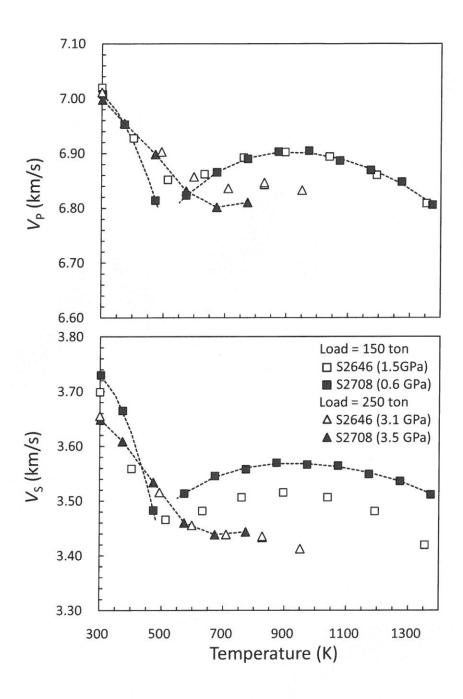


Figure 12. Comparison of V_P and V_S in the crustal portion of the subducting slab in the Japan arcs with those of gabbro and eclogite. Solid circles show V_P and V_S values along A-A' and B-B' sections of northeastern and central Japan of Nakajima et al. (2009). Open diamonds denote averaged V_P in northeastern Japan estimated using P-to-S-converted waves (Shiina et al. 2013).



Appendix **Figure A1.** Elastic wave velocities as a function of temperature in S2646 and S2708.