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3 **Determination of the concentration of asbestos minerals in highly**  
4 **contaminated mine tailings. The example of the abandoned mine waste**  
5 **of Crètaz and Èmarese (Valle d’Aosta, Italy).**

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## ABSTRACT

25 For the first time, this work reports concentration maps of asbestos minerals in  
26 contaminated mine tailings drawn using the results of Rietveld quantitative phase  
27 analysis (QPA). The investigated sites are located in the Valle d'Aosta region (Italy):  
28 Crètaz, the most important Italian magnetite mine, active until 1979 and Emarèse, one the  
29 most important chrysotile asbestos mines in Italy, active until 1968. The results of the  
30 study permit to draw the spatial distribution of the asbestos (chrysotile and tremolite in  
31 this specific case) concentration, useful to plan reclamation of the sites, with priority  
32 given to the areas with the highest asbestos concentration. Because of the complexity of  
33 the mineral assemblage which includes, among the others, antigorite, chlorite, talc, and  
34 tremolite, the concentration of chrysotile was cross-checked using different experimental  
35 techniques such as X-Ray powder diffraction (XRPD), Fourier Transform Infra-Red  
36 (FTIR) spectroscopy, scanning electron microscopy (SEM), Polarized light Optical  
37 Microscopy (PCOM), and differential thermal analysis (DTA). The accuracy of the  
38 results was validated by analyzing standard samples with known concentrations of  
39 chrysotile and tremolite. The comparison allowed to point out the advantages and  
40 disadvantages of each experimental method.

41 At Crètaz, chrysotile ranges from 4.4 to 22.8 wt% and tremolite from 1.0 to 10.3 wt%  
42 whereas at Emarèse the concentration of chrysotile varies from 3.3 to 39.5 wt% and  
43 tremolite from 5.9 to 12.4 wt%. Antigorite and chlorite are the major accompanying  
44 phases with variable amounts of other accessory minerals including magnetite,  
45 carbonates, talc, olivine, pyroxene, talc, and brucite. The results of our study are of key  
46 importance for the local environmental policies as the knowledge of the spatial

47 distribution of the asbestos concentration allows to plan a detailed reclamation agenda of  
48 the contaminated sites. The spots with the highest surface contamination of both  
49 chrysotile and tremolite were identified and classified as priority area in the reclamation  
50 plan.

51 **Keywords:** Chrysotile, tremolite, serpentine, mine tailings, quantitative determination

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## INTRODUCTION

54 Commercial asbestos minerals are classified into two groups: serpentine and  
55 amphibole asbestos, both sharing the same fibrous-asbestiform crystal habit but different  
56 structural arrangements at the molecular scale (Bailey 1988). The fibrous-asbestiform  
57 variety of serpentine is chrysotile (*white asbestos*) with ideal chemical formula  
58  $Mg_3(OH)_4Si_2O_5$ . The amphibole asbestos family includes five minerals: actinolite,  
59 amosite (fibrous variety of grunerite, *brown asbestos*), anthophyllite, crocidolite (fibrous  
60 variety of riebeckite, *blue asbestos*), and tremolite. Amphiboles' have general formula  
61  $AB_2C_5T_8O_{22}W_2$  with  $A =$ , Ca, K, Li, Na;  $B =$  Ca,  $Fe^{2+}$ , Li, Mg,  $Mn^{2+}$ , Na;  $C =$   $Fe^{2+}$ ,  
62  $Fe^{3+}$ , Li, Mg,  $Mn^{2+}$ ,  $Mn^{3+}$ ,  $Ti^{4+}$ ;  $T =$  Al, Si,  $Ti^{4+}$ ;  $W =$  Cl, F,  $O^{2-}$ , (OH). The ideal  
63 (approximated) formulas of the five asbestos species are: actinolite  $A =$ ,  $B =$  Ca,  $C =$   
64  $(Fe^{2+}, Mg)$ ,  $T =$  Si;  $W =$  (OH); amosite  $A =$ ,  $B = C =$   $Fe^{2+}$ ,  $T =$  Si;  $W =$  (OH);  
65 anthophyllite  $A =$ ,  $B = C =$  Mg,  $T =$  Si;  $W =$  (OH); crocidolite  $A =$ ,  $B =$  Na,  $C =$   
66  $(Fe^{2+}, Fe^{3+}, Mg)$ ,  $T =$  Si;  $W =$  (OH); tremolite  $A =$ ,  $B =$  Ca,  $C =$  Mg,  $T =$  Si;  $W =$  (OH).

67 The six asbestos minerals exhibit outstanding chemical-physical and technological  
68 properties exploited for various industrial applications. Chrysotile has been the most  
69 commonly used form of asbestos. Regrettably, the unique fibrous-asbestiform crystal

70 habit and surface activity, responsible for the excellent technological properties, also  
71 seem to be the cause of asbestos minerals' potential health hazard. The history of the  
72 epidemiological reports of asbestos related diseases is well described in Skinner et al.  
73 (1988). Although many epidemiological studies provided evidence that amphibole  
74 asbestos minerals are more hazardous than chrysotile (Hodgson and Darnton 2000), all  
75 six asbestos mineral species are assumed to be harmful to human health. Exposure  
76 through inhalation of asbestos minerals may provoke lung diseases (Skinner et al. 1988;  
77 Dilek and Newcomb 2003) and during the 1980s, asbestos minerals were declared proven  
78 human carcinogens by the US Environmental Protection Agency, the International  
79 Agency for Research on cancer (IARC) of the World Health Organization and the  
80 National Toxicology Program (Nicholson 1986; IARC 1977; Collegium Ramazzini  
81 2010). Later, many countries worldwide banned or restricted the use of asbestos  
82 containing materials (ACMs). In the countries where asbestos minerals are banned,  
83 ACMs have been progressively removed from the environment to minimize exposure of  
84 the population (Gualtieri 2012). Despite the huge efforts to entirely remove asbestos from  
85 the living environment, some issues are still matter of concern. One of these regards the  
86 correct evaluation of the health hazard associated with abandoned asbestos mines  
87 (especially dumps and tailings) and natural occurring asbestos (NOA). Mining of  
88 asbestos and asbestos associated minerals (e.g., iron oxides) generated vast amounts of  
89 residue material, chemically not very different from the original rock (Meyer 1980). An  
90 important difference is the fineness of the residue material composing the dumps, making  
91 it more prone to weathering or erosion by wind, with subsequent release of fibers in air or  
92 in the percolating surface hydrological system. With this premise, Viti et al. (2011)

93 suggested that the correct evaluation of the hazard associated with serpentinite outcrops  
94 (and asbestos containing mine tailings) would require a careful geo-statistical and geo-  
95 mechanical investigation of the fracturing/vein system, followed by careful qualitative  
96 and quantitative determinations. Unfortunately, the accomplishment of accurate and  
97 precise quantitative figures of the actual asbestos minerals content is made difficult by  
98 two major factors: (1) asbestos bearing serpentinite textures typically consist of fine-to-  
99 ultrafine intergrowths of fibrous and non-fibrous minerals, often difficult to identify by  
100 conventional methods, such as X-ray diffraction or microanalytical approaches. To this  
101 aim, DTA seems to be an effective and promising tool of analysis although only a case  
102 study has been reported so far (Viti et al. 2011) and further experimental evidence is  
103 needed; (2) asbestos and asbestos containing mine dumps and tailings invariably contain  
104 high concentrations of asbestos minerals (generally higher than 1wt%). In countries like  
105 Italy, regulated experimental techniques for massive materials (X-Ray Powder  
106 Diffraction - XRPD, Fourier Transform Infra-red spectroscopy - FTIR, Scanning Electron  
107 Microscopy - SEM, and Phase Contrast Optical Microscopy - PCOM) are applied to  
108 determine whether asbestos concentration is higher than 0.1 wt% (Italian D.Lgs  
109 10/03/2010 nr. 205 All. D). On the same line, Airborne Toxic Control Measure (ATCM)  
110 restricted in the USA the use of serpentine and ultramafic rock aggregates for surface  
111 applications to materials containing less than 0.25 wt% asbestos determined according to  
112 the Californian Air Resources Board method (CARB method 435, 1991). This scenario  
113 reveals that accuracy of these quantitative methods have not been tested for highly  
114 contaminated massive materials.

115 This work was prompted by the actual need to map the concentration of asbestos

116 minerals in abandoned asbestos or asbestos containing open mines with dumps and  
117 tailings possibly containing very high concentrations of asbestos fibers (well above 1  
118 wt% ). Because the costly reclamation of these sites usually conflicts with the tight  
119 budgets of the regional/national funds allocated to such expense items, a step by step long  
120 term plan must be proposed with priority given to the reclamation of the areas with the  
121 highest asbestos concentration. Therefore, the knowledge of the spatial distribution of the  
122 asbestos concentration in such sites is relevant from an economic standpoint as it may  
123 permit to prioritize the most hazardous areas within a long term reclamation plan. For the  
124 first time, we report the mapping of the concentration of asbestos minerals (chrysotile and  
125 tremolite) in abandoned sites with highly contaminated mine tailings. The sites under  
126 investigation are located in the Valle d'Aosta region in Italy: Crètaz is situated near the  
127 town of Cogne where the most important Italian magnetite mine was active until 1979;  
128 Emarèse was one the most important chrysotile asbestos mines in Italy, now abandoned  
129 since 1968.

130 The determination of the chrysotile concentration was accomplished using all the  
131 experimental techniques permitted by current Italian regulations for massive materials  
132 XRPD, FTIR, SEM, and PCOM. Quantitative PCOM results were integrated by  
133 qualitative observations in polarized light (PLOM) and with chromatic dispersion (CD).  
134 Although the Italian D.M. 09/06/1994 suggests the use of SEM when the estimated  
135 asbestos concentration is lower than 1wt% and MOCF, XRPD and FTIR when it is  
136 higher than 1wt%, there is actually no indications on the upper concentration limits of  
137 application of SEM. Hence, the analysis of samples with very high concentration of  
138 asbestos offers a proper case study. The accuracy of the results was validated by the

139 analysis of especially prepared standard samples with known concentration of chrysotile  
140 and tremolite and cross-checked by the application of the Differential Thermal Analysis  
141 (DTA) and the Rietveld methods. The comparison allowed to point out the advantages  
142 and disadvantages of each experimental method.

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#### **GEOLOGY OF THE INVESTIGATED AREA**

145 The main structural lines of the Aosta valley (Fig. 1) are aligned N-W to S-E. They  
146 represent a natural cross-section through the structural units of the Italian Alps where the  
147 ancient European and African continental margins are divided by relict portions of the  
148 oceanic floor. The margin of the European basement emerges in correspondence to the  
149 crystalline massif externally represented by the Monte Bianco (Elvetico - Ultraelvetico  
150 Domain) and internally by the Gran San Bernardo – Ruitor, the Monte Rosa and the Gran  
151 Paradiso (Pennidico Domain). The ancient portion of the Piedmont ocean (the Piedmont  
152 Zone with Calc-schists and Green Stones) occurs in the middle stretch of the Valley,  
153 spanning the inner margin of the crystalline massifs. The area of the lowest valley is  
154 mainly composed by the Pennidic and Austroalpine domains, characterized by folded  
155 structures and marked by metamorphism in association with large thrust of the Piedmont  
156 Zone of Calc-schists with Green Stones (ophiolites). The mining area of Cogne is found  
157 in that formation and emerges in the localities of Aymavilles, Urtier river, and Grauson  
158 river (Elter 1987). The rocks of that unit are known to host mineralizations of Cr, Pt, Ni  
159 and Fe and various phases such as asbestos minerals and talc. Iron is mostly hosted in  
160 magnetite, one of the main constituents of the serpentinite accessories, with the genesis  
161 attributed to the processes of serpentinization (hydrothermal alteration of the original

162 mantle peridotites) which began probably in the oceanic environment and continued  
163 during the Alpine orogeny (Dal Piaz 1971). The magnetite mines of Cogne, and in  
164 particular the site of Crètaz, are located inside the area of the Piedmont Calc-schists and  
165 Green Stones (Diella et al. 1994). The ore mining of Cogne was both underground and  
166 open (Di Colbertaldo et. al. 1967) and the iron ore has been industrially exploited from  
167 1900 to 1979.

168 The asbestos Émarèse mine is located inside the Piedmont Zone with Calc-schists and  
169 Greenstones (ophiolites). The metabasites which occur at the Émarèse area are mainly  
170 composed of serpentinites locally mineralized to antigorite and chrysotile. The latter  
171 sometimes occurs in very long fibers that can reach a length of 1 m. Serpentine are  
172 accompanied by carbonates, magnetite and talc. The mining activity regarded an open pit  
173 and several wells and tunnels (Cavinato 1964). The Émarèse mine is abandoned since  
174 1968.

175

## 176 **EXPERIMENTAL PROCEDURE**

### 177 **Materials sampling and sample preparation**

178 A number of representative samples were collected in the areas of Crètaz and  
179 Emarèse. Given the extremely hazardous situation due to the proximity of the  
180 contaminated site with the touristic town of Cogne, both surface and core sampling was  
181 conducted at Crètaz whereas only surface sampling was conducted at Emarèse. As far as  
182 the surface sampling is concerned, the materials were collected under a 25-30 mm thick  
183 surface layer of soil manually removed with the aid of a shovel. All the operations were  
184 conducted using protective equipments. Samples were collected down to a depth of about

185 60 mm with the aid of a manual corer. A representative amount of material, depending on  
186 the average grain size, was sampled and manually divided with the aid of a Retsch  
187 stainless steel sample splitter (maximum capacity = 16 l; maximum grain size = 50 mm)  
188 to obtain about 2 Kg of homogeneous final raw product that was finally stored inside  
189 glass bottles and sealed. At Crêtaz, core sampling was possible using a rotary core drill  
190 with a penetration length of about 20 m. Raw core samples were divided and  
191 homogenized using the same procedure described above for surface samples. All the  
192 surface and core collection spots for both localities are reported in Figure 1. The figure  
193 legend also specifies the sampling mode.

194 In the lab, using protective equipments and under extractor fan, raw samples were  
195 quartered to obtained about 50 g of homogeneous representative material and  
196 subsequently dried at 105 °C for 2 h. Powders were obtained by mechanical milling using  
197 Retsch mm 200 SiC jars, with a capacity of 10 ml each, for 10 min at a speed of 25  
198 oscillations/s. Further fine manual powdering was conducted using an agate mortar for 5  
199 min.

200 Two standard samples (labeled STD1 and STD2) with known contents of chrysotile  
201 and tremolite asbestos and mineralogical composition similar to that of the natural  
202 samples were prepared as weighed mixtures of artificial tailings to assess the accuracy of  
203 the quantitative determination. The compositions of the standard samples (wt%) were:  
204 STD1 (chrysotile 5.11, tremolite 5.11, talc 5.11, clinocllore 5.11, antigorite 8.07, calcite  
205 40.86, dolomite 10.2, magnesite 4.09, albite 16.34); STD2 (chrysotile 10.44, tremolite  
206 1.04, talc 5.22, clinocllore 5.22, antigorite 16.49, calcite 36.53, dolomite 10.44,  
207 magnesite 4.18, albite 10.44). Specimens of the mineral phases talc, clinocllore, calcite,

208 dolomite, magnesite, and albite were taken from the mineral collection of the MUSEO  
209 GEMMA 1786, Dipartimento di Scienze Chimiche e Geologiche of the University of  
210 Modena e Reggio Emilia (Modena, Italy).

211 The asbestos samples used to prepare the weighed mixtures of artificial tailings are the  
212 NIST chrysotile standard SRM1866a and the NIST tremolite sample SRM 1867a, the  
213 highest purity available. Bunches of fibers were ultrasonicated and selected under optical  
214 microscope. The antigorite sample is from a pale-green, splintery vein occurring in a  
215 massive serpentinite from Elba Island (Italy): sample ATG18 in Viti and Mellini (1996).  
216 The sample predominantly consists of antigorite lamellae with an average composition of  
217  $\text{Mg}_{2.62}\text{Fe}^{2+}_{0.16}\text{Fe}^{3+}_{0.03}(\text{OH})_4\text{Al}_{0.01}\text{Si}_2\text{O}_5$  (normalized to 2 apfu in the tetrahedral site; Loss  
218 of Ignition (L.O.I.) of 11.9 wt%) and super-periodicity of 48.8 Å; interstitial chrysotile  
219 occurs among antigorite lamellae. The total chrysotile content is 22.0(9) wt% (Viti et al.  
220 2011). Obviously, the chrysotile content of antigorite was considered during the  
221 preparation of the mixture. Sample purity of all the selected specimens was checked by  
222 XRPD. Prior to the weighing procedure, all the specimens were dried at 105 °C for 2 h  
223 and reduced to powders by manual grinding in an agate mortar using acetone as solvent.  
224 After weighing the various components, the coarse powders were homogenized and  
225 powdered in an agate mortar for 5 min.

226

## 227 **Analytical methods**

### 228 *Sample selection*

229 Two natural samples (C2 and E1) and the two standards STD1 and STD2 were  
230 selected for the determination of the concentration of chrysotile and tremolite (when

231 possible) asbestos, following the experimental methods considered in the Italian D.M.  
232 09/06/1994, differential thermal analysis (DTA) and X-ray powder diffraction (XRPD)  
233 with the Rietveld method. All the other samples collected during the surface and core  
234 sampling at Crètaz and Èmarese were analyzed with the Rietveld method to determine  
235 both the concentration of chrysotile and tremolite asbestos.

### 236 *Optical microscopy*

237 As far as the optical microscopy observations are concerned, prior to the qualitative  
238 analysis, the samples underwent a chemical attack using a very diluted HCl solution to  
239 remove the carbonate fraction and eventually hydrated carbonate alteration products and  
240 release the fibers from the matrix. About 0.2 mg of the powder was then suspended in a  
241 drop of acetone on a slide, put in a bottle, and inserted in an ultrasonic bath for 15 min. A  
242 liquid of proper refraction index is added to the powder on the slide for the polarized light  
243 optical microscopy (PCOM) observations. The following refractive indices were used for  
244 the identification of the asbestos fibres in the samples: 1.550 for chrysotile; 1.610-1.615  
245 for tremolite; 1.620 for anthophyllite, and 1.640 for actinolite. The microscopes used for  
246 the qualitative observations were a Leica DMLP and Leica DIALUX 20.

247 The samples for the quantitative analyses were prepared by suspending 0.1 mg of  
248 powder in 10 ml of distilled water. The suspension was ultrasonicated for 15 min. The  
249 suspension was filtered using a 47 mm cellulose nitrate filter to obtain an homogeneous  
250 dispersion. The filter was subsequently diaphanized for the PCOM observation. The  
251 observations were performed at 500x over a 1.57 mm<sup>2</sup> area of the membrane (0.5% of the  
252 total membrane area of 152 mm<sup>2</sup>). Each regulated (length > 5 µm and aspect ratio of 3)  
253 fiber of both chrysotile and tremolite asbestos observed on the membrane was reported

254 and summed up to calculate the total volume. Assuming theoretical density values of  
255 crysotile and tremolite fibers of 2.6 g/cm<sup>3</sup> and 3.0 g/cm<sup>3</sup>, respectively, it was possible to  
256 calculate the total weight of both asbestos species counted on the membrane. This value  
257 was then rescaled to the entire powdered area of the membrane and therefore to the  
258 amount (0.1 mg) of powder on it.

### 259 *Electron microscopy*

260 The scanning electron microscopy (SEM) determinations were performed following  
261 the method described in the Italian legislation (Italian Ministry of Health, 1994), which is  
262 very similar to the International Standard Organization (ISO) Method 14966 (ISO, 2002).  
263 The fiber-definition criteria are those cited above and are compliant with the ISO 14966  
264 and WHO standard (WHO 1997; ISO 2002). An amount of 10 mg of each powder sample  
265 was suspended in 200 ml of deionised and micro-filtered water with 0.1 vol.% surfactant  
266 additive (dioctyl sodium sulfocinate, C<sub>20</sub>H<sub>37</sub>NaO<sub>7</sub>S, CAS nr. 577-11-7), and  
267 ultrasonicated to facilitate the particle dispersion. A volume of 3 x 2.5 ml of this  
268 suspension was collected at different levels and put into a filtering system, allowing  
269 random deposition of the particles on polycarbonate filters (Osmonics 25 mm diameter,  
270 0.6 µm porosity). Filters were dried in an oven (at 55 °C), weighed using a high-precision  
271 balance and coated with graphite. The samples were analyzed with a Vega TS Tescan  
272 5163 XM SEM (LAB1, the University of Milan-Bicocca), in combination with an  
273 energy-dispersive X-ray spectrometer (EDAX Genesis 400), operating at 20 kV and 200  
274 pA, at a working distance of about 20 mm. For each sample, 1 mm<sup>2</sup> of the filter surface  
275 was investigated, working at 2000–6000 magnification. The technique is based on point-  
276 counting statistics, reporting the occurrence, number, dimension and chemical

277 composition of asbestos fibers in each measured spot. The volume of the single fiber is  
278 approximated to that of a cylinder, and the weight is calculated assuming an average  
279 density of 2.6 g/cm<sup>3</sup> for chrysotile and 3.0 g/cm<sup>3</sup> for all amphibole species. The fiber-  
280 concentration C (in ppm) was then calculated as follows:

$$281 \quad C = [A \cdot (w_c + w_a) / n \cdot a \cdot W] \cdot 10^6$$

282 with A = filter surface (mm<sup>2</sup>); w<sub>c</sub> = total weight of counted chrysotile fibers (mg); w<sub>a</sub> =  
283 total weight of counted amphibole fibers (mg); n = number of analyzed spots; a = spot  
284 surface (mm<sup>2</sup>); W = weight of the sample on the filter (mg).

285 The samples (following the protocol of LAB1) were also prepared separately and  
286 analyzed in another laboratory (LAB2, A.R.P.A. Aosta). An amount of 10 mg of each  
287 powder was suspended in 200 ml of deionised and micro-filtered water with 0.2 vol.%  
288 surfactant additive (sodium lauryl sulfate, CH<sub>3</sub>(CH<sub>2</sub>)<sub>11</sub>OSO<sub>3</sub>Na, CAS nr. 151-21-3), and  
289 ultrasonically treated. All the other experimental conditions were identical to those  
290 reported above. The instrument used for the observations was a Zeiss EVO MA10 with a  
291 LaB6 cathode, in combination with an EDAX system, operating at 20 kV and 200 pA,  
292 with 500–16000 magnification, at a working distance of 12 mm.

### 293 *Infrared spectroscopy*

294 For the Fourier transform Infrared (FTIR) spectroscopy measurements, pellets were  
295 prepared using 198 mg of dried KBr powder and 2 mg of dried sample powder, with a  
296 final load of 12 t for 12 s. A blank standard sample composed of KBr powder was also  
297 prepared. Prior to the analysis, a calibration curve with increasing amounts of chrysotile  
298 was prepared using the same experimental conditions to apply the Linear Calibration  
299 Curve Method (see a through description of the method applied to FTIR in De Stefano et

300 al. 2012). All the spectra were collected from 4000 to 600  $\text{cm}^{-1}$  at 4  $\text{cm}^{-1}$  resolution using  
301 a Bruker VECTOR 22 spectrometer with a Globar source, aperture 3 mm, 32 scans,  
302 velocity 10 kHz, detector DTGS with a window of KBr (8000–400  $\text{cm}^{-1}$ ), Mertz  
303 correction phase, apodization function 3-term Blackman-Harris. Data were collected and  
304 analyzed using the Bruker OPUS 6.5 software. The intensity of the absorption band at  
305 3697  $\text{cm}^{-1}$ , the stretching vibration of chrysotile non-hydrogen-bonded  $\text{OH}^-$  (Farmer  
306 1974), was measured and plotted versus the added chrysotile weight percent. It is  
307 assumed that the detection limit is 0.01 wt% (Foresti et al. 2003). This procedure is rather  
308 straightforward when no interference phases such as serpentine polymorphs and chlorite  
309 are present (see for example, Balducci and Valerio 1986). The presence in the  
310 investigated samples of the layer silicates antigorite and chlorite, which may overlap their  
311 absorption bands to that of chrysotile, requires a careful discrimination to obtain the  
312 intensity of the chrysotile band to plot vs. chrysotile concentration. For the investigated  
313 samples with unknown concentration of chrysotile, the intensity of the absorption band  
314 was measured as closest as possible to the ideal value of 3697  $\text{cm}^{-1}$  (Farmer 1974), using  
315 the “R” type integration process available in the software (the value of the intensity  
316 measured above the base line taken between 3720 e 3661  $\text{cm}^{-1}$ ). The resulting values  
317 were then used to extract the chrysotile concentraion from the calibration curve.

#### 318 *X-ray powder diffraction*

319 The XRPD quantitative analysis of chrysotile asbestos has been long conducted using  
320 the external standard method (Klug and Alexander 1974) where the peak  
321 intensity/integrated area of the major diffraction peak of an analyte (001 peak for  
322 chrysotile) is usually plotted vs. its concentration to build a calibration curve. The

323 concentration of the analyte in an unknown sample requires the determination of the  
324 correct unbiased intensity/integrated area of the major diffraction peak. Puledda and  
325 Marconi (1990) developed a method used for chrysotile legitimated by the Italian D.M.  
326 09/06/1994 which is based on the use of a silver membrane filter. The method also  
327 proposes an appropriate value for the deposition area of the sample on filter, which  
328 optimizes the diffraction response of the analyte. To apply the method described above,  
329 the powder patterns of the samples C2, E1 and the two standards STD1 and STD2 were  
330 collected using a Bragg–Brentano  $\theta$ - $\theta$  diffractometer (PANalytical X'Pert Pro), Cu  $K\alpha$   
331 radiation, 40 kV and 40 mA) equipped with a curved graphite monochromator and a gas  
332 proportional detector. The divergence, receiving and anti-scattering slits were of  $1^\circ$ , 0.1  
333 mm, and  $1^\circ$ , respectively. The scans were collected with a step scan of  $0.02^\circ 2\theta$ , and 5  
334 s/step in the range  $4$ - $64^\circ 2\theta$ . The data analysis was performed using the PANalytical  
335 HighScore Plus software version 2.2c.

336 The mineralogical quantitative phase analyses (QPA) of all the Crèta and Èmarese  
337 samples and the two standard samples STD1 and STD2 using the Rietveld method  
338 (Rietveld 1969) were performed through X-ray powder diffraction (XRPD). Data were  
339 collected using a Bragg–Brentano  $\theta$ - $2\theta$  diffractometer (Bruker D8 Advance, Cu  $K\alpha$   
340 radiation, 40 kV and 40 mA) equipped with a Göbel mirror on the incident beam and a  
341 solid state detector. The width of the divergence, receiving and anti-scattering slits were  
342 of 0.6, 0.1, and 0.2 mm, respectively. The scans were collected with a step scan of  $0.02^\circ 2\theta$   
343 and 20 s/step in the range  $4$ - $65^\circ 2\theta$ . QPAs were performed following the refinement  
344 strategies described in Gualtieri (2000). Refinements were accomplished with the GSAS  
345 (Larson and Von Dreele 1999) package and its graphical interface EXPGUI (Toby 2001).

346 Whenever the samples showed the presence of more than the maximum number of 9  
347 phases available to the program, the procedure described by Winbur et al. (2000) was  
348 used. Although samples were prepared with the side loading technique to minimize *a*  
349 *priori* preferred orientation of crystallites, residual preferred orientation effects were  
350 modelled with the March model (Dollase 1986). The following structural models were  
351 used: albite (Downs et al. 1994) and plagioclase (Wenk et al. 1980), antigorite (Uehara  
352 1998), calcite, dolomite, quartz (Le Saoût et al. 2011), chrysotile (Falini et al. 2004),  
353 chlinochlore (Zanazzi et al. 2009), illite (Birle 1968), magnetite (Haavik et al. 2000),  
354 microcline (Ribbe 1979) and orthoclase (Prince et al. 1973), talc (Perdikatsis and Burzclaff  
355 1981), and tremolite (Ungaretti and Oberti 2000).

#### 356 *Thermal analysis*

357 The collection of the thermal analysis curves thermogravimetry (TG), differential  
358 thermogravimetry (DTG), and DTA of the C2, E1, and the two standard samples STD1  
359 and STD2 samples were performed using a simultaneous differential thermal analysis  
360 (SDTA) SEIKO SSC/5200 SII. Data were collected in air with a flow rate of 2 µl/min, in  
361 the range 20–1000 °C, and heating rate of 10 °C/min. Instrumental precision was checked  
362 by repeated collections on reference samples, revealing good reproducibility  
363 (instrumental theoretical precision of ±0.5 °C); theoretical weight sensitivity is down to  
364 0.1 µg. The DTA traces of the reference samples are reported in Viti (2010). The DTA  
365 peak deconvolution has been performed following the method described in Viti et al.  
366 (2011). Fitting of the DTA peaks was performed using the deconvolution process avail-  
367 able in PeakFit4 (PeakFit Version 4.12, SPSS Inc., AISN Software). The temperature  
368 range selected for data analysis is 550–800 °C, that is the main dehydroxylation range, in

369 the attempt to avoid possible interferences with the 820 °C exothermic signals (Viti  
370 2010). The Loess algorithm was used to smooth the data and to remove local  
371 perturbations with a ratio of 5%. After that, baseline subtraction was accomplished by a  
372 second derivative test procedure.

373

374 Although it is well known that both transmission electron microscopy (TEM) and  
375 Raman are powerful diagnostic techniques for the study of asbestos containing materials,  
376 the two techniques have not been used in this work where the preference was given to the  
377 custom methods specifically developed for the determination of the concentration of  
378 asbestos minerals in massive materials.

379

380

## RESULTS

381 This section is divided into two parts. The first part summarizes the results of each  
382 experimental technique applied to the determination of the chrysotile asbestos content in  
383 the natural samples C2 and E1, and the two especially prepared standards STD1 and  
384 STD2. The second part describes the results of the quantitative phase analysis of the  
385 samples collected at Crêtaz and Emarèse sites.

386

### Optical observations

388 The application of the PCOM technique to the study of our samples was of great help  
389 from a qualitative point of view as it easily permitted to identify the nature of fiber  
390 bunches thanks to the chromatic dispersion (CD) of the fibers (McCrone 1987; NIOSH  
391 1989; EPA 1993; Moss 1994; Bellopede et al. 2009; Cavariani et al. 2010). Figure 2a,b

392 shows beautiful fiber aggregates of chrysotile and tremolite. From a quantitative  
393 standpoint, because the fiber counting is usually conducted at a low magnification  
394 (generally from 40 to 500x), only fibers or fiber aggregates of considerable size can be  
395 observed and counted, and it is sometimes difficult to distinguish between an asbestos  
396 fiber and crystal of different nature with acicular crystal habit (Fig. 2c). As a matter of  
397 fact, fibers with diameter smaller than 0.8-1  $\mu\text{m}$  are not visible with optical microscopy at  
398 500x with a resolution power of 0.2  $\mu\text{m}$  (Cavariani et al. 2010).

399

#### 400 **SEM analyses**

401 The SEM determinations allowed to detect even very small fibers (with a length  
402 shorter than 1-2  $\mu\text{m}$ ) invisible under optical microscope. We have observed that the  
403 sample preparation is crucial and a correct fiber counting is possible only if the stub is not  
404 overloaded, that is the case when particle aggregates are formed (see an example in Fig.  
405 3a). Small fibers are frequent in the standard samples STD1 and STD2 which were  
406 prepared using dry powders. Figure 3b,c reports selected SEM images of the investigated  
407 samples showing the presence of fibers with heterogeneous size of both chrysotile and  
408 tremolite asbestos fibers. The use of the EDS spot analysis is essential for the correct  
409 identification and counting of the chrysotile and tremolite asbestos fibers (Fig. 3c with  
410 the relative EDS analysis in the white box). The use of this technique also permits to  
411 detect fiber aggregates with distinctive habit and texture and consequently hard to  
412 classify otherwise. One such example is portrayed in Figure 3d where an intergrowth of  
413 chrysotile-antigorite aggregate is likely observed. Such aggregates may witness low  
414 grade regional progressive metamorphism of serpentinite which involves formation of

415 antigorite at expenses of chrysotile (Mellini et al. 1987). This paragenesis has already  
416 been reported for the Val Malenco serpentinites in the Italian Alps where chrysotile has  
417 been identified as antigorite precursor (Mellini et al. 1987; Wicks and O'Hanley 1988).  
418 Notwithstanding, the study of the serpentinite texture and its origin is out of the aims of  
419 this paper and should be accomplished using TEM.

420

#### 421 **FTIR data**

422 Figure 4 portrays the selected region between 3400 and 3950  $\text{cm}^{-1}$  of the FTIR spectra  
423 of the four investigated samples, where the absorption bands of chrysotile and  
424 interference layer silicates occur. The calculated absorbance of the stretching vibration of  
425 chrysotile and the resulting chrysotile content (wt%), using the calibration curve with  
426 equation  $y = 0.78269x$  with  $y = \text{absorbance (\%)}$  and  $x = \text{chrysotile weight fraction}$ , are  
427 reported in Table 1.

428

#### 429 **XRPD and the Rietveld method**

430 Because the quantitative analysis of chrysotile asbestos using XRPD with the external  
431 standard method (Klug and Alexander 1974) requires the determination of the peak  
432 intensity or the integrated area of the major diffraction peak of the (001) peak of  
433 chrysotile in the unknown sample, the severe peak overlap due to the interference of both  
434 antigorite and chlorite, as observed the investigated samples makes this method  
435 impracticable. The application of blind single peak fitting procedures to deconvolute each  
436 single contribution resulted in unreliable outcomes. For this reason, the use of this  
437 method for the determination of the chrysotile asbestos concentration in these matrices

438 turns out to be unfeasible. On the other hand, if a more robust fitting procedure with  
439 independent observations and (crystallographic) constraints is applied, deconvolution of  
440 each contribution seemed viable. As a matter of fact, the application of the Rietveld  
441 method guarantees both the use of independent observations (it is a full profile fitting  
442 procedure where all the peaks of each phase in the system are modeled) and robust  
443 crystallographic constraints (symmetry, unit cell, profile coefficients, background and  
444 others). Accuracy of estimates for multiple serpentine minerals can be achieved because  
445 all independent reflections belonging to the different serpentine polymorphs in the  
446 collected range can be fitted during the refinement procedure. To this aim, areas where  
447 overlap of the peaks of chrysotile and antigorite is limited, with distinct observations that  
448 help constrain refinement of these phases, exists: see for example the major  $(13 \bar{1})$  peak  
449 of antigorite at 2.52 Å which is well separated from chrysotile reflections  $(\bar{2} 0 2)$  and  $(2$   
450  $0 2)$ , at 2.45 Å and 2.55 Å, respectively. In addition, the fit of a number of peaks of each  
451 phase in the systems makes it possible to correct *a posteriori* any preferred orientation  
452 effect, if present, which may bias the major intensities of chrysotile, antigorite and  
453 chlorite. The result is a very good fit which allows the calculation of the weight estimates  
454 of each phase in the system including chrysotile (see the example of sample E1 in Fig 5).

455 As far as the four analyzed samples, agreement factors of the Rietveld refinements  
456 (Larson and Von Dreele 1999) were in the range:  $R_{wp} = 9.44-10.33\%$ ,  $R_p = 7.07-8.02\%$ ,  
457  $\chi^2 = 2.709-5.397$ . The Rietveld graphical outputs are shown in Figure 6 and Table 2  
458 reports the agreement factors and quantitative phase composition.

459

460 **DTA analyses**

461 Figure 7a-d shows the DTA endothermic signals of the four investigated samples and  
462 standard pure chrysotile, due to serpentine dehydroxylation processes in the range 550–  
463 850 °C and relative result of the fitting procedure. DTA signals were fit using two peaks  
464 for chrysotile (dehydroxylation I and II, weak and strong, respectively) and one peak for  
465 antigorite (I in Viti et al. 2011). Figure 7e shows the DTA trace and result of the fitting  
466 procedure for the pure chrysotile sample. Only fits with a regression coefficient  $R^2 \geq$   
467 0.990 were considered acceptable. The calculated peak areas were reduced to area ratios  
468 (ratio between the calculated area in the unknown and that of the pure chrysotile  
469 standard) and plotted in the curve obtained by Viti et al. (2011) for antigorite+chrysotile  
470 mixtures with a calculated equation of  $y = 0.0108x - 0.016$ . The results of the procedure  
471 are reported in Table 3.

472

### 473 **Mineralogical composition and asbestos concentration of the Crètaz and Emarèse** 474 **samples**

475 Agreement factors of the Rietveld refinements (Larson and Von Dreele 1999) were in  
476 the range  $R_{wp} = 6.46$ -11.32%,  $R_p = 4.08$ -8.01%,  $\chi^2 = 3.437$ –8.639. As an example, the  
477 Rietveld graphical output of sample C5 is reported in Figure 8. It should be noted the low  
478 peak to background ratio observed in a few samples was due to the fluorescent scattering  
479 induced by the large iron content (large magnetite amounts) which excites the copper  
480 atoms of the anticathode. Table 4 reports all the quantitative phase compositions of the  
481 investigated samples. Because of the low content of magnetite determined in the  
482 investigated samples, the microabsorption effects are presumably negligible and no  
483 Brindley (1945) correction was applied during the refinement procedure.

484

485

## DISCUSSION

### 486 **Accuracy of the results from the various experimental techniques**

487 The accuracy of the concentration figures of chrysotile asbestos in the abandoned sites  
488 of Crêtaz and Emarèse was assessed for two selected samples C2 and E1 and the two  
489 especially prepared standards STD1 and STD2, with known asbestos concentration, using  
490 the experimental techniques permitted by the current Italian regulation for massive  
491 materials (XRPD, FTIR, SEM, and PCOM), the Rietveld method and DTA. Table 5  
492 summarizes the resulting calculated concentration of chrysotile asbestos in the four  
493 samples.

#### 494 *Optical microscopy*

495 Although the application of the PCOM technique was qualitatively very useful, the  
496 figures obtained with this technique are underestimated with respect to the values  
497 provided by the application of the other experimental methods (Table 5): an estimate of  
498 3.7 wt% was calculated for STD1 sample against an expect value of 5.11 wt% and 1.4  
499 wt% for STD2 sample against an expected value of 10.44 wt%. It is possible that limited  
500 resolution power did not allow to reckon and quantify small fibers or fiber aggregates  
501 predominantly present in the fine powder standards. Analogous conclusions can be found  
502 in the literature for the analysis of both airborne and massive sample. As an example, for  
503 airborne asbestos fibers in steam tunnels, Dufresne et al. (2002) reported that  
504 concentrations found by the TEM/EDS method were higher than those determined by  
505 PCOM, especially when chrysotile fibers were present, probably because the TEM/EDS  
506 method has a higher resolution than PCOM. Airborne asbestos fibers were present in all

507 the steam tunnels and were barely detectable by the PCOM technique only. Our work  
508 confirms the limits of the optical methods and it is not surprising that in many countries  
509 optical microscopy was adopted only for the visual qualitative identification of asbestos  
510 fibers (see for example, EPA 1989; OSHA 1994; HSE 1994).

511 *Electron microscopy*

512 The chrysotile concentrations obtained from the SEM LAB1 analyses are fairly more  
513 accurate than the figures determined from the optical observations. On the other hand, the  
514 concentrations provided by SEM LAB2 evidence disagreement (see Table 5). The values  
515 provided by LAB1 laboratory are accurate and in agreement with the expected values: an  
516 estimate of 4.6 wt% was calculated for STD1 sample against an expect value of 5.11 wt%  
517 and 13.1 wt% for STD2 sample against an expected value of 10.44 wt%. Both calculated  
518 weights of natural samples C2 and E1 are also in agreement with the values provided by  
519 other methods. The values provided by the SEM LAB2 laboratory are largely  
520 underestimated as far as the standards are concerned.

521 Due to the higher resolution power and larger field depth, SEM allows to detect small  
522 fibers invisible using the optical microscope. The standards were prepared using  
523 chrysotile powders with fine chrysotile fibers hard to detect with an optical microscope  
524 even at 500x but clearly visible at 2000x with an electron microscope. This may explain  
525 the underestimation in the chrysotile concentration determined with the PCOM  
526 technique. The difference in the performance of the two microscopic techniques in favor  
527 of the SEM method have already been reported by Cavariani et al. (2010). Cavallo and  
528 Rimoldi (2013) who studied similar matrices (asbestos serpentinite quarries in  
529 Valmalenco, Central Alps, Northern Italy) reported that the small thickness of chrysotile

530 fibrils produced during quarrying activities, and the abundance of pseudo-fibrous  
531 antigorite cleavage fragments proved the SEM-EDS analytical procedure to be the most  
532 suitable.

533 Concerning the results provide by LAB2, Davies et al. (1996) have previously  
534 reported that the method for quantitative analysis of asbestos with SEM for both airborne  
535 and massive samples presents difficulties, primarily because of the complexity in  
536 standardizing the many operating parameters that are controlled. Although some modern  
537 SEMs are capable of resolving surface features at the angstrom level, there is still much  
538 difficulty in obtaining adequate images of unit fibrils. Such images are necessary to  
539 perform quantitative analyses. Moreover, even though routine SEM/EDS analysis allows  
540 evaluation of sample morphology and provides semi-quantitative chemical data, it does  
541 not allow determination of crystal structure like TEM. Although TEM is a very time-  
542 consuming method, seldom providing concentrations that can be considered  
543 representative of the whole sample, it is well known that its use makes it possible to  
544 resolve many of the ambiguities that lead to incorrect fiber concentration (Beaman and  
545 File 1976; Cattaneo et al. 2012).

546 The underestimation of the fiber concentration from LAB2 may be explained by  
547 sample powder heterogeneity. We have observed ball of thread like aggregates of  
548 chrysotile fibers within the powders (see Fig. 3e) of the samples prepared in LAB2  
549 determining areas with a great concentration of asbestos fibers and areas with no fibers at  
550 all. A possible explanation is the use of the surfactant sodium lauryl sulfate for the  
551 preparation of the suspension in place of dioctyl sodium sulfocinate. Sodium lauryl  
552 sulfate rises the pH of aqueous solution to about 9.5 whereas dioctyl sodium sulfocinate

553 stabilizes pH values of the aqueous solution to about 6.2. In distilled water, the zeta  
554 potential of chrysotile displays strongly positive values (about +40 mV) at 6.2 and  
555 decreases down to about 20 mV at pH = 9.5 (Light and Wei 1977). The decrease of the  
556 zeta potential towards the isoelectric point determines instability of the suspension  
557 because the particles with low zeta potential have no force to prevent them coming  
558 together and flocculating. Jolicouer et al. (1981) used these principles to retard  
559 sedimentation of asbestos fibers in NaCl solutions. Hence, the flocculating chrysotile  
560 fibers tend to lump together to form the observed ball of thread like aggregate. On a  
561 speculative level, it is not possible to rule out that aggregation also occurs in the dry  
562 powders, especially those that have undergone physical long-distance transportation and  
563 mechanical shaking. In both situations, fiber segregation may bias the sample  
564 homogeneity and the representativeness of the microscopic observation usually  
565 performed using a tiny amount of powder.

566 It was already said that the Italian D.M. 09/06/1994 advises the use of SEM when the  
567 estimated asbestos concentration is lower than 1wt% giving no indications on the upper  
568 concentration limits of application of the method. The concern about asbestos  
569 quantitative determination in the case of massive concentration of fibers is related to  
570 difficulties for fiber identification and counting due to severe fiber overlap, aggregation  
571 and stratification. The results of this study point out that SEM method, if a careful sample  
572 preparation, dispersion and representativeness are obtained, can be used even for the  
573 quantitative determination of chrysotile asbestos in massive materials with concentrations  
574 much larger than 1 wt% with a fairly good accuracy.

575 *Infrared spectroscopy*

576 There are many examples of application of FTIR for the quantitative determination of  
577 chrysotile asbestos in bulk materials (see for example, Balducci and Valerio 1986;  
578 Massola 1997) although some criticism still exists. According to Davies et al. (1996),  
579 FTIR is not a sensitive method for quantifying asbestos in loose aggregates and does not  
580 produce useful results when multi-component mixtures are analyzed. Oppositely, a  
581 positive report is provided by Foresti et al. (2003) who described the application of the  
582 FTIR Linear Calibration Curve method for the determination of low levels (0.01–1 wt%)  
583 of free fibers of chrysotile in contaminated clayey, sandy and sandy-organic soils. The  
584 detection limit of 0.01 wt% was reached with an enrichment of free fibers of chrysotile in  
585 the samples using a standard laboratory elutriator for sedimentation analysis. The  
586 linearity of the calibration curves obtained for samples having different soil matrices,  
587 indicates that the matrix effects can be accounted for. Recently, De Stefano et al. (2012)  
588 compared the accuracy and precision of the Linear Calibration Curve Method and the  
589 Method of Addition for the FTIR quantitative determination of asbestos bulk matrices  
590 and found that, providing careful samples preparation, both techniques quantify the  
591 asbestos content at the level of 1-2 wt% with good precision. The results provided in this  
592 study are very important as this is one of the few examples of application of the FTIR  
593 quantitative method to complex serpentinite samples.

594 The figures obtained with this method seem rather accurate and in agreement with the  
595 values provided by the application of the other experimental methods (Table 5): an  
596 estimate of 4.9(2) wt% was calculated for STD1 sample against an expect value of 5.11  
597 wt% and 13.8(2) wt% for STD2 sample against an expected value of 10.44 wt%. Both  
598 calculated weights of sample STD2 and E1 (32.9 wt% vs. 25.3 wt% from the Rietveld

599 method and 27.5 wt% from the SEM analysis) are apparently overestimated. This  
600 discrepancy may be possibly due to: (i) interference of antigorite and chlorite present in  
601 the investigated samples with a severe overlap of their absorption bands to that of  
602 chrysotile; (ii) deviation of the linearity of the curve intensity of the chrysotile absorption  
603 band vs. concentration for large chrysotile contents. The breaking of the linearity  
604 assumption has been already observed by De Stefano et al. (2012). They observed that in  
605 the case of chrysotile, the peak intensity is a quadratic-like function of the concentration  
606 in the extended concentration range.

#### 607 *X-ray powder diffraction*

608 In this work it was found that the XRPD external standard method (Klug and  
609 Alexander 1974) is unsuitable for such complex matrices as the observation (e.g., the  
610 (001) peak of chrysotile) overlaps to the major peaks of both antigorite (001) and chlorite  
611 (002) present in the samples (see sample E1 in Fig. 5), resulting in biased estimates of the  
612 integrated area using the single peak fitting deconvolution. The limits of the single peak  
613 method for such complex matrices were obvious to Giacomini et al. (2010) who obtained  
614 only qualitative estimates of the relative abundance of serpentine polymorphs in samples  
615 from the metaophiolites of the the Voltri Massif and Sestri–Vtaggio Zone (Liguria, NW  
616 Italy), using the internal standard (20 wt% of corundum) technique and the reference  
617 intensity ratio (RIR) method (Snyder and Bish 1989). On the same line, only qualitative  
618 determination via XRPD methods were reported by Rigopoulos et al. (2010) for  
619 asbestiform minerals in basic and ultrabasic rocks from ophiolite suites of central and  
620 northern Greece, Beneduce et al. (2012) for the ophiolites of the Pollino National Park  
621 (Calabria-Lucania border, southern Italy), and Cavallo and Rimoldi (2013) for

622 serpentinites of the Valmalenco area (Central Alps, Northern Italy). Puledra and Marconi  
623 (1990) assessed the validity of the XRPD external standard method with the Ag filter for  
624 synthetic mixtures prepared for the determination of chrysotile content in bulk and  
625 airborne samples but did not apply it to natural complex samples such as the one  
626 investigated here. It is difficult to evaluate the analytical accuracy of the estimates  
627 reported by Davis (1990) who used the RIR method to determine the content of asbestos  
628 minerals in synthetic multicomponent mixtures prepared samples demonstrate that the  
629 lower limit of detection for most asbestos minerals falls in the range 0.5% to 2 wt%.

630 On the contrary, the figures obtained with the Rietveld method seem accurate (Table  
631 5): an estimate of 5.8(2) wt% was calculated for STD1 sample against an expected value of  
632 5.11 wt% and 11.2(2) wt% for STD2 sample against an expected value of 10.44 wt%.  
633 Sample E1 and C2 also contain fairly large amounts of chrysotile asbestos. The results  
634 reported here witness that the Rietveld method can be successfully applied for the  
635 determination of chrysotile asbestos in these complex multicomponent samples  
636 characterized by the presence of two serpentine polymorphs (antigorite and chrysotile)  
637 and chlorite. It is not possible to generalize the outcome of this study and assert that the  
638 method is accurate even for more complex samples which contain all the serpentine  
639 polymorphs including lizardite, although this mineral association is not rare in green  
640 stones. Another critical aspect regards the range of applicability of the Rietveld method in  
641 terms of chrysotile content. It is well known that chrysotile possesses a cylindrical lattice  
642 with layers curled concentrically or spirally, usually around the  $x$  axis (*clinochrysotile*  
643 and *orthochrysotile*) and seldom around the  $y$  axis (*parachrysotile*), to form a tubular  
644 structure (roll) of about  $22\pm 27$  nm in diameter (Whittaker 1957; Yada 1971). Moreover,

645 the cylindrical lattice also displays extensive structural defectivity (e.g., random shift  
646 component along the fiber axis to yield cone lattices). Such structure complexity results  
647 in anisotropic broadening effects of the peak profiles in the X-ray powder patterns which  
648 can only be properly fit by recursive models such as the one implemented in DIFFaX+  
649 (Leoni et al. 2004) and not simply by Rietveld-based deterministic codes. For this reason,  
650 Wilson et al. (2006) developed an alternative approach to determine the chrysotile  
651 content in mine tailings at Clinton Creek (Yukon Territory) and Cassiar (British  
652 Columbia) using structure-less pattern fitting and with the addition of a known quantity  
653 of a well-crystallized material, by considering the serpentine minerals as amorphous  
654 phases, and tested the accuracy of the method using synthetic serpentine-rich mine  
655 tailings of known composition. Notwithstanding, marked peak profile broadening effects  
656 (especially the asymmetric band in the 19-26 °2θ range due to the destructive interference  
657 of non-basal diffraction peaks) are obvious when chrysotile fraction is high. Usually,  
658 when chrysotile is present in medium-low concentration, the powder pattern generally  
659 displays only the major basal (00*l*) peaks of chrysotile and minor non-basal peaks. Thus,  
660 peak profile broadening effects are less obvious and may be modeled by empirical  
661 Lorentzian profile anisotropic broadening functions such as the one implemented in  
662 GSAS (Larson and Von Dreele 1999):  $\gamma = Y + Y_e \cos\phi$  with  $\gamma$  = Lorentzian contribution to  
663 the pseudo-Voigt profile function;  $Y$  = isotropic strain broadening term;  $Y_e$  = anisotropic  
664 strain broadening term;  $\phi$  = angle which defines the direction of the broadening axis  
665 within the crystallographic setting. Wilson et al. (2006) reported chrysotile estimates as  
666 high as 90.8 wt%. For those samples resembling monophasic systems, the accuracy of the  
667 Rietveld method is expected to be low. On the other hand, for the systems investigated

668 here with chrysotile contents below 40 wt%, the accuracy of the Rietveld method is still  
669 expected to hold. As a matter of fact, the refinement of the anisotropic strain broadening  
670 term of the Lorentzian part of the pseudo-Voigt function was attempted for chrysotile for  
671 samples C2 and E1, setting the broadening direction along the  $b$  axis and the stacking  
672 fault model but did not significantly changed the final calculated weights.

### 673 *Thermal analysis*

674 The quantitative concentration of chrysotile in the standard samples using the DTA  
675 method, reported in Table 5, are rather accurate: 7.6 wt% was calculated for STD1  
676 sample against an expect value of 5.11 wt% and 12.2 wt% for STD2 sample against an  
677 expected value of 10.44 wt%. Although the successful application of the method has  
678 already been reported in the literature for simple systems such as pharmaceutical grade  
679 talc where the minimum level of detection was 1% by weight of chrysotile asbestos  
680 (Shelz 1974) and in cosmetic grade talc/body products where the minimum level of  
681 detection was 0.5% by weight of chrysotile asbestos (Luckewicz 1975), only the recent  
682 contribution of Viti et al. (2011) reported accurate estimates for complex natural systems  
683 such as serpentinites. Hence, the results of this work are a further confirmation of the  
684 reliability of this method. The apparently systematic overestimation should be due to the  
685 presence of chlorite in the system. As a matter of fact, Viti et al. (2011) reported that the  
686 possible occurrence of other minerals does not hamper the attainment of reliable  
687 qualitative determinations, with the only exception of chlorite whose typical DTA curve  
688 is characterized by an endothermic peak at 600–650 °C (Smykatz-Kloss 1974), and  
689 suggested that, prior to DTA data collection, qualitative XRPD should be performed to  
690 ascertain the presence and the amount of chlorite. The presence of chlorite was assessed

691 in the first instance via XRPD suggesting the addition of a further peak due to chlorite  
692 during the process of fit deconvolution. Notwithstanding, it is possible that a systematic  
693 correlation occurred for the calculation of the integrated areas and that the resulting  
694 integrated area of chrysotile was underestimated in favor of that of chlorite.

695

#### 696 **Crètaz and Emarèse samples**

697 According to the Italian law D.Lgs 12/03/2010 nr. 205 All. D, if the asbestos  
698 concentration is higher than 0.1 wt%, a site should be classified as “contaminated by  
699 asbestos”. Hence, both Crètaz and Emarèse sites should be reclaimed as all the collected  
700 samples consistently show asbestos concentration much higher than the imposed limit  
701 (Tables 4 and 5).

702 Figure 9a,b reports maps of concentration of chrysotile and tremolite obtained from  
703 the Rietveld analysis for the site of Crètaz (Tables 4 and 5). The maps were obtained by  
704 geostatistical analysis of the raw data in GIS environment using the software ArcGIS  
705 10.1 and the ArcGIS Geostatistical Analyst application. The Inverse Distance Weighting  
706 (IDW) model (Shepard 1968) was used for the deterministic interpolation of the data  
707 points. For the chrysotile concentration maps, the following parameters were used: first  
708 order model (power = 1), maximum neighbors = 10, minimum neighbors = 5, total  
709 calculation area = circular with radius of 138 m. For the tremolite concentration maps, the  
710 following parameters were used: second order model (power = 2), maximum neighbors =  
711 10, minimum neighbors = 5, total calculation area = circular with radius of 200 m.

712 The chrysotile content varies from 4.4 to 22.8 wt%. In all the samples, chrysotile is  
713 associated to antigorite (6.3-50.1 wt%), chlorite (clinochlore 4.1-15.2 wt%), muscovite

714 (2.2-15.8 wt%), quartz (1.9-29.1 wt%), and calcite (4.2-27.5 wt%). The mineral  
715 assemblage reflects the overall composition of the mine tailings, a mixing of three  
716 different rock types all belonging to the Piedmont Zone unit: the antigorite-serpentinites  
717 with magnetite formation, the calc-schists s.l. and local quartzites formation, and the  
718 exotic continental Permian-mesozoic succession with schists, quartzites, marbles, and  
719 limestones. The distinctive minerals of the latter formation are quartz, mica, carbonates  
720 calcite and dolomite, plagioclase, K-feldspar and tremolite (1.8-10.3 wt%). Apparently  
721 no trends of the chrysotile and tremolite concentration with (horizontal or vertical) space  
722 are observed. This random distribution is due to the history of the various deposits  
723 emplaced and mixed with other source of rock dumps during the activity of the mine and  
724 not to physical factors such as fiber leaching, transport, and re-deposition. The area with  
725 the highest surface concentration of asbestos minerals (see Figure 9) should be  
726 considered as priority of intervention in the reclamation plan.

727 Figure 9c, not elaborated with the geostatistical analysis due to the limited number of  
728 raw data points, reports the concentration of chrysotile in the site of Emarèse as obtained  
729 from the Rietveld analysis (see also Tables 4 and 5). The chrysotile content varies from  
730 3.3 to 39.5 wt% with the concentration linearly decreasing with the distance from the  
731 main mining sites. Chrysotile is invariably accompanied by antigorite (3.3-44.7 wt%) and  
732 chlorite (clinochlore 12.4-32.6 wt%). On the basis of the mineral assemblage, two groups  
733 of rock samples are observed: (i) samples E1-E5 belongs to the antigorite-serpentinites  
734 with magnetite formation of the Piedmont Zone. Besides the typical serpentinite phases  
735 antigorite, chrysotile and chlorite, magnetite, talc and brucite are found together with the  
736 minerals of the original ultramaphic rock forsterite and enstatite. Hematite, calcite and

737 dolomite are likely secondary phases formed after surface alteration in contact with  
738 water; (ii) samples E6-E7 are collected from the surface debris and soil composed of a  
739 mixing of the former antigorite-serpentinites with magnetite formation and the calc-  
740 schists s.l. and local quartzites formation both belonging to the Piedmont Zone unit. The  
741 distinctive minerals of the calc-schists s.l.- quartzite formation are quartz, mica, K-  
742 feldspar, plagioclase, and tremolite (5.9-12.4 wt%).

743 The area with the highest surface concentration of chrysotile (see Figure 9c) should  
744 also be classified as priority in the reclamation plan.

745

746

#### CONCLUDING REMARKS

747 A map of the concentration of asbestos minerals chrysotile and tremolite in  
748 contaminated mine tailings of the Valle d'Aosta region (Northern Italy) was drawn. The  
749 results of this study are strategic as the knowledge of the spatial distribution of the  
750 asbestos concentration allows to plan reclamation agenda of the sites. The area with the  
751 highest surface concentration of both chrysotile and tremolite were identified and  
752 classified as priority in the reclamation plan. The results of our study have general  
753 implications as the protocol of sampling, data analysis and mapping of the asbestos  
754 concentration can be used in any region of the globe where such deposits are found.  
755 Besides that, another general implication of our work regards the assessment of the  
756 accuracy of the results obtained using state of the art experimental techniques for the  
757 determination of the concentration of asbestos in massive materials. Although we focused  
758 on the methods recommended by the Italian laws, the Rietveld method and DTA (Viti et  
759 al. 2011) were also included so that a comparison among the various methods used

760 worldwide was possible. Accurate estimates were obtained using the SEM, FTIR, XRPD  
761 Rietveld and DTA methods whereas poor figures were obtained with the optical  
762 microscopy. The single peak fitting XRPD method revealed to be unsuitable because of  
763 the interference effects of antigorite and chlorite. It should be remarked that the Rietveld  
764 method has been successfully applied for the determination of chrysotile concentrations  
765 as high as about 40 wt% in these complex serpentinite multi-component samples  
766 characterized by the presence of two serpentine polymorphs (antigorite and chrysotile)  
767 and chlorite. Further investigations are required to test the accuracy of the method for  
768 even more difficult systems which contain all the three serpentine polymorphs, including  
769 lizardite. With higher concentrations (>50 wt%?), it is recommended to apply other  
770 methods such as the one proposed by Wilson et al. (2006) who determined the chrysotile  
771 content in asbestos mine tailings using structure-less pattern fitting and with the addition  
772 of a known quantity of a standard material so to be able to consider the serpentine  
773 minerals as amorphous phases.

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775

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1062 **Figure Captions**

1063 **FIGURE 1.** Sketch map of the geology of Valle d'Aosta (1:1.000.000 scale; Legend:  
1064 Austroalpine System (1 = Diorite–Kinzigite Zone, 2 = Gneiss Complex, 3 = Eclogitic  
1065 Micaschist Complex); Piedmont Zone with Calc-schists and Greenstones (4 = Oceanic  
1066 Metasedimentary overlay, 5 = Metabasites); Penninic System (6 = Upper Penninic Zone,  
1067 7 = Middle Penninic Zone, 8 = Outer Penninic Zone; 9 = Ultra-elvetic System; 10 =  
1068 Elvetic System) and location of the surface and core collection spots for the two  
1069 investigated sites: (a) Crètaz with C1-C8 = surface samples, depth 0.25-0.3 m; C9v1 =  
1070 drill 1, depth 0.7 m; C10v1 = drill 1, depth 7.5 m; C11v1 = drill 1, depth 15 m; C12v1 =  
1071 drill 1, depth 19.2 m; C13v2 = drill 2, depth 0.5 m; C14v2 = drill 2, depth 11 m; C15v2 =  
1072 drill 2, depth 22.4 m. (b) Emarèse with E1-E8 = surface samples.

1073 **FIGURE 2.** Selected PCOM images of the investigated samples (see text for details).  
1074 Legend: (a) chrysotile at 10x in liquid with refractive index of 1.550; (b) tremolite fibers  
1075 at 10x in liquid with refractive index of 1.615; (c) non-asbestos acicular-like crystal:  
1076 diopside at 10x in liquid with refractive index of 1.670.

1077 **FIGURE 3.** Selected SEM images of the investigated samples (see text for details).  
1078 Legend: (a) particle and fiber aggregates difficult to detect and count; (b) chrysotile fibers  
1079 of heterogeneous size; (c) tremolite fibers with the relative EDS point analysis which  
1080 permits the discrimination from other fibrous phases; (d) example of intergrowth of  
1081 mixed chrysotile-antigorite aggregate; (e) ball of thread like aggregates of chrysotile  
1082 fibers within the powders of the samples prepared in LAB2 (see the text for details).

1083 **FIGURE 4.** The FTIR spectra region between 3400 and 3950  $\text{cm}^{-1}$  of the four  
1084 investigated samples where the absorption bands of chrysotile and interference layer

1085 silicates occur. (a) STD1; (b) STD2; (c) C2; (d) E1.

1086 **FIGURE 5.** Selected low angle region of sample E1 showing severe overlap of the  
1087 peaks of chrysotile, antigorite and clinochlore and the result of the peak fitting using the  
1088 Rietveld method.

1089 **FIGURE 6.** The Rietveld graphical outputs of the samples selected for the accuracy  
1090 verification. Crosses represent the observed pattern, the thin line represents the calculated  
1091 pattern and the grey bottom line is the difference line. Markers of all the peaks of the  
1092 each crystalline phase included in the refinement procedure are also shown in different  
1093 lines. Legend: (a) STD1, lines of peak markers from the bottom: antigorite, chrysotile,  
1094 tremolite, talc, clinochlore, calcite, dolomite, magnesite, albite. (b) STD2, lines of peak  
1095 markers from the bottom: antigorite, chrysotile, tremolite, talc, clinochlore, calcite,  
1096 dolomite, magnesite, albite; (c) C2, lines of peak markers from the bottom: antigorite,  
1097 chrysotile, tremolite, muscovite, clinochlore, calcite, quartz, K-feldspar, albite; (d) E1,  
1098 lines of peak markers from the bottom: antigorite, chrysotile, clinochlore, calcite,  
1099 magnetite, forsterite, hematite, enstatite, brucite.

1100 **FIGURE 7.** DTA traces of the four investigated samples and standard pure chrysotile,  
1101 in the range 550–850 °C (see the text for details) with the result of the deconvolution  
1102 fitting procedure. Legend: (a) STD1; (b) STD2; (c) C2; (d) E1; (e) pure chrysotile.

1103 **FIGURE 8.** The Rietveld graphical output of sample C5. Crosses represent the observed  
1104 pattern, the thin line represents the calculated pattern and the grey bottom line is the  
1105 difference line. Markers of all the peaks of the each crystalline phase included in the  
1106 refinement procedure are shown. Legend from the bottom line: antigorite, chrysotile,  
1107 tremolite, muscovite, clinochlore, calcite, quartz, K-feldspar, dolomite.

1108        **FIGURE 9.** Maps of concentration of chrysotile (a) and tremolite (b) obtained from the  
1109 Rietveld analysis for the site of Crètaz obtained by geostatistical analysis of the raw data  
1110 in GIS environment using the software ArcGIS 10.1 and the ArcGIS Geostatistical  
1111 Analyst application. The raw concentrations of chrysotile for the site of Emarèse are  
1112 plotted in (c).  
1113

1114 **Tables**

1115 **TABLE 1.** The calculated absorbance (%) of the stretching vibration of chrysotile and the  
1116 resulting chrysotile content (wt%), using the calibration curve with equation  $y =$   
1117  $0.78269x$  with  $y =$  absorbance (%) and  $x =$  chrysotile weight fraction.

Sample	absorbance (%)	chrysotile content (wt%)	chrysotile content error
STD1	3.856	4.9	0.03
STD2	10.801	13.8	0.02
C2	4.931	6.3	0.02
E1	25.75	32.9	0.02

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1126 **TABLE 2.** The results of the Rietveld quantitative phase composition. The definition of  
 1127 the agreement factors is reported in Larson and Von Dreele (1999).

	STD1	STD2	C2	E1
R <sub>wp</sub> (%)	10.3	9.73	10.33	9.44
R <sub>p</sub> (%)	8.02	7.51	7.57	7.07
$\chi^2$	5.40	2.71	5.40	6.32
antigorite	6.9(3)	14.9(3)	3.5(3)	44.7(6)
brucite	-	-	-	0.3(1)
calcite	39.3(2)	36.7(3)	8.0(3)	2.6(2)
chrysotile	5.8(2)	11.2(3)	6.3(4)	25.1(9)
clinochlore	4.6(2)	5.9(3)	7.5(3)	15.8(4)
dolomite	11.9(2)	11.7(2)	-	-
hematite	-	-	-	1.3(2)
enstatite	-	-	-	1.2(3)
forsterite	-	-	-	4.0(6)
K-feldspar	-	-	4.2(3)	-
magnesite	4.3(2)	4.6(2)	-	-
magnetite	-	-	-	5.0(2)
muscovite/illite	-	-	10.9(5)	-
plagioclase	18.3(2)	10.8(3)	22.4(3)	-
quartz	-	-	32.6(2)	-
talc	3.3(5)	3.2(3)	-	-
tremolite	5.6(3)	1.0(3)	4.6(3)	-
TOT	100	100	100	100

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1130 **TABLE 3.** The results of the chrysotile determination using the DTA method (see the text  
1131 for details).

Sample	Sum of the peak area	Error	$r^2$	Calculated wt% chrysotile	Area ratio	Error
Pure chrysotile standard	1358	39	0.9909	100	1	n.d.
STD1	89	16	0.9927	7.6	0.06	0.03
STD2	155	8	0.9921	12.2	0.11	0.02
C2	108	3	0.9952	8.9	0.08	0.07
E1	300	26	0.9955	22.2	0.22	0.07

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1136 **TABLE 4.** The quantitative phase compositions of the Crêtaz and Emarèse samples. The definition of the agreement factors is reported  
 1137 in Larson and Von Dreele (1999).

	C1	C2	C3	C4	C5	C6	C7	C8	C9V1	C10V1	C11V1	C12V1
R <sub>wp</sub> (%)	7.68	10.33	6.46	8.8	5.64	11.32	7.12	6.98	7.8	7.86	3.29	10.48
R <sub>p</sub> (%)	5.58	7.57	4.81	6.22	4.08	8.01	5.51	5.42	5.94	6.02	5.55	7.93
$\chi^2$	5.55	5.40	3.44	6.60	3.87	8.64	2.55	2.96	3.33	3.35	2.64	5.61
antigorite	32.4(8)	3.5(3)	29.5(6)	47.1(6)	39.3(7)	50.1(6)	30.9(5)	23.6(4)	40.4(4)	19.5(7)	14.3(4)	52.4(5)
calcite	20.5(3)	8.0(3)	4.2(2)	9.0(3)	11.5(3)	8.1(3)	12.0(2)	8.0(2)	14.7(2)	27.5(2)	16.6(2)	5.0 (2)
chrysotile	6.2(9)	6.3(4)	4.4(6)	11.4(8)	9.7(8)	11.2(7)	15.8(4)	11.6(4)	11.8(4)	12.1(5)	6.0(3)	22.8(5)
clinochlore	13.4(5)	7.5(3)	10.0(3)	12.0(6)	6.1(5)	11.4(5)	15.2(5)	7.0(3)	10.1(5)	9.4(6)	9.0(3)	7.9(7)
dolomite	4.3(5)	-	-	2.3(5)	2.8(5)	2.2(5)	15.0(3)	-	-	4.6(2)	-	-
K-feldspar	3.5(4)	4.2(3)	1.7(3)	2.9(4)	5.1(4)	2.1(3)	-	-	-	-	-	-
magnetite	-	-	-	-	-	-	4.9(1)	2.5(1)	3.6(2)	4.7(1)	1.3(1)	4.2(2)
muscovite/illite	9.6(8)	10.9(5)	15.8(5)	8.3(8)	13.1(1)	8.4(7)	2.2(6)	10.3(5)	7.8(6)	9.3(6)	20.4(4)	2.2(7)
plagioclase	-	22.4(3)	3.3(4)	-	-	-	-	6.3(3)	-	-	-	-
quartz	2.0(3)	32.6(2)	28.8(2)	3.8 (4)	5.4(2)	4.1(1)	3.5(2)	17.6(1)	5.6(1)	10.2(2)	29.1(2)	3.4(2)
talc	0.5(2)	-	0.5(2)	-	-	-	0.5(2)	2.8(2)	2.1(2)	2.7(3)	3.3(3)	2.1(3)
tremolite	7.6(6)	4.6(3)	1.8(4)	3.2(5)	7.0(5)	2.4(5)	-	10.3(1)	3.9(5)	-	-	-
TOT	100	100	100	100	100	100	100	100	100	100	100	100

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	C13V2	C14V2	C15V2	E1	E2	E3	E4	E5	E6	E7	E8
R <sub>wp</sub> (%)	8.29	8.61	7.6	9.44	10.9	10.44	9.23	10.31	8.36	6.6	6.01
R <sub>p</sub> (%)	6.29	6.73	5.77	7.07	8.13	7.64	6.94	7.53	6.26	5.09	4.67
$\chi^2$	3.99	3.54	2.87	6.32	8.68	8.89	6.35	10.41	5.87	3.32	3.20
antigorite	35.4(5)	35.1(4)	6.3(4)	44.7(6)	21.4(5)	13.5(3)	7.1(2)	17.1(3)	3.3(2)	7.7(3)	8.8(3)
brucite	-	-	-	0.3(1)	2.0(3)	1.0(3)	1.1(2)	1.4(3)	-	-	-
calcite	12.0(3)	9.8(2)	5.0(2)	2.6(2)	-	3.5(2)	2.9(2)	4.9(3)	2.0(5)	1.3(2)	-
chrysotile	14.7(5)	21.9(4)	5.2(3)	25.1(9)	39.5(9)	35.0(9)	39.5(8)	36.4(9)	15.2(9)	4.6(4)	3.3(5)
clinochlore	6.7(4)	18.4(5)	4.1(5)	15.8(4)	25.4(7)	23.8(8)	32.6(6)	22.6(7)	23.5(8)	12.4(5)	20.0(5)
dolomite	5.9(3)	5.8(2)	-	-	-	1.4(2)	-	-	0.4(4)	-	-
hematite	-	-	-	1.3(2)	1.2(2)	1.5(2)	1.7(2)	1.8(2)	2.6(3)	-	-
enstatite	-	-	-	1.2(3)	1.9(4)	3.1(6)	2.5(7)	1.4(4)	10.5(9)	8.3(5)	6.0(5)
forsterite	-	-	-	4.0(6)	3.2(7)	6.4(7)	5.7(4)	9.3(6)	6.0(7)	-	-
K-feldspar	-	-	-	-	-	-	-	-	-	5.6(5)	5.5(4)
magnetite	4.6(1)	3.5(1)	-	5.0(2)	5.4(2)	6.2(2)	6.9(2)	5.1(2)	-	-	-
muscovite/illite	6.8(7)	2.3(6)	15.0(5)	-	-	-	-	-	7.6(9)	13.9(9)	13.3(9)
plagioclase	3.7(4)	-	18.7(4)	-	-	-	-	-	8.3(4)	12.9(4)	11.3(4)
quartz	5.5(2)	1.9(1)	40.1(1)	-	-	-	-	-	6.3(2)	12.5(2)	10.4(2)
talc	-	1.3(3)	1.6(2)	-	-	4.6(9)	-	-	8.4(9)	9.6(4)	9.0(4)
tremolite	4.7(4)	-	4.0(3)	-	-	-	-	-	5.9(5)	11.2(3)	12.4(3)
TOT	100	100	100	100	100	100	100	100	100	100	100

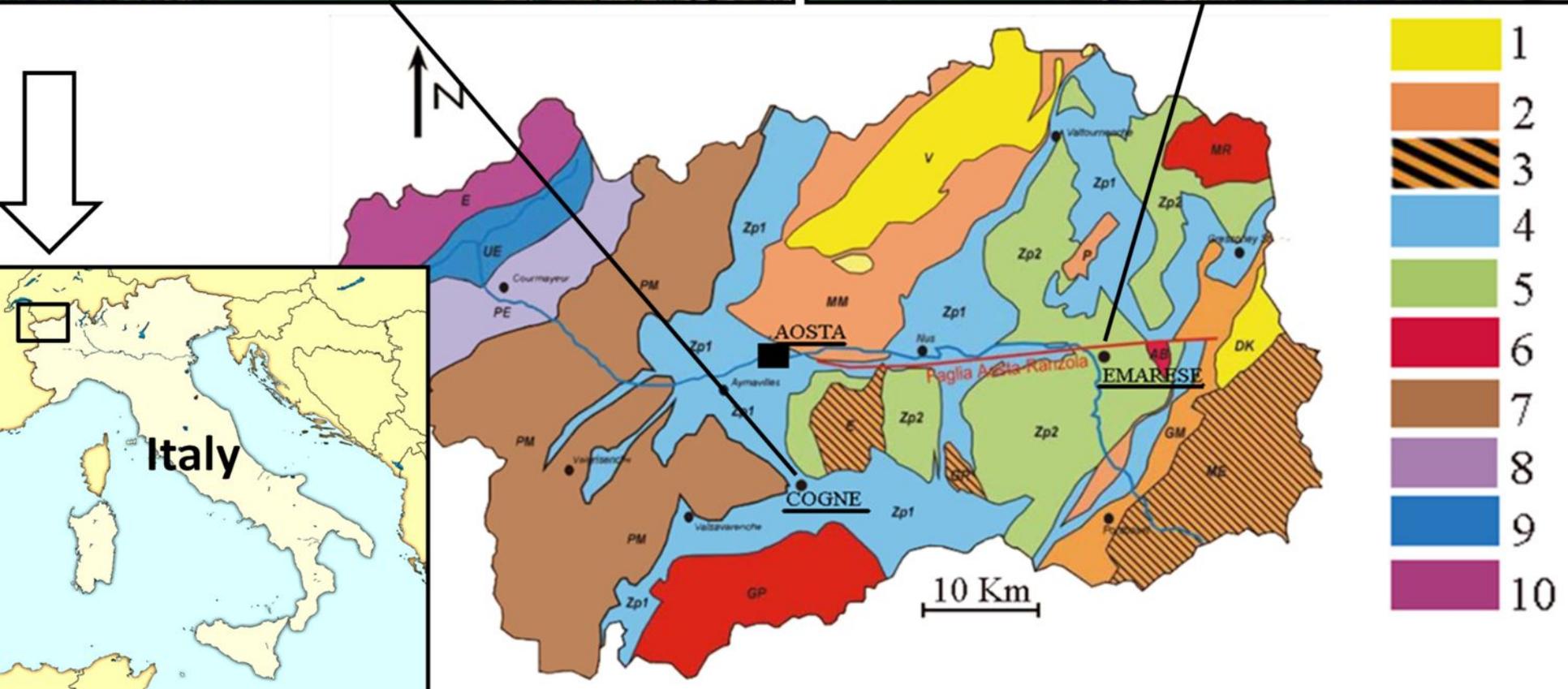
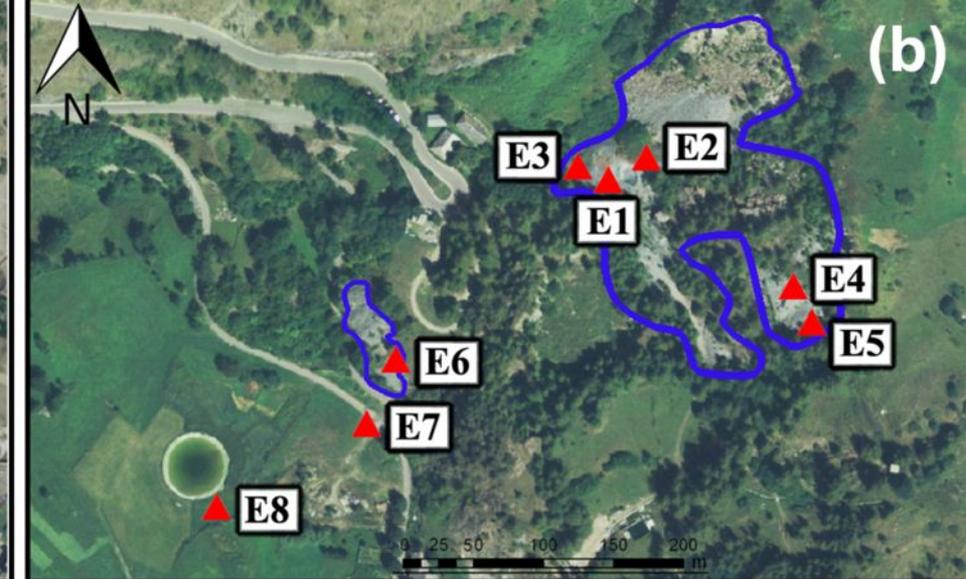
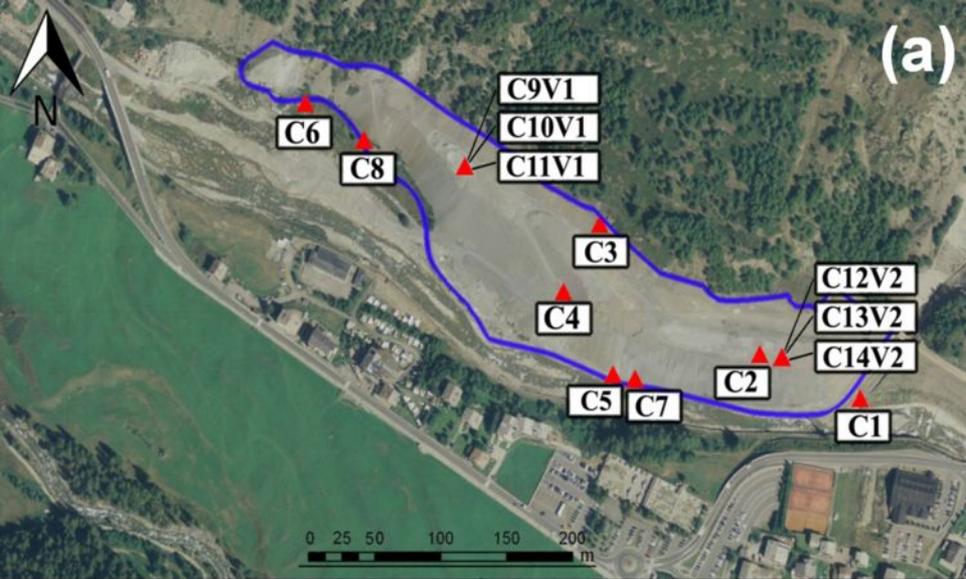
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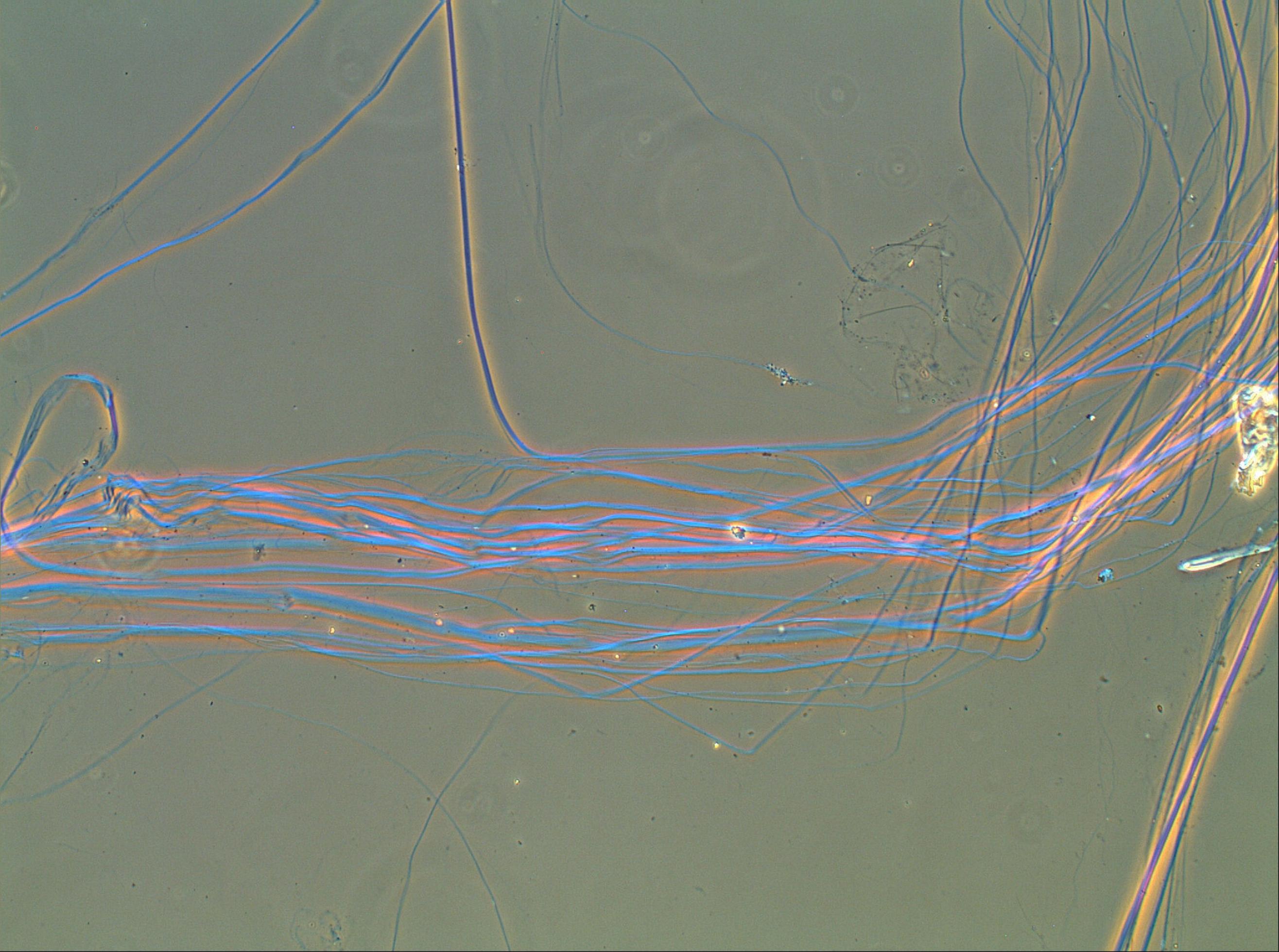
1146 **TABLE 5.** Summary of the calculated concentration of chrysotile asbestos in the  
 1147 four samples selected for the cross-check analysis.

Method	Normative reference	Chrysotile	Chrysotile	Chrysotile	Chrysotile
		wt% STD1	wt% STD2	wt% C2	wt% E1
		Weighed value 5.11	Weighed value 10.44		
DTA	-	7.6	12.2	8.9	22.2
FTIR	Italian D.M. 09-06-94	4.9	13.8	6.3	32.9
PCOM	Italian D.M. 09-06-94	3.7	1.4	n.d.	7.5
SEM	Italian D.M. 09-06-94 - LAB1	4.6	13.1	4.7	27.5
SEM	Italian D.M. 09-06-94 - LAB2	1.9	0.6	5.9	26.4
XRPD external standard method XRPD	Italian D.M. 09-06-94	n.d.	n.d.	n.d.	n.d.
Rietveld method	-	5.8	11.2	6.3	25.3

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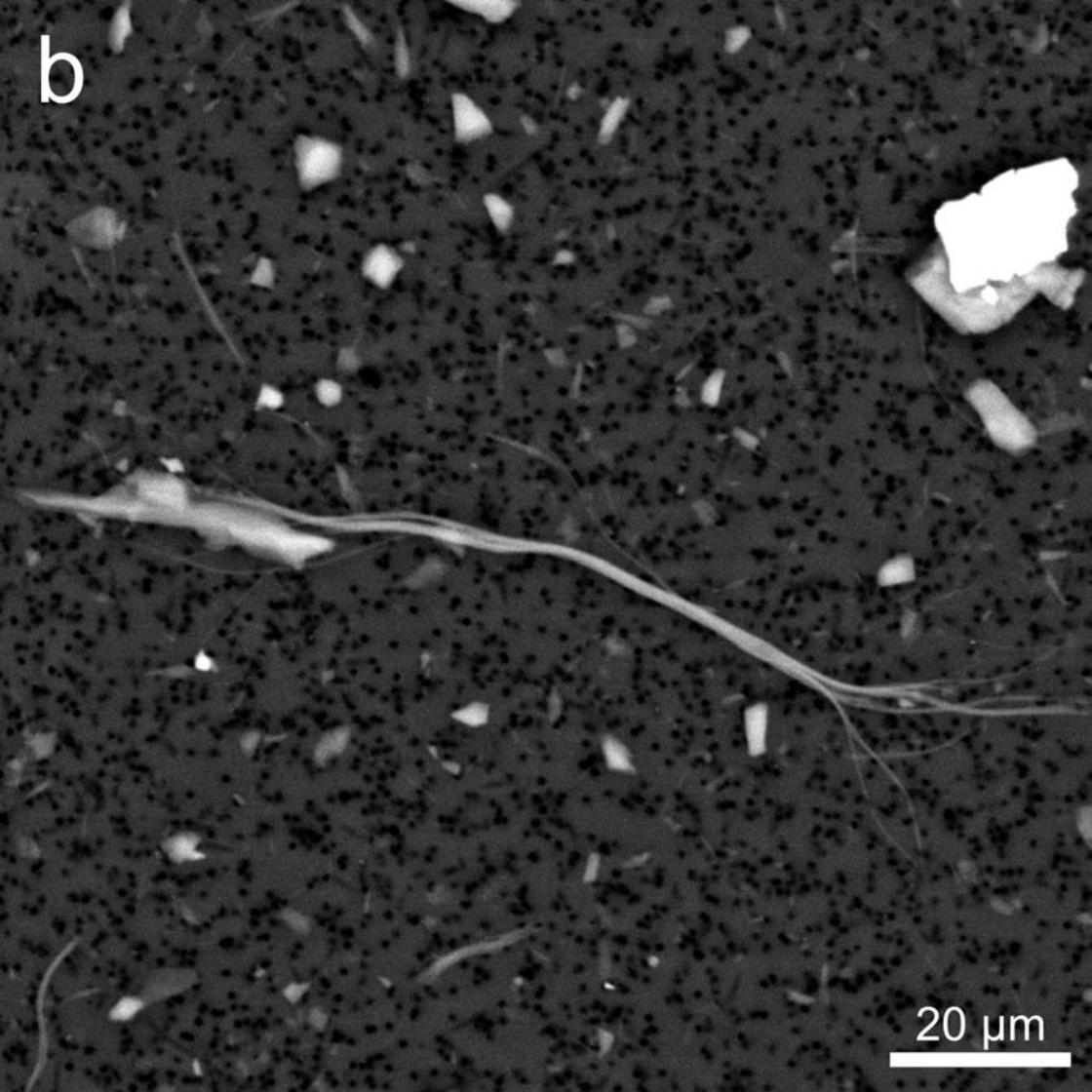


a



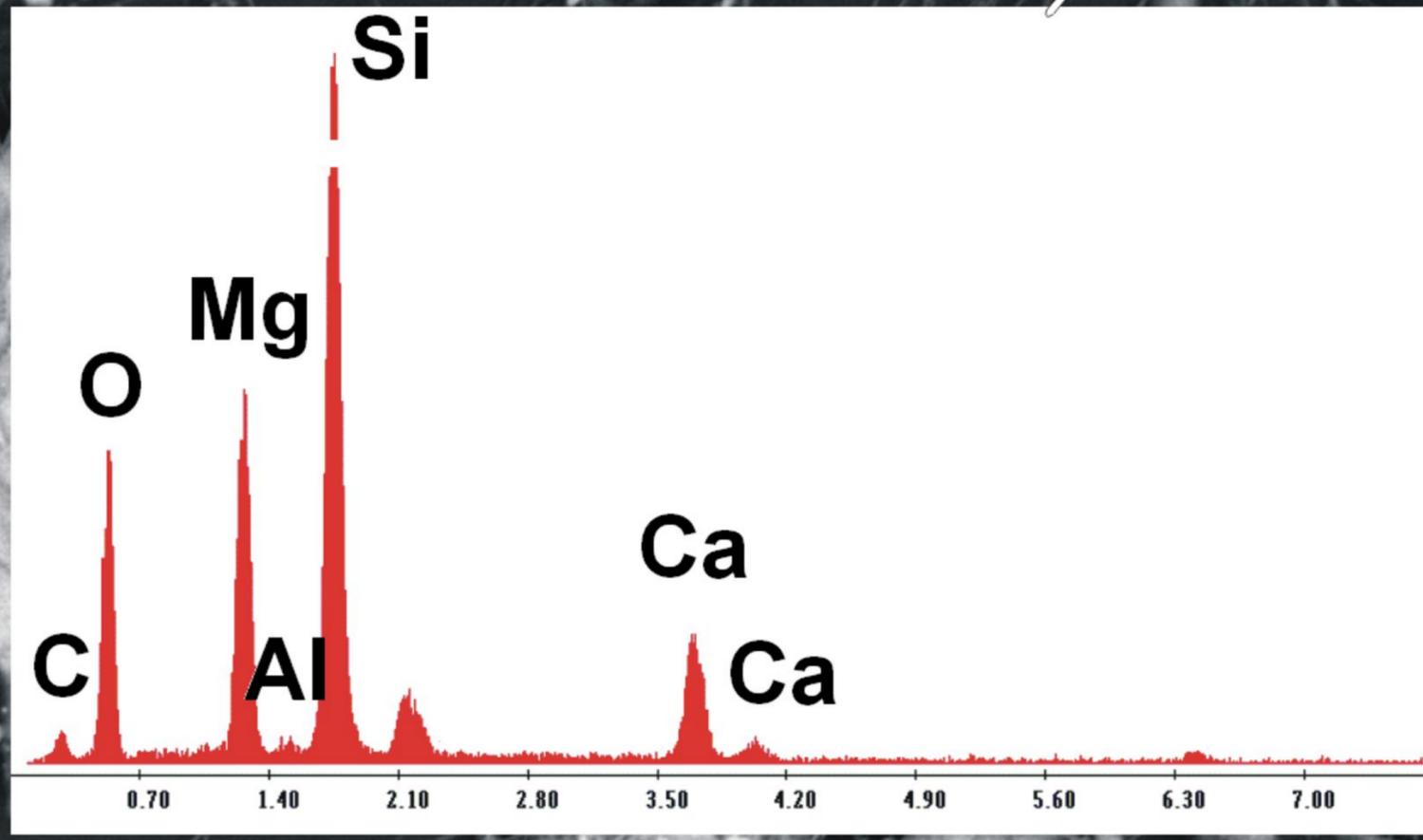
40  $\mu\text{m}$

b

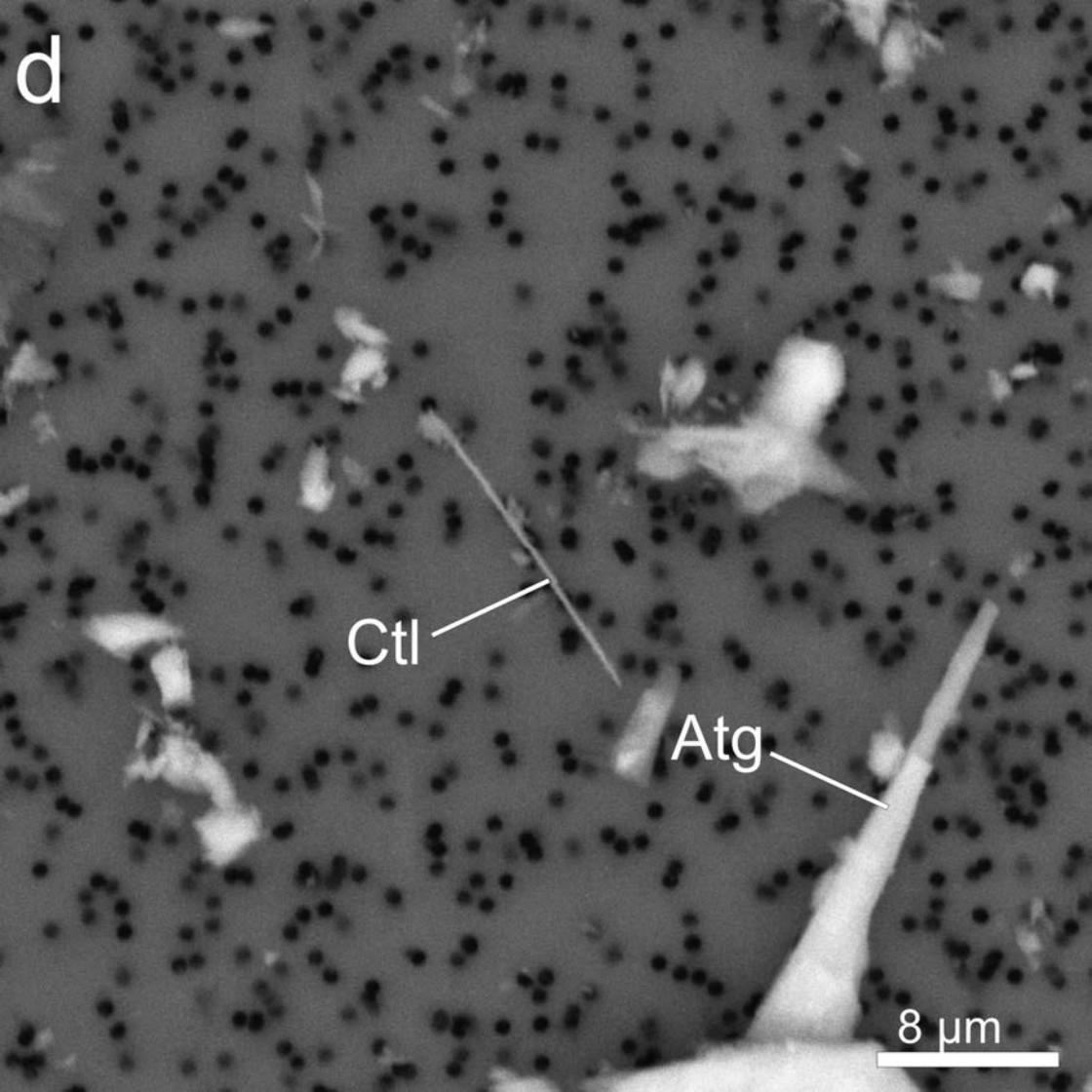


20  $\mu\text{m}$

C



50  $\mu\text{m}$



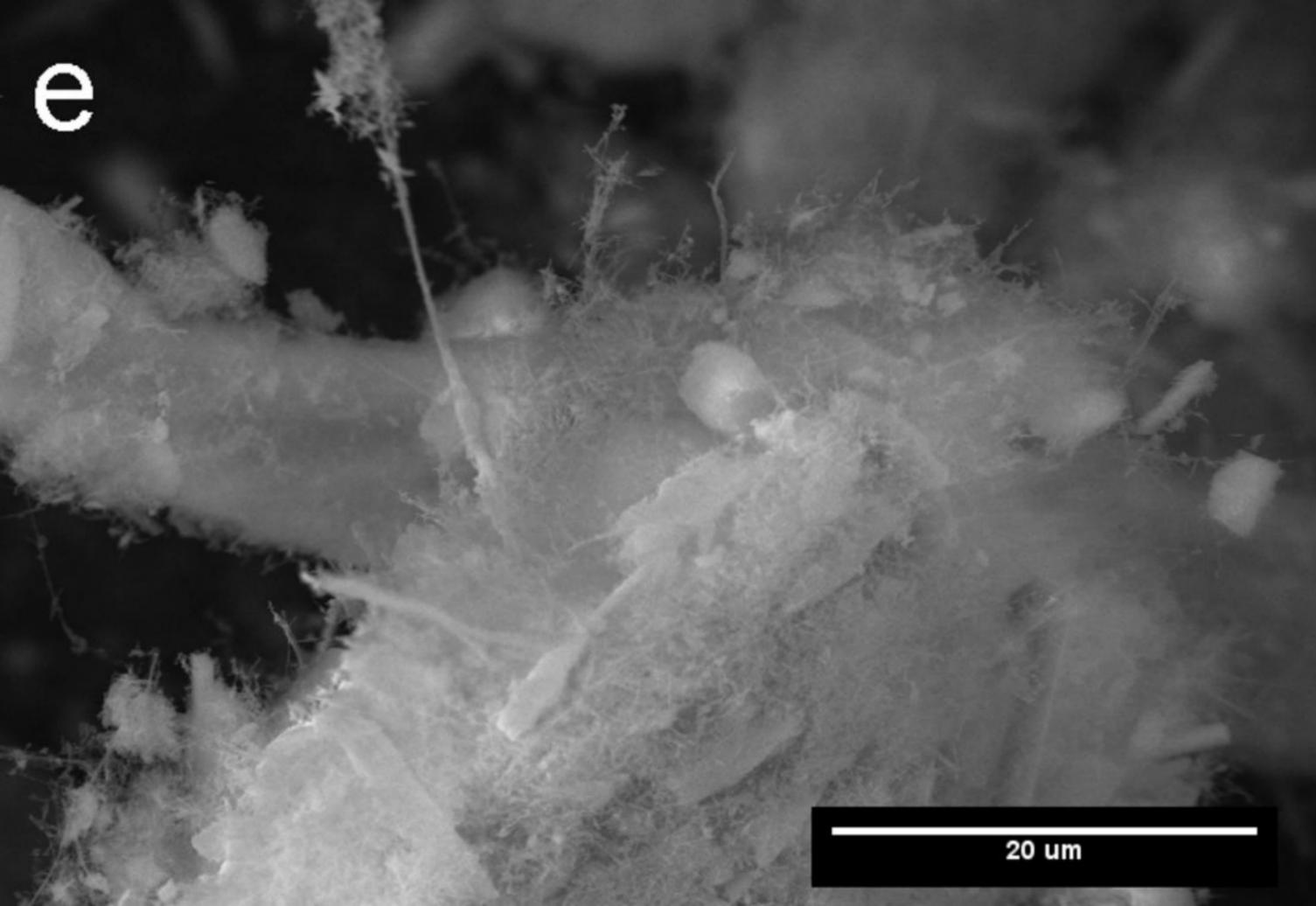
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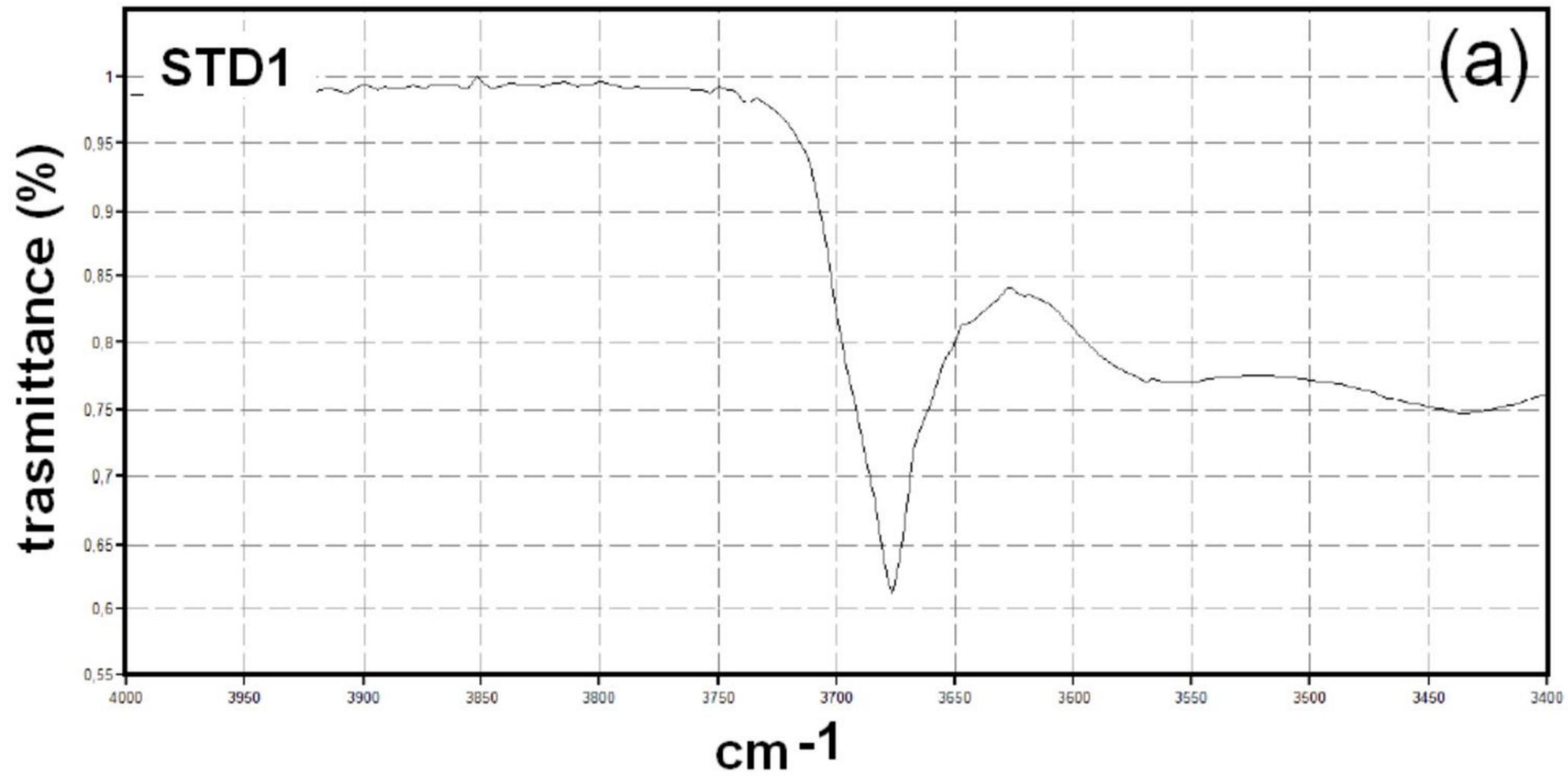
Ctl

Atg

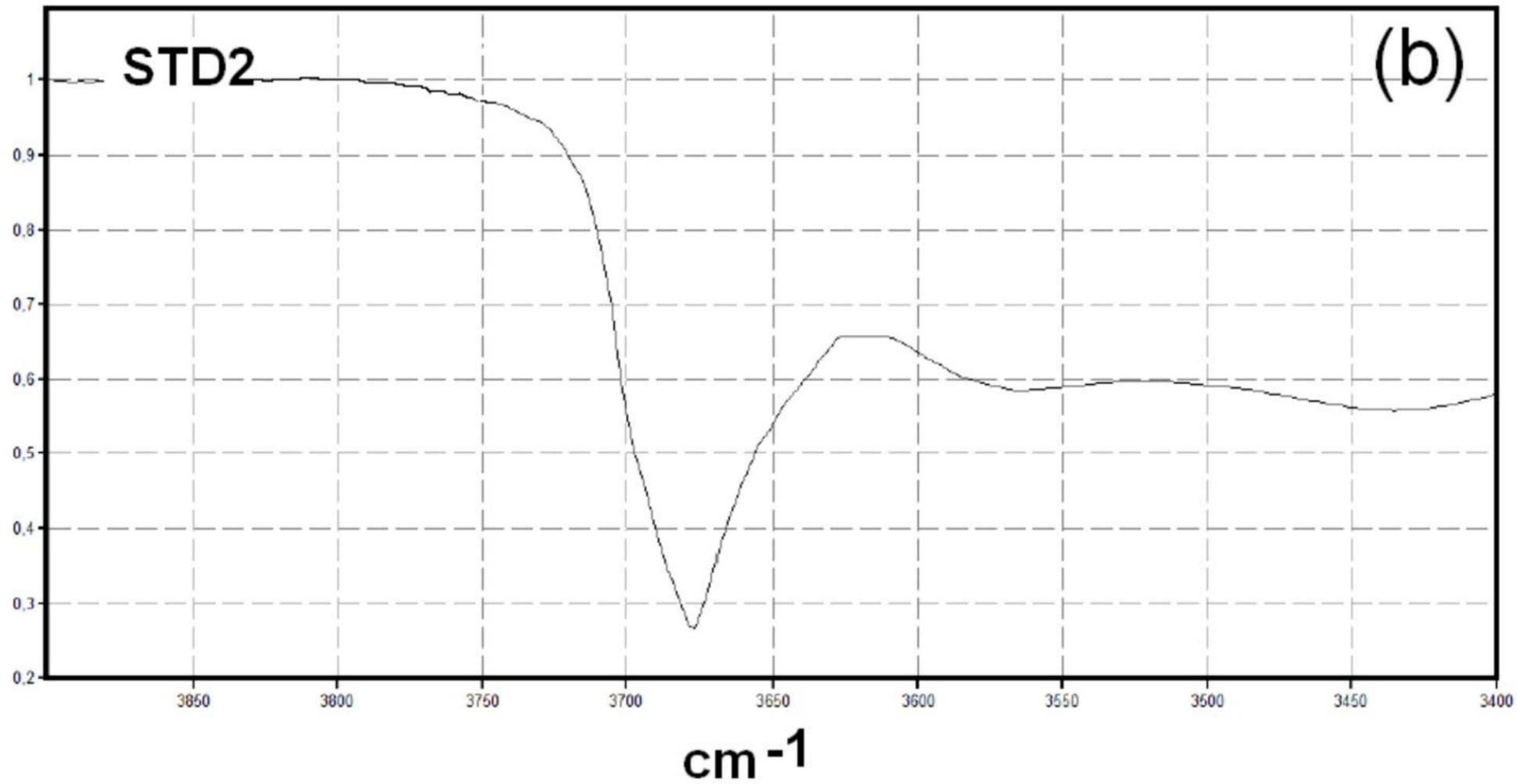
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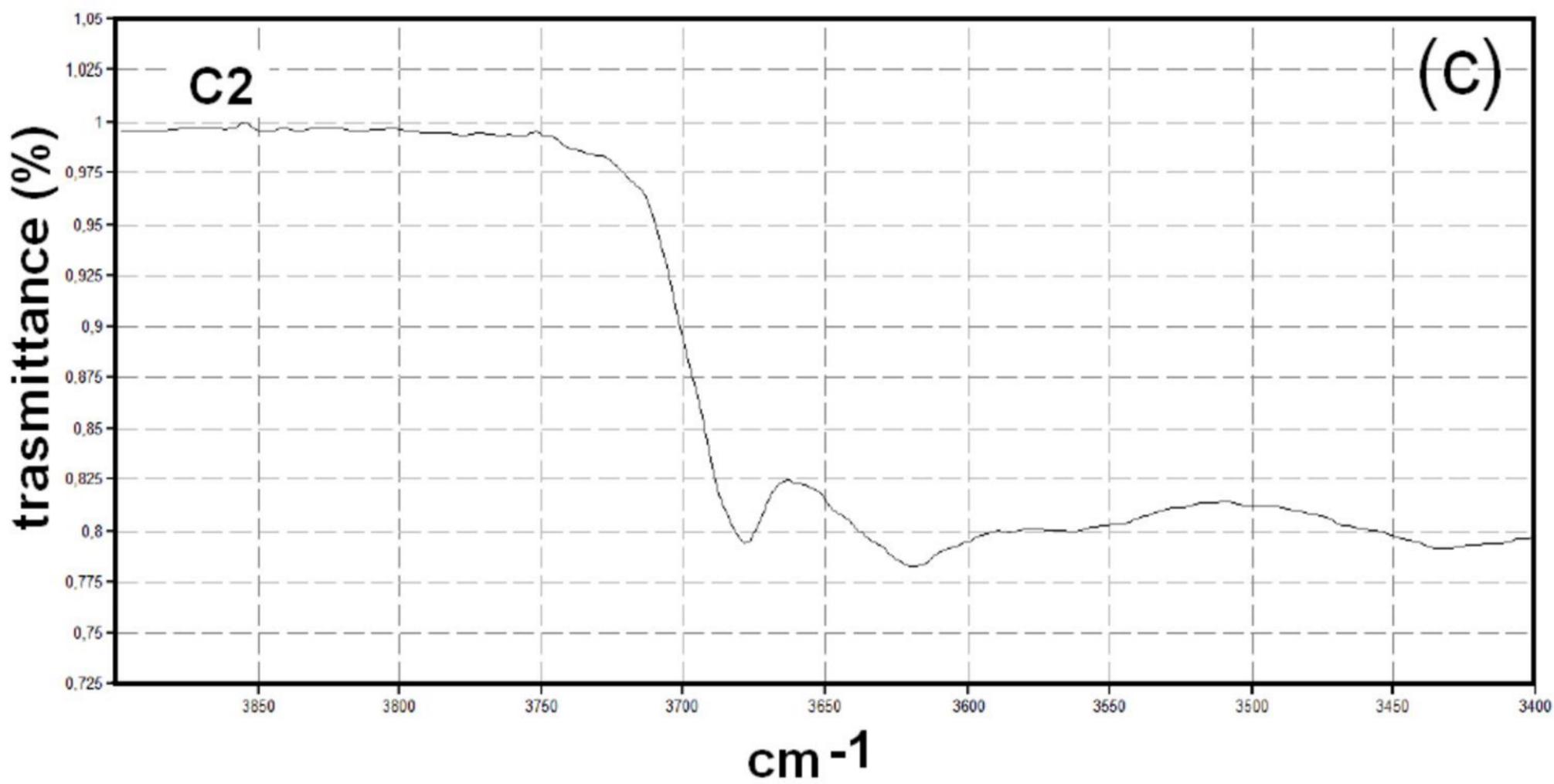
e

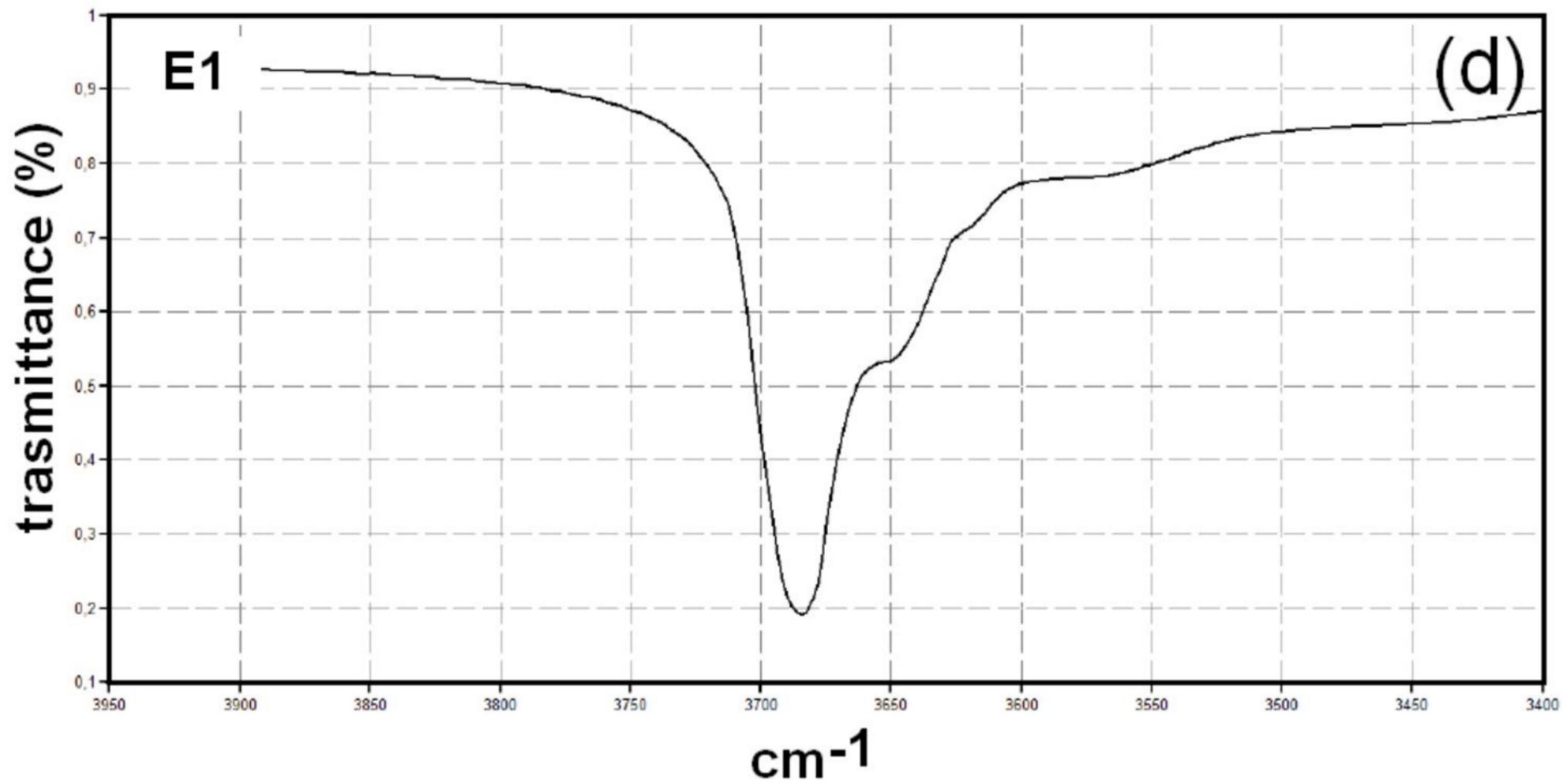


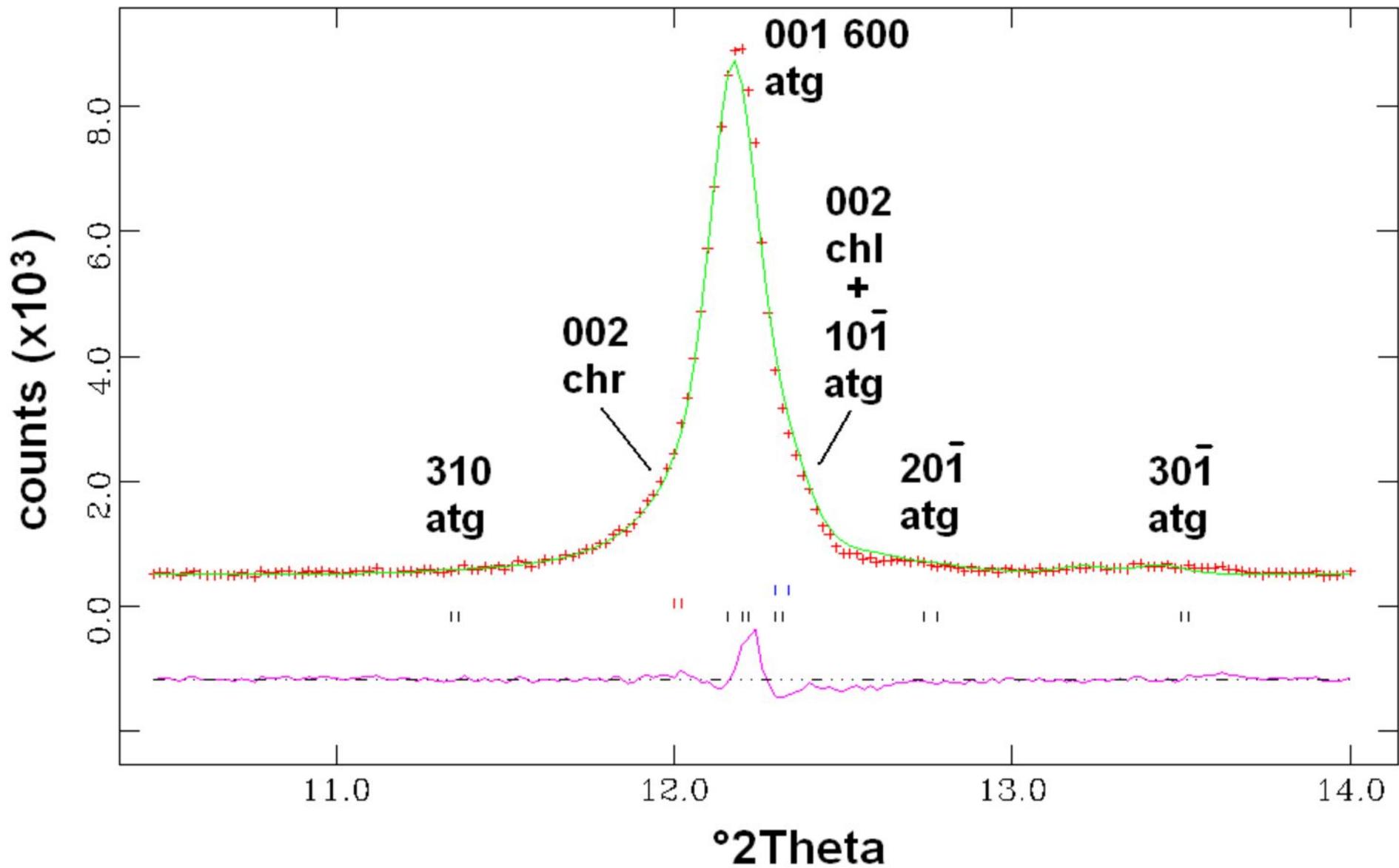


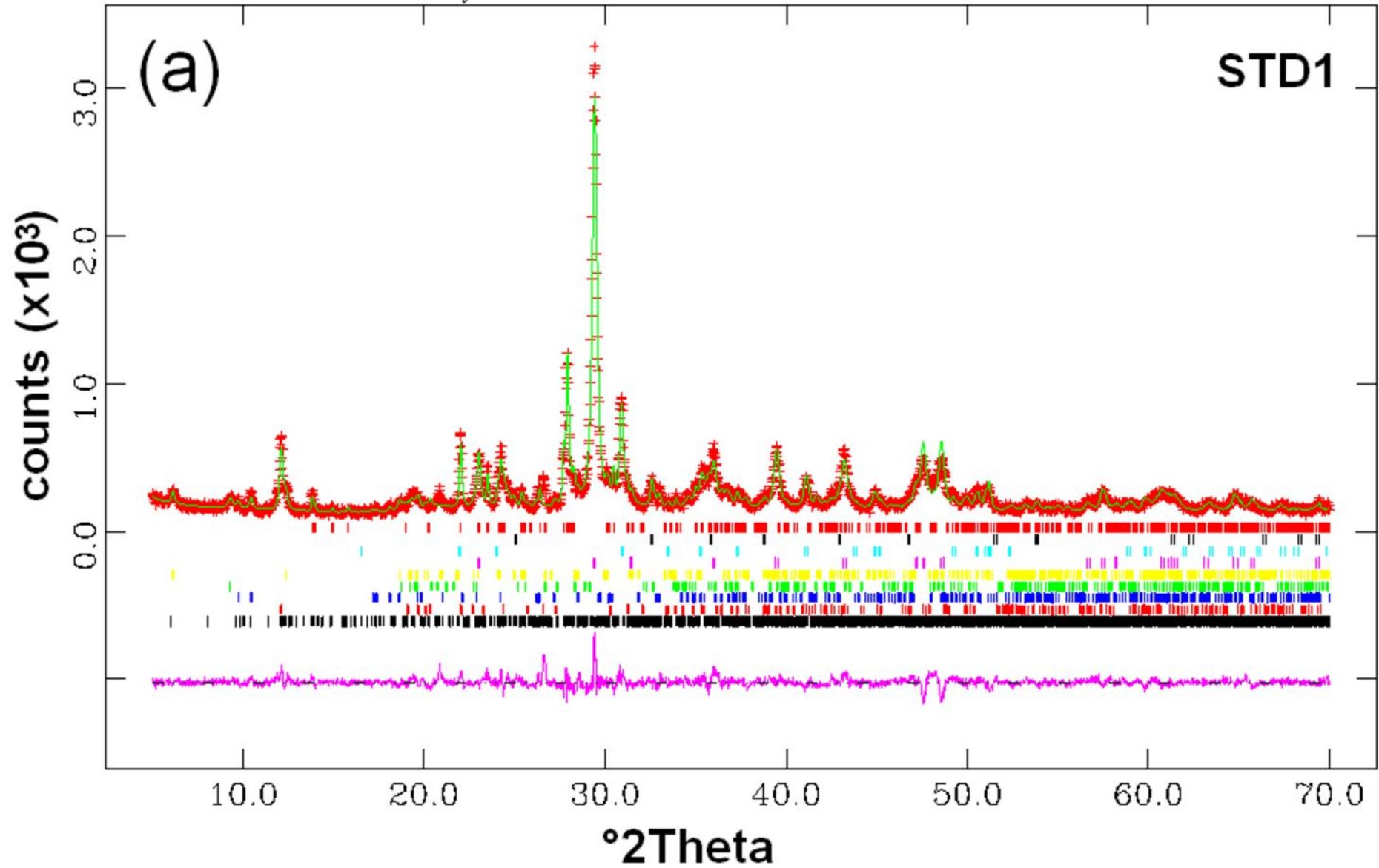
transmittance (%)

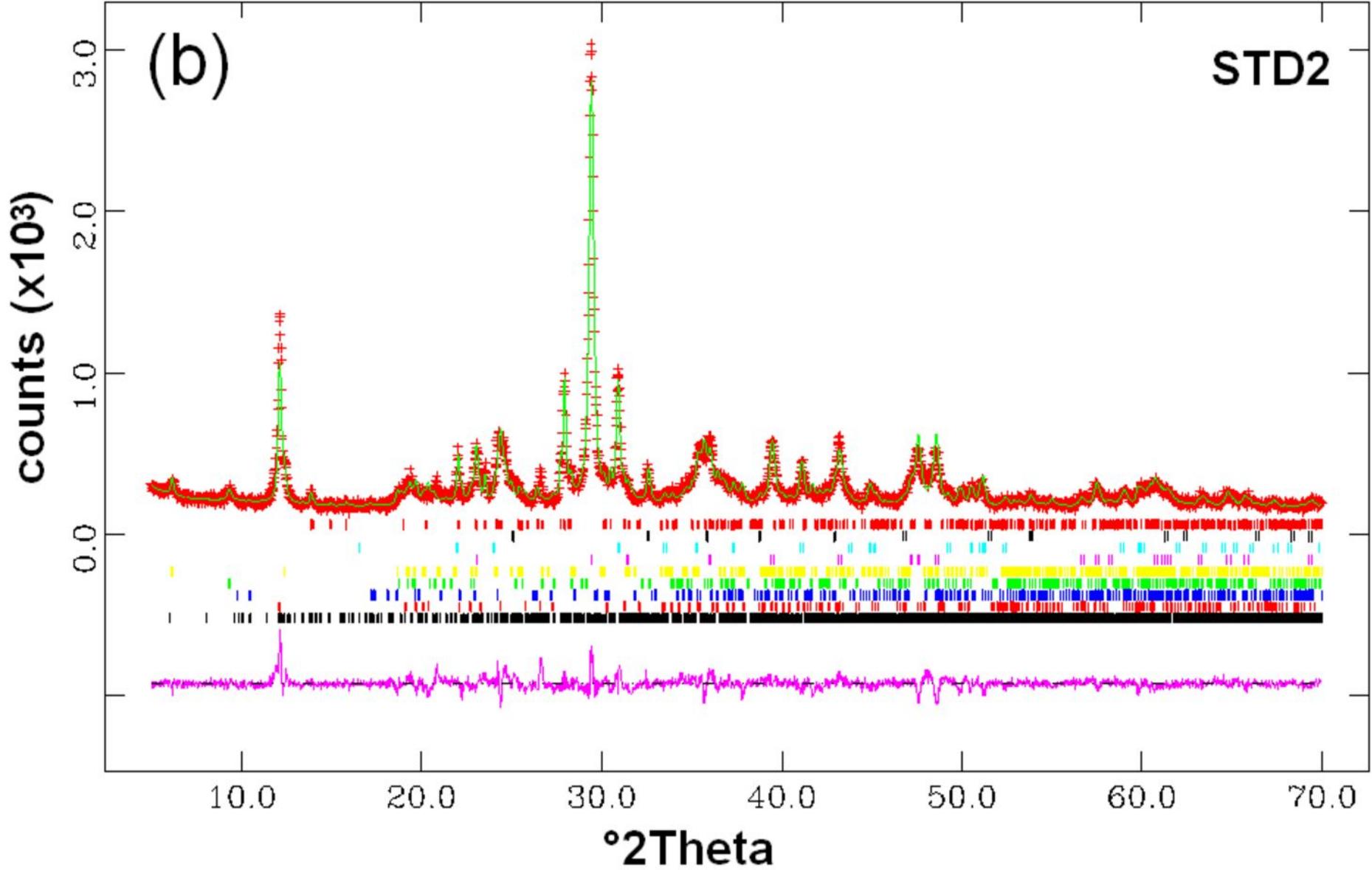






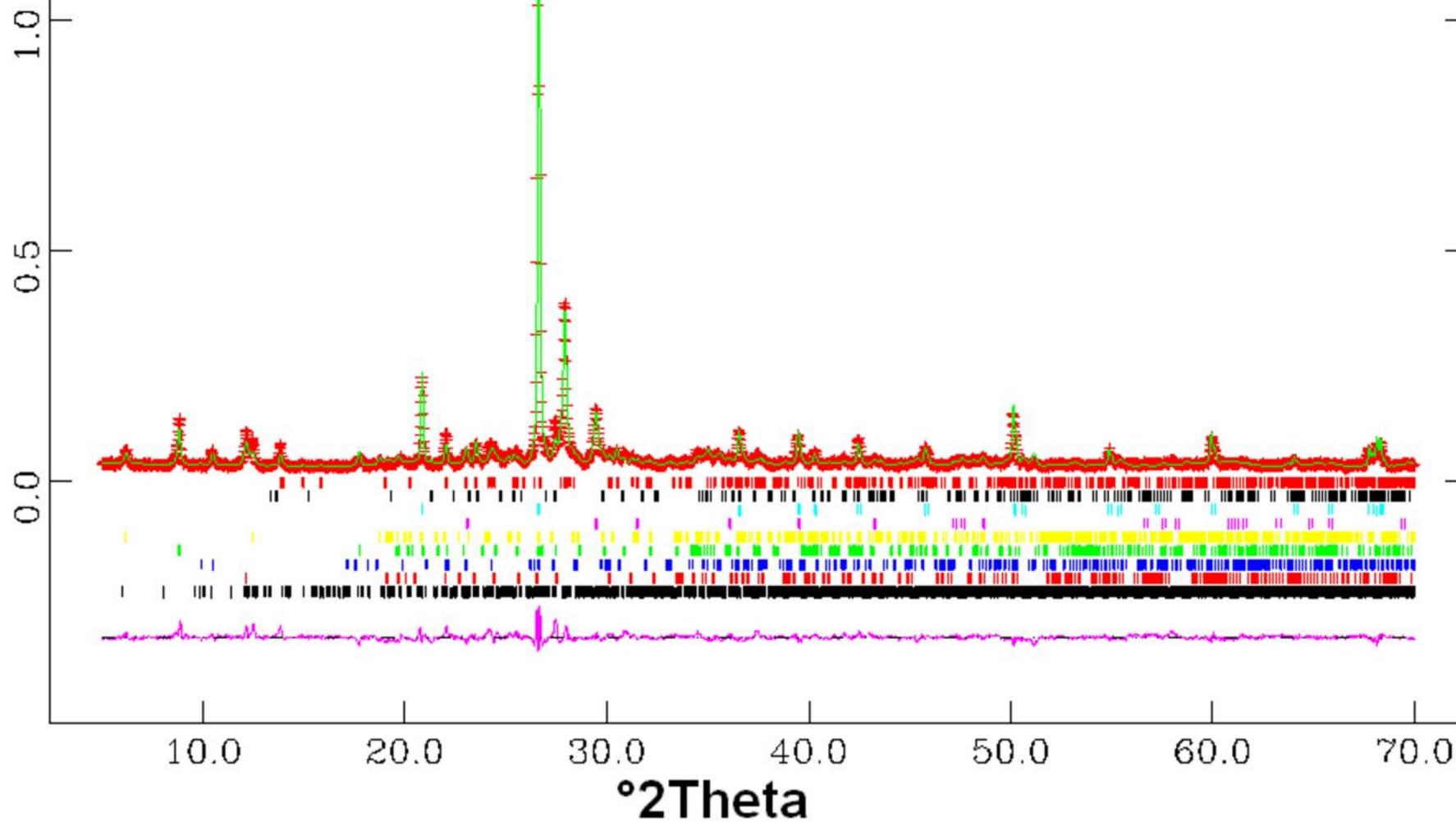






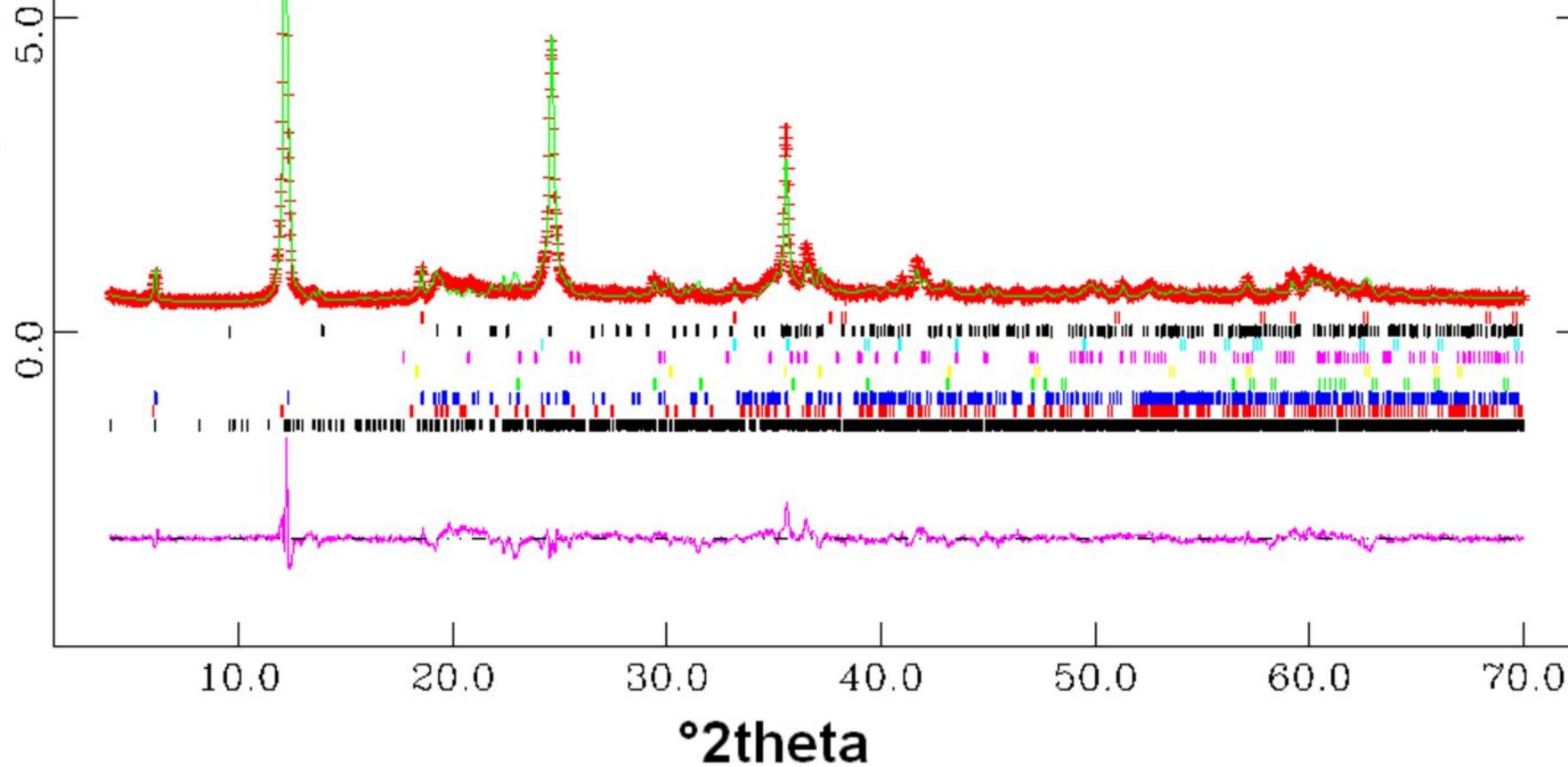
C2

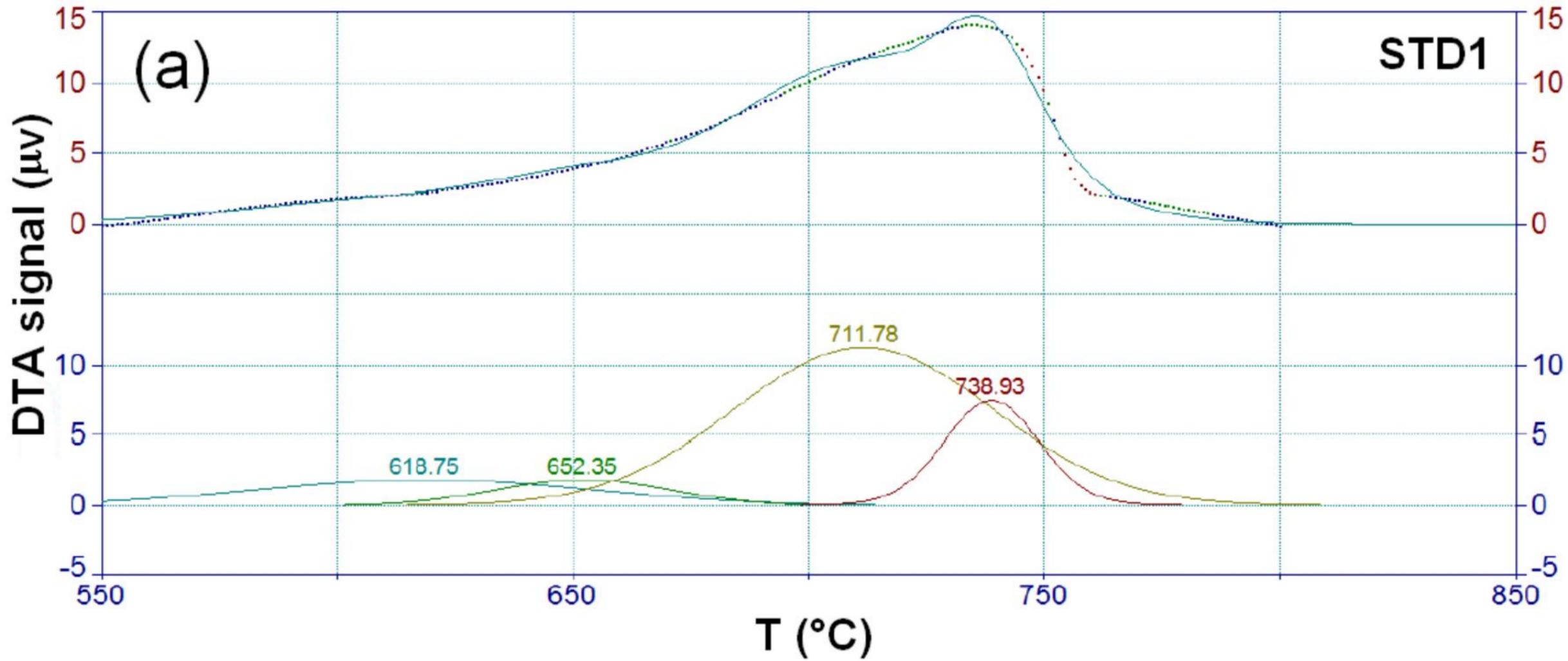
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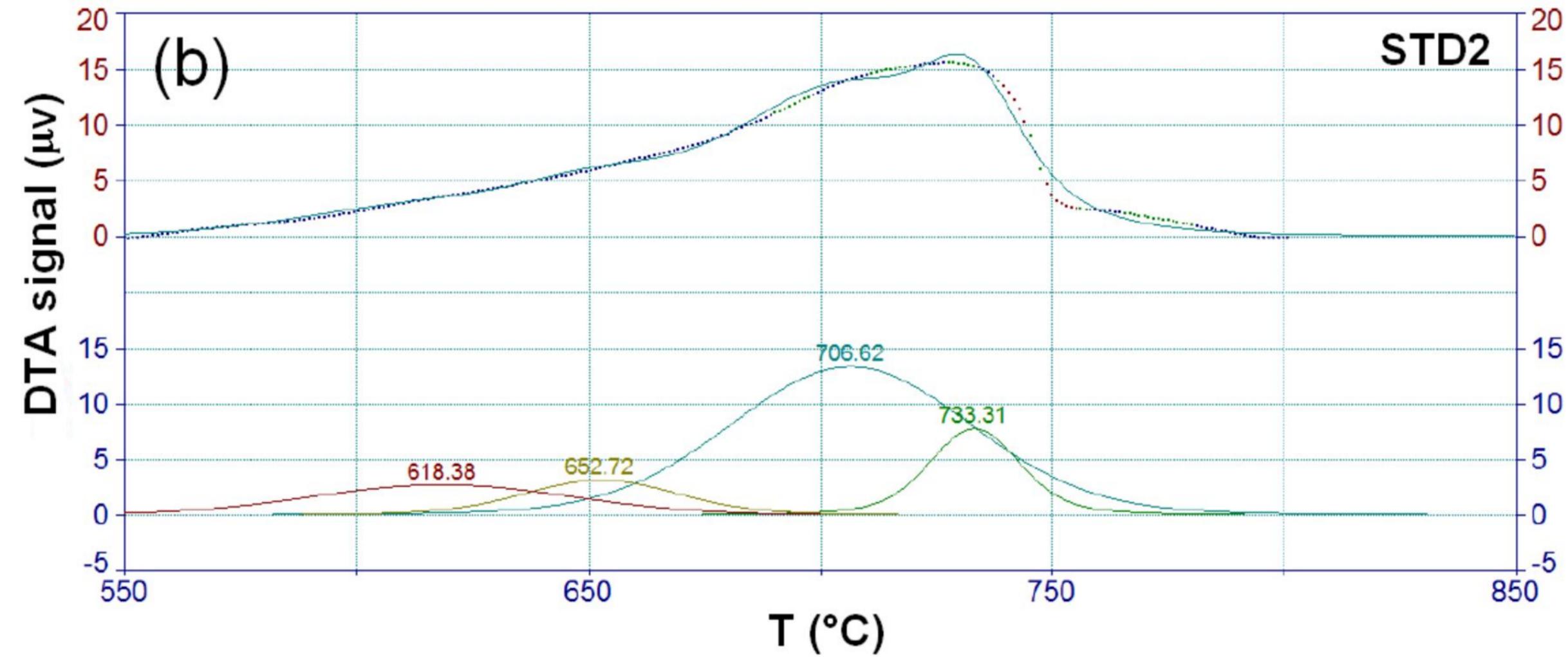
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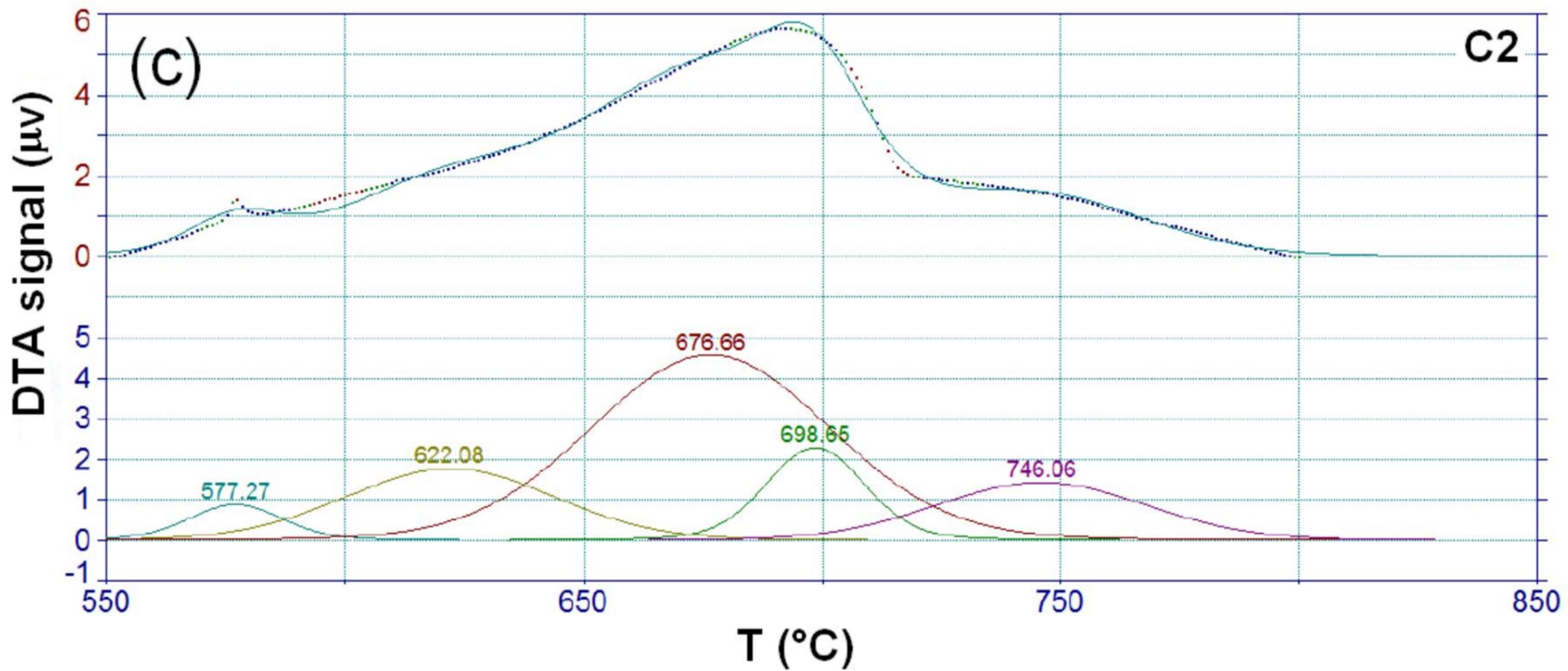
E1

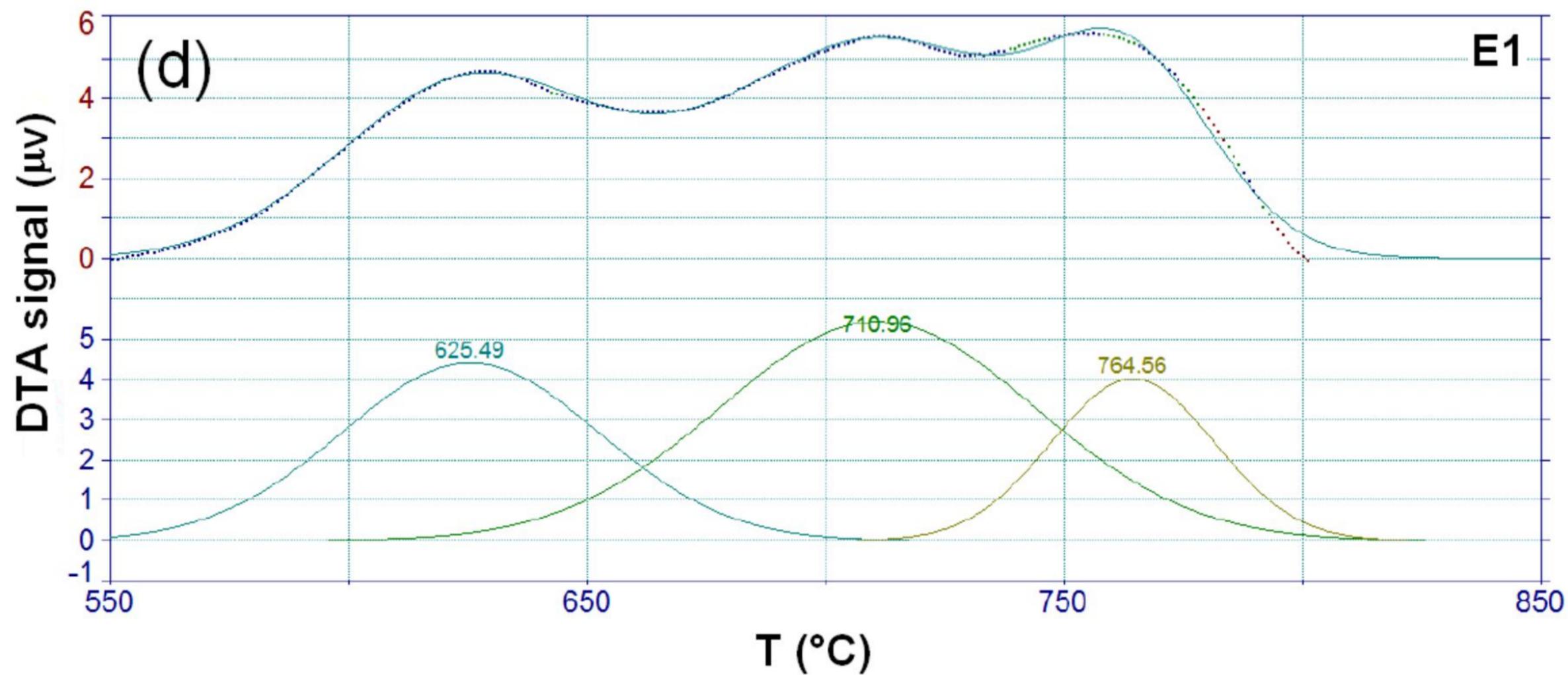
(d)

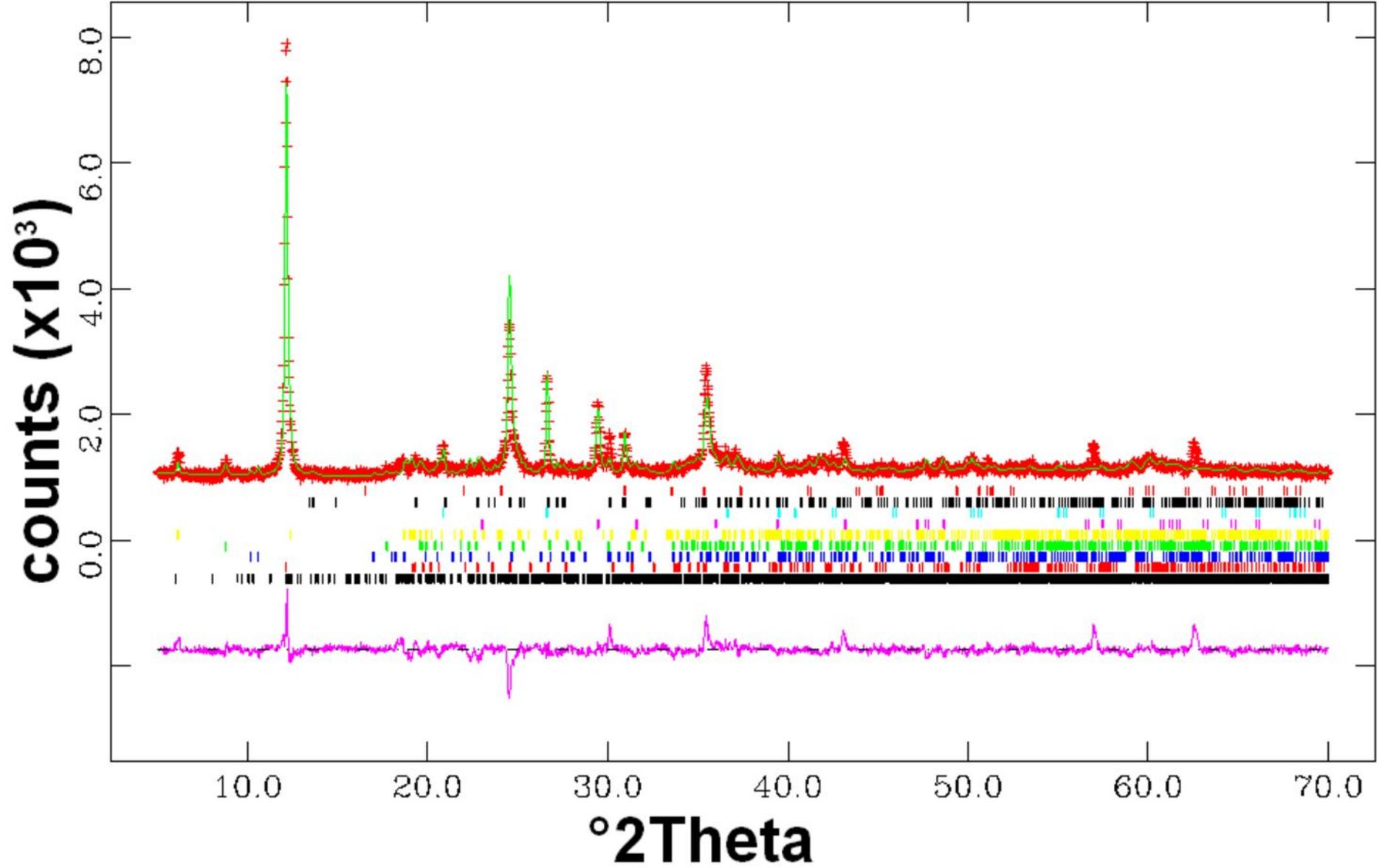
counts ( $\times 10^3$ )

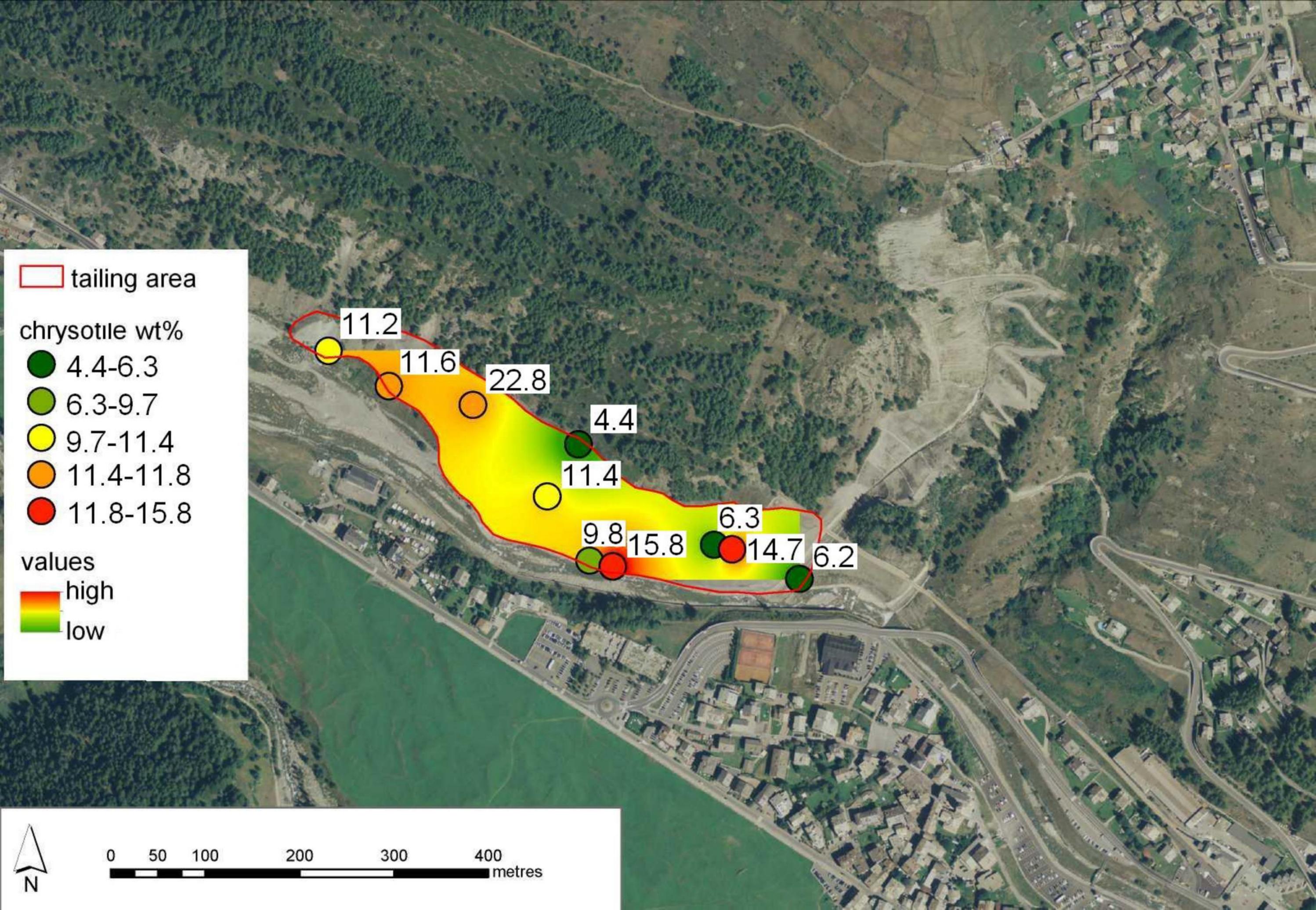






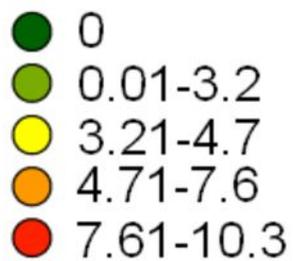




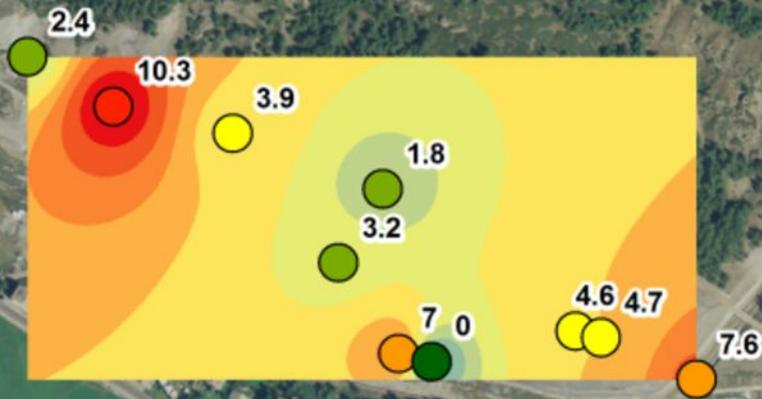
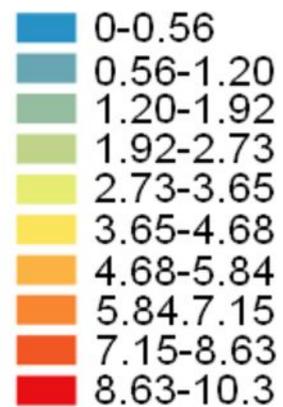


tremolite

wt%



prediction  
map - areas



0 50 100 200 300 400 metres

chrysotile  
wt%

