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Physio-chemical properties of gamma-irradiated vermiculite and their significance for

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radiation protection and thermoluminescence

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7

Abstract

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The present work reports the effects of gamma irradiation of vermiculite for the first time. The

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radiation induced changes of vermiculite were studied using different techniques viz: ultraviolet-

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visible spectroscopy (UV-Vis), dielectric measurements, X-ray diffraction (XRD), Fourier

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Transform Infrared spectroscopy (FTIR) and thermoluminescence (TL). In UV-Vis analysis, the

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Cody model was employed to calculate structural disorder from Urbach energy which explained

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the variation of the optical band gap (direct and indirect) with different (1- 2000 kGy) gamma

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doses. XRD analysis of the pristine and irradiated samples shows that the crystallinity improved

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upon irradiation at gamma dose upto 1000 kGy and deteriorated on further increase of the

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gamma dose. A significant change was observed in the dielectric properties after gamma

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irradiation. Data shows that a.c. conductivity is proportional to the nth power of frequency (f^n)

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in pristine and irradiated vermiculite, with a slope n ranging between 0.52 and 0.76 which

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indicates that electronic conduction takes place through an electron hopping process. No

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appreciable changes in characteristic bands (FTIR) have been observed after irradiation,

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indicating that natural vermiculite is chemically stable. A well defined TL peak around 132 °C

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and enhancement in its intensity with gamma dose (1- 1000 kGy) make vermiculite a perfect

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thermoluminescence dosimeter and indicates usefulness applications in radiation dosimetry.

24 **Keywords:** Natural Vermiculite · Gamma irradiation· Optical properties · Dielectric properties·
25 XRD· FTIR · Thermoluminescence

26 **Introduction**

27 Natural vermiculite is a constituent of the phyllosilicate or sheet silicate group of mineral
28 including Al_2O_3 , H_2O , MgO , FeO and SiO_2 which resembles mica in appearance (Brown 1961).
29 Vermiculite is a very important secondary mineral that is formed primarily by the hydrothermal
30 alteration of mica (biotite, phlogopite) or other clay minerals (Basset 1963). Vermiculites are a
31 significant group of clay minerals that consists of two tetrahedral layers as silica and alumina and
32 an octahedral layer of oxygen, magnesium, iron and hydroxyl molecules. Water located between
33 layers is an important part of the vermiculite structure. Vermiculite has long history of
34 applications in insulation, fire resistance, advanced materials, building industry, ceramics,
35 agriculture, horticulture and industrial markets (Strand and Stewart 1983; Hindman 1992;
36 Bergaya et al. 2006; Klein and Dutrow 2007). Nowadays, these natural minerals are in demand
37 due to its high thermal and insulation applications and sensitivity to ionizing radiation.
38 Irradiation of phyllosilicate minerals with gamma and heavy ions leads to remarkable changes in
39 their structural, chemical, optical and electrical properties (Kaur et al. 2013a, 2013b, 2013c).
40 A significant amount of work has been done on the effect of microwaves on exfoliation
41 properties and thermal behaviour of vermiculite by several groups (Marcos et al. 2009; Muiambo
42 et al. 2010; Marcos and Rodriguez 2011; Folorunso et al. 2013; Campos et al. 2009; Sakharov et
43 al. 2001), but literature survey, on the other hand, revealed no work on gamma irradiated induced
44 modifications of vermiculite. In the present work, effect of gamma irradiation on optical,
45 chemical, structural and dielectric properties of natural vermiculite minerals has been studied for

46 the first time to identify good materials which have high radiation shielding capacity and thermal
47 insulation property.

48 Thermo luminescence (TL) is a crucial phenomenon that takes place in various irradiated natural
49 minerals (Kaur et al. 2013c). In order to investigate the thermo luminescent characteristics of
50 vermiculite and to ascertain whether this mineral could qualify as a local dosimeter, the samples
51 of this work were subjected to TL studies. The present research investigates the thermo
52 luminescence response of gamma irradiated natural vermiculite to determine their relevance to
53 dating and radiation dosimetry.

54 **Experimental material and methods**

55 The natural Vermiculite mineral used in the present work was procured from Viozag area in
56 Andhra Pradesh, state of India. The original raw material was cleaved into thin sheets, uniform in
57 thickness and color by using a scalpel. Thickness of each film was measured with a micrometer
58 and is approximately $55 \pm 1 \mu\text{m}$.

59 The sheets of vermiculite were further analysed for elemental composition using Energy
60 Dispersive X-ray Spectroscopy (EDS) system attached to a Carl Zesis supra 55 field emission
61 SEM. A thin sheet of vermiculite was placed on a sample holder using silver tape and coated
62 with a layer of gold using a Quorum sputtering coater to make them conductive. Then the sample
63 was transferred to a microscope chamber for EDS measurements. The minimum detectable limit
64 of the EDS system attached with Carl Zesis Supra 55 field emission SEM is 0.2 wt%.

65 Thin sheets of vermiculite samples of $1 \times 1 \text{ cm}^2$ were irradiated at room temperature by γ -
66 radiation using ^{60}Co source from Gamma Chamber-1200 at IUAC, New Delhi to different doses
67 in the range of 1- 2000 kGy. γ -rays emitted by ^{60}Co have a mean energy of 1.25 MeV and the
68 dose rate in the irradiator was 7.5 kGy/h.

69 The nature of the changes induced by the gamma radiation in vermiculite were examined using
70 HITACHI U-3300 UV-Visible spectrophotometer in the range 200-1100 nm in order to
71 investigate the variation in optical energy gap and Urbach energy. Dielectric properties were
72 analyzed using Hewlett- Packard 4284A LCR meter in the frequency range of 20 Hz- 1 MHz at
73 room temperature. The accuracy of LCR meter for Capacitance- $\tan\delta$ measurements is 0.05%-
74 0.005% respectively at all text frequencies. The structural studies were carried out by a Bruker
75 AXS X-ray (CuK_{α} , $\lambda = 0.154\text{nm}$) diffractometer for a range of Bragg's angle 2θ ($5 < \theta < 60$) with
76 scanning speed of 2° per minute. FTIR spectra of pristine and irradiated vermiculite were
77 measured by NEXUS-670 FTIR spectrometer, in the range of $4000\text{-}400\text{ cm}^{-1}$. The sheets of
78 vermiculite were crushed and ground in high energy ball mill to prepare the powder samples for
79 TL properties. TL glow curves were recorded at a heating rate of 5°Cs^{-1} on a Harshaw TLD
80 reader (Model 3500) taking 5.0 ± 0.2 mg of each sample.

81 **Results and Discussion**

82 **EDS Analysis and Optical properties**

83 The EDS analysis indicates that the natural vermiculite mainly consists of silicon (Si),
84 magnesium (Mg), aluminium (Al), iron (Fe) and with some amounts of calcium (Ca) (Table 1).
85 Although the EDS elemental composition analysis are semiquantitative, provided an iron content
86 higher than samples from other sources (Marcos and Rodriguez 2010; Muiambo et al. 2010).
87 Figure 1 presents the UV-visible absorption spectra of the pristine and gamma irradiated (10,
88 100, 500, 1000, 1500 and 2000 kGy dose) vermiculite mineral in the wavelength (λ) range 350-
89 1100 nm. A broad absorption band was observed at 704 nm in pristine and gamma irradiated
90 samples which is characteristics of biotite impurity (Karickhoff and Bailey 1973). A shift of
91 absorption edge towards the lower wavelength side was readily observed upto to dose of 1000

92 kGy, but it shifts towards a much higher wavelength with further increase in dose greater than
93 1500 kGy (Figure 1). The vermiculite exhibits a blue shift upto gamma dose of 1000 kGy,
94 whereas it exhibit red shift at higher gamma doses upto 2000 kGy. The blue shift shows that the
95 optical band gap of the vermiculite was highly enhanced upto gamma dose of 1000 kGy. Davis
96 and Mott (1979) gave an expression for the absorption coefficient, $\alpha(h\nu)$, as a function of photon
97 energy ($h\nu$) for direct and indirect optical transitions through the following equation

$$98 \quad \alpha(h\nu) = B(h\nu - E_g)^n / h\nu \quad (1)$$

99 where $h\nu$ is the energy of incident photons, E_g is the value of the optical energy gap between the
100 valence band and the conduction band, B is a constant and the exponent n characterizes the type
101 of electronic transition whether it is a direct or indirect band transition. Specifically, n is $1/2$ and
102 2 for direct and indirect optical band gap respectively. The $(\alpha h\nu)^{1/2}$ and $(\alpha h\nu)^2$ were plotted as a
103 function of photon energy ($h\nu$) respectively, to determine indirect and direct band gap. The
104 calculated optical energy band gaps E_g (direct and indirect) of the pristine and gamma irradiated
105 samples tabulated in Table 2. The direct and indirect energy band gap in vermiculite has been
106 observed for the first time as no such report was found in literature survey. The value of the
107 optical direct and indirect band gap increases with the increase in gamma dose upto 1000 kGy,
108 however decreases with further increase in gamma dose upto 2000 kGy (Table 2).

109 Urbach energy (E_u) is another parameter which gets significantly affected by irradiation. The
110 irregularities in the band gap level can be defined in terms of Urbach energy. It depends on
111 induced disorder, static disorder, temperature and thermal vibrations in the lattice and on average
112 photon energies. The Urbach energy is estimated using relation (Urbach 1953)

$$113 \quad \alpha(\nu) = \alpha_0 \exp(h\nu/E_u) \quad (2)$$

114 where $h\nu$ is the photon energy, α_0 is a constant, E_u is the Urbach's energy and $\alpha(\nu)$ is an
115 absorption coefficient. The logarithm of the absorption coefficient $\alpha(\nu)$ as a function of the
116 photon energy ($h\nu$) for the pristine and gamma irradiated samples is plotted and Urbach energy
117 can be calculated by taking the slope of the linear part of the plot. The calculated values are
118 listed in Table 2.

119 Cody et al. (1981) first established the relationship between the structural disorder, the Urbach's
120 energy and the temperature (T) for explaining the optical absorption characteristics and has the
121 form

$$122 \quad E_u = \frac{E_p}{2\sigma_0} \left[X + \coth \left(\frac{E_p}{2k_B T} \right) \right] \quad (3)$$

123 where E_p is the phonon energy, σ_0 is Urbach edge parameter of order unity, k_B is Boltzmann
124 constant and X is a measure of structural disorder which is defined as the ratio of the mean
125 square deviation of atomic position $\langle U^2 \rangle$ to the zero-point uncertainty in the atomic position
126 squared $\langle U_0^2 \rangle$

$$127 \quad X = \frac{\langle U^2 \rangle}{\langle U_0^2 \rangle} \quad (4)$$

128 An average value of phonon energy in natural vermiculite is 73.7 meV, as determined by Raman
129 spectroscopy (Rakhshani, 2000). The structural disorder parameter 'X' for each Urbach energy
130 value at temperature of 22 °C is calculated using equation (3) and given in Table 2. The decrease
131 of the Urbach energy with increase in gamma dose upto 1000 kGy indicates the decrease in the
132 structural disorder (Table 2). By further increasing the gamma dose upto 2000 kGy, the increase
133 in the Urbach energy may be attributed to increase in the structural disorder.

134 The Urbach energy decreases with increase of gamma dose upto 1000 kGy (Table 2), which may
135 be due to the decrease of structural disorder (X), which in turns leads to increases the optical
136 band gaps (direct and indirect). The observed increase in the optical band gap and decrease in the
137 Urbach energy both arise due to decrease in structural disorder of vermiculite. However, with
138 further increase in gamma dose upto 2000 kGy the Urbach energy increases abruptly due to a
139 remarkable increase in the structural disorder. This means that the higher gamma dose produces
140 defects, dislocations and disorder in the vermiculite structure, which in turn leads to decrease the
141 optical band gaps (direct and indirect).

142 **Dielectric Properties**

143 The capacitance and $\tan\delta$ (loss factor) of the vermiculite were measured for varying frequencies
144 from 20 Hz to 1 MHz at room temperature. Two probe method involves sandwiching a thin sheet
145 of vermiculite between the two electrodes to form a capacitor (Kaur et al. 2013 c). The dielectric
146 constant (ϵ') and the dielectric loss (ϵ'') of the vermiculite were calculated by using the
147 relations:

$$148 \quad \epsilon' = \frac{Cd}{A\epsilon_o} \quad (5)$$

$$149 \quad \epsilon'' = \epsilon' \tan \delta \quad (6)$$

150 where C is capacitance (F), d is the thickness (m), ϵ_o is the free space permittivity ($\epsilon_o = 8.854 \times$
151 10^{-12} F/m) and A the area of the sample (m^2). The ac electrical conductivity (σ_{ac}) of the dielectric
152 material is given by the following equation:

$$153 \quad \sigma_{ac} = \epsilon' \epsilon_o \omega \tan \delta \quad (7)$$

154 where $\omega = 2\pi f$ is the angular frequency. The electric modulus (M^*) is defined as

$$155 \quad M^* = M' + iM'' \quad (8)$$

156 The real (M') and imaginary (M'') parts of electric modulus (M^*) can be calculated by using the
157 following relation:

158
$$M' = \frac{\epsilon'}{(\epsilon')^2 + (\epsilon'')^2} \quad (9)$$

159
$$M'' = \frac{\epsilon''}{(\epsilon')^2 + (\epsilon'')^2} \quad (10)$$

160 The calculated values of dielectric parameters using the above relations (5-10) are listed in Table
161 3 for different gamma doses at constant frequency of 400 kHz. The dielectric constant decreases
162 with increase of frequency (Figure 2). The dielectric constant is high in the low frequency region
163 and decreases continuously with increase in frequency before and after irradiation. The high
164 value of dielectric constant in the lower frequency region can be assigned to the presence of
165 interfacial i.e., space charge polarization mechanism and ionic conduction. The reduction of the
166 value of dielectric constant at higher frequencies may be due to the loss of significance of these
167 polarizations gradually. Figure 2 also shows that the dielectric constant increases with increasing
168 gamma dose up to 1000 kGy, however it decreases with further increase of gamma dose upto
169 2000 kGy. The loss factor ($\tan\delta$) versus frequency plots for pristine and gamma irradiated
170 vermiculite show positive values that indicate the inductive dominance (Figure 3). The low loss
171 tangent is observed in the gamma dose range of 10- 1000 kGy which concludes that the
172 vermiculite irradiated in this range possesses good optical quality and defect number is very low.
173 The dielectric loss shows the same behaviour as that of $\tan\delta$ with increase of gamma dose
174 (Figure 4). The values of ϵ'' , $\tan\delta$, σ_{ac} , M' and M'' are all lower in the dose range 10- 1000
175 kGy (Table 3). However, these dielectric parameters show an increase with further increase in
176 gamma dose upto 2000 kGy. The reduction of dielectric parameters at gamma dose of 10- 1000
177 kGy may be assigned to the healing effects of radiation, however irradiation induced defect (see

178 above- optical properties) becomes more prominent at higher gamma doses of 2000 kGy which
179 in turn raises the values of these parameters. This means that there is coexistence for the creation
180 and annihilation of defects at gamma doses upto 1000 kGy; at higher gamma doses (2000 kGy),
181 the number of defects induced by irradiation becomes larger than the number of defects
182 annihilated.

183 The complex impedance Z^* of the present material can be described by the following equation:

$$184 \quad Z^* = Z' - iZ'' \quad (11)$$

185 where Z' and Z'' are ascribed to real and imaginary parts of impedance. The relation between
186 electric modulus and complex impedance is given by:

$$187 \quad M^* = i\omega C_0 Z^* \quad (12)$$

188 In the above equations ω is the angular frequency ($=2\pi f$), C_0 is the vacuum capacitance of the
189 measuring cell and electrodes with an air gap of the dimensions of the sample thickness.

$$190 \quad C_0 = \frac{\epsilon_0 A}{d} \quad (13)$$

191 where ϵ_0 is the free space permittivity, d is the thickness and A the area of the sample. The
192 values of Z' and Z'' calculated by substituting the values of M^* and Z^* from equations (9) and
193 (11) in to equation (12) and are as follow:

$$194 \quad Z' = \frac{\omega C_0}{M''} \quad (14)$$

$$195 \quad Z'' = \frac{\omega C_0}{M'} \quad (15)$$

196 The calculated values of Z' and Z'' from above relation for different gamma dose are enlisted in
197 Table 3. It is observed that the values of Z' and Z'' increases with increase in gamma dose upto
198 1000 kGy, which means the gamma dose enhances its insulation properties upto this gamma

199 dose (Table 3). These results also support the enhancement in dielectric constant with gamma
200 dose. However, with further increase of gamma dose upto 2000 kGy, the values of Z' and Z''
201 decreases which indicates the reduction of dielectric constant.

202 A convenient formalism to investigate the frequency dependence of ac conductivity in a material
203 is based on the power law relation proposed by Jonscher (1996)

$$204 \quad \sigma_{ac} \propto f^n \quad (16)$$

205 where the frequency exponent n is temperature independent. The frequency dependence of the ac
206 conductivity of pristine and gamma irradiated vermiculite samples are shown in Figure 5 and the
207 values of n have been calculated from the slope of these straight lines. The calculated values of n
208 lies between 0.52 to 0.76. These are less than unity and together with the linear dependence of ac
209 conductivity indicates that the electrical conductivity may be ruled by an electron hopping
210 process. The similar result was observed previously in other minerals (Kaur et al. 2013b, 2013c).

211 **Structural Analysis**

212 Figure 6 shows the XRD pattern for vermiculite in pristine and irradiated up to the gamma dose
213 of 2000 kGy. In pristine vermiculite five peaks are observed at $2\theta = 8.87^\circ$, 26.67° , 35.79° ,
214 45.14° and 54.83° . These peaks do not show any change in their positions on irradiation with
215 different gamma doses. This indicates that there is no change in the basic crystalline structure of
216 vermiculite. The peaks become sharper and intensity of these peaks also increases with
217 increasing of gamma dose upto 1000 kGy, indicating the enhancement of crystallinity and
218 ordering in the vermiculite. However, with further increase in gamma dose upto 2000 kGy the
219 peaks become broader (Table 4) and intensity decreases which denotes some destruction of the
220 orderliness of the original crystal. The mean crystallite size (D), micro strain (S) and dislocation
221 density (δ) have been be estimated using Williamson-Hall (W-H) analysis from the XRD. W-H

222 analysis is a simplified integral breadth method where both size-induced and strain-induced
223 broadening is deconvoluted by considering the peak width as a function of 2θ (Williamson and
224 Hall 1953). The crystallite size (D) was calculated using Debye Scherrer formula from the full
225 width half maximum (FWHM) measurements

$$226 \quad D = \frac{K\lambda}{\beta \cos \theta} \quad (17)$$

227 here, K is a constant which depends on diffractometer setup, λ is wavelength of X-ray, θ is
228 diffraction angle or Bragg's angle and β denotes FWHM. In the present setup $K = 0.9$ and $\lambda =$
229 1.541 \AA . The values of β and θ can be obtained from the diffraction pattern (Figure 6). The
230 microstrain (S) arising from crystal imperfection and distortion was calculated using formula

$$231 \quad S = \frac{\beta}{4 \tan \theta} \quad (18)$$

232 The observed line width is simply the sum of equations (17) and (18)

$$233 \quad \beta = \frac{\lambda}{D \cos \theta} + 4S \tan \theta \quad (19)$$

234 By rearranging the above equation,

$$235 \quad \frac{\beta \cos \theta}{\lambda} = \frac{1}{D} + S \left(\frac{4 \sin \theta}{\lambda} \right) \quad (20)$$

236 The above equation is known as Williamson-Hall equation. The plot of $\left(\frac{\beta \cos \theta}{\lambda} \right)$ versus
237 $\left(\frac{4 \sin \theta}{\lambda} \right)$ gives the value of the microstrain from slope and the crystallite size from the ordinate
238 intercept. Figure 7 shows the W-H plots for the pristine and the gamma irradiated vermiculite at
239 different gamma doses. The values of crystallite size (D) and microstrain (S) obtained from the
240 intercept and slope of W-H plots are listed in Table 4. A negative slope in the W-H plot indicates

241 the presence of compressive strain experienced by the particle of vermiculite. The dislocation
242 density (δ) calculated using the following formula (Kulg and Alexander 1974) is listed in Table
243 4:

$$244 \quad \delta = \frac{15S}{aD} \quad (21)$$

245 A significant enhancement in crystallite size in irradiated vermiculite samples is observed upto to
246 gamma dose of 1000 kGy; micro strain (S) and dislocation density (δ) decreases in the gamma
247 dose range of 10- 1000 kGy (Table 4). With further increase in gamma dose upto 2000 kGy, the
248 crystallite size decreases, however, unlike the microstrain and dislocation density which increase.
249 This increase in crystallite size in the dose 10- 1000 kGy is due to the formation of some large
250 size crystallites due to irradiation. This binding of small crystallites increases the crystallite size
251 and decreases the micro strain in the structure which in turn decreases the dislocation density
252 (Table 4). This means that increased ordering of vermiculite structure takes place in the gamma
253 dose range 10- 1000 kGy. This is attributed to the improvement in the crystallinity due to the
254 increase in crystallite size, reduction in the defects and the decrease in structural disorder and
255 microstrain. A higher gamma doses upto 2000 kGy, the crystallite size decreases and micro
256 strain and dislocation density increases which indicate convergence of the vermiculite towards a
257 more disordered system. These results are in good agreement with the UV-Vis and dielectric
258 results.

259 **FTIR Analysis**

260 The FTIR spectrum of the pristine and gamma irradiated natural vermiculite illustrates the
261 characteristics bands of natural vermiculite (Muiambo et al. 2010) (Figure 8). A strong
262 absorption band is observed corresponding to the stretching vibration of the Mg₃-OH groups
263 located at 3443 cm⁻¹ related to OH hydroxyl groups of interlamellar water in the pristine

264 vermiculite sample. The infrared band observed at 2350 cm^{-1} revealed the presence of
265 chemisorbed CO_2 . The absorption band appearing at 1642 cm^{-1} is attributed to O–H bending
266 vibrations of the hydration water molecules. A strong band observed at 1004 cm^{-1} is assigned to
267 Si–O–Si and Si–O–Al stretching vibrations. The band appearing at 682 cm^{-1} is associated with
268 Si–O deformation. No new band has been observed in irradiated samples. Hence, gamma
269 irradiation did not induce any significant changes in the bonds, however, the changes in the
270 overall intensity of the transmittance after irradiation was observed (Figure 8). The intensity of
271 the transmittance increases with increase of gamma dose upto 1000 kGy. With further increase
272 of gamma dose upto 2000 kGy, the intensity decreases which indicate structural deformation of
273 vermiculite after irradiation (Kaur et al. 2013c). This agrees well with the results of UV-VIS
274 absorption spectra and XRD as discussed earlier.

275 **Thermoluminescence Analysis**

276 The TL glow curves of vermiculite exposed to gamma radiation at different doses (1- 2000 kGy)
277 are depicted in Figure 9. A glow curve of the vermiculite irradiated at different gamma doses
278 consists of one prominent peak around $132\text{ }^\circ\text{C}$, known as a dosimetry peak. The glow curves
279 obtained at all gamma doses show almost the same shape and peak position (Figure 9). The TL
280 intensity of the glow peak enhances systematically with the increase of gamma dose, which
281 indicates that this peak is a radiation sensitive peak. Increase in intensity of glow peaks by
282 gamma irradiation may be due to the increase in the number of traps into the lattice. It is also
283 noticed that though the TL intensity of $132\text{ }^\circ\text{C}$ glow peak of vermiculite increases with increase
284 in gamma dose but there is no appreciable shift in peak temperature.

285 The TL response curve of vermiculite irradiated by various gamma doses is exhibited in Figure
286 10 in which the height of $132\text{ }^\circ\text{C}$ main TL glow peak was used for measuring the TL intensity

287 and the error bars indicates the standard deviation of three measurements from their average
288 value. The TL dose response curve shows almost linear behavior in the dose range of 1 kGy to
289 1000 kGy. This linear behavior over a wide range of gamma dose (1- 1000 kGy) may be due to
290 the fact that number of electrons in the excited state increases with increase of the gamma dose
291 and hence there is a possibility of increase of recombination as well as TL emission. After a
292 gamma exposure of 1500 kGy, a decrease in TL intensity has been noticed with further increase
293 in dose. This fall in the TL intensity at higher doses might be due to the damage of the TL trap.
294 The activation energy of the trap depth (E) calculated using the empirical formulae (El-Kolaly et
295 al. 1994):

$$296 \quad E(eV) = \frac{T_p(K)}{464} \quad (22)$$

297 where T_p is the peak temperature of the TL glow curve. Since the position of the obtained glow
298 peaks is more or less independent of gamma dose, therefore the values of the activation energy
299 of all the trap depths are same and was found to be $E = 0.87$ eV. Thus a distinct radiation
300 sensitive glow peak and linear dose response upto wide range (1- 1000 kGy) leads us to conclude
301 for the first time that natural vermiculite is a suitable material for TL dosimeter in radiation rich
302 environment.

303 **Implications**

304 In the present work, effect of gamma irradiation on optical, chemical, structural and dielectric
305 properties of natural vermiculite minerals have been studied for the first time and indicates that
306 vermiculite is a material that has high radiation shielding capacity and thermal insulation
307 properties. The enhancement in opto-electric properties of vermiculite with gamma irradiation
308 also makes it suitable for optoelectronic devices. Gamma irradiation did not induce any
309 significant changes in the bonds, hence vermiculite is chemically stable. Thermoluminescence

310 properties demonstrate the applicability of natural vermiculite as a cost effective TL dosimeter
311 for innovational dosimetry in radiation rich environs.

312 **Conclusions**

313 The results of opto-structural, dielectric, chemical and thermoluminescence properties of γ -
314 irradiated vermiculite indicates that irradiation induces structural changes that lead to the
315 creation of new material with the enhanced properties. It is concluded from UV-Vis analysis that
316 the increase of optical band gap with gamma irradiation makes natural vermiculite mineral a
317 good candidate for efficient optoelectronic devices. A notable increase in the dielectric constant
318 of the vermiculite with gamma dose of 1000 kGy has been observed which improves its utility as
319 an electronic insulator. A negative slope in the Williamson-Hall plot indicates the presence of
320 compressive strain experienced by the particle of vermiculite. Gamma irradiation upto 1000 kGy
321 leads to the improvement in the crystallinity due to the increase in crystallite size, reduction in
322 the defects and the decrease in structural disorder and microstrain. The results at higher doses
323 reveal that the microstrain and dislocation density increases which leads to the production of
324 defects and structural disorder. The outcomes of FTIR analysis are in good agreement with the
325 UV-Vis, Dielectric and XRD results. From the TL studies, it is confirmed that natural
326 vermiculite is a good material for TL dosimeter for innovative dosimetry applications in
327 radiation rich environment. Thus we can conclude that the gamma irradiation can be utilized as a
328 tool for tailoring the various properties of vermiculite in the field of insulation and radiation rich
329 environment.

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Figure Captions

- Figure 1.** UV-Vis spectra of pristine and gamma irradiated vermiculite.
- Figure 2.** Dielectric constant (ϵ') versus frequency for pristine and gamma irradiated vermiculite.
- Figure 3.** $\text{Tan}\delta$ versus frequency for pristine and gamma irradiated vermiculite.
- Figure 4.** Dielectric loss (ϵ'') versus frequency of pristine and gamma vermiculite.
- Figure 5.** Frequency dependence of ac conductivity for pristine and gamma irradiated vermiculite.
- Figure 6.** X- ray diffraction pattern for pristine and gamma irradiated vermiculite.
- Figure 7.** Williamson Hall plots for the pristine and gamma irradiated vermiculite.
- Figure 8.** FTIR spectra of pristine and gamma irradiated vermiculite.
- Figure 9.** TL glow curves of vermiculite irradiated with gamma rays at different doses.
- Figure 10.** TL glow peak intensity vs gamma dose for vermiculite.

424 **Table 1.** EDS elemental composition analysis of natural vermiculite

425

Element	Weight%	Atomic%
Si	24.45	19.03
Al	10.69	8.67
Mg	10.19	9.16
Fe	10.51	4.12
Ca	1.61	0.88
O	42.55	58.15
Total	100.00	100.00

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428 **Table 2.** The variation of optical band gap energy, Urbach's energy and structural disorder with
429 gamma dose in the pristine and gamma irradiated vermiculite

Gamma Dose (kGy)	*Band Gap Energy (eV)		*Urbach 's energy (eV)	Structural Disorder (X)
	Indirect	Direct		
Pristine	1.04	2.02	1.28	33.57
10	1.19	2.23	1.24	32.48
100	1.32	2.33	1.18	30.86
500	1.58	2.57	1.04	27.06
1000	1.68	2.71	0.98	25.44
1500	0.78	1.83	1.32	34.65
2000	0.71	1.78	1.35	35.46

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*Standard deviation is ± 0.01

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434 **Table 3.** The variation of dielectric parameters with gamma dose in the pristine and Gamma
 435 irradiated vermiculite.

Gamma Dose (kGy)	ϵ'	ϵ''	Tan δ (1×10^{-1})	σ_{ac} (1×10^{-10}) ($\Omega^{-1}m^{-1}$)	M' (1×10^{-1})	M'' (1×10^{-4})	Z' (Ω) (1×10^{-3})	Z'' (Ω) (1×10^{-5})
Pristine	21.96	0.44	0.20	9.78	0.46	9.11	2.92	2.72
10	22.04	0.38	0.17	8.48	0.45	7.85	3.50	2.77
100	22.90	0.34	0.15	7.58	0.44	6.49	6.05	2.86
500	24.05	0.20	0.08	4.49	0.42	3.49	13.89	3.00
1000	24.72	0.13	0.05	2.94	0.40	2.17	60.39	3.11
1500	23.69	0.27	0.11	5.93	0.42	4.75	8.54	2.94
2000	22.25	0.37	0.17	8.30	0.45	7.54	4.46	2.79

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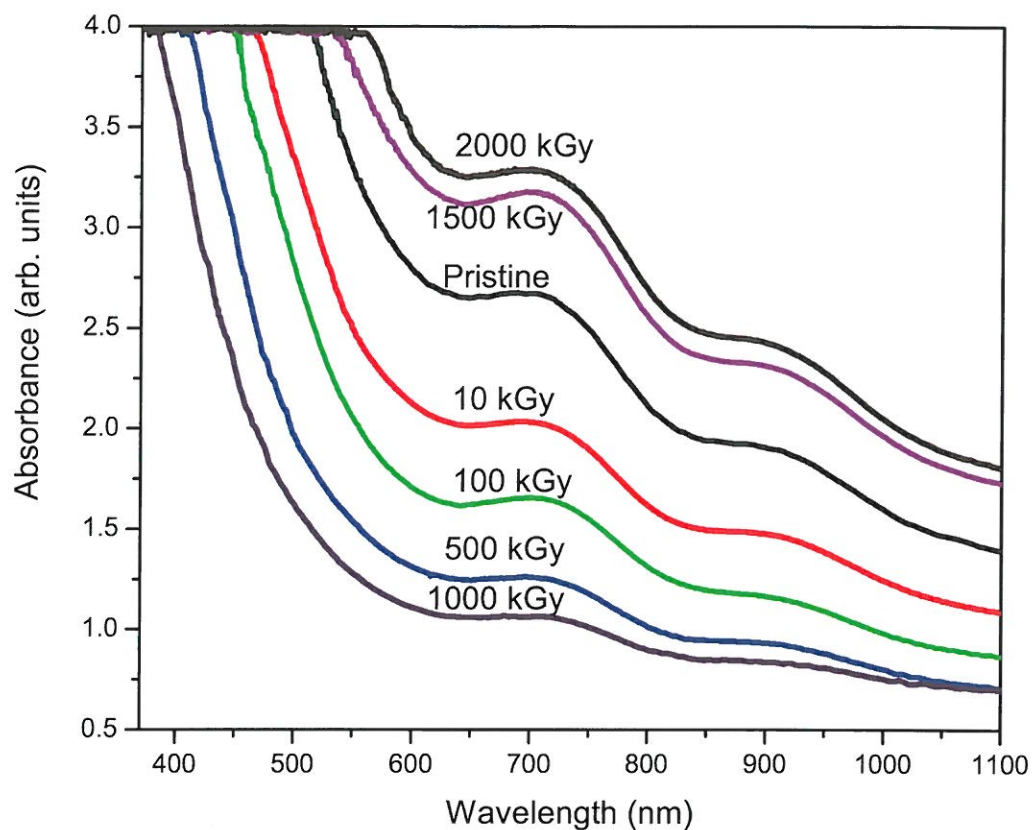
438 **Table 4.** Results of Williamson –Hall plots

Gamma Dose (kGy)	Peak Width (β) (10^{-2}) ($2\theta = 8.87^{\circ}$)	*Average Domain Size (D) nm	Microstrain (S) (10^{-4})	Dislocation Density (δ) (10^{-5})
Pristine	13.942	63.25	7.79	3.48
10	9.768	72.88	6.10	2.36
100	7.632	88.96	3.40	1.08
500	6.008	109.05	2.37	0.61
1000	5.011	132.62	1.79	0.38
1500	7.034	75.99	3.19	1.18
2000	10.288	55.58	8.77	4.46

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*Standard deviation is ± 0.05

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441

442 **Figure 1.** UV-Vis spectra of pristine and gamma irradiated vermiculite.

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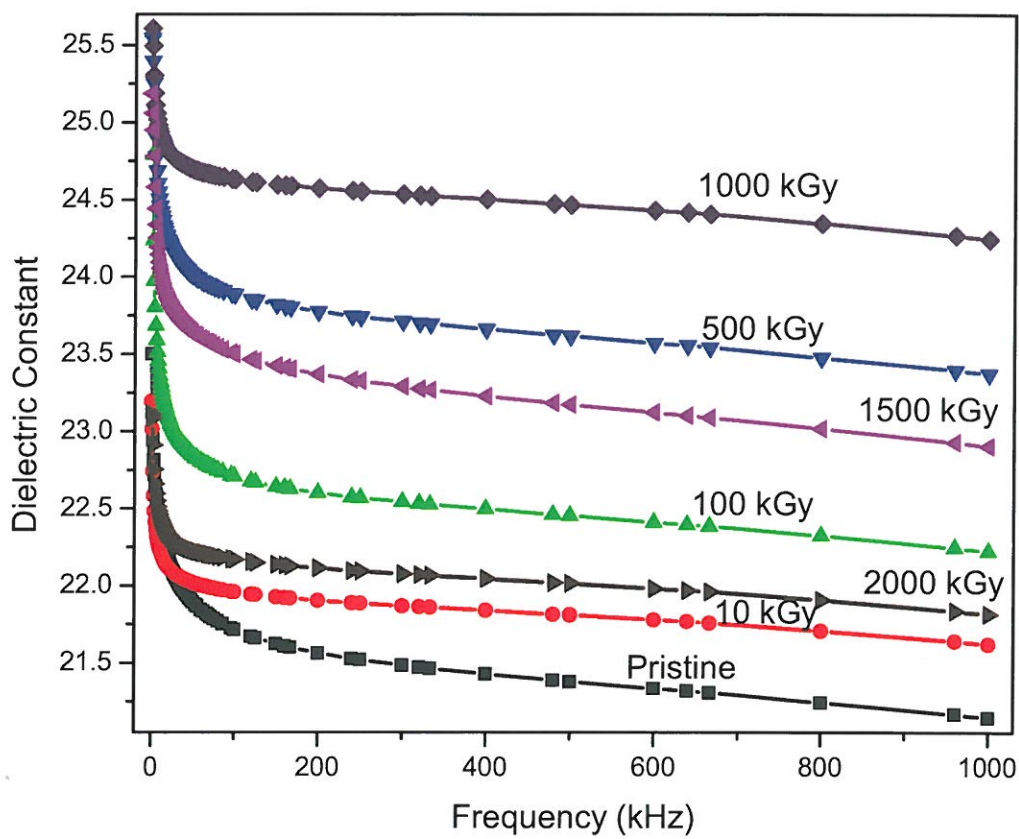
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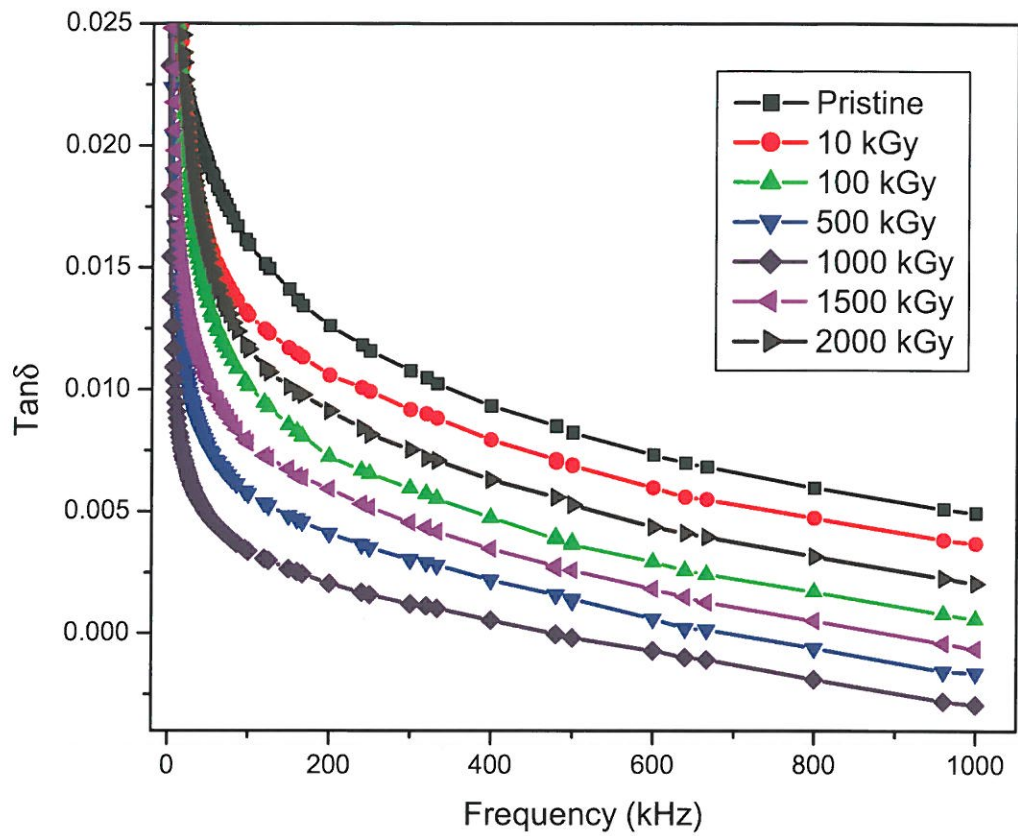
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450 **Figure 2.** Dielectric constant (ϵ') versus frequency for pristine and gamma irradiated
451 vermiculite.

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454 **Figure 3.** $\text{Tan}\delta$ versus frequency for pristine and gamma irradiated vermiculite.

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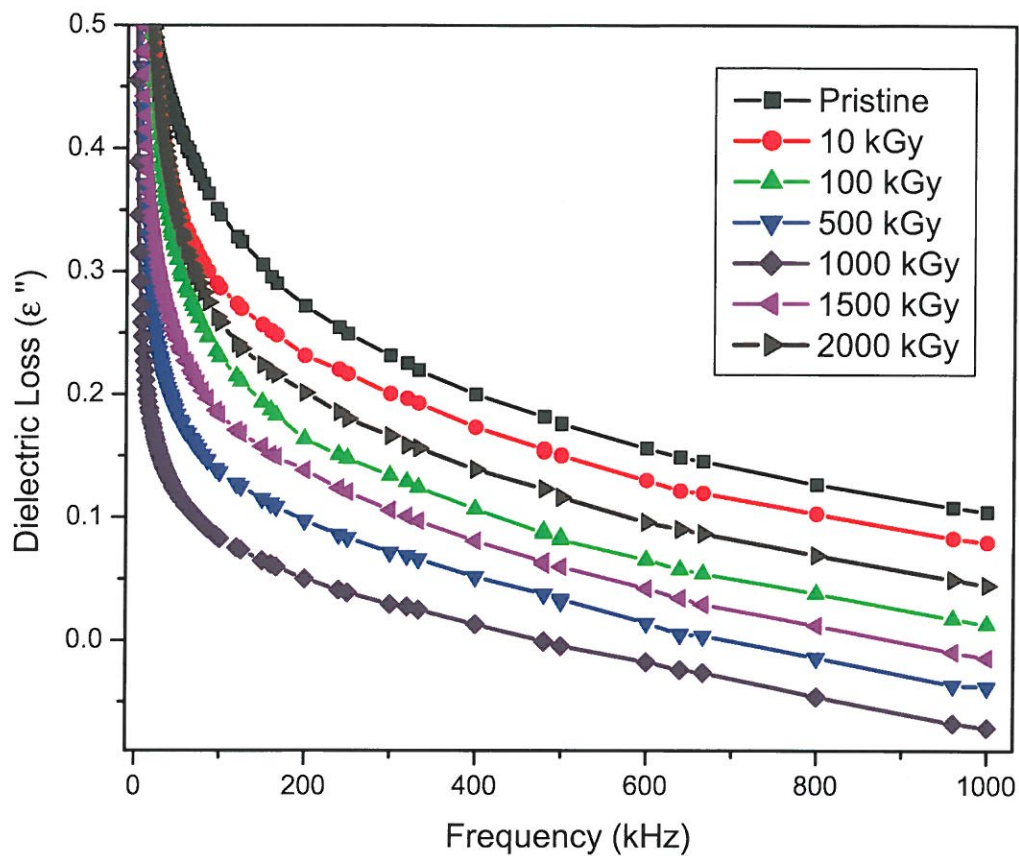
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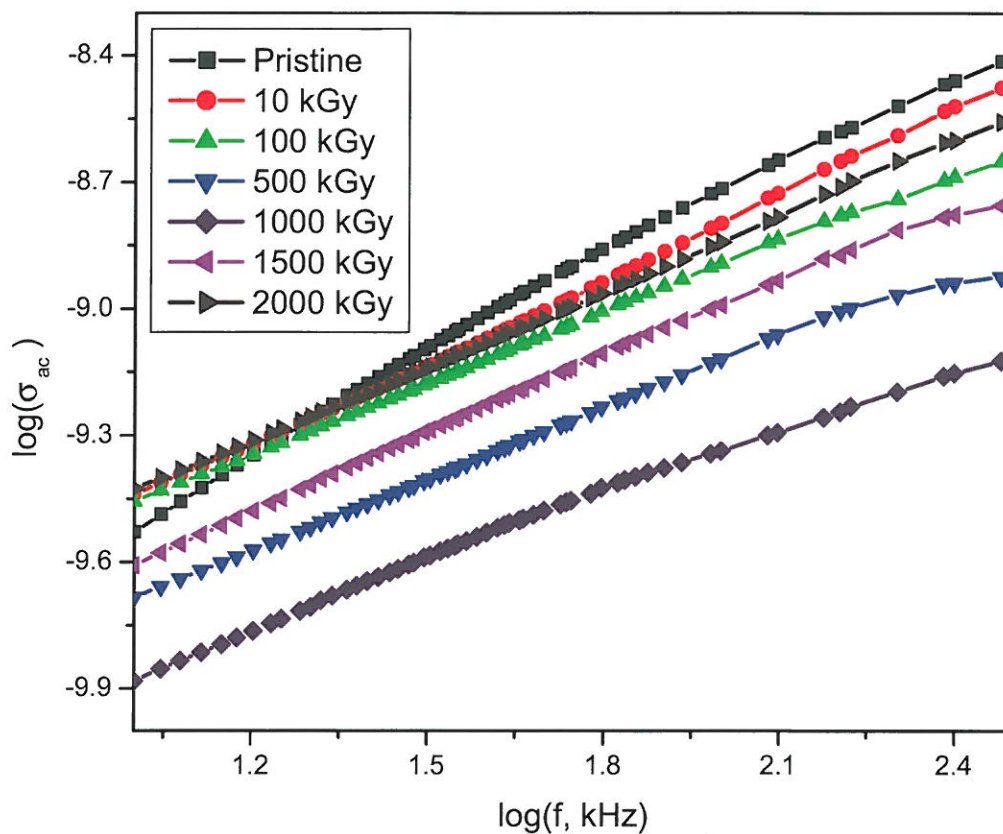
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463 **Figure 4.** Dielectric loss (ϵ'') versus frequency of pristine and gamma vermiculite.

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466 **Figure 5.** Frequency dependence of ac conductivity for pristine and gamma irradiated
467 vermiculite.

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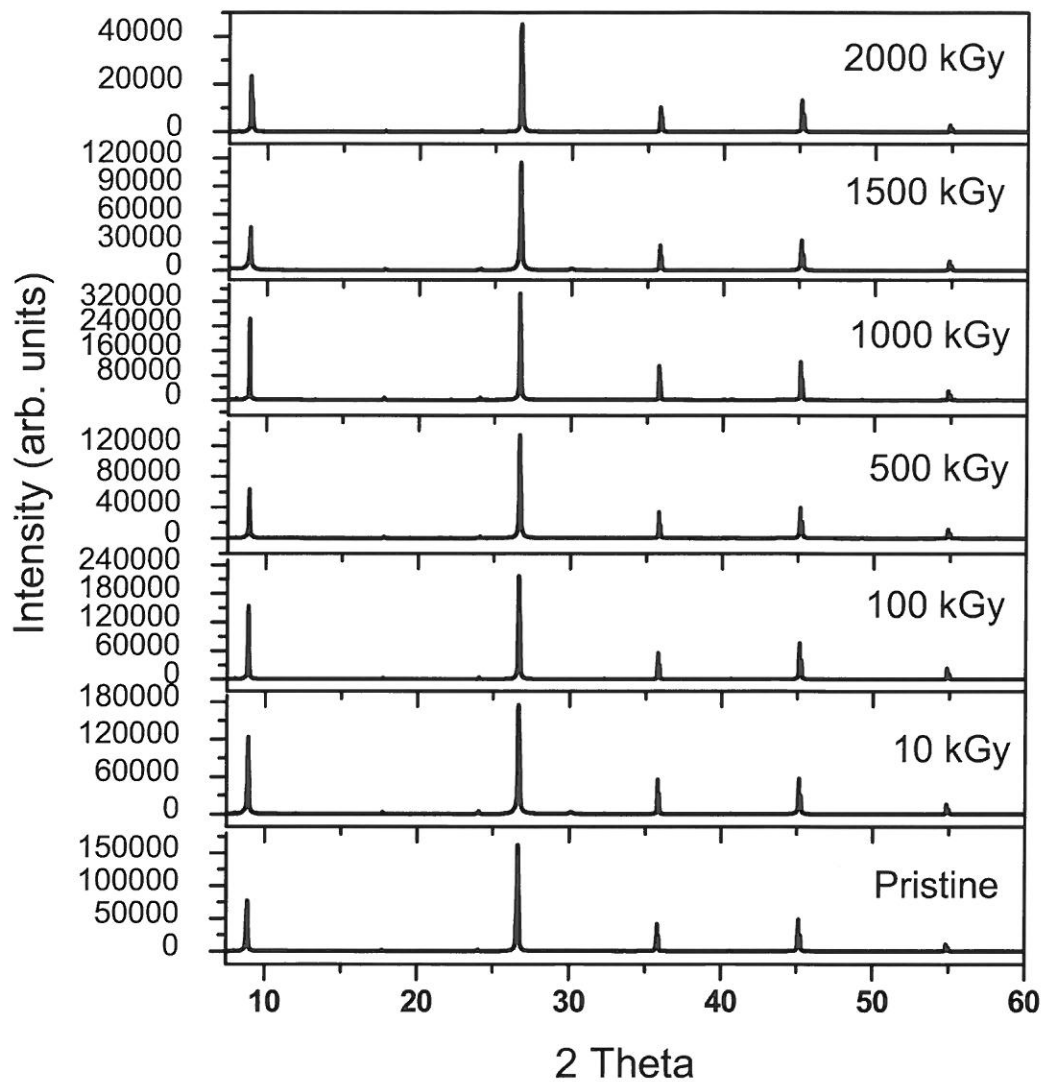
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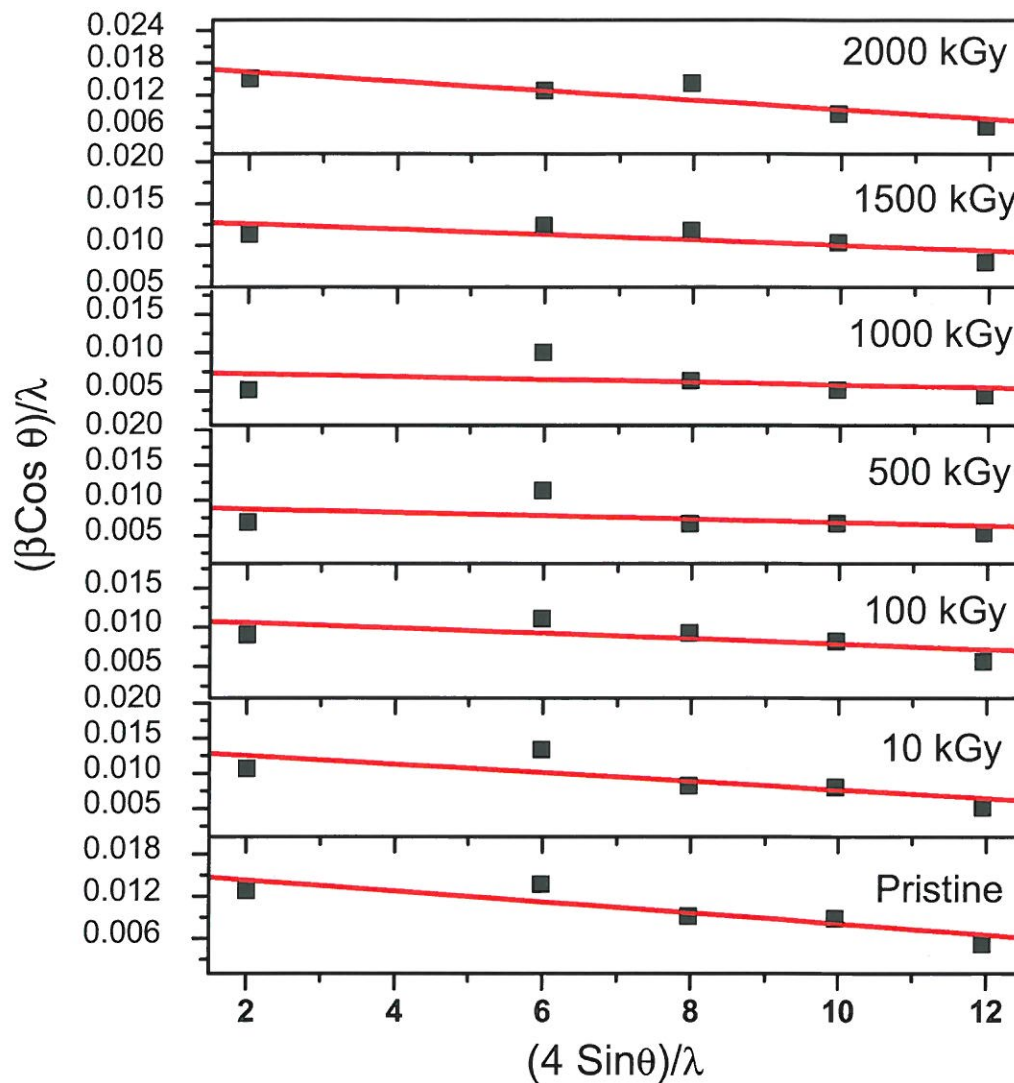
478 **Figure 6a.** X- ray diffraction pattern for pristine and gamma irradiated vermiculite.

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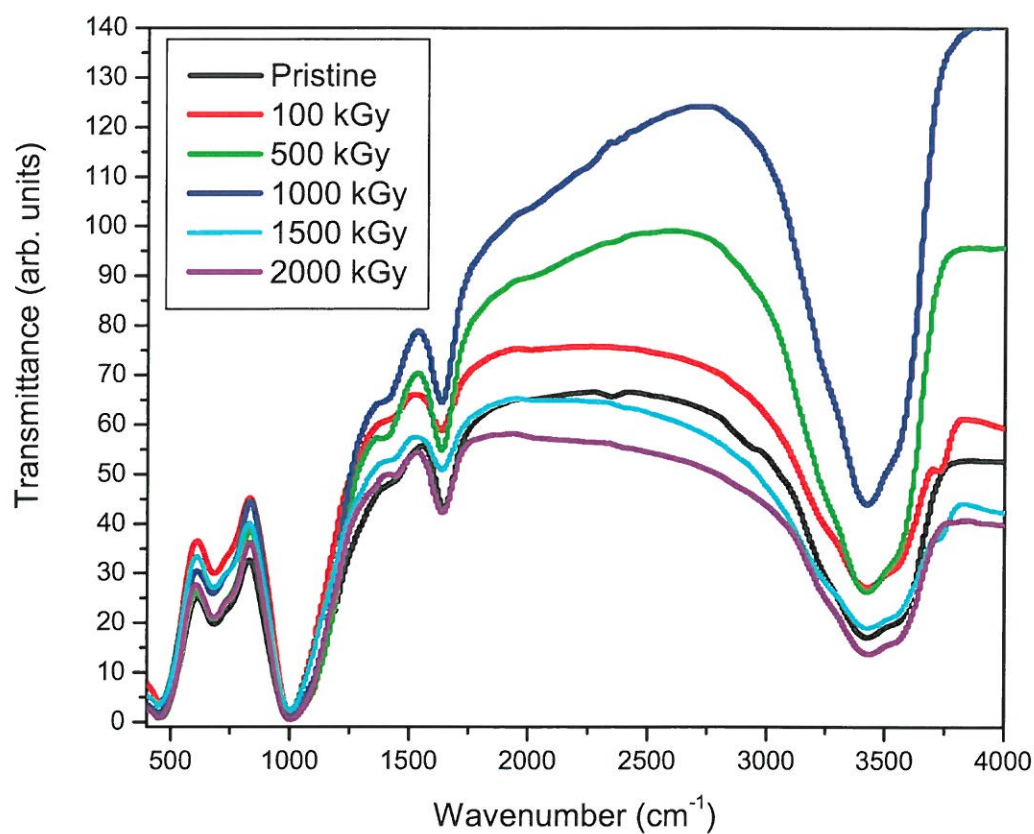
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484 **Figure 7.** Williamson Hall plots for the pristine and gamma irradiated vermiculite.

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489 **Figure 8.** FTIR spectra of pristine and gamma irradiated vermiculite.

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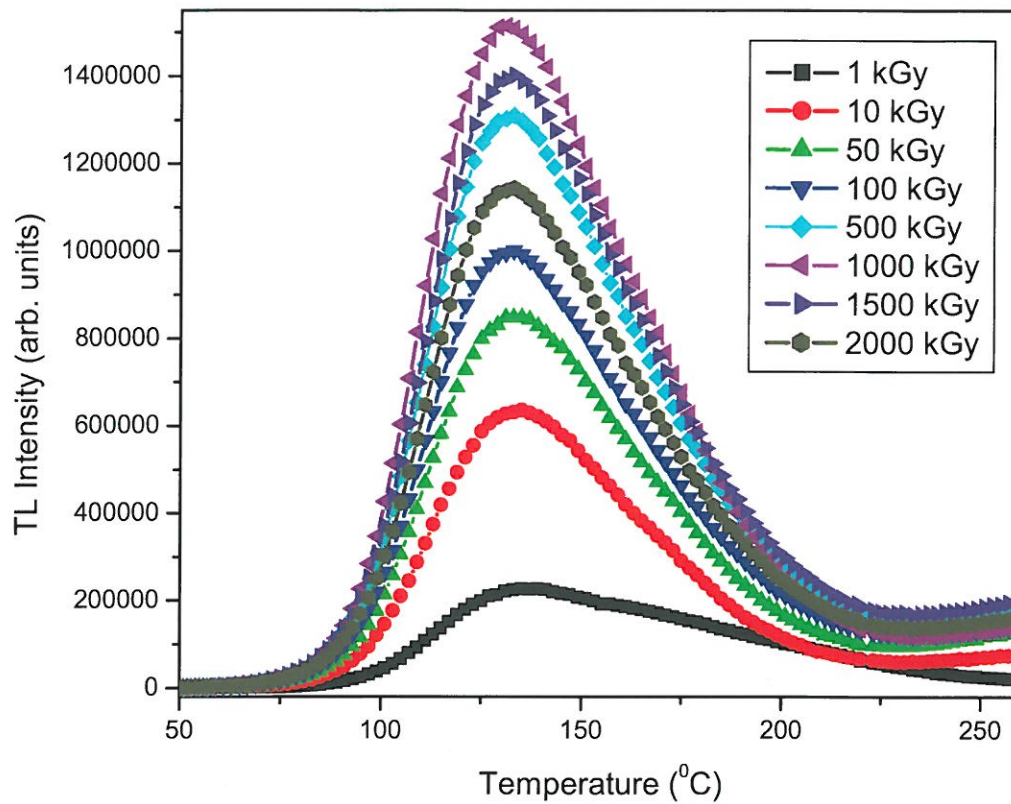
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499 **Figure 9.** TL glow curves of vermiculite irradiated with gamma rays at different doses.

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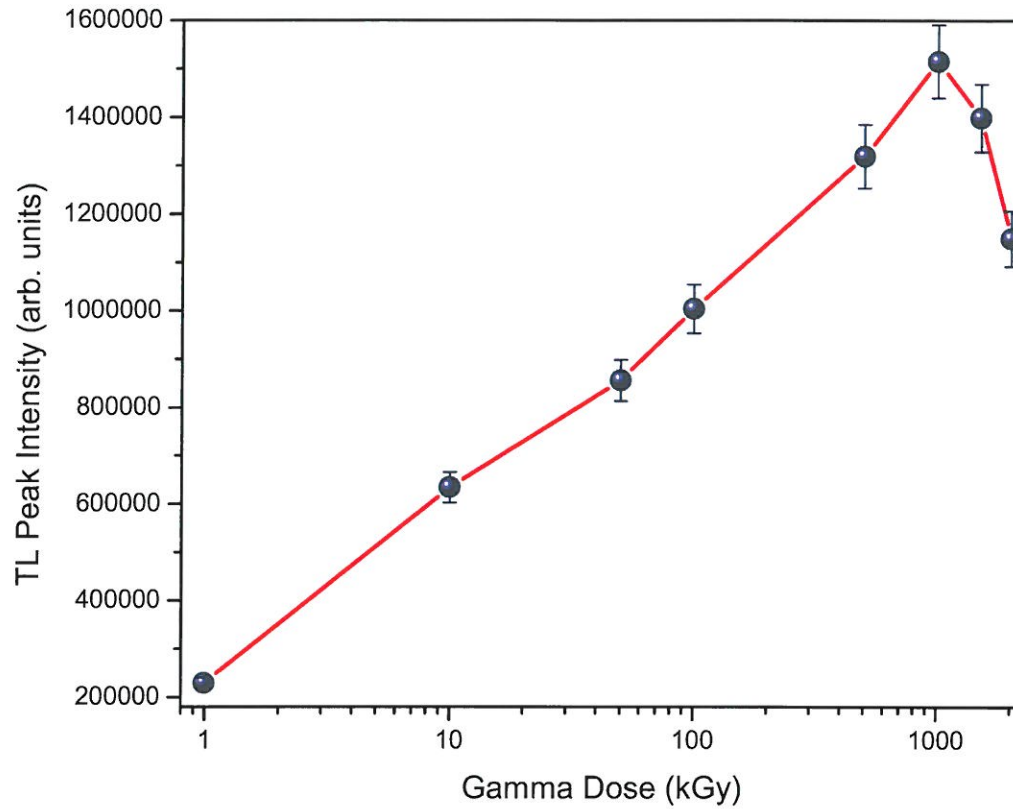
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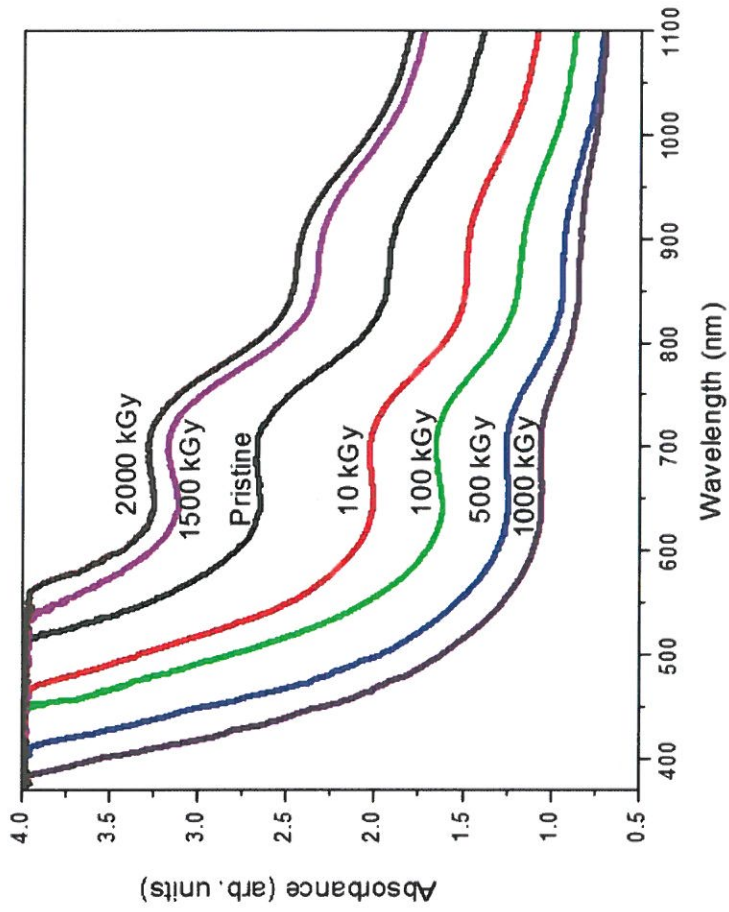


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510 **Figure 10.** TL glow peak intensity vs gamma dose for vermiculite.

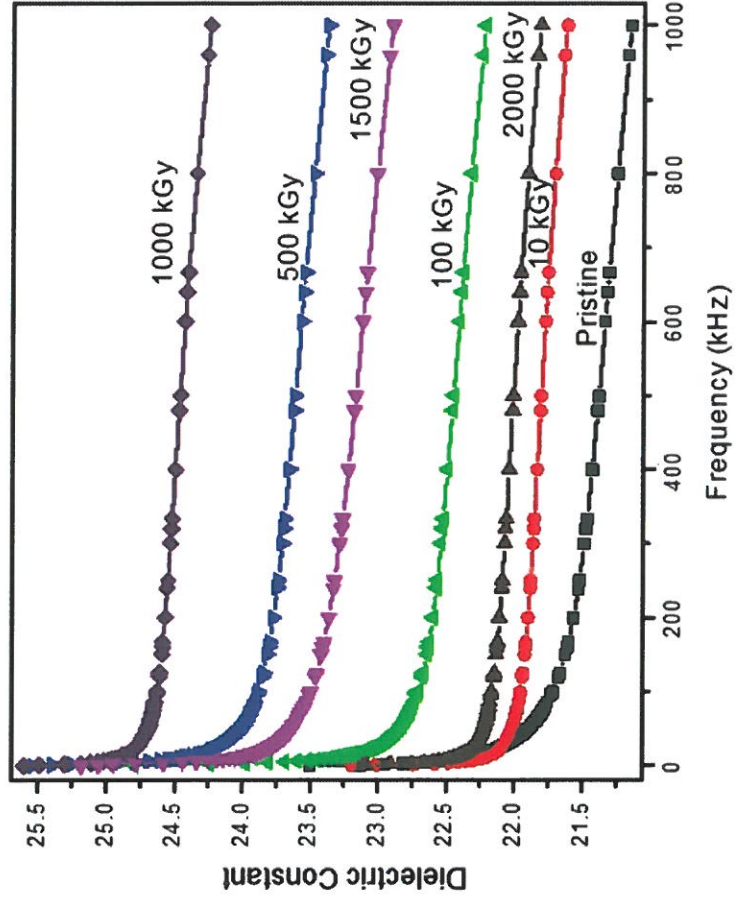
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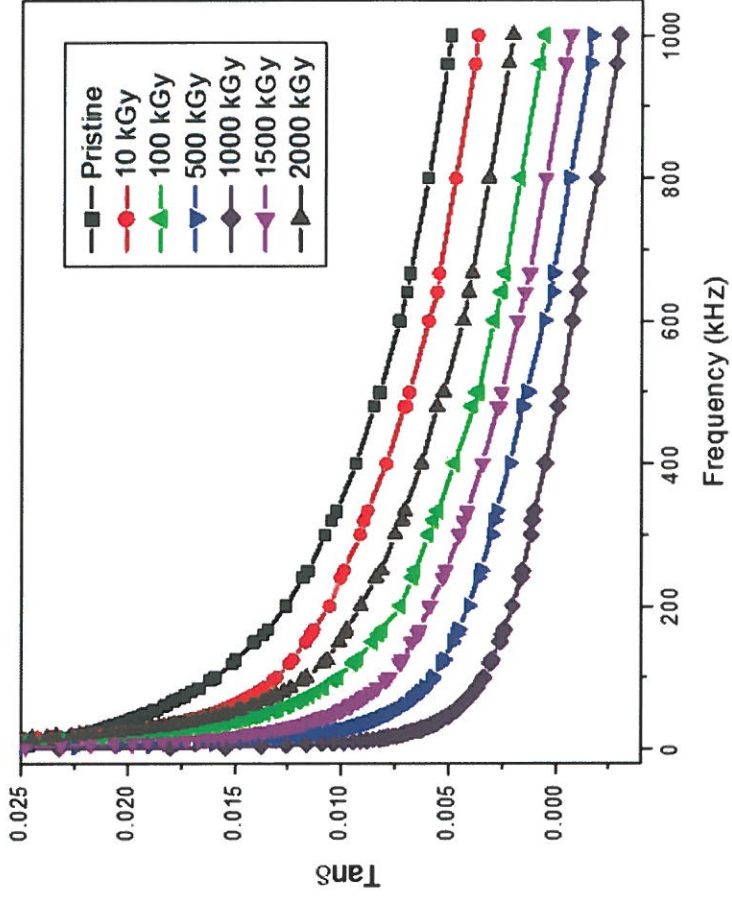
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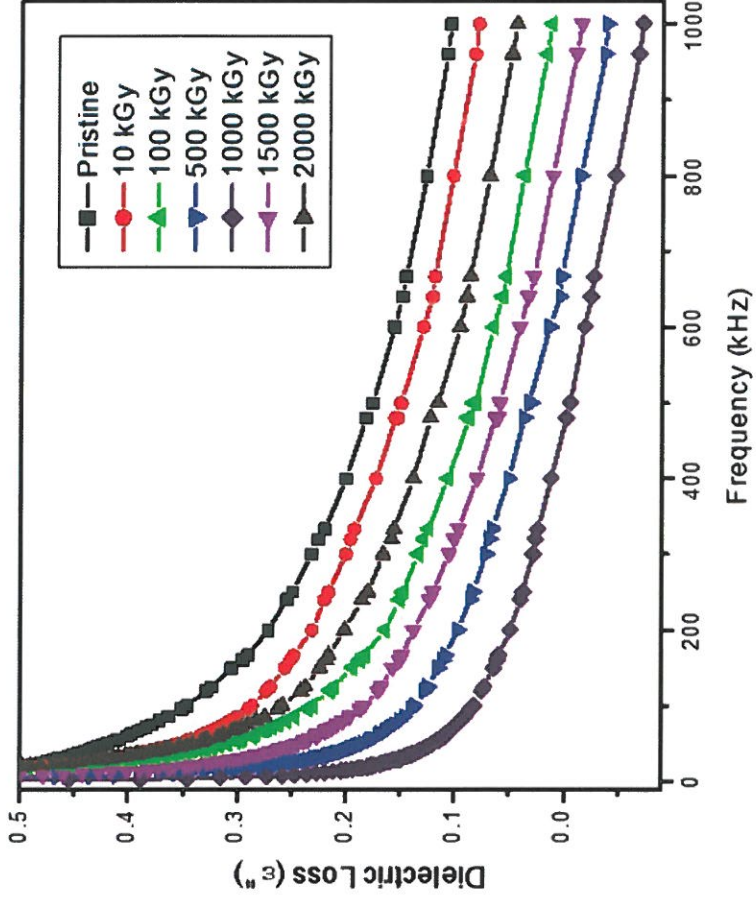
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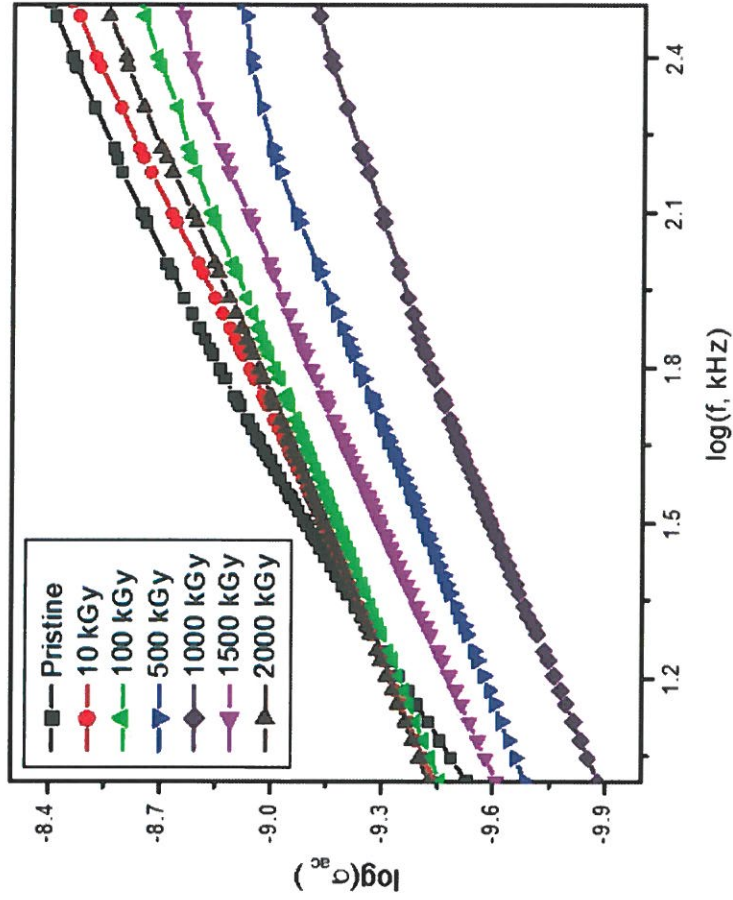
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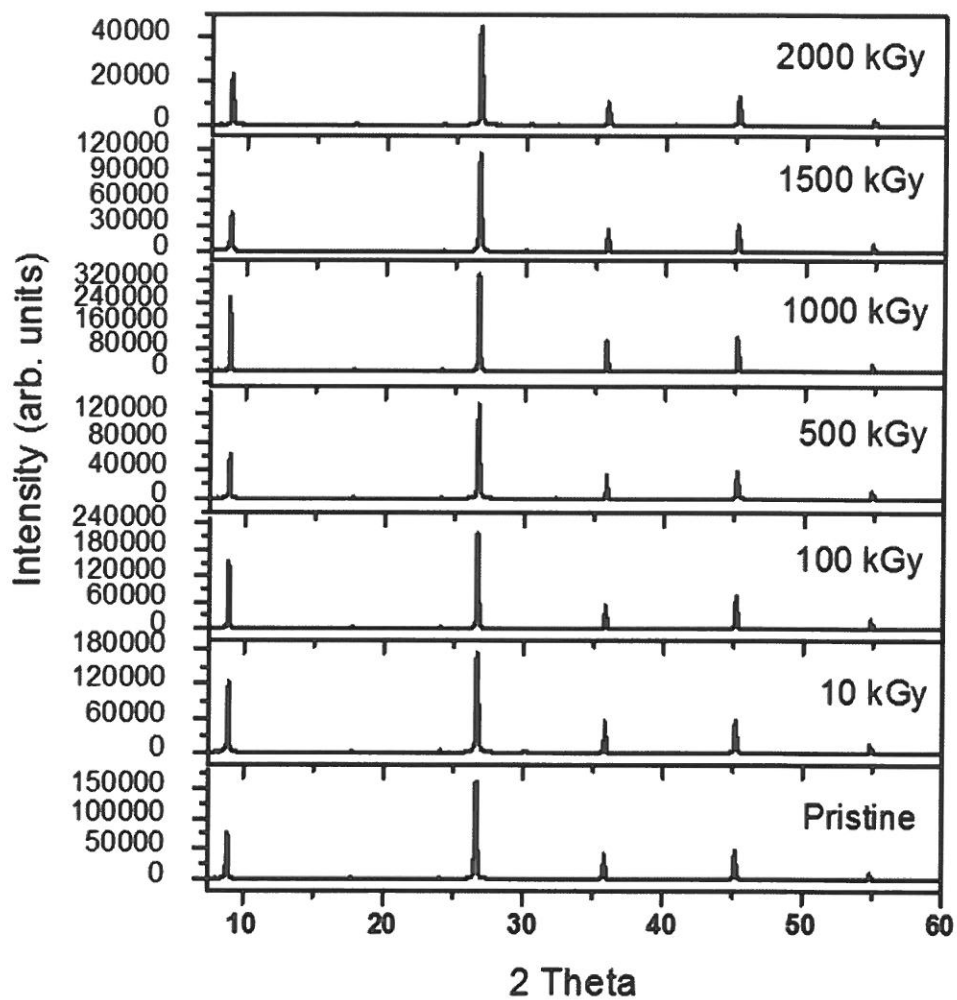
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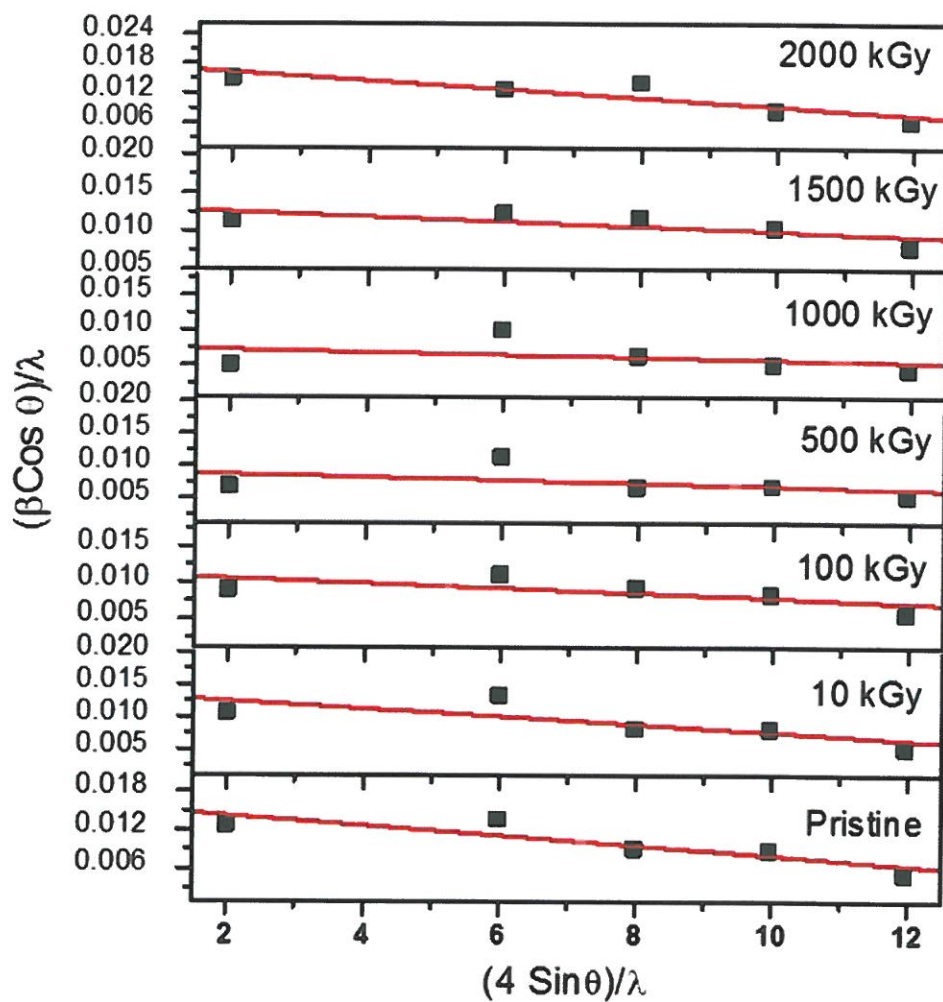
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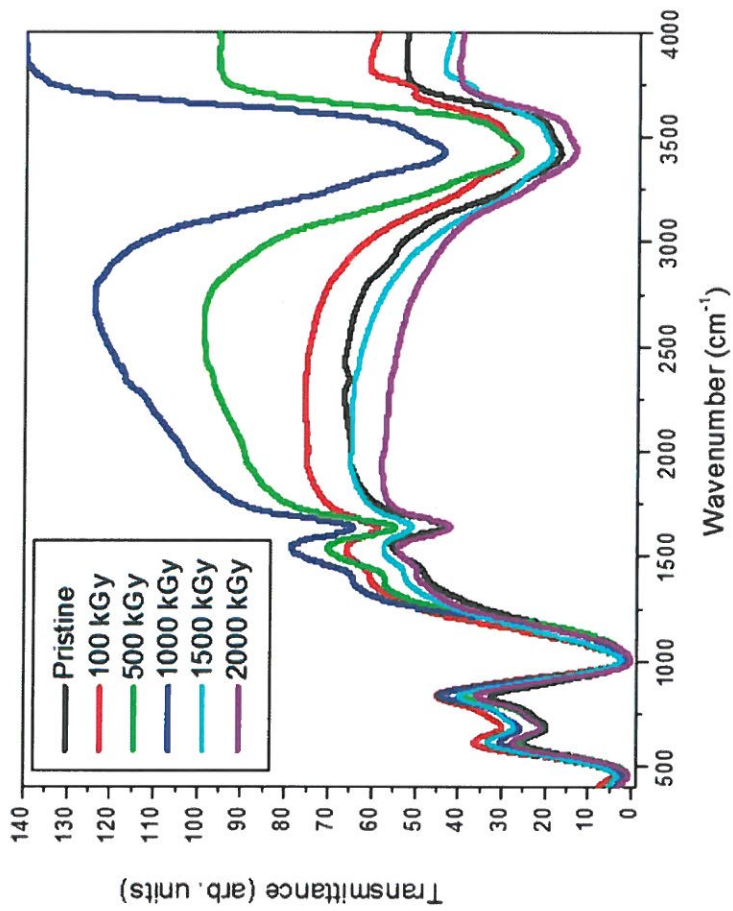
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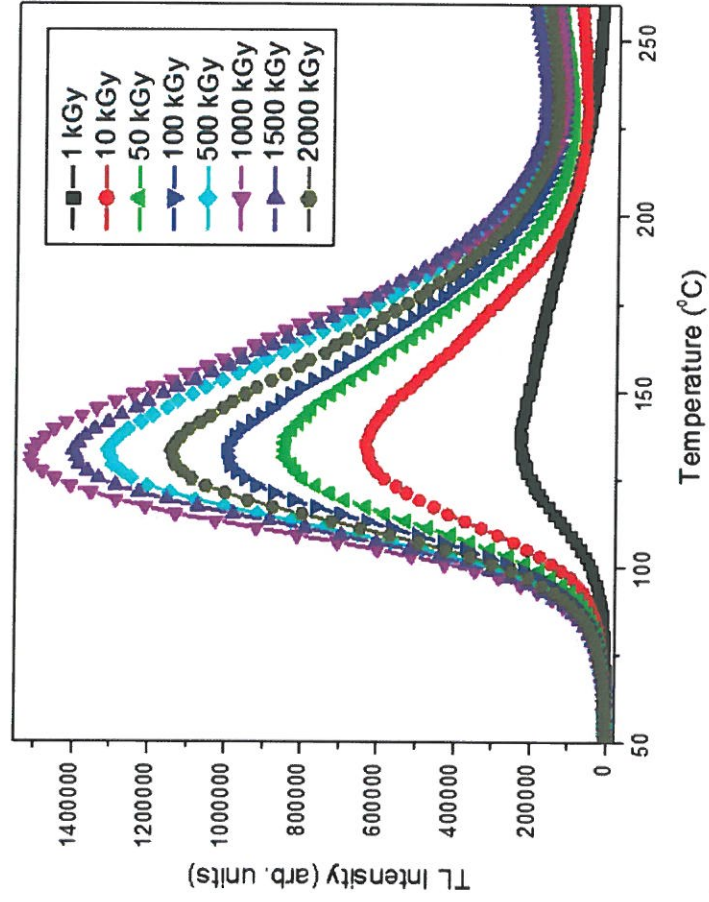
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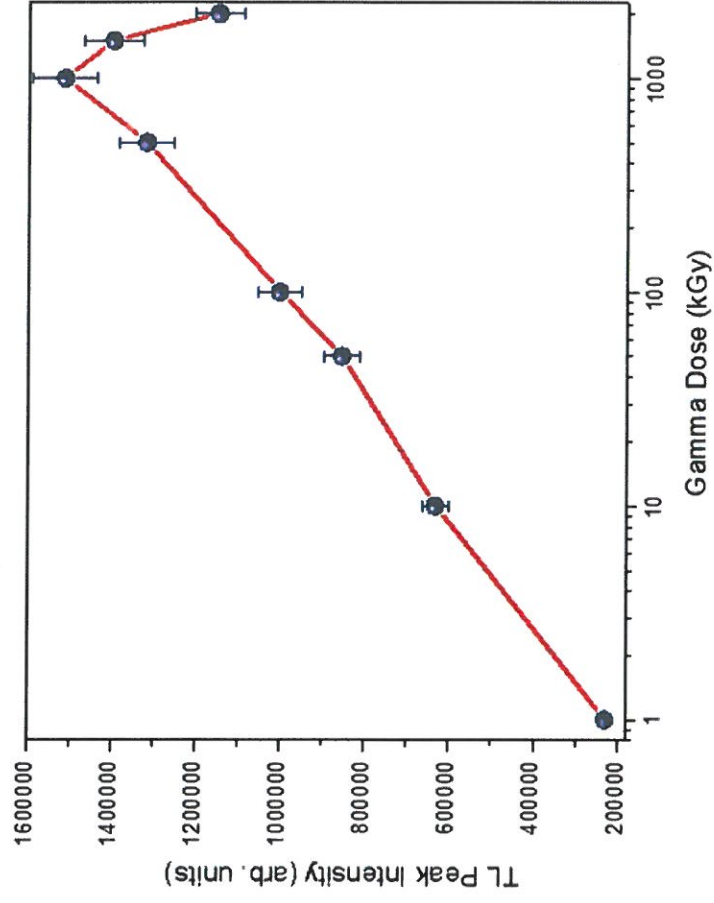
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