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*Revision 1*

**NanoSIMS Pb/Pb dating of tranquillityite in high-Ti lunar basalts:  
Implications for the chronology of high-Ti volcanism on the Moon**

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## Abstract

In this study, we carried out Pb/Pb dating of tranquillityite in high-Ti mare basalts 10044, 75055 and 74255, using a Cameca NanoSIMS 50 at a spatial resolution of  $\sim 3 \mu\text{m}$ . The analyses yielded  $^{207}\text{Pb}/^{206}\text{Pb}$  dates of  $3722 \pm 11 \text{ Ma}$  for sample 10044,  $3772 \pm 9 \text{ Ma}$  for sample 75055 and  $3739 \pm 10 \text{ Ma}$  for sample 74255, at 95% confidence level. These dates are consistent with previously determined crystallization and emplacement ages of these samples using different radiogenic systems. These high-precision ages allow refinement of the timing of some of the high-Ti basaltic volcanism on the Moon. Crystallization ages of three different high-Ti basalt units, integrating these new Pb/Pb ages with previous Rb-Sr and Sm-Nd age determinations, are consistent with previous estimates but associated with uncertainties 3 to 5 times lower. In addition, the data obtained in this study confirm that tranquillityite contains very low amounts of initial common Pb and has a high Pb ionization efficiency, making it an excellent candidate for Pb/Pb dating by ion microprobe. The higher spatial resolution afforded by NanoSIMS 50 and the recent discovery of tranquillityite in several terrestrial mafic rocks opens up a new area of research allowing an independent and rapid age dating of basaltic rocks in polished sections.

**Keywords:** Ion probe; High-Ti mare basalts; NanoSIMS 50; Pb/Pb dating; Tranquillityite

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## Introduction

52 U-Th-Pb geochronology by ion microprobe was first carried out three decades ago by  
53 Andersen and Hinthorne (1972, 1973) to date phosphates and Zr-rich minerals from lunar  
54 samples. These pioneering studies demonstrated the potential of Secondary Ion Mass  
55 Spectrometry (SIMS) for *in-situ* geochronology, but the low mass resolution of the instrument  
56 used introduced some major limitations, notably related to isobaric interferences on the Pb  
57 isotopes. The development of the Sensitive High Resolution Ion Micro Probe (SHRIMP)  
58 permitted an efficient separation of isobaric interferences on the Pb isotopes (see the review  
59 of Compston and Clement 2006), which allowed precise U-Pb age dating of lunar zircons  
60 (Compston et al. 1984). Since these initial studies, *in-situ* U-Pb age dating of lunar samples  
61 has often been focused on zircons (e.g., Nemchin et al. 2012, and references therein), except  
62 for the SHRIMP U-Pb age dating of phosphates from lunar meteorites (Anand et al. 2003;  
63 Terada et al. 2005; 2007a, 2007b). In lunar samples, zircon occurs in several rock types such  
64 as anorthosite, troctolite, gabbro-norite, quartz-monzodiorite, granite and felsite (e.g.,  
65 Nemchin et al., 2012), but is absent in basaltic rocks, in which Zr, as well as, U and Th, are  
66 commonly hosted by baddeleyite, zirconolite and tranquillityite.

67 Ion microprobe studies in terrestrial and extraterrestrial samples have shown that *in-situ*  
68 analysis of baddeleyite (Wingate and Compston 2000; Wingate et al. 1998; Chamberlain et al.  
69 2010; Li et al. 2010; Schmitt et al. 2010; Rodionov et al. 2012), zirconolite (Rasmussen and  
70 Fletcher 2004; Rasmussen et al. 2008; Zhang et al. 2010; Wang et al. 2012), and  
71 tranquillityite (Rasmussen et al. 2008; 2012) can yield precise and accurate U-Pb and/or  
72 Pb/Pb ages. Particularly, the suitability of the NanoSIMS 50 for dating purposes has been  
73 demonstrated on zircon (Takahata et al. 2008; Yang et al. 2012), monazite (Sano et al., 2006),  
74 baddeleyite (Yang et al. 2012) and zirconolite (Stern et al. 2005). U-Pb age dating of these  
75 minerals are however still hampered by crystal orientation effects for baddeleyite (e.g.,

76 Wingate and Compston 2000) or the lack of reference zirconolite and tranquillityite for which  
77 U and Pb isotopic compositions have been determined by an independent method. Among  
78 these minerals, tranquillityite might be most suitable for U-Pb and Pb/Pb geochronology on  
79 account of 1) its very low initial common Pb and 2) its high Pb ionization efficiency; ~ 5  
80 times that for zirconolite and an order of magnitude higher than zircon (Rasmussen et al.  
81 2008). Tranquillityite is commonly found in mare basalts from all the Apollo landing sites  
82 and has also been described from KREEP basalts (see Rasmussen et al. 2008 and references  
83 therein). Moreover, it has recently been identified in dolerite rocks from six localities in  
84 Western Australia (Rasmussen et al. 2012), suggesting that it could be more widely present in  
85 terrestrial mafic rocks than previously recognized. However, tranquillityite commonly occurs  
86 as relatively thin grains (< 5  $\mu\text{m}$  width) in lunar basalts, posing serious limitations for  
87 application of traditional geochronological techniques.

88 In this contribution, we present the results of Pb/Pb analyses carried out using the  
89 Cameca NanoSIMS 50 ion microprobe on tranquillityite grains occurring in high-Ti lunar  
90 basalts collected during Apollo 11 (10044) and Apollo 17 (75055 and 74255) missions. The  
91 Pb/Pb dates acquired in this study have been integrated into a review of previous age  
92 determinations for high-Ti mare basalts, allowing for a detailed discussion of the chronology  
93 of high-Ti mare volcanism at the Apollo 11 and 17 landing sites.

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### **Samples**

96 We have carried out Pb/Pb age dating of discrete tranquillityite grains from Apollo 11  
97 sample 10044 and Apollo 17 samples 75055 and 74255. Sample 10044 is a coarse-grained  
98 porphyritic ilmenite basalt, with high-Ti and low-K contents (Schmitt et al. 1970; Beatty and  
99 Albee 1978a, 1978b). Rb-Sr and Ar/Ar age dating indicate a crystallization age of ~ 3.65 -  
100 3.70 Ga (Papanastassiou et al. 1970; Turner 1970; Guggisberg et al. 1979), although Davis et

101 al. (1971) reported an older Ar/Ar date of about 4.0 Ga. Sample 75055 is a fine- to medium-  
102 grained intergranular to subophitic ilmenite basalt (e.g., Dymek et al. 1975), with high-Ti and  
103 low-K contents (e.g., Rhodes et al. 1976). Its crystallization age has been determined using  
104 the Rb-Sr and Ar/Ar techniques at ~ 3.75 - 3.80 Ga (Huneke et al. 1973; Tatsumoto et al.  
105 1973; Tera et al. 1974). Sample 74255 is a medium- to coarse-grained vesicular porphyritic  
106 basalt with abundant ilmenite (e.g., Dymek et al. 1975), with also high-Ti and low-K contents  
107 (e.g., Rhodes et al. 1976). Rb-Sr data gave an imprecise crystallization age of ~ 3.70 - 3.80 Ga  
108 (Murthy and Coscio 1976; Nyquist et al. 1976).

109 In samples 10044 and 75055, tranquillityite, with an ideal formula  $[\text{Fe}_8^{2+}$   
110  $(\text{Zr},\text{Y})_2\text{Ti}_3\text{Si}_3\text{O}_{24}]$ , commonly occurs as intergrowths of laths 20 to 50  $\mu\text{m}$  long that  
111 crystallized in late-stage mesostasis areas and in voids in plagioclase. Tranquillityite crystals  
112 were relatively smaller in size in sample 74255. Tranquillityite composition is similar in  
113 samples 10044 and 75055, roughly comprising of 14 - 15 wt.%  $\text{SiO}_2$ , 43 - 44 wt.% FeO and ~  
114 16 wt.%  $\text{ZrO}_2$ , with  $\text{TiO}_2$  content slightly higher in 75055 than in 10044 (Table 1).  
115 Tranquillityite in 74255 differs in terms of having higher  $\text{SiO}_2$  and lower FeO contents  
116 compared to those in samples 10044 and 75055 (Table 1). These chemical compositions are  
117 consistent with those reported from other high-Ti mare basalts, but differ considerably from  
118 terrestrial tranquillityite (Table 1 and Fig. 1).

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### **Analytical setup**

121 Polished thin sections 10044,645, 75055,49 and 74255,48 were obtained from the Lunar  
122 Sample Curation facility at the Johnson Space Centre (NASA). Back-scattered electron (BSE)  
123 images and X-ray maps of the entire polished sections were acquired using a Quanta 3D dual  
124 beam Focused Ion Beam (FIB) Secondary Electron Microscope (SEM) at the Open  
125 University, UK, fitted with an Oxford Instruments INCA energy dispersive X-ray detector.

126 For this purpose, sections were carbon coated and examined with an electron beam current of  
127 0.6 nA at an accelerating voltage of 20 kV. We used the whole-section X-ray maps of Zr to  
128 locate Zr-bearing minerals, and then acquired high magnification BSE images of areas  
129 containing these Zr-bearing minerals. X-ray spectra were then acquired for 60 seconds to get  
130 quantitative chemical compositions of tranquillityite grains (see Tables S2 and S3 in the  
131 Supplementary Material).

132 The Pb/Pb analyses were carried out using the Cameca NanoSIMS 50 installed at the  
133 University of Rennes 1, France. The NanoSIMS is a scanning ion microscope equipped with a  
134 double-focusing mass analyzer allowing high mass and spatial resolution. The NanoSIMS  
135 differs from other ion microscopes by its co-axial design that comprises a normal primary ion  
136 beam and a co-axial secondary ion extraction, optimizing simultaneously the objective lens  
137 performance and collection of secondary ions. The NanoSIMS design also reduces the  
138 distance between the sample and the objective, allowing for a smaller spot size for a given  
139 probe current compared to other ion microscopes. General descriptions of the instrument and  
140 of the tuning specific to Pb/Pb analyses can be found in Slodzian et al. (1992) and in Stern et  
141 al. (2005), respectively.

142 In the NanoSIMS 50, the primary column integrates a duoplasmatron ion source in  
143 which the negative oxygen ions are accelerated through a potential of - 8 kV, passed through  
144 a Wien mass filter for isolation of  $^{16}\text{O}^-$  ions, which are then focused to the target held at a  
145 positive potential of + 8 kV, resulting in a total impact energy of 16 keV. For tranquillityite  
146 analyses, the  $\text{O}^-$  beam was optimized using apertures D0-1 and D1-2, resulting in a current of  
147  $\sim 120 - 140$  pA on the sample surface. Each spot size on the sample surface was  $\sim 3.5 \mu\text{m} \times$   
148  $2.5 \mu\text{m}$  (Fig. 2), corresponding to a beam density of  $\sim 15 \text{ pA} \cdot \mu\text{m}^{-2}$ , which is similar to the  
149 value reported by Stern et al. (2005). During the analytical sessions, the vacuum in the  
150 analysis chamber remained constant at  $\sim 8 \times 10^{-10}$  torr.

151 Secondary ions excavated from the sample were extracted by the co-axial immersion  
152 lens and focused onto a fixed-width analyzer entrance slit (ES). In this study, we used ES-3  
153 (30  $\mu\text{m}$ ), which cuts about 60% of the secondary ions relative to the full transmission without  
154 slits. An additional 10% of the secondary ions were lost at the aperture slit (AS-1). No energy  
155 slit was used. The net relative transmission at high resolution was therefore  $\sim 35\%$ . With this  
156 setup, we achieved a mass resolving power of  $\sim 5000$  (Cameca definition, equivalent to a  
157 mass resolution of  $\sim 3500$ ), sufficient to isolate the major  $^{178-179}\text{Hf}^{28}\text{Si}$  interferences on the  
158  $^{206}\text{Pb}$  and  $^{207}\text{Pb}$  peaks in tranquillityite. The mass spectrometer of the NanoSIMS 50 installed  
159 in Rennes comprises of 5 detectors equipped with electron multipliers (EM), allowing for 5  
160 species to be simultaneously counted under a static magnetic field. Although for Zr-bearing  
161 minerals it is possible to simultaneously measure Pb, UO and UO<sub>2</sub>, it is not possible to  
162 measure  $^{204}\text{Pb}$ ,  $^{206}\text{Pb}$  and  $^{207}\text{Pb}$  at the same time because the dispersion distance between the  
163 Pb isotopes at the focal plane is less than the minimum separation of detectors. The Pb  
164 isotopes were therefore analyzed in the magnetic peak-switching mode on the detector EM#4.  
165 Between each analysis, the magnet field settings for the Pb peak positions were precisely  
166 determined manually and the  $^{206}\text{Pb}$  peak was used to set the automatic centering procedure.

167 Before analysis, the probe was rastered over a  $6\ \mu\text{m} \times 6\ \mu\text{m}$  area for  $\sim 5$  min to remove  
168 any potential surface contamination. For analysis, the scanning mode of the probe was  
169 switched off and the magnetic field was cyclically switched to measure  $^{204}\text{Pb}$ ,  $^{206}\text{Pb}$  and  $^{207}\text{Pb}$   
170 in an up-mass sequence. Automatic centering was performed on the  $^{206}\text{Pb}$  peak at the  
171 beginning of each analysis and repeated three times during a run. Counting times were set to 4  
172 s for  $^{206}\text{Pb}$  and 10 s for  $^{204}\text{Pb}$  and  $^{207}\text{Pb}$ , with a 2 s waiting time between each isotopic  
173 measurement, resulting in a total analysis time of about 30 minutes. The EM background is  
174 set to 0 in the NanoSIMS setup, and it was not further monitored in this study. Raw data were  
175 corrected for EM dead time of 44 ns by the NanoSIMS software. The secondary ion signals

176 were processed using the NanoSIMS DataEditor software developed by Frank Gyngard  
177 (Washington University, St Louis, USA), and isotope ratios were calculated from total counts.  
178 Final date calculations were carried out using the Isoplot 3.7 add-in for Excel (Ludwig 2008).  
179 No corrections were applied for instrumental mass fractionation. Stern and Amelin (2003)  
180 discussed up to 0.2%/amu fractionation of Pb isotopes during SHRIMP analyses, which tends  
181 to lower the measured  $^{207}\text{Pb}/^{206}\text{Pb}$  ratio. On the other hand, the presence of unresolved hydride  
182  $^{206}\text{PbH}$  mixed with  $^{207}\text{Pb}$  has an opposite effect, raising the  $^{207}\text{Pb}/^{206}\text{Pb}$  ratio. Thus, mass  
183 fractionation and hydride interferences are probably cancelling effects of each other (e.g.  
184 Ireland and Williams 2003). Consistently, instrumental mass fractionation for Pb isotopes has  
185 been shown to be negligible for NanoSIMS measurements in mono-collection mode (Stern et  
186 al. 2005), even with a static probe (Yang et al. 2012).

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## Results

189 Before analysis, we acquired a 128 pixels  $\times$  128 pixels ion image of the 45  $\mu\text{m}$   $\times$  45  $\mu\text{m}$   
190 area enclosing tranquillityite #14 and the associated baddeleyite in sample 10044. Secondary  
191 species  $^{47}\text{Ti}$ ,  $^{91}\text{Zr}$ ,  $^{206}\text{Pb}$  and  $^{238}\text{U}^{16}\text{O}$  were simultaneously monitored during 5 min 33 s, with a  
192 dwell time of 20 ms/px. The data were processed offline using the L'Image software package  
193 (L.R. Nittler, Carnegie Institution, USA) to produce the maps displayed in Figure 3. In  
194 agreement with their general chemical compositions, images show that tranquillityite is richer  
195 in Ti and poorer in Zr than baddeleyite. Baddeleyite also appears richer in  $^{238}\text{U}$  than  
196 tranquillityite, whereas count rates for  $^{206}\text{Pb}$  is higher in tranquillityite than in baddeleyite.  
197 This illustrates that Pb is more efficiently ionized in tranquillityite than in baddeleyite. A high  
198 Pb ionization efficiency in tranquillityite is consistent with the observation of Rasmussen et  
199 al. (2008), who found Pb ionization efficiency for tranquillityite about 5 times that for  
200 zirconolite during their SHRIMP analyses.



201 The Pb isotopic compositions of tranquillityite in the three mare basalt samples are  
202 given in Table 2. The common Pb content of tranquillityites analyzed in this study is very  
203 low, as indicated by measured  $^{204}\text{Pb}/^{206}\text{Pb}$  ratios of lower than 0.0002 (Table 2). As a result, it  
204 precludes calculation of  $^{207}\text{Pb}/^{206}\text{Pb}$  vs.  $^{204}\text{Pb}/^{206}\text{Pb}$  isochron ages due to the low dispersion in  
205 the  $^{204}\text{Pb}/^{206}\text{Pb}$  ratios (Fig. 4). Arrows pointing to the Pb isotopic composition of two possible  
206 contaminants are depicted in the  $^{207}\text{Pb}/^{206}\text{Pb}$  vs.  $^{204}\text{Pb}/^{206}\text{Pb}$  mixing diagrams (Fig. 4). When  
207 the measured ratios are highly radiogenic, which is the case here for lunar tranquillityites, any  
208 common Pb is likely to be mainly associated with polishing compounds and/or the conductive  
209 coating (e.g., Williams 1998). The Broken Hill galena Pb composition ( $^{204}\text{Pb}/^{206}\text{Pb} = 0.0626$ ,  
210  $^{207}\text{Pb}/^{206}\text{Pb} = 0.961$ ; Sangster et al. 2000) has been used to model this laboratory  
211 contamination, which is depicted by a grey arrow in Figure 4. The second arrow points  
212 towards an estimated lunar initial Pb. For this purpose, the values of Rasmussen et al. (2008)  
213 ( $^{207}\text{Pb}/^{206}\text{Pb} = 1.43$  and  $^{204}\text{Pb}/^{206}\text{Pb} = 0.0074$ ) have been used. These authors carried out  
214 several Pb/Pb analyses in K-feldspar and other matrix minerals in the high-Ti basalt 10047,  
215 and found that such a composition, computed for a single-stage evolution from the Canyon  
216 Diablo troilite primordial isotopic composition until 3.7 Ga and a  $\mu$  value of 500, matches  
217 well their measured data. In the three samples analyzed in this study, the majority of the Pb  
218 isotope ratios could be interpreted as mixing between a  $^{204}\text{Pb}$ -free  $^{207}\text{Pb}/^{206}\text{Pb}$  ratio,  
219 corresponding to the average of the measured  $^{207}\text{Pb}/^{206}\text{Pb}$  ratios, and a very small component  
220 of Pb introduced by laboratory contamination (Fig. 4). The fraction of  $^{206}\text{Pb}$  ( $f_{206}$ ) comprising  
221 unradiogenic Pb, calculated using the measured  $^{204}\text{Pb}/^{206}\text{Pb}$  and the Broken Hill Pb  
222 composition, never exceeds 0.3% (Table 2). As a result, the dates calculated using the  
223 measured  $^{207}\text{Pb}/^{206}\text{Pb}$  ratios and those corrected for unradiogenic Pb differ by a maximum of  
224 0.2%. Uncertainties in age determinations are thus largely dominated by counting

225 uncertainties, which is the reason why we used the measured  $^{207}\text{Pb}/^{206}\text{Pb}$  ratios to calculate  
226 individual dates.

227 In Apollo 11 sample 10044, the 9 analyses carried out on 5 different grains define a  
228 weighted mean  $^{207}\text{Pb}/^{206}\text{Pb}$  date of  $3722 \pm 11$  Ma (95% confidence level), with a MSWD of  
229 1.3 (Fig. 4a). In Apollo 17 sample 75055, 12 analyses carried out on 5 different tranquillityite  
230 grains yield a weighted mean  $^{207}\text{Pb}/^{206}\text{Pb}$  date of  $3772 \pm 9$  Ma (MSWD = 1.9; 95% confidence  
231 level; Fig. 4b). In sample 74255, 7 analyses carried out on 4 grains display more scatter  
232 compared to other two samples. Excluding the two outliers (Trq4#1 and Trq18#1; Fig. 4c),  
233 the remaining 5 analyses, carried out on 3 grains, define a weighted mean  $^{207}\text{Pb}/^{206}\text{Pb}$  date of  
234  $3739 \pm 10$  Ma (MSWD = 1.9; 95% confidence level, Fig. 4c). One analysis was also carried  
235 out on a baddeleyite grain associated with tranquillityite #14 in sample 10044 (Figs. 2 and 3).  
236 The measured  $^{204}\text{Pb}/^{206}\text{Pb}$  and  $^{207}\text{Pb}/^{206}\text{Pb}$  ratios are slightly higher and lower than in  
237 tranquillityite, respectively. These ratios are also associated with higher uncertainties (Table  
238 2), due to lower count rates for both  $^{206}\text{Pb}$  and  $^{207}\text{Pb}$  in baddeleyite (about an order of  
239 magnitude lower than in tranquillityite). Nevertheless, the calculated  $^{207}\text{Pb}/^{206}\text{Pb}$  date of  $3682$   
240  $\pm 90$  Ma is consistent within errors of the weighted mean  $^{207}\text{Pb}/^{206}\text{Pb}$  date calculated from  
241 tranquillityite analyses.

242 As an external control, analyses were also carried out on a chip of the well characterized  
243 Phalaborwa baddeleyite standard (details are given in the Supplementary Material). The 15  
244 individual  $^{207}\text{Pb}/^{206}\text{Pb}$  analyses define a weighted mean  $^{207}\text{Pb}/^{206}\text{Pb}$  date of  $2065 \pm 15$  Ma  
245 (Supplementary Fig. S1 and Table S1), which is consistent with the  $^{207}\text{Pb}/^{206}\text{Pb}$  crystallization  
246 ages of  $\sim 2060$  Ma determined for Phalaborwa baddeleyite by ID-TIMS (Reischmann 1995;  
247 Heaman 2009).

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## Discussion

### 251 Accuracy of the tranquillityite Pb/Pb dates

252 Since Pb/Pb dating is not amenable to the concordance test afforded by U-Pb dating, it  
253 is imperative to ensure that the tranquillityite Pb/Pb data accurately date the crystallization  
254 ages of the parental magmas. Indeed, these small tranquillityite grains might have been  
255 affected by diffusive loss of Pb after crystallization, for example, as a result of impact  
256 processes. Moreover, the closure temperature for diffusion of Pb isotopes in tranquillityite is  
257 currently unknown. Rasmussen et al. (2008) discussed the issue of possible disturbance of the  
258 Pb isotopic systematics in Zr-rich minerals in Apollo 11 sample 10047. Based notably on the  
259 comparison between zirconolite and tranquillityite Pb/Pb ages determined in sample 10047,  
260 these authors argued that  $^{207}\text{Pb}/^{206}\text{Pb}$  systematics in these two minerals provided an accurate  
261 and precise chronometer, and that dating lunar samples using these minerals appears to be  
262 more limited by the  $^{207}\text{Pb}/^{206}\text{Pb}$  data obtained on reference materials than by data obtained on  
263 the minerals themselves. As we only dated tranquillityite, we cannot evaluate consistency  
264 between dates given by tranquillityite and by another mineral. However, the obtained Pb/Pb  
265 ages can be compared with previous age determinations carried out on the studied samples  
266 using other chronometers. In the following, all previously published Rb-Sr and Sm-Nd ages  
267 have been recalculated using the Isoplot add-in for Excel (Ludwig 2008) to ensure a thorough  
268 propagation of uncertainties. Moreover, the new decay constant for  $^{87}\text{Rb}$  of  $1.3968 \times 10^{-11} \text{ yr}^{-1}$   
269 (Rotenberg et al. 2012) has been used to recalculate Rb-Sr isochron ages.

270 Sample 10044 was first dated soon after the Apollo 11 astronauts brought it back to the  
271 Earth. Papanastassiou et al. (1970) determined a Rb-Sr isochron date of  $3694 \pm 68 \text{ Ma}$   
272 (whole-rock and mineral fractions, excluding their Pyroxene A fraction; Table 3). Turner  
273 (1970) and Guggisberg et al. (1979) used the Ar/Ar chronometer and obtained whole-rock  
274 plateau dates of  $3.73 \pm 0.05 \text{ Ga}$  and  $3.71 \pm 0.04 \text{ Ga}$ , respectively. The tranquillityite Pb/Pb

275 date of  $3722 \pm 11$  Ma obtained in this study is in good agreement with these previous Rb-Sr  
276 and Ar/Ar age determinations. Moreover, our tranquillityite Pb/Pb date of  $3722 \pm 11$  Ma is  
277 consistent with the precise  $^{207}\text{Pb}/^{206}\text{Pb}$  date of  $3710 \pm 5$  Ma determined on sample 10047 by  
278 Rasmussen et al. (2008), which is petrologically identical to sample 10044 and probably came  
279 from the same lava flow (Beaty and Albee 1978b). The Pb/Pb date of  $3772 \pm 9$  Ma  
280 determined on tranquillityite in sample 75055 is consistent with the Rb-Sr isochron date of  
281  $3752 \pm 60$  Ma determined by Tera et al. (1974) (Table 3). It is also consistent with the well-  
282 defined plateau Ar/Ar date of  $3.78 \pm 0.04$  Ga obtained on a plagioclase separate by Huneke et  
283 al. (1973). In sample 74255,  $^{207}\text{Pb}/^{206}\text{Pb}$  ratios measured in tranquillityite grains display more  
284 scatter (Fig. 4). Two previous studies determined whole-rock mineral Rb-Sr isochron dates of  
285  $3751 \pm 77$  Ma (Nyquist et al. 1976) and  $3694 \pm 140$  Ma (Murthy and Coscio 1976),  
286 respectively. Combination of these two dataset yields a Rb-Sr isochron date of  $3750 \pm 120$  Ma  
287 (Table 3). Considering the very large uncertainty associated with the date determined by  
288 Murthy and Coscio (1976), this date, apart from being consistent with our tranquillityite  
289 Pb/Pb date of  $3739 \pm 10$  Ma, is too imprecise to be useful in the present context of age  
290 comparisons. The large uncertainties associated with Rb-Sr isochron dates and the presence of  
291 two outliers in our Pb isotopic data may indicate that this sample experienced some geological  
292 disturbance.

293

#### 294 **Chronology of high-Ti volcanism at Apollo 11 and 17 landing sites**

295 Mare-basalts returned from the Apollo 11 landing site have been classified in five  
296 different types, namely A, B1, B2, B3 and D (Beaty and Albee 1978a; Beaty et al. 1979; Ma  
297 et al. 1980; Rhodes and Blanchard 1980). The crystallization ages of these mare-basalts span  
298 the range from  $\sim 3.60$  to  $3.90$  Ga (e.g., Papanastassiou et al. 1977; Guggisberg et al. 1979).  
299 Previous Rb-Sr and Sm-Nd age determinations yielded a weighted mean crystallization age of

300 3695 ± 50 Ma for mare basalts from the B1 type. The new Pb/Pb dates obtained in this study  
301 and by Rasmussen et al. (2008) provide a more precise crystallization age of 3711 ± 4 Ma for  
302 basalts of the Apollo 11 B1 type (Fig. 5 and Table 3). Published Rb-Sr and Sm-Nd dates have  
303 been used to calculate the best estimates of the crystallization ages two other Apollo 11 types:  
304 the type B2 basalts are the oldest, with a crystallization age of 3852 ± 65 Ma, whereas type A  
305 basalts are the youngest with a well-defined crystallization age of 3597 ± 20 Ma (Fig. 5).  
306 These crystallization ages compare well with those summarized in previous synthesis (e.g.,  
307 Jerde et al. 1994; Stöffler and Ryder 2001, and references therein), except for type B2 basalts  
308 that appears to be ~ 50 Ma older than the age considered by Stöffler and Ryder (2001).  
309 However, due to a relatively large uncertainty associated with the age determination of type  
310 B2 basalts, the exact timing of Apollo 11 type B2 volcanism remains uncertain. For future  
311 work, it is important to better constrain the crystallization ages of type B2 basalts as their ages  
312 have been used as an important unit for crater retention age determinations (e.g., Stöffler and  
313 Ryder 2001).

314 At the Apollo 17 landing site, basalts have been grouped into three types A, B and C  
315 based on their whole-rock chemical compositions (Rhodes et al. 1976). Neal et al. (1990)  
316 further divided type B basalts into groups B1 and B2. Sample 75055 belongs to the type A.  
317 Rb-Sr and Sm-Nd dates obtained on six samples from Apollo 17 type A basalts (Table 3)  
318 define a weighted mean date of 3769 ± 30 Ma, which is identical to the Pb/Pb date of 3772 ±  
319 9 Ma obtained on tranquillityite in sample 75055. Integrating this date with previous  
320 determinations yield a weighted mean date of 3772 ± 9 Ma (Fig. 5), which we confidently  
321 interpret as the crystallization age of the Apollo 17 type A lava flow. This suggests that  
322 Apollo 17 type A lavas might have erupted ~ 20 Ma earlier than estimated previously (e.g.  
323 Paces et al. 1991; Stöffler and Ryder 2001). Sample 74255 belongs to the rare Apollo 17 Type  
324 C basalts, sampled at station 4 near Shorty crater. The three previous Rb-Sr age

325 determinations carried out on sample 74255 and 74275 (Table 3) yielded an imprecise  
326 weighted mean date of  $3755 \pm 61$  Ma, which is within error of the Pb/Pb tranquillityite date of  
327  $3739 \pm 10$  Ma. Based on the broad consistency between Rb-Sr age determinations and the  
328 Pb/Pb tranquillityite date obtained in this study on sample 74255, we propose that the  
329 crystallization age of the Apollo 17 type C lava flow is  $3739 \text{ Ma} \pm 10 \text{ Ma}$ . Our Pb/Pb  
330 crystallization age obtained on tranquillityite in type C sample 74255 provides a better  
331 constraint on the eruption age of this type, which was previously loosely constrained at  $3750$   
332  $\pm 70$  Ma (Stöffler and Ryder 2001, and references therein). However, additional age  
333 determinations on some other Apollo 17 type C basalts are desirable to firmly establish this  
334 crystallization age. Apollo 17 types A and C basalts have different crystallization ages of  $\sim$   
335  $3772$  Ma and  $\sim 3739$  Ma, respectively. Eruption ages of these two types of lavas are important  
336 as they also serve as age constraints for crater retention ages (Stöffler and Ryder 2001, and  
337 references therein). Finally, published Rb-Sr and Sm-Nd age determinations suggest that  
338 Apollo 17 type B basalts might have been emplaced at  $3719 \pm 44$  Ma. Unfortunately, the  
339 uncertainty on this age is fairly large and new age determinations should be carried out on  
340 Apollo 17 type B basalts.

341 Figure 5 shows that Apollo 11 type B1 and Apollo 17 type B units have similar  
342 crystallization ages, suggesting that they could represent a unique and very widespread  
343 volcanic event. However, some of their whole-rock geochemical characteristics indicate that  
344 they constituted two different lava flows (Fig. 6). Nevertheless, their similarities in ages imply  
345 widespread volcanism on the Moon at the same time.

346

#### 347 **Implications for Pb/Pb age dating with the NanoSIMS**

348 The Pb/Pb dates obtained on tranquillityites from three Apollo 11 and 17 high-Ti mare  
349 basalts are consistent with previous age determinations carried out using other radiogenic

350 systems. Hence, we consider that these Pb/Pb dates are reliable and provide accurate  
351 crystallization ages. Provision of a suitable tranquillityite standard for U-Pb dating will help  
352 further confirm and refine the crystallization ages of mare-basalts containing tranquillityite.  
353 Tranquillityite, with its low initial Pb content and its high Pb ionization efficiency, would  
354 constitute an excellent reference material, allowing for the use of even smaller primary  
355 current and thus smaller spot sizes. As tranquillityite has now been identified in several  
356 terrestrial dolerite samples (Rasmussen et al. 2012), in some cases in large amounts (more  
357 than 100 individual crystals in some polished thin sections, as reported by Rasmussen et al.  
358 2012), it may be possible to consider separating these crystals from large samples and use  
359 them as reference materials. It would require ID-TIMS analyses of both the Pb and the U  
360 isotopes, and provided that they are homogeneous, would constitute a good reference  
361 material.

362

363

### Conclusion

364 The Pb/Pb crystallization ages determined on tranquillityite in this study indicate that  
365 high-Ti mare basalts 10044, 75055 and 74255 were emplaced on the lunar surface at  $3722 \pm$   
366  $11$  Ma,  $3772 \pm 9$  Ma and  $3739 \pm 10$  Ma, respectively. These ages are associated with a  
367 precision better than 0.3% at the 95% confidence level, for a beam size of  $\sim 2$ -3  $\mu\text{m}$ ,  
368 confirming the suitability of tranquillityite for Pb/Pb dating. These ages are also consistent  
369 with crystallization ages determined previously using different chronometers, but provide  
370 tighter constraints on the timing of eruption of three different types of high-Ti mare basalts.  
371 Additionally, our new high-precision Pb/Pb crystallization ages obtained on tranquillityite  
372 seem to indicate concurrent mafic volcanic activity at  $\sim 3.71$  Ga on the lunar surface,  
373 represented by the Apollo 11 and Apollo 17 landing sites.

374 The small spatial resolution achieved in our dating protocol with the NanoSIMS 50 ion  
375 probe and the widespread occurrence of tranquillityite grains in mare basalts will allow better  
376 age constraints to be placed on different basaltic units sampled during the Apollo missions.  
377 This will ultimately lead to a better adjustment of the lunar calibration curve used for crater  
378 counting studies, which underpins all the ages estimated for other planetary bodies in the  
379 inner solar system.

380

381

### Acknowledgments

382 We would like to acknowledge the support of the Science and Technology Facilities  
383 Council (STFC) for funding this research (Grant no. ST/I001298/1 to MA). We also thank  
384 NASA CAPTEM for allocation of lunar samples, Diane Johnson for her help with SEM  
385 analysis, and Marc Poujol for proofreading the manuscript. Critical and constructive  
386 comments made by Trevor Ireland and an anonymous reviewer helped significantly improve  
387 the manuscript. Finally, we thank the FEDER, the Région Bretagne, the Conseil Général  
388 d'Ille-et-Vilaine and Rennes Métropole for financial support of the imaging platform ONIS  
389 (GIS Europia).

390

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571

572

573 **Figure captions**

574 **Figure 1.** Chemical composition of tranquillityite in Apollo 11 and 17 high-Ti mare basalts  
575 10044, 75055 and 74255, plotted in a ternary diagram  $\text{TiO}_2\text{-SiO}_2\text{-ZrO}_2$ . Data for lunar  
576 tranquillityite in basalts collected during the Apollo 11, 12 and 17 missions, as well as for  
577 terrestrial tranquillityite have been plotted for comparisons (data sources same as given in  
578 Table 1).

579

580 **Figure 2.** Secondary electron image of a sputtered crater produced in tranquillityite by a 30  
581 minutes analysis. Abbreviations are bdy: baddeleyite and trq: tranquillityite.

582

583 **Figure 3.** Secondary ion images of  $^{47}\text{Ti}$ ,  $^{91}\text{Zr}$ ,  $^{206}\text{Pb}$  and  $^{238}\text{U}^{16}\text{O}$  on the area enclosing  
584 tranquillityite #14 and an associated baddeleyite grain in sample 10044 (data processed with  
585 the L'Image software package, L.R. Nittler, Carnegie Institution, USA). UO intensity is  
586 higher and  $^{206}\text{Pb}$  intensity is lower in baddeleyite compared to tranquillityite. This is related to  
587 the better Pb ionization efficiency in tranquillityite compared to baddeleyite.

588

589 **Figure 4.**  $^{207}\text{Pb}/^{206}\text{Pb}$  vs.  $^{204}\text{Pb}/^{206}\text{Pb}$  diagram and weighted mean  $^{207}\text{Pb}/^{206}\text{Pb}$  dates calculated  
590 for tranquillityites in high-Ti mare basalts 10044, 75055 and 74255. In the right panels, data  
591 are plotted in the same order as in Table 2. No common Pb correction has been applied to the  
592 data (see text for details). The thick grey curves on extreme right are the probability  
593 distribution curves of the  $^{207}\text{Pb}/^{206}\text{Pb}$  dates. The two light grey analyses for samples 74255  
594 have been excluded from date calculation. In  $^{207}\text{Pb}/^{206}\text{Pb}$  vs.  $^{204}\text{Pb}/^{206}\text{Pb}$  diagrams,

595 uncertainties are portrayed at the  $1\sigma$  level, whereas uncertainty bars in the weighted mean  
596 date diagrams are at the  $2\sigma$  level.

597

598 **Figure 5.** Synthesis of the crystallization ages of the different types of high-Ti Apollo 11 and  
599 17 mare-basalts. Rb-Sr (squares), Sm-Nd (circles) and Pb/Pb (diamonds) ages displayed in the  
600 figure are compiled in Table 3, where references are given. The blue stars represent  
601 previously estimated crystallization ages of different types of Apollo 11 and 17 units (see  
602 Stöffler and Ryder 2001, and references therein). All error bars represent the  $2\sigma$  uncertainties  
603 associated with dates. Dates displayed next to each group correspond to the weighted average  
604 of individual dates, and are interpreted as crystallization ages of each group.

605

606 **Figure 6.** La (ppm) vs. K (ppm) diagram showing the composition of the samples from the  
607 different types of high-Ti Apollo 11 and 17 mare basalts included in the age synthesis (data  
608 have been compiled from the Lunar Sample Compendium -  
609 <http://curator.jsc.nasa.gov/lunar/compendium.cfm>).

610 **Table 1.** Representative chemical compositions of lunar and terrestrial tranquillityites.

Oxide	10044		75055		74255		Apollo 11*		Apollo 11†		Apollo 12*		Apollo 12‡		Apollo 17§		Terrestrial	
	n = 10		n = 10		n = 3		n = 7		n = 15		n = 6		n = 2		n = 3		n = 4	
	wt.%	sd	wt.%	sd	1.9	0.5	wt.%	sd	wt.%	sd	wt.%	sd	wt.%	sd	wt.%	sd	wt.%	sd
SiO <sub>2</sub>	14.1	0.5	14.7	0.5	16.0	0.6	14.2	1.2	15.0	0.4	14.2	1.1	14.7	1.0	14.0	0.7	16.6	1.5
TiO <sub>2</sub>	20.3	2.1	21.8	1.9	21.3	0.7	19.8	0.8	20.4	1.0	18.9	1.5	19.0	0.3	20.8	0.4	23.4	0.8
Al <sub>2</sub> O <sub>3</sub>	1.0	0.1	1.0	0.4	1.9	0.5	0.8	0.1	1.1	0.6	1.4	0.4	4.8	0.4	1.0	0.2	0.9	0.6
FeO	43.0	1.2	43.9	1.2	41.1	2.4	42.4	0.7	42.7	0.6	42.3	0.6	39.9	2.8	41.5	0.8	44.7	1.0
CaO	1.2	0.3	1.4	1.0	1.3	0.2	1.1	0.1	1.1	0.1	1.4	0.2	1.9	0.4	0.9	0.1	1.4	0.2
ZrO <sub>2</sub>	16.4	2.1	15.3	0.9	16.4	0.3	16.8	0.6	14.2	1.3	17.3	0.6	19.5	4.9	17.2	1.2	9.2	1.7
HfO <sub>2</sub>	0.4	0.4	0.7	0.6	0.5	0.3	0.1	0.2	0.3	0.0	0.3	0.4	-	-	0.2	0.0	0.1	0.0
Y <sub>2</sub> O <sub>3</sub>	2.7	1.4	2.3	0.7	2.2	0.3	3.0	0.9	2.5	0.4	2.8	1.8	-	-	3.1	0.4	0.6	0.5
Nb <sub>2</sub> O <sub>5</sub>	0.9	0.5	1.0	0.5	0.6	0.4	-	-	0.7	0.2	0.3	-	-	-	0.7	0.1	0.2	0.1

611 Note: Data are from Lovering et al. (1971)\*, Rasmussen et al. (2008)†, Simpson and Bowie (1971)‡, Meyer and Boctor (1974)§ and Rasmussen  
 612 et al. (2012)||.

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618 **Table 2.** NanoSIMS Pb/Pb data for analyses carried out in high-Ti mare basalts 10044, 75055  
 619 and 74255.

Analysis	$^{204}\text{Pb}/^{206}\text{Pb}$	1 $\sigma$ (%)	$^{207}\text{Pb}/^{206}\text{Pb}$	1 $\sigma$ (%)	$f_{206}$ (%) <sup>†</sup>	$^{207}\text{Pb}/^{206}\text{Pb}$ age (Ma)	1 $\sigma$ (Ma)
Trq5#1	0.00007	36	0.3493	0.7	0.12	3704	11
Trq10#1	0.00010	38	0.3601	0.9	0.15	3750	13
Trq10#2	0.00018	41	0.3567	1.2	0.28	3736	19
Trq8#1	0.00010	33	0.3549	0.7	0.16	3728	11
Trq8#2	0.00014	33	0.3551	0.9	0.22	3729	14
Trq11#5	0.00010	28	0.3505	0.7	0.17	3709	10
Trq11#3	0.00013	30	0.3525	0.8	0.21	3718	12
Trq11#4	0.00004	58	0.3542	0.9	0.07	3725	13
Trq14#1	0.00013	32	0.3531	0.9	0.21	3720	13
Bad14#2	0.00034	71	0.3444	3.0	0.54	3682	45
Trq34#1	0.00012	100	0.3545	2.5	0.19	3726	38
Trq34#2	0.00018	58	0.3614	1.8	0.28	3756	26
Trq34#3	0.00006	32	0.3606	0.6	0.10	3752	9
Trq33#1	0.00006	26	0.3672	0.5	0.10	3780	7
Trq33#2	0.00005	33	0.3662	0.5	0.07	3776	8
Trq33#3	0.00017	35	0.3687	1.0	0.26	3786	16
Trq33#4	0.00008	38	0.3651	0.8	0.13	3771	12
Trq23#3	0.00007	24	0.3668	0.5	0.12	3778	7
Trq24#1	0.00004	100	0.3758	1.5	0.07	3815	22
Trq17#1	0.00000	50	0.3637	0.6	0.00	3765	10
Trq17#2	0.00007	41	0.3692	0.8	0.11	3788	12
Trq17#3	0.00002	55	0.3608	0.5	0.03	3753	8
Trq7#3	0.00007	19	0.3543	0.4	0.11	3726	6
Trq7#2	0.00009	19	0.3572	0.4	0.14	3738	6
Trq7#1	0.00008	33	0.3590	0.7	0.13	3746	10
Trq4#1‡	0.00009	32	0.3665	0.7	0.14	3777	10
Trq4#2	0.00004	21	0.3583	0.3	0.06	3743	4
Trq9#2	0.00002	35	0.3589	0.4	0.04	3745	6
Trq18#1‡	0.00003	54	0.3488	0.7	0.05	3702	11

620 <sup>†</sup> $f_{206}$  denotes the unradiogenic fraction of  $^{206}\text{Pb}$ , calculated using the measured  $^{204}\text{Pb}/^{206}\text{Pb}$   
 621 ratios and the Broken Hill galena Pb isotope composition reported by Sangster et al. (2000)  
 622 (see text for details).

623 ‡Analyses excluded from date calculations.

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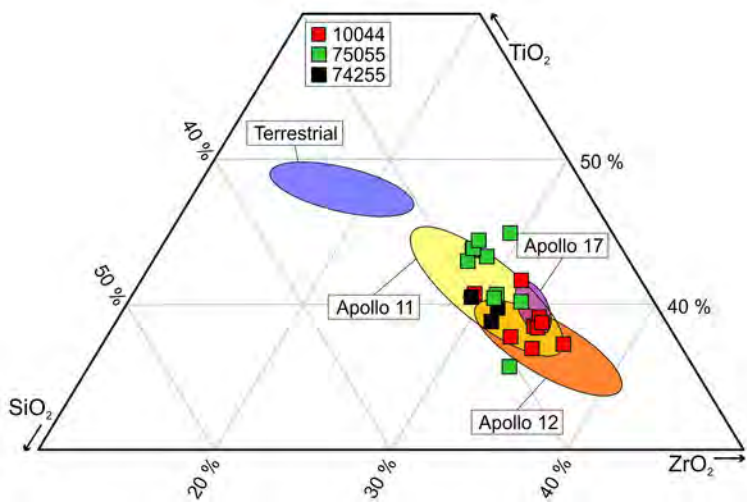
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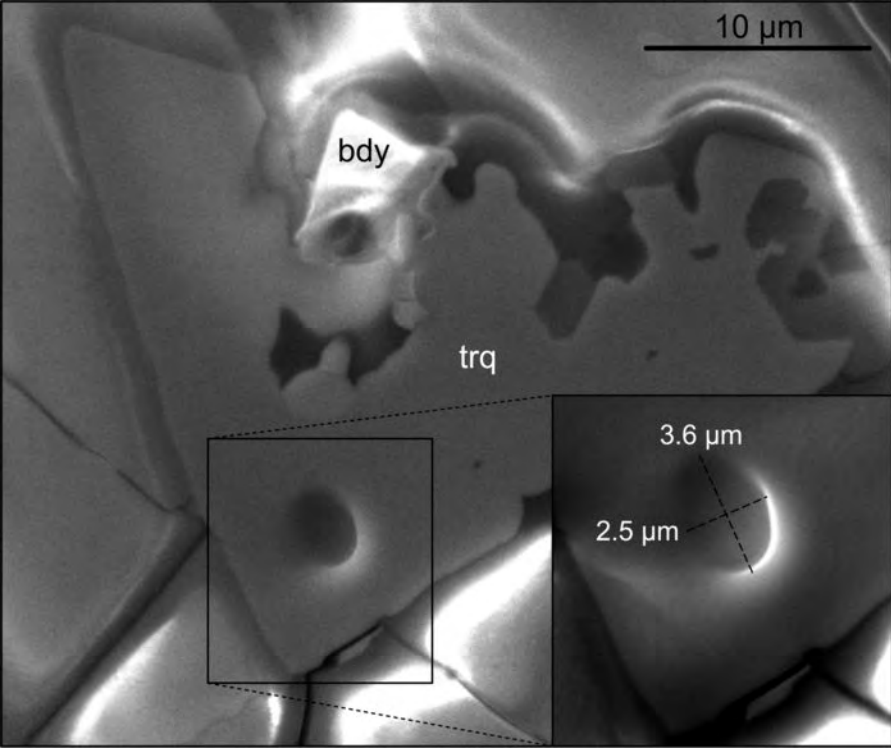
631 **Table 3.** A compilation of previous age determinations carried out on Apollo 11 and 17 high-  
 632 Ti mare basalts.

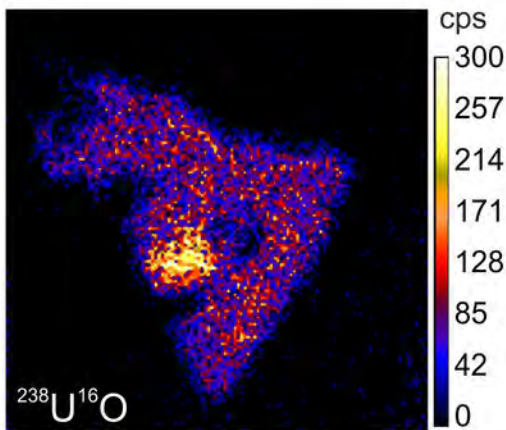
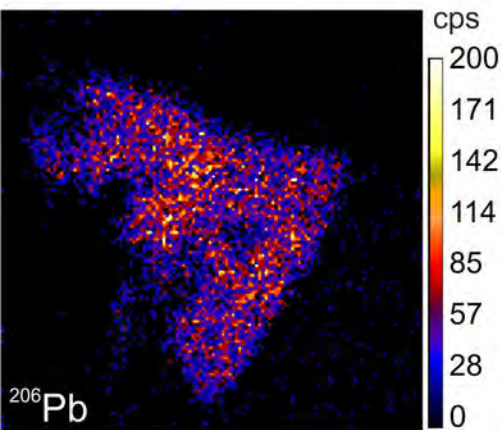
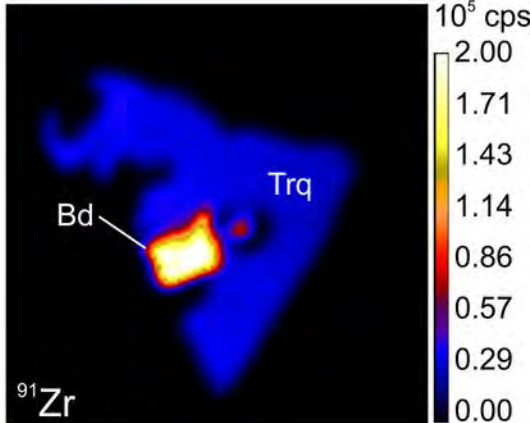
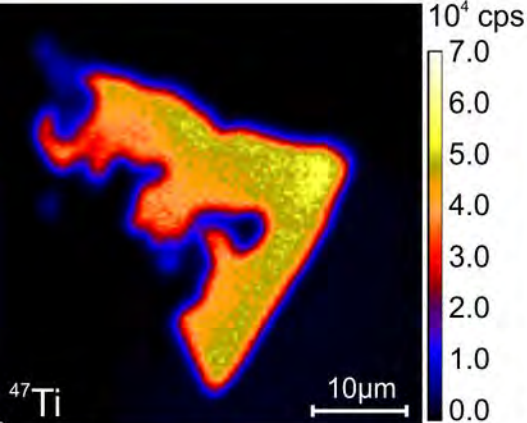
Sample	Type	Age (Ma)	2 $\sigma$ (Ma)	Method	References
10017	A	3576	56	Rb-Sr	Papanastassiou et al. (1970)
10024	A	3595	110	Rb-Sr	Papanastassiou and Wasserburg (1971)
10057	A	3610	73	Rb-Sr	Papanastassiou et al. (1970)
10069	A	3656	99	Rb-Sr	Papanastassiou et al. (1970)
10071	A	3662	63	Rb-Sr	Papanastassiou et al. (1970)
10072	A	3642	60	Rb-Sr	Papanastassiou et al. (1977)
		3575	27	Sm-Nd	Papanastassiou et al. (1977)
10058	B1	3715	110	Rb-Sr	Papanastassiou et al. (1970)
		3616	220	Rb-Sr	Snyder et al. (1994)
		3697	110	Sm-Nd	Snyder et al. (1994)
10044	B1	3694	68	Rb-Sr	Papanastassiou et al. (1970)
		3713	8	Pb/Pb	This study
10047	B1	3709.5	4.2	Pb/Pb	Rasmussen et al. (2008)
10062	B2	3911	140	Rb-Sr	Papanastassiou et al. (1977)
		3883	100	Sm-Nd	Papanastassiou et al. (1977)
10003	B2	3779	110	Rb-Sr	Papanastassiou and Wasserburg (1975)
70135	A	3825	73	Rb-Sr	Nyquist et al. (1975)
		3770	60	Sm-Nd	Nyquist et al. (1979)
71539	A	3733	110	Rb-Sr	Paces et al. (1991)
		3744	67	Sm-Nd	Paces et al. (1991)
75035	A	3808	130	Rb-Sr	Murthy and Coscio (1976)
75055	A	3752	60	Rb-Sr	Tera et al. (1974)
		3769	8	Pb/Pb	This study
75075	B	3809	120	Rb-Sr	Murthy and Coscio (1976)
		3839	91	Rb-Sr	Nyquist et al. (1975)
		3827	73	Rb-Sr	Combined datasets
		3709	65	Sm-Nd	Lugmair et al. (1975)
70139	B	3713	120	Rb-Sr	Paces et al. (1991)
		3710	120	Sm-Nd	Paces et al. (1991)
70035	B	3790	190	Rb-Sr	Evensen et al. (1973)
		3726	110	Rb-Sr	Nyquist et al. (1974)
		3734	93	Rb-Sr	Combined datasets
70017	B	3746	210	Rb-Sr	Nyquist et al. (1975)
74255	C	3751	77	Rb-Sr	Nyquist et al. (1976)
		3694	140	Rb-Sr	Murthy and Coscio (1976)
		3750	120	Rb-Sr	Combined datasets
74275	C	3829	140	Rb-Sr	Murthy and Coscio (1977)
		3736	10	Pb/Pb	This study

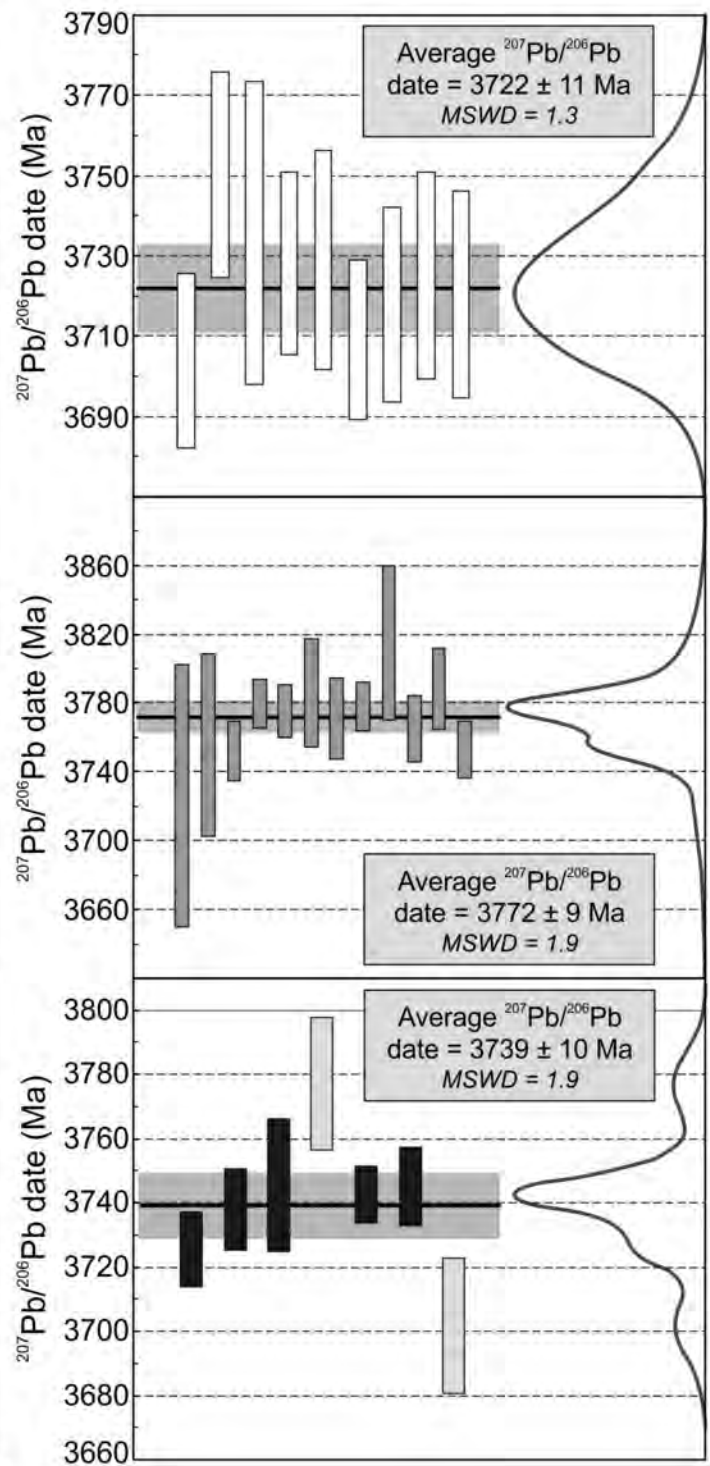
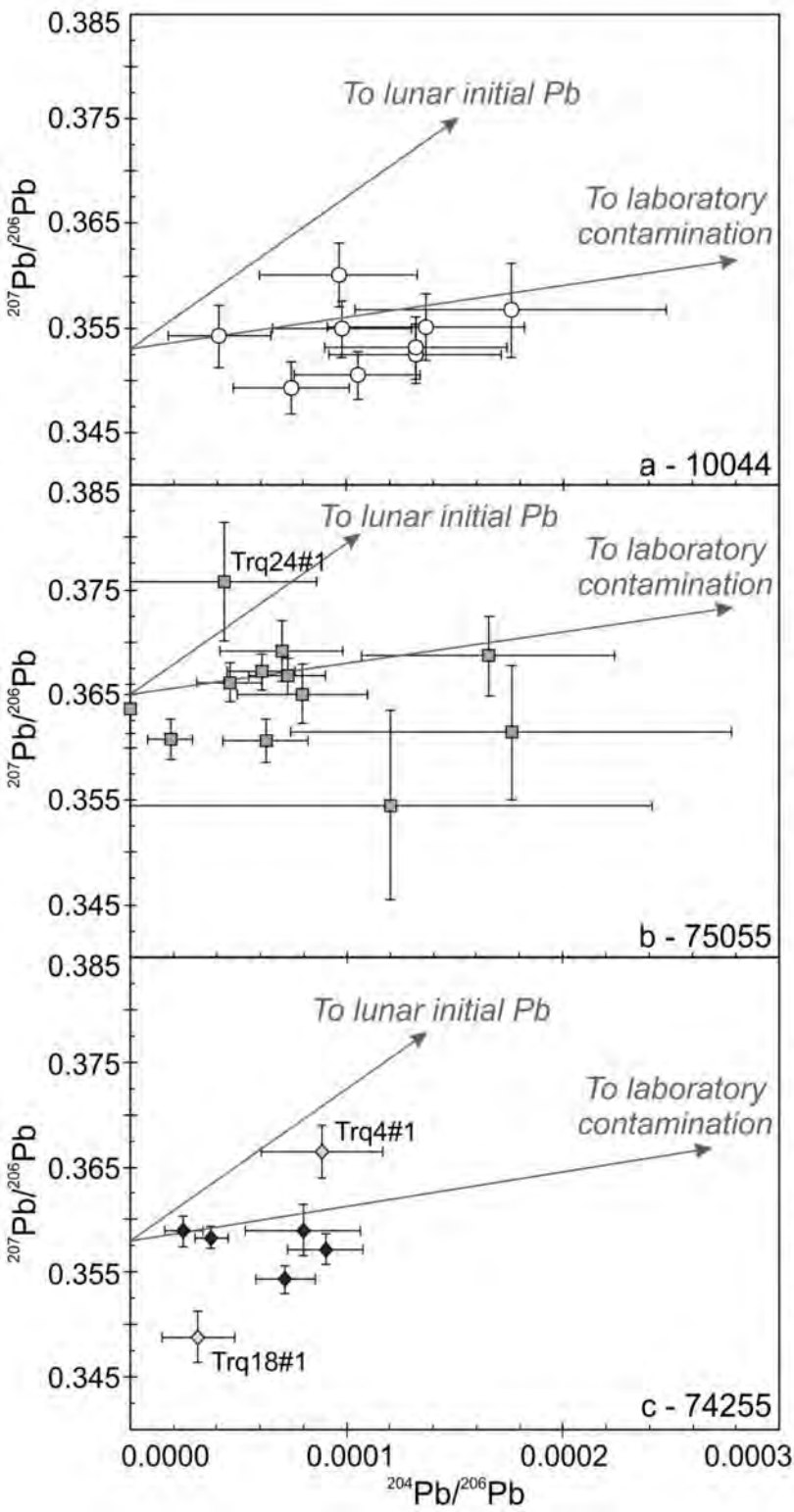
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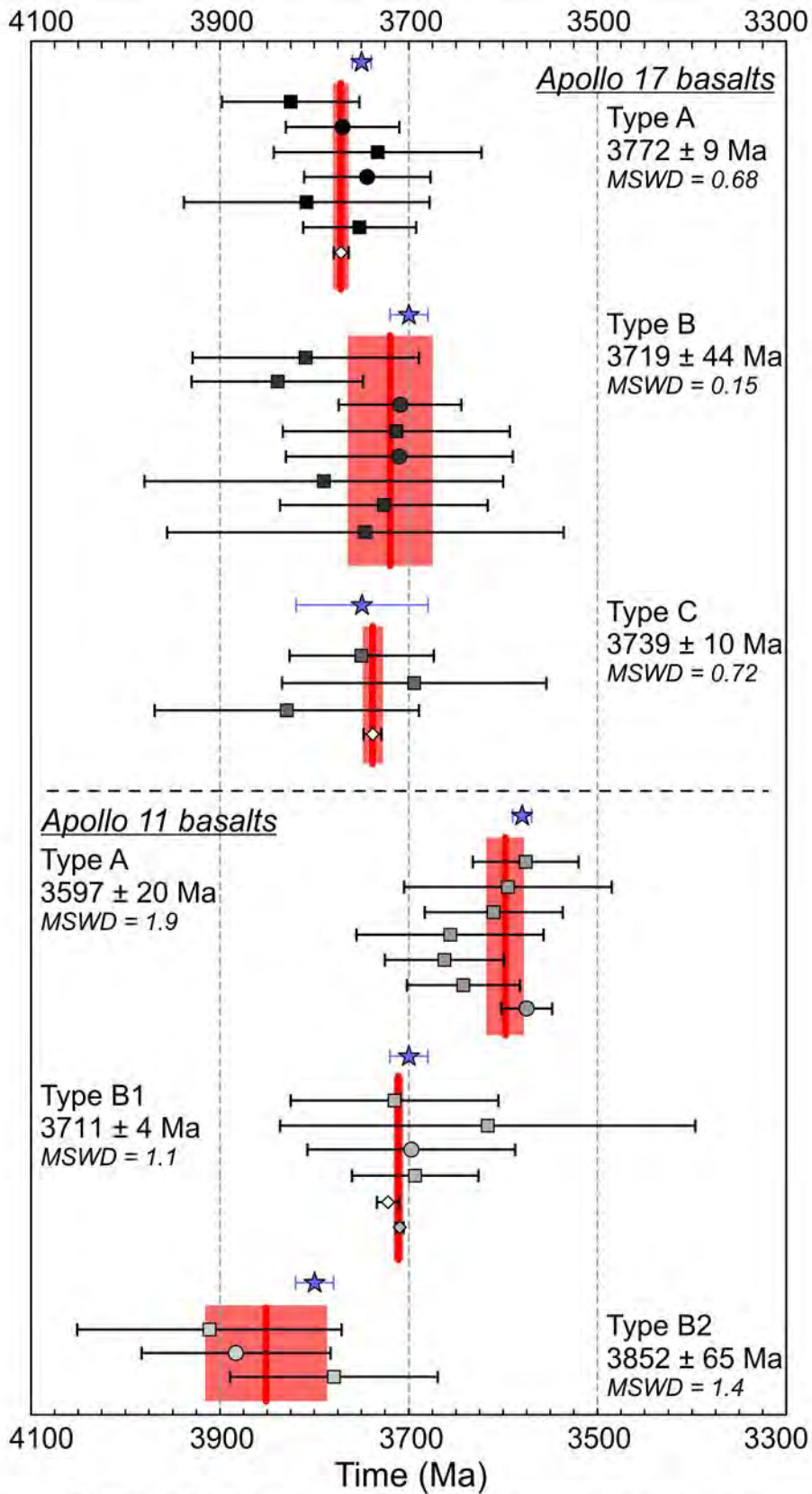
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Dating techniques

□ Rb-Sr    ○ Sm-Nd    ◇ Pb/Pb

★ Previous group crystallisation age estimate (e.g., Stöfler and Ryder 2001)

