1	Revision 2
2	On the use of unpolarized infrared spectroscopy for quantitative analysis of absorbing
3	species in birefringent crystals
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8	
9	Abstract
10	There is an understandable desire to use simple unpolarized infrared analysis of unoriented
11	anisotropic samples to extract quantitative information, rather than using more demanding
12	polarized techniques. Owing to the fact that unpolarized infrared absorbance in birefringent
13	media deviates from the Beer-Lambert law, previous studies have either warned against using
14	unpolarized spectroscopy for quantitative purposes, or have used flawed error analysis to justify
15	using simple averages of integrated absorbance of multiple absorbance bands as a proxy for total
16	integrated polarized absorbance in the principal spectra. It is shown here that unpolarized
17	infrared absorbance is correctly calculated by averaging in the transmission domain. The errors
18	in estimates of principal absorbance by averaging of unpolarized absorbance spectra are
19	evaluated using correct theory of unpolarized infrared transmission. Correction schemes for
20	integrated absorbance based on linear-absorbance error calculations are shown to be

spectra to be estimated from multiple unpolarized measurements of randomly oriented samples.

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inappropriate. A theory is developed that allows the sum of the polarized principal absorbance

23 The systematic errors that arise when averaging in the absorbance domain are avoided by use of

exact theory rather than an approximation. Numerical simulation shows that applying the new procedure to 10 unpolarized measurements of OH stretching bands in olivine results in convergence of the estimated total integrated principal polarized absorbance to within 10% of the true value for a sample size of 10 measurements, but the technique is limited to spectral regions that do not contain absorption bands that are simultaneously intensely absorbing and strongly anisotropic.

30 Keywords: Infrared spectroscopy, unpolarized, quantitative, hydroxyl, olivine

31

INTRODUCTION

32 Infrared (IR) spectroscopy performed in the most commonly used transmission geometry 33 enables the quantitative analysis of absorbing species by application of the Beer-Lambert law. A 34 common application in the Earth sciences is the measurement of hydroxyl, often at trace 35 concentrations in nominally anhydrous minerals (e.g., Rossman 1996). One drawback to such 36 measurements is that birefringent crystals can have strongly anisotropic absorption, leading to 37 large differences in absorbance as a function of the direction of light propagation through the 38 crystal. In such cases, it has been shown that the Beer-Lambert law applies only when the 39 absorbance is measured as the sum of polarized absorbances with the directions of polarization 40 parallel to each of the principal axes of the absorption indicatrix (e.g., Libowitzky and Rossman 41 1996). Consequently, several authors have emphasized that unpolarized infrared spectroscopy 42 cannot be used to make quantitative measurements in anisotropic materials (e.g., Libowitzky and 43 Rossman 1996; Bell et al. 2003). However, a robust determination of the errors associated with 44 unpolarized analysis would enable the use of unpolarized spectroscopy for quantitative 45 measurement in cases where polarized measurement of oriented samples is impractical. 46 Sambridge et al. (2008) and Kovács et al. (2008) developed a theory where the average

47	unpolarized absorbance is used as a proxy for total polarized absorbance. In calculating the
48	errors associated with the approximation, Sambridge et al. (2008) developed an alternative
49	theory for unpolarized absorbance that differs from that given by Libowitzky and Rossman
50	(1996). Here I will show that the theory originally presented in Libowitzky and Rossman (1996)
51	is, in fact, correct. The errors associated with unpolarized measurements will be calculated for
52	some specific examples.
53	Theory
54	The intensity of light, I , measured at a distance d after passing through an absorbing medium
55	is described by the relationship
56	$I = I_0 e^{-\alpha cd}, \tag{1}$
57	where I_0 is the intensity of light entering the medium, α is the molar absorption coefficient and c
58	is the concentration of the absorbing species. The dimensionless quantities transmittance (T) and
59	absorbance (A_n) are defined as
60	$T = I/I_0 \tag{2}$
61	and
62	$A_{\rm n} = -\ln(T) = \alpha c d. \tag{3}$
63	By convention, the decadic absorbance, A, is used for many practical applications:
64	$A = -\log_{10}(T) = \varepsilon c d, \tag{4}$
65	where ε is the decadic molar absorption coefficient, such that

66
$$I = I_0 \, 10^{-\varepsilon cd}$$

(5)

The natural absorbance (A_n) and corresponding molar absorption coefficient (α) differ from the base 10 equivalents (*A* and ε , respectively) by a constant factor of ln(10). Since the latter are used in the commercial software provided by the manufacturers of Fourier transform infrared (FTIR) spectrometers and by convention are preferred in many fields of quantitative analysis, absorbance hereafter will refer to base 10 absorbance. The choice of logarithm base has no effect on any of the results presented here.

73 The Beer-Lambert law, which is expressed in Equation 4, states that absorbance is 74 proportional to the product of concentration and sample thickness. It is important to note here 75 that absorbance (the exponent in Eq. 5) cannot be measured directly: the physical measurements 76 made in IR absorption spectroscopy are of the intensities of light, detected either after 77 transmission through the sample (I), or without a sample in place as a measurement of the 'background' intensity (I_0) . The FTIR spectrometer allows the analyst to measure the intensity of 78 79 light as a function of frequency, resulting in a spectrum that may be displayed in units of 80 transmittance or absorbance. The measurement, however, is made in the transmission domain.

Unpolarized light can be thought of as a mixture of plane waves with linear polarizations whose vibration directions are uniformly distributed in the plane perpendicular to the propagation direction. When an unpolarized beam of intensity I_0 is incident on an absorbing birefringent crystal, light of a given frequency is split into two polarized rays that can in general be considered as orthogonal, linearly polarized waves (e.g., Sambridge et al. 2008). The incident light intensity in each of the wave vibration directions then is equal to $I_0/2$. The intensity of light emerging from the crystal after travelling along either one of these ray paths may be measured by inserting a polarizer into the beam path. Transmittance as a function of polarization direction isgiven by:

90
$$T(\theta) = T_{\text{max}}\cos^2\theta + T_{\text{min}}\sin^2\theta,$$
 (6)

91 where T_{max} and T_{min} are the maximum and minimum transmittance values measured as the 92 polarizer is rotated, and θ is the angle between the direction of polarization during the 93 measurement and the polarizer orientation where $T = T_{\text{max}}$ (Libowitzky and Rossman 1996; 94 Sambridge et al. 2008). This relationship was derived elegantly by Sambridge et al. (2008) and 95 can be extended to calculate *T* for any vibration direction as a function of the values of *T* for 96 vibration directions parallel to the 3 principal axes of the absorption indicatrix (e.g., Asimow et 97 al. 2006; Sambridge et al. 2008).

98 Combining Equations 2 and 6, for any two mutually perpendicular polarization directions $\theta =$ 99 ξ and $\theta = \xi + 90^\circ$, the total intensity of the recombined light emerging from the crystal will be

100
$$I = I_0/2 \times (T_{\text{max}} \cos^2 \xi + T_{\text{min}} \sin^2 \xi + T_{\text{max}} \cos^2 (\xi + 90^\circ) + T_{\text{min}} \sin^2 (\xi + 90^\circ)) = I_0/2 \times (T_{\text{max}} + T_{\text{min}})$$
(7)

101 Thus, the intensity of the recombined light emerging from the crystal with no polarizer in place 102 leads to an unpolarized transmittance (T_{unpol}) of

103
$$T_{\text{unpol}} = (T_{\text{max}} + T_{\text{min}})/2.$$
 (8)

104 The value of *A* calculated from the measured light intensity is therefore:

105
$$A_{\text{unpol}} = -\log_{10}\left(\frac{T_{\text{max}} + T_{\text{min}}}{2}\right).$$
 (9)

106 This theoretical development arrives at a result identical to Libowitzky and Rossman (1996), 107 who stated that T_{unpol} is the average value of transmittance for all vibration directions in the plane 108 of the sample, given by integration of Equation 6:

109
$$T_{\text{unpol}} = \frac{1}{2\pi} \int_0^{2\pi} T(\theta) \, d\theta = \frac{(T_{\text{max}} + T_{\text{min}})}{2}.$$
 (10)

110 The foregoing theory is in contrast, however, to the suggestion of Sambridge et al. (2008) that A_{unpol} should be calculated by integration of $A(\theta)$, rather than $T(\theta)$. The formulation of 111 112 Sambridge et al. (2008) has been adopted by Kovács et al. (2008), who claim that Libowitzky 113 and Rossman (1996) are incorrect in their statement of Equation 10. The justification given by 114 Sambridge et al. (2008) for their alternative theory of unpolarized absorption is that 115 "spectroscopic measurements of IR radiation are often carried out using absorbance, A, rather 116 than transmittance, T". As I have shown in the preceding theoretical development, the reverse is 117 true: when a measurement is made with unpolarized light it is the absorbances in the directions 118 of the electric vectors of the two rays that determine the intensity of light transmitted through the 119 crystal, but the unpolarized absorbance spectrum is calculated (via the commercial software 120 provided with the spectrometer) from the measured intensity of transmitted light. In other words, 121 the intensity of light transmitted through the crystal is correctly calculated by addition in the 122 transmission domain.

Note that Equations 7 and 8 imply that the sum of light intensities measured sequentially with any two perpendicular polarizer orientations is equal to the intensity measured without the polarizer in place. In contrast, the theory of Sambridge et al. (2008) contravenes this physical expectation. Consider, also, the case of a highly polarizing sample where T_{max} approaches 1 and T_{min} approaches 0 (As $A \rightarrow \infty$, $10^{-A} \rightarrow 0$). Clearly, Equation 8 in this extreme case returns the 128 expected result of $T_{unpol}=0.5$ and hence $I=I_0/2$, while integration of A implies $T_{unpol}=0.25$, i.e., 129 $I=I_0/4$. The analytical solution leading to this result is presented in the Appendix. 130 The principal aim of Sambridge et al. (2008) and the companion paper, Kovács et al. (2008), 131 is to show the conditions under which averages of unpolarized infrared spectra make a good 132 approximation to the total polarized absorbance, and to evaluate the errors associated with that 133 approximation. Adopting an inaccurate expression for calculating A_{unpol} leads to underestimation 134 of errors. In the following section I will demonstrate the correctness of the theory outlined above 135 by comparison with measured spectra from the literature, and then use the correct theory to 136 reevaluate the errors associated with unpolarized infrared spectroscopy. Finally, a new protocol 137 is developed for estimating the sum of principal polarized absorbances from unpolarized spectra.

138

RESULTS

139 Comparison with measured spectra

140 When an infrared measurement is made in transmission mode, the absorbance measured 141 using unpolarized infrared light is close to the average of the maximum and minimum values for 142 polarized absorption only when the maximum absorption intensity is sufficiently small (e.g., 143 Sambridge et al. 2008). The extent of deviation of the unpolarized absorption from the average 144 of maximum and minimum polarized absorbances was calculated by Sambridge et al. (2008) 145 using an inaccurate formulation for unpolarized absorption, as expressed in the appendix of that 146 paper. In order to illustrate that the expressions for unpolarized absorbance given by Libowitzky 147 and Rossman (1996) and in Equations 8 and 9 of this study are consistent with observation, and 148 to show the real deviation of unpolarized absorption from the average polarized absorption, it is 149 necessary to compare polarized and unpolarized measurements. The ideal sample for this 150 purpose would exhibit strongly anisotropic absorbance and high maximum absorbance values, 151 while remaining within the realm of analysis conditions that are applied to real samples. One of 152 the most common applications of FTIR spectroscopy in Earth sciences is the measurement of H 153 in olivine, a mineral that exhibits exactly these properties. Figure 1 compares polarized and 154 unpolarized FTIR measurements in the OH stretching region of an olivine crystal from Kohlstedt 155 et al. (1996). In this case, where the olivine is oriented such that the direction of light 156 propagation is parallel to the crystallographic **b** axis, the maximum and minimum transmittance 157 values at each frequency are always observed with the polarizer oriented parallel to either the **a** 158 or \mathbf{c} axis. The unpolarized absorbance, according to Libowitzky and Rossman (1996), is then 159 given by:

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$$A_{\text{unpol}} = -\log_{10}\left(\frac{10^{-A_{a}}+10^{-A_{c}}}{2}\right),$$
 (11)

where A_a and A_c are the absorbance values for polarization directions parallel to crystallographic a and c axes, respectively. Note that the absorbance spectra in Figure 1 are normalized by sample thickness. Owing to the fact that unpolarized absorbance does not scale linearly with thickness, the values of A_a and A_c used in the calculation are those recorded from the original 126 µm thick sample (the sample thickness is recorded in the electronic data file provided by David Kohlstedt) and can be read from the right axis of Figure 1. The unpolarized absorbance according to Sambridge et al. (2008) is calculated as

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$$A_{\text{unpol}} = \frac{1}{2\pi} \int_0^{2\pi} A(\theta) \, d\theta,$$
 (12)

169 where

170
$$A(\theta) = -\log_{10}(10^{-A_a}\cos^2\theta + 10^{-A_c}\sin^2\theta),$$
 (13)

171 and θ is the angle between the **a** axis and the vibration direction. A Gaussian quadrature 172 algorithm was used for the numerical integration of $A(\theta)$.

173 The measured unpolarized spectrum is matched exactly by the unpolarized spectrum 174 calculated from the measured polarized spectra using Equation 11. In contrast, the alternative 175 theory of Sambridge et al. (2008) overestimates the measured spectrum significantly in the region between 3570 and 3620 cm⁻¹ where the maximum and minimum absorbances are strongly 176 offset. Note also that for much of the OH stretching region, the unpolarized absorption in this 177 178 example is indeed close to the average of the polarized values, suggesting that the difference 179 between integrated unpolarized absorbances and average polarized absorbances may be smaller 180 than the difference in the linear absorbances close to the maxima. A similar verification of 181 Equation 11 is demonstrated in Figure 6 of Libowitzky and Rossman (1996), using a calcite section that displays even more intensely anisotropic absorption bands than does the olivine in 182 183 Figure 1.

184 Variations with orientation and thickness

The usefulness of unpolarized spectra from randomly oriented anisotropic samples for quantitative analysis is determined principally by two factors: the convergence of the unpolarized absorbance to a single value, which determines the precision, and the deviation of that value from the polarized absorbance, which detracts from the accuracy of the measurement. To address each of these sources of uncertainty, large populations of unpolarized spectra have been calculated from a set of reference polarized spectra.

191 The reference spectra used for most of the following calculations are taken from Withers et 192 al. (2011), and consist of principal polarized spectra covering the OH stretching region from an 193 olivine that was hydrated in a high-pressure hydrothermal experiment, resulting in the 194 incorporation of 0.2 wt.% H₂O as structurally bound OH groups (Withers et al. 2012). Owing to 195 the high concentration of OH, the sample exhibits intense absorption bands (Figure 2). The highest-frequency peaks at 3612, 3598 and 3579 cm⁻¹ have strong absorbance in the $\mathbf{E} \parallel \mathbf{a}$ 196 197 spectrum and weak absorbance in the other principal axis spectra. Another sharp peak at 3566 cm^{-1} appears prevalently in the **E**||**c** spectrum, while the broader absorption band centered around 198 3550 cm⁻¹ has strongest absorption in the E||b| spectrum, and shows less anisotropy than the 199 200 higher-frequency peaks. This pattern of relative absorption intensity in the principal polarized 201 absorbance spectra is typical of olivines from high-pressure hydrothermal experiments (e.g., 202 Lemaire et al. 2004; Mosenfelder et al. 2006; Smyth et al. 2006; Litasov et al. 2007), but quite 203 distinct from many natural and lower-pressure experimental samples (e.g., Miller et al. 1987; Bai 204 and Kohlstedt 1993; Bell et al. 2003).

The principal polarized spectra shown in Figure 2 are normalized to a thickness of 1 cm. Because principal polarized (but not unpolarized) absorbance follows the Beer-Lambert law (Libowitzky and Rossmann 1996), the spectra can be scaled to any thickness (or concentration) of interest. The following describes how unpolarized spectra are then calculated from the principal axis spectra.

A spherical coordinate system with azimuthal angle ψ and polar angle φ is defined such that for an olivine crystal the crystallographic **c** direction corresponds to $\varphi=0$, and **a** and **b** directions correspond to (ψ , φ) = ($\pi/2$, 0) and ($\pi/2$, $\pi/2$), respectively. Taking the principal polarized absorbances $A_{\mathbf{a}}$, $A_{\mathbf{b}}$ and $A_{\mathbf{c}}$ as inputs, Sambridge et al. (2008) derive an equation that describes A_{unpol} as a function of incident light direction (their Eq. 60). In the derivation of that equation, an approximation is made whereby the functional form that describes the orientational dependence of transmittance is used for absorbance. Because unpolarized absorbance is correctly calculated

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$$A_{unpol}(\psi,\varphi) = -\log_{10}\left(\frac{1}{2}(T_{a}(\cos^{2}\varphi \ \cos^{2}\psi + \sin^{2}\psi) + T_{b}(\cos^{2}\varphi \ \sin^{2}\psi + \cos^{2}\psi) + \right)$$

$$221 \quad T_{\rm c}\sin^2\phi)\Big) \tag{14}$$

222 Choosing any two orthogonal vibration directions in the plane of the sample and calculating 223 *T* from the equation that describes the three dimensional transmission indicatrix (e.g., Asimow et 224 al. 2006; Sambridge et al. 2008) and then applying Equations 8 and 9 leads to the same result as 225 Equation 14. The unpolarized spectrum is calculated from the reference spectra by applying 226 Equation 14 at each frequency of the IR spectrum.

227 A pseudo-uniform distribution of 10,242 incident light directions was calculated by recursive 228 icosahedral subdivision with projection onto a spherical surface. For a given scaling of the 229 reference spectra, an unpolarized spectrum was calculated for each of the 10,242 directions of 230 incident light. Each of the unpolarized spectra shown in Figure 2 represents the average of 231 10,242 spectra calculated in this way. With increasing absorbance (i.e., increasing thickness or 232 concentration), absorbance bands that are strongly anisotropic in the principal polarized spectra 233 decrease in intensity in the unpolarized spectrum relative to more isotropic absorption bands. 234 Thus, in Figure 2 the highest peak in the 30 µm spectrum corresponds to the most intense but most strongly anisotropic 3612 cm^{-1} peak in the principal spectra, while at 50 µm the 3612 and 235 3579 cm⁻¹ peaks have equal height, and in the 100 μ m spectrum the maximum absorbance is 236 found at 3566 cm⁻¹, the position of the absorption band that is strongest in the $\mathbf{E} \parallel \mathbf{c}$ principal 237 238 spectrum, but which is convolved with the broader but more isotropic absorption band centered

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239 around 3550 cm⁻¹. When multiple absorption bands are convolved, as is usually the case for OH 240 bands in olivine, it is convenient to use integrated absorbance, which will be denoted by $\int A_{1}$ as a 241 measure of their combined absorbance. Figure 2 illustrates how it is the general degree of 242 anisotropy over all of the absorption bands in the spectrum that determines how far the integrated 243 absorbance of the unpolarized spectrum deviates from the average integrated principal spectra. 244 Thus, in an unpolarized spectrum that consists principally of absorption bands related to highly 245 anisotropic peaks such as the high-frequency peaks in the reference spectra in this example, the average integrated unpolarized absorbance ($A_{unpol, av}$) will deviate strongly from the average of 246 the integrated principal polarized absorbances ($\int A_{tot}/3$), while a spectrum that includes a 247 248 significant component of more isotropic absorbance, like the broader absorbance peaks at lower frequency in the example spectra, will have $\int A_{unpol, av}$ closer to $\int A_{tot}/3$. 249

Figure 3 shows histograms of A_{unpol} for the three populations of unpolarized spectra 250 251 corresponding to 30, 50 and 100 µm thicknesses of the olivine represented by the reference spectra in shown in Figure 2. With increasing absorbance in the reference spectra, $\int A_{unpol, av}$ 252 253 deviates further from $\int A_{tot}/3$. For this example, at 30 µm thickness the maximum linear absorbance in the unpolarized spectra is in the range of 0.13-0.27 absorbance units, and $\int A_{unpol, av}$ 254 is within 8% of $\int A_{tot}/3$. In this particular case, even though the analytical conditions are such that 255 256 the maximum linear polarized absorbance is greater than the limit of 0.3 absorbance units proposed by Kovács et al. (2008), $A_{unpol, av}$ does in fact provide a reasonable estimate of $A_{tot}/3$. 257 As the total absorbance increases, $\int A_{unpol, av}$ provides a progressively inaccurate estimate of 258 $\int A_{tot}/3$. The entire population of $\int A_{unpol}$ for unpolarized spectra calculated for 100 µm thick 259 samples is lower than $A_{tot}/3$, and A_{unpol} av underestimates $A_{tot}/3$ by 19%. The maximum linear 260 absorbance in the unpolarized spectra varies between 0.39 and 0.60 absorbance units which, 261

262 while higher than ideal, is within the range of values of published measurements of strongly H-263 enriched olivines from high-pressure experiments (e.g., Kohlstedt et al. 1996). Many such 264 olivines, even when prepared with the minimum practicable sample thickness, have sufficiently strong absorbance in the region of OH stretching that $A_{unpol, av}$ must depart significantly from 265 $\int A_{tot}/3$ (e.g., Mosenfelder et al. 2006; Smyth et al. 2006; Litasov et al. 2007). If unpolarized 266 267 measurements of strongly absorbing samples could be corrected to allow accurate estimation of 268 total absorbance the ease of measurement would be greatly reduced. The merits of one such 269 method of correction are discussed below.

270 **Corrections based on linear absorbance**

271 It has been proposed that the maximum linear absorbance observed in unpolarized spectra may be used to calculate a correction factor to account for the divergence of A_{unpol} av from $A_{tot}/3$ 272 273 when non-ideal analytical conditions result in high absorbance values in the regions where strong 274 absorption bands occur (Kovács et al. 2010). A measure of linear absorbance, however, contains 275 no information about absorbance bands in other regions of the spectrum that do not extend to the frequency of the linear measurement. As I have demonstrated above, $\int A_{unpol, av}$ becomes closer to 276 $\int A_{tot}/3$ when a significant proportion of the spectrum is more isotropic in nature, so basing a 277 278 correction on a maximum linear absorbance value that is influenced principally by a strongly 279 absorbing and highly anisotropic absorbance band can lead to overestimation of the correction 280 factor. When the correct theory of unpolarized measurement is used for the calculation of linear 281 absorbance, the two-dimensional linear-absorbance error estimates of Kovács et al. (2008) and 282 Kovács et al. (2010) increase by about a factor of 2 (Note that in the plots of linear-absorbance 283 errors calculated using the A-integral theory for unpolarized measurement shown in Figures 1 284 and 10 of Kovács et al. 2008, in the electronic supplement to that paper, and also in the 285 Appendix of Sambridge et al. 2008, several of the contours are wrongly labeled). The correction 286 for integrated absorbance, on the other hand, would be significantly overestimated by use of a 287 linear-absorbance error estimate. This is illustrated in Figure 4, wherein the correction factor, defined as the ratio of A_{tot} to $3 \times A_{unpol, av}$, is plotted against the average maximum linear 288 289 absorption for the entire population of calculated unpolarized spectra with a given thickness. The 290 curve labeled 'experimental' is defined by the reference spectra shown in Figure 2. Note the 291 point of inflection in the curve at maximum linear unpolarized absorbance of 0.24, which 292 corresponds to the point at which the maximum linear unpolarized absorbance switches from 3612 cm⁻¹ to lower-frequency peaks (see also Fig. 2). The correction factors calculated from the 293 294 average of calculated unpolarized spectra are valid only for the specific principal polarized 295 spectra that were used to calculate the unpolarized spectra. When the same calculation is 296 performed using the principal polarized spectra of a natural olivine (KLV23 of Bell et al. 2003), 297 the correction factors fall on a different curve to that defined by the experimental sample (Fig. 4). 298 Other samples with different relative populations of OH species would each have their own 299 sample-specific correction scheme. For example, the Pakistani olivine of Kovács et al. (2008), 300 which has very strong absorbance in $\mathbf{E} || \mathbf{a}$ and $\mathbf{E} || \mathbf{b}$ spectra relative to the $\mathbf{E} || \mathbf{c}$ spectrum over much 301 of the range of OH absorbance, would be expected to have a very large correction factor for high 302 absorption intensities. Of course, in a natural olivine with low H concentrations the sample 303 thickness may be reduced to limit the peak absorption intensities. The purpose of the calculation, 304 however, is to illustrate how the correction factor varies as the absorption profiles of the 305 reference spectra change. The different trends in Figure 4 show that, even though the correction 306 factor does appear to change somewhat systematically with the average maximum linear 307 unpolarized absorbance, the relationship varies from one sample to another.

While unpolarized olivine spectra with maximum polarized absorbances that do not exceed 0.3 absorbance units provide good estimates for $\int A_{tot}/3$, it is difficult to estimate the maximum polarized absorbance from unpolarized spectra. Indeed, for a very strongly polarized band that

has a minimum polarized absorbance of 0 in the plane of the sample, the unpolarized absorption cannot exceed 0.3 absorbance units (Eq. 9), regardless of the maximum value of polarized absorbance, and how correspondingly large a correction factor would be needed. It is consequently difficult to make a general recommendation for an acceptable level of unpolarized absorbance (Fig. 4).

The important point is that correction factors based on empirical correlations with a linear absorbance are not transferrable between samples. In the following section a protocol is described that allows the estimation of $\int A_{tot}$ from multiple unpolarized spectra, without need for empirical correction.

320 Estimation of $\int A_{tot}$ from unpolarized measurements

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Despite the offset between $\int A_{unpol, av}$ and $\int A_{tot}$, it is theoretically possible to estimate $\int A_{tot}$ from 321 322 unpolarized measurements provided that one works in the transmission domain. The technique described by Sambridge et al. (2008) for estimating $\int A_{tot}$ from unpolarized measurements is 323 based on the approximation of describing the absorbance indicatrix with the equation that 324 325 correctly describes the transmittance indicatrix. By working in the transmission domain, and using the correct theory of unpolarized absorbance (Eq. 8 and 9), it is possible to estimate $\int A_{tot}$ 326 327 without approximation. The following theoretical development follows that set out in Sambridge 328 et al. (2008) for absorbance, i.e., the equations applied as an approximation in the absorbance 329 domain are in fact applicable without approximation in the transmission domain.

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330 As established previously (Eq. 14), unpolarized transmittance, T_{unpol} , is related to the 331 principal polarized transmittances as:

332
$$T_{\text{unpol}}(\psi,\varphi) = \frac{1}{2}(T_{a}(\cos^{2}\varphi \ \cos^{2}\psi + \sin^{2}\psi) + T_{b}(\cos^{2}\varphi \ \sin^{2}\psi + \cos^{2}\psi) + T_{c}\sin^{2}\phi)$$
 (15)

Following the same reasoning that Sambridge et al. (2008) used to estimate A_{unpol} , the principal polarized transmittances can be related to the maximum, minimum and average unpolarized transmittance ($T_{unpol, max}$, $T_{unpol, min}$, and $T_{unpol, av}$, respectively). By integrating Equation 15 over all angles (ψ, ϕ) one obtains $T_{unpol,av} = \frac{1}{3}(T_{max} + T_{int} + T_{min})$, where T_{max} , T_{int} and T_{min} are principal polarized transmittance and $T_{max} > T_{int} > T_{min}$. By differentiation of Equation 15 with respect to ψ and ϕ the maximum and minimum unpolarized transmittances, $T_{unpol, max}$, and $T_{unpol, min}$, are shown to be equal to $\frac{1}{2}(T_{max} + T_{int})$ and $\frac{1}{2}(T_{min} + T_{int})$, respectively. Thus,

$$340 T_{\min} = 3T_{\text{unpol, av}} - 2T_{\text{unpol, max}} (16a)$$

$$341 T_{\rm int} = 2(T_{\rm unpol,\,min} + T_{\rm unpol,\,max}) - 3T_{\rm unpol,\,av} (16b)$$

$$342 T_{\text{max}} = 3T_{\text{unpol, av}} - 2T_{\text{unpol, min}} (16c)$$

Given a sufficiently large number of unpolarized spectra, the total polarized absorbance can be estimated. Note that simply averaging in the transmission domain does not provide an estimate of $\int A_{tot}$. The average polarized absorption cannot be derived from $T_{unpol, av}$, which is unfortunate, since the latter can be estimated relatively precisely. By applying Equations 16a-16c, each of the three principal polarized absorbances can be estimated, though at a given frequency it is not possible to assign those absorbances to particular principal spectra, since the distribution of relative absorbances varies with frequency. The sum of the principal absorbances 350 (A_{tot}) then is given by addition in the absorbance domain, so that an estimate of $\int A_{tot}$ can be 351 obtained in the following way:

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Collect multiple unpolarized FTIR spectra from randomly oriented samples with ideally
 constant, or else near-constant, thickness. Small variations in sample thickness may be
 corrected in the absorbance domain.

- 356 2. Determine the average $(T_{unpol,av})$, maximum $(T_{unpol,max})$ and minimum $(T_{unpol,min})$ transmittance
- 357 at each recorded frequency.
- 358 3. Create a total absorbance spectrum by calculating absorbance at each frequency as:

$$A_{\text{tot}} = -\log_{10} \left((3T_{\text{unpol},\text{av}} - 2T_{\text{unpol},\text{max}}) (2T_{\text{unpol},\text{min}} + 2T_{\text{unpol},\text{max}} - 3T_{\text{unpol},\text{av}}) (3T_{\text{unpol},\text{av}}) - 2T_{\text{unpol},\text{min}}) \right)$$

$$(17)$$

4. Integrate the total absorbance spectrum to obtain an estimate of $\int A_{tot}$.

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361 Unfortunately, the errors in estimated $\int A_{tot}$ tend to be much greater than those calculated by 362 Sambridge et al. (2008) for absorbance domain calculations. In fact, when measuring spectra 363 with strongly anisotropic absorbance, it is possible in a small sampling of unpolarized 364 measurements for $2T_{unpol, max}$ to exceed $3T_{unpol, av}$ in the region of the most intense absorption 365 bands, in which case Equation 17 fails. In these cases adding further measurements may allow 366 the calculation to succeed, but reducing the sample thickness is more likely to improve the 367 estimate of A_{tot} . In general, unpolarized measurements of samples where the maximum principal 368 polarized linear absorbance does not exceed 1 absorbance unit are unlikely to succumb to this 369 limitation. Under such conditions, though, simply averaging the integrated unpolarized 370 absorbance is likely to provide an acceptable estimate for $\int A_{tot}$. Applying Equation 17 removes 371 the bias toward underestimation of areas, but for a realistically low number of measurements it is 372 doubtful that the new theory improves significantly over taking an average of integrated 373 unpolarized absorption.

374 Convergence of unpolarized measurements

The number of spectra required for convergence of $\int A_{unpol, av}$ can be estimated through a 375 376 numerical simulation (Fig. 5). Random directions of incidence with spherical coordinates $\theta = m\pi$ and $\phi = \cos^{-1}(n)$, where *m* and *n* are random variates on [0,2) and [1,-1], respectively, were 377 378 chosen using the linear congruential generator by L'Ecuyer with added Bayes-Durham shuffle 379 (Press et al. 2007). For each incident direction, an unpolarized spectrum was calculated from the 380 reference spectra shown in Figure 2 after normalization to a thickness that varied randomly 381 between 45 and 55 µm. Thickness was corrected to 50 µm in the absorbance domain, which adds 382 a small source of simulated measurement error owing to deviation from the Beer-Lambert 383 relationship, and normally distributed noise with a standard deviation equivalent to 0.001 384 absorbance units was added in the absorbance domain. Convergence of A_{unpol} as is shown in Figure 5a. In this example, $3 \times \int A_{unpol, av}$ converges to 70.9 cm⁻¹, which underestimates $\int A_{tot}$ by 385 12%, in agreement with the population average calculated in the preceding section (cf. Fig. 2). 386 Using Equation 17 to estimate A_{tot} , on the other hand, results in convergence to within 0.5% of 387 the expected value of 80.2 cm⁻¹ (Fig. 5b). Note, though, that there is a tradeoff between the 388 389 standard error resulting from Equation 17 and the systematic error of averaging absorption (Fig. 5). Thus, with realistic numbers of measurements (n<30), $\int A_{tot}$ estimated by Equation 17 offers 390 391 only a minor improvement over the average of integrated unpolarized absorbances.

392 Uncertainties associated with unpolarized measurements

393 Without knowledge of the principal polarized spectra, it is impossible to know how far a 394 measurement of unpolarized absorbance at any particular frequency deviates from the average 395 principal polarized absorbance. This is nicely illustrated by the unpolarized spectrum shown in Figure 1, where unpolarized absorbance measured in the region around 3600 cm⁻¹ deviates 396 397 strongly from the average of A_a and A_c , while the same measured absorbance values at ~3560 cm⁻¹ are very close to the average of A_a and A_c . Using multiple measurements to estimate the 398 399 total principal polarized absorbance by way of Equation 17 can lead to large errors over the 400 frequency ranges where principal absorbance is strongly anisotropic. This is because anisotropic 401 principal absorbance can lead to a long tailed and potentially skewed distribution of T_{unpol} , from 402 which it is difficult to estimate maximum, minimum and average values with a small sampling. 403 Moreover, with no knowledge of the principal polarized spectra and a small sampling of 404 unpolarized measurements, the uncertainty in Atot estimated using Equation 17 cannot be 405 assessed. Thus, using unpolarized measurements to estimate A_{tot} , either by applying Equation 17 406 or by averaging integrated absorbance, relies on the unknown principal polarized spectra having 407 an acceptable degree of anisotropy. For this reason, incircumspect application of Equation 17 is 408 to be avoided.

Partial polarization of unpolarized beams within the spectrometer can provide an additional
source of uncertainty that is not considered in the foregoing theoretical calculations. The effects
of partial polarization and polarizer efficiency are discussed by Libowitzky and Rossman (1996).

412

CONCLUSIONS

413 1. Unpolarized absorbance is equal to the logarithm of the average of any two perpendicular414 measurements of transmittance with polarization directions in the plane of the sample, as

previously stated by Libowitzky and Rossman (1996). The systematic deviation of linear
unpolarized absorbance from the average principal polarized absorbance is consequently
underestimated by alternative theories of unpolarized absorbance (Sambridge et al. 2008;
Kovács et al. 2008; Kovács et al. 2010).

419 2. There is no theoretical justification for applying a correction based on the deviation of a 420 linear absorbance in one part of the spectrum from the average polarized absorbance at that 421 frequency to the integral of multiple absorbance bands. The correction scheme proposed by 422 Kovács et al. (2010), while based on an underestimation of the correction factor for linear 423 unpolarized absorbance, results in over-correction when applied to integrated unpolarized 424 OH absorption bands in the olivines considered here.

3. Based on calculations of unpolarized spectra in the OH-stretching region from natural and hydrothermally annealed olivines, averages of >10 unpolarized spectra where the maximum linear absorbance does not exceed 0.3 absorbance units are expected to give a reasonable estimate of $\int A_{tot}$. This conclusion is valid only for samples where the average degree of anisotropy in the principal polarized spectra is not significantly greater than that of the examples used in the simulations.

4. Multiple unpolarized spectra of randomly oriented samples can be used to estimate $\int A_{tot}$ by 432 application of Equation 17. Given a sufficient number of measurements, the new theory 433 reproduces $\int A_{tot}$ without systematic bias, but is limited by sensitivity to the shape of the 434 unpolarized transmittance distribution. For reasonably low numbers of measurements, the 435 new procedure is not a significant improvement over averaging integrated unpolarized 436 absorbance. 437 5. The error of an estimate of $\int A_{tot}$ based on unpolarized measurements varies with the degree of 438 anisotropy of the principal polarized spectra, and cannot be estimated accurately from a small 439 sampling of unpolarized spectra.

440

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495

FIGURE CAPTIONS

496 **Figure 1**

497 Comparison between measured and calculated unpolarized infrared spectra of olivine in the OH 498 stretching region. Polarized and unpolarized spectra from an olivine that was hydrothermally 499 annealed at 5 GPa and 1100 °C are from Kohlstedt et al. (1996) and were measured in a 126 µm 500 thick section that was oriented such that the direction of light propagation was parallel to the **b** 501 axis. Polarized spectra recorded with the directions of polarization parallel to the a and c axes are 502 labeled $\mathbf{E} \| \mathbf{a}$ and $\mathbf{E} \| \mathbf{c}$, respectively, and the unpolarized spectrum is plotted as a dashed line. 503 Unpolarized spectra were calculated from the $\mathbf{E} \parallel \mathbf{a}$ and $\mathbf{E} \parallel \mathbf{c}$ spectra using the 'A integral' theory 504 of Sambridge et al. (2008) and the 'T integral' theory of Libowitzky and Rossman (1996). The 505 average of the E||a and E||c spectra, corresponding to the 'A-theory' approximation of Sambridge 506 et al. (2008), is shown for comparison. Absorbance values read from the right axis are for a 507 sample thickness of 126 µm.

508 **Figure 2**

Calculation of average unpolarized absorbance spectra for olivine in the OH stretching region. Following the method described in the text, unpolarized spectra were calculated from the principal polarized spectra of an olivine containing 0.2 wt. % H_2O (M475 of Withers et al. 2011), shown in the inset. For each sample thickness labeled on the plot, 10⁴ unpolarized spectra were synthesized for evenly distributed directions of light propagation. Averages of calculated spectra are shown.

515 **Figure 3**

Histograms of integrated absorbance in the range $3100-3800 \text{ cm}^{-1}$ for populations of unpolarized spectra calculated with 10^4 evenly distributed directions of light propagation in an olivine. The polarized absorbance spectra used in the calculation (see Fig. 2) were scaled to the equivalent of 30, 50 and $100 \mu \text{m}$ thicknesses of olivine with 0.2 wt. % H₂O (or equivalent products of thickness and concentration). Unpolarized absorbances were normalized to 1 cm sample thickness for comparison with the average integrated absorbance of the normalized principal

522 polarized spectra, indicated by the vertical dashed line.

523 **Figure 4**

524 Correction factors, defined as the ratio of the average integrated absorbance of principal polarized spectra ($[A_{tot}/3)$) to the average integrated unpolarized absorbance ($[A_{unpol}]_{av}$). Unbroken 525 526 curves represent averages of large (10^4) populations of calculated spectra with evenly distributed 527 light propagation directions (see text for details). The population-average correction factors are 528 plotted against the average maximum linear absorbance in the unpolarized spectrum. The 529 calculation is unique to a set of principal polarized spectra. Spectra from an experimental sample 530 (M475, shown in Fig. 2) and a natural sample (KLV23 of Bell et al. 2003) are used to create two 531 examples. These correction factors are not transferrable between samples that have different 532 patterns of principal polarized absorption. The correction scheme described by Kovács et al 533 (2010), which is based on the linear-absorbance error estimates of Sambridge et al. (2008) 534 calculated using the 'A integral', is also shown. The grey hatched area falls outside of the space 535 of possible population-average correction factors for a single absorption band with a Gaussian 536 shape.

537 **Figure 5**

(a) Convergence of averaged integrated absorbance of unpolarized infrared spectra of olivine ($3 \times [A_{unpol})$), calculated from the reference spectra shown in Figure 2 after normalization to 50 µm thickness. Unpolarized spectra were calculated using Equation 14 for random directions of light propagation (see text for details). Error bars are standard error of the mean. The total integrated absorbance of the principal polarized spectra ($[A_{tot}]$) is marked by the dashed line. (b) Convergence of estimated $[A_{tot}] calculated from unpolarized spectra using Equation 17.$ 545

APPENDIX 1

546 Calculation of T_{unpol} for a perfect polarizer by integration of absorbance.

547 According to the alternative theory of unpolarized absorption, unpolarized absorbance is 548 given by:

549
$$A_{\text{unpol}} = -\frac{1}{2\pi} \int_0^{2\pi} \log_{10} (10^{-A_{\text{max}}} \sin^2 \theta + 10^{-A_{\text{min}}} \cos^2 \theta) \ d\theta, \tag{18}$$

550 where θ is defined with respect to the direction of minimum absorbance (Sambridge et al. 2008).

551 For a perfect polarizer, $A_{\text{max}} = \infty$ and $A_{\text{min}} = 0$. The term inside the integral then reduces to 552 $\log_{10}(\cos^2 \theta)$:

553
$$A_{\text{unpol}} = -\frac{1}{2\pi} \int_0^{2\pi} \log_{10}(\cos^2 \theta) \ d\theta$$
(19)

554 The symmetry of the function $\log_{10}(\cos^2(\theta))$ suggests

555
$$A_{\text{unpol}} = -\frac{2}{\pi} \int_0^{\pi/2} \log_{10}(\sin^2 \theta) \ d\theta$$
 (20)

556
$$= -\frac{4}{\pi} \int_0^{\pi/2} \log_{10}(\sin \theta) \ d\theta$$
 (21)

557 =
$$-\frac{4}{\pi \ln 10} \int_0^{\pi/2} \ln(\sin \theta) \, d\theta.$$
 (22)

558 The solution to the definite integral in Equation 22 is $-\pi/2 \ln(2)$ (Euler 1769), so:

559
$$A_{\text{unpol}} = 2 \log_{10}(2) = -\log_{10}(4^{-1}),$$
 (23)

560 leading to an incorrect value for the transmittance of a polarizer:

561
$$T_{\text{unpol}} = \frac{1}{4}$$
 (24)









Average maximum linear unpolarized absorbance



