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Pressure induced structural transformations in the low cristobalite form of AlPO_4

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Abstract

We have investigated the high pressure behavior of low cristobalite form of AlPO_4 (c- AlPO_4) using a combination of Raman scattering, synchrotron powder x-ray diffraction and classical molecular dynamics simulations. Our experiments indicate that under non-hydrostatic conditions c- AlPO_4 initially transforms to a monoclinic phase, which then transforms to the Cmc m phase via an intermediate, disordered structure. In contrast, x-ray diffraction measurements made under hydrostatic conditions show that the ambient structure transforms directly to the Cmc m phase. Our classical molecular dynamics simulations, carried out under hydrostatic conditions, also show that c- AlPO_4 directly transforms to the Cmc m phase at ~ 13 GPa.

Keywords: Cristobalite, high pressure, Raman scattering, synchrotron x-ray diffraction, classical molecular dynamics, AlPO_4

Introduction

AlPO_4 and GaPO_4 are iso-electronic analogs of silica and hence are iso-structural to silica polymorphs. Due to this, polymorphs of these phosphates have several properties similar to the silica polymorphs and hence have been studied to understand a number of geophysically relevant phenomena. In fact high pressure studies on AlPO_4 in the quartz form created a lot of excitement because of the claims of a memory glass effect, which was however, disproved subsequently (Kruger and Jeanloz, 1990; Sharma et al., 2000).

Apart from its geophysical importance, c- AlPO_4 is extensively used in industry due to its excellent properties like high melting point, low Young's modulus, low oxygen permeability and good erosion resistance (Huang et al., 2010; Morris et al., 1977). For example it is an integral part of hardened alumino phosphate cements (Bakunov and Shayakhmetov, 2007) and has been used as a binder for mullite silica fibres which are used as insulators in thermal power plants (Pitak and Churilova, 2004). It has recently been used as a protective coating on C-C composites which are used in aircraft and aerospace industries (Huang et al., 2010). Since the property of materials significantly depends on their structure, and the usability of materials in industry is limited by their

37 phase stability, it is important to investigate the structural stability of *c*-AlPO₄. There have been
38 extensive theoretical and experimental studies on silica and its polymorphs, however, phosphates of
39 aluminium and gallium have been relatively less explored, in particular in the low cristobalite form.

40 Experiments on α -cristobalite SiO₂ (*c*-SiO₂) have shown that, depending upon the degree of
41 hydrostaticity of the applied pressure, it may amorphize or undergo several symmetry lowering
42 crystal to crystal phase transitions (Gratz et al., 1993; Palmer and Finger, 1994; Palmer et al., 1994;
43 Tsuchida and Yagi, 1989). These results were validated by classical molecular dynamics (MD)
44 simulations where Garg and Sharma (2007) showed that *c*-SiO₂ becomes disordered when simulated
45 at P,T conditions representative of shock Hugoniot and Liang et al., (2007) showed that under non-
46 hydrostatic conditions the transition pathways and the high pressure phases are very sensitive to the
47 presence of anisotropic stresses. MD calculations of several authors indicate that the inter-atomic
48 potentials and the number of unit cells used for the simulation constrain whether *c*-SiO₂ transforms
49 directly to six coordinated stishovite or via an orthorhombic Cmc₂m phase (Dove et al., 2000a; Dove
50 et al., 2000b; Keskar and Chelikowsky, 1992; Tse and Klug, 1991; Tsuneyuki et al., 1989;
51 Tsuneyuki et al., 1988). Though Klug and Tse (ab-initio studies) did observe that a lattice distortion
52 precedes the *c*-SiO₂ to stishovite transformation (Klug et al., 2001), subsequent theoretical
53 calculations (both ab-initio and MD) of Huang et al. did not support the formation of the Cmc₂m
54 phase at high pressure. They have shown that the first high pressure phase of cristobalite silica has
55 an orthorhombic structure (S. G. : C222₁) (Huang et al., 2006).

56 In contrast to *c*-SiO₂, one of the early studies on cristobalite phosphates showed that GaPO₄
57 transforms directly to the orthorhombic Cmc₂m phase at ~ 16.5 GPa whereas *c*-AlPO₄ amorphizes
58 (Robeson et al., 1994). These authors had attributed amorphization in *c*-AlPO₄ to the large Al-O-P
59 angle. Subsequent MD simulations (Murashov et al., 1995) on *c*-GaPO₄ were in qualitative
60 agreement with the experimental results. The discrepancy in the transformation pressures was
61 attributed to the non directional nature of the pair potentials. However, recent experiments (Chio et
62 al., 2008; Ming et al., 2007) on *c*-GaPO₄ indicate that its high pressure behavior is somewhat similar
63 to that of *c*-SiO₂ though the monoclinic structures of the high pressure phases are different. Both x-
64 ray diffraction (XRD) and Raman spectroscopic studies show that *c*-GaPO₄ undergoes either one or
65 two phase transformations to a monoclinic phase prior to the transformation to the Cmc₂m phase.

66 Unlike *c*-SiO₂, so far none of these studies have shown that the Cmc₂m phase of GaPO₄
67 further transforms to a stishovite phase. Since all the cations in stishovite are six coordinated it is
68 possible that the high stability of the PO₄ tetrahedra (due to the strong P-O covalent bond) inhibits
69 the formation of the denser phases like stishovite or post stishovite. Recent studies (Pellicer-Porres et
70 al., 2007) on quartz structured α -AlPO₄ have however, shown that when it is pressurized to > 46
71 GPa the Cmc₂m phase transforms to a monoclinic CaCl₂ type of phase where the phosphorous atoms

72 are six coordinated. Thus in these APO_4 (A : Al, Ga etc.) compounds the Cmc \bar{m} phase may be
73 thought of as an intermediate phase, between four-fold and six-fold structures, stabilized by the
74 resistance to the breaking of the strong PO_4 bonds. Hence it is possible that if pressurized to
75 sufficiently high pressures GaPO_4 may also transform to a denser phase similar to c- SiO_2 .

76 Since the quartz forms of AlPO_4 and GaPO_4 show similar phase transitions at high pressure,
77 it is surprising that c- AlPO_4 should amorphize instead of undergoing crystal to crystal phase
78 transitions. It is possible that like c- SiO_2 the high pressure behavior of the other APO_4 compounds
79 may also be sensitive to the hydrostaticity of the pressure transmitting medium. Hence for a
80 consistent understanding of these iso-electronic, iso-structural compounds it is necessary to do a
81 careful investigation on the high pressure behavior of c- AlPO_4 . We have therefore studied the high
82 pressure behavior of this compound employing x-ray diffraction, Raman scattering and MD
83 simulations.

84

85 **Methodology:**

86 **Experimental Details**

87 The orthorhombic low-cristobalite form of AlPO_4 was prepared by co-precipitation followed
88 by high temperature annealing of the precipitate. Amorphous precipitate of AlPO_4 was prepared by
89 adding NH_4OH to an aqueous solution of aluminum chloride and H_3PO_4 . The precipitate was dried
90 at about 175°C and was then annealed at 1300°C for about 24 hrs. The sample was well ground and
91 was characterized with the help of x-ray diffraction and Raman spectroscopy. The lattice constants
92 of c- AlPO_4 were found to be $a=7.0673(6)$ Å, $b=7.0763(6)$ Å, $c=6.9751(3)$ Å in agreement with
93 earlier published data (Achary et al., 2003).

94 High pressure x-ray diffraction experiments were carried out on beamline 16ID-B of HPCAT
95 at the Advanced Photon Source using a diamond anvil cell (DAC). c- AlPO_4 powder and a tiny chip
96 of ruby were loaded in a gasket hole (with diameter ~ 0.2 mm) drilled in a pre-indented rhenium
97 foil. Two sets of experiments were carried out on c- AlPO_4 upto 11.3 and 43 GPa using x-ray
98 wavelength of 0.4028 Å and 0.4298 Å respectively. In the first experiment silicone oil was the
99 pressure transmitting medium (PTM) whereas in the second experiment, nitrogen was used as the
100 PTM. In both these experiments, pressure inside the sample chamber was monitored by the ruby
101 fluorescence line shift (Mao et al., 1986). The diffraction data which was collected using an image
102 plate detector was converted into one dimensional diffraction profiles with the help of FIT2D
103 software (Hammersley et al., 1996).

104 Raman measurements were carried out using a 460 mm single stage double pass
105 monochromator coupled to a liquid N_2 cooled CCD detector. The Raman scattering was measured
106 using the 488 nm or the 514.5 nm lines of an Ar ion laser as a light source and a supernotch filter
107 was used to cut off the Rayleigh scattered light. The size of the laser spot in these micro Raman

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108 experiments was $< 5 \mu\text{m}$. The powder sample was loaded in a diamond anvil cell along with a tiny
109 chip of ruby for pressure calibration. Typical relative error in measurement of pressure in hydrostatic
110 environment using our system is about 0.01 GPa. Since the sample was found to interact with
111 methanol-ethanol mixture, experiments were carried out using either Argon, silicone oil or without
112 any PTM.

113

114 **Computational Details**

115 Molecular dynamics simulations were carried out using a modified Nose-Hoover algorithm
116 (N, P, T) as implemented in the DL_POLY (Smith et al., 2003). The equations of motion were
117 integrated every 2.0 fs using the Verlet leapfrog algorithm. Initial macro cell was generated by
118 applying the translation symmetry on the experimentally determined structure of c- AlPO_4 in the
119 space group $C222_1$. The macro cell contains $(6 \times 6 \times 4)$ unit cells with 3456 atoms (576 Al + 576 P +
120 2304 O). The simulations were carried out at 300 K upto the maximum pressure of 160 GPa, (with
121 $\Delta P = 2$ GPa) employing semi empirical pair potentials of van Beest et al. (1990). We should also
122 note that these pair potentials have been earlier used to understand the high pressure behavior of
123 berlinite and have been able to stabilize the Cmc phase at high pressures. Ramaniah et al. (2003)
124 have shown that the results obtained from both MD and ab-initio calculations of α -quartz form of
125 AlPO_4 were close, indicating that these pair potentials could capture the main physics of both the α -
126 quartz and Cmc phases of AlPO_4 . These potentials are also known to simulate the different
127 polymorphs of the AlPO_4 phase (van Beest et al., 1990) and therefore we feel that they are good
128 enough to understand the high pressure behavior of cristobalite AlPO_4 . In the present calculations
129 the system was considered as equilibrated when the fluctuations in the macroscopic parameters like
130 total energy, cell volume etc. were $\sim 0.001\%$. The system was equilibrated at most of the pressures
131 for 100 ps but close to the transition pressures it was allowed to equilibrate for much longer time of
132 300 ps. The equilibrium properties were calculated by averaging over 2000 time steps.

133

134 **Results and Discussion:**

135

136 **Experimental**

137

138 The low temperature phase of c- AlPO_4 crystallizes in the orthorhombic space group $C222_1$
139 (D_2^5) with four formula units per unit cell. (The primitive cell contains two formula units.) Here it is
140 worth mentioning that the space group of the first high pressure phase of c- SiO_2 has also been
141 determined to be $C222_1$ (Liang et al., 2007). According to the factor group analysis, the symmetry
142 classification of all the 36 modes is (Rokita et al., 2000) $\Gamma_{\text{AlPO}_4} = 8A + 10B_1 + 9B_2 + 9B_3$ with acoustic
143 modes belonging to $B_1 + B_2 + B_3$. All the A-modes are Raman active, whereas only B_1 , B_2 and B_3 are

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144 IR active. At ambient conditions, internal and external modes observed by us are in good agreement
145 with earlier reported values (Nicola et al., 1978; Rokita et al., 2000). Raman spectrum of c-AlPO₄
146 with silicone oil as PTM shows higher background and hence poorer signal to noise ratio than
147 without any PTM. Though recently Argon has been shown to be quasi hydrostatic till ~ 10 GPa
148 (Klotz et al., 2009) our results obtained with Ar or without any PTM were found to be similar. This
149 similarity of the results may be arising from non-hydrostatic environment caused by overfilling of
150 the sample in the gasket, even when Argon was used as a PTM. This was necessary due to inherent
151 weak Raman signal of the sample.

152 Figure 1 shows the Raman spectra recorded without PTM at a few representative pressures.
153 At ambient conditions, consistent with the results of Rokita et al. (2000), the lattice vibration mode
154 was observed at 278 (P1) cm⁻¹ and a doublet of two overlapping modes (corresponding to the two
155 dimensional (2D) bending modes of [PO₄]³⁻ tetrahedra) were observed at ~ 380 (P2) cm⁻¹ and 390
156 (P3) cm⁻¹. The mode at ~ 482 (P4) cm⁻¹ can be assigned to the 3-dimensional bending mode of
157 [PO₄]³⁻ tetrahedra. The intense vibrational mode observed at 1119 (P6) cm⁻¹ corresponds to the one
158 dimensional symmetric stretch mode of [PO₄]³⁻ tetrahedra and the shoulder at ~ 1109 (P5) cm⁻¹ may
159 represent the corresponding asymmetric stretch mode. In our high pressure DAC experiments the
160 intensity of the aluminum sub-lattice band at 566-735 cm⁻¹ was found to be quite weak.

161 With increase in the pressure to ~ 0.4 GPa, Raman modes at 278 (P1) and 1109 (P5) cm⁻¹
162 gained intensity, accompanied by a reduction in the intensity of the low frequency mode (P3) in the
163 doublet of the PO₄ 2D bending mode. In addition to this, at ~ 0.6 GPa a new mode was observed
164 at ~ 432 cm⁻¹ accompanied with the relative intensity increase of mode P3 compared to mode P2.
165 Emergence of only one mode at ~ 432 cm⁻¹ and a weak mode close to mode P1 suggests that these
166 modes might have not been observed (since they were very weak) at ambient conditions but may
167 have gained intensity on the application of pressure. In our ab-initio calculations (Poswal and
168 Sharma) we have seen that this mode exists even at ambient pressure. On further increasing the
169 pressure to ~3.2 GPa most of the modes became broad and weak and merged above 4 GPa (figure 1)
170 to form broad Raman bands centered at ~1100 cm⁻¹ and 400 cm⁻¹. These observations seem to
171 suggest tetrahedral distortions, probably indicative of a sluggish first order phase transformation
172 from the ambient orthorhombic structure to another phase (HP-I). This is supported by the fact that
173 the broad Raman bands evolved into well resolved sharp peaks and gained intensity when further
174 pressurized to ~10.7 GPa. The observation of a larger number of Raman modes in HP-1 (compared
175 to the ambient phase) suggests that the high pressure phase may have a lower symmetry.

176 On further increase of pressure broadening and diminishing of intensity of the Raman modes
177 is observed above ~ 14.5 GPa. This could be due to the inhomogeneous stress distribution caused by
178 the non-hydrostatic environment, or it could be an indication of a second phase transition to a
179 disordered or poorly crystallized daughter phase. To remove this ambiguity, the sample was

annealed at this pressure by increasing the laser power. This resulted in the appearance of several new broad Raman modes (figure. 1) in the range of 500 to 1200 cm^{-1} . In addition to this, the weak mode observed at $\sim 450 \text{ cm}^{-1}$ before annealing became sharp after annealing. Even on lowering the laser power these modes were clearly visible. Here it is worth mentioning that this is not an artifact of any reaction between the sample and the PTM as this experiment was carried out without any PTM. The appearance of these new Raman modes shows that the changes in the Raman spectrum at this pressure were essentially due to a phase transformation to a new high pressure phase (HP-II). Robeson et al. (1994) have shown that c-GaPO₄ transforms to a CrVO₄ type structure at ~ 16.5 GPa. The comparison of these Raman modes with earlier reported Raman spectrum of CrVO₄ type structure (in particular that of FePO₄ (Pasternak et al., 1997) and CuCrO₄ (Baran, 1994)) suggest that the structure of the second high pressure phase could be similar to the CrVO₄ structure (Baran, 1994). These studies are in agreement with our (Poswal and Sharma) ab-initio Plane-Wave Self Consistent-Field (PWSCF) calculations where we observed that c-AlPO₄ transforms to the Cmc₂ structure at high pressure. In that study we did not observe the first monoclinic phase as the calculations were carried out under hydrostatic conditions. These observed phase transitions are clearly visible in the Raman frequency shift versus pressure, as shown in figure 2.

To identify the structures of the different high pressure phases, we will now discuss the results of our x-ray diffraction experiments. Figure 3 shows one dimensional x-ray diffraction patterns of c-AlPO₄ at various pressures. The lower two diffraction profiles are from the experiments carried out with silicone oil as a PTM (which remains hydrostatic upto ~ 2 GPa) and the rest are from the experiments where N₂ was used as a pressure transmitter. The study on these PTM shows that silicone oil remain hydrostatic up to ~ 2 GPa and N₂ up to ~ 10 GPa (Klotz et al., 2009).

The x-ray diffraction pattern at 4.7 GPa (using silicone oil as PTM) shows several new diffraction peaks in comparison with that observed at ambient conditions indicating a phase transition to a lower symmetry phase. This result is consistent with our Raman scattering measurements where we observed that the initial phase is stable below ~ 4 GPa beyond which it transforms. Even in GaPO₄ a similar symmetry lowering transition was observed at ~ 2 GPa. This new high pressure phase (HP1) was indexed using the Crysfire software (Shirley, 2004). Crysfire generated several similar monoclinic solutions with high figure of merit (7.4). These solutions were tested with Le-Bail refinement. The lattice parameters obtained from the best Le-Bail refinement are $a = 8.0914(8) \text{ \AA}$, $b = 4.5432(4) \text{ \AA}$, $c = 8.838(1)$ and $\beta = 120.018(9)^\circ$ with goodness of fit parameters $R_p = 0.03$ and $wR_p = 0.26$. The space group determined from the systematic absences in the diffraction profile was found to be P2. Figure 4 shows the Le-Bail refinement of the experimental data at ~ 4.7 GPa. Density considerations suggest that this monoclinic phase also has four formula units in each unit cell. To determine whether the structure of this monoclinic phase was similar to any of the known monoclinic polymorphs of silica like Badeyylite etc. we generated these structures

216 and their respective diffraction patterns, by scaling the volume (the cell constants determined from
217 the Le Bail fit were used) to that of the monoclinic phase observed by us. However, we found that
218 none of the generated diffraction patterns were similar to that of the HP1 phase. Though we could
219 not determine the exact structure of this HP1 phase the ratio of the lattice constants suggest that this
220 phase is distinct from the monoclinic phase observed in c-GaPO₄ as well as in c-SiO₂ (Dove et al.,
221 2000b; Ming et al., 2007).

222 On increasing the pressure beyond 11 GPa the x-ray diffraction pattern showed broad
223 diffraction peaks, indicative of destabilization of the monoclinic structure as also observed in our
224 Raman experiments. In addition to this, when the pressure is fully released from the monoclinic
225 phase, the observed d-values suggest that the compound transforms back to the initial structure,
226 indicating the reversibility of the structural changes. Raman scattering measurements also suggest
227 that it transforms back to the ambient phase.

228 With Nitrogen as PTM, the diffraction pattern at ~8.1 GPa displayed several new diffraction
229 peaks (figure 3) indicating a phase transformation. This diffraction pattern is distinct from the HP1
230 diffraction pattern observed in the experiment where silicone oil was used as the PTM. The
231 diffraction pattern of this high pressure phase of c-AlPO₄ could be indexed to an orthorhombic
232 structure in the Cmc₂m space group. This structure was found to be similar to the high pressure phase
233 of c-GaPO₄ (CrVO₄ type structure) (Robeson et al., 1994). To determine the lattice constants of this
234 phase the diffraction patterns were analysed using the Rietveld refinement method implemented in
235 the GSAS (Larson and Von Dreele, 2000) software. Since nitrogen is used as a PTM all the recorded
236 diffraction patterns were analyzed using two phases (AlPO₄ and N₂). Figure 5 shows the Rietveld
237 refined x-ray diffraction pattern at 8.1 GPa with AlPO₄ in Cmc₂m phase and N₂ in δ-phase (Olijnyk,
238 1990). The lattice parameters obtained from the Rietveld refinement are a= 5.085(1) Å b= 7.360(1)
239 Å and c= 5.838(1) Å (Table 1) with goodness of fit parameters Rp = 0.66 and wRp = 0.18 at 8.1
240 GPa. The comparison of lattice parameters with the earlier reported values are shown in table 2. The
241 Cmc₂m phase was found to be stable upto 33.3 GPa as shown in figure 3. The diffraction peaks
242 broadened and diminished in intensity at 41.1 GPa. This could be due to disordering or due to the
243 inhomogeneous distribution of stresses. To reduce the probable non-hydrostatic stresses the sample
244 was annealed to 310°C at this pressure using external heating (and the pressure measured after
245 annealing was ~ 43.9 GPa). This resulted in sharpening of the diffraction peaks implying the
246 stability of the Cmc₂m phase upto this pressure.

247 These x-ray diffraction and Raman scattering measurements show that under non-hydrostatic
248 conditions c-AlPO₄ transforms to a monoclinic structure at ~ 4 GPa. This structure becomes unstable
249 above 11 GPa and starts transforming to the Cmc₂m phase via a disordered phase (which may be due
250 to a thermodynamic barrier affecting the kinetics adversely). However in hydrostatic pressure
251 experiments ambient cristobalite phase directly transforms to the Cmc₂m phase above 8 GPa. These

252 observations indicate that, under hydrostatic pressures, cristobalite to Cmc m phase transformation in
253 c-AlPO $_4$ is not kinetically hindered. Even in c-SiO $_2$ it has been observed that when the simulations
254 were carried out under hydrostatic conditions it transformed to the Cmc m phase (Garg and Sharma,
255 2007; Tsuneyuki et al., 1989). These simulations also showed that even when c-SiO $_2$ directly
256 transformed to the stishovite phase it was always via a transient Cmc m phase. However, experiments
257 (Downs and Palmer, 1994; Tsuchida and Yagi, 1989) have shown that if non-hydrostatic stresses are
258 present then c-SiO $_2$ transforms to a monoclinic phase. This shows that there is a similarity in the
259 high pressure behavior of these APO $_4$ oxides.

260

261 **Computational**

262

263 To understand the atomistic mechanism of phase transitions, we have carried out classical
264 molecular dynamics simulations on c-AlPO $_4$. Cristobalite-AlPO $_4$ was equilibrated at 0.1 MPa and
265 300 K. The equilibrated final volume determined from these simulations is 328.95 Å 3 which
266 compares well with the experimental value 348.83 Å 3 i.e., within ~ 6 %. The calculated lattice
267 parameters at this pressure are $a = 7.1652$ Å, $b = 6.9428$ Å and $c = 6.6125$ Å in comparison to the
268 experimentally observed lattice parameters, $a = 7.0673(6)$ Å, $b = 7.0763(6)$ Å, $c = 6.9751(3)$ Å (i.e.
269 within 1.4 %, 1.9% and 5.2 % respectively). This equilibrated structure was subjected to increasing
270 pressure in steps of 2 GPa.

271 Figure 6 shows the evolution of calculated lattice parameters (calculated are shown by lines
272 and experimental by symbols) of c-AlPO $_4$ with pressure from MD simulations. On increasing
273 pressure the lattice parameter **b** shows less compression compared to **a** and **c**. MD simulations at ~
274 13 GPa show a large compression in **a** and **c** lattice parameters accompanied by an abrupt expansion
275 in the **b** parameter, indicating a phase transformation. Evolution of lattice parameters at 13 GPa
276 during MD simulation is given in supplementary information (Figure S1) . These changes result in a
277 volume drop of ~12.4 % at this pressure, confirming it to be a first order phase transition. This
278 phase transformation is at a slightly higher pressure than that found in the experiments. Though the
279 compression trends of calculated and the experimental lattice parameters are same in the Cmc m
280 phase, the calculated lattice parameters are slightly over estimated compared to the experimental
281 ones, as shown in figure 6. Even on release of pressure the Cmc m phase was found to be stable upto
282 ~5 GPa. Below this pressure it transform to a four coordinated phase distinct from the initial phase.
283 This is understandable as c-AlPO $_4$ is a metastable phase at ambient conditions and its formation may
284 depend on the thermodynamic path. For the ambient phase, the bulk modulus and it's derivative
285 obtained by fitting the calculated P-V data to 3rd order Birch Murnghan equation of state gives $B_0 =$
286 $24(3)$ GPa and $B' = 6(1)$ and $B_0 = 29(1)$ with B' fixed to 4. As we did not have sufficient experimental

287 data points to determine the bulk modulus of the ambient cristobalite phase, these values could not
288 be compared to the experimental one.

289 The phase transition displays itself through significant changes in the radial distribution
290 function (RDF) $g(r)$. The RDFs of Al-O, P-O and O-O as a function of pressure is given in
291 supplementary information (Figure S2). At ambient conditions, Al-O RDF shows a sharp peak at
292 1.74 Å which integrates to 4 implying that Al atoms have four oxygen nearest neighbors having a
293 bond length of ~ 1.74 Å. However, at ~ 13 GPa this sharp peak splits into two peaks, integrating to 6
294 implying that at this pressure Al atoms have six nearest neighbors. Analysis of the structure shows
295 that out of the six oxygen atoms coordinated to Al, four are at 1.79 Å and two are at 1.95 Å. These
296 features are similar to that of the Cmcm phase of SiO₂ observed by Tsuneyuki et al. (1989). In
297 contrast to the RDF of Al-O, RDFs of P-O do not show any dramatic changes in the first nearest
298 neighbor peak implying that the PO₄ tetrahedra remain intact up to 160 GPa. In the radial
299 distribution function of O-O there are two well separated peaks, the first peak being more intense
300 and sharp compared to the second one. Both these peaks integrate to 3, indicating that there are 3
301 first and 3 second nearest neighbors. After the transition, first coordination becomes 4 and second
302 coordination becomes ~ 9 . Structure analysis shows that this is due to the collapse of corner shared
303 tetrahedral network system (cristobalite) to a mixed network of corner shared octahedra and
304 tetrahedra (where individual octahedra are edge shared).

305 RDF and structural analysis at different pressures suggest that Al becomes six coordinated at
306 13 GPa while P continues to be four coordinated upto the highest pressure in our simulations. The
307 structure of the high pressure phase was found to be similar to the one observed by Tsuneyuki et al
308 in c-SiO₂ (CrVO₄ type) (Tsuneyuki et al., 1988). (Space group of the high pressure structure
309 obtained in the simulation is determined using the FINDSYM software H. T. Stokes, D. M. Hatch,
310 <http://stokes.byu.edu/isotropy.html>, 2004). It is made of corner linked AlO₆ octahedra and PO₄
311 tetrahedra. For direct comparison to the x-ray diffraction experiments we generated powder x-ray
312 diffraction patterns from the averaged coordinates at each pressure. At ambient conditions,
313 calculated and experimental diffraction patterns display good agreement in terms of intensity ratios.

314 At 13 GPa the computed diffraction pattern changed abruptly and it matches well with the
315 diffraction pattern of the 'daughter' phase (CrVO₄ type) obtained under hydrostatic conditions. The
316 lattice parameters just across the phase change also compare favorably, viz., $\mathbf{a} = 7.5875$ Å $\mathbf{b} =$
317 5.1979 Å and $\mathbf{c} = 5.6688$ Å (calculated, 13 GPa) versus $\mathbf{a} = 5.0784(1)$ Å $\mathbf{b} = 7.3627(1)$ Å and $\mathbf{c} =$
318 $5.8363(1)$ Å (experimental at ~ 8 GPa) (note that in simulations \mathbf{a} and \mathbf{b} axes were interchanged as
319 compared to the experimental setting). These results confirm that the high pressure phase observed
320 in the simulations is the same as observed in our experiments. The new structure i.e. Cmcm (CrVO₄
321 type) is predicted to be stable at least upto 160 GPa in agreement with our experiments where we
322 observed that the Cmcm phase is stable upto ~ 43 GPa (highest pressure achieved in this

323 experiment). Consistent with our structural analysis presented above, the RDF patterns of released
324 run show that Cmc m phase is stable up to 5 GPa below which it transforms to a four coordinated
325 phase. The calculated bulk modulus and its derivative determined by fitting 3rd order Birch
326 Murnaghan equation of state are $B_0 = 136(4)$ GPa and $B' = 4.7(2)$ with $V_0 = 240.29(1)$ Å³ fairly
327 matching with our experimental values $B_0 = 137(5)$ GPa and $B' = 4$ (fixed) with $V_0 = 230.571(1)$ Å³
328 . Earlier reported bulk modulus and its derivative are $B_0 = 118(7)$ GPa and $B' = 4$ (fixed) with $V_0 =$
329 $244.8(6)$ Å³ (Pellicer-Porres et al., 2007). The observed discrepancy in our experimental and earlier
330 reported values (Pellicer-Porres et al., 2007) could be due to a better (neon vis-a-vis nitrogen)
331 hydrostatic environment as compared to our studies. It is also possible that since in those studies
332 quartz was transforming to Cmc m and in our study cristobalite was transforming to Cmc m the end
333 product may have a different amount of internal stress or defects which may affect the bulk modulus.

334 Our MD results show that the cristobalite to Cmc m transition takes place at a higher pressure
335 when compared to the experiments carried out under hydrostatic conditions. In c-GaPO₄ the
336 discrepancy in transformation pressures has been attributed to the non-directional nature of the pair
337 potentials. It is possible that even in c-AlPO₄ the non-directional nature of the pair potentials may be
338 responsible for the differences in the transformation pressures. However, in the experiments one can
339 not rule out the possibility that even a small amount of non-hydrostaticity may reduce the
340 transformation pressure.

341

342 To summarize, our investigations show that the high pressure behavior of c-AlPO₄ depends
343 on the degree of hydrostaticity of pressure in the diamond anvil cell. Our Raman scattering and
344 XRD results show that under non-hydrostatic conditions the cristobalite phase first transforms to a
345 monoclinic phase and then to the Cmc m phase, via a disordered state. For this compound the
346 monoclinic and Cmc m phases seem to be separated by a kinetic barrier. However, our XRD and MD
347 simulations show that under hydrostatic pressures the ambient phase transforms directly to the
348 Cmc m phase, a result that is similar to c-SiO₂.

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FIGURE CAPTIONS

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352 Figure 1: Raman spectra of c-AlPO₄ at a few representative pressures. The presented data are from
353 the high pressure run performed without any pressure transmitting medium. Raman modes of
354 ambient phase are marked by P1, P2, P3, P4, P5 and P6. The mode assignment for Cmcm phase is
355 taken from the ab-initio calculations (Poswal and Sharma, 2011). The mode marked as asterisk does
356 not belong to Cmcm phase and could not be assigned.

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358 Figure 2: Experimentally observed shift of Raman modes of c-AlPO₄ with pressure in a non-
359 hydrostatic environment.

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361 Figure 3: Background subtracted x-ray diffraction patterns of c-AlPO₄ at a few representative
362 pressures. Lowest two diffraction patterns were recorded with silicone oil as a pressure transmitting
363 medium ($\lambda=0.4028$ Å) and the rest of the patterns are recorded with N₂ as pressure transmitter
364 ($\lambda=0.4298$ Å). The calculated peak positions of the c-AlPO₄ phases at ambient conditions have been
365 indicated by the vertical bars in the figure. The calculated diffraction patterns of N₂ at various
366 pressures are shown in the figure with the red line (Olijnyk, 1990).

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369 Figure 4 Le-Bail refinement of x-ray diffraction pattern at 4.7 GPa (Silicone oil as pressure
370 transmitting medium, $\lambda=0.4028$ Å). The experimental data are represented by plus sign, the Le-Bail
371 fit by red line and the difference is denoted by cyan. Vertical bars represent the diffraction peak
372 positions from solid high pressure monoclinic phase of c-AlPO₄. The goodness of fit parameters are
373 $R_p = 0.03$ and $wR_p = 0.26$.

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375 Figure 5 Rietveld refinement of x-ray diffraction pattern at ~8.1 GPa (N₂ as pressure transmitting
376 medium, $\lambda=0.4298$ Å). The experimental data are represented by plus sign, the Rietveld fit by red
377 line and the difference is denoted by cyan. Vertical bars represent the diffraction peak positions
378 from solid N₂ (δ -phase) (Olijnyk, 1990) and high pressure Cmcm phase of c-AlPO₄. The goodness of
379 fit parameters are $R_p = 0.66$ and $wR_p = 0.18$

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381 Figure 6 Evolution of the calculated lattice parameters of **a** (black line), **b** (red line) and **c** (blue line)
382 of the c-AlPO₄ phase with pressure as determined from our MD simulations. The experimental
383 lattice parameters at ambient conditions are shown by asterisk symbols and the high pressure lattice
384 parameters **a** (triangle), **b** (square) and **c** (circle) beyond the phase transition are shown as open
385 symbols. Arrows indicate the compression and decompression cycle. The drastic change in the cell
386 constants at ~ 13 GPa indicates the onset of the phase transition in MD.

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Table 1. Measured lattice parameters of AlPO_4 (cristobalite and Cmc m phase) obtained from Rietveld refinement of diffraction patterns recorded at various pressures with nitrogen as PTM. Lattice parameters in the monoclinic phase (with silicone oil as PTM) are determined using Le-Bail refinements.

Pressure (GPa)	Phase	a (Å)	b (Å)	c (Å)	Beta
0.05	C222 $_1$	7.0672(6)	7.0762(7)	6.9752(3)	
4.7	monoclinic	8.0914(8)	4.5431(4)	8.838(1)	120.018(9)
8.1	Cmc m	5.085(1)	7.360(1)	5.838(1)	
12.3	Cmc m	5.065(2)	7.296(3)	5.765(3)	
16.9	Cmc m	5.036(3)	7.220(4)	5.696(3)	
21.4	Cmc m	5.015(2)	7.172(4)	5.667(3)	
26.1	Cmc m	5.009(1)	7.131(2)	5.603(2)	
33.3	Cmc m	4.989(2)	7.023(3)	5.529(2)	
41.1	Cmc m	4.968(2)	6.948(3)	5.463(3)	
43.9*	Cmc m	4.967(2)	6.950(3)	5.441(3)	

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* Annealed at 310° C at 41.1 GPa

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Table 2. Comparison of lattice parameters in CrVO_4 (Cmc m) type structure with earlier reported values.

	P (GPa)	a (Å)	b (Å)	c (Å)	b/a	c/a
* AlPO_4	8.1	5.085(1)	7.360(1)	5.838(1)	1.4474	1.1481
† AlPO_4	13.9	5.0365(7)	7.2908(9)	5.7491(9)	1.4476	1.1415
‡ GaPO_4	15.9	5.12	7.31	5.92	1.4277	1.1563
§ SiO_2	15	5.20	7.44	5.58	1.4308	1.0731

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* (Present study)

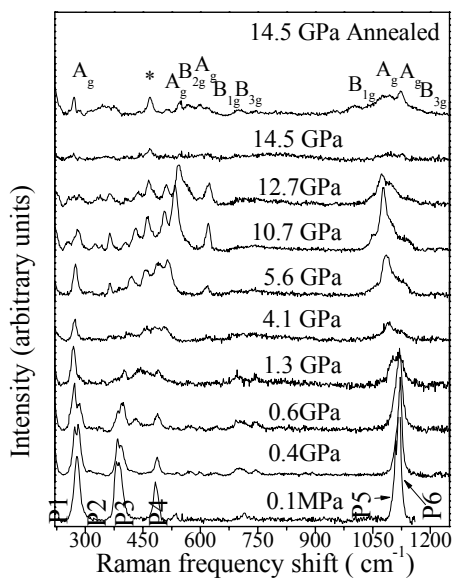
† (Pellicer-Porres et al., 2007)

‡ (Robeson et al., 1994)

§ (Tsuneyuki et al., 1989)

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Figure 1

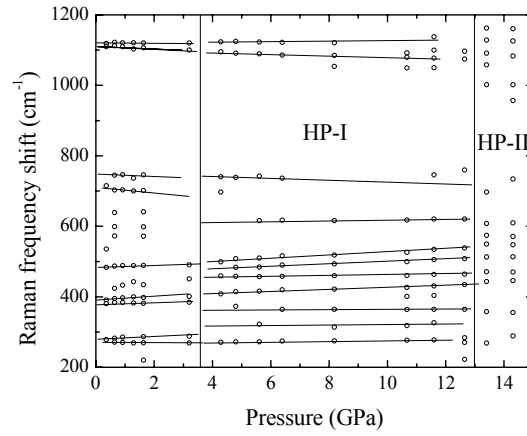


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Figure 2

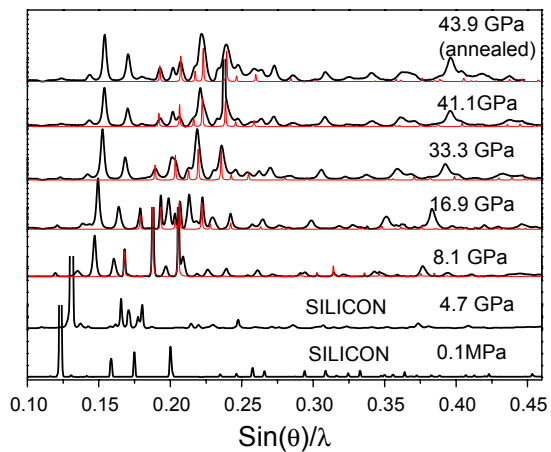


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Figure 3

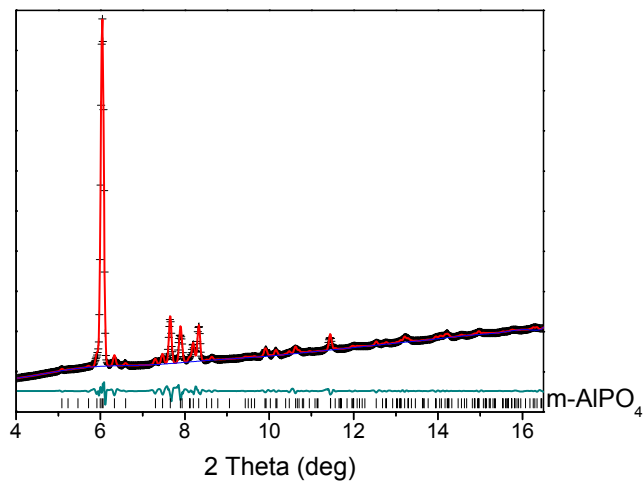


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Figure 4

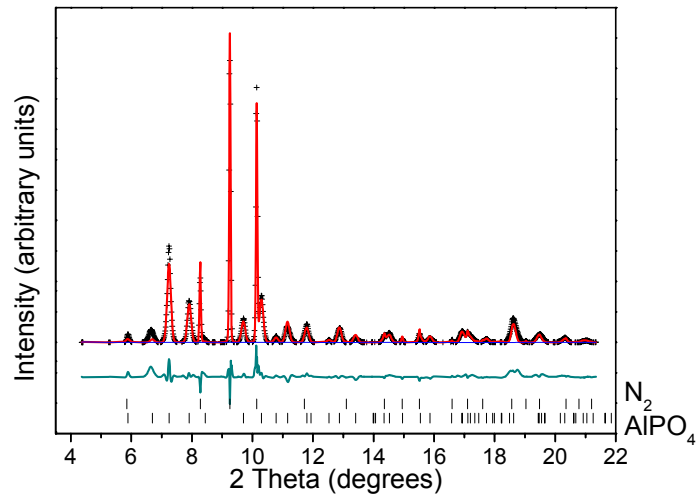


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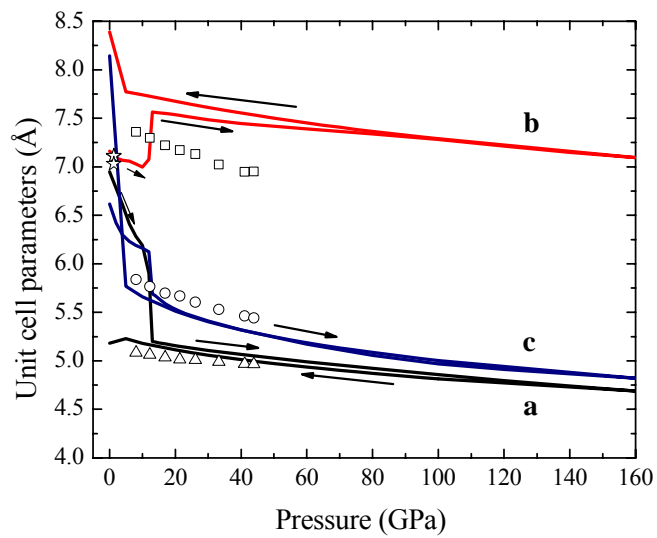
Figure 5



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Figure 6



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441 References:

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- 444 Achary, S.N., Jayakumar, O.D., Tyagi, A.K., and Kulshresththa, S.K. (2003) Preparation, phase
445 transition and thermal expansion studies on low-cristobalite type $Al_{1-x}Ga_xPO_4$ ($x=0.0, 0.20,$
446 $0.50, 0.80$ and 1.00). *Journal of Solid State Chemistry*, 176(1), 37-46.
- 447 Bakunov, V.S., and Shayakhmetov, U.S. (2007) Deformation of unfired phosphate-bound
448 refractories. Part 3. Deformation of hardened cements on heating and colling. *Refractories*
449 *and Industrial Ceramics*, 48, 73-76.
- 450 Baran, E.J. (1994) Vibrational and Electronic-Spectra of Copper(II) Chromate. *Spectrochimica Acta*
451 *Part a-Molecular and Biomolecular Spectroscopy*, 50(14), 2385-2389.
- 452 Chio, C.H., Sharma, S.K., Ming, L.C., Nakamoto, Y., and Endo, S. (2008) High pressure Raman
453 spectroscopic study of low-cristobalite $GaPO_4$. *Journal of Physics: Conference Series*, 121,
454 022003.
- 455 Dove, M., Tucker, M., Redfern, S., Craig, M., Trachenko, K., Marshall, W., and Keen, D. (2000a)
456 High-pressure studies of cristobalite. *ISIS annual Report*, 32.
- 457 Dove, M.T., Craig, M.S., Keen, D.A., Marshall, W.G., Redfern, S.A.T., Trachenko, K.O., and
458 Tucker, M.G. (2000b) Crystal structure of the high-pressure monoclinic phase-II of
459 cristobalite, SiO_2 . *Mineralogical Magazine*, 64(3), 569-576.
- 460 Downs, R.T., and Palmer, D.C. (1994) The Pressure Behavior of Alpha-Cristobalite. *American*
461 *Mineralogist*, 79(1-2), 9-14.
- 462 Garg, N., and Sharma, S.M. (2007) Classical molecular dynamical simulations of high pressure
463 behavior of alpha cristobalite (SiO_2). *Journal of Physics-Condensed Matter*, 19(45), 456201.
- 464 Gratz, A.J., Deloach, L.D., Clough, T.M., and Nellis, W.J. (1993) Shock Amorphization of
465 Cristobalite. *Science*, 259(5095), 663-666.
- 466 Hammersley, A.P., Svensson, S.O., Hanfland, M., Fitch, A.N., and Hausermann, D. (1996) Two-
467 dimensional detector software: From real detector to idealised image or two-theta scan. *High*
468 *Pressure Research*, 14(4-6), 235-248.
- 469 Huang, J.F., Yang, W., and Cao, L. (2010) Preparation of a $SiC/Cristobalite-AlPO_4$ Multi-layer
470 Protective Coating on Carbon/Carbon Composites and Resultant Oxidation Kinetics and
471 Mechanism. *J. Mater. Sci. Technol.*, 26(11), 1021-1026.
- 472 Huang, L., Durandurdu, M., and Kieffer, J. (2006) Transformation pathways of silica under high
473 pressure. *Nature Materials*, 5(12), 977-981.
- 474 Keskar, N.R., and Chelikowsky, J.R. (1992) Structural-Properties of 9 Silica Polymorphs. *Physical*
475 *Review B*, 46(1), 1-13.
- 476 Klotz, S., Chervin, J.C., Munsch, P., and Le Marchand, G. (2009) Hydrostatic limits of 11 pressure
477 transmitting media. *Journal of Physics D-Applied Physics*, 42(7).
- 478 Klug, D.D., Rousseau, R., Uehara, K., Bernasconi, M., Le Page, Y., and Tse, J.S. (2001) Ab initio
479 molecular dynamics study of the pressure-induced phase transformations in cristobalite.
480 *Physical Review B*, 63(10).
- 481 Kruger, M.B., and Jeanloz, R. (1990) Memory Glass - an Amorphous Material Formed from $AlPO_4$.
482 *Science*, 249(4969), 647-649.
- 483 Larson, A.C., and Dreele, R.B.V. (2000) GSAS: General structure analysis system. Los Alamos
484 National Laboratory Report LAUR, 86.
- 485 Liang, Y., Miranda, C.R., and Scandolo, S. (2007) Tuning oxygen packing in silica by
486 nonhydrostatic pressure. *Physical Review Letters*, 99(21), 215504-1.
- 487 Mao, H.K., Xu, J., and Bell, P.M. (1986) Calibration of the Ruby Pressure Gauge to 800-Kbar under
488 Quasi-Hydrostatic Conditions. *Journal of Geophysical Research-Solid Earth and Planets*,
489 91(B5), 4673-4676.
- 490 Ming, L.C., Nakamoto, Y., Endo, S., Chio, C.H., and Sharma, S.K. (2007) Phase transformations of
491 α -cristobalite $GaPO_4$ at pressures up to 52 GPa. *Journal of Physics: Condensed Matter*, 19,
492 425202.

- 493 Morris, J.H., Perkins, P.G., Rose, A.E.A., and Smith, W.E. (1977) Chemistry and Binding Properties
494 of Aluminum Phosphates. *Chemical Society Reviews*, 6(2), 173-194.
- 495 Murashov, V.V., Dubrovinsky, L.S., Tse, J.S., and Lepage, Y. (1995) The pressure induced phase
496 transition of the α cristobalite form of CaPO_4 . *Journal of Physics: Condensed Matter*, 7,
497 8279.
- 498 Nicola, J.H., Scott, J.F., and Ng, H.N. (1978) Raman study of α - β cristobalite phase transition in
499 AlPO_4 . *Physical Review B*, 18, 1972-1976.
- 500 Olijnyk, H. (1990) High pressure x-ray diffraction studies on solid N_2 upto 43.9 GPa. *J. Chem.*
501 *Phys*, 93, 8968.
- 502 Palmer, D.C., and Finger, L.W. (1994) Pressure-Induced Phase-Transition in Cristobalite - an X-
503 Ray-Powder Diffraction Study to 4.4 GPa. *American Mineralogist*, 79(1-2), 1-8.
- 504 Palmer, D.C., Hemley, R.J., and Prewitt, C.T. (1994) Raman-Spectroscopic Study of High-Pressure
505 Phase-Transitions in Cristobalite. *Physics and Chemistry of Minerals*, 21(8), 481-488.
- 506 Pasternak, M.P., Rozenberg, G.K., Milner, A.P., Amanowicz, M., Zhou, T., Schwarz, U., Syassen,
507 K., Taylor, R.D., Hanfland, M., and Brister, K. (1997) Pressure-Induced Concurrent
508 Transformation to an Amorphous and Crystalline Phase in Berlinite-Type FePO_4 . *Physical*
509 *Review Letters*, 79, 4409-4412.
- 510 Pellicer-Porres, J., Saitta, A.M., Polian, A., Itie, J.P., and Hanfland, M. (2007) Six-fold-coordinated
511 phosphorus by oxygen in AlPO_4 quartz homeotype under high pressure. *Nature Materials*,
512 6(9), 698-702.
- 513 Pitak, Y.N., and Churilova, Y.V. (2004) The Effect of Aluminum Phosphate Binder on the
514 Crystallization of Mullite-Silica Fibers. *Refr. and Ind. Cera.*, 45, 275-276.
- 515 Poswal, H.K., and Sharma, S.M. (2011) First principles investigations on cristobalite- AlPO_4 .
516 Manuscript under preparation.
- 517 Ramaniah, L.M., Sharma, S.M., Kunc, K., Garg, N., and Laghate, M. (2003) First-principles
518 determination of the relative stability of the alpha and Cmc m structures of AlPO_4 . *Physical*
519 *Review B*, 68, 014119,1-7.
- 520 Robeson, J.L., Winters, R.R., and Hammack, W.S. (1994) Pressure induced transformations of the
521 low cristobalite phase of GaPO_4 . *Physical Review Letters*, 73, 1644-1647.
- 522 Rokita, M., Handke, M., and Monzgawa, W. (2000) The AlPO_4 polymorphs structure in the light of
523 Raman and IR spectroscopy studies. *Journal of Molecular Structure*, 555, 251.
- 524 Sharma, S.M., Garg, N., and Sikka, S.K. (2000) High-pressure x-ray-diffraction study of alpha-
525 AlPO_4 . *Physical Review B*, 62(13), 8824-8827.
- 526 Shirley, R. (2004) Crysfire.
- 527 Smith, W., Leslie, M., and Forester, T.R. (2003) Computer code DL_POLY_2.14
528 (CCLRC,Daresbury laboratory, Daresbury, England).
- 529 Tse, J.S., and Klug, D.D. (1991) The Structure and Dynamics of Silica Polymorphs Using a 2-Body
530 Effective Potential Model. *Journal of Chemical Physics*, 95(12), 9176-9185.
- 531 Tsuchida, Y., and Yagi, T. (1989) A New, Post-Stishovite High-Pressure Polymorph of Silica.
532 *Nature*, 340(6230), 217-220.
- 533 Tsuneyuki, S., Matsui, Y., Aoki, H., and Tsukada, T. (1989) New pressure-induced structural
534 transformations in silica obtained by computer simulation. *Nature*, 339, 209.
- 535 Tsuneyuki, S., Tsukada, M., Aoki, H., and Matsui, Y. (1988) First-Principles Interatomic Potential
536 of Silica Applied to Molecular Dynamics. *Physical Review Letters*, 61(7), 869.
- 537 van Beest, B.W.H., Kramer, G.J., and Vansanten, R.A. (1990) Force-Fields for Silicas and
538 Aluminophosphates Based on Abinitio Calculations. *Physical Review Letters*, 64(16), 1955-
539 1958.
- 540
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