

1 Revision 2

2 **Garnet as a major carrier of the Y and REE in the granitic rocks: An example from the**
3 **layered anorogenic granite in the Brno Batholith, Czech Republic**

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ABSTRACT

18 Garnet and other rock-forming minerals from A-type granite dykes in the Pre-Variscan Brno
19 Batholith were analysed to determine relative contributions of individual minerals to whole rock
20 Y and REE budget and to assess incorporation mechanisms of these elements in garnet. Minor to
21 accessory garnet (< 2 vol%) is the essential reservoir for Y+REE in the Hlína granite accounting
22 ~84 % Y and 61 % REE of the total whole rock budget. Zircon is another important carrier of

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23 REE with ~13 % Y and ~11 % REE. At least ~21 % REE and 1 % Y were probably hosted by
24 Th- and U-rich monazite that has been completely altered to a mixture of secondary REE-bearing
25 phases. The contribution of major rock-forming minerals (quartz and feldspars) is low (~1 % Y;
26 10 % LREE; ~1 % HREE) excluding Eu, which is hosted predominantly by feldspars (~90 %).
27 Minor to accessory muscovite and magnetite incorporate ~1 % Y and ~2 % REE of the whole
28 rock budget. Magmatic garnet $\text{Sps}_{41-46}\text{Alm}_{28-44}\text{And}_{0-13}\text{Grs}_{6-12}\text{Prp}_{0-1}$ is Y- and HREE-rich (up
29 1.54 wt% Y; up ~1 wt% ΣREE), and the Y+REE enter the garnet structure via the menzerite-(Y)
30 substitution. The Y and REE show complex zoning patterns and represent sensitive indicator of
31 garnet evolution, in contrast to a rather homogeneous distribution of major divalent cations.
32 General outward decrease of Y+REE is a common feature due to the strong partitioning of
33 Y+HREE in the garnet relative to the other phases. REE underwent significant fractionation
34 during growth of early garnet I; the Yb_N/Nd_N ratio generally decreases from the core to rim of
35 garnet I. Higher Mn and Al, lower Ca, and Y+REE contents, as well as higher Yb_N/Nd_N ratio and
36 more negative Eu anomaly in garnet II overgrowths indicate its crystallization from a more
37 evolved melt. Application of zircon saturation geothermometry provides upper temperature limit
38 of 734 ± 14 °C for the closed-system crystallization. Mineral equilibria reveal that crystallization
39 started at QFM + 1.2, and preferential sequestration of Fe^{3+} into garnet and magnetite was
40 responsible for progressively reducing conditions. Equilibrium between magnetite, garnet, quartz
41 and plagioclase, representing the final crystallization stage of the granitic magma, occurred at
42 658-663 °C and QFM 0 to + 0.8, hence at undercooling of ~75 °C.

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44 **Key words:** garnet; EMP; LA-ICP-MS; Y+REE mass-balance calculations; A-type granite; Brno
45 Batholith

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INTRODUCTION

48 Magmatic garnet (almandine-spessartine) occurs mainly in granitic pegmatites, aplites and felsic
49 peraluminous S-type granitoids (e.g. Allan and Clarke 1981; Miller and Stoddard 1981; Černý
50 and Hawthorne 1982; Deer et al. 1982; Du Bray 1988; Harrison 1988; Dahlquist et al. 2007;
51 London 2008; Sheibi et al. 2010; Müller et al. 2012); it is rare in A-type (Zhang et al. 2009, 2012)
52 or I-type (Wu et al. 2004) granites. Garnet is a useful indicator of changing magmatic conditions
53 during its crystallization and peritectic reactions (e.g. Dahlquist 2001; Wang et al. 2003a;
54 Dahlquist et al. 2007; Erdmann et al. 2009; Lackey et al. 2012). It has also been identified as an
55 important carrier of Y+HREE (Jaffe 1951; Dudykina 1959; Vorma et al. 1966; Wakita et al.
56 1969; Grew et al. 2010; Marsh et al. 2012) with up to ~3 wt% Y+HREE in magmatic garnets
57 (Wang et al. 2003a), and with up to ~13 wt% Y in metamorphic garnets (Grew et al. 2010; Marsh
58 et al. 2012). The compositional zoning of Y+REE as a sensitive indicator of garnet growth and
59 crystallization history has been recently discussed particularly in metamorphic rocks (e.g.
60 Hickmott and Shimizu 1990; Lanzirotti 1995; de Lima et al. 1995; Spear and Kohn 1996; Bea et
61 al. 1997; Holten et al. 1997; Kotková and Harley 1999; Pyle and Spear 1999; Hermann and
62 Rubatto 2003; Kohn 2004; Čopjaková et al. 2005; Borghi et al. 2006; Buick et al. 2006; Røhr et
63 al. 2007; Spry et al. 2007; Marsh et al. 2012). By contrast, there are only a few studies dealing
64 with Y+REE zoning texture in magmatic garnets (Sevigny 1993; Whitworth and Freely 1994;
65 Wang et al. 2003a; Dorais and Tubrett 2012; Müller et al. 2012) and evolution of
66 Y+HREE/LREE ratios during garnet growth (Sevigny 1993; Dorais and Tubrett 2012).
67 We describe complexly zoned garnet with Y+REE-enrichment from microgranite portions of
68 subaluminous A-type Hlína granite in order to better understand: (i) the distribution of Y and

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69 REE in garnet, (ii) the substitution mechanisms of these elements, and (iii) the role of garnet in
70 controlling Y+REE content in this magmatic rock in particular. Moreover, contributions of the
71 individual minerals to the Y+REE budget in the granite were calculated and discussed.

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GEOLOGICAL SETTING

74 The large Cadomian (Pan-African, 650-550 Ma) Brunovistulian microcontinent is a portion of the
75 Precambrian lithosphere that is located at the eastern margin of the Bohemian Massif and
76 occupies parts of Czech, Austrian and Polish territories (Fig. 1). It consists of three major N-S
77 trending tectonic units of distinct character: (i) an ophiolite belt that crops out in the central part
78 and it divides the Brunovistulicum into (ii) the Slavkov terrane in the NE and (iii) the Thaya
79 terrane in the SW (Finger et al. 2000). The most critical and best exposed portion of the
80 Precambrian basement in central Europe is the Brno Batholith (Finger et al. 2000; Leichmann and
81 Höck 2008). Steep geological, tectonic and geophysical boundaries divide the Brno Batholith into
82 three different N-S oriented units mirroring the zoned structure of the whole Brunovistulicum
83 (Fig. 1). These units are (from west to east): the Western Granitoid Complex (WGC, as part of
84 the Thaya Terrane), the Central Ophiolite Belt (COB) and the Eastern Granitoid Complex (EGC,
85 as part of the Slavkov Terrane). The Central Ophiolite Belt (COB; dated at 725 ± 15 Ma by
86 Finger et al. 2000) originated in supra-subduction environment (Finger et al. 2000; Kalvoda et al.
87 2008; Leichmann and Höck 2008), but was subsequently intruded by younger (600-580 Ma)
88 granitoids (WGC and EGC).

89 Three granitic suites were distinguished within the WGC by Leichmann and Höck (2008): (i) the
90 Tetčice (S-type) suite, (ii) the Réna (I-type) suite, both represent evolved rocks of an active
91 continental margin, and (iii) the examined Hlína (A-type) suite (Fig. 1). The Tetčice suite consists

92 of biotite granite with frequent xenoliths of biotite-amphibole diorites and of metasedimentary
93 rocks. Zircon morphology and alumina saturation index (ASI) = 1.14-1.20 suggest S-type granite
94 affinity (Leichmann and Höck 2008). The Réna suite consists of felsic ($\text{SiO}_2 > 69$ wt%, ASI <
95 1.1), I-type amphibole–biotite granodiorites to biotite granites interpreted as volcanic-arc
96 products (Leichmann and Höck 2008). Numerous small (tens to hundred meters) and irregular
97 bodies and dykes of the Hlína suite intruded I-type granites of the Réna suite and, less frequently,
98 S-type granites of the Tetčice suite (Fig. 1). Recent mapping (Hönig et al. 2012) confirmed their
99 frequent occurrence in the SW part of the Brno Batholith and within the NE part of the Thaya
100 Batholith (Fig. 1), where the Hlína suite spreads out more than 80 km in north-south direction.
101 The strike of the bodies is generally parallel to the NW-SE trending regional fault zone. Both
102 hanging wall and foot wall contacts of the Hlína granite bodies with the host granodiorite are
103 sharp. The Hlína suite consists of felsic, garnet-bearing leucogranite with anorogenic, post-
104 collisional affinity, and often showing complex internal structure (Hönig et al. 2010).

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ANALYTICAL METHODS

107 **Sampling and modal counting**

108 Samples of the A-type Hlína granite for whole-rock chemical analysis and EMP study were
109 collected at 4 localities (A to D in Fig. 1). The detailed EMP and LA-ICP-MS study combined
110 with Y+REE mass balance calculations were performed using the samples of microgranitic units
111 from the granite dyke at the locality A (Moravské Bránice; Fig. 1). Twenty thin sections from this
112 locality were studied in detail using conventional petrographic methods including modal
113 counting.

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115 **EMP**

116 Garnet and other rock-forming and accessory minerals from the A-type Hlína granite were

117 analysed with a Cameca SX100 electron microprobe at the Joint Laboratory of Electron

118 Microscopy and Microanalysis, Department of Geological Sciences, Masaryk University, Brno

119 and Czech Geological Survey. Operating conditions for analyses were as follows: wavelength

120 dispersive mode, an accelerating voltage of 15 kV, a beam current of 40 nA for garnet, 20 nA for

121 magnetite and other accessory minerals and 10 nA for muscovite and feldspars; a beam diameter

122 of 2 μm for garnet and accessory minerals, and 5 μm for muscovite and feldspars. The Y+REE in

123 the zircon were analyzed at 150 nA. The P and Sr in feldspars were analyzed at 100nA. The

124 element concentrations in silicate minerals were analysed using the following standards and X-

125 ray lines: K_{α} lines - sanidine (Si, Al, K), albite (Na), pyrope (Mg), andradite (Ca, Fe), spessartine

126 (Mn), titanite (Ti), chromite (Cr), fluorapatite (P), topaz (F), L_{α} lines - YAG (Y), Yb (YbPO₄), Er

127 (ErPO₄), SrSO₄ (Sr), baryt (Ba). For magnetite the following standards and X-ray lines were

128 used: K_{α} lines - MgAl₂O₄ (Mg, Al), hematite (Fe), chromite (Cr), titanite (Ca, Ti), Ni₂SiO₄ (Ni),

129 ScVO₄ (V), sanidine (Si), Mn₂SiO₄ (Mn); L_{α} line - gahnite (Zn). The peak counting time was 10

130 seconds for major elements and 20-80 seconds for minor to trace elements. Average detection

131 limits in garnet are: for Y (340 ppm), Er and Yb (970 and 790 ppm, respectively). The calculated

132 relative errors (2 σ error) are about <3 % at the >10 wt% level, <5 % at the ~5 wt% level and <15

133 % at the ~1 wt% level. Analytical conditions and standards used for zircon and REE-bearing

134 phases were similar to those described by Breiter et al. (2009) and Čopjaková et al. (2011).

135 Chemical analyses of garnet were recalculated to the formula $^{XII}X_3^{VI}Y_2^{IV}T_3O_{12}$ on the basis of Σ

136 cations = 8 and the Fe³⁺/Fe²⁺ ratios were calculated by charge balance.

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138 **LA-ICP-MS**

139 Trace element contents in major (plagioclase, K-feldspar, quartz) and minor to accessory
140 minerals with a grain-size greater than 25 μm (garnet, magnetite, muscovite, epidote) were
141 investigated by LA-ICP-MS (CEITEC and Department of Chemistry, Faculty of Science,
142 Masaryk University, Brno) using an Agilent 7500ce (Agilent Technologies) quadrupole ICP-MS
143 with an attached UP 213 laser ablation system (New Wave Research). The samples were ablated
144 using a commercial Q-switched Nd:YAG laser operated at a wavelength of 213 nm (pulse
145 duration of 4.2 ns). Ablated material was transported from the sample chamber using helium
146 carrier gas (1 L/min) and mixed with argon (0.6 L/min) prior to the torch. Optimization of LA-
147 ICP-MS parameters (gas flow rates, sampling depth, voltage of ion optics) was performed using
148 glass reference material NIST SRM 612 to maximize the S/N ratio and minimum oxide formation
149 (ThO^+/Th^+ counts ratio 0.2 %) and U^+/Th^+ counts ratio 1.1 %. Potential polyatomic interferences
150 were minimized by collision reaction cell in He mode (2.5 mL/min).

151 The contents of major and trace elements in garnet, magnetite, muscovite, epidote, plagioclase
152 and K-feldspar were determined after laser ablation of individual spots with various diameter
153 (garnet - 55 μm ; magnetite - 80 μm ; muscovite - 40 μm ; epidote - 25 μm ; plagioclase and K-
154 feldspar - 110 μm) and fluence of laser beam (garnet, magnetite and plagioclase - 5 $\text{J}\cdot\text{cm}^{-2}$;
155 muscovite, epidote and K-feldspar - 3 $\text{J}\cdot\text{cm}^{-2}$) depending on the size of analyzed grains. Analyses
156 of mineral samples were carried out at frequency of 10 Hz and each spot was ablated for 60 s.
157 The element content in quartz sample was traced by line scanning mode at the diameter of 65 μm
158 and 5 $\mu\text{m}\cdot\text{s}^{-1}$ scan speed. The fluence of laser beam was set to 20 $\text{J}\cdot\text{cm}^{-2}$ and the analysis was
159 carried out at frequency of 5 Hz. The contents of elements of interest were estimated using SRM

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160 NIST 610 and 612 and internal reference element (Si – silicates, Fe – magnetite). The calculation
161 was performed after baseline correction and integration of peak area.

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163 **Whole rock chemical analyses**

164 Whole rock chemical analyses (8 samples) were determined at the Acme Chemical Laboratories
165 Ltd, Vancouver, Canada. Major elements were determined by using ICP-OES after fusion with
166 lithium borate flux. Trace elements, including the REE, were analyzed by ICP-MS with
167 additional lithium tetraborate fusion.

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169 **PETROGRAPHIC DESCRIPTION AND GEOCHEMISTRY OF THE HLÍNA GRANITES**

170 **Internal structure and petrography**

171 The Hlína granite bodies show relatively complex internal fabric. The intrusions are
172 predominantly built by fine-grained leucocratic microgranite with layering defined by large
173 proportion of garnet trains parallel to the contacts (Fig. 2; the train texture is used to characterize
174 garnets arranged into the subparallel strips, where individual garnet crystals are almost in a direct
175 contact with each other; e.g. Bogoch et al. 1997; Macleod 1992). The garnet trains are up to
176 several millimeters thick layers or layer sequences within the microgranite portion of the body. In
177 some cases the microgranite is interlayered with comb-shaped large crystals of K-feldspar,
178 plagioclase and quartz (more than 10 centimeters large) designated as unidirectional solidification
179 textures, UST (Shannon et al. 1982; for more detail see Hönig et al. 2010). The USTs are
180 developed mostly in marginal and apical parts of the intrusions and are parallel to the intrusive

181 contacts (Fig. 2a). Alternations of the garnet-bearing microgranite and garnet-free UST layers
182 extend over several meters within some of the Hlína bodies.

183 The fine- to medium-grained (0.5 to 2 mm) microgranite is composed of plagioclase An₈₋₁₅ (35-
184 45 vol%), K-feldspar (23-30 vol%) and quartz (20-32 vol%). The common minor to accessory
185 minerals are muscovite (3-5 vol%), garnet and magnetite (<2 vol%). Other accessory minerals:
186 epidote > zircon > ilmenite > secondary REE-bearing phases >>> titanite > Nb-Ta oxides >
187 xenotime-(Y) are rare to extremely rare. The grain size of the UST zones varies from 1 to 10 cm.
188 The mineral assemblage is represented by K-feldspar, plagioclase (An₁₀₋₁₅) and quartz with minor
189 muscovite; accessory minerals are extremely rare or absent.

190

191 **Garnet**

192 Garnet is present in two distinct morphological types: (i) the most abundant euhedral garnet (<0.5
193 mm) is arranged in long trains (from several dm to several m long; Fig. 2). These trains locally
194 alternate with similar but less abundant zones of individual magnetite grains. Euhedral garnet is
195 commonly inclusions-free (Fig. 3a-c); but it rarely encloses magnetite or minute zircon grains;
196 (ii) rather rare glomerophyric garnet (enclosing K-feldspar, quartz, plagioclase) forms isolated
197 phenocrysts (0.5-4 mm) with no apparent spatial relation to the garnet trains (Fig. 3d).

198 The euhedral garnet exhibits complex zoning patterns in BSE images characterized by core with
199 pronounced sector zoning usually overgrown by zones with sharply-defined fine oscillation –
200 garnet I (Fig. 3a-c). Both texturally distinct domains of garnet I, that is, volumetrically minor
201 sector zoned cores and volumetrically dominant domain with oscillatory zoning, show similar
202 chemical composition with gradual changes in major and trace elements. The euhedral garnet I

203 corresponds to $\text{Sp}_{45-46}\text{Alm}_{27-38}\text{And}_{4-13}\text{Grs}_{0-12}\text{Mzr}_{2-5}\text{Prp}_{0-1}$ (Fig. 4a; Table 1). Major cations
204 (1.27-1.36 apfu Mn, 1.13-1.27 apfu Fe, molar $\text{Mn}/(\text{Fe}_{\text{tot}}+\text{Mn})$ ratio = 0.50-0.52) are rather
205 constant and display negligible variations across garnet I grains (Fig. 4b). Garnet I shows low Al
206 (1.63-1.84 apfu), but it is enriched in Fe^{3+} (0.09-0.26 apfu) and particularly in Ca (0.49-0.70
207 apfu). The contents of Al and Fe^{2+} (0.90-1.18 apfu) gradually increase, whereas Fe^{3+} decreases
208 from the core outwards with no discontinuity during transition from sector to oscillatory zoned
209 domain (Fig. 4b). The content of Ca is constant through the central part of the garnet and depleted
210 in its outer part. The content of Mg (0.03-0.06 apfu) decreases slightly from the core outwards.
211 Along with Y (0.03-0.09 apfu; 0.61-1.96 wt% Y_2O_3), other HREE were detected by EMPA
212 ($\text{Er}_2\text{O}_3 \leq 0.30$ wt%, $\text{Yb}_2\text{O}_3 \leq 0.57$ wt%). The contents of Y+HREE generally decrease from the
213 core outwards (Fig. 4b), but in detail their values finely oscillate and correlate well with the
214 zoning observed in the BSE images (Fig. 4c). The content of Ti is rather high (≤ 0.44 wt% TiO_2)
215 and it is negatively correlated with Al. The content of Na_2O does not exceed 0.16 wt%.

216 Recalculation of the end-members in euhedral garnet shows rather constant spessartine, increase
217 of almandine and decrease of andradite from the core outwards. The content of the grossular end-
218 member increases generally from the core outwards and the Y+HREE content display the
219 opposite behavior (Fig. 4b).

220 In some of the garnet-bearing granite samples, the outermost parts of some garnet I crystals were
221 locally overgrown or replaced by later garnet II (Fig. 3b,c). The replacement irregularly
222 propagates from the rim inwards and the boundary between garnet I and garnet II is texturally
223 and chemically sharp. The later garnet II is slightly enriched in Mn, Al and Fe^{2+} and depleted in
224 Ca, Y+HREE, Fe^{3+} and Ti compared with the garnet I (Fig. 4; Table 1).

225 The rare glomerophytic garnet is only slightly enriched in andradite and spessartine and depleted
226 in almandine and grossular ($\text{Sp}_{\text{S}44-46}\text{Alm}_{28-38}\text{And}_{6-15}\text{Grs}_{6-10}\text{Mzr}_{0-4}\text{Prp}_{0-1}$) when compared to the
227 euhedral garnet (Fig. 4a; Table 1). The contents of Y (0.01-0.07 apfu; 0.19-1.59 wt% Y_2O_3) and
228 HREE ($\text{Er}_2\text{O}_3 \leq 0.30$ wt%, $\text{Yb}_2\text{O}_3 \leq 0.45$ wt%) in the glomerophytic garnet are similar to those in
229 the euhedral garnet. The compositional profile through glomerophytic garnet shows similar
230 chemical zoning as the euhedral garnet.

231

232 **Other rock-forming and accessory minerals**

233 Pink subhedral crystals of K-feldspar ($\text{Or}_{81}\text{Ab}_{17}\text{An}_{0.5}$) show two or more crosscutting perthitic
234 systems. Plagioclase (An_{8-15}) is simply zoned with typical magmatic zoning trend with more
235 calcic core and a myrmekitic texture commonly developed along the contacts with K-feldspar.
236 Both K-feldspar and plagioclase have low concentrations of P (≤ 0.01 wt% P_2O_5); low Sr
237 abundances (≤ 0.08 wt% SrO) in plagioclase are typical.

238 Muscovite forms rather anhedral small flakes. It is Fe-rich (Fe_{tot} up to 0.45 apfu) with Si up to
239 3.23 apfu, and Al ~ 2.3 apfu. Low contents of Na (≤ 0.03 apfu), Mg (≤ 0.04 apfu) and F (≤ 0.04
240 apfu) are typical. Muscovite is partly replaced by chlorite (chamosite with $\text{Fe}/(\text{Fe}+\text{Mg}) = 0.70-$
241 0.97). Wormy epidote (0.71-0.88 apfu Fe^{3+}) intergrowing with quartz usually overgrows
242 muscovite or magnetite. Euhedral crystals of magnetite (20-300 μm) are close to the ideal
243 composition with only trace amounts of Ti, Al and Si.

244 Euhedral to subhedral crystals of zircon (5-50 μm) are remarkably altered (Fig. 3e). Relics of
245 relatively "fresh" zircon ($\text{Zr}/\text{Hf} = \sim 19$) are very rare and contain only low concentrations of P (\leq
246 0.33 wt% P_2O_5), (Y,REE) $_2\text{O}_3$ (up 1.4 wt%), U (0.12-1.50 wt% UO_2), and Th (0.05-0.30 wt%

247 ThO₂). On the other hand, altered domains (Table 2) are depleted in SiO₂ (17.11-27.27) and ZrO₂
248 (38.21-52.08 wt%) and enriched in P₂O₅ (1.59- 6.77 wt%), Nb₂O₅ (0.08 - 3.57 wt%), CaO (0.24-
249 1.67 wt%), ThO₂ (0.21 - 28.19 wt%), UO₂ (0.23 - 2.39 wt%), FeO (0.76 - 3.93 wt%), Al₂O₃
250 (0.75-3.76 wt%) and (Y,REE)₂O₃ (2.94 - 9.72 wt%). The low sum of oxides (83-93 wt%) and the
251 presence of non-essential elements (Fe, Ca, Al, Nb, F) suggests amorphous and hydrated nature
252 of the altered parts of zircon. Incorporation of these elements into the amorphous zircon during
253 interaction with hydrothermal fluids was documented, for instance, by Geisler et al. (2003, 2007).
254 The chondrite-normalized REE pattern of the zircon is HREE-enriched; nevertheless, in the
255 altered parts it is rather flat ($Yb_N/Nd_N = 10-15$) with a weak positive Ce anomaly (Fig. 5a).
256 Aggregates of secondary REE-bearing phases and clay minerals (10-70 μ m large) occur
257 sporadically in thin sections, they follow primary grain boundaries or radiation damage haloes
258 (Fig. 3g,h). Needle-like REE-bearing secondary phase corresponds to Th-rich rhabdophane-(Nd)
259 (0.06-0.23 apfu Th) with variable negative Ce anomaly ($Ce/Ce^* = 0.1-0.6$; Fig. 5a), and with
260 0.07-0.12 apfu Y (Table 2). Another secondary phase is uranium-rich (up 6.5 % UO₂; Fig. 3g,h)
261 and it show very low REE contents close to the detection limit of EMP excluding Ce, yielding
262 significant positive Ce anomaly ($Ce/Ce^* \sim 4$; Fig. 5a). Xenotime-(Y) forms sporadically
263 pyramidal overgrowths on the zircon crystals. Other accessory minerals (ferromite, columbite-
264 (Fe), plumbopyrochlore and titanite) are extremely rare, with the contents of Y+REE below the
265 detection limit of EMP.

266

267 **Whole rock chemistry**

268 The felsic character of all samples (SiO₂ 74.61-75.86 wt%, K₂O 4.11-4.94 wt%, Na₂O 3.82-4.94
269 wt%) is consistent with the high-K calc-alkaline and a metaluminous to slightly peraluminous

270 nature (ASI = 0.98-1.08). Moderate concentrations of CaO (0.61-1.20 wt%), and low
271 concentrations of Fe₂O_{3tot} (0.62-0.93 wt%), MgO (0.02-0.03 wt%), TiO₂ (0.03-0.04 wt%) and
272 P₂O₅ (0.01-0.05 wt%) indicate a generally evolved character of these rocks (Table 3). Rather high
273 concentrations of MnO (0.04-0.90 wt%), Rb (156-311 ppm) and Nb (11-51 ppm) are typical for
274 garnet-enriched rocks and positively correlate with high contents of Y (14-116 ppm) and HREE
275 (Yb 4-15 ppm, Er 2-11 ppm). Compared to the granites of the Réna and Tetčice suites
276 (Leichmann and Höck 2008), the contents of Zr (66-107 ppm), Sr (13-30 ppm), and Ba (4-46
277 ppm) are low. Low K/Rb (110-241), fairly low mg# 4-7 (defined as molar ratio of 100
278 MgO/(FeO_t+MgO)), strong negative Eu anomaly (Eu/(Sm × Gd)^{1/2} = 0.15-0.36) and very high
279 K/Ba (742-2544) indicate a high degree of fractionation. The rock is depleted in LREE (only ~10
280 times enriched over chondritic abundances) relative to HREE (70-90 times enrichment relative to
281 chondrite), the La_N/Yb_N ratios vary from 0.53 in garnet-free rocks to 0.13 in garnet-rich zones
282 (Fig. 5f). The Hlína suite falls mostly into the field of within-plate granites (Fig. 6a,b) and,
283 according to the Ga/Al ratio (Fig. 6c), the rocks plot into the field of A-type granites. Based on
284 the concentrations of Y, Nb and Ga*3 (Fig. 6d) the Hlína granites are classified as the rocks with
285 a post-collision, post-orogenic and anorogenic affinity.

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LA-ICP-MS DATA

288 **Garnet**

289 Yttrium and REE were measured across 10 crystals of euhedral garnet, 3 to 11 analyses were
290 obtained along each analytical profile (total 72 analytical points). Yttrium contents vary from
291 2630 to 15530 ppm. Total rare earth element concentrations vary from 1710 to 9990 ppm and
292 positively correlate with Y contents (Table 4). The concentrations of the Y, Yb and Er obtained

293 by LA-ICP-MS are in good agreement with the results of EMPA. All analyses show enrichment
294 in HREE ($Yb_N/Sm_N = 17-441$) and significant Eu anomaly ($Eu/Eu^* = 0.01-0.05$; Fig. 5b; Table
295 4). The Yb_N/Sm_N ratio decreases smoothly with decreasing $\Sigma Y+REE$ contents from core towards
296 the rim of the garnet I (Fig. 7a) with no disruption during transition between domains with sector
297 and oscillatory zoning. By contrast, garnet II shows significantly higher Yb_N/Sm_N compared with
298 the outer zones of garnet I (Fig. 7a,b). Moreover, garnet II has more negative Eu anomaly
299 compared with garnet I (Fig. 7c,d).

300

301 **Other rock-forming and accessory minerals**

302 The Y+REE contents in magnetite are relatively high (90-256 ppm; Table 4). Chondrite-
303 normalized REE patterns (Fig. 5c) are nearly flat ($La_N/Yb_N = 0.4-2.7$) with a significant negative
304 Eu anomaly ($Eu/Eu^*=0.10-0.15$) and a negative Ce anomaly ($Ce/Ce^* = 0.47-0.77$). The Y+REE
305 concentrations in primary muscovite are rather low (8-95 ppm; Table 4). The chondrite
306 normalized REE patterns (Fig. 5d) are flat to slightly LREE-enriched ($La_N/Yb_N=0.5-5.3$), with a
307 negative Eu anomaly ($Eu/Eu^*=0.22-0.77$) and a positive Ce anomaly ($Ce/Ce^*=0.93-4.94$). The
308 highest $\square Y+REE$ contents ($\square 26$ ppm) in muscovite correlate with increased contents of Nb and
309 Ta and could indicate presence of submicroscopic inclusions of Y+HREE-rich Nb- and Ta-
310 containing phases. The Y+REE in plagioclase range from 1.0 to 6.7 ppm; the LREE (0.6-4.0
311 ppm) predominate over HREE (0.08-1.4 ppm), and there is a distinct positive Eu anomaly
312 ($Eu/Eu^* = 1.9-10.5$; Fig. 5e; Table 4). K-feldspar (Y+REE = 0.6-3 ppm) has only the LREE
313 significantly above the detection limit and positive Eu anomaly. The HREE are close to or below
314 the detection limit (Fig. 5e; Table 4). The Y+REE contents in quartz are low and close to the
315 chondritic values (Y+REE = 3-11 ppm), with flat Ce_N/Yb_N ratio close to 1 (Table 4). and.

316 Epidote has low contents of LREE (Σ LREE \sim 10-30 ppm) with a steep LREE-rich chondrite
317 normalized pattern (HREE are below the detection limit) and a weak negative Ce anomaly (Fig.
318 5d; Table 4).

319

320

MINERAL MASS BALANCE FOR Y+REE

321 Mineral mass calculation

322 The Y+REE mineral mass-balance calculations for three granite samples (MB1-1, MB1-52 and
323 MB1-55; Fig. 5f; Table 3) were carried out in order to assess the relative contribution of garnet
324 and other rock-forming minerals to the whole-rock Y+REE budget. These calculations ignore
325 rare small accessory minerals ($<10 \mu\text{m}$) with Y+REE contents below the detection limit of EMP
326 (ilmenite, titanite, Nb-Ta oxides) and extremely rare xenotime (one 5 by 10 μm -large xenotime-
327 (Y) grain per thin section contributes less than 0.05 % of the total Y+REE in granite). The
328 proportional contents of quartz, K-feldspar, plagioclase, muscovite and garnet were calculated by
329 mass balance using average EMP data and whole rock chemical compositions. The garnet
330 fraction was calculated from the MnO content in the whole rock lowered by the MnO content
331 retained in muscovite, magnetite and ilmenite. The magnetite, ilmenite and epidote fractions were
332 calculated using BSE image analysis. The fraction of zircon was calculated from the Zr content in
333 the rock lowered by the total Zr content located in other rock-forming minerals. The Y+REE
334 mass-balance calculations were performed as arithmetic mean of the Y+REE in individual
335 minerals (excluding garnet) and their modal content in the rock. Average Y+REE contents for
336 garnet could not have been calculated from the rim-core-rim profiles of the LA-ICP-MS analyses
337 as arithmetic mean because this approach would provide significantly higher Y+REE budget (in
338 this case 1.24 times higher than spherical layer model) due to significant core-rim zoning in

15

339 Y+REE. Thus, the average Y+REE content of the garnet was calculated by the spherical layer
340 model (see Fig. 3f). The calculated Y+REE budgets are presented in Tables 5 and 6 and
341 graphically in Fig. 8.

342

343 **Distribution of Y+REE among rock-forming minerals**

344 The mineral mode and average Y+REE contents in individual primary minerals (garnet, zircon,
345 muscovite, magnetite, quartz, plagioclase, K-feldspar, epidote) were used to calculate the
346 theoretical Y+REE pattern of the whole rock (model A). The calculated Y+REE pattern of the
347 model A is in a good agreement with the pattern obtained from the whole-rock analysis (Fig. 8a)
348 excluding LREE from La-Sm. Due to the strongly heterogeneous chemical composition of
349 aggregates of secondary REE-bearing phases their modal amount was difficult to determine.
350 Therefore, the fraction of secondary REE-bearing phases was calculated according to the
351 unbalanced La-Sm content between the model A and the whole rock chemical analysis and
352 average chemical composition of the secondary REE-bearing phases. The mode of secondary
353 phases corresponds to 10 grains, ~50 μm in size, per thin section, which seems to be consistent
354 with the observations of secondary REE-bearing phases in our samples. Model B (Fig. 8a)
355 expresses a theoretical Y+REE pattern obtained by addition of the secondary REE-bearing phases
356 to the model A. The difference between these two models suggests that ~63 % of the whole rock
357 LREE budget is hosted by secondary REE-bearing phases, which also contribute 1 % to the
358 whole rock Y and 21 % to the whole rock REE.

359 The results shown in Table 5 demonstrate that ~84 % of the Y and ~61 % of the REE are hosted
360 in garnet. The garnet contribution to the HREE is ~86 % and it systematically increases from Gd
361 to Yb (Fig. 8b; Table 6). Zircon, the second important carrier of Y+REE in the rock (Fig. 8b;

362 Table 5), contains ~13 % of whole rock Y and ~11 % REE (11 % LREE and 11 % HREE). The
363 fraction of Y+REE residing in the major rock-forming minerals (plagioclase, K-feldspar, quartz),
364 is with the exception of Eu rather low ~1 % Y and 4 % REE (of 10 % LREE and 1 % HREE)
365 despite the predominance of these minerals (94 vol%) in the rock. Approximately 89 % of Eu is
366 contained in feldspars (Fig. 8b; Table 6). Surprisingly, a relatively high content of Y+REE in
367 magnetite makes it a relatively important reservoir of Y (1 %) and REE (2 % of the total REE and
368 4 % of the total LREE), despite its low modal amount (Table 5). Moreover, ~1 % Y and 2 % REE
369 (4 % LREE>0.5 % HREE) reside in muscovite (Table 5). The contribution of epidote to the
370 Y+REE whole rock budget is only negligible (0.3 % of the total LREE; Table 5).

371

372

DISCUSSION

373 **Crystallization temperature**

374 We have used the zircon saturation thermometry to evaluate the liquidus temperature of the Hlína
375 A-type granite. Application of the zircon solubility model in silicate melts (Harrison and Watson
376 1983) yields temperature range between 711-751 °C, that is, 734 ± 14 °C. The zircon saturation
377 thermometry is applicable provided that Zr remained immobile on the sample scale during
378 postmagmatic alteration of zircon. Several lines of evidence suggest its generally conservative
379 behavior: (i) the Zr concentrations in six whole rocks neither correlate with the degree of zircon
380 alteration nor they show irregular fluctuations, (ii) the estimated temperatures are not scattered
381 but form a rather narrow range, and (iii) if portion of Zr was removed by aqueous fluids, the
382 original crystallization (saturation) temperature would have to be higher than ~730 °C, which is
383 unexpected for highly evolved granitic magmas (e.g., Anderson et al. 2008; Huang and Audétat

384 2012; London 2014). We thus interpret results of the zircon saturation thermometry as maximum
385 liquidus temperature, and subsequent crystallization must have proceeded in a closed system.

386 The minimum temperature limit of granite crystallization was estimated by two-feldspar solvus
387 geothermometry using non-ideal asymmetric mixing model of Holland and Powell (2003). K-
388 feldspar composition exhibits slight systematic decrease in Na concentrations from core to rim
389 (Ab_{14-09} to Ab_{05-04}). Plagioclase composition shows minor but systematic decrease from An_{15} to
390 An_{14-12} (core to rim), with Or_{00-01} . Cores of corresponding plagioclase and K-feldspar grains
391 indicate equilibrium temperatures from 441 ± 14 to 418 ± 18 °C at 1 kbar. By contrast, rims of
392 plagioclase and K-feldspar grains yield variable, poorly constrained temperatures from 413 ± 97
393 to 294 ± 15 °C at 1 kbar. We interpret the range of 441-418 °C as a minimum temperature limit
394 for granite crystallization, possibly under undercooled conditions, whereas the latter datum of
395 294 ± 15 °C is appears to indicate closure of the Na-K exchange *via* grain boundary diffusion.

396 Layered textures (alternating garnet-bearing aplite and garnet-free coarse-grained comb-like UST
397 layers) observed in the Hlína granite together with hour-glass sector cathodoluminescence-zoning
398 of K-feldspar and sector zoning in central parts of the garnets indicate a crystallization from non-
399 equilibrium supersaturated undercooled melt (*cf.* Webber et al. 1999; London 2009, 2014; Hönig
400 et al. 2010), broadly bracketed by 734 and 441-418 °C.

401

402 **Igneous origin of garnet**

403 The origin of garnet in granitic magmas (magmatic phenocrysts, xenocrysts, restite or secondary
404 reaction products) is still intensively debated (e.g. Miller and Stoddard 1981; Kontak and Corey
405 1988; Hogan 1996; Dahlquist et al. 2007; Erdmann et al. 2009; Dorais and Tubrett 2012; Lackey
406 et al. 2012). The chemical composition and zoning, euhedral morphology (Figs. 3a-c, 4) and

407 textural relations of garnet from the Hlína granite are consistent with its igneous origin. Garnet I
408 crystallized early from the felsic magma that was highly evolved, metaluminous to weakly
409 peraluminous (ASI = 0.98-1.08; Table 3) and rich in Y+HREE, Mn and poor in Mg, P, and Eu.
410 By contrast, formation of garnet in response to a biotite-consuming reaction, which would have
411 reflected an increase in the melt peraluminosity due to the progressive crystallization (Hogan
412 1996), appears to be unlikely. Interstitial, rather anhedral primary muscovite and wormy epidote
413 crystallized later, when H₂O concentration in the residual melt increased slightly, and this is
414 consistent with mineral sequences observed in A-type magmas crystallizing under initially low
415 H₂O fugacities (e.g. Bonin 2007).

416 The excess of Al in the melt is usually considered to be a prerequisite for garnet stability (du
417 Bray 1988). This is in agreement with common occurrence of accessory garnet of almandine-
418 spessartine composition with low contents of Mg and Ca (<2 wt% CaO) in the S-type (ASI \square
419 1.1) granites. Commonly, such garnets are Al-rich (1.95-2.00 apfu Al; Miller and Stoddard 1981;
420 du Bray 1988; Sevigny 1993; Dahlquist et al. 2007; Sheibi et al. 2010). By contrast, magmatic
421 garnet in the A-type Hlína granites shows lower Al concentrations (~1.6 apfu in garnet cores),
422 which are compensated by Fe³⁺ \pm Ti in the octahedral site (Fig. 4b; Table 2). The decreased Al
423 and elevated Ca, Fe³⁺ and Ti concentrations in garnet enable garnet crystallization from weakly
424 peraluminous melt (with ASI ~1). The garnet I composition resembles those described from
425 highly fractionated metaluminous to weakly peraluminous (ASI = 0.96-1.05) A-type granites at
426 Erlangmiao (Zhang et al. 2012).

427 Chemical composition (both major and trace elements) indicates that the transition from sector
428 zoned cores of garnet I to oscillatory zoned garnet I domains reflects continuous garnet
429 crystallization with no disruption in garnet growth (Fig. 4b, 7). Sector-zoned cores of garnet I

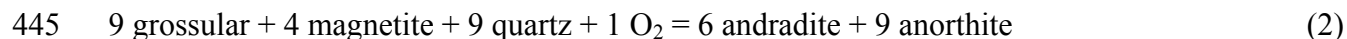
430 crystallized rapidly from undercooled melt far from equilibrium with coexisting minerals.
431 Moreover, rim of oscillatory zoned garnet I crystallized close to equilibrium with coexisting
432 minerals. Flat Mn and Fe_{tot} profiles (Fig. 4b) for garnet I could reflect rapid self-diffusion of
433 these elements, which is characteristic for garnets that have crystallized at high temperatures ~
434 700 °C (Yardley 1977; Manning 1983; Harrison 1988; Dahlquist et al. 2007). This observation
435 agrees well with the results of the zircon saturation geothermometry and confirms that garnet I is
436 an early, high-temperature liquidus phase in A-type granitic magmas (Bonin 2007).

437

438 **Redox conditions during crystallization**

439 We used the composition of garnet I (from a sample with no other garnet varieties present), in
440 combination with that of other coexisting minerals, to reconstruct the redox conditions during
441 magma crystallization (Fig. 9). The following mineral equilibria are applicable to garnet- and
442 magnetite-bearing leucogranites:

443



446

447 We used the internally self-consistent thermodynamic dataset of Holland and Powell (1998), as
448 updated in 2004, the compensated Redlich-Kwong equation of state for H₂O (Holland and Powell
449 1991, 1998) and the asymmetric non-ideal mixing models for plagioclase (Holland and Powell
450 2003) and garnet (White et al. 2007, as revised by Diener et al. 2008 and incorporating ideal Fe-
451 Mn mixing in the sense of White et al. 2005) to evaluate redox conditions of these equilibria with

20

452 respect to fayalite or anthophyllite-magnetite-quartz and magnetite-hematite oxygen-fugacity
453 buffers. Thermodynamic data for the skiagite end-member ($\text{Fe}_3\text{Fe}_2\text{Si}_3\text{O}_{12}$) were calibrated by
454 Malaspina et al. (2009), however, their exploratory application indicated equilibrium redox
455 conditions in excess of the magnetite-hematite buffer, in a strong disagreement with the observed
456 mineral assemblage and outside the range of common igneous rocks (e.g., Carmichael 1991,
457 Frost and Lindsley 1992). We prefer to estimate the thermodynamic data for the skiagite end-
458 member from the reciprocal composition space of the garnet solid solution in a self-consistent
459 manner with non-ideal excess mixing contributions (Diener et al. 2007).

460 Eq. (1) applied to the composition of garnet I core (sector zoned domain of the garnet I) yielded
461 $\log f\text{O}_2 = -14.9$ bar (Fig. 9), which corresponds to QFM + 1.2 at 734 °C (zircon saturation
462 temperature). Progressive crystallization of Fe^{3+} -bearing garnet I decreases the $\text{Fe}^{3+}/\text{Fe}^{2+}$ ratio of
463 the residual melt, thus effectively shifting the evolving system towards more reducing conditions.
464 The equilibrium between garnet I rim (homogeneous rim of oscillatory zoned domain of the
465 garnet I), magnetite and quartz yielded $\log f\text{O}_2 = -16.2$ bar at 718 °C, which is the liquidus
466 temperature for the crystallization of garnet and magnetite, as opposed to stability of garnet and
467 fayalite at lower temperatures. The equilibrium between quartz, magnetite, plagioclase and rim of
468 garnet I was reached at 657 °C as indicated by the intersection of Eq. (2) with the QFM buffer,
469 whereas the equilibrium between the same phase but garnet II occurred under similar conditions,
470 at 663 °C and QFM + 0.8 (Fig. 9). The location and course of the garnet-magnetite equilibrium
471 (Eq. 1; Fig. 9) indicate that the garnet I core composition will change only little if the
472 crystallization has occurred metastably during undercooling below ~730 °C. Following the
473 crystallization of garnet I, compositional changes of the residual melt caused its temporary
474 instability, it was partly dissolved and overgrown by Ca,Y+REE-poor and Mn,Al-rich garnet II

475 (Figs. 3b,c, 4b, 7). Despite these compositional fluctuations the compositions of both garnet types
476 yield consistent crystallization conditions, 657-663 °C and QFM 0 to + 0.8 at 1 kbar, which
477 represent an undercooling of ~75 °C below the zircon saturation temperature. The estimated
478 redox conditions (QFM 0 to +1.2) indicate that the Hlína suite belongs to moderately oxidized A-
479 type granites (Frost et al. 1999), but are consistent with oxygen fugacity recorded by magnetite,
480 titanite and ilmenite assemblages (Wones 1989), which are all observed in our samples.

481

482 **Y+REE substitutions in garnet**

483 Four substitution mechanisms controlling incorporation of Y+HREE into the garnet structure
484 have been proposed previously: (i) YAG substitution, [${}^X(\text{Y,REE}){}^{3+}_1\text{T}(\text{Al}^{3+},\text{Fe}^{3+})_1\text{X}(\text{R}^{2+})_1\text{T}(\text{Si}_{-1})$], is
485 considered to be common in spessartine-rich garnets from pegmatites (e.g., Jaffe 1951) and it has
486 been confirmed in metamorphic garnet as well (Røhr et al. 2007); (ii) coupled substitution linked
487 to the incorporation of Na into the X-site [$\text{Na}^+_1(\text{Y,REE}){}^{3+}_1(\text{R}^{2+})_{-2}$] described by Enami et al.
488 (1995) from metamorphic garnet in orthogneisses; (iii) menzerite-(Y) substitution
489 ${}^X(\text{Y,REE}){}^{3+}_1\text{Y}\text{R}^{2+}_1\text{X}\text{R}^{2+}_{-1}\text{Y}(\text{Al,Fe}){}^{3+}_{-1}$ has been described from metamorphic garnets in felsic
490 granulites only (Grew et al. 2010; Marsh et al. 2012); (iv) $(\text{Y,REE}){}^{3+}_1\text{Al}^{3+}_{-1}$ homovalent
491 substitution for the garnet from the Xihuashan granitic complex (Wang et al. 2003b), although
492 the same trend may be explained by the menzerite substitution (Grew et al. 2010).
493 The absence of Al in the T-site and negative correlation of Y vs. ${}^Y\text{Al}$ in garnet from the Hlína
494 granites (Fig. 10a) implies the absence of the YAG substitution. Owing to the low Na contents
495 (0.026 apfu) and the lack of correlation with the Y+REE contents (Fig. 10b), only minor
496 contribution from the substitution ${}^X\text{Na}{}^X\text{Y}{}^X\text{R}^{2+}_{-2}$ is expected. Yttrium concentrations negatively
497 correlate with ${}^X\text{R}^{2+}$ (Fig. 9c), and positively correlate with ${}^Y\text{R}^{2+}$ (Fig. 10d). These observations

498 suggest participation of the “ferro-menzerite”-(Y) substitution in garnet from the Hlína granite.
499 Positive correlations of Y and ^XCa and $^Y\text{Fe}^{3+}$ (Figs. 10e,f) further demonstrated that incorporation
500 of Y into the garnet structure is promoted by the andradite component. Menzerite (or “ferro-
501 menzerite”) substitution operates in low-pressure magmatic spessartine-rich garnet (this study)
502 along with high-grade metamorphic rocks (Grew et al. 2010; Marsh et al. 2012).

503

504 **Garnet and further carriers of Y+REE in the rock**

505 The whole-rock budget of Y+REE elements is principally controlled by accessory minerals and
506 their assemblage reflects the ASI and activity of P_2O_5 , F and other volatiles (e.g. Bea 1996;
507 Dahlquist 2001; Trumbull et al. 2010). Nevertheless, mineral mass-balance calculations for
508 Y+REE are rather scarce (Ward et al. 1992; Bea et al. 1994; Bea 1996; Dahlquist 2001; Förster
509 and Rhede 2006; Trumbull et al. 2010). In peraluminous granites more than 90 wt% of the
510 whole-rock Y+REE budget reside in the Y+REE-bearing phosphates (monazite, apatite,
511 xenotime) and zircon. The presence of REE-bearing phosphates reflects high phosphorus
512 solubility in peraluminous melts (Ward et al. 1992; Bea et al. 1994; Bea 1996). In metaluminous
513 to weakly peraluminous I-type granites zircon, allanite, titanite with amphibole are the main
514 carriers of Y+REE (Bea 1996; Dahlquist 2001; Trumbull et al. 2010). The sporadic mineral
515 mass-balance calculations for A-type granites indicate that REE-carbonates, allanite, niobo-
516 tantalates and zircon are the main carriers of Y+REE together with minor titanite, xenotime and
517 monazite (Bea 1996).

518 Garnet in peraluminous S-type granites contains only low Y (45-700 ppm) and Σ REE
519 concentrations (40-600 ppm; du Bray 1988; Sevigny 1993; Bea 1996) and therefore hosts only
520 subordinate amounts of the whole-rock Y+HREE budget (Bea 1996). On the contrary, it seems

521 that high Y (≤ 1.32 wt% Y- Harrison 1988; ≤ 1.89 wt% Y- Wang et al. 2003b; ≤ 1.54 wt% Y- this
522 study) and Σ REE contents (≤ 0.50 wt% - du Bray 1988; ≤ 1.00 wt% - this study) are common in
523 magmatic garnet found in fractionated SiO₂-rich and P-poor A-type and I-type granites. Low
524 contents of P in the melt prevented precipitation of REE-bearing phosphates in the Hlína granite
525 and the Y+REE remained incompatible in the melt until the saturation in garnet was reached.
526 Garnet is a major carrier of Y and REE in the A-type Hlína granite and accounts for about 84 %
527 of the Y and 61 % of the REE whole-rock budget (Fig. 8b, Table 5,6). Despite rather low ASI in
528 the A-type Hlína granites (Table 3), the allanite is absent and titanite is very rare. This is
529 consistent with its absence in the most siliceous granites elsewhere due to their low Ca contents
530 (Frost et al. 2000). The negligible contribution of epidote to the LREE whole rock budget in the
531 Hlína granite (Fig. 8b) is in contrast to observations from metaluminous granites and mafic
532 metamorphic rocks, where epidote-group minerals can host 50-95 % of LREE regardless of the
533 species (allanite, epidote, zoisite; Bea 1996; Spandler et al. 2003). Epidote in the Hlína granite
534 formed as a late mineral from REE-depleted melt. Spandler et al. (2003) confirmed that garnet
535 represents the dominant host for Y+HREE in metamorphosed pelitic and mafic rocks under
536 epidote blueschist to eclogite facies conditions as well. The REE pattern of zircon is relatively
537 flat, LREE-rich (Fig. 5a) when compared to typical REE patterns in zircon from granitic rocks
538 (cf. Hoskin and Schaltegger 2003). This unusual chemical composition makes zircon an
539 important carrier not only of Y and HREE (13 % Y and 11 % HREE of the whole-rock budget)
540 but also of LREE (11 % of the whole-rock budget; Fig. 8b). This LREE enrichment is probably a
541 result of interaction of amorphous and hydrated zircon-breakdown products with LREE-bearing
542 hydrothermal fluids.

543 Secondary REE-phases (Figs. 3g,h) represent alteration products after a primary REE-bearing
544 mineral enriched in U and Th, probably Th- and U-rich monazite or cheralite. This primary REE
545 mineral hosted 21 % of REE and 1 % of Y (assuming that REE remained immobile during
546 hydrothermal alteration), and represented important carrier of LREE (63 %) mainly from La to
547 Nd (Fig. 8b; Table 5,6). Trivalent cerium from primary phosphate was probably partly oxidized
548 to Ce⁴⁺ during the phosphate alteration and the released REE were redistributed between Ce-
549 depleted rhabdophane-(Nd) accommodating REE³⁺ and a Ce⁴⁺+U-enriched phase (Fig. 5a).
550 Only a few attempts have been made to investigate the Y+REE contents of magnetite from
551 granites. Magnetite in the Hlína granite has similar Y+REE content and pattern (Fig. 5c) to the
552 magnetite from the Rostberget granite (Öhlander et al. 1989) excluding the presence of negative
553 Ce anomaly. The negative Ce anomaly in magnetite is accompanied by positive Ce anomaly in
554 zircon and muscovite suggesting that Ce³⁺ was partly oxidized to Ce⁴⁺ and behaved distinctly
555 from the other REE.

556

557 **Processes controlling Y+REE variations in garnet**

558 Bulk diffusivities for many trivalent trace elements (including Y+REE) in garnet are slower than
559 those for major divalent (Mg, Fe, Mn) cations or for other octahedrally or tetrahedrally cations
560 (Lyasevich et al. 1977; Yardley 1977; Loomis et al.1985; Chakraborty and Ganguly 1992;
561 Cherniak 1998; Carlson 2006, 2012).

562 General outward decrease of Y and REE is a common feature of garnet from the A-type Hlína
563 granite (Figs. 4b,c,7), and it is common in garnet from some highly fractionated I-type granites
564 (Wang et al. 2003a) and granitic pegmatites (Smeds 1994; Whitworth and Freely 1994; Müller et

565 al. 2012) as well as from biotite granodiorites (Dorais and Tubrett 2012). In contrast, garnet from
566 peraluminous S-type granites shows increasing Y+REE contents from the core towards the rim or
567 zoning is absent (Sevigny 1993; Bea 1996). The Y+REE contents in garnet from the Hlína
568 granite correspond well with zoning patterns observed in the BSE images (Fig. 4c) and they seem
569 to be a sensitive indicator of magmatic garnet evolution even though the self-diffusion of the
570 major divalent cations (Mn, Fe) was sufficiently rapid to eliminate any original compositional
571 zoning. Rapid growth of the garnet from undercooled melt, indicated by the sector zoning in the
572 core (Fig. 3a,b; Shelley 1992), consumed a large part of Y+HREE from the residual melt. The
573 observed Y+REE oscillatory zoning in the Hlína garnet (Fig. 3a,b) could have been produced
574 during the alternating changes in the growth and diffusion rates near the crystal-melt interface, as
575 modelled by Allègre et al. (1981) or Tsune and Toramaru (2007). Garnet/melt partition
576 coefficients for REE generally increase with increasing atomic number (Irving and Frey 1978;
577 Fujimaki et al. 1984; Green et al. 2000). Hence, the HREE are preferentially incorporated into the
578 garnet structure, residual melt is continuously impoverished during the growth of garnet I and the
579 outer part of garnet I does not reach such high Y+REE contents and the REE pattern is flatter
580 relative to the core (Fig. 7a). The outward decreasing Yb_N/Nd_N ratio during crystallization of the
581 garnet I (Fig. 7a) indicates that growth of magmatic garnet is capable of fractionating REE in the
582 melt towards the HREE-poorer pattern as shown for high-grade metamorphic garnet as well
583 (Otamendi et al. 2002). Opposite compositional Y+REE trends in garnet from peraluminous S-
584 type granites (core-rim increasing Yb_N/Nd_N ratios) reflect the growth and fractionation of
585 monazite, which depleted the liquid in LREE simultaneously with the garnet growth (Sevigny
586 1993).

587 The replacement textures observed at the outer rims of euhedral garnet (Fig. 3b,c) were
588 interpreted as products of magmatic corrosion. Higher Mn and Al, lower Ca, Y+REE content, as
589 well as higher Yb_N/Nd_N ratio and more pronounced negative Eu anomaly in the garnet II (Fig.
590 4b,7) indicate its crystallization from the more evolved melt (London 2008; Müller et al. 2012).
591 Garnet has the highest Yb_N/Nd_N ratio from the studied mineral assemblage (Fig. 5, Table 4).
592 Crystallization of other phases (with lower Yb_N/Nd_N ratio relative to garnet) during the time gap
593 between the crystallization of garnet I and garnet II increased the Yb_N/Nd_N ratio in the melt.
594 Crystallization of feldspars consuming Eu yielded more pronounced negative anomaly in the
595 more evolved melt. Consequently, garnet II shows lower $\Sigma Y+REE$ abundance, higher Yb_N/Nd_N
596 ratio and more negative Eu anomaly when compared to garnet I (Fig. 7). Alternatively, the lower
597 $\Sigma Y+REE$ and higher Yb_N/Nd_N ratio in garnet II can be attributed to the crystallization from the
598 melt relatively enriched in the volatile constituents (H_2O , B_2O_3 , F) at lower temperature. Higher
599 B contents in garnet II (up to 250 ppm in garnet II and 25-67 ppm B in garnet I; Table 4) support
600 the progressive volatile enrichment. Distribution coefficients $D_{Grt/melt}$ for Y+REE in garnet
601 decrease with increasing H_2O content in the melt (Green et al. 2000) and decreasing temperature
602 in the experiments, hence hydrous magma tends to increase fractionation of HREE from LREE
603 into garnet (steeper pattern) compared to anhydrous conditions (Green et al. 2000). According to
604 Wang et al. (2003a) and Müller et al. (2012), the formation of Y-poor garnet margins
605 overgrowing the Y-rich cores indicates crystallization from volatile-enriched residual melt.

606

607 **IMPLICATIONS**

608 We documented early magmatic garnet from A-type, metaluminous to weakly peraluminous
609 granites. This garnet (<2 vol.%) is a major carrier of the Y and REE, and it incorporates 84 % Y

610 and 61 % REE of the whole-rock budget of these constituents. Detail mineral mass-balance of Y
611 and REE also revealed minor contributions of zircon and monazite. This is in contrast with the
612 majority of medium- to high-P granitic rocks where phosphates (e.g., monazite, xenotime) and
613 zircon are the major Y+REE carriers.

614 Our mass-balance calculations indicate that a spherical layer model or a more sophisticated
615 computation for determination of the average content of major to trace elements in minerals with
616 pronounced core-rim zonation is required to obtain meaningful results. By contrast, conventional
617 arithmetic mean can provide significantly higher (in case of enriched core and depleted rim) or
618 lower (in case of depleted core and enriched rim) values.

619 Our observations indicate menzerite (and "fero-menzerite") substitution to be the main
620 incorporation mechanism of Y+REE into the Hlína garnet. It demonstrates that this substitution is
621 viable not only in higher-grade metamorphic rocks but also in low-pressure magmatic rocks.

622 Elucidation of the substitution vector introducing higher Y+REE content into the garnet is crucial
623 for proper mineral-formula calculation, including Fe^{2+}/Fe^{3+} ratio and determination of garnet
624 end-members proportions.

625 Participation of the menzerite molecule, as well as incorporation of Ca, Ti^{4+} , along with high Fe^{3+}
626 content, all promote crystallization of garnet from metaluminous to weakly peraluminous melts.

627 Crystallization of the initially Ca-rich garnet effectively eliminated Ca from the residual melt,
628 which indirectly led to an increase in ASI. This magmatic garnet is an additional example of the
629 mineral typical for peraluminous granitic rocks stabilized in weakly metaluminous granitic rock
630 by elevated contents of Ca, Fe^{3+} and Ti^{4+} . By analogy, tourmaline (schorl-dravite), another
631 characteristic mineral of strongly peraluminous granites (London 2008), is stabilized by elevated
632 contents of Ca, Fe^{3+} and Ti^{4+} in metaluminous NYF pegmatites (Novák et al. 2011; Čopjaková et
633 al. 2013) and even in alkaline pegmatite (Filip et al. 2012).

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940

941 Figure captions

942 Fig. 1. Up left corner – Geological map of the Brunovistulicum, AF - Alpine Front, MSFZ –
943 Moravo-Silesian Fault Zone, HKFZ – Hamburg-Krakow Fault Zone, PPFZ – Peri-Pieninian Fault
944 Zone, TESZ – Transeuropean Suture Zone ; Center – simplified regional geological map of the
945 Brno and Thaya batholiths highlighting numerous of Hlína A-type granite bodies and dykes.
946 Down right corner – detail geological map of the Hlína A-type granite. Letters refer to the
947 individual localities: A. Moravské Bránice; B. U stavení; C. Pod amfiteatrem; D. Jihlava river
948 bank. The size of the Hlína bodies is slightly enlarged compared to its real volume to better
949 display. Projection – JSTK Krovak.

950

951 Fig. 2. Photographs of the Hlína rocks: a) Contact of laminated Hlína microgranite with hosted
952 Réna I-type granite bordered by marginal UST zone. b) Garnet-bearing Hlína microgranite with
953 sharp contacts to the hosted I-type Réna granite. c) Garnet trains parallel to the lamination of the
954 Hlína laminated microgranite body. d) Detail of microgranite with garnet trains alternating with
955 unidirectional Pl, Qtz and Kfs crystals (UST).

956

957 Fig. 3. BSE images of minerals; a) euhedral garnet I with sector zoning in central part and
958 oscillatory zoning in outer part; b,c) euhedral garnet with sector and oscillatory zoning; garnet I

959 (brighter) is partially replaced and overgrown by garnet II (darker); d) glomerophyric garnet
960 enclosing K-feldspar, quartz and plagioclase; e) "altered" zircon with small relics of relatively
961 "fresh" zircon; f) euhedral garnet after LA-ICP-MS measurement with marked spherical layers
962 used for calculation of average Y+REE content. The garnet crystals are considered as spheres that
963 are divided in to $n-1$ concentric spherical layers and the central sphere. The n vary from 4 to 7
964 and depends on the number of LA-ICP-MS analyses in the individual garnet crystal. The amounts
965 of Y+REE in specific spherical layers and for the central sphere were calculated as a function of
966 the volume and the Y+REE content; g) aggregate of secondary REE-bearing phases; needle-like
967 rhabdophane-(Nd) (Ph I = phase I) predominates over U-rich secondary phase (Ph II = phase II)
968 and clay minerals after unknown primary REE-bearing mineral; h) aggregate of secondary REE-
969 bearing phases; U-rich secondary phase (Ph II) predominates over needle-like rhabdophane-(Nd)
970 (Ph I) and clay minerals respecting boundary of radiation damage halo around primary accessory
971 mineral. Primary accessory mineral is almost exclusively replaced by clay minerals.

972

973 Fig. 4. Electron microprobe results of garnet. a) Ternary plots of garnet; upper: almandine-
974 spessartine-Ca-garnet (Ca-garnet=andradite+grossular+menzerite-(Y)); lower: grossular-
975 andradite-menzerite-(Y). b) The profiles through euhedral garnet I replaced by garnet II. The
976 upper plot shows variation of major and minor elements, the lower plot shows variation in the
977 end-members. c) Correlation between zonation in BSE image and Y content in the garnet; upper:
978 garnet I - not corroded rim; lower: garnet I partially replaced by garnet II.

979

980 Fig. 5. The REE patterns of individual minerals based on EMP data (a) and LA-ICP-MS data (b-
981 e) and the A-type Hlína granite (f). The REE data are normalized by chondrite (Taylor and
982 McLennan 1985). a) Zircon, rhabdophane-(Nd), U-rich phase, b) euhedral garnet, c) magnetite,

983 d) muscovite and epidote, e) plagioclase and K-feldspar. Arithmetic mean for Y+REE contents or
984 compositional range for individual minerals are plotted depending on the REE data heterogeneity.

985

986 Fig. 6. Intrusive setting of the Hlína A-type granite shown in various tectonic discrimination plots
987 after Pearce et al. (1984), Whalen et al. (1987) and Eby (1990); a) (Y+Nb) vs. Rb plot; b) Y vs.
988 Nb plot; c) (10000*Ga/Al) vs. Zr plot; d) ternary 3*Ga-Nb-Y diagram. Abbreviations: WPG -
989 Within Plate Granites; ORG - Ocean Ridge Granites, VAG - Volcanic Arc Granites, syn-COLG -
990 Syn Collisional Granites; A1 - A-type granite field associated with hotspots, mantle plumes
991 and/or continental rift zones; A2 - A-type granite field with post-collision, post-orogenic and
992 anorogenic affinity.

993

994 Fig. 7. The 2D plots for Y+REE in garnet with simplified garnet crystallization scheme;
995 Yb_N/Sm_N ratio vs. Y+REE content for a) all garnet analysis and b) average values for garnet
996 grains; Yb_N/Sm_N ratio vs. Eu/Eu^* for c) all garnet analysis and d) average values for garnet
997 grains. The dotted lines in b) and d) join average values of garnet I and garnet II within each
998 garnet grain.

999

1000 Fig. 8. a) The chondrite normalized REE+Y pattern of the whole rock (WR - blue symbols;
1001 average of the MB1-1, MB1-52 and MB1-55 samples) and patterns calculated according to
1002 models A and B. Dashed red line and open symbols - model A - calculated from primary mineral
1003 mass fractions and their average Y+REE content (Grt, Zrn, Ms, Mag, Qtz, Pl, Kfs and Ep; used
1004 abbreviations after Kretz 1983). Solid red symbols and line - model B - including REE-bearing
1005 secondary phases in addition to the primary minerals involved in the model A.

1006 b) Contribution of the rock-forming minerals (Grt, Zrn, Ms, Mag, Qtz, Pl, Kfs, Ep and secondary
1007 REE-bearing phases) to the budget of individual REE+Y. The height of the bar is normalized to
1008 100 % (scale on the left side Y axis). For the comparison, chondrite normalized REE+Y content
1009 is drawn as the gray underlying field (scale on the right side Y axis).

1010

1011 Fig. 9. The temperature-log oxygen fugacity diagram at $P = 1$ kbar and H_2O saturation
1012 illustrating the location of common fO_2 buffers as well as equilibria relevant to the garnet- and
1013 magnetite-bearing granites in the present study. The zircon saturation temperature (734 ± 14 °C)
1014 constrains the maximum temperature limit for closed-system crystallization. At these conditions,
1015 composition of garnet I and the absence of fayalite restrict the plausible redox conditions up to
1016 1.2 log units above the QFM buffer. Intercepts of multiple equilibria between magnetite, garnet,
1017 quartz and plagioclase at 657-663 °C (for garnet I rim and garnet II overgrowths) are interpreted
1018 as being close to the final solidification temperature. The course of the skiagite-magnetite-quartz
1019 equilibrium is nearly parallel to the QFM buffer, therefore, the crystallization could have
1020 commenced metastably below ~ 730 °C and/or isothermally near 660 °C. Mineral and end-
1021 member abbreviations: an – anorthite, andr – andradite, anth – anthophyllite, fa – fayalite, gr –
1022 grossular, hm – hematite, mt – magnetite, q – quartz, ski – skiagite.

1023

1024 Fig. 10. Binary plots showing major substitution trends involving Y in euhedral garnet. Full
1025 circles - garnet I, open circles - garnet II.

1026

1027 Table caption

1028 Table 1. Electron microprobe analysis of the euhedral and glomerophytic garnet and calculation
1029 of proportional end-members content (in mol. %).

45

1030

1031 Table 2. Electron microprobe analysis of fresh and altered zircon (Zrn) and rhabdophane-(Nd)
1032 (Rhb). Mineral formulae of both phases were calculated using normalization on 4 anions.

1033

1034 Table 3. Whole rock analyses of the A-type Hlína granite.

1035

1036 Table 4. Trace element analyses (LA-ICP-MS data) of rock forming minerals from the Hlína
1037 granite: Grt I-primary garnet, Grt II-secondary garnet, (c-core, r-rim), Ms-muscovite, Mag-
1038 magnetite, Pl-plagioclase, Kfs-K-feldspar, Qtz-quartz (average of 15 analyses), Ep-epidote
1039 (average of 7 analyses) All data are in wt. ppm. na-not analysed.

1040

1041 Table 5. Mineral fractions in the Hlína granite calculated expressed in vol. % and proportional
1042 fraction of Y, \sum REE, \sum LREE and \sum HREE in individual minerals.

1043

1044 Table 6. Average Y+REE content calculated from MB1-1, MB1-52 and MB1-55 samples and
1045 calculated proportional Y and REE content of individual minerals. LREE phs-secondary LREE-
1046 bearing phases, other abbreviations are the same as in table 4.

1047

1048

pyrope	0.0	0.0	0.0	0.4	0.5	1.0	1.0	0.0	0.8
spessartine	44.1	44.6	44.7	43.0	43.0	43.8	45.9	44.2	44.7
Na-Y garnet	0.7	1.2	1.7	1.1	1.1	0.5	0.5	0.9	0.6
morimotoite	2.5	2.7	2.8	1.2	1.2	0.2	0.2	3.2	1.8
menzerite	4.5	3.8	3.4	1.7	1.8	0.6	0.7	3.8	0.9
almandine	29.1	30.2	29.1	37.3	38.0	42.6	39.4	28.4	34.1
andradite	10.7	8.5	10.7	5.7	5.1	0.0	4.4	12.1	9.2
grossular	8.2	9.5	7.5	10.4	10.0	11.4	7.6	7.3	8.0

	"fresh"	"altered"				"fresh"	"altered"		
	Zrn	Zrn	Rhb	Rhb		Zrn	Zrn	Rhb	Rhb
SO ₃	0.02	0.18	0.29	0.00	S ⁶⁺	0.001	0.005	0.009	0.000
P ₂ O ₅	bdl	6.77	20.76	19.49	P ⁵⁺	0.000	0.208	0.703	0.695
Nb ₂ O ₅	bdl	1.00	bdl	bdl	Si ⁴⁺	0.951	0.678	0.307	0.294
SiO ₂	28.53	18.71	7.68	6.98	<i>subtotal</i>	<i>0.952</i>	<i>0.891</i>	<i>1.019</i>	<i>0.989</i>
TiO ₂	na	na	bdl	0.05	Nb ⁵⁺	0.000	0.016	0.000	0.000
ZrO ₂	57.54	41.04	1.48	0.39	Ti ⁴⁺	na	na	0.000	0.001
HfO ₂	7.82	1.77	bdl	bdl	Zr ⁴⁺	0.935	0.725	0.029	0.008
ThO ₂	0.30	1.28	7.53	24.49	Hf ⁴⁺	0.074	0.018	0.000	0.000
UO ₂	1.53	0.63	0.25	0.96	Th ⁴⁺	0.002	0.011	0.069	0.235
AlO ₃	0.08	3.76	na	na	U ⁴⁺	0.011	0.005	0.002	0.009
Sc ₂ O ₃	0.02	0.07	na	na	Al ³⁺	0.003	0.161	na	na
As ₂ O ₃	0.12	0.08	bdl	bdl	Sc ³⁺	0.001	0.002	na	na
Y ₂ O ₃	0.62	5.88	3.36	5.44	As ³⁺	0.002	0.002	0.000	0.000
La ₂ O ₃	bdl	0.04	6.86	4.88	Y ³⁺	0.011	0.113	0.071	0.122
Ce ₂ O ₃	bdl	0.28	11.18	1.97	La ³⁺	0.000	0.001	0.101	0.076
Pr ₂ O ₃	bdl	0.05	3.04	1.69	Ce ³⁺	0.000	0.004	0.164	0.030
Nd ₂ O ₃	bdl	0.21	12.93	8.24	Pr ³⁺	0.000	0.001	0.044	0.026
Sm ₂ O ₃	0.01	0.23	5.04	3.12	Nd ³⁺	0.000	0.003	0.185	0.124
Eu ₂ O ₃	bdl	bdl	bdl	bdl	Sm ³⁺	0.000	0.003	0.069	0.045
Gd ₂ O ₃	0.02	0.52	3.67	2.98	Eu ³⁺	0.000	0.000	0.000	0.000
Tb ₂ O ₃	bdl	0.09	na	na	Gd ³⁺	0.000	0.006	0.049	0.042
Dy ₂ O ₃	0.04	0.92	1.22	1.27	Tb ³⁺	0.000	0.001	na	na
Ho ₂ O ₃	0.02	0.17	na	na	Dy ³⁺	0.000	0.011	0.016	0.017
Er ₂ O ₃	0.11	0.54	0.33	0.32	Ho ³⁺	0.000	0.002	na	na
Yb ₂ O ₃	0.17	0.61	na	na	Er ³⁺	0.001	0.006	0.004	0.004
Lu ₂ O ₃	0.35	0.24	na	na	Yb ³⁺	0.002	0.007	na	na
CaO	0.09	0.30	5.54	7.62	Lu ³⁺	0.004	0.003	na	na
SrO	na	na	0.33	0.25	Ca ²⁺	0.003	0.012	0.237	0.344
FeO	0.33	1.08	2.06	1.92	Sr ²⁺	na	na	0.008	0.006

PbO	0.11	0.13	0.49	0.56	Fe ²⁺	0.009	0.033	0.069	0.067
Na ₂ O	na	na	0.21	0.08	Pb ²⁺	0.001	0.001	0.005	0.006
F	bdl	0.37	0.31	0.18	Na ⁺	na	na	0.017	0.007
O=F		-0.16	-0.12	-0.08	<i>subtotal</i>	<i>1.059</i>	<i>1.147</i>	<i>1.139</i>	<i>1.169</i>
∑ oxides	97.83	86.79	94.41	92.77	∑ cations	2.013	2.035	2.158	2.158
					F ⁻	0.000	0.042	0.039	0.024
					O ²⁻	4.000	3.958	3.961	3.976
					∑ anions	4.000	4.000	4.000	4.000

Locality	A	A	A	B	B	C	D
Sample	MB1-1	MB1-52	MB1-55	J/21b	J/21c	J Re	J Iv
SiO ₂	75.72	74.90	74.61	75.79	75.86	74.92	75.46
TiO ₂	0.03	0.03	0.03	0.03	0.04	0.04	0.04
Al ₂ O ₃	13.60	14.90	13.83	13.64	16.60	13.48	13.27
Fe ₂ O ₃ ^{tot}	0.62	0.71	0.93	0.90	0.71	0.79	0.84
MnO	0.12	0.13	0.32	0.10	0.09	0.04	0.90
MgO	0.02	0.02	0.03	0.03	0.03	0.03	0.02
CaO	0.94	1.11	1.10	0.96	1.20	1.16	0.61
Na ₂ O	4.30	4.21	3.82	3.85	3.85	3.93	4.18
K ₂ O	4.69	4.61	4.94	4.21	4.29	4.15	4.11
P ₂ O ₅	0.02	0.04	0.05	0.02	0.01	0.01	0.01
LOI	0.4	0.3	0.5	0.3	0.3	0.7	0.5
Σ oxides	100.25	100.22	100.27	99.71	99.8	99.25	99.13
Ba	17	4	31	18	14	17	46
Cs	3.9	3.1	2.9	na	na	na	na
Hf	4.5	6.3	6.8	na	na	na	na
Rb	182	180	170	174	156	174	311
Sr	13	13	22	22	21	30	16
Ta	4	2	2.1	na	na	na	na
Zr	66	101	107	79	74	76	72
Y	45	64	116	56	47	23	14
Ga	18	19	18	24	21	20	19
U	7	6	8	12	17	7	16
Th	13	15	18	12	17	7	16
Nb	28	11	13	51	46	19	23
La	2.30	2.30	3.00	2.00	3.00	3.50	1.90
Ce	5.70	5.90	8.50	5.00	6.10	6.70	5.20
Pr	0.75	0.92	1.25	0.87	1.16	1.25	0.53
Nd	3.40	4.90	5.90	4.30	5.00	5.10	2.40
Sm	1.85	2.15	2.93	2.20	2.80	2.30	1.40
Eu	0.14	0.17	0.18	0.21	0.19	0.27	0.10
Gd	2.91	3.58	5.62	3.44	3.26	2.34	2.97
Tb	0.76	0.94	1.66	0.74	0.71	0.44	0.79
Dy	5.79	7.56	12.97	6.51	5.53	2.94	6.48
Ho	1.35	1.90	3.14	1.64	1.45	0.81	1.80
Er	4.91	7.34	11.32	6.30	5.80	2.58	6.26

Tm	1.20	1.47	2.24	1.23	1.20	0.46	1.90
Yb	7.30	10.40	15.90	9.18	7.80	4.43	10.30
Lu	1.25	1.95	2.75	1.75	1.44	0.76	1.74
ASI	0.98	1.07	1.01	1.08	1.03	1.03	1.07
K/Rb	214	212	241	201	228	198	110
K/Ba	2304	1034	1327	1941	2544	2026	742
mg#	5	5	5	6	7	6	4
La _N /Yb _N	0.21	0.15	0.13	0.15	0.29	0.53	0.13
Eu/Eu*	0.18	0.19	0.14	0.23	0.19	0.36	0.15

	garnet grain 8 Grt I-c	garnet grain 8 Grt I-r	garnet grain 5 Grt I-c	garnet grain 5 Grt I-r	garnet grain 5 Grt II	garnet grain 10 Grt I-c	garnet grain 10 Grt I-r	garnet grain 10 Grt II	garnet grain 9 Grt I-c	garnet grain 9 Grt I-r	garnet grain 9 Grt II
B	bdl	67	na	na	na	66	25	93	47	bdl	250
Ti	2513	1165	na	na	na	1625	1106	151	1690	809	198
Zr	25.7	4.8	22.7	4.9	8.5	4.6	2.7	0.5	4.9	3.4	0.6
Nb	6.7	0.7	na	na	na	2.0	0.8	bdl	2.8	1.0	0.1
Ta	1.9	1.0	na	na	na	0.5	0.4	bdl	1.1	0.5	0.1
Nb/Ta	3.55	0.72				3.64	1.91		2.65	2.19	0.84
Y	15534	3756	14447	6780	3151	10090	5951	2973	11069	9438	4704
La	0.08	0.02	0.02	0.03	0.01	0.01	bdl	bdl	bdl	bdl	0.01
Ce	0.31	0.08	0.24	0.48	0.10	0.08	0.04	0.01	0.05	bdl	bdl
Pr	0.18	0.06	0.05	0.12	0.03	0.07	0.05	0.00	0.06	0.04	0.01
Nd	3.67	1.55	3.27	2.71	0.65	2.66	1.89	0.24	2.98	2.17	0.43
Sm	35.06	17.94	23.22	26.93	4.74	26.12	18.67	3.09	28.67	23.02	5.62
Eu	1.48	0.64	0.88	0.84	0.09	1.09	0.58	0.06	1.24	0.88	0.20
Gd	305.4	153.8	191.2	197.9	43.2	228.2	176.5	47.0	276.4	222.5	63.4
Tb	131.7	57.7	89.2	76.0	21.0	106.4	72.5	23.2	123.7	98.5	30.7
Dy	1738	532	1134	683	250	1125	673	302	1481	1084	378
Ho	548	112	391	157	93	327	168	120	488	372	160
Er	2043	291	1636	444	440	1101	532	578	1927	1345	685
Tm	399	72	366	72	103	205	107	141	406	268	183
Yb	4157	720	3288	528	1058	1808	943	1388	3526	2466	1811
Lu	629	91	613	78	211	271	145	260	566	391	346
ΣREE	9992	2049	7735	2267	2224	5202	2838	2863	8829	6273	3663
Yb _N /Sm _N	105	36	126	17	198	61	45	398	109	95	285
Eu/Eu*	0.044	0.038	0.041	0.035	0.020	0.044	0.031	0.016	0.043	0.038	0.032

	Ms	Ms	Mag	Mag	PI	PI	Kfs	Kfs	Qtz avg	Ep avg	
B	7343	2829	Mn	6413	1153						
Ti	754	1190	Ti	19209	3690						
Zr	17.2	49.5	Zr	373	477	0.81	1.51	0.78	0.52		
Y	4.07	38.41	Y	58.7	73.5	1.14	0.40	0.12	1.08	1.72	10.25
La	2.03	7.71	La	17.5	13.1	0.56	0.40	0.06	0.15	0.32	13.07
Ce	11.77	24.76	Ce	23.6	19.5	0.54	0.84	0.08	0.31	0.73	10.01
Pr	0.34	1.47	Pr	4.3	3.8	0.11	0.12	0.04	0.03	0.13	1.82
Nd	1.19	6.01	Nd	18.9	16.9	0.30	0.34	bdl	0.15	0.87	5.80
Sm	0.30	2.22	Sm	8.7	8.9	0.10	0.05	bdl	0.08	0.22	1.80
Eu	0.12	0.41	Eu	0.34	0.38	0.23	0.32	0.23	0.22	0.01	0.00

Gd	1.41	2.53	Gd	9.7	12.3	0.13	0.16	0.21	0.12	0.17	na
Tb	0.32	0.76	Tb	1.75	2.45	0.04	0.02	0.03	0.12	0.05	na
Dy	1.57	4.14	Dy	11.5	16.4	0.15	0.11	bdl	0.18	0.09	na
Ho	0.33	0.79	Ho	2.35	3.32	0.02	bdl	bdl	0.09	0.26	na
Er	1.16	2.25	Er	6.15	9.67	0.02	0.05	0.05	0.11	0.25	na
Tm	0.19	0.33	Tm	0.85	1.38	bdl	bdl	bdl	bdl	0.12	na
Yb	1.15	2.67	Yb	5.95	10.47	0.03	bdl	bdl	0.11	0.13	na
Lu	0.21	0.56	Lu	0.91	1.79	bdl	bdl	bdl	bdl	0.18	na
ΣREE	22	57	ΣREE	112	120	3.29	2.82	0.81	2.48	5.26	42.75
La _N /Yb _N	1.14	1.87	La _N /Yb _N	1.90	0.81						
Eu/Eu*	0.56	0.53	Eu/Eu*	0.11	0.11	6.6	10.5		6.9		
Ce/Ce*	3.14	1.64	Ce/Ce*	0.61	0.62						

	mineral fraction in the WR [%]	proportional fraction in individual minerals			
		Y [%]	REE [%]	La-Gd [%]	Tb-Lu [%]
Garnet	0.9	83.6	61.0	8.8	85.8
Zircon	0.026	12.5	10.7	10.8	10.6
Muscovite	3.8	0.8	1.6	3.8	0.5
Magnetite	0.9	0.7	1.7	3.5	0.8
Plagioclase	36.7	0.3	1.8	4.9	0.4
K-feldspar	26.8	0.1	0.5	1.2	0.1
Quartz	30.8	0.8	1.8	3.9	0.8
Epidote	0.120	0.0	0.1	0.3	n.d.
LREE phs	0.003	1.1	20.9	62.8	1.0

LREE phs = secondary REE-bearing phases

	average Y+REE in the WR [ppm]	proportional fraction of Y and REE of individual minerals [%]									Total
		Grt	Zrn	Ms	Mag	Pl	Kfs	Qtz	Ep	LREE phs	
La	7.45	0.0	2.9	4.4	5.7	8.1	1.2	4.4	0.9	72.4	100.0
Ce	7.36	0.0	7.0	6.5	2.7	4.4	0.8	3.4	0.2	75.0	100.0
Pr	8.04	0.1	4.3	2.9	3.8	4.1	0.9	4.4	0.3	79.2	100.0
Nd	7.40	0.4	7.4	2.3	3.7	4.8	1.0	6.5	0.2	73.8	100.0
Sm	11.85	8.1	18.3	1.9	3.8	3.1	1.2	3.4	0.1	60.1	100.0
Eu	2.24	3.8	0.0	3.1	2.1	53.4	35.3	2.3	0.0	0.0	100.0
Gd	15.53	42.7	23.6	1.3	2.9	2.4	0.6	1.5	0.0	25.0	100.0
Tb	23.83	73.6	19.2	1.3	2.1	1.5	0.8	1.6	0.0	0.0	100.0
Dy	29.24	79.2	14.2	0.8	1.4	0.8	0.1	0.3	0.0	3.3	100.0
Ho	27.31	82.8	11.9	0.5	0.9	0.5	0.3	3.1	0.0	0.0	100.0
Er	39.28	85.8	10.9	0.5	0.7	0.2	0.1	0.9	0.0	0.9	100.0
Tm	49.27	87.2	9.5	0.4	0.6	0.2	0.1	2.0	0.0	0.0	100.0
Yb	50.94	90.6	8.0	0.3	0.6	0.2	0.0	0.3	0.0	0.0	100.0
Lu	59.31	87.8	9.1	0.3	0.5	0.1	0.0	2.2	0.0	0.0	100.0
Y	44.27	83.6	12.5	0.8	0.7	0.3	0.1	0.8	0.0	1.1	100.0

LREE phs = secondary REE-bearing phases

