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2 **Elasticity of franklinite and trends for transition-metal oxide spinels**

3 Hans J. Reichmann¹, Steven D. Jacobsen², Tiziana Boffa Ballaran³

4 ¹Helmholtz Centre Potsdam, German Research Centre for Geosciences GFZ, 14473 Potsdam,
5 Germany

6 ²Department of Earth and Planetary Sciences, Northwestern University, Evanston, IL 60208

7 ³Bayerisches Geoinstitut, Universität Bayreuth, D-95440 Bayreuth, Germany

8

9 **Abstract**

10 The pressure dependence of single-crystal elastic moduli of a natural Mn-rich franklinite,
11 $(\text{Mn}_{0.40}\text{Fe}^{2+}_{0.16}\text{Zn}_{0.37}\text{Mg}_{0.03})(\text{Fe}^{3+}_{1.94}\text{Al}_{0.08})\text{O}_4$, has been determined by GHz-ultrasonic
12 interferometry in a diamond-anvil cell to 9.8 GPa. The room-pressure elastic constants of
13 franklinite are $C_{11} = 244(3)$ GPa, $C_{12} = 142(4)$ GPa, and $C_{44} = 77(2)$ GPa. Linear pressure
14 derivatives of C_{11} , and C_{12} are 4.3(3) and 3.8(3), respectively, whereas the C_{44} modulus
15 exhibits softening, fitted in the $P \leq 10$ GPa pressure range to $C_{44} = 77(2) + 0.29(2)P -$
16 $0.018(2)P^2$ GPa. The average of Hashin-Shtrikman bounds on the adiabatic bulk modulus
17 (K_{S0}) of franklinite is 175(3) GPa, with pressure derivative $K'_S = 4.3(3)$, and the shear modulus
18 $G_0 = 66(2)$ GPa with $G' = 0.09(3)$. The isothermal compressibility of franklinite was
19 determined from a separate high-pressure, single-crystal X-ray diffraction experiment to 7.8
20 GPa, yielding $K_{T0} = 173.5(7)$ GPa fitted with a fixed pressure derivative of $K'_T = 4$. When K' is
21 fixed to the ultrasonic value of 4.3, we obtain $K_{T0} = 172.2(7)$ GPa. In contrast to iron-free
22 gahnite (ZnAl_2O_4), franklinite exhibits pressure-induced mode softening of C_{44} similar to
23 magnetite (Fe_3O_4). Between end-member compositions ZnFe_2O_4 (franklinite) and MnFe_2O_4
24 (jacobsite) the bulk modulus decreases linearly with increasing %Mn, however we observe
25 non-linear behavior in other elastic moduli, especially C_{44} , which displays a pronounced
26 negative anomaly for the mid-range Mn composition. Applying Birch's law to $AB_2\text{O}_4$ -type
27 spinels reveals that oxide spinels containing transition metals on both *A* and *B* sites follow a
28 distinct trend from other spinels.

29 Keywords: spinel, elasticity, transition metals, equation of state

30 **1. Introduction**

31 Spinel phases with generalized formula AB_2O_4 (8 formula units per unit cell, space
32 group $Fd\bar{3}m$), are common non-silicate oxides in the crust and upper mantle. A special
33 feature of this group is the ability to host a wide range of multi-valence transition metals. The
34 compositional flexibility of spinels results in a multitude of natural and synthetic phases,
35 many of which are used in magnetic, ferroelectric, semiconducting, and superconducting
36 applications. MnZn-ferrites are among the most widely used soft ferrites in various types of
37 transformers and magnetic recording heads (Sugimoto 1999). In petrological application,
38 spinel solid solutions are recognized in marble and skarn deposits (Carvalho and Sclar 1988;
39 Frondel and Baum 1974) and used in many geothermobarometers (e.g. Nichols et al. 1992).
40 The $(Mg,Fe)_2SiO_4$ silicate spinel, ringwoodite, is likely a dominant phase in the lower-part of
41 the mantle transition zone from 520-660 km depth.

42 Many spinels undergo phase transitions at elevated pressures (e.g. Fei et al. 1999;
43 Funamori et al. 1998; Irifune et al. 2002; Levy et al. 2000) whereas other minerals with spinel
44 structure do not exhibit a phase change to very high pressure. For example, gahnite ($ZnAl_2O_4$)
45 does not exhibit a pressure-induced phase change up to 43 GPa (Levy et al. 2001). A closer
46 look at the relation between phase transition and mineral composition reveals that spinels with
47 two transition metals on the cation sites, like magnetite ($Fe^{2+}Fe^{3+}_2O_4$), tend to undergo the
48 phase changes at a lower pressure than those with none or only one transition metal, such as
49 spinel ($MgAl_2O_4$) and gahnite ($ZnAl_2O_4$).

50 Pressure-induced softening of shear-mode elastic moduli, such as C_{44} , is often
51 associated with pressure-induced phase changes (e.g. Carpenter and Salje 1998). Magnetite
52 undergoes a phase transition at 18-20 GPa (e.g. Dubrovinsky et al. 2003; Haavik et al. 2000;
53 Pasternak et al. 1994) and exhibits a negative pressure dependence of C_{44} (Reichmann and
54 Jacobsen 2004), whereas gahnite exhibits no phase transitions to 43 GPa (Levy et al. 2001)

55 and has a positive pressure dependence of C_{44} (Reichmann and Jacobsen 2006). In this paper
56 we further study transition-metal oxide spinels and the high-pressure behavior of their elastic
57 constants (C_{ij}), which reflect critical differences in interatomic bonding forces with varying
58 composition and pressure.

59 Li and Fisher (1990) measured the elastic constants of single-crystal ZnFe_2O_4 at ambient
60 conditions using the ultrasonic phase comparison method from 20-75 MHz. The elastic
61 constants of MnFe_2O_4 under high magnetic field were measured with ultrasonic methods by
62 Sakurai (1964). Li et al. (1991) determined the elastic constants of Co-ferrite spinels and
63 summarized the elastic constants of end-member transition-metal spinels. To further
64 characterize the elastic properties of MnZn-ferrite solid solutions, especially at high pressures,
65 we have studied a natural franklinite single crystal of mid-range composition,
66 $(\text{Mn}_{0.40}\text{Fe}^{2+}_{0.16}\text{Zn}_{0.37}\text{Mg}_{0.03})(\text{Fe}^{3+}_{1.94}\text{Al}_{0.08})\text{O}_4$, using GHz-ultrasonic interferometry in
67 diamond-anvil cells. We measured the complete elastic tensor of franklinite to 9.8 GPa and
68 report a non-linear pressure derivative of C_{44} , which approaches zero at above ~ 7 GPa. We
69 also observe non-linear dependence of C_{44} on composition between Zn and Mn end-members.
70 The isothermal compressibility of franklinite was also determined from separate single-crystal
71 X-ray diffraction experiments. Finally, Birch's law is evaluated for spinels containing one or
72 two transition metals on the *A* and *B* sites compared with other spinels.

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75 **2. Experimental**

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77 **Sample characterization**

78 Franklinite is an ordered normal spinel (Andreozzi et al. 2001; Pavese et al. 2000) and
79 exhibits two transition metals (Zn and Fe) on the cation sites. The natural franklinite sample
80 examined in this study originates from its type-locality in Franklin, New Jersey. Because of
81 the crystal's octahedral form, it was possible to cut and polish sections parallel to (100) and
82 (111) planes to within a few degrees. The basal plane of an octahedron represents the (100)

83 plane and the facets represent the (111) plane. We estimate the uncertainty in orientation of
84 the polished sections is about $\pm 2^\circ$. A (100) section was double-side polished to within a few
85 arc-seconds of parallel and having a thickness of $\sim 246 \mu\text{m}$, measured using a micrometer, for
86 room-pressure measurements of the P-wave velocity. For high-pressure experiments, (100)
87 and (111) sections of the franklinite crystal were polished with an optical-quality finish
88 together into thin section on the same slide to a thickness of $\sim 45 \mu\text{m}$, estimated by micrometer
89 measurement. Electron microprobe analysis revealed spatial variations of the iron and zinc
90 content. The Fe^{3+} content was determined by charge balance, and varies from 1.89 to 1.98
91 atoms per formula unit (apfu) and the Zn content varies from 0.35 to 0.41 apfu. The average
92 composition over 10 measurement points of the sample is $(\text{Mn}_{0.40}\text{Fe}^{2+}_{0.16}\text{Zn}_{0.37}\text{Mg}_{0.03})(\text{Fe}^{3+}_{1.94}$
93 $\text{Al}_{0.08})\text{O}_4$ (see **Table 1**). The density, $\rho = 5.038 \text{ g/cm}^3$ was calculated using the measured
94 lattice parameters and the stoichiometric composition determined from microprobe analysis

96 **GHz-ultrasonic interferometry**

97 The pressure-dependence of sound-wave velocities in franklinite were determined by
98 GHz-ultrasonic interferometry in a diamond anvil cell (Jacobsen et al. 2002; Reichmann et al.
99 1998; Spetzler et al. 1993). In this technique, P-wave ultrasonic tone bursts are generated by
100 thin-film ZnO transducers, coupled to the DAC through a single-crystal Al_2O_3 buffer rod,
101 attached to the table face of one diamond anvil (Reichmann et al. 1998). To produce S-waves,
102 a P-to-S conversion buffer rod made of yttrium-aluminum-garnet is attached to the table of
103 the diamond anvil (Jacobsen et al. 2004). Using tone bursts of 20-100 ns, strain waves
104 reflected at the front side overlap with reflections from the back side of the cylinder-shaped
105 sample. Scanning the frequency between about 800 and 1200 MHz results in an interference
106 pattern, from which the round trip travel time through the sample is determined analytically
107 from the frequency spacing (Δf) between adjacent maxima and minima, corresponding to
108 integer and half-integer acoustic wavelengths in the round-trip sample distance (e.g. Spetzler

109 et al. 1993). Longitudinal and transverse sound wave velocities $v_{P,S}$ are calculated using the
110 equation

$$111 \quad v_{P,S} = 2L/t \quad (\text{eq. 1})$$

112 where L is the sample thickness and t is the measured round-trip travel time.

113 To obtain transmission and quality acoustic reflections at GHz-frequencies where
114 acoustic wavelengths at 1-GHz are in the order of 10 μm , the samples must be optically
115 scratch-free with parallel front and back sides. The sample is placed on the diamond, which is
116 coupled to the acoustic buffer rod with the transducer (Jacobsen et al. 2004; Reichmann et al.
117 1998). No bonding agent between diamond and sample was used. A 16:3:1 methanol -
118 ethanol - water mixture was used as the pressure-transmitting medium, and a small amount of
119 silica aerogel was included in the sample chamber to gently press the sample against the
120 acoustic anvil for coupling and to prevent the alcohol mixture from flowing between the anvil
121 and sample (Reichmann and Jacobsen 2004). The pressure in the sample chamber was
122 determined before and after each run by the ruby fluorescence method (Mao et al. 1986).

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124 **X-ray Diffraction**

125 A crystal plate polished parallel to (111) measuring 45 μm thick and 150 μm across
126 was loaded into a BGI-designed diamond-anvil cell (Allan et al. 1996) with a culet size of 600
127 μm together with a ruby chip as the pressure calibrant (Mao et al. 1986). A 4:1 mixture of
128 methanol : ethanol was used as pressure transmitting medium. A steel-plate (T301) gasket
129 was employed between the anvils, pre-indented to 90-100 μm and drilled with a 280 μm
130 sample chamber. The unit-cell parameters were determined at various pressures up to 7.8 GPa
131 at ambient temperature on a Huber four-circle diffractometer (MoK α radiation) using the 8-
132 position centring procedure according to King and Finger (1979) and Angel et al. (2000). The
133 centring procedure and vector least squares refinement of the unit cell constants were

134 performed using SINGLE04 (Angel and Finger 2011). Unit cell parameters at various
135 pressures are reported in **Table 2**.

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137 **3 Results**

138 The unit-cell lattice parameter of the franklinite crystal measured at ambient
139 conditions is 8.4456(2) Å, compared to $a = 8.4412(2)$ Å of synthetic (ZnFe₂O₄) spinel (Levy
140 et al. 2000) and $a = 8.506(1)$ Å for MnFe₂O₄ (Butler and Buessem 1962). Compression data
141 from the single-crystal diffraction experiment are shown in **Figure 1a**. From analysis of the
142 normalized stress $F_E = P/[3f_E (1+2f_E)^{5/2}]$ versus Eulerian strain $f_E = [(V_0/V)^{2/3} - 1]/2$ plot (Angel
143 2000), it is clear that a 2nd order Birch–Murnaghan equation of state (BMII-EoS), i.e. with K'
144 fixed to the value of 4, is appropriate for fitting the P - V data (**Figure 1b**). The BMII-EoS
145 error-weighted fit resulted in an isothermal bulk modulus of $K_{T0} = 173.5(7)$ GPa with a room
146 pressure volume $V_0 = 602.41(4)$ Å³, identical to that measured by single-crystal X-ray
147 diffraction, with $V_0 = 602.40(4)$ Å³. The observation that the franklinite P - V data at $P \leq 10$
148 GPa can be described using a BMII-EoS, i.e. $K' = 4$ is in good agreement with the $K' = 4.3(3)$
149 obtained from the ultrasonic measurements (see next section). Fixing the value of K' to 4.3 we
150 obtain $K_{T0} = 172.2(7)$ GPa and an identical value of $V_0 = 602.41(4)$ Å³. Considering that the
151 values of bulk moduli obtained from the two different fitting procedures coincident within
152 one standard deviation, the BMII-EoS values were used to calculate the length-changes on
153 compression during the ultrasonic experiments.

154 The elastic tensor (C_{ij}) of cubic minerals contains three independent moduli: C_{11} , C_{12}
155 and C_{44} . The elastic constants are related to the sound velocities by:

$$156 \quad \rho(v_p^{[100]})^2 = C_{11} \quad (\text{eq. 2})$$

$$157 \quad \rho(v_s^{[100]})^2 = C_{44} \quad (\text{eq. 3})$$

$$158 \quad \rho(v_p^{[111]})^2 = (C_{11} + 4C_{44} + 2C_{12})/3 \quad (\text{eq. 4})$$

159 where ρ is the density and $v_{p,s}$ are longitudinal or transversal sound wave velocities in the
160 $[uvw]$ direction, denoted as superscript (e.g. Brugger 1965).

161 The room-pressure sample used to measure v_p in the $[100]$ direction was $246 (\pm 1) \mu\text{m}$.
162 The measured P-wave travel time at room-pressure was $70.65(2)$ ns, resulting in $v_p^{[100]} =$
163 $6960(30)$ m/s The thickness of samples used in the high-pressure diamond-anvil cell
164 experiments was much less, about $45 \mu\text{m}$, as determined by thickness measurements with a
165 micrometer. Both (100) and (111) direction plates were polished together on the same slide.
166 To determine the initial length (L_0) of the high-pressure samples, we measured the (100) thin
167 section P-wave travel time and then calculated L_0 using equation (1) from the measured room-
168 pressure velocity (**Table 3**). The calculated thickness L_0 of the high-pressure samples was
169 $44.83(3) \mu\text{m}$, in agreement with the micrometer measurement of $45(1) \mu\text{m}$. The sample
170 thickness L at high pressure was determined using the relation $(V/V_0)^{1/3} = (L/L_0)$, where V is
171 the volume and V_0 is the volume at $P=0$. This relation holds for cubic (and isotropic) crystals.
172 To determine the volume V at high pressure in the DAC, we utilized the isothermal equation
173 of state determined in this study, with $K_{T0}=173.5$ GPa and fixed $K' = 4$. Knowing $L_0(44.83$
174 $\mu\text{m})$, $V_0 = (L_0)^3$ and $V(P)$ from BMII-EoS, we calculated from the above equation $L(P)$.

175 In three separate experiments, the longitudinal and shear wave velocities of franklinite
176 were determined in $[100]$ and $[111]$ directions as a function of pressure (**Table 3**). An S-wave
177 acoustic interferogram and resulting sample travel times in franklinite at 9.8 GPa are shown in
178 **Figure 2**. The room pressure values of elastic constants are $C_{11} = 244(3)$ GPa, $C_{12} = 142(4)$
179 GPa and $C_{44} = 77(2)$ GPa. Variation of the elastic constants with pressure are shown in
180 **Figure 3**. The pressure derivatives of C_{11} and C_{12} are linear over the measured pressure range
181 of $P \leq 10$ GPa, with $dC_{11}/dP = 4.3(3)$ and $dC_{12}/dP = 3.8(3)$, shown by the solid lines in
182 **Figure 3** representing linear fits to the data. Whereas, the C_{44} shear-mode sensitive modulus
183 exhibits non-linear behavior indicative of mode softening, fitted to a polynomial equation at
184 pressures $P \leq 10$ GPa with $C_{44} = 77(2) + 0.29(2)P - 0.018(2)P^2$ GPa.

185 The shear and adiabatic bulk moduli were calculated from the average of Hashin-
186 Shtrikman bounds (Hashin and Shtrikman 1962; Hill 1952). Hashin and Shtrikman (1962)
187 developed variational principles in elasticity to constrain aggregate moduli (K and G) of
188 anisotropic media. In our study G_0 was found to be 66(2) GPa and $K_{S0} = 175(3)$ GPa.
189 Variation of the bulk and shear moduli of franklinite with pressure are shown in **Figure 4**.
190 Due to the absence of any curvature in the bulk and shear moduli as a function of pressure, we
191 fitted linear equations to the data with $K' = 4.3(3)$ and $G' = 0.09(3)$ (**Figure 4**). The aggregate
192 sound velocities, which are calculated according to the equations

$$193 \quad v_p = ((K+4/3 G)/\rho)^{1/2} \quad (\text{eq. 5})$$

$$194 \quad v_s = (G/\rho)^{1/2} \quad (\text{eq. 6})$$

195 vary with pressure according to the relations $v_p = 7233(40)$ (m/s) + $38.3(1) P$ and $v_s =$
196 $3630(25)$ (m/s) – $7.0(5) P$.

197 **4. Discussion**

198 The natural franklinite sample in this study ($\text{Mn}_{0.40}\text{Fe}^{2+}_{0.16}\text{Zn}_{0.37}\text{Mg}_{0.03}(\text{Fe}^{3+}_{1.94}\text{Al}_{0.08})$
199 O_4 , represents a solid solution between ideal franklinite (ZnFe_2O_4) and jacobsite (MnFe_2O_4).
200 Because the Mn-Zn ferrites are so widely used as soft ferrite cores in electronics (e.g.
201 Sugimoto 1999), we first examine trends of elasticity along the Mn-Zn join.

202 The adiabatic bulk modulus of this natural franklinite ($K_{S0} = 175$ GPa) is about the
203 average of bulk moduli for ZnFe_2O_4 ($K_{S0} = 182.5$ GPa) from Li and Fisher (1990) and
204 MnFe_2O_4 ($K_{S0} = 161$ GPa) from Sakurai (1964), resulting in a variation of bulk modulus with
205 Zn-Mn content that is approximately linear (**Figure 5**). The shear modulus of this natural
206 franklinite ($G = 66$ GPa) lies 5% below the linear extrapolation between end-member
207 compositions (**Figure 5**), but is not lower than pure MnFe_2O_4 .

208 In contrast, individual elastic constants show non-linear compositional dependence,
209 especially C_{44} (**Figure 5**). With increasing Mn-content relative to ZnFe_2O_4 , the C_{11} elastic
210 constant of the mid-range composition is about 8 GPa (~3%) above the linear extrapolation

211 between end members, whereas C_{12} is only 1% below the linear extrapolation between Zn and
212 Mn end members. The C_{44} elastic constant for this mid-range composition falls about 14 GPa
213 (16%) below the trend line extrapolated between end members, and furthermore exhibits a C_{44}
214 = 77 GPa that is about 11% lower than the MnFe_2O_4 spinel with $C_{44} = 86$ GPa (Sakurai 1964).
215 This trend implies that there is a minimum in C_{44} somewhere between ZnFe_2O_4 and MnFe_2O_4 ,
216 possibly near this measured mid-range composition (**Figure 5**). A high-pressure modification
217 of ZnFe_2O_4 was reported by Levy et al. (2000) above ~ 25 GPa, but on the basis of in-situ
218 high-pressure powder X-ray diffraction a definitive structure could not be identified, although
219 CaTi_2O_4 or CaMn_2O_4 -type structures were determined to be the best candidates. Because of
220 the minimum in C_{44} between ZnFe_2O_4 and MnFe_2O_4 we predict a similar phase transition to
221 occur in natural franklinite but at lower pressures.

222 Compared with ZnAl_2O_4 (gahnite), which has $K_{S0} = 209(5)$ GPa and $G_0 = 104(3)$ GPa
223 (Reichmann and Jacobsen 2006), the bulk and shear moduli of this mid-range franklinite are
224 17% and 36% lower. Whereas the elastic constants C_{11} , C_{12} of this franklinite are about 15%
225 lower than ZnAl_2O_4 , the C_{44} elastic constant is nearly 50% lower (**Table 4**). The pressure
226 derivatives C_{11}' and C_{12}' of franklinite and gahnite are similar, however, C_{44}' of franklinite
227 approaches zero above ~ 7 GPa, more similar to magnetite than gahnite (Reichmann and
228 Jacobsen 2004). In **Figure 6**, the variation of C_{44} with pressure for franklinite, gahnite,
229 magnetite, and spinel (MgAl_2O_4) are shown. The slopes of C_{44} of MgAl_2O_4 and ZnAl_2O_4 and
230 of ZnFe_2O_4 and Fe_3O_4 , respectively, are similar. The fact that dC_{44}/dP (franklinite) $\approx dC_{44}/dP$
231 (magnetite) $< dC_{44}/dP$ (gahnite) is consistent with the observation that dC_{44}/dP decreases with
232 increasing Fe- or transition metal content (Reichmann and Jacobsen 2006). A similar
233 relationship of dC_{44}/dP with Fe-substitution was observed in the (Mg,Fe)O solid solution by
234 Jacobsen et al. (2004), where compositions with less than $\sim 50\%$ Fe had positive dC_{44}/dP ,
235 mid-range compositions $dC_{44}/dP \sim 0$, and (Mg,Fe)O containing $\sim 80\%$ Fe and FeO show
236 negative dC_{44}/dP .

237 In **Table 4**, we summarize the elastic properties of a wide compositional range of
238 spinels. From **Table 4** it is apparent that the bulk moduli of spinels vary less with composition
239 than does the shear modulus. In general, the shear modulus decreases with increasing
240 transition metal occupancy of the A and B sites, with the exceptions of ringwoodite and
241 chromite. Ringwoodite (γ -phase) is the only silicate spinel in our comparison. Chromite has a
242 surprisingly high shear modulus considering that transition elements occupy both cation sites.

243 Liebermann (1970) showed that substituting Mg or Al in olivine and spinels with
244 transition metals decreases v_p and v_s linearly. In **Figure 7**, the longitudinal and transversal
245 sound wave velocities of various spinels from **Table 4** are plotted as a function of density.
246 Both v_p and v_s decrease linearly with increasing density for spinels not containing transition
247 metals, or only one transition metal on the A or B site (with the exception of Ni_2SiO_4). The
248 solid line in **Figure 7** represents a linear fit to the v_p and v_s data for spinels hosting no or only
249 one transition metal on the A or B site. The slopes are $dv_p/d\rho = -0.9 \text{ (ms}^{-1}/\text{kgm}^{-3}\text{)}$ and $dv_s/d\rho$
250 $= -0.7 \text{ (ms}^{-1}/\text{kgm}^{-3}\text{)}$, respectively. Spinel with two transition metals on the A and B sites
251 show a distinct deviation from this trend (with the exception of FeCr_2O_4). In **Figure 8** the bulk
252 sound velocity $V_\Phi = (K/\rho)^{1/2}$ is plotted versus density, ρ . The bulk sound velocity decreases
253 with higher density, which is in accordance with Birch's law (Birch 1952; Birch 1961) for
254 isostructural minerals. However, it is also obvious that spinel structured minerals with two
255 transition metals deviate from the general trend. The large deviation of spinels with two
256 transition metals (on both A and B sites) from the linear trends in **Figure 7 and 8** suggest that
257 non-ionic bonding forces begin to dominate elastic behaviour for those spinels.

258 The Cauchy relation, $0.5(C_{12} - C_{44}) = P$, is fulfilled when the solid exhibits Coulomb
259 interaction between the ions. For stable ionic structures, the value of $0.5(C_{12} - C_{44})$ should be
260 ~ 0 at room pressure. Spinel with the highest Cauchy number (i.e. those which violate the
261 most this relation) are franklinite and magnetite (**Table 4**), a sign that covalent bonding
262 increases at the expense of the Coulomb interaction when transition metals occupy both the

263 tetrahedral and octahedral sites. A peculiar feature of covalent bonding is the strong
264 directional dependence compared to Coulomb interaction. This asymmetry in the bonding
265 force may be one of the reasons for decreasing C_{44} of Fe-bearing spinel franklinite and
266 magnetite as well as for the deviation from Birch's law (**Figure 8**).

267 Another effect of the complex interatomic bonding forces for the Mn-Zn ferrites is
268 depicted in **Figure 5**. The highly non linear decrease of the single crystal elastic moduli C_{ij} for
269 the Mn-Zn ferrites with increasing Mn - content also suggest complex interatomic bonding
270 forces with mixed valence transition metal spinels. Evaluating the bonding properties of
271 transition metals is crucial to understand the elastic behavior of spinel-type structures with
272 application to a wide range of problems both in geophysical and engineering sciences.

273

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279 **References**

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- 281 Allan, D.R., Miletich, R., and Angel, R.J. (1996) A diamond-anvil cell for single crystal X-
282 Ray diffraction studies to pressures in excess of 10 GPa. *Review of Scientific*
283 *Instruments*, 67, 840-842.
- 284 Andreozzi, G.B., Lucchesi, S., Skogby, H., and Della Giusta, A. (2001) Compositional
285 dependence of cation distribution in some synthetic $(\text{Mg,Zn})(\text{Al,Fe}^{3+})_2\text{O}_4$ spinels.
286 *European Journal of Mineralogy*, 13, 391-402.
- 287 Angel, R.J. (2000) Equations of state. In R.M. Hazen and R.T. Downs, Eds. *High-*
288 *Temperature and High-Pressure Crystal Chemistry. Reviews in Mineralogy and*
289 *Geochemistry*. 41, 35-59.
- 290 Angel, R.J., Downs, R.T., and Finger, L.W. (2000) High-temperature high-pressure
291 diffractometry. In R.M. Hazen and R.T. Downs, Eds. *High-Temperature and High-*
292 *Pressure Crystal Chemistry. Reviews in Mineralogy and Geochemistry*, 41, 559-597.
- 293 Angel, R.J. and Finger, L.W. (2011) SINGLE: a program to control single-crystal
294 diffractometers. *Journal of Applied Crystallography*, 44, 247-251.
- 295 Antao, S.M., Hassan, I., Crichton, W.A., and Parise, J.B. (2005) Effects of high pressure and
296 high temperature on cation ordering in magnesioferrite, MgFe_2O_4 , using in situ

- 297 synchrotron X-ray powder diffraction up to 1430 K and 6 GPa. American
298 Mineralogist, 90, 1500-1505.
- 299 Bass, J.D., Weidner, D.J., Hamaya, N., Ozima, M., and Akimoto, S. (1984) Elasticity of the
300 olivine and spinel polymorphs of Ni_2SiO_4 . Physics and Chemistry of Minerals, 10,
301 261-272.
- 302 Birch, F. (1952) Elasticity and constitution of the Earth's interior. Journal of Geophysical
303 Research, 57, 227.
- 304 Birch, F. (1961) The velocity of compressional waves in rocks to 10 kilobars. Journal of
305 Geophysical Research, 66, 2199-2224.
- 306 Brugger, K.D. (1965) Pure modes for elastic waves in crystals. Journal of Applied Physics,
307 36, 759-768.
- 308 Butler, S.R. and Buessem, W.R. (1962) Variations of the lattice constant and line width of X-
309 ray reflections from manganese ferrite. Journal of Physics and Chemistry of Solids,
310 23, 1661-1669.
- 311 Carpenter, M.A. and Salje, E.K.H. (1998) Elastic anomalies in minerals due to structural
312 phase transitions. European Journal of Mineralogy, 10, 693-812.
- 313 Carvalho, A.V. and Sclar, C.B. (1988) Experimental determination of the ZnFe_2O_4 - ZnAl_2O_4
314 miscibility gap with application to franklinite-gahnite exsolution intergrowths from
315 the Sterling Hill zinc deposit, New Jersey. Economic Geology and the Bulletin of the
316 Society of Economic Geologists, 83, 1447-1452.
- 317 Dubrovinsky, L.S., Dubrovinskaia, N.A., McCammon, C.A., Rozenberg, G.K., Ahuja, R.,
318 Osorio-Guillen, J.M., Dmitriev, V., Weber, H.P., Le Bihan, T., and Johansson, B.
319 (2003) The structure of the metallic high-pressure Fe_3O_4 polymorph: experimental and
320 theoretical study. Journal of Physics: Condensed Matter, 15, 7697-7706.
- 321 Fei, Y., Frost, D.J., Mao, H.-K., Prewitt, C.T., and Häusermann, D. (1999) In situ structure
322 determination of the high-pressure phase of Fe_3O_4 . American Mineralogist, 84, 203-
323 206.
- 324 Frondel, C. and Baum, J.L. (1974) Structure and mineralogy of the Franklin zinc-iron-
325 manganese deposit, New Jersey. Economic Geology and the Bulletin of the Society of
326 Economic Geologists, 69, 157-180.
- 327 Funamori, N., Jeanloz, R., Nguyen, J.H., Kavner, A., and Caldwell, W. A. (1998) High-
328 pressure transformations in MgAl_2O_4 . Journal of Geophysical Research, 103(B9),
329 20813-20818.
- 330 Haavik, C., Stolen, S., Fjellag, H., Hanfland, M., and Häusermann, D. (2000) Equation of
331 state of magnetite and its high-pressure modification: Thermodynamics of the FeO-
332 system at high pressure. American Mineralogist, 85, 514-523.
- 333 Hashin, Z. and Shtrikman, S. (1962) On some variational principles in anisotropic and
334 nonhomogeneous elasticity. Journal of the Mechanics and Physics of Solids, 10, 335-
335 342.
- 336 Hazen, R.M. (1993) Comparative Compressibilities of Silicate Spinel: Anomalous Behaviour
337 of $(\text{Mg,Fe})_2\text{SiO}_4$. Science, 259, 206-209.
- 338 Hearmon, R.F.S. (1984) The elastic constants of crystals and other anisotropic materials. In
339 K.H.Hellwege, and A.M.Hellwege, Eds. Landolt-Börnstein Tables, III/18, p. 559.
340 Springer Verlag, Berlin.
- 341 Hill, R. (1952) The elastic behaviour of a crystalline aggregate. Proceedings of the Physical
342 Society of London A, 65, 349-354.
- 343 Irifune, T., Naka, H., Sanehira, T., Inoue, T., and Funakoshi, K. (2002) In situ X-ray
344 observations of phase transitions in MgAl_2O_4 spinel to 40 GPa using multianvil
345 apparatus with sintered diamond anvils. Physics and Chemistry of Minerals, 29, 645-
346 654.

- 347 Jackson, J.M., Sinogeikin, S.V., and Bass, J.D. (2000) Sound velocities and elastic properties
348 of γ - Mg_2SiO_4 to 873 K by Brillouin spectroscopy. *American Mineralogist*, 85, 296-
349 303.
- 350 Jacobsen, S.D., Reichmann, H.J., Spetzler, H.A., Mackwell, S.J., Smyth, J.R., Angel, R.A.,
351 and McCammon, C.A. (2002) Structure and elasticity of single-crystal (Mg,Fe)O and
352 a new method of generating shear waves for gigahertz ultrasonic interferometry.
353 *Journal of Geophysical Research*, 107(B2), 2037.
- 354 Jacobsen, S.D., Spetzler, H.A., Reichmann, H.J. and Smyth, J.R. (2004) Shear waves in the
355 diamond-anvil cell reveal pressure - induced instability in (Mg,Fe)O. *Proceedings of*
356 *the National Academy of Science*, 101, 5967-5871.
- 357 King, H.E. and Finger, L.W. (1979) Diffracted beam crystal centring and its application to
358 high-pressure crystallography. *Journal of Applied Crystallography*, 12, 374-378.
- 359 Levy, D., Diella, V., Pavese, A., Dapiaggi, M., and Sani, A. (2005) *P-V* equation of State,
360 thermal expansion, and *P-T* stability of synthetic (ZnCr_2O_4 spinel). *American*
361 *Mineralogist*, 90(7), 1157-1162.
- 362 Levy, D., Pavese, A., and Hanfland, M. (2000) Phase transition of synthetic zinc ferrite spinel
363 (ZnFe_2O_4) at high pressure, from synchrotron X-ray powder diffraction. *Physics and*
364 *Chemistry of Minerals*, 27, 638-644.
- 365 Levy, D., Pavese, A., Sani, A., and Pischedda, V. (2001) Structure and compressibility of
366 synthetic ZnAl_2O_4 (gahnite) under high-pressure conditions from synchrotron X-ray
367 powder diffraction. *Physics and Chemistry of Minerals*, 28, 612-618.
- 368 Li, Z. and Fisher, E.S. (1990) Single crystal elastic constants of zinc ferrite (ZnFe_2O_4).
369 *Journal of Material Science Letters*, 9, 759-760.
- 370 Li, Z., Fisher, E.S., Liu, J.Z., and Nevitt, M.V. (1991) Single crystal elastic constants of Co-
371 Al and Co-Fe spinels. *Journal of Material Science*, 26, 2621-2624.
- 372 Liebermann, R.C. (1970) Velocity-density systematics for the olivine and spinel phases of
373 Mg_2SiO_4 - Fe_2SiO_4 . *Journal of Geophysical Research*, 75(20), 4029-4034.
- 374 Mao, H.K., Xu, J., and Bell, P.M. (1986) Calibration of the ruby pressure gauge to 800 Kbar
375 under quasi hydrostatic conditions. *Journal of Geophysical Research*, 91(B5), 4673-
376 4676.
- 377 Nichols, G.T., Berry, R.F., and Green, D.H. (1992) Internally consistent gahnitic spinel -
378 cordierite-garnet equilibria in the FMASHZn system: geothermobarometry and
379 applications. *Contributions to Mineralogy and Petrology*, 111, 362-377.
- 380 Pasternak, M.P., Nasu, S., Wada, K., and Endo, S. (1994) High-pressure phase of magnetite.
381 *Physical Review B*, 50, 6446-6449.
- 382 Pavese, A., Levy, D., and Hoser, A. (2000) Cation distribution in zinc ferrite ($\text{Zn}_{0.97}\text{Fe}_{2.02}\text{O}_4$)
383 from in situ high-temperature neutron diffraction. *American Mineralogist*, 85, 1497-
384 1502.
- 385 Reichmann, H.J., Angel, R.A., Spetzler, H.A., and Bassett, W.A. (1998) Ultrasonic
386 interferometry and X-ray measurements on MgO in a diamond anvil cell. *American*
387 *Mineralogist*, 83, 1357-1360.
- 388 Reichmann, H.J. and Jacobsen, S.D. (2004) High-pressure elasticity of a natural magnetite
389 crystal. *American Mineralogist*, 89, 1061-1066.
- 390 Reichmann, H.J. and Jacobsen, S.D. (2006) Sound wave velocities and elastic constants of
391 ZnAl_2O_4 spinel and implications for spinel-elasticity systematics. *American*
392 *Mineralogist*, 91, 1049-1054.
- 393 Sakurai, J. (1964) Ultrasonic propagation in nickel and Mn-ferrite at high magnetic fields.
394 *Journal of the Physical Society of Japan*, 19(3), 311-317.
- 395 Sinogeikin, S.V., Bass, J.D., and Katsura, T. (2003) Single-crystal elasticity of ringwoodite to
396 high pressures and high temperatures: implications for 520km seismic discontinuity.
397 *Physics of the Earth and Planetary Interiors*, 136, 41-66.

398 Spetzler, H.A., Chen, G., Whitehead, S., and Getting, I.C. (1993) A new ultrasonic
399 interferometer for the determination of equation of state parameters of sub-millimeter
400 single crystals. *Pure and Applied Geophysics*, 141, 341-377.
401 Sugimoto, M. (1999) The past, present, and future of ferrites. *Journal of the American*
402 *Ceramic Society*, 82, 269-280.
403 Wang, H. and Simmons, G. (1972) Elasticity of some mantle crystal structures 1. Pleonaste
404 and hercynite spinel. *Journal of Geophysical Research*, 77, 4379 -4392.
405 Yoneda, A. (1990) Pressure derivatives of elastic constants of single crystal MgO and
406 MgAl₂O₄. *Journal of Physics of the Earth*, 38, 19-55.
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408

409 **Figure Captions**

410 **Figure 1.** (A) Variation of the unit-cell volume of franklinite with pressure (this study),
411 normalized to the room-pressure experimental volume (V_0). The solid line represents a BMII-
412 EoS fit to the data. The EoS curves of magnetite (Reichmann and Jacobsen 2004, dotted line)
413 and ZnAl₂O₄ (Reichmann and Jacobsen 2006, dashed line) are also shown for comparison.
414 (B) Normalized pressure F_E versus Eulerian strain f_E for franklinite compression, where $f_E =$
415 $((V_0/V)^{2/3} - 1)/2$ and $F_E = P/(3f_E(1+2f_E)^{5/2})$

416
417 **Figure 2.** (A) S-wave acoustic interferogram measured at 9.8 GPa by interfering the
418 diamond-culet echo with the sample echo (solid line). The dashed line shows variation of the
419 diamond-culet echo with frequency, used to remove system interference in the demodulated
420 spectrum, shown in (B). The travel time (C) is determined from the frequency spacing (Δf)
421 between maxima and minima for the best fit of m , the number of integer acoustic wavelengths
422 in the round-trip distance for the first frequency maximum. In (C), the result of choosing
423 adjacent values of m is shown.

424

425 **Figure 3.** Variation of the elastic tensor C_{ij} of franklinite with pressure.

426

427 **Figure 4.** Variation of the shear modulus G and adiabatic bulk modulus K_S of franklinite with
428 pressure. The shear modulus is invariant with pressure within the uncertainty of the
429 measurements.

430

431 **Figure 5.** Trends of various elastic moduli with composition between end-members $ZnFe_2O_4$
432 (Li and Fisher 1990) and $MnFe_2O_4$ (Sakurai 1964). The dashed lines are linear extrapolations
433 between the two end-member points. The mid-range franklinite (this study) shows linear
434 trends with composition for the bulk modulus and C_{12} elastic modulus, whereas other elastic
435 moduli display non-linear behavior. C_{44} of the mid-range franklinite (this study) is lower than
436 either end-member, suggesting that there is an intermediate composition with minimum C_{44} .

437

438 **Figure 6.** Comparison of C_{44} between franklinite (this study), magnetite (Reichmann and
439 Jacobsen 2004), gahnite (Reichmann and Jacobsen 2006), and spinel (Yoneda 1990).
440 Minerals with transition elements on their tetrahedral and octahedral sites (franklinite and
441 magnetite) exhibit invariant or slightly negative C_{44} dependence with pressure.

442

443 **Figure 7.** The compressional (circles) and longitudinal (squares) sound wave velocities of
444 spinels as a function of density (Birch's law). References are given in Table 4. Spinel
445 containing transition metals on both the A and B sites deviate from the trend of spinels
446 containing no transition metals, or only transition metals on the A or B sites.

447

448 **Figure 8.** The bulk sound velocity $V_\Phi = (K/\rho)^{1/2}$ versus density ρ . According to Birch's law V_Φ
449 decreases (linearly) with increasing ρ for isostructural minerals. Spinel with two transition
450 metals deviate significantly from the trend of those spinels with one or no transition metals.

451

452

453 **Table 1** Weight percent oxides from ten different microprobe analysis spots on the franklinite
454 sample.

455

Measurement	1	2	3	4	5	6	7	8	9	10
Oxide										
MgO	0.60	0.73	0.77	0.56	0.63	0.69	0.74	0.84	0.84	0.88
Al ₂ O ₃	1.75	0.95	1.03	1.95	1.74	1.51	1.16	0.79	0.90	0.44
SiO ₂	0.04	0.01	0.01	0.01	0.00	0.00	0.01	0.00	0.00	0.00
CaO	0.00	0.00	0.00	0.00	0.00	0.01	0.02	0.00	0.01	0.02
TiO ₂	0.18	0.12	0.10	0.10	0.17	0.16	0.12	0.10	0.12	0.07
Cr ₂ O ₃	0.00	0.00	0.02	0.02	0.01	0.01	0.00	0.00	0.01	0.00
MnO	12.34	10.69	10.60	13	12.52	12.15	11.34	10.45	10.66	9.72
FeO	67.85	75.92	74.91	64.96	68.06	68.86	71.36	77.12	75.52	80.62
ZnO	13.34	6.42	8.18	14.60	12.91	12.06	10.65	6.70	6.70	3.60
Total (%)	96.09	94.84	95.62	95.33	96.03	95.44	95.39	96.02	94.40	95.36

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461 **Table 2.** Compression data for franklinite from single-crystal X-ray diffraction. Uncertainties
462 are given in parentheses.

P (GPa)*	a (Å)	V (Å ³)
0.0001	8.4456 (2)	602.40 (4)
0.04	8.4446 (2)	602.19 (5)
1.68	8.4178 (2)	596.48 (5)
2.11	8.4122 (2)	595.28 (5)
2.47	8.4070 (3)	594.18 (6)
3.66	8.3894 (2)	590.47 (5)
4.43	8.3783 (2)	588.12 (5)
5.40	8.3638 (3)	585.07 (5)
6.54	8.3472 (3)	581.60 (6)
7.34	8.3354 (3)	579.13 (6)
7.36	8.3350 (2)	579.05 (5)
7.78	8.3306 (2)	578.12 (5)
3.15 [§]	8.3975 (3)	592.18 (6)

463

464 *Uncertainty in pressure is 0.05 GPa.

465 [§]Measured on decompression

466

467 **3. High-pressure sound velocity and elastic moduli data for Mn-franklinite.**

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P^{\S} (GPa)	$\rho^{\S\S}$ (kg/m ³)	$v_p^{[100]}$ (km/s)	C_{11} (GPa)	K_S^* (GPa)	G^* (GPa)	P (GPa)	ρ (kg/m ³)	$v_p^{[111]}$ (km/s)	C_{12} (GPa)	P (GPa)	ρ (kg/m ³)	$v_s^{[100]}$ (km/s)	C_{44} (GPa)
0.0001	5038	6.96(3)	244(3)	175(3)	66(2)	0.0001	5038	7.44 [†]	142(4)	0.0001	5038	3.92 [†]	77(2)
0.72	5059	6.98(3)	247(3)	178(3)	66(2)	2.97	5122	7.56(3)	154(4)	1.5	5081	3.91(1)	78(2)
1.25	5074	7.01(3)	249(3)	180(3)	66(2)	4.34	5160	7.60(3)	159(4)	2.1	5098	3.91(1)	78(2)
1.96	5094	7.03(3)	252(3)	183(3)	66(2)	4.71	5170	7.62(3)	161(4)	2.6	5112	3.91(1)	78(2)
3.62	5140	7.12(3)	261(3)	190(3)	67(2)	5.05	5179	7.63(3)	162(4)	4.2	5156	3.90(1)	78(2)
5.23	5183	7.16(3)	266(3)	197(3)	66(2)	5.75	5197	7.66(3)	166(4)	4.8	5172	3.89(1)	78(2)
5.55	5192	7.18(3)	268(3)	198(3)	66(2)	6.40	5215	7.68(3)	168(4)	5.4	5188	3.89(1)	78(2)
5.72	5197	7.18(3)	268(3)	199(3)	66(2)	6.66	5221	7.68(3)	169(4)	6.4	5215	3.88(1)	79(2)
6.43	5215	7.22(3)	272(3)	203(3)	66(4)	6.78	5225	7.69(3)	170(4)	7.1	5233	3.87(1)	79(2)
7.38	5240	7.25(3)	276(3)	207(3)	66(2)	7.09	5232	7.71(3)	172(4)	7.8	5251	3.87(1)	79(2)
						7.31	5238	7.72(3)	173(3)	8.3	5264	3.86(1)	79(2)
										8.5	5269	3.86(1)	78(2)
										8.9	5279	3.86(1)	78(2)
										9.1	5284	3.86(1)	79(2)
										9.5	5294	3.85(1)	79(2)
										9.8	5302	3.85(1)	79(2)

469

§ The uncertainty in pressure is estimated to be 0.05 (GPa)

470

§§ The uncertainties in densities are estimated to be 10(kg/m³)

471

* Calculated at pressures corresponding to C_{11} measurements using values of C_{12} and C_{44} from fits.

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† $v_p^{[111]}$ and $v_s^{[100]}$ at $P = 0.0001$ (GPa) determined from extrapolation of high pressure data.

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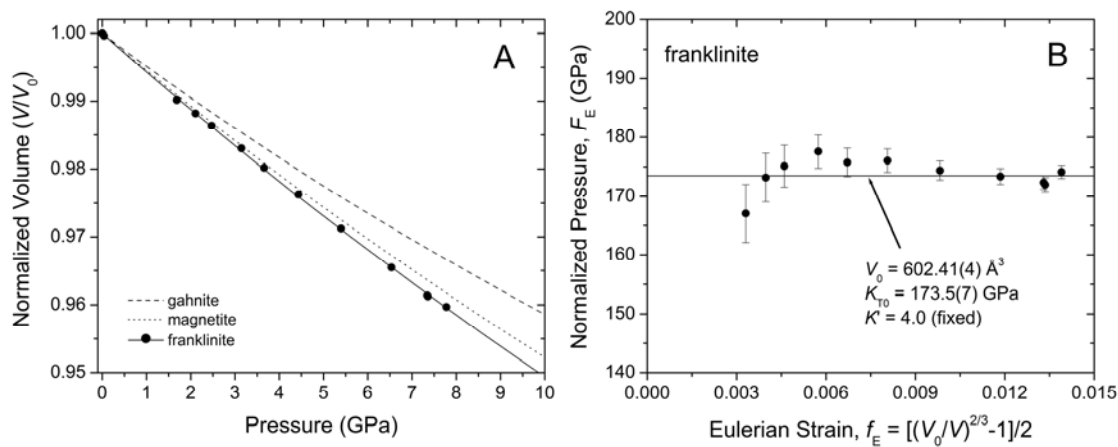
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476 **Table 4.** Elastic properties of spinel-structured minerals.

Formula (mineral name)	Ref.*	Method† <i>P</i> _{max}	ρ_0 (kg/m ³)	<i>C</i> ₁₁ (GPa) <i>C</i> ₁₁ ' <i>C</i> ₁₁ " (GPa ⁻¹)	<i>C</i> ₁₂ (GPa) <i>C</i> ₁₂ ' <i>C</i> ₁₂ " (GPa ⁻¹)	<i>C</i> ₄₄ (GPa) <i>C</i> ₄₄ ' <i>C</i> ₄₄ " (GPa ⁻¹)	$[(2C_{44}+C_{12})/$ <i>C</i> ₁₁ - 1]	Cauchy Relation (<i>C</i> ₁₂ - <i>C</i> ₄₄)/2	<i>K</i> _S (GPa) <i>K</i> _S ' <i>K</i> _S " (GPa ⁻¹)	<i>K</i> _t (GPa) <i>K</i> _t '	<i>G</i> (GPa) <i>G</i> '	<i>v</i> _p (m/s) <i>dv</i> _p / <i>dP</i> (ms ⁻¹ /GPa)	<i>v</i> _S (m/s) <i>dv</i> _S / <i>dP</i> (ms ⁻¹ /GPa)
MgAl ₂ O ₄ (spinel)	(1)	MHz 6 GPa	3578	282.9 5.59 -0.65	155.4 5.69 -0.64	154.8 1.44 -0.19	0.644	0.3	197.9 5.66 -0.65		110.0 0.61		
γ (Mg _{0.91} Fe _{0.09}) ₂ SiO ₄ (ringwoodite)	(2)	BS 16 GPa	3701	329(2) 6.2(2)	118(3) 2.8(3)	130(2) 0.8(2)	0.149(17)	-6.0	188(3) 4.1(3)		120(2) 1.3(2)	9690(20)	5680(10)
FeAl ₂ O ₄ (hercynite)	(3)	MHz 0 GPa	4280	266.0	182.5	133.5	0.690	24.5	210.3		84.5	8677	4428
(Mn _{0.40} Fe _{0.16} Zn _{0.37} Mg _{0.03}) (Fe ³⁺ _{1.94} Al _{0.08}) O ₄ (franklinite)	This study	GHz 10 GPa	5038	244(3) 4.3 (3)	142(4) 3.8 (3)	77(2) 0.29(2) -0.018(2)	0.21(25)	31	175(3) 4.3(3)	173.5(7) 4 (fixed)	66(2) 0.09(3)	7233(40) 38.3(1)	3630(25) -7.0(5)
(Zn _{0.74} Fe _{0.18} Mg _{0.08})Al ₂ O ₄ (gahnite)	(4)	GHz 9 GPa	4464	290(3) 4.48(10)	169(4) 5.0(8)	146(2) 1.47(3)	0.590(25)	11.5	209(5) 4.8(3)		104(3) 0.5(2)	8830(40) 47.2(5)	4824(40) 0.68(7)
ZnAl ₂ O ₄ (gahnite)	(5)	XRD 43 GPa	4597							201.7(9) 7.62(9)			
γ -Fe ₂ SiO ₄	(6)	XRD 5 GPa	4845							207(3) 4.8 (fixed)	104.9		
FeCr ₂ O ₄ (chromite)	(7)		5090	322	144	117	0.174	13.5	203.3		104.9		
FeFe ₂ O ₄ (magnetite)	(8)	GHz 9 GPa	5196	260.5(1.0) 5.14(13)	148.3(3.0) 5.39(10)	63.3(1.5) -0.13(4)	0.055(17)	42.5	185.7(3.0) 5.1(1)	180.0(1.0) 5.2(4)	60.3(3.0) -0.1(1)	7157(30) 47.4(4)	3407(20) -12.5(3)
ZnFe ₂ O ₄ (franklinite)	(9)	XRD 37 GPa	5319							166(3) 9.3(6)			
γ -Ni ₂ SiO ₄	(6)	XRD 5 GPa	5339							233(2) 4.8 (fixed)			
ZnCr ₂ O ₄ (zincchromite)	(10)	XRD 21 GPa	5366							183(4) 7.9(6)			
MgFe ₂ O ₄ (magnesioferrite)	(11)	MHz 0.3GPa	4554							165.6(6)	76.2(2)	7698(11)	4111(5)
Ni ₂ SiO ₄	(12)	BS	5351	366(3)	155(3)	106(1)	0.0027	24.5	226		106	8275	4447
γ -Mg ₂ SiO ₄	(13)	BS	3559	327(3)	114(2)	131(2)	0.149	-8.5	185(3)		120(2)	9850(60)	5820(30)
CoAl ₂ O ₄	(14)	MHz 0 GPa	4416	290.5	170.3	138.6	0.540	15.8	210.4		99.1	8807	4738
CoFe ₂ O ₄	(14)	Ultrasonic 0 GPa	5304	257.1	150.0	85.3	0.247	32.3	185.7		70.8	7267	3653
MnFe ₂ O ₄	(15)	Ultrasonic 0 GPa	5000	213	135	86	0.441	24.5	161		62.6	6993	3539

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*(1) (Yoneda 1990); (2) (Sinogeikin et al. 2003); (3) (Wang and Simmons 1972); (4) (Reichmann and Jacobsen 2006); (5) (Levy et al. 2001); (6) (Hazen 1993); (7) (Hearmon 1984); (8) (Reichmann and Jacobsen 2004); (9) (Levy et al. 2000); (10) (Levy et al. 2005); (11) (Antao et al. 2005); (12) (Bass et al. 1984); (13) (Jackson et al. 2000); (14) (Li et al. 1991); (15) (Sakurai 1964).
 † MHz = MHz-ultrasonic interferometry; GHz = GHz-ultrasonic interferometry; BS = Brillouin scattering; XRD = X-ray diffraction (static compression).



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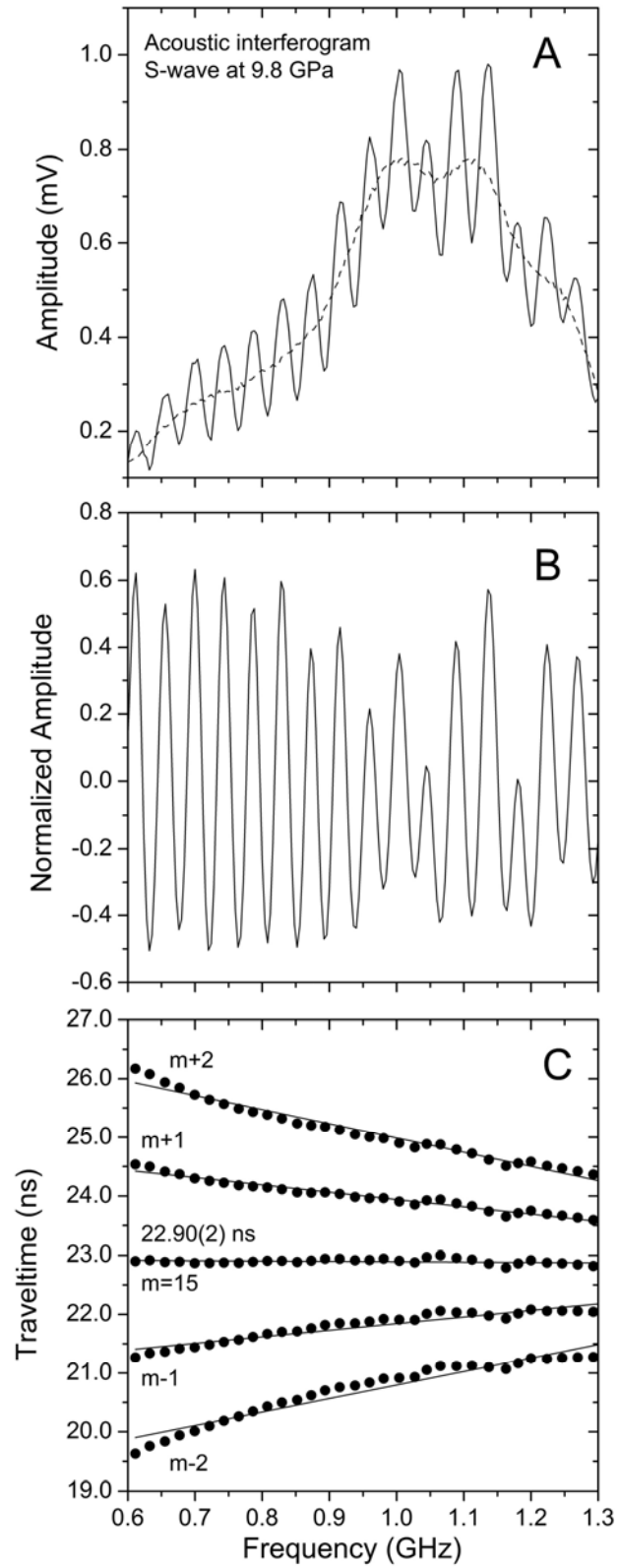
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483 **Figure 1.**

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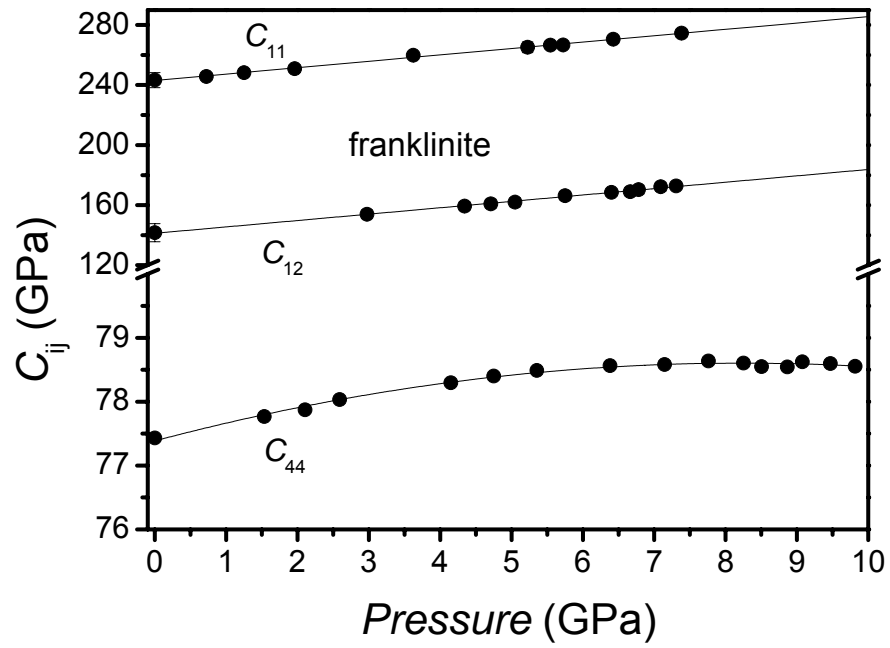


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489 **Figure 2.**

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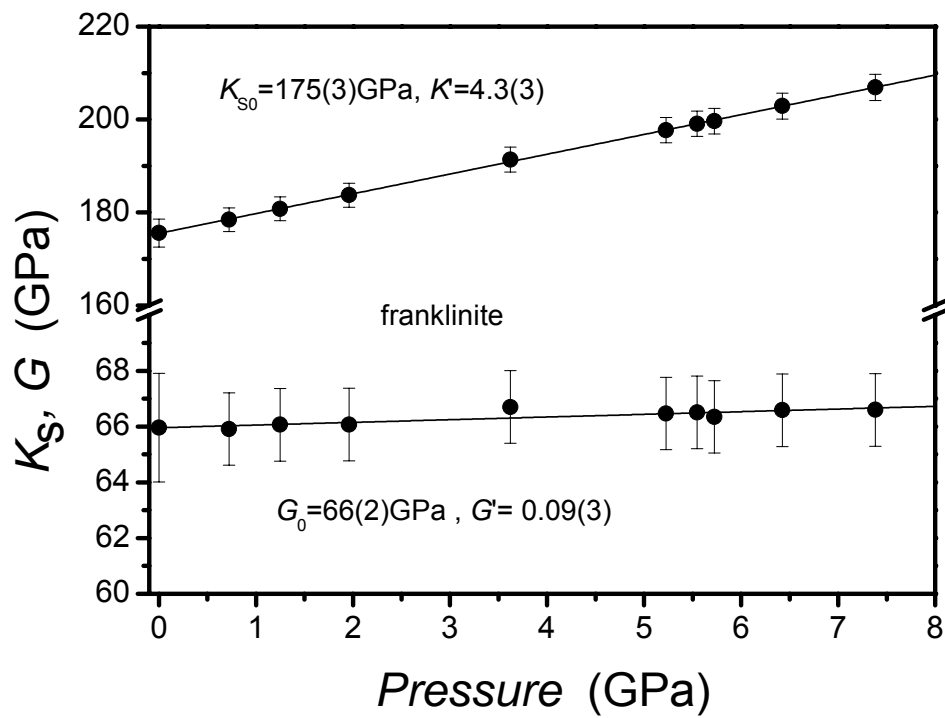


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493 **Figure 3.**

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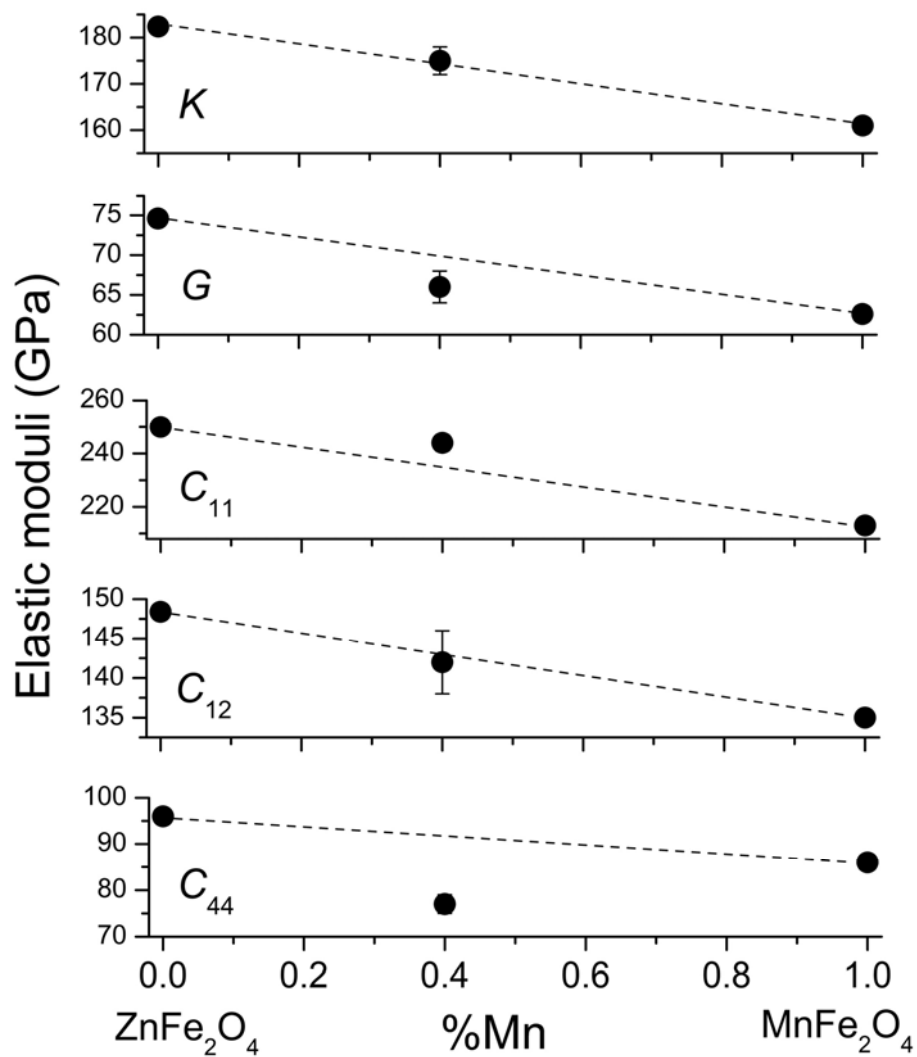


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497 **Figure 4.**

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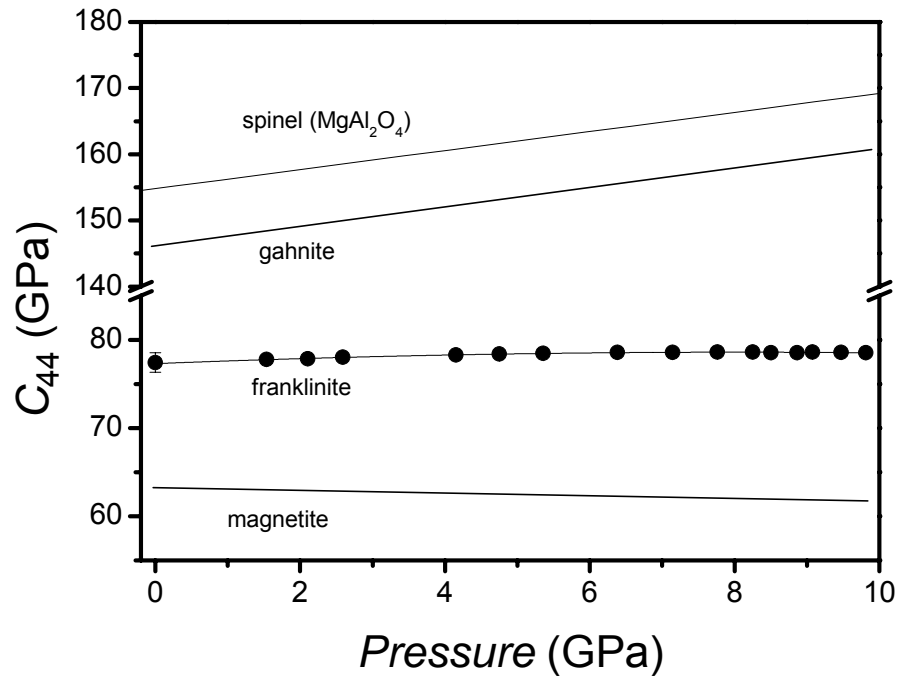
502 **Figure 5.**

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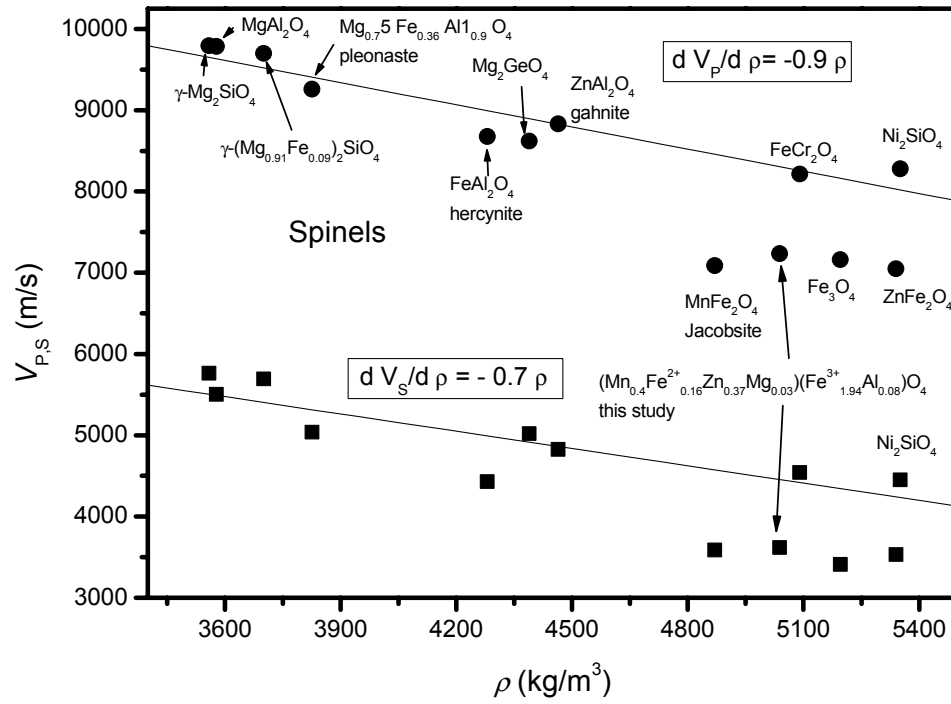
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509 **Figure 6.**

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514 **Figure 7.**

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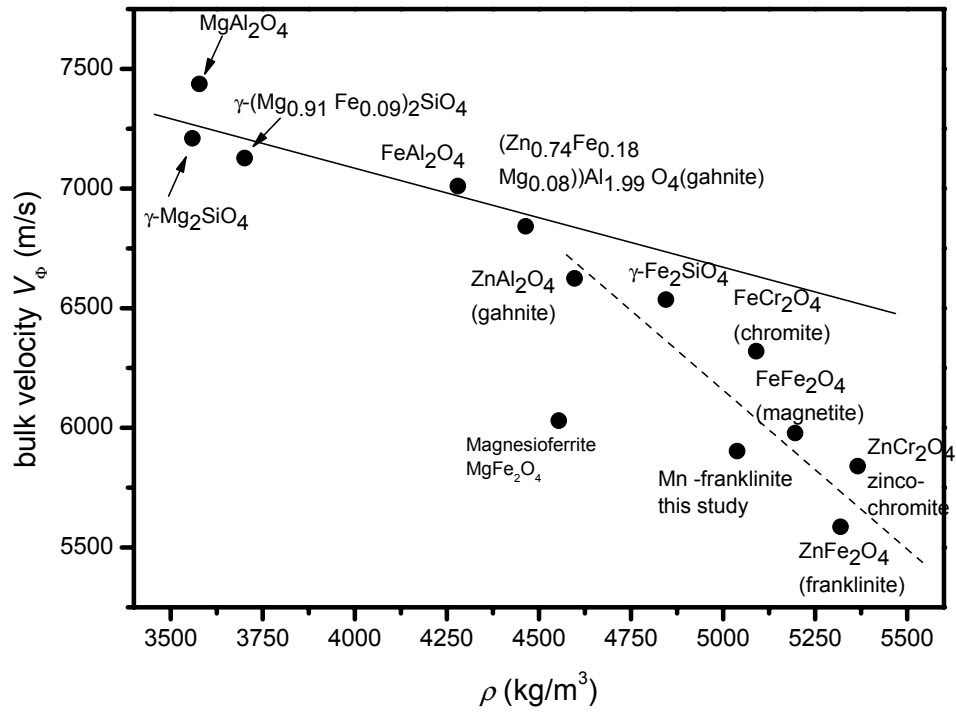
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529 **Figure 8.**

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