

## Revision 1

# Kumdykolite, a high-temperature feldspar from an enstatite chondrite

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22

24 ***Abstract***

25           We report the first occurrence of kumdykolite in a meteorite (Sahara 97072, EH3). This  
26 orthorhombic form of albite occurs in the core of a concentrically zoned metal-sulfide nodule. In  
27 contrast to the terrestrial kumdykolite, the meteoritic sample has a domain structure that is  
28 consistent with either orthorhombic ( $Pmnn$ ) or monoclinic ( $P2_1$ ) space groups. The two  
29 symmetries are indicated by the presence or lack, respectively, of  $h + k = 2n+1$  reflections in  
30 [001] selected-area electron diffraction patterns, effects that likely result from different Si-Al  
31 ordering.  $Pmnn$  kumdykolite has only one tetrahedral site for Si and Al, whereas  $P2_1$   
32 kumdykolite would have three tetrahedral sites for Si and one for Al. We propose that  
33 kumdykolite formed above 1300K and cooled rapidly enough to preserve its unique structure.  
34 Apparently, the cooling rate varied on the scale of nanometers allowing the local development of  
35 Si-Al ordering.

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43 **Key words:** kumdykolite, albite polymorph, enstatite chondrite, domain structure, Si-Al ordering

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## 45 ***Introduction***

46 Albite is an important Na-bearing mineral in unequilibrated, metal-rich enstatite (EH3)  
47 chondrites (Schneieder et al. 2002; Lehner et al. 2013a), although it is rare in carbonaceous and  
48 ordinary chondrites. It is intimately intergrown with sulfides and silica produced through silicate  
49 sulfidation in EH3 chondrites (Lehner et al. 2013a, b). Therefore, albite may carry important  
50 information about the sulfidation environment.

51 The albite structure has the potential to record thermal history (e.g., Smith 1974). At  
52 high-temperature, it crystallizes with monoclinic symmetry. Below 1273 K, Si-Al ordering  
53 occurs (e.g., Salje 1985; Salje et al. 1989), which results in the triclinic high-albite structure.  
54 Upon cooling to below 973 K, the ordering leads to unit-cell distortion and formation of low  
55 albite (also triclinic). The known high-pressure (>10 GPa) albite polymorph is lingunite (Liu  
56 1978; Liu and El Goresy 2007), with hollandite-type structure. Recently an albite polymorph  
57 with orthorhombic symmetry was described from eclogite in the ultrahigh-pressure Kumdy Kol,  
58 Kokchetav massif. The polymorph was named kumdykolite and proposed to be metastable,  
59 formed by rapid cooling from high temperature (Hwang et al. 2009).

60 We report the first occurrence of kumdykolite in a meteorite (SAH 97072 EH3). In  
61 contrast to the terrestrial example, this kumdykolite has a domain structure and occurs in two  
62 forms with different Si-Al ordering. The goals of this paper are to describe its occurrence in an  
63 EH3 chondrite, discuss its crystal structure, Si-Al ordering, and its possible formation condition.

## 64 ***Experimental method***

65 Chemical analysis and back-scattered electron (BSE) imaging of a concentrically zoned  
66 metal-sulfide nodule were performed using a thin section of SAH 97072 with an FEI NOVA  
67 scanning electron microscope (SEM). The Si- and Al-rich core of the nodule was extracted and

68 thinned to electron transparency using a focused-ion beam (FIB). Transmission electron  
69 microscope (TEM) data, bright-field TEM (BFTEM) images, and selected-area electron  
70 diffraction (SAED) patterns were acquired with JEOL2000FX and JEOL JEM 4000EX TEMs.  
71 The compositions of the grains were measured with an energy-dispersive X-ray spectrometer  
72 attached to a JEOL 2010F TEM. We used the Cliff-Lorimer thin-film approximation and ZAF  
73 correction to quantify the analysis using Genesis software. We used a  $\sim 0.1\text{-}\mu\text{m}$  electron beam  
74 and corrected the analyses for Na-loss by using an albite standard under identical conditions.

## 75 **Results**

76 The nodule consists of a kamacite-troilite mantle and a core containing oldhamite,  
77 niningerite, Zn-daubreelite, and S-rich porous silica (Fig. 1a-c). The porous silica contains grains  
78 of  $\text{NaAlSi}_3\text{O}_8$  (Table 1) that have SAED patterns consistent with kumdykolite (Figs. 2, 3).

79 Kumdykolite was reported in space group  $Pmnn$  (Hwang et al. 2009), a symmetry that  
80 requires systematic absences for reflections with  $h + k = 2n + 1$  in SAED patterns along  $[001]$ .  
81 Patterns from some areas of the meteoritic kumdykolite are consistent with these systematic  
82 absences (Figs. 2b, d, 3b), but others display faint  $h + k = 2n + 1$  reflections (Figs. 2c, 3c)  
83 indicating that kumdykolite can occur with two symmetries. Diffraction of dynamically scattered  
84 electrons can result in reflections with  $h = 2n + 1$  along  $[100]^*$  and with  $k = 2n + 1$  along  $[010]^*$ .  
85 However,  $h + k = 2n + 1$  reflections with  $h \neq 0$  and  $k \neq 0$ , as observed in Figs. 2c and 3c, can not  
86 be interpreted as diffraction of dynamically scattered electrons and therefore violate the  
87 selection rule for  $Pmnn$ . Multiple space groups are consistent with these SAED patterns (Figs.  
88 2c, 3c). However, the patterns overlap for  $n$  glide reflections (the strong ones in Figs. 2c, 3c),  
89 indicating one may be a derivative symmetry of the other (Buerger 1947). Assuming the same  
90 unit cell, a possible derivative of  $Pmnn$  is  $P2_1$ , which is also the space group of the high-



91 temperature synthetic anorthite published by Takeuchi et al. (1973) and svyatoslovite  
92 (Krivovichev et al. 2012). Based on these anorthite analogues, we propose that the space group  
93 of kumdykolite having  $h + k = 2n + 1$  reflections is  $P2_1$ .

94 The occurrence of different space groups for the same mineral is common in feldspars  
95 and is indicated by  $a$ ,  $b$ ,  $c$ , and  $d$ - type reflections in SAED patterns (Kroll and Ribbe 1980;  
96 Carpenter et al. 1985; Redfern 1992; Tribaudino and Angel 2012) as well as by antiphase  
97 domains and boundaries in TEM images (e.g., Carpenter 1991; Van Tendeloo et al. 1989;  
98 Németh et al. 2007). Therefore, we conclude that the dark and bright areas in kumdykolite  
99 BFTEM images (Figs 2a, 3a) corresponding to SAED patterns with and without the  $h + k = 2n$   
100  $+1$  reflections, are antiphase domains and boundaries.

## 101 ***Discussion***

### 102 **Structure and Si-Al ordering**

103 The fractional coordinates for the Si-Al and O sites for  $Pmnn$  kumdykolite based on the  
104 Takeuchi et al. (1973) structure model for  $P2_1$  synthetic  $\text{CaAl}_2\text{Si}_2\text{O}_8$  were proposed by Hwang et  
105 al. (2009). They generated the coordinates for one Si-Al site and 3 O sites by shifting the origin  
106 in the  $P2_1$  structure and eliminating the equivalent positions. However, in this structure all O are  
107 in general positions, which results in unbalanced charge and 6-coordinated Si-Al. This problem  
108 can be solved if two O occupy special positions. Furthermore, following the method of Hwang et  
109 al. (2009), a special Na site can be generated from the Takeuchi structure. Thus, we propose new  
110 atomic coordinates for  $Pmnn$  kumdykolite (Table 2) by modifying the structure of Hwang et al.  
111 (2009) and generating a Na site.

112  $Pmnn$  kumdykolite contains one framework cation site (Fig. 4a), with a disordered  
113 distribution for Al and Si. However, the Na likely occupies a split site similar to Ca in the

114 Takeuchi structure. We assume that  $P2_1$  kumdykolite contains ordered Si and Al in a structure  
115 analogous to that of synthetic  $P2_1$  anorthite and its natural analogue, svyatoslovite (Takeuchi et  
116 al. 1973; Krivovichev et al. 2012). Although Si and Al ordering occurs via two Si and two Al  
117 sites in the Takeuchi structure (Fig. 4b), there would be one site for Al and three for Si in  $P2_1$   
118 kumdykolite. By analogy to feldspars, we assume the low-symmetry kumdykolite structure  
119 develops during cooling.

### 120 **EH3 kumdykolite formation conditions**

121 The kumdykolite molar volume ( $104 \text{ cm}^3 \text{ mol}^{-1}$ ) may provide an estimate of formation  
122 pressure. It is significantly greater than that of lingunite ( $70 \text{ cm}^3 \text{ mol}^{-1}$ ), the high-pressure albite  
123 polymorph, but similar to that of low albite ( $100 \text{ cm}^3 \text{ mol}^{-1}$ ). Therefore, we presume that  
124 kumdykolite did not form at high pressure.

125 The kumdykolite formation temperature can be postulated by comparison with the high-  
126 temperature  $\text{CaAl}_2\text{Si}_2\text{O}_8$  polymorphs svyatoslavite and misteinbergite and by considering the  
127 proposed conditions of silicate sulfidation in EH3 chondrites (Lehner et al. 2013b). Svyatoslavite  
128 and misteinbergite crystallize between 1373 and 1673 K (Abe et al. 1991; Abe and Sungawa  
129 1995) and have been used as indicators of rapid cooling (Sokol et al. 1998; Krivovichev et al.  
130 2012; Nestola et al. 2010). Lehner et al. (2013a, b) reported that  $\text{NaAlSi}_3\text{O}_8$  could form during  
131 sulfidation of Al-bearing pyroxenes through release of the  $\text{Al}_2\text{O}_3$  and  $\text{SiO}_2$  components in a Na-  
132 enriched environment. Thermodynamic modeling of phase relations indicates that silicate  
133 sulfidation occurred at  $\sim 1400$  to  $1600$  K, a range consistent with svyatoslavite crystallization.  
134 Therefore we propose that kumdykolite formed at high-temperature, presumably  $> 1300$  K, and  
135 cooled rapidly preserving its structure. Furthermore, Si-Al ordering and both high- and low-  
136 symmetry kumdykolite suggest the cooling rate varied on the scale of nanometers.

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199 319.

## 200 **Figure captions**

201 **FIGURE 1.** Petrographic setting of kumdykolite from EH3 chondrite SAH 97072. (a) BSE  
202 image of kumdykolite and S-rich porous silica at the center of a metal-sulfide nodule (Lehner et  
203 al. 2011). (b) Location of the FIB extraction (white rectangle). (c) BFTEM image of  
204 kumdykolite and porous silica. kum= kumdykolite, px = pyroxene, k = kamacite, ng =  
205 niningerite, od = oldhamite, psil = porous silica, tr = troilite, zdb = zincian daubreelite.

206

207 **FIGURE 2.** BFTEM image and SAED patterns of the kumdykolite grain in the bottom center of  
208 Fig. 1c. Reflections with  $h = 2n + 1$  along  $[100]^*$ , with  $k = 2n + 1$  along  $[010]^*$ , and with  $l = 2n + 1$   
209 along  $[201]^*$  and  $[001]^*$  can occur because of diffraction of dynamically scattered electrons and  
210 are therefore not considered for the symmetry analysis. (a) Light and dark areas of the BFTEM  
211 image are indicative of antiphase domains. (b) SAED pattern along  $[10\bar{2}]$ . (c) SAED pattern  
212 along  $[001]$  tilted  $-40^\circ$  around  $[010]^*$  with respect to the pattern in panel (b). Faint  $h + k = 2n + 1$   
213 reflections (white arrows) violate the selection rule for  $Pmnn$  but are consistent with  $P2_1$ . (d)  
214 SAED pattern along  $[100]$  tilted  $+50^\circ$  around  $[010]^*$  with respect to the pattern in panel (b).

215

216 **FIGURE 3.** BFTEM image and SAED patterns of kumdykolite in the top center of Fig. 1c.  
217 Reflections with  $h = 2n + 1$  along  $[100]^*$  and with  $k = 2n + 1$  along  $[010]^*$  in SAED patterns can  
218 occur because of diffraction of dynamically scattered electrons, and are therefore not considered  
219 for the symmetry analysis. (a) Areas of the BFTEM image with and without antiphase domains  
220 (black and white circles, respectively). The black stripes parallel to  $\langle 100 \rangle$  and  $\langle 110 \rangle$  are caused  
221 by amorphization. They increase in width with exposure to the electron beam. (b) SAED pattern  
222 along  $[001]$  from the area in the white circle in (a) shows only strong  $h + k = 2n$  reflections. (c)  
223 SAED pattern  $[001]$  from the area in the black circle of (a) shows faint  $h + k = 2n + 1$  reflections  
224 (white arrows) that violate the selection rule for  $Pmnn$  space group and are consistent with  $P2_1$ .

225

226 **FIGURE 4.** Crystal structures of  $Pmnn$  kumdykolite and  $P2_1$  svyatoslavite along  $[001]$ . Black  
227 rectangles outline the unit cells. The extra-framework cations (black balls) occur in split sites. (a)  
228 Kumdykolite in  $Pmnn$  space group has a disordered Si-Al framework as Si and Al are distributed  
229 in one site. Grey polyhedra represent both  $\text{SiO}_4$  and  $\text{AlO}_4$  tetrahedra. (b) Svyatoslavite in  $P2_1$   
230 space group has an ordered Si-Al framework in which Si and Al occupy separate  
231 crystallographic sites. Black and light grey polyhedra represent  $\text{SiO}_4$  and  $\text{AlO}_4$  tetrahedra,  
232 respectively.

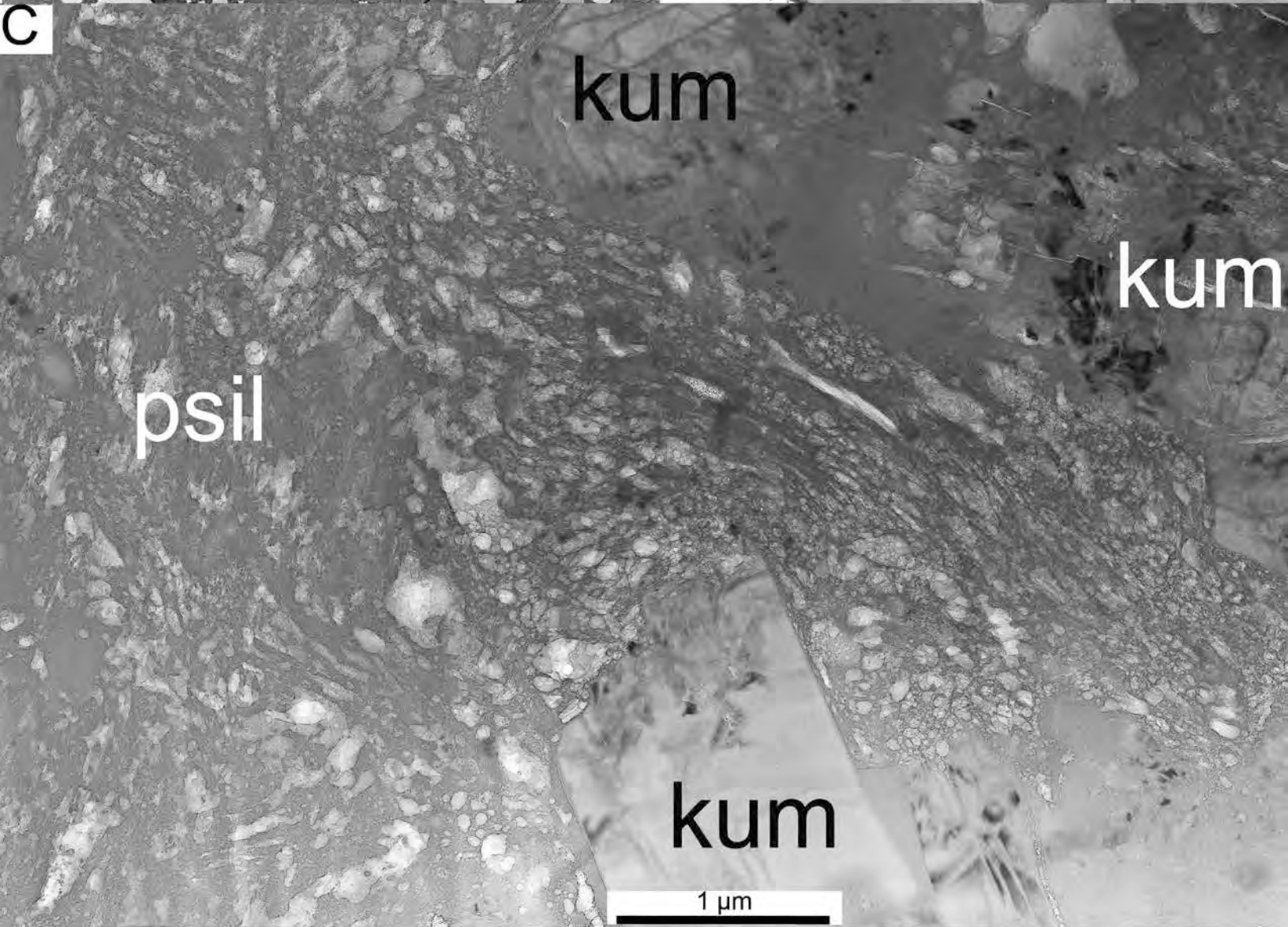
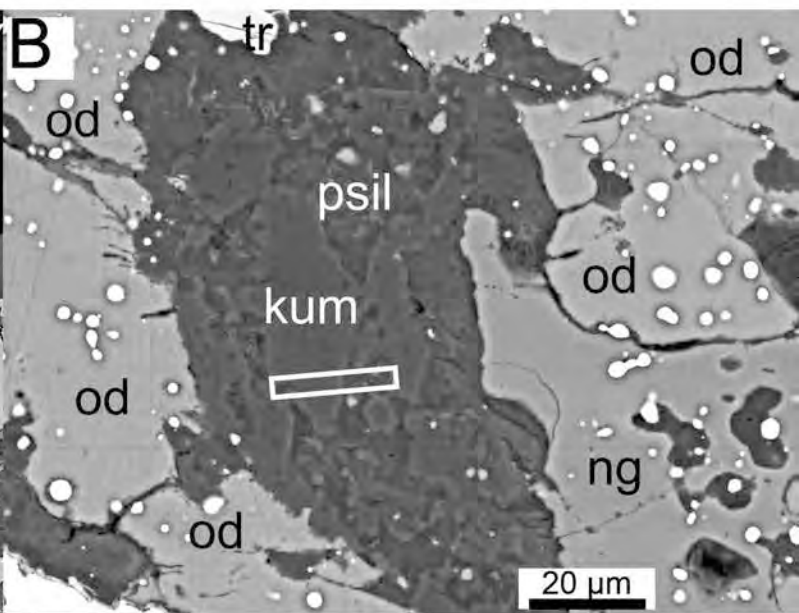
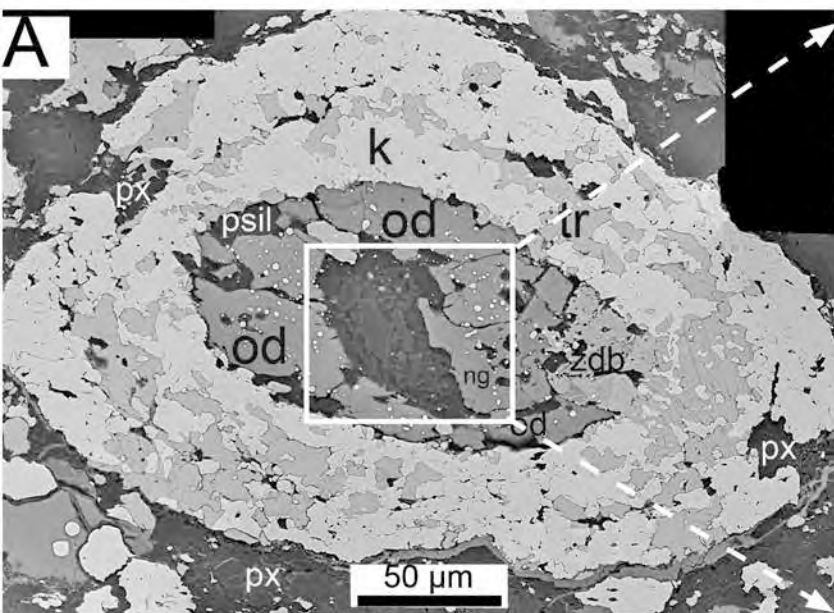
233

**TABLE 1.** Semi-quantitative chemical composition of kumdykolite. K and Ca were below the detection limit. The Na-content of <1 atom per formula unit points to Na-loss. The analytical error is estimated as  $\pm 5\%$ , relative.

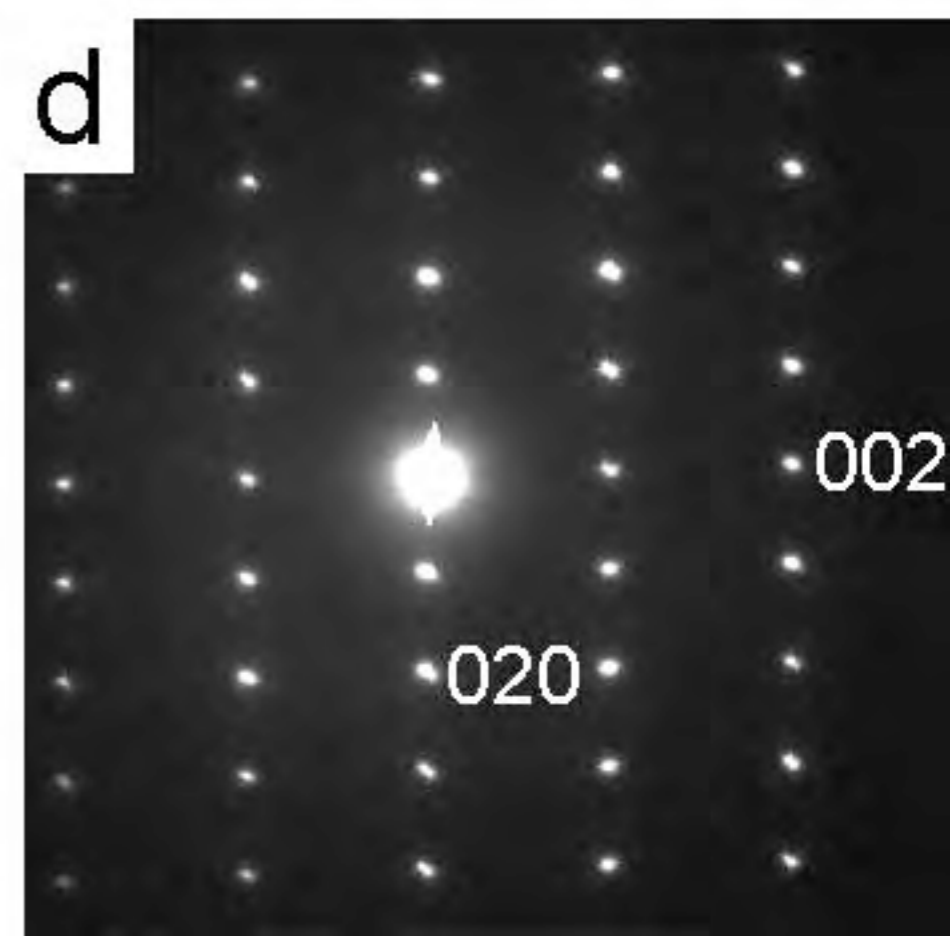
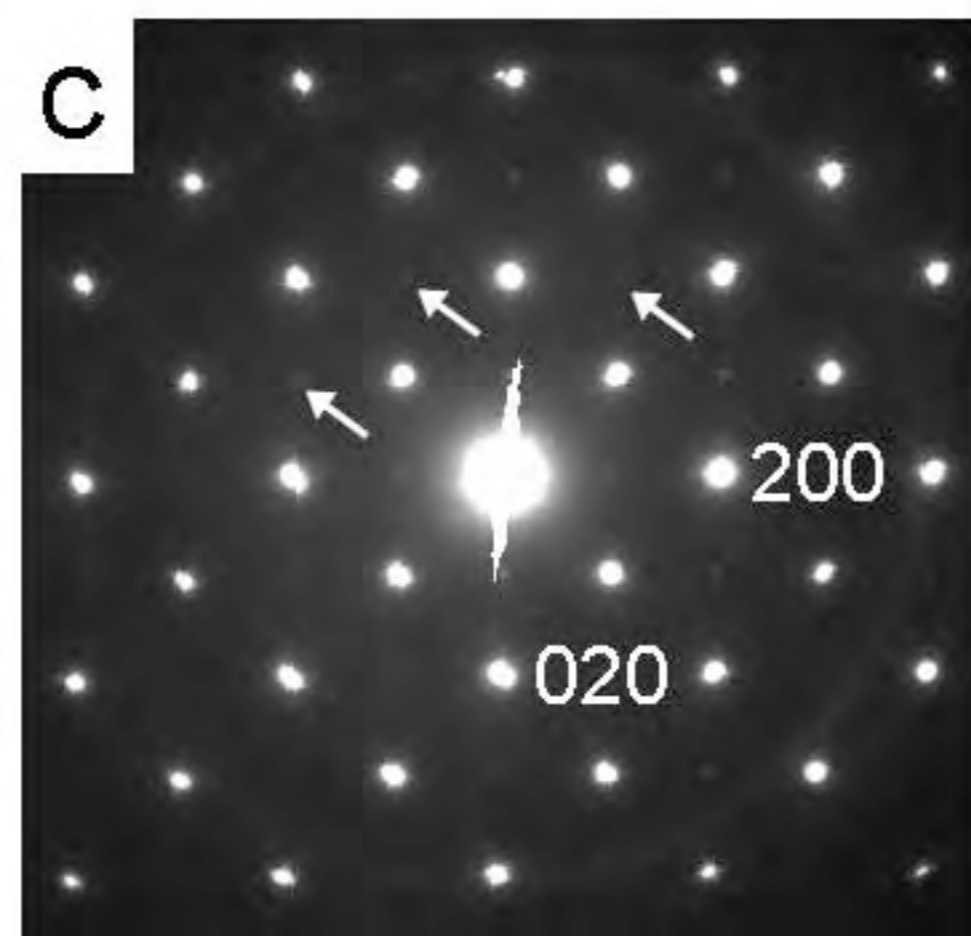
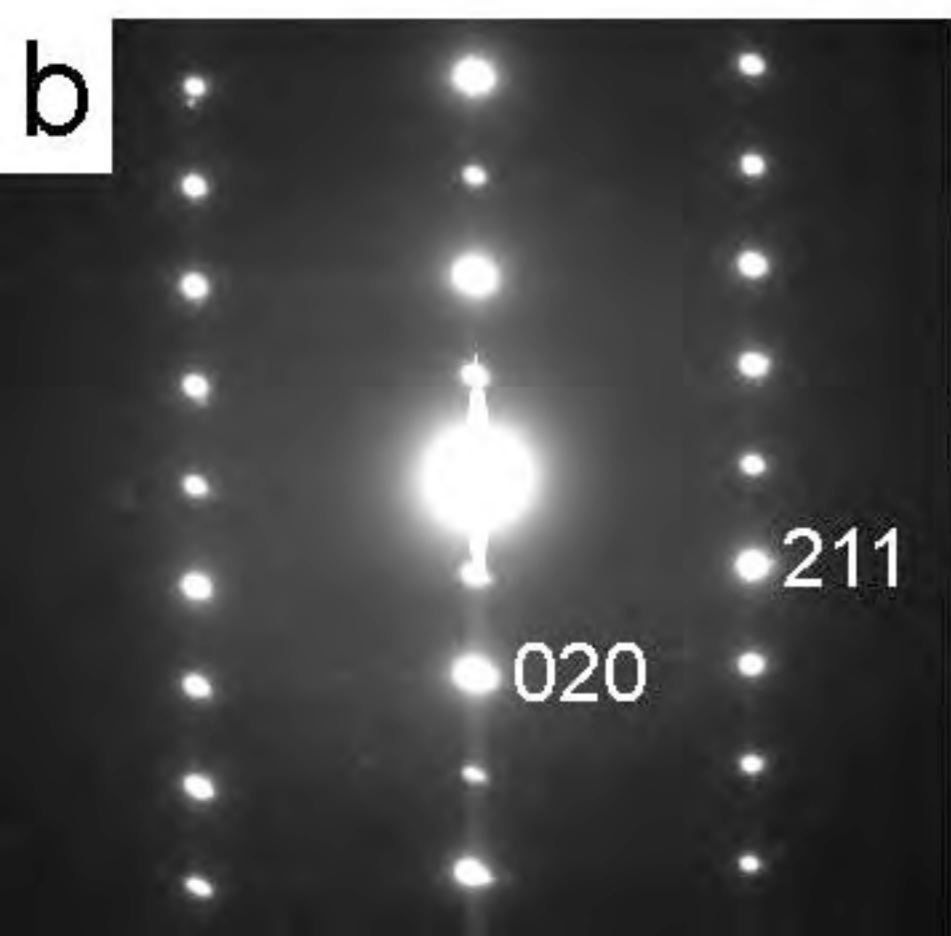
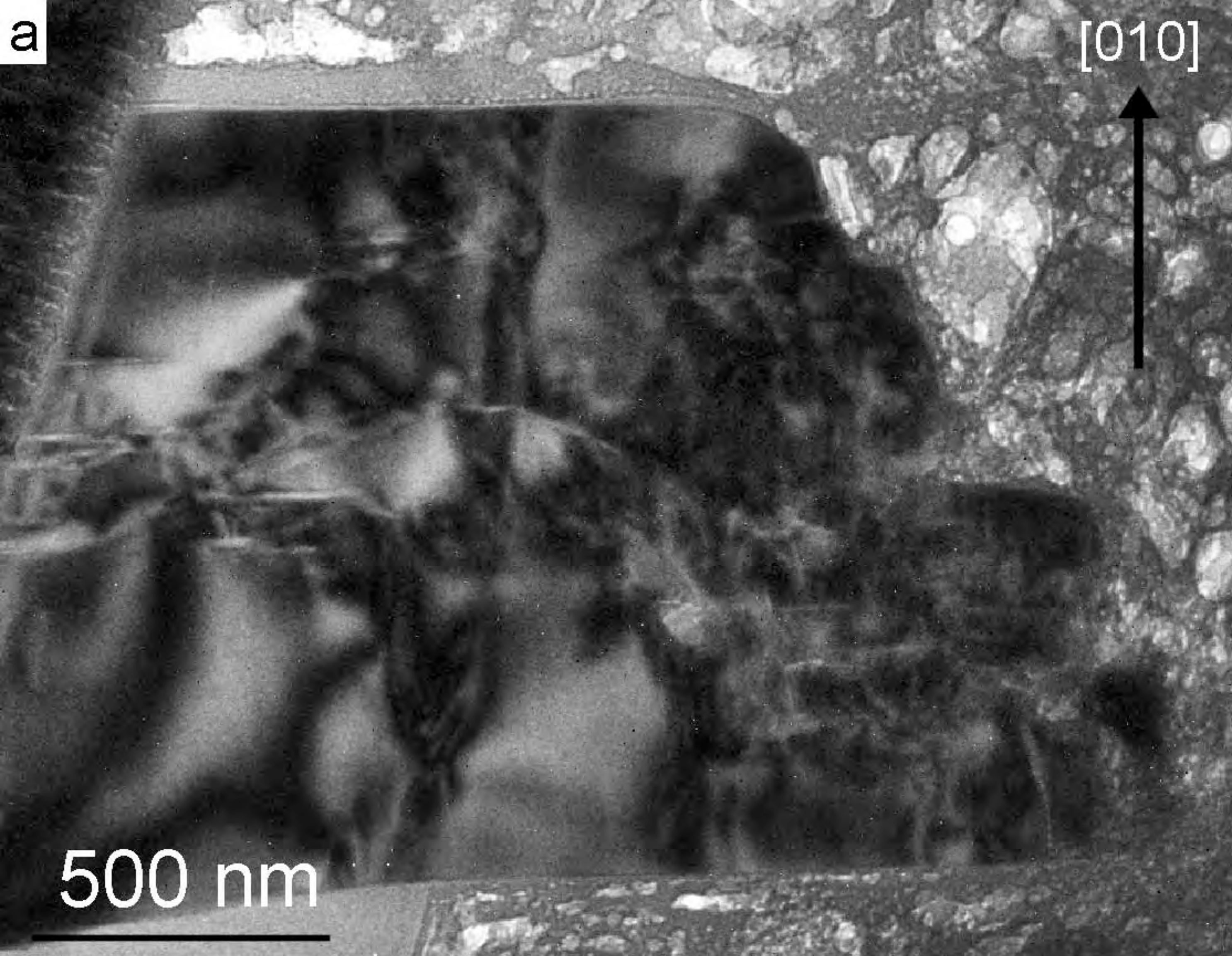
	Spot 1	Spot 2	Spot 3	Spot 4	Spot 5	Spot 6
SiO <sub>2</sub>	70.60	69.65	71.30	70.53	69.27	70.52
Al <sub>2</sub> O <sub>3</sub>	19.99	19.72	20.19	19.97	19.61	19.96
Na <sub>2</sub> O	9.41	10.63	8.51	9.50	11.12	9.52
Based on 8 O						
Si	3.04	3.02	3.06	3.04	3.01	3.04
Al	1.01	1.01	1.02	1.02	1.00	1.01
Na	0.79	0.89	0.71	0.79	0.94	0.80
$\Sigma$	4.84	4.92	4.79	4.85	4.95	4.85

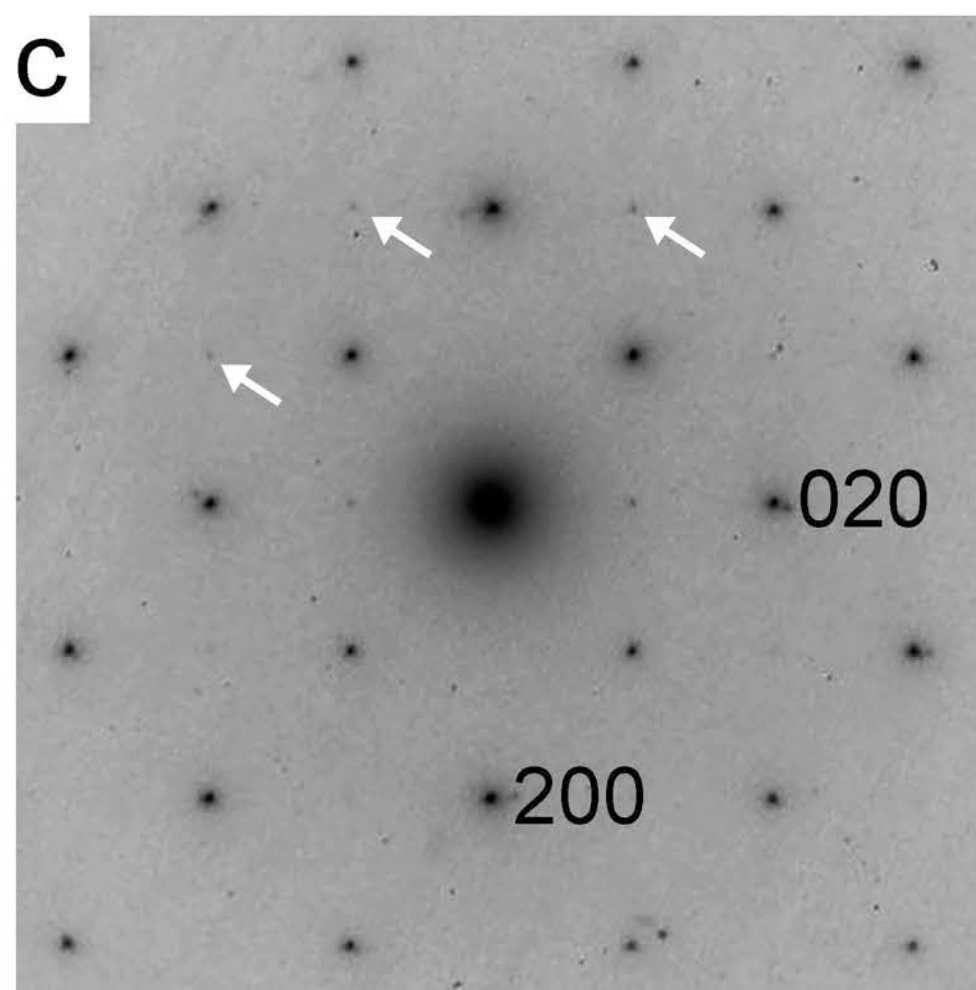
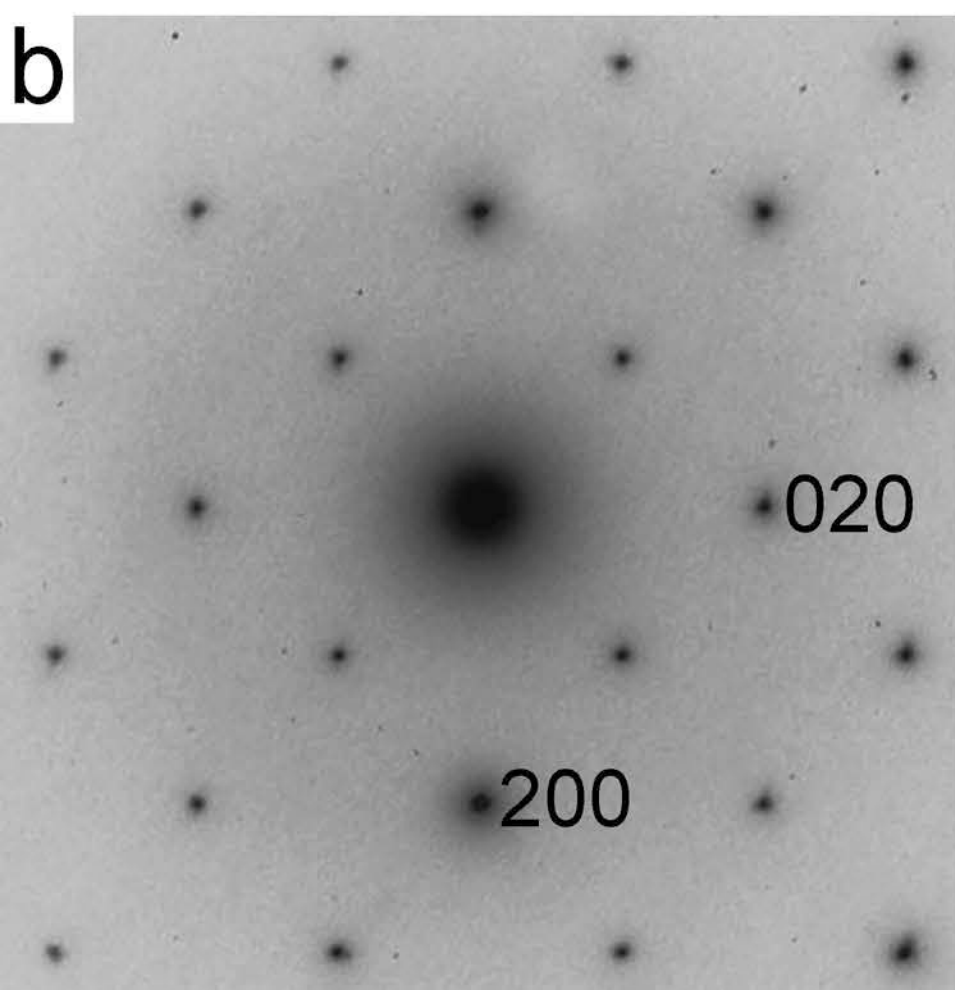
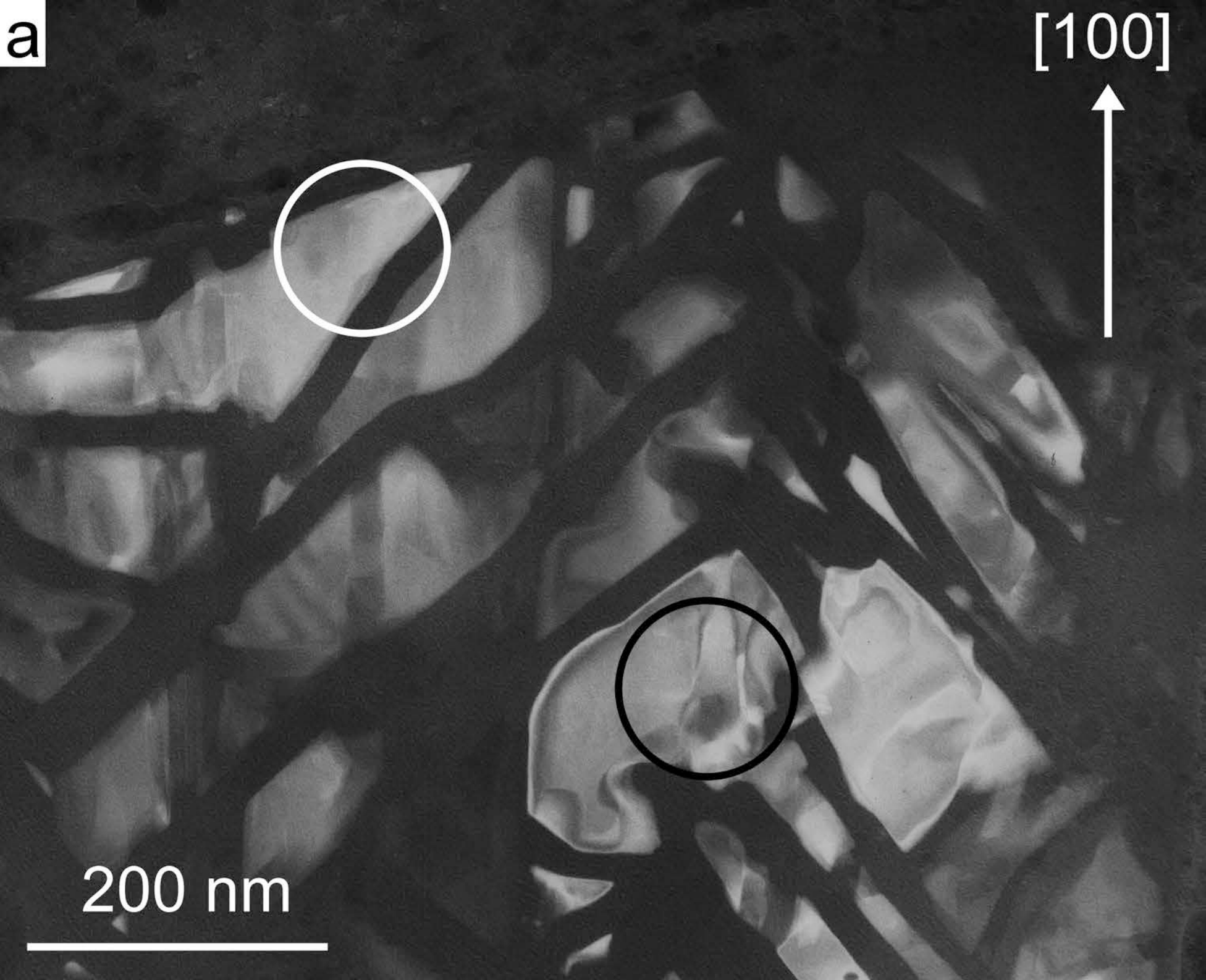
**TABLE 2.** Atomic coordinates for *Pmnn* kumdykolite derived from the structure of synthetic  $\text{CaAl}_2\text{Si}_2\text{O}_8$  given by Takeuchi et al. (1973). T is the tetrahedral site and can be occupied by Si or Al.

Atom	<i>x</i>	<i>y</i>	<i>z</i>
Na	0	0.4738	0.1142
T	0.2003	0.1498	0.1445
O1	0.3109	0.1819	0.4237
O2	0.2880	0	0
O3	0	0.1387	0.2521



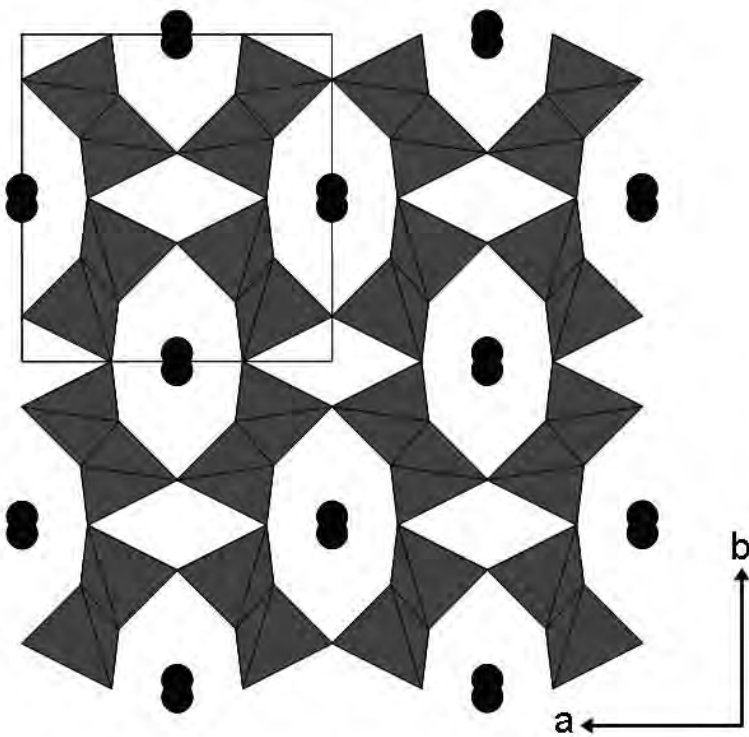






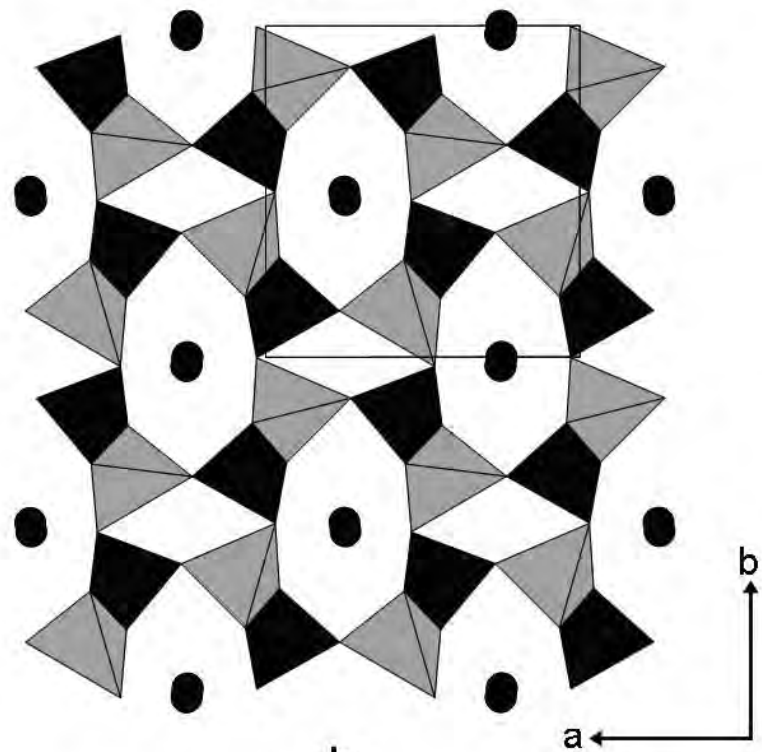


Kumdykolite ( $Pm\bar{m}n$ )



a

Svyatoslavite ( $P2_1$ )



b